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Selective Enrichment of Metal-binding Proteins Based on Magnetic Core/Shell Microspheres Functionalized with Metal Cations

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Metal binding proteins play many important roles in a broad range of biological processes. Characterization of metal binding proteins is important for understanding their structure and biological functions, thus leading to a clear understanding of metal associated diseases. T ϵ present study is the first to investigate the effectiveness of magnetic microspher functionalized with metal cations $(Ca^{2+}, Cu^{2+}, Zn^{2+}$ and $Fe^{3+})$ as the absorbent matrix in IMA^{\sim} technology to enrich metal containing/binding proteins. The putative metal binding proteins in rat liver was then globally characterized by using this strategy which was very easy to handle and could capture a number of metal binding proteins effectively. In total, 185 putative metal. binding proteins were identified from rat liver including some known lower abundant and membrane-bound metal binding proteins such as Plcg1, Acsl5, etc. The identified proteins were involved in many important processes including binding, catalytic activity, translation elongation factor activity, electron carrier activity, and so on.

1 Introduction

Metals play pivotal roles in a broad range of biological processes in all living organisms. It is estimated that approximately one-third of all proteins and enzymes require a metal cofactor for functionality (thus named "metalloproteins"), such as copper, zinc and iron to achieve their respective catalytic, regulatory and structural roles [1-2], and many diseases are also closely associated with metal metabolism disorders. Therefore, characterization of metalloproteins is important for understanding the structure and biological functions of such proteins, thus leading to a clear understanding of metal associated diseases. To this end, a variety of analytical methods have been used to analyse metalloproteome including bioinformatic approach [3], radioactive metal detection methods [4-6], colorimetric assays [7, 8], two-dimensional diagonal sodium dodecyl sulphatepolyacrylamide gel electrophoresis [9], and a combination of ICP and MALDI-MS methods [10, 11].

 Immobilized metal ion affinity chromatography (IMAC) is one of the best strategies in protein pre-fractionation in which specific protein fractions can be efficiently isolated from protein mixtures by designing particular binding ligand metal interactions [12]. Hence, a large number of IMAC strategies have been developed in the case of pre-fractionate proteins for metalloproteomic investigation by immobilizing different metal ions [13], such as Zn^{2+} , Ni²⁺, Cu²⁺ and Co²⁺ [14-16]. By using chelating Sepharose Fast Flow resin (Amersham Biosciences),

She *et al.* [17] analysed Cu and Zn metalloproteomes in thr human hepatoma cell lines (Hep G2, Mz-Hep-1 and SKHep-1), and Smith *et al.* [18] identified 67 copper-binding protein the cytoplasmic and microsomal of Hep G2 cells. Wang *et* \overline{q} ^{*t*} [19] carried out a systematic screen for copper-binding protei. s in soybean seeds by using IMAC technology. Barnett *et al.* [20] used the metal ion-charged IMAC columns (GE Healthca ϵ , UK) to probe the major cobalt, iron, manganese, and nickelbinding proteins in the marine cyanobacterium Synechococc_{is} sp. Heiss *et al.* [21] identified 22 Ni-interacting proteins in human B cells by IMAC based on Ni- Nitrilotriacetate (NTA) agarose beads (Qiagen). Ge et al. [22] identified seven bismuthbinding proteins from Helicobacter pylori cell extracts based on the NTA derivatized resin (Roche). Millar *et al.* [23] ident. 34 83 copper-interacting and 74 zinc-interacting proteins from Arabidopsis thaliana by IMAC. Sun *et al.* [24] prepared the Cuand Zn-immobilized metal affinity chromatography columns loading immobilized iminodiacetic acid (IDA, Pierce) into poly-prep columns (Bio-Rad) and analysed 232 and $1⁶$ putative Cu- and Zn-binding proteins from Streptococc pneumonia, in which 133 proteins were present in both preparations. In these studies, agarose/sephasore beads were the typical matrix, while Ren *et al.* [25] found that the type of support medium had a major impact on selectivity a d specificity of peptide selection in IMAC strategies by examining the selectivity of four different support materials with an IDA stationary phase to histidine-containing peptides. **Analysts** and T
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It is necessary, therefore, to look for a more efficient matrix to study metal-binding proteins by IMAC technology.

 Nanoparticles, especially magnetic composite microspheres, have been widely used in proteomic research including the enrichment of low abundant proteins/peptides, protein phosphorylation and glycolation [26, 27] due to their high surface areas and unique magnetic responsiveness. The present study uses, for the first time, magnetic core/shell microspheres functionalized with metal cations as absorbent matrix to isolate selectively putative metal-containing or metal-binding proteins by using IMAC approach. Firstly, we used transferrin as a model protein to investigate the effectiveness of this method, and found that it could be enriched effectively by the magnetic microspheres functionalized with metal ions. Then, the putative metal-binding proteins of rat liver were analysed. In total, 185 putative metal-binding proteins were identified, which were mainly binding proteins (including many known lower abundant and membrane-bound metal ion binding proteins), proteins with catalytic activity, structural molecules, transporter, and so on.

2 Experimental

2.1 Reagents and Materials

Transferrin, ammonium bicarbonate, dithiothreitol (DTT), iodoacetamide (IAA), urea, sodium dodecyl sulfate, trizma maleate, and α-cyano-4-hydroxycinnamic acid (CHCA) were obtained from Sigma-Aldrich (St. Louis, MO, USA). Acetonitrile (ACN, 99.9%), methanol, trifluoroacetic acid (TFA, 99.8%) and formic acid (FA) were purchased from Merck (Darmstadt, Germany). Sequencing grade modified trypsin was from Promega (Madison, WI, USA). Bradford assay reagent was obtained from Bio-Rad (Hercules, CA, USA). Imidazole was purchased from Aladdin. Glycidyl methacrylate (GMA) was obtained from Aldrich and vacuum distilled. N, N′- Methylenebisacrylamide (MBA) was bought from Fluka and recrystallized from acetone. 2, 2-Azobisisobutyronitrile (AIBN) was purchased from Sinopharm Chemical Reagents Company and recrystallized from ethanol. Deionized water used for all experiments was obtained from a Milli-Q system (Millipore, MA, USA). All other chemicals and reagents were of analytical grade and obtained from Shanghai Chemical Reagent Company.

2.2 Preparation of Magnetic Core/Shell Microspheres Fe3O4/PMG (poly (N, N′-methylenebisacrylamideco- glycidyl methacrylate)) / IDA Functionalized with Metal Cations

The magnetic core/shell microspheres Fe3O4/PMG/IDA were prepared according to ref. [28]. Briefly, magnetic colloidal nanocrystal clusters were prepared by a modified solvothermal reaction at first. γ-methacryloxypropyltrimethoxysilane were modified to the Fe3O4 surface and formed abundant double bonds. Following the addition of IDA to open the epoxy ring of GMA on the surface of composite microspheres, the core/shell Fe3O4/PMG microspheres were synthesized via one-step distillation-precipitation polymerization of GMA in ACN, by adding MBA as cross-linker and AIBN as initiator. After being washed with H2O, Fe3O4/PMG/IDA (50 mg) was incubated with 10 mL of mixture of CaCl₂, CuSO₄, ZnCl₂ and FeCl₃ solution (0.1 M) and stirred for 2 hr at room temperature (R.T.). The resultant cocktail of Fe₃O₄/PMG/IDA microspheres with different metal cations was collected, washed several times with water to remove excessive unbound metal ions and then dried in a vacuum oven at 40 °C.

2.3 Protein Extraction and Sample Preparation

Rat liver tissues were cut into small pieces and washed thr times with ice-cold phosphate-buffered saline to remove bloo². The tissues were dried by pledget and weighed, then homogenized on ice in lysis buffer (containing 25 mM Tris-HCl, pH8.0 and EDTA-free Protease Inhibitor Cocktail). The total lysate was centrifuged at 12 000 \times g for 30 min at 4 \degree C and the supernatant was collected. To deplete the metal ions in proteins, the lysate was treated for 24 hr by EDTA solution with molar ratio of 1:10 at 4 °C. The protein mixtures were then dialyzed in deionized water overnight at $4 \degree C$. To remove EDTA and metal ions, the water was changed several times ar^{-1} constantly stirred during dialysis. Finally, the water in samples was replaced to binding buffer in YM-3 centrifuge columns (Millipore) and the protein concentration was measured θ , Bradford assay using bovine serum albumin as a standard.

2.4 Selective Enrichment of Metal Containing / Binding Proteins

The microsphere cocktail of Fe₃O₄/PMG/IDA with Ca²⁺, Cu², Zn^{2+} and Fe³⁺ ions was first washed with lysis buffer three times. Protein mixtures in which metal ions had been remove by EDTA were then added to the magnetic microsphere together with binding buffer (25 mM Tris-HCl, pH8.0 and 5 mM imidazole). After incubating overnight at 4 °C, the proteinbound composite microspheres were separated from the remaining protein solutions by magnetic sedimentation and washed three times with binding buffer. Finally, the bound a proteins were eluted from the microspheres with elution buffe. (500 mM imidazole) and concentrated with YM-3 centrifuge columns. At the same time, equivalent microspheres without metal ions were used as a control in all experiments. For t' . model protein transferrin, the incubation was performed for hrs at 4 °C and the eluate fraction was separated by sodium dodecyl sulfate (SDS)-polyacrylamide gel electropho <is (PAGE) and visualized by silver staining, while for rat liver samples, the resultant eluates were subjected to in-solution digestion as follows: the proteins were treated with 10 mM DTT at 57 °C for 30 min, alkylated with 20 mM IAA at R.T. for 1 hr in the dark and digested with trypsin at an enzyme-tosubstrate ratio of 1:50 (w/w). The digestion procedure was allowed to proceed at 37 °C overnight, followed by lyophilisation. Finally, the products were analysed by LC-MS. **Analystan Controllering the Controllering Controlleri**

2.5 LC-MS Analysis and Database Search

The nano-LC MS/MS analysis was performed on a HPLC system composed of two LC-20AD nano-flow LC pumps, SIL-20 AC auto-sampler and an LC-20AB micro-flow $I^{\text{-}}$ pump (Shimadzu, Tokyo, Japan) connected to a LTQ Orbitrap

mass spectrometer (Thermo Scientific, Germany). Samples were re-suspended with buffer A (5% ACN containing 0.1% FA) and injected into a CAPTRAP column (0.5×2 mm, MICHROM Bioresources, CA) in 4 min with a flow rate of 25 μL min-1. Subsequently, samples were separated on a Magic C18-AQ reverse phase column (100 μm id \times 15 cm, Michrom Bioresources, USA) with a linear gradient from 5 to 45% buffer B (90% ACN in 1% FA) in 60 min at a flow rate of 500 nL min-1. The separated samples were introduced into the mass spectrometer via an ADVANCE 30 μm silica tip (MICHROM Bioresources, CA). The spray voltage was 1.8 kV, and capillary temperature was 180 °C. The mass spectrometer was operated in the data-dependent mode to switch automatically between MS and MS/MS acquisition. Survey full-scan MS spectra with one microscan (400-1800 Da) was acquired in the Orbitrap with a mass resolution of 100,000 at m/z 400, followed by MS/MS of the eight most-intense peptide ions in the LTQ analyser. The automatic gain control (AGC) during full-MS and MS/MS was set to 1000 000 ions and 10 000, respectively. Single charge state was rejected and dynamic exclusion was used with two microscans in 10 s and 90 s exclusion duration. For MS/MS, precursor ions were activated using 35% normalized collision energy at the default activation q of 0.25 and an activation time of 30 ms.

Database search was performed using Mascot v2.3.2 (Matrix Science, UK) as previously described against a composite database, including original and reversed protein database assuming the digestion enzyme trypsin [29]; a maximum of two missed cleavages were allowed, and full tryptic specificity required. Mass value was set as monoisotopic, and peptide charges 2+ and 3+ were taken into account. All samples were searched with oxidation (M) and carbamidomethyl (C) as variable modifications. Precursor ion mass tolerance was set to 15 ppm and product ion tolerance to 1.0 Da. The Searching database contained 7858 rat protein entries extracted from UniProt Knowledgebase (Release 2013_08) with an in-house perl script. The results were further filtered by Scaffold (Proteome Software, V4.2.0). An additional database search by X! Tandem was performed with parameters similar to Mascot except that Glu->pyro-Glu, ammonia-loss and Gln->pyro-Glu of the N-terminus were specified additionally. Peptide identifications by X! Tandem were accepted with >95.0% probability assigned by the PeptideProphet algorithm [30]. Probabilities of Mascot identifications were assigned by the Scaffold Local false discovery rate algorithm. Protein identifications with >99.0% probability by the ProteinProphet algorithm [31] and > 2 unique peptides were accepted. Then, label-free quantitation for proteins eluted from metal ions - IMAC (Exp) and control (Con) was carried out by Scaffold according to identified protein' total spectra count. The proteins with ratio of Exp / Con greater than 2 were defined as metal binding proteins. If a protein was not identified in the Con group, it was also defined as metal containing / binding proteins.

2.6 Bioinformatic Analysis

Functional classification and subcellular localization analysis for the identified proteins were carried out by gene ontology (GO) (http://www.geneontology.org/), uniprot (http://www.uniprot.org/uniprot) and DAVID Bioinformatics Resources (http://david.abcc.ncifcrf.gov/).

3 Results and discussion

3.1 Selective Enrichment of Putative Metal Containing/Binding Proteins by Magnetic Core/Shell Microspheres Functiona..z.u **with Metal Cations**

To investigate the enrichment efficiency of the magne c microspheres to metal containing/binding proteins, transferrin was used as a model protein. Transferrin is an iron- binding protein, which can bind iron very tightly, but reversibly, and contains two specific high affinity Fe (III) binding sites. At figure, 5 μg of transferrin was treated by EDTA solution to remove from it the bounded metal ions, so that it could be captured more efficiently by the magnetic nanoparticles with iron ions. The enrichment strategy was illustrated in Fig. 1.

Fig.1 Flowchart of the enrichment of metal containing/bind proteins in rat liver based on magnetic core/shell microsphere functionalized with metal cations followed by LC-MS analysi.

The Fe₃O₄/PMG/IDA-Fe³⁺ microspheres was added transferrin solutions and incubated for 4 hrs at 4° C, before the microspheres were separated from solution by magnetic sedimentation. The proteins were eluted by 500 mM imidazole, separated by 10% SDS-PAGE and stained by silver staining. the same time, the bare Fe3O4/PMG/IDA microspheres we. used as control by following exactly the same procedure. As shown in Fig. 2, transferrin could be effectively capture Fe3O4/PMG/IDA-Fe3+ microspheres (shown in the 'Exp-elute' lane), while in the control experiment, the signal was relatively very weak. Therefore, the Fe₃O₄/PMG/IDA-Fe³⁺ microsph could be used as the IMAC matrix to enrich effectively the iron binding proteins and very little non-specific binding w_{α} observed.

Fig.2 SDS-PAGE analysis for transferrin eluted from Fe3O4/PMG/IDA-Fe3+ microspheres (Exp: experiment group) and from bare Fe3O4/PMG/IDA microspheres (Con: control group). Transferrin, in which $Fe³⁺$ ions were delepted at first, was incubated with the magnetic microspheres and eluted with 500 mM imidazole. Then, the eluates were separated by SDS-PAGE and stained by silver staining. In Con group, transferrin existed mainly in the flow through fraction, while it was mainly in the elute fraction in Exp group.

 IMAC is a promising tool for rapidly capturing proteins that are involved in metal sensing and trafficking, because the binding sites in such proteins enable facile metal exchange (i.e. binding and release) and are often located on the protein surface [2]. Compared with the regular matrix (such as agarose and sepharose) in the study of metal binding proteins by IMAC technology, nanoparticles have prominent features of high surface area, a large number of functional groups and welldefined particle sizes. Thus, it has larger adsorption capacity and a much lower limit of detection towards metal containing/binding proteins. In addition, it can make the enrichment strategy very convenient and easy to handle by using an external magnetic field, which makes composite microspheres fascinating and promising for high throughput metal binding/containing protein research.

3.2 High-throughput Profiling of Putative Metal-Binding Proteins in Rat Liver

The liver has a wide range of functions including detoxification, protein synthesis, metabolism, etc., so hepatocytes are ideal for study of metal containing/binding proteins. We carried out the putative metal-binding proteins of rat liver by IMAC based on the magnetic core/shell microsphere cocktail functionalized with Ca^{2+} , Cu^{2+} , Zn^{2+} and Fe^{3+} ions. Metal binding proteins and metalloproteins are sensitive to various experimental conditions such as the pH and constitution of lysis buffer, and the changes in their tertiary structure often result in disturbance of the buried metal binding sites. The proteins were extracted in a state of physiological conformation by using a non-denaturing lysis buffer in our study. Additionally, to release the metal binding sites already occupied by metal ions and ensure proteins' binding capacity, the extracted proteins were firstly treated with EDTA and incubated with the cocktail of

microspheres functionalized with metal cations $(Ca^{2+}, Cu^{2+}, Qa^{2+})$ Zn^{2+} and Fe³⁺). The eluted proteins were digested with trypsin and then subjected to LC-MS analysis.

In total, 295 proteins were identified ambiguously (Shown in Supplementary Table), including many known metal bin⁴¹ proteins. After analysing with label-free quantitation, $18⁵$ proteins (Table 1) were designated putative metal binding proteins from rat liver in which 85 proteins had been reported as metal binding proteins by GO, BIND and other proteomic studies $[17, 18, 32]$. The subcellular localization and molecular function of identified proteins were annotated by gene ontology. As shown in Fig. 3 (A), the candidates were mainly localized \cdot cytoplasm, nucleus, mitochondrial, plasma membrane, and so on. Actually, the remaining 110 proteins, which the ratio of E_{T} / Con was less than 2 by label-free relative quantitation, also included 24 known metal binding proteins reported in UniProt database, such as Cytochrome P450 protein family, Bifunctional ATP-dependent dihydroxyacetone kinase/FAD-AMP lyase (cyclizing), and so on. Maybe the unoptimized binding/elution conditions or cut-off criterion for label fre quantatition caused that these proteins were detected only low significance levels. These proteins were also listed in Table 2. A^2
 A^2

It was reported that transition metal ions were required for many aspects of mitochondrial physiology, in which copper, iron and zinc were cofactors in metalloenzymes and metalloproteins $[33, 34]$. In this study a range of known met all binding proteins in the mitochondrial electron transport chain, metabolism and proteolysis machinery were identified. F. example, carbamoyl-phosphate synthase [ammonia] mitochondrial, which is a known calcium ion binding protein and involved in the urea cycle of ureotelic animals where the enzyme plays an important role in removing excess ammon a from the cell, was identified ambiguously. In particular, several known metal binding proteins in mitochondrion inner membrane were also detected. Cytochrome c oxidase subunit 5A mitochondrial, which contains heme moiety and can bi d iron ions, lies in the mitochondrion inner membrane and is the terminal oxidase in mitochondrial electron transport. A^T synthase subunit beta and NADH dehydrogenase [ubiquinone] flavoprotein 2, which are known calcium ion binding and iron binding protein respectively, were also ambiguously ident. \sim in our experiment.

Interestingly, 24 ribosomal proteins were identified in our experiment. Among which, three proteins from 40S subunit 10 from 60S subunit were firstly identified by proteomic techniques as putative metal-binding proteins in rat liver, which indicates that the ribosome may act as a metal ion storage site. Ribosomes are larger ribonucleoprotein complexes that mediate protein synthesis in all organisms. Zinc plays key catalytic roles in enzymes, as well as a structural role in numerous transcriptional activators and regulators. It was reported that t e majority of intracellular Zn(II) reside in the ribosome. Escherichia coli 70S ribosomes tightly bound 8 equiv of Zn [S] and ribosomes from Bacillus subtilis closely bound 2.5 equiv \sim Zn [36]. Actually, the high percentage of ribosomal proteins

was also identified in Streptococcus pneumonia by proteomic method [24].

 Notably, several membrane proteins and lower abundant proteins were identified in our study. For example, 1 phosphatidylinositol 4,5-bisphosphate phosphodiesterase gamma-1 (Plcg1) was ambiguously identified, which is a known calcium binding protein and plays an important role in the regulation of intracellular signalling cascades, actin reorganization and cell migration. According to the results of protein abundance analysis (http://pax-db.org), Plcg1's abundance is 2.25 ppm (ranked bottom 25% of the rat dataset). Long-chain-fatty-acid--CoA ligase 5 (Acsl5) is another known metal binding protein and located in mitochondrion outer membrane. Acsl5 is involved in lipid metabolism and has transmembrane domain. Its abundance is 96.6 ppm (ranked top 25%) and belongs to the medium/high abundant protein. Therefore, this strategy described here for the study of metal binding proteins could be used to isolate lower abundance and membrane-bound proteins to some extent, which was probably due to the following reasons: firstly, the absorbent carriers used in this method were nanoparticles with high surface area and a large number of functional moieties, leading to a higher binding capacity and much lower limit of detection towards metal binding proteins; secondly, the metal cofactor in the metal binding proteins had been delepted by EDTA before the enrichment procedure, so the putative metal binding sites in these proteins could interact with the absorbent materials of IMAC easily.

 The identified proteins were involved in multiple biochemical processes such as amino acid and protein biosynthesis, metabolism and redox homeostasis. According to their molecular functions, the identified proteins were mainly binding proteins, proteins with catalytic activity, structural molecule, transporter, proteins with electron carrier activity, etc. As shown in Fig 3 (B), 152 proteins were binding proteins among the identified proteins and the largest class. Several kinds of binding proteins were profiled including many known metal binding proteins, protein binding proteins, nucleic acid binding proteins and drug binding proteins. There were 48 known metal binding proteins reported by Uniprot, GO and BIND. For example, Parathymosin, which can mediate immune function, is a known zinc ion binding protein. Serum albumin can also bind a variety of essential and toxic metal ions, including Ca(II), Co(II), Ni(II), Cu(II), Zn(II) and Cd(II) [37]. It is believed to be the major zinc transporter in plasma [38] and has an excellent binding site for Cu(II) or Ni(II) at the Nterminus (N-terminal copper and nickel-binding motif, ATCUN) [39].

 One of the most important roles of metal species in biological organisms is acting as cofactors of diverse enzymes. In this study, 113 identified proteins had catalytic activity, mainly including oxidoreductase activity, transferase activity, hydrolase activity, dehydrogenase activity, GTPase activity and kinase activity. Peroxiredoxins, a class of thioredoxindependent peroxide reductase, are multifunctional thiol-specific antioxidant proteins that are highly abundant in the cytoplasm of mammalian cells where they regulate antioxidant defence, tumour suppression and control hydrogen peroxide signalling. Here, three members of peroxiredoxins (peroxiredoxin-1, 2 and 6) were identified, which were reported as metal binding proteins in UniProt database and previous proteomic literatures [17, 18, 32]. Protein disulfide-isomerase can catalyse the formation, breakage and rearrangement of disulfide bonds; four members of this protein (protein disulfide-isomerase, protein disulfide-isomerase A3, A4 and A6) were ambiguously identified in our study.

However, it should be noted that cytosolic proteins were still the largest group in our dataset such as cytoplasmic aconitation hydratase, isocitrate dehydrogenase [NADP] cytoplasmic and superoxide dismutase $[Cu-Zn]$, which were known me i^{\dagger} binding proteins. In addition, according to the results of protein abundance analysis by http://pax-db.org, the identified proteins were mainly high abundance proteins (ranked from above $25₁₀$) to above 5%). It was mainly because: 1) we used the moderation lysis buffer at almost physiological environment without detergents such as urea and SDS to keep the proteins in a staof physiological conformation and can still bind metal ions; \mathbf{t} t in such buffer conditions, it was usually ideal for the identification of higher abundant and more hydrophilic proteins; 2) the metal-binding sites in many proteins, for example, metalloenzymes, are often deeply buried in the interior of the protein, so they were not accessible for the EDTA treatment and interacting with the immobilized metal ions; 3) the $w_1 e$ dynamic range of relative protein expression also made ... difficult to detect those proteins expressed at low levels. Thus, combined with the enrichment strategy, improvement techniques for isolation of low abundant proteins and $m²$ hydrophobic proteins may further enhance the inclusiveness \overline{e}^{ϵ} putative metal-binding proteins. Furthermore, it should be mentioned that this method can only study those proteins with metal binding capacity in vitro and can't provide informati n on in vivo metal-protein interactions exactly, which is also the common limition of available IMAC methods. So furth x experimental evidence should be required to confirm whether a protein is a genuine metal binding protein in vivo or not. **Analysts of the Contract of the Memorial of the Memorial of the Memorial of the Winding of the Wi**

Fig.3 Results of large-scale protein annotation through Gene ontology. (A) protein subcellular localization analysis; B) protei molecular and functional analysis.

4 Conclusions

In summary, we first tested the effectiveness of magnetic core/shell microspheres functionalized with metal ions ($Ca²$

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 Cu^{2+} , Zn^{2+} and Fe^{3+}) to enrich metal containing/binding proteins, and then globally characterized the putative metalbinding proteins of rat liver by using this strategy. It was very easy to handle and capture a number of putative metal binding proteins effectively, although only those proteins with metal binding capacity in vitro can be studied by this method. In total, 185 proteins were identified as putative metal binding proteins from rat liver including some known lower abundant and membrane-bound metal binding proteins such as Plcg1, Acsl5, etc. The identified proteins were involved in many important processes including binding, catalytic activity, translation elongation factor activity and electron carrier activity. Therefore, our results indicated that the magnetic microspheres with metal ions was an ideal absorbent matrix in IMAC technology to capture proteins with metal binding capacity in vitro, and the enhanced selective recognition for putative metalbinding proteins would facilitate a complementary identification of the under-explored metal binding proteins and provide new insight into the dynamics of metal binding proteins. In parallel, it is also a relatively flexible method and can be used to study both the specific putative metal-binding proteins and its binding peptides. If a specific kind of metal ions, e.g. zinc cations, were immobilized onto the microspheres, the zincbinding proteins could be effectively captured in vitro. If protein mixture digests were incubated with the microspheres, the zinc binding peptides could be analyzed.

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- Table 1. Identification of metal-binding proteins in rat liver

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Table 2. Summary of known metal binding proteins detected at low significance levels

An new method based on magnetic microspheres with metal ions was developed and applied to characterize putative metal binding proteins.