

Analyst

Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this *Accepted Manuscript* with the edited and formatted *Advance Article* as soon as it is available.

You can find more information about *Accepted Manuscripts* in the [Information for Authors](#).

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard [Terms & Conditions](#) and the [Ethical guidelines](#) still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.

Cite this: DOI: 10.1039/x0xx00000x

Received 00th January 2014,
Accepted 00th January 2014

DOI: 10.1039/x0xx00000x

www.rsc.org/

Conductivity-Based Detection Techniques in Nanofluidic Devices

Zachary D. Harms, Daniel G. Haywood, Andrew R. Kneller, and Stephen C. Jacobson*

This review covers conductivity detection in fabricated nanochannels and nanopores. Improvements in nanoscale sensing are a direct result of advances in fabrication techniques, which produce devices with channels and pores with reproducible dimensions and in a variety of materials. Analytes of interest are detected by measuring changes in conductance as the analyte accumulates in the channel or passes transiently through the pore. These detection methods take advantage of phenomena enhanced at the nanoscale, such as ion current rectification, surface conductance, and dimensions comparable to the analytes of interest. The end result is the development of sensing technologies for a broad range of analytes, e.g., ions, small molecules, proteins, nucleic acids, and particles.

Introduction

We review the use of nanofluidic devices for chemical analysis with a special focus on conductivity-based detection. Conductivity-based detection methods have the benefit of being label-free and are able to sense a wide range of analytes from ions to proteins to nucleic acids to particles when coupled to a nanoscale channel. We limit this review to experiments that use synthetic and solid state nanofluidic pores and channels for chemical analysis. Reviews that cover biological nanopores^{1,2} and the theory of nanoscale transport^{3,4} can be found elsewhere.

Nanofluidic channels and pores can have one, two, or three dimensions with nanometre length scales. For this review, we refer to nanochannels as conduits having at least one dimension confined to the nanoscale, usually channel depth or width. Nanopores are defined simply as having at least their lateral dimensions (i.e., diameter or width and depth) confined to the nanoscale. We survey a range of device formats from conventional nanopore sensors, in which a three-dimensional pore is used to sense changes in conductivity, to in-plane nanopore devices, in which the pore is integrated directly into a micro- or nanofluidic channel.

To accomplish label-free conductivity detection, some element must be incorporated into the nanofluidic device that is sensitive to the presence of the analyte of interest. Nanopores

have constrictions through which particles pass transiently and typically cause an increase in the pore resistance that is proportional to particle volume. In-plane nanopores are fabricated directly into micro- and nanofluidic devices and can be used to sense particles multiple times. An alternative strategy to axial current measurement is lateral current measurement with electrodes positioned perpendicularly to the nanofeature and measure current as the analyte passes the electrodes. Electrodes for transverse measurements are placed on both sides of a nanochannel or nanopore and are composed of metal electrodes or additional nanofluidic channels. Ion current rectification is exploited to build detectors based on modification of nanopore surface charge, and the on/off behaviour is used to build diodes and transistors for enhanced device functionality. Time-delayed electrical responses of nanofluidic devices provide impedance measurements for the determination of device dimensions and reaction kinetics. Lastly, conductivity measurements with these devices are complemented by optical visualization.

Resistive-Pulse Sensing with Out-of-Plane Nanopores

Resistive-pulse sensing with nanopores is used in a wide variety of research areas including virology, bacteriology, and DNA studies. Resistive-pulse sensing uses a constriction (pore) with dimensions comparable to the analyte of interest, and the pore separates two conductive electrolyte solutions. As analyte

is driven through the pore by an electrical potential or pressure difference, changes in conductivity from transient blockages in ion current are measured, as shown in Fig. 1. Each transient decrease in current or ‘pulse’ corresponds to the transit of a single particle through the pore. Change in current is proportional to the change in nanopore resistance during translocation: thus, the amplitude of the current pulse increases with particle volume and decreases with pore volume.⁵ Pore resistance is the sum of the geometrical and access resistance terms, and both must be taken into account for accurate particle sizing.⁶ However, pulse amplitude is also affected by particle charge, and the counter ions moving through the pore. At low salt concentration, the conductivity of counter ions solvating DNA strands is greater than the solution conductivity, resulting in an increase in baseline current.⁷ Charged polystyrene particles in low conductivity solutions show an increase in baseline current at the exit of the nanopore.⁸ Pulse shape is a convolution of pore and particle geometry, and pulse shape reflects pore topography.⁹ Pulse width is the residence time within the pore, decreases with particle velocity,¹⁰ and increases with particle-pore interactions, such as adsorption.¹¹ Particle trajectory through the pore also plays an important role in both pulse width and amplitude.¹²

Over the last 60 years, a number of different types of sensors have been developed, and today, nanopores are fabricated in a variety of materials, e.g., polymers, SiN and SiO₂ membranes, Si and SiO₂ substrates, glass nanopipettes, carbon nanotubes, and graphene sheets (see Fig. 2). The original resistive-pulse sensor is described in a patent by Coulter in 1953 and consists of a glass test tube with a pinhole in the bottom.¹³ The ‘Coulter Counting’ principle was originally developed to count blood cells,¹⁴ and within two years, these devices had sufficient sensitivity to detect bacteria such as *E. coli* and *B. megaterium*.^{15,16} The ability of early Coulter counting devices to study dynamic processes is reported for the determination of PbCO₃ and CaC₂O₄ crystal growth rates.¹⁷

One of the first examples of nanoscale resistive-pulse sensing is detection of virus capsids passing through a nanopore in a polymer membrane. The Nucleopore method was developed to create filters for the screening of biological materials.¹⁸ Damage tracks are created by bombarding a membrane with high-energy fission products and are wet chemically etched to create pores with diameters ranging from nm to μm . In initial experiments, sub 100-nm particles are detected with a polycarbonate pore.¹⁹ Electrokinetic mobility measurements are made on latex nanoparticles and viruses, and a 6-nm difference in particle diameter is resolved from the difference in pulse amplitudes.⁵ Additionally, polycarbonate pores are used to study several types of bacteriophages and type C Oncornavirus, demonstrating the usefulness of resistive-pulse sensing to accurately count and size biologically relevant species in the 100-nm diameter range.²⁰

Track-etched polymer membranes continue to offer a low-cost, high-performance option for nanopore fabrication. Commonly used polymers include poly(ethylene terephthalate) (PET) and polyimide (PI). These commercially available membranes have a single heavy-ion damage track and are chemically etched with a two-step process to produce pores with reproducible base and tip diameters.²¹ A polymer membrane with a single pore is used for resistive-pulse sensing of individual DNA fragments²² and porphyrin molecules.¹⁰ PET pores are coated with gold, functionalized with poly(ethylene glycol), and used to study binding of bovine serum albumin to antibodies.^{23,24} PET pores contain carboxyl groups on the surface, and the pore surface is covalently modified with triethylene glycol to minimize electroosmotic flow and particle adsorption. These modified pores are used to resolve T = 3 and T = 4 Hepatitis B Virus capsids.²⁵

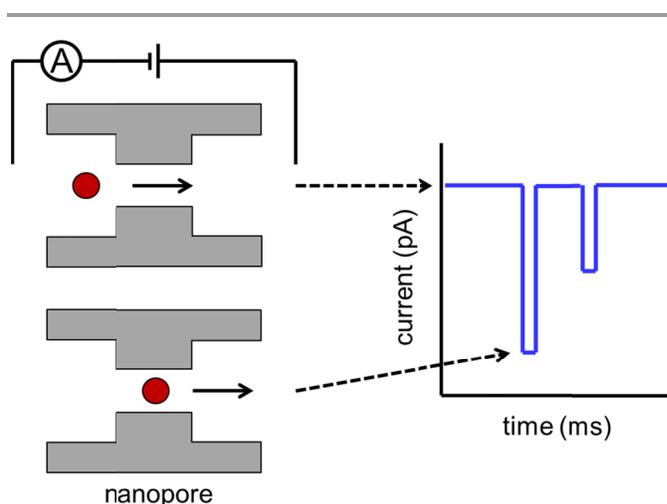


Fig. 1. Principle of resistive-pulse sensing. A potential is applied across a nanopore to electrokinetically drive the particle through the pore. As a particle transits the pore, the resistance typically increases, and current decreases. Transient decreases in current are observed as a series of ‘pulses’ that are proportional to particle size. The width of each pulse is proportional to the velocity of the particle through the pore.

Nanopores fabricated in glass offer low noise²⁶ and are more resistant to collapse than pores formed in polymers, and the surfaces of glass pores are easily modified by various silane chemistries. Recently, glass nanopores are used to characterize the transport of polystyrene nanoparticles^{27,28} and to examine the deformation of microgels as they pass through the nanopore.²⁹ Laser-pulled glass nanopipettes are easy and inexpensive to fabricate as resistive-pulse sensors and are capable of analysing single DNA strands labelled with nanoparticles.³⁰ Femtosecond-pulsed lasers are used to fabricate conical nanopores in glass for studying immune complexes.³¹ Antibody binding in complex media is accomplished at protein concentrations down to 30 nM. Laser pulling combined with wet chemical etching provides excellent control over pore size, and pipettes formed with this method are used to distinguish

Analyst

between enantiomers by immobilization of β -cyclodextrin within the pore.³² Glass pipettes are also functionalized to study biotin-streptavidin and antibody-antigen interactions.³³

Resistive-pulse sensors are also made from carbon nanotubes embedded in non-conducting polymer blocks. The pore length is defined by the thickness of the polymer block, and the nanotube diameter is tuned from a few nanometres to hundreds of nanometres by the synthesis process. Carbon nanotube sensors accurately measure size and electrokinetic mobility of individual polystyrene nanoparticles.^{34,35} Because carbon nanotubes are smooth-walled cylinders, calculations of particle size and mobility are accurate and precise and compare favourably to traditional methods such as dynamic light scattering, transmission electron microscopy, and phase analysis light scattering.³⁶ Carbon nanotubes are grown on Si wafers and isolated in polymer blocks for the analysis of single-stranded (ss) DNA.³⁷

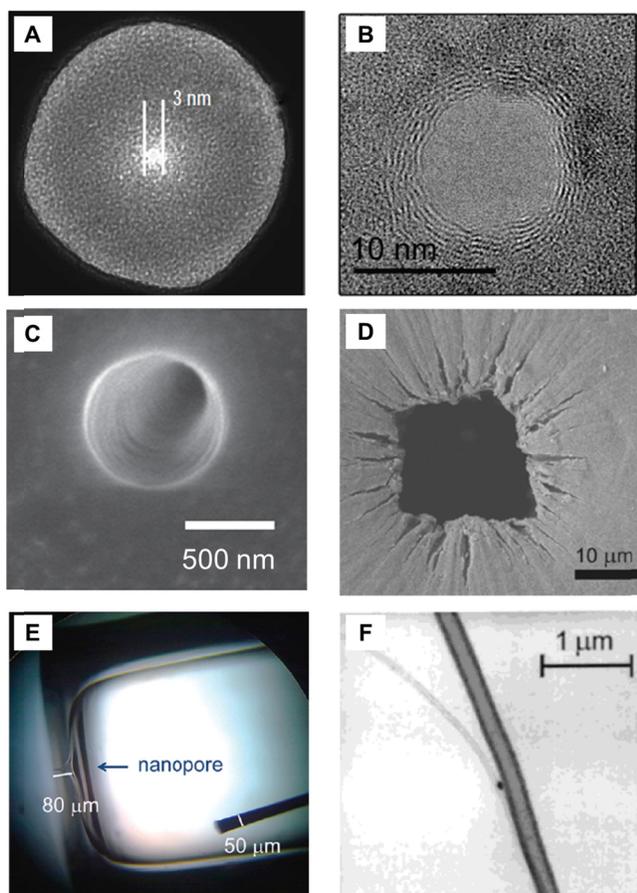


Fig. 2. Nanopores for resistive-pulse sensing. The variety of fabrication techniques and available substrates permit the pore and its dimensions to be tailored to suit the application of interest. Shown are nanopores in (A) SiN membrane, (B) graphene sheet, (C) poly(ethylene terephthalate) membrane, (D) tunable polyurethane membrane, (E) glass substrate, and (F) carbon nanotube. Panel A adapted with permission from Li *et al.*³⁸ Copyright 2003 Macmillan Publishers Ltd. Panel B adapted with permission from Merchant *et al.*³⁹ Copyright 2010 American Chemical Society. Panel D adapted with permission from Vogel *et al.*⁴⁰ Copyright 2011 American Chemical Society. Panel E adapted with permission from Lan *et al.*²⁸ Copyright 2012 American Chemical Society.

Panel F adapted with permission from Sun *et al.*³⁴ Copyright 2000 American Chemical Society.

Solid state nanopores are typically fabricated in Si or SiN membranes with the advantages of ease of handling, well-characterized material properties, and well-established fabrication protocols. Electron and ion beams provide fine control over nanopore diameter. SiN pores with single nanometre resolution are fabricated with a focused ion beam (FIB) instrument, and pore size is reduced by ion beam exposure due to a thermal annealing process.⁴¹ The size of ion-beam milled pores can also be tuned by electron-beam (e-beam) assisted deposition of SiO₂ in which the electron beam catalyses a reaction between a precursor gas and a SiN substrate containing a nanopore.⁴² This fabrication method can be easily scaled up. An array of nanopores drilled in a SiN membrane with a TEM beam record current measurements in parallel.⁴³

SiN pores are used to analyse DNA, and DNA conformation is determined from the magnitude of current displacement in multi-level events.³⁸ SiN pores are also fabricated by TEM drilling for rapid analysis of pM concentrations of DNA, assisted by a salt gradient across the pore.⁴⁴ Additionally, TEM-drilled pores permit control over pore thickness for discrimination between small nucleic acids.⁴⁵ Of particular interest is a SiN pore coated with a lipid bi-layer. Ligands incorporated into the bi-layer interact with targeted proteins to slow their transport and to increase the resolution between individual events.⁴⁶ Additional studies include antibiotic/RNA interactions,⁴⁷ protein unfolding studies,⁴⁸ and sizing of polystyrene nanoparticles.⁴⁹

Solid state nanopores are also created in SiO₂ membranes. The diameter of SiO₂ pores is tuned by imaging with a TEM to promote oxide growth and create 10-nm diameter pores for analysis of ss-DNA and double-stranded (ds) DNA.^{50,51} The magnitude of current displacement for DNA translocations depends on salt concentration, and an increase in current is observed at low salt concentration, whereas a decrease in current occurs at high salt concentration.⁷ E-beam lithography and wet chemical etching are used to create a pyramidal pore in SiO₂ for simultaneous measurement of electric charge and zeta potential for nanoparticles of varying composition.⁵²

Recently, nanopores are fabricated in graphene membranes. These sensors are being developed to discriminate individual DNA nucleotides, because pores in atomically thin graphene sheets provides higher axial resolution than pores formed in conventional membranes.^{39,53} A stack of alternating graphene and aluminium oxide sheets enables each graphene layer to be biased individually, where local potentials retard the transport of DNA through these pores, thus increasing resolution of individual base pairs.⁵⁴

One limitation of conventional sensors is that pore size must be tuned to the size of the analyte. Polyurethane films are used to

1 create pores with diameters that can be tuned in real time to
2 match analyte size. These membranes are punctured with
3 tungsten needles to create the pore, and then pressure is applied
4 to the membrane which causes the pore to stretch or relax as a
5 function of the pressure. Pore diameters range from hundreds of
6 nanometres to tens of microns. Accurate sizing of polymer
7 nanoparticles and Adenovirus particles is accomplished, as
8 confirmed by light scattering and TEM measurements.⁴⁰
9 Mobility of polystyrene nanoparticles is measured as a function
10 of nanopore size and pH for analysis of pulse shape.⁵⁵ Tuneable
11 pores in commercially available nanopore sensors are shown to
12 detect a wide size range of polystyrene beads⁵⁶ and have also
13 monitored the binding of DNA to streptavidin-functionalized
14 beads.⁵⁷

15 16 17 **Resistive-Pulse Sensing with In-Plane Nanopores**

18 Integration of in-plane nanopores into microfluidic devices
19 enhances device portability and ease of use, improves mass
20 transfer of the analyte to the pore, lowers sample consumption,
21 and allows integration of enhanced functions. Furthermore,
22 pore geometry is directly characterized, and experiments can
23 also be monitored optically. Planar pores are fabricated in glass,
24 silicon, or polymers with fabrication processes tuned to suit the
25 application of interest.
26

27
28 FIB techniques provide precise fabrication of sub-50 nm
29 features.⁵⁸ Nanofluidic channels are patterned directly into
30 silicon⁵⁹ or quartz⁶⁰ substrates to study the transport dynamics
31 of λ -DNA. Low surface roughness and uniformity of these
32 channels permit a thorough investigation of DNA mobility as a
33 function of potential, measuring the energy barrier to elongate
34 DNA strands as they enter confined space. FIB-milled channels
35 are also replicated in soft polymers, e.g., poly(dimethyl-
36 siloxane) (PDMS), such that a single master may yield multiple
37 replicas that are used to track DNA strands with a high degree
38 of device-to-device reproducibility.⁶¹ In addition, the
39 deformability of the polymer channels permits the channel
40 cross-section to be dynamically controlled during the
41 experiment by the application of mechanical force to the
42 substrate.⁶² Compression of the channel cross-section slows
43 DNA translocation by an order of magnitude with implications
44 for increasing measurement sensitivity towards discrimination
45 of DNA nucleotides.
46

47 Traditional lithographic processes pattern in-plane features that
48 can be hundreds to thousands of nanometres in width. E-beam
49 lithography is used to write metal lines, which are subsequently
50 replicated in PDMS.⁶³ These devices provide hours of
51 measurement time, and the effect of driving potential on DNA
52 conformation during translocation is studied. Multi-level events
53 are correlated to the folded shapes of the DNA strands. PDMS-
54 replicated nanopores are also used to perform immunoassays in
55 series and parallel. A protein, such as streptavidin, is attached
56 to a latex colloid.⁶⁴ Binding of anti-streptavidin increases the
57 diameter of the particle, and this size change is easily detected.

Furthermore, channels can be patterned in parallel and allows
for several immunoassays to be performed simultaneously.⁶⁵ A
microfluidic device in PDMS combines a nanopore with a
fluidic balancing arm and is used for high-throughput analysis
of nanoparticles and bacteriophage T7.⁶⁶ Accurate sub-100 nm
particle sizing is accomplished at a rate of 500,000 particles per
second.

Three-dimensional pores are also incorporated into planar
devices. A glass nanopipette is inserted into a PDMS device for
detection of λ -DNA folding inside the pore and probes the
effects of voltage and salt concentration on the translocation
pathways.^{67,68} Glass devices exhibit a high tolerance to pH and
electrical potential, and the close proximity of the pipette to a
glass cover slip enables simultaneous optical imaging.
Nanopipette sensors have also been used to detect latex⁶⁹ and
polystyrene nanoparticles.⁷⁰

In-plane sensors may be scaled up to micrometre dimensions,
such that cells or bacteria can be studied. A simple PDMS
micropore device tracks *E. coli*, allows accurate measurement
of the number and size of bacteria, and may hold promise for
early detection of food and waterborne diseases.⁷¹ In addition,
the lithographic process used to create these pores permits
integration of sensing electrodes. Microchannels are filled with
a liquid gallium-indium alloy that creates an electrode surface
when cured. These devices are used for experiments at low
frequencies, as the enhanced surface area alleviates polarization
effects.⁷² Recently, a planar PDMS micropore is used to detect
microparticles functionalized with antibodies.⁷³

52 53 54 **Multiple Resistive-Pulse Measurements on Single Particles**

55 A particle can be driven back and forth through a single pore by
56 reversal of the applied pressure or potential to allow multiple
57 measurements to be made on a single particle. Measurement
58 precision is increased, and the time between events yields
59 additional information. This principle is first demonstrated by
60 passing 27- μm diameter polystyrene beads back and forth
through a glass micropipette.⁷⁴ The same device is used to
measure the dissolution rate of air bubbles and the effect of
bovine serum albumin on surface tension of the bubble.⁷⁵
Multiple measurements made on ss-DNA with a PDMS
nanopore increases measurement precision.⁷⁶ Recapture of ss-
DNA is accomplished with a solid state nanopore for dynamic
information of strand conformation.⁷⁷ Nanoparticle dynamics
are measured by multiple passes through a single glass
nanopipette⁷⁸ and a single track-etched pore in a PET
membrane.⁷⁹

When multiple pores are arranged in series, a single particle is
measured multiple times without the need to pass the particle
back and forth through a single pore. Two different types of
multi-pore devices are shown in Fig. 3, one fabricated with out-
of-plane pores and one fabricated with in-plane pores. In Fig.

Analyst

3A, traditional, out-of-plane pores are used. A SiN/Si bilayer is fabricated by e-beam lithography and wet chemical etching, such that two pores of comparable diameter are stacked in series.⁸⁰ DNA is trapped in the area between the pores, and diffusion out of the device can be precisely monitored. The same device also tracks DNA time-of-flight from the first pore to the second pore.⁸¹ DNA conformation can be tracked by comparing changes in pulse amplitudes at both pores, indicating conformational changes that take place during translocation.

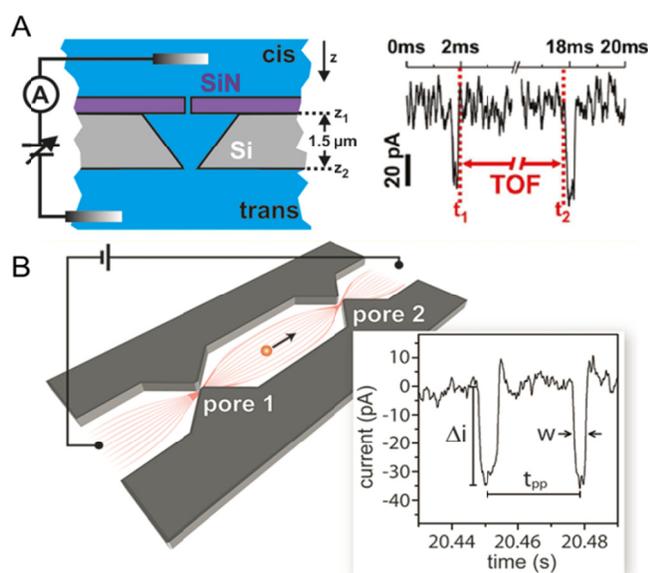


Fig 3. Resistive-pulse sensing with two nanopores in series. (A) Two solid state pores are stacked in a pore-cavity-pore configuration for measuring DNA time-of-flight (TOF). A single event is shown on the right, where two pulses are observed for the transit of a single DNA strand, and the time between pulses is the transit time between pores. (B) Schematic of an in-plane nanochannel device with two pores in series fabricated in SiO₂. Two pores measure the transit of single Hepatitis B Virus capsids, where each capsid is measured twice, as shown on the current trace on the right. Pulse amplitude (Δi) is proportional to capsid size, and the time between pulses (t_{pp}) is used to calculate the electrophoretic mobility of the capsid. Panel A adapted with permission from Langecker *et al.*⁸¹ Copyright 2011 American Chemical Society. Panel B adapted with permission from Harms *et al.*⁸² Copyright 2011 American Chemical Society.

One key advantage of in-plane nanopore devices is the freedom to control the number and arrangement of sensing nanopores within a device. Two nanopores in series are fabricated by e-beam lithography and reactive-ion etching in SiO₂ to measure the electrophoretic mobility of Hepatitis B Virus (HBV) capsids (Fig. 3B).⁸² Capsid velocity is shown to increase linearly with field strength and is used to calculate the electrokinetic mobility of the capsids. Nanochannels with two pores in series are milled with an FIB instrument to form channels and pores with three-dimensional topography.⁸³ These devices track both the size and mobility of individual T = 3 and T = 4 HBV capsids and are well-suited to study virus capsid assembly in which the assembly process can be captured with single particle resolution and in real time.

Another serial sensing design uses a long nanochannel that features an arbitrary pattern of expansions (or nodes) that is encoded on each current pulse.⁸⁴ With this 'node-pore' technique, the current trace exhibits a local minimum for each constriction encountered and a local maximum for each expansion. Unique current signatures permit identification of events with a low signal-to-noise ratio, expanding the useful dynamic range of the experiment.

As with single-pore sensors, multi-pore techniques can also be scaled up to interrogate cells. A series of 5 to 10 vertical electrode pairs are patterned along the length of a PDMS microchannel, and multiple impedance measurements are made on latex microparticles and fibroblast cells.^{85,86} A passive mixing region and dielectrophoretic focusing of the analyte prior to detection highlight the capability of microfluidic devices that incorporate additional on-device functions.

Transverse Conductivity Sensing

Efforts to sequence the human genome more rapidly and cost effectively have inspired new electrical sensing strategies that couple transverse conductance measurements with nanopore devices. Such devices incorporate electrodes perpendicular to the translocation pathway that measure the transverse current as a function of time. Two nanoelectrodes embedded in a nanopore and functioning as conducting probes are predicted to differentiate individual nucleotides.⁸⁷ Similar theories have suggested this method could significantly reduce sequencing times⁸⁸ and have high sensitivity.^{89,90}

Transverse sensing devices consist of exposed nanogap electrodes separated by 1–2 nm within a nanopore (Fig. 4). Several fabrication methods are used to create pores: FIB milling through metal⁹¹ and carbon nanotubes,⁹² nanofabricated mechanically controllable break junctions,⁹³ and TEM drilling.^{94–96} Experimentally, analytes are detected as they translocate through the pore by monitoring the tunnelling current across the nanogap electrodes as a function of time. Similar to sensing ionic current, translocation events are observed as pulses in a current trace. Tunnelling current through a 0.55 nm diameter gold nanojunction has been used to experimentally differentiate between single amino acids.⁹⁷

Sensing platforms are also designed to alleviate the need for a nanopore size-matched to the analyte of interest, easing fabrication constraints and increasing the detectable range of particle sizes.⁹⁸ Similar to the transverse conductivity sensing with nanoelectrodes, straight nanofluidic channels through which molecules transit are bisected by a conductivity sensor. Nanogap electrodes within a nanofluidic channel are capable of sensing ds-DNA when the electrode gap is < 13 nm.⁹⁹ Another device consists of a nanochannel milled by an FIB instrument through a metal nanoelectrode.¹⁰⁰ Recently, two FIB-milled nanochannels are designed to intersect and sense DNA translocation.¹⁰¹ Unlike the other sensors, this particular device

monitors the ionic conductivity across a short nanochannel that intersects a longer nanochannel through which the DNA strands pass. Nanogap electrodes adjacent to TEM-drilled nanopores also detect single DNA molecules.¹⁰² Simulations suggest that the current signature from each base pair is different, and a bisected nanochannel geometry may be useful for future DNA sequencing.¹⁰³

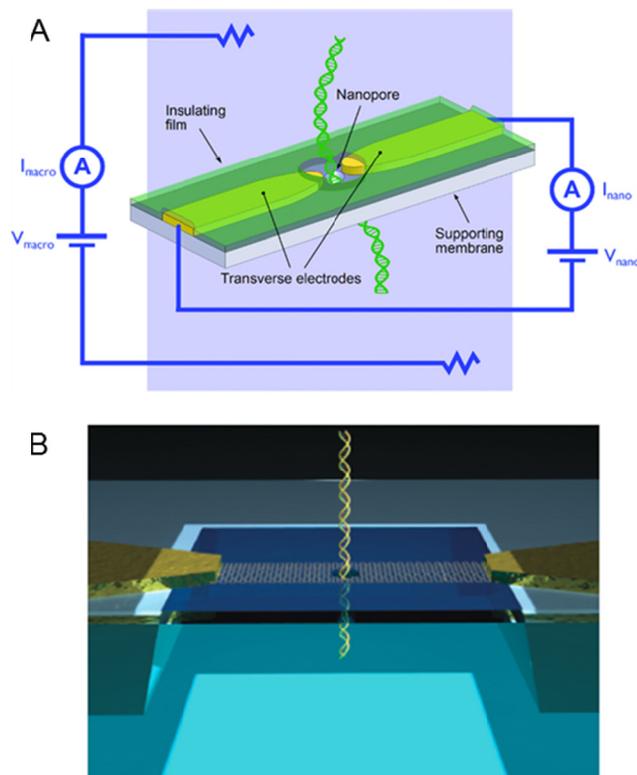


Fig. 4. Principle of transverse conductivity sensing. (A) As DNA is driven through the nanopore, the base pairs modulate the electric field between the transverse electrodes. Pores are often fabricated in SiN, and the metal electrodes are typically gold. (B) Recently, nanopore devices are formed in graphene which acts as both the nanopore substrate and electrode. Panel A adapted with permission from Healy *et al.*⁹⁵ Copyright 2012 John Wiley and Sons, Inc. Panel B adapted with permission from Radenovic *et al.*¹⁰⁴ Copyright 2013 Macmillan Publishers Ltd.

Established sensing methods have reported unique current signatures for specific bases.^{93,105} However, due to the finite size of metal electrodes and ionic channels, single-base resolution of DNA strands has eluded transverse conductivity sensors. Recent simulations suggest that graphene, because of its single atom thickness and high conductance, is capable of single-base resolution by transverse conductivity measurements during DNA translocation through a nanogap in the graphene sheet.¹⁰⁶⁻¹⁰⁸ These initial models conclude that the differing structures of each nucleotide is sufficient for base differentiation,¹⁰⁷ irrespective of the orientation of the nucleotide within the pore.¹⁰⁶ Several studies verify that atomically thin graphene is capable of sensing DNA by axial conductance^{39,53,109} as well as by lateral conductance.¹⁰⁴ These results suggest next-generation devices with graphene sheets

may be capable of both longitudinal sensing of DNA translocation and single-base discrimination by transverse conductivity measurement.

Ion Current Rectification

Ion current rectification (ICR) results from an uneven flux of cations and anions through nanofluidic structures with an asymmetry in surface charge or geometry^{110,111,112} (Fig. 5) or through physical blockages.^{113,114} Rectification is manifested as high and low conductance states that approximate the 'on' and 'off' behaviour of electronic diodes, respectively.^{115,116,117} The magnitude of ICR is reported as the rectification ratio (RR), which is the absolute value of the ion current at one potential (typically -1 V) divided by the ion current at an equal but opposite potential (+1 V), i.e., $RR = |I_- / I_+|$. A $RR > 1$ is typically observed when the surface charge is negative and, conversely, $RR < 1$ is commonly observed when the surface charge is positive. The degree of rectification also depends on the identity of the cations in solution and is shown experimentally to increase with increasing cation diameter.¹¹⁸ ICR is highly dependent on local surface charge,¹¹⁹ which can be controlled by pH and surface functionalization to build nanofluidic sensors with a high degree of sensitivity and selectivity. Furthermore, diode-like behaviour permits use of nanofluidics in simple logic circuits for enhanced device functionality.^{120,121} Additional terminals, e.g., electrodes or fluidic channels, can further modulate ion current for transistor-like behaviour.

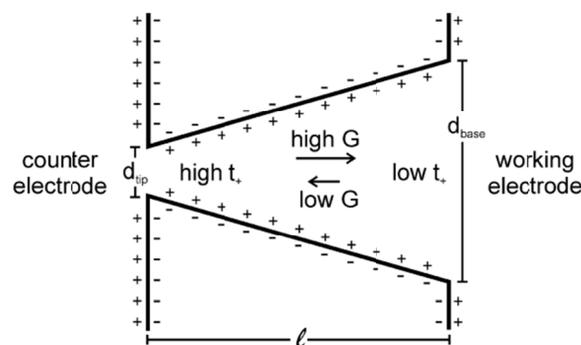


Fig. 5. Schematic of an asymmetric nanopore with negative surface charge. Arrows show direction of cation flux in high and low conductance states (G), which, when present, leads to ion current rectification. Figure adapted with permission from Kovarik *et al.*¹¹² Copyright 2009 American Chemical Society.

To regulate the rectification of ion current, conical polymer nanopores are functionalized with groups that respond to light or heat. Temperature-responsive polymers attached to the surface of a polyimide pore shrink or swell to control ion current.¹²² Current decreases as temperature increases due to collapse of the polymer brushes within the pore. Nanopores are modified with copolymer brushes that are responsive to both temperature and pH for independent control of ion flux and rectification ratio.¹²³ Photolabile protecting groups that act to

Analyst

neutralize surface charge are removed by UV radiation to return the nanopore to a charged state by an 'off' to 'on' activation.¹²⁴ UV radiation is also used to control pore size in a membrane that is modified with C4-DNA motors, due to a pH triggered conformational change initiated by photo-induced OH⁻ ions.¹²⁵ Light activation is combined with pH-modulation in a pore modified with spiropyran, where irradiation creates a negatively charged surface that is tuned by solution pH.¹²⁶

The magnitude and direction of rectification depends strongly on surface charge density and polarity, which are conveniently manipulated by solution pH. Unmodified polymer nanopores have a carboxyl groups on the surface with an isoelectric point around pH 3, where no rectification is observed. At a pH below the isoelectric point, RR < 1, whereas at a pH above the isoelectric point, RR > 1, due to the resulting net positive or negative surface charge, respectively.¹²⁷ Surface modification of polymer pores with L-lysine,^{128,129} L-histidine,¹²⁹ and phosphate brushes¹³⁰ shifts the isoelectric point and, consequently, the pH-dependent behaviour. Conical glass nanopores are also modified with an aminated silane^{131,132} and amphoteric chitosan¹³³ for pH-dependent rectification.

Rectification-Based Sensing

The surfaces of nanopores can be functionalized with groups that bind ions to disrupt surface charge and, in turn, the degree of rectification. Pores fabricated in polymer membranes, such as poly(ethylene) terephthalate (PET), contain carboxyl groups on their surface, that may be functionalized. These systems provide a biomimetic platform for studying ion channel regulation. Polymer pores functionalized with G4 DNA respond to K⁺ by undergoing a conformational change.¹³⁴ With K⁺ in the pore, the DNA folds into a densely packed structure that blocks ion current. Ca²⁺ and cobalt sepulchrate invert rectification on a negatively charged pore,¹³⁵ and Ca²⁺ binds to a phosphonic acid-modified pore for pH dependent rectification as shown in Fig. 6A.¹³⁶ pH-dependent rectification is similarly achieved by modifying PET nanopores with ss-DNA to create electrostatic meshes.¹³⁷ Nanopore surfaces functionalized with zinc finger peptides bind Zn²⁺ to induce a conformation change and cause a fluctuation in ion current.¹³⁸ Glass nanopipettes are modified with imidazole-terminated silanes¹³⁹ to bind Co²⁺ and calcium binding protein (calmodulin)¹⁴⁰ to bind Ca²⁺ and Mg²⁺. Ca²⁺ is also detected with calcein-modified PET pores that exhibit a lower RR in the presence of Ca²⁺ at high pH.¹⁴¹

Pores with asymmetric geometries that exhibit ICR can also be used to study small molecules. ICR increases cation concentration within a polymer nanopore above the bulk solubility product to precipitate CoHPO₄ and CaHPO₄, demonstrating the ability of nanopores to probe kinetics and early stages of crystallization.^{142,143} As the precipitation occurs within the nanopore, millisecond timescales are accessed, and the reversibility of the reaction is highly repeatable to obtain good counting statistics. Quartz nanopores containing

poly(vinylpyridine) modified with benzylboronic acid groups respond to monosaccharide polymers by collapsing to a zwitterionic state, and ICR reverses.¹⁴⁴ A track-etched polymer pore modified with boronic acid reversibly binds saccharides and glycoproteins.¹⁴⁵ Sensors that can accurately determine carbohydrate and glycoprotein concentrations have applications in clinical settings.

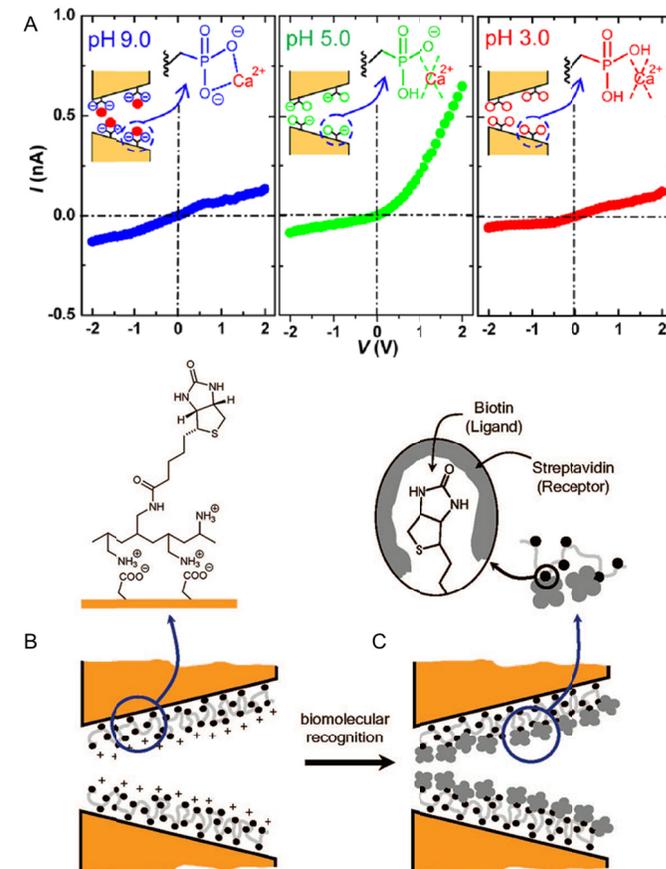


Fig. 6. Two strategies for rectification-based sensing. (A) A polymer-based nanopore is modified with phosphonic polyacid chains. These pores exhibit pH-dependent ion current rectification (ICR) and are used to detect calcium binding. (B-C) A nanopore modified with a carboxylated polymer electrostatically binds biotin ligand. Subsequently, the ligand binds streptavidin, which minimizes the excess surface charge and results in decreased ICR as well as decreased total ion current. Panel A adapted with permission from Ali *et al.*¹³⁶ Copyright 2012 American Chemical Society. Panels B-C adapted with permission from Ali *et al.*¹⁴⁶ Copyright 2008 American Chemical Society.

Pores with modified surfaces are well suited to study binding of biomolecules. Polymer pores coated with Au are readily modified with substituted thiol groups and are used to study the binding of biotin-streptavidin, protein-G-immunoglobulin, and antibody-ricin.¹¹⁹ PET pores are also directly functionalized with biotin to study streptavidin binding.^{146,147,148} An example of rectification-based detection is shown in Fig. 6B, where biotin-streptavidin binding eliminates surface charge on the pore, and binding events are detected as a decrease in ICR.¹⁴⁶ Pores are modified with horseradish peroxidase enzymes that feature sugar groups for monitoring the binding of the lectin protein, concanavalin A.¹⁴⁹ Functionalization of glass and

polymer pores with dendrimers¹⁵⁰ and peptide nucleic acid probes,¹⁵¹ respectively, enables detection of complementary DNA strands with high selectivity. Peptide nucleic acids are DNA analogues containing repeating glycine units bearing nucleotide bases and can be used to discriminate between complementary DNA strands and those containing a single base mismatch.

Addition of an extra terminal perpendicular to a nanofluidic channel is used to modulate ion current for transistor-like behaviour. The nanochannel functions as the source and drain, and the extra metal contact acts as the gate. These devices have been realized with arrays of silica nanochannels and with isolated silica nanotubes.^{152,153,154} Field-effect control of proteins is promising for preconcentration and separation applications. A silicon nanochannel in a transistor circuit is used to concentrate avidin.¹⁵⁵ Nanofluidic transistors are also fabricated in 3D solid state pores¹⁵⁶ for more efficient capture of DNA by reducing electroosmotic flow through application of a gate voltage.¹⁵⁷ These pores are fabricated in SiN membranes that sandwich a metal film and are coated with an oxide layer. Alternatively, an all-silicon device, where the membrane itself is used as the gate electrode, demonstrates that voltage gating enhances the rectification ratio by > 100-fold for high on-off ratios.¹⁵⁸ Manipulation of bovine serum albumin demonstrates the potential of this system to control protein transport.

Conductance-Based Detection

Conductance measurements provide additional means to probe nanofluidic phenomena. Measurements of nanochannel and nanopore conductance are used to characterize device geometry and surface chemistry and to probe surface binding reactions. Furthermore, if conductance is measured as a function of time when driven with AC waveforms, dynamic processes such as reaction kinetics can be tracked.

Nanochannel conductance depends on both channel geometry and surface charge. As the channel depth (or width) decreases and the surface-to-volume ratio increases, channel conductance deviates from bulk conductance due to surface conductance.^{159,160} Device conductance is proportional to number of nanochannels and is found to reach a plateau at low ionic strength due to conductance of surface charge.¹⁶¹ Geometric parameters of solid state nanopores, such as pore diameter, are estimated by conductance measurements.¹⁶²

Electrochemical impedance spectroscopy (EIS) measures the impedance response of nanochannel devices as a function of frequency. EIS spectra taken as a function of solution conductivity are used to model arrays of glass nanocapillaries as equivalent circuit elements for characterization of pore dimensions and the electrical double layer. Equivalent circuit models are useful tools for understanding the behaviour of nanofluidic devices and provide accurate estimates of pore shape and diameter.¹⁶³

EIS is used to investigate the role of surface charge on ion transport through 50-nm deep amorphous-Si/glass channels. Surface charge as a function of salt concentration is measured in a 50-nm wide quartz nanochannel with an embedded transverse Ti/Au electrode, where double layer overlap is observed at low ion concentration.¹⁶⁴ Mass transport through an amine-modified glass nanopore is affected by the surface charge as well as the geometry of the pore itself.¹⁶⁵

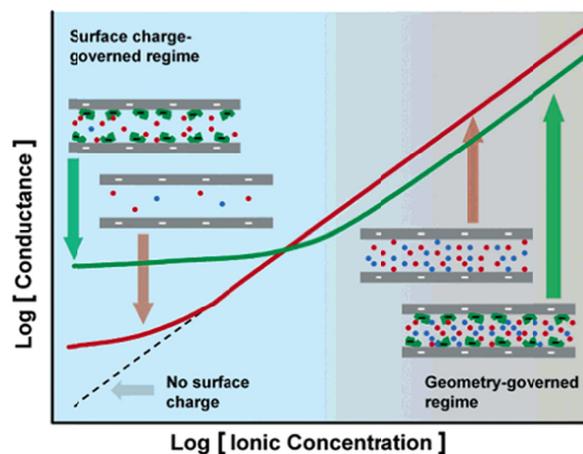


Fig. 7. Conductance-based detection. Nanochannels are biotinylated and bind streptavidin, which leads to a decrease in conductance at high ion concentration and an increase in conductance at low ion concentration. Figure adapted with permission from Karnik et al.¹⁶⁶ Copyright 2005 American Chemical Society

At low ionic strength, nanochannel conductance is dominated by surface charge, and as BSA adsorbs to the channel surface, the channel conductance decreases.¹⁶⁷ An array of 30-nm deep silica nanochannels modified with streptavidin is used to measure avidin binding, in which nanochannel conductance increases with biomolecule charge and decreases with biomolecule volume, as shown in Fig. 7.¹⁶⁶ Binding of streptavidin to a PEG-biotin in a 50-nm deep Pyrex nanochannel is measured under convective flow and shows the ability of nanofluidics to probe diffusion-limited reactions with a significantly reduced response time.¹⁶⁸ An array of nanochannels in a PET membrane is modified with streptavidin for monitoring T4 polynucleotide kinase activity through a cleavage reaction with a biotin-labelled ds-DNA probe.¹⁶⁹

Simultaneous Optical and Electrical Measurements

Despite the high sensitivity of conductivity-based sensing methods, detection of analytes of interest is often performed blindly. To overcome this limitation, optical measurements made simultaneously with electrical measurements of particles provide additional and complementary information that electrical measurements alone cannot provide. Tracking fluorescently tagged particles permits unambiguous correlation of a deviation in baseline current to a fluorescent signal, verifying the passing of a particle through the sensing region. Optical detection also provides information about analyte

Analyst

1 movement within the sensing region, which may explain the
2 variation between individual current pulses. Particle trajectory
3 is also monitored to improve device design for optimized
4 sensing.
5

6 Simultaneous sensing of ds-DNA molecules transiting a 4-nm
7 SiN pore is performed in a home-built cell by total internal
8 reflectance fluorescence microscopy.¹⁷⁰ Translocation events
9 are monitored by electrical measurements made across the pore
10 showing typical current deviations at the same time
11 fluorescence spikes are detected. Similar measurements are
12 made by confocal fluorescence microscopy in three dimensions
13 to visualize the motion of DNA through a SiN pore embedded
14 into a microfluidic device¹⁷¹ while current is simultaneously
15 measured to confirm the passing of the particle. Fluorescence
16 detection of DNA is also performed with in-plane FIB-milled
17 nanochannels¹⁰¹ to verify that deviations in current are caused
18 by transiting DNA molecules.
19

20 Carbon nanotubes positioned between two reservoirs are used
21 to detect single dye molecules by measuring the ionic current
22 passing through the tube as a function of time.¹⁷² Electrical
23 detection of a molecule is again measured as a deviation from
24 the baseline current, and optical detection is made possible by a
25 laser focused just past the end of the tube that excites the dye
26 exiting the tube. Simultaneous fluorescence and conductivity
27 measurements of small, fluorescently tagged polystyrene beads
28 transiting through a SiO₂-PDMS pore correlate the dependence
29 of the electrical peak shape to the characteristic shape of the
30 pore.¹⁷³
31
32

33 Conclusions and Outlook

34 Advances in fabrication techniques have driven research in
35 nanofluidics over the past decade. Conductivity measurements
36 offer a simple, yet powerful means to investigate nanoscale
37 systems. These measurements have the added benefit of being
38 label-free and can detect a wide variety of analyte types and
39 sizes. Furthermore, nanofluidic phenomena are often
40 exploited to enhance sensing by selective transport and pre-
41 concentration.
42

43 The ability to fabricate nanopores with dimensions similar to an
44 analyte of interest makes them appealing for resistive-pulse
45 sensing. One or more nanopores may be used to sense the
46 transport of single particles by the change in resistance upon
47 nanopore translocation. Nanopores can also be used to measure
48 single particles multiple times, and multiple nanopores can be
49 arranged in series to obtain more information and increase
50 measurement precision. Electrodes are added to nanochannels
51 or nanopores for transverse conductivity measurements,
52 providing extra detection functionality. Metal or fluidic
53 electrodes are used to monitor changes in current as analyte
54 passes through a nanogap. Enhanced spatial resolution over
55 traditional resistive-pulse sensing measurements is obtained
56 with the potential application of DNA sequencing. A simple
57
58
59
60

bulk conductance measurement in nanochannels is also used to
probe analyte binding to nanochannel walls, and time-
dependant signals are used to characterize device geometry.

Nanoscale conduits with an asymmetry in geometry or surface
charge may be exploited for ion current rectification-based
sensing. Surface charge is modulated by pH and adsorption for
control over rectification, and functional groups are used to
decorate the surface for specific interactions with target
analytes. Finally, optical measurements are used to complement
electrical measurements.

Conductivity-based sensing methods are poised to become cost-
effective bioanalytical tools and will continue to develop as
nanofabrication techniques improve. Future generations of
conductivity sensors will build upon the work described above
in order to improve detection efficiency and address a broad
range of analytical applications. A number of challenges
currently face electrical sensing technologies and include
increasing device-to-device reproducibility, decreasing
fabrication costs, and improving detection sensitivity. As these
challenges are being met, conductivity-based detection
becomes a practical and cost-effective alternative for a number
of applications. Most notably, the development of reliable lab-
on-a-chip devices capable of quickly and inexpensively
analysing nanoparticles and DNA are an exciting breakthrough
in the field. Commercial nanopore systems have been realized
for DNA analysis (www.nanoporetech.com and
www.nabsys.com), high throughput nanoparticle counting
(www.spectradynellc.com), and extended dynamic range
particle sizing (www.izon.com).

Acknowledgements

This work was supported in part by NIH R01 GM100071 and
NSF CHE-1308484.

Notes and References

Department of Chemistry, Indiana University, Bloomington, IN 47405

1. S. Howorka and Z. Siwy, *Chem. Soc. Rev.*, 2009, **38**, 2360-2384.
2. A. Oukhaled, L. Bacri, M. Pastoriza-Gallego, J.-M. Betton and J. Pelta, *ACS Chem. Bio.*, 2012, **7**, 1935-1949.
3. W. Sparreboom, A. van den Berg and J. C. T. Eijkel, *Nat. Nanotech.*, 2009, **4**, 713-720.
4. R. B. Schoch, J. Han and P. Renaud, *Rev. Mod. Phys.*, 2008, **80**, 839-883.
5. R. W. Deblois, C. P. Bean and R. K. A. Wesley, *J. Colloid Interface Sci.*, 1977, **61**, 323-335.
6. M. Davenport, K. Healy, M. Pevarnik, N. Teslich, S. Cabrini, A. P. Morrison, Z. S. Siwy and S. E. Letant, *ACS Nano*, 2012, **6**, 8366-8380.
7. R. M. M. Smeets, U. F. Keyser, D. Krapf, M. Y. Wu, N. H. Dekker and C. Dekker, *Nano Lett.*, 2006, **6**, 89-95.
8. J. Menestrina, C. Yang, M. Schiel, I. Vlasiouk and Z. S. Siwy, *J. Phys. Chem. C*, 2014, **118**, 2391-2398.
9. M. Pevarnik, K. Healy, M. E. Toimil-Molares, A. Morrison, S. E. Letant and Z. S. Siwy, *ACS Nano*, 2012, **6**, 7295-7302.
10. E. A. Heins, Z. S. Siwy, L. A. Baker and C. R. Martin, *Nano Lett.*, 2005, **5**, 1824-1829.

11. L. T. Sexton, H. Mukaibo, P. Katira, H. Hess, S. A. Sherrill, L. P. Horne and C. R. Martin, *J. Am. Chem. Soc.*, 2010, **132**, 6755-6763.
12. O. A. Saleh and L. L. Sohn, *Rev. Sci. Instrum.*, 2002, **73**, 4396-4398.
13. *U.S.A. Pat.*, U.S. patent 2656508, 1953.
14. W. H. Coulter, Proceedings of the National Electronics Conference, Chicago, 1957.
15. H. E. Kubitschek, *Nature*, 1958, **182**, 234-235.
16. R. Smither, *J. Appl. Bacteriol.*, 1975, **39**, 157-165.
17. M. Marković and L. Komunjer, *J. Cryst. Growth*, 1979, **46**, 701-705.
18. R. L. Fleischer, P. B. Price and E. M. Symes, *Science*, 1964, **143**, 249-250.
19. R. W. DeBlois and C. P. Bean, *Rev. Sci. Instrum.*, 1970, **41**, 909.
20. R. W. DeBlois and R. K. A. Wesley, *J. Virol.*, 1977, **23**, 227-233.
21. P. Y. Apel, Y. E. Korchev, Z. Siwy, R. Spohr and M. Yoshida, *Nucl. Instrum. Methods Phys. Res. B*, 2001, **184**, 337-346.
22. A. Mara, Z. Siwy, C. Trautmann, J. Wan and F. Kamme, *Nano Lett.*, 2004, **4**, 497-501.
23. J. E. Wharton, P. Jin, L. T. Sexton, L. P. Horne, S. A. Sherrill, W. K. Mino and C. R. Martin, *Small*, 2007, **3**, 1424-1430.
24. L. T. Sexton, L. P. Horne, S. A. Sherrill, G. W. Bishop, L. A. Baker and C. R. Martin, *J. Am. Chem. Soc.*, 2007, **129**, 13144-13152.
25. K. M. Zhou, L. C. Li, Z. N. Tan, A. Zlotnick and S. C. Jacobson, *J. Am. Chem. Soc.*, 2011, **133**, 1618-1621.
26. J. D. Uram, K. Ke and M. Mayer, *ACS Nano*, 2008, **2**, 857-872.
27. W.-J. Lan, D. A. Holden, J. Liu and H. S. White, *J. Phys. Chem. C*, 2011, **115**, 18445-18452.
28. W.-J. Lan and H. S. White, *ACS Nano*, 2012, **6**, 1757-1765.
29. D. A. Holden, G. Hendrickson, L. A. Lyon and H. S. White, *J. Phys. Chem. C*, 2011, **115**, 2999-3004.
30. M. Karhanek, J. T. Kemp, N. Pourmand, R. W. Davis and C. D. Webb, *Nano Lett.*, 2005, **5**, 403-407.
31. J. D. Uram, K. Ke, A. J. Hunt and M. Mayer, *Angew. Chem. Int. Ed.*, 2006, **45**, 2281-2285.
32. C. Gao, S. Ding, Q. Tan and L.-Q. Gu, *Anal. Chem.*, 2009, **81**, 80-86.
33. S. Umehara, M. Karhanek, R. W. Davis and N. Pourmand, *Proc. Natl. Acad. Sci. U. S. A.*, 2009, **106**, 4611-4616.
34. L. Sun and R. M. Crooks, *J. Am. Chem. Soc.*, 2000, **122**, 12340-12345.
35. T. Ito, L. Sun, R. R. Henriquez and R. M. Crooks, *Acc. Chem. Res.*, 2004, **37**, 937-945.
36. T. Ito, L. Sun, M. A. Bevan and R. M. Crooks, *Langmuir*, 2004, **20**, 6940-6945.
37. H. Liu, J. He, J. Tang, H. Liu, P. Pang, D. Cao, P. Krstic, S. Joseph, S. Lindsay and C. Nuckolls, *Science*, 2010, **327**, 64-67.
38. J. Li, M. Gershow, D. Stein, E. Brandin and J. A. Golovchenko, *Nat. Mater.*, 2003, **2**, 611-615.
39. C. A. Merchant, K. Healy, M. Wanunu, V. Ray, N. Peterman, J. Bartel, M. D. Fischbein, K. Venta, Z. T. Luo, A. T. C. Johnson and M. Drndić, *Nano Lett.*, 2010, **10**, 2915-2921.
40. R. Vogel, G. Willmott, D. Kozak, G. S. Roberts, W. Anderson, L. Groenewegen, B. Glossop, A. Barnett, A. Turner and M. Trau, *Anal. Chem.*, 2011, **83**, 3499-3506.
41. J. Li, D. Stein, C. McMullan, D. Branton, M. J. Aziz and J. A. Golovchenko, *Nature*, 2001, **412**, 166-169.
42. C. Danelon, C. Santschi, J. Brugger and H. Vogel, *Langmuir*, 2006, **22**, 10711-10715.
43. M. J. Kim, M. Wanunu, D. C. Bell and A. Meller, *Adv. Mater.*, 2006, **18**, 3149-3153.
44. M. Wanunu, W. Morrison, Y. Rabin, A. Y. Grosberg and A. Meller, *Nat. Nanotechnol.*, 2010, **5**, 160-165.
45. M. Wanunu, T. Dadosh, V. Ray, J. Jin, L. McReynolds and M. Drndić, *Nat. Nanotechnol.*, 2010, **5**, 807-814.
46. E. C. Yusko, J. M. Johnson, S. Majd, P. Prangkio, R. C. Rollings, J. Li, J. Yang and M. Mayer, *Nat. Nanotechnol.*, 2011, **6**, 253-260.
47. M. Wanunu, S. Bhattacharya, Y. Xie, Y. Tor, A. Aksimentiev and M. Drndić, *ACS Nano*, 2011, **5**, 9345-9353.
48. K. J. Freedman, M. Jurgens, A. Prabhu, C. W. Ahn, P. Jemth, J. B. Edell and M. J. Kim, *Anal. Chem.*, 2011, **83**, 5137-5144.
49. M. Tsutsui, S. Hongo, Y. He, M. Taniguchi, N. Gemma and T. Kawai, *ACS Nano*, 2012, **6**, 3499-3505.
50. A. J. Storm, C. Storm, J. H. Chen, H. Zandbergen, J. F. Joanny and C. Dekker, *Nano Lett.*, 2005, **5**, 1193-1197.
51. A. J. Storm, J. H. Chen, H. W. Zandbergen and C. Dekker, *Phys. Rev. E*, 2005, **71**, 051903.
52. N. Arjmandi, W. Van Roy, L. Lagae and G. Borghs, *Anal. Chem.*, 2012, **84**, 8490-8496.
53. G. F. Schneider, S. W. Kowalczyk, V. E. Calado, G. Pandraud, H. W. Zandbergen, L. M. K. Vandersypen and C. Dekker, *Nano Lett.*, 2010, **10**, 3163-3167.
54. B. M. Venkatesan, D. Estrada, S. Banerjee, X. Jin, V. E. Dorgan, M.-H. Bae, N. R. Aluru, E. Pop and R. Bashir, *ACS Nano*, 2012, **6**, 441-450.
55. G. R. Willmott and B. E. T. Parry, *J. Appl. Phys.*, 2011, **109**, 094307.
56. J. A. Eldridge, G. R. Willmott, W. Anderson and R. Vogel, *J. Colloid Interface Sci.*, 2014, **429**, 45-52.
57. E. R. Billinge, M. Broom and M. Platt, *Anal. Chem.*, 2014, **86**, 1030-1037.
58. L. D. Menard and J. M. Ramsey, *Nano Lett.*, 2011, **11**, 512-517.
59. L. C. Campbell, M. J. Wilkinson, A. Manz, P. Camilleri and C. J. Humphreys, *Lab Chip*, 2004, **4**, 225-229.
60. L. D. Menard and J. M. Ramsey, *Anal. Chem.*, 2013, **85**, 1146-1153.
61. P. Fanzio, V. Mussi, C. Manneschi, E. Angeli, G. Firpo, L. Repetto and U. Valbusa, *Lab Chip*, 2011, **11**, 2961-2966.
62. P. Fanzio, C. Manneschi, E. Angeli, V. Mussi, G. Firpo, L. Ceseracciu, L. Repetto and U. Valbusa, *Sci. Rep.*, 2012, **2**, 791.
63. Y.-H. Sen and R. Karnik, *Anal. Bioanal. Chem.*, 2009, **394**, 437-446.
64. O. A. Saleh and L. L. Sohn, *Proc. Natl. Acad. Sci. U. S. A.*, 2003, **100**, 820-824.
65. A. Carbonaro and L. L. Sohn, *Lab Chip*, 2005, **5**, 1155-1160.
66. J.-L. Fraikin, T. Teesalu, C. M. McKeeney, E. Ruoslahti and A. N. Cleland, *Nat. Nanotech.*, 2011, **6**, 308-313.
67. L. J. Steinbock, O. Otto, C. Chimere, J. Gornall and U. F. Keyser, *Nano Lett.*, 2010, **10**, 2493-2497.
68. L. J. Steinbock, A. Lucas, O. Otto and U. F. Keyser, *Electrophoresis*, 2012, **33**, 3480-3487.
69. P. Terejanszky, I. Makra, P. Fuerjes and R. E. Gyurcsanyi, *Anal. Chem.*, 2014, **86**, 4688-4697.
70. W.-J. Lan, C. Kubeil, J.-W. Xiong, A. Bund and H. S. White, *J. Phys. Chem. C*, 2014, **118**, 2726-2734.
71. J. Sun, Y. Kang, E. M. Boczeko and X. Jiang, *Electroanalysis*, 2013, **25**, 1023-1028.
72. A. L. Richards, M. D. Dickey, A. S. Kennedy and G. D. Buckner, *J. Micromech. Microeng.*, 2012, **22**, 115012.
73. Y. Han, H. Wu, F. Liu, G. Cheng and J. Zhe, *Anal. Chem.*, 2014, **86**, 9717-9722.
74. L. I. Berge, J. Feder and T. Jøssang, *Rev. Sci. Instrum.*, 1989, **60**, 2756-2763.
75. L. I. Berge, *J. Colloid Interface Sci.*, 1990, **134**, 548-562.
76. Y.-H. Sen, T. Jain, C. A. Aguilar and R. Karnik, *Lab Chip*, 2012, **12**, 1094-1101.
77. M. Gershow and J. A. Golovchenko, *Nat. Nanotechnol.*, 2007, **2**, 775-779.
78. S. R. German, L. Luo, H. S. White and T. L. Mega, *J. Phys. Chem. C*, 2013, **117**, 703-711.
79. M. Schiel and Z. S. Siwy, *J. Phys. Chem. C*, 2014, **118**, 19214-19223.
80. D. Pedone, M. Langecker, G. Abstreiter and U. Rant, *Nano Lett.*, 2011, **11**, 1561-1567.
81. M. Langecker, D. Pedone, F. C. Simmel and U. Rant, *Nano Lett.*, 2011, **11**, 5002-5007.
82. Z. D. Harms, K. B. Mogensen, P. S. Nunes, K. Zhou, B. W. Hildenbrand, I. Mitra, Z. Tan, A. Zlotnick, J. P. Kutter and S. C. Jacobson, *Anal. Chem.*, 2011, **83**, 9573-9578.
83. Z. D. Harms, D. G. Haywood, A. R. Kneller, L. Selzer, A. Zlotnick and S. C. Jacobson, *Anal. Chem.*, 2015, **87**, 699-705.
84. K. R. Balakrishnan, G. Anwar, M. R. Chapman, N. Trongtuong, A. Kesavaraju and L. L. Sohn, *Lab Chip*, 2013, **13**, 1302-1307.

- 1 85. Y. Wu, J. D. Benson and M. Almasri, *Biomed. Microdevices*, 2012, **14**, 739-750.
- 2 86. Y. Wu, J. D. Benson, J. K. Critser and M. Almasri, *J. Micromech. Microeng.*, 2010, **20**, 085035.
- 3 87. M. Zwolak and M. Di Ventra, *Nano Lett.*, 2005, **5**, 421-424.
- 4 88. J. Lagerqvist, M. Zwolak and M. Di Ventra, *Nano Lett.*, 2006, **6**, 779-782.
- 5 89. M. Krems, M. Zwolak, Y. V. Pershin and M. Di Ventra, *Biophys. J.*, 2009, **97**, 1990-1996.
- 6 90. G. Sigalov, J. Comer, G. Timp and A. Aksimentiev, *Nano Lett.*, 2008, **8**, 56-63.
- 7 91. A. P. Ivanov, E. Instuli, C. M. McGilvery, G. Baldwin, D. W. McComb, T. Albrecht and J. B. Edel, *Nano Lett.*, 2011, **11**, 279-285.
- 8 92. Z. Jiang, M. Mihovilovic, J. Chan and D. Stein, *J. Phys. Condens. Matter*, 2010, **22**, 454114.
- 9 93. M. Tsutsui, S. Rahong, Y. Iizumi, T. Okazaki, M. Taniguchi and T. Kawai, *Sci. Rep.*, 2011, **1**, 46.
- 10 94. B. C. Gierhart, D. G. Flowitt, S. J. Chen, Z. Zhu, D. E. Kotecki, R. L. Smith and S. D. Collins, *Sensors Actuat. B-Chem.*, 2008, **132**, 593-600.
- 11 95. K. Healy, V. Ray, L. J. Willis, N. Peterman, J. Bartel and M. Drndić, *Electrophoresis*, 2012, **33**, 3488-3496.
- 12 96. P. Xie, Q. H. Xiong, Y. Fang, Q. Qing and C. M. Lieber, *Nat. Nanotechnol.*, 2012, **7**, 119-125.
- 13 97. T. Ohshiro, M. Tsutsui, K. Yokota, M. Furuhashi, M. Taniguchi and T. Kawai, *Nat. Nanotechnol.*, 2014, **9**, 835-840.
- 14 98. M. Di Ventra, M. Krems, J. Wilson and Y. V. Pershin, *Methods Mol. Biol.*, 2012, **870**, 149-163.
- 15 99. X. G. Liang and S. Y. Chou, *Nano Lett.*, 2008, **8**, 1472-1476.
- 16 100. T. Maleki, S. Mohammadi and B. Ziaie, *Nanotechnology*, 2009, **20**, 105302.
- 17 101. L. D. Menard, C. E. Mair, M. E. Woodson, J. P. Alarie and J. M. Ramsey, *ACS Nano*, 2012, **6**, 9087-9094.
- 18 102. A. Fanget, F. Traversi, S. Khlybov, P. Granjon, A. Magrez, L. Forró and A. Radenovic, *Nano Letters*, 2013, **14**, 244-249.
- 19 103. M. Di Ventra, *Nanotechnology*, 2013, **24**, 342501.
- 20 104. F. Traversi, C. Raillon, S. M. Benameur, K. Liu, S. Khlybov, M. Tosun, D. Krasnozhan, A. Kis and A. Radenovic, *Nat. Nanotechnol.*, 2013, **8**, 939-945.
- 21 105. M. Tsutsui, M. Taniguchi, K. Yokota and T. Kawai, *Nat. Nanotechnol.*, 2010, **5**, 286-290.
- 22 106. T. Nelson, B. Zhang and O. V. Prezhdo, *Nano Lett.*, 2010, **10**, 3237-3242.
- 23 107. H. W. C. Postma, *Nano Lett.*, 2010, **10**, 420-425.
- 24 108. J. Prasongkit, A. Grigoriev, B. Pathak, R. Ahuja and R. H. Scheicher, *J. Phys. Chem. C*, 2013, **117**, 15421-15428.
- 25 109. S. Garaj, W. Hubbard, A. Reina, J. Kong, D. Branton and J. A. Golovchenko, *Nature*, 2010, **467**, 190-193.
- 26 110. C. Wei, A. J. Bard and S. W. Feldberg, *Anal. Chem.*, 1997, **69**, 4627-4633.
- 27 111. J. Cervera, B. Schiedt, R. Neumann, S. Mafé and P. Ramírez, *J. Chem. Phys.*, 2006, **124**, 104706.
- 28 112. M. L. Kovarik, K. M. Zhou and S. C. Jacobson, *J. Phys. Chem. B*, 2009, **113**, 15960-15966.
- 29 113. C. C. Harrell, P. Kohli, Z. Siwy and C. R. Martin, *J. Am. Chem. Soc.*, 2004, **126**, 15646-15647.
- 30 114. M. Ali, P. Ramirez, S. Nasir, Q.-H. Nguyen, W. Ensinger and S. Mafé, *Appl. Phys. Lett.*, 2014, **104**, 043703.
- 31 115. R. Karnik, C. Duan, K. Castelino, H. Daiguji and A. Majumdar, *Nano Lett.*, 2007, **7**, 547-551.
- 32 116. I. Vlassioux and Z. S. Siwy, *Nano Lett.*, 2007, **7**, 552-556.
- 33 117. R. Yan, W. Liang, R. Fan and P. Yang, *Nano Lett.*, 2009, **9**, 3820-3825.
- 34 118. T. Gamble, K. Decker, T. S. Plett, M. Pevarnik, J.-F. Pietschmann, I. Vlassioux, A. Aksimentiev and Z. S. Siwy, *J. Phys. Chem. C*, 2014, **118**, 9809-9819.
- 35 119. Z. Siwy, E. Heins, C. C. Harrell, P. Kohli and C. R. Martin, *J. Am. Chem. Soc.*, 2004, **126**, 10850-10851.
- 36 120. M. Ali, S. Mafé, P. Ramirez, R. Neumann and W. Ensinger, *Langmuir*, 2009, **25**, 11993-11997.
- 37 121. M. Ali, P. Ramirez, N. Hung Quoc, S. Nasir, J. Cervera, S. Mafé and W. Ensinger, *ACS Nano*, 2012, **6**, 3631-3640.
- 38 122. B. Yameen, M. Ali, R. Neumann, W. Ensinger, W. Knoll and O. Azzaroni, *Small*, 2009, **5**, 1287-1291.
- 39 123. W. Guo, H. Xia, L. Cao, F. Xia, S. Wang, G. Zhang, Y. Song, Y. Wang, L. Jiang and D. Zhu, *Adv. Funct. Mater.*, 2010, **20**, 3561-3567.
- 40 124. M. Ali, S. Nasir, P. Ramirez, I. Ahmed, N. Quoc Hung, L. Fruk, S. Mafé and W. Ensinger, *Adv. Funct. Mater.*, 2012, **22**, 390-396.
- 41 125. L. Wen, J. Ma, Y. Tian, J. Zhai and L. Jiang, *Small*, 2012, **8**, 838-842.
- 42 126. M. Zhang, X. Hou, J. Wang, Y. Tian, X. Fan, J. Zhai and L. Jiang, *Adv. Mater.*, 2012, **24**, 2424-2428.
- 43 127. Z. Guo, J. Wang, J. Ren and E. Wang, *Nanoscale*, 2011, **3**, 3767-3773.
- 44 128. B. Yameen, M. Ali, R. Neumann, W. Ensinger, W. Knoll and O. Azzaroni, *J. Am. Chem. Soc.*, 2009, **131**, 2070-2071.
- 45 129. M. Ali, P. Ramirez, S. Mafé, R. Neumann and W. Ensinger, *ACS Nano*, 2009, **3**, 603-608.
- 46 130. B. Yameen, M. Ali, R. Neumann, W. Ensinger, W. Knoll and O. Azzaroni, *Chem. Comm.*, 2010, **46**, 1908-1910.
- 47 131. J. Liu, D. Wang, M. Kvetny, W. Brown, Y. Li and G. Wang, *Anal. Chem.*, 2012, **84**, 6926-6929.
- 48 132. J. Liu, M. Kvetny, J. Feng, D. Wang, B. Wu, W. Brown and G. Wang, *Langmuir*, 2012, **28**, 1588-1595.
- 49 133. L.-X. Zhang, X.-H. Cao, Y.-B. Zheng and Y.-Q. Li, *Electrochem. Commun.*, 2010, **12**, 1249-1252.
- 50 134. X. Hou, W. Guo, F. Xia, F.-Q. Nie, H. Dong, Y. Tian, L. Wen, L. Wang, L. Cao, Y. Yang, J. Xue, Y. Song, Y. Wang, D. Liu and L. Jiang, *J. Am. Chem. Soc.*, 2009, **131**, 7800-7805.
- 51 135. Y. He, D. Gillespie, D. Boda, I. Vlassioux, R. S. Eisenberg and Z. S. Siwy, *J. Am. Chem. Soc.*, 2009, **131**, 5194-5202.
- 52 136. M. Ali, S. Nasir, P. Ramirez, J. Cervera, S. Mafé and W. Ensinger, *ACS Nano*, 2012, **6**, 9247-9257.
- 53 137. S. F. Buchsbaum, G. Nguyen, S. Howorka and Z. S. Siwy, *J. Am. Chem. Soc.*, 2014, **136**, 9902-9905.
- 54 138. Y. Tian, X. Hou, L. Wen, W. Guo, Y. Song, H. Sun, Y. Wang, L. Jiang and D. Zhu, *Chem. Comm.*, 2010, **46**, 1682-1684.
- 55 139. N. Sa, Y. Fu and L. A. Baker, *Anal. Chem.*, 2010, **82**, 9963-9966.
- 56 140. B. Vilozny, P. Actis, R. A. Seger, Q. Vallmajo-Martin and N. Pourmand, *Anal. Chem.*, 2011, **83**, 6121-6126.
- 57 141. Z. Meng, C. Jiang, X. Li and J. Zhai, *ACS Appl. Mater. Interfaces*, 2014, **6**, 3794-3798.
- 58 142. M. R. Powell, M. Sullivan, I. Vlassioux, D. Constantin, O. Sudre, C. C. Martens, R. S. Eisenberg and Z. S. Siwy, *Nat. Nanotech.*, 2008, **3**, 51-57.
- 59 143. L. Innes, M. R. Powell, I. Vlassioux, C. Martens and Z. S. Siwy, *J. Phys. Chem. C*, 2010, **114**, 8126-8134.
- 60 144. B. Vilozny, A. L. Wollenberg, P. Actis, D. Hwang, B. Singaram and N. Pourmand, *Nanoscale*, 2013, **5**, 9214-9221.
145. N. Quoc Hung, M. Ali, R. Neumann and W. Ensinger, *Sensors Actuat. B-Chem.*, 2012, **162**, 216-222.
146. M. Ali, B. Yameen, R. Neumann, W. Ensinger, W. Knoll and O. Azzaroni, *J. Am. Chem. Soc.*, 2008, **130**, 16351-16357.
147. I. Vlassioux, T. R. Kozel and Z. S. Siwy, *J. Am. Chem. Soc.*, 2009, **131**, 8211-8220.
148. M. Ali, B. Schiedt, R. Neumann and W. Ensinger, *Macromol. Biosci.*, 2010, **10**, 28-32.
149. M. Ali, P. Ramirez, M. N. Tahir, S. Mafé, Z. Siwy, R. Neumann, W. Tremel and W. Ensinger, *Nanoscale*, 2011, **3**, 1894-1903.
150. Y. Fu, H. Tokuhisa and L. A. Baker, *Chem. Comm.*, 2009, 4877-4879.
151. M. Ali, R. Neumann and W. Ensinger, *ACS Nano*, 2010, **4**, 7267-7274.
152. R. Karnik, R. Fan, M. Yue, D. Y. Li, P. D. Yang and A. Majumdar, *Nano Lett.*, 2005, **5**, 943-948.
153. R. Fan, M. Yue, R. Karnik, A. Majumdar and P. D. Yang, *Phys. Rev. Lett.*, 2005, **95**, 086607.
154. W. Guan, R. Fan and M. A. Reed, *Nat. Commun.*, 2011, **2**, 506.
155. R. Karnik, K. Castelino and A. Majumdar, *Appl. Phys. Lett.*, 2006, **88**, 123114.
156. S.-W. Nam, M. J. Rooks, K.-B. Kim and S. M. Rossnagel, *Nano Lett.*, 2009, **9**, 2044-2048.

Critical Review

Analyst

- 1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
157. K.-H. Paik, Y. Liu, V. Tabard-Cossa, M. J. Waugh, D. E. Huber, J. Provine, R. T. Howe, R. W. Dutton and R. W. Davis, *ACS Nano*, 2012, **6**, 6767-6775.
158. T. James, Y. V. Kalinin, C.-C. Chan, J. S. Randhawa, M. Gaevski and D. H. Gracias, *Nano Lett.*, 2012, **12**, 3437-3442.
159. D. Stein, M. Kruithof and C. Dekker, *Phys. Rev. Lett.*, 2004, **93**.
160. D. G. Haywood, Z. D. Harms and S. C. Jacobson, *Analytical chemistry*, 2014, **86**, 11174-11180.
161. R. B. Schoch, H. van Lintel and P. Renaud, *Phys. Fluids* 2005, **17**, 100604.
162. C. M. Frament and J. R. Dwyer, *J. Phys. Chem. C*, 2012, **116**, 23315-23321.
163. M. J. Vitarelli, Jr., S. Prakash and D. S. Talaga, *Anal. Chem.*, 2011, **83**, 533-541.
164. R. Hatsuki, F. Yujiro and T. Yamamoto, *Microfluid. Nanofluidics*, 2013, **14**, 983-988.
165. J. Feng, J. Liu, B. Wu and G. Wang, *Anal. Chem.*, 2010, **82**, 4520-4528.
166. R. Karnik, K. Castelino, R. Fan, P. Yang and A. Majumdar, *Nano Lett.*, 2005, **5**, 1638-1642.
167. N. F. Y. Durand and P. Renaud, *Lab Chip*, 2009, **9**, 319-324.
168. R. B. Schoch, L. F. Cheow and J. Han, *Nano Lett.*, 2007, **7**, 3895-3900.
169. L. Lin, Y. Liu, J. Yan, X. Wang and J. Li, *Anal. Chem.*, 2013, **85**, 334-340.
170. G. V. Soni, A. Singer, Z. Yu, Y. Sun, B. McNally and A. Meller, *Rev. Sci. Instrum.*, 2010, **81**, 014301.
171. V. Kurz, E. M. Nelson, J. Shim and G. Timp, *ACS Nano*, 2013, **7**, 4057-4069.
172. W. Song, P. Pang, J. He and S. Lindsay, *ACS Nano*, 2013, **7**, 689-694.
173. N. Yukimoto, M. Tsutsui, Y. He, H. Shintaku, S. Tanaka, S. Kawano, T. Kawai and M. Taniguchi, *Sci. Rep.*, 2013, **3**, 1-7.