

Analyst

Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this *Accepted Manuscript* with the edited and formatted *Advance Article* as soon as it is available.

You can find more information about *Accepted Manuscripts* in the [Information for Authors](#).

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard [Terms & Conditions](#) and the [Ethical guidelines](#) still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.

1
2
3 **Label-free Detection of Missense Mutations and Methylation Differences in the p53 Gene**
4
5
6 **Using Optically Diffracting Hydrogels**
7

8
9 Kelsey I. MacConaghy, Duncan M. Chadly, Mark P. Stoykovich*, and Joel L. Kaar*

10
11 Department of Chemical and Biological Engineering, University of Colorado, Boulder, CO
12 80309, USA
13

14
15
16 **Corresponding Author:*
17

18
19 Joel L. Kaar
20 University of Colorado Boulder
21 Department of Chemical and Biological Engineering
22 Campus Box 596
23 Boulder, CO 80309
24 Tel: (303) 492-6031
25 Fax: (303) 492-4341
26 Email: joel.kaar@colorado.edu
27
28
29
30

31
32 Mark P. Stoykovich
33 University of Colorado Boulder
34 Department of Chemical and Biological Engineering
35 Campus Box 596
36 Boulder, CO 80309
37 Tel: (303) 492-6522
38 Fax: (303) 492-4341
39 Email: mark.stoykovich@colorado.edu
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

ABSTRACT

We have developed a novel approach for DNA detection as well as genetic screening of mutations by uniquely combining DNA-responsive and optically diffracting materials. This approach entails the polymerization of a photonic crystal within a hydrogel network that alters the diffraction of light in response to a target DNA strand. The utility of this approach, which permits label-free sensing, was demonstrated via the detection of a target sequence from the DNA binding domain of the major tumor suppressor protein p53. Using a complementary capture probe strand, we were able to detect down to picomole concentrations of the target p53 sequence. Moreover, we demonstrated that this approach could readily detect a single base pair mutation in the target strand, which corresponds to the hotspot cancer mutation R175H in p53. The sensitivity of detection was increased by lowering the rate of annealing of the target strand and adjusting the solution ionic strength during optical characterization. Changes in ionic strength during characterization impact the melting temperature of the bound target DNA and the Donnan potential between the hydrogel and solution, which influence detection. We further showed that this approach is sensitive to epigenetic changes via the detection of a fully methylated form of the target p53 sequence. Ultimately, this approach represents a new paradigm for DNA detection and specifically genetic screening of p53 as well as other disease markers and nucleotide modifications that alter the properties of DNA (e.g., epigenetic alterations and adducts with chemical carcinogens).

INTRODUCTION

Due to the importance of DNA detection in a myriad of fields, including genetic screening, forensics, pathogen identification, and biotechnology (i.e., genome engineering), the development of new technologies for DNA sensing is critical. Of particular interest is the development of new approaches that accelerate DNA detection with high fidelity and reduce the cost of traditional DNA sequencing. In addition to traditional DNA sequencing (e.g., capillary electrophoresis), which, although precise, requires specialized instrumentation, other methods of detecting DNA include electrochemical impedance spectroscopy,¹⁻³ surface enhanced Raman spectroscopy,⁴ nanoparticle aggregation assays,⁵⁻⁷ analysis by quartz crystal microbalance,⁸⁻⁹ surface plasmon resonance,¹⁰⁻¹² and scanning tunneling spectroscopy.¹³⁻¹⁵ Additionally, fluorescent¹⁶⁻¹⁷ and chemiluminescent¹⁸⁻¹⁹ based techniques for DNA sensing have been reported. However, many of these techniques, as with traditional sequencing, require highly specialized instrumentation as well as exogenous labels or reagents and, moreover, are unable to detect down to single nucleotide changes.²⁰⁻²¹

A novel approach for the detection of DNA, which may overcome many of the limitations of current sensing methods, entails combining DNA-responsive and optically diffracting materials. In one such approach, a photonic crystal may be polymerized within a hydrogel matrix that can swell or contract in response to the presence of an analyte. The hydrogel matrix can be rationally tuned to change volume in response to a specific analyte by tethering a receptor for the analyte within the hydrogel along with the photonic crystal.²²⁻²⁵ Depending on the properties of the analyte (i.e., charge or hydrophobic character), receptor binding may trigger a volume change of the hydrogel by creating a Donnan potential or altering the interaction of the hydrogel with water. Changes in hydrogel volume are accompanied by an alteration in the lattice spacing of the photonic crystal that may be readily measured by

1
2
3 reflectance spectroscopy or, if large enough, visually through changes in coloration of the
4 hydrogel. We and others have previously exploited this approach to develop sensors for a broad
5 spectrum of small molecule analytes,²⁶⁻²⁸ metals,²⁹⁻³¹ changes in solution conditions,³²⁻³³ and,
6 more recently, protein kinase activity.³⁴⁻³⁵ Notably, in all cases, this approach enabled the
7 detection of environmental cues in the absence of exogenous reagents, using changes in optical
8 properties of the hydrogel as the primary readout.
9

10
11 In this work, we explored the utility of this approach as a sensing platform for label-free
12 DNA detection via encapsulation of a crystalline colloidal array (CCA) within a DNA-
13 responsive hydrogel. Specifically, we fabricated DNA-responsive hydrogel films that alter the
14 diffraction of light upon hybridization of a specific “target” DNA strand to a capture “probe”
15 sequence (**Figure 1**). We reasoned that hybridization of the target strand would cause a change in
16 the volume of the hydrogel network by increasing the concentration of immobilized negative
17 charges within the hydrogel. The addition of negative charges upon hybridization is due to the
18 backbone of the target DNA being comprised of negatively-charged phosphate groups. Once
19 these charges become immobilized within the hydrogel network, the Donnan potential between
20 the hydrogel and surrounding solution is modified, resulting in swelling of the hydrogel and
21 ultimately a change in the diffraction spectrum of the encapsulated CCA.
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42

43 The utility of this approach for DNA sensing was demonstrated via the detection of the
44 gene for the major tumor suppressor protein p53. A key transcription factor involved in cell
45 regulation, p53, which is inactivated in virtually all human cancers, is of specific interest as a
46 marker for early cancer detection.³⁶⁻⁴¹ For sensing, a short 18-mer sequence that is
47 complementary to the DNA-binding domain of the p53 gene was used as the capture probe and
48 was conjugated to the hydrogel network (**Figure 1**). The perfect matching 18-mer sequence from
49
50
51
52
53
54
55
56
57
58
59
60

the wild-type p53 gene was used as the target strand. Furthermore, we were interested in determining if the sensing approach was sensitive to mutations in p53, which is the most frequently mutated gene in cancer. To determine the sensitivity of the optical response of the detection approach to DNA mutations, a single base in the target strand was changed (G→A). This genetic alteration corresponds to mutation of arginine at position 175 to a histidine (i.e., R175H), which is one of the more frequent oncogenic mutations in p53. Finally, we tested if the detection scheme was also sensitive to DNA methylation by using a methylated form of the wild-type target strand (**Table 1**). We show that changes in methylation can also be readily detected, which may have additional implications in screening for epigenetic disease markers.

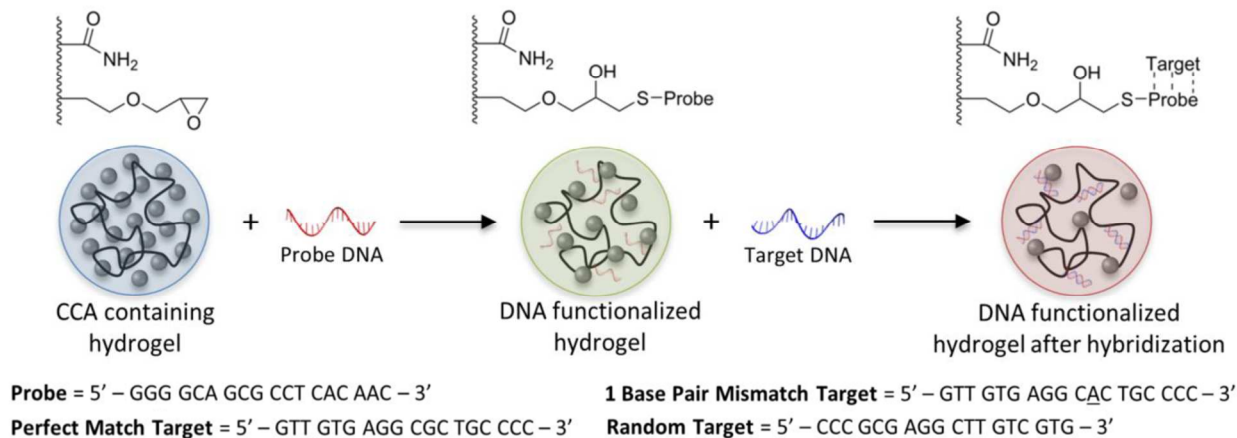


Figure 1. Schematic of hydrogel functionalization with “probe” DNA and subsequent hybridization of “target” DNA strands. Color changes in the optically diffracting hydrogel are representative of those observed upon functionalization and hybridization due to changes in the lattice spacing of the encapsulated CCA. The sequences of the probe and target DNA strands that were used are shown below the schematic.

EXPERIMENTAL METHODS

Materials

Acrylamide (AA), *N,N'*-methylenebis(acrylamide) (BA), and allyl glycidyl ether (AGE) monomers were purchased from Sigma-Aldrich (St. Louis, MO) and used without further purification. DNA oligos (Table 1) were purchased and used as received from Integrated DNA Technologies (Coralville, IA). The crosslinker SPDP-PEG4-NHS was purchased and used without further purification from Conju-Probe (San Diego, CA).

Table 1. Names and sequences of DNA oligos

Name	Sequence
Probe	5' – GGG GCA GCG CCT CAC AAC – 3'
Perfect Match (PM) Target	5' – GTT GTG AGG CGC TGC CCC – 3'
1 Base Pair Mismatch (1bpMM) Target	5' – GTT GTG AGG <u>C</u> AC TGC CCC – 3'
Random Target	5' – CCC GCG AGG CTT GTC GTG – 3'
Methylated Perfect Match (mPM)Target	5' – GTT GTG AGG <u>mCGmC</u> TG <u>mC</u> <u>mCmCmC</u> – 3'

Synthesis of Polystyrene Nanospheres

Highly-charged polystyrene (PS) nanospheres were synthesized via emulsion polymerization as previously described.⁴²⁻⁴³ The PS nanospheres used in all experiments had a concentration of 12 wt% in water and were 87 nm in diameter with a polydispersity of 4.9%, as determined by dynamic light scattering (Titan DynaPro with Dyna V6.3.4 software package, Wyatt Technology, Inc.; Santa Barbara, CA). Particles were stored at room temperature over a BioRad (Hercules, CA) AG501-X8 mixed bed resin.

Hydrogel Polymerization and Crystalline Colloidal Array Formation

Hydrogels were photopolymerized by solubilizing 35 mg (0.98 M) AA in 480 μ L of the CCA. To this mixture, a solution of 1 mg (0.015 M) BA and 24 mg (0.42 M) AGE in 20 μ L DMSO was added. The photoinitiator Irgacure 2959 (BASF; Florham Park, NJ) (10 wt% in

1
2
3 DMSO) was added at a final concentration of 0.05 wt% to the CCA-monomer solution. The
4
5 solution was then pipetted into a mold formed by two glass slides separated by a $273 \pm 2 \mu\text{m}$
6
7 Parafilm spacer. Samples were flood exposed with 365 nm light at 15 mW/cm^2 from a UV
8
9 mercury lamp for 45 min. Films were subsequently equilibrated and stored in ultrapure water for
10
11 a minimum of 24 h prior to DNA functionalization.
12
13

14 15 16 17 *Hydrogel Functionalization with DNA Probe* 18

19
20 DNA probe was solubilized in 100 mM NaPO_4 buffer, pH 8, at a concentration of 5 mM.
21
22 A fifteen-fold molar excess of the SPDP-PEG4-NHS linker relative to the DNA probe was
23
24 solubilized in 10 μL DMSO and added to the DNA solution. The solution was incubated at 4°C
25
26 and reacted for 2 h. The solution was then desalted to remove excess linker after which 100 mM
27
28 tris(2-carboxyethyl)phosphine (TCEP) was added and reacted at room temperature for 1 h to
29
30 reduce the linker disulfide bond. To determine the amount of linker containing DNA, the
31
32 absorbance of the cleaved pyridine-2-thione was measured at 343 nm. The concentration of the
33
34 reduced linker was calculated using the pyridine-2-thione extinction coefficient of $8080 \text{ M}^{-1} \text{ cm}^{-1}$.
35
36^{1,44} After determining the linker concentration, a final desalt was performed to remove the
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
cleaved pyridine-2-thione group.

52
53
54
55
56
57
58
59
60
Prior to functionalization, hydrogels were equilibrated in 100 mM NaPO_4 buffer, pH 8.
To each sample, 25 μL of 75 μM linker-modified DNA probe in 100 mM NaPO_4 buffer was
added and reacted at room temperature for 36 h. Post-functionalization, the samples were
thoroughly rinsed and stored in 100 mM NaCl , pH 6.

DNA Hybridization

Samples to be annealed were equilibrated in 100 mM NaCl solution, pH 6, prior to the addition of perfect match (PM), single base pair mismatch (1bpMM), random DNA, or methylated perfect match (mPM) target DNA. Target DNA was added to the hydrogels at concentrations ranging from 0.5 to 1000 μM and annealed by heating the hydrogel samples to 85 $^{\circ}\text{C}$, holding that temperature for 30 min, and ramping from 85 to 55 $^{\circ}\text{C}$ at rates ranging from 0.05 to 1 $^{\circ}\text{C}/\text{min}$. Once the temperature reached 55 $^{\circ}\text{C}$, the system temperature was no longer controlled and samples were permitted to naturally cool to room temperature.

Measurement of DNA Melting Point

The melting point for both the PM and 1bpMM samples was determined by first annealing samples with 50 μM PM target or 500 μM 1bpMM target in 100 mM NaCl followed by equilibration in 2.5 mM NaCl solution. Melting points were subsequently measured by submerging the hydrogels a in 2.5 mM NaCl solution and increasing the solution temperature from room temperature to 65 $^{\circ}\text{C}$ at a rate of 0.25 $^{\circ}\text{C}/\text{min}$.

Optical Diffraction Measurements

The optical response of the equilibrated hydrogel-encapsulated CCA biosensors was measured with an Ocean Optics (Dunedin, FL) USB-4000 fiber-optic spectrophotometer operated in reflectance mode set to an angle of incidence of 15 $^{\circ}$ from the sample surface normal. This optical setup allowed for characterization of the peak diffraction wavelength *in situ* and in real-time, enabling measurements as a function of temperature (*e.g.*, to generate melting curves). In all experiments, the hydrogels were initially rinsed with 100 mM NaCl solution, pH 6.0, to

1
2
3 remove non-hybridized DNA. For experiments utilizing a single ionic strength condition,
4
5 samples were introduced to the desired ionic strength by a stepwise decrease in NaCl
6
7 concentration. For experiments investigating sensor response to solution ionic strength, samples
8
9 were first equilibrated and characterized in 10 mM NaCl, pH 6, followed by a stepwise reduction
10
11 in ionic strength to 0.01 mM NaCl with measurements taken after equilibration at each condition.
12
13
14

15 16 17 **RESULTS AND DISCUSSION**

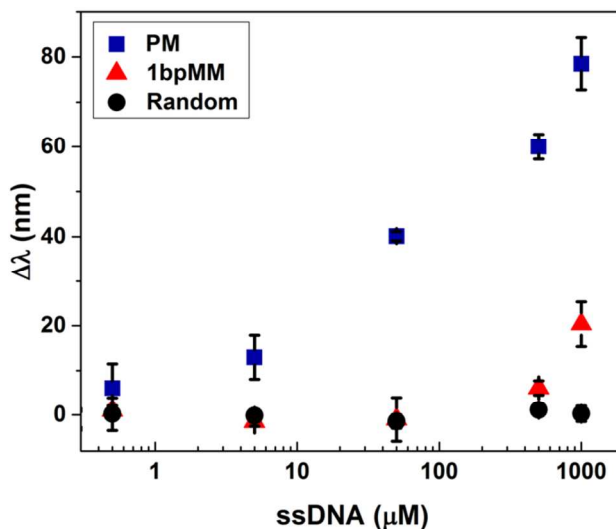
18 19 *Effect of Target DNA Concentration on Sensor Response*

20
21
22 Upon hydrogel functionalization with the p53 probe strand, the sensitivity and selectivity
23
24 of the sensing platform was investigated using a fully complementary (i.e., perfect match) and
25
26 random sequence. **Figure 2** shows the sensor response to concentrations of the perfect match
27
28 (PM) and random sequence ranging from 0.5 to 1000 μM , which corresponds to 25 pmole to 50
29
30 nmole of target DNA. Addition of the PM causes a dose-dependent redshift in the diffraction
31
32 peak with increasing target concentrations, resulting in a nearly 8-fold change in signal over the
33
34 concentration range. Conversely, the random sequence, which has the same GC content as the
35
36 PM, does not elicit a response at even the highest concentrations used. These results highlight the
37
38 sensitivity and selectivity of the approach, which presumably is due to differences in the
39
40 hybridization of the PM relative to the random sequence. Hybridization of the PM would result
41
42 in the immobilization of negative charges from the target DNA strand within the hydrogel,
43
44 triggering a change in the Donnan potential of the system. This change would, as a result, cause
45
46 the hydrogel network to expand and, in turn, the lattice spacing of the embedded CCA to
47
48 increase, leading to the observed redshift in peak diffraction. Accordingly, given that the random
49
50 sequence would not be expected to hybridize with the capture probe, the lack of response that
51
52
53
54
55
56
57
58
59
60

1
2
3 was generated by the random sequence is not surprising. In these measurements, the amount of
4 target DNA was less than that of the probe strand within the hydrogel such that the
5 immobilization of additional probe would not enhance the response.
6
7
8
9

10 Due to the sensitivity of the approach to the hybridization between the target and probe
11 strands, we also hypothesized that mutations in the target that weakened hybridization would be
12 detectable. Of particular interest was if single base pair mutations could be detected using this
13 approach. The detection of single base pair missense mutations is of practical importance for the
14 identification and screening of genetic markers that are associated with various diseases. In the
15 case of p53, screening for specific known hotspot mutations allows for the detection of genetic
16 hallmarks of cancer, which permits rational treatment using cancer-specific therapies. To
17 understand if our approach is sensitive to single base pair mutations, the optical response of
18 functionalized hydrogels to the target sequence with a base pair mismatch (1bpMM) was
19 measured. Notably, the mutation that was introduced results in the hotspot mutation R175H,
20 which perturbs the structure of the DNA binding domain of p53, resulting in a loss of p53
21 function.⁴⁵⁻⁴⁶ Although a redshift in the diffraction peak relative to the probe functionalized
22 sensor (i.e., the signal $\Delta\lambda$) was not observed at low concentrations of 1bpMM, a significant
23 change in the diffraction peak was apparent at higher concentrations, indicating that, despite
24 weaker affinity for the probe, a target with a single base pair substitution can be detected.
25 Presumably, because the binding affinity would increase, the response to a base pair change with
26 longer target and probe sequences would theoretically increase relative to the control. Moreover,
27 the apparent selectivity of the approach for the PM relative to 1bpMM suggests that, in principle,
28 a target strand may be differentiated from similar sequences in complex DNA mixtures. Such
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3 mixtures may include a multitude of strands with differing sequences and lengths, which may
4
5 arise from the digestion of cellular DNA.
6
7
8
9



10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29 **Figure 2.** Change in wavelength of peak diffraction relative to that of the probe functionalized
30 sensor as a function of the concentration of target ssDNA. Samples were annealed at a rate of 0.2
31 °C/min with PM, 1bpMM, or random target DNA concentrations ranging from 0.5 to 1000 μM.
32
33 Diffraction measurements were taken in a 0.25 mM NaCl solution at pH 6 and at room
34
35 temperature. Error bars represent $\pm 1\sigma$ from the mean of 3 samples.
36
37
38
39
40
41
42

43 *Effect of Annealing Rate and Temperature on Sensor Response*

44
45 Annealing conditions can greatly affect the ability of DNA to form the most
46 thermodynamically favorable duplexes. Accordingly, the impact of annealing conditions on the
47 sensitivity of the detection of the target sequence was investigated. The conditions for annealing
48 were varied by heating hydrogels that contained the probe in the presence of the target DNA to
49 approximately 10 °C above the theoretical T_m of the bound PM (~76 °C in 100 mM NaCl). After
50 heating, the solution containing the free PM target and hydrogel was cooled to well below the T_m
51
52
53
54
55
56
57
58
59
60

1
2
3 at different cooling rates. The final temperature to which the solution was cooled was 55 °C at
4
5 which the PM sequence should be mostly bound within the hydrogel. As shown in **Figure 3a**, the
6
7 magnitude of the redshift in diffraction is greatest at low cooling rates and drops off as the rate of
8
9 cooling increases beyond 0.2 °C/min. The decline in sensitivity at high cooling rates may be
10
11 attributed to imperfect hybridization of the PM to the probe sequence, which would result in
12
13 increased dissociation of the PM strand.
14
15

16
17 The impact of annealing conditions on the sensitivity of target detection was also
18
19 investigated by using a fixed cooling rate, but varying the annealing time. In this case, samples
20
21 were heated to 85 °C and subsequently cooled at a rate of 0.2 °C/min for different times, which
22
23 resulted in different final annealing temperatures, ranging from 85-55 °C. At the final annealing
24
25 temperature, the hydrogels were quickly cooled in ice water followed by rinsing in 100 mM
26
27 NaCl solution at room temperature to quench any further hybridization. **Figure 3b** shows the
28
29 optical response to the PM sequence as a function of the final annealing temperature. As
30
31 anticipated, at high final annealing temperatures, where the amount of bound PM is expected to
32
33 be low, the response of the sensing approach to the PM is low. A significant increase in the
34
35 detection of the PM sequence is observed at longer times and thus lower final annealing
36
37 temperatures. Annealing of the hydrogels below 55 °C resulted in no further change in the sensor
38
39 response, indicating further lowering the annealing temperature has negligible impact on
40
41 hybridization and sensitivity of the approach.
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

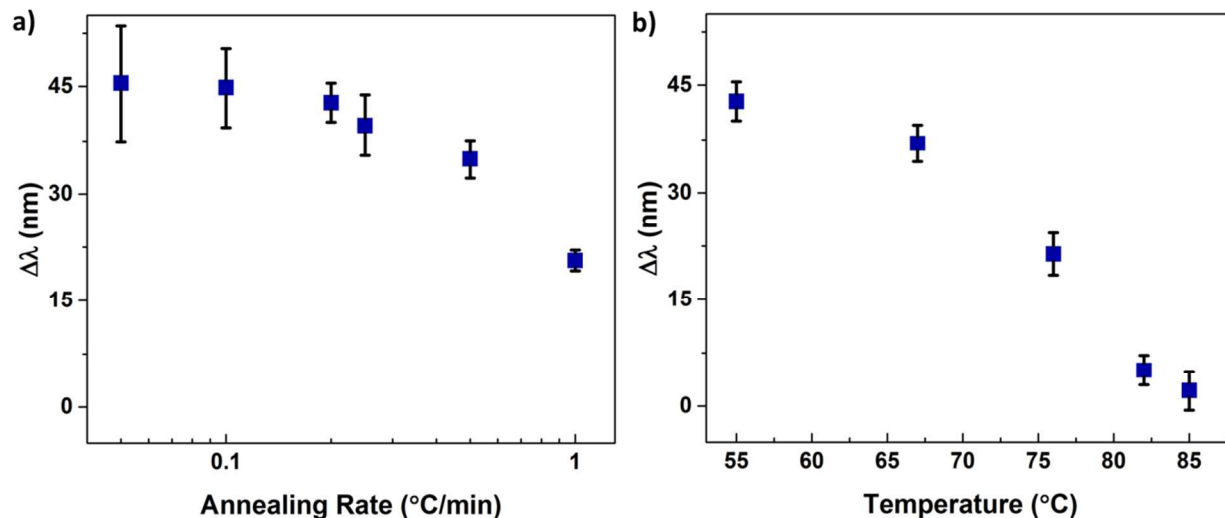


Figure 3. Optical response $\Delta\lambda$ observed for DNA hybridization as a function of a) annealing rate and b) annealing temperature. All samples were annealed in the presence of 50 μM PM target DNA at rates ranging from 0.05 to 1 $^{\circ}\text{C}/\text{min}$. In a) the annealing rate was maintained from 85 to 55 $^{\circ}\text{C}$, followed by uncontrolled but consistent cooling from 55 $^{\circ}\text{C}$ to room temperature. Samples presented in b) were annealed at a rate of 0.2 $^{\circ}\text{C}/\text{min}$ and, at the specified temperatures, were removed and immediately cooled in ice water followed by rinsing in 100 mM NaCl to quench hybridization. Diffraction measurements were taken in a 0.5 mM NaCl solution at pH 6 and at room temperature. Error bars represent $\pm 1\sigma$ from the mean of 3-6 samples.

Characterization of the Critical Melting Temperature of Target DNA

A fundamentally interesting question related to DNA detection using our sensing approach is if the T_m of the bound target DNA is the same in the hydrogel as in solution. Dramatic changes in T_m of the bound target DNA, relative to the annealing or characterization temperatures, may significantly lower the magnitude of the observed response to the target and thus the detection limit. To determine the T_m of the target sequences used in this work, the PM (at 50 μM) and 1bpMM (at 500 μM) sequences were annealed at a cooling rate of 0.2 $^{\circ}\text{C}/\text{min}$ to

1
2
3 the probe immobilized in hydrogels. For reference measurements, hydrogels were subject to
4 annealing under identical conditions without any target DNA. Of note, a larger concentration of
5 the 1bpMM sequence relative to the PM sequence was used to obtain optical responses of similar
6 magnitude for the two DNA targets. After annealing, the hydrogels were heated from room
7 temperature to 65 °C at a rate of 0.25 °C/min and the diffraction response of each sample was
8 measured over the entire temperature range. The diffraction response for the PM and 1bpMM
9 sequences is reported as a normalized response, which was determined as the difference of the
10 diffraction wavelength (i.e., $\Delta\lambda$) of the sample with target DNA from that of the reference (with
11 no target DNA) divided by the average maximum shift in peak diffraction wavelength (i.e.,
12 $\Delta\lambda_{\text{max,ave}}$) from the PM or 1bpMM. By reporting the normalized difference in diffraction
13 wavelength, volume changes in the hydrogel related to temperature increases and DNA
14 denaturation could be decoupled. In this case, changes in the equilibrium volume of the hydrogel
15 at elevated temperatures may be attributed to changes in the solvent density and Flory-Huggins
16 interaction parameter that alters the free energy of mixing of the system.⁴⁷

17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37 From the melting curve of the normalized diffraction response for the PM and 1bpMM
38 sequences (**Figure 4b**), the T_m of bound PM and 1bpMM could be approximated. The
39 approximated T_m of the bound PM and 1bpMM were 43.5 and 34.3 °C, respectively, as
40 determined from linear interpolation of the melting curves. For comparison, the theoretical T_m of
41 bound PM and 1bpMM in solution are 43 and 35 °C, respectively,⁴⁸ which are in good
42 agreement with the experimentally determined values when bound in the hydrogel. This close
43 agreement implies that the hybridization of the target DNA in the hydrogel is nearly identical to
44 that in solution and, moreover, that the theoretical T_m of the target DNA-probe duplex in solution
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

can be used to rationally design the probe strand to enhance sensitivity as well as to optimize the annealing and characterization temperatures.

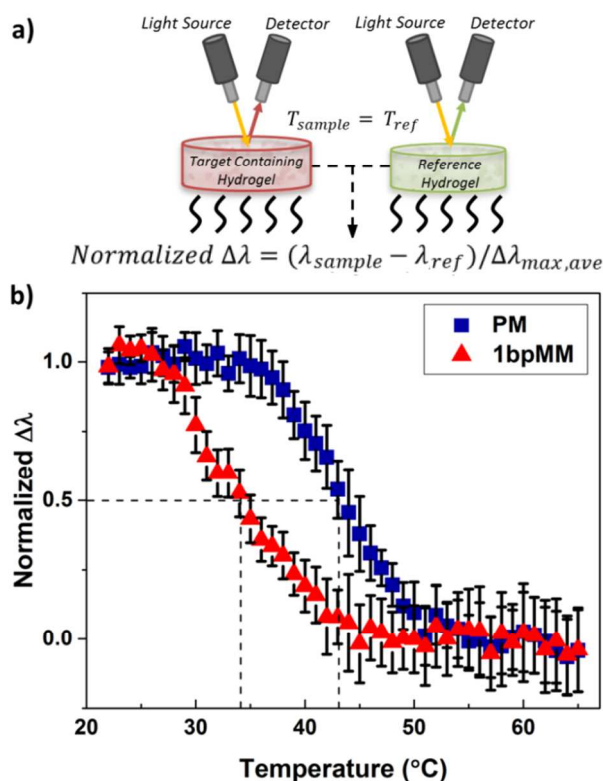


Figure 4. a) Schematic of the optical setup for characterizing melting curves and b) normalized melting curves for samples annealed with PM or 1bpMM target DNA. The normalized optical response was calculated as the difference in peak diffraction between the target containing samples and reference samples, normalized to the average maximum shift for the PM and 1bpMM samples. The T_m for the hybridized PM (blue squares) and 1bpMM (red triangles) was found to be 43.5 and 34.3 $^{\circ}C$, respectively. Sample annealing was performed prior to melting with 50 μM PM or 500 μM 1bpMM target DNA at a rate of 0.2 $^{\circ}C/min$. Melting was performed by ramping samples from room temperature to 65 $^{\circ}C$ at a rate of 0.25 $^{\circ}C/min$. Diffraction measurements were taken in a 2.5 mM NaCl solution at pH 6. Error bars represent $\pm 1\sigma$ from the mean of 3 samples.

Characterization of Sensor Response to Solution Ionic Strength

Due to the effect of ionic strength on DNA melting and the Donnan potential between the hydrogel and the surrounding environment, the detection of target DNA is strongly dependent on ionic strength during the diffraction measurements. To understand the magnitude of this effect, the sensor response as a function of ionic strength of the optical characterization solution was investigated for PM, 1bpMM, and control samples (**Figure 5**). For both PM and 1bpMM samples, starting at high ionic strengths, the response of the sensor increased significantly as the ionic strength of the characterization solution was lowered. However, as the ionic strength was further lowered, the sensor response decreased, resulting in two discernable regimes that describe the effect of ionic strength on the sensor response.

The distinction between these regimes is most notable for the PM sequence for which a maximum response was observed at an ionic strength of 0.25 mM. A decrease in the sensor response below this ionic strength is presumably due to the reduction in T_m for the hybridized PM to less than the characterization temperature, which results in weaker binding of the target strand. Notably, the predicted salt adjusted T_m for the hybridized PM is equivalent to room temperature at 0.15 mM salt (dashed line), which corresponds to the characterization temperature for these samples. As such, at ionic strengths less than 0.15 mM, one would expect that the amount of bound PM within the hydrogel at room temperature is very low. Despite the formation of the duplex being favored, the decrease in sensor response above the optimum ionic strength can be explained by a reduction in Donnan potential upon hybridization. This reduction is due to a smaller gradient in the concentration of mobile ions from the interior to the exterior of the hydrogel with increasing ionic strength.³⁵ For the 1bpMM sample, the optimum ionic strength

1
2
3 for detection was significantly higher than for the PM as expected given that the T_m for bound
4
5 1bpMM is presumably lower (than for bound PM) at all ionic strengths.
6
7

8 These results ultimately show the importance of considering ionic strength and
9
10 characterization temperature, which are intimately related, when expanding this sensing
11
12 approach to other sequences and mutations. For example, for probe and target strands with a
13
14 lower GC content, that when hybridized have a lower T_m , a lower characterization temperature
15
16 or a lower ionic strength could be used. Similarly, if longer probe and target strands are used, the
17
18 characterization temperature or ionic strength (or both) could be increased to enable detection.
19
20 However, the use of shorter probe and target strands enables a wider range of characterization
21
22 conditions due to a greater difference in the T_m between a strand that is a perfect match and one
23
24 that has a base pair mutation. The flexibility to rationally alter the characterization conditions for
25
26 the detection of virtually any sequence or length target strand represents a major strength of the
27
28 approach. In the case of longer target strands that form secondary structures (e.g., hairpin loops),
29
30 the annealing conditions could be altered to ensure melting of the target and hybridization with
31
32 the probe. Additionally, the location of a base pair mutation has little effect on the T_m of
33
34 hybridization unless the mutation is present at one of the end positions. The location of the
35
36 mutation at an end position would, in theory, result in a decrease in the difference in the response
37
38 between the perfect match and mutated strand. As such, this suggests that the probe strand
39
40 should be designed such that the anticipated mutation is internal within the probe sequence.
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

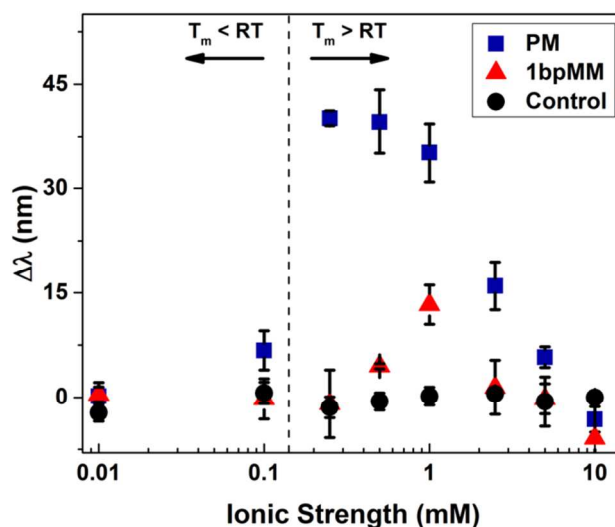


Figure 5. Optical response $\Delta\lambda$ as a function of ionic strength for samples annealed with 50 μM PM (blue squares) or 1bpMM (red triangles) target DNA annealed at a ramp rate of 0.2 $^{\circ}\text{C}/\text{min}$. Control samples (black circles) included probe functionalized hydrogels annealed in the absence of target DNA and hydrogels that were not functionalized with probe but annealed in the presence of target DNA. The vertical dashed line indicates the ionic strength conditions at which the melting temperature of the PM target DNA is equal to the optical characterization temperature. Diffraction measurements were taken after sample equilibration in 0.01 to 10 mM NaCl solutions at pH 6 and at room temperature. Error bars represent $\pm 1\sigma$ from the mean of 3-6 samples.

Detection of Methylated DNA

Having demonstrated the utility of our sensing approach to detect single base pair mutations, an interesting question to ask is if this approach is also sensitive to epigenetic DNA modifications. Such modifications include DNA methylation and hydroxymethylation, which play a crucial role in gene regulation and thus the development and progression of a variety of

1
2
3 diseases.⁴⁹⁻⁵⁰ Currently, the primary methods for detecting such modifications include mass
4 spectrometry⁵¹⁻⁵² and methylation-specific PCR,⁵³⁻⁵⁴ although newer methods, including
5 photopolymerization-based amplification,⁵⁵ have recently been reported. For conventional DNA
6 sensing methods, these modifications are difficult to detect due to the often negligible impact
7 these modifications generally have on DNA melting and thus hybridization.⁵⁶⁻⁵⁷
8
9

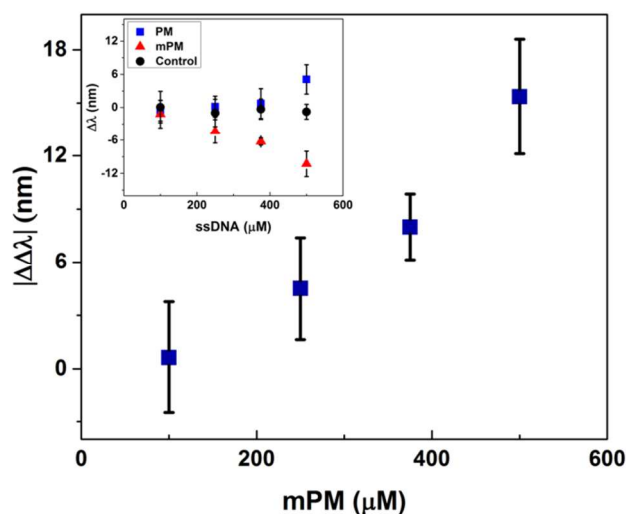
10
11 To explore the question of sensitivity to epigenetic changes, we characterized the optical
12 response using our approach to the fully methylated form of the PM sequence (mPM). The PM
13 sequence, when fully methylated, contains six methylated cytosine bases (**Table 1**). We
14 hypothesized that the methylated moieties in the mPM sequence would reduce the relative
15 hydrophilicity of the hydrogel (i.e., increase the strength of the Flory-Huggins interaction
16 parameter χ) and the extent of mixing in water, thereby causing the hydrogel to contract, rather
17 than swell. As expected, a dose-dependent response to the addition of the mPM target sequence
18 annealed at a rate of 0.2 °C/min was observed when measured at 10 mM ionic strength and room
19 temperature. This response, which is shown in **Figure 6**, is reported as the difference between
20 the optical peak shift due to hybridization of the mPM target strand and hybridization of the PM
21 target strand ($|\Delta\Delta\lambda|$). The raw response generated by the addition of the mPM sequence is shown
22 in the inset. Of note, optical characterization was performed in a high ionic strength solution to
23 minimize the electrostatic contributions to the observed response upon hybridization of the mPM
24 and PM sequences. By minimizing the contribution of electrostatics, the response that is
25 observed is primarily due to volume changes arising from modulation of the Flory-Huggins
26 interaction parameter and the addition of the methyl groups to the target DNA. Moreover, as
27 anticipated, the response generated by the addition of the mPM sequence resulted in a blueshift
28 in the diffraction peak of the encapsulated CCA, which is consistent with the hydrogel shrinking.
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

The relative impact of even a single methylation site on the target DNA may be predicted based on a theoretical model for hydrogel swelling,⁴⁷ in which the osmotic pressure term associated with hydrogel mixing in an aqueous solution scales as $\Pi_M \propto \chi$. To the simplest approximation $\chi_{water-hydrogel} \propto (\delta_{water} - \delta_{hydrogel})^2$ where δ are semiempirical solubility parameters related to the cohesive energy density as based on the approaches of Hildebrand or Hansen.⁵⁸⁻⁵⁹ Often, solubility parameters for polymers may be well estimated by using a group contribution approach, in which $\delta = (\sum F)/V$ where F are tabulated values of the molar attraction constants for common functional groups⁶⁰ and V is the molar volume of the repeat unit. Therefore, the dependence of the interaction parameter on the number of methylation sites n on the DNA target may be estimated by $\chi_{methylated} \approx \chi_{unmethylated} + 2n\delta_{CH_3}\chi_{unmethylated}^{1/2}$, and assuming a linear dependence of the optical response on χ (a reasonable approximation for small changes in χ , see Ref. 35), the blueshift in optical response based on a single methyl group may be estimated to be 2 ~ 3 nm. Based on the error of the optical response in **Figure 6**, the methylation of as few as two sites may be reasonably detected. One way to potentially enhance the response to a single methylation site may be to encapsulate the CCA in a hydrogel with a lower inherent χ .

Similarly, it may be predicted that other chemical modifications to DNA, including hydroxymethylation or more significantly functionalities that are highly hydrophobic, will also be readily detectable using this approach. For example, a single phenyl modification is predicted to have approximately three times the effect as a methyl modification. Chemical carcinogens that form DNA adducts may likewise be detected, including the classic example of benzo[a]pyrene which through a series of chemical reactions may be covalently linked to guanines in DNA.⁶¹⁻⁶³ In fact, the effect of benzo[a]pyrene on the p53 oncogene has been shown to lead to transversion

1
2
3 mutations, such as the single base pair mutation considered here.⁶⁴ The DNA biosensing scheme
4
5 presented here may then also provide opportunities to screen for chemical carcinogens and DNA
6
7 adducts that lead to mutations from which cancer originates.
8
9



10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31 **Figure 6.** Optical response as a function of the concentration of methylated DNA target. The
32 reported response ($|\Delta\Delta\lambda|$) is the difference between the optical shift in the wavelength of peak
33 diffraction upon hybridization of the methylated target and the detected shift upon hybridization
34 of the PM target. The inset shows the raw response generated by hybridization of the mPM
35 target. Samples were annealed from 85 to 55 °C utilizing a ramp rate of 0.2 °C/min and target
36 DNA concentrations ranging from 100 to 500 μM . Diffraction measurements were taken in a 10
37 mM NaCl solution at pH 6 and at room temperature. Error bars represent $\pm 1\sigma$ from the mean of
38 3-6 samples.
39
40
41
42
43
44
45
46
47
48
49
50
51

52 CONCLUSIONS

53
54 In summary, we have developed and demonstrated the utility of optically diffracting
55 hydrogels for the label-free detection of DNA, as well as missense mutations and methylated
56
57
58
59
60

1
2
3 sites ubiquitous to genes associated with a variety of diseases. Specifically, we showed that a
4 short target DNA sequence from p53 could be readily distinguished from an analogous sequence
5
6 that has a single base pair mutation that corresponds with the cancer hotspot mutation R175H in
7
8 a highly selective and dose-dependent manner. Furthermore, methylation of the native target
9
10 sequence could be detected, indicating the feasibility of using this approach to screen for
11
12 epigenetic modifications. Differences in the detection of the native versus mutant and methylated
13
14 sequences can be attributed to alterations in hybridization and polymer-solvent interactions,
15
16 respectively, and are sensitive to changes in ionic strength and hybridization conditions. This
17
18 approach ultimately represents a new paradigm for screening oncogenic hotspot mutations in p53
19
20 and other cancer-associated proteins. More broadly, this approach may be extended to screen for
21
22 genetic markers for other diseases as well as nucleotide modifications, stemming from epigenetic
23
24 changes as well as chemical modification. The high selectivity in differentiating between a target
25
26 strand and similar sequences would, in principle, permit the detection of the target strand from
27
28 complex DNA mixtures.
29
30
31
32
33
34
35
36
37

38 **ACKNOWLEDGMENTS**

39
40 Support was provided by the Univ. of Colorado Liquid Crystal Material Research Center (NSF
41
42 DMR0820579) and NIH Pharmaceutical Biotechnology Training Program (NIH 5T32GM8732).
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

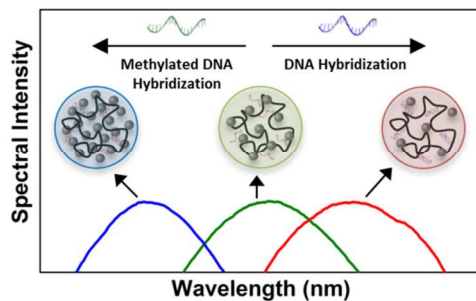
REFERENCES

- 1
 - 2
 - 3
 - 4
 - 5
 - 6
 - 7
 - 8
 - 9
 - 10
 - 11
 - 12
 - 13
 - 14
 - 15
 - 16
 - 17
 - 18
 - 19
 - 20
 - 21
 - 22
 - 23
 - 24
 - 25
 - 26
 - 27
 - 28
 - 29
 - 30
 - 31
 - 32
 - 33
 - 34
 - 35
 - 36
 - 37
 - 38
 - 39
 - 40
 - 41
 - 42
 - 43
 - 44
 - 45
 - 46
 - 47
 - 48
 - 49
 - 50
 - 51
 - 52
 - 53
 - 54
 - 55
 - 56
 - 57
 - 58
 - 59
 - 60
- (1) Benvidi, A.; Firouzabadi, A. D.; Moshtaghiun, S. M.; Mazloun-Ardakani, M.; Tezerjani, M. D., *Anal Biochem* **2015**. (In Press)
- (2) Ianeselli, L.; Greci, G.; Callegari, C.; Tormen, M.; Casalis, L., *Biosens Bioelectron* **2014**, *55*, 1-6.
- (3) Zhu, B. C.; Booth, M. A.; Shepherd, P.; Sheppard, A.; Travas-Sejdic, J., *Biosens Bioelectron* **2015**, *64*, 74-80.
- (4) Barhoumi, A.; Halas, N. J., *J Am Chem Soc* **2010**, *132* (37), 12792-12793.
- (5) Chak, C. P.; Lai, J. M. Y.; Sham, K. W. Y.; Cheng, C. H. K.; Leung, K. C. F., *Rsc Adv* **2011**, *1* (7), 1342-1348.
- (6) Bu, T.; Zako, T.; Fujita, M.; Maeda, M., *Chem Commun* **2013**, *49* (68), 7531-7533.
- (7) Wu, S.; Liang, P. P.; Yu, H. X.; Xu, X. W.; Liu, Y.; Lou, X. H.; Xiao, Y., *Anal Chem* **2014**, *86* (7), 3461-3467.
- (8) Yeri, A. S.; Gao, L. Z.; Gao, D., *J Phys Chem B* **2010**, *114* (2), 1064-1068.
- (9) Li, H.; Xiao, S. Y.; Yao, D. B.; Lam, M. H. W.; Liang, H. J., *Chem Commun* **2015**, *51* (22), 4670-4673.
- (10) Cheng, X. R.; Hau, B. Y.; Endo, T.; Kerman, K., *Biosens Bioelectron* **2014**, *53*, 513-518.
- (11) Ding, X. J.; Yan, Y. R.; Li, S. Q.; Zhang, Y.; Cheng, W.; Cheng, Q.; Ding, S. J., *Anal Chim Acta* **2015**, *874*, 59-65.
- (12) Zagorodko, O.; Spadavecchia, J.; Serrano, A. Y.; Larroulet, I.; Pesquera, A.; Zurutuza, A.; Boukherroub, R.; Szunerits, S., *Anal Chem* **2014**, *86* (22), 11211-11216.
- (13) Shapir, E.; Cohen, H.; Calzolari, A.; Cavazzoni, C.; Ryndyk, D. A.; Cuniberti, G.; Kotlyar, A.; Di Felice, R.; Porath, D., *Nat Mater* **2008**, *7* (1), 68-74.
- (14) Ryndyk, D. A.; Shapir, E.; Porath, D.; Calzolari, A.; Di Felice, R.; Cuniberti, G., *Acs Nano* **2009**, *3* (7), 1651-1656.
- (15) Tanaka, H.; Kawai, T., *Nat Nanotechnol* **2009**, *4* (8), 518-522.
- (16) Sun, W.; Yao, J.; Yao, T.; Shi, S., *Analyst* **2013**, *138* (2), 421-424.
- (17) Qiu, S.; Li, X.; Xiong, W.; Xie, L.; Guo, L.; Lin, Z.; Qiu, B.; Chen, G., *Biosens Bioelectron* **2013**, *41*, 403-408.
- (18) Wang, F.; Ma, C.; Zeng, X.; Li, C.; Deng, Y.; He, N., *J Biomed Nanotechnol* **2012**, *8* (5), 786-790.
- (19) Freeman, R.; Liu, X.; Willner, I., *J Am Chem Soc* **2011**, *133* (30), 11597-11604.
- (20) Sassolas, A.; Leca-Bouvier, B. D.; Blum, L. J., *Chem Rev* **2008**, *108* (1), 109-139.
- (21) Tosar, J. P.; Branas, G.; Laiz, J., *Biosens Bioelectron* **2010**, *26* (4), 1205-1217.
- (22) Cai, Z.; Smith, N. L.; Zhang, J. T.; Asher, S. A., *Anal Chem* **2015**, *87* (10), 5013-5025.
- (23) Zhang, J. T.; Wang, L.; Luo, J.; Tikhonov, A.; Kornienko, N.; Asher, S. A., *J Am Chem Soc* **2011**, *133* (24), 9152-9155.
- (24) Kamenjicki, M.; Kasavamoorthy, R.; Asher, S., *Ionics* **2005**, *10*, 233-236.
- (25) Nair, R. V.; Vijaya, R., *Prog Quant Electron* **2010**, *34* (3), 89-134.
- (26) Sharma, A. C.; Jana, T.; Kesavamoorthy, R.; Shi, L.; Virji, M. A.; Finegold, D. N.; Asher, S. A., *J Am Chem Soc* **2004**, *126* (9), 2971-2977.
- (27) Alexeev, V. L.; Sharma, A. C.; Goponenko, A. V.; Das, S.; Lednev, I. K.; Wilcox, C. S.; Finegold, D. N.; Asher, S. A., *Anal Chem* **2003**, *75* (10), 2316-2323.
- (28) Tian, E. T.; Wang, J. X.; Zheng, Y. M.; Song, Y. L.; Jiang, L.; Zhu, D. B., *J Mater Chem* **2008**, *18* (10), 1116-1122.
- (29) Yan, F. Y.; Asher, S., *Anal Bioanal Chem* **2007**, *387* (6), 2121-2130.

- 1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
- (30) Asher, S. A.; Sharma, A. C.; Goponenko, A. V.; Ward, M. M., *Anal Chem* **2003**, *75* (7), 1676-1683.
- (31) Arunbabu, D.; Sannigrahi, A.; Jana, T., *Soft Matter* **2011**, *7* (6), 2592-2599.
- (32) Xu, M.; Goponenko, A. V.; Asher, S. A., *J Am Chem Soc* **2008**, *130* (10), 3113-3119.
- (33) Lee, K.; Asher, S. A., *J Am Chem Soc* **2000**, *122* (39), 9534-9537.
- (34) MacConaghy, K. I.; Geary, C. I.; Kaar, J. L.; Stoykovich, M. P., *J Am Chem Soc* **2014**, *136* (19), 6896-6899.
- (35) MacConaghy, K. I.; Chadly, D. M.; Stoykovich, M. P.; Kaar, J. L., *Anal Chem* **2015**, *87* (6), 3467-3475.
- (36) Joerger, A. C.; Fersht, A. R., *Annu Rev Biochem* **2008**, *77*, 557-582.
- (37) Joerger, A. C.; Fersht, A. R., *Oncogene* **2007**, *26* (15), 2226-2242.
- (38) Freed-Pastor, W. A.; Prives, C., *Genes Dev* **2012**, *26* (12), 1268-1286.
- (39) Lane, D. P.; Benchimol, S., *Genes Dev* **1990**, *4* (1), 1-8.
- (40) Steele, R. J.; Thompson, A. M.; Hall, P. A.; Lane, D. P., *Br J Surg* **1998**, *85* (11), 1460-1467.
- (41) Soussi, T.; Wiman, K. G., *Cancer Cell* **2007**, *12* (4), 303-312.
- (42) Arunbabu, D.; Sannigrahi, A.; Jana, T., *J Appl Polym Sci* **2008**, *108* (4), 2718-2725.
- (43) Reese, C. E.; Guerrero, C. D.; Weissman, J. M.; Lee, K.; Asher, S. A., *J Colloid Interf Sci* **2000**, *232* (1), 76-80.
- (44) Stuchbury, T.; Shipton, M.; Norris, R.; Malthouse, J. P. G.; Brocklehurst, K.; Herbert, J. A. L.; Suschitzky, H., *Biochem J* **1975**, *151* (2), 417-432.
- (45) Liu, D. P.; Song, H.; Xu, Y., *Oncogene* **2010**, *29* (7), 949-956.
- (46) Soussi, T.; Lozano, G., *Biochem Biophys Res Commun* **2005**, *331* (3), 834-842.
- (47) Flory, P. J., *Principles of polymer chemistry*. Cornell University Press: Ithaca,, 1953; p 672 p.
- (48) Howley, P. M.; Israel, M. A.; Law, M. F.; Martin, M. A., *Journal of Biological Chemistry* **1979**, *254* (11), 4876-4883.
- (49) Robertson, K. D., *Nat Rev Genet* **2005**, *6* (8), 597-610.
- (50) Shukla, A.; Sehgal, M.; Singh, T. R., *Gene* **2015**, *564* (2), 109-118.
- (51) Ehrlich, M.; Nelson, M. R.; Stanssens, P.; Zabeau, M.; Liloglou, T.; Xinarianos, G.; Cantor, C. R.; Field, J. K.; van den Boom, D., *Proc Natl Acad Sci U S A* **2005**, *102* (44), 15785-15790.
- (52) Coolen, M. W.; Statham, A. L.; Gardiner-Garden, M.; Clark, S. J., *Nucleic Acids Res* **2007**, *35* (18), e119.
- (53) Herman, J. G.; Graff, J. R.; Myohanen, S.; Nelkin, B. D.; Baylin, S. B., *Proc Natl Acad Sci U S A* **1996**, *93* (18), 9821-9826.
- (54) Evron, E.; Dooley, W. C.; Umbricht, C. B.; Rosenthal, D.; Sacchi, N.; Gabrielson, E.; Soito, A. B.; Hung, D. T.; Ljung, B.; Davidson, N. E.; Sukumar, S., *Lancet* **2001**, *357* (9265), 1335-1336.
- (55) Heimer, B. W.; Shatova, T. A.; Lee, J. K.; Kaastrup, K.; Sikes, H. D., *Analyst* **2014**, *139* (15), 3695-3701.
- (56) Severin, P. M.; Zou, X.; Gaub, H. E.; Schulten, K., *Nucleic Acids Res* **2011**, *39* (20), 8740-8751.
- (57) Dahl, C.; Guldborg, P., *Biogerontology* **2003**, *4* (4), 233-250.
- (58) Hildebrand, J. H.; Scott, R. L., *Regular solutions*. Prentice-Hall: Englewood Cliffs, N.J., 1962; p 180 p.

- 1
2
3 (59) Hansen, C., *Hansen Solubility Parameters - A User's Handbook*. CRC Press: 1999.
4 (60) Cowie, J. M. G.; Arrighi, V., *Polymers: Chemistry and Physics of Modern materials*. 3
5 ed.; CRC Press: 2008.
6
7 (61) Jeffrey, A. M.; Jennette, K. W.; Blobstein, S. H.; Weinstein, I. B.; Beland, F. A.; Harvey,
8 R. G.; Kasal, H.; Miura, I.; Nakanishi, K., *J Am Chem Soc* **1976**, *98* (18), 5714-5715.
9 (62) Osborne, M. R.; Harvey, R. G.; Brookes, P., *Chem Biol Interact* **1978**, *20* (1), 123-130.
10 (63) Kriek, E.; Den Engelse, L.; Scherer, E.; Westra, J. G., *Biochim Biophys Acta* **1984**, *738*
11 (4), 181-201.
12 (64) Denissenko, M. F.; Pao, A.; Tang, M.; Pfeifer, G. P., *Science* **1996**, *274* (5286), 430-432.
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

Table of Contents Entry

**Highlight**

A novel approach for the optical detection of DNA was developed via polymerization of a photonic crystal within DNA-responsive hydrogel films, allowing for screening of genetic and epigenetic modifications.