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1     **Physicochemical characteristics of attached biofilm on granular activated**  
2                     **carbon for thermophilic biohydrogen production**

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13

14    **Abstract**

15             In this study, thermophilic biohydrogen production by mixed culture obtained from a  
16     continuous acidogenic reactor treating palm oil mill effluent were improved by using granular  
17     activated carbon (GAC) as support material. Batch experiments were carried out at 60 °C by  
18     feeding the anaerobic sludge bacteria with a sucrose-containing synthetic medium at an initial  
19     pH of 5.5 under anoxic conditions. The physico-chemical characteristics of attached biofilm  
20     were evaluated after extraction of the extracellular polymeric substances (EPS) of the biofilm  
21     by formaldehyde- NaOH method. The main component of the biofilm was protein (60%);  
22     while carbohydrate content accounted for 40% of the EPS. Two major absorption bands at

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23 approximately  $3400\text{ cm}^{-1}$  and  $1650\text{ cm}^{-1}$ , characteristics of stretching vibrations of hydroxyl  
24 and amino groups, respectively, were identified in the FT-IR spectra, confirming the  
25 composition of the EPS. Observations using scanning electron microscopy (SEM) illustrated  
26 the attachment of rod-shaped bacterial cells on GAC at  $60^{\circ}\text{C}$ . The maximum hydrogen  
27 production rate of  $4.3\text{ mmol/L/h}$  and hydrogen yield of  $5.6\text{ mol H}_2/\text{mol sucrose}$  were obtained  
28 from this attached biofilm system. The major soluble metabolites of fermentation were  
29 acetic acid and butyric acid. The results showed that the granular activated carbon enhanced  
30 the biohydrogen production by stabilizing the pH and microbial metabolites and therefore  
31 could be used as support material for fermentative hydrogen production under thermophilic  
32 conditions in large scale.

33

34 **Keywords:** Anaerobic sludge, Attached biofilm, Biohydrogen production, Granular activated  
35 carbon, Thermophilic anaerobic cultivation

## 36 **Introduction**

37 Hydrogen is a clean renewable energy with a great potential to substitute the depletion of  
38 limited fossil fuels. There are various ways to produce hydrogen. For example, anaerobic or  
39 fermentative hydrogen production is one of the most promising ways to stabilise waste  
40 organics and convert biomass into hydrogen using specific anaerobic bacteria. This process  
41 can fulfil a sustainable development as it only requires little energy input, utilises waste  
42 organics as substrates, and produces hydrogen, which a noncarbonaceous energy carrier.

43 Various microorganisms have been exploited to produce hydrogen, either in suspended or  
44 attached growth systems. Much have been focused on suspended culture systems to examine  
45 the performance of anaerobic biohydrogen production using either single/pure strain and/or  
46 mixed culture <sup>1,2</sup>. However, in case of mixed culture, several drawbacks have been reported  
47 with respect to the low hydraulic retention time (HRT), such as the washout of the biomass  
48 and unstable hydrogen-producing bacterial populations <sup>3</sup>. Therefore, stability of the microbial  
49 population has become one of the most crucial elements in biohydrogen production.

50 Some alternative strategies have been proposed e.g., by introducing attached growth  
51 system via immobilisation technique in an anaerobic fluidised bed (AFBR) <sup>4</sup>, fixed bed <sup>5</sup>, or  
52 upflow anaerobic sludge blanket (UASB) reactor <sup>6</sup> in order to enhance biomass retention time.  
53 Many studies have demonstrated that the yield and productivity in thermophilic fermentative  
54 hydrogen production is higher than in mesophilic hydrogen fermentation <sup>7</sup>. However, lower  
55 cell density has been a drawback of fermentation at thermophilic temperatures <sup>8</sup>. Therefore,  
56 attached biofilm instead of suspended growth seems to be very practical in keeping the  
57 culture density and at the same time increasing hydrogen performance <sup>9</sup>.

58 Generally, immobilisation involves in the formation of a biofilm or granulation of cells  
59 during the fermentation process. This technology has relatively good potential due to the high  
60 cell density and adherence to the support material which avoids cell washout at a low HRT  
61 and, subsequently improves performance in two-phase separation <sup>9</sup>. There exists only a few  
62 reports on the biohydrogen production using attached growth systems under thermophilic  
63 conditions <sup>9-10</sup> as most investigators have focused on mesophilic conditions <sup>11-12</sup>. A long  
64 period of time is required for the granulation process, as reported by Yu and Mu <sup>13</sup>. Zhang et  
65 al. <sup>14</sup>, on the other hand, demonstrated rapid formation of biofilms under anaerobic conditions  
66 in a fluidised bed reactor (FBR).

67 Biofilm formation depends on several factors such as (i) physical surface and chemical  
68 composition of support carrier, (ii) surrounding environment such as nutrient availability, pH,  
69 and temperature, and (iii) composition of microbial consortia <sup>15</sup>. Biofilm-based systems have  
70 been extensively used as immobilised cell systems as they enhanced the reaction rates and  
71 population dynamics <sup>16</sup>. However, these studies have been limited to the development of  
72 mixed microflora biofilms, and the physical conditions have not been well-characterised <sup>17</sup>.  
73 Further studies on the rapid development of biofilm and the characterisation of such films can  
74 minimise the mass transfer resistance and stabilise the hydrogen-producing bacteria on  
75 biofilm for a good hydrogen performance.

76 In this study, thermophilic fermentative hydrogen production was carried out by  
77 immobilization of the anaerobic sludge obtained from palm oil mill treatment plant on  
78 granular activated carbon (GAC) in batch mode. GAC has been known as an inert,  
79 hydrophobic sorbent favourable for cell attachment. The effect of pH on biohydrogen

80 production by thermophilic biofilm was compared with suspended cells (without support  
81 carrier of GAC) using sucrose-containing medium at 60°C. The characteristics and chemical  
82 composition of the GAC-attached biofilm developed under optimum pH condition was also  
83 examined. Finally, the biohydrogen production with this thermophilic biofilm using complex  
84 substrate like palm oil mill effluent (wastewater) as carbon source was also examined. This  
85 knowledge is important for the assessment of reactor performance with real wastewater in  
86 future studies.

## 87 **1. Materials and methods**

### 88 *2.1 Mixed culture, carrier supports, and fermentation medium*

89 Mixed microflora used in this study was obtained from a recent study<sup>18</sup>. The anaerobic  
90 sludge underwent heat shock (heating at 80 °C for 60 min) and then acclimatized in a 500ml  
91 anaerobic reactor at 60°C with HRT of 48-12h by feeding palm oil mill effluent in a  
92 continuous mode<sup>18</sup>. The reactor effluent was collected (5L) and stored in laboratory cold  
93 room at 4°C for further use.

94 The carrier used to attach the hydrogen-producing bacteria was granular activated carbon  
95 (GAC) grade VISORB with a mesh size of 10 × 16 VS 45 (Carbochem Inc., USA). The GAC  
96 was sieved using a sieve shaker (Model EFL 2000/2, Endecotts, London) to obtain the  
97 required particle size of 2–3 mm.

98 The medium used for biohydrogen production contained 10 g/L sucrose as the sole  
99 carbon and energy source and supplements (in g/L unless indicated otherwise) as follows:  
100 NH<sub>4</sub>Cl, 1; NaCl, 2; MgCl<sub>2</sub>.6H<sub>2</sub>O, 0.5; CaCl<sub>2</sub>.2H<sub>2</sub>O, 0.05; K<sub>2</sub>HPO<sub>4</sub>.3H<sub>2</sub>O, 1.5; KH<sub>2</sub>PO<sub>4</sub>, 0.75;  
101 NaHCO<sub>3</sub>, 2.6; cysteine hydrochloride, 0.5; yeast extract, 2; resazurin, 0.5 mg; and trace

102 elements, 1 mL (R&M Chemical, UK) <sup>19</sup>.

### 103 *2.2 Batch fermentation*

104 In this study, the effect of initial pH on thermophilic fermentative hydrogen production  
105 by anaerobic sludge was studied and compared using cells in suspended form as well as  
106 attached on granular activated carbon (GAC). The initial pH of the medium was varied from  
107 5.0 to 7.5 with 0.5 increments and was adjusted using either 1 M HCl or 1 M NaOH. Batch  
108 cultivation was performed in a 100 ml serum bottle with working volume of 55 ml and 10%  
109 heat treated POME sludge as inoculum. Pre-grown, 24 hr old culture (5 ml) was added to 50  
110 ml of aforementioned medium and the serum bottles were purged with nitrogen gas to create  
111 anaerobic condition. The immobilization of POME sludge were carried out by adding GAC  
112 in a ratio of 1:1 of heat treated POME sludge volume (ml) to GAC weight (g) in the serum  
113 bottle while fermentation without GAC was conducted in parallel. The serum bottles were  
114 incubated in a shaking water bath (Model SW22, Julabo, Germany) at 150 rpm, 60°C for 24  
115 hours. The repeated batch cultivation process was continued for maximum five successive  
116 batches to compare the productivity and stability of hydrogen production at different initial  
117 pH. These experiments were done in triplicates.

118 The optimal initial pH obtained in repeated batch cultivation for biohydrogen production  
119 was used to conduct another batch study for profiling cumulative hydrogen productivity and  
120 to examine the biofilm formation. The experiment was conducted in a 50 ml serum bottle  
121 with a working volume of 25 ml medium. The bacteria grown and attached to the GAC  
122 during the final batch of cultivation (initial pH 5.5) was collected and added to 25 ml of the  
123 fresh medium adjusted to pH 5.5. The volume of medium (ml) to weight (g) of GAC ratio is

124 10:1. The serum bottles were purged with nitrogen gas to create an anaerobic condition and  
 125 the serum bottles were incubated in a shaking water bath (Model SW22, Julabo, Germany) at  
 126 150 rpm, 60 °C. Samples were analyzed at every 3 hours interval for 48 hours. The biogas  
 127 produced was sampled using disposable syringe for further analysis on the gas composition.  
 128 The supernatant were analysed for soluble volatile fatty acids, total carbohydrates and zeta  
 129 potential analysis. Biofilm characteristics were determined by compositional analysis of  
 130 extracellular polymer substances (EPS) by chemical extraction and Fourier-transform infrared  
 131 (FTIR) spectra. Scanning electron microscopy (SEM) was employed for visualization of  
 132 biofilm on GAC. The experiment was conducted in triplicates.

### 133 *2.3 Calculation of hydrogen production*

134 The cumulative hydrogen production in the batch experiment was determined according  
 135 to a modified Gompertz equation using Sigma Plot Software 10.0 (Systat Software Inc.,  
 136 USA). Theoretically, the modified Gompertz equation is as follows<sup>20</sup>:

$$137 \quad H_t = H_m \cdot \exp \left\{ -\exp \left[ \frac{R_m \cdot e}{H_m} (\lambda - t) + 1 \right] \right\} \quad (1)$$

138 where  $H_t$  is the cumulative hydrogen production (mL),  $H_m$  is the maximum hydrogen  
 139 production (mL),  $R_m$  is the maximum hydrogen production rate (mL.h<sup>-1</sup>),  $e$  is Euler's number,  
 140  $e = 2.73$ ,  $\lambda$  is the lag phase time (h), and  $t$  is the incubation time (h).

### 141 *2.4 Monitoring and analysis of hydrogen gas, VFAs and sugar*

142 Hydrogen gas production was calculated from headspace measurements of gas  
 143 composition and the total volume of biogas produced at each time interval was determined  
 144 using equation 2:

$$145 \quad V_{H,i} = V_{H,i-1} + C_{H,i} (V_{G,i} - V_{G,i-1}) + V_H (C_{H,i} - C_{H,i-1}) \quad (2)$$



146 where  $V_{H,i}$  and  $V_{H,i-1}$  are cumulative hydrogen gas volumes at the current (i) and previous  
147 (i-1) time intervals,  $V_{G,i}$  and  $V_{G,i-1}$  are the total biogas volumes in the current and previous  
148 time intervals,  $C_{H,i}$  and  $C_{H,i-1}$  are the fraction of hydrogen gas in the headspace of the bottle  
149 measured using gas chromatography in the current and previous intervals, and  $V_H$  the total  
150 volume of headspace in the bottle <sup>21</sup>. The volume of biogas produced was measured using a  
151 gas-tight syringe (0.5 mL injection volume) and a gas chromatograph (GC, model SRI 8600C,  
152 USA) equipped with the helium ionization detector (HID) and the thermal conductivity  
153 detector (TCD). Helium (MOX 99.99%) gas was utilized as carrier gas for the GC and run at  
154 the flow rate of 25 mL/min. The initial oven temperature was set at 40°C with the pressure of  
155 2.7 psi. The temperature at 40°C was maintained for 5 min, followed by a ramping of 30°C  
156 per minute until the temperature achieved 220°C before it was held for another 10 min.

157 The fermentation liquid was filtered through a 0.22  $\mu$ m syringe filter before analysis.  
158 Soluble volatile fatty acids (VFAs) were analysed by HPLC analysis using an Agilent 1200  
159 HPLC system (California, USA) with a REZEX ROA column (Phenomenex, USA) equipped  
160 with a refractive index detector (RID). The flow rate of 5 mM  $H_2SO_4$  as the mobile phase was  
161 fixed at a constant 0.6 mL/min with isocratic elution at room temperature. The HPLC sample  
162 injection volume was 20  $\mu$ L, and the standard curves were generated using different  
163 concentrations of mixed organic acids. Sucrose was analysed using phenol-sulphuric acid  
164 method.

### 165 *2.5 Biofilm extraction, characterization and imaging*

166 The biofilm attached to the carrier (GAC) was mashed with a pestle to detach the  
167 extracellular polymeric substance (EPS), placed in a centrifuge tube, and stored at -20 °C for

168 further analysis. The attached EPS was extracted according to the formaldehyde-NaOH  
169 method as described by Liu and Fang<sup>22</sup>. Prior to extraction, EPS suspension was prepared by  
170 adding 1.5 g (dry weight) of suspended biofilm in a 10 mL of ultrapure water (Sartorius,  
171 Malaysia). Extraction was performed by treating the sample with 0.06 mL of 37% (w/w)  
172 formaldehyde at 4 °C for 1 h. Following the exposure to formaldehyde, the suspension was  
173 then treated with 5 mL of 1 M NaOH at 4 °C for about 3 h. Finally, the mixtures were  
174 centrifuged at 10,000 g for 15 min to obtain the soluble EPS.

175 Once the EPS was extracted, the soluble EPS was evaluated for the carbohydrates  
176 content using the phenol-sulphuric acid method. Proteins were determined based on the  
177 Bradford method using bovine serum albumin (BSA) as the standard. The total of  
178 carbohydrates and proteins were based on the quantity of volatile suspended solids (VSS) of  
179 the biofilm and measured according to the American Public Health Association (APHA)  
180 standard method.

181 The chemical composition of extracted EPS were also verified using FTIR spectrometry  
182 (Model Nicolet 6700, Thermo Scientific, USA) via the attenuated total reflectance (ATR)  
183 method. The FTIR spectra were acquired in the 4000–400 cm<sup>-1</sup> region with a resolution of 2  
184 cm<sup>-1</sup> using transmission mode.

185 The zeta potential was measured using Zetasizer Nano Particle Analyser (Nano ZS, Model  
186 ZEN 3600, Malvern Instrument Ltd., UK) and analysed using Dispersion Technology  
187 Software (DTS) version 5.02. The supernatant used for zeta potential analysis was maintained  
188 at about 0.02% on a wet-weight basis with ultrapure water (Sartorius Malaysia Sdn. Bhd.,  
189 Malaysia).

190 Biofilm developed on GAC was visualised using field emission scanning electron  
191 microscopy (FESEM) (Model Supra 55VP, Carl Zeiss AG, Germany). The GAC-immobilised  
192 cells were fixed with 2% (w/w) glutaraldehyde and left overnight at 4 °C. The fixed samples  
193 were washed with 0.1 M phosphate buffer solution for three times and left for 10 min each.  
194 Dehydration was done by successive passages through 30, 50, 70, 80, 90, and 100% (w/w)  
195 alcohol. The dehydrated particles were then transferred to a Critical Point Dryer (Model Leica  
196 EM CPD 300, Leica Microsystems, Germany) for 1 h and 30 min. The dried samples were  
197 sputter-coated with platinum and finally analysed using FESEM.

#### 198 *2.6 Validation of GAC-attached thermophilic biofilm using POME as substrate*

199 In order to validate the efficiency of GAC-attached thermophilic biofilm to produce hydrogen  
200 from palm oil mill effluent (POME), batch experiments were carried out in 50 ml serum bottles  
201 with 25 mL of working volume. For the purpose, synthetic medium as mentioned in Section 2.1  
202 was amended by adding raw POME (10% v/v) and replacing 10g/L sucrose with glucose and  
203 xylose(3:1) mimicking sugars available in raw POME. Hydrogen production performance was  
204 also compared using synthetic medium without addition of POME but with mixed sugars  
205 (Glucose and xylose) in the same ratio. Medium pH was adjusted to 5.5 and was purged with  
206 nitrogen gas to create an anaerobic condition. Serum bottles were incubated in a shaking water  
207 bath (Model SW22, Julabo, Germany) at 150 rpm, 60°C. Analysis of biogas, soluble volatile  
208 fatty acids, and total carbohydrates were carried out as described in section 2.4.

### 209 **3. Results and discussion**

#### 210 *3.1 Batch cultivation*

211 The effect of the initial pH of the mixed culture from heat treated POME sludge with  
212 GAC and without GAC (control) was studied in the pH range of 5.0 – 7.5. The effect of

213 initial pH in hydrogen fermentation is a crucial aspect because inappropriate pH could inhibit  
214 hydrogenase activity<sup>23</sup>. The results shown in Fig. 1 are the steady-state data of hydrogen  
215 performance of the last batch at different initial pH.

216 In suspended cells, the biohydrogen production rate gradually increases from 2.82 to  
217 4.28mmol/H<sub>2</sub>/L with the initial pH increasing from 5.0 to 6.5. However, further increment of  
218 initial pH to 7.0 reduced the biohydrogen production rate two-folds compared to that in pH  
219 5.0. The maximum H<sub>2</sub> content of 47.8% was obtained at pH 5.5. Generally, pH values  
220 between 5.0 and 6.5 are reported to be favorable for better microbial activity and hence better  
221 hydrogen production<sup>24</sup>.

222 In GAC-attached biofilm, both H<sub>2</sub> production rate and H<sub>2</sub> content increased with  
223 increasing initial pH from 5.0 to 5.5. With a further pH increase, the performance declined.  
224 Highest H<sub>2</sub> production rate and H<sub>2</sub> content of 5.32 mmol H<sub>2</sub>/L/h and 51.2%, respectively,  
225 were obtained at initial pH 5.5 with GAC. In this study, biohydrogen production rate could be  
226 enhanced up to 51% by using GAC as support material at initial pH of 5.5. The optimal  
227 cultivation pH of 5.5 for GAC is in agreement with reports from other researchers on  
228 thermophilic hydrogen fermentation at 60°C<sup>8</sup>.

229 Furthermore, the maximum hydrogen production rate at initial pH 7.0 and 7.5 for control  
230 experiment shows the lowest value among other initial pH with approximately 1.0 mmol  
231 H<sub>2</sub>/L/h. This observation is similar to Chong et al (2013), who also reported that the growth  
232 of bacteria as well as hydrogen production is restricted at pH above 7.0 in suspended culture  
233<sup>25</sup>. However, it is interesting to note that, higher value for hydrogen production was recorded  
234 at initial pH 7.0 (4.3 mmol H<sub>2</sub>/L/H) in presence of GAC in this study than those of initial pH

235 5.0 – 6.5 (in the range of 2.82 – 4.28 mmol H<sub>2</sub>/L/h) in control experiment (Fig. 1). This is due  
236 to the mechanical stability of the biofilms formed on the GAC that have great binding  
237 capacity for organic matter, hence provides an environment that is rich in nutrients promoting  
238 microbial adhesion <sup>26</sup>.

239 Therefore the experiment in presence of GAC even at a non-favorable pH condition exhibited  
240 better hydrogen production than the suspended cell in control experiments in all different  
241 initial pH studied. The immobilization system enabled cells to withstand considerable shear  
242 force and stay active towards a stressful environment <sup>26</sup>.

243 From the microbiological point of view, the suitable pH at which hydrogen production  
244 would be optimum is subjected to the type of hydrogen producing bacteria (HPB) in the  
245 inoculum. In our recent study <sup>18</sup> focusing on the community analysis of the anaerobic sludge  
246 from palm oil mill treatment plant, it was shown that the reactor effluent harbours species of  
247 *Bacillus* as dominant culture. *Bacillus smithii* CMB-B1 and *Bacillus coagulans* M36 species  
248 have been identified to be responsible for the production of hydrogen.

249 The same sludge was employed in this present study. Generally, basilli thermophilic  
250 (thermophilic bacilli) is aerobic or facultative anaerobic bacteria that live in the temperature  
251 range of 45-70 °C. *Bacillus* species have already been reported as producing hydrogen other  
252 than *Enterobacter* and *Clostridium* species <sup>27</sup>.

253 The temperature range and optimum pH for growth and hydrogen production of *Bacillus*  
254 *smithii* has been reported to be 25°C to 60°C and pH 5.7 by Nakamura et al. (1988) <sup>28</sup>. The  
255 potential of *Bacillus smithii* as hydrogen-producing bacteria was also studied by Grady et al.  
256 (1998) who used this species to convert waste biomass into hydrogen <sup>29</sup>. *Bacillus coagulans*

257 species has also been identified as a hydrogen producer in either mesophilic and thermophilic  
258 conditions<sup>30</sup>.

### 259 3.2 Kinetic analysis of batch fermentation using GAC-immobilized cells

260 Fig. 2 shows the kinetic analysis of biohydrogen and soluble metabolite production at  
261 initial pH 5.5 using GAC-attached POME sludge. The comparison between the experimental  
262 data and the predicted modified Gompertz model of the cumulative hydrogen production of  
263 GAC-attached biofilm is shown in Fig 2(a). The kinetic parameters obtained were based on  
264 the cumulative hydrogen production data fitted by the modified Gompertz model with  $H_m =$   
265  $58.6 \text{ ml}$ ,  $R_m = 2.8 \text{ ml.h}^{-1}$ , and  $\lambda = 0.003 \text{ h}$ . From the results, the hydrogen production rate  
266 (HPR) obtained was  $4.3 \text{ mmol/L/h}$ . An experimental  $\text{H}_2$  yield of  $5.6 \text{ mol H}_2/\text{mol sucrose}$ ,  
267 which equals  $2.8 \text{ mol H}_2/\text{mol hexose}$  was obtained by using acclimatized GAC-attached  
268 biofilm as inocula.

269 The concentration of soluble metabolite products (SMP) (as shown in Fig. 2b) increased with  
270 an increase in the fermentation time, which was reflected in more hydrogen production, as  
271 shown in Fig. 2(a). Primary SMP was acetic acid (HAc), comprising 40–55% of the SMP  
272 followed by butyric acid (HBu) accounted for approximately 30–40% of the fermentation  
273 liquor. In contrast, the production of ethanol (EtOH), which is considered an unfavourable  
274 metabolite for hydrogen production, was less than 20% of SMP. The amount of acetic acid  
275 produced at 48 h was  $24.2 \text{ mM}$ , followed by  $22.5 \text{ mM}$  butyric acid and  $11 \text{ mM}$  ethanol. The  
276 prevalence of acetic acid and butyric acid in the culture supernatant suggested that the  
277 acidogenesis pathway was favoured to produce higher biohydrogen<sup>31</sup>. Study by Kotay and  
278 Das (2007) using *Bacillus coagulans* IIT-BT SI isolated from sewage sludge also produced

279 acetic acid, butyric acid and ethanol as the primary metabolite during fermentative  
280 biohydrogen production<sup>30</sup>.

281 In this study, a higher yield of H<sub>2</sub> is obtained by using GAC-attached POME sludge in  
282 batch cultivation process under thermophilic condition, compared to Wu et al<sup>31</sup> using  
283 acclimatized, attached biofilm at mesophilic condition. The H<sub>2</sub> evolution was rapid with no  
284 lag time and consisted of 43.8 % hydrogen content of total biogas produced with almost  
285 threefold the working volume of fermentation. No methane was detected throughout the  
286 experiment.

287 Table 1 summarizes similar studies on biohydrogen production from mixed culture  
288 attached on activated carbon using sucrose as sole carbon source. Even though the carrier  
289 material and carbon source used are similar for all these studies, the current study seems to be  
290 unique in developing bioprocess for hydrogen production by acidogenic, thermophilic  
291 biofilm on GAC. It also revealed that employing repeated batch system, hydrogen yield even  
292 in smaller scale (serum bottle) can be improved up to 3-4 folds compared to Wu et al<sup>31</sup>.  
293 Further improvements on the fermentation performance can be achieved by using bioreactor  
294 systems where operational parameters such as pH and temperature can be controlled  
295 throughout the experiment.

296 The total volatile fatty acids (TVFAs) concentrations achieved at initial pH 5.5 after 48 h  
297 fermentation in presence of GAC were 46.7 mmol TVFAs (referring to the total amount of  
298 acetic and butyric acid) and hence contributed to higher hydrogen performance.  
299 Immobilization of hydrogen-producing bacteria on GAC gave protection from exposure to  
300 metabolite accumulation, low acid condition, and low substrate concentration during

301 fermentation.

### 302 *3.3 Biofilm characterization and imaging*

303 Biofilms are complex assemblages of microorganisms that are embedded in a matrix of  
304 extracellular polymeric substances (EPS). During repeated batch cultivation, mixed microbial  
305 cells grew in association with activated carbon surfaces resulted in the formation of biofilm.  
306 The characterization of biofilm was carried out by extracting EPS using chemical extraction  
307 followed by chemical content, zeta potential measurements and observation under FESEM.

308

#### 309 *3.3.1 Extraction and chemical composition of Extracellular polymeric substances (EPS)*

310 EPS is a component of aggregation of hydrogen-producing bacteria (HPB) accumulating  
311 in the biofilm on GAC. EPS produced at the solid surface of the GAC promote microbial  
312 adhesion by altering the physicochemical characteristics of the colonized surface. They create  
313 scaffolds with suitable physical characteristics and interconnected GAC pore structure that  
314 promote cell attachment<sup>26</sup>.

315 In this study, extraction was carried out using the formaldehyde-NaOH method as it is  
316 the most effective extraction method by Liu and Fang<sup>22</sup>. Composition of total carbohydrate  
317 (TC) and protein in extracellular polymeric substances (EPS) of GAC immobilised-cell  
318 biofilm at different interval time of fermentation are shown in Table 2. As can be seen in  
319 Table 2, the carbohydrate and protein levels in the extracted EPS constituents increased with  
320 an increase of the fermentation time, which may have been the result of enhanced adhesion  
321 between bacteria and GAC during biofilm formation. At the beginning of the fermentation  
322 (3-hour), both carbohydrate and protein contents were low. However, after 48 h of  
323 fermentation, carbohydrate and protein comprised of 9.2 mg/g VSS (40%) and 13.2 mg/g



324 VSS (60%) of the extracted EPS, respectively. The ratio of protein to carbohydrate (P/C) in  
325 the EPS for all contact times varied between 1.4 and 2.6. Similar results for  
326 protein/polysaccharide ratios of biofilm EPS between 1.8 and 5.4 was reported by Ras et al.  
327 <sup>34</sup>.

328 Moreover, there seemed a direct relationship between increment of total EPS  
329 concentration (given in Table 2) with total soluble metabolites produced (as shown in Figure  
330 2b) during hydrogen production. In this study, the total EPS concentration increased from  
331 13.6 mg/g VSS after 3 h fermentation to 22.4 mg/g VSS after 48 h fermentation which can be  
332 correlated with the increment of total SMP (SMP = TVFA + EtOH) from 9.1 mM to 57.7 mM  
333 during 48 h of fermentation. EPS secreted by mixed microflora can be subdivided into bound  
334 EPS (eg: attached organic materials) and soluble EPS or sometimes referred as soluble  
335 microbial products (SMP) <sup>35</sup>. It is well established that the production of this SMP reflects the  
336 hydrogenase metabolic pathway and hence, the performance of hydrogen <sup>23</sup>. Thus, EPS  
337 concentration plays an important role in establishing the structural and functional integrity of  
338 microbial biofilms and exhibit a direct relationship with hydrogen production <sup>36</sup>. However,  
339 the types of SMPs and the degree of microbial adhesion on the immobilized carrier will vary  
340 depending on the microbial species and operational conditions such as temperature and pH.

### 341 3.3.2 FTIR spectra

342 Fig. 3 shows the FTIR spectra of EPS formed after 48 h of fermentation by the  
343 GAC-attached biofilm. As depicted in Fig. 3, three major absorption bands were found in the  
344 sample. The broadest absorption band at approximately  $3400\text{ cm}^{-1}$  revealed the presence of –  
345 OH groups while absorption bands at approximately  $1650\text{ cm}^{-1}$  indicated the presence of

346 COOH groups. The band near  $700\text{ cm}^{-1}$  reflected the presence of unsaturated bonds in the  
347 samples. These functional groups confirmed that the samples were exopolysaccharide.  
348 Similar results have also been reported by Iyer et al (2005) and Kumar et al (2011)<sup>37-38</sup>.

### 349 3.3.3 Zeta potential

350 Zeta potential is an important index to characterize the surface charge of sludge, which  
351 would tend to decrease gradually with the reduction of negative charge of sludge surfaces<sup>39</sup>.

352 Fig. 4 shows changes in the zeta potential and pH of the culture supernatant during  
353 biohydrogen production under thermophillic condition. Before immobilization and  
354 cultivation, the zeta potential of raw sludge from POME at pH 5.5 was  $-2.04\text{ mV}$ .

355 After cultivation, the zeta potential or surface charge of the supernatant reduced to a negative  
356 value with increasing fermentation time. Just after 3 h of fermentation, the zeta potential was  
357 slightly reduced from  $-2.04\text{ mV}$  to  $-2.58\text{ mV}$ , and finally to  $-9.17\text{ mV}$  after 48 h, at the end of  
358 batch fermentation. Culture pH was shown to have similar trend as zeta potential and the pH  
359 value decreased from an initial pH of 5.5 to 4.74.

360 In contrast, Lin et al.<sup>40</sup>, reported that the zeta potential of the pure culture increased  
361 with decreasing culture pH. Similar phenomena was observed by Zhang et al.<sup>41</sup> during  
362 biofilm formation under acid incubation. In their study, an increase in the zeta potential of  
363 GAC from  $2.74\text{ mV}$  to  $76.41\text{ mV}$  accompanied with reduction in pH from 5.5 to 2.0. However,  
364 it is worthy to note that, in the present study, pH fluctuation was not abrupt as other studies  
365 and hence both pH and zeta potential showed similar trend. In addition, the increase in EPS  
366 content with increasing fermentation time (as shown in Table 2) suggested that the culture  
367 zeta potential could be influenced by adsorbing EPS that carried more negative charges<sup>14</sup>.

368 Recently, Su et al<sup>39</sup> have demonstrated that the decrease of zeta potential may speed up  
369 the granulation progress as a necessary condition. Bacterial adhesion is determined by an  
370 interplay between hydrophobic and electrostatic interactions. When bacteria approach the  
371 surface of the support material, they experience an electrostatic repulsion since both the  
372 bacteria and the GAC particle surface are negatively charged<sup>14</sup>. However, as the zeta  
373 potential decreased, microbial aggregation tends to strengthen because the low zeta potential  
374 decreased the repulsive electrostatic interactions. According to Gottenbos et al.<sup>42</sup>, the  
375 positively charged surface adversely affects biofilm formation. Thus, a negative surface  
376 charge as in cultivated sludge POME is an advantage for biofilm formation.

#### 377 *3.3.4 Microbial observation*

378 The GAC-attached biofilm at the end of fermentation (48 h) was selected for field  
379 emission scanning electron microscopy (FESEM) analysis. The surface morphology (crevices  
380 and micropores) of GAC before attachment are shown in Fig. 5(a) and 5(b) under different  
381 magnifications. The porous structure and irregular surface of GAC provided the space for cell  
382 growth. Fig. 5(c) and 5(d) illustrates the biofilm on the GAC carrier. As can be seen in Fig  
383 5(c), the rod-shaped bacterial cells with a length of 1.0–5.0  $\mu\text{m}$  and a width of 0.5–0.6  $\mu\text{m}$   
384 were dominated on the biofilm. A closer examination (10,000 $\times$  magnification) in Fig 5(d)  
385 revealed the attachment of individual cells onto the surfaces as well as within the cavities of  
386 GAC. These SEM images indicated that a stable and successful immobilization was achieved  
387 at thermophilic conditions. The surface porosity of GAC eased the bacteria to attach to the  
388 surface with the aid of the conditioning layer and the EPS formation at the substratum. EPS at

389 the substratum minimised the mass transfer resistance and stabilised the hydrogen-producing  
390 bacteria on biofilm for a good hydrogen performance.

### 391 *3.4 Validation of GAC-attached thermophilic biofilm using POME as substrate*

392 Sucrose (10g/L) has been used throughout the study to establish whether the GAC-  
393 immobilized system developed herein is successful for producing biohydrogen at  
394 thermophilic conditions or not. We have also examined biohydrogen production with this  
395 thermophilic biofilm using diluted palm oil mill effluent (POME) as carbon source. The  
396 POME used in this study had total suspended solids about  $47.9 \pm 0.4$  g/L. Generally, attached  
397 growth system is not suitable for substrates with high solid content. Therefore, we attempted  
398 to use diluted POME to study the performance of GAC-attached thermophilic biofilm on  
399 biohydrogen production. Fig. 6 depicts the experimental data and predicted profile by  
400 modified Gompertz model of cumulative hydrogen production from POME with GAC  
401 attached cells at 60°C. The maximum hydrogen produced in 25 mL medium is  $H_m$  of 40.3 mL  
402 or equivalent to  $R_m$  of  $4.8 \text{ mL}\cdot\text{h}^{-1}$ . This is slightly lower than control ( $H_m$  of 44.3 mL and  $R_m$   
403 of  $5.7 \text{ mL}\cdot\text{h}^{-1}$ ) which consisted of 100 % synthetic medium and glucose and xylose mimicking  
404 sugar composition in raw POME.

405 In this study, hydrogen yield in presence of POME was  $1.75 \text{ mol H}_2/\text{mol}$  sugar consumed,  
406 whereas the maximum hydrogen production rate was  $4.1 \text{ mmol H}_2/\text{L}\cdot\text{h}$ . The soluble  
407 metabolite products (SMP) produced was 33.3 mM, with acetate and butyrate comprised of  
408 99 % of the total SMP and 1% ethanol. The dominance of acetate and butyrate formation  
409 indicated that the pathway of hydrogen is the acidogenic pathway. Lower hydrogen yield was  
410 obtained when using POME ( $1.75 \text{ mol H}_2/\text{mol}$  sugar) compared to sucrose ( $2.8 \text{ mol H}_2/\text{mol}$

411 hexose) (Section 3.2), suggesting that the GAC immobilized cells need some adaptation time  
412 to a new combination of carbon source which consisted of hexose (glucose) and pentose  
413 (xylose)<sup>43</sup>. Nevertheless, the hydrogen production rate of POME (4.1 mmol/L/h) was almost  
414 similar when using sucrose (Section 3.2) with 4.3 mmol/L/h probably because the culture  
415 used in this study has already been acclimatized in POME. Moreover, as fermentation was  
416 performed under optimum pH and temperature, the enriched community remained same over  
417 the cultivation period with consistent fermentation performance.

#### 418 **4. Conclusions**

419 The biohydrogen production performance of suspended sludge and immobilized cell  
420 systems with GAC were investigated in this study. The optimal initial pH for thermophilic  
421 biohydrogen production was found to be 5.5. Adding GAC resulted in a prompt microbial  
422 colonization and biofilm development with a H<sub>2</sub> yield of 5.6 mol H<sub>2</sub>/mol sucrose and a  
423 hydrogen production rate of 4.3 mmol/L/h. This may be due to the protective effect of GAC  
424 carrier of the attached growth biofilm against the acidic environment compared to the  
425 suspended culture. Further improvements on the thermophilic biohydrogen production by  
426 GAC-immobilized cells are currently being attempted in fluidised bed anaerobic bioreactor to  
427 examine the microbial adhesion and tolerance at low hydraulic retention time (HRT).

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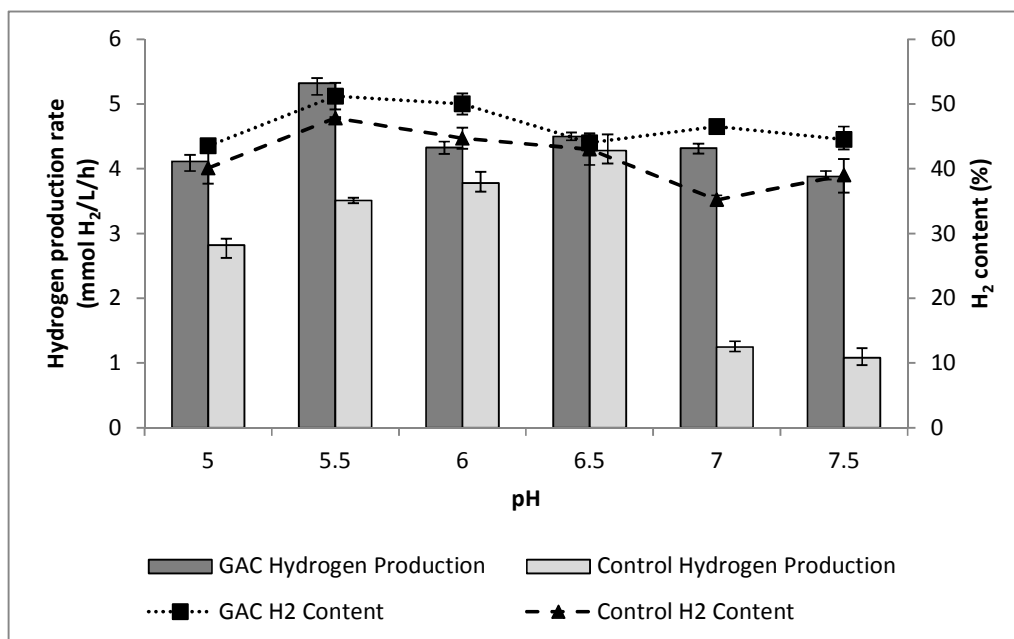
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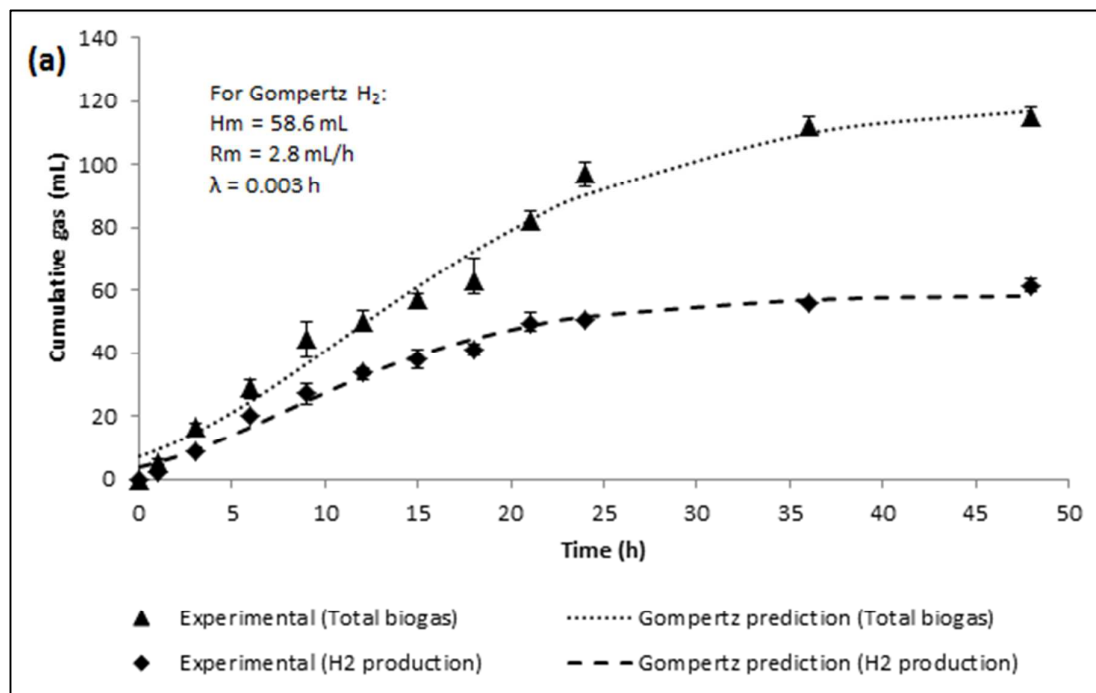


495 Fig 1 Hydrogen production rate and hydrogen content of final batch cultivation with GAC and without  
 496 GAC (control) at various initial pH

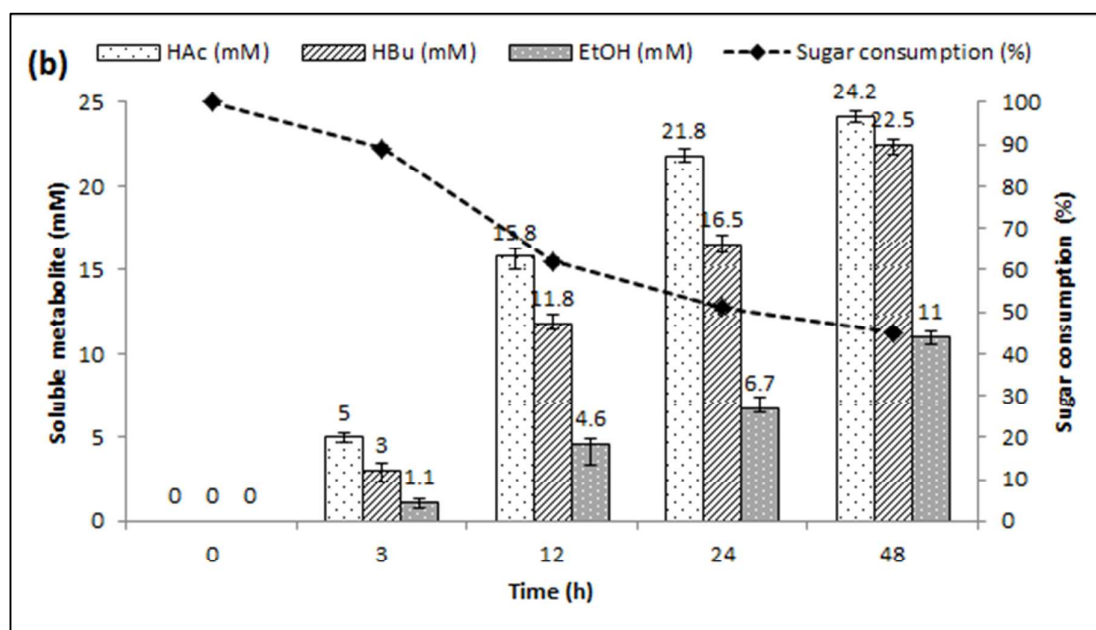
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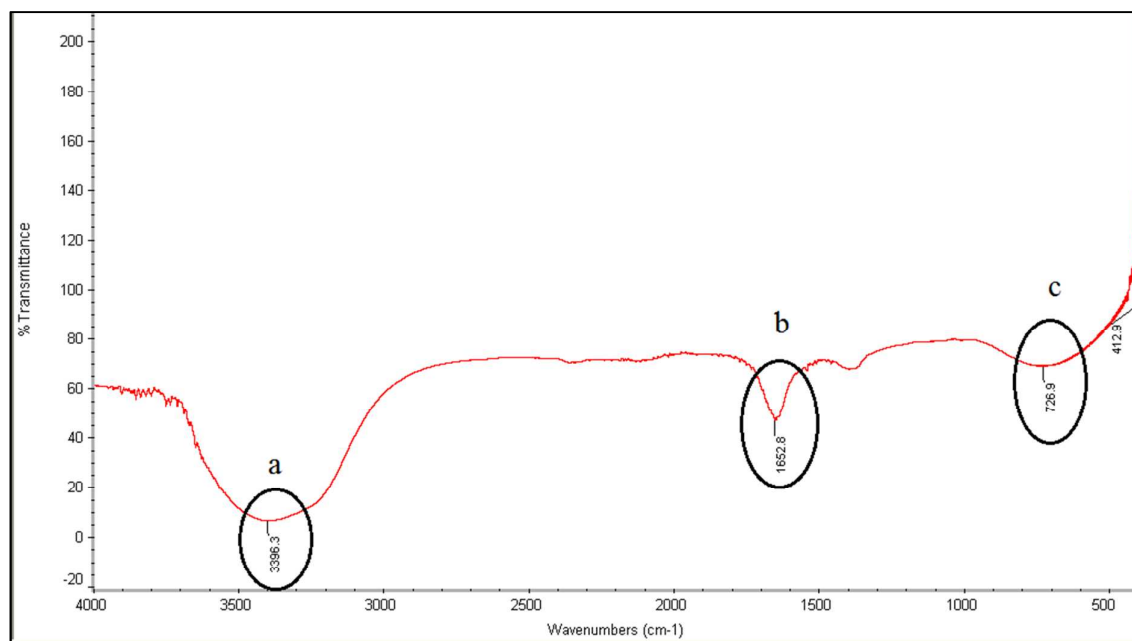


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502 Fig. 2 –Batch kinetics of hydrogen production from sucrose with GAC attached biofilm at 60°C at initial  
503 pH of 5.5 (a) Gompertz curve fitting graph of cumulative gas production and (b) Composition of soluble  
504 metabolites (Acetic acid-HAc, Butyric acid-HBu and Ethanol-EtOH) and sugar consumption  
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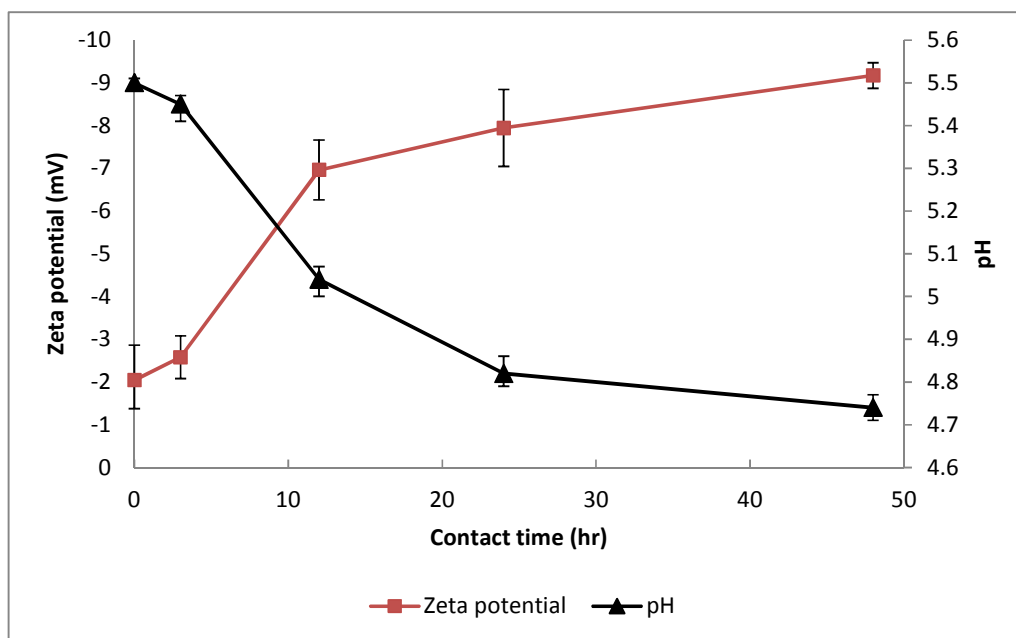


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507 Fig. 3 FTIR spectra of GAC attached biofilm at 48 h interval time of fermentation with the stretching  
508 vibrations of (a) combination of hydroxyl and amino groups, (b) C(N)=O and C–N+N–H groups in protein,  
509 and (c) unsaturated bonds.

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514 Fig. 4 Zeta potential and pH of supernatant at different interval time during biohydrogen production.

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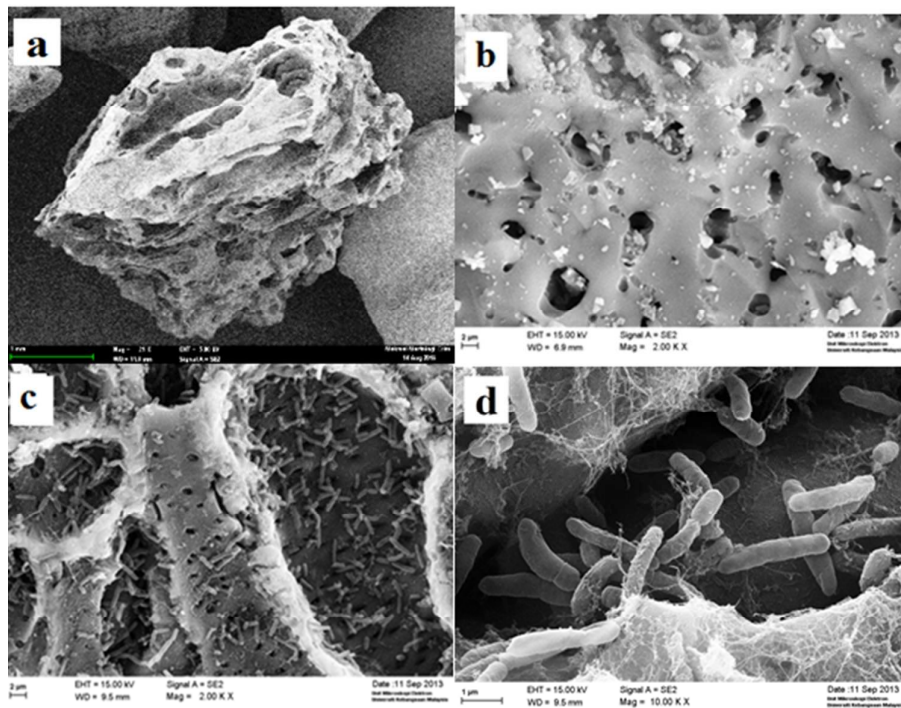
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523 Fig. 5 SEM images of (a) single GAC (magnification 21 $\times$ ) particle and (b) porous surface structure of  
524 GAC (magnification 2,000 $\times$ ) before attachment, (c) bacterial colonization onto GAC after 48 h of  
525 fermentation (magnification 2,000 $\times$ ), and (d) close-up view of rod-shaped bacteria (magnification 10,000 $\times$ )  
526 found as predominant species during biohydrogen production under thermophilic conditions (60  $^{\circ}$ C)

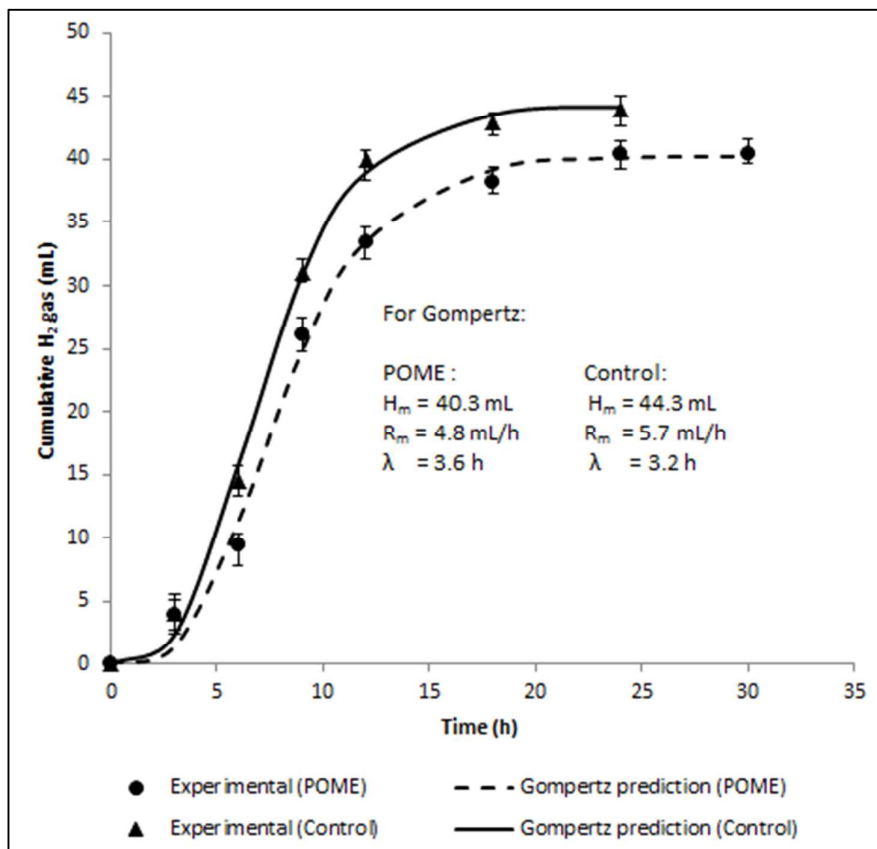
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533 Fig. 6 - Batch kinetics of hydrogen production from palm oil mill effluent (POME) with GAC attached

534 biofilm at 60°C

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543 **Table 1** Maximal hydrogen yield by mixed culture grown on activated carbon using sucrose as sole carbon  
 544 source under different operational conditions

Microorganisms	Substrate	Immobilization carrier	Mode/ Process	Initial pH	Optimal Operation Temperature (°C)	Max. H <sub>2</sub> Yield (mol H <sub>2</sub> / mol hexose)	Reference
Sewage sludge/ Mixed culture	Sucrose	Activated carbon	Continuous/ Fixed-bed	6.7	35	0.59	<sup>5</sup>
Sewage sludge/ Mixed culture	Sucrose	Activated carbon	Continuous/ Packed-bed	6.7	35	1.45	<sup>32</sup>
Sewage sludge/ Mixed culture	Sucrose	Activated carbon	Continuous/ CIGSB	6.7	35	1.5	<sup>33</sup>
Sewage sludge/ Mixed culture	Sucrose	Activated carbon	Continuous/ CSABR	6.6 ± 0.2	40	1.93	<sup>7</sup>
Sewage sludge/ Mixed culture	Sucrose	Activated carbon	Batch/ Serum vial	6.7	40	0.87	<sup>31</sup>
Sewage sludge/ Mixed culture	Sucrose	Activated carbon	Continuous/ FBR	6.4	40	1.88	<sup>11</sup>
POME sludge/ Mixed culture	Sucrose	Granular activated carbon	Batch/ Serum bottle	5.5	60	2.80	This study

CIGSB - carrier-induced granular sludge bed; CSABR - continuously stirred anaerobic bioreactor; FBR – Fluidized-bed reactor

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565 **Table 2** Composition of total carbohydrate and protein in EPS at different interval time of fermentation

Time (h)	Total EPS composition (mg/g VSS)		Ratio
	Carbohydrate	Protein	
3	3.8 ± 0.4	9.8 ± 0.8	2.6
12	7.3 ± 0.3	10.6 ± 0.9	1.5
24	7.7 ± 0.2	12.3 ± 0.5	1.6
48	9.2 ± 0.3	13.2 ± 0.4	1.4

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