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Tissue Cells Assisted Fabrication of Tubular Catalytic Platinum Microengines

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We report a facile platform for mass production of robust self-propelled tubular microengines. Tissue cells extracted from fruits of banana and apple, *Musa acuminate* and *Malus domestica*, are used as the support on which thin platinum film is deposited by means of physical vapor deposition. Upon sonication of the cells/Pt-coated substrate in water, microscrolls of highly uniform sizes are spontaneously formed.

¹⁰ Tubular microengines fabricated with the fruit cells assisted method exhibit fast motion of ~100 bodylengths/s (~1 mm/s). Extremely simple and affordable platform for mass production of the micromotors is crucial for the envisioned swarms of thousands and millions of autonomous micromotors performing biomedical and environmental remediation tasks.

Introduction

- ¹⁵ As a leading-edge nanotechnology, self-propelled micro-/nanoscale motors represent a fundamental step towards practical tiny machines.¹⁻⁵ These autonomous nano- and microsized devices are expected to perform a wide variety of tasks in biomedical field,^{6, 7} environmental remediation⁸⁻¹⁰ and natural
- ²⁰ resources discovery.¹¹ Self-propulsion at the nanoscale is achieved mainly by the mechanisms of self-electrophoresis^{12,13}, self-diffusiophoresis^{14, 15} or bubble ejection.¹⁶⁻¹⁸ The last mode of propulsion is of the greatest interest as it exhibits the strongest propulsion power and unprecedented velocity.^{19, 20} Bubble-²⁵ propelled micromotors are typical of microtubes with catalys^{17, 21}
- on the inner surface. Hydrogen peroxide fuel is decomposed into oxygen and the resulting bubbles are expelled from one end of the microtube, thus propelling it in the opposite direction.
- The fabrication of tubular micro/nanomotors is still a highly ³⁰ challenging task. The rolled-up nanotechnology²² includes topdown photolithography, angled e-beam evaporation and stressassisted rolling of nanomembranes upon etching away of the photoresist. The practical utility of this method is greatly hindered by the complicated fabrication process and high costs of
- ³⁵ related clean-room conditions. An alternative method of fabricating tubular micro/nanomotors is anodic aluminum oxide or polymer membrane-templated electrodeposition,^{19, 23} but this technique requires highly trained personnel and the cost of templates is not negligible. In addition, toxic electroplating
- ⁴⁰ solutions containing cyanide are usually needed in the electrodeposition process.Recently, there has been an intense interest in developing cheap
 - and easy plant-derived approaches for fabrication of motors. Millimeter sized biocatalytic motors based on plant tissues²⁴ and
- ⁴⁵ magnetically driven helical micromotors harnessing spiral waterconducting vessels of different plants²⁵ have been reported. Here,

we demonstrate an extremely simple and low-cost fabrication method of tubular microengines with the assistance of fruit cells. The use of cheap and widely accessible fruit cells instead of ⁵⁰ lithographically defined material as the support of metal layers omits the etching step and significantly reduces the fabrication costs. The resulting microtubes display ultrafast motion in the presence of hydrogen peroxide. The new fruit cells assisted method allows for rapid and cost-effective fabrication of ⁵⁵ thousands of copies of tubular microengines using banana and apple fruit tissue cells.

Results and Discussion

To prepare the tubular microengines, the aqueous dispersion of ⁶⁰ banana fruit cells was first prepared by homogenization of mashed banana tissue in water via ultrasonication. The homogenized dispersion of banana fruit cells was subsequently deposited on the glass cover slip. After complete evaporation of water, Pt was deposited on the glass cover slip coated with ⁶⁵ banana cells by sputtering. Upon ultrasonication of the cells and Pt coated cover slip in water, the multilayer film shattered and detached from the glass cover slip, rolling into scrolls. The whole fabrication process is illustrated in Scheme 1.



Scheme 1. Preparation of rolled-up tubular microengines using banana fruit cells: (a) deposition of banana fruit cells on the glass cover slip, (b) deposition of Pt layer by sputtering, (c) ⁵ ultrasonication of the cells/Pt-coated cover slip.

Figure 1 shows optical images of aqueous dispersion of banana fruit cells and the fabricated tubular microengines. As can been seen from Figure 1A, the banana fruit cells are several hundred ¹⁰ micrometers in length and are flat in shape. After the evaporation of water, the cells deposited on the cover slip became a layer of membrane that would detach from the surface upon coming into contact with water. There are no microtubes observed before and after deposition of Pt layer, where only some cracks of the ¹⁵ deposition layers emerged. However, microtubes were immediately obtained upon ultrasonication of the cells/Pt-coated glass cover slip in water, as shown in Figure 1B.





The SEM images of the resulting microtubes are presented in Figure 2 A and B. The microtubes exhibit excellent size ²⁵ uniformity and the rolled-up structure is clearly visible. A size distribution analysis of the obtained microtubes was carried out, as shown in Figure 2C. The majority of the fabricated tubular microengines are 10-15 μm long and the average opening diameters at opposite ends are around 2.5 μm and 3 μm, ³⁰ respectively. The uniformity of the tubular microengines fabricated by the new method is competitive with that of microtubes prepared by membrane-templated electrodeposition and rolled-up technology.²⁶



³⁵ **Figure 2.** (A, B) SEM images of the fabricated tubular microengines. Scale bars are 10 μ m and 1 μ m, respectively. (C) Length distribution of the tubular microengines, *n*=65.

EDX mapping analysis of the fabricated tubular microengines 40 (Figure 3) clearly shows the presence of platinum, carbon and oxygen, demonstrating the existence of banana fruit tissue in the microtube structure. For the fabrication of rolled-up microstructures, the strain in the film creates the bending moment²⁷ and strain relaxation is the driving force of the rolling process.²⁸ The strain in the fruit cells/Pt bilayer mainly comes ⁵ from two aspects. The Pt film deposited by sputtering is in a state of intrinsic stress, because of the accumulating effect of the crystallographic flaws that are built up during the deposition process.²⁹ The intrinsic stress is related to the deposition conditions and properties of the substrate surface.^{29, 30} In addition,

¹⁰ the different swelling properties of banana fruit cells and metal layer in water result in unequal changes in volume, thus generating swelling-induced strain.^{27, 28} The intrinsic stress of platinum film and the swelling-induced strain together lead to the formation of the fruit tissue cells/Pt microscrolls.



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Figure 3. SEM/EDX elemental characterization of the tubular microengines composition. Scale bar: $10 \ \mu m$.

Control experiment to prove the necessity of the banana fruit cells ²⁰ for successful formation of the microtubes was carried out. One half of the glass cover slip was covered with banana fruit cells while the other half of the glass was left bare. After deposition of Pt layer and ultrasonication in water, the Pt layer still covered the bare glass surface while the Pt layer and banana tissue cells on

- ²⁵ the other side detached from the glass cover slip upon ultrasonication(Figure 4, A, right and left side, respectively). The banana fruit cells provide an easily detachable interface between the metal layer and the glass substrate, which allows the instantaneous formation of microscrolls upon ultrasonication in
- ³⁰ water and eliminates the etching step using other chemical solutions. Additionally, we wish to show that banana fruit cells serve only as an example - other fruit cells can be employed to assist the preparation of tubular microengines as well. Figure 4B shows the optical image of the microtube fabricated using apple ³⁵ fruit cells.



Figure 4. (A) Optical image of the cover slip half coated with banana fruit cells (right side) after deposition of Pt and ultrasonication in water. Platinum is still on the uncoated cover ⁴⁰ slip after ultrasonication (left side). Scale bar: 50 μm (B) Optical image of tubular microengines fabricated using apple fruit cells. Scale bar: 20μm.

The tubular microengines fabricated with the fruit cells assisted 45 method exhibit high propulsion power in the presence of hydrogen peroxide fuel. Figure 5A shows a typical snapshot of the motion of the prepared microengines in 3% H₂O₂. The trajectory can be easily visualized from the oxygen bubble tail released from one end of the microtube. The speed of the tubular 50 microengines can be modulated by varying the concentration of hydrogen peroxide. As shown in Figure 5B, the average velocity of the fruit cells/Pt microtube engines increases from ~410 µm/s in 2% H_2O_2 to ~680 µm/s in 4% H_2O_2 Some microengines show ultrafast motion, up to 1mm/s in 3% H₂O₂ (Video S1; Supporting 55 Information). The mobility of the tubular microengines prepared with the introduced method is highly competitive with that of microengines fabricated by electrodeposition¹⁹ and rolled-up technology.²⁰ In addition, the very low cost of the fabrication process using common fruits, such as bananas and apples, offers 60 possibilities for mass production of micromotors. Thousands of microtube engines can be prepared with only milligrams of fruit tissue.



Figure 5. (A) Snapshot of the motion of a microtube engine in 3% H₂O₂ and 0.5% SDS. Scale bar: 50 µm. (B) Dependence of microjets' speed on H₂O₂ concentration in the presence of 0.5% ⁵ SDS. Tracking data was obtained from 15 independent running experiments in order to obtain the average speed.

Conclusions

- In summary, we demonstrated a very affordable and simple route ¹⁰ for fabrication of rolled-up tubular microengines, using fruit tissue cells as support of the deposited metal layer. Given the cheap price of fruits and small amounts needed for the preparation process, the cost of fabrication is only cut down to the consumption of Pt used for sputtering. One can envision the use
- ¹⁵ of the other soft tissues for assisted fabrication of microjets. Microtube engines fabricated by this method exhibited good uniformity in the dimensions and excellent mobility, up to ~100 bodylengths/s. The fruit tissue cells assisted fabrication process opens doors for mass fabrication of microscale devices at ²⁰ extremely low cost.

Experimental Section

Materials

Hydrogen peroxide (35%, lot no. 10172592) was purchased from

 $_{25}$ Alfa Aesar. Sodium dodecyl sulfate (SDS, lot no. 079K0335) was purchased from Sigma-Aldrich. VFM cover slips (22×22 mm, lot no. GP110220001) were purchased from Cellpath, UK. The platinum and silver targets for sputtering were purchased from Quorum Technologies Ltd, UK. Bananas and apples were ³⁰ purchased from local markets. The chemicals were used as received and ultrapure water (18.2 M Ω cm) from a Millipore Milli-Q purification system was used throughout the experiments.

Apparatus

- ³⁵ The ultrasonication process was carried out with a Fisherbrand FB 11203 ultrasonicator. Sputtering was carried out with JEOL JFC-1600 Auto Fine Coater. Scanning electron microscopy (SEM/EDX) analysis was obtained with a JEOL JSM 7600F instrument. Optical microscope images and videos were obtained
- ⁴⁰ with a Nikon Eclipse 50i microscope. Video sequences were processed with Nikon NIS-Elements software.

Methods

Preparation of microjets The cover slips were cleaned with water and dried with nitrogen gas. Fruit tissue (banana or apple, 500 45 mg) was dispersed in water (2 mL) by ultrasonication and 50 μL of the aqueous suspension was applied on the cover slip to spread over the whole surface. The cover slip was left over night to dry and platinum (15 nm) was sputtered on the cover slip. The cover slip was ultrasonicated subsequently in water and the aqueous 50 suspension of tubular microengines can be obtained.

Propulsion of microjets The experiments for the motion study of the fabricated tubular microengines were carried out in an aqueous solution containing different concentrations of hydrogen peroxide at a constant surfactant concentration (0.5 wt % of SDS). A mixture of microjets, SDS, H₂O₂ and water was applied on a glass slide freshly cleaned with nitrogen gas. Optical microscope videos and images were obtained with Nikon Eclipse 50i microscope. Video sequences were processed with Nikon NIS-ElementsTM software.

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Notes and references

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† Electronic Supplementary Information (ESI) available: Related video.

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¹ Sengupta, S.; Ibele, M. E.; Sen, A.; Angew. Chem. Int. Ed. 2012, 51, 8434-8445.

² G. Zhao, M. Pumera, *Chem. Asian J.* **2012**, *7*, 1994.

³ Wang, J. ACS Nano 2009, 3, 4-9.

⁴ Campuzano, S.; Kagan, D.; Orozco, J.; Wang, J. Analyst **2011**, 136, 4621-4630.

5 S. Sanchez, M. Pumera, Chem. Asian J. 2009, 4, 1402.

⁶ Wang, J.; Gao, W. ACS Nano 2012, 6, 5745-5751.

⁷ Wang, J. Lab Chip **2012**, 12, 1944-1950.

8 Gao, W.; Wang, J. ACS Nano 2014, 8, 3170-3180.

⁹ Soler, L.; Sanchez, S. Nanoscale 2014, DOI: 10.1039/c4nr01321b.

¹⁰ a) Zhao, G.; Seah, T. H.; Pumera, M. Chem. Eur. J. 2011, 17, 12020-12026. b) T. H. Seah, G. Zhao, M. Pumera, ChemPlusChem 2013, 78, 384.

¹¹ Sen, A.; Ibele, M.; Hong, Y.; Velegol, D. *Faraday Discuss.* **2009**, *143*, 15-27.

¹² Wang, Y.; Hernandez, R. M.; Bartlett, D. J.; Bingham, J. M.; Kline, T. R.; Sen, A.; Mallouk, T. E. *Langmuir* **2006**, *22*, 10451-10456.

¹³ M. Pumera, *Nanoscale* 2011, *2*, 1643.
¹⁴ Baraban, L.; Makarov, D.; Streubel, R.; Mönch, I.; Sanchez, S.; Schmidt, O. G. *ACS Nano* 2012, *6*, 3383-3389.

¹⁵ Gao, W.; Pei, A.; Dong, R.; Wang, J. J. Am. Chem. Soc. **2014**, 136, 2276-2279.

¹⁶ Solovev, A. A.; Mei, Y. F.; Ureña, E. B.; Huang, G.; Schmidt, O. G. Small 2009, 5, 1688-1692.

¹⁷ Solovev, A. A.; Sanchez, S.; Pumera, M.; Mei, Y. F.; Schmidt, O. G. *Adv. Funct. Mater.* **2010**, *20*, 2430-2435.

¹⁸ H. Wang, G. Zhao, M. Pumera, J. Am. Chem. Soc. 2014, 136, 2719.

¹⁹ Gao, W.; Sattayasamitsathit, S.; Orozco, J.; Wang, J. J. Am. Chem. Soc. **2011**, *133*, 11862-11864.

²⁰ Sanchez, S.; Ananth, A. N.; Fomin, V. M.; Viehrig, M.; Schmidt, O. G. J. Am. Chem. Soc. **2011**, *133*, 14860-14863.

²¹ Sanchez, S.; Solovev, A. A.; Mei, Y. F.; Schmidt, O. G. J. Am. Chem. Soc. **2010**, *132*, 13144.

18 Mei, Y.; Huang, G.; Solovev, A. A.; Ureña, E. B.; Mönch, I.; Ding, F.; Reindl, T.; Fu, R. K. Y.; Chu, P. K.; Schmidt, O. G. *Adv. Mater.* **2008**, *20*, 4085-4090.

²³ Zhao, G.; Pumera, M. RSC Adv. 2013, 3, 3963-3966.

²⁴ Gu, Y.; Sattayasamitsathit, S.; Kaufmann, K.; Vazquez-Duhalt, R.; Gao, W.; Wang, C.; Wang, J. Chem. Commun. **2013**, 49, 7307-7309.

²⁵ Gao, W.; Feng, X.; Pei, A.; Kane, C. R.; Tam, R.; Hennessy, C.; Wang, J. *Nano Lett.* **2014**, *14*, 305-310.

²⁶ Zhao, G.; Ambrosi, A.; Pumera, M. J. Mater. Chem. A **2014**, 2, 1219-1223.

²⁷ Luchnikov, V.; Sydorenko, O.; Stamm, M. *Adv. Mater.* **2005**, *17*, 1177-1182.

²⁸ Huang, M.; Cavallo, F.; Liu, F.; Lagally, M. G. *Nanoscale* **2011**, *3*, 96-120.

²⁹ Thornton, J. A. Thin Solid Films 1989, 171, 5-31

³⁰ Golod, S. V.; Prinz, V. Y. Appl. Phys. Lett. 2004, 84, 3391-3393.