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Exploring dual electrophiles in peptide-based proteasome inhibitors: carbonyls and epoxides

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Peptide epoxyketones are potent and selective proteasome inhibitors. Selectivity is governed by the epoxyketone dual electrophilic warhead, which reacts with the N-terminal threonine 1,2amino alcohol uniquely present in proteasome active sites. We studied a series of C-terminally modified oligopeptides featuring adjacent electrophiles based on the epoxyketone warhead. We found that the carbonyl moiety in the natural warhead is essential, but that the adjacent epoxide can be replaced by a carbonyl, though with considerable loss of activity.

Introduction

The ubiquitin-proteasome system (UPS) is the major cytosolic and nuclear protein degradation pathway in many organisms.¹ The 26S proteasome, composed of a catalytic 20S core particle and a 19S regulator particle, is responsible for the turnover of proteins tagged for degradation through poly-ubiquitin chains. In higher vertebrates, the constitutive proteasome core 20S particle contains three distinct catalytic activities, namely $\beta 1$ (cleaving after acidic amino acids), $\beta 2$ (cleaving after basic amino acids) and $\beta 5$ (cleaving after hydrophobic amino acids).² In immunoproteasomes, these catalytic subunits are replaced by B1i, B2i and B5i, respectively.³ Cortical thymic epithelial cells uniquely express ß5t subunits, which may replace $\beta 5i$ to form thymoproteasomes.⁴ To date numerous proteasome inhibitors have been reported. These include both natural products and synthetic compounds.⁵ Of note, proteasomes are validated drug targets in oncology and the peptide boronic acid, bortezomib⁶ as well as the peptide epoxyketone, carfilzomib⁷, are used in the clinic for the treatment of multiple myeloma and mantle cell lymphoma.



Figure 1. Structures of epoxomicin (1) and peptide α -ketoaldehyde (2) that are at the basis of the here presented studies.

Carfilzomib is a member of the family of peptide epoxyketone proteasome inhibitors. The first compound identified in this class is the natural product; epoxomicin (Figure 1, 1).⁸

Epoxomicin features an epoxyketone electrophilic trap, also referred to as the warhead, which renders this compound highly selective towards the proteasome catalytic sites. Proteasome catalytic activities are characterized by an N-terminal threonine in which the hydroxyl acts as the nucleophile with the amine as catalytic base. Such catalytic sites are rare in nature and peptide epoxyketones appear ideally suited to react with the 1,2-aminoalcohol moiety in these N-terminal threonine residues.⁹ In the inhibition mechanism, the carbonyl of the epoxyketone warhead is first attacked by the N-terminal threonine γ -hydroxyl (Figure 2) after which the α -amine attacks the epoxide ring, resulting in morpholine ring formation.



Figure 2. Mechanism of peptide epoxyketone (A) and peptide α -keto-aldehyde (B) mediated proteasome inactivation.

A large number of epoxyketone warhead containing inhibitors of proteasome have been reported¹⁰ including the aforementioned clinical drug, carfilzomib. A related class of proteasome inhibitors more recently investigated in detail and that also capitalize on the reactive 1,2-aminoalcohol present in proteasome active sites comprises the peptide α -keto-aldehydes (Figure 1, 2).¹¹ In a subsequent study, Groll and co-workers reported a series of N-benzyloxycarbonyl-trileucinyl-1,2dicarbonyl derivatives based on this motif and identified the α -

ketoamide electrophile as the most effective inhibitor of this series.¹² In α -keto-aldehyde 2, the carbonyl occupying the position normally occupied by the amide carbonyl of the scissile peptide bond is attacked first by the N-terminal threonine γ -hydroxyl to reversibly form the hemi-ketal. This then further condenses to form through a cyclic carbinolamine intermediate the corresponding 5,6-dihydro-2H-1,4-oxazine. In contrast to epoxyketone-mediated inhibition, this process is reversible, but the analogy is remarkable: two adjacent electrophilic carbons are brought near to the proteasome Nterminal threonine 1,2-amino alcohol to form after double nucleophilic attack a six-membered adduct. This analogy invites the question whether hybrid structures varying in the number and position of carbonyl moieties (featuring an sp² hybridized electrophilic carbon) and epoxides (with an sp³ electrophilic carbon) would be effective proteasome inhibitors. We decided to put this question to the test by the design, synthesis



evaluation as proteasome inhibitors of compounds **3-8** (Figure 3).

Figure 3. Target structures 3-8 that are at the basis of the herepresented studies

N-Benzyloxycarbonyl-trileucinyl epoxyketone (ZL3epoxyketone) **3** is a known^{10a} and easily accessible broadspectrum proteasome inhibitor and a close analogue of the natural product, epoxomicin. The compound features an α keto-epoxide as the electrophilic trap with the two electrophilic carbons hybridized as sp²-sp³. In compound 4 the direction of the epoxyketone is inverted and the two electrophilic carbons are now sp³-sp²-hybridized. Compound 5 features two adjacent epoxides (sp^3-sp^3) and compound 7, as close analogue to lead structure **2**, features two adjacent carbonyls (sp^2-sp^2) . Compounds 6 and 8 finally feature an enone system, thus Michael acceptors with the two electrophilic carbons now separated by one carbon. We thought to include these latter compounds because their ease of synthesis and because conjugate addition has been shown to be another valid strategy in the design of peptide-based proteasome inhibitors.¹³

Results and discussion

The synthesis of inverted epoxyketone 4 commenced with an enantioselective Mannich reaction (Scheme 1). Phosphonate 9 and sulfonate 10 were prepared following literature procedures¹⁴, and used in a one-pot procedure to form compound 11 via an enantioselective Mannich reaction catalyzed by quinine-based catalyst 12, followed by Horner olefination. According to chiral HPLC analysis, compound 11 was obtained as a 1:3 mixture of enantiomers and we continued the synthesis with this mixture. After reduction of 11 with



Scheme 1. Synthesis of 4ab. Reagents and conditions: (a) i) 12 (cat.), toluene/DCM, aq. 50% K_3PO_4 , -20 °C, 60h; ii) aq. 37% HCHO, aq. 50% K_3PO_4 , RT, 60h; (b) CeCl₃.7H₂O, NaBH₄, MeOH/; (c) i) *t*BuOOH, VO(acac)₂, DCM; ii) Dess-Martin periodinane, DCM; (d) 50%TFA/DCM; (e) i) *t*BuONO, HCl, DMF/DCM, -30 °C; ii) 15a or 15b, DiPEA/DMF.

After diastereoselective epoxidation (*t*BuOOH and VO(acac)₂) Dess-Martin oxidation of the hydroxyl group, and diastereomeric compounds 14a and 14b were obtained. Treatment with TFA yielded the leucine keto-epoxides 15a and 15b. Compound 16 was prepared according to literature methods.¹⁵ Finally, compounds 4a and 4b were prepared by standard azide peptide coupling of hydrazide 16 with either of the Leu-keto-epoxides 15a and 15b. During reverse HPLC purification of 4a and 4b, there was a small peak next to the main peak, which had the same mass as the product according to the LC/MS analysis. We argued that this most probably is the enantiomeric impurity which was introduced in compound 11. This impurity was removed HPLC purification, and NMRanalysis showed clearly that only one diasteromer was isolated for both compounds 4a and 4b. Therefore, we assume that the sole difference between 4a and 4b lies in the orientation of the epoxide ring. Again, we could not establish the absolute stereochemistry.

The synthesis of the di-epoxides commenced with Wittig olefination of the Cbz-Leu-epoxyketone **21**, prepared following the route of synthesis as published in the patent literature.¹⁶ Briefly, Boc-leucine **17** was transformed into Weinreb amide **18** to which was added 2-vinyllithium, prepared for this

purpose through transmetallation of 2-bromopropene with tertbutyllithium. The intermediate enone was stereoselectively reduced under Lüche conditions to provide after separation of the formed diastereomers allylic alcohol **19** in good yield and enantiomeric purity. Sharpless allylic epoxidation followed by Dess Martin oxidation gave leucine epoxyketone **20**. Removal of the Boc group and installment of the N-Cbz protective group gave key intermediate **21** in good overall yield (Scheme 2). Wittig olefination of Cbz-Leu-epoxyketone **21** gave alkene **22**, and ensuing epoxidation with mCPBA yielded epoxides **23a** and **24b** in equal amounts. The absolute stereochemistry of these two compounds, which were obtained as enantiomerically pure di-epoxides, again could not be determined.



Scheme 2. Synthesis of **5ab**. Reagents and conditions: (a) N,Odimethylhydroxylamine hydrochloride, HCTU, DiPEA, DCM; (b) i) 2-bromopropene, tBuLi, Et₂O; 2) NaBH₄, CeCl₃.7H₂O, MeOH; (c) i) VO(acac)₂, tBuOOH, DCM; 2) Dess Martin periodinane, DCM; (d) i) 50% TFA/DCM; ii) CbzCl, DiPEA, DCM; (e) [Ph₃PCH₃]⁺Br⁻, KHMDS, THF, RT to -78 °C; (f) mCPBA, DCM; (g) Pd/C, 1,4-cyclohexadiene, THF/MeOH; (h) i) *t*BuONO, HCl, DMF/DCM, -30 °C; ii) **24a** or **24b**, DiPEA/DMF.

After Cbz-removal (to give 24a/b) and condensation of these warheads with dipeptide 16, 5a and 5b were obtained. (S)-4-amino-2,6-dimethylhept-1-en-3-one 25, which we obtained in route to leucine epoxyketone 21, was condensation with 16 to obtain 6 (Scheme 3). Ozonolysis of 6 gave diketone 7. Compound 16 was converted in a similar sequence of events to give 8 as an inseparable mixture of diastereomers.

The inhibition properties of all compounds were determined in a competitive activity-based protein profiling assay using broad-spectrum proteasome activity-based probe BODIPY- epoxomicin 27^{17} as the read-out. The results (Figure 4) reveal that interchanging the position of the epoxide and the ketone part led to a dramatic drop in proteasome inhibitory potency (4a and 4b compared with 3).



Scheme 3. Synthesis of 6 and 7. Reagents and conditions: (a) i) *t*BuONO, HCl, DMF/DCM, -30 °C; ii) 25 or 26, DiPEA/DMF. (b) O₃, PPh₃, DCM, -78 °C to RT.

Compound **4b** proved almost completely inactive whereas diastereoisomer **4a** inhibited β 1 and β 2 at high concentration (>300 μ M). In case of di-epoxides **5a** and **5b**, inhibition of β 1 and β 2 subunits was observed at concentration above 100 μ M, while β 5 is not even completely blocked at 1 mM. Dicarbonyl derivative **7** displayed some β 5-selectivity in a manner similar to ZL3-epoxyketone **3**. Enones **6** and **8** showed similar inhibitory potential. At high concentration (100 μ M), β 1 was completely inhibited whereas β 2 and β 5 were only partially inhibited. Although being slightly less potent than **3**, diketo compound **7** showed complete β 5 inhibition already at 10 μ M, whereas β 1 and β 2 were inhibited at much higher concentrations.

Our results underscore the fact that the epoxyketone geometry as present in the natural product epoxomicin and synthetic compounds featuring the same adjacent sp³-sp² electrophilic carbons is the most effective design where it comes to sp²-carbonyl in peptide inhibition. The proteasome epoxyketones situated at the position resembling the amide carbonyl in a scissile amide bond of a proteasome substrate appears highly important for effective inhibition. Interestingly, peptide vinyl sulfones - another major class of proteasome inhibitors - position an sp²-hybridized electrophilic carbon for nucleophilic 1,4-addition at the same location whereas arguably the covalent but reversible inhibition effected by peptide boronic acids¹⁸ proceeds starting from an sp²-hybridized boron as well. Apparently, a situation in which the reactive carbon/carbon-replacing atom in a substrate/inhibitor is offered to the active site in an sp² geometry to yield after nucleophilic addition of the threonine-OH an sp³-adduct is ideal. The secondary electrophilic trap - the tertiary epoxide carbon in epoxomicin – is amenable for modification to provide sp² analogues, as is evidenced by peptide diketone 7 in a result that

30 100 300 1000 μM 10 3 3 General indicated 4a 4b 5a 5b 6 7 was used. 8 > B2

complements the literature findings^{11,12} on related compounds.

It is however also obvious that inhibitory potency is partially

compromised in such sp²-sp² hybridized compounds.

Figure 4. Inhibition profile of compounds 3-8 and the structure of ABP (27).

ABP, 27

Our set of compounds completes a series of peptide epoxyketone analogues as potential proteasome inhibitors that has been growing in the literature in recent years. We could not determine the absolute stereochemistry of the various epoxides we studied. However, in all cases we obtained the respective diastereoisomeric pairs as enantiomerically pure compounds, neither of which proved potent proteasome inhibitors. This result, coupled with our previous¹⁹ finding that a diastereomeric epoxyketone with respect to the chirality of the epoxide carbon

in epoxomicin derivatives also results in a drastic drop in activity, leads to the conclusion that the geometry and absolute stereochemistry of the epoxomicin warhead is indeed ideally suited for optimal proteasome inactivation.

Experimental

All reagents were of commercial grade and used as received unless otherwise. Methylene chloride (DCM), dimethylformamide (DMF) and tetrahydrofuran (THF) were stored over 4 Å molecular sieves. Reactions were conducted under an argon atmosphere. Reactions were monitored by TLC analysis by using DC-fertigfolien (Schleicher & Schuell, F1500, LS254) with detection by UV absorption (254 nm), spraying with an aqueous solution of $KMnO_4$ (7%) and KOH (2%). Column chromatography was performed on silica gel from Screening devices (0.040-0.063 mm). ¹H-NMR and ¹³C-APT-NMR spectra were recorded on Bruker AV-400 (400 MHz) or Bruker AV-600 (600 MHz) machines. Chemical shifts are given in ppm (δ) relative to tetramethylsilane as an internal standard. Coupling constants are given in Hz. Peak assignments are based on 2D ¹H-COSY and ¹³C-HSOC NMR experiments. All presented ¹³C-APT spectra are proton decoupled. LC/MS analysis was performed on a LCQ Advantage Max (Thermo Finnigan) equipped with a Gemini C18 column (Phenomenex). HRMS was recorded on a LTO Orbitrap (Thermo Finnigan). For reverse phase HPLC purification of the final compounds, an automated Gilson HPLC system equipped with a C18 semiprep column (Gemini C18, 250×10 mm, 5µ particle size, Phenomenex)

General procedure for Boc deprotection

The appropriate Boc-protected C-terminally modified leucine derivative was dissolved in TFA:DCM (1:1, v/v) and stirred for 20 min. Co-evaporation with toluene (3x) afforded the TFA-salt, which was used without further purification.

General procedure for azide peptide coupling

The hydrazide (16) was dissolved in 1:1 DMF:DCM (v/v) and cooled to -30 °C. tBuONO (1.1 eq.) and HCl (4M solution in 1,4dioxane, 2.8 eq.) were added, and the mixture was stirred for 3 hr. at -30 °C after which TLC analysis (10% MeOH/DCM, v/v) showed complete consumption of the starting material. The warhead-TFA salt was added to the reaction mixture as a solution in DMF with 5.0 eq. of DiPEA and this mixture was allowed to warm up to room temperature slowly overnight. The mixture was diluted with EA and extracted with H₂O (3x) and brine. The organic layer was dried over MgSO4 and purified by reverse phase HPLC.

Tert-butyl (S)-(6-methyl-3-methylene-2-oxoheptan-4-yl)carbamate (11)

Diphenylphosphorylketone 9 (798 mg, 1.54 mmol, 1.1 eq.), compound 10 (917 mg, 2.80 mmol), and catalyst 12 (134 mg, 0.28

mmol, 0.1 eq.) were added to the flask and the solids were suspended in a toluene/DCM (7:3) mixture (14.0 mL), and the mixture was cooled to -20 °C. Aqueous K₃PO₄ (50% w/w, 3.8 mL, 14.0 mmol, 5.0 eq.) was then added in one portion. After the resulting biphasic mixture had been vigorously stirred at the same temperature for 60 h, ag. formaldehyde (37% w/w, 1.0 mL, 14.0 mmol, 5.0 eq.) was added, followed by another portion of aq. K₃PO₄ (50% w/w, 15.4 mL, 56.0 mmol, 20.0 eq.). The mixture was stirred at room temperature for 60 h. The organic layer was then charged directly on a silica gel chromatographic column, the aqueous phase was extracted with toluene (2x7.0 mL), and the extracts were charged as well. Purification by column chromatography (1% EtOAc/pentane \rightarrow 5% EtOAc/pentane) gave 11 (469.7 mg, 1.8 mmol, 66% yield). The enantiomeric excess (ee) of the product was determined by HPLC (Chiralcel OD (250×4.6mm), n-hexane/i-PrOH 99:1, 1.0 mL/min, $\lambda = 254$ nm: $\tau_{max} = 5.612$ min, $\tau_{min} = 6.526$ min, 50.8% ee). ¹H NMR (400 MHz, CDCl₃): δ 6.04 (s, 1H), 5.94 (s, 1H), 5.18 (s, 1H), 4.44 (s, 1H) 2.33 (s, 3H), 1.54-1.44 (m, 3H), 1.42 (s, 9H), 2.04 (s, 3H), 0.93-0.89 (m, 6H). ¹³C NMR (100 MHz, CDCl₃): δ 199.91, 155.32, 148.78, 126.62, 79.23, 52.68, 43.96, 28.50, 25.32, 23.57, 22.65, 22.29. LC-MS (linear gradient $10 \rightarrow 90\%$ MeCN/H₂O, 0.1% TFA, 12.5 min): Rt (min): 7.68 (ESI-MS (m/z): 255.80 (M⁺)). HRMS calculated for $C_{14}H_{26}NO_3$ 256.19072 $[M+H]^+$; found 256.19085. $[\alpha]^{21}_{D}$ +9.8 (C=1, CHCl₃).

Tert-butyl ((2R/S,4R)-2-hydroxy-6-methyl-3-methyleneheptan-4-yl)carbamate (13a and 13b)

A solution of compound 11 (400 mg, 1.57 mmol) in methanol was put under argon atmosphere and CeCl₃·7H₂O (882 mg, 2.36 mmol, 1.5 eq.) was added. The solution was cooled to 0 °C, before NaBH₄ (71.2 mg, 1.88 mmol, 1.2 eq.) was added in 3 portions over 10 min. After stirring for 30 min., the reaction was quenched with glacial acetic acid and toluene was added after an additional 20 min at 0 °C. The solvents were removed in vacuo and the oily residue was dissolved in an EtOAc/H2O mixture. The organic layer was washed with H₂O and brine, dried over anhydrous MgSO4 and concentrated in vacuo. Purification by column chromatography (2%) EtOAc/pentane \rightarrow 10% EtOAc/pentane) gave 13a (153.4 mg, 0.6 mmol, 38% yield) and 13b (105.0 mg, 0.4 mmol, 26% yield). 13a: ¹H NMR (400 MHz, CDCl₃): δ 5.11 (s, 1H), 5.02 (s, 1H), 4.75 (s, 1H), 4.34 (q, J=6.8Hz, 1H), 4.21(s, 1H), 1.69-1.66 (m, 1H), 1.41 (s, 11H), 1.35 (d, J= 8Hz, 3H), 0.95 (d, J=6.8 Hz, 6H). ¹³C NMR (100 MHz, CDCl₃): δ 156.04, 155.07, 110.34, 79.88, 69.45, 48.96, 44.71, 28.52, 25.25, 23.35, 22.39, 22.00. **13b:** ¹H NMR (100 MHz, CDCl₃): δ 5.18 (s, 1H), 5.05 (s, 1H), 4.69 (s, 1H), 4.31 (q, J=6.4Hz, 1H), 4.20 (q, J=7.6 Hz, 1H), 1.70-1.65 (m, 1H), 1.45 (s, 11H), 1.38 (d, J= 6.4 Hz, 3H), 0.99 (d, J=6.8 Hz, 1H). ¹³C NMR (100 MHz, CDCl₃): δ 155.85, 155.54, 109.21, 79.82, 68.79, 50.48, 44.76, 28.54, 25.15, 23.22, 22.84, 22.24. HRMS calculated for C₁₄H₂₈NO₃ 258.20637 [M+H]⁺; found 258.20654.

*T*ert-butyl ((S)-1-((R/S)-2-acetyloxiran-2-yl)-3-methylbutyl) carbamate (14a and 14b)

Compound 13a (100 mg, 0.39 mmol) was dissolved in anhydrous DCM and cooled to 0 °C. Next, VO(acac)₂ (11 mg, 0.04 mmol, 0.1 eq.) and tBuOOH (0.21 ml 5.5 M in decane, 1.17 mmol, 3.0 eq.) were added and the reaction mixture was stirred for 2 h, while allowing the temperature to rise slowly to room temperature. The resulting purple solution was concentrated in vacuo and the residue was dissolved in EtOAc and washed with half saturated NaHCO₃ solution. The aqueous layer was extracted with EtOAc and the combined organic layer was washed with H₂O and brine. The organic layer was dried over anhydrous MgSO4 and concentrated in vacuo. The crude product was used in the next step without any purification. Dess-Martin periodinane (496 mg, 1.17 mmol, 3.0 eq.) was suspended in anhydrous DCM, put under argon atmosphere and cooled to 0 °C. The crude product was co-evaporated with toluene, dissolved in anhydrous DCM and added to the first solution. The reaction mixture was stirred overnight and allowed to warm up to room temperature. Sat. aq. NaHCO₃ was added and the layers were separated. The aqueous layer was extracted with EtOAc and the combined organic layers were washed with H₂O, brine, dried over anhydrous MgSO₄ and concentrated in vacuo. The crude product was purified by column chromatography (1% EtOAc/pentane \rightarrow 10% EtOAc/pentane) to yield the title compound (14a, 84.0 mg, 0.31 mmol, 79%). 14b was prepared in the same method and on the same scale, yielding the title compound (14b, 42.1 mg, 0.15 mmol, 40%). 14a: ¹H NMR (400 MHz, CDCl₃): δ 4.48-4.43 (m, 2H), 2.92-2.88 (m, 2H), 2.05 (s, 3H), 1.68-1.63 (m, 1H), 1.43 (s, 10H), 1.29-1.21 (m, 1H), 0.96-0.90 (m, 6H). ¹³C NMR (100 MHz, CDCl₃): δ 206.60, 155.48, 79.57, 64.92, 49.30, 47.00, 40.92, 28.33, 24.90, 24.59, 23.55, 21.25. LC-MS (linear gradient $10 \rightarrow 90\%$ MeCN/H₂O, 0.1% TFA, 12.5 min): R_t (min): 7.65 (ESI-MS (m/z): 271.80 (M⁺)). 14b: ¹H NMR (400 MHz, CDCl₃): δ 5.33 (d, *J*=10Hz, 1H), 3.50-3.44 (m, 1H), 3.15 (d, J=4.8Hz, 1H), 2.97 (d, J=5.2Hz, 1H), 2.01 (s, 3H), 1.67-1.51 (m, 2H), 1.44 (s, 10H), 1.29-1.21 (m, 1H), 0.93-0.89 (m, 6H). ¹³C NMR (100 MHz, CDCl₃): δ 208.64, 155.48, 79.41, 52.59, 51.26, 40.73, 28.43, 24.92, 24.48, 23.17, 22.04. LC-MS (linear gradient 10→90% MeCN/H₂O, 0.1% TFA, 12.5 min): R_t (min): 7.87 (ESI-MS (m/z): 271.80 (M⁺)). HRMS calculated for $C_{14}H_{26}NO_4$ 272.18563 [M+H]⁺; found 272.18592.

Benzyl ((S)-1-(((S)-1-(((R/S)-2-acetyloxiran-2-yl)-3methylbutyl)amino)-4-methyl-1-oxopentan-2-yl)amino)-4-methyl -1-oxopentan-2-yl)carbamate (4a and 4b)

Compound **15a** was prepared by the general procedure for Boc deprotection (**14a**, 14.9 mg. 55 µmol, 1.1 eq.), followed by the general procedure for azide coupling using **16** (19.6 mg, 50 µmol, 1 eq.). Purification by preparative HPLC ($60\% \rightarrow 90\%$ MeCN/H₂O) yielded the title compound (**4a**, 4.9 mg, 9.2 µmol, 18%). Compound **4b** was prepared in the same method and same scale, yielding the title compound (**4b**, 6.9 mg, 13.0 µmol, 26%). **4a**: ¹H NMR (600 MHz, CDCl₃): δ 7.26-7.15 (m, 5H), 5.03 (s, 2H), 4.75 (t, *J*=2.4 Hz, *J*=3.2 Hz, 1H), 4.20 (m, 1H), 4.03 (t, *J*=7.2 Hz, *J*=7.2 Hz, 1H), 2.80 (d, *J*=4.4 Hz, 1H), 2.74 (d, *J*=3.2 Hz, 1H), 1.91 (s, 3H), 1.61-1.38 (m, 8H), 1.08-1.01 (m, 1H), 0.85-0.78 (m, 18H). ¹³C NMR (150 MHz, CDCl₃): δ 208.80, 175.62, 174.88, 158.58, 129.48, 129.01, 128.85, 67.76, 65.82, 55.09, 53.31, 49.64, 45.94, 41.94, 41.52,

40.21, 26.04, 25.95, 25.87, 24.52, 23.94, 23.39, 23.36, 21.94, 21.24. LC-MS (linear gradient 10 \rightarrow 90% MeCN/H₂O, 0.1% TFA, 12.5 min): R_t (min): 8.39 (ESI-MS (m/z): 532.20, (M+H⁺)). **4b**: ¹H NMR (600 MHz, CDCl₃): δ 7.26-7.16 (m, 5H), 5.00 (s, 2H), 4.26 (t, *J*=5.2 Hz, *J*=4.8 Hz, 1H), 4.04 (m, 1H), 3.95 (m, 1H), 2.90 (m, 2H), 1.88 (s, 3H), 1.61-1.27 (m, 9H), 0.86-0.74 (m. 18H). ¹³C NMR (150 MHz, CDCl₃): δ 208.95, 175.57, 174.32, 158.58, 138.21, 129.47, 129.02, 128.89, 67.89, 63.26, 55.14, 53.29, 50.59, 49.57, 42.05, 41.50, 40.64, 25.92, 25.90, 24.45, 23.74, 23.41, 21.96, 21.86, 21.78. LC-MS (linear gradient 10 \rightarrow 90% MeCN/H₂O, 0.1% TFA, 12.5 min): R_t (min): 8.39 (ESI-MS (m/z): 532.20, (M+H⁺)). HRMS calculated for C₂₉H₄₆N₃O₆ 532.33811 [M+H]⁺; found 532.33796.

Tert-butyl (S)-(1-(methoxy(methyl)amino)-4-methyl-1-oxopentan-2-yl)carbamate (18)

Boc-Leu-OH (4.63 g, 20 mmol) was dissolved in DCM (200 mL), followed by addition of HBTU (9.10 g, 24 mmol, 1.2 eq.). After stirring for 5 min DiPEA (10.5 mL, 60 mmol, 3.0 eq.) and N,Odimethylhydroxylamine HCl (2.35 g, 24 mmol, 1.2 eq.) was added. TLC analysis indicated completion after 2 h. The reaction mixture was diluted with EtOAc, washed with 1M aqueous HCl solution, sat. aq. NaHCO3 and brine, dried over MgSO4 and concentrated in vacuo. Purification by column chromatography (10%)EtOAc/pentane \rightarrow 40% EtOAc/pentane) yielded the title compound as a clear oil (4.6 g, 17 mmol, 83%). ¹H NMR (400 MHz, CDCl₃): δ 5.07 (d, J = 9.0 Hz, 1H), 4.73 (s, 1H), 3.79 (s, 3H), 3.20 (s, 3H), 1.78 – 1.66 (m, 1H), 1.44 (s, 11H), 0.97 (d, J = 6.5 Hz, 3H), 0.93 (d, J = 6.7 Hz, 3H). ¹³C NMR (100 MHz, CDCl₃): δ 155.63, 79.43, 61.57, 48.97, 42.11, 32.08, 28.35, 24.73, 23.34, 21.58. LC-MS (linear gradient $10 \rightarrow 90\%$ MeCN/H₂O, 0.1% TFA, 15 min): R_t (min): 7.81 (ESI-MS (m/z): 274.93, (M+H⁺)).

Tert-butyl ((3R,4S)-3-hydroxy-2,6-dimethylhept-1-en-4-yl)carbamate (19)

A solution of 2-bromopropene (3.3 mL, 33.1 mmol, 3eq.) in dry Et₂O (40 mL) was cooled down to -78 °C under argon atmosphere and stirred for 15 min before adding tBuLi (31.0 mL 1.6M in pentane, 49.7 mmol, 4.5 eq.). The reaction mixture was stirred for 15 min. Weinreb amide 18 (3.0 g, 11.1 mmol) was coevaporated with toluene and dissolved in 10 mL Et₂O. This solution was added dropwise to the reaction mixture during 30 min. The resulting reaction mixture was allowed to warm up to rt and quenched after 2 h with sat. aq. NH₄Cl (40 ml). The water layer was extracted with EtOAc (3x) and the combined organics were washed with brine, dried over MgSO₄ and concentrated in vacuo. Purification by flash column chromatography (1% EtOAc/pentane \rightarrow 10% EtOAc/pentane) yielded the product as clear oil (1.60 g, 6.3 mmol, 57%). The product obtained from last step (1.60 g, 6.3 mmol) was dissolved in methanol (60 mL), followed by addition of CeCl₃ 7 H₂O (3.52 g, 9.5 mmol, 1.5 eq.). After the solution turned clear, it was cooled down to 0 °C and NaBH₄ (0.34 g, 8.8 mmol, 1.4 eq.) was added portion-wise. After 5 min TLC analysis indicated complete conversion and the reaction mixture was quenched with glacial AcOH (6 mL). The mixture was concentrated, coevaporated with toluene, dissolved in EtOAc, washed with H₂O and brine, dried over MgSO₄ and concentrated *in vacuo*. Purification by column chromatography (10% EtOAc/pentane \rightarrow 50% EtOAc/pentane) yielded the title compound (1.39 g, 5.4 mmol, 86%). ¹H NMR (400 MHz, CDCl₃): δ 5.03 (s, 1H), 4.93 (s, 1H), 4.14 (s, 1H), 3.83 (s, 1H), 1.75 (s, 3H), 1.63 (d, J = 5.9 Hz, 1H), 1.44 (s, 9H), 1.31-1.15 (m, 2H), 0.97-0.82 (m, 6H). ¹³C NMR (100 MHz, CDCl₃): δ 156.22, 144.96, 111.95, 111.34, 77.87, 51.07, 37.20, 28.48, 24.69, 23.87, 21.60, 19.52.

Tert-butyl ((S)-4-methyl-1-((R)-2-methyloxiran-2-yl)-1-oxopentan-2-yl)carbamate (20)

Compound 19 (0.786 g, 3.1 mmol) was coevaporated with toluene, dissolved in DCM (30 mL) and cooled down to 0 °C. After addition of VO(acac)₂ (0.041 g, 0.15 mmol, 0.05 eq.) the solution turned light blue-green and subsequent addition of tBuOOH (1.7 mL 5.5M solution, 9.4 mmol, 3.0 eq.) resulted in a dark brown-purple reaction mixture. After stirring for 1 h the reaction mixture was removed from the ice bath. After 15 min, the reaction mixture was concentrated in vacuo, dissolved in EtOAc and washed with a 1:1 mixture of sat. aq. NaHCO3 and H2O. The aqueous layer was extracted with EtOAc (3x) and the combined organics were washed with H₂O and brine, dried over MgSO₄ and concentrated in vacuo to give the crude intermediate as a yellow oil. Dess-Martin periodinane (1.974 g, 4.7 mmol, 1.5 eq.) was dissolved in DCM (30 mL) and cooled down to 0 °C. The intermediate was coevaporated with toluene, dissolved in DCM (30 mL) and added to the Dess-Martin periodinane solution. After 1.5 h the reaction mixture was removed from the ice bath. After another 1.5 h TLC analysis indicated full conversion of the intermediate and the reaction was quenched with sat. aq. NaHCO₃ (30 mL). The aqueous layer was extracted with DCM (2x15 mL). The combined organics were washed with H₂O (3x) and brine, dried over MgSO4 and concentrated in vacuo. Purification by column chromatography (1% EtOAc/pentane \rightarrow 5% EtOAc/pentane) yielded the title compound (391 mg, 1.4 mmol, 46%). ¹H NMR (400 MHz, CDCl₃): δ 4.89 (d, J = 8.4 Hz, 1H), 4.38 -4.27 (m, 1H), 3.29 (d, J = 4.9 Hz, 1H), 2.89 (d, J = 5.0 Hz, 1H), 1.78 – 1.66 (m, 1H), 1.52 (s, 3H), 1.48 (dd, J = 9.8, 3.2 Hz, 1H), 1.41 (s, 9H), 1.22 - 1.12 (m, 1H), 0.95 (dd, J = 12.8, 6.6 Hz, 6H). ¹³C NMR (101 MHz, CDCl₃): § 209.54, 155.63, 79.61, 58.98, 52.29, 51.37, 40.28, 28.31, 25.06, 23.41, 21.26, 16.75.

Benzyl ((S)-4-methyl-1-((R)-2-methyloxiran-2-yl)-1-oxopentan-2-yl)carbamate (21)

Tert-butyl ((S)-4-methyl-1-((R)-2-methyloxiran-2-yl)-1-oxopentan-2-yl)-carbamate (**20**) (312 mg, 1.15 mmol) was dissolved in 30% TFA in DCM and the reaction was stirred at room temperature for 30 min. All solvent was removed *in vacuo* and co-evaporated with toluene for three times. The deprotected compound was dissolved in 15 mL DCM and DiPEA (0.60 mL, 3.45 mmol, 3 eq.) was added. Then, CbzCl (0.25 mL, 1.73 mmol, 1.5 eq.) was added and the reaction was stirred at room temperature overnight. All solvent was removed *in vacuo* and the residue was dissolved in EtOAc. The

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organic layer was washed by H₂O, brine, dried over anhydrous MgSO₄ and concentrated *in vacuo*. Purification by column chromatography (1% EtOAc/pentane→5% EtOAc/pentane) yielded the title compound (290.0 mg, 0.95 mmol, 83%). ¹H NMR (400 MHz, CDCl₃): δ 7.37-7.23 (m, 5H), 6.90 (d, *J*=8.4 Hz, 1H), 5.42 (d, *J*= 8.4 Hz, 1H) 5.07-4.99 (m, 2H), 4.42-4.36 (m, 1H), 3.27 (d, *J*=5.2 Hz, 1H), 2.96 (d, *J*=4.8 Hz, 1H), 1.74-1.65 (m, 1H), 1.53-1.34 (m, 4H), 1.28-1.20 (m, 1H), 0.97-0.88 (m. 6H). ¹³C NMR (100 MHz, CDCl₃): δ 209.27, 156.22, 136.21, 128.70, 127.94, 127.41, 126.88, 66.84, 59.01, 52.20, 51.82, 40.21, 25.01, 23.37, 21.19, 16.67. LC-MS (linear gradient 10→90% MeCN/H₂O, 0.1% TFA, 12.5 min): R_t (min): 7.88 (ESI-MS (m/z): 305.93, (M+H⁺)). HRMS calculated for C₁₇H₂₄NO₄ 306.16998 [M+H]⁺; found 306.17003.

Benzyl ((S)-5-methyl-2-((S)-2-methyloxiran-2-yl)hex-1-en-3-yl)carbamate (22)

Methyltriphenylphosphonium bromide (543 mg, 1.52 mmol, 1.6 eq.) was suspended in 10 mL anhydrous THF and then potassium bis(trimethylsilyl)amide (2.7mL, 0.5M in toluene, 1.2 mmol, 1.4 eq.) was added drop by drop. The reaction was stirred at room temperature for 1 h and then cooled to -78 °C. Compound 18 (290.0 mg, 0.95 mmol) was dissolved in 5 mL anhydrous THF and added to the reaction mixture. The reaction was stirred at -78 °C for 1 h and then stirred overnight while allowing to warm up to room temperature. The reaction was quenched by saturated aqueous NH₄Cl (30 mL) and then extracted with EtOAc (3×30 mL). The combined organic layer was washed by brine, dried over anhydrous MgSO₄ and concentrated in vacuo. Purification by column chromatography (1% EtOAc/pentane→10% EtOAc/pentane) yielded the title compound (240 mg, 0.79 mmol, 83%). ¹H NMR (400 MHz, CDCl₃): δ 7.35-7.26 (m, 5H), 5.20-5.03 (m, 5H), 4.16-4.10 (m, 1H), 2.85 (d, J=4.8Hz, 1H), 2.73 (d, J=5.2 Hz, 1H), 1.70-1.60 (m, 1H), 1.49-1.25 (m, 5H), 0.96-0.89 (m, 6H). ¹³C NMR (100 MHz, CDCl₃): δ 155.68, 151.38, 136.53, 129.40, 128.47, 128.40, 128.05, 110.73, 66.83, 56.85, 54.25, 49.78, 44.98, 25.11, 23.28, 21.56. LC-MS (linear gradient 10 \rightarrow 90% MeCN/H₂O, 0.1% TFA, 12.5 min): R_t (min): 6.81 (ESI-MS (m/z): 304.07, (M+H⁺)). HRMS calculated for C₁₈H₂₆NO₃ 304.19072 [M+H]⁺; found 304.19086.

Benzyl ((S)-3-methyl-1-((2R/S, 2'R)-2'-methyl-[2,2'-bioxiran]-2-yl)butyl)-carbamate (23a and 23b)

Compound **22** (240 mg, 0.79 mmol) was dissolved in 15 mL anhydrous DCM and m-CPBA (585.0 mg. 2.37 mmol, 3 eq.) was added. The reaction was stirred at room temperature overnight. DCM was removed *in vacuo* and the residue was dissolved EtOAc and washed by saturated NaS₂O₃, saturated NaHCO₃, brine, dried over anhydrous MgSO₄ and concentrated *in vacuo*. Purification by column chromatography (1% EtOAc/pentane \rightarrow 10% EtOAc/pentane) yielded the title compounds (**23a**, 99 mg, 0.31 mmol, 39% and **23b**, 65.0 mg, 0.20 mmol, 26%). **23a**: ¹H NMR (400 MHz, CDCl₃): δ 7.41-7.29 (m, 5H), 5.17-5.08 (m, 2H), 4.72-4.69 (m, 1H), 4.34 -4.27 (m, 1H), 2.74 (d, *J*=5.2 Hz 1H), 2.68 (d, *J*=7.6 Hz, 1H), 2.65 -2.63 (m, 2H), 1.72-1.66 (m, 1H), 1.47(s, 3H), 1.44-1.35 (m, 2H), 0.99-0.85 (m, 6H). ¹³C NMR (100 MHz, CDCl₃): δ 156.37, 128.74,

128.62, 128.42, 128.27, 66.90, 60.85, 54.49, 51.89, 48.42, 47.56, 40.13, 23.79, 22.52, 21.20, 18.73. LC-MS (linear gradient 10 → 90% MeCN/H₂O, 0.1% TFA, 12.5 min): R_t (min): 7.56 (ESI-MS (m/z): 319.93, (M+H⁺)). **23b**: ¹H NMR (400 MHz, CDCl₃): δ 7.41-7.27 (m, 5H), 5.12-5.08 (m, 2H), 4.89 (d, *J*=9.2 Hz, 1H), 4.13 -4.07 (m, 1H), 2.80-2.73 (m, 3H), 2.53 (d, *J*=5.2 Hz, 1H), 1.68-1.61 (m, 1H), 1.47(s, 3H), 1.47-1.33 (m, 2H), 0.99-0.88 (m, 6H). ¹³C NMR (100 MHz, CDCl₃): δ 156.26, 128.70, 128.63, 128.56, 128.28, 66.84, 61.48, 57.03, 51.17, 49.76, 49.70, 40.60, 23.91, 22.77, 21.65, 19.04. LC-MS (linear gradient 10 → 90% MeCN/H₂O, 0.1% TFA, 12.5 min): R_t (min): 7.49 (ESI-MS (m/z): 319.93, (M+H⁺)). HRMS calculated for C₁₈H₂₆NO₄ 320.18563 [M+H]⁺; found 320.18573.

Benzyl ((S)-4-methyl-1-(((S)-4-methyl-1-(((S)-3-methyl-1-((2S or R, 2'R)-2'-methyl-[2,2'-bioxiran]-2-yl)butyl)amino)-1-oxopentan-2-yl)amino)-1-oxopentan-2-yl)carbamate (5a and 5b)

Compound 23a (39 mg, 0.12 mmol) was dissolved in 3 mL anhydrous THF and 3 mL anhydrous MeOH. 1,4-cyclohexadiene (0.4 mL, 5.52 mmol, 46 eq.) and 10% Pd-C (68 mg) were added. After stirring at room temperature for 2 h, the reaction mixture was filtered through celite and solvents were removed in vacuo. The crude products 24a was directly used in next step. 24b was prepared in the same method and the same scale. Compound 5a and 5b were prepared by the general procedure for azide peptide coupling used 16 (43.1 mg, 0.11 mol, 1 eq.), 24a and 24b. Purification by preparative HPLC (50% \rightarrow 100 % MeCN/H₂O) yielded the title compound (5a, 8.1 mg, 14.9 μmol, 14% and **5b**, 9.4 mg, 17.2 μmol, 16%). **5a**: ¹H NMR (400 MHz, MeOD): & 7.26-7.17 (m, 5H), 4.98 (s, 2H), 4.39-4.35 (m, 1H), 4.28-4.25 (m, 1H), 4.06-4.02 (m, 1H), 2.67 (d, J= 5.2 Hz, 1H), 2.54-2.52 (m, 2H), 2.45 (d, J=5.2 Hz, 1H), 1.61-1.37 (m, 9H), 1.29 (s, 3H), 0.88-0.79 (m, 18H). ¹³C NMR (100 MHz, MeOD): 8 175.37, 174.62, 158.52, 138.17, 129.47, 129.02, 128.82, 67.68, 61.59, 55.74, 54.87, 53.23, 52.59, 46.71, 41.94, 41.80, 40.10, 25.91, 25.85, 25.64, 24.18, 23.41, 23.37, 21.97, 21.86, 21.38, 18.93. LC-MS (linear gradient $10 \rightarrow 90\%$ MeCN/H₂O, 0.1% TFA, 12.5 min): R_t (min): 8.22 (ESI-MS (m/z): 546.27, (M+H⁺)). **5b**: ¹H NMR (600 MHz, MeOD): δ 7.25-7.16 (m, 5H), 5.00-4.94 (m, 2H), 4.29-4.25 (m, 2H), 4.06-4.01 (m, 1H), 2.59-2.49 (m, 4H), 1.58-1.20 (m, 12H), 0.83-0.73 (m, 18H). ¹³C NMR (150 MHz, MeOD): δ 175.17, 174.55, 158.53, 138.14, 129.48, 129.02, 128.82, 67.69, 63.11, 56.90, 54.97, 53.39, 52.61, 47.91, 47.68, 41.93, 41.76, 40.49, 25.84, 25.74, 25.67, 24.10, 23.39, 23.35, 22.18, 22.02, 21.53, 18.75. LC-MS (linear gradient 10 \rightarrow 90% MeCN/H₂O, 0.1% TFA, 12.5 min): R_t (min): 8.22 (ESI-MS (m/z): 546.27, $(M+H^+)$). HRMS calculated for C₃₀H₄₈N₃O₆ 546.35376 [M+H]⁺; found 546.35375.

Benzyl ((S)-1-(((S)-1-(((S)-2,6-dimethyl-3-oxohept-1-en-4-yl)amin -o)-4-methyl-1-oxopentan-2-yl)amino)-4-methyl-1-oxopentan-2yl)carbamate (6)

Compound 25 was prepared according to the literature procedures, followed by the general procedure for azide peptide coupling using 16 (121.7 mg, 0.31 mol, 1 eq.) and compound 25 (92.0 mg. 0.34 mmol, 1.1 eq.). Purification by column chromatography (10% EtOAc/pentane \rightarrow 30% EtOAc/pentane) yielded the title compound

(130.0 mg, 0.25 mmol, 81%). ¹H NMR (400 MHz, CDCl₃): δ 7.35-7.29 (m, 5H), 6.90 (d, *J*=8.4 Hz, 1H), 6.65 (d, *J*=7.6 Hz, 1H), 6.11 (s, 1H), 5.90 (s, 1H), 5.51 (d, *J*= 8.4 Hz, 1H) 5.40-5.35 (m, 1H), 5.13-5.06 (m, 2H), 4.55 (m, 1H), 4.27 (m, 1H), 1.89 (s, 3H), 1.68-1.34 (m, 9H), 0.96-0.86 (m. 18H). ¹³C NMR (100 MHz, CDCl₃): δ 200.71, 172.19, 171.41, 156.34, 142.26, 136.34, 128.64, 128.28, 128.12, 126.65, 67.12, 53.52, 51.82, 51.26, 42.73, 41.59, 41.29, 25.05, 24.79, 23.47, 22.96, 22.80, 22.37, 22.24, 21.86, 17.92. LC-MS (linear gradient 10 → 90% MeCN/H₂O, 0.1% TFA, 12.5 min): R_t (min): 8.69 (ESI-MS (m/z): 516.20, (M+H⁺)). HRMS calculated for C₂₉H₄₆N₃O₅ 516.34320 [M+H]⁺; found 516.34302.

Benzyl ((S)-4-methyl-1-(((S)-4-methyl-1-(((S)-6-methyl-2,3-dioxo -heptan-4-yl)amino)-1-oxopentan-2-yl)amino)-1-oxopentan-2-yl) -carbamate (7)

Ozone was bubbled into a stirred solution of 6 (130 mg, 0.25 mmol) in DCM at -78 °C for 15 min until the blue colour of the saturated ozone solution persisted. The solution was then purged with oxygen until the solution became colourless. Triphenylphosphine (131.1 mg, 0.50 mmol, 2 eq.) was added after purging with argon for 5 min and the reaction was stirred overnight while warming up to room temperature. Purification by preparative HPLC $(50\% \rightarrow 100\%)$ MeCN/H₂O) yielded the title compound (18.5 mg, 35.8 µmol, 14%). ¹H NMR (400 MHz, CDCl₃): δ 7.38-7.31 (m, 5H), 6.93 (d, J=5.6 Hz, 1H), 6.54 (d, J=8.0 Hz, 1H), 5.32 (d, J=7.2 Hz, 1H), 5.14-5.11 (m, 2H), 4.93-4.90 (m, 1H), 4.47-4.42 (m, 1H), 4.18 (m, 1H), 2.35 (s, 3H), 1.68-1.43 (m, 9H), 0.93-0.87 (m. 18H) ¹³C NMR (100 MHz, CDCl₃): 8 196.86, 172.50, 171.91, 156.53, 136.08, 128.74, 128.48, 128.23, 67.42, 53.82, 52.37, 51.43, 41.21, 40.33, 39.67, 25.13, 24.84, 24.20, 23.26, 23.04, 22.91, 22.12, 22.05, 21.61. LC-MS (linear gradient 10 \rightarrow 90% MeCN/H₂O, 0.1% TFA, 12.5 min): R_t (min): 8.52 (ESI-MS (m/z): 518.13, (M+H⁺)). HRMS calculated for $[C_{28}H_{44}N_{3}O_{6} 518.32246 [M+H]^{+};$ found 518.32245.

Benzyl ((S)-4-methyl-1-(((S)-4-methyl-1-(((S)-6-methyl-3-methyle -ne-2-oxoheptan-4-yl)amino)-1-oxopentan-2-yl)amino)-1-oxopen -tan-2-yl)carbamate (8)

Compound 26 was prepared by the general procedure for Boc deprotection (compound 11, 52 mg. 0.19 mmol, 1.1 eq.), followed by the general procedure for azide peptide coupling used 16 (67.8 mg, 0.17 mol, 1 eq.). Purification by column chromatography (10% EtOAc/pentane \rightarrow 30% EtOAc/pentane) yielded the title compound (68.2 mg, 0.13 mmol, 78%).), ¹H NMR (400 MHz, CDCl₃): δ 7.33-7.27 (m, 5H), 6.72 (d, J=8.4 Hz, 1H), 6.42 (d, J=7.6 Hz, 1H), 6.00 (s, 1H), 5.92 (s, 1H), 5.13-5.05 (m, 2H), 4.79-4.73 (m, 1H), 4.22 -4.39 (m, 1H), 4.25-4.24 (m, 1H), 2.30 (s, 3H), 1.68-1.29 (m, 9H), 0.92-0.85 (m. 18H). ¹³C NMR (100 MHz, CDCl₃): δ 199.71, 172.34, 170.85, 156.35, 148.23, 136.26, 128.63, 128.28, 128.10, 126.87, 67.13, 53.63, 52.01, 50.34, 43.57, 41.56, 41.27, 25.35, 24.90, 23.10, 22.90, 22.81, 22.27, 22.06, 22.00. LC-MS (linear gradient 10 → 90% MeCN/H₂O, 0.1% TFA, 12.5 min): R_t (min): 8.43 (ESI-MS (m/z): 516.20, $(M+H^+)$). HRMS calculated for $C_{29}H_{46}N_3O_5$ 516.34320 [M+H]⁺; found 516.34316.

Competition assay in cell lysate

Lysates of HEK-293T cells were prepared by sonication in 3 volumes of lysis buffer containing 50 mM Tris pH 7.5, 1 mM DTT, 5 mM MgCl₂, 250 mM sucrose, 2 mM ATP, and 0.025% digitonin. Protein concentration was determined by the Bradford assay. Cell lysates (15 μ g total protein) were incubated with the inhibitors for 1 h at 37 °C prior to incubation with green-BODIPY-epoxomicin **27** (0.5 μ M each) for an additional 1 h at 37 °C, followed by 3 min boiling with a reducing gel-loading buffer and fractionation on 12.5% SDS-PAGE. In-gel detection of residual proteasome activity was performed in the wet gel slabs directly on a BioRad Imager using the Cy2/Fam settings (λ ex 488 nm, λ em 520 nm).

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References

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3.

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5.

6.

- a) K. L. Rock, C. Gramm, L. Rothstein, K. Clark, R. Stein, L. Dick, D. Hwang and A. L. Goldberg, *Cell*, 1994, **78**, 761-771. b) A. Hershko and A. Chiechanover, *Annu. Rev. Biochem.*, 1998, **67**, 651-664.
 - D. Voges, P. Zwickl and W. Baumeister, *Annu. Rev. Biochem.*, 1999, **68**, 1015-1068.
 - P.-M. Kloetzel, Nat. Rev. Mol. Cell. Biol., 2001, 2, 179-187.
 - S. Murata, K. Sasaki, T. Kishimoto, S.-I. Niwa, H. Hayashi, Y. Takahama and K. Tanaka, *Science*, 2007, **316**, 1349-1353.
 - See for recent reviews on proteasome inhibitors: a) A. F.
 Kisselev, W. A van der Linden and H. S. Overkleeft, *Chem. Biol.*, 2012, **19**, 99-115. b) A. Rentsch, D.
 Landsberg, T. Brodman, L. Bulow, A. K. Girbig and M.
 Kalesse, *Angew. Chem. Int. Ed.*, 2013, **52**, 5450-5488. c)
 E. M. Huber and M. Groll, *Angew. Chem. Int. Ed.*, 2012, **51**, 8708-9720.
 - a) O. A. O'Connor, A. K. Stewart, M. Vallone, C. J. Molineaux, L. A. Kunkel, J. F. Gerecitano and R. Z. Orlowski, *Clin. Cancer Res.*, 2009, **15**, 7085-8091. b) J. Adams, M. Behnke, S. Chen, A. A. Cruickshank, L. R. Dick, L. Grenier, J. M. Klunder, Y.-T. Ma, L. Plamondon and R. L. Stein, *Bioorg. Med. Chem. Lett.*, 1998, **8**, 333-

18.

338. c) P. G. Richardson, P. Sonneveld, M. W. Schuster, 17. D. Irwin, E. A. Stadtmauer, T. Facon, J. L. Harousseau, D. Ben-Yehuda, S. Lonial, H. Goldschmidt, D. Reece, J. F. San-Miguel, J. Bladé, M. Boccadoro, J. Cavenagh, W. S. Dalton, A. L. Boral, D. L. Esseltine, J. B. Porter, D. Schenkein and K. D. Anderson, N. Engl. J. Med. 2005, 352, 2487-2498.

- 7. S. D. Demo, C. J. Kirk, M. A. Aujay, T. J. Buchholz, M. Dajee, M. N. Ho, J. Jiang, G. J. Laidig, E. R. Lewis, F. Parlati, K. D. Schenk, M. S. Smyth, C. M. Sun, M. K. Vallon, T. M. Woo, C. J. Molineaux and M. K. Bennett, Cancer Res., 2007, 67, 6383-6391.
- 8. a) L. Meng, R. Mohan, B. H. Kwok, M. Elofsson, N. Sin and C. M. Crews, Proc. Natl. Acad. Sci. USA, 1999, 96, 10403-10408. b) L. H. Meng, B. H. B. Kwok, N. Sin and C. M. Crews, Cancer Res. 1999, 59, 2798-2801.
- 9. M. Groll, K. B. Kim, N. Kairies, R. Huber and C. M. Crews, J. Am. Chem. Soc., 2000, 122, 1237-1238.
- 10. See for instance: a) M. Verdoes, B. I. Florea, W. A. van der Linden, D. Renou, A. M. C. H. van den Nieuwendijk, G. A. van der Marel, H. S. Overkleeft, Org. Biomol. Chem., 2007, 5, 1416-1426. b) P. P. Geurink, W. A. van der Linden, A. C. Mirabella, N. Gallastegui, G. de Bruin, A. R. M. Blom, M. J. Voges, E. D. Mock, B. I. Florea, G. A. Van der Marel, C. Driessen, M. van der Stelt, M. Groll, H. S. Overkleeft and A. F. Kisselev, J. Med. Chem., 2013, 56, 1262-1275. c) W. A. van der Linden, L. I. Willems, T. B. Shabaneh, N. Li, M. Ruben, B. I. Florea, G. A. van der Marel, M. Kaiser, A. F. Kisselev and H. S. Overkleeft, Org. Biomol. Chem., 2012, 10, 181-194. d) M. Verdoes, L. I. Willems, W. A. van der Linden, B. A. Duivenvoorden, G. A. van der Marel, B. I. Florea, A. F. Kisselev and H. S. Overkleeft, Org. Biomol. Chem., 2010, 8, 2719-2727. e) H. J. Zhou, M. A. Aujay, M. K. Bennett, M. Dajee, S. D. Demo, Y. Fang, M. N. Ho, J. Jiang, C. J. Kirk, G. J. Laidig, E. R. Lewis, Y. Lu, T. Muchamuel, F. Parlati, E. Ring, K. D. Shenk, J. Shields, P. J. Schwonek, T. Stanton, C. M. Sun, C. Sylvain, T. M. Woo and J. Yang, J. Med. Chem., 2009, 52, 3028-3038.
- 11. M. A. Gräwert, N. Gallastegui, M. Stein, B. Schmidt, P. M. Kloetzel, R. Huber and M. Groll, Angew. Chem. Int. Ed., 2011, 50, 542-544.
- 12. M. L. Stein, H. Cui, P. Beck, C. Dubiella, C. Voss, A. Krüger, B. Schmidt and M. Groll, Angew. Chem. Int. Ed., 2014, 53, 1679-1683.
- 13. M. Bogyo, J. S. McMaster, M. Gaczynska, D. Tortorella, A. L. Goldberg and H. Ploegh, Proc. Natl. Acad. Sci. USA., 1997, 94, 6629-6634.
- 14. a) S. Mazzotta, L. Gramigna, L. Bernardi and A. Ricci, Org. Process Res. Dev., 2010, 14, 687-691; b) E. Bernacka, A. Klepacz and A. Zwierzak, Tetrahedron Lett. 2001, 42, 5093-5094.
- 15. W. A. van der Linden, P. P. Geurink, C. Oskam, G. A. van der Marel, B. I. Florea and H. S. Overkleeft, Org. Biomol. Chem., 2010, 8, 1885-1893.
- 16. WO Pat. 149 512, 2007.

M. Verdoes, U. Hillaert, B. I. Florea, M. Sae-Heng, M. D. P. Risseeuw, D. V. Filippov, G. A. van der Marel and H. S. Overkleeft, Bioorg. Med. Chem. Lett. 2007, 17, 6169-6171

M. Groll, C. R. Berkers, H. L. Ploegh and H. Ovaa, Structure, 2006, 14, 451-456.

19. In our work on β 5-specific peptide epoxyketones we noted that an S-configured peptide epoxyketone (with the S stereochemistry pertaining the chirality of the quartenary epoxide carbon) inhibited proteasomes with much decreased potency when compared to its R-configured counterpart (that is, the configuration encountered in the natural product, epoxomicin). See: M. Britton, M. M. Lucas, S. L. Downey, M. Screen, A. A. Pletnev, M. Verdoes, R. A. Tokhunts, O. Amir, A. L. Goddard, P. M. Pelphrey, D. L. Wright, H. S. Overkleeft and A. F. Kisselev, Chem. Biol., 2009, 16, 1278-1289.