

ChemComm

Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this *Accepted Manuscript* with the edited and formatted *Advance Article* as soon as it is available.

You can find more information about *Accepted Manuscripts* in the [Information for Authors](#).

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard [Terms & Conditions](#) and the [Ethical guidelines](#) still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.

COMMUNICATION

Imaging the lipid bilayer of giant unilamellar vesicles using red-to-blue light upconversion †

Cite this: DOI: 10.1039/x0xx00000x

Received 00th January 2012,
Accepted 00th January 2012Sven H.C. Askes,^{a,‡} Néstor López Mora,^{a,‡} Rolf Harkes,^b Roman I. Koning,^c Bram Koster,^c Thomas Schmidt,^b Alexander Kros,^a Sylvestre Bonnet.^{a,*}

DOI: 10.1039/x0xx00000x

www.rsc.org/

Red-to-blue triplet-triplet annihilation upconversion was obtained in giant unilamellar vesicles. The upconverted light was homogeneously distributed across the membrane and could be utilized for the imaging of individual giant vesicles in three dimensions. These results show the great potential of TTA-UC for imaging applications under anoxic conditions.

Upconversion luminescence (bio)imaging offers great advantages over conventional imaging. The absence of autofluorescence results in high contrast images, while photons of low energy, i.e. within the phototherapeutic window (600-1000 nm), afford higher tissue penetration and negligible irradiation damage. For these reasons lanthanoid-based upconverting nanoparticles (UCNPs), for example, have attracted much interest.^{1,2} However, UCNPs suffer from several disadvantages, such as the need for high excitation power, the low absorption cross section of lanthanoid ions, and low upconversion efficiency in aqueous solution (typically $\leq 0.5\%$).² In contrast, triplet-triplet annihilation upconversion (TTA-UC) requires low excitation power ($< 100 \text{ mW}\cdot\text{cm}^{-2}$), employs sensitizers having high extinction coefficients in the phototherapeutic window, and has achieved upconversion quantum yields up to 14% in aqueous solution.^{2,3}

In TTA-UC, low-energy photons are converted into higher-energy photons by means of a photophysical mechanism involving a couple of molecular dyes called the sensitizer and annihilator (see Figure S1 for a qualitative Jablonski diagram).⁴⁻⁸ The sensitizer absorbs the low-energy light, undergoes intersystem crossing (ISC) to a triplet state, and transfers its energy to the annihilator molecule by triplet-triplet energy transfer. Further collision of two triplet annihilator molecules leads to triplet-triplet annihilation (TTA), whereby one annihilator molecule is promoted to the excited singlet state, whereas the other one falls back to the ground state. The singlet annihilator returns to the ground state by emission of a high-energy photon, thus realizing upconversion. Most molecular dyes used in TTA-UC are highly lipophilic and require supramolecular strategies to be used in aqueous solution.⁹⁻¹⁴ For example, sub-micrometer sized TTA-UC particles have been proposed for *in vitro* or *in vivo* imaging.^{2, 11, 12} We now

demonstrate that TTA-UC can also be used for the imaging of lipid membranes.

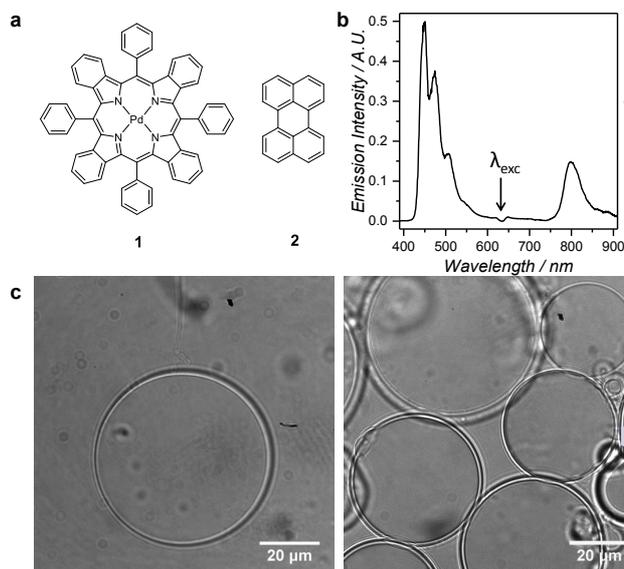


Figure 1. a) Chemical structures of palladium tetraphenyltetrabenzoporphyrin (1) and perylene (2). b) Emission spectra of DOPC upconverting GUVs with 30 mW 630 nm excitation ($0.24 \text{ W}\cdot\text{cm}^{-2}$ intensity) at 298 K in sulfite-supplemented (0.3 M) PBS buffer under air. c) Bright field micrographs of DOPC (left) and DMPC (right) upconverting giant vesicles at 298 K.

Giant Unilamellar Vesicles (GUVs) are classical tools for fluorescence imaging, as their large size (1–100 μm diameter), allows for direct observation of individual vesicles by optical microscopy techniques.¹⁵ GUVs have for example been used for visualizing lipid rafts, membrane fusion, or ion transport. In this study we functionalized PEGylated GUVs with palladium tetraphenyltetrabenzoporphyrin (1) as photosensitizer and perylene (2) as the annihilator (Figure 1a), and studied red-to-blue TTA-UC in the membrane of the vesicles by optical microscopy. The aim of the study was to investigate the dye distributions across the membrane, the homogeneity of

upconverted emission in the lipid bilayer, and the upconversion stability under imaging conditions. The growth of high-quality giant vesicles with a well-defined shape in physiologically relevant conditions, i.e., at high ionic strengths, was until recently considered as a challenge, but a new method was recently developed by some of us that is compatible with such conditions (up to 320 mOsm.kg⁻¹).¹⁷

Upconverting giant vesicles **GUV12** were thus prepared from a lipid mixture of 95 mol% phospholipid (either 1,2-dimyristoyl-sn-glycero-3-phosphocholine, i.e. DMPC, or 1,2-dioleoyl-sn-glycero-3-phosphocholine, i.e. DOPC), 4 mol% sodium N-(carbonyl-methoxypolyethylene glycol-2000)-1,2-distearoyl-sn-glycero-3-phosphoethanolamine (DSPE-MPEG-2000), 0.5 mol% compound **2**, and 0.02 mol% compound **1**. The complete procedure is described in the ESI†. Briefly, the dye-containing lipid mixture in chloroform was deposited on a chemically cross-linked dextran-poly(ethylene glycol) hydrogel substrate, dried to form a lipid film, and then the film was re-hydrated with phosphate buffered saline (PBS) supplemented with 0.3 M sodium sulfite (Na₂SO₃) and 0.2 M sucrose at 293 – 308 K. Transferring the solution onto a microscopy slide allowed for bright field imaging on a custom-build microscope based on an inverted microscopy setup. The images (Figure 1c) confirmed that for both lipid compositions (DMPC or DOPC) free-floating single vesicles were obtained, together with clusters of smaller vesicles. The images also show that the self-assembled vesicles were giant (diameter 1-100 μm), unilamellar, and spherical. The fact that almost identical procedures can be employed for preparing GUVs from lipids having a marked difference in their gel-to-liquid transition temperature ($T_m = -17.3$ °C and 23.9 °C for pure DOPC and DMPC, respectively)¹⁸, demonstrates the flexibility of the GUV preparation method. For comparison, much smaller LUVs (samples **LUV12**) with an average diameter of ca. 150 nm were prepared from the same lipid mixture but using a standard hydration-extrusion protocol (ESI†, Figure S2).

Sodium sulfite was added in the buffer as an oxygen-scavenging agent. Since the triplet states involved in TTA-UC are readily quenched by molecular oxygen, it is common practice to deoxygenate samples before measuring upconverted emission. With LUVs de-oxygenation can be achieved by, for example, bubbling the solution with argon or N₂. In the case of GUVs imaging however, bubbling an inert gas through the solution would at least impair visualization of single GUVs during a long time period of time due to convection, or even lead to damaging of the giant vesicles, so that supplementing the buffer with an oxygen scavenger is highly preferred. In a preliminary experiment, upconversion emission spectra of **LUV12** samples deoxygenated by either argon bubbling for 30 minutes or by adding 0.3 M sodium sulfite to the buffer, were compared (see Figure S4 and ESI† for details). When irradiated at 630 nm the emission spectrum of such LUVs at 298 K shows at 800 nm the phosphorescence band of **1**, and between 450 and 600 nm the blue singlet emission from **2** (Figure S4). The spectra from both deoxygenation methods were found to be very similar. It was thus concluded that Na₂SO₃ does not interfere with the photophysical processes at the origin of upconversion, and that sulfite might be used for scavenging dioxygen in a GUV-containing sample as well.

Indeed, even though addition of Na₂SO₃ significantly increased the ionic strength of the buffer (from 278 ± 1 mOsm.kg⁻¹ for PBS buffer to 884 ± 11 mOsm.kg⁻¹ when supplemented with

0.3 M sodium sulfite), as explained above sodium sulfite did not prevent the assembly of DMPC or DOPC **GUV12** using the hydrogel method. No differences in vesicle yield and morphology were observed in presence or absence of sodium sulfite in the buffer. This result demonstrates that the dextran-poly(ethylene glycol) hydrogel substrate is able to produce GUVs at high ionic strength, which is a significant advantage over alternative GUV preparation methods such as electroformation or gentle hydration, which often fail in such conditions. When irradiated at 630 nm under air, the emission spectrum of the DMPC or DOPC **GUV12** samples prepared in a sulfite-supplemented buffer was identical to the emission spectrum of the corresponding **LUV12** samples (Figure 1b and S4), showing that the dyes **1** and **2** were indeed incorporated in the lipid bilayer.

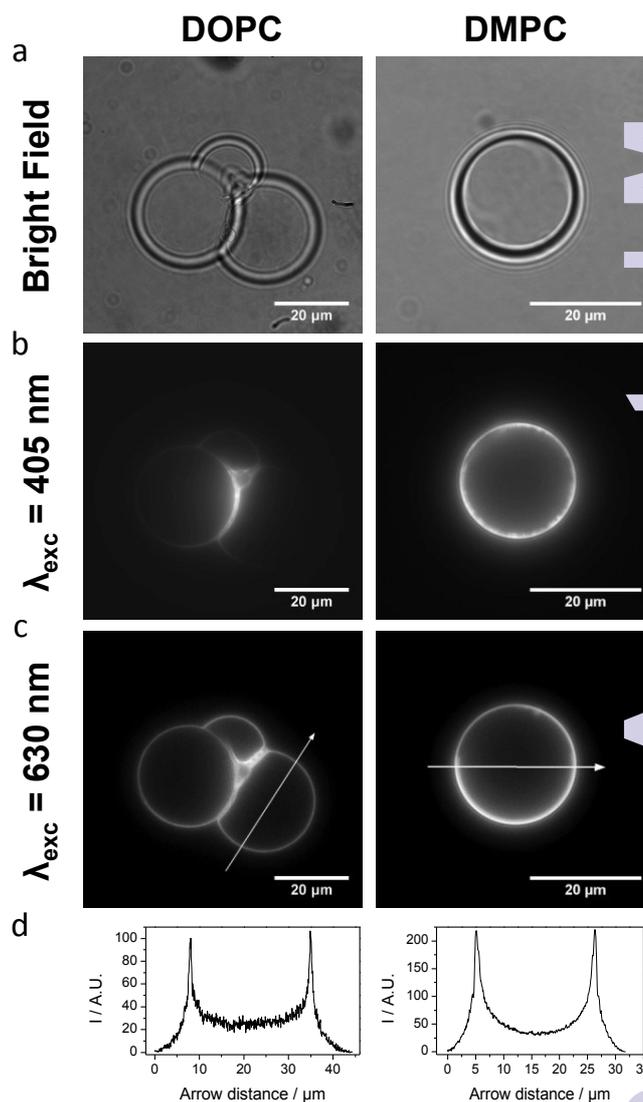


Figure 2. Imaging of DOPC (left) and DMPC (right) upconverting giant vesicles (**GUV12**) with a) bright field, b) 405 nm excitation and 450-500 nm detection, c) 630 nm excitation and 450-575 nm detection. d) Upconversion intensity profile plot following the arrows in the images directly above (c). At 630 nm: laser spot size diameter 39 μm, power 3.8 mW, intensity 320 W.cm⁻². At 405 nm: laser spot size diameter 22 μm (power 1 mW, intensity 60 W.cm⁻²) for DOPC image or 39 μm (power 1 mW, intensity 300 W.cm⁻²) for DMPC image. Images were acquired at 298 K in sulfite-supplemented (0.3 M) PBS buffer.

GUV12 samples were then visualized by emission microscopy at 298 K (Figure 2 and ESI†). When the vesicles were illuminated with violet light (405 nm), i.e. by direct excitation of perylene (**2**), fluorescence was clearly detected at the membrane (Figure 2b). To visualize upconversion, a 630 nm continuous wave PDT laser was coupled into the microscope and set at a power of a few milliwatts, resulting in the focal spot in an intensity of $\sim 300 \text{ W.cm}^{-2}$. All wavelengths other than 450 – 575 nm were strictly blocked by a combination of notch and short-pass filters (ESI†). High-quality images were obtained that were superimposable to the bright field images and to the fluorescence images recorded under white and violet light irradiation, respectively (Figure 2a-c). Control samples were prepared in which the porphyrin sensitizer **1** was omitted from the formulation (**GUV2**). Images recorded in identical conditions were black, i.e., no blue emission was observed (Figure S10). **GUV12** samples prepared in absence of sulfite oxygen scavenger and observed under air did not give any observable emission either (Figure S11). Altogether, these observations prove that the blue images recorded under 630 nm irradiation of **GUV12** samples supplemented with sulfite comes from the TTA upconversion process and are not the result of sensitizer emission (at 800 nm) or of two-photon absorption. Overall, all data conclude that both dyes **1** and **2** co-localize in the membrane and result in TTA upconversion. At this scale of observation the upconverted emission is homogeneous across the membrane and no phase separation of the lipids or dyes was observed.

or 5.2 W.cm^{-2} (empty circles), detection in the 450–575 nm region, $T=298 \text{ K}$. Snapshots were taken with an exposure time of 0.2 s (320 W.cm^{-2}) or 1.0 s (5.2 W.cm^{-2}). Error bars represent standard deviation based on six individual measurements. b) Upconversion emission microscopy images of **GUV12** sample at $t = 0 \text{ s}$ (left) and at $t = 10 \text{ s}$ (right) at an illumination intensity of 320 W.cm^{-2} (top) and 5.2 W.cm^{-2} (bottom). Excitation at 630 nm, detection at 450–575 nm.

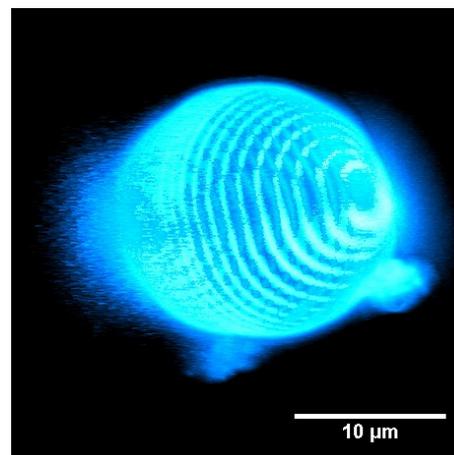


Figure 4. 3D reconstructed image of a DMPC **GUV12** sample, rotated counter-clockwise by 50° about the y-axis. Each z-slice was imaged at 298 K with 630 nm excitation (320 W.cm^{-2}) and detection in the 450–575 nm region. The z-distance between slices was $1.0 \mu\text{m}$. Video V1 (ESI†) exhibits a 360° rotation view of this image and of four other individual DMPC and DOPC **GUV12**.

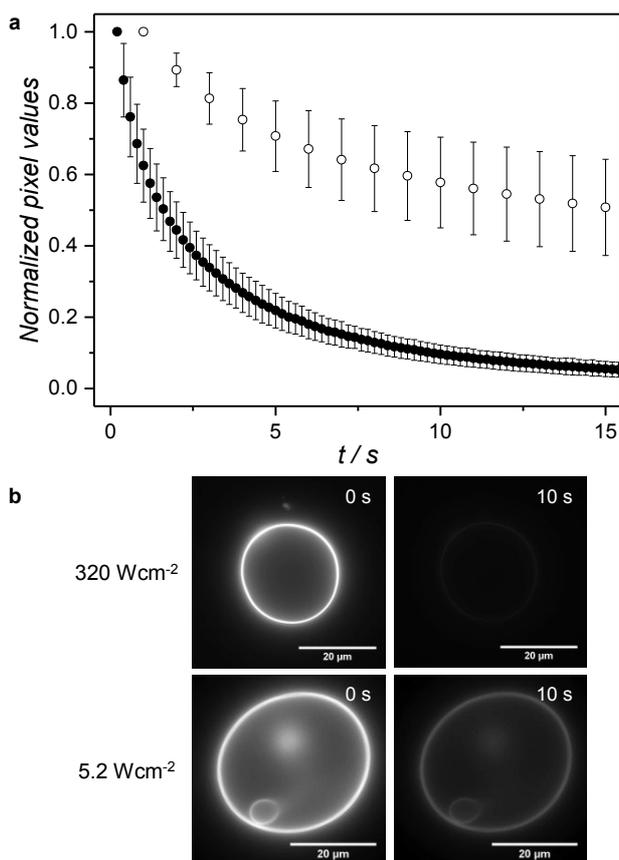


Figure 3. a) Averaged normalized pixel values as a function of red irradiation time during upconversion imaging of **GUV12** samples in sulfite-supplemented PBS buffer (0.3 M). Conditions: 630 nm excitation at 320 W.cm^{-2} (black filled circles)

Under the red-light irradiation conditions initially used in the microscopy setup (630 nm at an intensity of 320 W.cm^{-2}), substantial bleaching of the upconverted emission of **GUV12** samples was observed even in presence of 0.3 M of sulfite. A plot of the averaged normalized pixel values as a function of red irradiation time shows that the upconverted emission is halved after less than 3 seconds (Figure 3). When the light intensity was lowered 60 times (i.e., down to 5.2 W.cm^{-2}) clear upconversion images could still be recorded. In such conditions the bleaching rate was significantly lower (Figure 3), and the time necessary for halving the upconverted emission intensity of a pixel increased to approximately 15 seconds. The upconversion luminescence of **LUV-12** in a spectroscopy setup could be observed for less than 8 mW.cm^{-2} , with linear power dependency above 60 mW.cm^{-2} (Figure S6). Overall, these findings show that high power is not a requirement for the upconversion imaging of **GUV-12**.

In optimized conditions, we realized that the upconverted emission was intense enough to be utilized for reconstructing in 3D the membrane of the giant vesicles. Z-stack upconversion image acquisition was indeed performed on both DMPC and DOPC **GUV12** samples. The illumination intensity was deliberately chosen to be high (320 W.cm^{-2}) to make sure that z-stack image acquisition was short (200 ms exposure time per slice, ca. 45 slices per stack, total acquisition time $< 10 \text{ s}$). In such conditions, the slight lateral motion of the GUVs did not significantly affect the imaging process. From these stacks, 3D reconstructions were made (e.g. Figure 4), of which a video was compiled (Video V1 in the ESI†). This reconstruction demonstrates that the TTA-upconverted emission can be utilized for the three-dimensional reconstruction of an object that is 10 to 30 μm in size.

In conclusion, DOPC and DMPC giant vesicles capable of upconverting red light to blue light by means of triplet-triplet annihilation were prepared by lipid film hydration on a hydrogel substrate at high ionic strengths. The preparation method is facile and does not involve any specific equipment. Sodium sulphite added as an oxygen scavenger to the vesicle samples allows for observing upconversion even under air. According to optical microscopy, the upconverted emission allows for recording high quality images showing that upconversion is homogeneously realized across the lipid bilayer. The quality and stability of the upconverted images enabled the 3D reconstruction of upconverting GUVs. These results show the great potential of TTA upconversion for imaging applications under anoxic conditions, and open a route towards cell membrane imaging with upconverted light.

Notes and references

^a Leiden Institute of Chemistry, Leiden University, Einsteinweg 55, 2333 CC Leiden, The Netherlands

^b Leiden Institute of Physics, Leiden University, Niels Bohrweg 2, 2333 CA Leiden, The Netherlands

^c Leiden University Medical Center, Einthovenweg 20, 2333 ZC Leiden, The Netherlands

† Electronic Supplementary Information (ESI) available: Supplementary Information, and Video V1. See DOI: 10.1039/c000000x/

‡ These authors contributed equally to this work.

* Corresponding author: bonnet@chem.leidenuniv.nl

Prof. Dr. Lies Bouwman is kindly acknowledged for support and scientific discussion. NWO (the Netherlands Organisation for Scientific Research) is acknowledged for a VIDI grant to S.B. The European Research Council is acknowledged for an ERC starting grant to S.B. CONACYT is acknowledged for financial support to N.L.M.

1. J. Zhou, Q. Liu, W. Feng, Y. Sun and F. Li, *Chem. Rev.*, 2014.
2. Q. Liu, W. Feng, T. Yang, T. Yi and F. Li, *Nat. Protocols*, 2013, **8**, 2033-2044.
3. J.-H. Kim and J.-H. Kim, *J. Am. Chem. Soc.*, 2012, **134**, 17478-17481.
4. T. N. Singh-Rachford and F. N. Castellano, *Coord. Chem. Rev.*, 2010, **254**, 2560-2573.
5. J. Zhao, S. Ji and H. Guo, *RSC Advances*, 2011, **1**, 937-950.
6. Y. C. Simon and C. Weder, *J. Mater. Chem.*, 2012, **22**, 20817-20830.
7. Y. Y. Cheng, B. Fußckel, T. Khoury, R. I. G. C. R. Clady, N. J. Ekins-Daukes, M. J. Crossley and T. W. Schmidt, *J. Phys. Chem. A*, 2011, **115**, 1047-1053.
8. P. Duan, N. Yanai, H. Nagatomi and N. Kimizuka, *J. Am. Chem. Soc.*, 2015.
9. A. Turshatov, D. Busko, S. Baluschev, T. Miteva and K. Landfester, *New J. Phys.*, 2011, **13**, 083035.
10. M. Penconi, P. L. Gentili, G. Massaro, F. Elisei and F. Ortica, *Photochem. Photobiol. Sci.*, 2013.
11. Q. Liu, B. Yin, T. Yang, Y. Yang, Z. Shen, P. Yao and F. Li, *J. Am. Chem. Soc.*, 2013, **135**, 5029-5037.

12. C. Wohnhaas, V. Mailänder, M. Dröge, M. A. Filatov, D. Busko, Y. Avlasevich, S. Baluschev, T. Miteva, K. Landfester and A. Turshatov, *Macromol. Biosci.*, 2013, **13**, 1422-1430.
13. K. Tanaka, H. Okada, W. Ohashi, J.-H. Jeon, K. Inafuku and Y. Chujo, *Bioorg. Med. Chem.*, 2013, **21**, 2678-2681.
14. S. H. C. Askes, A. Bahreman and S. Bonnet, *Angew. Chem., Int. Ed.*, 2014, **53**, 1029-1033.
15. P. Walde, K. Cosentino, H. Engel and P. Stano, *ChemBioChem*, 2010, **11**, 848-865.
16. H. Valkenier, N. López Mora, A. Kros and A. P. Davis, *Angew. Chem., Int. Ed.*, 2015, **54**, 2137-2141.
17. N. Lopez Mora, J. S. Hansen, Y. Gao, A. A. Ronald, R. Kieltyka, L. Malmstadt and A. Kros, *Chem. Commun.*, 2014, **50**, 1957-1958.
18. D. Marsh, *Handbook of Lipid Bilayers*, Taylor & Francis Group LLC, Boca Raton, FL, USA, 2013.