Organic & Biomolecular Chemistry

Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this Accepted Manuscript with the edited and formatted Advance Article as soon as it is available.

You can find more information about *Accepted Manuscripts* in the **Information for Authors**.

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard <u>Terms & Conditions</u> and the <u>Ethical guidelines</u> still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.



www.rsc.org/obc

Cite this: DOI: 10.1039/c0xx00000x

www.rsc.org/xxxxxx

ARTICLE TYPE

N-Heterocyclic Carbene-Catalyzed Cyclocondensation of 2-Aryl Carboxylic Acid and Enones: Highly Enantioselective Synthesis of δ -Lactones.

50

Jin-Tang Cheng,^a Xiang-Yu Chen,^a and Song Ye*^{a,b}

s Received (in XXX, XXX) Xth XXXXXXXX 20XX, Accepted Xth XXXXXXXX 20XX DOI: 10.1039/b000000x

The enantioselective N-heterocyclic carbene-catalyzed [4 + 2] cyclocondensation of 2-aryl carboxylic acids and enones was developed, affording the corresponding chiral δ-lactones in 10 good yields with good diastereo- and high enantioselectivities.

The N-heterocyclic carbenes (NHCs) have been demonstrated as efficient organocatalysts for various reactions.¹ Particularly, the NHC-catalyzed generation of azolium enolates from α -chloroaldehydes,² formylcyclopropanes,³ α -aroyloxy ¹⁵ aldehydes,⁴ enals,⁵ ketenes,⁶ and esters⁷ have been extremely successful. However, the usage of stable and readily available carboxylic acids as the substrate for the NHC-catalyzed reactions has received little attention.8 Recently, Scheidt et al. reported the NHC-catalyzed enantioselective annulations for 20 dihydroquinolones by in situ generation of azolium enolates from carboxylic acid.9 Simultaneously, our group reported the NHCcatalyzed enantioselective [3 + 2] cyclocondensation of α,β unsaturated carboxylic acids with α -amino ketones and [3+3] cyclocondensation with imines.¹⁰



Scheme 1 NHC-catalyzed synthesis of δ -lactones

 δ -Lactones are widely found in numerous natural and unnatural bioactive compounds.¹¹ Thus, efficient construction of δ -lactones have been a continuing target for organic chemists.¹² ³⁰ In 2006, Bode *et al.* reported the pioneering NHC-catalyzed [4 + 2] cycloaddition of chloroaldehydes with enones to give δ lactones (Scheme 1, reaction a) ¹³ However, only the α -chloro- β , β -dihydroaldehydes were reported for the reaction, possible due to the sensitivity of α -chloroaldehydes to moisture and ³⁵ oxygen.¹⁴ Being readily available, stable and easy to manipulate, carboxylic acid would be useful alternative to chloroaldehyde for the NHC-catalyzed reaction. In this

This journal is © The Royal Society of Chemistry [year]

communication, we report the NHC-catalyzed generation of azolium enolate from carboxylic acids and the following [4 + 2]⁴⁰ cyclocondensation with enones to give δ -lactones (Scheme 1, reaction b).¹⁵

Table 1. Optimization of the reaction conditions.



1	A1	DCM	18	2:1	99(70)
2	A2	DCM	77	10:1	99(79)
3	B1	DCM	trace	/	/
4	B2	DCM	trace	/	/
5	A2	THF	25	4:1	99(77)
6	A2	ether	41	3:1	99(71)
7	A2	CH ₃ CN	19	1:2	92(97)
8	A2	toluene	39	16:1	99(45)
9	A2	DCM (0 °C)	68	3:1	99(87)
10	A2	DCM (rt)	1:6	3:1	99(70)

⁴⁵ ^a Isolated yield of the mixture of two diastereoisomers. ^b Determined by ¹H NMR (300 MHz) spectroscopy of the unpurified reaction mixture. ^c Enantiomeric excess of *cis*-**3a** followed of *trans*-**3a** in parenthesis. PivCl = Pivaloyl chloride, Mes = 2,4,6-trimethylphenyl, THF = tetrahydrofuran, TBS = *tert*-butyldimethylsilyl.

Initially, the reaction of 2-(2-methoxyphenyl)acetic acid 1a, via the in situ generated anhydride with pivaloyl chloride, with (enone 2a was investigated under NHC catalysis (Table 1). We were encouraged to find that the desired δ -lactone 3a was so obtained in 18% yield with 2:1 dr and 99% ee for *cis*-isomer and 70% ee for *trans*-isomer in the presence of 20 mol% of *N*-mesityl tetracyclic NHC precursor A1¹⁶ (entry 1). The yield and diastereoselectivity was dramatically improved with *N*-2,6diethylphenyl tetracyclic NHC precursor A2¹⁷ was empolyed as 45

Organic & Biomolecular Chemistry Accepted Manuscrip

the catalyst (entry 2). On the contrary, the reaction using the NHC precursor **B1** or **B2**, derived from *L*-pyroglutamic acid,¹⁸ gave only trace of desired cycloadduct (entries 3 and 4). Decreased yield and diastereoselectivity was resulted when the ⁵ reaction was carried out in THF, ether or acetonitrile (entries 5-7).

The reaction in toluene gave cycloadduct with improved diastereoselectivity but in low yield (entry 8). The diastereoselectivity was decreased when the reaction was carried out at 0 °C or rt, due to the partially epimerization of *cis-3a* to 10 *trans-3a* (entries 9 and 10).

Table 2. Substrate scope for enantioselective [4 + 2] cyclocondensation.

ArCO ₂ H + R	^O ^{VO2C} R	cat. A 2 (20 mol% PivCl (2.2 equiv. DIPEA (5.0 equiv DCM, -10 °C	(i) (i)
1a , Ar = 2-MeOC ₆ H ₄ 1b , Ar = C ₆ H ₅ 1c , Ar = 4-MeC ₆ H ₄ 1d , Ar = 4-MeC ₆ H ₄ 1d , Ar = 2,4-(MeO) ₂ C ₆ H ₃	2a, $R = Ph$, $R' = Et$ 2b, $R = 4-FC_6H_4$, R' 2c, $R = 4-GC_6H_4$, R' 2d, $R = 4-BrC_6H_4$, F 2e, $R = 4-MeC_6H_4$, 2g, $R = 2-CC_6H_4$, R' 2g, $R = 2-CC_6H_4$, R' 2h, $R = \beta$ -naphthyl, 2i, $R = Ph$, $R' = Me$ 2j, $R = 4-MeC_6H_4$, I 2k, $R = Me$, $R' = Et$? = Et ?' = Et R' = Et ,, R' = Et ,, R' = Et R' = Et R' = Me	3 (<i>cis</i> , major)

entry	1	2	3	y1eld" (%)	cis:trans [®]	ee (%) ^c
1	1a	2a	3aa	77	10:1	99
2	1a	2b	3ab	67	10:1	99
3	1a	2c	3ac	61	10:1	98
4	1a	2d	3ad	55	7:1	99
5	1a	2e	3ae	55	6:1	99
6	1a	2f	3af	80	10:1	99
7	1a	2g	3ag	63	6:1	98
8	1a	2h	3ah	33	3:1	99
9	1b	2a	3ba	88	1:1	ND
10	1c	2a	3ca	72	6:1	98
11	1d	2a	3da	58	3:1	99
12	1e	2a	3ea	54	6:1	99
13	1e	2e	3ee	66	20:1	99
14	1e	2f	3ef	50	>20:1	99
15	1a	2i	3ai	66	10:1	99
16	1a	2j	3aj	65	10:1	99
17	1a	2k	3ak	27	>20:1	97

^{*a*} Isolated yield of the mixture of two diastereoisomers. ^{*b*} Determined by ¹⁵ ¹H NMR (300 MHz) spectroscopy of the unpurified reaction mixture. ^{*c*} Enantiomeric excess of *cis*-**3**. ND = not determined.

With the optimized conditions in hand, the scope of the reaction substrates was briefly investigated (Table 2). It was found that both enones **2** with electron-withdrawing group (4-²⁰ FC₆H₄, 4-ClC₆H₄, 4-BrC₆H₄) and those with electron-donating group (4-MeC₆H₄, 4-MeOC₆H₄) were suitable substrates, furnishing the corresponding δ-lactones in good yields and good diastereoselectivities and excellent enantioselectivities (entries 2-6). The reaction of enone **2g** with 2-chlorophenyl gave the

- ²⁵ cycloadduct **3ag** in 63% yield, 6:1 dr and 98% ee (entry 7), while the reaction of enone **2h** with β -naphthyl resulted in decreased yield and diastereoselectivity, but exellent enantioselectivity was kept (entry 8). Several 2-arylacetic acids were also tested for the reaction. It was found that the reaction of simple phenylacetic
- 30 acid (1b) gave the δ-lactone 3ba in high yield but with 1:1 dr (entry 9). 2-Arylacetic acids with 4-methylphenyl or 4methoxylphenyl (1c and 1d) worked well for the reaction

This journal is © The Royal Society of Chemistry [year]

resulting in moderate diastereoselectivities but excellent enantioselectivites (entries 10 and 11). 2,4-35 Dimethoxyphenylacetic acid (1e) was also a workable substrate for the reaction with several enones (2a, 2e, 2f), giving the desired corresponding δ-lactones in good yield with good to high diastereoselectivities and excellent enantioselectivities (entries 12-14). The reaction of enones 2i and 2j with methyl esters 40 instead of ethyl esters went also well (entires 15 and 16). The reaction of enone 2k with alkyl instead of aryl substituent (R' = Et) gave the desired product 3ak in low yield albeit with high diastereo- and enantioselectivity (entry 17).



Figure 1. X-Ray structure of 3aa

The 3*S*, 4R-configuration of the cycloadduct **3aa** was determined by the X-ray analysis of its crystal. (Figure 1).¹⁹



Figure 2. Plausible catalytic cycle.

⁵⁰ A plausible catalytic cycle for the NHC-catalyzed [4 + 2] cyclocondensation of 2-arylacetic acid with enones is depicted in Figure 2. The addition of NHC to mixed anhydride I, which is formed in situ from the 2-arylacetic acid and pivaloyl chloride, gives acyl azolium II. The Michael addition of enolate III, which ⁵⁵ is equivalent with the acyl azolium II in the presence of base, affords adduct IV.²⁰ The intramolecular acylation of adduct IV

^{2 |} Journal Name, [year], [vol], 00–00

gives the final desired δ -lactone **3** and regenerates the NHC catalyst.

Conclusions

- In summary, carboxylic acids were demonstrated as suitable 5 substrates for the NHC-catalyzed reactions. The chiral NHCcatalyzed [4 + 2] cyclocondensation of carboxylic acid and enones gave the corresponding δ -lactones in good yields with good diastereo- and excellent ennatioselectivities. The readily availability, easy to manipulate and mild reaction conditions
- 10 make the NHC-catalyzed reaction of carboxylic acid potential useful for many related chemical transfromations involving azolium enolates. Other related NHC-catalyzed reactions of carboxylic acids are underway in our laboratory.

15 Acknowledgements

Financial support from the Ministry of Science and Technology of China (2011CB808600), National Science Foundation of China (21272237), the Open Project Program of National Engineering Research Center for Carbohydrate Synthesis of

20 Jiangxi Normal University, and the Chinese Academy of Sciences are greatly acknowledged.

Notes and references

30

^a Beijing National Laboratory for Molecular Sciences, Key Laboratory of

25 Molecular Recognition and Function. Institute of Chemistry, Chinese Academy of Sciences, Beijing 100190, China. Fax: (+86)10 6255 4449; Tel: (+86)10 6264 1156; E-mail: songye@iccas.ac.cn

^b National Engineering Research Center for Carbohydrate Synthesis, Jiangxi Normal University, Nanchang 330022, China

- † Electronic Supplementary Information (ESI) available: [details of any supplementary information available should be included here]. See DOI: 10.1039/b000000x/
- ‡ Footnotes should appear here. These might include comments relevant 35 to but not central to the matter under discussion, limited experimental and spectral data, and crystallographic data.
- 1. (a) D. Enders and T. Balensiefer, Acc. Chem. Res., 2004, 37, 534; (b) D. Enders, O. Niemeier and A. Henseler, Chem. Rev., 2007, 107, 5606; (c) X. Bugaut and F. Glorius, Chem. Soc. Rev., 2012, 41, 3511; (d) J.
- Douglas, G. Churchill and A. D. Smith, Synthesis, 2012, 44, 2295; (e) 40 A. Grossmann and D. Enders, Angew. Chem., Int. Ed., 2012, 51, 314; (f) J. Izquierdo, G. E. Hutson, D. T. Cohen and K. A. Scheidt, Angew. Chem., Int. Ed., 2012, 51, 11686; (g) H. U. Vora, P. Wheeler and T. Rovis, Adv. Synth. Catal., 2012, 354, 1617; (h) J. W. Bode, Nat.
- Chem., 2013, 5, 813; (i) S. J. Ryan, L. Candish and D. W. Lupton, 45 Chem. Soc. Rev., 2013, 42, 4906; (j) J. Mahatthananchai and J. W. Bode, Acc. Chem. Res., 2014, 47, 696.
- 2. (a) T. Y. Jian, L. H. Sun and S. Ye, Chem.Commun., 2012, 48, 10907; (b) L. Yang, F. Wang, P. J. Chua, Y. Lv, L. J. Zhong and G. Zhong,
- Org. Lett., 2012, 14, 2894; (c) Q. Ni, H. Zhang, A. Grossmann, C. C. 50 J. Loh, C. Merkens and D. Enders, Angew. Chem., Int. Ed., 2013, 52, 13562; (d) H. M. Zhang, H. Lv and S. Ye, Org. Biomol. Chem., 2013, 11, 6255; (e) L. Yang, F. Wang, R. Lee, Y. Lv, K. W. Huang and G. F. Zhong, Org. Lett., 2014, 16, 3872.
- 55 3. H. Lv, J. Mo, X. Fang and Y. R. Chi, Org. Lett., 2011, 13, 5366.
- 4. (a) A. T. Davies, J. E. Taylor, J. Douglas, C. J. Collett, L. C. Morrill, C. Fallan, A. M. Z. Slawin, G. Churchill and A. D. Smith, J. Org. Chem., 2013, 78, 9243; (b) J. E. Taylor, D. S. B. Daniels and A. D. Smith, Org. Lett., 2013, 15, 6058. (c) A. T. Davies, P. M. Pickett, A. M. Z. Slawin and A. D. Smith, ACS Catal., 2014, 4, 2696.

- (a) J. Kaeobamrung, M. C. Kozlowski and J. W. Bode, Proc. Natl. 5. Acad. Sci. U.S.A., 2010, 107, 20661. (b) X. Fang, X. Chen and Y. R. Chi, Org. Lett., 2011, 13, 4708; (c) X. Fang, K. Jiang, C. Xing, L. Hao and Y. R. Chi, Angew. Chem., Int. Ed., 2011, 50, 1910. (d) Y. M.
- Zhao, M. S. Cheung, Z. Lin and J. W. Sun, Angew. Chem., Int. Ed., 65 2012, 51, 10359; (e) X. Q. Dong and J. W. Sun, Org. Lett., 2014, 16, 2450
 - 6. (a) L. He, H. Lv, Y. R. Zhang and S. Ye, J. Org. Chem., 2008, 73, 8101; (b) H. Lv, Y. R. Zhang, X. L. Huang and S. Ye, Adv. Synth.
- Catal., 2008, 350, 2715; (c) H. Lv, L. You and S. Ye, Adv. Synth. Catal., 2009, 351, 2822; (d) X. N. Wang, H. Lv, X. L. Huang and S. Ye, Org. Biomol. Chem., 2009, 7, 346; (e) H. Lv, X. Y. Chen, L. H. Sun and S. Ye, J. Org. Chem., 2010, 75, 6973; (f) T. Y. Jian, L. He, C. Tang and S. Ye, Angew. Chem., Int. Ed., 2011, 123, 9270.
- (a) L. Hao, Y. Du, H. Lv, X. Chen, H. Jiang, Y. Shao and Y. R. Chi, 75 7. Org. Lett., 2012, 14, 2154. (b) S. J. Chen, L. Hao, Y. X. Zhang, B. Tiwari and Y. R. Chi, Org. Lett., 2013, 22, 5822; (c) L. Hao, S. Chen, J. Xu, B. Tiwari, Z. Fu, T. Li, J. Lim and Y. R. Chi, Org. Lett., 2013, 15.4956
- 80 8. For the direct catalytic α -functionalizations of carboxylic acid using amine-based nucleophilic catalyst, see: (a) G. S. Cortez, R. L. Tennyson and D. Romo, J. Am. Chem. Soc., 2001, 123, 7945; (b) V. C. Purohit, A. S. Matla and D. Romo, J. Am. Chem. Soc., 2008, 130, 10478; (c) S. H. Oh, G. S. Cortez and D. Romo, J. Org. Chem., 2005,
- 70, 2835; (d) H. Nguyen, G. Ma, T. Gladysheva, T. Fremgen and D. Romo, J. Org. Chem., 2010, 76, 2; (e) H. Henry-Riyad, C. Lee, V. C. Purohit and D. Romo, Org. Lett., 2006, 8, 4363; (f) G. Ma, H. Nguyen and D. Romo, Org. Lett., 2007, 9, 2143; (g) C. A. Leverett, V. C. Purohit and D. Romo, Angew. Chem., Int. Ed., 2010, 49, 9479;
- (h) D. Belmessieri, L. C. Morrill, C. Simal, A. M. Z. Slawin and A. D. 90 Smith, J. Am. Chem. Soc., 2011, 133, 2714; (i) L. C. Morrill, T. Lebl, A. M. Z. Slawin and A. D. Smith, Chem. Sci., 2012, 3, 2088; (j) C. Simal, T. Lebl, A. M. Z. Slawin and A. D. Smith, Angew. Chem., Int. Ed., 2012, 51, 3653; (k) L. C. Morrill, J. Douglas, T. Lebl, A. M. Z.
- Slawin, D. J. Fox and A. D. Smith, Chem. Sci. 2013, 4, 4146; (1) L. C. Morrill, S. M. Smith, A. M. Z. Slawin and A. D. Smith, J. Org. Chem., 2014, 79, 1640; (m) S. R. Smith, J. Douglas, H. Prevet, P. Shapland, A. M. Z. Slawin and A. D. Smith, J. Org. Chem., 2014, 79, 1626.
- 100 9. A. Lee, A. Younai, C. K. Price, J. Izquierdo, R. K. Mishra and K. A. Scheidt, J. Am. Chem. Soc., 2014, 136, 10589.

rganic & Biomolecular Chemistry Accepted Manuscrip

- 10. X. Y. Chen, Z. H. Gao, C. Y. Song, C. L. Zhang, Z. X. Wang and S. Ye, Angew. Chem., Int. Ed., 2014, 53, 11611.
- 11. (a) V. Boucard, G. Broustal and J. M. Campagne, Eur. J. Org. Chem., 2007, 2, 225; (b) J. Langen, C. Y. Wang, P. Slabizki, K. Wall and H. 105 G. Schmarr, Rapid. Commun. Mass Spectrom., 2013, 27, 2751.
 - 12. (a) K. A. Jørgensen, Angew. Chem., Int. Ed., 2000, 39, 3558; (b) K. Juhl and K. A. Jørgensen, Angew. Chem., Int. Ed., 2003, 42, 1498. (c) S. Aspin, L. López-Suárez, P. Larini, A. S. Goutierre, R. Jazzar and O. Baudoin, Org. Lett., 2013, 15, 5056. (d) T. Tozawa, H. Nagao, Y.
 - Yamane and T. Mukaiyama, Chem. Asian J., 2007, 2, 123. 13. M. He, G. J. Uc and J. W. Bode, J. Am. Chem. Soc., 2006, 128, 15088;
 - 14. (a) M. He, B. J. Beahm and J. W. Bode, Org. Lett., 2008, 10, 3817. (b)
 - N. De Kimpe and R. Verhe, The Chemistry of α -Haloketones, α -Haloaldehydes, and *a*-Haloimines; John Wiley & Sons: New York, 1988; Chapter 3.
 - 15. The NHC-catalyzed synthesis of δ -lactones from ketenes and enones was reported by our group: Y. R. Zhang, H. Lv, D. Zhou and S. Ye, Chem. Eur. J., 2008, 14, 8473.
- 120 16. (a) A. K. Ghosh, S. P. McKee and W. M. Sanders, Tetrahedron Lett. 1991, 32, 711; (b) M. S. Kerr, J. Read de Alaniz and T. Rovis, J. Org. Chem. 2005, 70, 5725; (c) M. He, J. R. Struble, and J. W. Bode, J. Am. Chem. Soc. 2006, 128, 8418.
- 17. B. Cardinal-David, D. E. A. Raup and K. A. Scheidt, J. Am. Chem. Soc., 2010, 132, 5345. 125
- 18. (a) Y. R. Zhang, L. He, X. Wu, P. L. Shao and S. Ye, Org. Lett., 2007, 10, 277; (b) D. Enders and J. Han, Tetrahedron: Asymmetry, 2008, 19,1367. (c) L. He, Y. R. Zhang, X. L. Huang and S. Ye, Synthesis, 2008, 2825; (d) X. L. Huang, L. He, P. L. Shao and S. Ye, Angew. Chem., Int. Ed., 2009, 48, 192; (e) H. M. Zhang, Z. H. Gao and S. Ye, 130 Org. Lett., 2014, 16, 3079.

Journal Name, [year], [vol], 00-00 | 3

This journal is © The Royal Society of Chemistry [year]

- 19, CCDC 1029598 (3aa) contains the supplementary crystallographic data for th0.is paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data request/cif.
- 5 20. The concerted [4 + 2] cycloaddition of enolate and enones is less likely according to the moderate diastereoselectivities observed for most cases.