

Organic & Biomolecular Chemistry

Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this *Accepted Manuscript* with the edited and formatted *Advance Article* as soon as it is available.

You can find more information about *Accepted Manuscripts* in the [Information for Authors](#).

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard [Terms & Conditions](#) and the [Ethical guidelines](#) still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.

Copper-Catalysed α -Selective Allylic Alkylation of Heteroaryllithium Reagents

Cite this: DOI: 10.1039/x0xx00000x

Carlos Vila, Valentín Hornillos, Martín Fañanás-Mastral and Ben L. Feringa*

Received 00th January 2012,
Accepted 00th January 2012

DOI: 10.1039/x0xx00000x

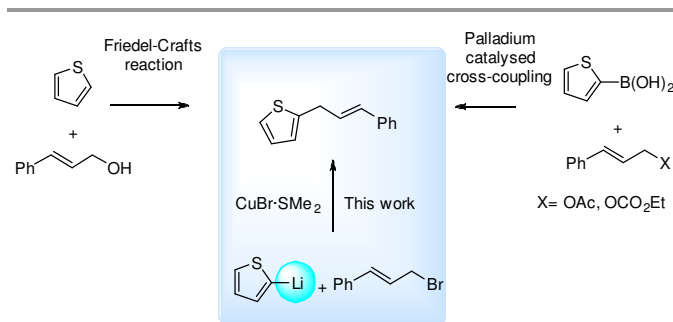
www.rsc.org/

2-Allyl-substituted thiophenes and furans are synthesised efficiently in a direct procedure using 2-heteroaryllithium reagents and allyl bromides and chlorides catalysed by ligand-free copper (I). The reactions take place under mild conditions, with excellent α -selectivity, high functional group tolerance and good yields for the S_N2 products.

Introduction

Heteroarenes such as thiophene and furan have attracted great attention in the last decades, due to their versatility in synthesis and specific properties.¹ These heterocyclic compounds have been used extensively in material science for organic dyes and electronic devices,² in agriculture and pharmaceutical chemistry³ or as intermediates for the synthesis of natural products or flavours.⁴ Therefore, the functionalization of furans and thiophenes at C2 represents an important target for organic synthesis.¹ In this context, the allylic substitution is a very convenient reaction,⁵ due to the fact that the incorporated olefin motif may serve as a latent group for further transformations. There are a number of methods for allylic substitutions with heterocycles (Scheme 1). For example, the Friedel-Crafts⁶ reaction represents a straightforward manner to access to 2-allylheteroarenes, but usually this transformation suffers from a lack of regioselectivity, both in the heteroarene and in the allyl electrophile, especially when thiophene is used as a nucleophile.⁷ On the other hand, several successful cross-coupling reactions catalysed by palladium⁸ and copper⁹ have been described, but in these cases a pre-functionalised thiophene or furan is needed. Usually, a boronic, organotin or organosilicon reagent is used in these cross-coupling reactions. As part of our continuing efforts to employ organolithium¹⁰ reagents in asymmetric allylic

substitutions¹¹ and palladium cross-coupling reactions,^{12,13} we were interested in the direct allylic alkylation of 2-heteroaryl lithium reagents. Herein, we present an α -selective allylic alkylation of 2-heteroaryl lithium reagents using copper as a catalyst.¹⁴ In this way, the initial functionalization of the organometallic reagent is avoided, which allows the synthesis of 2-allyl-substituted heterocycles in an efficient and straightforward procedure.



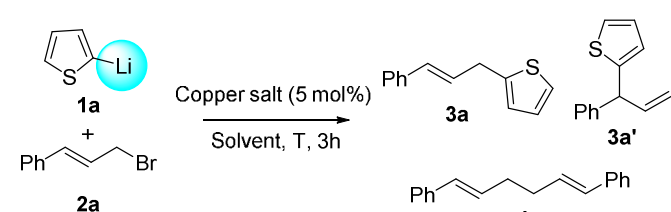
Scheme 1 Different methodologies for allylic alkylation of thiophene.

Results and discussion

Our studies began with the reaction of the 2-thienyllithium¹⁵ and cinnamyl bromide, both commercially available. Different solvents, temperatures and copper sources were evaluated (Table 1). Initially, solvents such as toluene, TBME and THF (entries 1-3, respectively), using $\text{CuBr}\cdot\text{SMe}_2$ as a catalyst at $-80\text{ }^\circ\text{C}$, were tested. Gratifyingly, full conversion and complete S_N2 selectivity (99:1, linear:branched) was achieved when THF was used as a solvent. This copper (I) catalysed allylic

alkylation reaction shows high regioselective to the α -substituted product, independently of the absence or presence of a ligand.¹⁶ Importantly, when copper(I) was not used the conversion dropped significantly (entry 4). We decided for practical reasons to carry out the reaction at higher temperature, therefore the allylic alkylation was tested at 0 °C. A screening of different copper (I) salts (entries 5-7) revealed that CuBr·SMe₂ is the most efficient catalyst for this transformation. When the reaction was run at 0 °C, a small amount of homocoupling product **4** was still formed (entry 5). Finally, when the reaction was performed at -5 °C in THF and using CuBr·SMe₂, full conversion was achieved (86% isolated yield of **3a**) without the presence of product **4**. When the reaction was carried out at room temperature, a complex mixture was obtained and the ¹H NMR of the crude mixture was difficult to analyse. Interestingly, when cinnamyl chloride was used (entry 12, table 1), full conversion and high regioselectivity to the linear product were also observed and the corresponding product **3a** was obtained in 93% yield.

Table 1 Optimization of the reaction conditions.^a

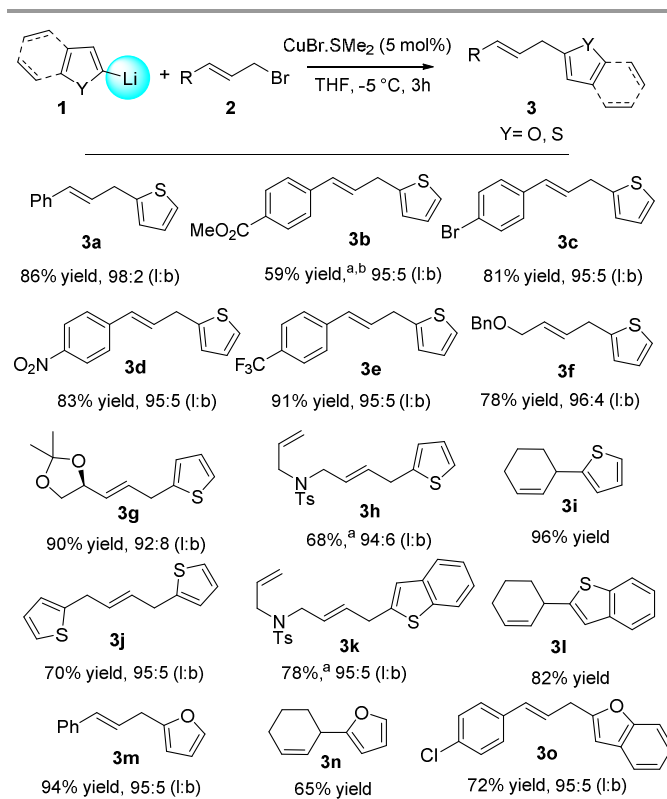


Entry	Solvent	copper	T (°C)	Conv. ^b (%)	3a+3a' ^b (l:b) ^c (yield) ^d	4 (%) ^b
1	Toluene	CuBr·SMe ₂	-80	95	90% (99:1)	5
2	TBME	CuBr·SMe ₂	-80	90	80% (99:1)	10
3	THF	CuBr·SMe ₂	-80	Full	Full (99:1)	0
4	THF	-	-80	35	25% (99:1)	10
5	THF	CuBr·SMe ₂	0	Full	94% (97:3)	6
6	THF	CuCl	0	55	50% (98:2)	5
7	THF	CuI	0	80	60% (98:2)	20
8	THF	-	0	80	60% (98:2)	20
9	THF	CuBr·SMe ₂	-5	Full	Full (98:2) (86%)	0
10 ^e	THF	CuBr·SMe ₂	-5	Full	Full (98:2) (93%)	0

^a Reaction conditions: copper salt (0.01 mmol, 5 mol%), 1.5 eq. of **1a** and 0.2 mmol of **2a** in 2 mL of solvent. ^b Conversions were determined by ¹H NMR. ^c Linear:branched ratio was determined by GC. ^d Isolated yield after column chromatography. ^e cinnamyl chloride was used as a allylic reagent.

With the optimised conditions in hand (Table 1, entry 9), the scope of this reaction was investigated (Scheme 2). It should be emphasized that high functional group tolerance was observed and cinnamyl bromides with different substituents, such as methyl ester, bromide,¹⁷ NO₂ or CF₃ groups in *para* position at the aromatic ring can be present, affording the corresponding products **3b-3e**, in good to high yields with excellent levels of regioselectivity. Other functional groups such as benzyl ether or a dioxalane ring were allowed and the corresponding products **3f** and **3g**, were obtained in 78 and 90% yield, respectively. Also, the presence of N-Ts-protected amines was tolerated, but in this case 1.1 eq. of organolithium reagent was used in order to obtain good yield. Furthermore,

multiple coupling of **1a** is shown in the twofold alkylation of (*E*)-1,4-dibromobut-2-ene, providing the corresponding dialkylated product **3j** in 70% isolated yield. Next, different heteroarylithium reagents were tested. For example benzo[*b*]thiophen-2-ylithium, easily prepared by direct metallation with *n*-BuLi,¹⁸ was successfully coupled with (*E*)-*N*-allyl-*N*-(4-bromobut-2-en-1-yl)-4-methylbenzenesulfonamide and 3-bromocyclohexene, affording the corresponding products **3k** (78%) and **3l** (81%) in high yields. 2-Furanylithium and 2-benzofuranylithium, freshly prepared by direct metallation, were also suitable partners for this reaction, resulting in products **3m-3o**, with high regioselectivity and good yields.



Scheme 2 Scope of copper(I)-catalysed allylic alkylation. Reaction conditions: allyl bromide **2** (0.2 mmol) was added to a stirred solution of CuBr·SMe₂ (0.01 mmol) in 2 mL of dry THF at -5 °C; 2-heteroarylithium reagent **1** (0.3 mmol) was added dropwise over 1 h. Isolated yield after column chromatography. Linear:branched ratio determined by GC. ^a 1.1 eq. of 2-heteroarylithium reagent **1** was used. ^b 1,2-addition to the carbonyl group was also observed as a side reaction.

Conclusions

In summary, we have developed a highly regioselective ligand-free copper(I) catalysed allylic alkylation using directly 2-heteroarylithium reagents. The corresponding α -substituted products are obtained in good to excellent yields (up to 94%). The reaction takes place under mild conditions and tolerates a wide range of functional groups and offers a method for direct access to various C2-substituted heterocycles.

Acknowledgements

Financial support from The Netherlands Organization for Scientific Research (NWO-CW), National Research School Catalysis (NRSC-Catalysis), the European Research Council (ERC advanced grant 227897), the Royal Netherland Academy of Arts and Sciences (KNAW) and the Ministry of Education Culture and Science (Gravity program 024.601035) is gratefully acknowledged. C.V. was supported by Intra-European Marie Curie fellowship (FP7-PEOPLE-2011-IEF-300826). We thank T. D. Tiemersma-Wegman for the HRMS analysis.

Notes and references

Stratingh Institute for Chemistry, University of Groningen, Nijenborgh 4, 9747 AG Groningen, The Netherlands. Fax: +31 50 363 4278; Tel: +31 50 3634296; E-mail: b.l.feringa@rug.nl.

Electronic Supplementary Information (ESI) available: [details of any supplementary information available should be included here]. See DOI: 10.1039/c000000x/

- (a) A. R. Katritzky and C. W. Rees, in *Comprehensive Heterocyclic Chemistry*, Pergamon Press, New York, Vol 4, 1984; (b) A. R. Katritzky, C. W. Rees and E. F.V. Scriven, in *Comprehensive Heterocyclic Chemistry II*, Pergamon Press, New York, Vol 2, 1996; (c) A. R. Katritzky, C. A. Ramsden, E. F.V. Scriven, and R. J.K. Taylor *Comprehensive Heterocyclic Chemistry III*, Pergamon Press, New York, Vol 3, 2008. (d) G. W. Gribble and T. L. Gilchrist, in *Progress in Heterocyclic Chemistry*, Vol. 15, Pergamon Press, Oxford, 2003.
- (a) J. Roncali, *Chem. Rev.*, 1992, **92**, 711-738. (b) D. T. McQuade, A. E. Pullen and M. T. Swager, *Chem. Rev.* 2000, **100**, 2537-2574.
- J. B. Sperry, D. L. Wright, *Curr. Opin. Drug Disco. Devel.* 2005, **8**, 723-740.
- B. H. Lipshutz, *Chem. Rev.*, 1986, **86**, 795-819.
- (a) L. S. Hegedus, in *Transition Metal in the Synthesis of Complex Organic Molecules*, University Science Books: Mill Valley, CA, 1994; (b) B. M. Trost and D. L. Van Vranken, *Chem. Rev.*, 1996, **96**, 395-422.
- R. M. Roberts and A. A. Khalaf, in *Friedel-Crafts Alkylation Chemistry. A Century of Discovery*, Marcel Dekker, New York, 1984.
- (a) G. Onodera, H. Imajima, M. Yamanashi, Y. Nishibayashi, M. Hidai and S. Uemura, *Organometallics* 2004, **23**, 5841-5848; (b) J. Le Bras and J. Muzart, *Tetrahedron* 2007, **63**, 7942-7948; (c) G.-Q. Chen, Z.-J. Xu, S. L.-F. Chan, C.-Y. Zhou and C.-M. Che, *Synlett*, 2001, 2713-2718; (d) P. N. Chatterjee and S. Roy, *Tetrahedron*, 2012, **68**, 3776-3785; (e) D. Das, S. Pratihar, U. K. Roy, D. Mal and S. Roy, *Org. Biomol. Chem.*, 2012, **10**, 4537-4542.
- (a) H. Matsubashi, S. Asai, K. Hirabayashi, Y. Hatanaka, A. Mori and T. Hiyama, *Bull. Chem. Soc. Jpn.*, 1997, **70**, 1943-1945; (b) S. Zhang, D. Marshall and L. S. Liebeskind, *J. Org. Chem.* 1999, **64**, 2796-2804; (c) D. Bouyssi, V. Gerusz and G. Balme, *Eur. J. Org. Chem.* 2002, 2445-2448; (d) C. Bonini, L. Chiummiento, M. De Bonis, M. Funicello, P. Lupatelli and R. Pandolfo, *Tetrahedron: Asymmetry*, 2006, **17**, 2919-2924; (e) M. L. N. Rao, D. Banerjee and S. Giri, *Tetrahedron Lett.*, 2009, **50**, 5757-5761; (f) M. L. N. Rao, D. Banerjee and S. Giri, *J. Organomet. Chem.*, 2010, **695**, 1518-1525; (g) F.-Q. Yuan, F.-Y. Sun and F.-S. Han, *Tetrahedron*, 2012, **68**, 6837-6842.
- (a) M. Ishikura, M. Kamada, I. Oda, T. Ohta and M. Terashima, *J. Heterocyclic Chem.*, 1987, **24**, 377-386; (b) N. S. Nudelman and C. Carro, *Synlett*, 1999, 1942-1944.
- (a) *The Chemistry of Organolithium Compounds*, Eds: Z. Rappoport, I. Marek, in WILEY-VCH, 2004; (b) *Lithium Compounds in Organic Synthesis*, Eds. R. Luisi, V. Capriati, WILEY-VCH: Weinheim, 2014; (c) V. Snieckus, *Chem. Rev.*, 1990, **90**, 879-933.
- (a) M. Pérez, M. Fañanás-Mastral, P. H. Bos, A. Rudolph, S. R. Harutyunyan and B. L. Feringa, *Nature Chem.* 2011, **3**, 377-381; (b) M. Fañanás-Mastral, M. Pérez, P. H. Bos, A. Rudolph, S. R. Harutyunyan and B. L. Feringa, *Angew. Chem. Int. Ed.*, 2012, **51**, 1922-1925; (c) P. H. Bos, A. Rudolph, M. Pérez, M. Fañanás-Mastral, S. R. Harutyunyan and B. L. Feringa, *Chem. Commun.* 2012, **48**, 1748-1750; (d) M. Pérez, M. Fañanás-Mastral, V. Hornillos, A. Rudolph, P. H. Bos, S. R. Harutyunyan and B. L. Feringa, *Chem. Eur. J.* 2012, **18**, 11880-11883. (e) S. Guduguntla, M. Fañanás-Mastral, and B. L. Feringa, *J. Org. Chem.* 2013, **78**, 8274-8280.
- (a) M. Giannerini, M. Fañanás-Mastral and B. L. Feringa, *Nature Chem.* 2013, **5**, 667-672; (b) V. Hornillos, M. Giannerini, C. Vila, M. Fañanás-Mastral and B. L. Feringa, *Org. Lett.* 2013, **15**, 5114-5117; (c) M. Giannerini, V. Hornillos, C. Vila, M. Fañanás-Mastral and B. L. Feringa, *Angew. Chem. Int. Ed.*, 2013, **52**, 13329-13333; (d) C. Vila, M. Giannerini, V. Hornillos, M. Fañanás-Mastral and B. L. Feringa, *Chem. Sci.*, 2014, **5**, 1361-1367. (e) C. Vila, V. Hornillos, M. Giannerini, M. Fañanás-Mastral and B. L. Feringa, *Chem. Eur. J.*, 2014, DOI: 10.1002/chem.201404398.
- For seminal work: (a) S.-I. Murahashi, M. Yamamura, K.-I. Yanagisawa, N. Mita and K. Kondo, *J. Org. Chem.* 1979, **44**, 2408-2417. For a highlights, see: (b) V. Pace and R. Luisi, *ChemCatChem*, 2014, **6**, 1516-1519. (c) V. Capriati, F. M. Perna and A. Salomone, *Dalton Trans.* **2014**, DOI:10.1039/c4dt01004c.
- (a) E. S. M. Persson, M. van Klaveren, D. M. Grove, J.-E. Bäckvall and G. van Koten, *Chem. Eur. J.* 1995, **1**, 351-359; (b) A. S. E. Karlström and J.-E. Bäckvall, *Chem. Eur. J.* 2001, **7**, 19811-1989; (c) M. Yamanaka, S. Kato and E. Nakamura, *J. Am. Chem. Soc.* 2004, **126**, 6287-6293.
- Comparison of commercially available 2-thienyl organometallic reagents (source: Sigma-Aldrich (2014)): 2-thienyllithium solution (1M) = 690 €/mol; (2-thienyl)trimethylsilane = 1300 €/mol; 2-(tributylstannyl)thiophene = 1330 €/mol; 2-thienylboronic acid = 3350 €/mol; 2-thiopheneboronic acid MIDA ester = 5970€/mol.
- Different phosphine based ligands such PPh₃, 1,3-bis(diphenylphosphanyl)propane, (rac)-BINAP, (R,R)-Taniaphos, (R,S)-Josiphos were tested and in all cases highly regioselectivity to the α -substituted product was observed.
- No dehalogenation was observed in the course of the reaction for the (E)-1-bromo-4-(3-bromoprop-1-en-1-yl)benzene.
- See supporting information for further details.