# Organic & Biomolecular Chemistry

Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this Accepted Manuscript with the edited and formatted Advance Article as soon as it is available.

You can find more information about *Accepted Manuscripts* in the **Information for Authors**.

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard <u>Terms & Conditions</u> and the <u>Ethical guidelines</u> still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.



www.rsc.org/obc

### ARTICLE

Cite this: DOI: 10.1039/x0xx00000x

### Reaction of $\beta$ -Enaminones and Acetylene Dicarboxylates: Synthesis of Substituted 1,2-Dihydropyridinones<sup>†</sup>

Vemu Nagaraju,<sup>a</sup> Dalovai Purnachander,<sup>a,b</sup> N. S. V. M. Rao Mangina,<sup>a,b</sup> Surisetti Suresh,<sup>c</sup> Balasubramaniam Sridhar<sup>d</sup> and Galla V. Karunakar<sup>\*a,b</sup>

Received 00th January 2012, Accepted 00th January 2012

DOI: 10.1039/x0xx00000x

www.rsc.org/

Synthesis of substituted 1,2-dihydropyridinones is described in one pot reaction of  $\beta$ enaminones and acetylene dicarboxylates where new C-C and C-N bonds were formed. The title compounds were obtained in moderate to good yields.

#### Introduction

Among various heterocyclic molecules, nitrogen heterocyclic molecules have proven as potential compounds<sup>1</sup> in crop protection chemicals, functional materials and medicinal chemistry.<sup>2</sup> Different methods were developed for synthesis of nitrogen heterocyclic molecules<sup>3</sup> via metal catalysed<sup>4</sup> and organocatalysed reactions.<sup>5</sup>

Nitrogen heterocyclic molecules, in particular synthesis<sup>6</sup> of 2-pyridinones<sup>7</sup> have received much attention because of their potential applications in various fields (Figure 1).<sup>8</sup> For example amrinone and milrinone were used as cardiotonics.<sup>9</sup> Perampanel is identified as an important molecule for the treatment of Parkinson's disease.<sup>10</sup> 2-Pyridinone derivatives were showing properties like antihypertensive,<sup>11</sup> antitumor,<sup>12</sup> antibiotic,<sup>13</sup> antiviral,<sup>14</sup> antibacterial,<sup>15</sup> thrombin inhibition,<sup>16</sup> tissue factor VIIa inhibition,<sup>17</sup> human chymase inhibition<sup>18</sup> and human leukocyte elastase inhibition.<sup>19</sup> Some of the 2-pyridinone derivatives have been used as dyes.<sup>20</sup> Significant number of natural products are having 2-pyridinone core unit in their chemical structure.<sup>21</sup> Most of these molecules are exhibiting interesting biological and pharmacological properties.<sup>22</sup>

- <sup>b</sup>Academy of Scientific and Innovative Research, CSIR-Indian Institute of Chemical Technology, Hyderabad, 500007, India
- <sup>c</sup>Organic and Biomolecular Chemistry Division,
- CSIR-Indian Institute of Chemical Technology,

CSIR-Indian Institute of Chemical Technology,

†*Electronic Supplementary Information (ESI) available: CCDC* 1004429. For ESI and crystallographic data CIF or other electronic format see DOI:10.1039/c000000x/

Synthesis of nitrogen heterocyclic molecules by exploiting the chemical reactivity of  $\beta$ -enaminones is of our current interest.



Figure 1 Selected examples of important molecules containing 2pyridinone core skeletons and their applications.

#### **Results and discussion**

Very recently, we have developed two different synthetic methods by exploitation of substituted  $\beta$ -enaminones.<sup>24</sup> Azabicyclo[4.1.0]hepta-2,4-dienes were efficiently synthesized in a reaction of *N*-propargylic  $\beta$ -enaminones with acetylene dicarboxylates by a novel and exceptionally catalyst free conditions.<sup>25a</sup> Synthesis of 3-methylene-3,4-dihydro-2*H*-pyrrolines were achieved by reaction of *N*-propargylic  $\beta$ -enaminones with arynes via gold-catalysis.<sup>25b</sup> In continuation to our efforts towards the exploration of enaminone reactivity, we became interested to test the reactivity of  $\beta$ -enaminones towards activated alkynes.

Liang *et al.*, reported that the reaction of dialkyl acetylene dicarboxylates and  $\beta$ -enaminone derivatives in the presence of copper catalyst to give polysubstituted pyrroles (Figure-2 Scheme-a).<sup>26a</sup> It was reported that tandem reaction of primary

<sup>&</sup>lt;sup>a</sup>Crop Protection Chemicals Division,

CSIR-Indian Institute of Chemical Technology, Hyderabad, 500007, India. gallavk@iict.res.in

Hyderabad, 500007, India.

<sup>&</sup>lt;sup>d</sup>Center for X-ray Crystallography,

Hyderabad, 500007, India.

amines and acetylene esters gave 2-pyridones featuring carboxylates as substitutents.<sup>26b</sup>



Herein, we describe reactivity of substituted  $\beta$ -enaminones 1 on dialkyl acetylene dicarboxylates 2 in the presence of base to accesses 1,2-dihydropyridinones 3 (Figure 2, Scheme-b). This reaction offers the synthesis of 2-pyridinones with significant molecular complexity where new C-C and C-N bonds were formed without using transition metals.

In an initial experiment,  $\beta$ -enaminone **1a** (1 equiv.) reacted with diethyl acetylene dicarboxylate **2a** (1 equiv.) in the presence of potassium carbonate (1.5 equiv.) in acetonitrile solvent at 70 °C for 5 h, it was observed that the starting materials were fully consumed, very interestingly 22% yields of 1,2-dihydropyridinone **3a** was isolated (Scheme 1).



Scheme 1 Synthesis of 2-pyridinone 3a by reaction of 1a and 2a.

We have conducted experiments to improve the yield of this transformation by utilizing **1a** and **2a** in the presence of  $Cs_2CO_3$  in acetonitrile solvent at 90 °C for 4 h. The starting material **1a** was disappeared (monitored by TLC) but the yield of the product **3a** was not improved (24%). In an another experiment, toluene was used as a solvent the above reaction was performed at 110 °C, the starting material **1a** was disappeared (monitored by TLC) but the yield (19%) of the product **3a** was poor.

Based on these observations we have conducted one more experiment by taking  $\beta$ -enaminone **1a** (1 equiv.) and diethyl acetylene dicarboxylate **2a** (1 equiv.) in CH<sub>3</sub>CN solvent at 70 °C for 3 h. It was observed that the both the starting materials were consumed. However, the pyridone **3a** was not formed in this reaction. It was observed that the addition product **4** was formed in this reaction in 87% yield (Scheme 2). We have further conducted a reaction by taking **4** (1 equiv.) in the presence of K<sub>2</sub>CO<sub>3</sub> (1.5 equiv.) to check the formation of **3a**. Very interestingly, we have isolated 64% yield of pyridinone derivative **3a** as a product (Scheme 2).

Later we thought of conducting this experiment in one pot to obtain the desired pyridone **3a**. Accordingly, the substrates **1a** and **2a** were heated at 70 °C for 3 h followed by the addition of  $K_2CO_3$  at 30 °C and continued the reaction for 16 h. As expected the desired pyridone **3a** was formed in good yields (Table 1, entry 1).

![](_page_2_Figure_11.jpeg)

Scheme 2 Synthesis of 2-pyridinone (3a) via intermediate 4.

Experiments were conducted using  $\beta$ -enaminone **1a** (1 equiv.) with **2a** (1 equiv.) in the presence of different bases (1 equiv.) such as Et<sub>3</sub>N, pyridine, piperidine, *N*,*N*-diethylamine, and *N*,*N*-diisopropylethylamine in acetonitrile. These conditions did not yield product **3a** (Table 1, entries 2-6). When the reaction was performed by using **1a** with **2a** in the presence of *N*,*N*-diisopropylamine (1 equiv.) and DABCO in acetonitrile solvent for 24 hours the product **3a** was isolated in lower yields (Table 1, entry 7-8). Without using any base, this reaction did not yield product **3a** (Table 1, entry 9).

Table 1 Optimization of reaction conditions.

Ph		(a) CH <sub>3</sub> CN, 70 °C (b) CS <sub>2</sub> CO <sub>3</sub> , rt	<mark>2, 3h</mark>	Ph Ph N O
Entry	Base	Solvent	Time	(h) Yield(%)
1	K <sub>2</sub> CO <sub>3</sub>	CH <sub>3</sub> CN	16	61
2	Et <sub>3</sub> N	CH <sub>3</sub> CN	24	pno
3	Pyridine	CH <sub>3</sub> CN	42	pno
4	Piperidine	CH <sub>3</sub> CN	42	pno
5	Diethylamine	CH <sub>3</sub> CN	42	pno
6	N,N-Diisopropyl ethylamine	CH <sub>3</sub> CN	42	pno
7	<i>N</i> , <i>N</i> -Diisopropyl amine	CH <sub>3</sub> CN	24	23
8	DABCO	CH <sub>3</sub> CN	24	31
9	No base	CH <sub>3</sub> CN	42	pno
10	$Cs_2CO_3$	CH <sub>3</sub> CN	9	73
11	Na <sub>2</sub> CO <sub>3</sub>	CH <sub>3</sub> CN	16	31
12	NaOH	CH <sub>3</sub> CN	13	33
13	KOH	CH <sub>3</sub> CN	13	28
14	$Cs_2CO_3$	$H_2O$	24	cm
15	$Cs_2CO_3$	MeOH	16	27
16	$Cs_2CO_3$	DMF	16	21
17	$Cs_2CO_3$	THF	18	61
18	$Cs_2CO_3$	1,4-dioxane	16	62
19	$Cs_2CO_3$	CHCl <sub>3</sub>	16	66
20	Cs <sub>2</sub> CO <sub>3</sub>	Toluene	18	64

Reaction conditions: **1a** (0.273 mmol), **2a** (0.273 mmol), solvent (3 mL), All reactions initially conducted at 70  $^{\circ}$ C for 3 h then base was added at room temperature; Base (0.409 mmol); pno: product not observed. cm: complex mixture; Yields are for isolated products.

Further experiments were conducted by utilizing 1a and 2a in acetonitrile solvent in the presence of Cs<sub>2</sub>CO<sub>3</sub>, the product 3a yield (73%) was improved (Table 1, entry 10). Reaction of 1a with 2a by utilizing bases such as Na<sub>2</sub>CO<sub>3</sub>, NaOH and KOH in acetonitrile solvent, these conditions gave product 3a in lower yields (Table 1, entries 11-13). Having these results in hand, we

have further screened for solvent choice. Reaction of 1a and 2a in the presence of  $Cs_2CO_3$  different solvents were used to get better yields of product 3a. In the presence of water this reaction did not yield the desired product 3a (Table 1, entry 14). Two reactions were performed by using 1a and 2a in the presence of  $Cs_2CO_3$  in polar solvents like methanol and DMF, the product 3a was isolated in poor yields (Table 1, entries 15-16). We have next performed the reaction of 1a with 2a in the presence of  $Cs_2CO_3$  in THF, 1,4-dioxane, toluene and chloroform, good yields of the product 3a was observed (Table 1, entries 17-20).

Based on the best optimized reaction conditions (Table 1, entry 10), various substituted  $\beta$ -enaminones **1a-u** and substituted acetylenedicarboxylates 2a and 2b were employed. The results are summarized in Table 2. When substituted  $\beta$ enaminone 1b reacted with diethyl acetylenedicoaroxylate 2a gave **3b** in 76% yield (Table 2, entry 2). Substrate **1c** which is having electron donating group  $(4-OMe-C_6H_4)$  in the R<sup>3</sup> position reacted with 2a gave 63% yield of 3c (Table 2, entry 3). Electron withdrawing substrate like 4-F- $C_6H_4$  at  $R^3$  position 1d reacted with 2a gave 74% yield of 3d (Table 2, entry 4).  $\beta$ -Enaminone substrate that contain both electron donating (R<sup>2</sup>:4- $Me-C_6H_4$ ) group and withdrawing (R<sup>3</sup>:4-NO<sub>2</sub>-C<sub>6</sub>H<sub>4</sub>) group like 1e reacted with 2a to give 71% yield of 3e (Table 2, entry 5). In the case of 1f reaction with 2a, the corresponding pyridone 3f was isolated in 75% yield (Table 2, entry 6). The structure of the product 3f was further confirmed by single crystal X-ray analysis (Figure 3).

![](_page_3_Figure_6.jpeg)

Figure 3 ORTEP representation of 1,2-dihydropyridinones (3f: CCDC 1004429)

Substrates which are having electron withdrawing groups like 1g  $(R^3:4-F-C_6H_4)$  and 1h  $(R^3:4-NO_2-C_6H_4)$  reacted with 2a gave 70% and 65% yields of 3g and 3h, respectively (Table 2, entries 7-8). Cyclohexyl substituted  $\beta$ -enaminone 1i reaction with 2a gave 73% yield of product 3i (Table 2, entry 9).  $\beta$ -Enaminone 1j reacted with 2a gave 63% yield of 3j (Table 2, entry 10). Electron donating functional groups at  $R^2$  ( $R^2$ :4-Me-C<sub>6</sub>H<sub>4</sub>) and  $R^3$  ( $R^3$ :4-OMe-C<sub>6</sub>H<sub>4</sub>) positions derived  $\beta$ -enaminone like 1k reacted with 2a gave 64% yield of 3k (Table 2, entry 11). Reaction of 1l with 2a gave 72% yield of **31** (Table 2, entry 12). The substituted  $\beta$ -enaminone having *n*-propane at  $R^1$  position and withdrawing group at  $R^3$  ( $R^3$ :4-F-C<sub>6</sub>H<sub>4</sub>) position like 1m reacted with 2a gave 67% yield of product 3m (Table 2, entry 13).  $\beta$ -Enaminone derivative containing withdrawing groups at  $R^2$  ( $R^2$ :4-F-C<sub>6</sub>H<sub>4</sub>) and  $R^3$  ( $R^3$ :4-CF<sub>3</sub>-C<sub>6</sub>H<sub>4</sub>) positions like 1n reacted with 2a gave 75% yield of product 3n (Table 2, entry no. 14).

Table 2 Scope of the synthesis of 1,2-dihydropyridinones.

![](_page_3_Figure_10.jpeg)

(Table 2 Contd.)

#### (Table 2 Contd.)

![](_page_4_Figure_3.jpeg)

![](_page_4_Figure_4.jpeg)

Reaction conditions: **1a** (0.273 mmol), **2a** (0.273 mmol) CH<sub>3</sub>CN (3 mL); All reactions initially conducted at 70  $^{\circ}$ C for 3 h then base was added at room temperature; Cs<sub>2</sub>CO<sub>3</sub> (0.409 mmol); Yields are for isolated products.

 $\beta$ -Enaminone substituted at R<sup>2</sup> (R<sup>2</sup>:4-Me-C<sub>6</sub>H<sub>4</sub>) and R<sup>3</sup> (4-<sup>t</sup>Bu-C<sub>6</sub>H<sub>4</sub>) positions like **10** reaction with **2a** gave 70% yields of product **30** (Table 2, entry 15). In case of  $\beta$ -enaminone **1b** reaction with diterterybutyl acetylene dicarboxylate **2b** gave 74% yield of **3p** (Table 2, entry 16).  $\beta$ -Enaminone **1f** reacted with **2b** gave 72% yield of **3q** (Table 2, entry 17).  $\beta$ -Enaminone derivative containing *n*-alkyl group at R<sup>2</sup> (R<sup>2</sup>:C<sub>6</sub>H<sub>13</sub>) position like **1p** reaction with **2a** gave 73% yield of **3r** (Table 2, entry 18). The substrate  $\beta$ -Enaminone which is having substitutions

Journal Name

at R<sup>1</sup> position (tert-butyl(ethoxy)dimethylsilane), R<sup>2</sup> position  $(4-Me-C_6H_4)$  and R<sup>3</sup> position  $(4-CMe_3-C_6H_4)$  like 1q reacted with 2a and 2b gave highly functionalised 2-pyridinones 3s and 3t in 68% and 72% yields, respectively (Table 2, entries 19-20).  $\beta$ -Enaminone derivative 1r that contain tertbutyl(ethoxy)dimethylsilane at  $R^1$  position and electron withdrawing group at  $R^2$  ( $R^2$ :4-F-C<sub>6</sub>H<sub>4</sub>) position reacted with 2a and 2b gave 78% and 75% yields of products 3u and 3v, respectively (Table 2, entries 21-22).  $\beta$ -Enaminone 1s, containing electron withdrawing substitutions at R<sup>3</sup> (R<sup>3</sup>:2,3-di  $Cl-C_6H_3$ ) position, reacted with **2b** gave 69% yield of product **3w** (Table 2, entry 23).  $\beta$ -Enamine derived from 1,4-diketone like 1t reacted with 2a and 2b gave the products 3x and 3y in 65% and 68% yields, respectively (Table 2, entries 24-25).  $\beta$ -Enaminone derivative that contain acetate substitution at  $R^1$  ( $R^1$ =  $-CH_2$ -COOEt) position like 1u reacted with 2a gave product **3z** in 71% yield (Table 2, entry 26).

We have further tested the scope of this transformation using  $\beta$ enamino esters instead of  $\beta$ -enaminones.  $\beta$ -Enamino ester **5a** reacted with **2a** and **2b** gave products **6a** and **6b** in 63% and 65% yields, respectively (Table 3, entries 1-2).

![](_page_5_Figure_5.jpeg)

Reaction conditions: 5 (0.355 mmol), 2 (0.355 mmol) CH<sub>3</sub>CN (3 mL); All reactions initially conducted at 70  $^\circ\text{C}$  for 3 h then base was added at room temperature; Cs<sub>2</sub>CO<sub>3</sub> (0.533 mmol); Yields are for isolated products.

 $\beta$ -Enamino ester derivative having alkyl substitutions at R<sup>1</sup> and R<sup>2</sup> positions and like **5b** reacted with **2a** gave moderate yield of product **6c** (Table 3, entry 3). Reaction of **5c** with **2a** gave 47% yield of product **6d** (Table 3, entry 4). Enamines derived from acetylene diesters like **5d** and **5e** reacted with **2a** gave 45% and 48% yields of highly functionalised 2-pyridones **6e** and **6f**, respectively (Table 3, entries 5-6).  $\beta$ -Enaminone derivative having alkyl substitution at R<sup>3</sup> position and aromatic group at R<sup>2</sup> position like **5f** reacted with **2a** gave very poor yield of product **6g** (Scheme 3). This reaction clearly indicate that the  $\beta$ -enaminone derivative containing alkyl substitution at R<sup>3</sup> position leads to produce very poor yield of 2-pyridinone (**6g**).

![](_page_5_Figure_8.jpeg)

A possible reaction mechanism may be explained for the formation of 1,2-dihydropyridinones (Scheme 3). Initially, nucleophilic addition of  $\beta$ -enaminones 1 to the electrophile 2 would take place to give intermediate I. The enolate II of intermediate I would give intermediate III. Then the intermediate III undergo cyclisation to give pyridinones 3. We have isolated an analogue of this intermediate I, for example compound 4 which was further converted to pyridone 3a (Scheme 4).

![](_page_5_Figure_10.jpeg)

#### Conclusions

In conclusion, we have developed a straightforward and efficient method for synthesis of dihydropyridone derivatives having significant molecular complexity with good yields. Importantly, this transformation was achieved in one pot without using transition metals or catalysts. Current research is focused on further exploitation of the reactivity of substituted  $\beta$ -enaminone derivatives.

#### Experimental

#### **General information**

All the reactions were carried out in oven dried reaction flasks under nitrogen atmosphere and also solvents and reagents were transferred by oven-dried syringes to ambient temperature. TLC was performed on Merck silica gel aluminium sheets using UV as a visualizing agent and a 0.5% aqueous potassium permanganate solution and heat as developing agents. Solvents were removed under reduced pressure. Columns were packed as slurry of silica gel in hexane and ethyl acetate solvent mixture. The elution was assisted by applying pressure with an air pump. <sup>13</sup>C NMR spectra were recorded on 75 and 125 MHz spectrometers. <sup>1</sup>HNMR spectra were recorded on 300 and 500 MHz spectrometers in appropriate solvents using TMS as internal standard. The following abbreviations were used to explain multiplicities: s = singlet, d = doublet, dd = doublet doublet, t = triplet, m = multiplet. All reactions were performed under nitrogen atmosphere with freshly distilled and dried solvents and solvents were distilled using standard procedures. Unless otherwise noted, reagents were obtained from Aldrich, Alfa Aesar, and TCI used without further purification. Substituted  $\beta$ -enaminones (**1a-u**) were prepared by following the reported procedure.<sup>27</sup>

X-ray Crystallography: X-ray data for the compound was collected at room temperature using a Bruker Smart Apex CCD diffractometer with graphite monochromated MoK $\alpha$  radiation ( $\lambda$ =0.71073Å) with  $\omega\text{-scan}$  method.  $^{28}$  Preliminary lattice parameters and orientation matrices were obtained from four sets of frames. Unit cell dimensions were determined using 6371 reflections in the range of  $2.51 \square < \theta < 23.79^{\circ}$  for **3f**. Integration and scaling of intensity data were accomplished using SAINT program.<sup>28</sup> The structure was solved by direct methods using SHELXS97<sup>29</sup> and refinement was carried out by full-matrix least-squares technique using SHELXL97.<sup>29</sup> Anisotropic displacement parameters were included for all non-hydrogen atoms. H atoms were positioned geometrically and treated as riding on their parent C atoms [C-H = 0.93-0.97 Å and $U_{iso}(H) = 1.2U_{eq}(C)$  or  $1.5U_{eq}(C)$ ]. The crystal was found to be twinned and the exact twin matrix was identified by the integration program as -1.002 0 0.004, 0 -1 0, 0.801 0 1.002. The structure was refined using the hklf 5 routine with all reflections, resulting in a BASF value of 0.178 (2).

### General procedure for Synthesis of 1,2-dihydropyridinones (3a)

In a 25 mL round-bottomed two-neck flask compound enaminone 1a (0.1g, 0.273 mmol, 1 equiv.) was taken then dissolved in acetonitrile (2 mL) to this reaction mixture compound 2a (0.046g 0.273 mmol, 1 equiv.) was added and allowed to stir at 70 °C for 3 h under nitrogen atmosphere ( yellow colour reaction mass was observed in the reaction flask). This reaction mixture was allowed to room temperature. Progress of the reaction was monitored by TLC. Then Cs<sub>2</sub>CO<sub>3</sub> (0.133g, 0.409 mmol, 1.5 equiv.) was added portion wise at room temperature to this reaction mixture. Reaction mixture colour was changed from yellow to brown colour. This reaction mixture was allowed to stir at room temperature for 9 h. Progress of the reaction was monitored by TLC. After completion of the reaction, 3 mL of water was added to the reaction mixture. Reaction mass was extracted with ethyl acetate (3 x 5 mL). The combined organic layers were washed with aqueous brine, dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, and concentrated under vacuum. The crude residue was purified through a silica gel column using hexane and ethyl acetate as eluent (10/3) to give pure 1,2-dihydropyridine-4-carboxylates 3a. The similar procedure was followed for the synthesis of all 2-pyridinone derivatives (3a-z).

#### Ethyl 1-(2-(1*H*-indol-3-yl)ethyl)-5-benzoyl-2-oxo-6-phenyl-1,2dihydropyridine-4-carboxylate (3a)

 $R_{f}$ : 0.3; Hexane: Ethyl acetate mixture (10:3); Yield: 73%; light brown colour solid; Melting Point: 143-145 °C; <sup>1</sup>H NMR (500 MHz,

CDCl<sub>3</sub>):  $\delta$  7.96 (brs, 1H), 7.49-7.45 (m, 2H), 7.43-7.38 (m, 1H), 7.32-7.27 (m. 3H), 7.25-7.23 (brs, 2H), 7.16-7.10 (m, 3H), 6.95-6.86 (m, 4H), 6.84 (brs, 1H), 4.13-4.06 (m, 4H), 3.06 (t, *J* = 7.6 Hz, 2H), 1.08 (t, *J* = 7.0 Hz, 3H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  194.0,164.4, 162.0, 148.6, 140.2, 138.1, 136.0, 132.6, 131.5, 129.6, 129.3, 128.7, 128.1, 127.1, 122.3, 121.9, 121.4, 119.2, 118.4, 111.8, 111.0, 62.2, 47.3, 23.9, 13.5; HRMS (ESI): calcd. for C<sub>31</sub>H<sub>27</sub>N<sub>2</sub> O<sub>4</sub> [M+H]<sup>+</sup> 491.1965; found 491.1975.

#### Ethyl 1-(2-(1H-indol-3-yl)ethyl)-5-(2-naphthoyl)-2-oxo-6-phenyl-1,2-dihydropyridine-4-carboxylate (3b)

R<sub>f</sub>: 0.3; Hexane: Ethyl acetate mixture (10:3); Yield: 76%; orange colour solid; Melting Point: 190-194 °C; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): *δ* 8.04-7.95 (brs, 2H), 7.89-7.77 (m, 2H), 7.74-7.67 (m, 1H), 7.66-7.45 (m, 3H), 7.37-7.16 (m, 3H), 7.16-7.03 (m, 3H), 7.02-6.81 (m, 5H), 4.17-4.00 (m, 4H), 3.07 (t, *J* = 8.1 Hz, 2H), 1.03 (t, *J* = 7.1 Hz, 3H), <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): *δ* 193.8, 164.4, 162.0, 148.7, 140.2, 135.9, 135.5, 135.3, 132.1, 131.5, 130.7, 129.5, 129.4, 128.4, 128.2, 127.7, 127.1, 126.6, 124.1, 122.2, 121.9, 121.5, 119.3, 118.4, 111.8, 111.0, 62.0, 47.3, 24.0, 13.5; HRMS (ESI): calcd for  $C_{35}H_{29}N_2O_4$  [M+H]<sup>+</sup> 541.2121; found 541.2131.

### Ethyl 1-(2-(1H-indol-3-yl)ethyl)-5-(4-methoxybenzoyl)-2-oxo-6-phenyl-1,2-dihydropyridine-4-carboxylate (3c)

R<sub>f</sub>: 0.2; Hexane: Ethyl acetate mixture (10:3); Yield: 63%; pale orange colour semisolid; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 8.13 (brs, 1H), 7.46 (d, *J* = 8.6 Hz, 2H), 7.30-7.25 (m, 3H), 7.18-7.10 (m, 3H), 6.94-6.86 (m, 4H), 6.82 (d, *J* = 1.8 Hz, 1H), 6.74 (d, *J* = 8.8 Hz, 2H), 4.13-4.06 (m, 4H), 3.81 (s, 3H), 3.05 (t, *J* = 7.7 Hz, 2H), 1.09 (t, *J* = 7.1 Hz, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 192.5,164.4, 163.1, 162.0, 148.3, 140.0, 135.9, 131.6, 131.2, 131.1, 129.5, 129.3, 128.1, 127.1, 122.4, 121.8, 121.3, 119.2, 118.5, 118.4, 113.3, 111.8, 111.0, 62.1, 55.3, 47.3, 23.9, 13.5; HRMS (ESI): calcd. for  $C_{32}H_{29}N_2O_5$  [M+H]<sup>+</sup> 521.2071; found 521.2065.

#### Ethyl 1-(2-(1H-indol-3-yl)ethyl)-5-(4-fluorobenzoyl)-2-oxo-6phenyl-1,2-dihydropyridine-4-carboxylate (3d)

R<sub>f</sub>: 0.4; Hexane: Ethyl acetate mixture (10:3); Yield: 74%; orange colour semisolid; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>): δ 8.09 (brs,1H), 7.45 (q, *J* = 5.2 Hz, 2H), 7.28 (q, *J* = 8.3 Hz, 3H), 7.13 (q, *J* = 6.7 Hz, 3H), 6.98-6.79 (m, 7H), 4.19-4.04 (m, 4H), 3.06 (t, *J* = 8.3 Hz, 2H), 1.12 (t, *J* = 6.7 Hz, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 192.6, 166.2, 164.4, 161.9, 148.4, 140.0, 135.9, 134.6, 131.4, 131.3, 131.2, 129.5, 129.4, 128.1, 127.2, 122.4, 122.0, 121.6, 119.3, 118.4, 115.3, 115.1, 111.0, 62.2, 47.4, 23.8, 13.6; HRMS (ESI): calcd for C<sub>31</sub>H<sub>26</sub>FN<sub>2</sub>O<sub>4</sub> [M+H]<sup>+</sup> 509.1871; found 509.1885.

#### Ethyl 1-(2-(1H-indol-3-yl)ethyl)-5-(4-nitrobenzoyl)-2-oxo-6-ptolyl-1,2-dihydropyridine-4-carboxylate (3e)

R<sub>f</sub>: 0.3; Hexane: Ethyl acetate mixture (10:3); Yield: 71%; light brown colour solid; Melting Point: 150-154 °C; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 8.10-8.01 (m, 3H), 7.5 (d, J = 9.0 Hz, 2H), 7.36-7.29 (m, 2H), 7.16 (t, J = 7.5 Hz, 1H), 6.97-6.82 (m, 5H), 6.61 (d, J = 8.3 Hz, 2H), 4.22-4.05 (m, 4H), 3.07 (t, J = 7.5 Hz, 2H), 2.26 (s, 3H), 1.18 (t, J = 6.7 Hz, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 192.7, 164.3, 162.0, 149.6, 149.1, 142.7, 140.0, 139.8, 136.0, 129.4, 128.9, 128.0, 127.2, 123.2, 122.5, 121.9, 121.6, 119.2, 118.3, 117.5, 111.8, 111.1, 62.4, 47.3, 23.6, 21.1, 13.7; HRMS (ESI): calcd. for C<sub>32</sub>H<sub>27</sub>N<sub>2</sub>O<sub>6</sub>Na [M+Na]<sup>+</sup> 572.1792; found 572.1795.

#### Ethyl 5-benzoyl-2-oxo-1-phenethyl-6-phenyl-1,2dihydropyridine-4-carboxylate (3f)

R<sub>f</sub>: 0.3; Hexane: Ethyl acetate mixture (10:3); Yield: 75%; light orange colour solid; Melting Point: 140-144 °C; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>): δ 7.54 (d, J = 6.7 Hz, 2H), 7.47-7.38 (m, 1H), 7.34-7.23 (m, 4H), 7.23-7.12 (m, 5H), 6.91 (d, J = 7.5 Hz, 2H), 6.87-6.80 (m, 2H), 4.10 (q, J = 6.7 Hz, 2H), 3.98 (t, J = 7.5 Hz, 2H), 2.88 (t, J = 7.5 Hz, 2H), 1.08 (t, J = 6.7 Hz, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 193.8, 164.3, 161.7, 148.4, 140.1, 138.0, 137.6, 132.7, 131.3, 129.5, 129.4, 128.7, 128.4, 128.2, 128.1, 126.5, 121.4, 118.4, 62.2, 47.9, 34.1, 13.5; HRMS (ESI): calcd. for C<sub>29</sub>H<sub>25</sub>NO<sub>4</sub>Na [M+Na]<sup>+</sup> 474.1675; found; 474.1673.

Crystal data for 3f:  $C_{29}H_{25}NO_4$ , M = 451.50, colorless block, 0.21 x 0.17 x 0.09 mm<sup>3</sup>, monoclinic, space group  $P2_1/c$  (No. 14), a =8.1432(16), b = 15.506(3), c = 19.315(4) Å,  $\Box \beta = 99.992(3)^{\circ}$ , V =2401.8(8) Å<sup>3</sup>, Z = 4,  $D_c = 1.249$  g/cm<sup>3</sup>,  $F_{000} = 952$ , Bruker SMART APEX CCD area-detector, MoK $\alpha$  radiation,  $\Box \lambda = 0.71073$  Å, T =294(2)K,  $2\theta_{\text{max}} = 50.0^{\circ}$ , 21908 reflections collected, 21908 unique  $(R_{int} = 0.0000)$ . Final GooF = 1.051, R1 = 0.0604, wR2 = 0.1685, R indices based on 15095 reflections with I>2 $\sigma$ (I) (refinement on  $F^2$ ), 309 parameters, 0 restraints,  $\mu = 0.083 \text{ mm}^{-1}$ . CCDC 1004429 contains supplementary Crystallographic data for the structure. be obtained free These data can of charge at www.ccdc.cam.ac.uk/conts/retrieving.html [or from the Cambridge Crystallographic Data Centre (CCDC), 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44(0) 1223 336 033; email: deposit@ccdc.cam.ac.uk.

#### Ethyl 1-benzyl-5-(4-fluorobenzoyl)-2-oxo-6-phenyl-1,2dihydropyridine-4-carboxylate (3g)

 $R_{\rm f}$ : 0.3; Hexane: Ethyl acetate mixture (10:3); Yield: 70%; brown liquid; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 7.58-7.54 (m, 2H), 7.31 (s, 1H), 7.22-7.15 (m, 4H), 7.07 (t, *J* = 7.6 Hz, 2H), 6.95-6.90 (m, 2H), 6.85-6.80 (m, 4H), 5.11 (brs, 2H), 4.15 (q, *J* = 7.1 Hz, 2H), 1.13 (t, *J* = 7.1 Hz, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 192.4, 166.2, 164.3, 162.1, 148.6, 140.4, 136.0, 134.5, 131.3, 131.2, 131.1, 129.5, 128.3, 127.9, 127.4, 126.8, 121.8, 118.4, 115.3, 115.1, 62.3, 48.9, 13.6; HRMS (ESI): calcd for C<sub>28</sub>H<sub>22</sub>FNO<sub>4</sub> [M+H]<sup>+</sup> 456.1605; Found 456.1611.

#### Ethyl 1-benzyl-5-(4-nitrobenzoyl)-2-oxo-6-phenyl-1,2dihydropyridine-4-carboxylate (3h)

R<sub>f</sub>: 0.4; Hexane: Ethyl acetate mixture (10:3); Yield: 65%; brown colour liquid; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>): δ 8.08 (d, *J* = 9.0 Hz, 2H), 7.65 (d, *J* = 9.0 Hz, 2H), 7.33 (s, 1H), 7.24-7.14 (m, 4H), 7.06 (t, *J* = 7.5 Hz, 2H), 6.84-6.77 (m, 4H), 5.11 (s, 2H), 4.19 (q, *J* = 7.5 Hz, 2H), 1.18 (t, *J* = 7.5 Hz, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 192.6, 164.2, 162.1, 149.6, 148.9, 142.6, 140.2, 135.8, 130.7, 129.9, 129.6, 129.3, 128.4, 128.1, 127.5, 126.8, 123.2, 122.0, 117.8, 62.5, 48.9, 13.7; HRMS (ESI): calcd. for C<sub>28</sub>H<sub>22</sub>N<sub>2</sub>O<sub>6</sub>Na [M+Na]<sup>+</sup> 505.1370; found 505.1375.

#### Ethyl 1-cyclohexyl-5-(4-fluorobenzoyl)-2-oxo-6-phenyl-1,2dihydropyridine-4-carboxylate (3i)

R<sub>f</sub>: 0.4; Hexane: Ethyl acetate mixture (10:3); Yield: 73%; yellow colour solid; Melting Point: 170-174 °C; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 7.70-7.65 (m, 2H), 7.27 (m, 3H), 7.21-7.17 (m, 2H), 6.93-6.88 (m, 2H), 6.7 (s, 1H), 3.91 (q, J = 7.1 Hz, 2H), 3.31 (m, 1H), 2.36-2.18 (m, 2H), 1.80-1.48 (m, 8H), 1.02 (t, J = 7.1 Hz, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 189.5, 168.1, 166.1, 165.1, 156.3, 137.3,

134.7, 131.4, 131.3, 130.3, 128.6, 128.5, 121.2, 115.2, 115.0, 60.9, 55.2, 29.7, 25.9, 24.8, 13.7; HRMS (ESI): calcd. for  $C_{27}H_{27}FNO_4$   $[M+H]^+$  448.1918; found 448.1927.

#### Ethyl 5-(1-naphthoyl)-1-butyl-2-oxo-6-phenyl-1,2dihydropyridine-4-carboxylate (3j)

R<sub>f</sub>: 0.4; Hexane: Ethyl acetate mixture (10:3); Yield: 63%; orange colour solid; Melting Point: 98-101 °C; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>); δ 8.42-8.34 (m, 1H), 7.83 (d, *J* = 8.3 Hz, 1H), 7.78-7.68 (m, 1H), 7.63 (d, *J* = 6.7 Hz, 1H), 7.48-7.39 (m, 2H), 7.33 (t, *J* = 7.5 Hz, 1H), 7.10 (s, 1H), 7.03-6.87 (m, 5H), 4.07 (q, *J* = 7.5 Hz, 2H), 3.72 (t, *J* = 7.5 Hz, 2H), 1.56-1.41 (m, 2H), 1.40-1.13 (m, 2H), 1.10-1.00 (m, 3H), 0.66 (t, *J* = 7.5 Hz, 3H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>): δ 195.3,165.3, 161.7,149.6, 141.7, 135.8, 133.3, 132.9, 131.5, 130.2, 129.0, 128.7, 127.9, 127.8, 127.4, 126.1,125.5, 123.7, 120.5, 62.1, 45.9, 30.3, 19.7, 13.5, 13.1; HRMS (ESI): calcd. for C<sub>29</sub>H<sub>28</sub>NO<sub>4</sub> [M+H]<sup>+</sup> 454.2012; found 454.2015.

#### Tert-butyl 1-butyl-5-(4-methoxybenzoyl)-2-oxo-6-p-tolyl-1,2dihydropyridine-4-carboxylate (3k)

R<sub>f</sub>: 0.2; Hexane: Ethyl acetate mixture (10:3); Yield: 64%; pale orange semisolid; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 7.71 (d, J = 8.6 Hz, 2H), 7.41 (s, 1H), 7.21-7.10 (m, 4H), 6.94 (d, J = 8.6 Hz, 2H), 3.96 (s, 3H), 3.89 (t, J = 7.9 Hz, 2H), 2.42 (s, 3H), 1.73-1.60 (m, 2H), 1.36 (s, 9H), 1.25 (q, J = 7.3 Hz, 2H), 0.86 (t, J = 7.1 Hz, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): 192.2, 163.5, 163.2, 162.0, 148.4, 141.5, 139.3, 131.2, 129.1, 128.7, 126.0, 120.6, 118.3, 113.3, 83.5, 55.3, 46.0, 30.4, 27.3, 21.2, 19.8, 13.3; HRMS (ESI): calcd for C<sub>29</sub>H<sub>34</sub>NO<sub>5</sub> [M+H]<sup>+</sup> 476.2431; found 476.2433.

#### Ethyl 1-butyl-5-(4-nitrobenzoyl)-2-oxo-6-phenyl-1,2dihydropyridine-4-carboxylate (31)

R<sub>f</sub>: 0.3; Hexane: Ethyl acetate mixture (10:3); Yield: 72%; brown colour liquid; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>): δ 8.12 (d, *J* = 9.0 Hz, 2 H), 7.68 (d, *J* = 9.0 Hz, 2H), 7.33-7.18 (m, 4H), 7.07 (d, *J* = 6.7 Hz, 2H), 4.17 (q, *J* = 6.7 Hz, 2H) 3.82-3.72 (m, 2H), 1.58-1.45 (m, 2H), 1.21-1.03 (m, 5H), 0.70 (t, *J* = 7.5 Hz, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 192.8, 164.3, 161.7, 149.6, 148.6, 142.7, 139.7, 131.1, 129.9, 129.5, 129.4, 128.3, 123.2, 121.7, 117.5, 62.4, 46.1, 30.3, 19.8, 13.7, 13.2; HRMS (ESI): calcd. for C<sub>25</sub>H<sub>25</sub>N<sub>2</sub>O<sub>6</sub> [M+H]<sup>+</sup> 449.1707; found 449.1713.

#### Ethyl 5-(4-fluorobenzoyl)-2-oxo-6-phenyl-1-propyl-1,2dihydropyridine-4-carboxylate (3m)

R<sub>f</sub>: 0.2; Hexane: Ethyl acetate mixture (10:3); Yield: 67%; brown colour semi solid; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 7.59-7.54 (m, 2H), 7.27 (m, 1H), 7.24-7.20 (m, 3H), 7.09-7.06 (m, 2H), 6.94 (t, *J* = 8.5 Hz, 2H), 4.12 (q, *J* = 7.1 Hz, 2H), 3.71 (t, *J* = 7.7 Hz, 2H), 1.6-1.5 (m, 2H), 1.11 (t, *J* = 7.1 Hz, 3H), 0.69 (t, *J* = 7.32 Hz, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 192.6, 166.3, 161.8, 148.3, 139.8, 134.6,131.4, 131.3, 131.2, 129.6, 129.4, 128.2, 121.5, 118.1, 115.3, 115.2, 62.2, 47.8, 21.8, 13.6, 11.0,; HRMS (ESI): calcd. for C<sub>24</sub>H<sub>23</sub>FNO<sub>4</sub> [M+H]<sup>+</sup> 408.1605; found 408.1603.

#### Ethyl 6-(4-fluorophenyl)-2-oxo-1-propyl-5-(4-(trifluoromethyl)benzoyl)-1,2-dihydropyridine-4-carboxylate (3n)

R<sub>f</sub>: 0.3; Hexane: Ethyl acetate mixture (10:3); Yield: 75%; pale yellow semisolid; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>):  $\delta$  7.67 (d, *J* = 8.2 Hz, 2H), 7.58 (d, *J* = 8.2 Hz, 2H), 7.29 (s, 1H), 7.11-7.06 (m, 2H),

6.94 (t, J = 8.5 Hz, 2H), 4.14 (q, J = 7.0 Hz, 2H), 3.74-3.69 (m, 2H), 1.61-1.52 (m, 2H), 1.14 (t, J = 7.1 Hz, 3H), 0.73 (t, J = 7.4 Hz, 3H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  193.0, 164.6, 164.1, 161.7, 147.4, 140.7, 139.6, 131.5, 131.4, 128.7, 125.3, 125.2, 121.9, 115.7, 115.4, 62.3, 47.8, 21.8, 13.5, 11.0; HRMS (ESI): calcd. for C<sub>25</sub>H<sub>21</sub>F<sub>4</sub>NO<sub>4</sub> [M+H]<sup>+</sup> 476.1479; found 476.1466.

#### Ethyl 5-(4-tert-butylbenzoyl)-1-methyl-2-oxo-6-p-tolyl-1,2dihydropyridine-4-carboxylate (30)

 $R_{\rm f}$ : 0.3; Hexane: Ethyl acetate mixture (10:3); Yield: 70%; yellow colour semisolid; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 7.50 (d, *J* = 8.5 Hz, 2H), 7.30 (d, *J* = 7.3 Hz, 2H),7.19 (s, 1H), 7.04 (d, *J* = 7.7 Hz, 2H), 6.98 (d, *J* = 7.9 Hz, 2H), 4.07 (q, *J* = 7.1 Hz, 2H), 3.29 (s, 3H), 2.26 (s, 3H), 1.28 (s, 9H), 1.04 (t, *J* = 7.1 Hz, 3H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>): δ 193.3, 164.4, 162.3, 156.3, 149.0, 140.3, 139.5, 135.4, 129.1, 129.0, 128.7, 128.6, 124.9, 120.2,118.4, 62.0, 34.8, 34.1, 30.8, 21.1, 13.3; HRMS (ESI): calcd. for C<sub>27</sub>H<sub>30</sub>NO<sub>4</sub> [M+H]<sup>+</sup> 432.2169; found 432.2153.

#### Tert-butyl 1-(2-(1H-indol-3-yl)ethyl)-5-(2-naphthoyl)-2-oxo-6phenyl-1,2-dihydropyridine-4-carboxylate (3p)

R<sub>f</sub>: 0.3; Hexane: Ethyl acetate mixture (10:3); Yield: 74%; pale orange colour solid; Melting Point: 185-188 °C; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 8.04-7.95 (m, 2H), 7.89-7.77 (m, 2H), 7.72 (d, J = 8.6 Hz, 1H), 7.67-7.47 (m, 3H), 7.31-7.22 (m, 2H), 7.19 (d, J = 7.5 Hz, 1H), 7.15-7.04 (m, 3H), 7.00-6.93 (m, 2H), 6.92-6.82 (m, 3H), 4.11 (t, J = 7.5 Hz, 2H), 3.06 (t, J = 8.1 Hz, 2H), 1.16 (s, 9H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 193.6, 163.6, 162.1, 148.7, 142.0, 136.0, 135.6, 135.2, 132.1, 131.7, 130.8, 129.5, 129.3, 128.4, 128.1, 127.6, 127.1, 123.6, 124.2, 122.3, 121.9, 121.0, 119.2, 118.4, 118.3, 111.8, 111.0, 83.8, 47.2, 27.3, 24.0; HRMS (ESI): calcd. for C<sub>37</sub>H<sub>33</sub>N<sub>2</sub>O<sub>4</sub> [M+H]<sup>+</sup> 569.2434; found; 569.2435.

## Tert-butyl5-benzoyl-2-oxo-1-phenethyl-6-phenyl-1,2-dihydropyridine-4-carboxylate (3q)

R<sub>f</sub>: 0.3; Hexane: Ethyl acetate mixture (10:3); Yield: 72%; pale yellow colour solid; Melting Point: 167-170 °C; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>): δ 7.59-7.53 (m, 2H), 7.48-7.39 (m, 1H), 7.35-7.24 (m, 3H), 7.24-7.10 (m, 6H), 6.92 (d, *J* = 6.7 Hz, 2H), 6.88-6.81 (m, 2H), 3.97 (t, *J* = 7.5 Hz, 2H), 2.86 (t, *J* = 7.5 Hz, 2H), 1.22 (s, 9H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 193.5, 163.5, 161.9, 148.4, 141.9, 138.3, 137.7, 132.7, 131.6, 129.4, 128.9, 128.7, 128.4, 128.1, 126.5, 121.0, 118.2, 83.8, 47.8, 34.2, 27.3; HRMS (ESI): calcd. for C<sub>31</sub>H<sub>30</sub>NO<sub>4</sub> [M+H]<sup>+</sup> 480.2169; found; 480.2171.

#### Ethyl 1-(2-(1H-indol-3-yl)ethyl)-5-(2-naphthoyl)-6-hexyl-2-oxo-1,2-dihydropyridine-4-carboxylate (3r)

R<sub>f</sub>: 0.2; Hexane: Ethyl acetate mixture (10:3); Yield: 73%; red colour solid; Melting Point: 140-144 °C; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 8.17-8.07, (m, 2H), 7.93-7.82 (m, 4H), 7.64 (d, *J* = 7.7 Hz, 1H), 7.61-7.49 (m, 2H), 7.38 (d, *J* = 7.5 Hz, 1H), 7.25-7.12 (m, 2H), 7.06 (d, *J* = 2.2 Hz, 1H), 6.64 (s, 1H), 3.9 (t, *J* = 6.9 Hz, 2H), 3.56 (q, *J* = 7.1 Hz, 2H), 3.17 (t, *J* = 7.7 Hz, 2H), 2.25-2.15 (m, 2H), 1.47-1.35 (m, 2H), 1.18-0.97 (m, 6H), 0.82-0.69 (m, 6H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 190.7, 168.5, 165.2, 160.3, 137.4, 136.4, 136.2, 135.3, 132.3, 129.9, 129.4, 128.3, 128.2, 127.6, 127.1, 126.6, 124.5, 122.3, 120.2, 119.7, 118.3, 112.1, 111.5, 111.4, 60.7, 41.7, 31.0, 29.1, 28.5, 25.8, 24.9, 22.3, 13.8, 13.4; HRMS (ESI): calcd. for  $C_{35}H_{37}N_2 O_4 [M+H]^+$  549.2747; found 549.2753.

#### Ethyl 5-(4-tert-butylbenzoyl)-1-(2-(tertbutyldimethylsilyloxy)ethyl)-2-oxo-6-p-tolyl-1,2dihydropyridine-4-carboxylate (3s)

 $R_{\rm f}$ : 0.4; Hexane: Ethyl acetate mixture (10:3); Yield: 68%; pale brown colour semi solid; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 7.55 (d, *J* = 8.39 Hz, 2H), 7.37 (t, *J* = 3.20 Hz, 3H), 7.08 (q, *J* = 7.93 Hz, 4H), 4.18 (q, *J* = 7.01 Hz, 2H), 4.06 (t, *J* = 6.25 Hz, 2H), 3.89 (t, *J* = 6.25 Hz, 2H), 2.34 (s, 3H), 1.38 (s, 12H), 0.92 (s, 9H), 0.06 (s, 6H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 193.6, 164.5, 162.0, 156.2, 149.5, 139.3, 135.7, 129.8, 128.74, 128.7, 128.6, 128.4, 124.9, 120.7, 118.7, 62.0, 59.4, 47.9, 34.9, 30.9, 25.8, 25.7, 21.1, 13.4, -5.5; HRMS (ESI): calcd. for C<sub>34</sub>H<sub>46</sub>O<sub>5</sub>NSi [M+H]<sup>+</sup> 576.3141; found 576.3141.

#### Tert-butyl 5-(4-tert-butylbenzoyl)-1-(2-(tertbutyldimethylsilyloxy)ethyl)-2-0x0-6-p-tolyl-1,2dihydropyridine-4-Carboxylate (3t)

R<sub>f</sub>: 0.4; Hexane: Ethyl acetate mixture (10:3); Yield: 72%; pale brown colour semi solid; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 7.65 (d, J = 8.24 Hz, 2H), 7.49-7.44 (m, 2H), 7.24 (s, 1H), 7.20-7.13 (m, 4H), 4.11 (t, J = 6.25 Hz, 2H), 3.93 (t, J = 6.25 Hz, 2H), 2.41 (s, 3H), 1.44 (s, 9H), 1.34 (s, 9H), 0.98 (s, 9H), 0.12 (s, 6H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 193.0, 163.6, 162.0, 156.3, 149.4, 142.1, 139.2, 135.7, 129.5, 128.8, 128.7, 128.5, 124.9, 120.1, 118.3, 83.4, 59.4, 47.7, 34.8, 30.8, 27.1, 25.7, 21.0, 18.1, -5.6; HRMS (ESI): calcd for C<sub>36</sub>H<sub>50</sub>O<sub>5</sub>NSi [M+H]<sup>+</sup> 604.3452; found 604.3455.

#### Ethyl 1-(2-(tert-butyldimethylsilyloxy)ethyl)-5-(4-fluorobenzoyl)-2-oxo-6-phenyl-1,2-dihydropyridine-4-carboxylate (3u)

R<sub>f</sub>: 0.4; Hexane: Ethyl acetate mixture (10:3); Yield: 78%; pale brown colour semi solid; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 7.63-7.58 (m, 2H), 7.31-7.28 (m, 1H), 7.27-7.22 (m, 3H), 7.19-7.15 (m, 2H), 6.97 (t, J = 8.54 Hz, 2H), 4.19 (q, J = 7.17 Hz, 2H), 4.00 (t, J = 6.2 Hz, 2H), 3.86 (t, J = 6.2 Hz, 2H), 1.17 (t, J = 7.1 Hz, 3H), 0.87 (s, 9H), 0.01 (s, 6H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 192.7, 166.2, 164.4, 161.9, 149.0, 140.2, 134.6, 131.2, 130.0, 129.5, 128.0, 121.3, 118.2, 115.2, 115.1, 62.2, 59.3, 48.1, 29.6, 25.8, 13.5, -5.4. HRMS (ESI): calcd for C<sub>29</sub>H<sub>35</sub>O<sub>5</sub>NFSi [M+H]<sup>+</sup> 524.2260; found 524.2260.

#### Tert-butyl 1-(2-(tert-butyldimethylsilyloxy)ethyl)-5-(4fluorobenzoyl)-2-oxo-6-phenyl-1,2-dihydropyridine-4carboxylate (3v)

R<sub>f</sub>: 0.4; Hexane: Ethyl acetate mixture(10:3); Yield: 75%; pale brown colour semi solid; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 7.57-7.51 (m, 2H), 7.25-7.20 (m,1H), 7.18 (t, J = 7.93 Hz, 2H), 7.12 (s, 1H), 7.08 (d, J = 8.24 Hz, 2H), 6.92 (t, J = 8.24 Hz, 2H), 3.92 (t, J = 6.1Hz, 2H), 3.77 (t, J = 6.1 Hz, 2H), 1.24 (s, 9H), 0.80 (s, 9H), 0.06 (s, 6H). <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 192.3, 166.2, 163.4, 161.9, 148.9, 141.9, 134.7, 131.5, 131.3, 129.9, 129.4, 128.0, 120.8, 118.0, 115.2, 83.7, 59.4, 47.9, 27.3, 25.8, 18.2, -5.4. HRMS (ESI): calcd for C<sub>31</sub>H<sub>39</sub>O<sub>5</sub>NFSi [M+H]<sup>+</sup> 552.2548; found 552.2550.

#### Tert-butyl 1-(2-(tert-butyldimethylsilyloxy)ethyl)-5-(2,3dichlorobenzoyl)-2-oxo-6-phenyl-1,2-dihydropyridine-4carboxylate (3w)

R<sub>f</sub>: 0.4; Hexane: Ethyl acetate mixture (10:3); Yield: 69%; pale brown colour semi solid; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 7.34-7.31 (m, 1H), 7.22-7.16 (m, 2H), 7.15-7.10 (m, 2H), 7.06 (d, J = 7.17 Hz, 2H), 7.01-6.96 (m, 2H), 3.85 (t, J = 6.25 Hz, 2H), 3.75 (t, J = 6.25Hz, 2H), 1.43 (s, 9H), 0.79 (s, 9H), 0.06 (s, 6H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 191.6, 164.4, 162.0, 149.8, 143.7, 139.5, 133.9,

132.4, 131.3, 130.8, 129.5, 129.2, 128.2, 126.4, 120.4, 83.6, 59.4, 47.8, 31.8, 27.7, 25.8, 14.0, -5.4. HRMS (ESI): calcd for  $C_{31}H_{38}O_5NCl_2Si$  [M+H]<sup>+</sup> 602.1914; found 602.1890.

#### Ethyl 5,6-dibenzoyl-1-benzyl-2-oxo-1,2-dihydropyridine-4carboxylate (3x)

**R**<sub>f</sub>: 0.4; Hexane: Ethyl acetate mixture (10:3); Yield: 65%; pale brown colour semi solid; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>):  $\delta$  7.58-7.37 (m, 6H), 7.32 (t, *J* = 7.5 Hz, 2H), 7.24-7.14 (m, 3H), 7.07 (s, 5H), 5.23 (brs, 2H), 3.91 (q, *J* = 7.1 Hz, 2H), 1.00 (t, *J* = 7.1 Hz, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>):  $\delta$  191.9, 189.3, 164.2, 161.3, 147.6, 141.5, 137.3, 135.2, 134.8, 134.4, 133.2, 129.4, 128.8, 128.4, 128.3, 128.1, 127.8, 122.3, 116.7, 62.4, 48.6, 13.4; HRMS (ESI): calcd for C<sub>29</sub>H<sub>24</sub>O<sub>5</sub>N [M+H]<sup>+</sup> 466.1624; found 466.1624.

### Tert-butyl 5,6-dibenzoyl-1-benzyl-2-oxo-1,2-dihydropyridine-4-carboxylate (3y)

 $R_{f:}$  0.4; Hexane: Ethyl acetate mixture (10:3); Yield: 68%; pale brown colour semi solid; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 7.54-7.45 (m, 2H), 7.44-7.31 (m, 3H), 7.31-7.21 (m,3 H), 7.15-7.05 (m, 3H), 7.00 (s, 5H), 5.15 (brs, 2H), 1.08 (s, 9H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 191.7, 189.4, 163.2, 161.4, 147.3, 142.9, 137.4, 135.2, 134.9, 134.4, 133.2, 129.4, 128.9, 128.4, 128.3, 128.1, 127.7, 122.1, 116.7, 84.1, 48.5, 27.1; HRMS (ESI): calcd for C<sub>31</sub>H<sub>28</sub>O<sub>5</sub>N [M+H]<sup>+</sup> 494.1934; found 494.1934.

#### Ethyl 1-(2-ethoxy-2-oxoethyl)-6-(4-fluorophenyl)-2-oxo-5-(4-(trifluoromethyl)benzoyl)-1,2-dihydropyridine-4-carboxylate (3z)

R<sub>f</sub>: 0.4; Hexane: Ethyl acetate mixture (10:3); Yield: 71%; pale brown colour semi solid; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 7.71 (d, J = 7.9 Hz, 2H), 7.58 (d, J = 8.0 Hz, 2H), 7.30 (s, 1H), 7.15-7.08 (m, 2H), 6.92 (t, J = 8.2 Hz, 2H), 4.46 (s, 2H), 4.22-4.13 (m, 4H), 1.23 (t, J = 7.0 Hz, 3H), 1.14 (t, J = 7.0 Hz, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ, 192.6, 167.3, 163.9, 161.5, 161.4, 147.2, 140.4, 131.5, 131.4, 128.8, 126.7, 125.3, 121.9, 118.1, 116.0, 115.8, 62.5, 61.9, 47.4, 13.9, 13.5; HRMS (ESI): calcd for C<sub>26</sub>H<sub>22</sub>O<sub>6</sub>NF<sub>4</sub> [M+H]<sup>+</sup> 520.1405; found 520.1377.

#### General procedure for Synthesis of intermediate (4)

In a 25 mL round-bottomed two-neck flask compound **1a** (0.1g, 0.273 mmol, 1equiv.) was taken then dissolved in acetonitrile (2 mL) after that compound **2a** (0.046gm 0.273 mmol, 1 equiv.) was added and allowed to stir the reaction mixture at 70 °C for 3 h. Progress of the reaction was monitored by TLC. After completion of the reaction, acetonitrile solvent was removed in vacuum. The crude residue was purified through a silica gel column using hexane and ethyl acetate as eluent (10/2) to give (87%) pure addition product **4**.

#### Diethyl 2-(1-(2-(1*H*-indol-3-yl)ethylamino)-3-oxo-1,3diphenylprop-1-en-2-yl)maleate (4)

 $R_{\rm f}$ : 0.3; Hexane: Ethyl acetate mixture(10:1); Yield: 82%; brown colour semisolid; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 12.26 (brs,1H), 8.15 (brs, 1H), 7.44-7.34 (m, 2H), 7.33-7.10 (m, 10H), 7.08-6.93 (m, 3H), 6.12 (s, 1H), 4.03 (q, *J* = 6.9 Hz, 2H), 3.83 (q, *J* = 6.9 Hz, 2H), 3.36 (q, *J* = 6.9 Hz, 2H), 3.01 (t, *J* = 6.9 Hz, 2H), 1.16 (t, *J* = 7.1 Hz, 3H), 1.05 (t, *J* = 7.1 Hz, 3H); <sup>13</sup>C NMR (300 MHz, CDCl<sub>3</sub>): δ 192.5, 166.1, 165.3, 145.2, 142.2, 136.2, 133.0, 129.0, 128.9, 128.6, 128.3, 127.9, 127.4, 127.1, 121.7, 119.1, 118.1, 111.6, 111.2, 101.7, 61.2,

This journal is © The Royal Society of Chemistry 2012

60.2, 45.3, 26.5, 14.0, 13.8; HRMS (ESI): calcd. for  $C_{33}H_{33}$   $O_5N_2$   $\left[M\!+\!H\right]^+$  537.2378; found 537.2384.

### General procedure for Synthesis of 1,2-dihydropyridinones (6a)

In a 25 mL round-bottomed two-neck flask compound carboxylate substituted enaminone 5a (0.1g, 0.355 mmol, 1 equiv.) was taken then dissolved in acetonitrile (2 mL) to this reaction mixture compound **2a** (0.06g 0.355 mmol, 1 equiv.) was added and allowed to stir at 70 °C for 3 h ( yellow colour reaction mass was observed in the reaction flask). This reaction mixture was allowed to room temperature. Progress of the reaction was monitored by TLC. Then Cs<sub>2</sub>CO<sub>3</sub> (0.173g, 0.533 mmol, 1.5 equiv.) was added portion wise at room temperature to this reaction mixture. Reaction mixture colour was changed from yellow to brown colour. This reaction mixture was allowed to stir at room temperature for 9 h. Progress of the reaction was monitored by TLC. After completion of the reaction, 3 mL of water was added to the reaction mixture. Reaction mass was extracted with ethyl acetate (3 x 5 mL). The combined organic layers were washed with aqueous brine, dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, and concentrated under vacuum. The crude residue was purified through a silica gel column using hexane and ethyl acetate as eluent (10/3) to give pure 1,2dihydropyridine-4-carboxylates 6a. The similar procedure was followed for the synthesis of all 2-pyridinone derivatives (6a-f).

#### Diethyl 1-benzyl-6-oxo-2-phenyl-1,6-dihydropyridine-3,4dicarboxylate (6a)

 $R_{f}$ : 0.4; Hexane: Ethyl acetate mixture (10:3); Yield: 63%; semi solid; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 7.41 (t, *J* = 7.3 Hz, 1H), 7.30 (t, *J* = 7.7 Hz, 2H), 7.22-7.13 (m, 4H), 7.05 (d, *J* = 7.1 Hz, 2H), 6.84-6.76 (m, 2H), 5.08 (brs, 2H), 4.35 (q, *J* = 7.1 Hz, 2H), 3.86 (q, *J* = 7.1 Hz, 2H), 1.35 (t, *J* = 7.1 Hz, 3H); 0.85 (t, *J* = 7.17 Hz, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 165.6, 164.6, 162.0, 149.9, 140.2, 136.0, 132.0, 129.6, 129.0, 128.3, 128.0, 127.3, 126.7, 120.9, 113.0, 62.2, 61.2, 49.0, 13.9, 13.3; HRMS (ESI): calcd for C<sub>24</sub>H<sub>24</sub>O<sub>5</sub>N [M+H]<sup>+</sup> 406.1652; found 406.1653.

#### 4-Tert-butyl 3-ethyl 1-benzyl-6-oxo-2-phenyl-1,6dihydropyridine-3,4-dicarboxylate (6b)

R<sub>f</sub>: 0.4; Hexane: Ethyl acetate mixture (10:3); Yield: 65%; semi solid; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>): δ 7.40 (t, *J* = 7.4 Hz, 1H), 7.30 (t, *J* = 7.7 Hz, 2H), 7.19-7.15 (m 3H), 7.10 (s, 1H), 7.05 (d, *J* = 7.1 Hz, 2H), 6.82-6.78 (m, 2H), 5.07 (brs, 2H), 3.86 (q, *J* = 7.1 Hz, 2H), 1.55 (s, 9H), 0.84 (t, *J* = 7.1 Hz, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 165.8, 163.6, 162.1, 149.5, 141.5, 136.0, 132.0, 129.6, 129.0, 128.3, 128.0, 127.3, 126.7, 120.7, 113.3, 83.4, 61.2, 48.9, 27.7, 13.3; HRMS (ESI): calcd for C<sub>26</sub>H<sub>28</sub>O<sub>5</sub>N [M+H]<sup>+</sup> 434.1943; found 434.1941.

#### Diethyl 1-butyl-2-methyl-6-oxo-1,6-dihydropyridine-3,4dicarboxylate (6c)

R<sub>f</sub>: 0.3; Hexane: Ethyl acetate mixture (10:3); Yield: 40%; pale brown colour semi solid; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 6.84 (s, 1H), 4.30 (m, 4H), 4.17-4.00 (m, 2H), 2.52 (s, 3H), 1.72-1.59 (m, 2H), 1.50-1.22 (m, 9H), 0.97 (t, *J* = 7.3 Hz, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 166.7, 165.1, 161.8, 147.4, 140.3, 118.6, 111.0, 61.9, 61.6, 44.8, 30.1, 20.1, 17.1, 13.8, 13.5, HRMS (ESI): calcd for C<sub>16</sub>H<sub>24</sub>O<sub>5</sub>N [M+H]<sup>+</sup> 310.1635; found 310.1637.

### Page 10 of 11

#### Diethyl 2-methyl-6-oxo-1-phenyl-1,6-dihydropyridine-3,4dicarboxylate (6d)

R<sub>f</sub>: 0.4; Hexane: Ethyl acetate mixture (10:3); Yield: 47%; solid; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 7.58-7.52 (m, 2H), 7.51-7.46 (m, 1H), 7.16 (m, 2H), 6.93 (s, 1H), 4.35 (q, *J* = 7.1 Hz, 2H), 4.30 (q, *J* = 7.1 Hz, 2H), 2.10 (s, 3H), 1.38 (t, *J* = 7.1 Hz, 3H), 1.33 (t, *J* = 7.1 Hz, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 166.3, 165.0, 162.2, 148.5, 141.5, 137.7, 129.9, 129.1, 127.4, 119.3, 110.5, 62.0, 61.6, 19.1, 13.8, 13.7; HRMS (ESI): calcd for C<sub>18</sub>H<sub>20</sub>O<sub>5</sub>N [M+H]<sup>+</sup> 330.1340; found 330.1339.

## Triethyl 1-benzyl-6-oxo-1,6-dihydropyridine-2,3,4-tricarboxylate (6e)

R<sub>f</sub>: 0.3; Hexane: Ethyl acetate mixture (10:3); Yield: 45%; semi solid; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): *δ* 7.40-7.23 (m, 3H), 7.22-7.14 (m, 2H), 6.84 (s, 1H), 5.31 (s, 2H), 4.35 (q, J = 7.1 Hz, 2H), 4.23 (q, J = 7.1 Hz, 2H), 4.15 (q, J = 7.1 Hz, 2H), 1.40-1.24 (m, 9H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): *δ* 165.1, 163.3, 161.5, 160.7, 144.5, 142.7, 134.9, 128.5, 127.8, 127.3, 120.8, 63.0, 62.2, 62.0, 48.8, 29.6, 13.9, 13.8, 13.2; HRMS (ESI): calcd for C<sub>21</sub>H<sub>24</sub>O<sub>7</sub>N [M+H]<sup>+</sup> 402.1556; found 402.1556.

## Triethyl 1-butyl-6-oxo-1,6-dihydropyridine-2,3,4-tricarboxylate (6f)

R<sub>f</sub>: 0.4; Hexane: Ethyl acetate mixture (10:3); Yield: 48%; pale brown colour semi solid ; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 6.71 (s, 1H), 4.44 (q, *J* = 7.0 Hz, 2H), 4.34 (q, *J* = 7.1 Hz, 2H), 4.25 (q, *J* = 7.1 Hz, 2H), 3.9 (t, *J* = 7.9 Hz, 2H), 2.73 (t, *J* = 7.1 Hz, 3H), 1.75-1.66 (m, 2H), 1.44-1.38 (m, 2H), 1.35 (t, *J* = 7.1 Hz, 3H) 1.30 (t, *J* = 7.1 Hz, 3H), 0.94 (t, *J* = 7.3 Hz, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 165.4, 163.3, 161.7, 160.4, 145.2, 142.9, 120.1, 106.9, 63.2, 62.2, 62.0, 47.4, 30.6, 20.0, 13.9, 13.8, 13.7, 13.5; HRMS (ESI): calcd for C<sub>16</sub>H<sub>27</sub>O<sub>7</sub>NNa [M+H]<sup>+</sup> 368.1690; found 368.1688.

#### Ethyl 5-acetyl-1-benzyl-2-oxo-6-phenyl-1,2-dihydropyridine-4carboxylate (6g)

R<sub>f</sub>: 0.4; Hexane: Ethyl acetate mixture (10:3); Yield: 11%; brown colour semi solid; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>): δ 7.81 (d, *J* = 7.78 Hz, 2H), 7.59-7.55 (m, 1H), 7.45 (t, *J* = 7.78 Hz, 2H), 7.33 (t, *J* = 7.47 Hz, 2H), 7.30-7.27 (m, 1H), 7.19-7.15 (m, 3H), 5.42 (s, 2H), 4.05 (q, *J* = 7.17 Hz, 2H), 2.18 (s, 3H), 1.08 (t, *J* = 7.17 Hz, 3H); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 194.6, 164.3, 162.5, 146.2, 140.4, 137.7, 135.2, 133.4, 128.9, 128.8, 127.6, 126.3, 119.7, 117.1, 62.1, 47.6, 18.0, 13.5; HRMS (ESI): calcd for C<sub>23</sub>H<sub>22</sub>NO<sub>4</sub> [M+H]+376.1543; found 376.1542.

#### Acknowledgements

We thank Department of Science and Technology (DST) India for Fast track grant-No: SR/FT/CS-146/2010 and CSIR, New Delhi for financial support as part of XII Five Year plan under title ORIGIN (CSC-0108). We also thank V. J. Rao and GS for their support. VN thanks to DST for Project Fellowship and DP thanks UGC-JRF, NSVMRM thanks to UGC-SRF and also thank to AcSIR.

#### Notes and references

 (a) Comprehensive Heterocyclic Chemistry II; eds., A. R. Katritzky, C. W. Ress and E. F. V Scriven. Pergamon: Oxford, U. K., 1996; Vol. 1-9; (b) D. Ayres and J. D. Loike, Chemistry and pharmacology of natural products. Lignans: chemical, biological and clinical properties, Cambridge University Press, Cambridge, **1990**; (c) T. Eicher and S. Hauptmann, *The Chemistry of Heterocycles: Structure, Reactions, Syntheses, and Applications*, Wiley-VCH, Weinheim, Germany, 2nd edn., 2003; (d) M. Saleem, H. J. Kim, M. S. Ali and Y. S. Lee, *Nat. Prod. Rep.*, 2005, **22**, 696; (e) G. M. Cragg and D. J. Newman, *Expert Opin. Investg. Drugs*, 2000, **9**, 2783.

- 2 (a) Refer special "Insight" on plant defence in *Nature*, 2001, 411, 825; (b) L. M.-Sanz and F. Susanne, J. Med. Chem., 2012, 55, 4062; (c) A. Thomas, *Angew. Chem., Int. Ed.*, 2010, 49, 8328.
- (a) Comprehensive Heterocyclic Chemistry III; eds., A. R. Katritzky, C. A. Ramsden, E. F. V. Scriven and R. J. K. Taylor. Pergamon: Oxford, U. K., 2008; Vol. 1-13; (b) E. J. Kang and E. Lee, Chem. Rev., 2005, 105, 4348; (c) A. Bermejo, B. Figadere, M.-C. Zafra-Polo, I. Barrachina, E. Estornell and D. Cortes, Nat. Prod. Rep., 2005, 22, 269; (d) E. G. Brown, Ring Nitrogen and Key Biomolecules: The Biochemistry of N-heterocycles, Kluwer Academic, Boston, 1998; (e) K. C. Majumdar, G. V. Karunakar and B. Sinha, Synthesis, 2012, 44, 2475; (f) J. W. Daly, T. F. Spande and H. M. Garraffo, J. Nat. Prod., 2005, 68, 1556; (g) Z. Ye, L. Shi, X. Shao, X. Xu, Z. Xu and Z. Li, J. Agric. Food Chem., 2013, 61, 312.
- 4 Selected references on metal catalysis: (a) A. S. K. Hashmi, Acc. Chem. Res., 2014, 47, 864; (b) M. P. Muňoz; Chem. Soc. Rev., 2014, 43, 3164; (c) F. López and J. L. Mascareňas, Chem. Soc. Rev., 2014, 43, 2904; (d) J. Bariwal and E. V. D. Eycken, Chem. Soc. Rev., 2013, 42, 9283; (e) C. -L. Sun, B. -J. Li, and Z. -J. Shi, Chem. Rev., 2011, 111, 1293; (f) I. P. Beletskaya and V. P. Ananikov, Chem. Rev., 2011, 111, 1596; (g) B. M. Rosen, K. W. Quasdorf, D. A. Wilson, N. Zhang, A.-M. Resmerita, N. K. Garg, and V. Percec, Chem. Rev., 2011, 111, 1346; (h) S. Cacchi and G. Fabrizi, Chem. *Rev.*, 2011, **111**, 215; (i) X. –F. Wu, H. Neumann, and M. Beller, *Chem. Rev.*, 2013, **113**, 1; (j) G. Zeni and R. C. Larock, *Chem. Rev.*, 2006, 106, 4644; (k) M. Rudolph and A. S. K. Hashmi, Chem. Commun., 2011, 47, 6536; (I) L. Fensterbank and M. Malacria, Acc. Chem. Res., 2014, 47, 953; (m) X. -F. Wu, X. Fang, L. Wu, R. Jackstell, H. Neumann, and M. Beller, Acc. Chem. Res., 2014, 47, 1041; (n) J. Ye and S. Ma, Acc. Chem. Res., 2014, 47, 989; (o) J. Mahatthananchai and J. W. Bode, Acc. Chem. Res., 2014, 47, 696; (p) A. Fürstner, Acc. Chem. Res., 2014, 47, 925; (q) C. Obradors and A. M. Echavarren, Acc. Chem. Res., 2014, 47, 902.
- 5 Selected references on organocatalysis: (a) M. Fèvre, J. Pinaud, Y. Gnanou, J. Vignolle and D. Taton, Chem. Soc. Rev., 2013, 42, 2142; (b) M. P. Carroll and P. J. Guiry, Chem. Soc. Rev., 2014, 43, 819; (c) A. Moyano and R. Rios, Chem. Rev., 2011, 111, 4703; (d) G. Desimoni, G. Faita, and K. A. Jørgensen, Chem. Rev., 2011, 111, 284; (e) M. Bartók, Chem. Rev., 2010, 110, 1663; (f) T. B. Poulsen and K. A. Jørgensen, Chem. Rev., 2008, 108, 2903; (g) Y. Wei and M. Shi, Chem. Rev., 2013, 113, 6659; (h) E. G. -Urdiales, I. Alfonso and V. Gotor, Chem. Rev., 2005, 105, 313; (i) D. -F. Chen, Z. -Y. Han, X. -L. Zhou and L. -Z. Gong, Acc. Chem. Res., 2014, 47, 2365; (j) L. Pu, Acc. Chem. Res., 2014, 47, 1523; (k) From Biomimetic Concepts to Applications in Asymmetric Synthesis; eds., A. Berkessel and H. Gröger, Wiley-VCH: Weinheim, 2005; (l) Enantioselective Organocatalysis; P. I. Dalko, Ed.; Wiley-VCH: Weinheim, 2007; (m) M Terada and N. Momiyama in Science of synthesis: Asymetric organocatalysis ed. K Markova, Georg Thieme Verlag, stuttgart, 2012, vol 2, p 219; (n) P. I. Dalko and L. Moisan, Angew. Chem., Int. Ed., 2004, 43, 5138.
- 6 (a) B. H. Patel, A. M. Mason and A. G. M. Barrett, Org. Lett. 2011,
  13, 5156; (b) T. Hondo, M. Warizaya, T. Niimi, I. Namatame, T. Yamaguchi, K. Nakanishi, T. Hamajima, K. Harada, H. Sakashita, Y. Matsumoto, M. Orita, and M. Takeuchi, J. Med. Chem. 2013, 56, 3582; (c) E. Knovenagel, A. Fries, Chem. Ber. 1898, 31, 767; (d) E. Ziegler, F. Hradetzky, Monatsh. Chem. 1964, 95, 1247; (e) S. N. Huckin, L. Weiler, Can. J. Chem. 1974, 52, 1343; (f) T. Kato, Y. Kubota, M. Tanaka, H. Takahashi, T. Chiba, Heterocycles 1978, 9, 841; (g) K. Ito, S. Miyajima, J. Heterocyclic Chem. 1992, 29, 1037.
- (a) L. V. Reis, A. M. Lobo and S. Prabhakar, *Tetrahedron Lett.*, 1994, 35, 2747; (b) P. Sanna, A. Nuvole, P. A. Sequi and G. Paglietti, *Heterocycles*, 1993, 36, 259; (c) A. Z. A. Elassar and A. A. El-Khair, *Tetrahedron*, 2003, 59, 8463; (d) S. M. Riyadh, I. A. Abdelhamid, H. M. Al-Matar, N. M. Hilmy and M. H. Elnagdi, *Heterocycles*, 2008, 75, 1849.

- 8 (a) Q. Li, L. A. Mitscher and L. L. Shen, *Med. Res. Rev.*, 2000, 20, 231; (b) N. R. Irlapati, J. E. Baldwin, R. M. Adlington, G. J. Pritchard and A. R. Cowley, *Tetrahedron*, 2006, 62, 4603; (c) H. Miyadera, K. Shiomi, H. Ui, Y. Yamaguchi, R. Masuma, H. Tomoda, H. Miyoshi, A. Osanai, K. Kita and S. Omura, *Proc. Natl.* Acad. Sci. U. S. A., 2003, 100, 473; (d) C. J. Cowden and I. Paterson, *Org. React.*, 1997, 51, 1; (e) P. Wipf and H. Jahn, *Tetrahedron*, 1996, 52, 12853.
- 9 (a) A. E. Farah and A. A. Alousi, *Life Sci.*, 1978, 1139; (b) A. A. Alousi, J. M. Canter, M. J. Monterano, D. J. Fort and R. A. Ferrari, *J. Cardiovasc Pharmacol.*, 1983, **5**, 1139.
- (a) C. J. McElhinny, F. I. Carroll and A. H. Lewin, *Synthesis*, 2012, 44, 57;
   (b) M. A. Rogawski, *Epilepsy Currents*, 2011, 11, 56;
   (c) O. Rascol, P. Barone, M. Behari, M. Emre, N. Giladi, C. W. Olanow, E. Ruzicka, F. Bibbiani, D. Squillacote, A. Patten and E. Tolosa, *Clin. Neuropharmacol.*, 2012, 35, 15.
- 11 (a) K. Paulvannan and T. Chen, J. Org. Chem., 2000, 65, 6160; (b) Source: MDL Drug Data registry, by MDL information systems, Inc.; San Leandro, CA.
- 12 (a) N. Hirano, J. Kohno, S. Tsunoda, M. Nishio, N. Kishi, T. Okuda, K. Kawano, S. Komatsubara and N. Nakanishi, *J. Antibiot.*, 2001, **54**, 42; (b) L. A. Hasvold, W. Wang, S. L. Gwaltney, II, T. W. Rockway, L. T. J. Nelson, R. A. Mantei, S. A. Fakhoury, G. M. Sullivan, Q. Li, N.-H. Lin, L. Wang, H. Zhang, J. Cohen, W. -Z. Gu, K. Marsh, J. Bauch, S. Rosenberg and H. L. Sham, *Bioorg. Med. Chem. Lett.*, 2003, **13**, 4001.
- (a) V. Åberg and F. Almqvist, Org. Biomol. Chem., 2007, 5, 1827;
  (b) L. Cegelski, J.S. Pinkner, N.D. Hammer, C.K. Cusumano, C.S. Hung, E. Chorell, V. Åberg, J.N. Walker, P.C. Seed, F. Almqvist, M.R. Chapman and S.J. Hultgren, Nat. Chem. Biol., 2009, 5, 913.
- 14 P. S. Dragovich, T. J. Prins, R. Zhou, E. L. Brown, F. C. Maldonado, S. A. Fuhrman, L. S. Zalman, T. Tuntland, C. A. Lee, A. K. Patick, D. A. Matthews, T. F. Hendrickson, M. B. Kosa, B. Liu, M. R. Batugo, J.-P. R. Gleeson, S. K. Sakata, L. Chen, M. C. Guzman, J. W. Meador, III, R. A. Ferre, and S. T. Worland, *J. Med. Chem.*, 2002, 45, 1607.
- 15 Q. Li, L. A. Mitscher, L. L. Shen, Med. Res. Rev., 2000, 20, 231.
- 16 (a) P. E. J. Sanderson, D. L. Dyer, A. M. Naylor-Olsen, J. P. Vacca, S. J. Gardell, S. D. Lewis, B. J. Jr. Lucas, E. A. Lyle, J. J. Lynch, Jr.; A. M. Mulichak, *Bioorg. Med. Chem. Lett.*, 1997, 7, 1497; (b) S. Y. Tamura, J. E. Semple, J. E. Reiner, E. A. Goldman, T. K. Brunck, M. S. Lim-Wilby, S. H. Carpenter, W. E. Rote, G. L. Oldeshulte, B. M. Richard, R. F. Nutt, W. C. Ripka, *Bioorg. Med. Chem. Lett.*, 1997, 7, 1543.
- 17 J. J. Parlow, R. G. Kurumbail, R. A. Stegeman, A. M. Stevens, W. C. Stallings and M. S. South, J. Med. Chem., 2003, 46, 4696.
- 18 T. Shimatani, N. Hosotani, M. Ohnishi, K. Kumagai, I. Saji, J. Antibiot., 2006, 59, 29.
- (a) L. G. Beholz, P. Benovsky, D. L. Ward, N. S. Barta, J. R. Stille, J. Org. Chem., 1997, 62, 1033; (b) P. R. Bernstein, B. C. Gomes, B. J. Kosmider, E. P. Vacek, J. C. Williams, J. Med. Chem., 1995, 38, 212.
- 20 (a) D. Ž. Mijin, G. S. Ušćumlić, N. V. Valentić, A. D. Marinković, *Hem. Ind.*, 2011, **65**, 517; (b) V. P. Litvinov, S. G. Krivokolysko, and V. D. Dyachenko, *Chemistry of Heterocyclic Compounds, Vol.35, No.* 5, 1999.
- (a) H. J. Jessen and K. Gademann, *Nat. Prod. Rep.*, 2010, 27, 1168;
  (b) W. A. Loughlin, J. D. A. Tyndall, M. P. Glenn, and D. P. Fairlie, *Chem. Rev.*, 2004, 104, 6085.
- 22 (a) D. Ž. Mijin, J. M. Marković, D. V. Brković, A. D. Marinković, Hem. Ind., 2014, 68, 1; (b) G. Jones, Pyridines and their benzo derivatives: synthesis. In: A. McKillop (Ed.), Comprehensive heterocyclic chemistry II, Pergamon Press, Oxford, 1996, p. 167; (c) G. Jones, Pyridines and their benzo derivatives: synthesis. In: Boulton A, A. McKillop (Ed.), Comprehensive heterocyclic chemistry, Pergamon Press, Oxford, 1984, p. 395; (d) H. Tieckelmann, In Hetrocyclic Compounds, Pyridine and Its Derivatives, Supplement Part Three; Abramovitch, R. A., Ed.; Wiley-Interscience: New York, 1974, p 597.
- 23 Literature on enaminones: (a) R. V. Edwankar, C. R. Edwankar, O. A. Namjoshi, J. R. Deschamps and J. M. Cook, *J. Nat. Prod.*, 2012, 75, 181; (b) T. Miura, Y. Funakoshi, M. Morimoto, T. Biyajima and M. Murakami, *J. Am. Chem. Soc.*, 2012, 134, 17440; (c) B. Stanovnik and J. Svete, *Chem. Rev.*, 2004, 104, 2433;

- 24 The Chemistry of Enamines; Z. Rappoport, ed.; John Wiley & Sons, New York, 1994, Part 1, 525; (e) A. Saito, T. Konishi and Y. Hanzawa, Org Lett., 2010, 12, 372; (f) S. Cacchi, G. Fabrizi and E. Filisti, Org Lett., 2008, 10, 2629; (g) Z. Chen, J. Zhu, H. Xie, S. Li, Y. Wu and Y. Gong, Chem. Commun., 2010, 46, 2145; (h) R. -L. Yan, J. Luo, C. -X. Wang, C. -W. Ma, G. -S. Huang and Y. -M. Liang, J. Org. Chem., 2010, 75, 5395; (i) M. Rudolph and A. S. K. Hashmi, Chem. Soc. Rev., 2012, 41, 2448.
- 25 (a) K. Goutham, V. Nagaraju, S. Suresh, P. Raghavaiah and G. V. Karunakar, *RSC Adv.*, 2014, 4, 21054; (b) K. Goutham, N. S. V. M. Rao. Mangina, S. Suresh, P. Raghavaiah and G. V. Karunakar, *Org. Biomol. Chem.*, 2014, 12, 2869.
- (a) R. -L. Yan, J. Luo, C. -X. Wang, C. -W. Ma, G.-S. Huang and Y. -M. Liang, *J. Org. Chem.*, 2010, **75**, 5395; (b) I. Yavari and M. J. Bayat, *Tetrahedron Lett.*, 2011, **52**, 6649.
- 27 (a) A. S. Karpov and T. J. J. Müller, Org. Lett., 2003, 5, 3451; (b) A. S. Karpov and T. J. J. Müller, Synthesis, 2003, 2815; (c) H. Yamamoto and K. Maruoka, J. Am. Chem. Soc. 1981, 103, 6133; (d) T. Naka and K. Koide, Tetrahedron Lett.; 2003, 44, 443; (e) R. J. Cox, D. J. Ritson, T. A. Dane, J. Berge, J. P. H Charmant and A. Kantacha, Chem. Commun., 2005, 1037; (f) G. -W. Wang and C. -B. Miao, Green Chem., 2006, 8, 1080.
- 28 Bruker (2001). SAINT (Version 6.28a) & SMART (Version 5.625). Bruker AXS Inc., Madison, Wisconsin, USA.
- 29 G. M. Sheldrick, Acta Crystallogr, 2008, A64, 112.