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ARTICLE TYPE

Stereoselective Synthesis of (E)-3,3-diaryl and (E)-3-aryl-3-aryloxy Allylamines and Allylalcohols from trans-Cinnamyl chloride and alcohol

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In this work we describe the regio and stereoselective synthesis of (E)-3,3-diaryl and (E)-3-aryl-3-Aryloxy allylamines and allylalcohols. The starting materials are the non-expensive commercially available cinnamyl alcohol and chloride. Bromination/dehydrobromination sequence furnished the (E)-3bromo-3-phenylallyl amines and alcohol as single regioisomers and high stereoselectivity (>98%). These 10 vinyl bromides were used as substrate in cross-coupling reactions furnishing the arylated products with good to excellent yields and total E-configuration retention. With this protocol, we were able to produce regio and stereospecifically trisubstituted olefins and vinyl ethers by Suzuki cross-coupling and Ullmann vinylation.

Introduction

15 Regio and stereocontrol in the production of multi-substituted unsaturated compounds remains an important challenge in organic synthesis.¹⁻⁴ Particularly, the controlled insertion of aryl and aryloxy fragments into olefin scaffolds is an important task for the synthesis of bioactive compounds, such as Z-tamoxifene,², ₂₀ ⁵⁻⁷ sertraline, ⁸ indatraline, ⁸ and tolterodine, ⁹ among others. ¹⁰⁻¹²

1,1-Diaryl trisubstituted alkenes, specifically, are usually synthesised by mono or di Heck arylation of cinnamates, $^{13\text{-}16}$ α , β unsaturated adehydes¹⁷ and acrylonitriles^{18, 19} However, these methodologies contain some drawbacks. For instance, the 25 presence of an unsaturated electron-withdrawing group conjugated to the olefin is imperative to drive the regioselectivity for the formation of the 1,1-diaryl product. In addition, and depending on the conditions, stereoisomeric mixtures are obtained. An example of lack of regio and stereoselectivity 30 caused by the absence of conjugation at double bond is the Heck arylation of cinnamic alcohol, where mixtures of regioisomers and double bound isomerisation are obtained. 20 Allyl alcohols can be also obtained by hydrogenation²¹ or transfer hydrogenation²²

of the corresponding α,β -unsaturated aldehyde. Considering 35 allylamines, to ensure regioselectivity in Heck arylation, is necessary the presence of an electron withdrawing group bounded to the nitrogen. 23, 24

3-Arylated allyl amines such as abamine-SG, abamine and naftifine are bioactive compounds.²³ In our ongoing research, we 40 need to obtain selective protocols for the synthesis of 3,3diarylated allyl amines and alcohols. In this context, we wish to describe here a highly regio and stereoselective approach to the synthesis of (E)-3,3-diaryl allyl amines and alcohols and (E)-3aryl-3-aryloxy allyl amines and alcohols. Our strategy is based on 45 the regio and stereoselective formation of vinyl bromides 5 from

commercially available and inexpensive starting materials (trans-

cinnamyl alcohol and trans-cinnamyl chloride). This strategy allowed us to synthesise the target trisubstituted-olefins via Suzuki and C-O cross-coupling reactions, in both cases as single 50 regioisomers with *E:Z* ratios greater than 98:2.

Initially, trans-cinnamyl chloride was allowed to react with

Results and Discussion

diisopropylamine and N,N-methylbenzylamine leading to the (E)-N,N-dialkyl-3-phenylprop-2-en-1-amines 3a and 3b in yields of 55 86% and 88%, respectively (Scheme 1). The products 3a and 3b were then protonated and submitted to bromination in CH₂Cl₂ at 0°C affording the antiperiplanar dibrominated products 4a and 4b as crystalline solids in good yields (81 and 77%, respectively). It must be mentioned that the protonation is imperative since 60 bromination of the free amine led to a complex mixture of products. Cinnamic alcohol was also brominated in the same conditions affording the intermediate 4c at a yield of 86%. In order to establish the best dehydrobromination conditions, a range of solvent, base and reaction temperature combinations 65 were tested to achieve high regio and stereoselectivity (Table 1 of Supporting Information). As a result, by using KOH, THF and 35°C, the vinyl bromides 5a and 5b were formed as single regiosomers in 88% and 84% yields, respectively, with stereoselectivity greater than 98%. It must be mentioned that the 70 work-up was very simple since no column chromatography was necessary. A simple filtration and solvent evaporation furnished the desired products with high purity (>99% as judged by GC and ¹H-NMR). Concerning the vinyl bromide **5c** obtained from the cinnamic alcohol, the regioselectivity was slightly lower since a 75 mixture 90:10 of regioisomers was obtained at the end of the reaction. In this case, chromatographic separation of regioisomers was necessary, and after that, vinylic compound 5c was obtained at a yield of 78% and with a E:Z ratio of 98:2. The stereochemistry E of all vinyl bromides was ascribed through

NOE experiments. Correlations between aromatic and allylic CH₂ hydrogens were observed for the olefins **5a-c** (see S.I.).

Scheme 1: Regio and stereoselective synthesis of vinyl bromides 5a, 5b and 5c.

In previous works, we have used the dehydrobromination of (1,2dibromomoethyl)benzene in order to obtain the α-bromostyrene with regioselectivities up to 95%. 25, 26 Here we have used the same simple KOH-promoted dehydrobromination protocol and 10 were delighted to see that for the dibromides 4a and 4b, only the α -bromo regioisomers were observed and that for 4c, a 90:10 regiosomeric ratio was obtained. It is important to mention that the dehydrobromination reaction does not require high temperatures, microwave irradiation, ionic liquid as media or 15 metal additives. 27, 28 We believe that the reaction is facilitated due to an anchimeric assistance of amine or alcohol group in the transition state that removes the benzylic proton and provides the 3-phenyl-3-bromo regioisomer (Path A in Figure 1). On the other hand, in the transition state that provides the opposite regiosomer 20 (3-phenyl-2-bromo product), this assistance is not possible (Path B); therefore, the Path A and consequently 3-phenyl-2-bromo allylamines and allyl alcohol 5 are favored. The high stereoselectivity obtained for the dehydrobromination product $(E/Z = \ge 98\%)$ is related to the strong preference for the anti-25 elimination process. 6, 29

Figure 1. Transition states that provide the major regioisomer (Path A) and minor regioisomer (Path B).

With vinyl bromides 5 in hands, we evaluated the Suzuki 30 cross-coupling, starting with the conditions already established for (*E*)-bromostilbene. 6, 30 The optimisation was

carried out at room temperature for one hour using substrate 5a and 2-methoxy-5-methylpheylboronic acid as coupling partners (Table 1). Initially, a phosphine-free system was 35 tested (entry 1), giving a high conversion but with a considerable amount of vinyl bromide reduction as a byproduct. In a sequence, systems based on Pd(OAc)₂/P(o-Tol)₃ and Pd(OAc)₂/BINAP were applied, leading to yields of 47% and 48%, respectively. The systems that achieved the best 40 results were those based on PPh3: 67% yield with PdCl₂(PPh₃)₂ and 73% yield with Pd(OAc)₂/PPh₃; these were the same conditions described by us as being the best for the coupling between (E)-bromostilbene and arylboronic acids. 6, 30 Finally, in order to obtain complete conversion, the reaction 45 time was increased to 2.5 h, giving the product with 98% GC yield, 93% isolated yield and with complete retention of the double bond configuration. Again, the stereochemistry of the product was determined by NOE, where a correlation between the olefinic hydrogen and the aryl ortho-methyl hydrogen was 50 observed (see S.I.). It must to be mentioned that these conditions are very practical and convenient since triphenylphosphine is a cheap phosphine and only 0.5 mol% of Pd(OAc)₂ is necessary for the complete conversion at room

scope of the reaction was tested. In this way, the reaction was performed with **5a**, **5b** and **5c** and a plethora of arylboronic acids (Table 2). Good to excellent results were observed, considering that yields ranging from 71 to 94% were attained with the products being obtained with complete (*E*)-stereochemistry retention. The best yields were observed with the electron-rich 4-methoxyphenylboronic acid (entries 1 and 9). Moreover, the reaction proceeds very well with both deactivated (entry 3) and steric demanding arylboronic acids (entry 5); however, in these cases a larger excess of arylboronic acid is needed to afford the complete conversion of vinyl bromide.

10

Table 1: Condition optimisation for Suzuki cross-coupling of vinyl bromide ${\bf 5a.}^a$

Conv. Yield^t Entry Catalyst Reduction (%) (%) (%) 1 Pd(OAc)₂ 70 42 28 Pd(OAc)2/P(o-Tol)3 10 2 62 47 3 Pd(OAc)2/BINAP 48 3 53 4 PdCl₂(PPh₃)₂ 70 67 3 75 2 5 Pd(OAc)₂/PPh₃ 73 Pd(OAc)₂/PPh₃ 100 98 (93)

a) Reaction Conditions: **5a** (0.5 mmol), arylboronic acid (0.6 mmol), KOH (1.0 mmol), THF (4 mL), [Pd] (0.5 mol %), phosphine (1 mol %), 25°C, 1 h. Internal standard: tetradecane. By-products detected: vinyl bromide reduction and arylboronic acid homocoupling. For isolated yield a 2.0 mmol scale were used. b) GC yields c) 2.5h.

Table 2: Stereoselective synthesis of 3,3-diaryl allylamines and allyl alcohols via Suzuki cross-coupling of 5a-c.

Br
Ph
$$CH_2X$$
 $Pd(OAc)_2/PPh_3$, $ArB(OH)_2$ Ph CH_2X
5a-c $r.t.$, 2.5h Ph CH_2X
5b: $X = N(i - Prop)_2$

5c: X = OH Yield Yield^t Entry Entry Product Entry Product Product (%) (%)MeO₂S MeO Me Me 7a 94 11a 7c 1 84 Me N(i-Pr)₂ N*(i*-Pr)₂ OH MeO₂S MeOC 2 12a 10 8c 8a 81 $N(i-Pr)_2$ N(i-Pr)2 MeO₂S 3° 9c 9a 83 6b 74 11 N(i-Pr)2 Ph N(Me)Bn MeO 10a 92 93 6c

a) Reaction Conditions: 5 (2.0 mmol), arylboronic acid (2.4 mmol), KOH (4.0 mmol), THF (16 mL), Pd(OAc)₂ (0,5 mol %), PPh₃ (1 mol %), 25°C, 2.5 h. b) Isolated yields. c) 3 mmol of arylboronic acid was necessary to achieve complete vinyl bromide conversion.

Ph

- 25 Beside Suzuki cross-coupling, the Ullmann reaction of phenols was also applied to the vinyl bromides 5b and 5c in order to obtain novel (E)-trisubstituted vinyl-ethers regio and stereospecifically. We applied the same conditions as described for Ullmann reaction of (E)-bromostilbene with phenols and
- 30 azole³¹ and the results obtained are described in Table 3. Initially, the reaction was performed with the electron rich 4-methoxyphenol and the substrate **5b**; thus, the product **8b** was obtained with 86% yield (Entry 1). After this, we tested the C-O coupling with a high steric demanding *o*-cresol and the result was

 $N(i-Pr)_2$

Ph

not affected, since the same yield was observed (Entry 2). Lower yields are usually obtained for the Ullmann coupling of aryl or vinyl halides with phenols containing electron-withdrawing groups. 32, 33 Indeed, when the reaction was performed with the s less nucleophilic 4'-hidroxyacetophenone, an important decrease in yield was observed for both substrates **5b** and **5c**. It is important to mention that similarly to Suzuki coupling, the products were obtained with total *E*-configuration retention.

Table 3: Stereospecific Synthesis of trisubstituted vinyl-ethers by Ullmann Coupling of 5b and 5c with substituted phenols.

Br
$$R^1$$
 R^1 R

Entry	R^1	Product		Yield (%)
1	4-OMe	MeO——O Ph——N(Me)Bn	8b	86
2	2-Me	Me O Ph N(Me)Bn	9b	86
3	4-COMe	MeOC — O — N(Me)Bn	10b	43
4	4-COMe	MeOC — O Ph — OH	10c	51
5	4-OMe	MeO————————————————————————————————————	11c	74

Reaction Conditions: 5 (2.0 mmol), phenol (3.0 mmol), K₃PO₄ (4.0 mmol), CuI (10 mol%), phenanthroline (10 mol%), dioxane (8 ml).

15 Conclusions

In summary, we described herein the regio and stereoselective synthesis of vinyl bromides derived from cinnamic alcohol and cinnamic amines. These bromides were used as the substrate for Suzuki cross-coupling and Ullmann vinylation, furnishing coupling products with good to excellent yields and total *E*-configuration retention. This strategy represents an innovative method for the stereo-controlled synthesis of (*E*)-3,3-diaryl allylamines and allylalcohols since no double bound conjugation is needed in the substrate to attain high regioselectivity. Moreover, by using this approach, it is also possible to access novel vinyl ethers with complete regio and stereocontrol.

Experimental

General Information

Allylic substitution, bromination and dehydrobromination reactions were performed in air atmosphere, using solvents and reagents as received. The catalytic Suzuki cross-couplings and copper-catalyzed Ullmann vinylation were performed under argon atmosphere. For these reactions THF and dioxane were distilled over Na/benzophenone and the MeOH was degassed before the use. NMR spectra were recorded on a Varian XL300 spectrometer and in a Bruker Avance 400 MHZ. IR spectra were recorded on a Shimadzu FTIR-8300 spectrometer. Mass spectra were obtained on a GC/MS Shimadzu QP-5050 (EI, 70 eV). Gas chromatography analyses were performed on a Shimadzu equipament equipped with a 30 meter DB-5 column and FID detector. ESI-(+) HRMS analyses were performed on a Q-Tof (Micromass) mass spectrometer.

Synthesis of the amines 3a and 3b

⁴⁵ A mixture of 10 g (66.0 mmol) of cinamyl chloride, 262 mmol of diisopropylamine (or *N,N*-benzylmethylamine) 36.2 g (262 mmol) of K₂CO₃, 120 mL of toluene and 120 mL of tert-butanol were stirred for 24h at 90 °C . The cinnamyl chloride consumption was monitored by GC. After, the reaction was filtered and the solvent was removed. The product was dissolved in Et₂O and extracted with HCl 1.0 M. The aqueous phase was neutralized with NaOH 1.0 M, extracted with Et₂O, washed with brine and dried over MgSO₄. The solvent was removed to give the corresponding (*E*)-N,N-dialkyl-3-phenylprop-2-en-1-amine as yellow oils.

(*E*)-*N*,*N*-diisopropyl-3-phenylprop-2-en-1-amine (3a): yellow oil, yield: 88%. IR (neat): \vee 3025, 2962, 2930, 1600, 1448, 1380, 1170, 963, 689 cm⁻¹. MS (IE 70 eV): m/z (relat. int.): 217 (18, M⁺), 202 (56), 174 (5), 144 (6), 126 (11), 117 (100), 115 (81), 91 (60), 77 (19). ¹H-NMR (300 MHz, CDCl₃) δ ppm 0.95 (d, *J* = 6.6 Hz, 12 H), 3.00 (hept, *J* = 6.6 Hz, 2 H), 3.18 (dd, *J* = 4.6 Hz, 1.5 Hz, 2H), 6.15 (m, 1H), 6.48 (d, *J* = 15.8 Hz, 1H), 7.06-7.32 (m, 5 H). ¹³C-NMR (75 MHz, CDCl₃) δ ppm 21.4; 48.3; 48.9; 126.8; 127.6; 129.1; 130.6; 132.4; 138.2. HRMS: m/z calcd for C₁₅H₂₄N (65 (M + H⁺): 218.1909; found: 218.1899.

(E)-N-benzyl-N-methyl-3-phenylprop-2-en-1-amine yellow oil, yield: 86%. IR (atr) v 3059, 3026, 2784, 1599, 1494, 1451, 1364, 1022, 967, 736 cm⁻¹. MS (IE 70 eV): m/z (relat. int.): 237 (14), 236 (10), 167 (4), 146 (100), 134 (7), 117 (41), 91 (77). Th-NMR (300 MHz, CDCl₃) δ ppm 2.0 (s, 3H), 3.25 (d, J=6.5 Hz, 2H) 3.67 (s, 2H), 6.37 (m, 1H), 6.60 (d, J=15.9 Hz, 1H), 7.20-7.56 (m, 10H). 13 C-NMR (75 MHz, CDCl₃) δ ppm 42.2; 59.8; 61.8; 126.2; 126.9; 127.3; 127.5; 128.2; 128.5; 129.0; 132.5; 137.0; 138.9. HRMS: m/z calcd for $C_{17}H_{20}N$ (M + H⁺): 75 238.1596; found: 238.1588.

Bromination of 3a and 3b

A mixture of HBr 1.0 M (50 mL) and 50 mmol of **3** was stirred for 10 min. After, the water was evaporated; the product was 80 dissolved in 50 mL of CH₂Cl₂, dried over MgSO₄ and filtered. Then, a solution of Br₂ (3.1 mL, 60 mmol) in CH₂Cl₂ (30 mL) was added dropwise at 0°C during 30 min and allowed to attain room temperature. The reaction mixture was stirred overnight in

the dark. After, the bromine color of the mixture was removed by addition an excess of aq NaHSO₃ 10% and the pH was raised to 9 by addition of KOH 10%. The organic layer was then dried (MgSO₄) and the solvent was evaporated to give the crude 5 products. The products were crystallized in ethanol to give the 2,3-dibromo-*N*,*N*-dialkyl-3-phenylpropan-1-amines.

2,3-dibromo-*N*,*N*-**diisopropyl-3-phenylpropan-1-amine (4a)**: White solid, Yield: 81%, IR (atr) v 3031, 2972, 2956, 2926, 1494, 1454, 1390, 1209, 1133, 921, 765 cm⁻¹. ¹H-NMR (300 mHz, CDCl₃) δ ppm 0,99 (dd, J = 14.9; 6.6 Hz, 12H), 2.79 (dd, J = 14.9; 9.0 Hz, 1H), 3.04 (hep, J = 6.6 Hz, 2H), 3.41 (dd, J = 14.9; 3.4 Hz, 1H), 4.47 (dt, J = 9.0; 3.4 Hz, 1H), 5.10 (d, J = 9.0 Hz, 1H), 7.20-7.41 (m, 5H). ¹³C-NMR (75 MHz, CDCl₃) δ ppm 20.5; 48.7; 52.6; 55.5; 59.9; 128.1; 128.4; 128.6; 140.0. HRMS: δ m/z calcd for $C_{15}H_{24}Br_{2}N$ (M + H⁺): 376.0275; found: 376.0281.

N-benzyl-2,3-dibromo-N-methyl-3-phenylpropan-1-amine (4b): White crystals, Yield: 77%. IR (atr) ν 2959, 2927, 1495, 1456, 1213, 1135, 691 cm⁻¹. ¹H-NMR (300 MHz, CDCl₃) δ ppm 2.37 (s, 3H), 3.03 (dd, J = 13.9, 7.3 Hz, 1H), 3.20 (dd, J = 13.9, 20 4.8 Hz, 1H), 3.70 (m, AB system, 2H), 4.67 (ddd, J = 8.5, 7.3, 4.8 Hz, 1H) 5.35 (d, J = 8.5 Hz, 1H) 7.25 – 7.51 (m, 12H). ¹³C-NMR (75 MHz, CDCl₃) δ ppm 42.7; 54.4; 56.7; 62.6; 62.7; 127.2; 128.2; 128.3; 129.0; 138.4; 139.1. HRMS: m/z calcd for $C_{17}H_{20}Br_2N$ (M + H⁺): 397.9942; found: 397.9969.

2,3-dibromo-3-fenilpropan-1-ol(4c)

A solution of cinnamic alcohol (30 mmol) in 100 mL of CH₂Cl₂ was cooled to 0°C in ice bath and kept in the dark. To this solution a mixture of 1.9 mL (36 mmol) of Br₂ in 30 mL of ³⁰ CH₂Cl₂ was added dropwise during 30 min. After, the reaction was allowed to attain room temperature and the mixture was stirred during 16h. After, a solution of NaHSO₃ 10% was added and the organic layer was separated, washed with brine, dried over MgSO₄, filtered and the CH₂Cl₂ was evaporated to give the ³⁵ crude product as a yellow solid. The product was recrystallized from a hot solution of hexane/toluene 1:1 furnishing white crystals. The crystals were then washed with cold hexane and dried under vacuum.

2,3-dibromo-3-phenylpropan-1-ol (4c)³⁴: White crystals, Yield: 86%, IV (atr) ν 3360, 3028, 2925, 2870, 1493, 1452, 1231, 1135, 1051, 762, 692, 624, 578 cm⁻¹. MS (EI 70 eV): m/z (relat. Int.): 295 (2), 293 (4), 291 (2), 215 (68), 213 (82), 185 (99), 183 (100), 133 (57), 105 (91), 91 (87). ¹H-NMR (300 MHz, CDCl₃) δ ppm 2.17 (t, J = 6.74, 1H), 4.21-4.36 (m, 2H), 4.70 (ddd, J = 45 11.1, 4.3, 2.7 Hz, 1H), 5.27 (d, J = 11.1 Hz, 1H), 7.18-7.56 (m, 5H). ¹³C-NMR (75 MHz, CDCl₃) δ ppm 52.2; 59.3; 65.8; 127.8; 128.7; 128.9; 139.8.

Dehydrobomination reactions of 4a-4c

50 A mixture of dibrominated prduct **4** (5,0 mmol), KOH (10,0 mmol) and THF (25 mL) was stirred at 35 °C for 48 h. The mixture was then filtered and the THF was evaporated. For the substrates **4a** and **4b** the mixture was dissolved in hexane, in order to precipitate inorganic salts and filtered again. Finally, the 55 hexane was evaporated to give the corresponding (*E*)-3-bromo-*N*,*N*-dialkyl-3-phenylprop-2-en-1-amines **5a** and **5b** as pale yellow oils. It is important to mention that single regiosomers and almost single diasteroisomer (*E*:*Z* ≥ 98%) were obtained. For **4c**,

after the end of the reaction, the mixture was filtered, the THF was evaporated and the crude product **5c** was analysed by NMR. It was observed a regioisomeric mixture of 9:1. The regioisomers were then separated by column chromatography in silica, with a hexane/ethyl acetate eluent. The (*E*)-configuration of the products **5a-c** were determined by 1D- and 2D-NOESY experiments (See

(E)-3-bromo-N,N-diisopropyl-3-phenylprop-2-en-1-amine

(5a): pale yellow oil, yield: 88%. IR (atr) v 3053, 2964, 2929, 1460, 1363, 1178, 844, 759, 696 cm⁻¹. MS (EI 70 eV): m/z (int. 70 relat.): 297 (4, M+2), 295 (4,M+), 282 (24), 280 (25), 197 (36), 195 (40), 144 (6), 115 (100), 102 (21), 70 (20), 56 (40). ¹H-NMR (300 MHz, CDCl₃) δ ppm 0.93 (d, 12 H, J = 6.5 Hz), 3.00 (hept, 2 H, J = 6.5 Hz), 3.08 (d, 2H, J = 6.7 Hz), 6.29 (t, 1H, J = 6.7 Hz), 7.25-7.38 (m, 5 H). ¹³C-NMR (75 MHz, CDCl₃) δ ppm 20.6; 75 44.8; 48.6; 119.9; 128.1; 128.4; 128.9; 135.9; 138.6. HRMS: m/z calcd for C₁₅H₂₃BrN (M + H⁺): 296.1014; found: 296.1008.

(*E*)-*N*-benzyl-3-bromo-*N*-metyl-3-phenylprop-2-en-1-amine (**5b**): pale yellow oil, yield: 84%. IR (atr) v 3058, 3028, 2941, 2788, 1632, 1452, 1025, 699 cm⁻¹. MS (EI 70 eV): m/z (relat. so int.): 317 (3), 315 (3), 236 (28), 226 (17), 224 (18), 134 (60), 115 (33), 91 (100). ¹H-NMR (300 MHz, CDCl3) δ ppm 2.08 (s, 3H), 3.09 (d, J = 7.3 Hz, 2H), 3.36 (s, 2H), 3.68 (s, 3H), 5.18 (t, J = 7.3 Hz, 1H), 6.74 (d, J = 9.0 Hz, 2H), 6.89 (d, J = 9.0 Hz, 2H), 7.15-7.32 (m, 8H), 7.43 (d, J = 9.0 Hz, 2H). ¹³C-NMR (100 MHz, 85 CDCl₃) δ 41.9; 56.2; 61.4; 123.0; 127.1; 128.1; 128.2; 128.6; 128.9; 128.9; 131.6; 138.3; 138.4. HRMS: m/z calcd for C₁₇H₁₉BrN (M + H⁺): 316.0701; found: 316.0702.

(*E*)-3-bromo-3-fenilprop-2-en-1-ol (5c): Colorless oil, Yield: 78%. MS (EI 70 eV): m/z (relat. Int.): 214 (7), 212 (8), 133 (55), 115 (54), 105 (100), 103 (85), 89 (25), 77 (51), 63 (48), 51 (80).

¹H-NMR (400 MHz, CDCl3) δ ppm 1.71 (s, 1H), 4.07 (d, J = 7.2 Hz, 2H), 6.42 (t, J = 7.2 Hz, 1H), 7.33-7.40 (m, 5H) ¹³C-NMR (100 MHz, CDCl3) δ ppm 60.6; 124.7; 128.2; 128.8; 129.0; 132.5; 137.7. HRMS: m/z calcd for C₉H₉BrO: 211.9837; found: 95 211.9850.

Suzuki Coupling of 5a-c with Arylboronic Acids

An oven-dried resealable Schlenk flask charged with **5** (2 mmol) was evacuated and back-filled with argon. Then, Pd(OAc)₂ (2.2 mg, 0.01 mmol), PPh₃ (5.2 mg, 0.02 mmol), arylboronic acid (2.4 mmol), KOH (224 mg, 4 mmol), MeOH (8 mL), and THF (8 mL) were added. The reaction mixture was stirred at room temperature for 2.5 h. After, the solution was then taken up in Et₂O (30 mL) and the Et₂O layer was washed with aq 1.0 M NaOH (10 mL) and brine (2 \times 5mL). The organic layer was dried (MgSO₄), filtered, concentrated under reduced pressure. The crude material was purified by column chromatography on silica gel using Hexane/Ethyl acetate as eluent. The *E*-configuration of the products was assigned by 2D-NOESY of selected compounds.

110 (*E*)-*N*,*N*-diisopropyl-3-(2-metoxy-5-methylphenyl)-3phenylprop-2-en-1-amine (6a): yellow oil, Yield: 93%. MS (IE 70 eV): m/z (relat. Int.): 337 (11, M⁺), 237 (70), 221 (20), 178 (32), 165 (24), 126 (100), 115 (51), 91 (68). ¹H-NMR (300 MHz, CDCl3) δ ppm 0.90 (d, J = 6.5 Hz, 12H), 2.20 (s, 3H), 3.00 (hep, 115 J = 6.5 Hz, 2H), 3.22 (d, J = 6.4 Hz, 2H), 3.43 (s, 3H), 5.84 (t, J =6.4 Hz, 1H), 6.64 (d, J = 8.9 Hz, 1H), 6.94 (d, J = 8.9 Hz, 1H)

- 6.95 (s, 1H), 7.05-7.24 (m, 5H). ¹³C-NMR (75 MHz, CDCl3) δ ppm 20.7; 20.9; 43.9; 49.3; 55.8; 115.5; 126.1; 127.1; 128.3; 128.7; 129.3; 131.1; 132.8; 138.4; 140.5; 154.6. HRMS: m/z calcd for $C_{23}H_{32}NO (M + H^+)$: 338.2484; found: 338.2474.
- 5 (E)-N,N-diisopropyl-3-(4-metoxyphenyl)-3-phenylprop-2-en-**1-amine** (7a): Yellow oil, Yield: 94%. IR (atr) v 3056, 2970, 2919, 1491, 1379, 1022, 860, 768, 696 cm⁻¹. ¹H-NMR (300 MHz, CDCl3) δ ppm 0.87 (d, J = 6.6 Hz, 12H), 2.97 (hep, J = 6.6 Hz, 2H), 3.07 (d, J = 6.5 Hz, 2H), 3.67 (s, 3H), 6.01 (t, J = 6.5 Hz, 10 1H), 6.70 (d, J = 8.9 Hz, 2H), 7.03-7.11 (m, 4H), 7.19-7.30 (m, 3H). ¹³C-NMR (75 MHz, CDCl3) δ ppm 20.7; 44.2; 48.7; 55.1; 113.3; 126.9; 127.9; 128.2; 129.7; 135.0; 140.0; 140.4; 158.6. HRMS: m/z calcd for C₂₂H₃₀NO (M + H⁺): 324.2327; found: 324.2338.
- 15 N,N-diisopropyl-3,3-diphenylprop-2-en-1-amine (8a): Yelllow oil, Yield: 81%. IR (atr) v 3058, 2965, 2830, 1662, 1592, 1345, 1125, 768, 699 cm⁻¹. MS (EI 70 eV): m/z (relat. Int.): 293 (9), 278 (13), 250 (5), 193 (100), 178 (23), 165 (11), 126 (50), 115 (5), 91 (23). ${}^{1}\text{H-NMR}$ (300 MHz, CDCl₃) δ ppm 0.88 (d, J = 6.6₂₀ Hz, 12H), 2.97 (hep, J = 6.6 Hz, 2H), 3.09 (d, J = 6.4 Hz, 2H), 6.10 (t, J = 6.4 Hz, 1H), 6.98-7.46 (m, 10H). ¹³C-NMR (75 MHz, CDCl3) δ ppm 20.7; 44.2; 48.6; 126.8; 127.1; 127.9; 129.7; 129.9; 131.6; 131.8; 139.9; 140.7; 142.4. HRMS: m/z calcd for $C_{21}H_{28}N (M + H^{+})$: 294.2222; found: 294.2212.
- 25 (E)-3-(4-(trifluoromethyl)phenyl)-N,N-diisopropyl-3phenylprop-2-en-1-amine (9a): Yellow oil, Yield: 83%. IR (atr) v 3057, 3027, 2969, 1668, 1616, 1325, 1167, 1125, 1067, 840, 701 cm⁻¹. MS (EI 70 eV): m/z (relat. Int.): 361 (9), 346 (24), 318 (4), 261 (100), 246 (8), 183 (29), 126 (30), 115 (27), 91 (13). 361 30 (9), 346 (24), 318 (4), 261 (100), 246 (8), 183 (29), 126 (30), 115 (27), 91 (13). ¹H-NMR (300 MHz, CDCl₃) δ ppm 0.95 (d, J = 6.5Hz, 12H), 3.03 (hep, J = 6.5 Hz, 2H), 3.17 (d, J = 6.3 Hz, 2H), 6.24 (t, J = 6.3 Hz, 1H), 7.14 (d, J = 8.0 Hz, 2H), 7.29-7.46 (m, 5H), 7.50 (d, J = 8.0 Hz, 2H). ¹³C-NMR (75 MHz, CDCl₃) δ 20.7; 35 44.4; 48.9; 118.9; 122.4; 124.9; 124.9; 125.0; 125.1; 126.0; 127.4; 128.3; 128.9; 129.7; 139.0; 139.7; 145.9. HRMS: m/z calcd for $C_{22}H_{27}F_3N$ (M + H⁺): 362.4583; found: 362.4593.
- (E)-N,N-diisopropyl-3-phenyl-3-m-toluylprop-2-en-1-amine (10a): Yellow oil, Yield: 92%. IR (atr) v 3021, 2966, 2924, 1665, 40 1599, 1318, 1175, 785, 701 cm⁻¹. MS (EI 70 eV): m/z (relat. Int.): 307 (12), 306 (5), 292 (10), 264 (6), 207 (100), 196 (14), 165 (8), 129 (37), 126 (64), 115 (53), 91 (14). ¹H-NMR (300 MHz, CDCl₃) δ ppm 0.88 (d, J = 6.6 Hz, 12H), 2.23 (s, 3H), 2.97 (hep, J = 6.6 Hz, 2H), 3.07 (d, J = 6.4 Hz, 2H), 6.07 (t, J = 6.4 Hz, 45 1H), 6.91-7.12 (m, 5H), 7.17-7.34 (m, 4H). ¹³C-NMR (75 MHz, CDCl₃) δ ppm 20.6; 21.3; 44.2; 48.6; 124.4; 126.8; 127.5; 127.7; 127.8; 129.7; 131.4; 137.4; 140.0; 140.8; 142.4. HRMS: m/z calcd for $C_{22}H_{30}N$ (M + H⁺): 308.2378; found: 308.2366.

(E)-N,N-diisopropyl-3-mesityl-3-phenylprop-2-en-1-amine 50 (11a): White solid, mp: 80 °C, Yield: 84%. IR (atr) v 2970, 1611, 1491, 1379, 1287, 1201, 1180, 1023, 696 cm⁻¹. MS (EI 70 eV): m/z (relat. Int.): 335 (26), 334 (16), 292 (6), 235 (100), 220 (12), 205 (10), 157 (21), 129 (13), 115 (41), 91 (27). ¹H-NMR (300 MHz, CDCl3) δ ppm 0.97 (d, J = 6.5 Hz, 12H), 2.17 (s, 6H), 2.28

 $_{55}$ (s, 3H), 3.05 (hep, J = 6.5 Hz, 2H), 3.44 (d, J = 6.3 Hz, 2H), 5.61 (t, J = 6.3 Hz, 1H), 6.86 (s, 2H), 7.14-7.31 (m, 5H). ¹³C-NMR (75) MHz, CDCl3) δ ppm 20.6; 20.8; 21.0; 43.8; 48.5; 126.5; 127.7; 128.2; 129.1; 134.1; 136.0; 136.2; 138.5; 139.3; 140.6. HRMS:

- m/z calcd for $C_{24}H_{34}N$ (M + H⁺): 336.2691; found: 336.2678.
- 60 (E)-N,N-diisopropyl-3-(4-(methylsulfonil)phenyl)-3phenylprop-2-en-1-amine (12a): White solid, mp: 91 °C Yield: 71%. IR (atr) v 3057, 2963, 1594, 1378, 1303, 1148, 952, 771, 702 cm⁻¹. ¹H-NMR (400 MHz, CDCl₃) δ 0.95 (d, J = 6.6 Hz, 12H), 2.95 - 3.08 (m, 5H, hep+s), 3.16 (d, J = 6.4 Hz, 2H), 6.2965 (t, J = 6.4 Hz, 1H), 7.07-7.18 (m, 2H), 7.30-7.46 (m, 5H), 7.81(d, J = 8.6 Hz, 2H). ¹³C-NMR (100 MHz, CDCl₃) δ 20.6; 44.2; 44.4; 48.8; 127.0; 127.4; 127.7; 128.2; 129.5; 135.5; 138.3; 138.5; 139.0; 147.7. HRMS: m/z calcd for $C_{22}H_{30}NO_2S$ (M + H⁺): 372.1997; found: 372.2007.
- $_{70}$ (E)-N-benzyl-N-methyl-3-(4-(methylsulfonil)phenyl)-3phenylprop-2-en-1-amine (6b): viscous brown oil, Yield: 74%. IR (atr) v 3057, 3025, 2836, 1661, 1592, 1452, 1311, 1144, 1092, 1023, 772, 701 cm⁻¹. ¹H-NMR (400 MHz, CDCl₃) δ 2.23 (s, 3H), 3.07 (s, 3H), 3.16 (d, J = 6.7 Hz, 2H), 3.49 (s, 2H), 6.39 (t, J =75 6.7 Hz, 1H), 7.11-7.18 (m, 2H), 7.26-7.44 (m, 10H), 7.84 (d, J = 8.5 Hz, 2H). ¹³C-NMR (75 MHz, CDCl₃) δ 42.6; 44.5; 56.2; 62.0; 127.0; 127.2; 127.6; 128.0; 128.2; 128.4; 128.9; 129.6; 130.8; 138.4; 138.8; 142.0; 147.6. HRMS: m/z calcd for C₂₄H₂₆NO₂S (M + H⁺): 392.5419; found: 392.5422.
- 80 (E)-3-(4-methoxyphenyl)-3-phenylprop-2-en-1-ol (6c): White wax, Yield: 93%. IR (atr) v 3310, 3048, 2946, 2834, 1605, 1508, 1441, 1247, 1177, 1016, 816, 646 cm-1. MS (IE 70 eV): m/z (int. relat): 240 (10), 222 (100), 207(79), 197 (40), 178 (65), 165 (22), 152 (27), 111 (12), 89 (18), 51 (21). ¹H-NMR (300 MHz, ₈₅ CDCl₃) δ ppm 1.57 (s, 1H), 3.80 (s, 3H), 4.19 (d, J = 6.9 Hz, 2H), 6.17 (t, J = 6.9 Hz, 1H), 6.82 (d, J = 8.9 Hz, 2H), 7.14-7.22 (m, 4H), 7.32-7.41 (m, 3H). ¹³C-NMR (75 MHz, CDCl₃) δ ppm 55.3; 60.8; 113.5; 125.6; 127.5; 128.2; 128.7; 129.7; 134.3; 139.2; 143.8; 159.2. HRMS: m/z calcd for $C_{16}H_{17}O_2$ (M + H⁺): 90 241.1229; found: 241.1237.
- (E)-3-(4-(methylsulfonil)fenil)-3-fenilprop-2-en-1-ol White crystals, mp: 89 °C Yield: 77%. IR (atr) v 3496, 3062, 3021, 2981, 1591, 1292, 1140, 1036, 816, 774, 701 cm⁻¹. ¹H-NMR (400 MHz, CDCl₃) δ 1.63 (s, 1H, OH), 3.05 (s, 3H), 4.26 95 (d, J = 6.6 Hz, 2H), 6.35 (t, J = 6.6 Hz, 1H), 7.09-7.19 (m, 2H), 7.39 (t, J = 6.2 Hz, 3H), 7.43 (d, J = 8.4 Hz, 2H), 7.84 (d, J = 8.4Hz, 2H). ¹³C-NMR (100 MHz, CDCl₃) δ 44.5; 60.5; 127.3; 128.1; 128.4; 128.6; 129.6; 130.8; 137.8; 139.1; 142.3; 147.3. HRMS: m/z calcd for $C_{16}H_{17}O_3S$ (M + H⁺): 289.3751; found: 289.3752.
- 100 1-(4-((E)-3-hydroxy-1-phenylprop-1-enyl)phenyl)ethanone (8c): White solid, Yield: 86%. ${}^{1}\text{H-NMR}$ (400 MHz, CDCl₃) δ 1.56 (s, 1H, OH), 2.59 (s, 3H), 4.25 (d, J = 6.7 Hz, 2H), 6.35 (t, J= 6.7 Hz, 1H, 7.11-7.18 (m, 2H), 7.34 (d, J = 8.5 Hz, 2H), 7.36-7.44 (m, 3H), 7.88 (d, J = 8.5 Hz, 2H). ¹³C-NMR (100 MHz, 105 CDCl₃) δ 26.6; 60.6; 127.7; 127.9; 128.3; 128.4; 129.6; 129.7; 136.0; 138.3; 143.1; 146.3; 197.7. HRMS: m/z calcd for $C_{17}H_{17}O_2(M + H^+)$: 253.1229; found: 253.1232.
- 3,3-diphenylprop-2-en-1-ol³⁵ (**9c**) Pale mp: 59 °C, Yield: 82% IR (atr) v 3316, 3052, 3023, 2918,1598, 110 1441, 1017, 759, 699 cm⁻¹. ¹H-NMR (400 MHz, CDCl₃) δ1.53 (s, 1H, OH), 4.14 (d, J = 6.8 Hz, 2H), 6.17 (t, J = 6.8 Hz, 2H), 7.09(dd, J = 7.8, 1.6 Hz, 2H), 7.13 - 7.33 (m, 8H). ¹³C-NMR (100 MHz, CDCl₃) δ 60.7; 127.4; 127.5; 127.6; 128.1; 128.2; 129.7; 139.0; 141.7; 144.2.
- 115 Copper-Catalyzed Ullmann-type Vinylation of Phenols with 5b and 5c

An oven-dried resealable Schlenk flask charged with **5b** or **5c** (2.0 mmol), phenol (3.0 mmol), 1,10-phenanthroline (36 mg,0.2 mmol), CuI (38 mg, 0.2 mmol) and K₃PO₄·H₂O (4.0 mmol, 920 mg), was evacuated and back-filled with argon. Then, dioxane (8 mL) was added and the reaction mixture was stirred at 100 °C for 16h. Next, the reaction mixture was cooled, filtered and analysed by GC and GC-MS. After, the solution was taken up in CH₂Cl₂ (30 mL) and the organic layer was washed with aq 1.0 M NaOH (to remove the phenol excess) and brine (2 × 5mL), and then dried over MgSO₄. Finally, the solvent was evaporated and the product purified by column chromatography (silica gel in hexane/ethylacetate) and analysed by ¹H-NMR and ¹³C-NMR. The *E*-configuration of the products was assigned by 2D-NOESY of selected compounds.

- 15 (*E*)-3-(4-methoxyphenoxy)-*N*-benzyl-*N*-methyl-3-phenylprop-2-en-1-amine (8b): Pale yellow oil, Yield: 86%. MS (EI 70 eV): m/z (relat. Int.): 359 (7), 358 (17), 268 (11), 251 (26), 239 (15), 161 (33), 146 (49), 134 (22), 105 (56), 91 (100). ¹H-NMR (300 MHz, CDCl₃) δ ppm 2.08 (s, 3H), 3.09 (d, *J* = 7.3 Hz, 2H), 3.36 (s, 2H), 3.68 (s, 3H), 5.18 (t, *J* = 7.3 Hz, 1H), 6.74 (d, *J* = 9.0 Hz,
- ²⁰ (8, 2H), 3.68 (8, 3H), 5.18 (1, J = 7.5 Hz, 1H), 6.74 (d, J = 9.0 Hz, 2H), 6.89 (d, J = 9.0 Hz, 2H), 7.15-7.32 (m, 8H), 7.40-7.45 (m, 2H). ¹³C-NMR (75 MHz, CDCl₃) δ ppm 41.9; 54.1; 55.5; 61.4; 108.3; 114.6; 120.3; 126.9; 128.0; 128.1; 128.5; 128.8; 129.1; 134.5; 138.6; 149.8; 155.3; 155.8. HRMS: m/z calcd for ²⁵ C₂₄H₂₆NO₂(M + H⁺): 360.1964; found: 360.1967.
- (*E*)-3-(*o*-toluyloxy)-*N*-benzyl-*N*-methyl-3-phenylprop-2-en-1-amine (**9b**): Pale yellow oil, Yield: 86%. IR (atr) v 3059, 3027, 2956, 1655, 1597, 1491, 1228, 1183, 1115, 845, 751, 699 cm⁻¹ MS (EI 70 eV): m/z (relat. int.): 343 (7), 342 (12), 252 (16), 236
- ³⁰ (12), 223 (12), 146 (26), 115 (32), 105 (55), 91 (100). ¹H-NMR (300 MHz, CDCl₃) δ ppm 2.01 (s, 3H), 2.21 (s, 3H), 3.03 (d, J = 7.3 Hz, 2H) 3.30 (s, 2H), 4.97 (t, J = 7.3 Hz, 1H), 6.85-6.89 (m, 2H), 6.94-7.31 (m, 10H), 7.44-7.47 (m, 2H). ¹³C-NMR (100 MHz, CDCl₃) δ 16.2; 41.9; 54.2; 61.4; 106.7; 119.6; 123.6;
- $_{35}$ 127.0; 128.1; 128.2; 128.6; 128.7; 129.1; 129.5; 131.2; 134.7; 153.9; 155.2. HRMS: m/z calcd for $C_{24}H_{26}NO$ (M + H $^+$): 344.2014; found: 344.2021.

$1\hbox{-}(4\hbox{-}(E)\hbox{-}3\hbox{-}(N\hbox{-}benzyl\hbox{-}N\hbox{-}methylamine)\hbox{-}1\hbox{-}phenylprop\hbox{-}1\hbox{-}$

- **enyloxy)phenyl)ethanone** (**10b):** Yellow oil, Yield: 43%. IR ₄₀ (atr) v 3058, 2933, 1716, 1589, 1503, 1207, 1031, 700 cm⁻¹. ¹H-NMR (400 MHz, CDCl₃) δ ppm 2.16 (s, 3H), 2.45 (s, 3H), 3.20 (d, J = 7.3 Hz, 2H), 3.44 (s, 2H), 5.63 (t, J = 7.3 Hz, 1H), 6.94 (d, J = 8.9 Hz, 2H), 7.16-7.30 (m, 8H), 7.37-7.46 (m, 2H), 7.79 (d, J = 8.9 Hz, 2H). ¹³C-NMR (75 MHz, CDCl₃) δ ppm 26.3; 42.1;
- $_{45}$ 53.7; 61.6; 114.9; 117.0; 127.0; 128.1; 128.2; 128.6; 128.8; 129.0; 130.3; 131.3; 133.5; 138.6; 152.1; 161.2; 196.6. HRMS: m/z calcd for $C_{25}H_{26}NO_2$ (M + H $^+$): 372.1964; found: 372.1954.
- **1-(4-(**(*E*)**-3-hidroxy-1-phenylprop-1-enyloxy)phenyl)ethanone (10c**): Brown oil, Yield: 51%. IR (atr) v 3345, 3057, 3027, 1681,
- ⁵⁰ 1601, 1359, 1267, 1023, 957, 702, cm⁻¹. ¹H-NMR (400 MHz, CDCl₃) δ ppm 2.50 (s, 3H), 4.33 (d, J = 7.6 Hz, 2H), 5.65 (t, J = 7.6 Hz, 1H), 7.01 (d, J = 8.9 Hz, 2H), 7.29-7.39 (m, 3H), 7.42-7.48 (m, 2H), 7.85 (d, J = 8.9 Hz, 2H). ¹³C-NMR (100 MHz, CDCl₃) δ ppm 26.4; 58.8; 114.7; 117.6; 128.4; 128.6; 129.3;
- ss 130.4; 131.7; 133.0; 153.8; 160.7; 196.9. HRMS: m/z calcd for C₁₇H₁₇O₃ (M + H⁺): 269.1178; found: 269.1155.
 - (*E*)-3-(4-methoxyphenoxy)-3-phenylprop-2-en-1-ol (11c) White solid, Yield: 74% mp: 54 °C IR (atr) v 3380, 3054, 2933,

1650, 1504, 1260, 1096, 982, 832, 780, 704 cm⁻¹. ¹H-NMR (400 MHz, CDCl₃) δ 3.79 (s, 3H), 4.21 (d, J = 7.8 Hz, 2H), 5.19 (t, J = 7.8 Hz, 1H) 6.85 (d, J = 9.0 Hz, 2H), 7.01 (d, J = 9.0 Hz, 2H), 7.36-7.45 (m, 3H), 7.51-7.57 (m, 2H). ¹³C-NMR (100 MHz, CDCl₃) δ 55.6; 59.3; 107.5; 114.7; 121.1; 128.2; 128.7; 129.06; 134.1; 149.1; 155.8; 158.1. HRMS: m/z calcd for $C_{17}H_{17}O_3$ (M + 65 H⁺): 257.1178; found: 257.1182.

Notes and references

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- 70 † Electronic Supplementary Information (ESI) available: [Dehydrobromination optimisation, NMR spectra and NOE-stereochemistry characterisation are described in the Supporting Information]. See DOI: 10.1039/b000000x/
- 75 1. K. Itami and J. Yoshida, Chem. Eur. J., 2006, 12, 3966-3974.
 - K. Itami, T. Kamei and J. Yoshida, J. Am. Chem. Soc., 2003, 125, 14670-14671.
 - C. X. Zhou, D. E. Emrich and R. C. Larock, Org. Lett., 2003, 5, 1579-1582.
- 80 4. A. B. Flynn and W. W. Ogilvie, *Chem. Rev.*, 2007, **107**, 4698-4745.
- 5. R. A. Pilli and L. G. Robello, J. Braz. Chem. Soc., 2004, 15, 938-944.
- C. M. Nunes, J. Limberger, S. Poersch, M. Seferin and A. L. Monteiro, Synthesis, 2009, 2761-2765.
- P. E. Tessier, A. J. Penwell, F. E. S. Souza and A. G. Fallis, *Org. Lett.*, 2003, 5, 2989-2992.
- J. C. Pastre and C. R. D. Correia, Adv. Synth. Catal., 2009, 351, 1217-1223.
- F. Ulgheri, M. Marchetti and O. Piccolo, J. Org. Chem., 2007, 72, 6056-6059.
- K. Gaukroger, J. A. Hadfield, L. A. Hepworth, N. J. Lawrence and A. T. McGown, J. Org. Chem., 2001, 66, 8135-8138.
 - T. J. N. Watson, S. W. Horgan, R. S. Shah, R. A. Farr, R. A. Schnettler, C. R. Nevill, F. J. Weiberth, E. W. Huber, B. M. Baron, M. E. Webster, R. K. Mishra, B. L. Harrison, P. L. Nyce, C. L. Rand and C. T. Goralski, *Org. Process Res. Dev.*, 2000, 4, 477-487.
 - S. M. Nobre, M. N. Muniz, M. Seferin, W. M. da Silva and A. L. Monteiro, Appl. Organomet. Chem., 2011, 25, 289-293.
- 13. C. Gurtler and S. L. Buchwald, Chem. Eur. J., 1999, 5, 3107-3112.
- 100 14. M. Ohff, A. Ohff and D. Milstein, Chem. Commun., 1999, 357-358.
 - 15. A. F. Littke and G. C. Fu, J. Am. Chem. Soc., 2001, 123, 6989-7000.
 - I. Kondolff, H. Doucet and M. Santelli, *Tetrahedron Lett*, 2003, 44, 8487-8491.
- 17. A. Nejjar, C. Pinel and L. Djakovitch, *Adv. Synth. Catal.*, 2003, **345**, 612-619.
- M. MorenoManas, R. Pleixats and A. Roglans, Synlett, 1997, 1157-1158.
- J. Masllorens, M. Moreno-Manas, A. Pla-Quintana, R. Pleixats and A. Roglans, Synthesis, 2002, 1903-1911.
- 110 20. L. Bagnell, U. Kreher and C. R. Strauss, *Chem. Commun.*, 2001, 29-30.
 - S. Fleischer, S. Zhou, K. Junge and M. Beller, *Angew. Chem. Int. Ed.*, 2013, 52, 5120-5124.

- G. Wienhöfer, F. A. Westerhaus, K. Junge and M. Beller, *Journal of Organometallic Chemistry*, 2013, 744, 156-159.
- P. c. Prediger, L. s. F. Barbosa, Y. Génisson and C. R. D. Correia, *New J. Chem.*, 2011, 76, 7737-7749.
- 5 24. P. Prediger, A. R. da Silva and C. R. D. Correia, *Tetrahedron*, 2014, 70, 3333-3341.
- 25. V. R. Lando and A. L. Monteiro, Org. Lett., 2003, 5, 2891-2894.
- P. B. Silveira and A. L. Monteiro, J. Mol. Catal. A: Chem., 2006, 247, 1-6.
- 10 27. B. C. Ranu and R. Jana, J. Org. Chem, 2005, 70, 8621-8624.
 - B. C. Ranu, S. K. Guchhait and A. Sarkar, *Chem. Commun.*, 1998, 2113-2114.
 - 29. J. Avraamides and A. Parker, Aust. J. Chem., 1983, 36, 1705-1717.
 - 30. C. M. Nunes, D. Steffens and A. L. Monteiro, Synlett, 2007, 103-106.
- 15 31. J. Limberger, B. C. Leal, D. F. Back, J. Dupont and A. L. Monteiro, Adv. Synth. Catal., 2012, 354, 1429-1436.
 - F. Monnier and M. Taillefer, Angew. Chem., Int. Ed., 2009, 48, 6954-6971.
- 33. A. Ouali, J. F. Spindler, A. Jutand and M. Taillefer, *Adv. Synth.* 20 *Catal.*, 2007, **349**, 1906-1916.
 - M. Zhu, S. Lin, G.-L. Zhao, J. Sun and A. Córdova, *Tetrahedron Lett*, 2010, 51, 2708-2712.
 - H. Zheng, M. Lejkowski and D. G. Hall, *Chem. Sci.*, 2011, 2, 1305-1310.