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Cite this: DOI: 10.1039/c0xx00000x

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ARTICLE TYPE

RSC Advances Accepted Manuscript

Smart functional polymers – a new route towards creating a sustainable environment

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Received (in XXX, XXX) Xth XXXXXXXX 20XX, Accepted Xth XXXXXXXX 20XX 5 DOI: 10.1039/b000000x

Smart functional polymers have gained huge amount of interest in recent times due to their innumerable applications in the areas including sensors, actuators, switchable wettability, bio-medical and environmental applications. Numerous intensive research has been carried out to develop smart functional polymers using stimuli responsive polymeric moieties. This review article encapsulates recent

¹⁰ developments in the area of smart functional polymers with a focus on various (physical, chemical and biological) stimuli responsive systems and their applications. Furthermore, this review also provides useful insights and in-depth analysis on the feasibility of utilizing stimuli responsive polymeric materials / composites in anti-fouling and water harvesting applications which holds tremendous potential to create a sustainable environment.

15 1. Introduction

Many living creatures in nature have perfected the art of responding to various adverse external stimuli. Few naturally existing stimuli responsive creatures include the leaves of Mimosa pudica (collapses immediately when touched), Venus

- ²⁰ flytrap (closes fast enough to catch its prey), sea cucumbers (changes their stiffness in face of danger), chameleons (changes colours according to the nature of the environment) and sunflower (follows the movement of sun) etc.¹⁻³ Researchers studied these naturally existing beings and discovered that the
- 25 responsive nature is primarily due to the presence of biomacromolecules or biopolymers such as proteins and collagens etc. For example, in sea cucumbers, the secreted chemical triggers the interaction of collagen nanofibers present inside the dermis layer which helps to reinforce the soft matrix and thereby
- ³⁰ increase the stiffness of the dermis membrane.² Inspired by such naturally existing materials, researchers have concentrated their efforts to design synthetic functional responsive polymers and polymer composites that are useful for various scientific and industrial applications. Over the past few years, several polymers
- ³⁵ and polymeric composites responding to stimuli such as light, temperature, pH etc. have been developed by making use of interactions between polymer moieties and by implementing erudite synthesis methodologies.^{1,3} A wide range of responsive polymers including bulk solids, thin films, gels, nanofibers etc.
- ⁴⁰ have been effectively fabricated. They demonstrate great potential in industrial applications such as coatings, sensors, actuators, electronic devices etc. and also in bio-medical applications including drug delivery, gene delivery, imaging and diagnosis.
- ⁴⁵ This review article orbits around the recent developments in the area of smart functional polymers primarily focussing on different (physical, chemical and biological) stimuli responsive

systems and their potential applications. Some unique applications of the responsive polymers, specifically in the areas of anti-fouling and water harvesting are also highlighted. We believe that by employing stimuli responsive polymeric materials/composites in environmental remediation applications such as anti-fouling and water harvesting applications may assist in creating a sustainable environment.

55 2. Smart Polymers

Polymers that possess the ability to respond to external stimuli are referred to as smart polymers or stimuli responsive polymers.³ These polymers can respond to stimuli in several ways by altering colour, light transmitting abilities, conductivity, shape, wettability ⁶⁰ etc.⁴⁻⁶ The degree of response of such polymers can be trigged and controlled by the intensity of applied stimuli. Researchers have developed many responsive polymers/surfaces and employed various stimuli such as temperature, intensity of light, humidity, pH, electric/magnetic fields and also combination of ⁶⁵ stimuli to induce and control changes in the physical/chemical properties of the polymers thereby making them smart, functional and highly suitable for numerous household and industrial applications.⁷ Stimuli responsive polymers can be broadly classified as (i) single-stimulus and (ii) multi-stimuli responsive ⁷⁰ polymers as exemplified in **Fig. 1**.

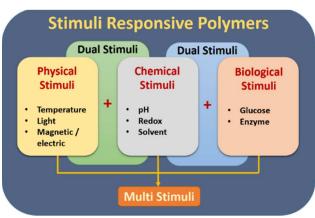


Fig. 1 Classification of Stimuli Responsive Polymers

2.1. Single stimulus-responsive polymers

Stimuli that induce changes in the polymer can be further s classified into three categories: physical, chemical and biological stimuli.⁸ Polymers respond to physical stimuli (light, temperature, magnetic and electrical) due to the modification of chain dynamics (i.e. the energy level of the polymer/solvent system) (**Fig. 2**).⁹ On the other hand, chemical stimuli modulate molecular interactions between polymer and solvent molecules or between polymer chains to induce changes in the polymer (**Fig. 3**).¹⁰ Biological stimuli correspond to the actual functioning of the molecules such as enzymatic reactions and receptor recognition etc.¹¹

15 2.1.1. Physical Stimuli

2.1.1.1 Temperature as a stimulus

Among all the existing physical stimuli, temperature stimulus has attracted a great deal of attention because this stimulus can be ²⁰ easily applied and monitored externally.¹²⁻¹⁴ Thermo-responsive polymer systems exhibit a critical solution temperature at which the polymer system undergoes a phase change within a small temperature range. This phenomenon is due to the disruption of intra and intermolecular interactions resulting in the expansion or

- ²⁵ contraction of polymer chains. A typical thermo-responsive polymeric solution possesses an upper critical temperature (UCST) above which one phase of the polymer exists and below which phase separation can be noticed. On the contrary, polymeric solutions that remain monophasic below a specific ³⁰ temperature and turn into bi-phasic above that temperature are
- generally considered to possess lower critical solution temperature (LCST). Based on the mechanism and end group chemistry, several thermo-responsive polymer systems have been reported: (a) poly(N-alkyl substituted acrylamides), e.g. poly(N-
- ³⁵ isopropylacrylamide) with an LCST of 32 °C ^{15,16} and (b) poly (N-vinylalkylamides), e.g. poly(N-vinylcaprolactam), with an LCST of about 32-35 °C.¹⁷ Other copolymers such as poly(Llactic acid)-poly(ethylene glycol)-poly(L-lactic acid) (PLLA-PEG-PLLA) triblock copolymers,¹⁸ and poly(ethylene oxide)-
- ⁴⁰ poly(propylene oxide)-poly (ethylene oxide) (PEO-PPO-PEO) copolymers have also been investigated for their thermoresponsive activities.¹⁹

2.1.1.2. Light as a stimulus

Light as a stimulus offers numerous flexibilities including 50 instantaneous application, high accuracy with tunable exposure wave-length control and it also enables the control of long distant applications by the use of fiber optic cables. Light stimulus beginning with hard ultra-violet to infra-red allows diversity in application which may not be offered by other stimuli. 55 Furthermore, light can be applied directly on the polymer surface to trigger a response.^{20,21} In light responsive polymers, the impact created by exposure to light induces photoisomerization and/or photochromism,^{22,23} which renders light as a highly versatile stimuli. Highly investigated photo-responsive polymers include 60 azobenzene (transcis isomerization), spiropyran (spiro to merocyanine form), spirooxazine (spiro to merocyanine form) and fulgide (photochromic behaviour) derivatives.²⁴⁻³⁰ Recently developed photosensitive block copolymer micelles have also attracted lot of attention.31

2.1.1.3. Magnetic / electric signals as a stimulus

Electrical stimulus can precisely control the response of polymers via the magnitude of the current, duration of an electrical pulse ⁷⁰ and intervals between the pulses. Electrically responsive polymers are typically conducting polymers that can transform their shape (swell, shrink or bend) when subjected to an electric field.³² Commonly explored electrically responsive polymers include polythiophene (PT) and sulphonated-polystyrene ⁷⁵ (PSS).^{33,34}

Polymer composites that respond to changes in magnetic fields are referred to as magnetically active polymer composites. These polymer composites are made of elastomers or gels filled with small magnetic particles. Typical fillers include metal particles, ⁸⁰ iron (III) oxide particles, ferromagnetic particles, NdFeB particles and nickel powders. Materials that are widely explored for developing the polymer matrix are poly(p-dioxanone)- poly(ɛcaprolactone) copolymer, cross-linking oligo (ɛ-caprolactone) dimethacrylate/butyl acrylate, and grafting polymer poly(ɛss caprolactone) diisocyanatoethyl methacrylate (PCLDIMA) and poly(ethylene glycol) mono-methylether-monomethacrylate (PEGMA).³⁵⁻⁴⁰

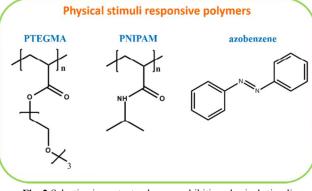


Fig. 2 Selective important polymers exhibiting physical stimuli responsive properties.

2.1.2. Chemical Stimuli

5 2.1.2.1. pH as a stimulus

pH responsive polymers consist of moieties that can donate or accept protons when there is an environmental change in pH.41-43 Any change in pH initiates ionic interactions leading to the 10 collapse or expansion of polymer chains in aqueous solution, induced by the electrostatic repulsion of the charges that are generated in this process.⁴⁴ Typical pH responsive materials include polyacids and polybases. Polyacids such as poly(acrylic acid) (pKa around 5) donates its protons and swells under basic 15 conditions, while polybases such as poly(N,N-dimethyl aminoethyl methacrylate) accepts protons under acidic conditions and expands due to coulomb repulsion.⁴⁵ Highly investigated pH responsive polymers include chitosan,⁴⁶ albumin,⁴⁷ gelatin,⁴⁸ poly(acrylic acid) (PAAc)/chitosan IPN,49 poly(methacrylic acid-²⁰ g-ethylene glycol) [P(MAA-g-EG)],⁵⁰ poly(ethylene imine) (PEI).51 poly(N,N-diakylamino ethylmethacrylates) (PDAAEMA), and poly(lysine) (PL).^{52,53}

2.1.2.2. Redox as a stimulus

25 Redox stimulus occurs due to the change in oxidation state of redox sensitive groups. Such stimulus can be mostly seen in inorganic chemistry particularly with transition metals. However, several organic compounds such as dithienylethenes,⁵⁴ ferrocene 30 or disulfides 55,56 also responds to redox sensitivity. Furthermore, acid moieties present in polyanhydrides,⁵⁷ liable poly(lactic/glycolic) acid⁵⁸ induces redox responsiveness due to their instability in reducing environment. Redox responsive polymer like poly(NiPAAm-co-Ru(bpy)3) can produce a 35 chemical wave due to the periodic redox change of Ru(bpy)3 into an oxidized state of lighter color. Such redox reaction results in

the alteration of hydrophobic and hydrophilic properties of the

polymer chain (by swelling/de-swelling the polymer).⁵⁹

40 2.1.2.3. Solvent as a stimulus

Solvent responsive polymeric systems can be synthesized from deformed polymers as solvent molecules cause swelling of the polymeric materials and increases the flexibility of the ⁴⁵ macromolecular polymer chains. Surfaces with switchable properties can be obtained by switching the conformation of surface grafted polymer chains. Structural transition of brushes by solvent treatment involves the preparation of patterned molecular brushes. Several polymers including poly(methyl ⁵⁰ methacrylate) (PMMA) and polystyrene (PS) patterns are synthesized and their responsiveness with different solvents are studied. It is found that the conformational transition of the synthesized polymer brushes largely depends on the quality of the solvent.⁶⁰ For example, Chen *et al.* ^{60,61} studied the deformation

⁵⁵ of PMMA line patterned brushes with different solvents. They found that the degree of deformation of the PMMA brushes can be varied when treated with different solvents. **Fig. 4** shows the schematic illustration of the reversible behaviour of PMMA brushes when immersed with water and THF (Tetrahydrofuran) ⁶⁰ resulting in brush and mushroom like regimes, respectively.

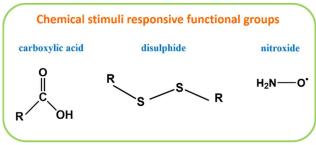


Fig. 3 Selective important functional groups exhibiting chemical stimuli responsive properties.

Several other polymers such as poly-(ethylene glycol) (PEG), 65 poly(butyl acrylate) and poly(2-dimethylaminoethyl methacrylate) have also been explored to fabricate solvent responsive polymers. Deformation of polymer brushes on solvent treatment has opened up many new possibilities in surface engineering concepts.^{62,63}

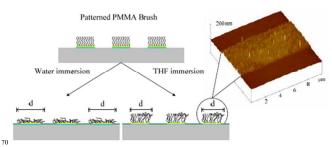


Fig. 4 Schematic illustrations of reversible poly(methyl methacrylate) (PMMA) brush treated with good and poor solvents.⁶⁰

2.1.3. Biological Stimuli

75 2.1.3.1. Glucose responsive

Glucose responsive polymers are widely explored due to their potential applications in drug delivery (insulin delivery).⁶⁴ Glucose responsive polymers are synthesized by conjugating ⁸⁰ glucose oxidase (GOx) with a pH responsive polymer. When such polymer comes in contact with glucose, GOx oxidizes glucose to gluconic acid which causes a change in pH of the environment. In response to the pH change, the pH responsive polymer shows a volume transition. This drastic change in the ⁸⁵ polymer is regulated by the body's glucose level which in turn affects the enzyme activity. Currently, there is huge amount of interest devoted in this area to develop sensitive, bio-degradable glucose responsive polymers.

90 2.1.3.2. Enzyme responsive

Naturally occurring bacteria located in the colon region secretes special enzymes such as azoreductase and glycosidases which are capable of degrading various polysaccharides including pectin, ⁹⁵ chitosan, dextrin etc.⁶⁵⁻⁶⁷ These bacterial enzymes generally destroy the polymer system completely. Henceforth, a typical enzyme responsive polymer system does not require any external

trigger for its decomposition. Because of this mechanism, enzyme responsive polymers attract great deal of attention in biological applications. Nonetheless, a major challenge the researchers face in employing these polymer systems is the difficulty of precisely s controlling the initial response time.^{68,69}

2.2. Dual stimuli-responsive polymers

2.2.1. Thermo and light responsive polymers

In the year 1988, Kungwatchakun *et al.*⁷⁰ reported the first dual responsive polymeric system (polymers that responds to both ¹⁰ thermal and light stimuli). In this work, thermo responsive copolymers are synthesized by polymerization of NIPAM (N-isopropylacrylamide) with an acrylamide monomer (N-(4-phenylazophenyl)acrylamide). Photochromic azobenzene moieties are introduced into the aqueous solution of PNIPAAM

- ¹⁵ and the phase separation temperature of the solution of 11th 11th and monitored by the intensity of light. During this process, a shift from 21 °C to 27 °C in phase separation temperature is seen after UV light irradiation. This reversible change in LCST is attributed to the change of dipole moment from 0 to 3 debye due
- ²⁰ to the trans-to-cis isomerization of the chromophoric moieties of azobenzene. The initial phase transition temperature of 21 °C is re-obtained by exposure to visible light. Based on this groundbreaking study, numerous studies focused their attention on using functionalized azobenzene moieties to develop dual responsive ²⁵ polymer systems.⁷¹⁻⁷⁴
- s porymer systems.

2.2.2. Thermo and pH responsive polymers

Polymeric systems that are responsive to both temperature and pH have attracted great deal of attention in the field of drug delivery, as both the entities are subjected to change inside the

- ³⁰ cancer tissue. These changes can be utilized to trigger autonomous response.⁷⁵⁻⁷⁷ The functional moieties that are capable of forming ionic groups by dissociation or association upon protonation are incorporated into the backbone chain of the LCST polymer (Examples: carboxylic acids and tertiary amines).
- ³⁵ Furthermore, certain homopolymers such as poly(2-(dimethylamino) ethyl methacrylate) (PDMAEMA) exhibit pH and thermo responsive behaviour. PDMAEMA exhibits a cloud point in the range of 50 °C in neutral aqueous solution.⁷⁸ Nonetheless, cloud points gets shifted to higher values with ⁴⁰ increasing pH due to the protonation of amino functionality
- thereby representing dual responsive behaviour.

2.3. Multi stimuli-responsive polymers

The success of dual responsive polymers encouraged researchers to investigate and develop polymeric systems that can respond to

⁴⁵ triple stimuli. The addition of another stimulus with the dual responsive polymeric systems can improve the precision of the response. Furthermore, the presence of additional stimuli can also enhance the switching windows or switching conditions due to the increased level of polymer complexities.

50 2.3.1. Thermo, light and pH responsive polymers

Tang *et al.*⁷⁹ synthesized polymeric system that responds to temperature, light and pH using azobenzene terminated PDMAEMA polymer via Atom-transfer radical-polymerization

(ATRP) (Fig. 5). They demonstrated that the LCST 55 characteristic of the developed polymeric system can be altered by changing the pH value. For example, when pH = 4, no LCST is seen due to an increased polarity caused by the protonation of dimethylamino functionality. Decreasing the proton concentration i.e. increasing the pH, caused deprotonation of dimethyl amino 60 group which resulted in the reduction of LCST to about 68 °C when pH = 7 and to 30 °C when pH = 11. Under UV light irradiation, the trans-to-cis photoisomerization of the azobenzene moiety resulted in a higher LCST. Nonetheless, it is recovered under visible light irradiation. This process is shown to be 65 completely reversible. Based on this work, several other studies were conducted to investigate the effect of azobenzene moieties located at the chain end of stimuli responsive polymers. The results of these studies suggest that the amount of azobenzene is proportional to the difference in LCST before and after 70 irradiation.80,81

Recently, several other polymeric systems including PNIPAAM with spirobenzopyran, hyper-branched polyethylenimine with isobutyramide groups and copolymeric systems prepared using N-hydroxymethylacrylamide (NHMA), ⁷⁵ NIPAM and 2-diazo-1,2-naphthoquinone-5sulfonylmethylacrylamide (DNQ) have also been investigated to produce triple responsive polymeric material systems.⁸²

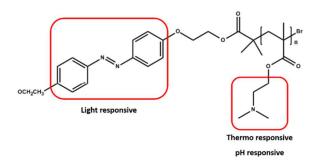


Fig. 5 PDMAEMA polymer end-functionalized with azobenzene, which can be stimulated by light, temperature and change of the pH value.⁷⁹

2.3.2. Thermo, light and redox responsive polymers

Various classes of triple responsive (thermo, light and redox) polymeric systems are synthesized by employing different redox active moieties. Theato et al.83 synthesized triple responsive 85 polymeric system using PNIPAAM copolymers containing 4amino-2,2,6,6-tetramethylpiperidine-1-oxyl (amino-TEMPO) and amino functionalized azobenzene moieties linked by an amide bond of the polymer backbone. The TEMPO moiety can be reduced by ascorbic acid, which is a mild reducing agent and can ⁹⁰ be re-oxidized by an oxidizing agent such as red prussiate. During the reduction process, the TEMPO moiety is reduced to the corresponding hydroxylamine. This results in a balance shift between hydrophilic/hydrophobic states, moving towards increased hydrophilicity thereby leading to an increase in LCST. 95 In order to stimulate the azobenzene chromophore, the polymeric system is irradiated with UV light. This process resulted in further increase of LCST. Although there is no difference in the final values of LCST irrespective of the sequence of applied stimuli to the two responsive groups, the intermediate values of

¹⁰⁰ LCST are seen to depend on the sequence of stimulation. When the azobenzene is stimulated first, the impact caused by azobenzene moiety is about 60% stronger compared to an already reduced copolymer. This is due to the increasing polarity influence of the isomerized azobenzene in hydrophobic copolymer state compared to the influence of the isomerization

- s existing within the already hydrophilic copolymer. Because of the stability of each state along with the reversibility phenomenon, this polymeric system is considered as a potential candidate for logic gate operations.
- Though there are several other dual and multi stimuli ¹⁰ responsive polymeric systems such as thermo-redox systems, environmental, pH and temperature responsive systems^{84,85} etc., the review within its scope has highlighted only the most widely used dual/multi stimuli systems.

3. Domain Specific Applications

15 Stimuli responsive polymers find their applications in numerous areas including sensors, actuators, biomedical and environmental applications. In this review article, we highlight few seminal applications of stimuli responsive polymeric systems in which most research activities are focused in.

20 3.1. Sensors

Sensors are primarily used to sense and provide information when there is a change in either physical, chemical or biological changes in an environment. The sensed information is then used to trigger the necessary actions. Few examples include ²⁵ monitoring toxic gasses and vapours in the working environment; constant check on the contaminants level in the industrial effluents etc. Thus novel developments in sensors technology will play a seminal role in maintaining a sustainable environment. As stimuli receptive polymers can sense and respond to the minor

³⁰ physical/chemical/biological changes, interest among the researchers has shifted to tailor make polymeric sensors. These sensors can address the needs of today's industrial demands to establish a sustainable environment.

35 3.1.1. Optical Sensors

Chen *et al.*⁸⁶ developed two-dimensional periodic concave grating (2DPCG) of tethered poly(N-isopropylacrylamide) (PNIPAAM). The fabricated array structures (200 nm on Si ⁴⁰ substrate) can be created and removed reversibly by changing the temperature (between 25 and 40° C). This leads to substantial changes in the effective refractive index (n_{eff}), resulting in the colour change from blue to red. The colour change can be observed by the naked eye at an incident angle of 10°-20°. These

⁴⁵ types of polymeric system can be potentially used in temperature responsive optical devices. Many other materials including poly(methyl methacrylate) (PMMA),⁸⁷ polymerized (2dimethylaminoethyl methacrylate) (PDMAEMA),⁸⁸ photonic crystals^{89,90} etc. have also been explored in recent years to ⁵⁰ fabricate effective optical sensors.

3.1.2. Chemical Sensors

El-Ragehy *et al.*,⁹¹ fabricated and investigated the electrochemical ⁵⁵ response characteristics of poly(vinyl chloride) (PVC) membrane sensors to determine the presence of fluphenazine hydrochloride

and nortriptyline hydrochloride. This technique is based on the formation of ion-pair between the two cations and sodium tetraphenylborate (NaTPB) or tetrakis (4-chlorophenyl) borate ⁶⁰ (KtpClPB). Liu *et al.*⁹² developed a novel poly(vinyl) chloride (PVC) membrane electrode based on pethidine-phosphotung state ion association as the electroactive material. This is used for determining pethidine hydrochloride drug in injections and tablets.

Stimuli responsive polymer surfaces can be used as a triggered "catch and release" surface for the fabrication of nanomaterials. Comrie *et al.*⁹³ employed this property to fabricate polymer-metal micro objects. They achieved this by incorporating hydrophobic functional group into poly(glycidyl methacrylate) (PGMA)
⁷⁰ brushes as a means of creating a robust, etch resistant film on chromium substrate. This amalgamation of resistance to etching and ease of subsequent lift-off enabled the fabrication of hybrid planar polymer-metal objects. These types of responsive polymeric systems can be employed in areas such as drug 75 delivery, biomimetic systems, MEMS (microelectromechanical systems).

In addition to PVC and PGMA, other polymeric materials such as PMMA⁹⁴, PNIPAAM⁹⁵ have also been investigated to produce chemical sensors.

80 3.2. Actuators

Recently, researchers have employed stimuli responsive polymers to fabricate self-oscillating systems which can generate periodic mechanical energy from the chemical energy (Belousov-Zhabotinsky reaction). Such self-oscillating systems can be 85 widely used in pulse generators, chemical pacemakers, actuators and micro-pumps.⁹⁶ Yoshida et al.⁹⁷ developed polymeric gels which can undergo peristaltic motion without the application of external stimuli. These unique polymeric gels are fabricated by co-polymerizing N-isopropylacrylamide (thermo responsive 90 polymer), with ruthenium tris(2,20-bipyridine) [Ru(bpy)3] as the catalyst for the BZ reaction. Maeda et al.98 fabricated selfoscillating gel actuators with gradient structures. In this work, the pendulum motion is created by fixing one end of the gel as shown in Fig. 6. Aizenberg et al. 99,100 produced thin hydrogel film 95 actuators with high aspect ratio rods incorporated within them. The expansion and collapse of the thin hydrogel layers affected the orientation of the incorporated rods. These films can be employed in designing surfaces with reverse and conventional switching behaviour.

Jager *et al.*¹⁰¹ fabricated micro-actuators based on polypyrrolegold bilayers to enable large movement of structures attached to these actuators. They can be of great interest for the manipulation of biological objects such as single cells. Recently, Pedrosa and co-workers employed polypyrrole to fabricate polypyrrole-gold ¹⁰⁵ nanofingers which can effectively act as nano-actuators.¹⁰²

3.3. Bio-medical applications

3.3.1. Diagnosis

Polymeric systems that can respond to specific biomolecules and ¹¹⁰ also to changes in temperature, pH etc. can be highly useful in the detection of diseases that are caused by the imbalance in the chemicals or variations of physical/biological variables in the environment. Henceforth, great efforts have been made in recent times to employ stimuli responsive polymeric systems in the field of disease diagnosis and biosensors.

Uchiyama *et al.*^{103,104} employed PNIPAAM to detect the presence of benzofurazan and observed a clear reversible

- ⁵ response to temperature cycles associated with PNIPAAM chain conformational changes and the polarity sensitivity of benzofurazan moieties. This behaviour is exemplified in Fig. 7 which clearly shows the change in the fluorescence intensity of the polymer at different temperatures. Several material systems
- ¹⁰ including poly(2-vinylpyridine) (P2VP), diblock copolymers of poly(ethylene glycol) and poly(sulfadimethoxine) (PEG-PSDM) have also been explored for the fabrication of nano-biosensors in recent times.¹⁰⁵⁻¹⁰⁷

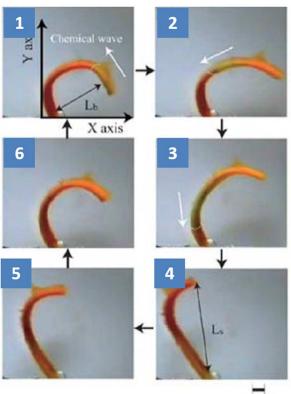




Fig. 6 Images of the repeated bending and stretching motion of the poly[NIPAAm-co-Ru(bpy)3-co-AMPS] gel strip (R10-A3) in the mixture solution of the BZ substrates; First, the whole gel strip was in homogeneous reduced state; Second, the gel strip is in a locally oxidized state when the chemical wave propagates in the gel from one edge to the 20 other edge (1 → 4); Finally, the whole gel strip change reduced state, and was bending (5 → 6).⁹⁸

3.3.2. Imaging

Stimuli responsive polymeric systems can be employed in ²⁵ imaging technique for the detection of diseased/damaged tissues. The damaged tissues show a moderately elevated temperature and pH when compared to normal healthy tissues. Stimuli responsive polymers can be used to sense this small temperature and pH differences to detect the diseased tissues.

Lee *et al.*¹⁰⁸ explored pluronic triblock copolymers [poly(ethylene oxide)-poly(propylene oxide)-poly(ethylene oxide) (PEO-PPO-PEO)] end-capped with a cyanine dye (Cy5.5) for imaging diseased tissues. Pluronic block copolymers responds to temperature stimulus via change in their supramolecular ³⁵ interactions. During heat treatment, polymer chains develop from a dissolved state to a micellar aggregation state. They found that the transition from dissolved chains to micelles is accompanied by fluorescence quenching of the Cy5.5 dye. These structures can be used as near infrared (NIR) thermo probes for imaging ⁴⁰ diseased/damaged tissues. In addition to block copolymers, poly(amidoamine) (PAMAM) dendrimers, peroxalate polymers embedded with fluorescent dye etc. have also been investigated for their potential use in imaging techniques.^{109,110}

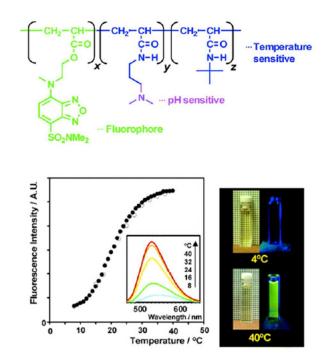


Fig. 7 A Fluorescent polymer sensor for temperature.¹⁰³

In addition to the above mentioned biomedical applications, stimuli responsive polymers have also been predominantly explored in the area of drug delivery systems including protein and enzyme delivery,¹¹¹⁻¹¹⁸ gene delivery¹¹⁹⁻¹²⁷ etc.

Thermo-responsive therapeutic grippers are widely used in drug delivery applications¹²⁸. Biopsy is another area which can make use of responsive polymers. Recently, Gultepe *et al.*¹²⁹ fabricated thermally responsive untethered microtools (μ grippers) which can effectively replace the conventional biopsy 55 forceps. Furthermore, research work have also been conducted on employing stimuli responsive polymers for other medical applications such as regenerative medicine, smart surfaces for tissue engineering, fabrication of biological interfaces, injectable implants¹³⁰⁻¹⁴⁵ etc. Table 1 summarizes the classes of polymer ⁶⁰ systems that are used for various stimuli responsive applications.

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Table 1: Various Stimuli Responsive Polymers and their applications

S.No	Types of Stimuli	Responsive polymers/co- polymers	Applications
1	Physical	PNIPAAM, PDMAEMA, PEGMA, PCLDIMA, azobenzene, spiropyran	Sensors, Actuators, Switchable wettability, Self-Healing,
2	Chemical	Chitosan, albumin, gelatin PDAAEMA, PAAc, PEI	Anti-fouling
3	Biological	Chitosan, polycaprolactone, PLA, PLGA	Diagnosis, Imaging, Drug Delivery, Bio- interface, Biopsy Studies

4. A new-route towards sustainable environment

- ¹⁵ In today's world environmental contamination is increasing drastically due to urbanization, increased industrial effluence and constant change in the life style of people. As a result of this, there is a substantial increase in global warming causing ad-hoc changes in the climatic conditions throughout the world.
 ²⁰ According to the recent report from NASA, the rate of increase of earth temperature has nearly doubled in the last 50 years.¹⁴⁶ In
- current scenario, providing clean air to breathe and pure drinking water for the world population itself poses immense challenge.¹⁴⁷ Nonetheless, the advent of new-age smart, functional polymeric
- ²⁵ materials has given enormous hope, scope and opportunities to establish a sustainable environment. Though, the stimuli responsive polymeric materials may not provide solution for all the environmental issues, they can still address vital issues like

reduction of sewage fouling, seawater fouling, fouling caused by ³⁰ industrial effluents and automobile exhaust etc.

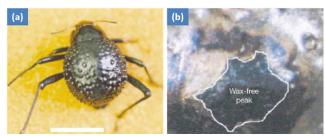
Furthermore, the ability of stimuli responsive polymeric materials to absorb moisture from the humid environment and release pure water on application of suitable stimuli has made them a potential candidate to address the growing demand for the ³⁵ supply of pure water in the humid geographical locations. Due to their diversified abilities, researchers believe that stimuli responsive polymeric materials can provide much needed remedy to the growing demands of today's environment. Henceforth, many research groups have focused their attention on developing

⁴⁰ polymeric systems for anti-fouling and water harvesting applications. The following two sections of the review article highlights some seminal research work that has been reported so far in the literature that employs stimuli responsive polymeric materials for anti-fouling and water harvesting applications.

45 4.1. Water harvesting

The very idea of harvesting water from the moist environment has been derived from nature. This behaviour of extracting water from humid air is first observed in a special type of desert beetle known as Stenocara (The tenebrionid beetle, **Fig. 8a**). This insect ⁵⁰ possesses a tailor made covering on its body for collecting water from early morning fog.¹⁴⁸ The beetle tilts its body forward into the direction of blowing winds and makes use of its tailor made carapace to extract water from air. It is believed that the large droplet of water is deposited on its carapace due to the insect's ⁵⁵ bumpy surface (near-random array of bumps 0.5-1.5 mm apart,

- each about 0.5 mm in diameter, **Fig. 8b**) which consist of alternating hydrophobic (waxy region) and hydrophilic (non-waxy) regions. The water harvesting system functions by 'growing' droplets on the hydrophilic seeding points of the peaks.
- ⁶⁰ The water present in the fog/moisture settles on these hydrophilic peaks gets accumulated to from large droplets of water. The water striking the hydrophobic slopes bounces towards hydrophilic region and gets collected.



⁶⁵ Fig. 8 The water-capturing surface of the fused over-wings (elytra) of the desert beetle Stenocara. a) Adult female, dorsal view; peaks and troughs are evident on the surface of the elytra; b) A 'bump' on the elytra, stained with Red O for 15 min and then with 60% isopropanol for 10 min, a procedure that tests for waxes. Depressed areas of the otherwise black
 ⁷⁰ elytra are stained positively (waxy, colored), whereas the peaks of the bumps remain unstained (wax-free; black).¹⁴⁸

In addition to dessert beetles, Zheng *et al.*¹⁴⁹ investigated and reported that spider silk also possesses the ability to extract water from moist air. Water harvesting ability of cribellate 'Uloborus 75 walckenaerius' spider's silk is attributed to the unique fiber structure that it attains after it comes in contact with moisture. The 'wet-rebuilt' fibers are characterized by periodic spindleknots made of random nanofibrils and separated by joints made **RSC Advances**

of aligned nanofibrils (**Fig. 9**). These unique structural features result in a surface energy gradient between the spindle-knots and the joints and also in a difference in Laplace pressure, with both factors acting together to achieve continuous condensation and ⁵ directional collection of water drops around spindle-knots. In this way, the spider silk harvest water from the humid air.

Inspired by these water harvesting strategies perfected by nature, over the past decade, researchers have investigated techniques to mimic the biological approach with the help of

- ¹⁰ synthetic materials to develop an efficient water harvesting systems that are capable of extracting water from foggy/moist environment. Recently, Yang *et al.*¹⁵⁰ employed "grafting form" approach to fabricate sponge-like cotton fabric using poly(Nisopropylacrylamide), PNIPAAM, brushes which has the ability
- ¹⁵ to autonomously collect and release water from humid atmosphere (**Fig. 10**). In this polymeric system, the collection and release of water from fog/moisture is triggered and controlled by applied temperature. The mechanism behind this phenomenon is attributed to the structural changes of temperature responsive
- ²⁰ PNIPAAM combined with the high surface roughness of the synthesized fabric. This mechanism is also responsible for reversible switching between two extreme wettability states (superhydrophilicity/superhydrophobicity) (see Fig. 11). Lot of research work is being carried out in this area using various ²⁵ stimuli responsive polymeric systems.¹⁵¹⁻¹⁵⁵

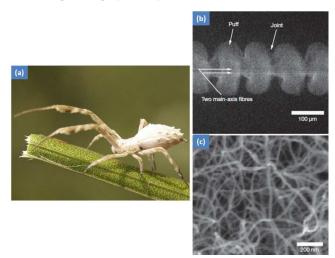


Fig. 9 Photograph of Uloborus walckenaerius (Photo Credit: Ferran Turmo Gort; Authors would like to convey special thanks to Ferran Turmo Gort for the photograph); (b) Low magnification environmental
 30 SEM image of periodic puffs and joints surrounding two main-axis fibers; (c) Magnified image of puff composed of countless nanofibrils.¹⁴⁹

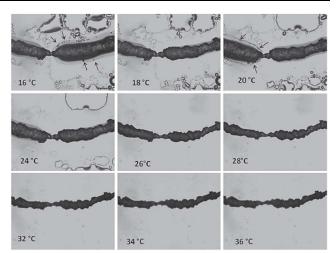
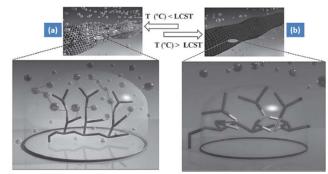


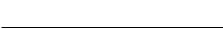
Fig. 10 Water collection/release by the PNIPAAM-cotton fiber upon temperature changes. OM sequential images of the PNIPAAM-cotton as the temperature rises from 16-36 °C under controlled atmosphere (96% humidity). Arrows indicate water droplets absorbed by the fiber.¹⁵⁰



9 Fig. 11 (a) water collection (superhydrophilic state) by hydrogen bonding between PNIPAAM; b) water release (superhydrophobic state) by the formation of PNIPAAM intermolecular bonds.¹⁵⁰

4.2. Anti-fouling

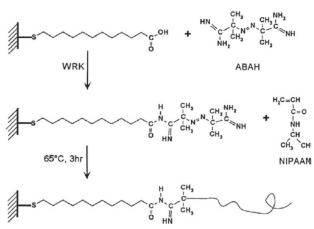
Currently, anti-fouling is another critical environmental 45 application in which stimuli responsive polymers are widely explored. In particular, poly(N-isopropylacrylamide) (PNIPAAM) is most widely studied environmentally sensitive polymer for anti-fouling applications.¹⁵⁶⁻¹⁶³ Lopez et al.¹⁵⁶ investigated the stimuli responsive wettability of PNIPAAM for 50 anti-fouling applications (fouling-release surfaces). They employed dip coating approach to fabricate anti-fouling coatings on glass substrates. PNIPAAM-coated glass slides were subjected to incubation (18 hr) in artificial sea water containing the marine bacterium Halomonas marina or in natural bay water at a 55 temperature above the LCST. The samples were then rinsed with artificial seawater (4° C) to induce fouling release properties. The authors report that 95% of the fouling bacterial cells that had attached to the PNIPAAM surfaces were removed effectively. Lopez and co-workers¹⁵⁷ also studied the anti-fouling abilities of 60 surfaces fabricated from PNIPAAM-PS (polystyrene) tethered surfaces. Furthermore, the same research group prepared antifouling films on gold surfaces by in situ polymerization of



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NIPAAM on initiator modified self-assembled monolayers (SAM) (**Fig. 12**).¹⁵⁸ From the attachment and detachment studies,

it is seen that more than 90% of cells released from PNIPAAM functionalized SAM surface.



⁵ Fig. 12 Schematic representation of the process of in situ polymerization of NIPAAM on initiator-derivatized SAMs. 2,2 ' -azobis(2amidopropane) hydrochloride (ABAH) is used as a free-radical initiator; NIPAAM = N –isopropylacrylamide.¹⁵⁸

¹⁰ In addition to PNIPAAM, poly(ethylene oxide) PEO is also explored for anti-fouling properties. Recently, Minko *et al.*¹⁶⁴ proposed a novel coating approach with 3D grafted polymer structure for rendering a surface with long-lasting antifouling properties (**Fig. 13**).

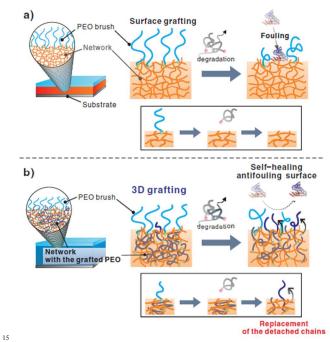


Fig. 13 Schematic illustration of the PEO-grafted P2VP network films: a) the grafting of PEO to the surface of a P2VP film and b) 3D polymer grafting on the surface and inside of a P2VP film. The self-healing aspect of the antifouling property is due to the rearrangement of internally grafted polymers to the interface (marked as dark blue chains).¹⁶⁴

The proposed coating with the 3D grafting consisting of polymeric chains grafted both to the surface and inside the host

material possesses the ability to retain the anti-fouling effect even when substantial fraction of the grafted polymers are degraded ²⁵ and detached. The self-healing mechanism of the proposed structure refers to the replacement of the detached or damaged polymeric chains by segments of the chains stored inside the film in proximity to the interface. A number of segments of the stored

grafted chains relocate from the film's interior driven by an ³⁰ emerging gradient in a chemical potential, and an antifouling effect in the exposed area can thus be recovered by itself. The authors claim that the pH-responsive poly(2-vinylpyridine) films with the 3D grafting of poly(ethylene oxide) in physiological conditions (pH 7.4 and 37 °C) demonstrated a 4-fold increase in ³⁵ longevity of antifouling behaviour than the material with the surface grafted polymer. Furthermore, it was observed that, the 3D grafted responsive films retain their pH responsive properties.

5. Conclusions and Future Outlook

In the recent past, stimuli responsive polymers/polymeric 40 composites have attracted great deal of attention due to their innumerable applications including sensors, actuators, switchable wettability, drug delivery, imaging, diagnosis, self-healing coatings and also in miniaturized electronic devices. The rapid development in this filed is due to (1) comprehensive 45 understanding of the responsive polymeric moieties interaction; (2) development of new erudite synthesising approaches; (3) fabrication of polymeric systems which can be controlled by employing more than one stimuli and (4) nanoscale fabrication of stimuli responsive surfaces etc. Future studies can focus on 50 doping stimuli responsive polymeric materials with metal oxide and noble metal nano-particles. Incorporation of these dopants into the polymeric matrix may result in interesting properties.

In yet another example, stimuli responsive polymers can potentially play a major role for development in the field of 55 information technology. The concept of orthogonal and independent addressing of the responsive group may result in achieving parallel writing of information. Furthermore, a dramatic increase in memory density may also be possible in future. These objectives can be achieved by the development of 60 new polymer chemistries and also by the precise control of several stimuli responsive groups. Medical application is another field, which is expected to be impacted by the developments in stimuli responsive polymers. Though abundant research has been carried out in employing stimuli responsive polymers in various 65 medical applications including diagnostic equipment, therapeutic treatments, tissue regeneration etc., application of stimuli responsive materials at nanometre scale is still an emerging area which will benefit greatly from extensive studies on biodisposability, bio-distribution, toxicity etc. Another area that is 70 gaining huge amount of interest is the development of CO₂ responsive polymers. Certain CO₂ responsive polymers can reversibly capture CO₂ from air resulting in the reduction of greenhouse effect. However, further extensive studies are needed to address the existing challenges in the area of CO₂ responsive 75 polymers including the development of facile synthetic approaches, exploring the behaviour of CO₂ responsive polymers by modifying the physical properties such as shape and size etc.

Recently, numerous studies have explored the feasibility of employing stimuli responsive polymeric materials in anti-fouling

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and water harvesting applications. In future, commercial success of stimuli responsive materials in anti-fouling and water harvesting applications may lead to establishing a sustainable environment. Above all, we believe that fabrication of cost 5 effective and highly efficient stimuli responsive materials for household, industrial, medicinal and environmental applications

household, industrial, medicinal and environmental applications may require successful collaboration among researchers from different fields, so that in near future, stimuli responsive materials may pave the way for a better tomorrow.

10 Acknowledgement

The authors, VAG and AB would like to acknowledge the financial support from Ministry of Education (MOE - Tier 2), Singapore [Grant T2-MOE-1302].

Notes and references

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References

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- 1. L. Zhai, Chem. Soc. Rev., 2013, 42, 7148-7160.
- J. R. Capadona, K. Shanmuganathan, D. J. Tyler, S. J. Rowan and C. Weder, *Science*, 2008, **319**, 1370-1374; R. Birenheide, M. Tamori, T. Motokawa, M. Ohtani, E. Iwakoshi, Y. Muneoka, T. Fujita, H. Minakata, K. Nomoto, *Biol Bull.*, 1998, **194**, 253-259.
- Handbook of Stimuli-Responsive Materials, ed. U. Marek, 100 Wiley-VCH Verlag GmbH&Co. KGaA, Weinheim, Germany, 2011.
- 4. L. Brunsveld, B. J. B. Folmer, E. W. Meijer and R. P. Sijbesma, *Chem. Rev.*, 2001, **101**, 4071-4098.
- D. Roy, J. N. Cambre and B. S. Sumerlin, *Prog. Polym. Sci.*, 105 2010, 35, 278-301.
- 6. F. Liu and M. W. Urban, Prog. Polym. Sci., 2010, 35, 3-23.
- M. A. C. Stuart, W. T. S. Huck, J. Genzer, M. Muller, C. Ober, M. Stamm, G. B. Sukhorukov, I. Szleifer, V. V. Tsukruk, M. Urban, F. Winnik, S. Zauscher, I. Luzinov and S. Minko, *Nat.* 110 *Mater.*, 2010, 9, 101-113.
- E. Cabane, X. Zhang, K. Langowska, C. G. Palivan and W. Meier, *Biointerphases*, 2012, 7, (1-4):9.
- 9. E. S. Gil and S. M. Hudson, Prog. Polym. Sci., 2004, 29, 1173-1222.
- M. Delcea, H. Mohwald, A. G. Skirtach, *Adv. Drug. Deliv. Rev*, 2011, 63, 730-747.
 - W. B. Liechty, D. R. Kryscio, B. V. Slaughter and N. A. Peppas, *Annu. Rev. Chem. Biomol. Eng.*, 2010, 1, 149-173.
 - N. Zhang, S. Salzinger and B. Rieger, *Macromolecules*, 2012, 120 45, 9751-9758.
 - X. Y. Liu, F. Cheng, Y. Liu, H. J. Liu and Y. Chen, J. Mater. Chem., 2010, 20, 360-368.
 - B. Xue, L. Gao, Y. Hou, Z. Liu and L. Jiang, *Adv. Mater.*, 2013, 25, 273-277.
 - 15. Y. Liu, L. Meng, X. Lu, L. Zhang and Y. He, *Polym. Adv. Technol.*, 2008, **19**, 137-143.
 - S. Ohya, H. Sonoda, Y. Nakayama and T. Matsuda, Biomaterials, 2005, 26, 655-659.
 - K. Suwa, K. Morishita, A. Kishida and M. Akashi, J. Polym. 130 Sci. Part A Polym. Chem., 1997, 35, 3087-3094.
 - K. Na, K. H. Lee, D. H. Lee and Y. H. Bae, *Eur. J. Pharm. Sci.*, 2006, 27, 115-122.

- 19. A. Sosnik and D. Cohn, *Biomaterials*, 2004, **25**, 2851-2858.
- 20. Y. Qiu and K. Park, Adv. Drug. Deliv. Rev, 2001, 53, 321-339.
- A. Roggan, M. Friebel, K. Dorschel, A. Hahn and G. Muller, *J. Biomed Opt.*, 1999, 4, 36-46.
- Y. Shiraishi, R. Miyamoto and T. Hirai, Org. Lett., 2009, 11, 1571-1574.
- A. Albini, E. Fasani and D. Faiardi, J. Org. Chem., 1987, 52, 155-157.
- 24. F. D. Jochum and P. Theato, *Polymer*, 2009, **50**, 3079-3085.
- 25. Z. Mahimwalla, K. Yager, J. Mamiya, A. Shishido, A. Priimagi and C. Barrett, *Polym. Bull.*, 2012, **69**, 967-1006.
- 26. C. Pietsch, U. S. Schubert and R. Hoogenboom, *Chem. Commun.*, 2011, 8750-8765.
- T. Hirakura, Y. Nomura, Y. Aoyama and K. Akiyoshi, Biomacromolecules, 2004, 5, 1804-1809.
- 28. M. Rini, A. K. Holm, E. T. J. Nibbering and H. Fidder, J. Am. Chem. Soc., 2003, **125**, 3028-3034.
- 29. Y. Huang, W. Liang, J. K. S. Poon, Y. Xu, R. K. Lee and A. Yariv, *Appl. Phys. Lett.*, 2006, **88**, 181102-181103.
- A. K. Chibisov and H. Gorner, *Phys. Chem. Chem. Phys.*, 2001, 3, 424-431.
- 31. Y. Zhao, J. Mater. Chem., 2009, 19, 4887-4895.
- 32. P. M. Mendes, Chem. Soc. Rev, 2008, 37, 2512-2529.
- D. P. Jones, J. L. Carlson, P. S. Samiec, P. Sternberg, V. C. Mody, R. L. Reed and L. A. S. Brown, *Clinica. Chimica. Acta*, 1998, **275**, 175-184.
- A. N. Koo, H. J. Lee, S. E. Kim, J. H. Chang, C. Park, C. Kim, J. H. Park and S. C. Lee, *Chem. Commun.*, 2008, 44, 6570-6572.
- X. Yu, S. Zhou, X. Zheng, T. Guo, Y. Xiao and B. Song, Nanotechnology, 2009, 20, 235702:1-9.
- J. M. Cuevas, J. Alonso, L. German, M. Iturrondobeitia, J. M. Laza and J. L. Vilas, *Smart Mater. Struct.*, 2009, 18, 075003:1-10.
- A. Golbang and M. Kokabi, *Eur. Polym. J.*, 2011, 47, 1709-1719.
- M. Zeng, S. W. Or and H. L. W. Chan, J. Appl. Phys., 2010, 96, 203502:1-3.
- D. W. Zhang, Y. J. Liu and J. S. Leng, *Appl. Phys. Lett.*, 2010, 96, 111905:1-3.
- 40. Z. Varga, G. Filipcsei and M. Zrinyi, *Polymer*, 2006, **47**, 227-233.
- S. P. Nunes, A. R. Behzad, B. Hooghan, R. Sougrat, M. Karunakaran, N. Pradeep, U. Vainio and K. V. Peinemann, *ACS Nano*, 2011, 5, 3516-3522.
- 42. H. Jia, A. Wildes and S. Titmuss, *Macromolecules*, 2012, **45**, 305-312.
- 43. R. Liu, P. Liao, J. Liu and P. Feng, *Langmuir*, 2011, **27**, 3095-3099.
- 44. S. Dai, P. Ravi and K. C. Tam, Soft Matter, 2008, 4, 435-449.
- 45. F. Liu and M. W. Urban, *Macromolecules*, 2008, **41**, 6531-6539.
- M. Y. Abdelaal, E. A. Abdel-Razik, E. M. Abdel-Bary and I. M. Sherbiny, *J. Appl. Polym. Sci.*, 2007, **103**, 2864-2874.
- 47. H. Y. Park, I. H. Song, J. H. Kim and W. S. Kim, *Int. J. Pharm.*, 1998, **175**, 231-236.
- 48. M. Kurisawa and N. Yui, *Macromol. Chem. Phys.*, 1998, **199**, 1547-1554.
- 49. J. W. Lee, S. Y. Kim, S. S. Kim, Y. M. Lee, K. H. Lee and S. J. Kim, *J. Appl. Polym. Sci.*, 1999, **73**, 113-120.
- J. Zhang and N. A. Peppas, *Macromolecules*, 1999, 33, 102-107.
- Z. Sideratou, D. Tsiourvas and C. M. Paleos, *Langmuir*, 2000, 16, 1766-1769.
- 52. S. E. Burke and C. J. Barrett, *Biomacromolecules*, 2003, 4, 1773-1783.
- V. Toncheva, M. A. Wolfert, P. R. Dash, D. Oupicky, K. Ulbrich, L. W. Seymour, *Biochim. Biophys. Acta*, 1998, 1380, 354-368.
- H. Logtenberg and W. R. Browne, Org. Biomol. Chem., 2013, 11, 233-243.

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RSC Advances

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95

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105

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120

125

130

140

- 55. M. Mazurowski, M. Gallei, J. Li, H. Didzoleit, B. Stuhn and M. Rehahn, Macromolecules, 2012, 45, 8970-8981.
- D. J. Phillips and M. I. Gibson, Chem. Commun., 2012, 48, 56. 1054-1056.
- K. W. Leong, B. C. Brott, R. Langer, J. Biomed. Mater. Res., 57. 1985, 19, 941-955.
 - 58. S. Cohen, T. Yoshioka, M. Lucarelli, L. H. Hwang, R. Langer, Pharm. Res, 1991, 8, 713-720.
- R. Yoshida, T. Yamaguchi and E. Kokufuta, J. Artif. Organs, 59. 1999, 2, 135-140.
- J. K. Chen, C. Y. Hsieh, C. F. Huang, P. M. Li, S. W. Kuo and 60. F. C. Chang, Macromolecules, 2008, 41, 8729-8736.
 - J. K. Chen, C. Y. Hsieh, C. F. Huang and P. M. Li, J. Colloid 61 Interface Sci., 2009, 338, 428-434.
- J. K. Chen and A. L. Zhuang, J. Phys. Chem. C, 2010, 114, 15 62. 11801-11809.
 - J. K. Chen and A. L. Zhuang, Colloid Polym. Sci., 2011, 289, 63. 1283-1294.
 - D. Velluto and J. A. Hubbell, 64. S. Cerritelli, Biomacromolecules, 2007, 8, 1966-1972.
 - O. Chambin, G. Dupuis, D. Champion, A. Voilley and Y. 65. Pourcelot, Int. J. Pharm., 2006, 321, 86-93.
 - V. R. Sinha, R. Kumria, Int. J. Pharm, 2001, 224, 19-38. 66
 - T. F. Vandamme, A. Lenourry, C. Charrueau and J. C. 67.
 - Chaumeil, Carbohydr. Polym., 2002, 48, 219-231. Y. Itoh, M. Matsusaki, T. Kida and M. Akashi, 68. Biomacromolecules, 2006, 7, 2715-2718.
 - E. O. Akala, P. Kopeckova, and J. Kopecek, Biomaterials, 69. 1998, 19, 1037-1047.
 - D. Kungwatchakun and M. Irie, Makromol. Chem., Rapid 70. Commun., 1988, 9, 243-246.
 - 71. X. Tao, Z. Gao, T. Satoh, Y. Cui, T. Kakuchi and Q. Duan, Polym. Chem., 2011, 2, 2068-2073.
 - 72. A. Dirani, X. Laloyaux, A. E. Fernandes, B. Mathy, O. Schicke, O. Riant, B. Nysten and A. M. Jonas, Macromolecules, 2012, 45, 9400-9408.
 - 73. F. D. Jochum, L. Zur Borg, P. J. Roth and P. Theato, Macromolecules, 2009, 42, 7854-7862.
 - 74. N. Ishii, J. Mamiya, T. Ikeda and F. M. Winnik, Chem. 110 Commun., 2011, 47, 1267-1269.
 - 75. B. Mu and P. Liu, React. Funct. Polym., 2012, 72, 983-989.
 - 76. C. M. Schilli, M. Zhang, E. Rizzardo, S. H. Thang, Y. K. Chong, K. Edwards, G. Karlsson and A. H. E. Muller, Macromolecules, 2004, 37, 7861-7866.
- S. Kulkarni, C. Schilli, B. Grin, A. H. E. Muller, A. S. 45 77. Hoffman and P. S. Stayton, Biomacromolecules, 2006, 7, 2736-2741.
 - 78. V. Butun, S. P. Armes and N. C. Billingham, Polymer, 2001, 42. 5993-6008.
 - X. Tang, X. Liang, L. Gao, X. Fan and Q. Zhou, Polym. Sci., 79. Part A: Polym. Chem., 2010, 48, 2564-2570.
 - 80. H. Akiyama and N. Tamaoki, Macromolecules, 2007, 40, 5129-5132.
 - H. Akiyama and N. Tamaoki, Polym. Sci., Part A: Polym. 81. Chem, 2004, 42, 5200-5214.
 - Y. Y. Yu, F. Tian, C. Wei and C. C. Wang, Polym. Sci., Part 82. A: Polym. Chem, 2009, 47, 2763-2773.
 - 83. P. Schattling, F. D. Jochum and P. Theato, Chem. Commun., 2011, 47, 8859-8861.
 - D. J. Phillips and M. I. Gibson, Biomacromolecules, 2012, 13, 84. 3200-3208.
 - 85 N. Kuramoto, Y. Shishido and K. Nagai, Polym. Sci., Part A: Polym. Chem, 1997, 35, 1967-1972.
 - 86. J. K. Chen, J. H. Wang, C. C. Cheng and J. Y. Chang, ACS 135 Appl. Mater. Interfaces, 2013, 5, 2959-2966.
 - 87. J. K. Chen, J. H. Wang, C. C. Cheng, J. W. Chang and F. C. Chang, Appl. Phys. Lett., 2013, 102, 151906:1-4.
 - 88 J. K. Chen and B. J. Bai, J. Phys. Chem. C, 2011, 115, 21341-21350
- J. H. Chen, J. H. Wang, C. J. Chang and C. F. Huang, Sens. 89. 70 Actuat. B, 2013, 188, 1123-1131.

- 90. K. Matsubara, M. Watanabe and Y. Takeoka, Angew. Chem. Int. Ed., 2007, 46, 1688-1692.
- N. A. El-Ragehy, A. M. El-Kosasy, S. S. Abbas and S. Z. El-91 Khateeb, Anal. Chim. Acta., 2000, 418, 93-100.
- 92. Z. H. Liu, M. L. Wen, Y. Yao and J. Xiong, Sens. Actuat. B, 2001, 77, 219-223.
- 93. J. E. Comrie and W. T. S. Huck, Langmuir, 2007, 23, 1569-1576.
- 94 S. Tugulu, M. Harms, M. Fricke, D. Volkmer and H. A. Klok, Angew. Chem. Int. Ed., 2006, 45, 7458-7461.
- 95. S. Y. Kim, T. Kanamori and T. Shinbo, J. Appl. Polym. Sci., 2002, 84, 1168-1177.
- 96 A. N. Zaikin and A. M. Zhabotinsky, Nature, 1970, 225, 535-537.
- Y. Murase, S. Maeda, S. Hashimoto and R. Yoshida, 97. Langmuir, 2009, 25, 483-489.
- S. Maeda, Y. Hara, R. Yoshida and S. Hashimoto, Macromol. 98 Rapid Commun., 2008, 29, 401-405.
- B. Pokroy, A. K. Epstein, M. C. M. Persson-Gulda and J. 99. Aizenberg, Adv. Mater., 2009, 21, 463-469.
- 100. A. Sidorenko, T. Krupenkin, A. Taylor, P. Fratzl and J. Aizenberg, Science, 2007, 315, 487-490.
- 101. E. W. H. Jager, O. Inganäs and I. Lundström, Science, 2000, 288, 2335-2338.
- 102. V. A. Pedrosa, X. Luo, J. Burdick and J. Wang, Small, 2008, 4, 738-741.
- 103. S. Uchiyama, N. Kawai, A. P. de Silva and K. Iwai, J. Am. Chem. Soc., 2004, 126, 3032-3033.
- 104. S. Uchiyama, Y. Matsumura, A. P. de Silva and K. Iwai, Anal. Chem., 2003, 75, 5926-5935.
- 105. I. Tokareva, S. Minko, J. H. Fendler and E. Hutter, J. Am. Chem. Soc., 2004, 126, 15950-15951.
- 106. S. W. Hong, C. H. Ahn, J. Huh and W. H. Jo, Macromolecules, 39, 7694-7700.
- 107. T. Wu, G. Zou, J. Hu and S. Liu, Chem. Mater., 2009, 21, 3788-3798.
- 108. S. Y. Lee, S. Lee, I. C. Youn, D. K. Yi, Y. T. Lim, B. H. Chung, J. F. Leary, I. C. Kwon, K. Kim and K. Choi, Chem. Eur. J., 2009, 15, 6103-6106.
- 109. J. M. Criscione, B. L. Le, E. Stern, M. Brennan, C. Rahner, X. Papademetris and T. M. Fahmy, Biomaterials, 2009, 30, 3946-3955.
- 110. D. Lee, S. Khaja, J. C. Velasquez-Castano, M. Dasari, C. Sun, J. Petros, W. R. Taylor and N. Murthy, Nat. Mater., 2007, 6, 765-769
- 111. E. Cabane, V. Malinova, S. Menon, C. G. Palivan and W. Meier, Soft Matter, 2011, 7, 9167-9176.
- 112. J. C. M. Lee, H. Bermudez, B. M. Discher, M. A. Sheehan, Y. Y. Won, F. S. Bates and D. E. Discher, Biotechnol. Bioeng., 2001. 73. 135-145.
- 113. Y. Lee and K. Kataoka, Soft Matter, 2009, 5, 3810-3817.
- 114. Y. Lee, S. Fukushima, Y. Bae, S. Hiki, T. Ishii and K. Kataoka, J. Am. Chem. Soc., 2007, 129, 5362-5363.
- 115. P. M. George, D. A. LaVan, J. A. Burdick, C. Y. Chen, E. Liang and R. Langer, Adv. Mat., 2006, 18, 577-581.
- 116. J. M. Pernaut and J. R. Reynolds, J. Phys. Chem. B., 2000, 104, 4080-4090.
- 117. M. S. Yavuz, Y. Cheng, J. Chen, C. M. Cobley, Q. Zhang, M. Rycenga, J. Xie, C. Kim, K. H. Song, A. G. Schwartz, L. V. Wang and Y. Xia, Nat. Mater., 2009, 8, 935-939
- 118. C. T. Huynh, S. W. Kang, Y. Li, B. S. Kim and D. S. Lee, Soft Matter, 2011, 7, 8984-8990.
- 119. D. Wakebayashi, N. Nishiyama, Y. Yamasaki, K. Itaka, N. Kanayama, A. Harada, Y. Nagasaki and K. Kataoka, J. Controlled Release, 2004, 95, 653-664.
- 120. X. B. Xiong, H. Uludag and A. Lavasanifar, Biomaterials, 2010, 31, 5886-5893.
- 121. S. Lin, F. Du, Y. Wang, S. Ji, D. Liang, L. Yu and Z. Li, Biomacromolecules, 2007, 9, 109-115.
- 122. J. Du, Y. Tang, A. L. Lewis and S. P. Armes, J. Am. Chem. Soc., 2005, 127, 17982-17983.

65

80

85

- 123. H. Lomas, I. Canton, S. MacNeil, J. Du, S. P. Armes, A. J. Ryan, A. L. Lewis and G. Battaglia, *Adv. Mater.*, 2007, **19**, 4238-4243.
- 124. H. Lomas, M. Massignani, K. A. Abdullah, I. Canton, C. Lo Presti, S. MacNeil, J. Du, A. Blanazs, J. Madsen, S. P. Armes, A. L. Lewis and G. Battaglia, *Faraday Discuss.*, 2008, **139**, 143-159.
- 125. H. Iatrou, H. Frielinghaus, S. Hanski, N. Ferderigos, J. Ruokolainen, O. Ikkala, D. Richter, J. Mays and N. Hadjichristidis, *Biomacromolecules*, 2007, 8, 2173-2181.
- 126. M. Oishi, S. Sasaki, Y. Nagasaki and K. Kataoka, *Biomacromolecules*, 2003, **4**, 1426-1432.

10

15

20

25

35

45

55

60

- 127. N. K. Jain and A. Asthana, *Expert Opin. Drug Deliv.*, 2007, 4, 495-512.
- 128. K. Malachowski, J. Breger, H. R. Kwag, M. O. Wang, J. P. Fisher, F. M. Selaru and Prof. D. H. Gracias, *Angew. Chem. Int. Ed.*, 2014, **53**, 8045-8049.
- 129. E. Gultepe, Jatinder S. Randhawa, S. Kadam, S.Yamanaka, F. M. Selaru, E. J. Shin, A. N. Kalloo and D. H. Gracias, *Adv. Mater.*, 2013, **25**, 514-519.
- 130. A. Kushida, M. Yamato, C. Konno, A. Kikuchi, Y. Sakurai and T. Okano, J. Biomed. Mater. Res., 1999, 45, 355-362.
- 131. M. Hirose, O. H. Kwon, M. Yamato, A. Kikuchi and T. Okano, *Biomacromolecules*, 2000, **1**, 377-381.
- 132. M. Yamato, C. Konno, M. Utsumi, A. Kikuchi and T. Okano, *Biomaterials*, 2002, **23**, 561-567.
 - 133. M. Ebara, M. Yamato, T. Aoyagi, A. Kikuchi, K. Sakai and T. Okano, *Biomacromolecules*, 2004, **5**, 505-510.
 - 134. K. Anikin, C. Rocker, A. Wittemann, J. Wiedenmann, M. Ballauff and G. U. Nienhaus, J. Phys Chem. B, 2005, 109, 5418-5420.
 - 135. N. Nath and A. Chilkoti, Anal. Chem., 2003, 75, 709-715.
 - 136. N. Comolli, B. Neuhuber, I. Fischer and A. Lowman, *Acta. Biomater.*, 2009, **5**, 1046-1055.
- 137. S. Y. Kim and S. C. Lee, J. Appl. Polym. Sci., 2009, 113, 3460-3469.
 - W. Frey, D. E. Meyer and A. Chilkoti, *Langmuir*, 2003, 19, 1641-1653.
 - J. Hyun, W. K. Lee, N. Nath, A. Chilkoti and S. Zauscher, J. Am. Chem. Soc., 2004, 126, 7330-7335.
 - 140. D. Cunliffe, C. de las Heras Alarcon, V. Peters, J. R. Smith and C. Alexander, *Langmuir*, 2003, **19**, 2888-2899.
 - 141. C. E. Schmidt, V. R. Shastri, J. P. Vacanti and R. Langer, *Proc. Nat. Acad. Sci.*, 1997, 94, 8948-8953.
- 142. J. Shi, N. M. Alves and J. F. Mano, *Adv. Funct. Mater.*, 2007, 17, 3312-3318.
 - 143. J. H. Cho, S. H. Kim, K. D. Park, M. C. Jung, W. I. Yang, S. W. Han, J. Y. Noh and J. W. Lee, *Biomaterials*, 2004, 25, 5743-5751.
- 50 144. J. P. Chen and T. H. Cheng, *Macromol. Biosci.*, 2006, 6, 1026-1039.
 - 145. J. H. Ryu, Y. Lee, W. H. Kong, T. G. Kim, T. G. Park and H. Lee, *Biomacromolecules*, 2011, **12**, 2653-2659.
 - http://earthobservatory.nasa.gov/Features/GlobalWarming
 M. A. Montgomery and M. Elimelech, *Environ. Sci. Technol.*,
 - 2007, **41**, 17-24.
 - 148. A. R. Parker and C. Lawrence, Nature, 2001, 414, 33-34.
 - 149. Y. Zheng, H. Bai, Z. Huang, X. Tian, F. Q. Nie, Y. Zhao, J. Zhai1 and L. Jiang, *Nature*, 2010, 463, 640-643.
 - 150. H. Yang, H. Zhu, M. R. M. Hendrix, J. H. G. M. Lousberg, G. de With, A. C. Esteves and J. H. Xin, *Adv. Mat.*, 2013, 25, 1150-1154.
 - T. Uakushiji, K. Sakai, A. Kikuchi, T. Aoyagi, Y. Sakurai and T. Okano, *Langmuir*, 1998, 14, 4657-4662.
- 65 152. J. K. Chen, J. H. Wang, J. Y. Chang and S. K. Fan, *Appl. Phys. Lett.*, 2012, **101**, 123701:1-5.
 - 153. J. K. Chen and J. Q. Qui, J. Nanopart. Res., 2012, 14, 942-956.
 - 154. C. M. Bruinink, M. Peter, P. A. Maury, M. de Boer, L. Kuipers, J. Hukens and D. N. Reinhoudt, *Adv. Funct. Mater.*, 2006, 16, 1555-1565.

- 155. K. Y. Suh and H. H. Lee, Adv. Funct. Mater., 2002, 12, 405-413.
- 156. L. K. Ista and G. P. Lopez, J. Ind. Microbiol. Biotechnol., 1998, 20, 121-125.
- 157. L. K. Ista, V. H. Perez-Luna and G. P. Lopez, *Appl. Environ. Microbiol.*, 1999, 65, 1603-1609.
- 158. L. K. Ista, S. Mendez, V. H. Perez-Luna and G. P. Lopez, *Langmuir*, 2001, **17**, 2552-2555.
- 159. S. Mendez, L. K. Ista and G. P. Lopez, *Langmuir*, 2003, **19**, 8115-8116.
- 160. L. K. Ista, S. Mendez and G. P. Lopez, *Biofouling*, 2010, 26, 111-118.
- 161. D. Cunliffe, C. D. Alarcon, V. Peters, J. R. Smith and C. Alexander, *Langmuir*, 2003, **19**, 2888-2899.
- 162. V. A. Ganesh, H. K. Raut, A. S. Nair and S. Ramakrishna, J. Mater. Chem., 2011, 21, 16304-16322.
- 163. A. Baji, M. Abtahi and S. Ramakrishna, J. Nanosci. Nanotechnol., 2014, 14, 1-18.
- H. Kuroki, I. Tokarev, D. Nykypanchuk, E. Zhulina and S. Minko, Adv. Funct. Mater., 2013, 23, 4593-4600.

12 | Journal Name, [year], [vol], oo-oo