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# *n*-Type Small Aromatic Core Diimides Flanked With Electron Donating Thienylethyl Moiety and Electrical Responses in Organic Devices

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## Abstract

We report *n*-type small aromatic core diimide derivatives functionalized with electron donating heteroaromatic ring bearing moiety, *N,N'*-bis[2-(2-thienyl)ethyl]-benzene-1,2,4,5-tetracarboxylic diimide and *N,N'*-bis[2-(2-thienyl)ethyl]-naphthalene-1,4,5,8-tetracarboxylic diimide. The synthesis, characterization of the newly synthesized diimides based organic semiconductors and electrical responses in electronic devices are investigated in detail. The structure, electrochemical and optical properties of the diimide derivatives were determined using <sup>1</sup>H NMR, <sup>13</sup>C NMR, mass spectrometry, UV-visible spectroscopy and cyclic voltammetry. The electron density distribution of the materials was also studied using density functional theory calculations to relate the structure and device's characteristics. The materials were employed for the preparation of organic thin-film transistors with inorganic and polymeric gate dielectrics at various substrate temperatures, and the device's electrical characteristics have been analyzed in detail. We also further explore its applicability in advanced electronic devices by fabricating a resistance load-type organic inverter, and the dynamic response behavior was investigated.

**Keywords:** Diimide, polymer gate dielectric, organic thin-film transistor, *n*-channel semiconductor

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## 1 **1. Introduction**

2 Over the last few decades, organic thin-film transistors (OTFTs) have attracted widespread interest and  
3 huge progress has been achieved on their application in versatile electronic devices [1]. Accordingly,  
4 significant efforts have been targeted toward the development of a wide variety of organic/polymers for  
5 the ultimate goal of generating inexpensive organic devices [2]. In organic small-molecule-based TFTs,  
6 polycyclic aromatic compounds are preferentially used as the semiconducting materials [3]; e.g., acene  
7 derivatives [4] and diketopyrrolopyrrole (DPP) derivatives [5] as the *p*-channel components and  
8 fullerene derivatives [6] and rylene diimide derivatives [7] as the *n*-channel semiconductors. Recently,  
9 we demonstrated that easily accessible, small molecular DPPs could be used for *p*-channel OTFTs [8].  
10 In contrast to *p*-channel small molecules, where a large variety of chemical structures have been  
11 presented, there remains a need to explore easily derivable, inexpensive small molecule-based *n*-channel  
12 semiconductors.

13 Rylene diimide derivatives are well-known aromatic electron-transport materials that are used  
14 in a variety of organic electronic devices because of their excellent thermal and environmental stability  
15 [9]. In particular, pyromellitic diimide (PMDI), naphthalene diimide (NDI) and perylene diimide (PDI)  
16 are the most extensively studied organic semiconductor (OSC) materials for *n*-channel OTFTs owing to  
17 their relatively high electron affinities and the versatile tunability of their optical and electrochemical  
18 properties via parent core extension or functionalization [10]. Unlike the core alteration, chemical  
19 structure engineering through a functionalization at the *N*-position of imide is a simple one-pot synthetic  
20 approach to a variety of aromatic derivatives for *n*-channel OTFTs [11]. The relatively facile synthesis  
21 of pyromellitic diimide (PMDI) and NDI derivatives in comparison to that of PDI derivatives enables  
22 more efficient large-scale production. Hence, a large number of *N*-functionalized PMDI and NDI  
23 derivatives with various side chains have been reported as a channel layer material for OTFTs [9a]. A  
24 series of NDI derivatives with alkylthienyl moieties have been reported and their spectroscopic,  
25 electrochemical, electronic, and structural properties were investigated in detail [12]. The Asha's group

1 also prepared a series of amide-functionalized NDIs for improved charge-carrier mobility, and described  
2 the effect of hydrogen bonding on the self-assembly of NDIs [13]. Lee *et al.* investigated two *n*-type  
3 soluble NDI derivatives with phenylmethyl and (trifluoromethyl)benzyl groups in the *N*-position of NDI  
4 [14]. An alkylphenyl functionalized NDI [15] and a 2-phenylethyl functionalized PDI derivative (BPE-  
5 PTCDI) [16] for a high performance *n*-channel OTFTs have been reported by Katz's and Bao's group,  
6 respectively. It is also reported that the aromatic/heteroaromatic ring bearing side chains in the rylene  
7 diimide derivatives have a potential role in controlling molecular packing and optoelectronic properties  
8 [17].

9 Here, we report the synthesis and characteristics of new small aromatic core diimide derivatives  
10 which have different core size and are functionalized with heteroaromatic ring (thienylethyl group) at  
11 the *N*-position of the core. We also report the static/dynamic characteristics of the aromatic core diimide  
12 derivatives based organic electronic devices with inorganic/organic dielectric fabricated at different  
13 substrate temperature.

## 15 2. Experimental

16 Details of the materials and instruments used in the synthesis and characterization are described  
17 in the Materials and Instruments section of the Supporting Information.

### 18 2.1 Synthesis of *N,N'*-bis[2-(2-thienyl)ethyl]-benzene-1,2,4,5-tetracarboxylic diimide (**3a**):

19 Benzene-1,2,4,5-tetracarboxylic dianhydride (500 mg, 2.29 mmol) was dissolved in  
20 dimethylformamide (DMF; 10 mL). 2-Thiophene ethylamine (640.8 mg, 5.04 mmol) was added and the  
21 resultant mixture was stirred at 110 °C overnight under a nitrogen atmosphere. After cooling to ambient  
22 temperature, the mixture was filtered and washed with methanol. A white flake like solid (920 mg) was  
23 obtained with a yield of 92%. The material was purified by vacuum sublimation. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400  
24 MHz): δ = 8.25 (s, 2H), 7.15 (m, 2H), 6.91 (d, 2H), 6.86 (d, 2H), 4.04 (t, 4H), 3.28 (t, 4H). <sup>13</sup>C (CDCl<sub>3</sub>,  
25 100 MHz): δ = 165.91, 139.39, 137.13, 127.08, 127.08, 125.83, 124.40, 118.37, 39.99, 28.31. Mp: 276.4

1 °C.

## 2 **2.2 Synthesis of *N,N'*-bis[2-(2-thienyl)ethyl]-naphthalene-1,4,5,8-tetracarboxylic diimide (3b):**

3 1,4,5,8-Naphthalene-1,4,5,8-tetracarboxylic dianhydride (200 mg, 0.745 mmol) was dissolved  
4 in dimethylformamide (DMF; 6 mL). 2-Thiophene ethylamine (208.7 mg, 1.64 mmol) was added and  
5 the resultant mixture was stirred at 110 °C overnight under a nitrogen atmosphere. After cooling to  
6 ambient temperature, the mixture was filtered and washed with methanol. A yellowish brown solid (300  
7 mg) was obtained with a yield of 83%. The material was purified by vacuum sublimation. Mp: 277.1 °C.  
8 <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz): δ = 8.76 (s, 4H), 7.14-6.92 (m, 6H), 4.48 (t, 4H), 3.29 (t, 4H). <sup>13</sup>C (CDCl<sub>3</sub>,  
9 100 MHz): δ = 162.67, 140.16, 131.08, 127.08, 127.01, 126.59, 125.71, 124.13, 42.05, 28.11. HRMS  
10 [M+H]<sup>+</sup>, m/z calcd for C<sub>26</sub>H<sub>18</sub>N<sub>2</sub>O<sub>4</sub>S<sub>2</sub>: 487.562, found: 487.3613.

## 11 **2.3 Thin-film transistor fabrication and characterization:**

12 Three types of gate dielectric were used for the fabrication of OTFTs with the synthesized  
13 aromatic diimide small molecules: (i) *p*-doped Si/bare SiO<sub>2</sub> (300 nm) (capacitance ( $C_i$ ) = 100 pF/mm<sup>2</sup>),  
14 (ii) *p*-doped Si/OTS-treated SiO<sub>2</sub> (300 nm) ( $C_i$  = 110 pF/mm<sup>2</sup>), and (iii) ITO/cross-linked poly(4-  
15 vinylphenol) (CL-PVP) ( $C_i$  = 61.04 pF/mm<sup>2</sup>). A metal-insulator-metal (MIM, active area of 0.203 cm<sup>2</sup>)  
16 device was used to measure the capacitance of the gate dielectrics. The Si/SiO<sub>2</sub> and glass/ITO substrates  
17 were cleaned as per a previously reported procedure [8]. An *n*-octadecyltrichlorosilane self-assembled  
18 monolayer was deposited over Si/SiO<sub>2</sub> by spin-coating (3000 rpm, 10 s) a 0.05 M OTS solution (in  
19 toluene) immediately after UV-ozone treatment; the OTS-treated substrate was then baked at 80 °C for 3  
20 min. Using our previously reported method, a robust uniform thin-film of cross-linked poly(4-  
21 vinylphenol) (CL-PVP) on a glass/ITO substrate was obtained as a gate dielectric [18]. The active layer  
22 (50 nm) of **3a** or **3b** was deposited by thermal vacuum evaporation on top of the dielectrics at varied  
23 substrate temperatures (25, 45, and 60 °C) at a fixed rate of 0.2 Å/s. Finally, gold source and drain  
24 electrodes (50 nm) were thermally deposited under high vacuum ( $6 \times 10^{-6}$  Torr) at a rate of 0.3 Å/s.  
25 During Au deposition, a shadow mask was used to define the channel length ( $L$ ) and width ( $W$ ) of 50

1 and 1000  $\mu\text{m}$ , respectively. An HP 4156A semiconductor analyzer was used to determine the electrical  
2 characteristics of the OTFTs under vacuum.

### 3 **2.4 Organic inverter fabrication and characterization:**

4 A resistive-load-type inverter was constructed by connecting the **3b** OTFT fabricated at  $T_s$  of 45  
5  $^{\circ}\text{C}$  with the OTS-treated  $\text{SiO}_2$  and an external resistive load (100  $\text{M}\Omega$ ). To characterize the dynamic  
6 response of the transistor, a function generator (maximum 60 MHz; Rigol DG4062) was used to supply  
7 the input voltage. The output voltage response was detected using a digital oscilloscope (100 MHz;  
8 Rigol DS2102). A high-voltage amplifier (1 MHz, 400 V<sub>p-p</sub>, Pintek HA-405) was used to amplify the  
9 signal and a Keithley 6517A was used as the power supply.

## 10 **3. Results and discussion**

11 The synthetic schematic of thienylethyl-functionalized pyrromelittic (**3a**) and naphthalene  
12 diimides (**3b**) are shown in Fig. 1. The materials were successfully prepared using an  
13 amine/DMF/110  $^{\circ}\text{C}$  protocol [19] with a high yield (92% for **3a** and 83% for **3b**). The chemical  
14 structures were identified by spectroscopic analysis (Supporting Information, Fig. S1-S2) and the  
15 corresponding data are provided in the experimental section.

16 DFT calculations [B3LYP/6-31G(d,p) level] of the energy levels using SPARTAN10 [20],  
17 predicted a LUMO energy level of -3.21 eV for **3a** and -3.47 eV for **3b** in vacuum and their respective  
18 HOMO energy levels are at -6.30 and -6.17 eV, respectively. Electron density distribution of the LUMO  
19 (a and b) and HOMO (c and d) of **3a** and **3b** is depicted in Fig. 2. It is found that the lowest unoccupied  
20 molecular orbital density is distributed mainly on the core of the molecule and the highest occupied  
21 molecular orbital density is localized on thiophene unit (side chain) suggesting that the core of the  
22 material is electron accepting in nature. Similar predominant localization of the LUMO and HOMO  
23 wave functions on the acceptor and donor units, respectively, has been reported for a set of rylene and  
24 related diimide derivatives [21, 22]. The LUMO energy of **3a** and **3b** was also experimentally calculated  
25 by cyclic voltammetry (Instrument section of the Supporting Information for details), as shown in inset

1 of Fig. 2 (right side) and Fig. S3a, respectively. The LUMO value of **3a** and **3b**, calculated by  $E_{\text{LUMO}} = -$   
2  $[4.8 - E_{1/2, \text{Fc/Fc}^+} + E_{\text{red, onset}}]$  eV [23], were found to be -3.29 and -3.91 eV, respectively. The onset  
3 potential of the first reduction wave of **3a** and **3b** are -1.13 and -0.51 V (**3b**), respectively, where  $E_{1/2}$  of  
4 ferrocene is 0.38 V. The optical properties of **3a** and **3b** in chloroform were investigated via UV-visible  
5 spectroscopy and the corresponding absorption spectra are shown in Fig. S3. The absorption bands for  
6 **3a** are centered at 320 and 310 nm where as for **3b** the peaks are at 362 and 382 nm respectively.  
7 Optical bandgaps calculated from absorption cut off values are 3.65 and 3.05 eV for **3a** and **3b**,  
8 respectively. The thermal properties of the materials were investigated through a thermogravimetric  
9 analysis (TGA) and differential scanning calorimetry (DSC). The thermal decomposition temperatures  
10 with a 5% weight loss were calculated to be 328 °C and 395 °C for **3a** and **3b**, respectively (Fig. 3a).  
11 Differential scanning calorimetry (DSC) analysis revealed that the materials showed clear exothermic  
12 crystallization and endothermic melting temperature at 267 and 277.1 °C for **3b** (Fig. 3b) and at 273 and  
13 276.4 °C for **3a**, respectively (inset, Fig. 3b). These results reveal that their thin-film could be easily  
14 prepared through a thermal vacuum evaporation over a broad range of substrate temperatures without  
15 any thermal degradation during the film formation.

16 The OTFTs based on **3b** show well-defined *n*-channel characteristics (discussed in the later  
17 part). On the other hand, the device based on **3a** showed no such characteristics. It is clearly different  
18 observation from other diimide derivatives with pyromellitic diimide core (**4a** and **4b**, Fig. 1) exhibiting  
19 *n*-channel transport behavior [19]. The reported diimides (**4a** and **4b**) have fluorinated side chain that  
20 stabilizes the LUMO energy level. However, in our pyromellitic diimide (**3a**), electron donating  
21 thienylethyl group is introduced as a side chain, resulting in the higher LUMO energy level. The higher  
22 LUMO level of **3a** may increase charge injection barrier (1.99 eV) from Au electrode ( $\phi_m = 5.2$  eV) [24],  
23 that is relatively higher than those of **4a** (1.74 eV) and **4b** (1.66 eV) (See the inset of Fig. 2). On the  
24 other hand, the **3b** with an extended diimide core shows lower lying LUMO energy compared to **3a**, and  
25 the devices based on **3b** showed good *n*-channel OTFT properties. It can be inferred from the above

1 reasons that the electrical characteristics of small aromatic core diimide derivatives could be affected by  
2 the core size and the type of introduced side group. The theoretically calculated LUMO energy levels of  
3 all the discussed materials are depicted in inset of Fig. 2 (left side). Based on the results, we further  
4 investigated in detail the performance of OTFTs with **3b** deposited on different gate dielectrics such as  
5 OTS-treated SiO<sub>2</sub>, and cross-linked poly(4-vinylphenol) at various substrate temperature in order to  
6 explore its potential applicability as an organic semiconductor in advanced OTFTs.

7 It could be inferred from the device results (discussed in the later section) that the electron  
8 mobilities of the device are dependent on the type of gate dielectric and  $T_s$  [11b, 25, 26]. Atomic force  
9 microscopy (AFM) and X-ray diffraction (XRD) analysis were performed on the **3b** thin-film to  
10 understand the variation in the performance of the OTFTs with different gate dielectrics and  $T_s$ . Fig. 4(a-  
11 c) show the AFM height profiles of the **3b** thin-film deposited at a  $T_s$  of 45 °C onto bare SiO<sub>2</sub>, OTS-  
12 treated SiO<sub>2</sub>, and CL-PVP, respectively, while those on the various gate dielectrics at  $T_s$  of 25 and 60 °C  
13 are depicted in Fig. S5. The images reveal that the grain size increases slightly with increasing  $T_s$  but  
14 retains an almost identical crystal shape. Although there are many reports on the increased charge-  
15 carrier mobility with increasing grain size [16c, 19, 27], our initial studies showed the opposite trend;  
16 the mobility decreased with increasing grain size that are also observed by Park group [28]. It has been  
17 suggested by them that interconnection and improved contact between grains results in the enhancement  
18 of mobility, while the layer with larger voids could reduce the carrier mobility inspite of the formation  
19 of large grains. In our study we also observed larger grains of **3b** with bigger voids at higher  $T_s$  (60 °C)  
20 in the film may leads the decrease in the hole mobility. The AFM image of the **3b** thin-film on OTS-  
21 treated SiO<sub>2</sub> (Fig. 4b) prepared at  $T_s$  of 45 °C revealed improved contact between the grains with needle-  
22 like crystals, which were not observed on bare SiO<sub>2</sub> and CL-PVP. In addition, the device fabricated with  
23 OTS-treated SiO<sub>2</sub> at  $T_s$  of 45 °C showed the best device performance. This improvement in electron  
24 mobility may also be because of an increase in the crystallinity of the **3b** thin-film, which was evident  
25 from the XRD analyses. Fig. 4d shows the XRD patterns of the **3b** thin-films prepared at  $T_s$  of 45 °C on



1 bare SiO<sub>2</sub>, OTS-treated SiO<sub>2</sub>, and CL-PVP. The XRD patterns reveal that the **3b** on OTS-treated SiO<sub>2</sub>  
2 gives the highest diffraction intensity, which indicates improved crystallinity. Similar observations were  
3 reported by the Bao's group regarding BPE-PTCDI thin films on an OTS-treated SiO<sub>2</sub> [16c]. The XRD  
4 patterns of **3b** film (50 nm) on bare SiO<sub>2</sub>, CL-PVP and OTS-treated SiO<sub>2</sub> substrates at various  $T_s$  are  
5 also depicted in Fig. S6. For all samples a diffraction peak at  $2\theta = 6.47^\circ$  was detected and the  
6 corresponding  $d$ -spacing was calculated to be 1.364 nm using Bragg's equation [ $n\lambda = 2d \sin \theta$ , where  $\lambda$   
7 is the incident X-ray wavelength ( $\lambda = 1.540598 \text{ \AA}$ )].

8 We have investigated the characteristics of the OTFTs with **3b** (naphthalene diimide core) as  
9 channel layer material on bare SiO<sub>2</sub>, OTS-treated SiO<sub>2</sub>, and CL-PVP at different substrate temperatures  
10 ( $T_s$ ) of 25, 45, and 60 °C and a representative bottom-gate, top-contact OTFT is shown in Fig. 5a. Fig.  
11 S4(a-c) show the representative output characteristics of the OTFTs prepared on bare SiO<sub>2</sub>, OTS-treated  
12 SiO<sub>2</sub>, and CL-PVP, respectively at  $T_s$  of 25 °C. The output characteristic curves for all the devices  
13 clearly showed typical  $n$ -type characteristics. The transfer curves of all the devices fabricated with  
14 different gate dielectrics and  $T_s$  are shown in Fig. 5(b-d), and the device performance parameters are  
15 listed in Table 1. Fig. 5b shows the transfer characteristics of the OTFT fabricated on a bare SiO<sub>2</sub> at  
16 various  $T_s$ . The electron mobility extracted from the saturation regime were  $6.2 \times 10^{-4}$ ,  $5.5 \times 10^{-4}$ , and  
17  $5.9 \times 10^{-6} \text{ cm}^2/\text{Vs}$  for the devices prepared at  $T_s$  of 25, 45, and 60 °C, respectively. Similarly, OTFT  
18 devices based on an OTS-treated SiO<sub>2</sub> (Fig. 5c) at  $T_s$  of 25, 45, and 60 °C, exhibited charge carrier  
19 mobilities of  $1.2 \times 10^{-2}$ ,  $1.7 \times 10^{-2}$ , and  $3.8 \times 10^{-3} \text{ cm}^2/\text{Vs}$ , respectively. The OTFT devices with an OTS-  
20 treated SiO<sub>2</sub> showed much better performance with respect to charge-carrier mobility and  $I_{\text{on}}/I_{\text{off}}$  than  
21 that of devices with bare SiO<sub>2</sub>. Devices with an OTS-treated SiO<sub>2</sub> exhibit  $I_{\text{on}}/I_{\text{off}}$  values in the range  $10^4$ -  
22  $10^5$  and electron mobilities varying from  $10^{-2}$  to  $10^{-3} \text{ cm}^2/\text{Vs}$ , which is almost 100 times higher than that  
23 of devices with a bare SiO<sub>2</sub>. The enhanced performance may be caused by lower charge-carrier trap  
24 density at the interface between the gate dielectric and semiconducting channel. The effect of treatment  
25 of the SiO<sub>2</sub> surface with a self-assembled monolayer has been discussed in literature both for  $p$ -type [29]

1 and *n*-type semiconductors [16c, 30]. The highest mobility  $\sim 0.02 \text{ cm}^2/\text{Vs}$  was achieved for the device  
2 fabricated at a substrate temperature of  $45 \text{ }^\circ\text{C}$  on an OTS-treated  $\text{SiO}_2$  gate dielectric. We also have  
3 analyzed devices with a polymer gate dielectric, poly(4-vinylphenol) (CL-PVP) (Fig. 5d) fabricated at  
4  $T_s = 25, 45, \text{ and } 60 \text{ }^\circ\text{C}$ . The devices showed mobilities of  $3.6 \times 10^{-4}$ ,  $1.9 \times 10^{-4}$ , and  $7.6 \times 10^{-5} \text{ cm}^2/\text{Vs}$ ,  
5 respectively, which are comparable or higher than those of bare  $\text{SiO}_2$  devices. It indicates that the newly  
6 synthesized NDI derivative is potential active candidate for device with inorganic as well as organic  
7 gate dielectrics and flexible substrates.

8 To demonstrate the potential of the synthesized material for application in advanced organic  
9 electronics, load-type inverters were fabricated and characterized. The inverter, which is the  
10 fundamental logic element in an integrated circuit (IC) that converts low input (logic 0) to high output  
11 (logic 1) or vice versa, was constructed using a **3b** based OTFT fabricated at  $T_s$  of  $45 \text{ }^\circ\text{C}$  with an OTS-  
12 treated  $\text{SiO}_2$ , the device exhibiting the best overall performance. A schematic diagram of the resistive-  
13 load inverter with an external resistive load ( $100 \text{ M}\Omega$ ) is shown in Fig. 6a. We measured the dynamic  
14 response characteristics of the inverter at frequencies of 1 (Fig. S7a), 0.5 (Fig. 6b), and 0.1 Hz (Fig.  
15 S7b), with a fixed  $V_{\text{DD}} = 50 \text{ V}$ . As shown in Fig. 6b, when a square wave of  $V_{\text{IN}}$  switching from 0 to 50  
16 V at a fixed frequency,  $V_{\text{OUT}}$  exhibits a dynamic switching response to  $V_{\text{IN}}$  with a transient behavior just  
17 after the changes in the  $V_{\text{IN}}$  [31]. However, it does not generate square-like output waves because of a  
18 large overlap capacitance [32], which could be avoided by using a high dielectric insulator and/or  
19 designing an appropriate device structure. Similar dynamic behavior was reported by Lee et al. in a  
20 resistance load-type organic inverter [33].

#### 21 **4. Conclusions**

22 In summary, we adopted the simplest one-pot synthetic procedure to obtain small aromatic core diimide  
23 molecules **3a** and **3b**. The best performance was observed from the **3b** device prepared using an OTS-  
24 treated  $\text{SiO}_2$  substrate. The AFM images and XRD patterns reveal that the morphology and crystalline  
25 nature of the **3b** film varies according to the surface properties and substrate temperature during

1 deposition. The highest electron mobility of  $\sim 0.02 \text{ cm}^2/\text{Vs}$  with an  $I_{\text{on}}/I_{\text{off}}$  of  $3.14 \times 10^4$  was achieved for  
2 the **3b** TFT. We also demonstrated decent dynamic inverter action from our **3b**-TFT-based resistive  
3 load-type organic inverter. Moreover, the synthesized NDI derivative is compatible with polymer gate  
4 dielectric and hence could be used for advanced organic electronic device applications.

## 5 **Appendix A. Supplementary material**

6  
7 Supplementary data associated with this article can be found in the online version.

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## Legends to Figures

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**Table 1.** Summary of performance of OTFTs with **3b**

**Figure 1.** Synthesis scheme of new small core aromatic diimides. The structure of **4a** and **4b** are taken from reference no. 19.

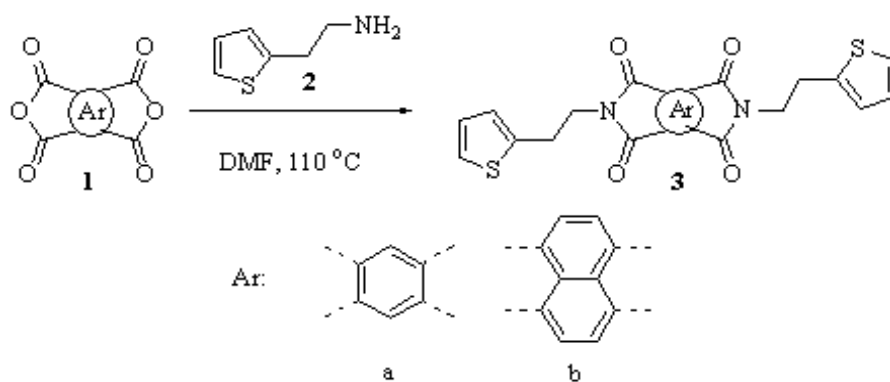
**Figure 2.** Electron density distribution in (a, b) LUMO and (c, d) HOMO of **3a** and **3b**, respectively. Left inset: the energy level diagram of synthesized material compared with reported diimides. \*These are taken from reference no. 19. Right inset: the CV curve of **3b** in DCM (scan rate: 50 mV s<sup>-1</sup>).

**Figure 3.** (a) Thermogravimetric analysis (TGA) of synthesized materials and (b) Differential scanning calorimetry (DSC) curve of **3a** and **3b** in nitrogen with a heating and scanning rate of 10 °C/min, respectively.

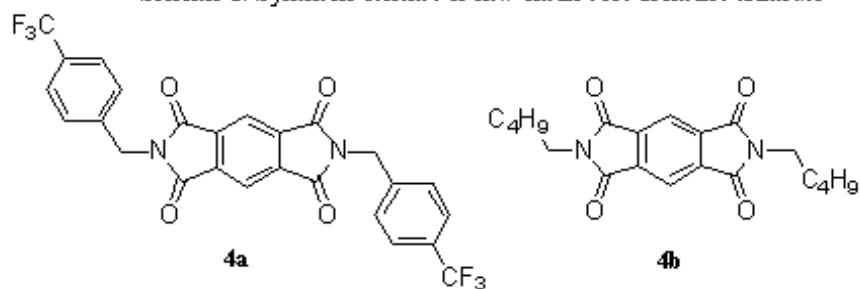
**Figure 4.** Tapping-mode atomic force microscopy (AFM) images of **3b** thin-film (50 nm) on (a) bare SiO<sub>2</sub>, (b) OTS treated SiO<sub>2</sub> and (c) CL-PVP at  $T_s = 45$  °C. (d) Corresponding out-of-plane XRD patterns at the same substrate temperature.

**Figure 5.** (a) Schematic diagram of bottom gate, top-contact OTFT devices. Transfer characteristics of OTFTs with **3b** at different substrate temperature ( $T_s$ ) ( $V_{DS} = 80$  V) on various gate dielectrics: (b) bare SiO<sub>2</sub>, (c) OTS-treated SiO<sub>2</sub> and (d) cross-linked polymer gate dielectric (CL-PVP).

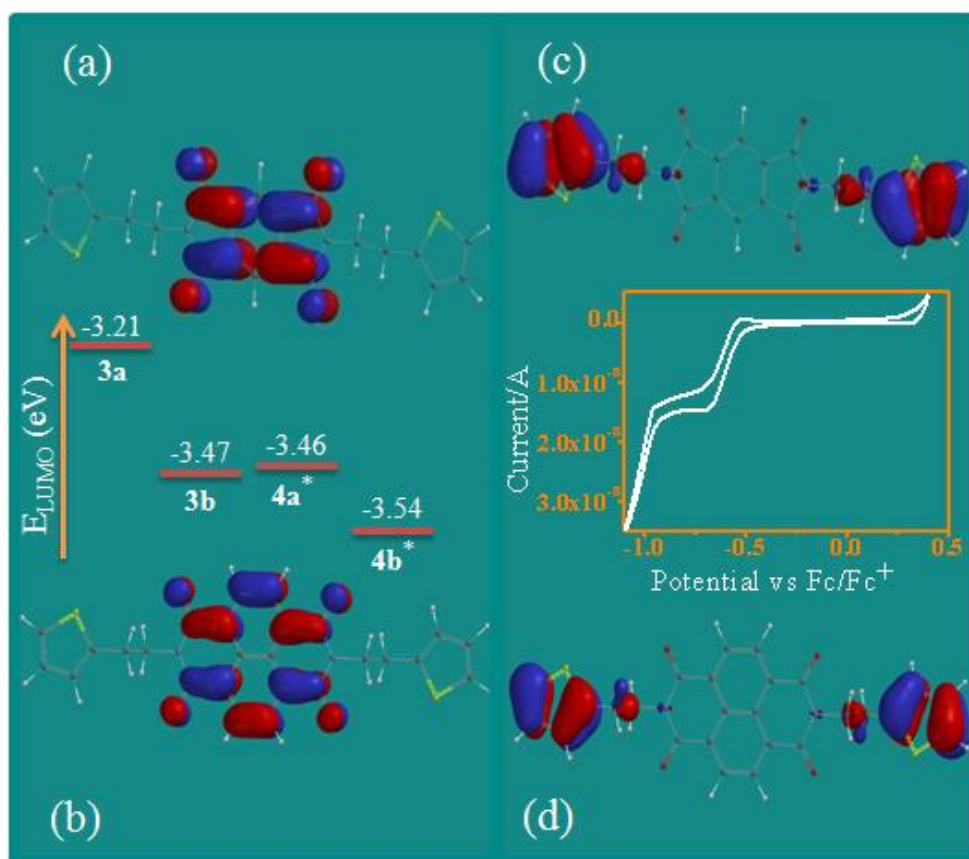
**Figure 6.** (a) Schematic diagram of resistance-load type inverter based on **3b** OTFT (OTS treated SiO<sub>2</sub> substrate,  $T_s = 45$  °C), (b) Dynamic behavior of the corresponding device operated under a supply voltage  $V_{DD} = 50$  V, resistor  $R = 100$  M $\Omega$  and input voltage pulse = 0.5 Hz.



**Scheme 1.** Synthesis scheme of new small core aromatic diimides



**Figure 1**



**Figure 2**

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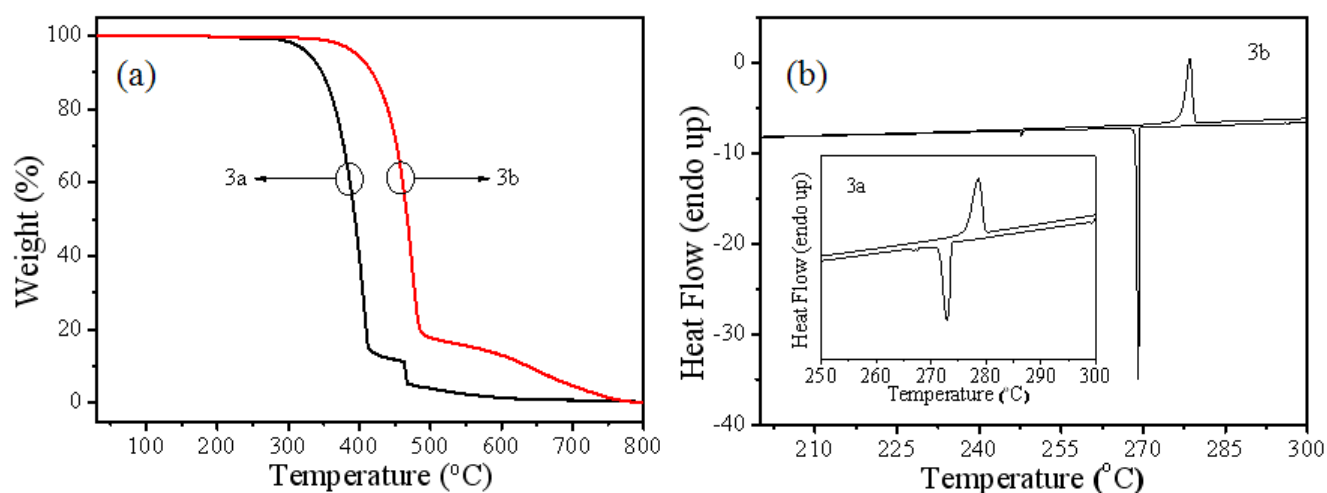


Figure 3

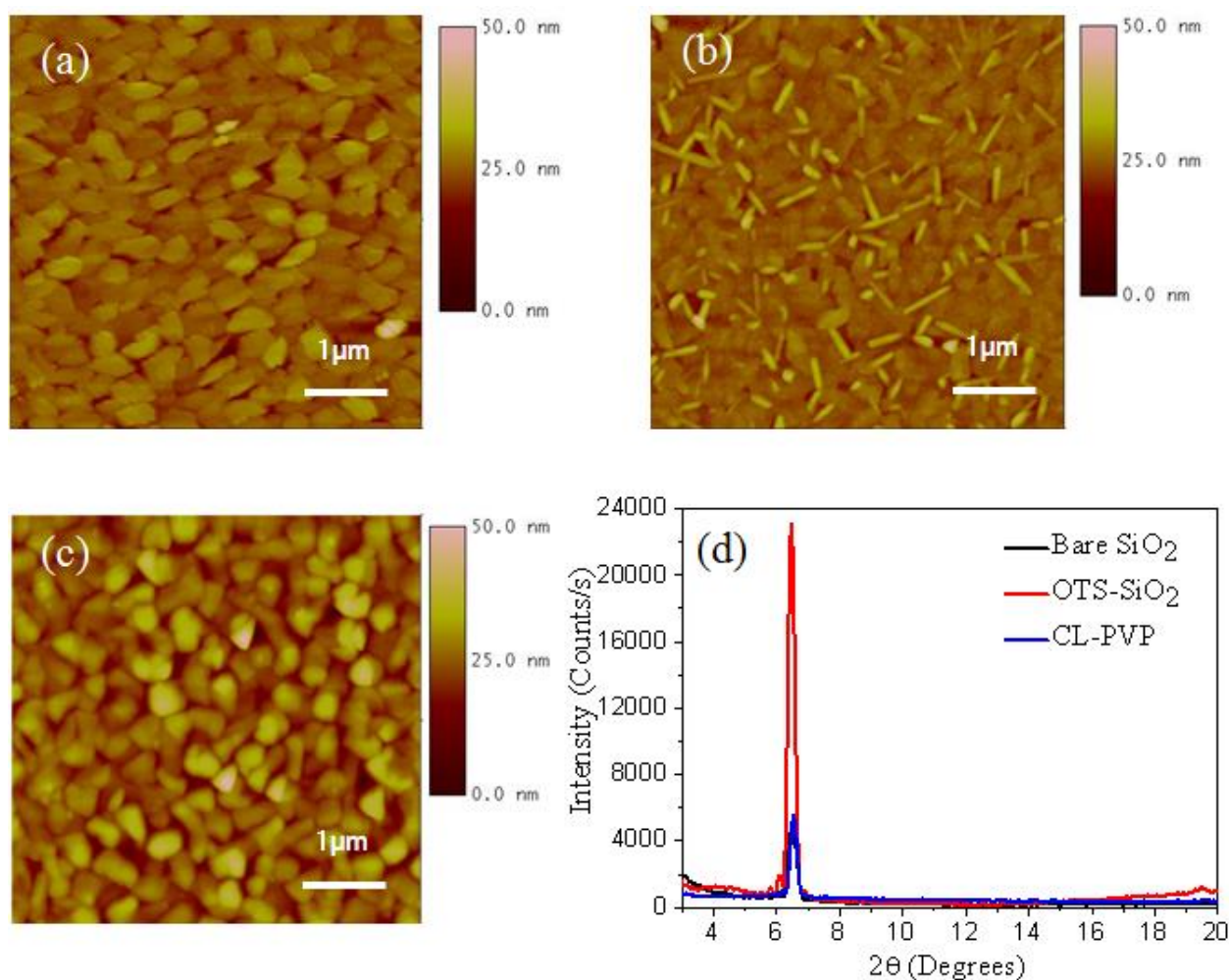


Figure 4

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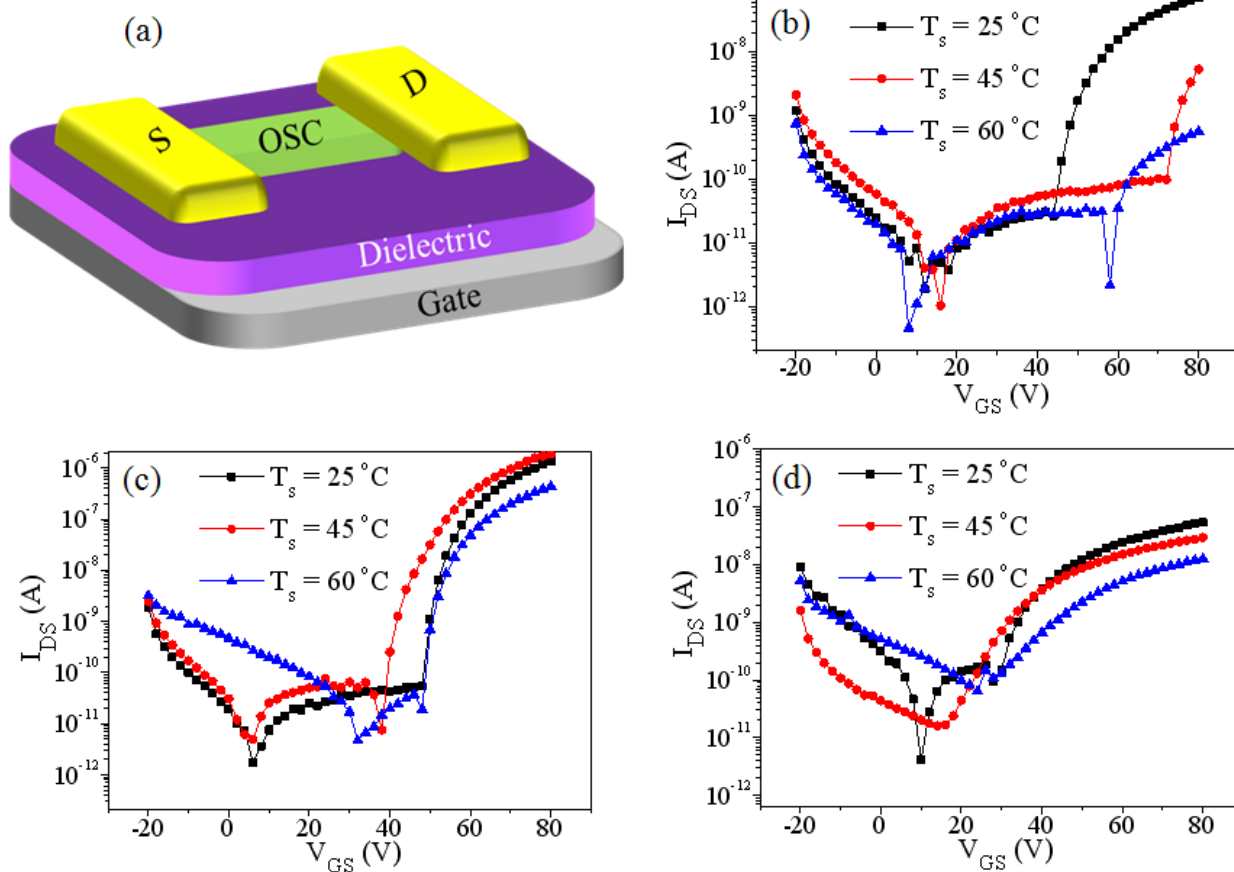


Figure 5

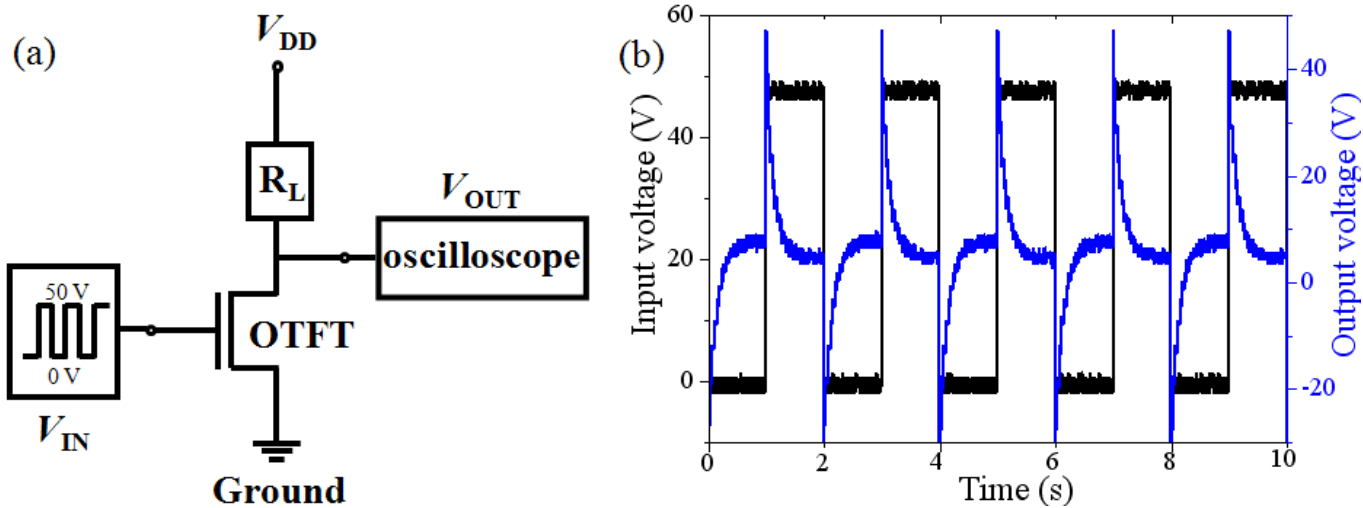


Figure 6

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1 **Table 1.** Summary of performance of OTFTs with **3b**

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Gate dielectric	Substrate Temp. (°C)	Mobility (cm <sup>2</sup> /Vs)	$I_{on}/I_{off}$ ratio	$V_{th}$ (V)	S.S (V/dec)
Bare SiO <sub>2</sub>	25	$6.2 \times 10^{-4}$	$3.8 \times 10^3$	44.8	2.4
Bare SiO <sub>2</sub>	45	$5.5 \times 10^{-4}$	$7.5 \times 10^1$	70.7	2.5
Bare SiO <sub>2</sub>	60	$5.9 \times 10^{-6}$	$1.6 \times 10^1$	50.0	13.9
OTS-SiO <sub>2</sub>	25	$1.2 \times 10^{-2}$	$1.2 \times 10^5$	51.0	1.3
OTS-SiO <sub>2</sub>	45	$1.7 \times 10^{-2}$	$3.1 \times 10^4$	48.2	2.1
OTS-SiO <sub>2</sub>	60	$3.8 \times 10^{-3}$	$2.5 \times 10^4$	50.1	3.4
CL-PVP*	25	$3.6 \times 10^{-4}$	$5.8 \times 10^2$	26.7	5.6
CL-PVP*	45	$1.9 \times 10^{-4}$	$3.4 \times 10^3$	26.5	5.6
CL-PVP*	60	$7.6 \times 10^{-5}$	$1.2 \times 10^2$	28.5	15.4

3 \*CL-PVP: Cross-linked poly(4-vinylphenol)

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