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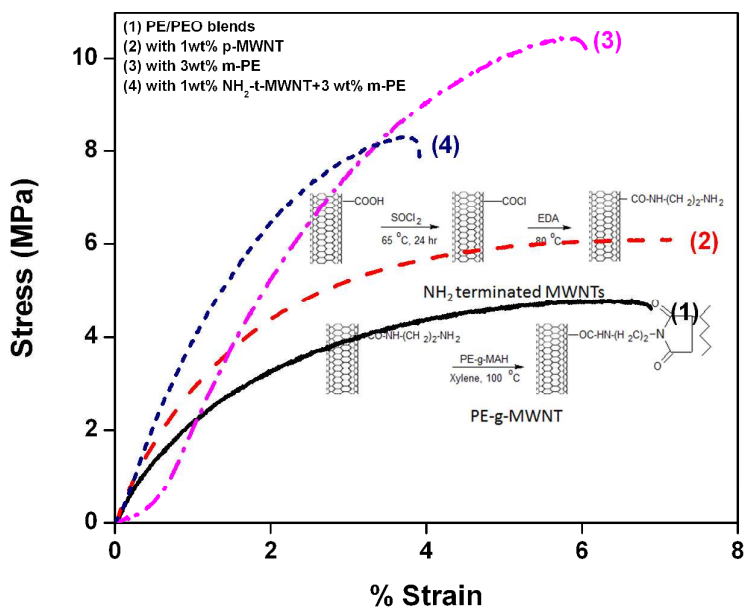
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Graphical Abstract



1 wt.% MWNT grafted with PE showed significant increase in structural properties in PE/PEO blends.

PE/PEO blends compatibilized by PE brush immobilized on MWNTs: improved interface and structural properties

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Abstract

Polyolefin based blends have tremendous commercial importance in view of their exceptional properties. In this study the interface of a biphasic polymer blend of PE (polyethylene) and PEO (polyethylene oxide) has been tailored to reduce the interfacial tension between the phases and to render finer morphology. This was accomplished by employing various strategies like addition of maleated PE (PE grafted maleic anhydride), immobilizing PE chains, *ex-situ*, onto MWNTs by covalent grafting, and *in-situ* grafting of PE chains on to MWNTs during melt processing. Multiwalled nanotubes (MWNTs) with different surface functional groups have been synthesized either *a priori* or were facilitated during melt mixing at higher temperature. NH₂ terminated MWNTs were synthesized by grafting ethylene diamine (EDA) on to carboxyl functionalized carbon nanotubes (COOH-MWNTs) and further, was used to reactively couple with maleated PE to immobilize PE chains on the surface of MWNTs. The covalent coupling of maleated PE with NH₂ terminated MWNTs was also realized *in-situ* in the melt extruder at high temperature. Both NH₂ terminated MWNTs and the *in-situ* formed PE brush on MWNTs during melt mixing, revealed a significant improvement in the mechanical properties of the blend besides remarkably improving the dispersion of the minor phase (PEO) in the blends. Structural properties of the composites were evaluated and the tensile fractured morphology was assessed using scanning electron microscopy.

Keywords:- PE/PEO blends, polymer brush, PE-g-MWNT, structural properties

Introduction

Blending of polymers is a versatile tool to achieve targeted properties with the existing polymers.^{1, 2} In general, most polymeric blends are often thermodynamically immiscible in nature due to small gain in entropy on mixing. The final morphology of the blends is contingent on composition, viscosity ratio, interfacial tension, processing etc., resulting in a wide variety of microstructures such as matrix- droplet, fibrillar, lamellar or continuous. A compatibilizer like a block or a graft co-polymer is often added to reduce the interfacial tension between the constituents and to generate a finer morphology.³⁻⁵ For instance, Hashmi et al.,⁶ showed that the addition of styrene-butadiene-styrene (SBS) copolymer to polypropylene/polystyrene blends exhibited a finer morphology with respect to blends without SBS. Meltzer et al.,⁷ observed that addition of maleated PE to PE/PEO blend led to a significant refinement in morphology due to reactive compatibilization between maleic anhydride (MAH) and the hydroxyl ends of PEO.

Polyolefins represent a class of important commodity polymers because of its good processability, low cost and good chemical resistance and hence, has tremendous commercial importance. Current research focuses to develop polyolefin based new materials with desired performance in view of its commercial importance. However, due to inherent immiscibility with other macromolecules, the desired properties are often not realized by mere blending. Further, the nonpolar nature of the polyolefins add to the challenge of compatibilizing polyolefinic blends. Therefore, various methods have been explored in the recent past to compatibilize polyolefinic blends.

The different morphologies generated during melt processing of polymer blends can be tailored to explore various strategic applications. Among the different morphologies, the matrix-droplet type offer a unique route to design substrates for separation technology. Polymeric membranes, in general, are derived from phase inversion process, thermally induced phase separation (TIPS), track etching or by solvent free methods like stretching melt-cast polymer films etc. While these methods have their own merits and demerits, an alternative route in designing polymer based membranes, which has recently attracted lot of attention, is by selectively etching one of the phases from a bi-phasic blend.^{7 8}

Recent studies report the key role of nanoparticles (NPs) in compatibilizing binary blends. However, the underlying mechanism is significantly different from the classical approach (i.e. block/graft copolymer) of compatibilizing binary blends. Many factors such as surface free

energy of the component, melt viscosity, processing parameters etc. govern the localization of NPs in a binary blend. Besides rendering finer microstructures, the NPs also offer unique functional properties which make them a potential candidate in designing polymer based nanocomposites. Among the NPs, carbon nanotubes (CNTs) have attracted a great deal of interest owing to their exceptional properties. While the majority of the work reported till date focuses on the electrical, mechanical and thermal properties of the composites, their key role in stabilizing binary blends has also been realized recently.

Therefore, in this study, we have determined the mechanical properties and dispersion of the phases in the presence of functionalized multiwalled nanotubes (MWNTs). The mechanical properties and the morphology of the blends were determined. We have attempted to reduce the interfacial tension between the phases by employing different strategies. These include the addition of PE grafted maleic anhydride, *ex-situ* immobilizing PE chains on to MWNTs and *in-situ* grafting of PE chains on to amine functionalized MWNTs. We have compared the various strategies and determined the blend that exhibits enhanced mechanical properties and shows increased dispersion of PEO in PE. The reasons for these observations are also discussed in detail.

Experimental

Materials

Low density polyethylene (PE) (melt flow index of 25 g/10 min and density of 0.925 g/cc), polyethylene oxide (PEO, $M_w = 400,000$), maleated polyethylene (PE grafted with maleic anhydride), and ethylene diamine (EDA) were obtained from Sigma Aldrich. The pristine MWNTs (p-MWNTs, NC 7000 of 90 % purity and 1.5 μm long) and acid functionalized MWNTs (COOH-MWNTs, NC 3150 of 95 % purity and $< 1 \mu\text{m}$ long) were obtained from Nanocyl, Belgium. Solvents like tetrahydrofuran (THF), xylene, and thionyl chloride (SOCl_2) were obtained from Merck, India and were used without further purification.

Synthesis of NH_2 terminated MWNTs

NH_2 terminated MWNTs were synthesized from carboxyl (-COOH) functionalized MWNTs using EDA; a convenient method that has been widely applied in functionalizing MWNTs. Typically, 800 mg of MWNT-COOH was suspended in THF and was mixed with 80 mL of

SOCl_2 in a 250 mL round bottomed flask. The mixture was then stirred at 80 °C for 24 h in presence of EDA. The powder was isolated by vacuum filtration and was washed with anhydrous THF to remove excess SOCl_2 . This purification cycle was repeated couple of times to ensure complete removal of excess SOCl_2 and the final product was vacuum dried for 24 h at 80 °C.

Immobilizing PE on the surface of MWNTs

PE grafted MWNT were prepared by grafting maleated PE on to NH_2 terminated MWNTs via an amidation reaction between the amine groups of NH_2 terminated MWNTs and the maleic anhydride groups of maleated PE. In a typical reaction, 50 mg of NH_2 terminated MWNTs was initially suspended in 10 mL of xylene. The mixture was sonicated for half an hour to exfoliate the nanotubes. 120 mg of maleated PE was then added to this mixture and was refluxed at 100 °C under N_2 atmosphere. The mixture was then filtered with a 0.22 μm PTFE membrane. The collected powder was extracted with 200 mL of boiling xylene for 48 h to remove un-grafted maleated PE, and dried at 90 °C under vacuum overnight, thus immobilizing PE chains on the surface of MWNTs (PE-g-MWNTs).

Characterization of NH_2 terminated MWNTs and PE-g-MWNT

X-ray Photon Scattering (XPS) scans were recorded on a Kratos Analytical instrument using Al monochromatic source (1.486 keV) and the Fourier transform infrared (FTIR) spectra were recorded on a Thermo Nicolet 6700 in the wavelength range of 4000–400 cm^{-1} using KBr pellets. TGA was carried out under N_2 atmosphere with a Pyris Diamond analyzer at a heating rate of 10 °C min^{-1} . Transmission Electron Microscopy (TEM) images were acquired using Tecnai G2 F30 at 300 KV.

Preparation of compatibilized blends

Blends with and without compatibilizers (maleated PE), and MWNTs (p-MWNT, NH_2 terminated MWNTs and PE-g-MWNT) were prepared by melt mixing under N_2 atmosphere in a laboratory scale twin screw extruder (Polylab, Thermohaake Minilab II) at 150 °C and 60 rpm for 20 min. The mini extruder has a recirculation channel which ensures complete mixing for

desired time before the material can be taken out. Prior to melt mixing all the samples were vacuum dried overnight.

Characterization of blends

For tensile properties, dumbbell shaped specimens were prepared by compression molding at 150 °C. The tensile tests were done on Instron Universal Testing Machine at room temperature with crosshead speed of 5 mm.min⁻¹. Morphological analysis for tensile fractured samples were assessed using field emission scanning electron microscopy (FE-SEM) on a ULTRA 55, FESEM, Carl Zesis with accelerating voltage of 5 KV. The morphology of the as pressed films were analyzed using SEM. Prior to SEM, the films were etched with cold sterilized distilled (DI) water to remove the PEO phase.

The melting temperature (T_m), crystallization temperature (T_c) and % crystallinity (X_c) of PE (normalized with respect to weight and weight fraction) was measured using a Mettler Toledo DSC instrument with a heating and cooling rate of 10 °C/min. The degree of crystallinity of PA6 phase was calculated from the heat of fusion of second heating cycle. The heat of fusion (ΔH_m) of PE phase was normalized to the fraction of polymer present in the blends. The degree of crystallinity (X_c) of PE phase was determined from the ratio of normalized heat of fusion (ΔH_m , norm) to the heat of fusion of 100% crystalline PE, (ΔH_f^0), which was taken as 293 J/g⁹.

Results and discussion

Synthesis and characterization of NH₂ terminated MWNTs and PE-g-MWNT

As shown in Scheme 1, a two-step procedure was adopted to prepare NH₂ terminated MWNTs. It consisted of NH₂ attachment on to MWNT surface by acylation with thionyl chloride (SOCl₂) at 65°C for 24 h. This was followed by amidation of the resulted MWNT-COCl with an excess of EDA at room temperature for 48 h to complete the functionalization process. The excess of EDA was then washed with anhydrous THF and filtered using 0.22 μm PTFE membrane. In the second step, maleated PE was coupled to NH₂ terminated MWNTs through the formation of amide groups, yielding PE-g-MWNTs, as shown in Scheme 1.

The direct evidence for EDA functionalized MWNTs and successful grafting of PE was provided by FTIR and XPS. Figure 1a-b illustrates the FTIR and XPS scans of NH₂ terminated MWNTs

and PE-g-MWNTs. A broad peak at 3424 cm^{-1} in the FTIR spectra indicates the presence of (-OH) hydroxyl groups in MWNT-COOH. Upon coupling with EDA, a peak appears at 1653 cm^{-1} indicating the amide carbonyl (C=O) stretch. Similar observations were reported earlier^{10, 11}. The presence of C-N stretching vibrations at 1044 cm^{-1} and the symmetric stretching at 3359 cm^{-1} confirms the presence of saturated primary amine on the MWNT surface. In the FTIR spectrum of PE-g-MWNTs, peaks at 1375 cm^{-1} and 1455 cm^{-1} are assigned to bending vibration of CH_3 , while peak at 1719 cm^{-1} indicates C=O stretching vibration of maleic anhydride. In addition, peak at 2916 cm^{-1} shows stretching vibrations of PE. A shift of C=O stretching vibration from 1709 to 1656 cm^{-1} confirms the covalent bonding of maleated PE on to MWNT surface^{12, 13}.

Functionalization of MWNTs with EDA is further confirmed by XPS. The presence of N1s peak in functionalized MWNTs indicates the presence of amine. Furthermore, N1s peak of NH_2 terminated MWNTs was fitted to deconvolute the two species (see Figure 1b). The lower binding energy (BE) peak at 396.9 eV is attributed to terminal amine and the higher BE peak at 397.8 eV could be attributed to the amide nitrogen. The latter showed higher BE due to its electropositive nature and manifests the electron withdrawing nature of the carbonyl group^{14, 15}.

In general, immobilization of macromolecular chains on MWNTs can influence their thickness, which can be characterized by TEM and the fraction of polymer tethered on the MWNT surface can be evaluated by TGA. Figure 1c shows the TEM micrographs of MWNT-COOH and PE-g-MWNTs. The latter showed a layer of polymer on the MWNTs as manifested from its increased diameter. This can result in improved dispersion of MWNTs in the PE phase of the blends. Figure 1d shows the TGA scans for MWNT-COOH, maleated PE and PE-g-MWNT. At $550\text{ }^\circ\text{C}$, the relative weight losses for MWNT-COOH, maleated PE and PE-g-MWNTs are 17% , 91 and 54 wt.% respectively. Large weight loss of PE-g-MWNTs is attributed to the thermal degradation of attached PE polymer chains. Thus confirming the maleated PE being grafted successfully on MWNTs¹³.

Structure-property relationship in PE-g-MWNTs compatibilized PE/PEO blends

An efficient route to tailor the interface in binary polymer blends is by adding a compatibilizer. Next to the classical block copolymers, recent studies have revealed the key role of NPs in compatibilizing binary blends as discussed in the introduction section. In our earlier work, we have shown the potential of MWNTs in suppressing the coalescence of PMMA droplets in the P α MSAN matrix and the refinement of bi-continuous structures in PA6/ABS blends. Although,

MWNTs render finer morphology irrespective of its localization in the blends, the mechanism is very different from the classical block copolymer route. It is understood from one of our previous work¹⁶ that MWNTs localize in the thermodynamically less favored phase (PE) in PE/PEO blends which is driven by the flow characteristic of the components (the melt viscosity of PEO is significantly higher than PE). The key role of MWNTs as a compatibilizer may not be realized in PE/PEO blends due to lack of interfacial adhesion of MWNTs with the non-polar PE phase. Hence, in this study, we adopted two different approach to compatibilize PE/PEO blends: by *in situ* and *ex situ* immobilization of PE chains on to MWNTs as discussed before. However, both *in situ* and *ex situ* grafting of PE chains on to MWNTs can serve as an effective compatibilizer in PE/PEO blends. Recall, that the *in situ* grafting of PE chains on to MWNTs was realized by reactive coupling of maleated PE with NH₂ terminated MWNTs during melt-mixing and the *ex situ* was realized by grafting maleated PE with NH₂ terminated MWNTs *a priori* before melt mixing. Blends with pristine MWNT (p-MWNTs) was also prepared and investigated for their compatibilizing effects in PE/PEO blends.

The improved compatibility is revealed by SEM and is illustrated in Figure 2. To identify the minor phase, the specimens for SEM were immersed in DI water so that the dispersed PEO phase appears as holes in the PE matrix. Blends of PE/PEO are immiscible in nature and typically exhibit coarse morphology (see Figure 2a). Clearly, the PEO holes in PE/PEO blends with different MWNTs are much smaller than those of the neat blends, indicating that PEO is more finely dispersed in PE/PEO/MWNT blends. Interestingly, with addition of 3 wt% maleated PE in PE/PEO blends, the size of the dispersed PEO phase drastically reduced (see Figure 2b and the inset of Figure 2b) in striking contrast to the neat blends. This reduction in the PEO droplets clearly indicate the role of maleated PE in compatibilizing PE/PEO blends by possibly reducing the interfacial tension between PE and PEO¹⁷. More interestingly, the *in situ* grafting of maleated PE on to NH₂ terminated MWNTs, through amidation during processing, has led to finer dispersion of PEO droplets in PE (see Figure 2d) and which is not realized with p-MWNTs (see Figure 2c). A schematic description of the compatibilizing mechanism of maleated PE and *in situ* grafting of PE on to MWNTs is displayed in Scheme 2. Due to high interfacial tension, macrophase separation exists in uncompatibilized PE/PEO blends. With the introduction of maleated PE and PE-g-MWNTs, the interface is modified thereby reducing the unfavorable interactions. Consequently, the *ex situ* grafting of PE chains on to MWNTs led to better

dispersion of MWNTs in the PE phase and the combination of maleated PE further resulted in a finer dispersion of PEO droplets in the blends (see Figure 2e-f).

The reinforcing effect of MWNTs, maleated PE and PE-g-MWNTs is evaluated using tensile tests. Figure 3 demonstrates the typical stress-strain curves for uncompatibilized and compatibilized by maleated PE and PE-g-MWNTs. Both PE and PEO show ductile behavior during uniaxial tensile test (not shown here). The control PEO exhibit a yield point followed by cold drawing and strain hardening (not shown here) whereas, the 70/30 PE/PEO blends exhibit a typical brittle failure with poor ultimate tensile strength and modulus. This is due to poor interfacial adhesion between PE/PEO. The corresponding tensile fracture morphology is shown in Figure 4a. Clearly, the debonding at the interface is a cause of premature failure, thus lowering the mechanical properties with respect to the components (see Table 1). This is discussed in more details in the next section. A classical route to reduce the interfacial tension and improve the stress transfer at the interface is by compatibilizing¹⁸. In the present work, we employed two different approaches; *in situ* and *ex situ* grafting of PE chains on to MWNTs. The use of maleated PE showed an increase of 109% in the ultimate tensile strength and 106% in the Young's modulus (YM) in the blends. This dramatic improvement in the mechanical properties is attributed to the specific interactions between maleic anhydride (MAH) and the terminal hydroxyl (OH) groups in PEO⁷.

Mechanical properties of the blends largely depend on the phase morphology and interfacial adhesion between the constituents. It is envisaged that the elongation at break is an important parameter to judge the interfacial adhesion between the phases, whereas the tensile strength is an indication of finer morphology, domain size and size homogeneity¹⁹. Incorporation of p-MWNTs and NH₂ terminated MWNTs showed an increase in YM by 55% and 178% respectively; and in the ultimate tensile strength by 20% and 60% respectively. Although the MWNTs exhibit a reinforcement effect in the blends but the elongation at break is greatly sacrificed. This observation is a clear mandate to the fact that premature failure originates from the CNT aggregates¹⁹ and due to lack of interfacial adhesion with the matrix (here PE) polymer. Interestingly, NH₂ terminated MWNTs showed higher mechanical strength than the p-MWNT and can be attributed to possible interfacial interactions with PEO as well thereby acting as an interfacial modifier. In order to improve the interfacial adhesion and to tailor the interface, *in situ* grafting of maleated PE on to NH₂ terminated MWNTs was facilitated during processing. The

thus formed PE-g-MWNTs showed exceptionally high YM (186%) and ultimate tensile strength (60%) and more interestingly, retained the elongational properties. It is observed that for 1 wt.% of PE-g-MWNT an increase of 78% in ultimate tensile strength and 143% in YM is recorded.

In order to support the observations above and to understand the mechanism of fracture, the tensile fractured morphology is investigated in detail here and is shown in Figure 4a-e. The tensile fractured surfaces of the neat 70/30 PE/PEO blends reveal that PEO droplets is embedded in the PE matrix and with a weak interface. PE is a semicrystalline and ductile material, the lack of interfacial adhesion between PE and PEO caused an early failure with respect to the constituents. Peterlin²⁰ proposed that the component which has low strength and ductility (here PE), comprises of stacks of parallel lamellae with fewer links in-between which in turn deform plastically and spreads to maximum compliance before fracture. The typical failure in ductile semicrystalline polymer comprises of fibrillar morphology in the final stage of deformation. Hence, the fibrillar morphology observed in Figure 4b can be related with the fact that the PE phase shares the most load and due to poor interfacial adhesion, the stress transfer at the interface is relatively poor. Two interesting observations are noted here. Firstly, with the addition of maleated PE, the fibrils appear to be thinner with respect to the fractured surfaces of the neat blends manifesting the compatibilization rendered in the PE/PEO blends. Secondly, the fine dispersion of PEO phase in the PE matrix.^{19, 21}

Incorporation of MWNTs in the blends led to reduction in the fibrillar morphology again suggesting poor interfacial adhesion between the phases. This observation is also reflected in the elongational properties of the blends. It is now understood that both fibrillar morphology and finer distribution of PEO phase are important to obtain improved mechanical properties in PE/PEO blends. In addition, MWNTs improve the YM of the blend significantly in striking contrast to the improvement in ultimate tensile strength. The latter property is strongly contingent on the interfacial adhesion between the phases; however, the reinforcement effected by MWNTs is well realized by the improvements in YM. A combination of MWNTs and maleated PE could result in simultaneous improvement in YM and ultimate tensile strength provided PE grafts are formed on the surface of MWNTs during melt mixing. In order to realize this, we compared the mechanical properties of the blends with *ex situ* PE-g-MWNTs with that of the *in situ* formed PE grafts on to MWNTs during processing. Interestingly, although *ex situ*

formed PE-g-MWNTs are relatively well dispersed in the PE phase, the *in situ* formed PE-g-MWNTs resulted in significant improvement in YM and ultimate tensile strength. These observations are also supported by the tensile fractured morphology where finer dispersion of PEO phase is well evident next to improved interfacial adhesion, as seen in Figure 5 where *in situ* formed grafts were observed to bridge the phases.

Melting and Crystallization behavior of the blends: effect of MWNTs

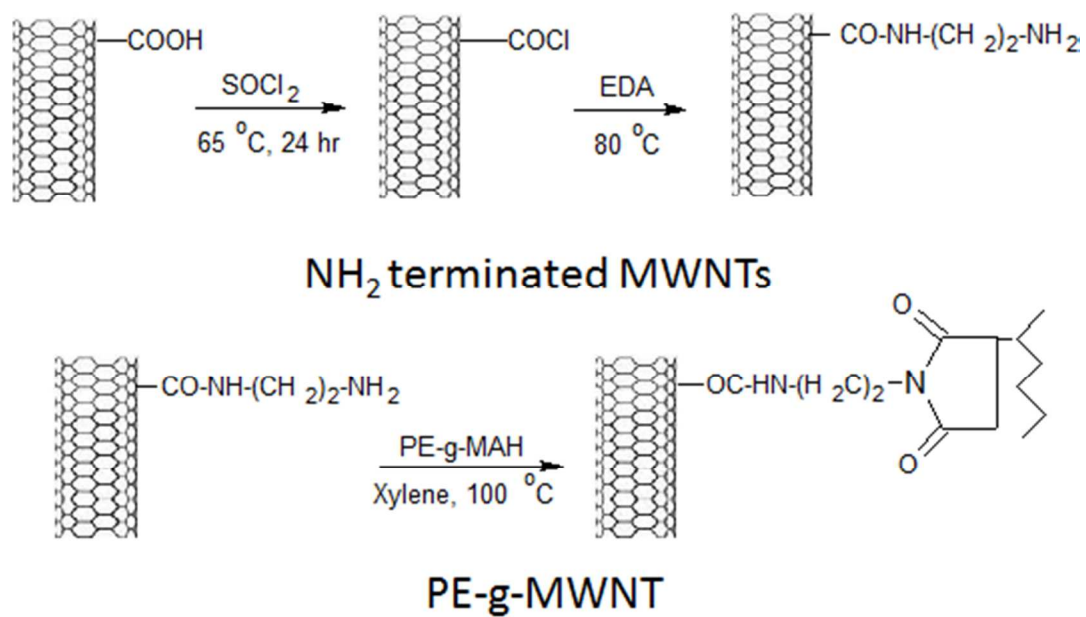
Figure 6 shows the cooling (Figure 6a) and the heating scans (Figure 6b) of PE/PEO blends and the melting and crystallization temperatures for various blends investigated here are listed in Table 2. The corresponding exotherms of PE and PEO is well evident in the cooling scans of the blends; a typical characteristic of un-compatible blend²². The crystallization temperatures (T_c) of PE and PEO are observed to be 100.4 °C and 44.8 °C, respectively. Two interesting observations draw our attention. Firstly, with addition of 3 wt.% maleated PE in PE/PEO blends, the T_c of PEO has been delayed by ca. 10 °C with respect to neat blends while the T_c of PE remains unchanged. Secondly, we observed fractionated crystallization in PE/PEO blends in the presence of maleated PE. The melt interfacial reaction between maleated PE and the terminal hydroxyl group of PEO possibly has led to delayed crystallization in PEO²³. It is worth recalling that maleated PE has remarkably improved the dispersion of PEO in PE matrix (see Figure 2d). It is envisaged that the heterogeneities are strongly contingent on the droplet size. For instance, the smaller droplets might contain less efficient heterogeneities and will only nucleate at the temperature when they become active.²³ A moderate decrease in the melting temperature of PEO (see Figure 6b) also suggests amorphous-amorphous miscibility in the presence of maleated PE. In the presence of MWNTs, no appreciable change is observed in the crystallization temperature of either PE or PEO. It is envisaged that CNTs act as hetero nucleating agent and influence the rate of crystallization in semicrystalline polymer^{24,25}. In this case, MWNTs or modified MWNTs did not show any appreciable change in the crystallization and melting temperature. Hence, it can be concluded that the observed changes in the structural properties are closely related with the strengthened interface on account of melt-interfacial interactions.

Conclusions

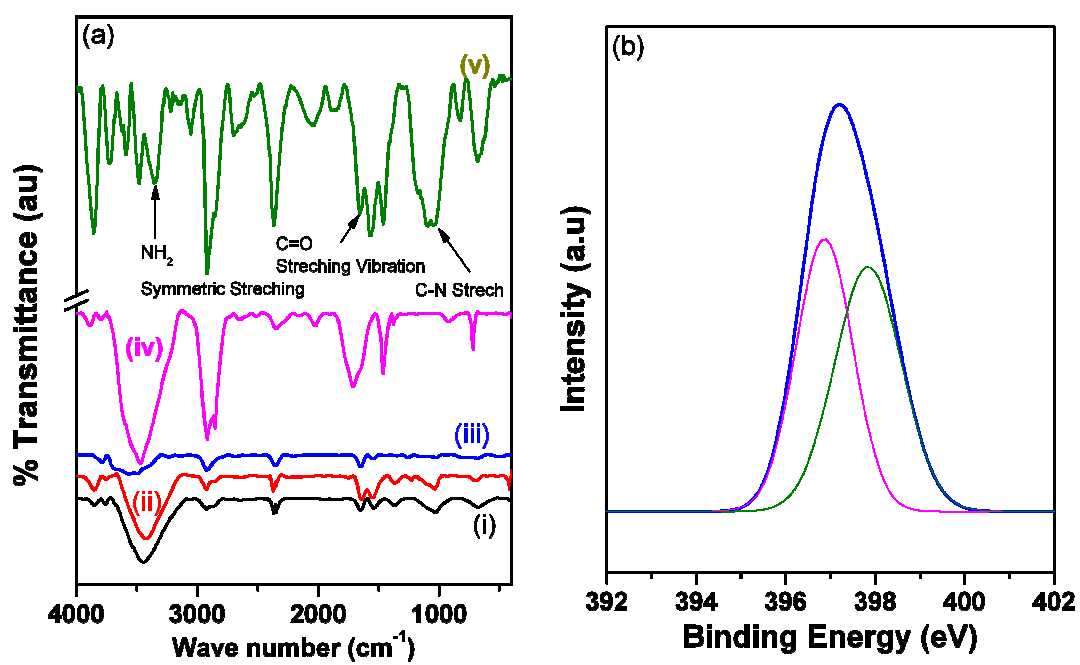
In order to render finer morphology in the blends, various routes for compatibilization such as addition of maleated PE (PE grafted maleic anhydride), immobilizing PE chains, *ex-situ*, onto MWNTs by covalent grafting, and *in situ* grafting of PE chains on to MWNTs during melt processing were adopted. Blends with modified MWNTs revealed a significant improvement in the mechanical properties of the blend besides remarkably improving the dispersion of the minor phase (PEO) in the blends. Blends with *in situ* formed PE-g-MWNTs showed exceptionally high Young's modulus (186% higher with respect to neat blends) and ultimate tensile strength (60% as compared to neat blends) and more interestingly, retained the elongational properties.

Acknowledgements

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Scheme 1. Immobilizing PE on to MWNTs via reactive coupling of maleated PE with NH₂ terminated MWNTs



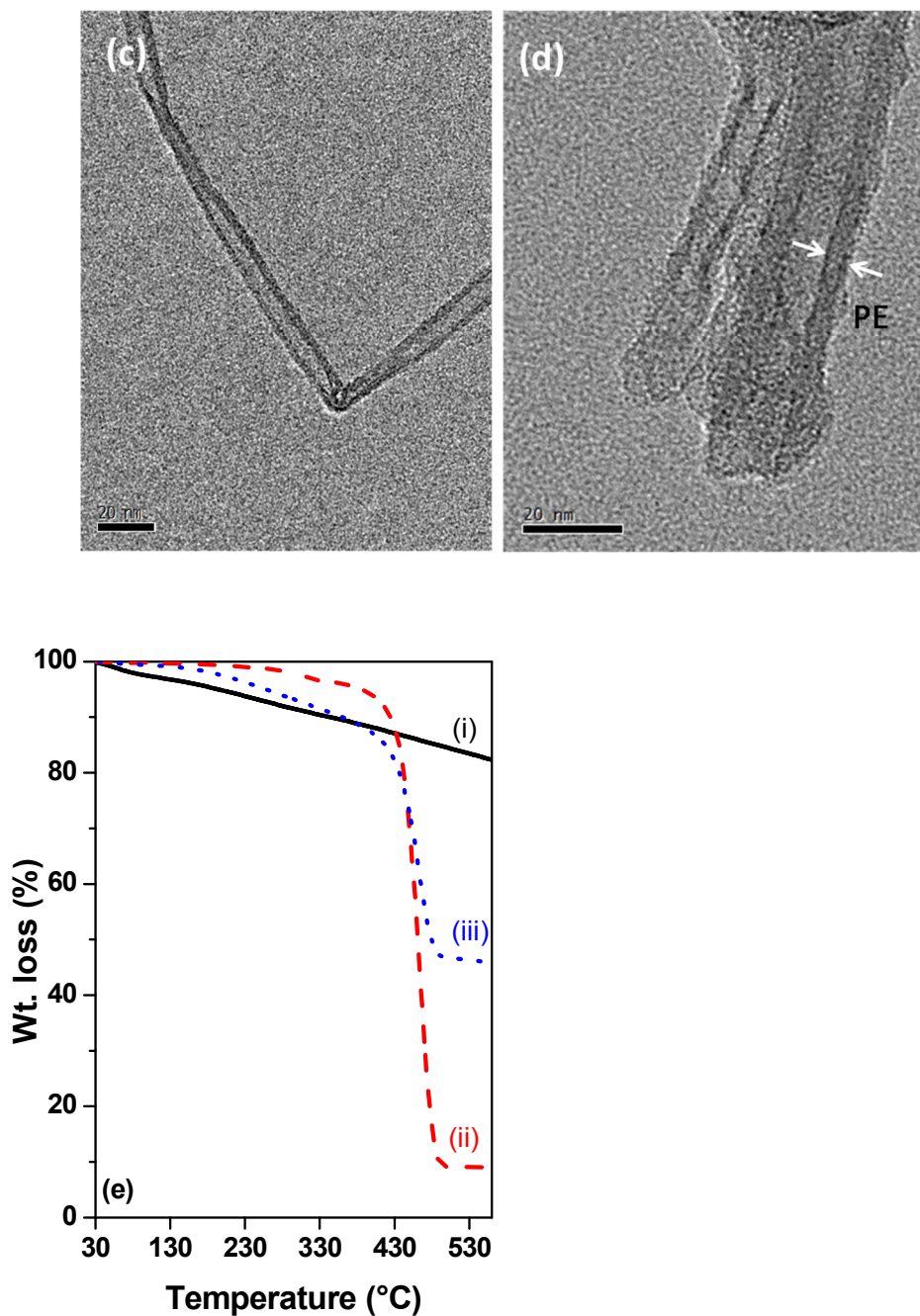
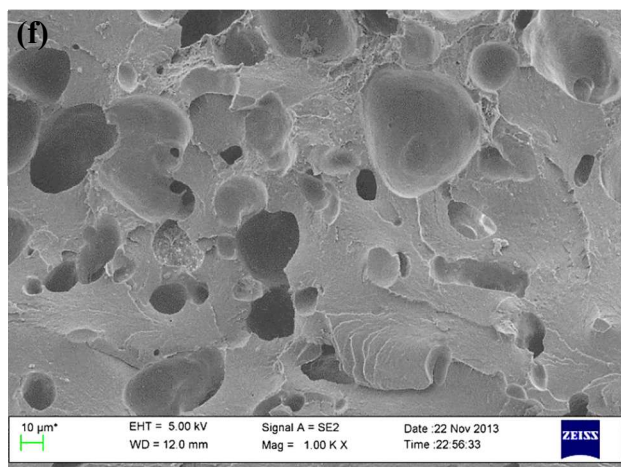
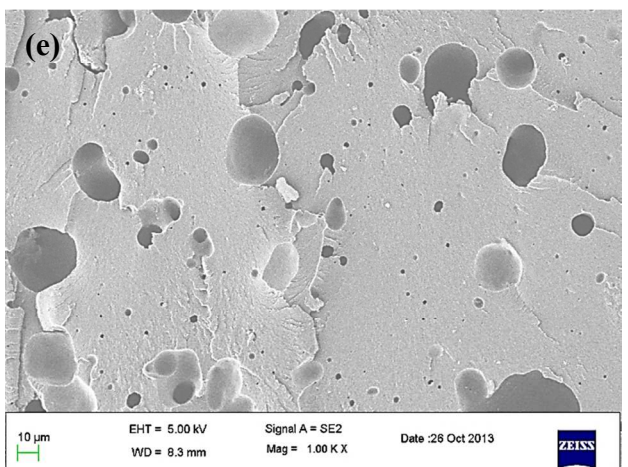
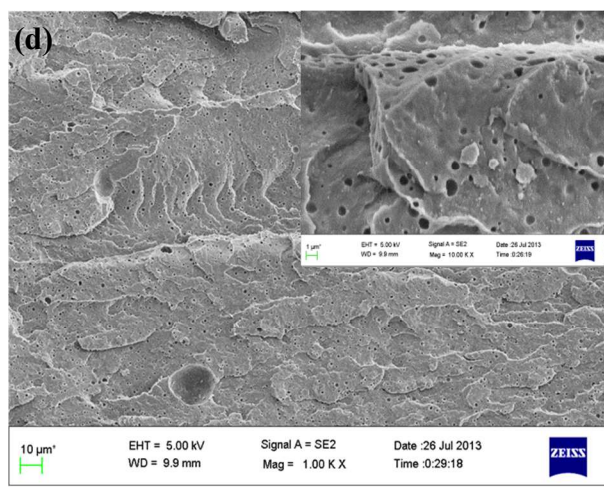
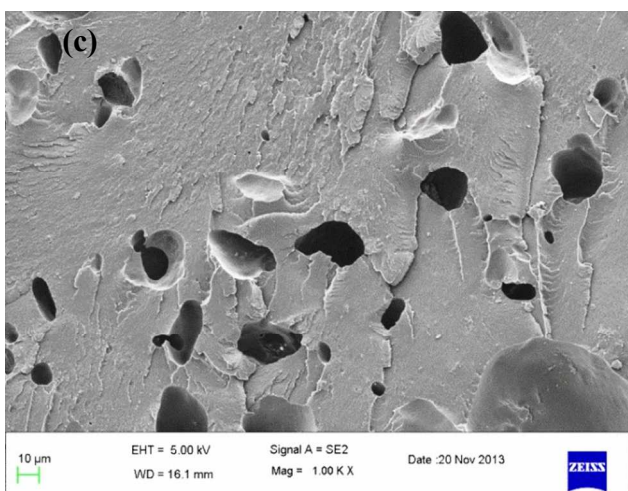
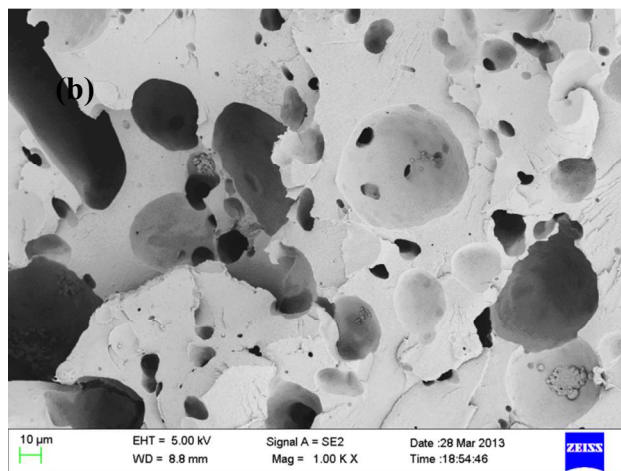
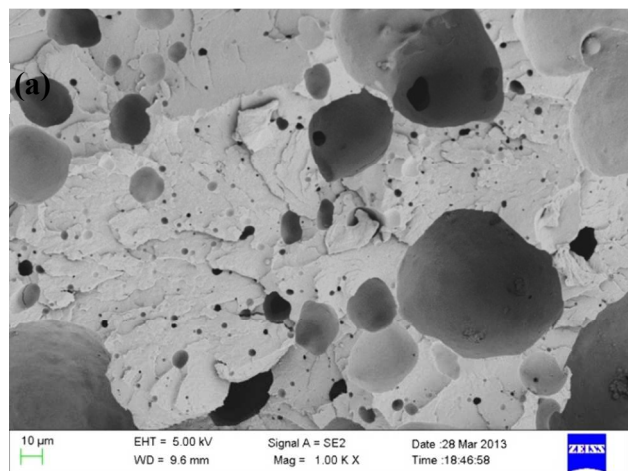


Figure 1 (a) FT-IR spectra of (i) p-MWNTs, (ii) MWNT-COOH, (iii) NH_2 terminated MWNTs, (iv) maleated PE and (v) PE-g-MWNT; (b) XPS scans of NH_2 terminated MWNTs; TEM image of (c) MWNTs-COOH and (d) PE-g-MWNT (*ex situ*); (e) TGA thermogram of (i) neat COOH-MWNT (ii) maleated PE and (iii) PE-g-MWNT.



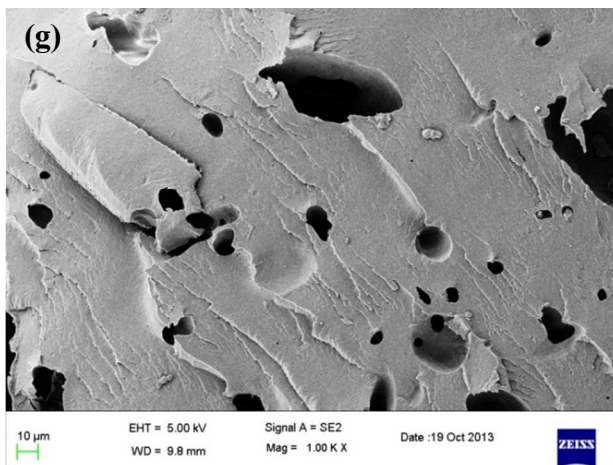
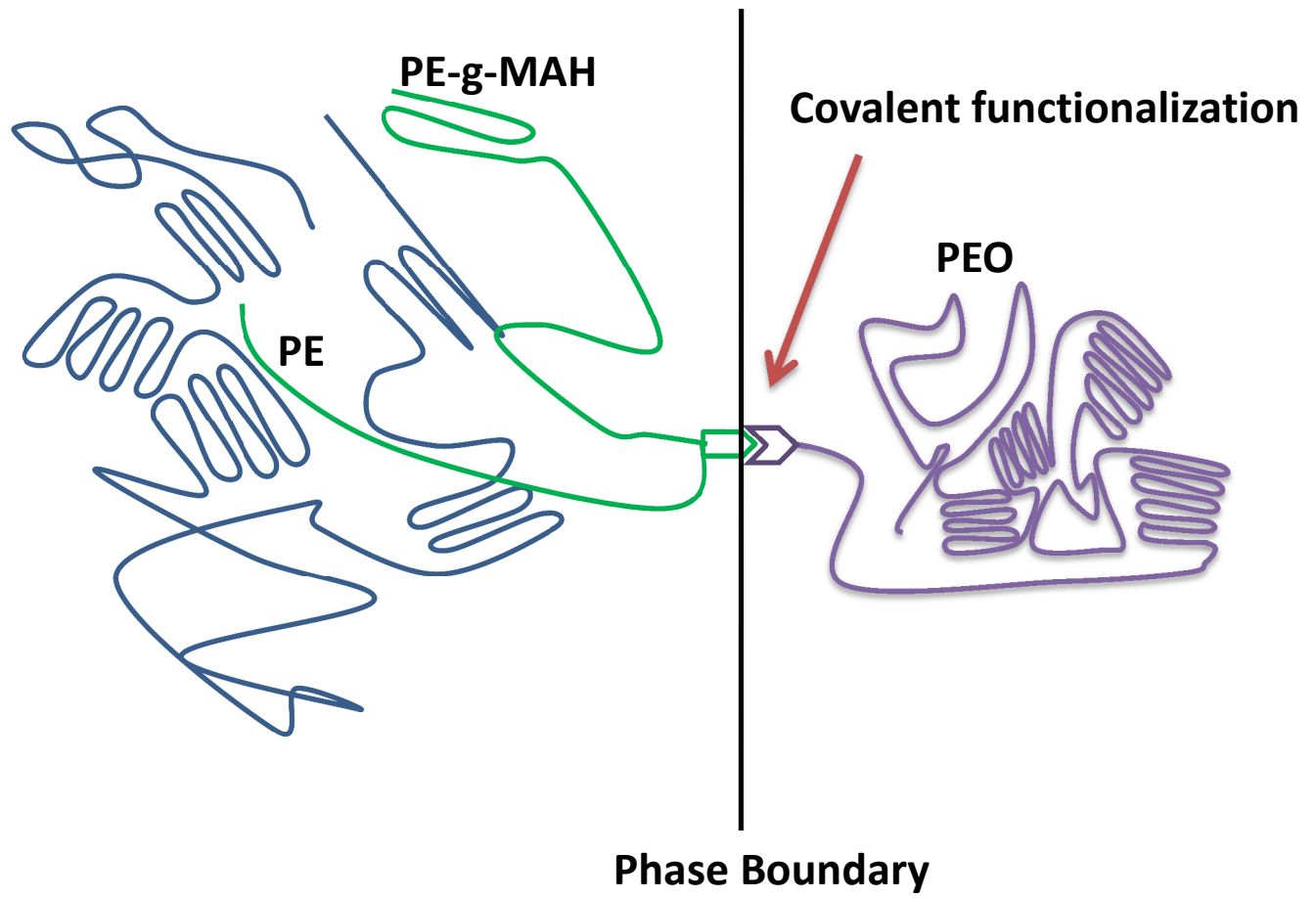
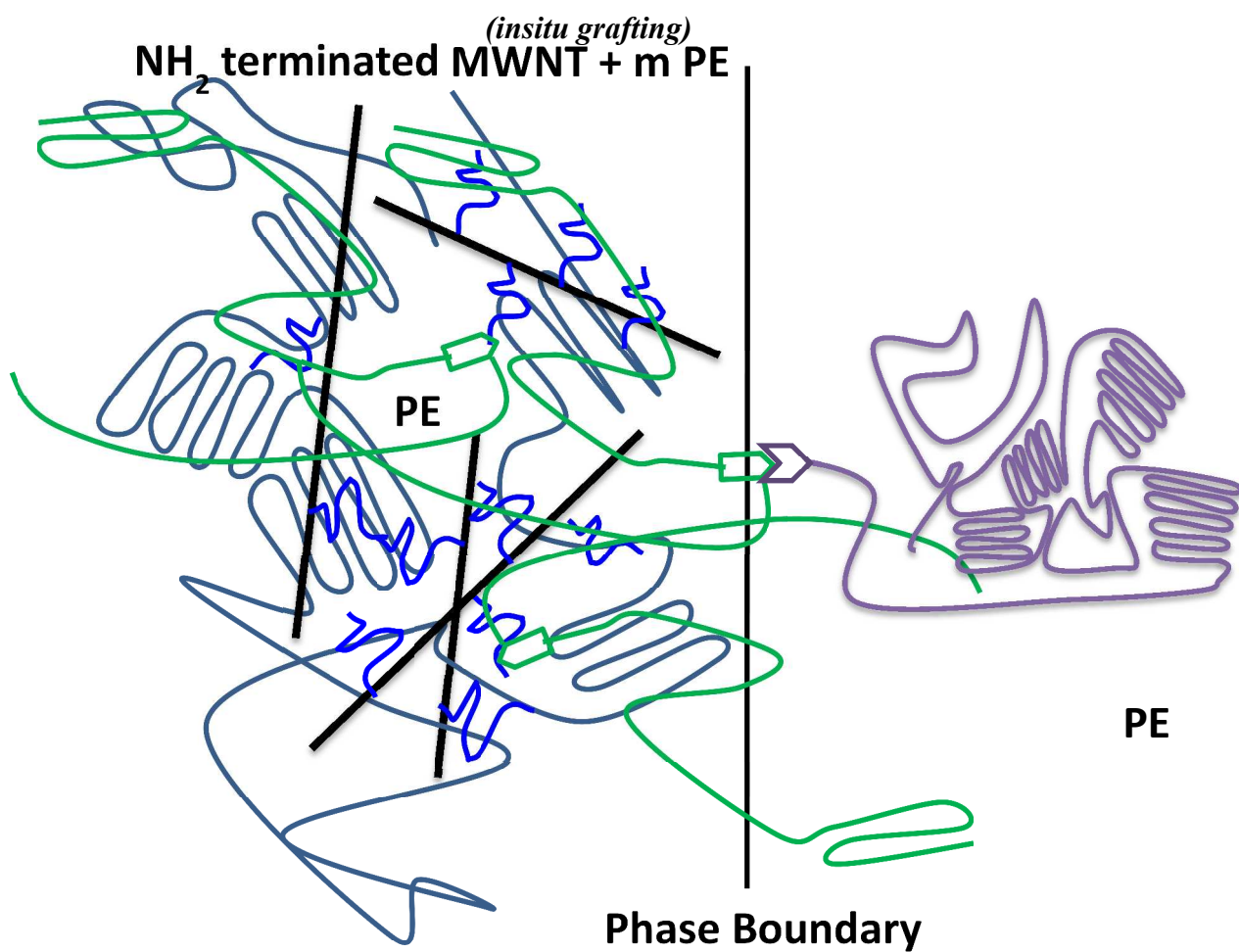
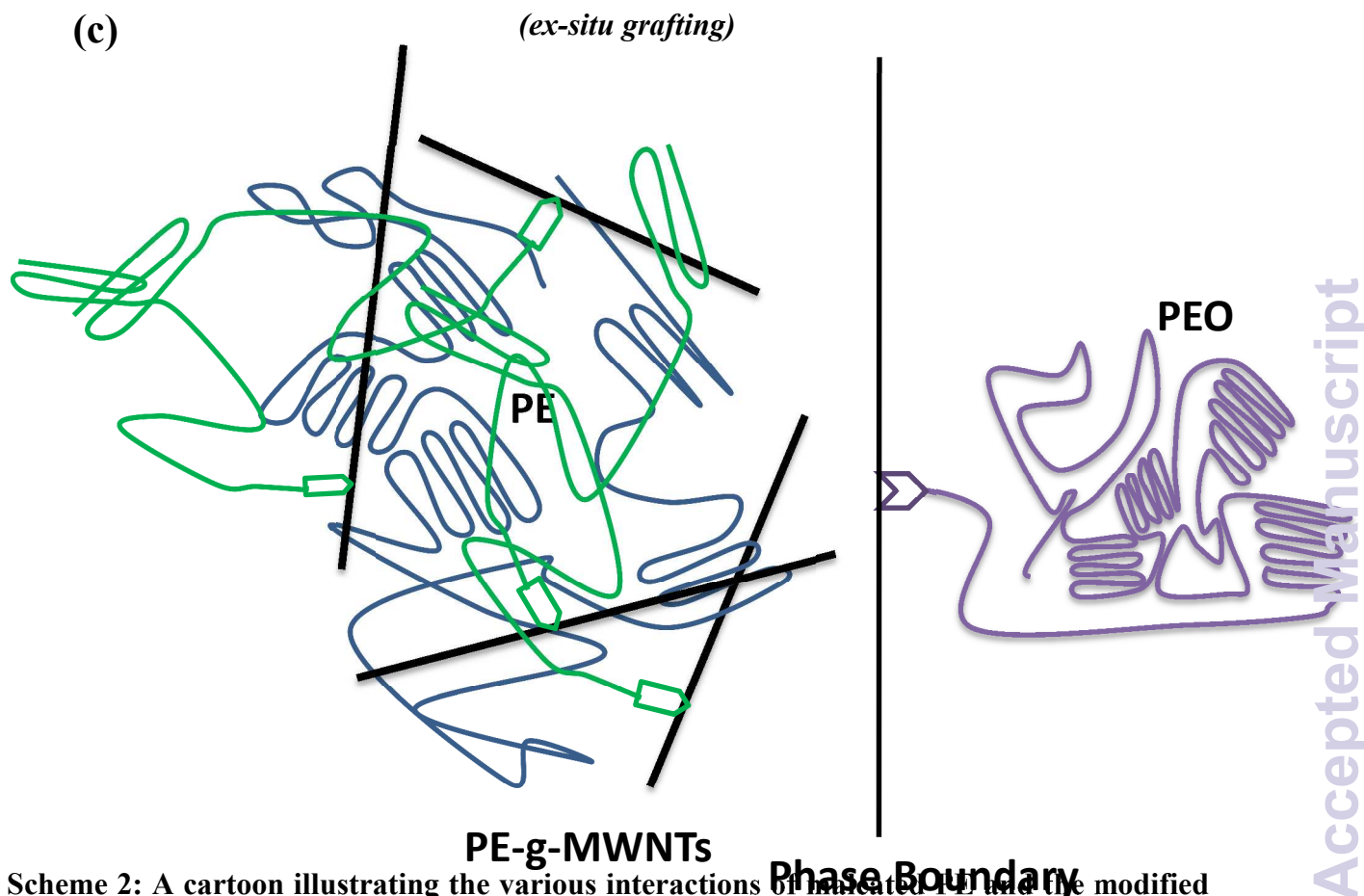


Figure 2: Cryofractured morphology of (a) 70/30 PE/PEO blend (b) with 1 wt.% p-MWNTs (c) with 1 wt.% NH₂ terminated MWNT (d) with 3 wt.% maleated PE (e) with 3 wt.% maleated PE and 1 wt.% p-MWNTs (f) 3 wt.% maleated PE and with 1 wt.% NH₂ terminated MWNT (g) with 3 wt.% maleated PE and 1 wt.% PE-g-MWNTs.



(b)





Scheme 2: A cartoon illustrating the various interactions of maleated PE and the modified MWNTs with the constituents (PE and PEO); a) maleated PE (mPE) can interact with the hydroxyl end groups of PEO; b) maleated PE can interact with NH_2 terminated MWNTs *in situ* during melt-mixing and also with PEO c) *ex situ* grafting of maleated PE with NH_2 terminated MWNTs *a priori* before melt mixing.

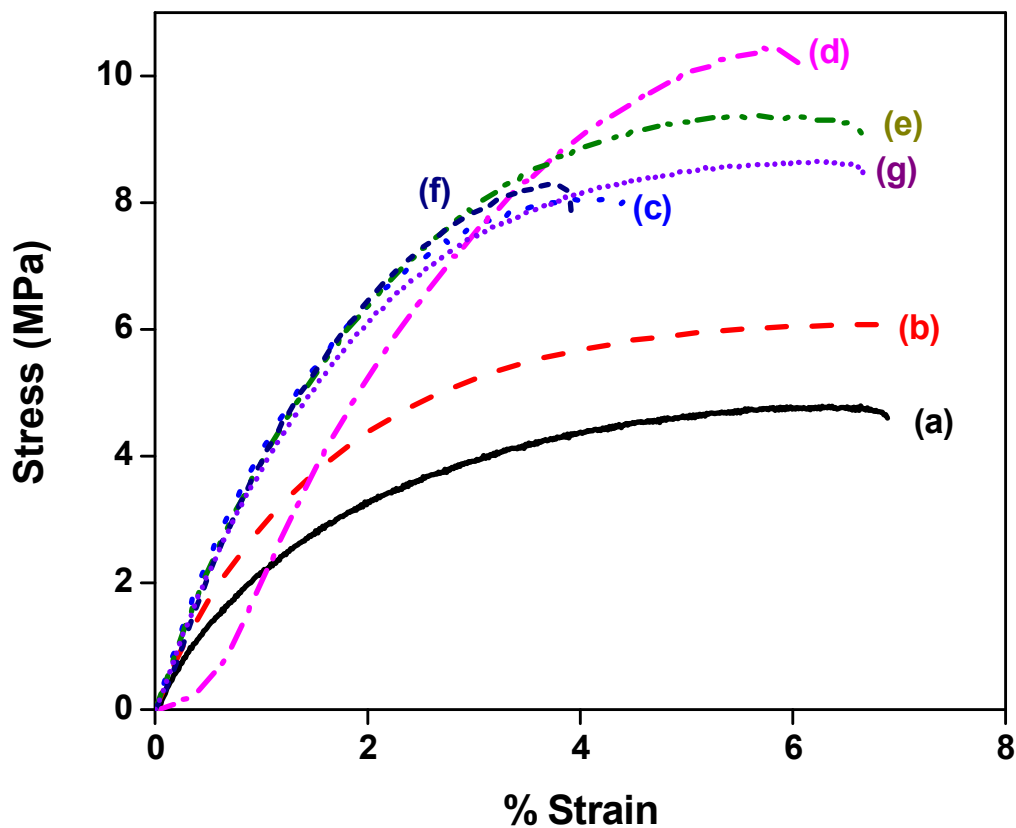


Figure 3: Stress strain of (a) 70/30 PE/PEO blend (b) 1 wt.% p-MWNT (c) 1 wt.% NH_2 terminated MWNT (d) with 3 wt.% maleated PE, (e) with 3 wt.% maleated PE and 1 wt.% p-MWNT, (f) with 3 wt.% maleated PE and 1 wt.% NH_2 terminated MWNT and (g) with 3 wt.% maleated PE and 1 wt.% PE-g-MWNTs

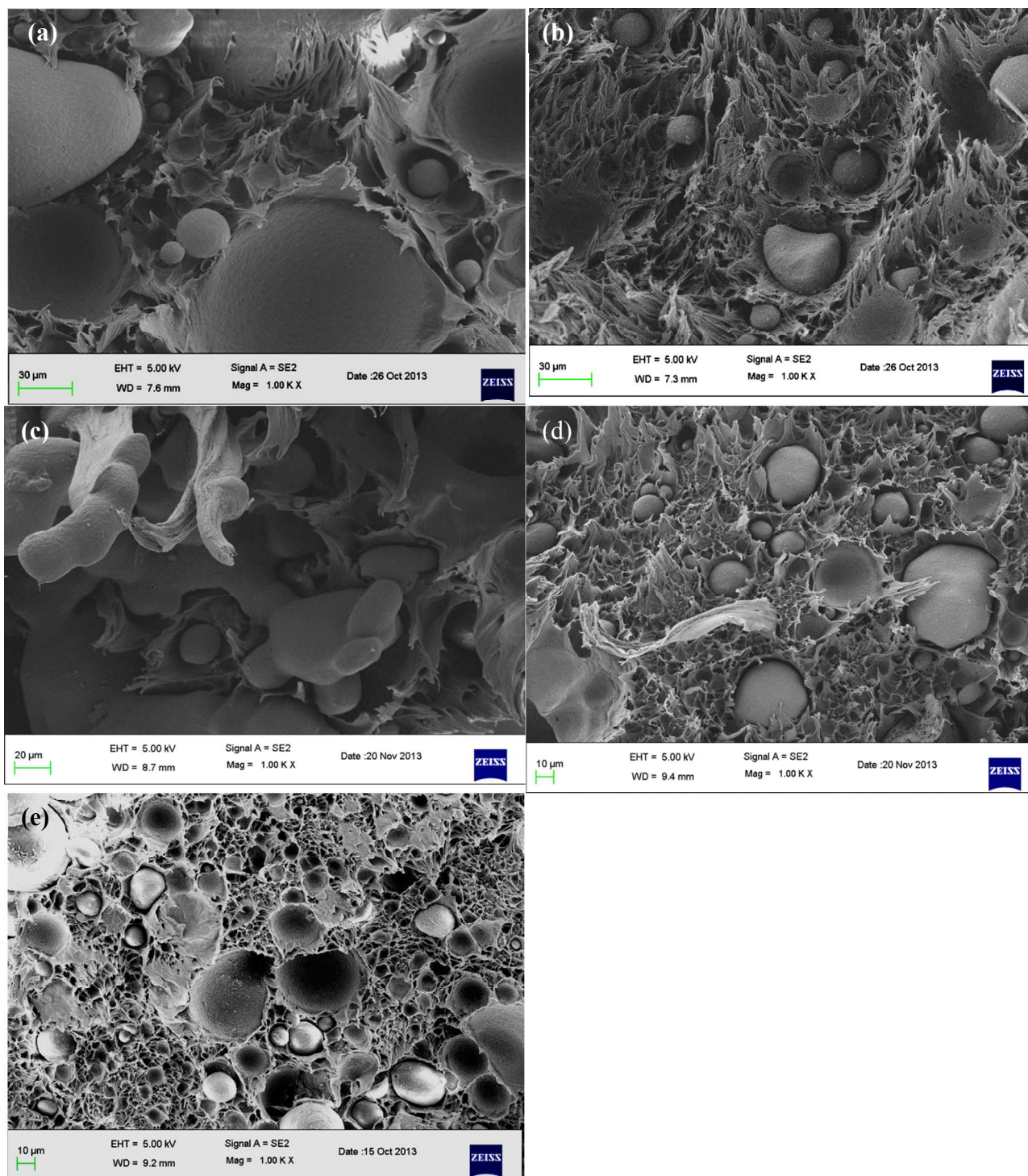


Figure 4: Tensile fracture of (a) 70/30 PE/PEO blend (b) with 3 wt.% maleated PE (c) with 1 wt.% NH_2 terminated MWNTs (d) 3 wt.% maleated PE and 1 wt.% NH_2 terminated MWNTs (e) with 3 wt.% maleated PE and 1 wt. % PE-g-MWNT.

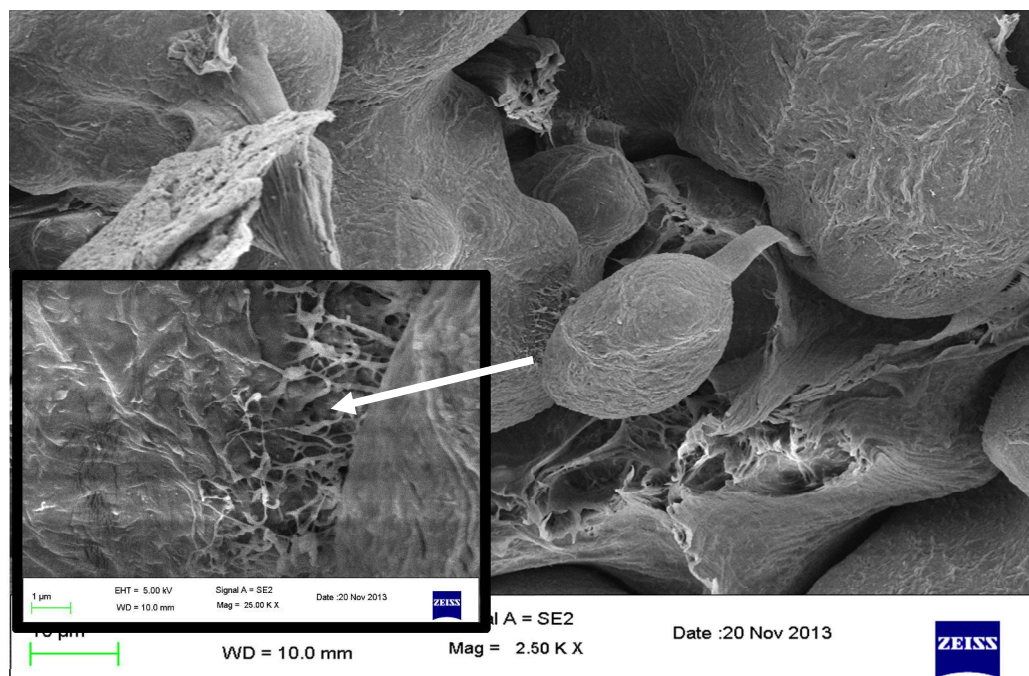


Figure 5: High resolution SEM image of tensile fractured surface of 3 wt.% maleated PE and 1 wt.% NH_2 terminated MWNTs. Inset shows MWNT bridging the phases.

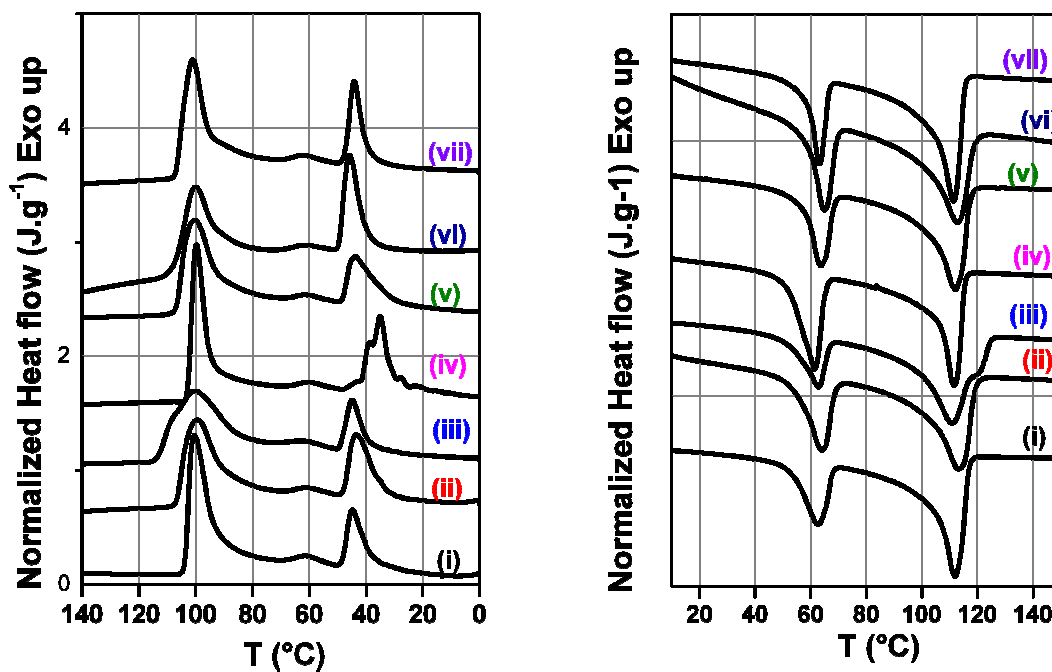


Figure 6: DSC: (a) cooling and (b) second heating scans of blends of (i) 70/30 PE/PEO blend (ii) 1 wt.% p-MWNT (iii) 1 wt.% NH₂ terminated MWNT (iv) with 3 wt.% maleated PE, (v) with 3 wt.% maleated PE and 1 wt.% p-MWNT, (vi) with 3 wt.% maleated PE and 1 wt.% NH₂ terminated MWNT and (vii) with 3 wt.% maleated PE and 1 wt.% PE-g-MWNTs

Table 1: Mechanical properties of the 70/30 PE/PEO blend with compatibilizer and MWNTs

Sample	Ultimate Tensile Strength (MPa)	Tensile Modulus (MPa)	Elongation at break (%)
Neat 70/30 Blend	5 ± 0.6	173 ± 8.0	7.0 ± 0.5
With 1 wt.% p-MWNTs	6 ± 0.1	268 ± 7.6	6.9 ± 1.1
With 1 wt.% NH ₂ - t-MWNTs	8 ± 0.1	481 ± 20	4.4 ± 0.3
With 3 wt.% PE-g-MAH	10 ± 0.1	356 ± 4.6	4.8 ± 0.8
With 3 wt.% PE-g-MAH and 1 wt.% p-MWNT	9 ± 0.6	399 ± 8.7	6.0 ± 0.8
With 3 wt.% PE-g-MAH and 1 wt.% NH ₂ - t-MWNTs	8 ± 0.2	496 ± 9.2	3.9 ± 0.5
With 3 wt.% PE-g-MAH and 1 wt.% PE-g-MWNT	8 ± 0.2	421 ± 6.5	6.5 ± 1.1

Table 2: DSC melting and crystallization parameter of the 70/30 PE/PEO blend with and without compatibilizer and MWNTs

Sample	Melting Temperature (T _m)(°C)		Crystallization Temperature (T _c)(°C)	
	PE	PEO	PE	PEO
Neat 70/30 Blend	112	63	100	45
With 1 wt.% p-MWNTs	114	64	100	44
With 1 wt.% NH ₂ - t-MWNTs	111	63	101	45
With 3 wt. % PE-g-MAH	111	61	100	35
With 3 wt.% PE-g-MAH and 1 wt.% p-MWNT	112	64	101	44
With 3 wt.% PE-g-MAH and 1 wt.% NH ₂ - t-MWNTs	113	65	100	46
With 3 wt.% PE-g-MAH and 1 wt.% PE-g-MWNT	111	63	101	44

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