

Analytical Methods

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4 1 Strategy for rapid structural characterization of saponins and flavonoids from
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6 2 the testa of *Camellia oleifera* Abel. seed by ultra-high-pressure liquid
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8 3 chromatography combined with electrospray ionization liner ion trap-Orbitrap
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10 4 mass spectrometry

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4 19 **Abstract:** Saponins and flavonoids are the main bioactive ingredients in
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6 20 *Camellia oleifera* Abel.. In this study, the fragmentation pathways of triterpene
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8 21 saponins in *Camellia oleifera* were investigated and confirmed by electrospray
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10 22 ionization liner ion trap-Orbitrap (LTQ-orbitrap), and a new strategy for rapid
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12 23 characterization of saponins by ultra-high-pressure liquid chromatography with
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14 24 LTQ-orbitrap mass spectrometry (UHPLC-LTQ-orbitrap-MSⁿ) was developed. Based
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16 25 on the summarized fragmentation rules, 36 triterpene saponins from testa of *C.*
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18 26 *oleifera* extracted by n-butanol were found, 23 of which were completely and
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20 27 tentatively characterized. Besides, 35 saponins were identified as novel saponins.
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22 28 Additionally, 8 flavonol-*O*-glycosides were separated from *C. oleifera*, with 7 being
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24 29 tentatively characterized but one unambiguously identified, and one of the 7
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26 30 flavonol-*O*-glycoside was first reported in genus *Camellia*. This study provided a
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28 31 systematic strategy in identification of saponins by UPLC-MSⁿ method in genus
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30 32 *Camellia* and summarized essential data for phytochemical studies of *C. oleifera*.

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34 33 **Keywords:** *Camellia oleifera*; UHPLC-LTQ-orbitrap-MSⁿ; Flavonoids; Triterpenoid
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36 34 saponins; Orbitrap
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36 1. Introduction

37 *Camellia oleifera* Abel, a traditional oil crop in south China, has been widely planted
38 as an economic crop. The seeds of *C. oleifera* Abel. are the raw materials for the extraction
39 of edible oil containing its unsaturated fatty acids contents and nutrients¹. However, as its
40 byproduct, the testa of seeds is commonly used as a detergent or an organic fertilizer with
41 low economic value, resulting in the waste of many bioactive compounds, such as
42 flavonoids and saponins. Flavonoids is a kind of important bioactive compounds with
43 antioxidant activity^{2,3} and anticancer activity⁴, which has been isolated and structurally
44 identified in literatures²⁻⁶. Saponins are the major active ingredients in *C. oleifera* Abel,
45 Which have various bioactivities, such as antimicrobial activity⁷, antioxidant activity^{8,9}
46 and cytotoxic activity¹⁰. However, few studies elucidated the structural of saponins from *C.*
47 *oleifera*. As far as we are concerned, only 9 types of triterpenoid saponins^{3,9-11}, which were
48 the same (2 types) as or similar to the saponins from genus *Camellia*¹², have been isolated
49 and structurally identified by nuclear magnetic resonance spectroscopy (NMR).

50 It is available to isolate and identify the high-content components by NMR with, but
51 not appropriate for the minor compounds¹³. Multi-stage mass spectrometry (MSⁿ) is a
52 valuable method in characterization of compounds¹⁴⁻¹⁷, when combined with liquid
53 chromatography, especially ultra-high-pressure Liquid chromatography (UHPLC), it is
54 suitable for rapid structural analysis in plant extracts^{18,19} such as flavonoids^{20,21} and
55 triterpene saponins^{22,23}, due to its high sensitivity, short time and low consumption²⁴⁻²⁶,
56 Better performance can be obtained when the high resolution mass spectrometric analyzer
57 are (HR-MS) applied in the UHPLC-MSⁿ system by providing exact mass and possible

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3 58 chemical formula^{13, 27}, which can overcome the difficulties in distinction of isobaric
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5 59 compounds²⁸. Recently, the LTQ-orbitrap, an emerging high resolution mass spectrum
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7 60 detector which can provide high mass accuracy measurements for precursor and product
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9 61 ions rapidly and simultaneously, has been developed as an useful technology in detection
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11 62 and identification of plant extracts^{29, 30}.

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16 63 The saponins characterized by LC-MS² in some genera have been reported, such as
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18 64 *Glycyrrhiza*²³, *Symplocos*²⁶, *Momordica*³¹ and *Albizia*³². Their typical fragment pathway is
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20 65 loss of monosaccharide residues and acyl groups, but different structures exhibit difference
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22 66 in fragment behaviors. Few literatures studied on genus *Camellia*^{8, 33}. Scoparo have
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24 67 identified about 20 different saponins derived from tea leaves without describing the
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26 68 process of identification³³. Hu deduced 9 saponins from *C. oleifera* in positive ion mode⁸
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28 69 and gave a brief explanation for the identification. However, the fragmentation behavior
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30 70 and pathway of triterpene saponins in genus *Camellia* has not been clearly and
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32 71 systematically studied yet.

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38 72 In this paper, the fragmentation pathways of triterpene saponins in *C. oleifera* were
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40 73 systematically studied and confirmed by LTQ orbitrap. Based on the summarized
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42 74 fragmentation rules, we developed a new strategy for rapid characterization of saponins in
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44 75 n-butanol extract by UPLC-LTQ-orbitrap-MSⁿ. Also, flavonol-*O*-glycosides were separated
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46 76 and identified. This study contributed to the identification of saponins in genus *Camellia*
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48 77 and provided essential data for further phytochemical study of *C. oleifera*.

52 53 78 **2. Materials and methods**

54 55 56 79 **2.1 Chemicals and materials**

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3 80 The testa of *C. oleifera* Abel seed was purchased from Realpark Company (Hunan,
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5 81 China) and collected from Hunan province, China. Camelliaside A was purchased from
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8 82 Biopurify Phytochemicals Ltd (Chengdu, China), and total saponins (purity > 96%) of *C.*
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10 83 *oleifera* were purchased from Solarbio Co. (Beijing, China).

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13 84 Analytical grade formic acid, HPLC-grade methanol (MeOH) and acetonitrile (ACN)
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15 85 were purchased from Merck (Darmstadt, Germany). Analytical grade ethanol, ethyl acetate
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17 86 (EtOAc), n-butanol (BuOH) and trichloromethane (CHCl₃) were purchased from Beijing
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19 87 Chemical Works (Beijing,China). Water was purified by a Milli-Q system (Milford, MA,
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21 88 USA).

26 89 **2.2 Sample preparation**

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30 90 The testa of *C. oleifera* seeds (5 g) were powdered and then refluxed at 55 °C with 70%
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32 91 ethanol (3×150 ml, 1 h) to get the crude extract. The extract was evaporated to dry, and
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34 92 then partitioned in CHCl₃, EtOAc, BuOH, and H₂O. The BuOH extract was filtered,
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36 93 evaporated and freeze dried, then dissolved with 50% MeOH to a concentration of about
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38 94 2mg/ml. The solution was filtered through a 0.22 μm membrane, and a volume of 5 μL was
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40 95 injected for UPLC-MS³ analysis.

44 96 **2.3 UHPLC-MS³ Method**

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48 97 UHPLC-MS³ analysis was carried out with a Thermo UltiMate 3000 UHPLC system
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50 98 and an LTQ Orbitrap XL mass detector (ThermoFisher, CA, USA). The UHPLC system
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52 99 consisted with a quaternary pump, a diode-array-detector, an autosampler and a column
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54 100 compartment. An Agilent ZORBAX SB-C18 column (2.1×100 mm, 1.8 μm; Agilent, Palo
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56 101 Alto, USA) was used at the flow rate of 0.2 ml/min. The column temperature was 35 °C,
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3 102 and the mobile phase was water containing 0.1% formic acid (A) and the mixture of MeOH
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5 103 and ACN (50:50, v/v) containing 0.1% formic acid (B). A linear gradient elution program
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8 104 was as follows: 20% B (v/v) in 0-5 min, 20-80% B in 5 min-12 min, 80% B in 12-18 min,
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10 105 100% B in 18.1-30 min, and 20% B in 30.1-40 min. The ESI source was operated both in
11
12 106 positive and negative ion mode. High purity nitrogen (N₂) was used as sheath gas (30arb)
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14 107 and auxiliary gas (10arb). High purity helium (He) was used as collision gas. In positive ion
15
16 108 mode, the other conditions were as follows: source temperature, 300 °C; Source voltage,
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18 109 4.5 kv; Capillary voltage, 35 V; Tube lens, 110 V. For negative ion mode, source
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20 110 temperature was 350 °C and tube lens was 110 V, the other conditions were the same as in
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22 111 the positive mode. Full scan data acquisition (mass range: m/z 100-2000) and data
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24 112 dependant MS³ were acquired. The resolution was 30,000 for full mass scan and 15,000 for
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26 113 data dependant MS scan. The CID collision energy was adjusted to 35% of the maximum
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28 114 and the isolation width of precursor ions was set at m/z 2.0.
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35 115 **3. Results and discussion**

36 37 38 116 **3.1 Extraction and Partition**

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41 117 To remove unnecessary compounds, liquid-liquid partition was used in separation based
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43 118 on the polarity. Each of the four extracts (CHCl₃, EtOAc, BuOH, and H₂O) was detected by
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45 119 UHPLC-MS, and the results showed the main bioactive compounds, flavonol-*O*-glycosides
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47 120 and triterpene saponins mostly existed in BuOH extract. Thus, BuOH extract were used for
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49 121 analysis. The base peak chromatograms of BuOH extract were shown in Fig. 1. However,
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51 122 saponins were hard to be separated by UHPLC because of their variability and similar
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53 123 structures³³, so extracted ion chromatogram (EIC) was applied to identify their structures.
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59 124 **3.2 Characterization of flavonol-*O*-glycosides**

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3 125 Totally, 8 flavonoids were identified in *C. oleifera* Abel., and all of them were
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6 126 flavonol-*O*-glycosides, including one found in this genus for the first time. The
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8 127 fragmentation pathway for flavonol-*O*-glycosides has been studied in detail ²⁰, in which
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10 128 loss of monosaccharide residues are a typical way to obtain the aglycone structure and then
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12 129 the carbohydrate sequence. The structures of flavonol-*O*-glycosides in *C. oleifera* Abel. are
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14
15 130 usually characterized by the linkage of α -L-rhamnopyranose, β -D-glucopyranose,
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17 131 acetyl- α -L-rhamnopyranose and xylopyranose to the flavonoid skeleton through the C-3
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20 132 group.

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23 133 Both positive and negative ion modes were collected to identify the
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25 134 flavonol-*O*-glycosides in *C. oleifera* Abel (Table 1). All 8 flavonol-*O*-glycosides gave an
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27 135 identical fragment ions in the MS² spectra at m/z 287.05637 (C₁₅H₁₁O₆; [Aglycone+H]⁺) in
28
29 136 positive ion mode and m/z 285.03961 (C₁₅H₉O₆; [Aglycone-H]⁻) in negative ion mode
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31 137 (compound **5** showed a much higher relative abundance of [Aglycone-H][•] ion at m/z 284
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33 138 than that of [Aglycone-H]⁻ ion ³⁴, which was formed by the hemolytic cleavage probably
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35 139 due to the character of the sugar substitution³⁵. Although it could not be observed in all
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37 140 flavonol-3-*O*-glycosides, the existence of remarkably higher relative abundance of
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39 141 [Aglycone-H][•] ion was very pronounced for flavonol-3-*O*-glycosides³⁶). Moreover, the
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42 142 MS³ spectra of m/z 287 and m/z 285 were consistent (fig.2), which proved the same
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45 143 aglycone.

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50 144 Compared with standard compound, as well as the literature data ³³ of UV spectra data
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52 145 λ_{\max} at 266nm and 348nm and high resolution accurate molecular weight (ppm < 5),
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54 146 Compound **1** was unequivocally identified as camelliaside A. Since all
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56 147 flavonol-*O*-glycosides showed similar fragmentation pathway by loss of monosaccharide
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58 148 residues, camelliaside A was used as a standard. In positive ion mode, camelliaside A

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3 149 displayed a $[M+H]^+$ ion at m/z 757.22092 ($C_{33}H_{41}O_{20}$) and MS^2 fragment at m/z 595.16770
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5 150 ($C_{27}H_{31}O_{15}$; $[M+H-162]^+$), m/z 449.10884 ($C_{21}H_{21}O_{11}$; $[M+H-162-146]^+$) and m/z
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7 151 287.05641 ($C_{15}H_{11}O_6$; $[M+H-162-146-162]^+$), corresponding to the cleavage of glucose
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9 152 (Glc), rhamnose (Rha) and glucose unit successively. Aglycone was further confirmed to be
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11 153 kaempferol by product ions of m/z 287.05641 ($C_{15}H_{11}O_6$), including neutral loss of CO
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13 154 (m/z 259, the product ion of kaempferol; m/z 231, the product ion of m/z 259), H_2O (m/z
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15 155 269, the product ion of kaempferol; m/z 213, the product ion of m/z 231), retro-Diels-Alder
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17 156 fragmentation (m/z 153), some other rearrangements (m/z 259, m/z 258 and m/z 241), and
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19 157 the cleavage of the C ring from kaempferol protonated at C-3 position (m/z 165) which
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21 158 were consistent with former reports³⁷ (fig. S3C, shown in supporting materials). Moreover,
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23 159 the MS^3 spectra could be used to distinguish kaempferol and its isomers,
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25 160 2'-hydroxygenistein and luteolin (fig. S4, shown in supporting materials). Though they
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27 161 showed the same m/z value at 287 in MS^2 spectra, the product ions in MS^3 spectra were
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29 162 different. The methoxy unit at C-3 position in the C ring for kaempferol had more
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31 163 fragmentation pathways than others. The typical fragment ion $^{0,2}A^+$ ion at m/z 165, from
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33 164 protonation at C-3 position, could only be yielded by kaempferol (Scheme S1, shown in
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35 165 supporting materials)^{28, 38}. Thus, MS^n strategy was necessary in confirmation of aglycone.

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44 166 In negative ion mode, the fragmentation pathway of aglycone at m/z 285.03961 was
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46 167 similar to that in positive ion mode (fig. S3D, shown in supporting materials), but MS^2
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48 168 spectrum was different. Loss of disaccharide residues made it hard to deduce the sequence
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51 169 of glycosides. Meanwhile, flavonol-*O*-diglycosides, rutinose
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53 170 (L-rhamnopyranosyl-(1→6)-_D-glucopyranose) and neohesperidose
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55 171 (L-rhamnopyranosyl-(1→2)-_D-glucopyranose) could be distinguished by MS^2 spectrum in
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57 172 negative ion mode. Rutinose mainly caused aglycone fragment (for compound 7), while
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3 173 neohesperidose yielded more product ions at higher m/z values, corresponding to the
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5 174 $[M-H-Rha]^-$ ion, $[M-H-Rha-H_2O]^-$ ion, $[M-H-Rha-120]^-$ ion and $[M-H-120]^-$ ion³⁹. Besides,
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8 175 for flavonoids with two hexoses, the 1→2 and 1→6 interglycosidic linkages could be
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10 176 distinguished by the $[M-H-180]^-$ ion⁴⁰. Therefore, positive ion mode was generally used to
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12 177 identify the possible constituents of other flavonol-*O*-glycosides and negative ion mode
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14 178 was used to distinguish the linkage type.

179 The carbohydrate sequence could be easily determined by the loss of monosaccharide
180 residues in positive ion modes: the loss of 162Da (C₆H₁₀O₅), 146Da (C₆H₁₀O₄), 132Da
181 (C₅H₈O₄) and 188Da (C₈H₁₂O₅) corresponding to the cleavage of hexose, deoxyhexose,
182 pentose and acetyl-deoxyhexose respectively. However, for most compounds,
183 systematically structural identification of the sugar chain, including the linkage type and
184 steric structure, was unavailable from MS data. Their full structures were tentatively
185 deduced by literature (Table 1). Therefore, the hexose could be determined as glucose (for
186 compounds **1**, **2**, **4**, **5**, **6**, **7** and **8**) or galactose (Gal) (for compounds **2** and **3**) and the
187 deoxyhexose was supposed to be rhamnose and the pentose was identified as xylose (Xyl)
188 (Fig. 2). Additionally, the linkage type could be tentatively deduced. It was also used for
189 the identification of saponins.

190 By investigating the reported compounds, compounds **2-8** were tentatively identified.
191 All of them were reported in *C. oleifera*, except compounds **2** and **3**.

192 Compound **2** was comprised of a kaempferol unit and two hexose units. The high
193 relative abundance of $[M-H-180]^-$ ion at m/z 429 indicated the (1→2) interglycosidic
194 linkage⁴⁰. Compared with literature data⁴¹, the MS spectra data was the same as for
195 camelliaside C (kaempferol-3-*O*-β-D-galacopyranosyl-(1→2)-β-D-glucopyranoside) in *C*.

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3 196 *sinensis*. However, the stereoisomer, Kaempferol-3-*O*-sophoroside (kaempferol-3-*O*- β -D-
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5 197 glucopyranosyl-(1 \rightarrow 2)- β -D-glucopyranoside), has also been reported in *C. oleifera* Abel².
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8 198 Since the stereochemical structure of the hexose could not be identified only by MSⁿ
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10 199 spectra, compound **2** was tentatively identified as Kaempferol-3-*O*-sophoroside or
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12 200 camelliaside C.

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16 201 Compound **3** was comprised of a kaempferol unit, a hexose unit and two
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18 202 methylpentose units, which has not been reported for this genus before. Based on the
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20 203 Chemical Abstracts database (<https://scifinder.cas.org/>) , Only one compound ,
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22 204 kaempferol-3-*O*- α -L-rhamnopyranosyl-(1 \rightarrow 4)-[α -L-rhamnopyranosyl-(1 \rightarrow 6)- β -D-galactopyr-
23
24 205 anoside]⁴², fits the MS spectra data. Thus, compound **3** was presumed to be this
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26 206 compound.

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31 207 Of all the flavonol-*O*-glycosides, half of them (compounds **1**, **4**, **5** and **7**) have been
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33 208 reported with bioactivity. Camelliaside A and B have proved to have inhibition on the
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35 209 growth of hepatocellular carcinoma cells significantly⁴. Besides, compounds **5** and **7** have
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37 210 been reported to have moderate DPPH radical scavenging activity². They showed potential
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39 211 use in medicine.

212 **3.3 Characterization of triterpenoid saponins**

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214 To date, over 100 kinds of saponins in this genus, including acylated pentacyclic
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216 215 triterpenoid saponins, have been reported, and most of them are oleanane-type triterpenoid
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218 216 saponins. They have been characterized by glycosides at C-3 and one or more acyl groups
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220 217 at C-16, C-21, C-22 and C-28 in aglycones^{9, 10, 12, 43-46}. Since there was no previous study,
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222 218 we observed the mass spectrometric behaviors of the saponins in *C. oleifera* and
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224 summarized their fragmentation rules systemically.

3.3.1 Fragmentation behavior of triterpenoid saponins

Compared with total saponin in fragmentation behavior and literature data⁹, 36 compounds in *C. oleifera* Abel were tentatively identified as triterpenoid saponins. Their mass spectrometric behavior were summarized and the element composition of the fragment ions were confirmed by Orbitrap.

In positive ion mode, most saponin gave a $[M+NH_4]^+$ ion but sometimes a $[M+Na]^+$ ion as base peak in MS spectra. In MS² spectra, $[M+Na]^+$ ion yielded produce ions $[Glycoside+Na]^+$ at m/z 655.17120 ($C_{23}H_{36}O_{20}Na$; $[Hexose+hexose+pentose+gluconic\ acid\ (GlcA)+Na]^+$) or m/z 685.17975 ($C_{24}H_{38}O_{21}Na$; $[Glc+Gal+Gal+GlcA+Na]^+$), but for $[M+NH_4]^+$ ion, the data was messy in the MS² spectrum.

In negative ion mode, most components had similar fragmentation behavior by losing acyl groups and sugar residues. Moreover, most saponins, except the desacyl-saponins, displayed a sugar residue ion in low abundance at m/z 451.10842 ($C_{17}H_{23}O_{14}$; $[Pentose+hexose+GlcA-H]^-$) or m/z 481.11891 ($C_{18}H_{25}O_{15}$; $[Gal+Gal+GlcA-H]^-$) by loss of a hexose or pentose in glycosides, while the desacyl-saponins showed $[Aglycone-H]^-$ ion instead of sugar residue ion due to lack of acyl groups. It could be used to distinguish normal saponins and desacyl-saponins, and the ion at m/z 451 or m/z 481 could be considered as a diagnose ion for normal saponins.

From low resolution mass spectra data, the sugar residue ion at m/z 451 ($C_{17}H_{23}O_{14}$) was hard to be distinguished from the $[Aglycone-H]^-$ ion ($C_{30}H_{43}O_3$; m/z 451) (the aglycone of Camelliasaponin B₁⁴⁷), corresponding to two different fragmentation pathways, which were reported in triterpenoid saponins^{26, 31}. However, with LC-LTQ-orbitrap, the ion at m/z 451.10842 could be confirmed ($C_{17}H_{23}O_{14}$; $[Pentose+hexose+GlcA-H]^-$). The MS³

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3 242 spectra could also help to support this point. For the saponins which glycosides consisted of
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5 243 2 hexose, a pentose and a GlcA, the $[M-H-hexose]^-$ ion could yield this ion in MS^3 spectra,
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7
8 244 while the $[M-H-pentose]^-$ ion could not. Therefore, the ion at m/z 451 could not be the
9
10 245 $[Aglycone-H]^-$ ion. In conclusion, UHPLC-LTQ-orbitrap- MS^3 was critical by providing the
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12 246 high accuracy mass measurements for the product ions.

16 247 **3.3.2 Strategy for identification of triterpenoid saponins**

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19 248 Due to the various kinds of structure, the separation and identification of saponins were
20
21 249 challenging. Thus, a strategy for rapid identification of triterpenoid saponins was proposed
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23 250 as shown in Fig. 3. The identification of saponins could be divided into three parts: acyl
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25 251 groups, glycosides, and aglycones. Firstly, for the acyl groups, whether the sugar residue
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27 252 ion at m/z 451 or m/z 481 existed or not could distinguish normal saponins and
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29 253 desacyl-saponins. For desacyl-saponins, the acyl group did not exist, and for normal
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31 254 saponins, it could be inferred from $[M-H-NL]^-$ ion by neutral loss (NL) of acyl group. The
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33 255 second part was glycoside. They could be deduced from the loss of monosaccharide
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35 256 residues successively, which were similar to flavonol-*O*-glycosides. The last part was
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37 257 aglycones. For desacyl-saponins, the formula for aglycones could be deduced from the
38
39 258 $[Aglycone-H]^-$ ion, while for normal saponins, it could be calculated from the
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41 259 $[Aglycone+acyl-H]^-$ ion or $[Aglycone+acyl+H_2O-H]^-$ ion after inferred the acyl group. As a
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43 260 result, their aglycones could tentively identify based on the same formula as previous
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45 261 reports. This method is useful to identify the aglycones in saponins, such as in *Glycyrrhiza*
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47 262 *yunnanensis*²³.

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55 263 The fragmentation pathway of compound **29** was shown in Fig. 4, as an example for the
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57 264 identification of triterpenoid saponins. The MS^2 spectra gave a $[M-H-102]^-$ ion at m/z
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3 265 1101.50884 ($C_{53}H_{81}O_{24}$), indicating a MBOH (methylbutyryl-OH) unit or IsoVOH
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5 266 (isovaleryl-OH) unit as an acyl group. The sequence constitution of sugar chain could also
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8 267 be deduced as S_1 (shown in Fig. 5) by the $[M-H-C_5H_8O_4]^-$ ion at m/z 1071.53418,
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10 268 $[M-H-C_6H_{10}O_5]^-$ ion at m/z 1023.51263, $[M-H-C_5H_8O_4-C_6H_{12}O_6]^-$ ion at m/z 891.47162 and
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12 269 $[M-H-C_5H_8O_4-C_6H_{12}O_6-C_6H_{12}O_6-CO_2]^-$ ion at m/z 667.41888, corresponding to the
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14 270 cleavages of a pentose unit, two hexose unit and CO_2 . The molecular weight of aglycones
15
16 271 could be obtained by $[Aglycone+MBOH-H]^-$ ion at m/z 553.38806 ($C_{35}H_{53}O_5$), and the
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18 272 result ($C_{30}H_{44}O_3$; 452Da) indicated the formula of its aglycone was the same as
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20 273 Camelliasaponin B_1 ⁴⁸. Thus, its structure was tentatively inferred as shown in Fig. 5.

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25 274 Following the above procedure, 23 saponins were tentatively deduced, while 13
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27 275 saponins could not be completely inferred. Based on the formula of aglycones, the
28
29 276 triterpenoid saponins in *C. oleifera* could be classified into 9 different types (Fig 5). Type I:
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31 277 Compounds **17**, **24**, **25**, **29**, **30**, **35**, **36** and **37**, showed the same formula in aglycones as
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33 278 Camelliasaponin B_1 ($C_{30}H_{44}O_3$; 452Da)⁴⁸. Type II: Compounds **18**, **19**, **31**, **32** and **38**, gave
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35 279 the $[aglycone+acyl+H_2O-H]^-$ ion instead of $[aglycone+acyl-H]^-$ ion, and their aglycones
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37 280 resembled those of Camelliasaponin A_1 ($C_{30}H_{46}O_2$; 438Da)⁴⁹. Type III: Compounds **9** and
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39 281 **10**, which are desacyl-saponins, had the same aglycones as desacyl-assamsaponin F⁵⁰
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41 282 ($C_{30}H_{46}O_5$; 486Da). Type IV: Compounds **13** and **14** (desacyl-saponins), corresponded to
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43 283 desacyl-camelliasaponin B⁴⁸ ($C_{30}H_{46}O_4$; 470Da). Type V: Compounds **15** and **16**
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45 284 ($C_{30}H_{48}O_3$; 456 Da), whose behaviors were similar to type II, received a
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47 285 $[aglycone+H_2O-H]^-$ ion instead of $[aglycone-H]^-$ ion, and their aglycones were tentatively
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49 286 identified the same as desacyl-Camelliasaponin A⁴⁹. Type VI: Compound **11** ($C_{30}H_{44}O_4$;
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51 287 468Da), just like the aglycone of Assamsaponin F⁵⁰. Type VII: Compound **12**, was the
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53 288 isomer of Theasaponin A4⁵¹ (different in carbohydrate sequence) ($C_{30}H_{46}O_4$; 470Da),
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3 289 and can be distinguished from Type IV by the different fragmentation behaviors of normal
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5 290 saponins and desacyl-saponins. Type VIII: Compounds **20** and **21**, the desacyl-saponins,
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7 291 had the same aglycone as Camallidin II ⁵² (C₃₀H₄₆O₃; 454Da). Type IX: Other 13
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9 292 compounds that could not be calculated in their aglycones for the cleavage of acyl groups
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11 293 cannot be observed in those compounds, which was probably because of their special
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13 294 structure.
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18 295 Totally, 36 saponins were separated and characterized in the BuOH extract, 23 of which
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20 296 could completely inferred but 13 of them could not. All the saponins, except compound **37**,
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22 297 were identified as novel saponin. About half of them are the isomer of reported saponins,
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24 298 for they are different in structures. In addition, a new kind of glycosides (S₁) containing
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26 299 many compounds was found. S₁ consisted of the same monosaccharides with a known
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28 300 glycoside (S₄), but the sequence of sugar chain (S₁) was different. In MS² spectra, the
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30 301 compounds which had this kind of glycoside, such as compound **29**, showed high
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32 302 abundance of [M-H-C₅H₈O₄]⁻ ions, which could not be produced by S₄. Such fragment
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34 303 behavior could also be observed in TOF-MS ^{8,53}. Though this kind of glycoside (S₁) was
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36 304 not reported in genus *Camellia*, it was reported in other triterpenoid saponins, such as in
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38 305 *Antonia ovate* ⁵⁴.
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45 306 **4. Conclusion**

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48 307 In conclusion, the main bioactive compounds in the BuOH extract from the testa of
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50 308 *C.oleifera* Abel. seeds were first characterized by UHPLC-LTQ-orbitrap-MSⁿ. To the best
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52 309 of our knowledge, it was the first time to summarize the available fragmentation pathways
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54 310 of triterpenoid saponins and developed a strategy for characterizing the saponins in
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56 311 *C.oleifera*. In total, one flavonol-*O*-glycoside was unambiguously identified and 7
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3 312 flavonol-*O*-glycosides and 23 triterpene saponins were tentatively characterized, as well as
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5 313 13 saponins could not be completely inferred. In all these 44 compounds, 35 saponins were
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8 314 characterized as novel compounds and one flavonol-*O*-glycoside was first reported in genus
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10 315 *Camellia*.

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13 316 The UHPLC-LTQ-orbitrap-MSⁿ showed advantages in identification of flavonoids and
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16 317 saponins. The high accuracy mass measurements provided by orbitrap could help to
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18 318 propose the elemental composition for ions, and make it possible to distinguish the sugar
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20 319 residue ion and the [Aglycone-H] ion, while the MSⁿ spectra could be used to confirm the
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22 320 aglycones of flavonoid and find the possible fragmentation pathways of saponins with
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25 321 HR-MSⁿ.

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27
28 322 This work was helpful for the identification of other triterpenoid saponins in this genus
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30 323 by LC-MSⁿ for their similar structure, and provided essential data for further photochemical
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32 324 studies of *C. oleifera*. However, the UHPLC-LTQ-orbitrap-MSⁿ could be hardly used in
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34 325 distinguishing the linkage positions and stereoisomers. NMR data was necessary for
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36 326 completely structural characterization.

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References:

1. H. Zhong, D. R. Bedgood Jr., A. G. Bishop, P. D. Prenzler and K. Robards, *Food Chem*, 2007, **100**, 1544-1551.
2. D. Gao, M. Xu, P. Zhao, X. Zhang, Y. Wang, C. Yang and Y. Zhang, *Food Chem*, 2011, **124**, 432-436.
3. J. Chen, H. Wu, B. Liau, C. J. Chang, T. Jong and L. Wu, *Food Chem.*, 2010, **121**, 1246-1254.
4. L. C. Du, B. L. Wu and J. M. Chen, *Chinese Chem. Lett.*, 2008, **19**, 1315-1318.
5. J. Chen, B. Liau, T. Jong and C. J. Chang, *Se.p Purif. Technol.*, 2009, **67**, 31-37.
6. Y. Ye, Y. Guo and Y. Luo, *Int. J. mol. sci.*, 2012, **13**, 12401-12411.
7. J. Hu, S. Nie, D. Huang, C. Li, M. Xie and Y. Wan, *J. Sci. Food Agr.*, 2012, **92**, 2443-2449.
8. J. Hu, S. Nie, D. Huang, C. Li and M. Xie, *Int. J. Food Sci. Tech*, 2012, **47**, 1676-1687.
9. X. Zhang, Y. Han, G. Bao, T. Ling, L. Zhang, L. Gao and T. Xia, *Molecules*, 2012, **17**, 11721-11728.
10. H. Zhou, C. Z. Wang, J. Z. Ye and H. X. Chen, *Phytochem. Lett.*, 2014, **8**, 46-51.
11. S. Sugimoto, G. Chi, Y. Kato, S. Nakamura, H. Matsuda and M. Yoshikawa, *Chem. Pharm. Bull.*, 2009, **57**, 269-275.
12. P. Zhao, D. Gao, M. Xu, Z. Shi, D. Wang, C. Yang and Y. Zhang, *Chem. Biodivers.*, 2011, **8**, 1931-1942.
13. S. Wang, L. Chen, J. Leng, P. Chen, X. Fan and Y. Cheng, *J. Pharmaceut. Biomed.*, 2014, **98**, 22-35.
14. Y. Gao and S. A. McLuckey, *Rapid Commun. mass Sp.*, 2013, **27**, 249-257.
15. Y. Gao and S. A. McLuckey, *J. Mass Spectrom.*, 2012, **47**, 364-369.
16. Y. Gao, J. Yang, M. T. Cancilla, F. Meng and S. A. McLuckey, *Anal. Chem.*, 2013, **85**, 4713-4720.
17. I. K. Webb, Y. Gao, F. A. Londry and S. A. McLuckey, *J. Mass Spectrom*, 2013, **48**, 1059-1065.
18. G. Zheng, P. Zhou, H. Yang, Y. Li, P. Li and E. H. Liu, *Food Chem*, 2013, **136**, 604-611.
19. X. Chen, L. Hu, X. Su, L. Kong, M. Ye and H. Zou, *J Pharmaceut Biomed*, 2006, **40**, 559-570.
20. F. Cuyckens and M. Claeys, *J Mass Spectrom*, 2004, **39**, 1-15.
21. G. B. Gonzales, K. Raes, S. Coelus, K. Struijs, G. Smaghe and J. Van Camp, *J Chromatogr A*, 2014, **1323**, 39-48.
22. L. Qi, H. Wang, H. Zhang, C. Wang, P. Li and C. Yuan, *J Chromatogr A*, 2012, **1230**, 93-99.
23. S. Ji, Q. Wang, X. Qiao, H. Guo, Y. Yang, T. Bo, C. Xiang, D. Guo and M. Ye, *J Pharmaceut Biomed*, 2014, **90**, 15-26.
24. M. Cui, W. Sun, F. Song, Z. Liu and S. Liu, *Rapid Commun Mass Sp*, 1999, **13**, 873-879.
25. B. Li, Z. Abliz, G. Fu, M. Tang and S. Yu, *Rapid Commun Mass Sp*, 2005, **19**, 381-390.
26. B. Li, Z. Abliz, M. Tang, G. Fu and S. Yu, *J Chromatogr A*, 2006, **1101**, 53-62.
27. S. Wang, P. Chen, Y. Xu, X. Li and X. Fan, *J. Sep. Sci.*, 2014, **37**, 1748-1761.
28. M. Stobiecki, P. Kachlicki, A. Wojakowska and L. Marczak, *Phytochem. Lett.*, 2015, **11**, 358-367.

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57
58
59
60
29. X. Qiu, J. Zhang, Z. Huang, D. Zhu and W. Xu, *J. Chromatogr. A*, 2013, **1292**, 121-131.
30. W. Songsong, X. Haiyu, M. Yan, W. Xuguang, S. Yang, H. Bin, T. Shihuan, Z. Yi, L. Defeng, L. Rixin and Y. Hongjun, *J. Pharmaceut. Biomed.*, 2015, **111**, 104-118.
31. Z. Lin, X. Liu, F. Yang and Y. Yu, *Int. J. Mass Spectrom.*, 2012, **328-329**, 43-66.
32. L. Han, G. Pan, Y. Wang, X. Song, X. Gao, B. Ma and L. Kang, *J. Pharmaceut. Biomed.*, 2011, **55**, 996-1009.
33. C. T. Scoparo, L. M. de Souza, N. Dartora, G. L. Sasaki, P. A. J. Gorin and M. Iacomini, *J. Chromatogr. A*, 2012, **1222**, 29-37.
34. F. Ferreres, D. Gomes, P. Valentão, R. Gonçalves, R. Pio, E. A. Chagas, R. M. Seabra and P. B. Andrade, *Food Chem.*, 2009, **114**, 1019-1027.
35. E. Hvattum and D. Ekeberg, *J Mass Spectrom.*, 2003, **38**, 43-49.
36. F. Cuyckens and M. Claeys, *J Mass Spectrom.*, 2005, **40**, 364-372.
37. R. E. March and X. Miao, *Int J Mass Spectrom.*, 2004, **231**, 157-167.
38. Y. L. Ma, Q. M. Li, H. Van den Heuvel and M. Claeys, *Rapid Commun. Mass Sp.*, 1997, **11**, 1357-1364.
39. F. Cuyckens, R. Rozenberg, E. de Hoffmann and M. Claeys, *J. Mass Spectrom.*, 2001, **36**, 1203-1210.
40. F. Ferreres, R. Llorach and A. Gil-Izquierdo, *J. Mass Spectrom.*, 2004, **39**, 312-321.
41. T. Sekine, Y. Arai, F. Ikegami, Y. Fujii, S. Shindo, T. Yanagisawa, Y. Ishida, S. Okonogi and I. Murakoshi, *Chem. Pharm. Bull.*, 1993, **41**, 1185-1187.
42. Y. Murai, R. Takahashi, F. R. Rodas, J. Kitajima and T. Iwashina, *Nat. Prod. Commun.*, 2013, **8**, 453-456.
43. H. Matsuda, M. Hamao, S. Nakamura, H. Kon'I, M. Murata and M. Yoshikawa, *Chem. Pharm. Bull.*, 2012, **60**, 674-680.
44. K. Fujimoto, S. Nakamura, S. Nakashima, T. Matsumoto, K. Uno, T. Ohta, T. Miura, H. Matsuda and M. Yoshikawa, *Chem. Pharm. Bull.*, 2012, **60**, 1188-1194.
45. N. Li, Z. Ma, Y. Chu, Y. Wang and X. Li, *Fitoterapia*, 2013, **84**, 321-325.
46. M. Myose, T. Warashina and T. Miyase, *Chem. Pharm. Bull.*, 2012, **60**, 612-623.
47. M. Yoshikawa, E. Harada, T. Murakami, H. Matsuda, J. Yamahara and N. Murakami, *Chem. Pharm. Bull.*, 1994, **42**, 742-744.
48. M. Yoshikawa, T. Murakami, S. Yoshizumi, N. Murakami, J. Yamahara and H. Matsuda, *Chem. Pharm. Bull.*, 1996, **44**, 1899-1907.
49. M. Yoshikawa, T. Murakami, S. Yoshizumi, N. Murakami, J. Yamahara and H. Matsuda, *Chem. Pharm. Bull.*, 1996, **44**, 1899-1907.
50. T. Murakami, J. Nakamura, T. Kageura, H. Matsuda and M. Yoshikawa, *Chem. Pharm. Bull.*, 2000, **48**, 1720-1725.
51. M. Yoshikawa, T. Morikawa, S. Nakamura, N. Li, X. Li and H. Matsuda, *Chem. Pharm. Bull.*, 2007, **55**, 57.
52. C. Nishino, S. Manabe, N. Enoki, T. Nagata, T. Tsushida and E. Hamaya, *J. Chem. Soc., Chem. Commun.*, 1986, 720-723.
53. X. Zhang, X. Zhu, Y. Liu, T. Xia, X. Du, L. Gao and D. Zhang, *Shipin Kexue (Beijing, China)*,

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54
55
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57
58
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60

2012, **33**, 7-11.

54. A. A. Magid, H. Bobichon, N. Borie, N. Lalun, C. Long, C. Moretti and C. Lavaud, *Phytochemistry*, 2010, **71**, 429-434.

55. B. Li and Y. Luo, *Chemical Journal on Internet*, 2003, **5**, No pp. given.

56. C. Ma, K. Dastmalchi, G. Flores, S. Wu, P. Pedraza-Peñalosa, C. Long and E. J. Kennelly, *J. Agr. Food Chem.*, 2013, **61**, 3548-3559.

57. M. P. Rodríguez-Rivera, E. Lugo-Cervantes, P. Winterhalter and G. Jerz, *Food Chem.*, 2014, **158**, 139-152.

58. H. Fu, C. Li, J. Yang, X. Chen and D. Zhang, *Phytochemistry*, 2013, **85**, 167-174.

59. T. Nagata, T. Tsushida, E. Hamaya, N. Enoki, S. Manabe and C. Nishino, *Agr. Bio. Chem.*, 1985, **49**, 1181-1186.

60. H. Matsuda, S. Nakamura, K. Fujimoto, R. Moriuchi, Y. Kimura, N. Ikoma, Y. Hata, O. Muraoka and M. Yoshikawa, *Chem. Pharm. Bull.*, 2010, **58**, 1617-1621.

Figure Captions

Fig.1 Base peak chromatograms of the following fractions: (A) BuOH extract in negative ion mode (T70Bu); (B) Partial enlarged drawing of BuOH extract in negative ion mode.

Fig. 2 The procedure to identify flavonol-O-glycosides. Glu: glucose; Gal: galactose; Rha: rhamnose; Xyl: xylose; Ac: acetyl.

Fig. 3 The procedure to identify saponins. NL: neutral loss; MB: methylbutyryl; MBOH: methylbutyryl-OH; Ang: angeloyl; Tig: tigeloyl; Ac: acetyl; Hex: hexenoyl; IsoV: isovaleryl.

Fig. 4 Proposed fragmentation pathway for compound **29**.

Fig. 5 Probable structure of triterpenoid saponins identified in T70Bu. S₁, S₂, S₃ and S₄ are the sugar residues in saponins.

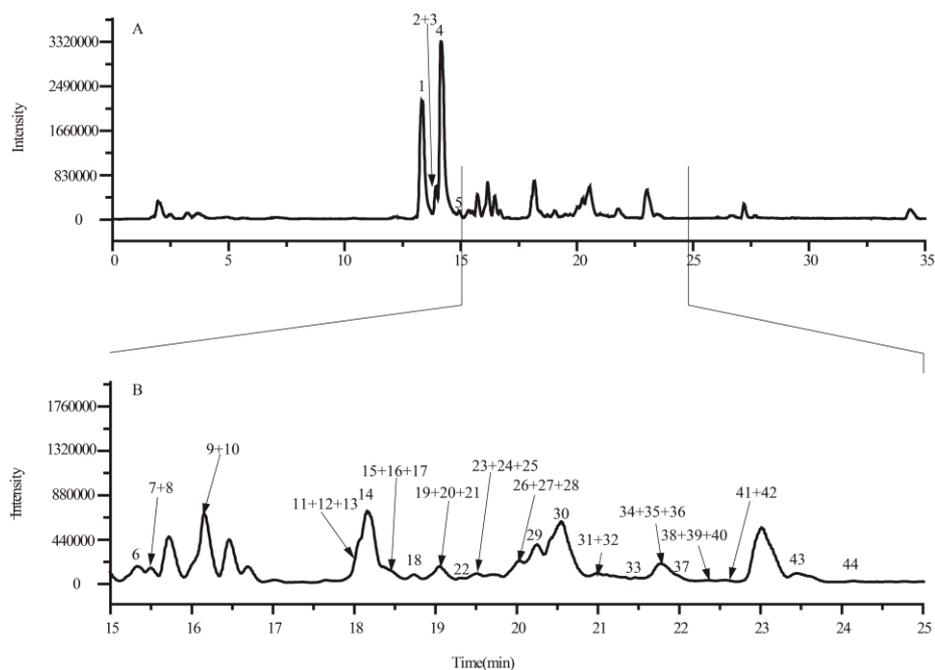


Fig.1 Base peak chromatograms of the following fractions: (A) BuOH extract in negative ion mode (T70Bu); (B) Partial enlarged drawing of BuOH extract in negative ion mode.

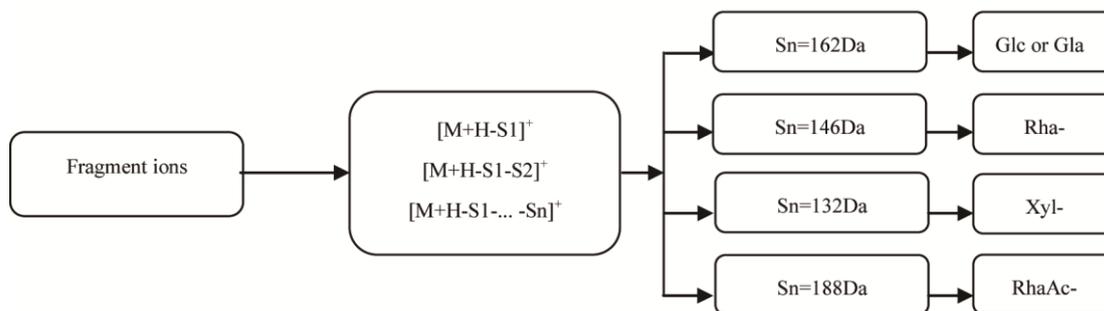


Fig. 2 The procedure to identify flavonol-O-glycosides. Glu: glucose; Gal: galactose;

Rha: rhamnose; Xyl: xylose; Ac: acetyl.

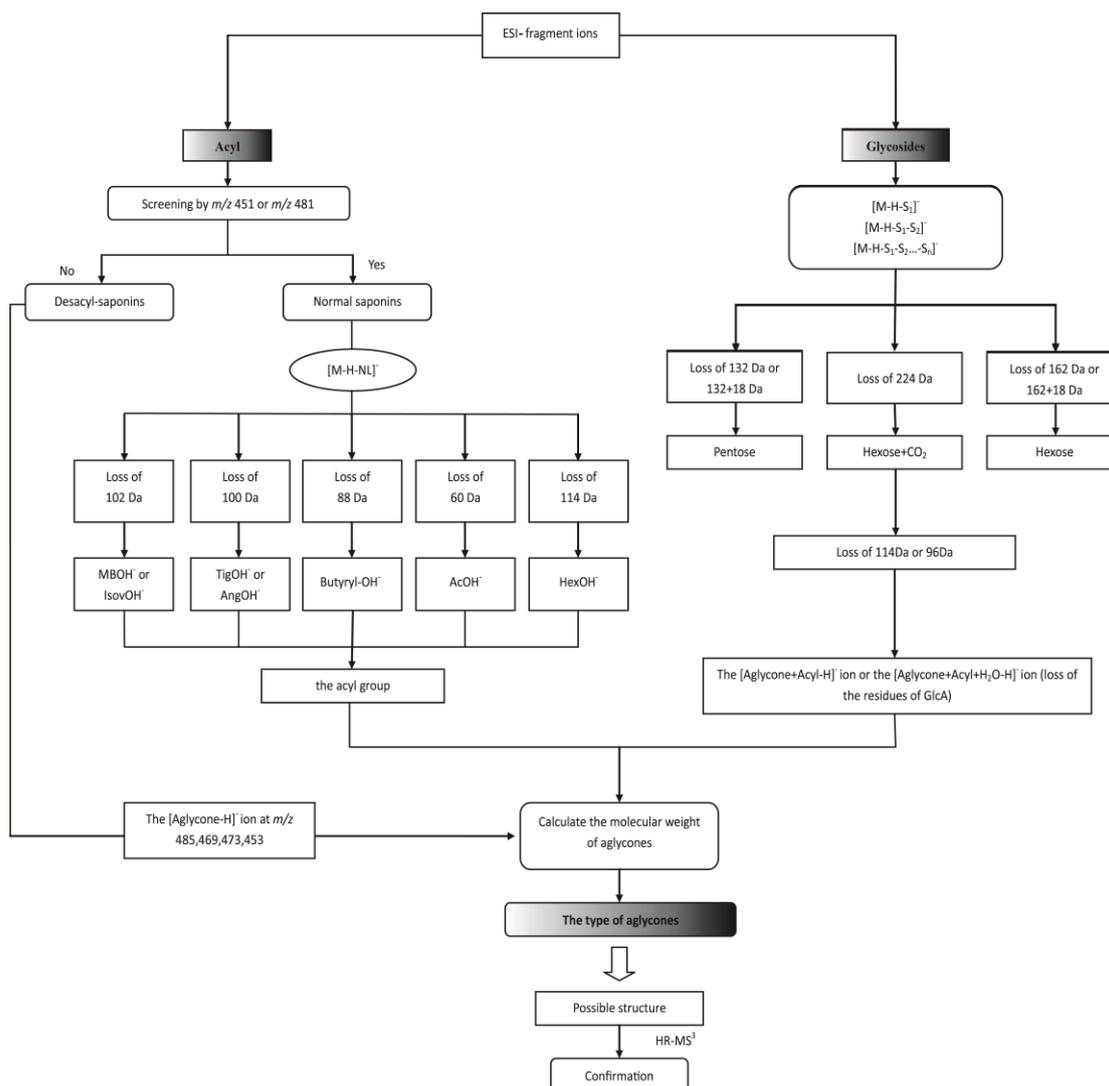


Fig. 3 The procedure to identify saponins. NL: neutral loss; MB: methylbutyryl; MBOH: methylbutyryl-OH; Ang: angeloyl; Tig: tigeloyl; Ac: acetyl; Hex: hexenoyl; Isov: isovaleryl.

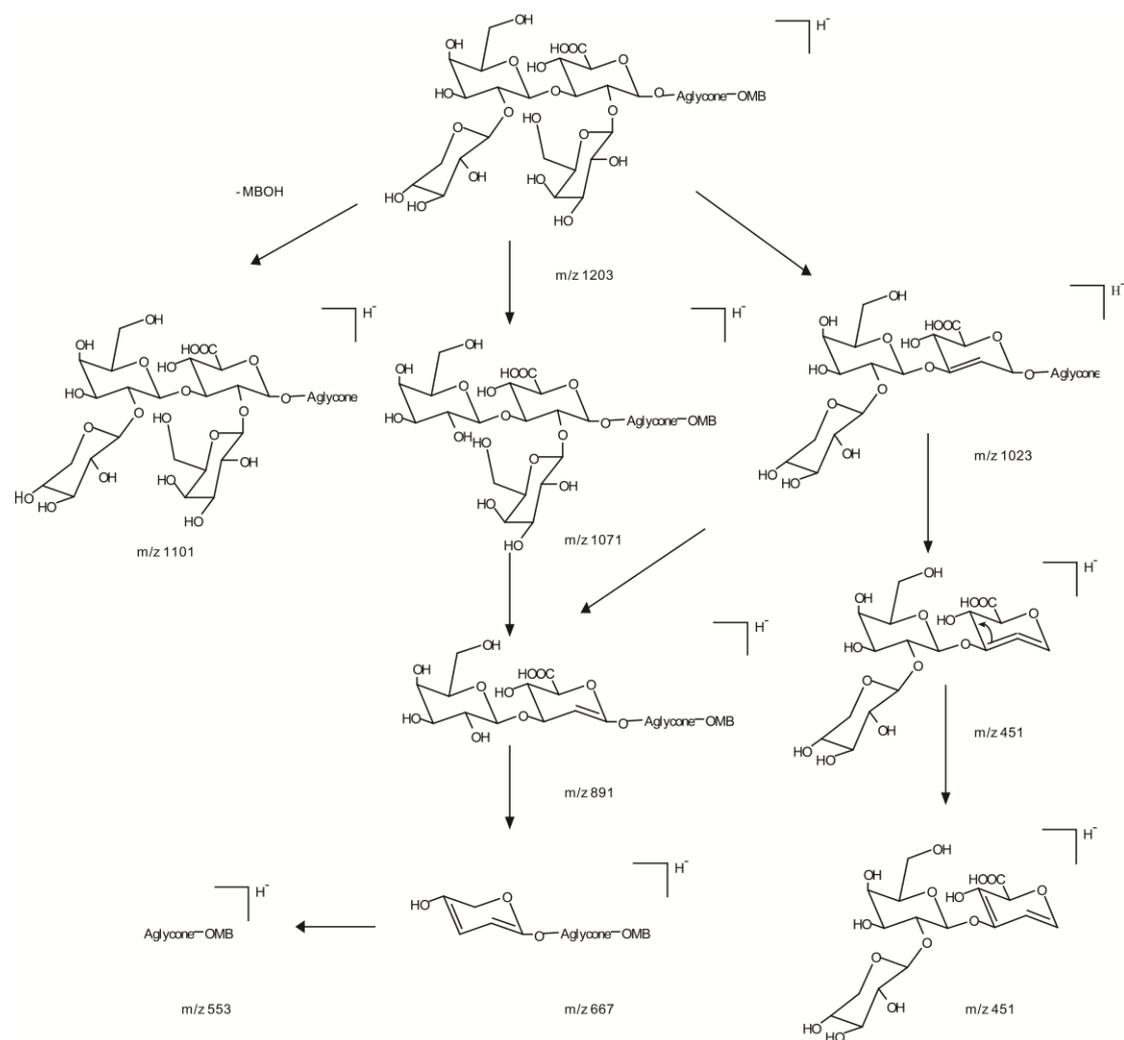


Fig. 4 Proposed fragmentation pathway for compound 29.

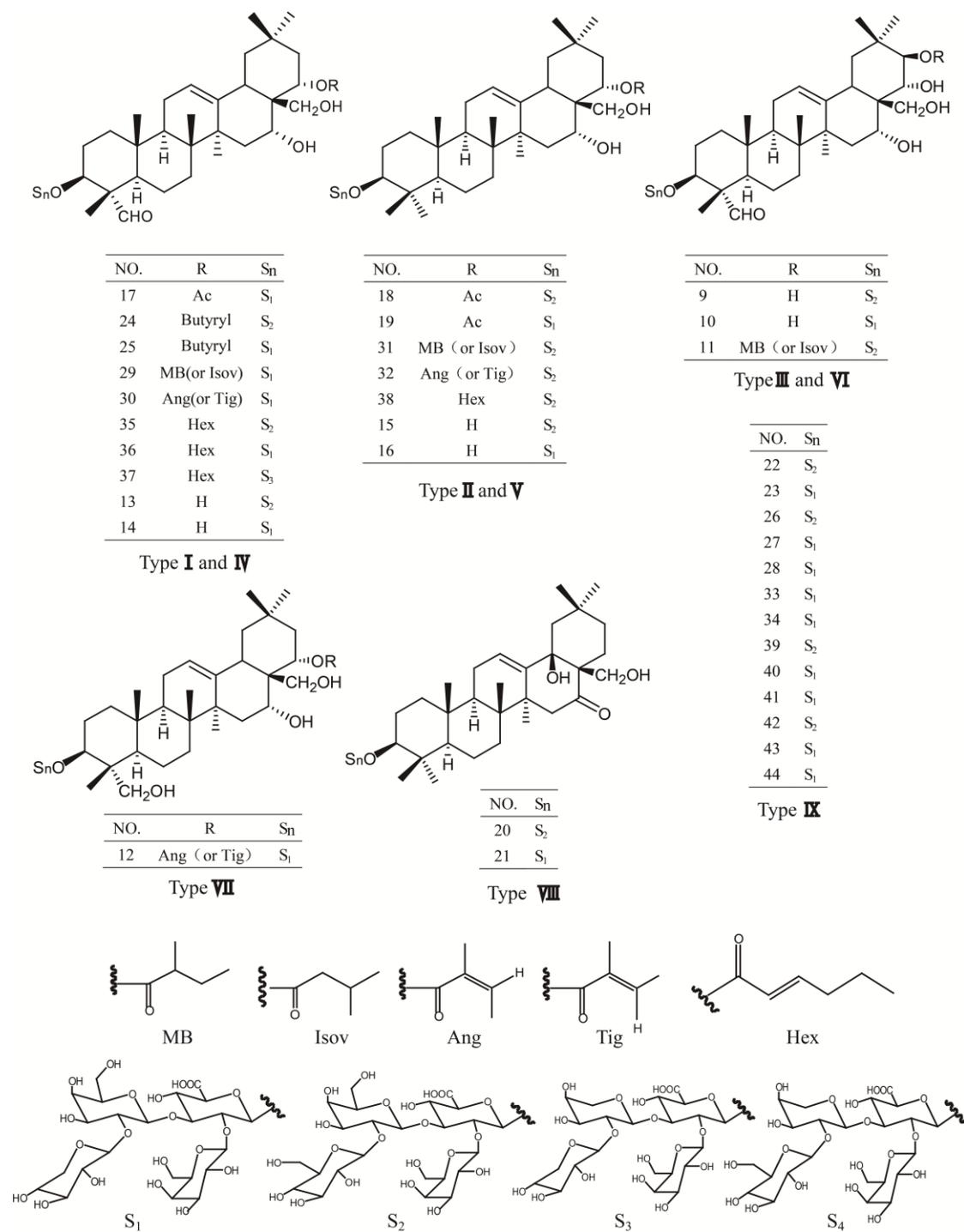


Fig. 5 Probable structure of triterpenoid saponins identified in BuOH extract. S₁, S₂,

S₃ and S₄ are the sugar residues in saponins.

Table 1 Characterization of flavonol-O-glycosides.

N o.	tR(+/-)	Identification	Formula	ESI-[M-H]-			ESI+[M+H]+			reference
				Mean measured mass	Major MS2 ions (m/z)%	Error (ppm)	Mean measured mass	Major MS2 ions (m/z)%	Error (ppm)	
1	13.44/ 13.31	camelliaside A ^a	C ₃₃ H ₄₀ O ₂₀	755.20292	593 (27),575 (100),285 (39)	1.36	757.21938	287 (100),449 (29),595 (28)	1.07	4, 33
2	13.98/ 13.88	kaempferol-3-O-sophoroside or camelliaside C	C ₂₇ H ₃₀ O ₁₆	609.14656	447 (13),429 (100),285 (99)	2.55	611.16066	287(100),449(39)	0.53	2, 41
3	14.03/ 13.94	Kaempferol-3-O- α -L-rhamnopyranosyl-(1 \rightarrow 4)-[α -L-rhamnopyranosyl-(1 \rightarrow 6)- β -D-galactopyranoside]. ^b	C ₃₃ H ₄₀ O ₁₉	739.20959	593 (23),575 (100),285 (27),284 (39)	2.15	741.22437	287(100),449(10),595(32)	0.97	42
4	14.28/ 14.15	camelliaside B	C ₃₂ H ₃₈ O ₁₉	725.19236	593(28),575(100),285(39)	1.22	727.20801	287(100),449(20),595(25)	0.90	4, 55
5	14.73/ 14.63	leucoside	C ₂₆ H ₂₉ O ₁₅	579.13445	447(28),429(86),285(88),284(100)	2.21	581.15079	287(100),449(29)	1.19	2, 34
6	15.03/ 14.93	kaempferol-3-O-[4''''-O-acetyl- α -L-rhamnopyranosyl-(1 \rightarrow 6)]-[β -D-glucopyranosyl-(1 \rightarrow 2)]- β -D-glucopyranoside	C ₃₅ H ₄₂ O ₂₁	797.21477	755(100),617(31),285(16)	1.61	799.22979	287(100),449(12),637(15)	0.82	2
7	15.42/ 15.32	kaempferol-3-O-rutinoside	C ₂₇ H ₃₁ O ₁₅	593.15131	285(100)	2.04	595.16629	287(100),449(19)	0.92	2, 56, 57
8	15.59/ 15.50	kaempferol-3-O-[4''''-O-acetyl- α -L-rhamnopyranosyl-(1 \rightarrow 6)]-[β -D-xylopyranosyl-(1 \rightarrow 2)]- β -D-glucopyranoside	C ₃₄ H ₄₁ O ₂₀	767.20411	725(100),617(20),285(12)	1.56	769.21937	287(100),449(22),581(25),637(16)	1.04	2

a Compared with standard compounds

b Found in this genus for the first time.

Table 2 Characterization of triterpenoid saponins.

No.	tR(-)	Identification	Formula	ESI-			reference
				Mean measured mass[M-H]-	Major MS2 ions (m/z)%	Error (ppm)	
9	16.03	novel saponin	C ₅₄ H ₈₆ O ₂₇	1165.52825	1103 (46), 1003 (46), 985 (100), 823 (16), 599 (12), 485 (12)	0.84	
10	16.14	Isomer of desacyl-assamsaponin F	C ₅₃ H ₈₄ O ₂₆	1135.51769	1003 (46), 955 (100), 823 (22), 599 (12), 485 (11)	0.87	50
11	17.94	novel saponin	C ₅₉ H ₉₄ O ₂₈	1249.58608	1147 (23), 1087 (50), 1069 (100), 907 (19), 683 (8), 569 (8), 481 (2)	1.03	
12	18.03	Isomer of Theasaponin A4	C ₅₈ H ₉₂ O ₂₇	1219.57522	1119 (11), 1087 (16), 1039 (100), 907 (21), 683 (8), 587 (7), 569 (4), 451 (2)	0.81	51
13	18.07	novel saponin	C ₅₄ H ₈₆ O ₂₆	1149.53381	987 (46), 969 (100), 807 (18), 583 (11), 469 (15)	1.26	
14	18.15	Isomer of desacyl-camelliasaponin B	C ₅₃ H ₈₄ O ₂₅	1119.52192	987 (46), 939 (100), 807 (23), 583 (11), 469 (14)	0.11	47
15	18.35	novel saponin	C ₅₄ H ₈₈ O ₂₅	1135.55450	973 (52), 955 (100), 793 (14), 569 (8), 473 (6)	1.24	
16	18.43	Isomer of desacyl- Gordonsaponin F	C ₅₃ H ₈₆ O ₂₄	1105.54358	973 (53), 925 (100), 793 (15), 569 (7), 473 (6)	0.95	58
17	18.45	novel saponin	C ₅₅ H ₈₆ O ₂₆	1161.53236	1101 (33), 1029 (41), 981 (100), 849 (17), 625 (8), 511 (7), 451 (1)	0.50	
18	18.73	novel saponin	C ₅₆ H ₉₀ O ₂₆	1177.56398	1117 (30), 1015 (52), 997 (100), 835 (15), 611 (7), 515 (4), 481 (4)	0.27	
19	18.83	Isomer of Gordonsaponin F	C ₅₅ H ₈₈ O ₂₅	1147.55348	1087 (20), 1015 (33), 967 (100), 835 (22), 611 (8), 515 (5), 451 (4)	0.27	58
20	18.95	novel saponin	C ₅₄ H ₈₆ O ₂₅	1133.53761	953 (100), 791 (14), 567 (10), 453 (13)	0.14	

Table 2 Continued

No.	tR(-)	Identification	Formula	ESI-			reference
				Mean measured mass[M-H]-	Major MS ² ions (m/z)%	Error (ppm)	
21	19.03	isomer of Camallidin II	C ₅₃ H ₈₄ O ₂₄	1103.52722	971 (50), 923 (100), 791(19),567(10),453 (14)	0.31	52, 59
22	19.23	novel saponin	C ₆₀ H ₉₆ O ₂₉	1279.59535	1243 (5), 1191 (4), 1131 (3), 1117 (50), 1099 (100), 937 (15), 713 (4), 585 (19), 481(4)	1.60	
23	19.36	novel saponin	C ₅₉ H ₉₄ O ₂₈	1249.58338	1117 (52), 1069 (100), 937 (18), 713 (8), 617 (12), 451 (4)	1.12	
24	19.48	Isomer of Theasaponin A4	C ₅₈ H ₉₂ O ₂₇	1219.57397	1131 (45), 1057 (45), 1039 (100), 877 (17), 653 (7), 539 (5), 481 (6)	0.20	51
25	19.50	Isomer of Theasaponin A1	C ₅₇ H ₉₀ O ₂₆	1189.56396	1101 (30), 1057 (47), 1039 (6), 1009 (100), 877 (20), 653 (10), 539 (7), 451 (2)	0.26	26
26	19.95	novel saponin	C ₆₁ H ₉₆ O ₂₉	1291.59361	1129 (42), 1111 (100), 949 (16), 725 (7), 597 (9), 481 (3)	1.35	
27	20.04	Isomer of TheasaponinA5	C ₆₀ H ₉₄ O ₂₈	1261.58366	1129 (42), 1081 (100), 949 (20), 725 (9), 629 (6), 597(9), 451 (2)	1.13	51
28	20.07	novel saponin	C ₆₀ H ₉₆ O ₂₈	1263.59705	1161 (6), 1131 (46), 1083 (100), 951 (23), 727 (5), 631 (11), 599 (18), 451 (2)	2.68	
29	20.48	Isomer of CamelliasaponinC1	C ₅₈ H ₉₂ O ₂₆	1203.57786	1101 (8), 1071 (43), 1023 (100), 891 (21), 667 (8), 553 (6), 451(5)	1.20	48
30	20.56	Isomer of CamelliasaponinB1	C ₅₈ H ₉₀ O ₂₆	1201.56368	1101 (18), 1069 (49), 1021 (100), 889 (19), 665 (11), 551 (8), 451 (3)	0.02	48
31	20.92	novel saponin	C ₅₉ H ₉₆ O ₂₆	1219.60913	1117 (17), 1057 (59), 1039 (100), 895 (12), 877 (11), 653 (2), 557 (5), 481 (8)	1.21	
32	20.98	Isomer of Sasanquasaponins II	C ₅₉ H ₉₄ O ₂₆	1217.59530	1117 (13), 1037 (100), 875 (13), 651 (7), 555 (4), 481 (3)	0.28	60

Table 2 Continued

No.	tR(-)	Identification	Formula	ESI-			reference
				Mean measured mass[M-H]-	Major MS ² ions (m/z)%	Error (ppm)	
33	21.44	novel saponin	C ₅₅ H ₉₄ O ₃₁	1249.56348	1135 (1), 1117 (39), 1069 (100), 937 (19), 713 (9), 599(3), 451 (3)	4.85	
34	21.51	novel saponin	C ₅₉ H ₉₂ O ₂₈	1247.56934	1115 (44), 1067 (100), 935 (16), 711 (9), 597 (6), 451 (3)	0.17	
35	21.65	novel saponin	C ₆₀ H ₉₄ O ₂₇	1245.59045	1131 (19), 1083 (46), 1065 (100), 921 (17), 903 (20), 679 (14), 565 (16), 481 (5)	0.47	
36	21.77	Isomer of Oleifera A	C ₅₉ H ₉₂ O ₂₆	1215.57931	1101 (20), 1083 (48), 1035 (100), 903 (19), 679 (13), 565(9), 451 (3)	1.04	9
37	22.12	TeaseedsaponinG or its isomer	C ₅₈ H ₉₀ O ₂₅	1185.56874	1071 (29), 1053 (56), 1005 (100), 903 (12), 679 (12), 565 (6), 451 (2)	0.80	46
38	22.27	Isomer of sasanquasaponin I	C ₆₀ H ₉₆ O ₂₆	1231.57422	1117 (9), 1069 (39), 1051 (100), 907 (15), 889 (13), 665 (6), 569 (4), 481 (6)	0.88	60
39	22.32	Isomer of Teaseedsaponin A	C ₆₂ H ₉₆ O ₂₈	1287.60099	1201 (2), 1155 (58), 1107 (100), 975 (19), 751 (16), 637 (12), 451 (6)	0.43	46
40	22.35	novel saponin	C ₆₂ H ₉₈ O ₂₈	1289.61523	1203 (3), 1157 (50), 1109 (100), 977 (25), 753 (38), 639 (4), 451 (2)	0.66	
41	22.55	novel saponin	C ₆₃ H ₉₆ O ₂₈	1299.60114	1167 (53), 1119 (100), 987 (22), 763 (13), 649 (10), 451 (2)	0.54	
42	22.79	novel saponin	C ₆₄ H ₉₈ O ₂₉	1329.61200	1167 (57), 1149 (100), 1005 (20), 987 (10), 763 (10), 649 (7), 481 (5)	0.75	
43	23.45	Isomer of TeaseedsaponinD	C ₆₃ H ₉₈ O ₂₈	1301.61589	1169 (51), 1121 (100), 989 (22), 765 (12), 651 (9), 451 (2)	0.15	46
44	24.14	novel saponin	C ₆₇ H ₉₆ O ₂₉	1347.60107	1215 (43) ,1185 (12), 1167 (100), 1035 (23), 811 (9), 697 (5), 451 (1)	0.47	