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**Anaerobic ammonium oxidation (ANAMMOX) sludge immobilized by waterborne polyurethane and its nitrogen removal performance-a lab scale study** 

**Guanghui Chen<sup>a</sup> , Jun Lia, \*, Salma Tabassumb,\*, Zhenjia Zhang<sup>b</sup>**

*<sup>a</sup>Key Laboratory of Beijing for Water Quality Science and Water Environment Recovery Engineering, Beijing University of Technology, Beijing 100124, China. E-mail: ghui0066@163.com b School of Environmental Science and Engineering, Shanghai Jiao Tong University,* 

*Shanghai 200240, China.* 

*Corresponding authors Tel.:* **+86 15221195745, +86 13611249208; fax:** 

**+8601067391726-8002, +86 021-54740836** 

**E-mail addresses: salmazenith@gmail.com (S. Tabassum), lijunbjut@163.com (Li.** 

**Jun)** 

# **ABSTRACT:**

In the present study, to maintain the biomass of anaerobic ammonium oxidation bacteria (ANAMMOX) in water, waterborne polyurethane (WPU) was used to immobilize ANAMMOX sludge. The ANAMMOX granules immobilized by WPU exhibited best entrapment support, superb bioactivity and highest mechanical stability when compared with the performance of other ANAMMOX immobilized granules of polyvinyl alcohol (PVA), sodium alginate (SA), and the mixed material, PVA-SA. In a continuous-flow experiment, during high volume-load WPU-immobilized granules exhibited a relatively high total nitrogen removal rate, no effluent suspended solids or granule crushing were observed within 100days. The highest total nitrogen removal rate reached 80.98% was observed when the volume load of TN was  $1.697 \text{ kg/(m}^3\text{-d)}$  with HRT 1.5hr. WPU-immobilized granules exhibited better correlation  $(R^2: 0.945 NH_4^{\text{-}}N$  and 0.989 NO<sub>2</sub>-N) with experimental kinetic data for biological nitrogen removal. According to experimental data, good sludge retaining ability, strong resistance to shock-loading and mechanical stability during long-term operation of the granules immobilized by WPU was observed. The formation of volcanic crater-like concaves on the two sides of each bacterium and large amount of ANAMMOX bacteria grew along the channels inside of the WPU granule were observed by scanning electron microscopy. Finally, the microbial community analysis 16S rDNA cloning revealed that Candidatus *Brocadiafulgida*  (JX243641.1) was the primary ANAMMOX bacteria inside the WPU-immobilized granule*.* As carriers, immobilization materials protect ANAMMOX bacteria and

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increase biomass; however, they have no effect on bacteria and bacterial community structure.

**Keywords:** anaerobic ammonium oxidation, immobilization technique, biological

nitrogen removal, waterborne polyurethane

# **1. Introduction**

Traditional biological nitrogen removal techniques primarily remove the nitrogen element in effluent through nitrification-denitrification<sup>1</sup>. Anaerobic ammonium oxidation (ANAMMOX) can oxidize ammonia nitrogen to  $N_2$  using nitrite as the electron acceptor under anaerobic or anoxic conditions<sup>2</sup>. Compared with traditional biological nitrogen removal techniques, ANAMMOX technique have the following advantages: small site area, low energy consumption, no additional organic carbon source, and low surplus sludge volume<sup>3</sup>. However, ANAMMOX bacteria are autotrophic bacteria; they grow slowly and have a low cell yield. In addition, they are also affected by environmental conditions like temperature, water quality, etc.<sup>4, 5</sup> Therefore, starting and operation of ANAMMOX technique have been extremely difficult, which limits its development<sup>6</sup>. Researchers all around the world have conducted extensive studies on the ANAMMOX technique and proposed techniques such as the ANAMMOX strain fed-batch<sup>7, 8</sup>. However, these techniques generally require culturing sufficient high-rate ANAMMOX granules under optimal conditions. There are still issues in the culturing process from flocculent ANAMMOX sludge (FAS) to high-rate ANAMMOX granules (HAG), including loss of bacteria and slow granule formation<sup>9</sup>. Therefore, retaining microbes and ensuring the biomass of ANAMMOX bacteria have become a primary objective for developing the ANAMMOX technique<sup>10</sup>.

The embedded immobilization technique is a new microbe immobilization technique in the modern bioengineering field. The embedded immobilization technique

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immobilizes free cells or enzymes in a constrained area through immobilization materials, such that the activity of the cells or the enzymes can be maintained and reused. The embedded immobilization technique provides a good retaining effect on microbes<sup>11-13</sup>. Studies have shown that immobilizing ANAMMOX sludge using the embedded immobilization technique can effectively prevent the loss of ANAMMOX bacteria and maintains the biomass in the reactor, and thus, this technique appears promising $14-16$ . However, there are disadvantages of the immobilized ANAMMOX granules prepared using the existing immobilization materials, such as insufficient mechanical strength, low bioactivity, and lack of stability during long-term operation. Several methods have been used to enhance immobilized granules; however, there has been no significant improvement<sup>17-19</sup>. Therefore, a new material, waterborne polyurethane (WPU)<sup>20</sup>, was selected in this study. In addition, natural polymer, sodium alginate (SA), synthesized polymer, polyvinyl alcohol (PVA), and mixed polymer PVA-SA were compared. ANAMMOX sludge was immobilized and then investigated in the aspects of mechanical stability, ANAMMOX performance with different immobilization materials, and shock-loading resistance during long-term operation to provide a basis for further study. This study also aims to investigate the removal efficiencies of WPU-, SA-, PVA-SA-, PVA-immobilized granules for  $NH_4^+$ -N and  $NO<sub>2</sub>$ -N.

#### **2. Materials and methods**

#### **2.1. Immobilization materials**

ANAMMOX sludge used in this study was obtained from an up flow anaerobic sludge blanket (UASB) reactor that was operated stably for more than 2 years in our laboratory<sup>21</sup>. Sequencing analysis revealed that the primary bacteria in the sludge were Candidatus *Brocadiafulgida* (JX243641.1). Sludge was washed three times with phosphate buffered saline (PBS,  $pH=7.4$ , 0.1 M) solution to remove any residual substrate on the sludge surface. Afterwards, the sludge was centrifuged for 10min at 4000 rpm.

Analytical pure immobilization agents (WPU, PVA, SA, and PVA-SA) and cross linking agents/initiators  $(CaCl_2, H_3BO_3, \text{pot}$  potassium persulfate (KPS), and tetramethylethylenediamine (TMEDA)) were used.

# **2.2. Water quality**

Synthetic wastewater was used. The main components of wastewater are listed in Table 1. Trace elements I and II were according to the reference<sup>22</sup>. The pH of the synthetic wastewater ranged from 7.1 to 7.84.

#### **2.3. Immobilization procedure**

The preparation methods for all the immobilized granules were described as follows (mass percentage): *WPU-immobilized granules*: 10% WPU solution and ANAMMOX sludge concentrate of the equivalent volume were mixed homogeneously. To initiate polymerization, the initiators TMEDA and KPS were added, and the solution

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was stirred rapidly. It will take approximately 30min to become a gel. The resulting solid gel was cut into  $3 \times 3 \times 3$ mm cubes to obtain WPU-immobilized granules.

The Molecular Weight (MW) distribution of WPU in hydrogel had been investigated in our previous research<sup>23</sup>. The MW of WPU is determined to be  $5100g/mol$  under the condition that gelation reaction occurred in 10% solution. We selected 10% solution medium to form WPU polymer since such generated WPU polymer could ensure sufficient mechanical strength and mass transfer property, so the WPU immobilized granules prepared was more significant.

*PVA-immobilized granules:* Equivalent volume of 10% PVA solution and ANAMMOX sludge concentrate were mixed homogeneously. Using a pipette the mixed solution (PVA solution and ANAMMOX sludge) was then added drop by drop in  $H_3BO_3$  solution. The obtained solution was stored at  $4^{\circ}$ C for 12h in refrigerator for cross linking to obtain spherical PVA-immobilized granules.

*SA-immobilized granules:* Equivalent volumes of 2% SA solution and ANAMMOX sludge concentrate were mixed. Using a pipette the mixed solution was added drop by drop in CaCl<sub>2</sub> solution (4%). The obtained solution was stored at  $4^{\circ}$ C for 12h in refrigerator for cross linking to obtain spherical SA-immobilized granules. *PVA-SA-immobilized granules*: Firstly 8% PVA solution and 2% SA solution were mixed, and then mixed with an equivalent volume of ANAMMOX sludge concentrate. Using a pipette the mixed solution was added drop by drop in CaCl<sub>2</sub> solution  $(4\%)$ . The obtained solution was stored at 4˚C for 12h in refrigerator for cross linking to obtain spherical PVA-SA-immobilized granules.

#### **2.4. Activation of immobilized granules**

To recover the activity of the ANAMMOX bacteria in the immobilized granules, activated culture was performed on the prepared immobilized granules. The prepared immobilized granules were collected. Deionised (DI) water was used to wash the immobilized granules thoroughly until the effluent cleared. Immobilized granules were then transferred into the synthetic wastewater for activated culture and batch culture $^{24}$ . One activation cycle was completed, when the concentration of  $NH_4^+$ -N in the effluent decreased to 20% with respect to the concentration of  $NH_4^+$ -N in the influent. After the completion of one cycle, the wastewater was replaced with fresh synthetic wastewater. After 1w of activation, the immobilized granules were taken out for later use.

# **2.5. Determination of mechanical stability of the immobilized granules**

Mechanical strength, expansion coefficient, tensile strength and swelling properties were used as evaluation indices to determine the mechanical stability of the activated immobilized granule. *Mechanical strength:* In a serum bottle reactor immobilized granules (30) of similar size of each type were selected and placed (Figure 1a). DI water 400mL was added to the reactor. The mixture was then stirred magnetically at 600rpm for 48h, after that we observed the ratio of the number of intact immobilized granules to the original number of immobilized granules. *Expansion coefficient:* In a 500mL conical flask immobilized granules (20) of similar size of each types were added, approximately 400mL DI water was added in the flask. The flask was then slowly shaken at 32˚C for

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72h. Diameters of the immobilized granules before and after the treatment were measured with vernier caliper. The expansion coefficient was determined by the ratio of the mean diameter of immobilized granules after 72 h of treatment to the mean diameter of the original immobilized granules.

*Tensile strength:* Tensile strength for four types of immobilized granules was evaluated using a particle strength tester (YKHC-3A, Heng odd, China). Immobilized granules with similar sizes were placed on a pan located in the middle of the tester. The physical stress was exerted on the granules, the maximum stress value was automatically recorded and tensile strength was calculated. Each experiment was repeated 10 times, and the average tensile strength was recorded. *Swelling properties*: In a 500mL conical flask immobilized granules (20) with similar sizes of each type were added, followed by the addition of 400mL synthetic wastewater. Then the flask was slowly shaken at 32˚C for 1w, changes in the immobilized granules were observed.

# **2.6. Measurement of ANAMMOX performance**

Apparatus for measuring the ANAMMOX performance of immobilized granules is shown in Figure 1a. Forty (40) millilitres activated immobilized granules (including approximately 20mL concentrated sludge) of each type was added to a 500mL serum bottle with a cap. Serum bottle was covered with black reflector paper. Then, 400mL of synthetic wastewater was added to the serum bottle. The bottle was then placed on the constant temperature magnetic stirrer. To ensure an anaerobic environment, high purity  $N<sub>2</sub>$  flowed into the bottle through the air inlet for 30min to air-strip the dissolved oxygen

in the water. Temperature was controlled at  $32^{\circ}$ C, Magnetic stirrer (regulated speed 150 rpm) while the pH was not controlled. Sampling was conducted every 8h through the air inlet using a syringe. Twenty (20) millilitres un-immobilized ANAMMOX sludge concentrate was used to prepare a 40mL mixed solution as a control group for synchronous experiments. Two to three repeats were conducted for each experiment. Later, the results obtained were averaged.

#### **2.7. Continuous-flow experiment**

An up flow stirred reactor was used for the continuous-flow experiment as shown in Figure 1b. The reactor was made of polymethyl methacrylate (PMMA) and the effective reactor volume was 2L. A stirring device was placed on top of the reactor to ensure that the immobilized granules were stirred homogeneously. The effluent flows through the overflow weir. To prevent the immobilized granules from flowing out with the water a mesh was placed at the water outlet. There was a water bath layer on the external wall of the reactor. The water bath was used to maintain the temperature inside the reactor at 32˚C. After 1w of activation, immobilized granules (the amount was based on 10% of the volume filling ratio) were added to the reactor. Synthetic wastewater was used in the experiment; the concentrations of  $NH_4^+$ -N and  $NO_2$ -N were 40~60mg/L and 40~60mg/L, respectively. The concentrations of other components are listed in Table 1. During the experiment, the hydraulic retention time (HRT) was continuously decreased; such indices as  $NH_4^+$ -N, NO<sub>2</sub> -N, NO<sub>3</sub> -N, total nitrogen (TN), mixed liquor suspended solids (MLSS), and mixed liquor volatile suspended solids (MLVSS) in the effluent

were determined.

#### **2.8 Kinetic measurements**

20 mL of WPU- immobilized granules and 200 mL of synthetic wastewater were mixed in a 300mL serum bottle. The  $NH_4^+$ -N and  $NO_2^-$ -N concentrations in the simulated wastewater corresponded to the required concentrations. The serum bottles with mixed solutions were then sealed, vented by high-purity  $N_2$  to replace the dissolved oxygen (DO) and then placed in an incubator shaker at a constant temperature of 32 °C with a speed of 80 RPM. Samples were taken every six hours to calculate the  $NH_4^+$ -N and NO<sub>2</sub>-N removal rates. All tests were repeated three times.

# **2.9. Scanning electron microscopy (SEM) and molecular biological analysis**

*SEM analysis of immobilized granules:* immobilized granules were taken out of the reactor, washed, and pentanediol 25%was used to immobilize the granules for 1.5h. Then, granules were washed three times in PBS. Subsequently, the granules were dehydrated in an ascending series of ethanol (volume fraction: 50%, 70%, 80%, 90% and 100%); dehydration period was 10~15min each time. Lastly, isoamyl acetate was used for displacement. After the granules were freeze-dried for 24h, an approximately1500-nm-thick layer of metallic membrane was electroplated on the surfaces of the samples

*For molecular biological analysis:* 16S rDNA-cloning was used. First, immobilized granules were ground in a mortar. Then, an ultrasonic processor (amplitude: 50%) was used to crush the granules for 60s. DNA was extracted. Nested polymerase chain

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reaction (PCR) was used to qualitatively amplify DNA fragment: a primer consisting of the universal primer for planctomyces, pla46f (5'-GGATTAGGCATGCAAGTC-3'), and the universal primer for bacteria, 630r (5'-CAKAAAGGAGGTGATCC-3'), was used for the first run of PCR amplification. The reaction system of PCR (total volume: 50 µL) was as follows:  $5\mu L$  10×PCR buffer, 2 $\mu$ L MgCl<sub>2</sub>, 1.5 $\mu$ L dNTP (2.5mmol•L<sup>-1</sup> each),  $0.5\mu L$  pla46f ( $20\mu$ mol $\text{L}^{-1}$ ),  $0.5\mu L$  630r ( $20\mu$ mol $\text{L}^{-1}$ ), 1 $\mu$ L Taq DNA polymerase (5U), and 0.5µL DNA template; finally, 50µL ddH20 was added. The reaction program for PCR was as follows: initial denaturation at 96˚C for 10min; denaturation at 96˚C for 1min, renaturation at 72˚C for 1min and extension at 72˚C for 1min, 35 cycles in total; and last extension at 72˚C for 10min.Subsequently, the specific primer for ANAMMOX bacteria was used to amplify Amx368f (5'-TTCGCAATGCCCGAAAGG-3') and Amx820r (5'-AAAACCCCTCTACTTAGTGCCC-3'). The reaction system and program of PCR were the same as in the first-run amplification system. The reaction products were analyzed using 1.2% agarose gel electrophoresis detection<sup>25, 26</sup>.

# **2.10. Analytical methods**

 $NH_4^+$ -N, NO<sub>2</sub> -N, NO<sub>3</sub> -N, TN, MLSS and MLVSS were determined according to standard methods<sup>27</sup>. Temperature and pH values were determined by WTW/Multi 3420 multiparameter. For SEM Hitachi S-4300 was used.

# **3. Results and discussion**

#### **3.1. Preparation of immobilized granules**

Figure 2 shows digital images of the 4 types of immobilized granules. The PVA, SA,

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and PVA-SA immobilized granules were all spherical, and their diameters ranged from 2 to 4mm. WPU-immobilized granules were  $3\times3\times3$ mm cubes. The SA-immobilized granules were gel-like and showed a red brown colour, they were elastic. The surfaces of the PVA-immobilized granules and the PVA-SA-immobilized granules exhibited a slight white colour; their elasticity was relatively poor. A severe conglutinating phenomenon occurred in the PVA-immobilized granules during the cross linking period and the later operation period. The WPU-immobilized granules were transparent gels and exhibited a jacinth colour, which is the colour of ANAMMOX sludge; they were very elastic.

# **3.2. Stability of the immobilized granules**

Table 2 lists the mechanical stability indices for the 4 types of immobilized granules. After 1w of stirring in synthetic wastewater, PVA-SA-immobilized granules and SA-immobilized granules became soft and fragile.

 The mechanical strength of SA-immobilized granules was the lowest; after 48h of high-speed rotation, over 50% of the SA-immobilized granules broke. The WPU-immobilized granules exhibited the best stability; after 1w of stirring, their swelling properties remained essentially the same. The expansion coefficient of the WPU-immobilized granules was 1.04; after 48h of high-speed rotation, none of the granules broke. The PVA-immobilized granules also exhibited relatively good stability; after high-speed rotation, 26% of the granules broke. The expansion coefficient of the PVA-immobilized granules was 1.19. However, the conglutinating phenomenon in the

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PVA-immobilized granules was the most severe, and they had the worst balling property. After calculating the tensile strength for each immobilised granule (Table 2), WPU showed the best tensile strength  $(6.1 \text{ kg/m}^2)$  while SA was lowest  $(2.2 \text{ kg/m}^2)$ . The tensile strength for PVA and PVA-SA was 4.6 kg/m<sup>2</sup> and 3.5 kg/m<sup>2</sup>, respectively. Considering all the four indices, the order of mechanical stability was as follows (from strongest to weakest): WPU>PVA>PVA-SA>SA.

The differences in the mechanical stability of the immobilized granules were primarily caused by the properties of the immobilization materials. PVA is a synthetic polymer and has good physical properties; the immobilized granules prepared using PVA exhibited good mechanical strength and swelling properties. However, there are many hydrophilic hydroxyl groups in PVA molecules, which have certain water swelling properties and auto-condensing tendencies<sup>28</sup>. Therefore, PVA-immobilized granules tended to conglutinate, which decreased the specific surface area of the immobilized granules and thus affected the mass transfer property. SA is a natural polymer. The size of the network structure formed by SA and the  $Ca^{2+}$  in the cross linking agent was relatively large, while the mechanical strength of the network structure was low. If there are  $Mg^{2+}$  and phosphates in the water, then this network structure will be damaged, resulting in the swelling and crushing of SA-immobilized granules. SA was introduced into PVA, and PVA-SA-immobilized granules were prepared. The macro-porous network structure of SA molecules can improve the mass transfer property of PVA, while the -OH bonds on the molecular chains of PVA and the

-COO<sup>-</sup> in SA form hydrogen bonds, which can also improve the conglutination behaviour of PVA-immobilized granules. However, due to the existence of SA, the swelling properties of immobilized granules will be affected; the granules will become soft, swollen, and crushed during long-term operation.

WPU is a synthetic polymeric material; it has good biocompatibility and excellent mechanical strength<sup>29</sup>. When cross-linked, WPU forms a stable gel. Figure 3 shows that WPU gel is a polymer with a three-dimensional network structure; the relative molecular mass and density are relatively large; the space of this three-dimensional network structure is relatively small, which not only ensures the mechanical strength of the WPU-immobilized granules but also absorbs less water when swelling; therefore, the WPU-immobilized granules exhibited the best mechanical stability among the 4 types of immobilized granules.

# **3.3. ANAMMOX performance of the immobilized granules**

Figure 4a shows the variation curve of the concentration of  $NH_4^+$ -N with respect to time in each reactor. The concentrations of  $NH_4^+$ -N decreased first rapidly and then slowly in both the control group and the immobilization group. As the initial concentration of  $NH_4^+$ -N was high and the ANAMMOX rate was fast, while the later decrease in the substrate concentration decreased the reaction rate. During 0~8h, the concentration of  $NH_4^+$ -N in the control group decreases swiftly. After 16h, the removal rate for the immobilization group surpassed the removal rate for the control group. After 40h, the removal efficiency decreased in both the control group and the immobilization

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group. Eventually, at 50h, NH<sub>4</sub><sup>+</sup>-N was completely removed in all reactors. The variation pattern of the concentration of  $NO<sub>2</sub>$ -N was basically the same as the variation pattern of the NH<sub>4</sub><sup>+</sup>-N. During 0~8h, the removal rates for NH<sub>4</sub><sup>+</sup>-N and NO<sub>2</sub><sup>-</sup>-N in the control group were higher than in the immobilization group, which was mainly because the existence of the immobilization material prevented the mass transfer between the materials in the water, such as  $NH_4^+$ -N, and the ANAMMOX bacteria in the immobilized granules. After 16h, the removal rates for  $NH_4^+$ -N and  $NO_2^-$ -N in the immobilization group were higher than in the control group due to the dispersion of the substrate into immobilized granules, the more concentrated biomass of the immobilized microbes resulted in higher activity of the ANAMMOX bacteria.

The removal efficiencies of WPU-, SA-, PVA-SA-, PVA-immobilized granules for  $NH_4^+$ -N and  $NO_2^-$ -N were also different. During 8~24h, the substrate completely dispersed into the immobilized granules. Degradation of  $NH_4^+$ -N and  $NO_2^-$ -N exhibited good linear correlation. By computation, the average removal rates of WPU-, SA-, PVA-SA-, PVA-immobilized granules, and the control group for NH<sub>4</sub><sup>+</sup>-N were obtained, which were 3.194, 3.122, 2.683, 2.419 and 2.363mg/h, respectively; removal rates for NO<sub>2</sub>-N were 4.053, 3.896, 3.353, 3.051 and 3.021mg/h, respectively. The order of removal efficiency for  $NH_4^+$ -N and  $NO_2^-$ -N (from fastest to slowest): WPU>SA>PVA-SA>PVA>control group.

By comparison with the variation yield curve of  $NO<sub>3</sub>$ -N with respect to time in Figure 4c, the ratio of the removed amount of  $NO<sub>2</sub>$ -N in the control group to the

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removed amount of  $NH_4^+$ -N was obtained (1.201), which was slightly lower than the theoretical value (1.32). The mean ratio of the yield of  $NO<sub>3</sub>$ -N to the removal amount of  $NH_4^+$ -N was 0.266, which was slightly higher than the theoretical value (0.26) as, during the preparation of immobilized granules and the subsequent activated culture, a small amount of oxygen was brought into the reaction system; during initial reaction period, the oxygen caused aerobic ammonium oxidation reaction, resulting in a relatively low removal amount of  $NO<sub>2</sub>$ -N and a relatively high yield of  $NO<sub>3</sub>$ -N. If the deviation caused by the oxygen was removed, then the measured ratio would be very close to the theoretical ANAMMOX value.

The ratios of the removed amount of  $NO<sub>2</sub>$ -N to the removed amount of  $NH<sub>4</sub>$ <sup>+</sup>-N of WPU-, SA-, PVA-SA-, and PVA-immobilized granules were 1.227, 1.228, 1.211, and 1.235, respectively; the ratios of the yield of  $NO<sub>3</sub>$ -N to the removed amount of  $NH<sub>4</sub>$ <sup>+</sup>-N were 0.237, 0.232, 0.246, and 0.236, respectively, which were both very close to the ratios in the control group. Thus, these immobilized granules exhibited good ANAMMOX performance, and the order of the ANAMMOX performance was as follows: WPU-immobilized granules>SA-immobilized granules>PVA-SA-immobilized granules>PVA-immobilized granules.

The ANAMMOX properties of different immobilized granules were different and were related to the form of the cross-linked immobilized granules. For each PVA-immobilized granule, there was a homogenous spherical shell; the exterior was relatively dense, and the mass transfer resistance was large. In addition, the

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conglutination of granules resulted in a decrease in specific area, leading to a poor mass transfer property; therefore, the ANAMMOX performance of the PVA-immobilized granules was poorest.

The spherical shell formed on the surface of each SA-immobilized granule was relatively thin, and the mass transfer resistance was relatively small<sup>30</sup>. After cross linking, size of the network structure formed inside was relatively large; therefore, SAimmobilized granules exhibited relatively good ANAMMOX performance. There was also a dense spherical shell on the surface of each PVA-SA-immobilized granule; however, due to the introduction of SA, the macro-porous network structure of the SA molecules improved the mass transfer property of PVA. Therefore, the ANAMMOX performance of PVA-SA-immobilized granules was superior to the performance of PVA-immobilized granules but inferior to the performance of SA-immobilized granules. There was no obvious shell on the surface of each immobilized granule formed from WPU condensation; WPU-immobilized granule was soft, and the mass transfer property was good. Figure 5a and 5b show  $\times$ 1.0k and  $\times$ 4.0k SEM images of the surfaces of WPU-immobilized granules. Figure 5a shows many channels on the surface of each WPU-immobilized granule; bacteria were divided into different areas by the channels. Figure 5b shows that these bacteria were in a spherical shape, and there were volcanic crater-like concaves on the two sides of each bacterium, indicating that these were typical ANAMMOX bacteria. Figure 5c shows the SEM image of the inside of a WPU-immobilized granule after it was cut open; it can be seen that a large amount of

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ANAMMOX bacteria grew along the channels inside of the granule. So, ANNAMOX bacteria were confirmed to exist within and on the granule and the WPU-immobilized granules had the best mass transfer property and ANAMMOX activity.

# **3.4. Comparison and Selection of optimal immobilization materials.**

The different immobilized granules have different properties with different immobilization materials. The key factor of embedding immobilized technology is to select a suitable immobilization material. Table 3 shows the study of imbedded immobilization by different materials, In traditional materials, PVA had the best mechanical strength and stability, whereas the biological activity and mass transfer performance were worst. PVA-SA was more suitable as immobilization material for ANNAMOX. Wang et al.<sup>31</sup> found that sodium carboxymethyl cellulose (CMC) had better mass transfer performance and biological activity in comparison to SA.

After two years Zhu et al.<sup>24</sup> found that the mechanical strength and stability of CMC was bad although it had good mass transfer performance , and CMC was not suitable as immobilization material. Compared to other, WPU had obvious advantages on mechanical strength and mass transfer performance. The results of the present study are also consistent with previous studies<sup>32, 33</sup>. Hence, WPU is a excellent immobilization material for ANAMMOX sludge.

# **3.5. WPU-immobilized granules activity using Continuous-flow experiment**

To further study the nitrogen removal performance and stability of WPU-immobilized granules during long-term operation, a method in which the HRT

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was gradually decreased was used for the continuous-flow experiment.

The HRT was decreased from 8h to 1.5h, and the corresponding volume load of TN was increased from  $0.328\text{kg/(m}^3\text{-d})$  to  $1.697\text{kg/(m}^3\text{-d})$ . Figure 6 shows the variations of the influent and effluent TN of the reactor during the experimental period. During 0~35d, HRT of the reactor was 8h. The concentration of TN decreases rapidly, and the removal rate reached 22% in the first 7 days, after which the removal rate levelled off and the effluent TN remained at approximately 80mg/L. After the activity of the immobilized granules was recovered, the removal rate for TN increased rapidly starting at the  $14<sup>th</sup>$ day; on the  $35<sup>th</sup>$  day, the effluent TN was 14.18mg/L, and the removal rate reached 85%. On the  $36<sup>th</sup>$  day, the HRT was decreased to 5h, and the concentration of the effluent TN increased to 50.34mg/L. However, the removal rate still remained over 50%. As the experiment continued, the removal rate gradually returned to a relatively high level. Afterwards, the HRT was continuously decreased to 3h and 1.5h. The concentration of the effluent TN first increased and then gradually decreased. On 100<sup>th</sup> day, HRT was 1.5h, the effluent TN conc. was 20.17mg/L, and the removal rate reached 80.98%, while the volume load of TN was  $1.697 \text{kg/(m}^3 \cdot \text{d})$ .

Figure 6 shows that in the initial stage of the experiment, in which the HRT was decreased and the volume load increased, there was a stable period for the removal rate for TN in all cases, indicating that the WPU-immobilized granules exhibited very strong shock-loading resistance. When the load increased, the existence of the immobilization material would slow down the impact of the suddenly increased TN on the ANAMMOX

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bacteria, and thus, the removal efficiency was stable. After a while, when the ANAMMOX bacteria gradually adapted to the load variation, the activity of ANAMMOX increased, and the removal rate for TN also increased rapidly, which was consistent with the phenomenon reported by Bae et al.<sup>34</sup> in which the immobilization technique was used to treat textile wastewater. In addition, the effluent of the reactor was always clear during the experimentation period, and no suspended solids (SS) were detected. None of the immobilized granules broke. The WPU-immobilized granules exhibited good sludge retaining ability and mechanical stability.

# **3.6. Biological nitrogen removal kinetics**

The kinetic characteristics of WPU-immobilized granules are also studied using the Haldane model (Equation  $(1)$ )<sup>35</sup>.

$$
V = \frac{V_{\text{max}}}{1 + \frac{K_{\text{m}}}{S} + \frac{S}{K_{\text{i}}}}
$$
(1)

Where, V is the reaction rate, S is the substrate concentration,  $V_{\text{max}}$  is the maximum reaction rate, Km is the half-saturation constant and Ki is the half-suppression constant. Non-linear fitting was performed on the experimental data using the Haldane model with Origin 8.0 to derive the kinetic functions for NH4+-N (Figure 7(a)) and NO2--N (Figure 7(b)). The fitting constants  $R^2$  were 0.945 and 0.989, indicating good correlations. The data in Figure 7(a) showed that the obtained maximum nitration reaction rate of the immobilized granules was  $1.97 \text{ mg/(mg/d)}$ , the ammonium half-saturation constant was 1.42 mmol/L, and the ammonium half-suppression constant

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was 817.8mmol/L. The data in Figure 7(b) showed that the maximum nitrate reaction rate was 2.39mg/(mg/d), the nitrate half-saturation constant was 1.042 mmol/L, and the nitrate half-suppression constant was 89.95 mmol/L.

Compared to ANAMMOX sludge, the kinetic of immobilized granules showed higher maximum reaction rate, the increased reaction rate promotes the conversion of substrates, relaxes the self-suppression of substrates, and increases ANAMMOX sludge production. The results from the kinetic study also showed that the half-saturation constants of the immobilized particles for ammonium and nitrite were lower than ANAMMOX sludge, indicating that the immobilized particles better utilize the substrate in adverse environments. The half-suppression constants of the immobilized particles for ammonia and nitrite were higher than ANAMMOX sludge, so the immobilized particles could survive longer in high-concentration substrates.

#### **3.7. Bacterial community structure analysis**

To study the effect of the immobilization material (WPU) on the ANAMMOX bacteria, as well as the variation of the bacterial community structure in the immobilized granules, WPU-immobilized granules that had been in stable operation for 100d in the reactor were selected for the 16S rDNA-cloning experiment. Forty-eight (48) 16SrDNA clones were obtained. After PCR detection, 43 positive clones were obtained. The 43 positive clones were sent to a sequencing company for sequence analysis. After the sequence analysis was completed, the result was Blast-aligned against the gene bank.

Clones with the same sequence alignment result were defined as an operational taxonomic unit (OTU). Four (4) OTUs were obtained after sequence alignment. A phylogenetic tree was built using the final sequences and similar sequences to these OTUs (Figure 8).

The analysis through the phylogenetic tree and alignment against the gene bank revealed that the homologies between the 43 clones and the ANAMMOX bacteria with known sequences were all over 99%. The 4 OTUs were closest to 4 types of ANAMMOX bacteria (Table 4).

The sequence for the ANAMMOX bacteria with the highest similarity among all the cloning and sequencing results was Candidatus *Brocadiafulgida* (JX243641.1), for which there were 26 sequencing results with 100% similarity. The sequence similarities between the remaining 17 sequencing results and the uncultured ANAMMOX bacteria were all 99%. Therefore, the primary ANAMMOX bacteria inside the immobilized granules were Candidatus *Brocadiafulgida* (JX243641.1), which was consistent with the enriched type of ANAMMOX bacteria inside the reactor before immobilization. It can be seen from the experimental result of 16S rDNA cloning that the immobilized ANAMMOX bacteria exhibited relatively strong shock-loading resistance and ANAMMOX performance, which was primarily due to the effect of the immobilization material instead of the variation of the bacteria. The immobilization material served as a carrier and provided protection; it cushioned the external load variation through the layered network structure, reduced the impact on the ANAMMOX bacteria, and increased the shock-loading ability. In addition, the immobilization material served as a carrier inside the immobilized granules; the ANAMMOX bacteria proliferated copiously, and thus, the biomass concentration increased. Therefore, the ANAMMOX performance of the immobilized granules increased.

#### **Conclusions**

Comprehensively considering mechanical stability and ANAMMOX performance, WPU has significant advantages compared to the other materials. WPU was an excellent immobilization material for ANAMMOX sludge. The WPU-immobilized granules maintained good sludge retaining ability and mechanical stability during long-term operation and exhibited very strong shock-loading resistance. The removal of total nitrogen was higher than 80% at HRT 1.5h, and continuous flow experiment was stable. Biological nitrogen removal kinetics showed that WPU-immobilized granules exhibited better ANNAMOX kinetic characteristics. Microbial community analysis showed that the dominant ANAMMOX bacteria inside the WPU-immobilized granules were Candidatus *Brocadiafulgida* (JX243641.1). As the carrier, the immobilization material protected the bacteria and increased the biomass; it had no effect on the ANAMMOX bacteria and the bacterial community structure, Consequently, ANAMMOX sludge immobilized by waterborne polyurethane could be a promising strategy for treating wastewater containing high level of nitrogen.

# **Acknowledgements**

This research was supported by National Critical Project for Science and Technology on Water Pollution Prevention and Control of China (Nos. 2014ZX07201-011) and Natural Science Foundation of Bei Jing (BJNSF, Grant Nos.8122005)

# References

- 1. U. V. Dongen, M. S. M. Jetten and M. C. M. VanLoosdrecht, *Water Science & Technology* 2001, 44 153-160.
- 2. A. A. V. Graaf, P. D. Bruijn, L. A. Robertson, M. S. M. Jetten and J. G. Kuenen, *Microbiology*, 1997, 143, 2415-2421.
- 3. M. S. M. Jetten, M. Wagner, J. Fuerst, M. V. Loosdrecht, G. Kuenen and M. Strous, *Current Opinion in biotechnology*, 2001, 12, 283-288.
- 4. C. Tang, P. Zheng, Q. Mahmood and J. Chen, *Journal of Central South University of Technology*, 2010, 17, 79-84.
- 5. C. J. Tang, P. Zheng, T. T. Chen, J. Q. Zhang, Q. Mahmood, S. Ding, X. G.Chen, J. W. Chen and D. T. Wu, *Water Res*, 2011, 45, 201-210.
- 6. C. Tang, L. Xiong, Y. Wang and P. Zheng, *Environmental Science*, 2013, 34, 3544-3551.
- 7. B. Wett, *Water Science & Technology*, 2006, 53, 121-128.
- 8. W. R. L. V. Star, W. R. Abma, D. Blommers, J.-W. Mulder, T. Tokutomi, M. Strous, C. Picioreanu and M. C. M. V. Loosdrecht, *Water Research*, 2007, 41, 4149-4163.
- 9. B. Zu, D. Zahang and Q. Yan, *Environmental Science*, 2008, 29, 683-687.
- 10. S. Qiao, T. Tian, X. Duan, J. ZHOU and Y. CHENG, *Chemical Engineering Journal,*, 2013, 230, 19-26.
- 11. Y. Chen, T. Zhou and S. Yao, *Journal of microbiology and biotechnology*, 2013, 23, 511-517.
- 12. J. Huang, M. Tang and C. M. Li, *RSC Advances*, 2014, 4, 46498-46501.
- 13. D. Zhang, Y. Zhang, F. Shen, J. Wang, W. Li, E. Li and J. Falandysz, *Desalination and Water Treatment*, 2014, 52, 4792-4801.
- 14. S. Riham and H. E. Muftah, *Journal of Water Process Engineering*, 2014, 1, 84-90.
- 15. R. Chen, Y. Ma, C. Zhao, Z. Lin, X. Zhu, L. Zhang and W. Yang, *RSC Advances*, 2014, 4, 46653-46661.
- 16. D. Yamei, Z. Zhenjia, J. Yongwei, L. Zhirong and L. Jian, *Journal of Environmental Sciences*, 2011, 23, 366-371.
- 17. A. Magri, M. B. Vanotti and A. A. Szogi, *Bioresource Technology*, 2012, 114, 231-240.
- 18. Y. Z. Zhou, J. Yang, X. L. Wang, Y. Q. Pan, H. Li, D. Zhou, Y. D. Liu, P. Wang, J. D. Gu, Q. Lu, Y. F. Qiu and K. F. Lin, *Environ Sci Pollut Res*, 2014, 1-10.
- 19. C. Zhong, P. Liu, W. Zhang, J. Liu and X. Zeng, *Chinese Journal of Environmental Engineering*, 2013, 7, 2837-2843.
- 20. Y. Dong, Z. Zhang, Y. Jin, J. Lu, X. Cheng, J. Li, Y.-y. Deng, Y.-n. Feng and D. Chen, *Journal of Environmental Sciences,*, 2012, 24, 999-1005.
- 21. Z. Li, C. Liu, B. Zhao, J. Ma, X. Wang and J. Li, *China Environmental Science* 2013, 4, 648-654.
- 22. M. Strous, E. V. Gerven, P. Zheng, J. G. Kuenen and M. S. M. Jetten, *Water Research*, 1997, 31, 1955-1962.
- 23. Y. M. Dong, Ph.D Shanghai jiaotong university, 2012.
- 24. G. L. Zhu, Y. Y. Hu and Q. R. Wang, *Water Sci Technol*, 2009, 59, 2379-2386.
- 25. G. Zhu, S. Wang, Y. Wang, C. Wang, N. Risgaard-Petersen, M. S. Jetten and C. Yin, *ISME J.* , 2011,

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5, 1905-1912.

- 26. M. Z. Khan, S. Singh, T. R. Sreekrishnan and S. Z. Ahammad, *RSC Advances*, 2014, 4, 46851-46859.
- 27. D. E. Andrew and H. F. M. Ann, 21st edition edn., 2005.
- 28. J. C. J. F. Tacx, H. M. Schoffeleers, A. G. M. Brands and L. Teuwen, *Polymer*, 2000, 41, 947-957.
- 29. S. J. Lee, S. K. Lee and B. K. Kim, *Journal of Macromolecular Science, Part B*, 2014, 53, 254-264.
- 30. Z. H. Xiong, C. Z. Sun and Q. C. Liu, *Environmental Science and Technology*, 2005, 28, 82-84.
- 31. B. E. Wang and Y. Y. Hu, *Journal of Environmental Sciences*, 2007, 19, 451-457.
- 32. M. Y. Zhao and Z. F. Lei, *Water Purification Technology*, 2006, 25, 45-48.
- 33. H. Y. Fu, P. C. XU, T. Chai, P. F. Gao, G. H. Huang and F. W. Chen, *Chinese Journal of Environmental Engineering*, 2013, 7, 3256-3262.
- 34. W. Bae, D. Han, F. Cui and M. Kim, *KSCE Journal of Civil Engineering*, 2014, 18, 964-970.
- 35. C. S. Gee, M. T. Suidan and J. T. Pfeffer, *Journal of Environmental Engineering*, 1990, 116, 18-31.

# **Figure captions**

Figure 1. (a) Apparatus for measuring ANAMMOX performance (b) Stirred continuous-flow reactor

**Figure 2.** Digital images of the 4 types of immobilized granules (a) PVA (b) SA (c) PVA-SA (d) WPU

**Figure 3.** Chemical structure of WPU gel

**Figure 4.** Variation curves of the concentrations of  $NH_4^+$ -N,  $NO_2^-$ -N and  $NO_3^-$ -N with respect to time (a)  $NH_4^+$ -N (b)  $NO_2^-$ -N (c)  $NO_3^-$ -N

**Figure 5.** SEM images of WPU-immobilized granules (a) surface×1.0k; (b)

surface $\times$ 4.0k; (c) inside $\times$ 6.0k

**Figure 6.** Variation curves of the TN removal effects of the WPU-immobilized granules with respect to time

**Figure 7.** Biological nitrogen removal kinetics of immobilized granules (a) kinetic equation of  $NH_4^+$ -N (b) kinetic equation of NO<sub>2</sub>-N

**Figure 8.** Phylogenetic tree of the ANAMMOX bacteria



Figure 1. (a) Apparatus for measuring ANAMMOX performance (b) Stirred continuous-flow reactor 318x171mm (96 x 96 DPI)



Figure 2. Digital images of the 4 types of immobilized granules (a) PVA (b) SA (c) PVA-SA (d) WPU 965x233mm (96 x 96 DPI)



Figure 3. Chemical structure of WPU gel 167x115mm (300 x 300 DPI)



Figure 4. Variation curves of the concentrations of NH4+-N、NO2--N and NO3--N with respect to time (a) NH4+-N (b) NO2--N (c) NO3--N 463x266mm (96 x 96 DPI)

![](_page_33_Figure_2.jpeg)

Figure 4. Variation curves of the concentrations of NH4+-N、NO2--N and NO3--N with respect to time (a) NH4+-N (b) NO2--N (c) NO3--N 463x261mm (96 x 96 DPI)

![](_page_34_Figure_2.jpeg)

Figure 4. Variation curves of the concentrations of NH4+-N、NO2--N and NO3--N with respect to time (a) NH4+-N (b) NO2--N (c) NO3--N 463x271mm (96 x 96 DPI)

![](_page_35_Figure_2.jpeg)

Figure 5. SEM images of WPU-immobilized granules (a) surface×1.0k; (b) surface×4.0k; (c) inside×6.0k 1026x259mm (96 x 96 DPI)

![](_page_36_Figure_2.jpeg)

Figure 6. Variation curves of the TN removal effects of the WPU-immobilized granules with respect to time 297x209mm (150 x 150 DPI)

![](_page_37_Figure_2.jpeg)

Figure 7. Biological nitrogen removal kinetics of immobilized granules (a) kinetic equation of NH4+-N (b) kinetic equation of NO2--N 466x563mm (96 x 96 DPI)

![](_page_38_Figure_2.jpeg)

Figure 8. Phylogenetic tree of the ANAMMOX bacteria 254x106mm (96 x 96 DPI)

![](_page_39_Picture_156.jpeg)

Composition of the synthetic wastewater

Notes: The content of Trace element I and II are 1mL/L

![](_page_40_Picture_87.jpeg)

Stability of immobilized granules

Notes:1. (-) – unchanged; (+) – slightly softened and loosened; (++) – softened and fragile

2. Unit of tensile strength is  $kg/m<sup>2</sup>$ 

Study of imbedded immobilization by different materials

![](_page_41_Picture_75.jpeg)

Notes: CMC- carboxymethylcellulose; CTS-Chitosan

The 16S rDNA library of ANAMMOX bacteria and analysis of the phylogenesis of OTUs

![](_page_42_Picture_100.jpeg)