# ChemComm

## Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this Accepted Manuscript with the edited and formatted Advance Article as soon as it is available.

You can find more information about *Accepted Manuscripts* in the **Information for Authors**.

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard <u>Terms & Conditions</u> and the <u>Ethical guidelines</u> still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.



www.rsc.org/chemcomm

## COMMUNICATION

## 1Cite this: DOI: 10.1039/xoxx00000x

# Enantioselective Synthesis of Planar-Chiral Benzosiloloferrocenes by Rh-Catalyzed Intramolecular C-H Silylation

Received ooth January 2015, Accepted ooth January 2015

DOI: 10.1039/x0xx00000x

www.rsc.org/

Takanori Shibata,<sup>*a,b,\**</sup> Tsubasa Shizuno,<sup>*a*</sup> and Tomoya Sasaki<sup>*a*</sup>

Theintramolecularreactionof2-(dimethylhydrosilyl)arylferrocenesachievedthefirstsynthesisofplanar-chiralbenzosiloloferrocenes.Theenantioselectivecrossdehydrogenativecouplingofan sp<sup>2</sup> C-Hbondofferrocenewith a Si-H bondproceeded efficiently withtheuse of a Rh-chiral dienecatalyst.

Dibenzosilole (9-silafluorene) is an important skeleton of many functional molecules that are useful as conjugated organosilicon materials in organic electronics and photonics.<sup>1</sup> Therefore, a facile method for the synthesis of multi-substituted dibenzosilole is strongly desired. Among various catalytic protocols, Pd-catalyzed intramolecular C-H arylation of diarylsilane<sup>2</sup> and consecutive coupling of 2.2'-diiodo-1,1'-biphenyl with dihydrosilane<sup>3</sup> are typical examples. However, these reactions require halogenated substrates and suffer from the formation of by-products derived from halogens, based on atom-economical concerns. The Ir-catalyzed [2+2+2] cycloaddition of silanylbenzene-tethered 1,6-diyne with alkyne affords a perfectly atom-economical transformation for the synthesis of multi-substituted dibenzosiloles.<sup>4</sup> As another approach, the Rhcatalyzed intramolecular dehydrogenative coupling of sp<sup>2</sup> C-H and Si-H bonds has been reported<sup>5</sup> and asymmetric variants followed, in which a chiral spirocyclic silicon center was created by a Rh-chiral diphosphine catalyst.<sup>6</sup>

In contrast, the catalytic creation of planar chirality in disubstituted ferrocenes initiated by enantioselective C-H bond cleavage has attracted much attention over the past few years.<sup>7</sup> Aminomethyl group-directed asymmetric C-H arylation, Heck-type reaction, and annulation have been realized through the use of Pd-chiral amino acid catalysts.<sup>8</sup> We also reported isoquinolyl directed-C-H alkylation with alkenes by using an Ir-chiral diene catalyst.<sup>9</sup> As for the intramolecular reaction, carbene C-H insertion of diazoketone-tethered ferrocenes by a chiral Cu catalyst is a pioneering work, and planar-chiral cyclic ketone-fused ferrocenes

have been prepared (Scheme 1-a).<sup>10</sup> In 2014, three groups independently reported C-H arylation with haloarenes by using chiral Pd catalysts in the presence of more than a stoichiometric amount of base. Planar-chiral indenone- and quinolinone-fused ferrocenes were obtained in excellent ee (Scheme 1-b).<sup>11</sup> In this communication, we report the cross dehydrogenative coupling (CDC) of ferrocene and hydrosilane or hydrogermane by using a Rh-chiral diene catalyst. This is the first example of the enantioselective synthesis of planar-chiral heterole-fused ferrocenes (Scheme 1-c).<sup>12</sup>

a) C-H insertion of carbene (ref. 10)



b) C-H arylation by haloarene (ref. 11)



c) This work: Si-H/C-H coupling



**Scheme 1** Enantioselective and intramolecular reaction for the synthesis of planar-chiral ring-fused ferrocenes

Page 2 of 4

We chose 2-(dimethylhydrosilyl)phenylferrocene (**1aa**) as a model substrate, which was readily prepared in two steps (Scheme 2). Diazotization of 2-bromoaniline followed the reaction with ferrocene,  $^{13}$  and aryllithium prepared from the obtained bromide underwent silylation by dimethylhydrosilane.



Scheme 2 Preparation of 2-(dimethylhydrosilyl)phenylferrocene 1aa

According to the literature,<sup>6</sup> we intended to perform the intramolecular reaction of ferrocene **1aa** in the presence of a chiral Rh catalyst prepared from [RhCl(cod)]<sub>2</sub> and (*S*)-BINAP (Table 1, Entry 1). Desired CDC product **2aa** was obtained in good yield and enantioinduction was ascertained, albeit in low selectivity. The major by-product was disilane **4aa**, which resulted from the self-coupling of hydrosilanes. The iridium counterpart was less effective with respect to both yield and ee, and a significant amount of desilylated product **3aa** was formed (Entry 2).<sup>14</sup> Cationic Rh catalyst gave a higher yield along with the formation of **3aa** (Entry 3). We also investigated several chiral diphosphine ligands, but without success.<sup>15</sup>

### Table 1 Rh- or Ir-BINAP-catalyzed C-H silylation of 1aa



Entry	Catalyst	Yield $(\%)^{a}$	2aa:3aa°	Ee (%) <sup>c</sup>	Yield $(\%)^{d}$
1	$[\mathbf{Rh}Cl(cod)]_2$	78	1:0	-27	31
2	$[IrCl(cod)]_2$	65	1:0.42	7	6
3	$[\mathbf{Rh}(\mathrm{cod})_2]\mathrm{BF_4}^{\mathrm{e}}$	97	1:0.20	-21	ND
<sup>a</sup> Total yield of <b>2</b> as and <b>3</b> as <sup>b</sup> The ratio was determined by NMP					

<sup>a</sup> Total yield of **2aa** and **3aa**. <sup>a</sup> The ratio was determined by NMR analysis. <sup>c</sup> Ee of **2aa**. <sup>d</sup> NMR yield of **4aa**. <sup>e</sup> (*S*)-BINAP (12 mol%) was used.

We next examined chiral dienes as ligands (Table 2). We were pleased to find that Carreira's dienes<sup>16</sup> drastically improved the enantioselectivity: chiral Rh catalysts prepared from [RhCl(coe)<sub>2</sub>]<sub>2</sub> and dienes generally realized moderate enantioselectivity (ca. 60%), but the ratio of **2aa** and **3aa** differed depending on the aryl group (Entries 1-8). In each entry, disilane **4aa** was formed as a by-product to some extent. We determined that **L1** was the best ligand, and most strongly suppressed the formation of **3aa**. Hayashi's dienes **L9**<sup>17</sup> and **L10**<sup>18</sup> were ineffective in the present reaction.





Chiral dienes

Entry	Chiral diene	Yield (%) <sup>a</sup>	2aa:3aa <sup>b</sup>	Ee (%) <sup>c</sup>
1	L1	78	1:0.17	58
2	L2	80	1:0.29	63
3	L3	82	1:0.23	65
4	L4	89	1:0.24	59
5	L5	73	1:0.46	35
6	L6	61	1:1.31	62
7	L7	75	1:0.39	52
8	L8	61	1:0.24	60
9	L9	64	1:0.29	-33
10	L10	82	1:0.12	<2

<sup>a</sup> Total yield of **2aa** and **3aa**. <sup>b</sup> The ratio was determined by NMR analysis. <sup>c</sup> Ee of **2aa**.

To improve the yield and ee, we screened several alkenes as hydrogen acceptors (Table 3).<sup>19</sup> Norbornene surely increased the yield of **2aa** and decreased the yield of **3aa** with comparable ee (Entry 2). Unstrained cyclic alkene gave negative results (Entry 3). In the presence of 4-vinylcyclohexene, the reaction proceeded quantitatively with the almost perfect suppression of disilane formation, but the ee was significantly decreased (Entry 4). In contrast, 3,3-dimethylbut-1-ene realized a drastic increase in enantiomeric excess and greater amounts of the alkene slightly suppressed the formation of **3aa** (Entries 5 and 6).<sup>20</sup> The hydrogen acceptor probably prevented the hydrogenation of the chiral diene ligand and the enantioselectivity was improved.

#### Table 3 Effect of alkenes as a hydrogen acceptor

[RhCl(coe)<sub>2</sub>]<sub>2</sub> (10 mol%) L1 (24 mol%)

122	alkene (5 equiv)	222	Ŧ	300
1aa	toluene, 135 °C, 1.5 h	Zdd	т	Jdd

Entry	Alkene	Yield (%) <sup>a</sup>	2aa:3aa <sup>b</sup>	Ee (%) <sup>c</sup>
1	none	78	1:0.17	58
2	norbornene	89	1:0.03	57
3	cyclooctene	50	1:0.34	59
4	4-vinylcyclohexene	>99	1:0.01	40
5	3,3-dimethylbut-1-ene	72	1:0.11	82
6 <sup>d</sup>	3,3-dimethylbut-1-ene	67	1:0.07	85

Journal Name

<sup>a</sup> Total yield of **2aa** and **3aa**. <sup>b</sup> The ratio was determined by NMR analysis. <sup>c</sup> Ee of **2aa**. <sup>d</sup> Alkene (10 equiv) was added.

The present reaction was sensitive to the bulkiness around the silicon atom (Scheme 3). Both the yield and ee decreased as the substituents (R) increase in size. 2-(Diphenylhydrosilyl)phenylferrocene (1ad) was transformed into the corresponding benzosiloloferrocene 2ad in good yield, but the ee was low.



Scheme 3 Effect of substituents on the silicon atom

We checked the substrate scope by introducing several substituents on the benzene ring of 2-(dimethylhydrosilyl)phenylferrocene (Table 4). Both of electron-donating and –withdrawing groups could be installed into the paraand meta-positions with respect to the silyl group: moderate yield and good ee were achieved (Entries 1-5). A methyl group at the 3-position slightly affected the enantioselectivity (Entry 6). Introduction adjacent to the silyl group gave a poor ee (Entry 7)

 
 Table 4
 Substrate scope of 2-(dimethylhydrosilyl)arylferrocenes



Entry	R	Time (h)	Yield (%) <sup>a</sup>	2:3 <sup>b</sup>	Ee (%) <sup>c</sup>
1	4-Me (1ba)	24	56	1:0.11	74
2	4-CF <sub>3</sub> (1ca)	4	55	1:0.14	78
3	5-Me (1da)	4	44	1:0.06	86
4	5-CF <sub>3</sub> (1ea)	9	53	1:0.17	80
5	5-OCF <sub>3</sub> (1fa)	2	55	1:0.16	76
6	3,5-Me <sub>2</sub> (1ga)	5	75	1:0.13	55
7	6-Me (1ha)	24	67	1:0.09	5

<sup>a</sup> Total yield of **2** and **3**. <sup>b</sup> The ratio was determined by NMR analysis. <sup>c</sup> Ee of **2**.

The reaction of (dimethylhydrogermyl)phenylferrocene (4) also proceeded to afford benzogermoloferrocene 5 (Scheme 4).<sup>21</sup> The low yield and ee were probably due to the bulkiness of a germanium atom compared with a silicon atom.



**Scheme 4** Reaction of 2-(dimethylhydrogermyl)phenyl-ferrocene (**4**)

In conclusion, we first prepared ferrocene-fused benzosiloles and benzogermole by enantioselective and intramolecular cross dehydrogenative coupling of an sp<sup>2</sup> C-H bond of ferrocene with a Si-H or Ge-H bond. We will evaluate the potential of a new class of planar-chiral molecules as electronic organic materials.

## Acknowledgement

This work was supported by Grant-in-Aid for Scientific Research on Innovative Areas, "Molecular Activation Directed toward Straightforward Synthesis," MEXT and JST, ACT-C, Japan.

## Notes and references

<sup>a</sup> Department of Chemistry and Biochemistry, School of Advanced Science and Engineering, Waseda University, Shinjuku, Tokyo 169-8555, Japan, Fax: (+81) 3-5286-8098; e-mail: tshibata@waseda.jp

<sup>b</sup> JST, ACT-C, 4-1-8 Honcho, Kawaguchi, Saitama 332-0012, Japan.

† Electronic Supplementary Information (ESI) available: [details of any supplementary information available should be included here]. See DOI: 10.1039/c000000x/

- For recent reviews, see: (a) M. Shimizu and T. Hiyama, Synlett, 2012, 23, 973; (b) J. Y. Corey, Adv. Organomet. Chem., 2011, 59, 181.
- (a) M. Shimizu, K. Mochida and T. Hiyama, *Angew. Chem., Int. Ed.*, 2008, **47**, 9760; asymmetric variant; (b) R. Shintani, H. Otomo, K. Ota and T. Hayashi, *J. Am. Chem. Soc.*, 2012, **134**, 7305.
- 3 Y. Yabusaki, N. Ohshima, H. Kondo, T. Kusamoto, Y. Yamanoi and H. Nishihara, *Chem.-Eur. J.*, 2010, **16**, 5581.
- 4 (a) T. Matsuda, S. Kadowaki, T. Goya and M. Murakami, Org. Lett., 2007, 9, 133; asymmetric variant: (b) R. Shintani, C. Takagi, T. Ito, M. Naito and K. Nozaki, Angew. Chem., Int. Ed., 2015, 54, 1616.
- 5 T. Ureshino, T. Yoshida, Y. Kuninobu and K. Takai, J. Am. Chem. Soc., 2010, **132**, 14324.
- 6 Y. Kuninobu, K. Yamauchi, N. Tamura, T. Seiki and K. Takai, Angew. Chem., Int. Ed., 2013, 52, 1520.
- 7 Reviews of planar-chiral ferrocenes, see: (a) D. Schaarschmidt and H. Lang, Organometallics, 2013, 32, 5668; (b) S. Arae and M. Ogasawara, Tetrahedron Lett., 2015, 56, in press (doi:10.1016/j.tetlet.2015.01.130).
- 8 (a) D.-W. Gao, Y.-C. Shi, Q. Gu, Z.-L. Zhao and S.-L. You, J. Am. Chem. Soc., 2013, **135**, 86; (b) C. Pi, Y. Li, X. Cui, H. Zhang, Y. Han and Y. Wu, Chem. Sci., 2013, **4**, 2675; (c) Y.-C. Shi, R.-F. Yang, D.-W. Gao and S.-L. You, Beil. J. Org. Chem., 2013, **9**, 1891.
- 9 T. Shibata and T. Shizuno, Angew. Chem., Int. Ed., 2014, 53, 5410.
- 10 S. Siegel and H. Schmalz, Angew. Chem., Int. Ed., 1997, **36**, 2456.
- (a) R. Deng, Y. Huang, X. Ma, G. Li, R. Zhu, B. Wang, Y.-B. Kang and Z. Gu, J. Am. Chem. Soc., 2014, **136**, 4472; (b) D.-W. Gao, Q. Yin, Q. Gu, and S.-L. You, J. Am. Chem. Soc., 2014, **136**, 4841; (c) L. Liu, A.-A. Zhang, R.-J. Zhao, F. Li, T.-J. Meng, N. Ishida, M. Murakami and W.-X. Zhao, Org. Lett., 2014, **16**, 5336.
- 12 Only an example of the asymmetric synthesis of heterole-fused ferrocenes by diastereoselective reaction: S. Yasuike, J. Hagiwara, H. Danjo, M. Kawahata, N. Kakusawa, K. Yamaguchi and J. Kurita, *Heterocycles*, 2009, **78**, 3001.
- (a) K.G. Rutherford and W. Redmond, Org. Synth., 1963, 43, 12; (b)
   W. F. Little, C. N. Reilley, J. D. Johnson, K. N. Lynn and A. P. Sanders, J. Am. Chem. Soc., 1964, 86, 1376.

- 14 Compounds **2aa** and **3aa** have similar polarity and could not be separated by preparative TLC, but could be separated by GPC (gel permeation chromatography), and **2aa** was fully characterized.
- 15 (a) The representative ee of 2aa by using chiral diphosphines: tolBINAP (9% ee); xylylBINAP (15% ee); Me-DUPHOS (<2% ee).</li>
  (b) We have recently learned that Prof. Kazuhiko Takai (Okayama University) also examined a rhodium-cihral diphosphine-catalyzed asymmetric synthesis of ferrocene-fused benzosiloles. We thank Prof. Takai for exchanging information prior to the publication.
- 16 (a) C. Fischer, C. Defieber, T. Suzuki and E. M. Carreira, J. Am. Chem. Soc., 2004, **126**, 1628; (b) C. Defieber, H. Grützmacher and E. M. Carreira, Angew. Chem., Int. Ed., 2008, **47**, 4482.
- 17 K. Okamoto, T. Hayashi and V. H. Rawal, Chem. Commun., 2009, 4815.
- 18 (a) N. Tokunaga, Y. Otomaru, K. Okamoto, K. Ueyama, R. Shintani and T. Hayashi, J. Am. Chem. Soc., 2004, **126**, 13584; (b) R. Shintani and T. Hayashi, Aldrichimica Acta, 2009, **42**, 38.
- 19 C. Cheng and J. F. Hartwig, Science, 2014, 343, 853.
- 20 Recrystallization of **2aa** gave a single crystal with improved ee (99% ee), and its absolute structure was determined by X-ray analysis; The crystal data of compound **2aa**:  $C_{18}H_{18}FeSi$ , M = 318.27, orthorhombic, space group  $P2_12_12_1$  (no. 19), a = 9.6741(4) Å, b = 10.9465(6) Å, c = 29.0292(11) Å, V = 3074.1(2) Å<sup>3</sup>, Z = 8,  $\mu$ (Mo-K $\alpha$ ) = 10.447 cm<sup>-1</sup>; number of reflections measured: total 29429 and unique 6966, R1 = 0.0371, wR2 = 0.1036, Flack parameter (Friedel pairs = 2036) 0.003(8). CCDC 1045375.
- 21 M. Murai, K. Matsumoto, R. Okada and K. Takai, *Org. Lett.*, 2014, 16, 6492.