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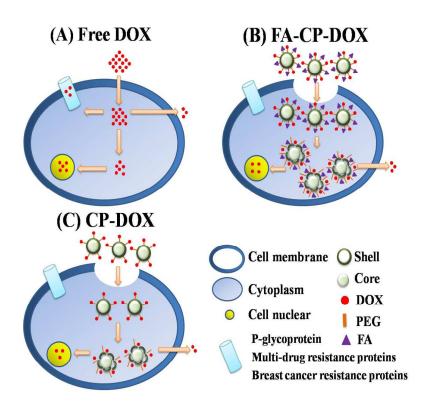
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# **Graphic abstract**



1	Folate-Mediated and Doxorubicin-conjugated
2	Poly(E-Caprolactone)-g-Chondroitin Sulfate Copolymers for Enhanced
3	Intracellular Drug Delivery
4	
5	
6	Yu-Sheng Liu <sup>1</sup> , Hsuan-Ying Chen <sup>1</sup> , Jay-An Yeh <sup>1</sup> , and Li-Fang Wang <sup>1,2,3*</sup>
7	<sup>1</sup> Department of Medicinal and Applied Chemistry, Kaohsiung Medical University,
8	Kaohsiung 807, Taiwan
9	<sup>2</sup> Department of Biotechnology, Kaohsiung Medical University, Kaohsiung 807, Taiwan
10	<sup>3</sup> Institute of Medical Science and Technology, National Sun Yat-Sen University,
11	Kaohsiung 804, Taiwan
12	
13	
14	
15	
16	
17	
18	
19	Correspondence to: Li-Fang Wang, Professor of Medicinal & Applied Chemistry
20	Kaohsiung Medical University
21	School of Life Science
22	100, Shih-Chuan 1 <sup>st</sup> Rd, Kaohsiung City 807, Taiwan
23	<b>Tel:</b> 011-886-7-3121101-2217
24	Fax: 011-886-7-3125339
25	E-mail: <u>lfwang@kmu.edu.tw</u>
26	
27	
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30	

# 1 Abstract

- 2 The aim of this study was to conjugate an anticancer drug, doxorubicin (DOX) and a folate targeting
- 3 moiety, folic acid (FA), to self-assembled polycaprolactone (PCL)-graft-chondroitin sulfate (CS)
- 4 copolymers for enhanced chemotherapy. The PCL-graft-CS copolymer was abbreviated as CP. DOX
- 5 was conjugated to CP using a bifunctional polyethylene glycol as a spacer (CP-DOX). FA was
- 6 conjugated to the CP-DOX to yield FA-CP-DOX that could enhance the cellular uptake in folate
- 7 receptor (FR)-overexpressing cancer cells. The CP-DOX and FA-CP-DOX copolymers were
- 8 confirmed using <sup>1</sup>H-nuclear magnetic resonance (<sup>1</sup>H-NMR) and Fourier transform infrared (FTIR)
- 9 spectrophotometers. CP-DOX was spherical and FA-CP-DOX was worm-like. The copolymers
- 10 without DOX were non-cytotoxic against U87 cells. The IC<sub>50</sub> value (an inhibitory concentration of
- 11 50% cell growth) of FA-CP-DOX was comparable to that of free DOX but much lower than that of
- 12 CP-DOX against U87 cells following 24, 48 and 72 h of post incubation. Because of recognition to
- 13 the FR, the magnificently cellular uptake of FA-CP-DOX into U87 cells was observed using flow
- 14 cytometry and confocal laser scanning microscopy.
- 15

Keywords: chondroitin sulfate; poly(ε-caprolactone); doxorubicin; folic acid; amphiphilic copolymers;
 micelles.

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# 1 Introduction

2	The most considerable challenges facing effective cancer therapy are the solubility and systemic
3	toxicity of anticancer agents, the lack of tumor localization and an even distribution throughout the
4	whole body including tumor tissues. <sup>1, 2</sup> In addition, anticancer drugs' short half-lives in blood
5	circulation and their undesirable pharmacokinetic behavior are among other drawbacks which are
6	present in the way of cancer chemotherapy. <sup>3</sup> Conjugation of low molecular weight (MW) drugs to
7	macromolecular carriers is therefore considered a promising approach for improving the efficacy of
8	cytotoxic drugs on tumor cells along with fewer side effects on normal tissues. <sup>3-5</sup> A good example of
9	a rational drug-conjugate design is the history of CRLX101, <sup>6</sup> which was developed by covalently
10	conjugating camptothecin (CPT) to a linear, cyclodextrin-polyethylene glycol (CD-PEG) copolymer
11	that self-assembles into nanoparticles.
12	The macromolecular conjugated prodrugs have several advantages over their low MW
13	precursors. <sup>7</sup> The main advantages are: (1) an increase in the water solubility of low soluble or
14	insoluble anticancer drugs and enhanced biodistribution and therapeutic efficacy; (2) an
15	accumulation of anticancer drugs in tumor tissues through the enhanced permeation and retention
16	(EPR) effect and a reduction in systemic side effects; (3) an improvement in drug pharmacokinetics
17	and a more prolonged plasma half-life; (4) the potential of developing a multifunctional drug
18	delivery system, consisting of several active therapeutic or imaging molecules; (5) protection of

1	anticancer drugs against deactivation during blood circulation, transport to targeted organs or tissues
2	and intracellular trafficking. Polysaccharide-based macromolecular prodrugs have gained increasing
3	attention, such as hyaluronic acid (HA), <sup>8-10</sup> dextran, <sup>11</sup> chitosan, <sup>12</sup> and heparin, <sup>13, 14</sup> due to their cost
4	effectiveness, abundance in nature, remarkable physicochemical, and biological characteristics and
5	simplicity of chemical reactions required for specific modifications.
6	Most of the polysaccharides contain reactive functional groups, such as amino, hydroxyl and
7	carboxylic groups, which may readily be utilized as an active site for drug conjugation either directly
8	or via linkers. Chondroitin sulfate (CS) is a natural polysaccharide widely distributed in the human
9	organism. It binds endogenous proteins with different functional properties such as growth factors,
10	adhesion molecules or enzymes, which regulate the human immune system. <sup>15</sup> CS has a similar
11	chemical structure to HA, which can recognize CD44 receptors overexpressing in many solid tumor
12	cells. <sup>16, 17</sup> Besides, CS has many promising properties including biocompatibility, biodegradability,
13	an anti-inflammatory agent. <sup>18</sup>
14	Our previous study successfully introduced methacrylate groups to CS (CSMA) that could be
15	reacted with poly(caprolactone) (PCL) end-capped double bonds to yield a graft copolymer. Various
16	amounts of PCL have been grafted onto CSMA with different degrees of methacrylation and the
17	chemicophysical properties and morphologies of PCL-graft-CS copolymers (CP) were thoroughly
18	characterized. <sup>19, 20</sup> It appears worthwhile to use PCL as a core and CS as a shell because CS is a

highly water-soluble polymer and PCL is a highly hydrophobic polymer, leading to a lower critical	
micelle concentration that benefits the stabilization of micelles circulation in the bloodstream. Two	
anticancer drugs have been loaded in the core of the self-assembled CP micelles for delivering	
doxorubicin $(DOX)^{21}$ or $CPT^{22}$ to suppress the growth of non-small cell lung tumors. The CP	pt
micelles are expected to be biodegradable due to the susceptibility of the aliphatic ester of PCL and	<sup>S</sup> Cri
the sugar linkages of CS to hydrolysis.	Snu
In this contribution, an optimized composition of PCL on CP was synthesized via a free radical	Ma
reaction. Because CP bears both of carboxylic and hydroxyl groups to be utilized to conjugate with	ed
anticancer drugs, thus, an anticancer agent, DOX, was chemically linked to CP to produce CP-DOX.	ept
Further, a cancer mediated ligand, folic acid (FA), was grafted onto CP-DOX to improve cellular	Advances Accepted Manuscript
uptake into tumor cells, which overexpress folate receptors (FR). In future, same or different	es l
hydrophobic anticancer agents can be loaded into the core of the self-assembled CP micelles to test	nc
the synergetic effect in chemotherapy.	dva
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#### **Experimental** 15

#### 16 Materials

Sodium chondroitin sulfate was purchased from Tohoku Miyagi Pharmaceutical Co., Ltd. (Tokyo, 17

18 Japan). Methacrylic anhydride and acryloyl chloride were purchased from Lancaster (Lancashire, Page 7 of 36

1	UK) and used as received. Pyrene, ε-caprolactone, 2,2'-azobisisobutyronitrile (AIBN), and 1,
2	1'-carbonyldiimidazole (CDI) were purchased from Acros (Geel, NJ). Bi-functional poly(ethylene
3	glycol) (NH <sub>2</sub> -PEG-NH <sub>2</sub> , MW= 3350 g/mol) and (NH <sub>2</sub> -PEG-COOH, MW= 3500 g/mol) were
4	purchased from Sigma chemical company (St. Louis, MO) and JenKem technology (Beijing, China),
5	respectively. Doxorubicin (DOX), dimethyl sulfoxide-d (DMSO-d <sub>6</sub> ), chloroform-d (CDCl <sub>3</sub> ), and
6	deuterium oxide (D <sub>2</sub> O) were purchased from Aldrich (St. Louis, MO). Fetal bovine serum (FBS) was
7	purchased from Biological Industries (Beit Haemek, Israel). Potassium dihydrogen phosphate,
8	disodium hydrogen phosphate, glycine, boric acid, and hydrochloric acid were purchased from Fluka
9	(Buchs, Switzerland) and used for buffer preparation. Roswell park memorial institute medium
10	(RPMI 1640) and trypsin-EDTA were obtained from Invitrogen (Carlsbad, CA). All other unstated
11	chemicals were purchased from Sigma chemical company and used without further purification.
12	
13	Preparing methacrylated chondroitin sulfate (CSMA), poly(ε-caprolactone) end-capped with
14	acrylated groups (MeO-PCL-Ac) and poly( <i>ɛ</i> -caprolactone)- <i>g</i> -chondroitin sulfate (CP)
15	CSMA and MeO-PCL-Ac were synthesized according to our previous report. <sup>21</sup> The degree of
16	methacrylation on CS was controlled at 70%. CP copolymers were synthesized using a free radical
17	reaction as previously described. <sup>21</sup> Briefly, AIBN (1 wt%) in DMSO relative to the total weight of
18	MeO-PCL-Ac and CSMA was added to the solution, where CSMA (100 mg) with 1 mL of double

1	deionized (DD) water and MeO-PCL-Ac (100 mg) with 100 mL of dimethylsulfoxide (DMSO) were
2	stirred in a two-neck round-bottom flask under an argon atmosphere at 60°C. Following 8 hours
3	reaction at 60°C, the reaction solution was cooled to room temperature and purified by dialysis
4	against DD water using MW cut-off 6000~8000 membrane (Spectrum Labs, Rancho Dominguez,
5	CA) for 2 days. The dialyzed solution was removed from the dialysis tube to a serum bottle, followed
6	by freeze-drying. The residue of MeO-PCL-Ac was removed by washing three times with ethyl
7	acetate. Next, the precipitate was dissolved in DD water and centrifuged for 5 minutes at 12,000 rpm.
8	The supernatant was removed and lyophilized to yield the final CP (yield: ~53%).
9	
10	Preparing folic acid-poly(ethylene glycol) (FA-PEG) and doxorubicin-poly(ethylene glycol)
10 11	Preparing folic acid-poly(ethylene glycol) (FA-PEG) and doxorubicin-poly(ethylene glycol) (DOX-PEG)
11	(DOX-PEG)
11 12	(DOX-PEG) Folic acid-poly(ethylene glycol) (FA-PEG). Folic acid (FA, 6.588 mg, 0.015 mmol) and
11 12 13	(DOX-PEG) Folic acid-poly(ethylene glycol) (FA-PEG). Folic acid (FA, 6.588 mg, 0.015 mmol) and 1,1-carbonyldimidazole (CDI, 7.9 mg, 0.049 mmol) were dissolved in a two-neck round-bottom
11 12 13 14	(DOX-PEG) Folic acid-poly(ethylene glycol) (FA-PEG). Folic acid (FA, 6.588 mg, 0.015 mmol) and 1,1-carbonyldimidazole (CDI, 7.9 mg, 0.049 mmol) were dissolved in a two-neck round-bottom flask containing 4 mL of DMSO under an argon atmosphere with stirring. After 24 hours stirring,
<ol> <li>11</li> <li>12</li> <li>13</li> <li>14</li> <li>15</li> </ol>	(DOX-PEG) Folic acid-poly(ethylene glycol) (FA-PEG). Folic acid (FA, 6.588 mg, 0.015 mmol) and 1,1-carbonyldimidazole (CDI, 7.9 mg, 0.049 mmol) were dissolved in a two-neck round-bottom flask containing 4 mL of DMSO under an argon atmosphere with stirring. After 24 hours stirring, bi-functional NH <sub>2</sub> -PEG-NH <sub>2</sub> (MW= 3350 g/mol, 50 mg, 0.015 mmol) with 4 mL of DMSO was

1	$\sim$ 65%). To quantify FA, a dried sample was dissolved in PBS buffer (pH 7.4) and the absorbance was
2	measured using a UV-Vis spectrophotometer at 280 nm. The FA amount was calculated against a
3	standard calibration curve of free FA in the concentration range of 0.01–0.06 mg/mL, assuming that
4	the absorption coefficient of FA-PEG was the same as FA.
5	<b>Doxorubicin-poly(ethylene glycol) (DOX-PEG).</b> DOX · HCl (5.00 mg, 0.0086 mmol) was
6	desalted by stirring with triethylamine (TEA, 1.44 $\mu$ L, 0.0105 mmol) in DMSO. Next,
7	N-hydroxysuccinimide (NHS, 1.00 mg, 0.0086 mmole) was added into the DOX solution under an
8	argon atmosphere at room temperature with stirring for 24 hours. Bi-functional NH <sub>2</sub> -PEG-COOH
9	( MW= 3500 g/mol, 30.17 mg, 0.0086 mmol) and dicyclohexylcarbodiimde (DCC, 1.75 mg, 0.0086
10	mmol) were added to the above solution and the reaction was stirred at room temperature for 24
11	hours under an argon atmosphere. The solution was dialyzed to remove unreacted DOX using MW
12	cut-off 1000 membrane for 2 days. All steps were performed in darkness. The product was obtained
13	via freeze-drying (yield: 67%). The amount of DOX conjugated with PEG was measured at 485 nm
14	using the UV-Vis spectrophotometer. The DOX amount was calculated against a standard calibration
15	curve, assuming that the absorption coefficient of DOX-PEG was the same as DOX. A DOX
16	calibration curve was built by dissolving free DOX in DMSO with a concentration range of 5–100
17	μg/mL.
10	

18

1	Preparing folic acid-poly(ethylene glycol)-CP (FA-PEG-CP), CP-poly(ethylene
2	glycol)-doxorubicin (CP-DOX) and folic acid-poly(ethylene glycol)-CP-poly(ethylene
3	glycol)-doxorubicin (FA-CP-DOX)
4	Folic acid-poly(ethylene glycol)-CP (FA-PEG-CP). FA-PEG and CP were separately
5	dissolved in DD water at a concentration of 1 mg/mL as a stock solution.
6	N-(3-dimethylaminopropyl)-N'-ethylcarbodiimide hydrochloride (EDAC) was mixed with the
7	FA-PEG stock solution for activating carboxylate groups of FA-PEG. Following 24 hours, the CP
8	stock solution was slowly added into the above solution and the reaction was stirred at room
9	temperature for another 24 hours. The solution was dialyzed to remove unreacted FA-PEG against
10	DD water for 2 days using MW cut-off 6000~8000 membrane. All steps were performed in darkness.
11	The FA-PEG-CP products were obtained via freeze-drying. To determine the amount of FA-PEG in
12	FA-PEG-CP, the calibration curve of FA-PEG at a concentration range of 0.01–0.06 mg/mL in PBS
13	buffer (pH 7.4) was created at 280 nm using the UV-Vis spectrophotometer, assuming that the
14	absorption coefficient of FA-PEG-CP was the same as FA-PEG.
15	CP-poly(ethylene glycol)-doxorubicin (CP-DOX). DOX-PEG (4 mg, 0.001 mmole) and
16	EDAC (2 mg, 0.0104 mmole) were dissolved in 2 mL of DD water with stirring at room temperature
17	for 24 hours. Ten mg of CP in 5 mL of DD water was added to the above solution and the reaction
18	was stirred at room temperature for another 24 hours. The solution was dialyzed using MW cut-off

1	6000~8000 membrane. All steps were performed in darkness. The product was obtained via
2	freeze-drying (yield: 60%). The content of DOX was calculated against the standard calibration
3	curve of DOX as aforementioned.
4	Folic acid-poly(ethylene glycol)-CP-poly(ethylene glycol)-doxorubicin (FA-CP-DOX).
5	DOX-PEG (4 mg, 0.001 mmole), FA-PEG (10 mg, 0.0026 mmole) and EDAC (4.2mg, 0.0209
6	mmole) were dissolved in 7 mL of DD water with stirring at room temperature for 24 hours. Next, 10
7	mg of CP in 5 mL of DD water were added to the above solution and the reaction was stirred for
8	another 24 hours. The solution was dialyzed using MW cut-off 6000~8000 membrane. All steps were
9	performed in darkness. The product was obtained via freeze-drying (yield: ~50%). The FA-PEG and
10	DOX contents of FA-CP-DOX were quantified against the standard calibration curves of FA-PEG
11	and DOX, respectively.
12	
13	Characterization
14	FTIR spectra were acquired using a Perkin-Elmer-2000 spectrometer. Dried samples were mixed
15	with potassium bromide (KBr) powder and pressed into pellets. Sixty-four scans were
16	signal-averaged in the range of 4000 - 400 cm <sup>-1</sup> at a resolution of 4 cm <sup>-1</sup> . The chemical structures of
17	samples were determined using proton nuclear magnetic resonance ( <sup>1</sup> H-NMR) spectroscopy (Varian
18	Mercury plus-200 spectrometer, Varian; Palo Alto, CA), where CSMA, FA-PEG, FA-PEG-CP,

DOX-PEG, CP-DOX, and FA-CP-DOX were dissolved in D<sub>2</sub>O, CP in DMSO-*d*<sub>6</sub>, and MeO-PCL-Ac
 in CDCl<sub>3</sub>, respectively.

3

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# 4 **Preparing and characterizing micelles**

5 Micelles were prepared by a dialysis method. Ten mg of copolymers were dissolved in 5 mL of DMSO containing 4 µL of trifluoroacetic acid at 60 °C. The solution was dialyzed against DD water 6 7 and freeze-dried to yield a micellar product. A critical micelle concentration (CMC) value was obtained using pyrene as a hydrophobic fluorescence probe. Pyrene at  $3.0 \times 10^{-5}$  M in acetone was 8 9 prepared as a stock solution. Sixty µL of the stock solution were added into each vial and acetone was removed by hair-drier. A series of micelles in DD water  $(1.91 \times 10^{-6} - 1.0 \text{ mg/mL})$  were added 10 11 into the vial containing pyrene and stored in darkness overnight. Pyrene excitation spectra were 12 recorded using an emission wavelength at 390 nm. The emission and excitation slit widths were set at 2.5 and 2.5 nm, respectively. The CMC value was determined from the ratios of pyrene intensities 13 14 at 339 and 334 nm and calculated from the intersection of two tangent plots of  $I_{339}/I_{334}$  vs the log concentrations of micelles.<sup>19</sup> 15 16 Micelles were suspended in DD water at a concentration of 0.5 mg/mL and hydrodynamic

diameters were measured three times using dynamic light scattering (DLS, Zetasizer Nano ZS,

1	electron microscopy (JEM-2000 EXII; JEOL, Tokyo, Japan). A carbon coated 200 mesh copper
2	specimen grid (Agar Scientific Ltd. Essex, UK) was glow-discharged for 1.5 minutes. Each micelle
3	solution (0.5 mg/mL) was dropped on a copper grid and allowed to dry at room temperature for 3
4	days.
5	
6	Drug release
7	The <i>in vitro</i> release of DOX from CP-DOX was determined in 0.1 M PBS buffers of pH 5.6, 6.4, and
8	7.4 at 37 °C. One mg of CP-DOX was suspended in 1 mL of the above buffers. Each eppendorf was
9	kept in a shaker at 37 °C and 150 rpm. At predetermined time intervals, four eppendorfs were
10	removed from the shaker and centrifuged at 12,000 rpm and 4 °C for 5 min. The supernatants were
11	collected to estimate the amount of DOX release, which was correlated to a standard calibration
12	curve of DOX in the same buffer using a UV-Vis spectrophotometer, as aforementioned.
13	
14	Cell experiments
15	U87 cells (a human glioblastoma cell line) were cultured in RPMI 1640 supplemented with 10%
16	FBS and 1% penicillin-streptomycin at 37°C under humidified 5% CO <sub>2</sub> . The medium was
17	replenished every three days. The cells were sub-cultured after they had reached confluence.
18	Live cells of U87 were counted using a trypan-blue dye exclusion method <sup>23</sup> after the cells had been

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1	incubated with CI	', DOX, FA-C	P-DOX, and CP	-DOX, respectively.

2	The cells were seeded in 12-well culture plates at a density of $1 \times 10^5$ cells per well in RPMI
3	1640 containing 10% FBS for 24 hours. The culture medium was removed and replaced with 1 mL
4	medium containing various concentrations of test samples. Following 24, 48 and 72 hours of
5	incubation, the cells were washed three times with PBS and trypsinized. The cells were collected and
6	resuspended with 1 mL of PBS in an eppendorf. The $10\mu L$ PBS solution containing the cells was
7	mixed with $10\mu$ L trypan blue and live cells were observed using a microscope (Nikon Eclipse TS100;
8	Tokyo, Japan).
9	
10	Flow cytometry
10	
11	The cellular uptake of free DOX, CP-DOX and FA-CP-DOX was studied using a flow cytometer.
11	The cellular uptake of free DOX, CP-DOX and FA-CP-DOX was studied using a flow cytometer.
11 12	The cellular uptake of free DOX, CP-DOX and FA-CP-DOX was studied using a flow cytometer. U87 cells were seeded at a density of $1 \times 10^5$ cells per well in 6-well plates in RPMI 1640
11 12 13	The cellular uptake of free DOX, CP-DOX and FA-CP-DOX was studied using a flow cytometer. U87 cells were seeded at a density of $1 \times 10^5$ cells per well in 6-well plates in RPMI 1640 supplemented with 10% FBS and incubated for 24 hours. Next, the culture medium was removed
11 12 13 14	The cellular uptake of free DOX, CP-DOX and FA-CP-DOX was studied using a flow cytometer. U87 cells were seeded at a density of $1 \times 10^5$ cells per well in 6-well plates in RPMI 1640 supplemented with 10% FBS and incubated for 24 hours. Next, the culture medium was removed and replaced with 2mL RPMI 1640 containing test samples at a DOX concentration of 2 µg/mL. The
11 12 13 14	The cellular uptake of free DOX, CP-DOX and FA-CP-DOX was studied using a flow cytometer. U87 cells were seeded at a density of $1 \times 10^5$ cells per well in 6-well plates in RPMI 1640 supplemented with 10% FBS and incubated for 24 hours. Next, the culture medium was removed and replaced with 2mL RPMI 1640 containing test samples at a DOX concentration of 2 µg/mL. The cells were further incubated at 37 °C for 2, 6 and 12 hours, washed three times with PBS, collected

1	free FA (2 mg/mL) for 24 hours. The medium was removed before 2 mL of folic acid-deficient RPMI
2	1640 containing test samples at a DOX concentration of 2 $\mu$ g/mL were added and incubated for 2
3	hours. Otherwise the cells were treated as same procedures as described above.
4	
5	Confocal laser scanning microscopy (CLSM)
6	U87 cells were seeded at a density of $1 \times 10^5$ cells per well in 12-well plates containing one glass
7	coverslip per well, in RPMI 1640 supplemented with 10% FBS for 24 hours. The 1 mL of RPMI
8	1640 (without 10% FBS) containing free DOX, FA-CP-DOX and CP-DOX at a DOX concentration
9	of 2 $\mu$ g/mL was added into each well and incubated at 37 °C for 2 hours. The coverslips containing
10	the cells were removed and washed gently with 2 mL of 0.1 M PBS (twice), treated with 5 $\mu g/mL$
11	Hoechst 33342 to stain cell nuclei for 30 minutes and 100 nM Lysotracker Green (DND-26,
12	Invitrogene; Carlsbad, CA) to stain endolysosomes for 1 hour. Next, the cells were washed three
13	times with 0.1 M PBS and fixed with 3.7% paraformaldehyde for 30 minutes. The cells on the
14	coverslips were washed three times with 0.1 M PBS and mounted with fluorescent mounting
15	medium on glass slides for CLSM (Fv 1000; Olympus, Tokyo, Japan) observation.
16	
17	Statistical methods
18	The results were reported as mean $\pm$ standard deviation. Statistical significance was determined by
19	two-tailed Student <i>t</i> -test. *P $< 0.05$ was considered to be significant

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# 2 **Results and discussion**

# 3 Synthesis of CP-DOX and FA-CP-DOX

Amphiphilic copolymers combining extremely different hydrophilic and hydrophobic characteristics 4 between CS and PCL have been successfully synthesized.<sup>19, 21</sup> Instead of using a PCL molecular 5 weight of 2200 g/mol in previous studies, we used 3500 g/mol to increase the hydrophobic core area 6 7 to accommodate more hydrophobic drugs for future studies. The weight ratio of the modified CS and PCL at 1/1 in feed was adapted to produce PCL-graft-CS (CP). The FT-IR spectrum of CP shows the 8 characteristic –OH stretching signal in the 3200 - 3600 cm<sup>-1</sup> region and the maximum peak intensity 9 of -CONH at 1646 cm<sup>-1</sup> mainly attributed to CS. The carbonyl -COO absorbance at 1726 cm<sup>-1</sup> was 10 assigned to PCL. The PCL composition of CP was characterized by <sup>1</sup>H-NMR, that is 8.1 mol%. 11 12 When the hydrophilic content is larger than the hydrophobic content, a micellar architecture is commonly observed.<sup>24</sup> 13

14 To link FA to CP, PEG was utilized as a spacer. The PEG diamine and FA at a molar ratio of 1 15 was set to produce FA-PEG using CDI as a coupling agent, slightly modified from the method reported in the literature.<sup>25</sup> The <sup>1</sup>H-NMR spectrum of FA-PEG shows the peaks at 6.6, 7.7 and 8.5 16 17 ppm because of the FA moiety and 3.8 ppm because of the methylene units of PEG (Supporting 18 Information Figure S1). The peak intensity ratio of FA and PEG was measured and used to estimate 19 the degree of FA linked to PEG, i.e.  $\sim 83.3\%$ . In parallel, a standard calibration curve of FA was built 20 using a UV-Vis spectrophotometer at 280 nm. The percentage of FA in FA-PEG calculated against the FA calibration curve ( $R^2 = 0.9987$ ) is ~ 96.7%, assuming that the absorption coefficient of 21 FA-PEG was the same as FA.<sup>26</sup> Both NMR and UV-Vis results implied that approximately one end of 22 23 PEG was modified with FA.

FA-PEG-grafted CP (FA-PEG-CP) was synthesized using the amino groups on the end of FA-PEG to react with the carboxylic groups of CP using EDAC as a conjugation agent. The graft percentage of FA-PEG onto CP was tested with various feeding weight ratios of FA-PEG and CP as shown in Table 1. The FA-PEG content of FA-PEG-CP was determined against the calibration curve done with FA-PEG using the UV-Vis spectrophotometer at 280 nm ( $R^2$ = 0.9998). The result demonstrated that the FA-PEG content of FA-PEG-CP increases linearly through increasing the FA-PEG feeding amount ( $R^2$ = 0.9918, Supporting Information Figure S2).

The <sup>1</sup>H-NMR and FTIR spectra of FA-PEG-CP synthesized at the weight ratio of 30:30 of 8 9 FA-PEG and CP were shown in Figure 1. The peaks appearing at 6.6, 7.7, and 8.5 ppm are 10 attributable to FA, 3.8 ppm to PEG, and 1.9 ppm to the acetamide group of CS (Figure 1a). Since the 11 NMR spectrum was done in D<sub>2</sub>O, the characteristic peaks of PCL are unclear due to the solubility 12 issue. The FT-IR spectrum of FA-PEG-CP appears to be a combination from the spectra of CP and 13 FA-PEG (Figure 1b). The –OH stretching in the 3100 - 3600 cm<sup>-1</sup> region and the maximum signal of carbonyl -COO absorbance at 1730 cm<sup>-1</sup> were attributed to CP and the C-O-C stretching at 1103 14 cm<sup>-1</sup> was attributed to FA-PEG. The FA-PEG concentration of FA-PEG-CP obtained from the 15 UV-Vis measurement is 10.87 µM. This weight ratio of 30:30 of FA-PEG and CP was adapted to link 16 17 DOX molecules in the later study.

Doxorubicin-poly(ethylene glycol) (DOX-PEG) was synthesized using the method modified from the literature.<sup>27</sup> In the <sup>1</sup>H-NMR spectrum of DOX-PEG (Figure 2a), the presence of DOX was confirmed by the appearance of signals at 6.5 - 8.5 ppm (Peaks B, C, D), attributable to the aromatic ring, and 1.18 ppm (Peak A), attributable to the methyl groups of DOX. The MALDI-MS spectrum exhibits the maximum peak of COOH-PEG-NH<sub>2</sub> at 3593.1 (Figure 2b, left panel), close to the reported molecular weight of 3500 g/mol from the supplier (JenKem technology). The molecular

1 weight of DOX is 543.5 g/mol. After the conjugation reaction occurring between COOH-PEG-NH<sub>2</sub> 2 and DOX, one mole of water was given off. Thus, the theoretical molecular weight of DOX-PEG is 3 4018.6 g/mol. The MALDI-MS spectrum of DOX-PEG (Figure 2b, right panel) shows the maximum 4 peak at 3994.6 m/e, close to the theoretical value. A standard calibration curve of DOX was created 5 using the UV-Vis spectrophotometer at 485 nm. The percentage of DOX in DOX-PEG calculated 6 against the DOX calibration curve ( $R^2$ = 0.9998) is ~ 68.5%.

7 The amino groups on the end of DOX-PEG were utilized to react with the carboxylic groups of 8 CP to yield CP-DOX. Similarly, the amino groups of DOX-PEG and FA-PEG were simultaneously reacted with CP to yield FA-CP-DOX. Their <sup>1</sup>H-NMR spectra were shown in Figure 3. In the 9 10 <sup>1</sup>H-NMR of CP-DOX (Figure 3a), Peak A is a characteristic signal of the methyl group of DOX, 11 Peak B is attributed to the acetamide of CS and Peak C is the ethylene linkage of PEG. In addition to the characteristic peaks, attributable to CS, DOX and PEG, the <sup>1</sup>H-NMR spectrum of FA-CP-DOX 12 13 shows the FA peaks of para-aminobenzoic acid (Peaks D, E) and aromatic pteridine ring (Peak F) at 14  $\delta$  6.80, 7.60 and 8.60 ppm, respectively. The mol% of DOX measured against the DOX calibration 15 curve are 8.34% for CP-DOX and 8.12% for FA-CP-DOX, respectively.

# 16 Micellar properties and drug release

Hydrophobic amphiphilic copolymers could be assembled through a simple dialysis method to form
micelles in an aqueous solution because of easy segregation between hydrophobic and hydrophilic
segments.<sup>28</sup> The CMC value of CP is 3.08 μg/mL determined using pyrene as a fluorescent probe
(Table 2). Modification of CP with FA-PEG dramatically increases a CMC value. The CMC value of
FA-PEG-CP goes hand-in-hand with an increase in FA-PEG feeding amount. For example, the CMC

1	value of FA-PEG-CP is 8.13 $\mu$ g/mL with a weight feed ratio of FA-PEG and CP at 5:30 and increases
2	to 13.2 $\mu$ g/mL at 50:30 (Table 2 and Supporting Information Figure S3). The introduction of
3	FA-PEG to CP increases the solubility of CP as well (Supporting Information Figure S4). At a
4	concentration of 1 mg/mL in DD water, CSMA is completely soluble; CP is turbid and FA-PEG-CP
5	at a feed weight ratio of 30:30 becomes transparent.
6	The hydrodynamic diameter of CP is 231 nm and slightly deceases as FA-PEG is grafted onto it.
7	For example, the hydrodynamic diameter of FA-PEG-CP at a weight ratio of 50:30 is ~208 nm.
8	Incorporation of DOX-PEG to CP increases the hydrodynamic diameter to 253 nm for CP-DOX and
9	241 nm for FA-CP-DOX (Table 2 and Supporting Information Figure S5). The zeta potential of CP is
10	-42 mV and increases to between -20 and -25 mV depending on the amount of FA-PEG grafted onto
11	CP (Table 2). A further graft of DOX-PEG to FA-PEG-CP (FA-CP-DOX) changed the zeta potential
12	to a less negative value because some carboxylic groups of CP had been utilized to react with
13	DOX-PEG, leading to reduced negative charges on the exterior surface of CP.
14	The morphology of the self-assembled CP was examined by TEM as a spheroid (Figure 4 $a - c$ ).
15	Apparently, FA-PEG-CP is worm-like using a weight ratio of FA-PEG and CP at 30:30 (Figure 4 d –
16	f). After grafting DOX onto CP or FA-PEG-CP, CP-DOX still shows a spherical image (Figure 4 g –
17	i) and FA-CP-DOX is worm-like, a similar appearance to FA-PEG-CP (Figure 4 $j$ – 1). The average
18	particle sizes estimated from TEM images are $90.9 \pm 7.6$ , $82.2 \pm 9.6$ , $92.5 \pm 3.2$ , and $87.6 \pm 8.2$ nm,

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1	for CP, FA-PEG-CP, CP-DOX, and FA-CP-DOX, respectively. Since the TEM samples were done in
2	a dehydrated state and those of DLS in a solution, the larger particle diameters observed in DLS
3	could be attributed to the swelling behavior of the shell compartment. <sup>29</sup>
4	In vitro DOX release from CP-DOX was carried out in 0.1 M PBS and pH of 5.6, 6.4 and 7.4 at
5	37 °C. As seen in Figure 5, about 8% of DOX was released at pH 7.4, 50% at pH 6.4, and 65% at pH
6	5.6 within 180 hours. CP-DOX was expected to have an effective drug release in the tumor
7	microenvironment since the drug release rate accelerated with the decrease in pH. A similar drug
8	released profile was speculated to FA-CP-DOX because it has the same hydrolysable linkage as
9	CP-DOX. Although tumor pH may vary according to the tumor area, average extracellular tumor pH
10	is between 6.0 and 7.0, whereas in normal tissues and blood, the extracellular pH is around 7.4. <sup>30, 31</sup>
11	In addition, most of the nanoparticles are taken up by endocytosis into cells <sup>32</sup> and this endocytic
12	pathway begins near the physiological pH of 7.4, then it drops to pH 5.5–6.0 in endosomes. <sup>33</sup> Thus,
13	the pH-dependent release behavior of CP-DOX will benefit tumor-targeted DOX delivery.
14	

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#### 15 Cytotoxicity and intracellular uptake

16 Cytotoxic data demonstrated that U87 cells are slightly dose dependent on the CP concentration and show ~93% cell viability, even at a high concentration of 1000 µg/mL (Supporting Information 17 18 Figure S6). To test which DOX formulation is more sensitive to U87 cells, the cells exposed to

1

various concentrations of free DOX, CP-DOX and FA-CP-DOX were studied at three incubation	
time points of 24, 48 and 72 hours (Supporting Information Figure S7). The $IC_{50}$ values, defined as a	
concentration inhibiting the 50% cell proliferation, were summarized in Table 3. The $IC_{50}$ values	
decrease as the incubation time period increases. The values are 3.59, 0.97, and 0.39 $\mu$ g/mL of U87	bt
cells exposed to free DOX for 24, 48 and 72 hours, respectively. Free DOX shows the highest	CUI
cytotoxicity against U87 cells and FA-CP-DOX is close to free DOX. Both have a better cell-killing	Snu
effect than CP-DOX.	Mal
The folate receptor on the cell membrane was examined after U87 cells had been treated with	ea
folate receptor antibody using flow cytometry (Supporting Information Figure S8). The pattern was	ept
clearly observed a right shift as compared with the control group, explaining the folate receptor	vances Accepted Manuscript
antibody strongly interacts with the cells. The result explained that U87 is a folic acid-expressing cell	Sé l
line. Flow cytometric analysis was utilized to study the cellular uptake of free DOX, CP-DOX and	nce
FA-CP-DOX into U87 cells at different time points using an equivalent DOX concentration of 2	dva
μg/mL. A greater shift of a histogram to the right implies a larger amount of DOX was internalized	A
into the cells. The largest shift was always observed in FA-CP-DOX at 3 test time points of 2, 6 and	KSC Ad
12 hours (Figure 6) Clearly, the cellular untake of nanonarticles gradually increases with time	

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2	time points of 24, 48 and 72 hours (Supporting Information Figure S7). The $IC_{50}$ values, defined as a
3	concentration inhibiting the 50% cell proliferation, were summarized in Table 3. The $IC_{50}$ values
4	decrease as the incubation time period increases. The values are 3.59, 0.97, and 0.39 $\mu g/mL$ of U87
5	cells exposed to free DOX for 24, 48 and 72 hours, respectively. Free DOX shows the highest
6	cytotoxicity against U87 cells and FA-CP-DOX is close to free DOX. Both have a better cell-killing
7	effect than CP-DOX.
8	The folate receptor on the cell membrane was examined after U87 cells had been treated with
9	folate receptor antibody using flow cytometry (Supporting Information Figure S8). The pattern was
10	clearly observed a right shift as compared with the control group, explaining the folate receptor
11	antibody strongly interacts with the cells. The result explained that U87 is a folic acid-expressing cell
12	line. Flow cytometric analysis was utilized to study the cellular uptake of free DOX, CP-DOX and
13	FA-CP-DOX into U87 cells at different time points using an equivalent DOX concentration of 2
14	$\mu$ g/mL. A greater shift of a histogram to the right implies a larger amount of DOX was internalized
15	into the cells. The largest shift was always observed in FA-CP-DOX at 3 test time points of 2, 6 and
16	12 hours (Figure 6). Clearly, the cellular uptake of nanoparticles gradually increases with time.
17	A further comparison of cellular uptake among free DOX, CP-DOX and FA-CP-DOX was
18	carried out using CLSM following 2 h of incubation. As seen in Figure 7, both CP-DOX and

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1	FA-CP-DOX enter into the cytoplasm and localize at the endolysosomes of U87 cells. In contrast,
2	low fluorescence was observed for the cells incubated with free DOX. The possible explanation for
3	the higher uptake in the formulated DOX was that micellar particles are usually internalized inside
4	cells by endocytosis, while free drugs mainly do this by diffusion. <sup>34</sup> Moreover, the superior uptake of
5	FA-CP-DOX compared to free DOX as well as CP-DOX might be because FA-CP-DOX recognizes
6	FR, which facilitates FR-mediated endocytosis. To confirm this internalization route, the cells were
7	pretreated with FA to block the FR-mediated uptake process, prior to incubation with CP-DOX or
8	FA-CP-DOX. A clear decrease in uptake was observed in FA-CP-DOX alone (Figure 8), indicating
9	that the FR-mediated endocytosis pathway was indeed involved in the internalization of
10	FA-CP-DOX into U87 cells.
11	Optimizing the drug release from drug-encapsulated nanoparticles after entering inside cells is
12	also a key factor in designing a successful drug delivery system. In addition to chemically-loaded
13	DOX to CP-DOX and FA-CP-DOX, the hydrophobic core area of CP-DOX and FA-CP-DOX could
14	be utilized to physically load a hydrophobic drug to increase the synergistic effect in chemotherapy
15	in future.
16	
17	Conclusion
18	We successfully conjugated CP with FA-PEG and DOX-PEG to produce FA-CP-DOX as

19 macromolecular prodrugs as well as hydrophobic drug carriers. Modification of CP with FA-PEG

20 dramatically increased a CMC value, which increased with an increase in FA-PEG content. The

1	hydrodynamic diameter decreased with an increased FA-PEG content. Conjugation of DOX to CP or
2	FA-PEG-CP increased the hydrodynamic diameters as compared to those without DOX. The zeta
3	potential of CP became less negative as FA-PEG was grafted onto its surface. A graft of DOX to
4	FA-PEG-CP changed the zeta potential to a less negative value. From TEM images, CP and
5	DOX-CP were spherical, but FA-PEG-CP and FA-CP-DOX were worm-like. DOX-CP and
6	FA-CP-DOX were expected to have an effective drug release in the tumor microenvironment
7	because the drug release rate accelerated with the decrease in pH. FA-CP-DOX showed the
8	comparable cytotoxicity to free DOX against U87 cells. In addition, FA-CP-DOX displayed the
9	highest DOX internalization among the three test samples, DOX, CP-DOX and FA-CP-DOX. The
10	internalization pathway of FA-CP-DOX was involved with FR-mediated endocytosis in
11	FR-expressing U87 cells. In this study, a new type of amphiphilic FA-CP-DOX copolymer was
12	successfully prepared and its potential application as a macromolecular prodrug as well as a drug
13	carrier was illustrated.
14	
15	Conflict of interest
16	The authors declare no conflict of interest.
17	
18	Acknowledgements

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# 1 References

- H. Zhang, J. Wang, W. Mao, J. Huang, X. Wu, Y. Shen and M. Sui, *J. Controlled Release*, 2013, 166, 147-158.
- H. Xiao, H. Song, Q. Yang, H. Cai, R. Qi, L. Yan, S. Liu, Y. Zheng, Y. Huang, T. Liu and X. Jing, *Biomaterials*, 2012, 33, 6507-6519.
- 6 3. F. Kratz, Expert Opin. Inv. Drug, 2007, 16, 855-866.
- 7 4. S. Zhu, L. Qian, M. Hong, L. Zhang, Y. Pei and Y. Jiang, *Adv. Mater.*, 2011, 23, 84-89.
- W. L. Ye, J. B. Du, B. L. Zhang, R. Na, Y. F. Song, Q. B. Mei, M. G. Zhao and S. Y. Zhou,
   *PLoS One*, 2014, 9, e97358.
- 10 6. C. Young, T. Schluep, J. Hwang and S. Eliasof, Curr. Bioact. Compd., 2011, 7, 8-14.
- N. Goodarzi, R. Varshochian, G. Kamalinia, F. Atyabi and R. Dinarvand, *Carbohydr. Polym.*,
   2013, 92, 1280-1293.
- N. Goodarzi, M. H. Ghahremani, M. Amini, F. Atyabi, S. N. Ostad, N. Shabani Ravari, N.
   Nateghian and R. Dinarvand, *Chem. Biol. Drug Des.*, 2014, 83, 741-752.
- 15 9. S. Manju and K. Sreenivasan, J. Colloid Interf. Sci., 2011, **359**, 318-325.
- S. Cai, S. Thati, T. R. Bagby, H. M. Diab, N. M. Davies, M. S. Cohen and M. L. Forrest, J.
   *Controlled Release*, 2010, 146, 212-218.
- 18 11. S. Mitra, U. Gaur, P. C. Ghosh and A. N. Maitra, J. Controlled Release, 2001, 74, 317-323.
- 19 12. F. Q. Hu, L. N. Liu, Y. Z. Du and H. Yuan, *Biomaterials*, 2009, **30**, 6955-6963.
- 20 13. I.-K. Park, Y. J. Kim, T. H. Tran, K. M. Huh and Y.-k. Lee, *Polymer*, 2010, **51**, 3387-3393.
- 21 14. W. She, N. Li, K. Luo, C. Guo, G. Wang, Y. Geng and Z. Gu, *Biomaterials*, 2013, 34, 2252-2264.
- 23 15. V. R. Pipitone, Drugs Exp. Clin. Res., 1991, 17, 3-7.
- A. Pathak, P. Kumar, K. Chuttani, S. Jain, A. K. Mishra, S. P. Vyas and K. C. Gupta, ACS
   *Nano*, 2009, **3**, 1493-1505.
- T. Kurosaki, T. Kitahara, S. Kawakami, K. Nishida, J. Nakamura, M. Teshima, H. Nakagawa,
  Y. Kodama, H. To and H. Sasaki, *Biomaterials*, 2009, **30**, 4427-4434.
- 28 18. N. Volpi, Curr. Med. Chem-Anti-inflammatory & Anti-Allergy Agents, 2005, 4, 221-234.
- 29 19. A. L. Chen, H. C. Ni, L. F. Wang and J. S. Chen, *Biomacromolecules*, 2008, 9, 2447-2457.
- 30 20. L. F. Wang, H. C. Ni and C. C. Lin, J. Biomater. Sci. Polym. Ed., 2011, 23, 1821-1842.
- 31 21. Y. J. Lin, Y. S. Liu, H. H. Yeh, T. L. Cheng and L. F. Wang, *Int. J. Nanomed.*, 2012, 7,
   32 4169-4183.
- 33 22. Y. S. Liu, C. C. Chiu, H. Y. Chen, S. H. Chen and L. F. Wang, *Mol. Pharm.*, 2014, 11, 1164-1175.
- R. Khonkarn, S. Mankhetkorn, W. E. Hennink and S. Okonogi, *Eur. J. Pharm. Biopharm.*,
  2011, **79**, 268-275.
- 37 24. K. Letchford and H. Burt, Eur. J. Pharm. Biopharm., 2007, 65, 259-269.
- 38 25. G. Xiang, J. Wu, Y. Lu, Z. Liu and R. J. Lee, Int. J. Pharm., 2008, 356, 29-36.
- 39 26. H. Zhu, F. Liu, J. Guo, J. Xue, Z. Qian and Y. Gu, *Carbohydrate Polymers*, 2011, 86, 1118-1129.
- 41 27. H. S. Yoo and T. G. Park, *J Controlled Release*, 2004, **100**, 247-256.
- 42 28. C. Allen, D. Maysinger and A. Eisenberg, *Colloid Surface B*, 1999, 19, 3-27.
- 43 29. M. Bodnar, J. F. Hartmann and J. Borbely, *Biomacromolecules*, 2006, 7, 3030-3036.
- 44 30. F. Danhier, O. Feron and V. Préat, J. Controlled Release, 2010, 148, 135-146.
- 45 31. M. Breunig, S. Bauer and A. Goepferich, *Eur. J. Pharm. Biopharm.*, 2008, 68, 112-128.

- 1 32. T. Tanaka, S. Shiramoto, M. Miyashita, Y. Fujishima and Y. Kaneo, *Int. J. Pharm.*, 2004, **277**, 39-61.
- 3 33. J. S. Park, T. H. Han, K. Y. Lee, S. S. Han, J. J. Hwang, D. H. Moon, S. Y. Kim and Y. W.
   4 Cho, *J. Controlled Release*, 2006, 115, 37-45.
- 5 34. N. Nishiyama and K. Kataoka, Adv. Polym. Sci., 2006, 193, 67-101.
- 6 7

1	Figure Captions				
2	<b>Figure 1.</b> (a) <sup>1</sup> H-NMR spectrum of FA-PEG-CP done in $D_2O$ . Peaks A & B are respectively				
3	attributed to two protons of benzoic acid in folic acid, Peak C to the pteridine proton in folic acid,				
4	Peak D to the methylene units in PEG and Peak E to the methyl group of acetamide in CS. (b) FTIR				
5	spectra of CP, FA-PEG, and FA-PEG-CP.				
6					
7	Figure 2. (a) $^{1}$ H-NMR spectrum of DOX-PEG and (b) MALDI-MS spectra of COOH-PEG-NH <sub>2</sub> and				
8	DOX-PEG.				
9					
10	Figure 3. <sup>1</sup> H-NMR spectra of (a) CP-DOX and (b) FA-CP-DOX.				
11					
12	<b>Figure 4.</b> TEM images of (a) - (c), CP; (d) - (f), FA-PEG-CP; (g) - (i), CP-DOX and (j) - (l),				
13	FA-CP-DOX with different magnefications.				
14					
15	Figure 5. In vitro DOX released from CP-DOX at pH=7.4, 6.4, and 5.6 PBS.				
16	Et ( El				
17 18	<b>Figure 6.</b> Flow cytometric histograms of free DOX, FA-CP-DOX and CP-DOX relative to U87				
18	control cells at 2, 6 and 12 h of incubation. The equivalent DOX concentration was 2 $\mu$ g/mL.				
19 20	Figure 7. Confocal microscopic photographs of free DOX, FA-CP-DOX and CP-DOX in U87 cells				
20 21	at 2 h of incubation. The equivalent DOX concentration was 2 $\mu$ g/mL. Nuclei were stained by				
21	Hoechest 33342 in blue, endolysosomes stained by Lysotracker in green, and DOX in red.				
22	Hotenest 55542 in olde, endolysosomes stanted by Lysotracker in green, and DOX in red.				
23	Figure 8. Flow cytometric histograms of FA-CP-DOX and CP-DOX in the presence and absence of				
25	folic acid in U87 cells at 2 h of incubation. The equivalent DOX concentration was 2 $\mu$ g/mL.				
26					
27					

1 Table 1. The FA content of FA-PEG-CP using various weight ratios of FA-PEG and CP

FA-PEG:CP (Weight ratio)	5:30	10:30	15:30	30:30	50:30
Absorbance <sup>#</sup>	0.142	0.186	0.216	0.325	0.393
Concentration (mg/mL)	0.018	0.023	0.027	0.041	0.049
$\mu \mathbf{M}$	4.76±0.23	6.22±0.21	7.22±0.21	10.87±0.12	13.16±0.16
FA-PEG:CP (Weight ratio)	30:10	40:10	50:10	60:10	70:10
	<b>30:10</b> 0.462	<b>40:10</b> 0.493	<b>50:10</b> 0.505	<b>60:10</b> 0.528	<b>70:10</b> 0.570
(Weight ratio)					

2 (polycaprolactone-*graft*-chondroitin sulfate) in feed

<sup>#</sup>Measured at 280 nm by UV-visible spectrometry.

Table 2. Hydrodynamic diameter, zeta potentials, and critical micellar concentrations of

FA-PEG-CP at various weight ratios of FA-PEG and CP (polycaprolactone-graft-chondroitin

1

2 3

4

sulfate) in feed				
	$D_h (nm)^a$	PDI <sup>a</sup>	Zeta (mV)	CMC (mg/mL)
СР	231.2±30.5	$0.404 \pm 0.02$	-42.2±2.8	3.08×10 <sup>-3</sup>
FA-PEG:CP (5:30)	236.9±24.5	$0.352 \pm 0.03$	-21.5±2.6	8.13×10 <sup>-3</sup>
(10:30)	215.1±17.2	0.329±0.03	-22.6±3.1	8.33×10 <sup>-3</sup>
(15:30)	214.6±15.1	$0.298 \pm 0.02$	-24.2±5.7	8.61×10 <sup>-3</sup>
(30:30)	222.7±9.5	0.215±0.03	-22.7±5.1	9.67×10 <sup>-3</sup>
(50:30)	208.2±8.1	0.257±0.01	-20.5±2.1	1.32×10 <sup>-2</sup>
CP-DOX <sup>#</sup>	253.3±21.2	0.372±0.08	-19.5±4.2	ND
FA-CP-DOX <sup>#</sup>	241.2±24.5	0.318±0.09	-17.6±1.2	ND

<sup>#</sup> CP-DOX and FA-CP-DOX were prepared at the weight ratio of FA-PEG and CP at 30:30. 5

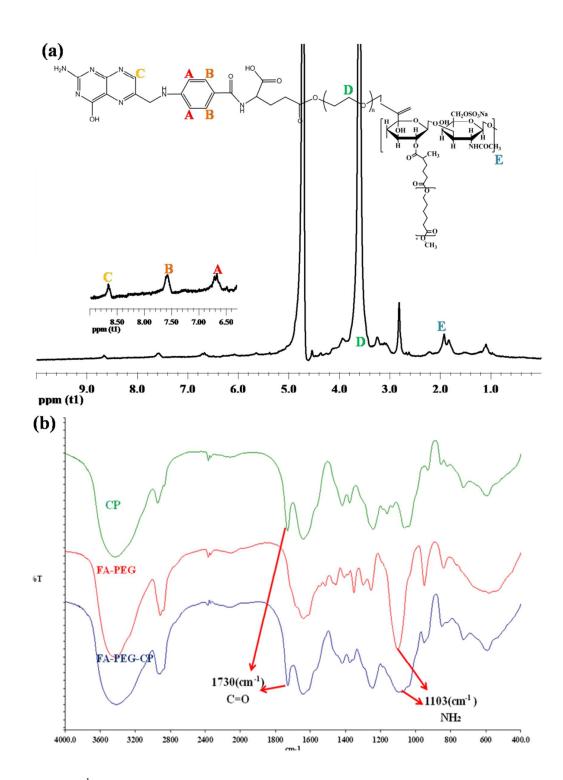
<sup>a</sup> Micelles sizes and polydispersity index (PDI) were measured at a concentration of 0.1 mg/mL in DD 6 7 water by DLS.

CMC: Critical micellar concentration; ND: Not determined. 8

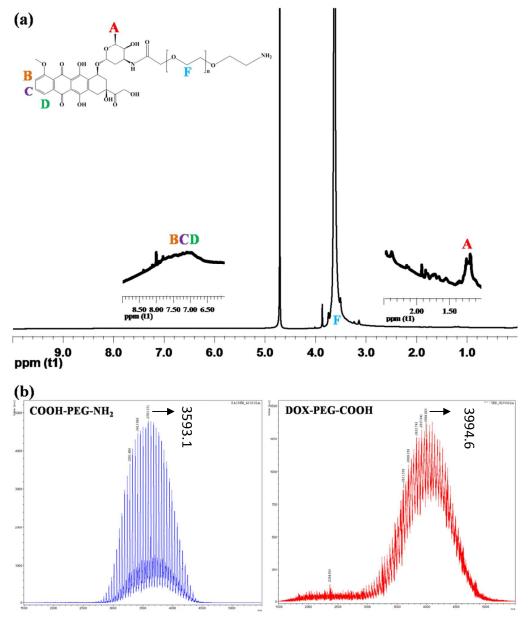
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- 10 11

Table 3. The IC<sub>50</sub> values of drug-treated U87 cells at three incubation time periods **Incubate time** 24 h 48 h 72 h Sample Free DOX 3.59 µg/mL  $0.97 \ \mu g/mL$ 0.39 µg/mL FA-CP-DOX 0.99 µg/mL 4.64 μg/mL 0.47 µg/mL **CP-DOX** 8.89 µg/mL 1.87 µg/mL 0.92 µg/mL

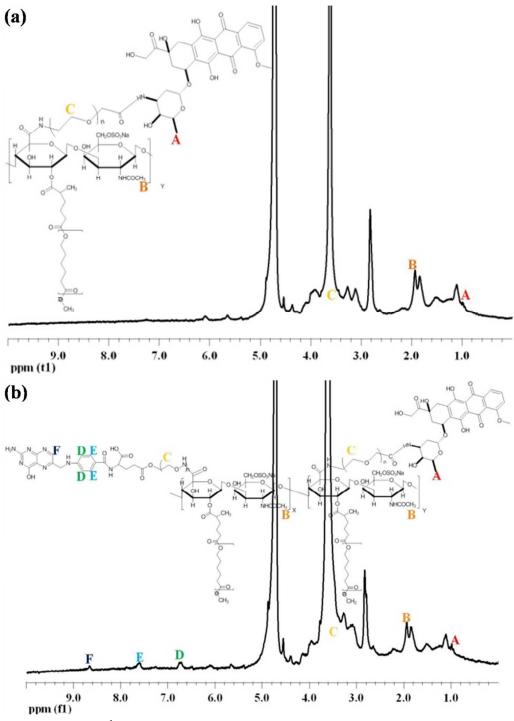
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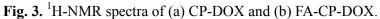


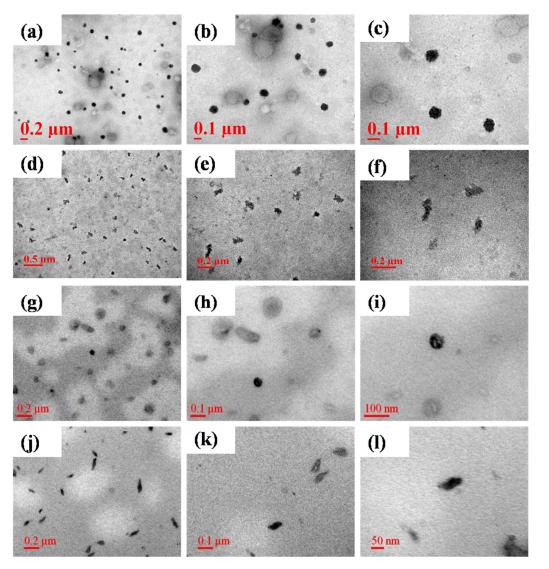
**Fig. 1.** (a) <sup>1</sup>H-NMR spectrum of FA-PEG-CP done in  $D_2O$ . Peaks A & B are respectively attributed to two protons of benzoic acid in folic acid, Peak C to the pteridine proton in folic acid, Peak D to the methylene unit in PEG and Peak E to the methyl group of acetamide in CS. (b) FTIR spectra of CP, FA-PEG, and FA-PEG-CP.



**Fig. 2.** (a) <sup>1</sup>H-NMR spectrum of DOX-PEG and (b) MALDI-MS spectra of COOH-PEG-NH<sub>2</sub> and DOX-PEG







**Fig. 4.** TEM images of (a) - (c) CP, (d) - (f) FA-PEG-CP, (g) - (i) CP-DOX, and (j) - (l) FA-CP-DOX with different magnefications.

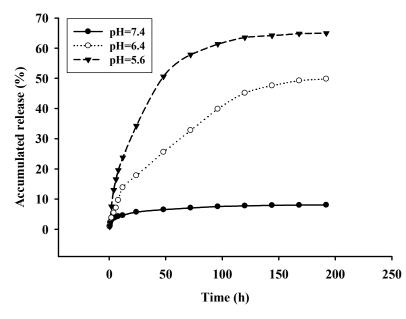


Fig. 5. In vitro DOX released from CP-DOX at pH=7.4, 6.4, and 5.6 PBS.

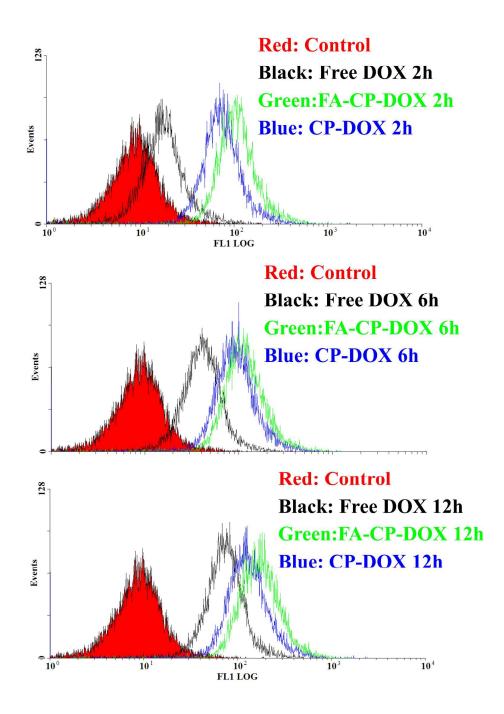
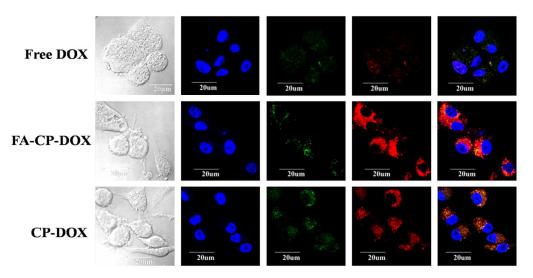
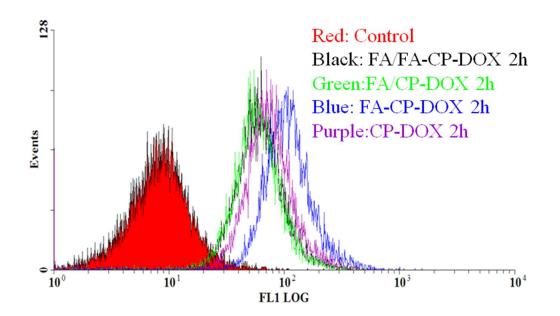


Fig. 6. Flow cytometric histograms of free DOX, FA-CP-DOX, and CP-DOX relative to U87 control cells at 2, 6, and 12 h of incubation. The equivalent DOX concentration was 2 μg/mL.



**Fig. 7.** Confocal microscopic photographs of free DOX, FA-CP-DOX and CP-DOX in U87 cells at 2 h of incubation. The equivalent DOX concentration was 2  $\mu$ g/mL. Nuclei stained by Hoechest 33342 in blue, endolysosomes stained by Lysotracker in green and DOX in red.



**Fig. 8.** Flow cytometric histograms of FA-CP-DOX and CP-DOX in the presence and absence of folic acid in U87 cells at 2 h of incubation. The equivalent DOX concentration was  $2 \mu g/mL$ .