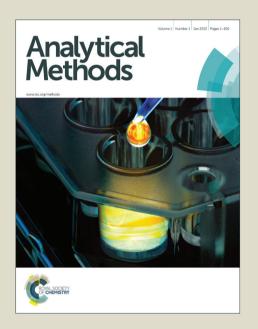
Analytical Methods

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Rapid Label-Free Detection of *E. coli* using Antimicrobial Peptide **Assisted Impedance Spectroscopy**

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There is an increasing demand for rapid detection of waterborne pathogens to monitor drinking water safety. We demonstrate a compact, label-free sensor array for rapid detection of Escherichia coli (E. coli) in contaminated water samples using antimicrobial peptide assisted impedimetric sensor platform. Interdigitated electrode arrays immobilized with the antimicrobial peptide Colicin V (ColV) were used to screen the affinity towards different bacterial strains by monitoring impedance variations in real-time. This ColV asssisted impedance biosensor exhibited high selectivity towards gram-negative strains particularly towards E. coli strains. This selective detection of E.coli from other strains was observed at 10² cfu mL⁻¹, which is clinically relevant. The sensor can detect E. coli from 10² to 10⁶ cfu mL⁻¹ in water sample at pH 7 to 9. These results show that the antimicrobial peptide ColV assisted impedimetric array is capable of rapid, specific detection of E. coli in contaminated water samples.

Introduction

Escherichia coli (E. coli) is a well-known type of Gram-negative bacteria with certain strains causing serious illness through the synthesis of the Shiga toxin.¹ The most commonly identified Shiga toxin-producing E. coli (STEC) is E. coli O157:H7.2 It is estimated that 265,000 STEC infections occur each year in the US.³ According to World Health Organization (WHO), E. coli testing is a paramount factor in verifying the microbial safety of water for human consumption. The guideline value of E. coli or thermotolerant coliform bacteria in drinking water is 0 cfu per 100 mL.4 It is becoming increasingly evident that biosensing detection methods should be rapid, sensitive and highly specific in order to avoid further damages caused by life threatening pathogens. The culture and colony counting method may take a long time due to the

In this study, we demonstrate an impedance sensor for rapid detection of E. coli in water samples utilizing antibacterial peptide Colicin V (ColV). ColV is an 88 amino acid, linear unmodified antimicrobial peptide (AMP).8 It is a type of bacteriocin produced by E. coli which acts against E. coli strains by binding to the outer membrane receptors and using them to translocate the cytoplasmic membrane, leading to membrane depolarisation, which ultimately kills the cell.^{8,9} In recent years, several studies reported the use of AMPs as pathogen recognition elements in biosensor arrays. They have been proven to be a robust receptor layer with high selectivity towards pathogenic strains of bacteria. 10 Therefore, we have used CoIV as a selective receptor layer for specific recognition of E. coli strains in the impedance spectroscopy. By monitoring the impedance amplitude change due to biomolecular binding on the electrodes, impedance spectroscopy offers a rapid, compact, lowcost, and label-free biosensing technique. The sensors have shown enhanced sensitivities and achieved lower detection limits compared to traditional impedance analysers. The impedance amplitude varies with the injected bacteria samples due to the number and type of adsorbed bacteria on the electrode surface. With the ColV functionalized impedance sensor, we have determined the selectivity, detection limit, and the detection capability for E. coli contaminated water samples.

Experimental Section

CoIV was genetically expressed and purified from cultures of E. coli strain MC4100 as previously described. E. coli strain ATCC

reproduction rate of the bacteria. Other methods including nucleic detection chain reaction/PCR), (e.g. polymerase enzyme/substrate method (e.g. bacteria enzymatic activity) and whole-cell recognition (e. g. immunoassay, bacteriophage) are efficient in terms of sensitivity and specificity, but require trained personnel to conduct tests.^{5,6} Currently, most bacteria detection methods are label-dependent, for example, immunoassay, which uses labelled monoclonal antibodies (Mab) to recognize unique bacterial specific antigens. In addition, most antibodies are expensive and unstable in harsh environment. Thus, there is an urgent need to develop rapid sensors for selective and sensitive detection of E. coli in suspected water samples.

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25922, L. monocytogenes strain ATCC 19116, S. enteritidis strain ATCC 13076, S. aureus strain ATCC 13566 and P. fluorescens strain CHAO were used in this study as detection targets.

The impedance sensor array consists of three microfluidic reaction chambers with five pairs of gold interdigitated electrodes (IDEs) in each chamber. The dimensions of each electrode are 3350 μ m/100 μ m/150 nm and the spacing between each electrode is 40 μ m (Fig 1A and 1B).

The gold electrodes were functionalized with ColV through a covalent bond formation between the free amine group of a thiol linker attached to the electrodes and the carboxylic group on the peptide. The IDEs were functionalized first with a self-assembled monolayer (SAM) of cysteamine linker (HSCH₂CH₂NH₂·HCl). The ColV immobilization was performed overnight at room temperature by injection of EDC/NHS activated ColV stock solution. Before each experiment, the functionalized sensor was rinsed with PBS to remove any unbound ColV.

A SHARP IA-2 impedance biosensor (SHARP Laboratory of America, USA) was used to measure the impedance response caused by the affinity binding of the bacteria. When the functionalized CoIV sensing layer captures the target bacteria (Fig 1C), the impedance variation caused by the binding of the bacteria is measured in real-time. In order to avoid any interference and eliminate any signal as a result of non-specific adsorption and buffer effects, we used a blank reaction chamber with corresponding buffer/solution as a reference channel for differential readout.

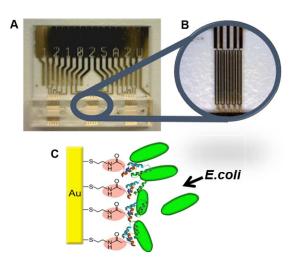


Fig 1. A. Image of the impedance sensor chip; B. Zoomed-in view of the interdigitated electrodes in one microreaction chamber. C. Schematic showing the adsorption of *E. coli* on the ColV functionalized gold electrode.

The sensitivity, resolution, and the selectivity of the sensor array were tested by injecting $\it E.~coli$ and $\it L.~monocytogenes$ with a serial concentration from 10^2 to 10^6 cfu mL $^{-1}$. The bacteria samples were injected into microreaction chambers separately. After the impedance magnitude variation reached an equilibrium state, the bacteria samples, with differing concentration, were injected. To rinse the micro-reaction chambers, PBS was injected into the micro-reaction chambers after the last sample injection. This also allowed us to test the binding stability between ColV and the bacteria target.

Contaminated artificial beverages and water samples with various pHs were injected into the reaction chambers to test the sensor for unknown solutions using real-world samples. All the test results were monitored and analysed using the Sharp BioZ software package. Impedance magnitude and variations in other parameters were compared to determine the sensor characteristics.

3. Results and Discussion

3.1 Data analysis method

A Fourier Domain scheme was used for calculating the impedance. The data analysis software package integrated in the Sharp IA-2 user interface was based on a previously described algorithm. Using surface chemistry theories, the target bacteria bound to the immobilized sensing layer at a rate that follows an exponential function. This rate decays with time after the bacteria is injected into reaction chambers. As a result, the electrode impedance varies as a function of time and it is modelled by:

$$|Z| = B + A(1 - e^{-st}) \tag{1}$$

where s, A and B are constants. B is the offset at which the exponential increase of the impedance modulus begins, and it stands for the impedance baseline of the instrument with the target-free PBS buffer solution. Parameter s expresses the decay time constant which tracks the concentration of the analyte in each reaction chamber. A is related to the sensor surface coverage which relates to the molecular affinity of target-probe interactions. A also represents the "end point" of the binding curve which indicates the density of the sensing layer molecules, size of the analyte particles (bacteria), and dielectric property of the target molecules. From the derivative of the equation (1), the initial slope of the binding curve can be expressed as

$$tan \ \gamma = s \times A \tag{2}$$

Parameter s can be obtained using the initial slope and the end point of the binding curve. Another parameter of interest is the area covered under the impedance modulus curve which represents the total adsorption amount of the target molecules on the electrode surface. By analysing parameters A, s and area, the probe-target interactions and adsorption kinetics can be studied. 12

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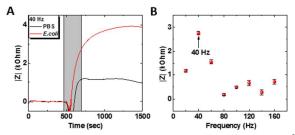


Fig 2. Impedance magnitude increase with the injection of 10^3 cfu mL⁻¹ *E. coli* sample and PBS buffer in the ColV functionalized microreaction chamber. A: impedance magnitude increase at 40 Hz. B: The impedance magnitude differential between *E. coli* and PBS sample, experiment performed at 40 Hz gives largest difference between the *E. coli* sample and PBS buffer.

3.2 Frequency optimization

The frequency for optimal operation of the impedance sensor was determined by injecting various samples with and without *E. coli* into the peptide functionalized sensor chambers, and monitoring the response as a function of frequency (20 to 160 Hz). **Fig 2A** presents the typical impedance magnitude change with 10³ cfu mL⁻¹ *E. coli* sample injection. The blank represented the signal of PBS injection in ColV functionalized chamber without target *E. coli* cells. The differential between the *E. coli* and the blank (net impedance magnitude) was compared at various frequencies (as shown in **Fig 2B**) and at 40 Hz the impedimetric magnitude difference was observed to be maximum at ~3700 Ohm between the *E. coli* injection and the PBS injection. Thus, 40 Hz was carefully chosen as the optimum frequency for *E. coli* detection with ColV sensing layer in this study.

3.3 Selectivity and sensitivity

The selectivity and sensitivity of this functionalized impedimetric sensor was investigated via net impedance magnitude variation by injecting bacterial solutions at various concentrations (Fig 3A).

From 10² to 10⁵ cfu mL⁻¹, the impedance magnitude increases from 1800 Ohm and reaches equilibrium in about 500 sec. Response variation for every order of magnitude change in *E. coli* concentration is shown in **Fig 3B**. After the injection of 10⁶ cfu mL⁻¹ sample of bacteria, the impedance magnitude showed an increase of over 4000 Ohm in about 1000 sec. This phenomenon indicates the ColV functionalized electrode surface is fully covered by the bacteria at high concentration and there is physisorption of the bacteria on the electrode surface. However the impedance magnitude variation is minimum with *S. enteritidis, S. aureus, P. fluorescens* and *L. monocytogenes* concentration lower than 10⁶ cfu mL⁻¹, indicating that the ColV is less sensitive to other bacteria strains compared to *E. coli*.

The exponential time constant s is another parameter for the measurement of the AMP-bacteria interaction. The time constant represents the adsorption rate of the bacteria on the electrodes surface. From **Fig 3C**, the adsorption rate of E. coli on the ColV functionalized surface decreases with the bacteria concentration from 10^2 to 10^5 cfu mL $^{-1}$. One possible reason is the adsorption of E. coli cells reduced the active sites thus the adsorption rate decrease with concentration. However, the time constants of S. enteritidis, enteritidis

Mixed samples of *E. coli* and *L. monocytogenes* were injected into the microreaction chambers in order to test the sensor performance with multiple bacteria strains (purple data curve in **Fig 3A, 3B and 3C**). The impedance magnitude variation and the time constant change of the mixed bacteria injection are similar with that of *E. coli* injections indicating the ColV sensing layer is specific to *E. coli*. For both *E. coli* and the mixed sample, the time constant decrease with the increase of bacteria concentration which suggests the sensor is selective to *E. coli*.

The ColV functionalized impedance sensor showed two orders of magnitude lower detection limit within the short detection time compared with the ELISA method (10⁴ cfu mL⁻¹).¹³ Also, compared with previous impedimetric bacteria sensors for heat-killed *E. coli* cells,¹⁴ this ColV functionalized sensor is able to detect live pathogenic *E. coli* cells at a similar concentration range in much shorter detection times. This detection limit is also comparable with impedimetric sensors for Gram-positive bacteria detection

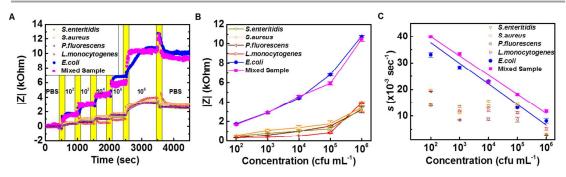


Fig 3. Impedance sensor signal variation via bacteria samples injections. Sensitivity and selectivity of the sensor was compared. A: impedance magnitude change with experiment time; the mixed sample of *E. coli* and *L. monocytogenes* represented the similar signal with *E. coli* samples. B: impedance magnitude increase with *E. coli* concentration from 10^2 to 10^5 cfu mL⁻¹. C: time constant *s* decrease with *E. coli* concentration, it did not show clear decrease with other strains.

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Table 1. Determination of *E. coli* concentration in PBS buffer. *E. coli* samples were prepared separately and the impedimetric magnitude change due to *E. coli* samples injection were compared with the previously determined impedimetric magnitude-*E. coli* concentration relationship in **Fig 3B.**

	Z (kOhm)	Measured Concentration (cfu mL ⁻¹)	OD ₆₀₀ Concentration (cfu mL ⁻¹)
SPL1	6.5558	110000	100000
SPL2	5.4815	22000	20000
SPL3	4.3296	9900	10000
SPL4	2.8782	1010	1200
SPL5	3.6715	2400	2000

developed by Etayash et al.¹⁵ With this detection method for Gramnegative bacteria, a fast-scan impedimetric sensor with low detection limit for pathogenic bacteria is expected.

3.4 Artificial sample detection

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E. coli-spiked PBS solutions were prepared as simulated contaminated water samples. The impedance variations due to artificial samples were compared with the impedance-concentration correlation developed in the sensitivity test (Fig 3B) for estimation of E. coli concentration in artificial samples. The E. coli concentrations measured using the impedimetric sensor are listed in Table 1.

Comparing with the OD_{600} concentration, the results measured from the impedance sensor have a 10% error for concentrations from 10^4 to 10^5 cfu mL⁻¹. But the error is higher with lower concentration around 10^3 cfu mL⁻¹, indicating the resolution limit of the sensor. However, from the perspective of bacteria screening, the CoIV functionalized impedance sensor is suitable for differentiating pre-concentrated samples at concentrations orders of magnitude greater than the conventional techniques and is accurate within the optimum concentration range.

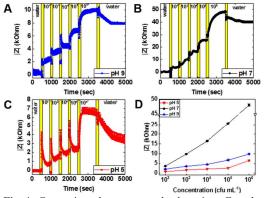


Fig 4. Contaminated water sample detection. *E. coli* was added into water samples with pH values of 9, 7 and 5. Impedance magnitude variation with increasing *E. coli* concentration in water at: A. pH 9; B. pH 7; C. pH 5. D. impedance magnitude increase much higher at pH 7, possibly due to the lower conductivity of pH 7 water. ColV binding activity was inhibited at pH 5.

3.5 Water sample detection

Most consumable beverages have a wide pH range, varying from 3 to 9.5. We prepared water samples with pH 5, 7 and 9 by adding HCI/NaOH into MilliQ water. Artificially contaminated samples were prepared by adding *E. coli* into the prepared water samples with various concentrations. The contaminated water samples were tested using the ColV functionalized impedance sensor as described in the previous section.

In **Fig 4A, 4B and 4C**, the impedance magnitudes increase with *E. coli* concentrations for all three samples. The pH 9 sample represented an identical result as that in PBS buffer which suggests that the sensor is capable of operation at pH values as high as 9. However, the impedance variation is about five times larger in the pH 7 sample than for the other two (**Fig 4D**). One possible reason is the lower conductivity at neutral pH. On the other hand, the impedance magnitude increase measured for the pH 5 sample is much lower than previous samples, especially in the concentration range from 10^2 to 10^5 cfu mL⁻¹. This indicates that ColV binds less at pH 5 than at pH 7 and pH 9. This dependence of ColV at varying pH levels was previously reported. Thus, our ColV functionalized sensor has high sensitivity for *E. coli* detection in water at pH 7 to 9, but has reduced sensitivity in the lower pH range.

4. Conclusion

Gram-negative antimicrobial peptide ColV demonstrated as a bioreceptor in a highly selective and sensitive impedimetric biosensor platform for fast-scan, label-free E. coli detection. Using the impedimetric modulus change, we demonstrated a lower detection limit for E. coli at 10² cfu mL⁻¹ which is clinically relevant. Furthermore, this sensor is able to distinguish E. coli from other bacteria strains including S. enteritidis, S. aureus, P. fluorescens and L. monocytogenes. The ColV functionalized sensor was successful in detecting E. coli cells in the artificial samples and water samples in the pH range of 7 to 9. These results suggest that a CoIV assisted impedimetric sensor has the potential to be used in the rapid detection of E. coli in water samples with high sensitivity and selectivity. Experiments are presently underway to establish stability and uniformity of this ColV assisted impedance sensor for field applications. . This study also suggests that narrow spectrum antimicrobial peptides can be a good sensing layer for different biosensor platforms such as surface plasmon resonance (SPR) and microcantilevers for label-free detection.

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Rapid Label-Free Detection of *E. coli* using Antimicrobial Peptide Assisted Impedance Spectroscopy

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Graphical Abstract

A compact, label-free sensor array for rapid detection of *Escherichia coli (E. coli)* using antimicrobial peptide assisted impedimetric sensor platform.

