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Solid-state NMR measurements and DFT calculations of the magnetic shielding tensors of protons of water trapped in barium chlorate monohydrate

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Abstract

The magnetic shielding tensors of protons of water in barium chlorate monohydrate are investigated at room temperature by means of solid-state NMR spectroscopy, both for static powders and under magic-angle spinning conditions, using one- and two-dimensional techniques. First-principles DFT calculations based on a periodic planewave pseudopotential formalism for a static periodic system provide support for our spectral interpretation and corroborate the experimental findings in the fast motion regime.



Key words

Solid-State NMR, Chemical Shift Anisotropy, Dipolar Interaction, Dipolar Echo, DFT calculations, Periodic Planewave Pseudopotential Methods.

Abbreviations

- MAS Magic-angle spinning
- CSA Chemical Shift Anisotropy
- rms root-mean square
- DFT Density Functional Theory

Introduction

Water is one of the most abundant molecules on earth and plays a fundamental role in a vast variety of chemical systems, ranging from the biochemical pathways underlying the most complex functions of living systems, to the mechanical properties of the Earth's crust. Whether considered as an isolated molecule in the gas phase or in liquid or solid condensed phases, many aspects of water remain not yet properly understood.[1-3] Hydrogen bonding, chemical exchange and dynamics complicate the description of the system.

Recently, attention has been drawn to the possibility of manipulating the populations of the proton spin eigenstates of water with the view of exciting a long-lived state in analogy with para-H₂.[4] A knowledge of all nuclear spin interactions which can affect and perturb the eigenstates of protons in water is crucial for the design of experimental strategies aiming at establishing long-lived states in any context, whether in liquid bulk, trapped in a crystal or in a fullerene cage.[5-7] These interactions, i.e., chemical shieldings, dipolar or quadrupolar couplings, are generally anisotropic or orientation dependent, and may affect NMR spectra to an extent that may render spectral interpretation difficult.[8] Nevertheless, the inhomogeneous broadenings which arise from such interactions in solids can be thoroughly studied by NMR spectroscopy. The use of magic-angle spinning (MAS) can partially remove this broadening to yield high-resolution spectra that benefit from a gain in signal intensity.[9,10] Specific experiments may be used to reintroduce the anisotropic information averaged out by the mechanical rotation. All these capabilities identify solid-state NMR spectroscopy as a method of choice for investigations of the interactions that can affect nuclear spin states. Density Functional Theory (DFT) calculations based on a planewavepseudopotential formalism [11,12] can nowadays be readily performed for periodic systems made up of a few hundred atoms. Such *in-silico* calculations enable insight into observable properties, such as chemical shielding tensors and electric field gradients, that have proven extremely useful to assist the interpretation of NMR spectra of solid samples.[13-15]

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In this context, we turned our attention to the protons of water molecules trapped in crystals of barium chlorate monohydrate, $Ba(ClO_3)_2 \cdot H_2O$. The anisotropy of the chemical shifts of the protons has been investigated previously by NMR spectroscopy, both in solution and solid state.[16,17] Here, we further explore the inhomogeneous CSA interaction by refocusing the much larger homonuclear dipolar couplings. The experimental findings are interpreted in the light of the results from DFT calculations.

Results and discussion

Figure 1a shows a static proton NMR spectrum of barium chlorate monohydrate, Ba(ClO₃)₂·H₂O, acquired at room temperature in a 9.4 T magnet by means of a $90^{\circ}_{v} - \tau - 90^{\circ}_{\phi} - \tau$ solid echo experiment. When phase cycling is applied to the second 90° pulse to select the p = $+1 \rightarrow p = -1$ coherence pathway, this experiment can refocus both linear and quadratic interactions, i.e., both shift anisotropies and dipolar couplings.[18] The features of the lineshape clearly indicate what is commonly referred to as a *Pake pattern* due to *isolated* homonuclear dipolar-coupled spin pairs, originally observed in gypsum that, in contrast with our system, occurs as a dihydrate, i.e., CaSO₄·2H₂O.[19] The mere observation of such a powder pattern indicates that the water molecules are sufficiently remote from each other in the crystal that intermolecular couplings between protons of different water molecules do not contribute significantly to the lineshape. It is interesting to note the asymmetry of the powder pattern, which was not observed in Pake's pioneering study on gypsum, presumably because he worked at a much lower field of 0.682 T. It can easily be appreciated that the low-frequency (right-hand) shoulder of the pattern in Fig. 1a is broader than its high-frequency counterpart. This asymmetry has been already observed in spinning powders and correctly ascribed to the chemical shift anisotropy of the proton spins.[17] Figure 1b shows, a series of numerical simulations, that include a chemical shift anisotropy Δ_{cs} = -10 and +10 ppm (4 kHz at 9.4 T), in black and red, respectively, and reproduce well the asymmetry of the experimental powder pattern in Fig. 1a. However, as indicated by the series of spectra stacked in Fig. 1b for different

asymmetry of the shielding tensor $\eta_{CS} = 0$, 0.5 and 1, the effect of this parameter on the lineshape is less pronounced, with only the 'horns' of the pattern significantly affected. When compared to the experimental spectrum of Fig 1a, it is clear that even a slight homogeneous broadening can easily hide the subtle variations in the lineshape that result from a variation in η_{CS} . All this means that an accurate determination of the asymmetry, η_{CS} , form the analysis of 1D static NMR spectra may be, at least in the case at hand, very difficult, if not possible.

One additional parameter that needs to be taken into account in order to analyse the spectrum in Fig. 1a is the relative orientation of the shielding and dipolar tensors of the proton spins. The latter is aligned along the H–H vector whereas the former, when they are axially-symmetric, i.e., in the limit where $\eta_{CS} = 0$, are usually assumed to have their unique axis aligned parallel to their respective H–O bonds. In solid-state systems, when dealing with more than one interaction, three Euler angles $\Omega = (\alpha, \beta, \gamma)$ are required to describe their relative orientations. Each interaction can be defined in its own principal axis frame (*P*), where the relevant tensor is diagonal. A crystal frame (*C*) may also be considered so that all interactions can be referred to a common frame of reference. Consequently, each interaction λ has a specific set of Euler angles $\Omega_{PC}^{\lambda} = (\alpha_{PC}, \beta_{PC}, \gamma_{PC})$. Consideration of the rotor frame (*R*) is also required for MAS experiments. Finally, the lab frame (*L*), where the experiment takes place, is also needed. It is legitimate, and adopted in this study, to assume the *P* frame of a given interaction λ to be coincident with the common *C* frame. This is simply done by choosing $\Omega_{PC}^{\lambda} = (0^{\circ}, 0^{\circ}, 0^{\circ})$. In our case, for the two-spin system of a single isolated water molecule, two shielding tensors and one dipolar tensor need be taken into account.

However, water molecules that are trapped in solids are known to undergo rapid reorientation by flipping around the C_2 axis defined by the H–O–H bisector.[20] In the fast motional regime at room temperature, it is commonly assumed that such motions result in an average shielding tensor projected onto the C_2 axis, so that its main axis is, therefore, orthogonal to that of the dipolar tensor. As a result, the shielding tensors of the two protons are equivalent and colinear at room temperature. In contrast, a rotation about the C_2 axis has

no effect on the dipolar tensor since a 180° flip does not alter the size of this interaction. Therefore, in order to simulate the lineshapes, we assume the spin system to be made up of two I = 1/2 spins, with equivalent shielding tensors that are colinear, and with two P frames that are coincident with the common C frame. Consequently, we have three sets of Euler angles, $^{H(1)}\Omega_{PC}^{CS} = (0^{\circ}, 0^{\circ}, 0^{\circ}) = ^{H(2)}\Omega_{PC}^{CS}$ and $\Omega_{PC}^{D} = (0^{\circ}, \beta_{PC}, \gamma_{PC})$. As the dipolar tensor is axiallysymmetric and traceless, only two angles, say, β_{PC} and γ_{PC} (henceforth simply referred to as β and γ) are relevant, i.e., α_{PC} is redundant and assumed to be 0° in this context.[21] In Figure 1b, static patterns are simulated for two cases of $\Omega_{PC}^{D} = (0^{\circ}, 0^{\circ}, 0^{\circ})$ and $(0^{\circ}, 90^{\circ}, 0^{\circ})$, in black and red, respectively. As previously discussed for the asymmetry parameter η_{CS} of the shielding tensor, the angle β has very little effect on the static lineshape. If a systematic fit of the spectrum of Fig. 1a is performed over the two-dimensional space spanned by the parameters Δ_{cs} and η_{cs} , for the case of β = 90°, one finds Δ_{cs} = 11 ± 3 ppm and η_{cs} = 0.3 ± 0.5. Clearly, the error associated with the asymmetry is too large to be reliable. An analogous fit for the case β = 0° produces substantially identical parameters, i.e., \varDelta_{CS} = –10 \pm 3 ppm and η_{CS} = 1.0 \pm 0.7, meaning once more that β cannot be determined. It is worth noting that a fit assuming Ω_{PC}^{D} = $(0^{\circ}, 90^{\circ}, 0^{\circ})$ results in $\Delta_{CS} > 0$ whereas the case of $\Omega_{PC}^{D} = (0^{\circ}, 0^{\circ}, 0^{\circ})$ produces $\Delta_{CS} < 0$.

Figure 1c shows a magic-angle spinning (MAS) spectrum recorded at 9.4 T with a rotorsynchronized solid echo using a spinning frequency of $v_{rot} = 10$ kHz. The intensities of the spinning sidebands are markedly asymmetric with respect to the isotropic shift (marked by *), which, in analogy with the static case, can be ascribed to the chemical shift anisotropy. Figure 1d shows two simulations for the two cases of $\beta = 0$ and 90°, in black and red, respectively. The black spectrum is slightly shifted to higher frequencies for clarity. The two spectra are again remarkably similar, revealing only tiny differences in the intensities of the spinning sidebands. Only the cases of $\eta_{CS} = 0$ are shown, since variations of this parameter produces even smaller effects than variations of β . Attempts to fit the spectrum of Fig 1c result in very large uncertainties of the relevant parameters, reflecting the fact that they have little effect on

the lineshape. More specifically, one obtains $\Delta_{CS} = -9 \pm 7$ ppm and $\eta_{CS} = 0 \pm 11$ in the case of $\Omega_{PC}^{D} = (0^{\circ}, 0^{\circ}, 0^{\circ})$ and $\Delta_{CS} = 9 \pm 10$ ppm and $\eta_{CS} = 1 \pm 2$ in the case of $\Omega_{PC}^{D} = (0^{\circ}, 90^{\circ}, 0^{\circ})$. As previously observed for the fits of Fig. 1b, the cases of $\beta = 0$ and 90° yield, respectively, negative and positive values for the shift anisotropy Δ_{CS} . The lack of both accuracy and precision which affects these measurements can be rationalized by considering that the inhomogeneity due to the shielding interaction is almost completely averaged by MAS, since $\Delta_{CS} \approx 4$ kHz and $v_{rot} = 10$ kHz. Therefore, an accurate measurement of the shielding tensor with one-dimensional NMR techniques seems to be difficult under both static and MAS conditions in this case, where the size of the predominant dipolar interaction, i.e., ca. 30 kHz, and homogeneous broadening mask the effects of the shielding anisotropy.

In order to gain insight into the system under investigation, and to corroborate and interpret the inhomogeneities that were measured experimentally, periodic planewave pseudopotential DFT calculations were carried out with the CASTEP code [22] on the periodic system. Figures 2a, b and c show the unit cell of $Ba(CIO_3)_2 \cdot H_2O$ viewed down the x-, y- and zaxis, respectively. The unit-cell lengths are a = 8.92 Å, b = 7.83 Å and c = 9.43 Å, and angle $\beta =$ 93.39°.[23] The space group is C2/c. Four water molecules can be seen, each of which is neighbor to a Ba^{2+} ion on its C_2 axis. When the magnetic shielding tensors of protons in $Ba(ClO_3)_2 H_2O$ are computed before geometry optimization, all protons of all water molecules are characterized by the same main components of their shielding tensors. In contrast, if geometry optimization is performed, this degeneracy is broken and two types of water can be identified. Nevertheless, the proton sites within each water molecule are always identical to one another. If the unit cell size is fixed and conservation of the symmetry is imposed in the geometry optimization step, the differences between the two types of water tend to disappear. The latter condition has a smaller effect on the calculated shielding tensors than the former. Figure 2d shows the magnetic shielding tensors of the proton sites represented as light-brown ellipsoids. Once expressed in their principal axis frames, all protons are characterized by the same main components of the shielding tensor. The relative orientation

between two tensors of each water molecule is described by the Euler angles (92.83°, 65.47°, 92.83°). The relevant NMR parameters obtained are summarized in Table 1. Generally, DFT calculations yield $\Delta_{CS} = -16.5$ ppm and $\eta_{CS} = 0.2$. Note that very little difference is obtained between the varying methods for structural optimization. The computational investigation is performed on a static system, so that motional averaging of the interactions is not taken into account. As we expect the two tensors to be averaged by fast dynamics at room temperature, we express the two shielding tensors of two protons, H(1) and H(2) belonging to a single water molecule in a common frame by means of the following transformations:

$$\sigma'_{H(1)} = R^{-1}(\alpha, \beta, \gamma) \sigma_{H(1)} R(\alpha, \beta, \gamma), \tag{1}$$

where the operator $R(\alpha, \beta, \gamma)$ can be decomposed into its constituent rotations:

$$R(\alpha,\beta,\gamma) = R_z(\alpha)R_y(\beta)R_z(\gamma).$$
⁽²⁾

A single rotation operator, say, $R_z(\alpha)$, performs a rotation of the shielding tensor of H(1), $\sigma_{H(1)}$, through an angle α around the z-axis. This produces $\sigma'_{H(1)}$, i.e., $\sigma_{H(1)}$ as expressed in the principal axis of $\sigma_{H(2)}$. The average tensor $\bar{\sigma}_{H(1,2)}$ is then simply given by:

$$\bar{\sigma}_{H(1,2)} = \left(\sigma'_{H(1)} + \sigma_{H(2)}\right) / 2 , \qquad (3)$$

By diagonalizing $\bar{\sigma}_{H(1,2)}$ one obtains the principal components of the averaged interaction tensor. The relevant NMR parameters under investigation thus produced are $\bar{\Delta}_{CS} = -8.7$ ppm and $\bar{\eta}_{CS} = 0.9$. It is worth notice that the chemical shift anisotropy calculated with DFT methods is negative. These parameters represent the averaged shielding tensor of an averaged ¹H site that one can measure in barium chlorate monohydrate in the fast motional regime. If one considers instantaneous jumps of the protons between their two orientations, the details of the dynamic process are irrelevant, so long as the relative orientation between the initial and final configurations is known. The size of the calculated average anisotropy $\bar{\Delta}_{CS}$ is in good agreement

with the experimental 1D spectra. On the other hand, the uncertainty associated with our measurements of the asymmetry η_{CS} does not allow any reasonable comparison with the averaged value $\bar{\eta}_{cS}$ calculated with DFT methods.

The optimal conditions for the measurement of the shielding tensors of protons in $Ba(ClO_3)_2 H_2O$ would be a two-dimensional technique capable of isolating the shift interaction in the indirect dimension, producing a pure-shift $F_1(\omega_1)$ projection where the predominant dipolar interaction has been removed by refocusing. Antonijevic and Wimperis have proposed a two-dimensional NMR method to refocus the first-order quadrupolar interaction of deuterium spins (I = 1) in the indirect dimension of a static 2D spectrum.[21] The basic principle of this experiment relies on a solid echo in the center of a t_1 evolution to refocus the quadratic or bilinear Hamiltonians such as dipolar or first-order quadrupolar couplings, whereas modulations due to linear terms such as that of the inhomogeneous Zeeman Hamiltonian are retained. This can be achieved if the second 90° pulse is phase cycled to select the $p = +1 \rightarrow p = +1$ coherence pathway.[21] A pure-shift F₁ dimension is thus achieved. Although specifically designed for 2 H spins (*I* = 1), it is easy to verify that the same result can be achieved for a I = 1/2 spin pair with equivalent shielding tensors subject to a homonuclear dipolar coupling. This condition applies in our case of two equivalent tensors whose different orientations are averaged in the fast motional regime at room temperature. Although the space parts are different, the first-order quadrupolar interaction and homonuclear dipolar interaction have same bilinear spin parts, i.e., $T = 3I_zS_z - IS$, where I = S for I = 1. The evolution during the t_1 interval of the $90^{\circ}_{y} - t_1 / 2 - 90^{\circ}_{x} - t_1 / 2$ experiment may be represented by the following transformations:

$$\rho_{1} = U_{CS}(t_{1}/2)U_{D}(t_{1}/2)\rho_{0}U_{D}^{\dagger}(t_{1}/2)U_{CS}^{\dagger}(t_{1}/2), \qquad (4)$$

$$\rho_2 = U_{rf} \left(90_x^{\circ}\right) \rho_1 U_{rf}^{\dagger} \left(90_x^{\circ}\right), \tag{5}$$

$$\rho_{3} = U_{CS}(t_{1}/2)U_{D}(t_{1}/2)\rho_{2}U_{D}^{\dagger}(t_{1}/2)U_{CS}^{\dagger}(t_{1}/2), \qquad (6)$$

where $\rho_0 = I_+ + S_+$ for a system with two spins I = S = 1/2 or I_+ for a single I = 1 spin. Ideal pulses are considered and the order in which shift and dipolar (or first-order quadrupolar) interactions are treated is irrelevant, as the corresponding Hamiltonians commute with one another. In the case of two I = 1/2 spins, the final state ρ_3 is characterized by single-quantum matrix elements $\rho_3^{(p=+1)}$ which are modulated exclusively by the shift interaction, i.e., $Tr[I_-\rho_3] = 2\exp(-i2\pi\omega_{CSA})$. It is worth noting that the single-quantum matrix elements $\rho_3^{(p=-1)}$ are instead completely unmodulated, i.e., $Tr[I_+\rho_3] = 2$. This latter feature is utilised in the experiments of Fig.1. Figure 3a shows a two-dimensional NMR spectrum of a static powdered sample of barium chlorate monohydrate. The $F_2(\omega_2)$ dimension is, of course, affected by both CSA and dipolar interactions, whereas pure-shift information appears in F₁. It is worth noting that the horizontal F₂ projection appears slightly distorted when compared to the static spectrum acquired with a solid echo and shown in Fig. 1a. This is due to the fact that a whole echo [24] has been acquired in the 2D experiment whereas, in the case of the 1D spectrum, acquisition is started at the very top of the echo. In the former case, but not in the latter, unavoidable homogeneous losses act 'asymmetrically' around the top of the echo. Moreover, and ideally, an equal number of data points on each side of the top of the echo would be needed for undistorted lineshapes to be obtained. All these deviations from an ideal whole echo result in the distortion of the Pake pattern as observed in the projection of the 2D spectrum of Fig. 3a. Figure 3b shows the projection of the 2D spectrum onto the F₁ axis with corresponding fit, in black and red, respectively. Despite its strength, the dipolar interaction does not affect the lineshape and the shielding tensor is much easier to access. It is important to stress at this point that the pure-shift F1 dimension thus produced is independent of the relative orientation of the dipolar and shielding tensors. Furthermore, due to the equivalence of the principal components and orientations of the two proton tensors as a result of fast motional averaging at room temperature, the numerical fit in Fig. 3b could be performed by assuming a single I = 1/2 spin species. All these simplifications result in a considerable reduction of the number of variables to consider. Moreover, there is no need for fitting the whole two-dimensional spectrum. Figure 3c shows the rms deviations associated with the above-mentioned numerical fit as a function of the shift anisotropy Δ_{CS} and asymmetry η_{CS} .[13] The surface shows clear minima that correspond to the optimal parameters, thus indicating a more accurate and precise measurement of the principal components than those obtained from the 1D spectra of Fig 1. The relevant parameters that we obtain are $\Delta_{CS} = -10.5 \pm 0.5$ ppm and $\eta_{CS} = 0.7 \pm 0.2$, with errors that are considerably smaller than those obtained by the fits of the spectra in Fig. 1. Most importantly, these values agree remarkably well with the averaged parameters calculated with DFT methods.

In cases where the shift anisotropy is small and can easily be removed by MAS, Orr et al. [25,26] have developed a two-dimensional method capable of amplifying the chemical shift anisotropy in the indirect dimension of a two-dimensional spectrum. This allows one to have simultaneously the high resolution typical of MAS spectra, and the reintroduction of anisotropic information otherwise averaged out by the mechanical rotation. This experiment is formally equivalent to that of Crockford et al. [27,28] and has been directly derived as an amplified version of the method of Antzutkin et al. [29] who revisited the Phase-Adjusted Spinning Sidebands (PASS) [30-31] experiment with a rigorous formalism. The effect of homonuclear dipolar interactions has been investigated and proven to be deleterious for the desired recoupling and amplification of the chemical shift anisotropy.[26] However, it can be shown that, in the case of two equivalent spins, the presence of a homonuclear dipolar interaction does not affect the experiment because the dipolar Hamiltonian is not affected by the series of π pulses which aim to recouple and amplify the shielding anisotropy. As a result, the dipolar coupling is refocused by rotation at the magic angle *provided* the observation is performed stroboscopically with the rotor period. This implies that, as in the previously-described static 2D experiment, the resulting indirect F_1 dimension is again free of dipolar effects. Furthermore, as a result of the amplification effect N, the F_1 projection mirrors the intensities of the spinning sidebands that one would acquire if a simple dipolar echo experiment could be performed with a fictitious spinning rate $v'_{Rot} = v_{Rot} / N$, i.e., if one could generate more spinning sidebands to improve the mapping of the anisotropy. The 2D-Amplified PASS spectrum of Ba(ClO₃)₂·H₂O is

shown in Figure 4a. The spinning sideband pattern in F_2 is affected both by the dipolar and shift interactions, as in a simple dipolar echo experiment. In contrast, F₁ produces a series of points which mimic the spinning sideband intensities due to the shift interaction only. Moreover, having been recorded at v_{Rot} = 3333 Hz with an amplification factor N = 3.333, the spinning sideband pattern presented in F₁ resembles that acquired at a fictitious spinning rate of $v'_{Rot} = 1$ kHz. This grants with a larger number of spinning sidebands to analyse. Figure 4b shows fitting of the F₁ projection calculated by summing F₁ slices extracted at each spinning sideband and numerical fit, in black and red, respectively. Again, in this spinning case, as in the static experiment of Fig. 3, only one single I = 1/2 spin needs to be taken into account in the fit, without any need for simulating the entire multiple-pulse two-dimensional experiment. As in Fig. 3, the pure-shift F_1 dimension is independent of the relative orientation of the dipolar and shielding tensors. The agreement between the fit and the experiment is generally good. This is particularly evident when one considers the 2D plot of the rms deviations in Fig 4c, which allows one to extract the relevant parameters Δ_{CS} = -8.5 ± 0.6 ppm and η_{CS} = 1 ± 0.1 with greatly improved accuracy and precision compared to those obtained from 1D spectra. As for the static spectrum of Fig. 3, we are able to measure accurately and fully characterize a relatively small chemical shift interaction of ca. 4 kHz by removing the predominant dipolar interaction of ca. 30 kHz also under spinning conditions. Furthermore, the agreement with the averaged parameters calculated by DFT methods is good. The principal components of the motionally-averaged proton shielding tensor measured experimentally in this study are summarized in Table 2.

The contour plots resulting from *both* pure-shift 2D NMR techniques adopted in this study also carry information about the relative orientation between the shielding and dipolar tensors. This feature can be easily appreciated in Figure 5, where numerical simulations of the Antonijevic-Wimperis and 2D-Amplified PASS experiments are shown for the three cases of Ω_{PC}^{D} = $(0^{\circ}, 0^{\circ}, 0^{\circ}), (0^{\circ}, 45^{\circ}, 0^{\circ})$ and $(0^{\circ}, 90^{\circ}, 0^{\circ})$. These simulations clearly show that i) the 2D correlation lineshapes depend on the relative orientation between the two interactions and ii) the F₁ projections are instead independent with respect to this feature. Although a thorough analysis

of the 2D correlation lineshapes has not been undertaken in this study, the comparison between our experimental evidence and numerical simulations performed with the SIMPSON code seems to suggest that the Euler angle which relates the main *z*-axes of dipolar and shielding tensors is close to zero. This is not consistent with an averaging motion given by 2-fold flips about the H–O–H bisector, which would instead yield $\beta = 90^{\circ}$. Further studies may be required to investigate and interpret this finding in terms of types of motion of water molecules in Ba(ClO₃)₂·H₂O.

Conclusions

The chemical shift tensors of the protons of isolated water molecules trapped in barium chlorate monohydrate Ba(ClO₃)₂·H₂O have been characterized by solid-state NMR under both spinning and static conditions. Two-dimensional techniques allow the measurement of the chemical shift anisotropy by selectively refocusing the predominant dipolar interaction from the indirect dimension. This produces a pure-shift indirect dimension in the resulting 2D spectrum from which the main components of the interaction tensor can be easily determined. By means of these techniques one can importantly reduce the number of parameters to consider when analysing the lineshapes. In particular, the relative orientation between the shielding and dipolar tensors can be neglected. The measured values of the anisotropy and asymmetry agree well with the motionally-averaged values calculated for a periodic system with planewave pseudopotential DFT methods, as implemented in the CASTEP code.

Experimental and computational details

Ba(ClO₃)₂·H₂O was purchased from Alfa Aesar and kept under vacuum overnight to minimize the adsorption of water which leads to a misleading sharp isotropic peak in the middle of the Pake pattern. All 1D ¹H MAS and 2D static spectra were recorded at T = 300 K on a Bruker 400 spectrometer (9.4 T) with an Avance-II console, using 3.2 mm rotors in a triple-resonance probe designed for low-temperature DNP measurements. The *rf*-field amplitude was $v_1 = 125$

kHz, corresponding to 90° pulses τ_p = 2 µs. The 2D spinning spectrum was recorded at room temperature on a Bruker 600 spectrometer (14.1 T) equipped with an AVANCE-III console and a widebore 4 mm probe. The *rf*-field amplitude was $v_1 = 100$ kHz, corresponding to 90° pulses $\tau_p =$ 2.5 µs. The proton chemical shifts were referenced to adamantane at 1.8 ppm. Spinning frequencies of v_{rot} = 3.333 or 10 kHz were used. Typical recycle intervals varied between 3 and 20 s. All 1D 90°- τ -90°- τ solid echoes were phase-cycled for the selection of the $p = 0 \rightarrow p = +1 \rightarrow p$ p = -1 pathway by nesting Exorcycle [32] with CYCLOPS [33] so to produce an overall 16-step phase cycle.[18] The refocusing delay was $\tau = 50 \ \mu s$ and 100 μs ($v_{rot} = \tau_{rot}^{-1} = 10 \ \text{kHz}$) for static and spinning 1D spectra, respectively. The 2D static pure-shift spectrum of Fig 3a was acquired by averaging 256 transients for each of 50 t_1 increments of 10 μ s, τ = 180 μ s and a recycling delay of 25 s. The 2D-Amplified PASS experiment of Fig. 4a resulted from the averaging of 14 transients for each of 32 t_1 increments, a spinning rate v_{rot} = 3. 333 kHz, an amplification factor N = 3.333 and a recycle delay of 3 s. For this latter experiment, cogwheel phase cycling [34] was used to minimize the number of transients required. Numerical simulations and fits were performed with SIMPSON, [35] using 2000 crystal orientations sampled with the REPULSION scheme [36] for static spectra, whereas 323 orientations were sampled with the CZW scheme [37-39] over 12 γ -angles for spinning spectra. The *rms* plotted in Figures 3c and 4c is that output by SIMPSON, as described in the manual.

Calculations of total energies and NMR parameters were carried out using the CASTEP DFT code (version 6),[22] employing the gauge-including projector augmented wave (GIPAW) [40] algorithm to reconstruct the all-electron wave function in the presence of a magnetic field. Calculations were performed using the GGA PBE functional [41] and core-valence interactions were described by ultrasoft pseudopotentials.[42] A planewave energy cutoff of 60 Ry was used, and integrals over the Brillouin zone were performed using a *k*-point spacing of 0.04 Å⁻¹. All calculations were converged as far as possible with respect to both *k*-point spacing and cutoff energy. Calculations were performed on a 198-node (2376 core) Intel Westmere cluster with 2 GB memory per core and QDR Infiniband interconnect at the University of St Andrews. The

reduced shielding anisotropy Δ_{CS} as used in this work is obtained by multiplying the full shielding tensor as calculated with CASTEP by the factor 2/3.

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References

- [1] G. Hura, D. Russo, R.M. Glaeser, T. Head-Gordon, M. Krack and M. Parrinello, *Phys. Chem. Chem. Phys.* 2003, 5, 1981.
- [2] P. Postorino, R.H. Tromp, M-A. Ricci, A.K. Soper and G.W. Neilson, *Nature* 1993, 366, 668.
- [3] H.E. Stanley and J. Texeira, J. Chem. Phys. 1980, 73(7), 3404.
- [4] C.R. Bowers and D.P. Weitekamp, J. Am. Chem. Soc. 1987, 109, 5541.
- [5] H. Eisendarth, W. Stone and J. Jeener, *Phys. Rev. B* 1978, 17, 47.
- [6] T. Kravchuk, M. Reznikov, P. Tichonov, N. Avidor, Y. Meir, A. Bekkerman and G. Alexandrowicz, *Science* 2011 331, 319.
- [7] C. Beduz, M. Carravetta, J. Y.-C. Chen, M. Concistrè, M. Denning, M. Frunzi, A.J. Horsewill, O.G. Johannessen, R. Lawler, X. Lei, M.H. Levitt, Y. Li, S. Mamone, Y. Murata, U. Nagel, T. Nishida, J. Ollivier, S. Rols, T. Rõõm, R. Sarkar, N.J. Turro and Y. Yang, *Proc. Natl. Acad. Sci.* 2012, 109, 32, 12894.
- [8] M.J. Duer, *Solid-State NMR Spectroscopy: Principles and Applications*, Blackwell Science, UK, 2002.
- [9] E. R. Andrew, A. Bradbury and R. G. Eades, *Nature* 1958, 182, 1659.
- [10] I. Lowe, Phys. Rev. Lett. 1959, 22, 133.

- [11] C. Bonhomme, C. Gervais, F. Babonneau, C. Coelho, F. Pourpoint, T. Azaïs, S.E. Ashbrook, J.M. Griffin, J.R. Yates, F. Mauri and C.J. Pickard, *Chem. Rev.* 2012, 112, 5733.
- [12] T. Charpentier, Solid State Nucl. Magn. Reson. 2011, 40, 1, 1.
- [13] M.R. Mitchell, D. Carnevale, R. Orr, K.R. Whittle and S.E. Ashbrook, J. Phys. Chem. C 2012, 116, 4273.
- [14] M. Castro, V.R. Seymour, D. Carnevale, J.M. Griffin, S.E. Ashbrook, P.A. Wright, D.C. Apperley, J.E. Parker, S.P. Thompson, A. Fecant and N. Bats, *J. Phys. Chem. C* 2010, 114, 12698.
- [15] D. Carnevale, V. del Amo, D. Philp and S.E. Ashbrook, *Tetrahedron* 2010, 66, 6238.
- [16] K. Modig and B. Halle, J. Am. Chem. Soc. 2002, 124, 12031.
- [17] P. Tekely, P. Palmas and P. Mutzenhardt, J. Magn. Reson. 1997, 127, 238.
- [18] S. Antonijevic and S. Wimperis, J. Magn. Reson. 2003, 164, 343.
- [19] G.E. Pake, J. Chem. Phys. 1948, 16, 4, 327.
- [20] J.R. Long, R. Ebelhäuser and R.G. Griffin, J. Phys. Chem. A 1997, 101, 988.
- [21] S. Antonijevic and S. Wimperis, J. Chem. Phys. 2004, 121, 22, 1.
- [22] M.D. Segall, P.J.D. Lindan, M.J. Probert, C.J.Pickard, P.J. Hasnip, S.J. Clark, M.C. Payne, J. Phys.: Condens. Matter 2002, 14, 2717.
- [23] S.K. Sikka, S.N. Momin, H. Rajagopal and R. Chidambaram, J. Chem. Phys. 1968, 48, 5, 1883.
- [24] S.P. Brown and S. Wimperis, J. Magn. Reson. 1997, 124, 279.
- [25] R.M. Orr, M.J. Duer and S.E. Ashbrook, J. Magn. Reson. 2005, 174, 301.
- [26] R.M. Orr and M.J. Duer, Solid State Nucl. Magn. Reson. 2006, 30, 1.
- [27] C. Crockford, H. Geen and J.J. Titman, Chem. Phys. Lett. 2001, 344, 367.
- [28] L. Shao, C. Crockford, H. Geen, G. Grasso and J.J. Titman, J. Magn. Reson. 2004, 167, 75.
- [29] O.N. Antzutkin, S.C. Shekar and M.H. Levitt, J. Magn. Reson. 1995, 115, 7.
- [30] W.T. Dixon, J. Magn. Reson. 1981, 44, 220.
- [31] W.T. Dixon, J. Chem. Phys. 1982, 77, 1800.
- [32] G. Bodenhausen, R. Freeman and D.L. Turner, J. Magn. Reson. 1977, 27, 511.

- [33] D.I. Hoult and R.E. Richards, Proc. Roy. Soc. (London) 1975, A344, 311.
- [34] M.H. Levitt, P.K. Madhu and C.E. Hughes, J. Magn. Reson. 2002, 155, 300.
- [35] M. Bak, J.T. Rasmussen and N.C. Nielsen, J. Magn. Reson. 2000, 147, 296.
- [36] M. Bak and N.C. Nielsen, J. Magn. Reson. 1997, 125, 132.
- [37] S.K. Zaremba, Ann. Mat. Pure. Appl. 1966, 293, 4.
- [38] H. Conroy, J. Chem. Phys. 1967, 47, 5307.
- [39] V.B. Cheng, H.H. Suzukawa Jr. and M. Wolfsberg, J. Chem. Phys. 1973, 59, 3992.
- [40] C.J. Pickard and F. Mauri, *Phys. Rev. B* 2001, 63, 245101.
- [41] J.P. Perdew, K. Burke and M. Ernzerhof, Phys. Rev. Lett. 1996, 77, 3865.
- [42] J.R. Yates, C.J. Pickard and F. Mauri, *Phys. Rev. B* 2007, 76, 024401.

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Solid-state NMR measurements and DFT calculations of the magnetic shielding tensors of protons of water in barium chlorate monohydrate

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Tables

Table 1 Reduced chemical shift anisotropy $\Delta_{CS} = \delta_{zz} - \delta_{iso}$, with isotropic shift $\delta_{iso} = (\delta_{xx} + \delta_{yy} + \delta_{zz})/3$, and asymmetry $\eta_{CS} = (\delta_{xx} - \delta_{yy})/\Delta_{CS}$ for the two proton sites in barium chlorate monohydrate as calculated with DFT methods as implemented in CASTEP in a periodic planewave-pseudopotential formalism. Calculations have been performed without and with geometry optimization (Opt). In this latter case the crystallographic degeneracy of two of the four water molecules is lifted. The possibilities of fixing the unit-cell size (Fix) and imposing conservation of symmetry (Sym) have also been considered.

	Site 1		Site 2	
	Δ_{cs} (ppm)	η _{cs} (ppm)	$\Delta_{ extsf{cs}}$ (ppm)	η _{cs}
No Opt	-16.69	0.15	-	-
Opt	-16.27	0.17	-16.19	0.17
Opt / Fix	-16.47	0.17	-16.48	0.17
Opt / Sym	-16.29	0.17	-16.20	0.17
Opt / Fix / Sym	-16.45	0.17	-16.45	0.17
Motionally averaged	-8.7	0.9	-	-

Table 2Experimental values of anisotropy Δ_{cs} and asymmetry η_{cs} as measured with 1D static and
spinning solid echoes and 2D static and spinning pure-shift methods.

NMR method		eta	Δ_{cs} (ppm)	η_{cs}
1D solid echo	static	0°	-10 ± 3	1 ± 0.7
		90°	11 ± 3	0.3 ± 0.5
	spinning	0°	-9±7	0 ± 11
		90°	9 ± 10	1 ± 2
2D Pure-shift	static	-	-10.5 ± 1.0	0.7 ± 0.2
	spinning	-	-8.5 ± 0.6	1 ± 0.1

Figure captions

- Figure 1 a) Experimental ¹H NMR spectrum of a static powder of Ba(ClO₃)₂·H₂O acquired with a solid echo on a 9.4 T magnet. b) Numerical simulations of Pake patterns resulting from two protons subject to a shielding anisotropy $\Delta_{CS} = 4$ kHz, $\eta_{CS} = 0$, 0.5 and 1.0, and a dipolar coupling constant d = -29 kHz, with relative orientations of the two tensors $\beta = 0$ and 90°, in black and red, respectively. c) Experimental ¹H NMR spectrum acquired with a rotor-synchronized solid echo on a 9.4 T magnet at $v_{Rot} = 10$ kHz. d) Simulations of the spinning sideband pattern of c) when $\beta = 0$ and 90°, in black and red, respectively. In all cases, the carrier frequency was set to coincide with the isotropic peak (*), i.e., $v_{rf} = \delta_{lso} = 1.68$ kHz = 4.2 ppm.
- Figure 2 Unit cell of Ba(ClO₃)₂·H₂O viewed down the x-, y- and z-axis is shown in a), b) and c), respectively. Oxygen, proton, barium and chlorine atoms are shown in red, white, blue and green, respectively. d) Representation of the magnetic shielding tensors of proton nuclei as light-brown ellipsoids.
- Figure 3 a) Experimental two-dimensional correlation between a dipole-shift dimension (horizontally in F₂) and a pure-shift dimension (vertically in F₁) of barium chlorate monohydrate as obtained with the experiment proposed by Antonijevic and Wimperis at 9.4 T.[21] b) F₁ projection of the spectrum (black) in a) with a fitted simulation (red). c) Two-dimensional contour plot showing the *rms* resulting from a systematic fit of the F₁ projection of the 2D spectrum in a) over the subspace spanned by the parameters Δ_{CS} and η_{CS} . The intensity scale has been arbitrarily limited to 50.
- Figure 4 a) Experimental two-dimensional correlation between a dipole and shift dimension (F_2) and an amplified pure-shift dimension (F_1) of barium chlorate monohydrate as obtained with the CSA-Amplified PASS experiment at 14.1 T.[25] b) F_1 projection of the spectrum (black) in a) with a fit (red). c) Two-dimensional contour plot showing the *rms* resulting from a systematic fit of the F_1 projection of the 2D spectrum in a) over the subspace spanned by the Δ_{cs} and η_{cs} parameters. The intensity scale has been arbitrarily limited to 50.

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Figure 5 Numerical simulations of the 2D correlation experiment shown in Fig. 3a with $\beta = 0$, 45 and 90°, in a), b) and c), respectively. Numerical simulations of the 2D correlation experiment shown in Fig. 4a with $\beta = 0$, 45 and 90°, in d), e) and f), respectively. Realistic pulses have been assumed in all cases and coherence selection was also taken into account. The isotropic shift was $\delta_{iso} = 0$ kHz whereas the anisotropy Δ_{cs} and asymmetry η_{cs} were those in Table 2 for the corresponding experiments. All simulations assumed an external magnetic field $B_0 = 9.4$ T and a dipolar coupling constant d = -29 kHz.

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Figures

Figure 1







Solid-state NMR measurements and DFT calculations of the magnetic shielding tensors of protons of water in barium chlorate monohydrate

Figure 3

a)

b)

c)









 Δ_{CS} (ppm)

Solid-state NMR measurements and DFT calculations of the magnetic shielding tensors of protons of water in barium chlorate monohydrate

Figure 5

