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## COMMUNICATION

## Facile Synthesis of 2-Amino-3-Bromoquinolines by Palladium-Catalyzed Isocyanide Insertion and Cyclization of *gem*-Dibromovinylanilines

Cite this: DOI: 10.1039/x0xx00000x

Received 00th January 2012,  
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DOI: 10.1039/x0xx00000x

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**A novel and efficient synthesis of 2-amino-3-bromoquinolines through palladium-catalyzed isocyanide insertion followed by intramolecular cyclization of *gem*-dibromovinylanilines was developed. The reactions were carried out in 1,4-dioxane at 100°C for 2-3h and the corresponding products were obtained in good isolated yields.**

The quinoline scaffold occurs in a large number of natural products and synthetic drugs with different biological activities<sup>1</sup>. Especially 2-aminoquinoline derivatives have been frequently studied during the past decades because of their pharmacological potential covering a range of possible applications, including anti-cancer<sup>2</sup>, anti-hypertensive<sup>3</sup> and anti-Alzheimer's<sup>4</sup> activities. In the past decades, while numerous synthetic approaches have been developed for quinoline synthesis<sup>5</sup>, only a few examples are available for the synthesis of 2-aminoquinolines<sup>6</sup>. Nevertheless, the existing methods generally suffer from limited availability of substrates, complicated multistep procedures and low regioselectivity in most cases. Meanwhile, aromatic halides are a very important class of compounds not only widely used in drug and natural product synthesis, but also providing good opportunities for further formation of C-C and/or C-heteroatom bonds by metal-catalyzed coupling reactions<sup>7</sup>. Therefore, simple and efficient new protocols for the preparation of halo-2-aminoquinolines are still desirable.

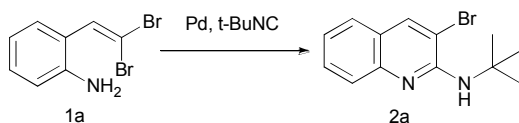
*Gem*-dihaloolefins as highly versatile reagents have attracted considerable attentions in transition metal-catalyzed cross-coupling chemistry due to their high reactivity and ready availability from inexpensive aldehydes<sup>8</sup>. Recently, a number of novel and elegant methods for the preparation of various substituted indoles from 2-(*gem*-dibromovinyl)anilines were developed via Pd and/or Cu-catalyzed tandem cross-coupling strategies, such as C-N/C-C<sup>9,10</sup>, C-N/C-N<sup>11</sup>, C-N/C-P<sup>10</sup>, C-N/C-H<sup>12</sup>, C-N/carbonylation<sup>13</sup>, and

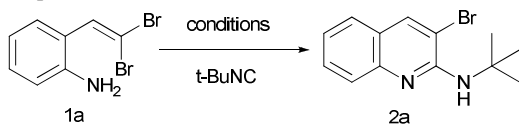
C-N/carbonylation/C-C reactions<sup>14</sup>. However, to the best of our knowledge, there is no example of constructing quinoline scaffold from *gem*-dihaloovinylanilines. In our continuous studies of 1,1-dibromoolefins<sup>15</sup>, we report herein the preparation of 2-amino-3-bromoquinoline derivatives through tandem palladium-catalyzed isocyanide insertion and intramolecular cyclization of *gem*-dibromovinylanilines. The reaction involves *t*-butyl isocyanide and 2-(*gem*-dibromovinyl)anilines using palladium catalysis (Scheme 1).

Our investigation started with the treatment of 2-(*gem*-dibromovinyl)aniline (**1a**) and *t*-BuNC in the presence of Pd(OAc)<sub>2</sub> (5 mol%) and PPh<sub>3</sub> (10 mol%) using K<sub>2</sub>CO<sub>3</sub> (2 equiv.) as the base in THF at 100°C. To our delight, the quinoline product **2a** was obtained in 70% yield after 3h (Table 1, entry 1). To improve the yield, screening of the reaction conditions was then carried out. Firstly, by using Pd(OAc)<sub>2</sub> as catalyst, the effect of different solvents including DMSO, DMF, Toluene, 1,4-dioxane and CH<sub>3</sub>CN was studied (Table 1, entry 2-6). Among them, 1,4-dioxane gave **2a** in the highest yield (Table 1, entry 5). Lautens<sup>16</sup> reported a convenient synthesis of 2-bromoindoles using PtBu<sub>3</sub>. When PtBu<sub>3</sub> was used in our studies, **2a** was obtained in 34% yield while 2-bromo indole in 29% yield. When *t*-BuNC (0.6eq) was used, **2a** was obtained in 17% yield, and 2-bromoindole in 33% yield (Table 1, entry 7, 8). After screening a series of palladium catalysts, Pd(dppf)Cl<sub>2</sub> turned out to be the best choice (Table 1, entry 6-12). We were delighted to find that the reaction went well without additional ligand PPh<sub>3</sub> (Table 1, entry 13). Other experiments with different bases showed that Cs<sub>2</sub>CO<sub>3</sub> was the most efficient one (Table 1, entry 13-16). In addition, elevating or lowering reaction temperature resulted in decreased yields of **2a** (Table 1, entry 17,18). On the basis of these findings, a wide variety of 2-amino-3-bromoquinolines were prepared in good yields from 2-(*gem*-dibromovinyl)anilines under the optimized conditions: Pd(dppf)Cl<sub>2</sub> as the catalyst, Cs<sub>2</sub>CO<sub>3</sub> as the base and 1,4-dioxane as the solvent at 100°C. The results are shown in Table 2.

Gratifyingly, *gem*-dibromovinylanilines bearing both electron-withdrawing and electron-donating moieties on the aromatic ring afforded the quinoline products in good yields (Table 2, entry 1-6). It is obvious that the electronic effects of substituted groups on the benzene rings in 2-(*gem*-dibromovinyl)anilines had little impact on the yields of the products. Halogen substitutes on the aromatic ring were well tolerated, giving polyhalogenated 2-amino-3-bromoquinolines (Table 2, entry 7-10), which provides attractive

**Scheme 1.** Strategic Approach to the Synthesis of 2-Amino-3-Bromoquinolines



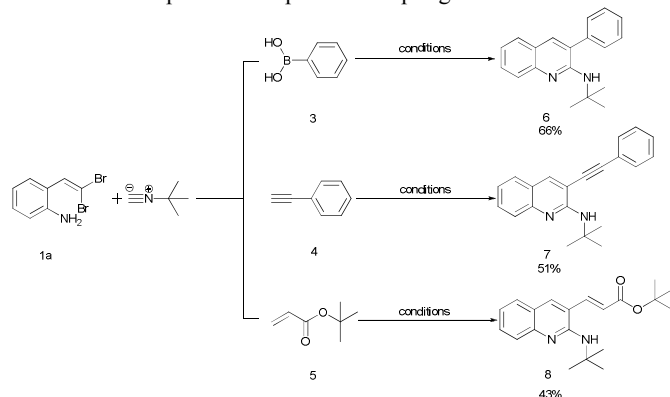
**Table 1.** Optimization of the reaction conditions<sup>a</sup>

Entry	Catalyst	Ligand	Solvent	Base	Yield <sup>b</sup> (%)
1	Pd(OAc) <sub>2</sub>	PPh <sub>3</sub>	THF	K <sub>2</sub> CO <sub>3</sub>	70
2	Pd(OAc) <sub>2</sub>	PPh <sub>3</sub>	DMSO	K <sub>2</sub> CO <sub>3</sub>	43
3	Pd(OAc) <sub>2</sub>	PPh <sub>3</sub>	DMF	K <sub>2</sub> CO <sub>3</sub>	46
4	Pd(OAc) <sub>2</sub>	PPh <sub>3</sub>	Toluene	K <sub>2</sub> CO <sub>3</sub>	51
5	Pd(OAc) <sub>2</sub>	PPh <sub>3</sub>	Dioxane	K <sub>2</sub> CO <sub>3</sub>	77
6	Pd(OAc) <sub>2</sub>	PPh <sub>3</sub>	CH <sub>3</sub> CN	K <sub>2</sub> CO <sub>3</sub>	72
7	Pd(OAc) <sub>2</sub>	PtBu <sub>3</sub>	Dioxane	K <sub>2</sub> CO <sub>3</sub>	34
8	Pd(OAc) <sub>2</sub>	PtBu <sub>3</sub>	Dioxane	K <sub>2</sub> CO <sub>3</sub>	17 <sup>c</sup>
9	PdCl <sub>2</sub>	PPh <sub>3</sub>	Dioxane	K <sub>2</sub> CO <sub>3</sub>	68
10	Pd(PPh <sub>3</sub> ) <sub>4</sub>	—	Dioxane	K <sub>2</sub> CO <sub>3</sub>	61
11	Pd(dppf)Cl <sub>2</sub>	PPh <sub>3</sub>	Dioxane	K <sub>2</sub> CO <sub>3</sub>	80
12	Pd(PPh <sub>3</sub> )Cl <sub>2</sub>	PPh <sub>3</sub>	Dioxane	K <sub>2</sub> CO <sub>3</sub>	79
13	Pd(PPh <sub>3</sub> )Cl <sub>2</sub>	—	Dioxane	K <sub>2</sub> CO <sub>3</sub>	81
14	Pd(dppf)Cl <sub>2</sub>	—	Dioxane	Cs <sub>2</sub> CO <sub>3</sub>	86
15	Pd(dppf)Cl <sub>2</sub>	—	Dioxane	Na <sub>2</sub> CO <sub>3</sub>	82
16	Pd(dppf)Cl <sub>2</sub>	—	Dioxane	K <sub>3</sub> PO <sub>4</sub>	65
17	Pd(dppf)Cl <sub>2</sub>	—	Dioxane	Cs <sub>2</sub> CO <sub>3</sub>	71 <sup>d</sup>
18	Pd(dppf)Cl <sub>2</sub>	—	Dioxane	Cs <sub>2</sub> CO <sub>3</sub>	58 <sup>e</sup>

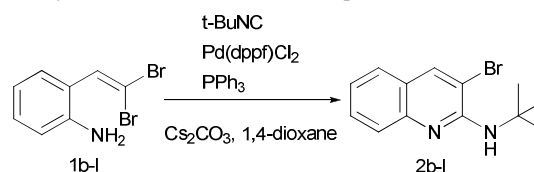
<sup>a</sup> Reaction conditions: **1a** (0.2 mmol), t-BuNC (0.3 mmol), base (0.4 mmol), catalyst (0.01 mmol), ligand (0.04 mmol), solvent (2.0 mL), sealed tube, 100°C, 3h. <sup>b</sup> isolated yields. <sup>c</sup> t-BuNC (0.12 mmol). <sup>d</sup> 120°C. <sup>e</sup> 80°C.

intermediates for their further transformation via transition-metal catalyzed C-C or C-heteroatom formation reactions.

Having successfully established the general method for the synthesis of 2-amino-3-bromoquinolines, we envisioned that 2-amino-3-bromoquinolines could easily undergo an intramolecular cross-coupling reaction to afford various 3-substituted-2-aminoquinolines, even through a one-pot two-step process. As expected, we found 3-substituted-2-aminoquinolines (**6/7/8**) could be conveniently prepared in 66%/51%/43% yields through palladium-catalyzed isocyanide insertion and typical Suzuki/Sonogashira/ Heck reaction, respectively (Scheme 2).

**Scheme 2.** One-pot Two-Step Cross-coupling Reaction of **2a**<sup>a</sup>

<sup>a</sup> Reaction conditions: step (1) **1a** (0.2 mmol), t-BuNC (0.3 mmol), Cs<sub>2</sub>CO<sub>3</sub> (0.4 mmol), Pd(dppf)Cl<sub>2</sub> (0.01 mmol), 1,4-dioxane (2.0 mL), sealed tube, 100°C, 3h; step (2) **3/4/5** (0.3 mmol), K<sub>3</sub>PO<sub>4</sub> (0.4 mmol), Pd(dppf)Cl<sub>2</sub> (0.01 mmol), Toluene (2.0 mL), sealed tube, 120°C, N<sub>2</sub>, 12h.

**Table 2.** Synthesis of 2-amino-3-bromoquinolines **2b-2l**<sup>a</sup>

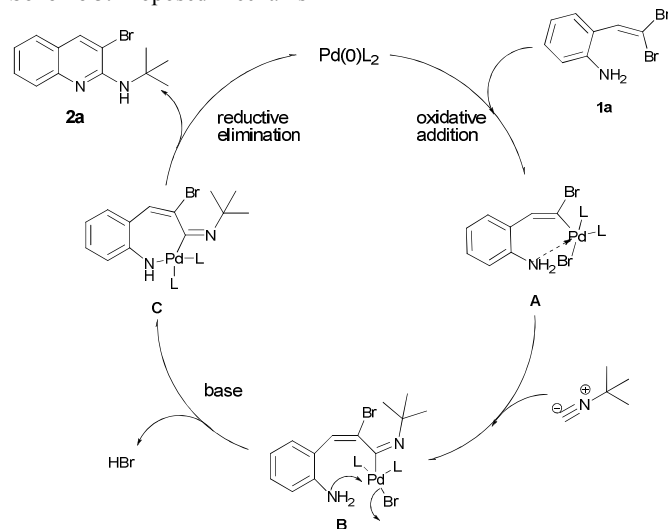
Entry	Substrate	Product	Yield <sup>b</sup> (%)
1			82
2			87
3			85
4			81
5			84
6			86
7			80
8			82
9			83
10			84
11			85

<sup>a</sup> Reaction conditions: **1** (0.2 mmol), t-BuNC (0.3 mmol), Cs<sub>2</sub>CO<sub>3</sub> (0.4 mmol), Pd(dppf)Cl<sub>2</sub> (0.01 mmol), 1,4-dioxane (2.0 mL), sealed tube, 100°C, 3h. <sup>b</sup> isolated yields

Since ortho effect exists in many metal-catalyzed reactions<sup>6a,15,17</sup>, a plausible mechanism for the formation of 2-amino-3-bromoquinoline

is proposed in Scheme 3. Initial oxidative addition of *cis*-Bromovinyl of **1a** to Pd(0) species formed vinylpalladium complex **A**. Subsequent migratory insertion of *t*-butyl isocyanide to **A** gave intermediate **B**, followed by intramolecular nucleophilic attack of the aniline to provide intermediate **C** and hydrogen bromide. Reductive elimination of **C** afforded final product **2a**, generating the Pd(0) catalyst. Further studies into the mechanism of this transformation and synthetic applications are still ongoing.

**Scheme 3.** Proposed Mechanism



## Conclusions

In summary, we have developed a novel and efficient method for the construction of 2-amino-3-bromoquinolines via palladium-catalyzed isocyanide insertion followed by intramolecular cyclization of *gem*-dibromovinylanilines. The palladium-catalyzed reactions proceeded in 1,4-dioxane at 100°C for 3h and the corresponding products were obtained in good isolated yields. More importantly, this method can be extended to the synthesis of various 3-substituted-2-aminoquinolines via palladium-catalyzed Suzuki/Sonogashira/Heck reactions starting from easily accessible *gem*-dibromovinylanilines in a one-pot two-step way.

This work was financially supported by Foundation of the Leading Talents of Guangdong Province (No. 1187000044).

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† Electronic Supplementary Information (ESI) available: [details of any supplementary information available should be included here]. See DOI: 10.1039/c000000x/

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