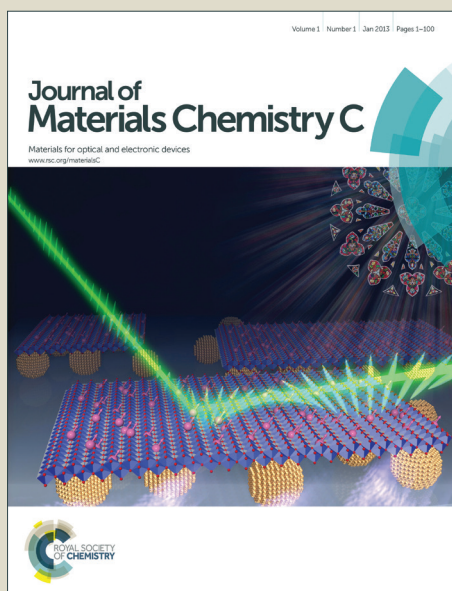


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Hierarchical SiO₂@Bi₂O₃ Core/Shell Electrospun Fibers for Infrared Stealthy Camouflage

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Novel, freestanding membranes composed of SiO₂@Bi₂O₃ hierarchical core/shell fibers were prepared by a combination of two fabrication methods: electrospinning and hydrothermal reaction. The SiO₂@Bi₂O₃ composite membranes were primarily supported by flexible SiO₂ fibers after calcination treatment of electrospun PVA/SiO₂ hybrid fibrous membranes. SiO₂@Bi₂O₃ composite fibers were fabricated via a process that entails hydrothermal growth of bismuth precursor nanocoating (Bi-PN) on the surface of SiO₂ fibers followed by the thermal treatment of the harvested SiO₂@Bi-PN fibers. It was observed that Bi₂O₃ nanoparticles were well anchored on the surface of SiO₂ fibers and the phase transition of Bi₂O₃ nanoparticles occurred during the thermal treatment of SiO₂@Bi-PN composite fibers at different temperatures. The infrared emission rates of the resultant SiO₂@Bi₂O₃ composite membranes were evaluated in comparison with pure SiO₂ fibers in 2 ~ 22 μm wavebands. It is theorized that the coating of Bi₂O₃ nanoparticles contributes to the decrease of infrared emissivity, and the infrared emission properties of SiO₂@Bi₂O₃ composite fibers are related to the α-Bi₂O₃ phase. The results favourably indicated prospects of SiO₂@Bi₂O₃ composite fibrous membranes for applications in infrared stealth camouflage.

Introduction

Stealth technology is indispensable in the modern military and detection industries. Infrared stealth has become a major focus with increasing demands for advanced detection and stealth technology. Infrared detectors discover the targets by contrasting infrared radiation between the target and background. In order to implement infrared stealth camouflage, the radiation intensity difference between the target and background should be small enough that it is negligible for detection.¹ Nevertheless, the radiation intensity of most military targets is higher than that of the natural background.² According to the Stefan-Boltzmann's law,³

$$W = \varepsilon \cdot \sigma \cdot T^4 \quad (1)$$

W: Emission energy
ε: Infrared emissivity
σ: Constant of Stefan-Boltzmann
T: Absolute temperature

the temperature (T) and infrared emissivity (ε) of the object are important parameters of infrared radiation. More specifically, infrared stealth camouflage is achieved by depressing the exterior temperature and decreasing the surface emissivity of the object.

Infrared stealth camouflage materials are a critical foundation for modern stealth technology. Infrared stealth fibers or fabrics are types of such infrared stealth materials, incorporating the infrared stealth materials into common fibers that retain the function of flexibility and weavability. For example, a military tank warmed-up by the sun may radiate at a higher temperature

than its surrounding long into a cold night and becomes easily detectable objects against the environment. Thermal-insulating materials, fiber or fabric generally acts as a cover for such objects, impeding the propagation of thermal energy. Alternatively, a low emissivity coating, polymer binder and pigment (comprised of metals, metal oxide and semiconductor) can be directly laid on the surface of the object to reduce infrared emission.⁴⁻⁹ However, thick and dense coatings increase the total weight of the materials. Instead, current infrared camouflage materials are required to be thin, lightweight and possess broad band characteristics.¹⁰⁻¹² Via a common melt-spinning method, Yu et al.¹⁰ prepared a bicomponent fiber with a sheath-core structure using polypropylene (PP) chips and various fillers (Ba/Mn-Zn ferrite and bronze particles). The bicomponent fiber possessed exceptional radar absorbing effect. Additionally, the bicomponent fiber possessing aluminium (Al) nanoparticles in the sheath showed better performance in radar absorption and infrared camouflage. The input of Al particles resulted in the infrared emissivity of the fibers decreasing from 0.79 to 0.62.

Over the past 20 years, electrospinning has been acknowledged as a simple but efficient method for producing fibers with diameters ranging from nano to micrometers, using a rich variety of materials (organic, inorganic or hybrid).¹³⁻¹⁷ Moreover, electrospun mats are advantageous due to their light weight nature and large surface area. Recently, surface coating or modifications of fibers have been utilized in inorganic nanofibrous membranes for practical applications of remediation and pollution control.¹⁸⁻²¹ The results of previous

studies imply that it may be possible to fabricate composite fibers with infrared stealth performance if the fibers act as the backbone for growth of low-emissive camouflage materials. Among fibrous materials, SiO₂ nanofibrous membrane is an excellent candidate for further preparation of hierarchical structures due to its high thermal stability and good flexibility.²² Nevertheless, SiO₂ is a conventional insulation material with high emissive. In order to achieve the lower emissivity, metallic pigments were suggested as ideal candidates²³⁻²⁶ for surface coating of SiO₂ fibers. The main disadvantage is the high reflectance from the surface of metal in the visual range, which significantly reduces the visual camouflage effectiveness. Furthermore, the metal is easily oxidized at high temperatures, leading to increased emissivity.

Due to its easily controlled infrared reflection spectroscopy and emissivity, semiconductor pigment has been found broad applications in the field of infrared stealth.^{27,28} Bismuth oxide (Bi₂O₃) is a semiconductor with a specific transparency and low infrared emissivity in the infrared light range. It can be used as the coating on SiO₂ fiber surfaces to reduce the emissivity of composite membranes. In addition, it is known that Bi₂O₃ exhibits polymorphism and exists in six polymorphs,²⁹⁻³³ where α phase is stable at low temperature and β phase is high-temperature metastable.³⁴ Metastable β -Bi₂O₃ may gradually change into α -Bi₂O₃ as the treated temperature increases.^{35,36} And each phase behaves in a special manner.³⁷⁻⁴² Therefore, it would be interesting to investigate the influence of Bi₂O₃ phase form on infrared emissivity properties of the composite fibers.

In this paper, we reported the preparation of SiO₂@Bi₂O₃ composite fibers through hydrothermal growth of bismuth precursor nanocoating (Bi-PN) on the surface of electrospun SiO₂ fibers followed by calcination at different temperatures. The incorporation of Bi₂O₃ nanoparticles with low infrared emissivity on the fibers potentially endows the fibrous membrane with properties that make it feasible for use in infrared stealth camouflage. The effects of Bi₂O₃ phase form on infrared emissivity properties of the composite fibers have been studied in the range of 2 ~ 22 μ m.

Experimental

Materials. All chemicals and solvents were commercially available and used as received without further purification. The starting materials included polyvinyl alcohol (PVA, molecular weight (M_w) is from 75000 to 80000 g/mol), tetraethyl orthosilicate (TEOS), diethylene glycol (DEG, 99.0%), phosphoric acid (H₃PO₄), concentrated nitric acid, concentrated sulfuric acid, hydrogen peroxide (H₂O₂, 30%), acetone and ethanol were purchased from Sinopharm Chemical Reagent Co., Ltd., (China). Bismuth acetate (99.999%) was purchased from Alfa Aesar (Tianjin) Co., Ltd., China. Deionized (DI) water was used in all experiments.

Preparation of SiO₂ fibrous membrane. The electrospun SiO₂ fibrous membrane was produced as following procedures according to the literature.⁴³ 3 g of 10 wt % PVA aqueous solution was dropped slowly into an equivalent weight of silica gel which was prepared by hydrolysis and polycondensation of a mixture with the molar composition of TEOS:H₃PO₄:H₂O = 1:0.01:11. After stirring, viscous precursor solution of the PVA/SiO₂ composite was obtained. Electrospinning was conducted using a microinjection pump at constant temperature (25 °C) and relative humidity (50%). Precursor solution was transferred into a 10 mL syringe with a 21 gauge (diameter = 0.5 mm) needle. The distance between the metallic needle and a squared metallic mesh collector (a = 15 cm) was 15 cm.

The hybrid fibrous membrane (PVA/SiO₂) was spun at a fixed electrical potential of 16 kV and fixed feeding rate of 1.0 mL/h precursor solution. The hybrid fibrous membrane was peeled off and then calcinated at 800 °C in air for 4 hrs to remove PVA. The fibrous membrane with the chemical composition SiO₂ was obtained. Subsequently, the as-prepared SiO₂ fibrous membrane was immersed in a bath of H₂SO₄/H₂O₂ (3:1, v/v) for about 6 hrs and thoroughly rinsed with DI water and ethanol several times.

Preparation of SiO₂@Bi-PN core/shell fibrous membrane. The synthesis process of SiO₂@Bi-PN fibers is described briefly as follows. First, 0.20 mmol (0.077 g) of bismuth(III) acetate was dissolved in diethylene glycol (DEG, 5ml) under vigorous stirring. Following this, 2 mL concentrated nitric acid and 35 mL of acetone were added. The resultant mixture solution was stirred for another 30 minutes.⁴⁴ Then, the SiO₂ fibrous membrane was immersed into the Bi-PN solution followed by hydrothermal treatment at 100 °C for 10 hrs in a Teflon-lined stainless steel vessel (50 mL). After cooling to room temperature, the obtained SiO₂@Bi-PN fibers were washed with ethanol three times and dried at 60 °C in the oven.

Preparation of SiO₂@Bi₂O₃ composite membranes. The above SiO₂@Bi-PN fibrous membranes were heated at different temperatures for 4 hrs. The resultant samples SiO₂@Bi₂O₃(400), SiO₂@Bi₂O₃(500) and SiO₂@Bi₂O₃(600) were obtained at heating temperature of 400 °C, 500 °C and 600 °C, respectively. A heating rate of 1 °C/min was used.

Characterization. Thermal property of the SiO₂@Bi-PN composite membrane was studied from 50 to 800 °C with a heating rate of 10 °C/min by a thermogravimetric analyzer (Diamond TG/DTA, PerkinElmer). The morphological analyses of the fibers were analyzed by a field emission scanning electron microscope (FE-SEM, S-4800, Hitachi) and a transmission electron microscopy (TEM, FEI Tecnai G20). All samples were coated by gold sputtering prior to SEM observations. The structural analyses of the fibers were conducted from 2 θ = 10° to 60° using X-ray diffractometer (XRD, X'Pert-Pro MRD, Philips) with Cu K α radiation (λ = 0.1542 nm) under a voltage of 40 kV and a current of 40 mA. X-ray photoelectron spectroscopy (XPS) spectra were obtained using XPS (Axis Ultra HSA, Kratos) with monochromatic Al K α radiation at a reduced power of 225 W. For the high-resolution scan, the step size was 0.1 eV and the pass energy was 40 eV. The vacuum in the analysis chamber was maintained at 4.0 \times 10⁻⁹ Pa for measurement. Infrared emission properties were examined using an infrared emission tester (TSS-5X, Japan) at room temperature. According to the Kirchoff's law,⁴⁵ the relationship between infrared emissivity (ϵ) and reflectance (R) is expressed as:

$$\epsilon + R = 1 \quad (2)$$

The emissivity was determined by converting reflectance of infrared rays which were emitted from hemispherical black body inside a sensor head at the surface of measurement target. The standard error of the instrument is \pm 0.01. The fibrous membranes were tailored into a square shape with dimensions of 50 \times 50 mm² for measurement.

Results and discussion

Fig. 1 shows the electrospun PVA/SiO₂ hybrid fibrous membrane and SiO₂ fibrous membrane as prepared, and after bending. Notably, the SiO₂ fibrous membrane presented the high flexibility upon bending as shown in Fig. 1c and 1d. The thickness of the SiO₂ fibrous membrane, as collected, is approximately 200 μ m in Fig. 2a. The morphology of SiO₂ fibers was investigated by SEM (Fig. 2b and 2c), which

presented that the surface of SiO_2 fibers is smooth and clean, and cross-section of SiO_2 fibers are solid and compact. In addition, the average diameter of the SiO_2 fibers is approximately 800 nm. It can be seen that the crosslinking between silica fibers, which was highlighted by the red circles

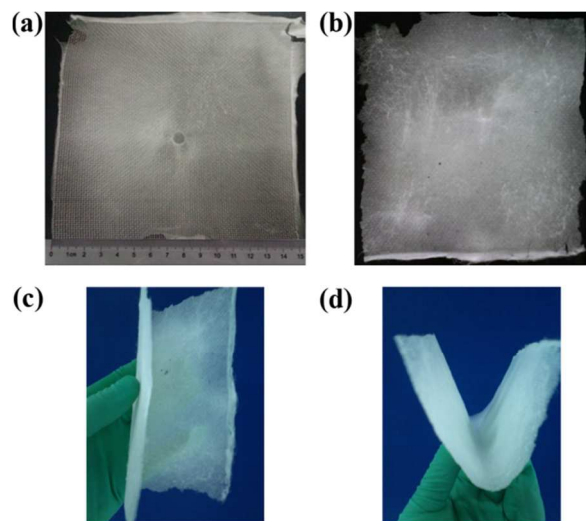


Fig. 1 Photos of (a) electrospun PVA/ SiO_2 hybrid fibrous membrane on a squared metallic mesh collector ($a = 15$ cm) and (b) SiO_2 fibrous membrane after removing of PVA via calcination, (c) and (d) bending flexibility of the SiO_2 fibrous membrane.

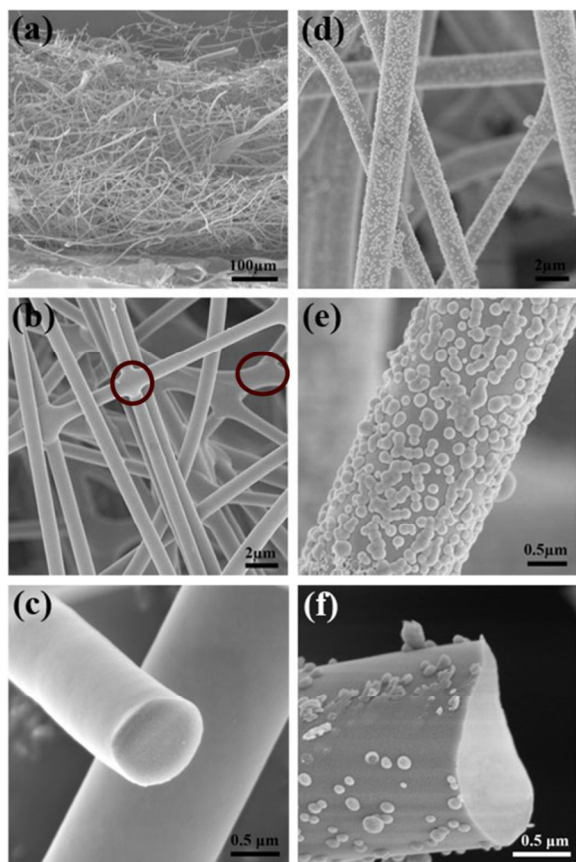


Fig. 2 SEM micrographs (a) cross-sectional view of SiO_2 fibrous membrane, (b) SiO_2 fibers, (c) cross-sectional view of SiO_2 fibers, (d) and (e) SiO_2 @Bi-PN core/shell fibers at low and high magnification, (f) cross-sectional view of SiO_2 @Bi-PN core/shell fibers.

in Fig. 2b. It is formed in the electrospun process and the hydroxyl groups of PVA on the surface of nanofibers interact to each other. And after calcinations, the melting phenomenon²² may reinforce the connection. In comparison, it can be observed in Fig. 2d-2f that the nanoparticles of bismuth precursor have been fixed and uniformly dispersed only on the surface of SiO_2 fibers. The activation of SiO_2 with -OH groups increases the combination of Bi_2O_3 nanoparticles with SiO_2 fibers, limiting dependence on adhesives.

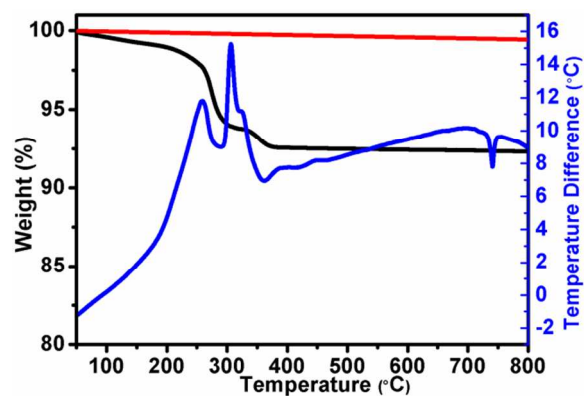


Fig. 3 The TGA of SiO_2 fibers (red) and TG-DTA of SiO_2 @Bi-PN fibers (black and blue).

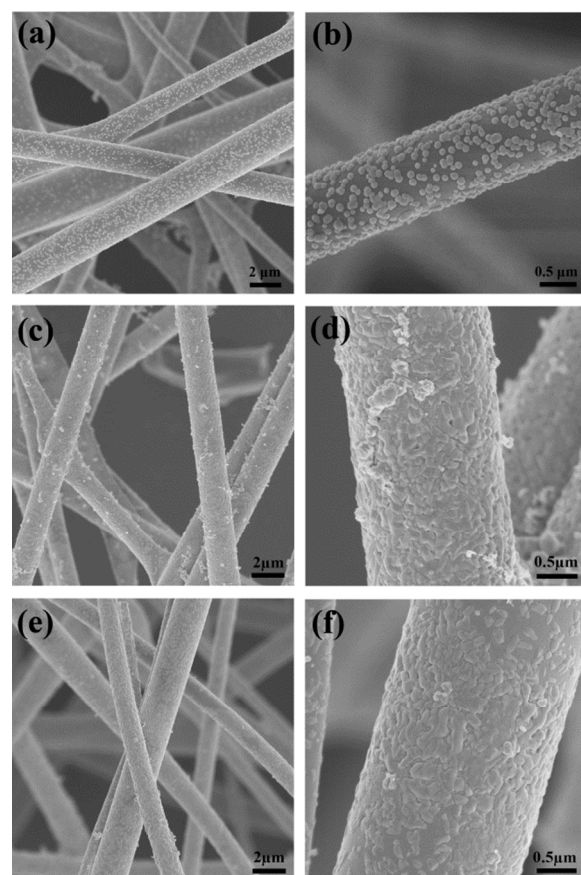


Fig. 4 SEM images of SiO_2 @ Bi_2O_3 composite fibers calinated at different temperatures: (a) and (b) 400 °C, denoted as SiO_2 @ Bi_2O_3 (400), (c) and (d) 500 °C, denoted as SiO_2 @ Bi_2O_3 (500), (e) and (f) 600 °C, denoted as SiO_2 @ Bi_2O_3 (600).

The thermal transformation of the precursor $\text{SiO}_2@ \text{Bi-PN}$ to $\text{SiO}_2@ \text{Bi}_2\text{O}_3$ was measured based on the TGA and DTA measurements, as shown in Fig. 3. First, the SiO_2 fibers retain constant weight from 50 to 800 °C, confirming the thermal stability of the SiO_2 membrane. The TGA curve of $\text{SiO}_2@ \text{Bi-PN}$ composite membrane showed the weight loss in the range of 200-350 °C, which can be attributed to the decomposition of bismuth precursor and the phase transformation from precursor to Bi_2O_3 . The temperature range of sharp peaks on DTA trace fits well with that of conversion in the TGA curve. Furthermore, the DTA curve gradually ascended as the temperature increased, which may be the evidence of Bi_2O_3 phase transition processes. According to the TGA and DTA curves, the calcination process was performed at a temperature of above 400 °C to ensure the phase transition occurred.

After being calcinated at 400°C, 500°C, and 600°C, $\text{SiO}_2@ \text{Bi}_2\text{O}_3$ composite fibers were obtained and observed by SEM, as shown in Fig. 4. It can be seen that the Bi_2O_3 particles form separate domains with a spherical shape and diameter of 100-300 nm on the surface of SiO_2 fibers after calcination at 400 °C, which is consistent with the result of TEM observation (Fig. S1 in ESI). As the temperature increased to 500 °C, fused domains connected and covered the surface of the fibers. At 600 °C, the surfaces of the fibers were completely covered by the Bi_2O_3 layer, which possessed the higher smoothness compared with previous conditions. The difference of the morphology of $\text{SiO}_2@ \text{Bi}_2\text{O}_3$ core-shell fibers indicates that the Bi_2O_3 may undergo phase transitions and unusual reactions as the temperature increases, resulting in different phase structures.

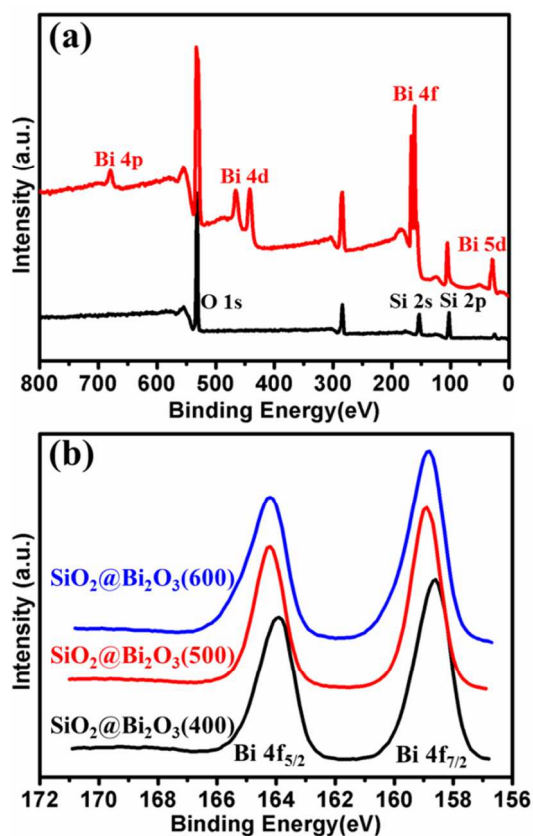


Fig. 5 (a) XPS spectra for the surface of SiO_2 fibers (black) and $\text{SiO}_2@ \text{Bi}_2\text{O}_3(400)$ composite fibers (red) for comparison. (b) High resolution spectra of Bi 4f peaks for $\text{SiO}_2@ \text{Bi}_2\text{O}_3$ composite fibers calcinated at different temperatures.

Fig. 5a shows the XPS spectra of the representative sample $\text{SiO}_2@ \text{Bi}_2\text{O}_3(400)$ in comparison with the pure SiO_2 fibers. It can be seen that apart from the Si 2p and O 1s peaks, peaks such as Bi 4p, Bi 4d, Bi 4f, Bi 5d emerged with strong relative intensities, indicating that SiO_2 fibers were mainly coated with Bi_2O_3 particles. The C 1s peak, located at 284.5 eV in the XPS spectra, is derived from a surface layer of carbon or organic contamination. The high resolution XPS spectra of Bi 4f exhibit narrow Gauss-shaped dual peaks, denoted as Bi 4f_{7/2} and Bi 4f_{5/2} spectra, as shown in Fig. 5b. The characteristic binding energies of Bi 4f_{7/2} and Bi 4f_{5/2} are 158.6 eV and 163.9 eV for $\text{SiO}_2@ \text{Bi}_2\text{O}_3(400)$ fibers, and 158.9 eV and 164.2 eV for both $\text{SiO}_2@ \text{Bi}_2\text{O}_3(500)$ and $\text{SiO}_2@ \text{Bi}_2\text{O}_3(600)$ fibers. These values are in agreement with the reported values,^{37,42} which indicate that the Bi ions are mainly in the oxidation state of Bi³⁺. Nevertheless, the peak position of Bi 4f shifts to a higher binding energy as the calcination temperature increases from 400 to 600 °C. The shifting of binding energies confirms the change of the oxidation state of Bi ions, which may attribute to the phase transition of Bi_2O_3 .

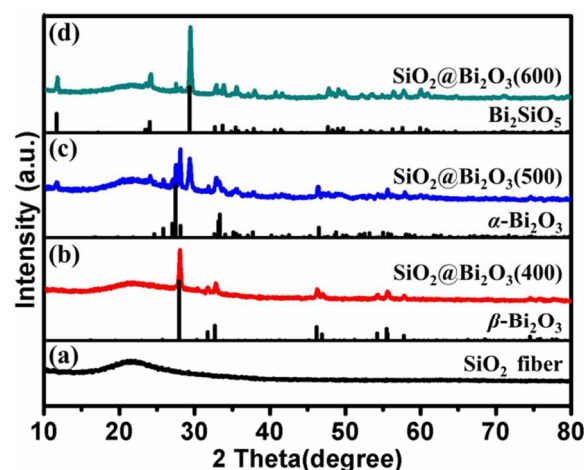


Fig. 6 XRD patterns: (a) SiO_2 fibers, (b) standard JCPDS data for $\beta\text{-Bi}_2\text{O}_3$ (black) and $\text{SiO}_2@ \text{Bi}_2\text{O}_3(400)$ composite fibers (red), (c) standard JCPDS data for $\alpha\text{-Bi}_2\text{O}_3$ (black) and $\text{SiO}_2@ \text{Bi}_2\text{O}_3(500)$ composite fibers (blue), (d) standard JCPDS data for Bi_2SiO_5 (black) and $\text{SiO}_2@ \text{Bi}_2\text{O}_3(500)$ composite fibers (green).

XRD patterns of the SiO_2 fibers and composite fibers are shown in Fig. 6. The pure SiO_2 fibers show a weak and broad diffraction peak around 22°, which can be attributed to the amorphous phase of SiO_2 in Fig. 6a. In Fig. 6b, the diffraction peaks of $\text{SiO}_2@ \text{Bi}_2\text{O}_3$ fibers are in accordance with a standard diffraction of $\beta\text{-Bi}_2\text{O}_3$ (JCPDS No. 78-1793). The sharp diffraction peaks indicate its high crystallinity. It is notable that in the XRD pattern of the $\text{SiO}_2@ \text{Bi}_2\text{O}_3(500)$ composite membrane, shown in Fig. 6c, the peaks of $\beta\text{-Bi}_2\text{O}_3$ decrease while the second phase of $\alpha\text{-Bi}_2\text{O}_3$ (JCPDS No. 76-1730) and another new phase with the chemical composition Bi_2SiO_5 (JCPDS No. 75-1483) appeared. Namely, the $\text{SiO}_2@ \text{Bi}_2\text{O}_3(500)$ composite membrane is containing $\alpha\text{-Bi}_2\text{O}_3$ and $\beta\text{-Bi}_2\text{O}_3$ phase junction. With the further increase of thermal treatment temperature up to 600 °C, the major phase can be indexed to the Bi_2SiO_5 while a trace amount of $\alpha\text{-Bi}_2\text{O}_3$ phase as shown in Fig. 6d. It is clear that the $\text{SiO}_2@ \text{Bi}_2\text{O}_3$ core-shell fibers undergo a phase transition from $\beta\text{-Bi}_2\text{O}_3$ to $\alpha\text{-Bi}_2\text{O}_3$ and interfacial reaction of Bi_2O_3 and SiO_2 ⁴⁶ occur as the temperature increases from 400 to 600 °C.

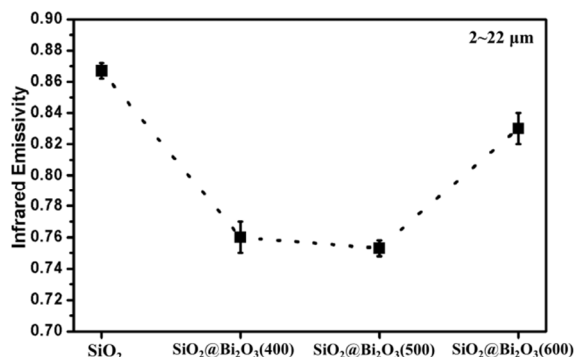


Fig. 7 Infrared emissivity of SiO₂ fibrous membrane and SiO₂@Bi₂O₃ composite membranes calcinated at different temperatures.

The infrared emissivity of SiO₂ fibrous membrane within the wavelength range of 2 ~ 22 μm is determined to be 0.87 (Fig. 7). For SiO₂@Bi₂O₃ composite membranes, there is a special layer between the core and shell interface. The motion modes of molecules, atoms and dangling bond at the interface layer are not same as those of bulk SiO₂ fibers,⁴⁷ which give rise to different infrared absorption. According to the Kirchhoff's law,⁴⁵ the emissivity is equal to the absorptivity of incident radiation for the same material. Therefore, it is observed that the infrared emission rates of SiO₂@Bi₂O₃ composite fibrous membranes clearly decrease compared to the pure SiO₂ fibrous membrane, which can be ascribed to the interfacial interactions between SiO₂ fibers and Bi₂O₃ particles. Particularly, the emissivity of SiO₂@Bi₂O₃(500) composite membrane is estimated to be 0.75, displaying the lower infrared emissivity than that of SiO₂@Bi₂O₃(400) composite membrane. It is supposed that the stable phase α-Bi₂O₃ owing to the phase transition of β-Bi₂O₃ is more beneficial for decreasing the infrared emissivity. Firstly, the β-Bi₂O₃ was formed at 400 °C, and was converted to α-Bi₂O₃ after higher calcination temperature. β-Bi₂O₃ is a metastable tetragonal phase that exists at moderate temperatures between 330 and 650 °C. It has a distorted defect structure with 25% of ordered vacant oxygen sites and could be stabilized to room temperature by partial substitution with other metal cations.^{48,49} Therefore, partial replacement of Bi atoms with Si atoms may occur while the SiO₂@Bi₂O₃(400) composite membrane was cooling from 400 °C to room temperature. Consequently, the lattice will be compressed around the Si impurity. Impurities are known to affect vibrational properties of crystals by modifying the distribution of normal mode frequencies and altering the nature of the atomic displacements in the neighbourhood of the impurities. The β-Bi₂O₃ with distorted lattice could enhance the infrared absorptivity. This enhancement may be also associated with CO₂ molecules in air adsorbed on these coordinatively unsaturated sites and interacted with β-Bi₂O₃ on the surface. As a result, it induces the higher infrared emissivity. In comparison, α-Bi₂O₃ displayed lower infrared emissivity owing to its stable and less-defect crystal structure. Instead of promoting complete transition from β-Bi₂O₃ to α-Bi₂O₃, the increasing of treatment temperature encourages the interfacial reaction of Bi₂O₃ and SiO₂, which gives rise to the less amount of α-Bi₂O₃ and slight higher infrared emissivity for SiO₂@Bi₂O₃(600) composite fibrous membrane. In addition, the repeated measurements of infrared emissivity of SiO₂@Bi₂O₃ composite fibrous membranes are under the ambient condition and last for a couple of weeks. The final results show the stability within the range of fluctuation (experimental and systemic errors), which

means the aging effect of the fibers is insignificant under the repeated infrared exposure.

Conclusions

In this work, we have prepared SiO₂@Bi₂O₃ composite fibrous membranes through a process that includes electrospinning and hydrothermal reaction, followed by calcination. Using thermal treatments at different temperatures, SiO₂@Bi₂O₃ composite fibrous membranes with different phase structural Bi₂O₃ were obtained. The coating of Bi₂O₃ nanoparticles contributes to the infrared emissivity of the composite fibrous membranes. And introduction of stable α-Bi₂O₃ nanoparticles to the SiO₂@Bi₂O₃ composite fibers could decrease the infrared emissivity of the composite membrane more effectively.

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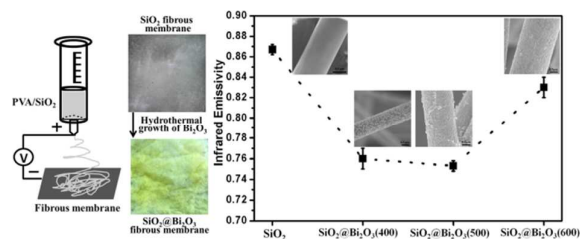
Notes and references

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- Electronic Supplementary Information (ESI) available: TEM images of SiO₂@Bi₂O₃ composite fibers. See DOI: 10.1039/x0xx00000x/

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TOC



Hierarchical $\text{SiO}_2@ \text{Bi}_2\text{O}_3$ core/shell electrospun fibrous membranes with low infrared emissivity were successfully prepared. And the infrared emissivity is influenced by the Bi_2O_3 nanoparticles with different phase structures on the surfaces of the SiO_2 fibers.