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In this work, a multifunctional drug delivery system was developed for potential application in NIR fluorescence imaging and targeting PDT.



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The multifunctional magnetic nanoparticles for simultaneous cancer near-infrared imaging and targeting photodynamic therapy †

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Cancer theranostics, the ability to simultaneous cancer diagnose and treatment, has become one of the major driving forces in nanobiotechnology. In the present work, a multifunctional system, methylene blue-incorporated folate-functionalized Fe_3O_4 /mesoporous silica core/shell magnetic nanoparticles (MNPs), for simultaneous near-infrared (NIR) fluorescence imaging and targeting photodynamic therapy (PDT) has been developed. The core Fe_3O_4 MNPs offers the function of magnetically guided drug delivery, the mesoporous silica shell acts as an efficient drug loaded carrier, the photosensitizer methylene blue (MB) exhibits excellent NIR fluorescence imaging and PDT efficiency, and the folic acid (FA) can effectively enhance the delivery of MB to the targeting cancer cells which overexpress the folate receptor. The results indicated that the multifunctional system could effectively be used in NIR fluorescence imaging. Moreover, it exhibited a synergistic effect of magnetic targeted PDT of cancer under NIR laser irradiation. Thus, the multifunctional system is promising for simultaneous cancer diagnosis and therapy.

1 Introduction

Cancer is a devastating disease with incidence increasing at an alarming rate and survival not improved substantially during the past several decades. Although enormous efforts have been made in early detection and comprehensive treatment for this disease, little or no survival improvement was obtained, which needs further development of novel strategies.^{1,2} According to the report from National Cancer Institute (NCI) in USA, nanobiotechnology, which not only carries multiple diagnostic/therapeutic payloads in the same package, but also facilitates the targeted delivery into specific sites and across complex biological barriers,3 has tremendous potential for cancer prevention, diagnosis, imaging, and treatment.⁴ The multifunctional integrated system combines different properties such as tumor targeting, imaging, and selective therapy in an all-in-one system, which will provide more useful multimodal approaches in the battle against cancer.⁵⁻⁹

Photodynamic therapy (PDT) is now well established as a technique for cancer treatment.¹⁰⁻¹³ In contrast to other conventional medical treatments, PDT doesn't need to release the used drugs and it is based on the concept that photosensitizers (PSs) are able to generate reactive oxygen species (ROS) upon irradiation, such as singlet oxygen ($^{1}O_{2}$) or free radicals, and can irreversibly damage the pathological cells without damaging adjacent healthy ones.¹⁴⁻¹⁶ Unfortunately, the

phototoxicity, hydrophobicity and the low selectivity of the PS agents limit the current applications of PDT in cancer therapy.¹⁷ Therefore, the development of new biocompatible delivery vehicle with stable aqueous dispersion, site-specific and timecontrolled delivery abilities is still urgently needed. Among the various delivery vehicles, mesoporous silica nanoparticles (MSNs) hold the promise to be a highly efficient PDT drug delivery platform owing to their attractive features such as uniform pore size, large surface area and high accessible pore ease of chemical modification, volume. excellent biocompatibility and avid uptake by cells.18-22 The porous structure of MSNs not only permits the accommodation of a large quantity of PSs, but also helps to enhance the permeability of oxygen and generate ${}^{1}O_{2}$, which is essential for PDT. Furthermore, their surfaces can be modified with special targeting moieties such as antibodies, folate and aptamers for site-specific behavior. However, to our knowledge, the application of multifunctional MSNs as photosensitizing vehicles that provides both MRI and fluorescence imaging diagnosis and photodynamic therapy has not been satisfied explored. There are very few reports available on the applications of MSNs as PSs vehicles.

Moreover, the accurate localization of PS-containing nanoparticles in cells or target tissues is very important for effective PDT. It will offer a powerful guidance for sitedirected irradiation of target diseased tissues without causing ARTICLE

damage to the healthy tissues. Recently, optical imaging probes have been incorporated into MSNs along with PSs to offer dual capability of imaging and therapy.23-25 Optical imaging can provide the highest sensitivity and obtain detailed information at subcellular levels, which allow accurate targeting and simultaneous phototherapy treatment. In fact, some PSs can emit fluorescence and generate ¹O₂ simultaneously under the irradiation.²⁶ Methylene blue (MB) is a hydrophilic phenothiazinium photosensitizer with promising applications in the PDT for its high quantum yield of ${}^{1}O_{2}$ generation ($\Phi \sim 0.5$) in the excitation of the therapeutic window (600-900 nm), and low dark toxicity.^{27,28} In addition, MB is also the most inexpensive of the commercially available near-infrared (NIR) fluorescent dyes, and has been widely used for bioanalysis.²⁹ Therefore, taking advantage of the intrinsic fluorescence of photosensitizer to develop single photosensitizer-encapsulated nanoparticles for simultaneous in vivo imaging and PDT is significant.

Enhancing tumor accumulation of therapeutic agents by physical forces such as an external magnetic field (MF) has emerged as a new tumor-targeting strategy. During this process, magnetic nanoparticles (MNPs) carrying therapeutics circulating in the bloodstream would be attracted by the MF applied on the tumor, resulting in greatly enhanced enrichment of therapeutic agents in targeted tumor region to improve the cancer treatment efficacy.³⁰ Moreover, MNPs as a magnetic resonance imaging (MRI) contrast agent exhibit a unique magnetic resonance (MR) contrast enhancement effect that enables noninvasive MRI of cell trafficking, gene expression, and cancer.^{31,32}

Herein, in this paper, we present a core–shell structured nanomaterial, namely, Fe_3O_4 @mSiO₂(MB)-FA, which is multifunctional in the fields of NIR fluorescence imaging and targeting PDT. The Fe_3O_4 nanoparticles were prepared *via* coprecipitation method, and mesoporous silica was coated as shell layer, in which photosensitizer was loaded. Then a layer of poly(ethylene glycol) (PEG) shell and folate receptors were synthesized. Systematic experiments *in vitro* and *in vivo* were designed to carefully evaluate the physical and chemical properties, cytotoxicity, cellular uptake, NIR light-induced cells killing, *in vivo* imaging, as well as targeting PDT of the multifunctional system (scheme 1).



Scheme 1. The cancer cell targeting PDT mechanism of the multifunctional system.

2 Experimental

2.1 Materials

2.1.1 Reagents and materials. Ferric chloride hexahydrate (FeCl₃•6H₂O), ferrous chloride tetrahydrate (FeCl₂•4H₂O), sodium oleate (C₁₈H₃₃ONa), ammonium hydroxide (25%, aqueous solution), tetraethoxysilane (TEOS), ammonium nitrate (NH₄NO₃), cetyltrimethylammonium bromide (CTAB), 3-amino-propyltriethoxysilane (APTES), Methylene blue (MB), acid (FA), 1-ethyl-3[3-dimethylaminopropyl] folic carbodiimide hydrochloride (EDC), dicyclohexylcarbodiimide (DCC), dimethyl sulfoxide (DMSO) and Nhydroxysuccinimide (NHS) were obtained from Sinopharm Chemical Reagent Co. Ltd., and used without further purification. 2-methoxy (polyethyleneoxy) propyltrimethoxysilane (PEG-silane) was from Gelest (USA). 3-(4, 5-dimethylthiazol-2-yl)-2, 5-diphenyltetrazolium bromide (MTT), 4, 6-diamidino-2-phenylindole (DAPI) and DPBF were purchased from Sigma-Aldrich (USA). Dulbecco's modified Eagle's medium (DMEM) cell culture medium, penicillin, streptomycin, trypsin, fetal bovine serum (FBS), and heparin sodium were bought from Gibco Invitrogen (USA). All other chemicals and reagents were of analytical grade and used as received.

2.1.2 Cell lines and animal. Human ovarian cancer cells (SK-OV-3 cells), human cervical cancer cells (HeLa cells), mouse sarcoma cells (S-180 cells) and mouse fibroblast cells (NIH 3T3 cells) were purchased from the Shanghai Institutes for Biological Sciences, Chinese Academy of Sciences (Shanghai, China). Kunming mice of clean grade (female, 4-6 weeks old) and BALB/c mice (female, 4-6 weeks old, 16-18 g of body weight) were purchased from Shanghai SLAC Laboratory Animal Center (Shanghai, China), and were used in accordance with approved institutional protocols established by the Shanghai Department of Experimental Animals Management.

2.2 Methods

2.2.1 Synthesis of Fe_3O_4@mSiO_2(MB)-FA nanoparticles. Fe₃O₄@mSiO₂(MB)-FA nanoparticles were synthesized following the method in our previous report.³³ Firstly, highly biocompatible monodisperse superparamagnetic Fe_3O_4 @mSiO₂ core/shell nanoparticles with mesoporous silica shells were synthesized. Then these particles were coated with the covalently bonded biocompatible polymer PEG and modified with the cancer targeting ligand FA. Finally, the water-soluble photosensitizer MB was loaded into the mesoporous silica shell.

2.2.2 Characterization of $Fe_3O_4@mSiO_2(MB)$ -FA nanoparticles. The particles were characterized by means of powder X-ray diffraction (XRD), transmission electron microscopy (TEM), dynamic light scattering (DLS), Fourier transformed infrared spectroscopy (FT-IR), Brunauer-Emmett-Teller (BET), vibrating sample magnetometer (VSM), UV-Vis and fluorescence spectroscopy. High-resolution TEM data were

collected on a JEOL model JEM 2100 electron microscope operating at an accelerating voltage of 200 kV. DLS data were obtained on an electrophoretic light scattering spectrophotometer (Beckman Coulter Delsa nano C, USA). XRD data were collected on a Rigaku corporation D/MAX 2550 VB/PC Multi-Purpose X-ray diffractometer with Cu Ka radiation (λ = 0.1542 nm). The specific surface area was calculated by the BET (ASAP 2020-M) method. The pore size distribution was obtained from the Barret-Joner-Halenda (BJH) method. The FT-IR spectra of the nanoparticles were obtained on a Nicolet 6700 spectrometer. Magnetic properties were recorded using a VSM (Lake Shore, USA). UV-Vis absorption spectra were measured on an UV-Vis Spectrophotometer (Evolution 220, Japan). Fluorescence spectra of liquid state were recorded on a Lumina Fluorescence spectrometer. (Thermo scientific, USA). The cellular images were acquired with a confocal laser scanning microscope (CLSM, Nikon AIR).

2.2.3 Detection of singlet oxygen. Commonly, there are two methods for the detection of ¹O₂, one the method by using luminescence emission spectra at 1270 nm,34-36 the second is based on an indirect method using a chemical ¹O₂ probe.^{36,37} In this study, DPBF is used as a probe^{34,37} to detect the $^{1}O_{2}$ quantum yield. DPBF reacts irreversibly with ¹O₂ that causes a decrease in the intensity of the DPBF absorption band at 400 nm. In a typical experiment, 15 mL of DPBF in acetonitrile (5.5 mM) was mixed with 2 mL, 1.5 mg/mL of Fe₃O₄@mSiO₂(MB)-FA nanoparticles in acetonitrile. The experiment mixed DPBF with free MB dispersed in acetonitrile used as the standard. The solutions were irradiated with a 650 nm laser source at 5 mW/cm², and their absorbencies at 400 nm were recorded at every 10 seconds, using a UV-Vis Spectrophotometer.

2.2.4 Cytotoxicity assessment. The in vitro cytotoxicity was measured by using the MTT assay in SK-OV-3 cells. Cells $(1 \times 10^5 \text{ well}^{-1})$ were inoculated into a 96-well cell-culture plate and then incubated at 37 °C in a 5% CO2-humidified incubator for 24 h. 200 µL of Fe₃O₄@mSiO₂(MB)-FA nanoparticles with different concentrations (10-200 µg/mL), DMEM were added to the wells, separately. After incubation for 24 h and 48 h at 37 ^oC under 5% CO₂, the supernatant was removed. Subsequently, MTT (20 µL, 5 mg/mL) solved in DMEM (200 µL) were added and the plates were incubated at 37 °C for another 4 h. Then supernatant was removed before DMSO was added to each well to dissolve the formazan. The absorbance at 492 nm and 630 nm was detected with spectrophotometric microplate reader (THERMO Multiskan MK3 spectrometer). Each data point was collected by averaging that of three wells, and the untreated cells were used as controls. The following equation was used to calculate the inhibition of cell viability.

Cell viability(%) =
$$\left(\frac{\text{mean absorption of value treatment group}}{\text{mean absorption value of control}}\right) \times 100\%$$

2.2.5 Cellular uptake study. The cellular uptake and distribution of $Fe_3O_4@mSiO_2(MB)$ -FA nanoparticles were performed using flow cytometry and CLSM. For flow

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cytometry, HeLa cells (1×10^6) that overexpress folate receptors were seeded in 6-well culture plates and cultured for 24 h. Following this, the cells were treated with Fe₃O₄@mSiO₂(MB)-FA nanoparticles ($200 \mu g/mL$) for 2 h and 4 h. After the preset time intervals, the culture medium was discarded, and cells were washed three times with PBS and harvested with trypsinization. The cell pellets were resuspended in PBS and measured for the fluorescence intensity (excitation: 650 nm; emission: 690 nm) on a BD FACSAria flow cytometer (Beckton Dickinson, USA), and Cell Quest software was used to analyze the data.

For CLSM studies, HeLa cells were propagated in DMEM containing FBS (10%) and penicillin/streptomycin (1%). Then the cells were digested and resuspended in the DMEM medium (without FARs). 1×10^4 cells were transferred into a 6-well tissue culture plates. After 24 h of incubation, the cells were carefully rinsed with PBS (pH 7.4). 2 mL DMEM medium of the Fe₃O₄@mSiO₂(MB)-FA (200 µg/mL) were added to the Petri dishes and incubated for 2 h, followed by three-time rinses with PBS. After fixation with 4% paraformaldehyde in PBS at room temperature for 30 min and then cells were treated with 1 mL 0.01% Triton X-100 (Sigma) for 10 min and the nuclei were stained with DAPI (1 µg/mL, Sigma) for 15 min. Each step of the above was washed with PBS three times. Then cover slips containing cells were mounted onto slides and were then observed under CLSM.

2.2.6 In vitro PDT effect. Two 96-well plates were divided into two groups: dark control and experimental group. SK-OV-3 cells were seeded in the 96-well plate at a density of 1×10^4 cells per well for 24 h. Then DMEM cell medium containing concentrations of Fe₃O₄@mSiO₂(MB)-FA different nanoparticles were added to the wells (200 µL per well, 0, 10, 50, 100, 150, 200 µg/mL). After incubation for 24 h, the cells were washed three times with 200 µL PBS to remove the unbound nanoparticles. Then 200 µL PBS was added and the cells were exposed to a 650 nm laser with a power density of 70 mW/ cm² for 4 min. After irradiation, cells were incubated another 24 h in a 5% CO₂, 95% air humidified incubator at 37 °C. Dark control group keeps identical to experimental group without irradiation. The cell viability was measured by a MTT assay mentioned above and expressed as a percentage of the control.

2.2.7 In vivo imaging of Fe_3O_4 @mSiO₂(MB)-FA nanoparticles in animals. The female BALB/c mice were maintained in a pathogen-free condition on arrival. To generate HeLa tumor xenografts, 1×10^7 cells suspended in 200 µL of salined were subcutaneously implanted into the right limb armpits of the mice. Tumors were allowed to grow until the average volume of the xenograft tumors was approximately 50 mm³ (about 10 days). The mice were randomly divided into three groups, three mice each group. For group I, the mice were administrated with saline; group II, Fe_3O_4 @mSiO₂(MB)-NH₂; group III, Fe_3O_4 @mSiO₂(MB)-FA. Fe_3O_4 @mSiO₂(MB)-NH₂ and Fe_3O_4 @mSiO₂(MB)-FA administrated were 5 mg MB equiv/kg through tail vein in all of the formulations. The whole body *in vivo* fluorescence imaging of the mice was performed an hour later utilizing a *In-Vivo* Imaging System IVIS Lumina II (Coliper, USA). The instrument equipped with fluorescent filter sets (excitation/emission wavelength of 650/690 nm).

2.2.8 In vivo anti-tumor properties. Xenograft tumor mouse model. All animal experiments were performed under a protocol approved by Shanghai laboratory animal center. Mice ascitic tumor models were generated by subcutaneous injection of 1×10^6 S-180 cells in 0.1 mL saline into the right armpit of female Kunming mice (18-22 g, Shanghai SLAC Laboratory Animal Center). The mice were used when the tumor volume reached 50×50 mm³.

In vivo PDT. For the in vivo antitumor experiments, the tumor-bearing mice were divided into five groups (n=5), minimizing the differences of weights and tumor sizes in each group. The mice were administered with (1) saline, (2) magnetic/650 nm laser, (3) Fe₃O₄@mSiO₂(MB)-FA/650 nm Fe₃O₄@mSiO₂(MB)-FA/magnetic, laser, (4) (5)Fe₃O₄@mSiO₂(MB)-FA/magnetic/650 nm laser, (Fe₃O₄@mSiO₂(MB)-FA dose: 5 mg/kg) in saline were intravenously injected into mice via the tail vein every 2 days, respectively, and then the tumor regions were irradiated with 650 nm laser (70 mW/cm², 10 min) at 4 h post-injection. For the purpose of in vivo magnetic targeting, a magnet was glued onto the tumor site of the mice. The mice were observed daily for clinical symptoms and the tumor sizes were measured by a caliper every 24 h.

2.2.9 Statistical analysis. Data were analyzed using the SPSS software package. Quantitative data are expressed as mean \pm SD and analyzed by use of Student's t test. P values < 0.05 were considered statistically significant.

3 Results and discussion

3.1 Synthesis and characterization of Fe₃O₄@mSiO₂(MB)-FA

In our experiments, the multifunctional Fe₃O₄@mSiO₂(MB)-FA nanoparticles were prepared by a multistep process (see Methods section for details in our previous report³⁷). The TEM image displayed in Fig. S1 (A) shows that the prepared hydrophobic Fe₃O₄ are monodispersed nanoparticles with a uniform average diameter of 5 nm. The MB-loaded Fe₃O₄@mSiO₂-FA nanoparticles are discrete and uniform with an average diameter of 70 ± 5 nm, and well-ordered mesopores were also clearly observed (Fig. 1S (B)). In addition, the DLS data (Fig. S2) shows Fe₃O₄@mSiO₂(MB)-FA has narrow size distribution with an overall hydrodynamic diameter of 89.8 nm in water without aggregation. The crystallinity of Fe₃O₄ does not change after modification of FA and load of MB (Fig. S3). Moreover, the uniform mesoporous pore size along with small particle size (<100 nm) are facilitative and favorable for drug delivery applications (Fig. S4). The representative hysteresis loop of Fe₃O₄@mSiO₂(MB)-FA at ambient temperature are shown in Fig. S5, and the saturation magnetization of the Fe₃O₄@mSiO₂(MB)-FA is ~8.46 emu/g, which suggested the superparamagnetism of the as-prepared nanoparticles.

3.2 Singlet oxygen detection

In this study, the ¹O₂-generation capability of the Fe₃O₄@mSiO₂(MB)-FA nanoparticles was assessed using DPBF, a ¹O₂ chemical probe, in acetonitrile under 650 nm laser irradiation (5 mw/cm²). DPBF reacts irreversibly with ${}^{1}O_{2}$ and the reaction can be followed by recording the decrease in the intensity of the DPBF absorption at around 400 nm. The changes in the absorption spectra of DPBF in the presence of Fe₃O₄@mSiO₂(MB)-FA nanoparticles after different irradiation times are shown in Fig. 5. Control tests were carried out to confirm that the decrease in the absorption of DPBF was induced by ${}^{1}O_{2}$ (Fig. 1, inset). In the presence of Fe₃O₄@mSiO₂(MB)-FA nanoparticles, the DPBF absorption at 400 nm dramatically decreases under NIR-laser irradiation (Fig. 1, curve a in the inset), thereby suggesting that these nanoparticles are highly efficient in the generation of reactive ¹O₂. In contrast, there are fewer decreases in DPBF absorbance for free MB (Fig. 1, curve b in the inset). The effective ${}^{1}O_{2}$ generating capability of our nanoparticles under NIR light makes it possible for them to be applied in NIR-induced PDT.



Fig. 1. Absorption spectra of $Fe_3O_4@mSiO_2(MB)$ -FA in the presence of DPBF after different times of irradiation with a 650 nm laser source at 5 mw/cm². Inset: Decay curves of absorption of DPBF as function of time of irradiation. DPBF with dispersion of $Fe_3O_4@mSiO_2(MB)$ -FA (a) and DPBF with MB free (b) in acetonitrile.

3.3 Cellular uptake

The ability to target nanoparticles to specific organelles or receptors is one of the most important factors for their prospective application in bioimaging and drug delivery. Various types of targeting agents, such as antibodies, aptamers and FA, have been developed for the specific identification antigens or receptors on targeting cancer cells. In this study, FA was modified onto the Fe₃O₄@mSiO₂ as the targeting component because folate receptors (FR) are overexpressed in many human cancerous cells.

Flow cytometry and CLSM were used to evaluate the effect of FA on the celluar uptake behavior of $Fe_3O_4@mSiO_2(MB)$ -FA against FR positive HeLa cells. To precisely observe the cellular distributions of the multifunctional nanoparticles, the double fluorescence–labeling experiments and visualized red fluorescence from MB and blue fluorescence from DAPI

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labeling the nucleus were performed. As described in Fig. S6, intense red fluorescence of MB loaded $Fe_3O_4@mSiO_2$ -FA is observed in the cytoplasm of FA-positive HeLa and SK-OV-3 cells in comparison with NIH 3T3 cells without folate receptors, whose fluorescence could be negligible. The results confirm that objective of increasing specificity and sensitivity of $Fe_3O_4@mSiO_2(MB)$ -FA image by labeling cancer cells with over-expression of folate receptors on the surface has been achieved. And with the increase of the incubation time, the red fluorescence in both cytoplasm and nuclei increased (Fig. 2).

The cellular uptake of Fe₃O₄@mSiO₂(MB)-FA into HeLa cells was futher quantitatively analyzed with flow cytometry. Fig. 3A shows the flow cytometry histograms of MB fluorescence from HeLa cells incubated with Fe₃O₄@mSiO₂(MB)-FA with the concentration of 200 µg/mL for 2 h and 4 h, respectively. Cells without any treatment were used as a negative control to detect autofluorescence. The flow cytometry analysis clearly demonstrated that the changes in fluorescence intensity of MB were observed in the cells after 2 h incubation with Fe₃O₄@mSiO₂(MB)-FA. With the increase of the incubation time, the relative geometrical mean fluorescence intensity of Fe₃O₄@mSiO₂(MB)-FA pretreated cells obvious increases (Fig. 3B).



Fig. 2. Confocal laser micrographs of HeLa (D, E, F) cells incubated with Fe₃O₄@mSiO₂(MB)-FA for 2 h and 4 h. For each panel, from left to right were the cells with unclear staining with DAPI, with MB fluorescence and overlays of images.



Fig. 3. Flow cytometry histogram profiles (A) and fluorescence intensity (B) of HeLa cells after incubation with $Fe_3O_4@mSiO_2(MB)$ -FA nanoparticles for 2 h and 4 h.

3.4 Laser induced in vitro PDT effects on cancer cells

In order to assess the dark cytotoxicity of our multifunctional Fe₃O₄@mSiO₂(MB)-FA nanoparticles for in vivo applications, we did an MTT assay using SK-OV-3 cells as model and the results indicate that Fe₃O₄@mSiO₂(MB)-FA displays noncytotoxicity and good biocompatibility on SK-OV-3 cells within 0-200 µg/mL (Figure S7). To investigate the PDT efficiency of Fe₃O₄@mSiO₂(MB)-FA, SK-OV-3 cells were first incubated with different concentrations of Fe₃O₄@mSiO₂(MB)-FA for 24 h and then treated with/without laser (650 nm) irradiation. The MTT assay was used to evaluate the cell viabilities. Cell viability was normalized to control cells (no drug and nonirradiated). The combination of 24 h exposure of tumor cells to Fe₃O₄@mSiO₂(MB)-FA followed by laser irradiation-induced dose and light-dependent cytotoxicity on the tumor cells, which was significantly different from nonirradiation control in statistics, as shown in Fig. 4A and Fig. 4B. With the increase of drug dose and light dose, the cell viability gradually decreases. To the irradiated group, 100 µg/mL Fe₃O₄@mSiO₂(MB)-FA causes approximately 50% cell viability lost, demonstrating an obvious photodynamic activity. When the concentration of Fe₃O₄@mSiO₂(MB)-FA reaches 200 µg/mL, the cells are almost death (about 80% loss of cell viability). The group treated with the drug without light exposure shows that the drug alone has no effects on tumor cells which coincides with the result of cytotoxicity assessment.



Fig. 4. Phototoxicity of Fe₃O₄@mSiO₂(MB)-FA to HeLa cells. (A) SK-OV-3 cells were incubated with 0-200 μ g/mL Fe₃O₄@mSiO₂(MB)-FA for 24 h at 37 °C in the dark prior to irradiation for 4 min with 650 nm laser (70 mW/cm²). (B) SK-OV-3 cells were incubated with 150 μ g/mL Fe₃O₄@mSiO₂(MB)-FA for 24 h at 37 °C in the dark prior to of a series of light doses (0, 4.2, 8.4, 12.6, 16.8 and 21 J/cm²).

3.5 In vivo imaging

The tumor uptake of Fe_3O_4 @mSiO_2(MB)-FA in the mice was determined by NIR fluorescence imaging of cancer tumor model mice after injection of nanoparticles *via* the tail vein. Figure 5 shows *in vivo* near infrared fluorescence imaging of saline (Fig. 5I), Fe_3O_4 @mSiO_2(MB)-NH₂ (Fig. 5 II) and Fe_3O_4 @mSiO_2(MB)-FA (Fig. 5III) in mice after 1.5 h, 3 h and 4.5 h post-injection through the tail vein. As expected, from Fig. 5, Fluorescence signal in mouse III (A, B, C) distributed throughout the whole body, and the intensity of mouse III (A, B, C) was higher than that of mouse II (A, B, C) at the tumor site. The results indicated that Fe_3O_4 @mSiO_2(MB)-FA exhibited a longer plasma circulation time than others. However, Fe_3O_4 @mSiO_2(MB)-NH₂ in blood decreased rapidly after the

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administration that intravenous due to the fact Fe₃O₄@mSiO₂(MB)-FA reduced the uptake of MB by the reticuloendothelial system (RES) and the mononuclear phagocyte system.²³ In the images of ex vivo (Figure 5D), the high accumulation of nanoparticles was observed in RES organs, including liver, spleen and kidney, possibly caused by the clearance of the RES. And a significant enhance in MB levels was found in tumor site of mouse III, indicating that the FA conjugated nanoparticles are able to be recognized by the FR of tumor cells. The prolonged circulation gave the opportunity for Fe₃O₄@mSiO₂(MB)-FA to accumulate to the tumor site by EPR effect and made FA conjugated nanoparticles contact with tumor cells and internalize into the tumor cell via receptor-mediated endocytosis.



Fig. 5. *In vivo* optical fluorescence imaging of HeLa tumor xenografted mude mice and their several organs and tumor, which were taken after sacrifice (at 4.5 h post-injection)

3.6 In vivo antitumor efficacy

To investigate *in vivo* PDT efficacy of Fe₃O₄@mSiO₂(MB)-FA, comparative efficacy studies were conducted. The S-180 tumorbearing mice were divided into 5 groups and were treated according to protocols as summarized in method Section 2.2.9. The changes of relative tumor volume as a function of time were plotted in (Fig. 6A). After 8 days treatment, control group shows a relative tumor volume (V/V₀) of 8.70 ± 0.89, Magnet/650 nm laser and Fe₃O₄@mSiO₂(MB)-FA/magnet results in V/V₀ of 9.53 ± 0.96 and 8.79 ± 0.67, the tumorbearing mice treated with Fe₃O₄@mSiO₂(MB)-FA/650 nm laser and Fe₃O₄@mSiO₂(MB)-FA/magnet/650 nm laser achieves (V/V₀) of 6.11 ± 0.6 and 5.45 ± 0.95. Fe₃O₄@mSiO₂(MB)-FA/magnet/650 nm laser has tumor growth inhibition (TGI) of 62.64%, it is significantly more effective than the other therapeutic groups (p < 0.05). In Figure 6B and Figure 6C, it was obvious that, compared with control group, mice treated with Fe₃O₄@mSiO₂(MB)-FA/650 nm laser, the tumor volume is greatly reduced, this is a successful application that MB is used as a photosensitizer in PDT to achieve in vivo tumor treatment efficacy. Remarkably enhanced PDT effect is observed for tumor-bearing mice treated by Fe₃O₄@mSiO₂(MB)-FA/magnet/650 nm laser, in comparison to that treated with Fe₃O₄@mSiO₂(MB)-FA/650 nm laser at the same Fe₃O₄@mSiO₂(MB)-FA and light doses. Because of the magnetic targeting property of Fe₃O₄@mSiO₂(MB)-FA, when a magnet is glued to the top of the tumor, more Fe₃O₄@mSiO₂(MB)-FA will go to the tumor site than nonmagnetic group, so the PDT efficacy of Fe₃O₄@mSiO₂(MB)-FA with a magnet is higher than that of non-magnetic group. The growth of tumor tissue is successfully suppressed by Fe₃O₄@mSiO₂(MB)-FA/magnet/650 nm laser. This high high therapeutic efficacy originates from the Fe₃O₄@mSiO₂(MB)-FA accumulation in tumor tissue.



Fig. 6. Antitumor efficacy of Fe3O4@mSiO2(MB)-FA in the Kunming mice bearing S-180 tumors at a dose of 5 mg/kg; each group was intravenously administered every day for a total of 8 times, error bars were based on SD of five tumors per group. Control: blank; M: magnet; D: drug; L: light. A: Tumor growth of mice in different treatment groups within 8 days. B: Quantitative results of tumor weight excised from the tumor-bearing mice sacrificed on day 8. C: The picture of the tumors in different groups at the end of the experiment.

Conclusions

In summary, a multifunctional system Fe₃O₄@mSiO₂(MB)-FA for simultaneous cancer NIR imaging and targeting PDT is and developed. successfully designed The prepared Fe₃O₄@mSiO₂(MB)-FA has high water-solubility, noncytotoxicity, good biocompatibility, and can serve as a powerful PDT agent for in vitro fluorescence imaging and PDT cancer cell killing under NIR light irradiation. In vivo targeting PDT is further demonstrated in our animal experiments, showing excellent tumor ablation therapeutic efficacy by using $Fe_3O_4(a)mSiO_2(MB)$ -FA as the targeting PDT agent. Consequently, this unique multifunctional system may have great potential application in early diagnosis and therapy of cancer.

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Notes and references

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references

- F. Yang, C. Jin, S. Subedi, C. L. Lee, Q. Wang, Y. J. Jiang, J. Li, Y. Di and D. L. Y, *Cancer. Treat. Rev.*, 2012, 38, 566-579.
- B. Sivakumar, R. G. Aswathy, Y. Nagaoka, S. Iwai, K. Venugopal, K. Kato, Y. Yoshida, T. Maekawa and D. N. Sakthi Kumar, *RSC Advances*, 2013, 3, 20579-20598.
- A. C. Bonoiu, S. D. Mahajan, H. Ding, I. Roy, K. T. Yong, R. Kumar, R. Hu, E. J. Bergey, S. A. Schwartz and P. N. Prasad, *Proc. Natl. Acad. Sci. USA*, 2009, **106**, 5546-5550.
- 4. A. Jemal, R. Siegel, E. Ward, Y. P. Hao, J. Q. Xu, T. Murray and M. J. Thun, *CA-Cancer J. Clin.*, 2008, **58**, 71-96.
- O. C. Farokhzad, J. Cheng, B. A. Teply, I. Sherifi, S. Jon, P. W. Kantoff, J. P. Richie and R. Langer, *Proc. Natl. Acad. Sci. U S A*, 2006, 103, 6315-6320.
- 6 . K. Yan, P. Li, H. Zhu, Y. Zhou, J. Ding, J. Shen, Z. Li, Z. Xu and P.K. Chu, *RSC Adv.*, 2013,3, 10598-10618
- 7. Z. H. Cao, R. Tong, A. Mishra, W. C. Xu, G. C. L. Wong, J. J. Cheng and Y. Lu, *Angew. Chem. Int. Edit.*, 2009, **121**, 6616-6620.
- J. Kim, H. S. Kim, N. Lee, T. Kim, H. Kim, T. Yu, I. C. Song, W. K. Moon and T. Hyeon, *Angew. Chem. Int. Ed.*, 2008, 47, 8438 –8441.
- X. Z. Shi, H. Gong, Y. J. Li, C. Wang, L. Cheng and Z. Liu, Biomaterials, 2013, 34, 4786-4793.
- D. Bechet, P. Couleaud, C. Frochot, M. L. Viriot, F. Guilleminand and M. Barberi-Heyob, *Trends. Biotechnol.*, 2008, 26: 612-621.
- Y. N. Konan, R. Gurny and E. Allemann, J. Photoch. Photobio. B, 2002, 66, 89-106.
- S. B. Brown, E. A. Brown and I. Walker, *Lancet. Oncol.*, 2004, 5, 497-508.
- A. Mazzaglia, C. Conte, A. Scala, G.Siracusano, N. Leone, S. Patanè, F. Ungaro, A. Miro, M. T. Sciortino and F. Quaglia, *RSC Adv.*, 2014, DOI: 10.1039/C4RA07847K.
- P. Couleaud, V. Morosini, C. Frochot, S. Richeter, L. Raehmand and J.O. Durand, *Nanoscale*, 2010, 2, 1083–1095.
- R. R. Allison, R. E. Cuenca, G. H. Downie, P. Camnitz and B. Brodish, *Photodiagn. Photodyn.*, 2005, 2, 205-222.
- 16. Z. W. Li, C. Wang, L. Cheng, H. Gong, S. N. Yin, Q. F. Gong, Y. Q. Li and Z. Liu, *Biomaterials*, 2013, 34, 9160-9170.
- Y. Cheng, J. D. Meyers, A. M. Broome, M. E. Kenney, J. P. Basilion and C. Burda, *J. Am. Chem. Soc.*, 2011, **133**, 2583-2591.

- B. I. Slowing, B. G. Trewyn, S. Giri and VS-Y. Lin, *Adv. Funct. Mater.*, 2007, 17, 1225-1236.
- M. Gary-Bobo, O. Hocine, D. Brevet, M. Maynadier, L. Raehm, S. Richeter, V. Charasson, B. Loock, A. Morère, P. Maillard, M. Garcia and J. O. Durand, *Int. J. Pharm.*, 2012, **423**, 509-515.
- 20. J. Wang, C.K. Tsung, R. C. Hayward, Y. Wu and G. D. Stucky, *Angew. Chem. Int. Edit.*, 2005, 44, 332-336.
- J. Lu, M. Liong, J. I. Zink and F. Tamanoi, Small, 2007, 3, 1341-1346.
- 22. S. L. Gai, P. P. Yang, C. X. Li, W. X. Wang, Y. L. Dai, N. Niu and J. Lin, *Adv. Funct. Mater.*, 2010, **20**, 1161-1172.
- 23. S. H. Cheng, C. H. Lee, M. C. Chen, J. S. Souris, F. G. Tseng, C. S. Yang, C. Y. Mou, C. T. Chen and L. W. Lo, *J. Mater. Chem.*, 2010, 20, 6149-6157.
- A. K. Parchur, A. A. Ansari, B. P. Singh, T. N. Hasan, N. A. Syed, S. B. Rai and R. S. Ningthoujam, *Integr. Biol.*, 2014, 6, 53–64.
- A. A. Ansari, T. N. Hasan, N. A. Syed, J. P. Labis, A.K. Parchur, G. Shafi, A. A. Alshatwi, *Nanomedicine*, 2013, 9, 1328–1335.
- N. Nishiyama, Y. Morimoto, W.D. Jang and K. Kataoka, *Adv. Drug. Deliver. Rev.*, 2009, **61**, 327-338.
- R. Dabestani and I. N. Ivanov, *Photochem. Photobiol.*, 1999, **70**, 10-34.
- S. H. Seo, B. M. Kim, A. Joe, H. W. Han, X. Chen and Z. Cheng, Biomaterials, 2014, 35, 3309-3318.
- T. Deng, J.S. Li, J.H. Jiang, G.L. Shen and R.Q. Yu, *Adv. Funct. Mater.*, 2006, 16, 2147-2155.
- Y. Chen, H. Chen, S. Zhang, F. Chen, S. Sun, Q. He, M. Ma, X. Wang, H. Wu, L. Zhang, J. Shi, *Biomaterials*, 2012, 33, 2388-2398.
- 31. B. P. Pan, D. X. Cui, Y. Sheng, C. Ozkan, F. Gao, R. He, Q. Li, P. Xu and T. Huang, *Cancer Res.*, 2007, 67, 8156-8163.
- 32. F. Hu, L. Wei, Z. Zhou, Y. Ran, Z. Li and M. Gao, *Adv. Mater.*, 2006, 18, 2553-2556.
- 33. X. L. Zhao, H. L. Zhao, Z. Y. Chen, D. H. Zhang and M. B. Lan, J. Nanosci. Nanotechnol, 2014; DOI: 10.1166/jnn.2014.9867.
- 34. P. K. Frederiksen, M. Jørgensen and P. R. Ogilby, J. Am. Chem. Soc., 2001, 123, 1215-1221.
- A. Karotki, M. Kruk, M. Drobizhev, A. Rebane, E. Nickel and C. W. Spangler, J. Sel. Top. Quant., 2001, 7, 971-975.
- W. Spilleer, H. Kliesch, D. Wohrle, S. Hackbarth, B. R. B. Roder and G. Schnurpfeil, *J. Porphyr. Phthalocya.*, 1998, 2, 145-158.
- X. L. Zhao, H. L. Zhao, H. H. Yuan and M. B. Lan, J. Biomed. Nanotechnol., 2014, 10, 262-270.