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Graphical Abstract:



I	Calcium mitigates the stress caused by $2nSO_4$ as support fertilizer and enhances
2	sulforaphane formation of broccoli sprouts
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15 Abstract

14 In order to improve the growth condition of broccoli sprouts under ZnSO₄ application, exogenous CaCl₂ was added in the cultural medium. Then the growth profiles, key bioactive 15 16 substances (glucosinolates, sulforaphane, ascorbic acid and phenolic compounds), antioxidant capacity, myrosinase activity and related genes expression of broccoli sprouts were evaluated. 17 Results showed that the stressful condition caused by ZnSO₄ was effectively mitigated by CaCl₂. 18 19 Consequently, ascorbic acid and total phenolics content as well as antioxidant capacity of broccoli 20 sprouts decreased compared with that of sole ZnSO₄ treatment. However, sulforaphane formation increased since the higher glucoraphanin content, myrosinase activity and the related genes 21 22 expression induced after CaCl₂ treatment. Glucoraphanin content and sulforaphane formation of the water-treated sprouts decreased steadily during germination. However, sulforaphane formation of 23 24 sprouts treated by ZnSO₄ plus CaCl₂ increased after germinating for 2 days. These results suggested 25 that CaCl₂ could mitigate the stress caused by ZnSO₄ and enhance the sulforaphane formation of broccoli sprouts. 26

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28 Keywords: Broccoli sprouts, calcium, ZnSO₄ stress, glucoraphanin, sulforaphane, myrosinase

30 **1 Introduction**

Compared with mature counterparts, broccoli sprouts are more globally consumed as a 31 promising food since they are rich in health-promoting compounds such as folic acid, phenolics, 32 ascorbic acid and glucosinolates.^{1,2} Regular consumption of broccoli sprouts can reduce the risk of 33 developing chronic diseases and different types of cancer.³ These beneficial effects are mainly due 34 to the presence of glucosinolates, especially glucoraphanin that accounts for over 50% of total 35 glucosinolate in most broccoli cultivars.^{4,5} Upon a disruption of plant tissues, glucoraphanin will be 36 37 hydrolyzed by myrosinase into sulforaphane. Sulforaphane is proved to be a natural inducer of the phase II detoxication enzyme to detoxify cancer-causing chemicals.³ In plants, sulforaphane 38 formation depends on the content of glucoraphanin biosynthesized by flavin-containing 39 monooxygenase (FMO). Except for being hydrolyzed by myrosinase, glucoraphanin also serves as a 40 precursor for biosynthesis of alkenyl-glucosinolates by AOP2.⁶ In addition, sulforaphane formation 41 also depends on myrosinase and epithiospecifier protein (ESP) activity. High ESP activity is 42 beneficial for sulforaphane nitrile formation. Hence, it is necessary to investigate these enzymes 43 gene expression and sulforaphane formation of broccoli sprouts. 44

Glucosinolates as one kind of sulphur-containing compound may be considered as a source of sulphur for sprouts growth under low-sulphur conditions.⁷ Simultaneously, the biosynthesis of glucosinolates also needs sulphur.⁷ Hence, sulphur has been applied to broccoli sprouts growth in order to accumulate glucosinolates.^{8, 9} However, the exceeded application of sulphur would decrease broccoli sprouts yield, significantly.⁸ In our previous study, we compared the 2 mmol/L of ZnSO₄, K₂SO₄ and methionine (Met) on growth and glucosinolates contents of broccoli sprouts. The results showed that only ZnSO₄ increased glucosinolate content but inhibited broccoli sprouts

growth, significantly (Fig. S1, Supplementary data). Hence, how to mitigate the stress caused by
ZnSO₄ and increase the yield of the sprouts is very important.
Calcium is known to be involved in signal transduction in plant through binding with
Ca²⁺-sensors such as calmodulin to induce appropriate physiological cellular responses.¹⁰ Besides,
Ca²⁺ plays an important role in maintaining the functional and structural integrity of cell
membranes.¹¹ Epstein ¹² clearly demonstrated that the addition of supplemental Ca²⁺ in the growth
medium could enhance salt tolerance of plant. Therefore, it is hypothesized that whether Ca²⁺ could

mitigate the stress caused by ZnSO₄ during broccoli sprouts growth and further improve the content
 of bioactive substances.

In the present study, we attempted to add Ca^{2+} in the growth medium of broccoli sprouts paralleled with ZnSO₄ application. Subsequently, the growth profiles, key bioactive substances (glucosinolates, sulforaphane, ascorbic acid and phenolic compounds), antioxidant capacity, myrosinase activity and related genes expression of broccoli sprouts were evaluated. The changing patterns of glucoraphanin content, sulforaphane formation and myrosinase activity of germinating broccoli seeds and sprouts were also investigated.

67 **2 Materials and methods**

68 2.1 Materials and reagents

Seeds of broccoli (*B. oleracea* L. var. *italica*) were purchased from Nanjing Jinshengda Seed
Co. Ltd. (Jiangsu, China). Standard samples of sulforaphane and glucotropaeolin were purchased
from Sigma Chemical Co., (St. Louis, MO, USA). Other chemicals and reagents were of analytical
grade and purchased from Shanghai Institute of Biochemistry (Shanghai, China).

73 **2.2 Seed germination**

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Dry seeds were immersed in 1.5% sodium hypochlorite for 15 min, then drained and washed with distilled water until they reached neutral pH. They were then placed in distilled water and soaked at 30 °C for 4 h. Soaked seeds were germinated on a filter paper in petri dishes (15 cm in diameter) filled with sterilized quartz sand and grew in the incubators at 30 °C in darkness for 4 days. Sprout samples were rapidly and gently collected from the surface of the filter paper. Then the fresh weight (FW) of the broccoli sprouts was weighed, after which they were immediately frozen in liquid nitrogen and kept in polyethylene bags at -70 °C for further experiments.

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2.3 ZnSO₄ and CaCl₂ treatment

Seeds in each petri dish were supplied with 60 mL distilled water for the first day of 82 germination and then every 12 h with 20 mL of the following testing solutions respectively: (i) 83 84 distilled water as the control (Water); (ii) 4 mmol/L ZnSO₄ (Zn); (iii) 11 mmol/L CaCl₂ (Ca); (iv) 4 85 mmol/L ZnSO₄ plus additional mixture of 11 mmol/L CaCl₂ (Zn+Ca). The ZnSO₄ concentration of 4 mmol/L was pre-selected with the highest sulforaphane formation while inhibited sprouts growth, 86 significantly (Fig. S2, Supplementary data). As for the CaCl₂ concentration, we did a series of 87 concentration including 0, 3, 5, 7, 9, 11, 13 and 15, respectively. Then the 11 mmol/L was chosen 88 89 since it could get the highest sulforaphane formation of sprouts and also increase the sprouts length 90 under stress of 4 mmol/L ZnSO₄ (Fig. S3, Supplementary data).

91 **2.4 Sprout length measurement**

Twenty sprouts were taken as one sample group. Their length was measured using a vernier
 caliper (Shanghai Precision Instruments Co., Ltd., Shanghai, China), directly.

94 **2.5 Malondialdehyde (MDA) level determination**

95 M

MDA was determined using the method of 2-thiobarbituric acid reaction. Fresh sprouts (0.5 g)

were homogenized with 5 mL of 5% trichloroacetic acid. The mixture was centrifuged at 10,000
rpm for 17 min (4 °C). The supernatant was used to determine MDA content as described by Yang *et al.*¹³ MDA content was expressed as nmol/g FW.

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2.6 Ascorbic acid content determination

Ascorbic acid content was determined according to Volden *et al.*¹⁴ Fresh broccoli sprouts (0.3
g) were extracted with 5 mL of 1.0% (w/v) oxalic acid. After centrifuged at 10,000 rpm for 15 min,
the supernatant was filtered through a 0.45 μm Millipore before injected into HPLC (Agilent 1200,
USA) with Agilent 1200 tunable absorbance detector and a reversed-phase C18 column (4.6×250
mm, 5 μm, ZORBAX. Eclips) and detected at 254 nm. Ascorbic acid was calculated by external
standard curve and expressed as mg per 100 g FW.

106 **2.7 Total phenolics content determination**

Total phenolics of the sprouts (0.2 g) was extracted with 5.0 mL of 50% methanol. The mixture was centrifuged at 10,000 rpm for 15 min. One mL of the supernatant was mixed with 1 mL of 0.2 mmol/L Folin-Ciocalteu reagent and 2.0 mL of 2% Na₂CO₃. The mixture reacted at room temperature in the darkness for 2 h. Its absorbance was read at 765 nm. Gallic acid was used as the standard. Results were expressed as mg of gallic acid equivalent (GAE) per 100 g FW.

112 **2.8 Antioxidant capacity test**

The sample extract was prepared the same as that of total phenolics. The antioxidant capacity was evaluated using the T-AOC Kit (Nanjing Jiancheng Biotech Inc., Jiangsu, China). One unit of antioxidant capacity was defined as an increase in absorbance of 0.01 at 520 nm per min per gram at 37 °C of the reaction system.

117 **2.9 Glucosinolates analysis**

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Glucosinolates were extracted and analyzed as previously reported by Guo et al.¹⁵ Briefly, 118 samples (500 mg) were extracted with 2 mL of 75% methanol at 80 °C for 15 min in a 119 120 temperature-controlled water bath, with vortexing at regular intervals. The supernatant was 121 collected after centrifugation (5 min, 10,000 rpm), and the residues were extracted once again by using 2 mL of 75% methanol, centrifuged. One milliliter of the combined supernatant was applied 122 123 to a 1 mL DEAE Sephadex A-25 column (acetic acid activated) and rinsed with 2 mL of 0.02 mol/L sodium acetate. After addition of 200 µL of arylsulfatase solution and incubation for 16 h at 35 °C, 124 125 the desulphoglucoraphanin was eluted with 4 mL of Milli-Q water and filtered through a 0.45 µm membrane filter. Separation and detection were performed on an Agilent 1200 HPLC system 126 (AgilentTechnologies Co. Ltd., Palo Alto, CA, USA) equipped with a G1314B UV detector and a 127 128 G1311A quat pump, using a Eclipse XDB-C18 column (5 μ mparticle size, 4.6 \times 150 mm; 129 AgilentTechnologies Co. Ltd.) at 226 nm. Determination was conducted at a flow-rate of 1.0 130 mL/min in a linear gradient starting with 0% acetonitrile for 1 min, reaching 20% acetonitrile at 21 131 min, 0% acetonitrile at 26 min. Sinigrin (2-propenyl glucosinolate) (Sigma, St. Louis, MO, USA) 132 was added to each sample as an internal standard before the first extraction. The total glucosinolate and glucoraphanin content was calculated and expressed as µmol/g fresh weight of broccoli sprouts. 133

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2.10 Sulforaphane formation determination

Sulforaphane formation was measured by the method of Gu et al.¹⁶ with some modifications.
Sprouts (0.2 g) were homogenized with 4.0 mL of distilled water. The mixture was hydrolyzed at
37 °C for 3 h in a shaking water bath. After hydrolyzing, it was extracted three times with 10 mL of
ethyl acetate. The ethyl acetate fraction was dried at 37 °C under vacuum on a rotary evaporator.
After dissolving in 2.0 mL of 10% acetonitrile, filtered through a 0.45 µm membrane filter before

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injected into HPLC. Agilent 1200 HPLC system (AgilentTechnologies Co. Ltd., Palo Alto, CA,

141	USA) equipped with a Eclipse XDB-C18 column (5 $\mu mparticle$ size, 4.6 \times 150 mm; Agilent
142	Technologies Co. Ltd.) was used.
143	2.11 Myrosinase activity determination
144	Myrosinase activity determination was conducted as described previously by Guo et al. ¹⁵ The
145	protein content of the supernatant was determined according to Bradford (1976) using bovine serum
146	albumin as the standard. One myrosinase activity unit corresponded to 1.0 nmol glucose formed per
147	minute. The specific activity was expressed as units per milligram of protein.
148	2.12 Gene expression assay (QRT-PCR, Quantitative Real-Time PCR)
149	Total RNA from broccoli sprouts was isolated using a E.Z.N.A.™ Plant RNA Kit (OMEGA,
150	R6827-01). Approximately 2 μ g total RNA were used as a template for the first strand cDNA
151	synthesis, which was performed with a RT-PCR Kit (TaKaRa: DR027S). First-strand cDNA was
152	synthesized as described previously. ¹⁷ The PCR amplification was performed using TaKaRa
153	Ex-TaqTM polymerase for target genes and Actin. For QRT-PCR analysis, the sequence-specific
154	primers used were shown in Table 1. Triplicate quantitative assays were performed on each cDNA
155	with the SYBR [®] Premix Ex Taq TM (TAKARA: RR420A) with the ABI 7500 sequence detection
156	system according to the manufacturer's protocol (Applied Biosystems, Foster City, CA, USA). The
157	PCR cycling conditions used were as follows: 1 cycle of 95 °C for 30 s, followed by 40 cycles of
158	95 °C for 3 s, 60 °C for 30 s.

159 Table 1 The primers used in the present study

Gene	Primer name	Primer sequences $(5' \rightarrow 3')$
EMO	Sense	TCGTCTTGGTGTCGTCGGTC
F MO _{GS-OX}	Ant-sense	ATCCTCATAAGCCTCCGCAAT
AOP2	Sense	GAGTAACGGAAAGAAAGAAGACAAGG

	Ant-sense	ATAAGCGTGAAGAGTAGAACGAGGT	
MVD	Sense	AAGGTCATCAGGGAGAAGGGTG	
MIK	Ant-sense	TGTTTGGCAGGGTTCTTAGTGG	
ECD	Sense	ACATTTGGGACCAGGGACG	
LSF	Ant-sense	TTTCCATACACGGTGGCAGTC	
Actin	Sense	CTGTTCCAATCTACGAGGGTTTCT	
Aciin	Ant-sense	GCTCGGCTGTGGTGGTGAA	

160 **2.13 Statistical analysis**

Experimental data were expressed as the mean ± standard deviation (SD) with three replications (n = 3) and subjected to statistical analysis with SPSS 18.0 (SPSS Inc., Chicago, IL). Means were performed by Duncan's multiple-range tests. Differences at p < 0.05 were considered to be significant. **3 Results and analysis**

166 **3.1 Growth profiles of broccoli sprouts**



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Fig. 1. The growth profiles of broccoli sprouts grown under different treatments. Values are the means of triplicate analyses. Error bars show the standard deviation. Those with different lower case letters are significantly different at p < 0.05. After 4-day germination, the sprouts were collected and washed, and measured the sprouts length using a vernier calipers, directly.

The growth profiles of 4 day-old broccoli sprouts in photo and sprout length were presented in Fig. 1. ZnSO₄ application significantly (p < 0.05) inhibited sprouts growth with the shortest length of 17.3 mm, whereas that of the water-treated was 38.9 mm. After spraying supplementary Ca²⁺, the ZnSO₄-induced stress was significantly mitigated, where the sprouts elongation was accelerated. Interestingly, the application of sole Ca²⁺ also increased the sprout length by 32.4% compared with the control.



179 **3.2 Fresh weight and MDA content of broccoli sprouts**

Fig. 2. The fresh weight (A) and MDA content (B) of broccoli sprouts grown under different treatments. Values are the means of triplicate analyses. Error bars show the standard deviation. Those with different lower case letters are significantly different at p < 0.05. After 4-day germination, the sprouts were collected, washed and sipped up. The fresh weight and MDA level were measured immediately.

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The fresh weight of sprouts grown under $ZnSO_4$ was the least, indicating the growth of sprouts was markedly inhibited (p < 0.05) (Fig. 2A). Spraying CaCl₂ alone favored sprout elongation, whose fresh weight increased by 27.9% compared with the control. CaCl₂ addition alleviated the ZnSO₄ stress, generating an increment of 47.0% in fresh weight in comparison with the 191 ZnSO₄-treated sprouts.

MDA content exhibited an opposite trend of fresh weight (Fig. 2B). ZnSO₄ application led to the highest accumulation of MDA in sprouts compared with the control. After adding CaCl₂ based on ZnSO₄ treatment, the sprouts' MDA content decreased by 26.9%. Water (Control) and CaCl₂ alone did not cause stress condition, where the MDA content of sprouts kept at a very low level. **3.3 Ascorbic acid, total phenolics content and antioxidant capacity of broccoli sprouts**



Fig. 3. Ascorbic acid content (A), total phenolics content (B) and antioxidant capacity (C) of broccoli sprouts grown under different treatments. Values are the means of triplicate analyses. Error

bars show the standard deviation. Those with different lower case letters are significantly different at p < 0.05. After 4-day germination, the sprouts were collected, washed and sipped up. The ascorbic acid content, total phenolics content and antioxidant capacity of fresh samples were measured immediately.

The sprouts treated with $ZnSO_4$ had the highest contents of ascorbic acid and total phenolics, followed by that of $ZnSO_4$ plus $CaCl_2$, water and $CaCl_2$ alone (Fig. 3A and 3B). $CaCl_2$ treatment alone had no significant effect (p > 0.05) on ascorbic acid content while decreased that of total phenolics as compared with the control. The contents of ascorbic acid and total phenolics of broccoli sprouts grown under $ZnSO_4$ were 3.17- and 3. 91-fold of the control, respectively. Whereas, additional $CaCl_2$ treatment decreased their contents by 36.47% and 44.23% compared with $ZnSO_4$ treatment, respectively.

As shown in Fig. 3C, the antioxidant capacity of $ZnSO_4$ treated sprouts was the highest, which was 2.55-fold of the control. Adding $CaCl_2$ on the basis of $ZnSO_4$ treatment decreased the antioxidant capacity. Compared with the control, spraying $CaCl_2$ alone also decreased the sprouts antioxidant capacity by 19.1%.

217 **3.4** Glucosinolate content, sulforaphane formation and myrosinase activity of broccoli sprouts

Compared with the control (Water), $CaCl_2$, $ZnSO_4$ and $ZnSO_4$ plus $CaCl_2$ treatments significantly increased total glucosinolate content in broccoli sprouts (Fig. 4A). $ZnSO_4$ led the highest content of total glucosinolate, which was 2.13-fold of the control. However, $ZnSO_4$ plus $CaCl_2$ treatment decreased content of total glucosinolate compared with $ZnSO_4$ treatment. In both hypocotyl and cotyledon, total glucosinolate content in $ZnSO_4$ plus $CaCl_2$ treated sprouts increased

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significantly by 1.01- and 0.43- fold, respectively, compared with the control (Fig. 4B). Similar to total glucosinolate content, $CaCl_2$, $ZnSO_4$ and $ZnSO_4$ plus $CaCl_2$ treatments all increased glucoraphanin content, but the treatment of $ZnSO_4$ plus $CaCl_2$ best favored the accumulation of glucoraphanin (Fig. 4C). In both hypocotyl and cotyledon, glucoraphanin content in $ZnSO_4$ plus $CaCl_2$ treated sprouts increased significantly by 1.59- and 0.55- fold, respectively, compared with the control (Fig. 4D).



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Fig. 4. Total glucosinolate (A, B) and glucoraphanin (C, D) content, sulforaphane formation (E, F) and myrosinase activity (G, H) of broccoli sprouts under different treatments. Values are the means of triplicate analyses. Error bars show the standard deviation. Those with different lower case letters

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are significantly different at p < 0.05. * Represents significant differences at p < 0.05. After 4-day germination, the sprouts were collected, washed and sipped up. Then frozen in liquid nitrogen. Total glucosinolate and glucoraphanin content, sulforaphane formation and myrosinase activity were measured within 2 days.

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Compared with the control, treatments of $CaCl_2$ alone, $ZnSO_4$ and $ZnSO_4$ plus $CaCl_2$ were all beneficial for the formation of sulforaphane (Fig. 4E). They increased sulforaphane formation by 18.9%, 43.5% and 91.2%, respectively. However, $ZnSO_4$ plus $CaCl_2$ decreased sulforaphane formation in hypocotyl by 33.62% and increased that in cotyledon significantly compared with the control (Fig. 4F).

Myrosinase activity of broccoli sprouts was enhanced by 3.40-fold under ZnSO₄ treatment in comparison with the control (Fig. 4G). Supplementary CaCl₂ further stimulated myrosinase activity by 15.3% compared with ZnSO₄ treatment. Spraying CaCl₂ alone increased the activity by 75.9% compared with the control. Similar to glucoraphanin content, myrosinase activity increased under ZnSO₄ plus CaCl₂ treatment significantly and by 0.78 and 2.57- fold in hypocotyl and cotyledon respectively compared with the control (Fig. 4H).

3.5 Genes expression related to sulforaphane formation

ZnSO₄ and ZnSO₄ plus CaCl₂ treatments induced FMO_{GS-OX} expressions which were 2.48- and 2.67-fold of the control, respectively (Fig. 5A). However, they decreased AOP2 expression (Fig. 5B), significantly. CaCl₂ addition significantly enhanced the expression of ESP (Fig. 5C). ZnSO₄ and ZnSO₄ plus CaCl₂ treatments all induced *MYR* expression which was 1.97-, 4.12- and 9.45-fold of the control, respectively (Fig. 5D).



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Fig. 5. The relative expression of FMO_{GS-OX} (A), AOP2 (B), ESP (C) and MYR (D) of broccoli sprouts under different treatments. Each datum is the mean and standard deviation of three replicates per treatment. Values not sharing the same letter are significantly different at p < 0.05.

After 4-day germination, the sprouts were collected, washed and sipped up. Total RNA wasextracted immediately using a kit.

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3.6 Changes of glucoraphanin content, sulforaphane formation and myrosinase activity of broccoli sprouts during germination

Germination brought a significant (p < 0.05) decrease in glucoraphanin content both in control and the ZnSO₄ plus CaCl₂ treatment sprouts (Fig. 6A). Glucoraphanin content of sprouts grown under water and ZnSO₄ plus CaCl₂ decreased by 96.5% and 90.4% within 4 days of germination, respectively. However, after germinating for 1 day, glucoraphanin content of ZnSO₄ plus CaCl₂ treated sprouts was significantly (p < 0.05) higher than that of the control at the same time point. Similar to glucoraphanin content, sulforaphane formation of the control sprouts decreased

270 steadily with time elongation (Fig. 6B). Whereas, ZnSO₄ plus CaCl₂ treatment led to an gradual

- increase in sulforaphane formation after 2 days of germination. No significant (p > 0.05) difference
- was observed between the control and $ZnSO_4$ plus $CaCl_2$ treatment at the first 2 days of germination.
- However, sulforaphane formation of 3 day- and 4 day-old sprouts were significantly (p < 0.05)
- higher than that of the control.



Fig. 6. Glucoraphanin content (A), sulforaphane formation (B) and myrosinase activity (C) of broccoli seeds and sprouts during germination period under control and ZnSO₄ plus CaCl₂ treatments. Values are the means of triplicate analyses. Error bars show the standard deviation. * Represents significant differences at p < 0.05. During germination, sprouts were sampling every day

and frozen in liquid nitrogen. Glucoraphanin content, sulforaphane formation and myrosinase
activity were measured within 2 days.

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Myrosinase activity of sprouts sprayed with water increased to 2^{nd} day of germination and decreased afterwards (Fig. 6C). In contrast, the activity increased continuously to 3^{rd} day of germination and kept stable when sprayed with ZnSO₄ plus CaCl₂. No significant (p > 0.05) change occurred at the first day of germination.

287 4 Discussion

The addition of CaCl₂ to seed germination medium under salinity stress have been reported to 288 enhance germination rate and seedling growth in wimmera ryegrass,¹⁸ cotton,¹⁹ tomato and 289 cucumber.²⁰ However, the concentration of CaCl₂ should be clearly defined since lower or higher 290 291 concentration would have significant different effects, which depends upon the salinity stress source and plant species or cultivars. In wimmera ryegrass, the 5.0 mmol/L Ca²⁺ showed the best effect on 292 NaCl stress, while 10.0 mmol/L Ca²⁺ was optimal on MgCl₂ stress.¹⁸ Besides, the author found that 293 increasing Ca^{2+} concentration to 10.0 mmol/L under NaCl stress decreased the radicle length 294 compared with that of 5.0 mmol/L. In this study, it was also detected that a concentration higher 295 296 than 11 mmol/L would decrease broccoli sprout length (Fig. S2, Supplementary data). Apart from directly adding Ca²⁺ to sprout growth medium, several studies chose to pre-treat seeds through 297 soaking in Ca²⁺ solutions.^{21, 22} They both found the salinity tolerance of tested seeds was greatly 298 enhanced. These results suggested that the mitigation effect of Ca^{2+} might also be achieved via 299 soaking seeds prior to germination. 300

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The salinity environment is usually created by NaCl treatment during sprouts growth.^{23, 24} In

broccoli sprouts cultivation, in order to enhance glucoraphanin biosynthesis, sulphur-containing 302 chemicals are commonly supplied.⁷ However, since the sprouts are very salt sensitive.²⁵ it is 303 difficult to find the optimal sulphur concentration without influencing sprouts growth. It has been 304 reported that unsuitable concentrations of sodium thiosulphate⁸ and K₂SO₄²⁵ inhibited sprouts 305 growth and thus decreased the positive effects of sulphur fertilization. Here, the stress was caused 306 307 by spraying ZnSO₄, which was originally expected to be a sulphur fertilizer. For one probable reason, the stress of $ZnSO_4$ was due to the excessive level of Zn^{2+} because Zn is a micronutrient for 308 sprout growth. The excessive Zn^{2+} was not utilized promptly and thus created an osmotic stress 309 condition where the water absorption system was interrupted. Another one might be that $ZnSO_4$ 310 presents a similar negative effect as NaCl does. Hence, towards different broccoli cultivars and 311 312 growth medium, the optimal sulphur type and concentration should be pre-selected depending on sprouts sulphur uptake abilities and partitioning of sulphur into glucoraphanin.²⁶ 313

314 Ascorbic acid content and the activities of its regenerating enzymes in plants are highly related to environmental stresses such as light, temperature, salt, and drought.²⁷ Nishikawa et al.²⁸ 315 investigated the ascorbic acid metabolism in postharvest broccoli florets, and found that 316 317 ascorbate-regenerating system could be activated to protect plant organs from environmental 318 stresses. In this study, the ascorbic acid content in broccoli sprouts was accelerated by ZnSO₄ 319 application, which could also be attributed to the activation of the regenerating enzymes under ZnSO₄ stress.²⁷ Phenolic compounds are secondary metabolites produced when plants are facing 320 environmental stresses.²⁹ In the present study, ZnSO₄ application significantly promoted the 321 biosynthesis of phenolic compounds as a result of the possible osmotic stress (Fig. 3B). However, 322 when the stress situation was likely mitigated by exogenous CaCl₂, the total phenolics content of 323

324 broccoli sprouts was decreased as compared with that of ZnSO₄.

Ascorbic acid and phenolic compounds both have high antioxidant activities in plants,^{30, 31} 325 hence, the antioxidant capacity of plant tissues is closely associated with their contents.³² 326 327 Significant positive correlation between ascorbic acid, phenolic compounds and antioxidant capacity was observed in radish³³ and broccoli sprouts.⁵ In the present study, the antioxidant 328 329 capacity of broccoli sprouts of the tested four treatments had a same changing trend with the contents of ascorbic acid and phenolic compounds (Fig. 3). The highest antioxidant capacity was 330 observed in sprouts having the highest ascorbic acid and phenolic compounds contents. Exogenous 331 Ca^{2+} decreased the antioxidant capacity since the contents of ascorbic acid and phenolic compounds 332 were decreased.⁵ Barillari *et al.*³⁴ reported that 4-methylthio-3-butenyl glucosinolate isolated from 333 Raphanus sativus sprouts was related to its total antioxidant capacity. Guo et al.⁵ also suggested 334 335 that glucosinolates in broccoli sprouts had some antioxidant capacity. However, in this study, no obvious relations were observed between total antioxidant and total glucosinolates or glucoraphanin 336 content (p > 0.05). The discrepancy could be attributed to different cultivars of broccoli in which 337 338 the content of antioxidant components is different.

Falk *et al.*³⁵ highlighted that sulphur fertilization could lead to an increase in glucosinolate content ranging from 0.25- to more than 50-fold, which relies on sulphur types and treating styles. Here, as a sulphur-containing chemical, $ZnSO_4$ increased total glucosinolate content, but $ZnSO_4$ plus CaCl₂ treatment decreased it content (Fig. 4A). Interestingly, the content of glucoraphanin was further enhanced by CaCl₂ addition (Fig. 4C). Applying CaCl₂ alone to sprouts increased glucoraphanin content compared with that of the control. Besides, it was noted that glucoraphanin content of sprouts under sole CaCl₂ treatment was higher than that of sole ZnSO₄ treatment. In

addition, compared with ZnSO₄ treatment, ZnSO₄ plus CaCl₂ treatment increased genes expression
related to glucoraphanin biosynthesis (Fig. 5A), but decreased *AOP2* expression which is related to
glucoraphanin degradation (Fig. 5B). These indicated that CaCl₂ could induce glucoraphanin
biosynthesis in broccoli sprouts.

In both ZnSO₄ plus CaCl₂ treated and control sprouts, glucoraphanin content decreased 350 steadily with germination time (Fig. 6A), which agreed with previous studies.^{2, 36} The decrease in 351 352 glucoraphanin content might result from 'dilution effects' of water absorption during growth.³⁷ 353 Another possibility could be that glucoraphanin was degraded as sulphur supply to support sprouts tissue expansion and formation of other sulphur-containing compounds.⁷ Under ZnSO₄ plus CaCl₂ 354 treatment, the decrease of glucoraphanin was delayed to some extent, which was probably a result 355 of sulphur supply. Pérez-Balibrea *et al.*⁹ had previously reported that glucoraphanin content of 356 357 broccoli sprouts treated with K₂SO₄ kept stable when grew from 6 to 12 days.

358 Sulforaphane is hydrolyzed from glucoraphanin by myrosinase, whose formation largely depends on both glucoraphanin content and myrosinase activity.⁷ In this study, sulforaphane 359 360 formation of broccoli sprouts treated with ZnSO₄ plus CaCl₂ was the highest (Fig. 4D). This might be due to a higher glucoraphanin content (Fig. 4B), myrosinase activity (Fig. 4E) and its higher 361 362 expression (Fig. 5D). Although the glucoraphanin content of sprouts under sole CaCl₂ treatment was 363 higher than that of sole ZnSO₄, its sulforaphane formation was less because of a lower myrosinase activity (Fig. 4E) and its lower expression (Fig. 5D) as well as the higher expression of ESP (Fig. 364 5C). Liang et al.³⁸ reported that Zn^{2+} could enhance myrosinase activity and thus increase 365 sulforaphane formation. The finding in the present study was similar to theirs. It was clearly noticed 366 that glucoraphanin content of broccoli sprouts under ZnSO₄ plus CaCl₂ and the control treatments 367

368	both decreased steadily during germination (Fig. 6A). However, the sulforaphane formation of
369	sprouts treated with ZnSO ₄ plus CaCl ₂ did not decrease steadily as expected (Fig. 6B). In contrast,
370	its content increased after germinating for 2 days. This could be due to the fact that myrosinase
371	activity increased to a high level after 1 day of germination. These results suggested that the
372	glucoraphanin-myrosinase system is complicated and needs further investigation.
373	5 Conclusion
374	CaCl ₂ application effectively mitigated stressful condition caused by ZnSO ₄ . Consequently, the
375	antioxidant compounds such as ascorbic acid and total phenolics content decreased. However,
376	sulforaphane formation increased since the higher glucoraphanin content, myrosinase activity and
377	the related genes expression induced after CaCl ₂ treatment.
378	Conflicts of interest
379	The authors declare no conflict of interest.
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