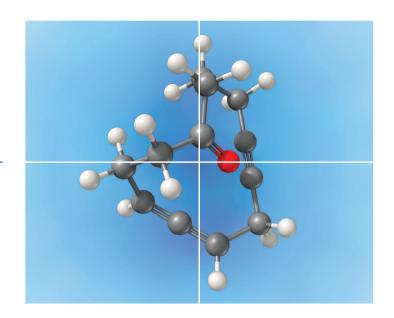
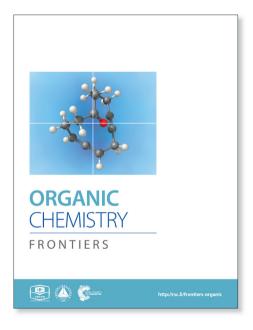
ORGANIC CHEMISTRY

FRONTIERS

Accepted Manuscript





This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this Accepted Manuscript with the edited and formatted Advance Article as soon as it is available.

You can find more information about *Accepted Manuscripts* in the **Information for Authors**.

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard **Terms & Conditions** and the **Ethical guidelines** still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.





http://rsc.li/frontiers-organic

6 7 8

9 10

11

12

13 14 15

16

17

18 19

20 21 22

23

24

25

26

27

28

29

30

31

32

33

34

35

36

37

38

39

40

41

42

43

44

45

46

47

48

49

50

51

52

53

54

55

56

57

58 59 60 Cite this: DOI: 10.1039/x0xx00000x

A Room-Temperature Synthesis of 2, 2'-Bisoxazoles through Palladium-Catalyzed Oxidative Coupling of α-Isocyanoacetamides

Jian Wang, Shuang Luo, Jing Li and Qiang Zhu*

DOI: 10.1039/x0xx00000x

Received ooth January 2012,

Accepted ooth January 2012

www.rsc.org/

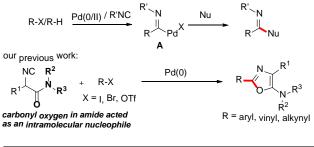
A palladium-catalyzed synthesis of symmetric and unsymmetric 2, 2'-bisoxazoles starting from readily available α -isocyanoacetamides was developed. The reaction was performed at room temperature in air which acted as the sole oxidant of Pd⁽⁰⁾. Mechanistic studies suggested that double isocyanide insertion into Pd^(II)-O bond was involved.

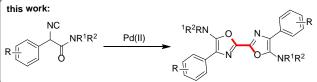
Acting as isoelectronic equivalent of carbon monoxide, isocvanide has shown its great potential in palladium-catalyzed isocyanide insertion reactions.¹ Imidoyl palladium(II) complex A was considered as a general intermediate in reaction with various nucleophiles followed by reductive elimination, generating amidines,² amides,³ ketimines,⁴ imidates, thioimidates⁵ and aldehydes⁶ correspondingly (Scheme 1). Functionalized heterocycles could be generated by linking a nucleophile to substrates R-X/R-H ready for imidoyl palladium(II) intermediate formation upon oxidative addition or C-H bond activation and isocyanide insertion.⁷ Another strategy for heterocycle construction involving isocyanide insertion as a key step employs bisnucleophiles and isocyanides under oxidative conditions.⁸ For instance, Orru and co-workers reported an efficient synthesis of cyclic guanidine derivatives and related heterocycles via palladium-catalyzed isocyanide insertion with diamines or amino alcohols.^{8a} Recently, our group developed a different strategy aiming at construction of heterocycles by linking a nucleophile to isocyanide substrate. α-Isocyanoacetamides, in which the carbonyl oxygen in the amide moiety acted as an intramolecular nucleophile, reacted with aryl, vinyl, or alkynyl halides under palladium catalysis to provide C2-diversified oxazoles.⁹ During the optimization of reaction conditions, a symmetric 2, 2'-bisoxazole byproduct was identified, albeit in very low yield (Scheme 1). In this novel process, two oxazole rings are formed in one pot through multiple bond formation including two C-O bonds and one C-C bond starting from acyclic substrates. This unprecedented and unexpected transformation intrigued us to investigate it in details.

C(sp²)-C(sp²) direct linked bisheterocycles are of vital importance in pharmaceuticals, natural products, and functional materials.¹⁰ Traditional approaches to these compounds are mostly based on heteroaryl (pseudo)halides and oganometallic reagents.¹¹ In recent years, more step-efficient and atom-economic strategies employing oxidative coupling of existing heterocyclic skeletons

through C-H bond activation were developed.¹² For instance, method directed towards 2, 2'-bisoxazoles were successfully developed by transition metal catalyzed coupling of dual C-H bonds.¹³ However, limitations of these methods, including high reaction temperatures, using stoichiometric or excess amount of Cu/Ag-based oxidant, still exist. Herein, we report a novel palladium-catalyzed synthesis of symmetric and unsymmetric 2, 2'-bisoxazoles by oxidative homoand cross-coupling of readily available α -isocyanoacetamides.¹⁴ This reaction occurs smoothly at room temperature and uses air as the sole oxidant.

palladium-catalyzed isocyanide insertion reactions:





Scheme 1. Palladium-catalyzed isocyanide insertion reactions

The reaction conditions were screened with 2-isocyano-2phenyl-1-(piperidin-1-yl)ethanone **1a** as a test substrate catalyzed by Pd(OAc)₂ (10 mol %) in air at room temperature (Table 1). Among various solvents tested, the reaction performed best in MeCN in the presence of Cs₂CO₃ (1.1 equiv) and PPh₃ (20 mol %), delivering the desired symmetric 2, 2'-bisoxazole **2a** in 70% yield (entries 1-4). Further investigations including changing reaction atmosphere from air to pure O₂ or replacing the base from Cs₂CO₃ to LiOtBu gave lower yields of **2a** (entries 5 and 7). In the absence of PPh₃, the transformation was much less efficient (51% yield, entry 6). When a solution of **1a** in MeCN (1.0 mL) was added slowly via a syringe

2

3

4

5 6

7

8

9

10

11

12

13

14

15

16

17

18

19

20

21

22

23

24

25

26

27

28

29

30

31

32

33

34

35

36

37

38

39

40 41

42

43

44

45

46

47

48

49

50

51

52

53

54

55

56

57

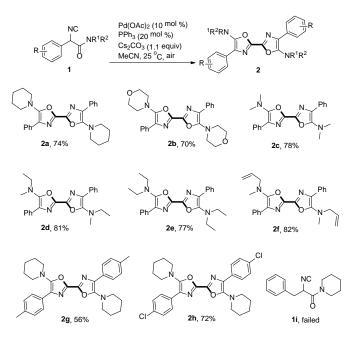
58 59 60 pump during 0.5 h to a mixture containing $Pd(OAc)_2$, PPh_3 , Cs_2CO_3 and 1 mL of MeCN, the yield of **2a** was increased slightly to 74% (entry 8).

Table 2. Optimization of reaction conditions^a

NC N		Pd(OAc) ₂ (10 mol %) PPh ₃ (20 mol %) base (1.1 equiv) solvent, air or O ₂ , 25 °C			
~	1a			2a	\checkmark
entry	solvent	base	ligand	atmosphere	yield ^b
1	DMF	Cs_2CO_3	PPh_3	air	26%
2	DCM	Cs_2CO_3	PPh ₃	air	66%
3	dioxane	Cs_2CO_3	PPh ₃	air	44%
4	MeCN	Cs ₂ CO ₃	PPh_3	air	70%
5	MeCN	Cs_2CO_3	PPh_3	O ₂	44%
6 ^{<i>c</i>}	MeCN	Cs_2CO_3	-	air	51%
7	MeCN	LiOtBu	PPh ₃	air	50%
8 ^d	MeCN	Cs ₂ CO ₃	PPh ₃	air	74%

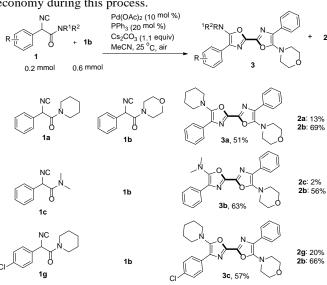
^{*a*} Reaction conditions: **1a** (0.2 mmol), Pd(OAc)₂ (10 mol %), base (0.22 mmol, 1.1 equiv), PPh₃ (20 mol %), solvent (2 mL), in air, 25 °C, 0.5 h. ^{*b*}Isolated yield. ^{*c*}2.0 h. ^{*d*}A solution of **1a** in MeCN (1 mL) was added to the reaction mixture via a syringe pump within 0.5 h.

With the optimized reaction conditions in hand, the scope of α isocyanoacetamides was then screened (Scheme 2). Besides piperidinyl amide, cyclic morpholino analogue of **1a** also generated the corresponding product **2b** smoothly in 70% yield. Other α isocyanoacetamides derived from acyclic secondary amines including *N*,*N*-dimethylamine (**1c**), *N*-methyl-*N*-ethylamine (**1d**), *N*,*N*-diethylamine (**1e**) and *N*-methylallylamine (**1f**) all homocoupled efficiently to produce the corresponding symmetric 2, 2'bisoxazoles (**2c-2f**) in good yields. It is noteworthy that the terminal alkene in **2f** survived the reaction well. Methyl and chloro substituted 2, 2'-bisoxazoles **2g** and **2h** were obtained in 56% and 72% yields respectively. Unfortunately, isocyanoacetamide bearing a benzyl group rather than an aryl one at the α -position was not a suitable substrate in this transformation (**1i**).



Scheme 2. Scope of symmetric 2, 2'-bisoxazoles. Reaction conditions: A solution of **1a** (0.20 mmol) in MeCN (1 mL) was added to the reaction mixture containing $Pd(OAc)_2$ (0.02 mmol, 10 mol %), PPh₃ (0.04 mmol, 20 mol %), Cs_2CO_3 (0.22 mmol, 1.1 equiv) and MeCN (1 mL) via a syringe pump within 0.5 h at 25 °C in air.

When two different α -isocyanoacetamides were present, an unsymmetric 2, 2'-bisoxazole product derived from crosscoupling together with two homo-coupling products was obtained (Scheme 3). For example, addition a solution of 1a (0.2 mmol, 1 equiv) and 1b (3 equiv) in 4 mL of MeCN to an open reaction tube containing catalyst, ligand, base and CH₃CN (1 mL) via a syringe pump in 1 h generated an unsymmetric 2, 2'-bisoxazole product **3a** in synthetically useful yield (51%) after careful chromatography isolation. Symmetric 2, 2'bisoxazoles 2a and 2b generated from homo-coupling were also obtained in 13% and 69% yields, respectively. The selectivity for cross-coupling was better in a reaction of 1c and 1b, generating unsymmetric product **3b** in 63% yield. Unsymmetric 2'-bisoxazole 3c containing an aromatic chloride 2. functionality was also isolated in 57% yield. The current strategy provides an efficient approach to both symmetric and unsymmetric 2, 2'-bisoxazoles in one step starting from simple acyclic α -isocyanoacetamides. It is notable that two heterocyclic rings are constructed simultaneously at ambient temperature in open air. Three chemical bonds including two C-O bonds and one C-C bond are formed with 100% atomeconomy during this process.



Scheme 3. Scope of unsymmetric 2, 2'-bisoxazoles. Reaction conditions: A solution of 1 (0.20 mmol) and 1b (0.6 mmol) in MeCN (4 mL) was added to the reaction mixture containing $Pd(OAc)_2$ (0.02 mmol, 10 mol %), PPh₃ (0.04 mmol, 20 mol %), Cs₂CO₃ (0.22 mmol, 1.1 equiv) and MeCN (1 mL) via a syringe pump within 1 h at 25 °C in air. Isolated yields of 2b are based on 1b. Other yields are based on another reactant 1.

This reaction was scalable, as exemplified by sub-gram preparation of 2a with equal efficiency (a, Scheme 4). Further diversification of the obtained oxazole product 2h was also performed. Transforming the chloride moiety to boronic acid ester through palladium catalysis was realized in 89% yield. The product 4 containing two aromatic boronic acid ester moieties is expected to be a useful precursor for more complicated symmetric 2, 2'-bisoxazole synthesis (b).¹⁵ Suzuki coupling of 2h with phenyl boronic acid also performed smoothly, giving highly conjugated product 5 in high yield (c).¹⁶

2

Organic Chemistry Frontiers

(b)

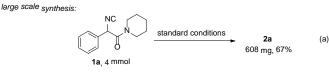
(c)

7

4,89%

5 93%





Pd2(dba)3 (10 mol %)

XPhos (40 mol %)

dioxane, 110 °C, Ar

Pd(OAc)2 (20 mol %)

JohnPhos ₍₄₀ mol %)

KF (6 equiv)

THF, 65 °C, Ar

PhB(OH)2 (6 equiv)

KOAc (6 equiv)

(Bpin)₂ (8 equiv)

further transformations:

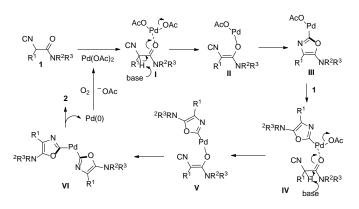
2h, 0.1 mmol

Scheme 4. Large scale synthesis and further transformations.

To verify the reaction pathway, C2 unsubstituted oxazole **6** was treated under the standard aerobic conditions. Most of the starting material **6** was recovered with no homo-coupling product **2a** being detected, which suggested that **6** was unlikely a reaction intermediate. Although the role of triphenyl phosphine was not fully understood, it may facilitate the process of reductive elimination and stabilize the Pd(0) species before being oxidized to Pd^(II) by O₂ in air.



A plausible reaction mechanism was proposed in Scheme 5. Coordination of the carbonyl oxygen in α -isocyanoacetamide **1** with Pd(OAc)₂ affords intermediate **I**. Deprotonation and the subsequent isocyanide insertion to the Pd-O bond forms the first oxazole ring in intermediate **III**. Repeating the same process furnishes the key bisoxazole ligated palladium(II) intermediate **VI**. Reductive elimination releases the homo-coupling product **2** and the Pd⁽⁰⁾ species which is reoxidized to Pd^(II) by O₂ in air. It is also possible that isocyanide insertion to Pd-O bond in Pd(OAc)₂ takes place before its coordination with the carbonyl oxygen.



Scheme 5. Proposed mechanism.

In summary, we have developed a novel palladium-catalyzed synthesis of symmetric and unsymmetric 2, 2'-bisoxazoles starting from readily available acyclic α -isocyanoacetamides. Double isocyanide insertion was believed as a key step in this transformation. The reaction was performed at room temperature in air which acted as the sole oxidant of Pd(0). The resulting symmetric or unsymmetric products were highly π -conjugated, showing their great potential in functional material synthesis.

This work was supported by National Science Foundation of China (21202167).

Notes and references

State Key Laboratory of Respiratory Disease, Guangzhou Institutes of Biomedicine and Health, Chinese Academy of Sciences, 190 Kaiyuan Avenue, Guangzhou 510530 (China), Fax: (+86) 20-3201-5299.

E-mail: zhu_qiang@gibh.ac.cn

Electronic Supplementary Information (ESI) available: [details of any supplementary information available should be included here]. See DOI: 10.1039/c000000x/

- For reviews,see: (a) T. Vlaar, B. W. Maes, E. Ruijter, R. V. A. Orru, Angew. Chem. Int. Ed. 2013, 52, 7084; (b) S. Lang, Chem. Soc. Rev. 2013, 42, 4867; (c) G. Qiu, Q. Ding, J. Wu, Chem. Soc. Rev. 2013, 42, 5257.
- C. G. Saluste, R. J. Whitby, M. Furber, *Angew. Chem. Int. Ed.* 2000, 39, 4156.
- 3 (a) H. Jiang, B. Liu, Y. Li, A. Wang, H. Huang, Org. Lett. 2011, 13, 1028; (b) J. Peng, L. Liu, Z. Hu, J. Huang, Q. Zhu, Chem. Commun. 2012, 48, 3772.
- 4 M. Tobisu, S. Imoto, S. Ito, N. Chatani, J. Org. Chem. 2010, 75, 4835.
- 5 C. G. Saluste, R. J. Whitby, M. Furber, *Tetrahedron Lett.* 2011, 42, 6191.
- 6 X. Jiang, J. M. Wang, Y. Zhang, Z. Chen, Y. M. Zhu, S.-J. Ji, Org. Lett., 2014, 16, 3492.
- (a) Y. Wang, H. Wang, J. Peng, Q. Zhu, Org. Lett. 2011, 13, 4604; (b) Y. Wang, Q. Zhu, Adv, Synth, Catal. 2012, 354, 1902; (c) G. van Baelen, S. Kuijer, L. Rýćek, S. Sergeyev, E. Janssen, F. J. J. de Kanter, B. U. W. Maes, E. Ruijter, R. V. A. Orru, Chem. Eur. J. 2011, 17, 15039; (d) T. Vlaar, E. Ruijter, A. Znabet, E. Jansson, F. J. J. de Kanter, B. U. W. Maes, R. V. A. Orru, Org. Lett. 2011, 13, 6496; (e) G. Qiu, G. Liu, S. Pu, J. Wu, Chem. Commun. 2012, 44, 2903; (f) G. Qiu, Y. Lu, J. Wu, Org. Biomol. Chem. 2013, 11, 798; (g) G. Qiu, Y. He, J. Wu, Chem. Commun. 2012, 48, 3836; (h) Y. Li, J. Zhao, H. Chen, B. Liu, H. Jiang, Chem. Commun. 2012, 48, 3545; (i) B. Liu, Y. Li, H. Jiang, M. Yin, H. Huang, Adv. Synth. Catal. 2012, 354, 2288; (j) B.Liu, Y. Li, M. Wu, H. Jiang, Chem. Commun. 2012, 48, 11446; (l) V. Tyagi, S. Khan, A. Giri, H. M. Gauniyal, B. Sridhar, P. M. S. Chauhan, Org. Lett. 2012, 14, 3126; (m) X.-D. Fei, Z.-Y. Ge, T. Tang, Y.-M. Zhu, S.-J. Ji, J. Org. Chem. 2012, 77, 10321; (n) Y. Ito, I. Ito, T. Hirao, T. Saegusa, Synth. Commun. 1974, 4, 97; (o) N. Thirupathi, M. H. Babu, V. Dwivedi, R. Kant, M. S. Reddy, Org. Lett. 2014, 16, 2908; (p) X. Jiang, T. Tang, J.-M. Wang, Z. Chen, Y.-M. Zhu, S.-J. Ji, J. Org. Chem., 2014, 79, 5082; (q) R. Mancuso, I. Ziccarelli, D. Armentano, N. Marino, S. V. Giofrè, B. Gabriele, J. Org. Chem., 2014, 79, 3506.
- 8 (a) T. Vlaar, R. C. Cioc, P.Mampuys, B.U.W. Maes, R. V. A. Orru, E. Ruijter, Angew. Chem. Int. Ed. 2012, 51, 13058; (b) B. Liu, M. Yin,

2

3

4

5

6

7

8

9

10

11 12

13

14

15

16

17

18

19

20

21

22

23

24

25

26

27

52

53

54

55

H. Gao, W. Wu, H. Jiang, J. Org. Chem., 2013, **78**, 3009; (c) T. Vlaar, R. V. A. Orru, B. U. W. Maes, E. Ruijter, J. Org. Chem., 2013, **78**, 10469. (d) T. Fang, Q. Tan, Z. Ding, B. Liu, B. Xu, Org. Lett., 2014, **16**, 2342. (e) T.-H. Zhu, S.-Y. Wang, G.-N. Wang, S.-J. Ji, Chem. Eur. J. 2013, **19**, 5850.

- 9 J. Wang, S. Luo, J. Huang, T. Mao, Q. Zhu, *Chem. Eur. J.* 2014, 20, 11220.
- 10 (a) Polycyclic Aromatic Hydrocarbons; R. G. Harvey, Eds.; Wiley-VCH: New York, 1996; (b) R. Capdeville, E. Buchdunger, J. Zimmermann, A. Matter, Nat. Rev. Drug Discovery 2002, 1, 493; (c) I. Cepanec, Synthesis of Biaryls, Elsevier, Amsterdam, 2004; (d) V. Balzani, A. Credi, M. Venturi, Molecular Devices and Machines, Wiley-VCH, Weinheim, 2008; (e) A. M. Norberg, L. Sanchez, R. E. Maleczka, Jr., Curr. Opin. Drug Discovery Dev. 2008, 11, 853; (f) Recent developments: S. Hiraoka, Y. Hisanaga, M. Shiro, M. Shionoya, Angew. Chem. Int. Ed. 2010, 49, 1669.
- (a) J. K. Stille, Angew. Chem., Int. Ed. 1986, 25, 508. (b) N. Miyaura, A. Suzuki, Chem. Rev. 1995, 95, 2457; (b) Diederich, F., Stang, P. J., Eds. Metal-catalyzed cross-coupling reactions; Wiley-VCH: New York, 1998; (c) S. P. Stanforth, Tetrahedron 1998, 54, 263; (d) J. Hassan, M. Se'vignon, C. Gozzi, E. Schulz, M. Lemaire, Chem. Rev. 2002, 102, 1359; (e) M. Schnu'rch, R. Flasik, A. F. Khan, M. Spina, M. D. Mihovilovic, P. Stanetty, Eur. J. Org. Chem. 2006, 3283; (f) S. Yanagisawa, T. Sudo, R. Noyori, K. Itami, J. Am. Chem. Soc. 2006, 128, 11748; (g) K. Billingsley, S. L. Buchwald, J. Am. Chem. Soc. 2007, 129, 3358.
- 28 12 For selected examples, see: (a) S.-L. You, J.-B. Xia, Top. Curr. Chem. 29 2010, 292, 165; (b) T.W. Lyons, M. S. Sanford, Chem. Rev. 2010, 30 110, 1147; (c) T. Newhouse, P. S. Baran, Angew. Chem. Int. Ed. 31 2011, 50, 3362; (d) C. S. Yeung, V. M. Dong, Chem. Rev. 2011, 111, 32 1215; (e) L. McMurray, F. O Hara, M. J. Gaunt, Chem. Soc. Rev. 33 2011, 40, 1885; (f) X. Bugaut, F. Glorius, Angew. Chem. Int. Ed. 34 2011, 50, 7479; (g) Z. Wang, K. Li, D. Zhao, J. Lan, J. You, Angew. 35 Chem. Int. Ed. 2011, 50, 5365; (h) J. Dong, Z. Long, F. Song, N. 36 Wu, Q. Guo, J. Lan, J. You, Angew. Chem. Int. Ed. 2013, 52, 580. (i) 37 S. Fan, Z. Chen, X. Zhang, Org. Lett., 2012, 14, 4950; (j) Z. Mao, Z. 38 Wang, Z. Xu, F. Huang, Z. Yu, R. Wang, Org. Lett., 2012, 14, 3854; 39 (k) N.-N. Li, Y.-L. Zhang, S. Mao, Y.-R. Gao, D.-D. Guo, Y.-Q. 40 Wang, Org. Lett., 2014, 16, 2732. 41
- 13 For selected examples, see: (a) W. Han, P. Mayer, A. R. Ofial, Angew.
 43 *Chem. Int. Ed.* 2011, **50**, 2178; (b) J. Dong, Y. Huang, X. Qin, Y.
 44 Cheng, J. Hao, D. Wan, W. Li, X. Liu, J. You, *Chem. Eur. J.* 2012,
 45 **18**, 6158; (c) M. Zhu, K. Fujita R. Yamaguchi, *Chem. Commun.*46 2011, **47**, 12876; (d) Y. Li, J. Jin, W. Qian, W. Bao, *Org. Biomol.*47 *Chem.*, 2010, **8**, 326.
- 48 14 (a) A. V. Gulevich, A. G. Zhdanko, R. V. A. Orru, V. G. Nenajdenko,
 49 *Chem. Rev.* 2010, 110, 5235; (b) A. Fayol, C. Housseman, X. Sun, P.
 50 Janvier, H. Bienaymé, J. Zhu, *Synthesis*, 2005, 1, 161; (c) C.
 51 Housseman, J. Zhu, *Synlett*, 2006, 11, 1777.
 - 15 K. L. Billingsley, T. E. Barder, S. L. Buchwald, Angew. Chem. Int. Ed. 2007, 46, 5359.
 - 16 J. P. Wolfe, R. A. Singer, B. H. Yang, S. L. Buchwald, J. Am. Chem. Soc. 1999, 121, 9550.

Page 4 of 4