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ARTICLE TYPE

Thermoelectric properties of p-type polycrystalline SnSe doped with Ag [†]

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Many IV-VI semiconductors tend to be good thermoelectric materials, this includes all Pb chalcogenides as well as the Pb-free SnTe: all of which crystalize in NaCl cubic structure. Another group of IV-VI compounds form layered orthorhombic structure. SnSe is one of these compounds, whose transport properties as polycrystalline thermoelectric materials have rarely been studied. Here we present our study

¹⁰ of p-type polycrystalline SnSe doped with Ag, prepared by melting and hot pressing. SnSe has anisotropic properties with hysteresis seen in resistivity between 300 and 650 K regardless of doping. Ag is not an ideal dopant but is able to increase the carrier density significantly, as a result a peak *zT* of 0.6 was observed at 750 K. Transport properties of doped SnSe can be explained with a single parabolic band model, which suggests promising potential for this compound together with its challenges.

15 1. Introduction

Thermoelectric materials are semiconductors that enable direct conversion between heat and electricity. Devices using these materials are being used in solid state cooling as well as power supply¹. The performance of a material as thermoelectrics can be ²⁰ evaluated with the dimensionless figure of merit, $zT = \sigma S^2 T/\kappa$, where *S*, *T*, σ and κ are, Seebeck coefficient, absolute temperature, electrical and thermal conductivity, respectively. Around 500 °C the highest *zT*s are seen in IV-VI compounds such as PbTe^{2, 3}, PbSe^{4, 5} and PbS⁶. Despite of the continuing ²⁵ progress in advancing performance of these materials⁷⁻¹², the containing of Pb has limited their future in domestic application.

- SnTe could be an alternative according to recent study¹³ but the use of Te is still a big disadvantage for domestic devices that are more cost-sensitive. The other of IV-VI compounds, including ³⁰ SnSe, have not been studied as carefully in polycrystalline form, compared with the compounds mentioned above. High
- thermoelectric performance has just recently been reported¹⁴ in undoped, single crystal SnSe accelerating interest in this material. Unlike its heavier analogs with NaCl type structure, SnSe forms ³⁵ a layered structure with zigzagged atomic chains with
- orthorhombic symmetry^{15, 16}. It undergoes a phase transition at high temperature, which alters the zigzag chains so that the atoms are stacked with a configuration that resembles the rock-salt but with stacking faults every other layers along one (001) ⁴⁰ direction¹⁷. This leads to a considerable difference in the band structure and transport properties between this compound and the rock-salt IV-VI compounds.

In this paper we discuss the synthesis and transport properties of p-type polycrystalline SnSe as a potential thermoelectric ⁴⁵ material. Stoichiometric SnSe is found to be a semiconductor with very low intrinsic defect level and high resistivity. Extra free carriers, in this case holes, can be introduced by substituting Sn with Ag. For the sample $Ag_{0.01}Sn_{0.99}Se$, zT is found to be 0.6 at 750 K, higher than that of an undoped SnSe (0.3) due to increase of charge carrier density. SnSe could be a promising thermoelectric material, given the doping challenges could be resolved.

2. Experimental

- ⁵⁵ Elements Ag, Sn, Se, all of purity of 99.999 % were loaded into quartz ampoules. The ampoules were then evacuated, sealed, and slowly heated up to 1200 K, and kept for 12 hours followed by water quench. The obtained ingots were further annealed at 800 K for 72 hours. The annealed ingots were ground into powders
 ⁶⁰ with an agate mortar, and then hot pressed using a induction heating rapid hot press¹⁸ at 800 K, 36 MPa under 1 atm argon for 10 minutes to form a dense pellet 12 mm in diameter. Two sheets of samples were cut along different directions for measurement,
- with relative density no less than 96 % of theoretical value (6.18 ⁶⁵ g/cm³). The grain size is estimated to be between 10 to 100µm, which according to our previous experience on lead chalcogenides⁶, leads to negligible influence on electron and phonon transport. The phase composition is analyzed by powder X-ray diffraction (XRD), performed with a diffractometer ⁷⁰ (PANalytical X'Pert Pro) equipped with Cu Kα radiation. The microstructures (estimated grain size of 10 to 100µm) of samples are investigated by scanning electron microscope (SEM, ZEISS). The compositions of the phase and the distributions of elements are determined by X-ray energy dispersive spectrometer (EDS, ⁷⁵ Oxford Instrument X-Max). Two-dimensional Seebeck
- coefficient maps and their corresponding histograms are determined from a scanning Seebeck coefficient measurement system¹⁹ developed at the California Institute of Technology.

Temperature dependent Seebeck coefficient is determined by measuring the voltage difference on two sides of the sample using chromel-niobium thermocouples under an oscillating temperature gradient of 0 to 15 K with constant ambient temperature²⁰. s Electrical resistivity and Hall effect are measured using the Van

- der Pauw method in a magnetic field up to $\pm 2T^{21}$. The thermal conductivity is calculated from $\kappa = \rho C_p D_T$. With thermal diffusivity, D_T measured by laser flash method (Netzsch LFA 457) under Ar flow using Cowan model plus pulse correction.
- ¹⁰ The heat capacity values C_p are obtained using suggested fitting equation given by Pashinkin et al. based experimental result²², which is consistent with experimental values reported for higher temperatures²³ from Yamaguchi et al. (Plot of C_p shown in ESI†). Density ρ is calculated using the measured dimensions and ¹⁵ weight. During all measurements data are acquired during both
- heating and cooling. The uncertainty of each measurement is estimated to be approximately 5 %, which leads to the combined experimental uncertainty of zT to be about ~20 %.



Fig. 1 (a) XRD patterns of SnSe alloying with different Ag concentrations indicate SnSe and a small amount of AgSnSe₂ phase (arrows). (b) EDS mapping images of 1 % Ag alloyed SnSe alloys. The Ag area is marked with pink coloring. (c) and (d) Scanning Seebeck coefficient measurement

25 results and histograms of 1% and 5% Ag- alloyed SnSe alloys, respectively.

3. Results and discussion

30 3.1 Synthesis and characterization

X-ray diffraction data of samples with the nominal Ag compositions staring from 0 to 7 % are shown in Fig. 1a. The patterns can be indexed to the low temperature SnSe phase (JCPDS #140159) with an orthorhombic structure. Small amount ³⁵ of secondary phase, denoted by arrows, is seen in Ag doped

samples and is identified to be the AgSnSe₂ phase (JCPDS

#331194) with cubic structure. The existence of AgSnSe₂ is also confirmed by the back-scattered electron (BSE) images of polished surfaces of all Ag doped samples. The SEM
⁴⁰ micrographs and EDS mapping images of 1% Ag samples are shown in Fig. 1b as an example. Ag segregates in regions of few microns even for as little as 1% Ag addition. Scanning Seebeck coefficient mapping on Ag_{0.01}Sn_{0.99}Se and Ag_{0.05}Sn_{0.95}Se (Fig. 1c and d) revealed small regions with abnormally low Seebeck
⁴⁵ coefficients that likely originate from the metallic secondary phase. As with SEM the Seebeck mapping shows all these regions are isolated without forming a linked network. Using an effective medium model^{24, 25} we estimated the effect of the secondary AgSnSe₂ phase on the electrical conductivity and Hall ⁵⁰ coefficient to be less than 10%.



Fig. 2 Seebeck coefficients (a) and electrical conductivity (b) as a function of temperature for Sn₁Se₁, Sn₁Se_{1.05}, and Sn₁Se_{0.95}. (c) Measured ⁵⁵ Hall carrier concentration as function of Ag content at 300 K.

Stoichiometric SnSe shows p-type intrinsic behavior with low carrier density on the order of low 10¹⁷ cm⁻³. With up to 5% of extra Se the compound shows largely the same transport behavior 60 with negligible carrier density change; whereas with 5% extra Sn the compound become slightly n-type (but not stable) (Fig. 2a and b). Unlike Pb chalcogenides, SnSe is fairly difficult to dope²⁶⁻ ²⁹. Various candidates as dopants were attempted (Br, Sb, Bi and In as donors; Na, Ag, and Tl as acceptors), but only Na and Ag 65 seems to be able to increase the carrier density to the concentrations usually required for good thermoelectric performance. While Na provides better efficiency and higher carrier density, samples doped with Na are not stable upon repeated heating and cooling. As a result, Ag is used as the 70 dopant for this study. As shown in Fig. 2c, with addition of Ag, the carrier density can be gradually increased and saturated at about 9×10^{18} cm⁻³ for the Ag_{0.07}Sn_{0.93}Se sample. The doping efficiency is poor on the order of 1%, which is consistent with microstructural observation. The solubility of Ag in SnSe is very 75 small but seems to be sensitive to actual composition so that the systematic change of Ag content has resulted in a very slight change of Ag solubility, which is reflected by the increasing of carrier density.

3.2 Electrical and thermal transport properties

All SnSe samples show hysteresis during heating and cooling (more in resistivity and less in Seebeck coefficient) at around 450 to 600 K. Altering the Sn:Se ratio, as well as doping with Ag fails 5 to eliminate the hysteresis. The Differential Scanning Calorimetry (DSC) Showed very small peaks around 430 K, 580 K and 610 K, respectively (See ESI). The phase transition in SnSe was reported to occur at 809 K¹⁷, Other processes, are thus responsible for hysteresis in resistivity as well as the peaks seen in DSC. Despite 10 the hysteresis, the transport properties of the Ag doped samples

- are consistent upon several repeated heating and cooling measurements (Fig. 3), suggesting that the hysteresis in these samples are more likely caused by reversible processes rather than non-equilibrium states of the samples. For clarity in the
- 15 following discussions only results measured during cooling are shown, the heating process reveals the same trend only with different absolute values (Fig. S1).



20 Fig. 3 Temperature dependent resistivity (a) and Seebeck coefficients (b) of 1% Ag alloyed SnSe after several repeated heating and cooling test.



Fig. 4 Temperature dependence of electrical resistivity (a), Seebeck ²⁵ coefficients (b), thermal conductivity (c) and zT (d) for Ag-alloyed SnSe.

Fig. 4 shows the temperature dependence of transport properties for samples with different Ag content. These are the properties measured along the direction perpendicular to the hot press ³⁰ direction. The resistivity at room temperature has dropped significantly with Ag doping, yet at 750 K the resistivity of doped samples are still several times higher than typical good thermoelectrics. The resistivity is anisotropic and the in-plane result (perpendicular to hot press direction) is about 40% lower 35 than the other direction (Fig. 5 and Figs. S2 to S4).

The temperature dependent Seebeck coefficients are shown in Fig. 4b. High *S* values are seen for all samples, even the one with largest carrier density 9×10^{18} cm⁻³ shows *S* around 200 µV/K at room temperature. A maximum is seen in results for SnSe and 40 Ag_{0.01}Sn_{0.99}Se, the other two samples with higher Ag content have very weak temperature dependence. A maximum in the Seebeck coefficient is usually a result of bipolar conduction (noticeable minority carriers being excited) where from the peak value a band gap of about 0.5 eV can be estimated³⁰, in roughly 45 consistent with the recently reported¹⁴ band gap of 0.8 eV. Bipolar conduction could contribute here, although there are also indications that point to other possibilities as discussed later.



Fig. 5 (a) Seebeck coefficients, (b) electrical resistivity, (c) thermal ⁵⁰ conductivity, and (d) zT of two slices cut from a single cylinder along different direction. Different coloring in a) b) and c) represents results measured from each slice of the cylinder, while the coloring in d) represents zT calculated from properties measured along each direction.

For all samples the Hall mobility decreases with temperature roughly following a $T^{1.5}$ relation, shown in Fig. 6 (data shown are smoothed fitting result, raw data showed considerable scattering due to large contact resistance in Van de Pauw measurement (Fig. S5)). This enables us to use the simple single parabolic band ⁶⁰ model with acoustic phonon scattering approximation to understand the change of transport properties with doping. At 300 K, the change of Seebeck coefficient with carrier density (Pisarenko relation) has suggested a density of state effective mass m_d^* of 0.75 m_e (Fig. 7a), where m_d^* was determined by Hall ⁶⁵ carrier density ($n_H = n/r_H$, assuming spherical Fermi surface) via:

$$n = \frac{(2m_d^* k_B T)^{3/2}}{2\pi^2 \hbar^3} F_{1/2}(\eta) \tag{1}$$

$$r_{H} = \frac{3}{4} \frac{F_{1/2}(\eta) F_{-1/2}(\eta)}{F_{0}(\eta)^{2}}$$
(2)

using reduced chemical potential η from Seebeck coefficient:

$$S = \frac{k_B}{e} \left[\left[\frac{2F_1(\eta)}{F_0(\eta)} - \eta \right] \right]$$
(3)

 $F_x(\eta)$ are the Fermi integrals:

$$F_{x}(\eta) = \int_{0}^{\infty} \frac{\varepsilon^{x}}{1 + \exp(\varepsilon - \eta)} d\varepsilon$$
⁽⁴⁾

The drift mobility governed by acoustic phonon scattering 5 process depends on chemical potential via:



Fig. 6 log μ -log T plots of SnSe alloys indicating the linear regime with a 10 slope of δ = -1.5.



Fig. 7 Seebeck coefficient (a) and Hall mobility (b) as function of Hall carrier density at 300 and 750 K, respectively.

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 C_l is the average longitudinal elastic constants, N_v is the valley degeneracy of the band, Ξ is the deformation potential coefficient. Measured Hall mobility ($\mu_H = \mu r_H$) versus Hall carrier density can be well explained with this equation (Fig. 7b). The

- ²⁰ parameter μ_0 stands for the mobility extrapolated to the nondegenerate limit, and is found to be 41 cm²/Vs by fitting experimental results. Band structure calculation³¹ has suggested that the maximum of the valence band in SnSe is on the Λ symmetry line between Gamma and Z, having a valley
- ²⁵ degeneracy of 2. Assuming C_l of SnSe is the same as SnTe (58 GPa) the deformation potential coefficient is estimated to be 24 eV, being comparable to those found in rock-salt Pb chalcogenides⁶.

Using the same parameters determined at 300 K, SPB model ³⁰ further explained the Pisarenko relation at 750 K (Fig. 7). Usually with bipolar conduction, the measured Seebeck coefficient should be lower than modeling result using SPB. This being not the case for Ag doped SnSe implies that there may be other possible reason for the decrease of Seebeck coefficients and increase of

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³⁵ carrier density with increasing temperature, such as temperature dependent defect level or Ag solubility. Similarly the observed mobility values are also comparable with SPB modeling result: no reduction due to bipolar conduction is seen.

The thermal conductivities κ measured up to 750 K are ⁴⁰ presented in Fig. 4c. Due to high resistivity the electronic contribution for all samples are negligible. The measurement generally reflects the change of lattice thermal conductivity with temperature. For undoped SnSe the κ_L is found to be 1.1 W/mK at 300 K and decreases with temperature to 0.55 W/mK at 750 K,

- ⁴⁵ We notice this result is lower than historical results on single crystals^{32, 33} (1.8 W/mK along cleavage plane), as would be expected from the polycrystalline nature of the sample, while the recent study¹⁴ on single crystal SnSe has reported even lower κ_L . Compared to other IV-VI compounds SnSe has a very low lattice
- ⁵⁰ thermal conductivity at room temperature, probably due to its complex atomic arrangement relative to the highly symmetric NaCl structure. However, the T^1 dependence, result of Umklapp process dominated phonon transport, which was seen in rock-salt Pb chalcogenides, is not observed here. The κ_L decreased slowly
- ⁵⁵ before saturated at 0.55 W/mK, suggesting a large residue thermal conductivity from optical phonon contribution³⁴. In contrast, κ_L for PbSe at 750 K is about 0.7 W/mK, only slightly larger than SnSe whereas the electrical properties of PbSe are much better. Similar to resistivity, anisotropy is also seen in ⁶⁰ thermal conductivity measured along different directions and the in-plane thermal conductivity is about 60% higher than those along the other direction (Fig. 5c and Fig. S4). Thermal conductivities increase with the increase of Ag contents. This is partly due to the averaging with a metallic secondary phase, ⁶⁵ which has high thermal conductivity. The higher thermal conductivity could also be related to the possible change in the degree of texturing due to the uniaxial pressure and the presence of ductile metallic AgSnSe₂ with low melting temperature.



Fig. 8 zT as a function of carrier density at 750 K. The experimental results (symbol) can be predicted by the modeled zT (solid and dash curves).

3.3 Figure of merit

75

The *zT* values (in-plane direction) as a function of temperature are calculated using data acquired during the cooling process and the result is shown in Fig. 4d. The *zT* of the 1% Ag sample ⁸⁰ reaches a maximum of ~0.6 at 750 K as compared to 0.3 for the

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stoichiometry SnSe. Continue increasing Ag content and carrier density does not further increase zT. This is because first the higher thermal conductivity due to increased volume fraction of secondary phase, and second the lower mobilities. Using the SPB $_{5}$ model with parameters determined above $(m_{d}^{*} = 0.75 m_{e}, N_{v} = 2, 1000 \text{ m})$

- s model with parameters determined above ($m_d = 0.75 m_e$, $N_v = 2$, $C_l = 58$ GPa, $\Xi = 24$ eV, and using $\kappa_L = 0.5$ W/mK), the modeled zT versus carrier density at 750 K is plotted in Fig. 8, which predicts a maximum of 0.4 when the carrier density is around 4 to 5×10^{19} cm⁻³. The calculated zT of 0.6 for Ag_{0.01}Sn_{0.99}Se being
- ¹⁰ higher than the predicted maximum, mainly due to the observed mobility being much higher than the model as well as other samples.

The predicted zT for SnSe is somewhat disappointing. There are however two factors not accounted for in current analysis that

- ¹⁵ could potentially justify higher expectation on zT values. First, the mobility of these samples may be not as high as the cleanest samples, especially when secondary phase has been found. Assuming the mobility of "clean" samples can be twice as high, the maximum zT would be 0.7 (dashed line in Fig. 8), which is a
- $_{20}$ good start for a Pb-free and Te-free thermoelectric material. Second, unlike the isotropic Pb-chalcogenides, the modeling of transport properties on polycrystalline SnSe in this study is subject to the degree of texturing: theoretically maximum *zT* can be only achieved when single crystal SnSe is grown and
- 25 measured along the preferred direction. We notice that independent studies $^{35-37}$ have reported for single crystal SnSe a Hall mobility of $\sim 120~{\rm cm^2/Vs}$ perpendicular to c-axis (denoted a-axis in reference [14]), a three-fold difference compared with the highest mobility found in this study, while only a quarter 35 to a
- ³⁰ half³⁷ of this along c-axis. Thus there could be up to a three-fold increase of zT (to 1.2,) if single crystals of these samples were grown and tested perpendicular to c-axis. Nevertheless, further material engineering is needed before p-type SnSe could compete, in terms of average zT, with the best available PbTe or
- ³⁵ PbSe thermoelectrics. At the same time, the hysteresis issue in properties seen here between 300 and 650 K has to be resolved. Until then, any practical value of SnSe as a thermoelectric material will remain an open question regardless of its peak *zT*.

Conclusions

- ⁴⁰ In summary, polycrystalline SnSe doped with Ag were prepared by melting followed by rapid hot pressing. Undoped SnSe has low carrier density. Doping with Ag increases this to 9×10^{18} cm⁻³. The transport properties, especially resistivity and thermal conductivity are anisotropic. Hysteresis is seen in resistivity
- ⁴⁵ between heating and cooling regardless of doping. Transport properties at 300 K and 750 K can be explained with a single parabolic band model and acoustic phonon scattering ¹⁰⁰ approximation. The density-of-states effective mass is found to be 0.75 m_e and the nondegenerate limit drift mobility is 41
- $_{50}$ cm²/Vs, both independent of temperature. The peak *zT* measured 19. at 750 K is 0.6. Although higher *zT* in doped p-type SnSe is still possible, material engineering is needed for SnSe to be a truly 105 20. promising thermoelectric material.

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References

- ⁶⁰ 1. L. E. Bell, *Science*, 2008, **321**, 1457-1461.
 - Y. Pei, A. LaLonde, S. Iwanaga and G. J. Snyder, *Energy Environ.* Sci., 2011, 4, 2085-2089.
- A. D. LaLonde, Y. Pei and G. J. Snyder, *Energy Environ. Sci.*, 2011, 4, 2090-2096.
- 65 4. H. Wang, Y. Pei, A. D. Lalonde and G. J. Snyder, *Adv. Mater.*, 2011, 23, 1366-1370.
 - H. Wang, Y. Pei, A. D. LaLonde and G. J. Snyder, *Proc. Natl. Acad.* Sci. U. S. A., 2012, **109**, 9705-9709.
- 6. H. Wang, E. Schechtel, Y. Pei and G. J. Snyder, *Advanced Energy Materials*, 2013, **3**, 488-495.
- Y. Pei, X. Shi, A. LaLonde, H. Wang, L. Chen and G. J. Snyder, *Nature*, 2011, 473, 66-69.
- K. Biswas, J. He, I. D. Blum, C. I. Wu, T. P. Hogan, D. N. Seidman, V. P. Dravid and M. G. Kanatzidis, *Nature*, 2012, 489, 414-418.
- 9. H. Wang, Z. M. Gibbs, Y. Takagiwa and G. J. Snyder, *Energy Environ. Sci.*, 2014, 7, 804.
- L. D. Zhao, S. Hao, S. H. Lo, C. I. Wu, X. Zhou, Y. Lee, H. Li, K. Biswas, T. P. Hogan, C. Uher, C. Wolverton, V. P. Dravid and M. G. Kanatzidis, *Journal of the American Chemical Society*, 2013, 135, 7364-7370.
- L. D. Zhao, J. He, C. I. Wu, T. P. Hogan, X. Zhou, C. Uher, V.
 P. Dravid and M. G. Kanatzidis, *Journal of the American Chemical Society*, 2012, **134**, 7902-7912.
- 85 12. L. D. Zhao, S. H. Lo, J. He, H. Li, K. Biswas, J. Androulakis, C. I. Wu, T. P. Hogan, D. Y. Chung, V. P. Dravid and M. G. Kanatzidis, *Journal of the American Chemical Society*, 2011, 133, 20476-20487.
 - Q. Zhang, B. Liao, Y. Lan, K. Lukas, W. Liu, K. Esfarjani, C. Opeil, D. Broido, G. Chen and Z. Ren, *Proc. Natl. Acad. Sci. U. S. A.*, 2013, **110**, 13261-13266.
 - L. D. Zhao, S. H. Lo, Y. Zhang, H. Sun, G. Tan, C. Uher, C. Wolverton, V. P. Dravid and M. G. Kanatzidis, *Nature*, 2014, 508, 373-377.
- 95 15. I. Lefebvre, M. A. Szymanski, J. Olivier-Fourcade and J. C. Jumas, *Phys Rev B*, 1998, **58**, 1896-1906.
 - T. Chattopadhyay, J. Pannetier and H. G. Vonschnering, J. Phys. Chem. Solids, 1986, 47, 879-885.
 - H. Wiedemeier and H. G. von Schnering, Zeitschrift für Kristallographie - Crystalline Materials, 1978, 148, 295-303.
 - A. D. LaLonde, T. Ikeda and G. J. Snyder, *Rev Sci Instrum*, 2011, **82**, 025104.
 - S. Iwanaga and G. J. Snyder, J. Electron. Mater., 2012, 41, 1667-1674.
 - S. Iwanaga, E. S. Toberer, A. LaLonde and G. J. Snyder, *Rev* Sci Instrum, 2011, **82**, 063905.
 - K. A. Borup, E. S. Toberer, L. D. Zoltan, G. Nakatsukasa, M. Errico, J. P. Fleurial, B. B. Iversen and G. J. Snyder, *Rev Sci Instrum*, 2012, 83, 123902.
 - A. S. Pashinkin, A. S. Malkova, V. A. Fedorov and M. S.

Journal of Materials Chemistry A Accepted Manuscript

Mikhailova, Inorg. Mater., 2006, 42, 593-595.

- K. Yamaguchi, K. Kameda, Y. Takeda and K. Itagaki, *Mater. Trans. JIM*, 1994, **35**, 118-124.
- 24. D. Stroud, Phys. Rev. B, 1975, 12, 3368-3373.
- 5 25. V. Guttal and D. Stroud, *Phys. Rev. B*, 2006, 73.
- 26. J. Umeda, J. Phys. Soc. Jpn., 1961, 16, 124.
- B. L. Evans and Hazelwoo.Ra, Journal of Physics D-Applied Physics, 1969, 2, 1507-1516.
- A. Agarwal, P. D. Patel and D. Lakshminarayana, J. Cryst.
 Growth, 1994, **142**, 344-348.
- 29. A. Agarwal, M. N. Vashi, D. Lakshminarayana and N. M. Batra, *J. Mater. Sci.-Mater. Electron.*, 2000, **11**, 67-71.
- H. J. Goldsmid and J. W. Sharp, J. Electron. Mater., 1999, 28, 869-872.
- R. Car, G. Ciucci and L. Quartapelle, *physica status solidi (b)*, 1978, 86, 471-478.
 - J. D. Wasscher, W. Albers and C. Haas, *Solid-State Electron.*, 1963, 6, 261-264.
- 33. D. P. Spitzer, J. Phys. Chem. Solids, 1970, **31**, 19-40.
- 20 34. E. S. Toberer, A. Zevalkink and G. J. Snyder, *Journal of Materials Chemistry*, 2011, 21, 15843-15852.
 - 35. S. Asanabe, J. Phys. Soc. Jpn., 1959, 14, 281.
 - W. Albers, C. Haas, H. Ober, G. R. Schodder and J. D. Wasscher, *J. Phys. Chem. Solids*, 1962, 23, 215-220.
- T. H. Patel, R. Vaidya and S. G. Patel, *Bull. Mat. Sci.*, 2003, 26, 569-574.

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