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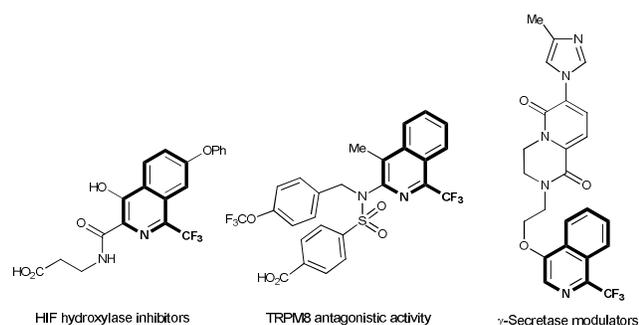
# 1-Trifluoromethylated isoquinolines *via* radical trifluoromethylation of isonitriles

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A simple and efficient approach to biologically important 1-trifluoromethylated isoquinolines starting with readily prepared  $\beta$ -aryl- $\alpha$ -isocyano-acrylates and commercially available Togni reagent as  $\text{CF}_3$  radical precursor is described. These transformations occur in the absence of any transition metal and the title compounds are obtained in moderate to excellent yields. This protocol comprises a trifluoromethylation with concomitant isoquinoline framework construction.

In the past decade, the synthesis of  $\text{CF}_3$ -containing aromatic compounds has caught great attention from the synthetic community because of their wide application not only in materials science and agrochemistry, but also in medicinal chemistry.<sup>1,2</sup> The arene solubility and lipophilicity can be significantly improved upon introducing a trifluoromethyl group as an arene substituent. As compared to their non  $\text{CF}_3$ -congeners, trifluoromethylated arenes show higher membrane permeability, increased bioavailability<sup>3</sup> and higher metabolic stability. Therefore, the development of novel and efficient methods for incorporation of the  $\text{CF}_3$  moiety into aromatic compounds is of importance.



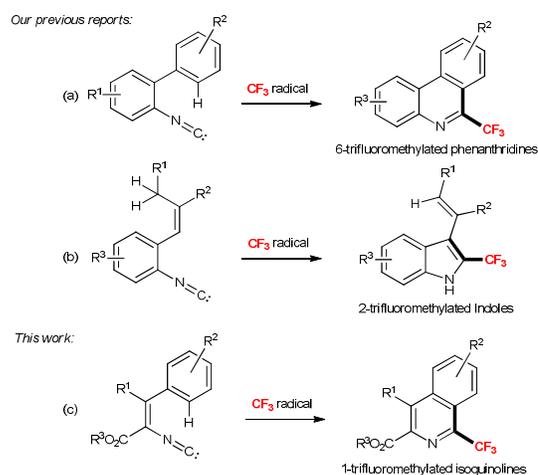
**Figure 1** Representative biologically active *N*-heterocycles containing the 1-trifluoromethylated isoquinoline scaffold.

The isoquinoline scaffold is a privileged chemical entity, which can be found in various natural products, drug candidates, and biologically active compounds.<sup>4</sup> There is continuing interest in the development of novel synthetic methods for the preparation of isoquinolines.<sup>5</sup> In particular, novel synthetic approaches to 1-trifluoromethylated isoquinolines are required due to their prominent biological

activities (Fig. 1).<sup>6</sup>

Conventional methods for preparation of 1-trifluoromethylated isoquinolines rely on coupling reactions of aryl halides with trifluoromethylation reagents<sup>6d,6e,6f,6g</sup> and on the Bischler-Napieralski reaction<sup>6a</sup>. However, the former approach requires halogenation of the preformed isoquinoline core and the latter approach suffers from multistep reaction sequences and generally harsh reaction conditions. As a complementary approach we present herein radical trifluoromethylation of isonitriles for preparation of 1-trifluoromethylated isoquinolines.

Recently, the radical isonitrile insertion reaction has emerged as a powerful strategy for the construction of various heteroarenes.<sup>7,8</sup> Along the lines, we have disclosed that isonitriles can be used as highly efficient  $\text{CF}_3$  radical acceptors for preparation of 6-trifluoromethylated phenanthridines<sup>8b</sup> and 2-trifluoromethylated indoles<sup>8g</sup> (Scheme 1, a and b).



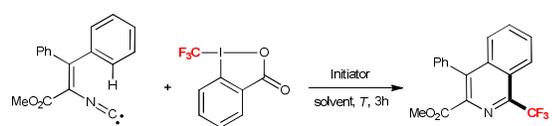
**Scheme 1** Construction of important *N*-heterocycles by radical trifluoromethylation of isonitriles.

Encouraged by these results, we envisioned that 1-trifluoromethylated isoquinolines can be obtained by  $\text{CF}_3$  radical addition to  $\beta$ -aryl- $\alpha$ -isocyano-acrylates with subsequent base-promoted homolytic aromatic substitution (Scheme 1, c).<sup>9,10</sup>

We have recently shown that the  $\text{CF}_3$  radical can be

efficiently generated from the commercially available Togni reagent **2** by using Bu<sub>4</sub>Ni as a radical initiator.<sup>8b,g</sup> We therefore initiated our studies by investigating radical trifluoromethylation of readily prepared  $\alpha$ -isocyano cinnamic acid ester **1a** with **2** (1.2 equiv) in the presence of Bu<sub>4</sub>Ni as an initiator in 1,4-dioxane at 80 °C for 3 h. Gratifyingly, the desired isoquinoline **3a** was obtained in 75% yield (Table 1, entry 1). A worse result was achieved by using Bu<sub>4</sub>NBr as an initiator (Table 1, entry 2). The Bu<sub>4</sub>Ni loading could be further lowered to 5 mol % without affecting the yield, but a further lowering of the Bu<sub>4</sub>Ni concentration to 1 mol % led to a decreased yield (62%, Table 1, entries 3 and 4). Other solvents, such as CH<sub>3</sub>CN, DCE, and EtOAc provided lower yields (Table 1, entries 5-7). Decreasing temperature to 60 °C also afforded a slightly lower yield (Table 1, entry 8) and increasing temperature to 100 °C did not affect the yield to a large extent (Table 1, entry 9). However, yield was further increased upon using 1.5 equiv of **2** (80%, Table 1, entry 10).

**Table 1** Optimization of reaction conditions<sup>a</sup>



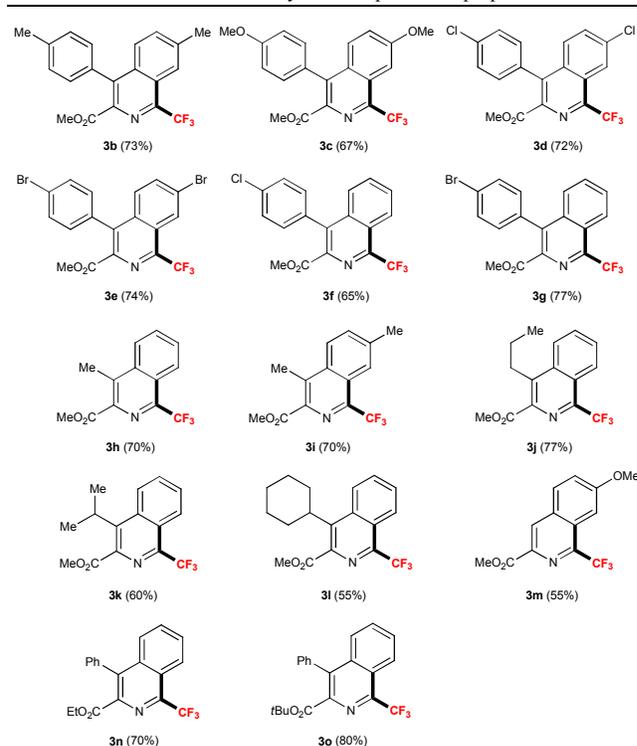
Entry	Initiator	mol %	Solvent	T [°C]	Yield <sup>b</sup> (%)
1	Bu <sub>4</sub> Ni	10	1,4-dioxane	80	75
2	Bu <sub>4</sub> NBr	10	1,4-dioxane	80	43
3	Bu <sub>4</sub> Ni	5	1,4-dioxane	80	76
4	Bu <sub>4</sub> Ni	1	1,4-dioxane	80	62
5	Bu <sub>4</sub> Ni	5	CH <sub>3</sub> CN	80	41
6	Bu <sub>4</sub> Ni	5	DCE	80	20
7	Bu <sub>4</sub> Ni	5	EtOAc	80	70
8	Bu <sub>4</sub> Ni	5	1,4-dioxane	60	70
9	Bu <sub>4</sub> Ni	5	1,4-dioxane	100	76
10 <sup>c</sup>	Bu <sub>4</sub> Ni	5	1,4-dioxane	80	80

<sup>a</sup> All reactions were carried out with **1a** (0.2 mmol), **2** (0.24 mmol), and initiator (0.02 mmol) in 1,4-dioxane (1.0 mL) at 80 °C under Ar for 3 h. <sup>b</sup> Isolated yields. <sup>c</sup> Using 1.5 equiv of **2**.

With optimized reaction conditions in hand, the scope and limitations of the isoquinoline synthesis were investigated (Table 2). Reactions with various  $\beta,\beta$ -diaryl- $\alpha$ -isocyanoacrylates derived from symmetrical and unsymmetrical diaryl ketones proceeded well (for the synthesis of the isonitriles, see Supporting Information), and the corresponding heteroarenes **3b-3g** were isolated in good yields (65-77%, Table 2). Electronic effects of the substituents at the arene rings did not affect the reaction outcome to a large extent. Importantly, olefin isomerization did not occur under the applied conditions and homolytic aromatic substitution occurred regioselectively at the arene ring *cis* to the isonitrile functionality (see **3f** and **3g**).  $\beta$ -Alkyl- $\alpha$ -isocyano cinnamic acid esters, where the aryl and isonitrile groups are *cis*-oriented, undergo the radical cascade reaction smoothly to provide the corresponding isoquinolines **3h-3l** in moderate to good yields (55-77%). A slightly lower but still good yield was obtained for an  $\alpha$ -isocyano cinnamic acid ester derivative lacking the additional  $\beta$ -substituent (**3m**: 55%, Table 2). The

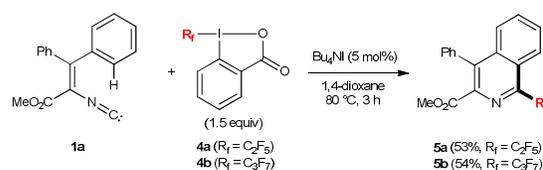
ester group in the  $\beta$ -phenyl- $\alpha$ -isocyano cinnamic acid esters (ethyl and *tert*-butyl) could also be varied to give the corresponding isoquinolines **3n** and **3o** in good yields (**3n**: 70%; **3o**: 80%). To document the practicability of the method, we repeated the final experiment at larger scale (2 mmol) and isolated isoquinoline **3o** in 71% yield (0.53 g, see Supporting Information).

**Table 2** Various 1-trifluoromethylated isoquinolines prepared.<sup>a,b</sup>



<sup>a</sup> All reactions were carried out with **1** (0.2 mmol), **2** (0.3 mmol), and Bu<sub>4</sub>Ni (0.01 mmol) in 1,4-dioxane (1.0 mL) at 80 °C under Ar for 3 h. <sup>b</sup> Isolated yields.

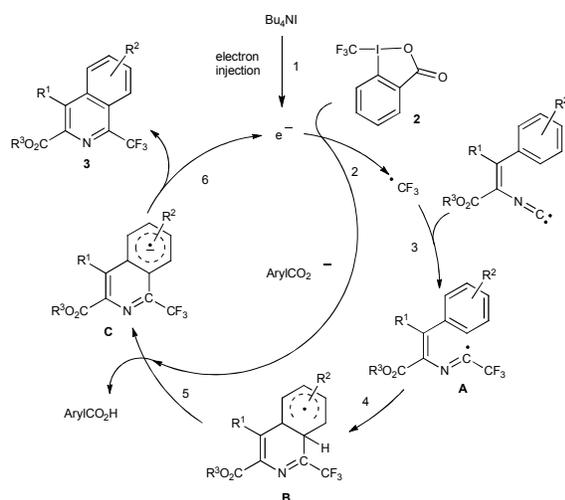
Notably, this method can also be applied to the synthesis of perfluoroalkylated isoquinolines. To this end, the I<sup>III</sup> reagents **4a**<sup>11</sup> and **4b**<sup>8b</sup> were applied to the radical cascade reaction using **1a** as a substrate to give **5a** and **5b**, documenting the potential of the new method for the preparation of 1-perfluoroalkylated isoquinolines.



**Scheme 2** Preparation of 1-perfluoroalkylated isoquinolines.

Preliminary mechanistic studies revealed that CF<sub>3</sub>-radicals are likely involved in these reactions. Under standard conditions, formation of **3a** from **1a** was fully suppressed in the presence of 2,2,6,6-tetramethylpiperidine-1-oxyl (TEMPO) as a radical scavenger, and the TEMPO-CF<sub>3</sub> adduct was detected by <sup>19</sup>F NMR spectroscopy (see Supporting Information). Based on this experiment, our previous reports,<sup>8b,8g</sup> and our recent conceptual article,<sup>12</sup> a plausible

mechanism considering the electron as a catalyst is proposed in Scheme 3. The catalytic cycle is started by electron injection (step 1) from the iodide anion to the Togni reagent **2** generating a CF<sub>3</sub>-radical and the *ortho*-iodobenzoic acid anion with Bu<sub>4</sub>N<sup>+</sup> as the counteranion (step 2). CF<sub>3</sub>-radical addition to the isonitrile generates an imidoyl radical **A** (step 3),<sup>7</sup> which cyclizes to the arene to give cyclohexadienyl radical **B** (step 4). We assume that **B** gets deprotonated by *ortho*-iodobenzoate (ArylCO<sub>2</sub><sup>-</sup>) to radical anion **C** (step 5).<sup>9</sup> This radical anion then further reacts to the product isoquinoline **3** thereby formally liberating the catalytic electron closing the catalytic cycle (step 6).



**Scheme 3** Proposed reaction mechanism documenting the role of the electron as a catalyst.

In summary, we have demonstrated a simple and efficient method for the synthesis of 1-trifluoromethylated isoquinolines starting with readily prepared  $\beta$ -aryl- $\alpha$ -isocyano-acrylates. The radical cascade uses the commercially available Togni reagent **2** as CF<sub>3</sub> radical precursor. Various 1-trifluoromethylated isoquinolines were successfully prepared in moderate to excellent yields. Importantly, the method is also applicable to the synthesis of 1-perfluoroalkylated isoquinolines. This transformation occurs without the help of any transition metals and products obtained are of biological importance.

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## Notes and references

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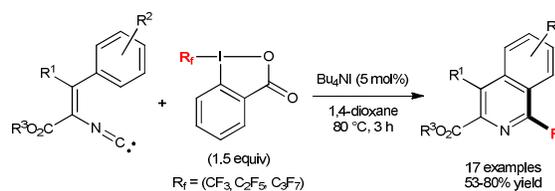
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ARTICLE TYPE

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A simple and efficient approach to biologically important 1-trifluoroalkylated isoquinolines starting with readily prepared  $\beta$ -aryl- $\alpha$ -isocyano-acrylates and  $R_f$ -I(III)-reagents ( $R_f = \text{CF}_3, \text{C}_2\text{F}_5, \text{C}_3\text{F}_7$ ) is described.