ORGANIC CHEMISTRY

FRONTIERS

Accepted Manuscript





This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this Accepted Manuscript with the edited and formatted Advance Article as soon as it is available.

You can find more information about *Accepted Manuscripts* in the **Information for Authors**.

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard **Terms & Conditions** and the **Ethical guidelines** still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.





http://rsc

http://rsc.li/frontiers-organic

1 2 3

4

5

6 7 8

9 10

11 12

13 14 15

16

17 18

19

20

21

22

23

24

25

26

27

28

29

30

31

32

33

34

35

36

37

38

39

40

41

42

43

44

45

46

47

48

49

50

51

52

53

54

55

56

57

58

59 60 **ARTICLE TYPE**

Cite this: DOI: 10.1039/c0xx00000x

Asymmetric Cobalt Catalysts for Hydroboration of 1,1-Disubstituted Alkenes

Jianhui Chen, Tuo Xi, Xiang Ren, Biao Cheng, Jun Guo and Zhan Lu*

Received (in XXX, XXX) Xth XXXXXXXX 20XX, Accepted Xth XXXXXXXX 20XX

5 DOI: 10.1039/b000000x

The chiral iminopyridine oxazoline (IPO) ligands were designed, synthesized and utilized for the first cobalt-catalyzed highly regio- and enantioselective *anti*-Markovnikov hydroboration of 1,1-disubstituted aryl alkenes. This novel IPO ligands will likely be of high value for asymmetric transformations with first-row transition metals.

Introduction

¹⁰ Asymmetric hydroboration of alkenes is one of the most useful methods to form chiral alkylboronic acid derivatives which are widely used in organic synthesis.¹ Hydroboration of terminal alkenes catalyzed by chiral transition metals is more favored to Markovinov regioselectivity.² Catalytic asymmetric anti-15 Markovnikov hydroboration of 1,1-disubtituted alkenes remains a challenge.³ Low enantioselectivity and in some case poor regioselectivity were obtained through Rh- and Ir-catalyzed reactions of 1,1-disubstituted alkenes with catecholborane. Recently, two catalytic systems, Iridium with chiral PN-ligand⁵ ²⁰ and copper with chiral NHC ligand,⁶ were reported, respectively, to realize asymmetric hydroboration of 1,1-disubstituted alkenes. However, noble transition metal was used, or $B_2(pin)_2$ was used to provide more waste, a few case shows high enantioselectivity $(\geq 90\%$ ee). To the best of our knowledge,⁷ there is no previous 25 report on asymmetric cobalt-catalyzed anti-Markovnikov hydroboration of 1,1-disubtituted alkenes.⁸

Noble metals play a very important role of asymmetric organic transformations in academia and industry, such as asymmetric hydrogenation of alkenes, ⁹ however, earth-abundant metal 30 catalysts often reacting via one-electron processes are limited in some type of reactions. Redox-active ligands which have been studies by the spectroscopy properties might provide the possibility for earth-abundant metals to go through two-electron redox processes to promote bond- breaking and making events.¹⁰ 35 The asymmetric applications of redox-active ligands are extremely rare. Recently, Chirik group reported highly enantioselective hydrogenation of alkenes¹¹ using C1-symmetric bis(imino)pyridine cobalt complexes¹² which show that chiral redox-active ligand is a potential good class of catalysis for 40 asymmetric organic synthesis, however, the chiral imine on the catalyst is not stable and easily to release. Based on the bi(imino)pyridine ligands, we introduced the chiral oxazoline units as stereodirecting elements ¹³ and designed the new iminopyridine oxazoline (IPO) cobalt complexes, in which, the 45 iminopyridine group is proposed to stabilize the cobalt and chiral oxazoline group to control enantioselectivity (Scheme 1).

Herein we report the synthesis of a series of chiral IPO ligands from the commercially available starting materials (**Scheme 1**). The cobalt complexes **2** could be synthesized by combining ⁵⁰ cobalt dichloride with the corresponding ligands and are bench-stable.

Scheme 1. Design and synthesis of chiral redox-acitve cobalt complexes.



a) (S)-aminoalochol, Zn(OTf)₂, toluene, reflux, 70-93% yield; b) nBuLi, Et₂O, -78 °C, 2 h, then DMA, 35-40% yield; c) 2,6-diisopropylaniline, cat. HCOOH, MeOH, relux, 24 h, 32-41% yield; d) CoCl₂, THF, rt, then Et₂O, 86-95% yield.

Results and discussion

We chose the hydroboration of styrene **6a** with HBpin as the model reaction to test the reactive of our designed chiral cobalt complexes (IPO-CoCl₂). Using only 0.5 mol% cobalt complexes ⁶⁰ and 1.5 mol% of NaHBEt₃ (1 M in THF solution) as the reductant without any additive solvent, high reactivity and regio- and enantioselectivities were observed in all cases among which complex **2c** gave the excellent yield and highest enantioselectivity (**Scheme 2**). The reaction was really slow using ⁶⁵ **1c** with iridium catalyst which might illustrate that IPO ligands worked better with first-row transition metals than late-transition

rganic Chemistry Frontiers Accepted Manuscrip

metals. Poor reactivities were shown in using the bisoxazoline ligand instead of IPO ligands which the iminopyridine group is proposed to stabilize the cobalt.

Scheme 2. Optimizations.



With the best complex 2c in hands, studies exploring the scope of this process are summarized in Chart 1. The reactions were operated under schlenk line in 2.5 mmol scale, not necessarily in glove box. 1) The reaction represented high enantioseletivities 10 with a variety of substituted α -methyl styrenes, including both electron-rich and electron-deficient styrene compounds; 2) Halides and protected heteroatoms can be tolerated at para, meta and ortho-position on the aryl rings; 3) Although a silghtly low ee were observed in the reaction of ortho-substituted styrenes, 15 high yields were obtained; 4) Long alkyl chain on α-position of styrenes, even with functionalized alkyl chain, were

Chart 1. Asymmetric anti-Markovnikov hydroboration of alkenes.^a

tolerated to prepare the corresponding hydroboration products in high enantioselectivities; 5) The cyclic styrene with terminal alkene was also reacted to afford 7ab in a slightly 20 low yield with 95% ee; 6) Gratifyingly, 1,1-dialkyl substituted alkenes 6ac and 6ad also participated in this reaction to give the desired hydroboration products in 72% yield with 33% ee and 70% ee, respectively.

The compounds 7c and 7w can be easily oxidized and ²⁵ further derivatized¹⁴ to (R)-naproxen and (R)-ibuprofen, which enantiomers are well-known non-steroid anti-inflammatory and analgesic drugs ¹⁵(Scheme 3).

Scheme 3. Further derivatizations.



30 Conclusions

In conclusions, we have developed a novel iminopyridine oxazoline cobalt-catalyzed highly enantioselective and regioselective anti-Markovnikov hydroboration of 1,1disubstituted alkenes with hydroborate. A series of useful 35 highly enantiopure borate compounds were easily synthesized from the simple alkenes without any directing group. Current



7aa, 90%, 94% ee^e ^a Standard condition: Unless otherwise noted, 6 (2.5 mmol), HBPin (2.5 mmol), 2c (0.5 mol%), NaBHEt₃ (1.5 mol%) at room temperature for 1 h; ^b 2 mol% 2c; ^c toluene (0.5 mL).^d 1 mol% 2c; ^c 5 mol% 2c

7ab, 50%, 95% ee

1

7v. 72%, 98% ee

7z 62% 96% ee

7ac 72% 33% ee

7ad 72% 70% ee

1

2

3

59 60 efforts in our lab are underway to explore the applications of IPO ligands in asymmetric reactions.

Acknowledgements

Financial support was provided by the "Thousand Youth ⁵ Talents Plan", the Fundamental Research Funds for the Central Universities (2013QNA3022), the Starting Funds from Zhejiang University

Notes and references

^a Department of Chemistry, Zhejiang University, 148 Tianmushan Road 10 Hangzhou, Zhejiang, 310028, China, E-mail: luzhan@zju.edu.cn

- † Electronic Supplementary Information (ESI) available: [details of any supplementary information available should be included here]. See DOI: 10.1039/b000000x/
- ¹ For some selected reviews, see: (a) H. C. Brown, B. Singaram, Acc. Chem. Res. **1988**, 21, 287. (b) M. Zaidlewicz, in Comprehensive Organometallic Chemistry; G. Wilkinson, F. G. A. Stone, E. W. Abel, Eds.; Pergamon: Oxford, **1982**, 7, 229, and references cited therein. For some selected examples, see: (c) H. C. Brown, D. B. Bigley, J. Am. Chem. Soc. **1961**, 83, 486. (d) A. Z. Gonzalez, J. G. Román, E. Gonzalez, J. Martinez, J. R. Medina, K. Matos, J. A. Soderquist, J. Am. Chem. Soc. **2008**, 130, 9218.
- ² (a) T. Hayashi, Y. Matsumoto, Y. Ito, J. Am. Chem. Soc. 1989, 111, 3426. (b) K. Burgess, M. J. Ohlmeyer, Chem. Rev. 1991, 91, 1179. (c) I. Beletskaya, A. Pelter, Tetrahedron 1997, 53, 4957. (d) H. Doucet, E. Fernandez, T. P. Layzell, J. M. Brown, Chem. Eur. J. 1999, 5, 1320. (e) C. M. Crudden, D. Edwards, Eur. J. Org. Chem. 2003, 4695. (f) C. M. Crudden, Y. B. Hleba, A. C. Chen, J. Am. Chem. Soc. 2004, 126, 9200. (g) S. A. Moteki, D. Wu, K. L. Chandra, D. S. Reddy, J. M. Takacs, Org. Lett. 2006, 8, 3097. (h) S. A. Moteki, J. M. Takacs, Angew. Chem. Int. Ed. 2008, 47, 894. (i) C. M. Crudden, B. M. Glasspoole, C. J. Lata, Chem. Commun. 2009, 6704. (j) S. A. Moteki, K. Toyama, Z. Liu, J. Ma, A. E. Holmes, J. M. Takacs, Chem. Commun. 2012, 263.
- 3 (a) S. P. Thomas, V. K. Aggarwal, Angew. Chem. Int. Ed. 2009, 48, 1896. For cobalt-catalyzed racemic hydroboration of alkenes, see: (b) M. Zaidlewicz, J. Meller, Tetrahedron Lett. 1997, 38, 7279. (c) J. V. Obligacion, P. J. Chirik, J. Am. Chem. Soc. 2013, 135, 19107. (d) L. Zhang, Z. Zuo, X. Leng, Z. Huang, Angew. Chem. Int. Ed. 2014, 53, 2696. For iron-catalyzed racemic hydroboration of alkenes, see: (e) J. Y. Wu, B. Moreau, T. J. Ritter, J. Am. Chem. Soc. 2009, 131, 12915. (f) A. M. Tondreau, C. C. H. Atienza, K. J. Weller, S. A. Nye, K. M. Lewis, J. G. P. Delis, P. J. Chirik, Science 2010, 327, 794. (g), J. V. Obligacion, P. J. Chirik, Org. Lett. 2013, 15, 2680; (h) L. Zhang, D. Peng, X. Leng, Z. Huang, Angew. Chem. Int. Ed. 2013, 52, 3676. (i) M. D. Greenhalgh, S. P. Thomas, Chem. Commun. 2013, 11230. (j) S. C. Bart, E. Lobkovsky, P. J. Chirik, J. Am. Chem. Soc. 2004, 126, 13794.
 - ⁴ T. Hayashi, Y. Matsumoto, *Tetrahedron. Asymmetry* **1991**, *2*, 601.
 - ⁵ C. Mazet, D. Gerard, *Chem. Commun.* 2011, 298.
 - ⁶ R. Corberan, N. W. Mszar, A. H. Hoveyda, *Angew. Chem. Int. Ed.* 2011, 50, 7079.
 - 7For the review on enantioselective cobalt-catalyzed tranformations: H. Pellissier, H. Clavier, *Chem. Rev.* **2014**, *114*, 2775.
 - ⁸ Unfortunately, during we were submitting the manuscript on our independent work, a similar result was reported by Huang, Z. et al L. Zhang, Z.-Q. Zuo, X.-L. Wan, Z. Huang, J. Am. Chem. Soc. 2014, 136, 15501. However, we used the different strategy for synthesis of chiral iminopyridine oxazoline since November 14th, 2013.
 - ⁹ For some reviews on asymmetric hydrogenation, see: (a) R. Noyori, Angew. Chem. Int. Ed. 2002, 41, 2008. (b) W. S. Knowles, Angew. Chem. Int. Ed. 2002, 41, 1999. (c) A. J. Minnaard, B. L. Feringa, L. Lefort, J. G. de Vries, Acc. Chem. Res. 2007, 40, 1267. (a) N. B. Johnson, I. C. Lennon, P. H. Moran, J. A. Ramsden, Acc. Chem. Res.

2007, 40, 1291. (b) C. S. Schultz, S. W. Krska, Acc. Chem. Res. **2007**, 40, 1320. (c) S. J. Roseblade, A. Pfaltz, Acc. Chem. Res. **2007**, 40, 1402.

- ¹⁰ P. J. Chirik, K. Wieghardt, Science 2010, 327, 794.
- ¹¹ (a) S. Monfette, Z. R. Turner, S. P. Semproni, P. J. Chirik, J. Am. Chem. Soc. 2012, 134, 4561. (b) C. Bianchini, G. Mantovani, A. Meli, F. Migliacci, F. Zanobini, F. Laschi, A. Sommazzi, Eur. J. Inorg. Chem. 2003, 1620.
- ¹² For the review on bis(imino)pyridines, see: (a) V. C. Gibson, C. Redshaw, G. A. Solan, *Chem. Rev.* 2007, 107, 1745. For selected bis(imino)pyridine cobalt-catalyzed examples, see: (b) Q. Knijnenburg, A. D. Horton, H. V. D. Heijden, T. M. Kooistra, D. G. H. Hetterscheid, J. M. M. Smits, B. de Bruin, P. H. M. Budzelaar, A. W. Gal, *J. Mol. Catal. A* 2005, 232, 151. (c) Q. Knijnenburg, D. Hetterscheid, T. M. Kooistra, P. H. M. Budzelaar, *Eur. J. Inorg. Chem.* 2004, 1204.
- 13 For some selected reviews on chiral oxazoline ligands, see: (a) P. Braunstein, F. Naud, Angew. Chem. Int. Ed. 2001, 40, 680. (b) H. A. McManus, P. J. Guiry, Chem. Rev. 2004, 104, 4151. (c) G. C. Hargaden, P. J. Guiry, Chem. Rev. 2009, 109, 2505. (d) G. Desimoni, G. Faita, , K. A. Jørgensen, Chem. Rev. 2011, 111, PR284.
- ¹⁴ P. Galletti, M. Pori, D. Giacomini, Synlett 2010, 2644.
- ¹⁵ (a) I. T. Harrison, B. Lewis, P. Nelson, W. Rooks, A. Roszkowski, A. Tomolonis, J. H. Fried, *J. Med. Chem.* **1970**, *13*, 203. (b) T. Y. Shen, *Angew. Chem. Int. Ed. Engl.* **1972**, *11*, 460. (c) D. Lednicer, L. A. Mitscher, *The Organic Chemistry of Drug Synthesis*, Wiley, New York, **1977**, vol. 1.