Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this Accepted Manuscript with the edited and formatted Advance Article as soon as it is available.

You can find more information about *Accepted Manuscripts* in the **Information for Authors**.

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard <u>Terms & Conditions</u> and the <u>Ethical guidelines</u> still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.



www.rsc.org/foodfunction

1				
1		,		
	ļ		1	
		9		
		ł	1	
	i	Ì		k
	l			
	(
	(1		Ś
Ì				
Ĩ				
	(C		5
	1		1	ζ
	1		ļ	2
1				2
		0		
	(Ľ)
		ì	1	í
			1	2
		S		
	e		ſ	
	1			
	(
1	ŝ			
1			1	
		9		
		ς		
	i			
	1			
l		l		
C	5	5		
1				
	(_		ķ
	(Ì	1	5
,				
l				

Food & Function Impact of boiling on phytochemicals and antioxidant activity of green vegetables 1 2 consumed in the Mediterranean diet 3 Ana F. Vinha^{1,2}, Rita C. Alves^{1,3}, Sérgio V. P. Barreira², Anabela S. G. Costa¹, M. 4 Beatriz P.P. Oliveira^{1*} 5 6 ¹REQUIMTE/Departamento de Ciências Químicas, Faculdade de Farmácia da 7 Universidade do Porto, Rua de Jorge Viterbo Ferreira n.º 228, 4050-313 Porto, Portugal. 8 ²FCS-UFP/Faculdade de Ciências da Saúde, Universidade Fernando Pessoa, Rua Carlos 9 10 da Maia n.º 296, 4200-150 Porto, Portugal. ³REQUIMTE/ Instituto Superior de Engenharia do Porto, Instituto Politécnico do Porto, 11 12 Rua Dr. António Bernardino de Almeida n.º 431, 4200-072 Porto, Portugal. 13 14 *Corresponding author: 15 M. Beatriz P.P. Oliveira ¹REQUIMTE/ Departamento de Ciências Químicas, Faculdade de Farmácia da 16 Universidade do Porto, Rua de Jorge Viterbo Ferreira n.º 228, 4050-313 Porto, Portugal. 17 Tel: + 351 220 428 640; Fax: + 351 226 093 390. 18 beatoliv@ff.up.pt 19 20 21

22 Abstract

The effect of boiling (10 minutes) on eleven green vegetables frequently consumed in the Mediterranean diet was evaluated. For that, some physicochemical parameters and the contents of vitamin C, phenolics and carotenoids, as well as the antioxidant activity, were determined in raw and boiled samples.

27 The raw vegetables analysed in this study were good sources of vitamin C, carotenoids 28 and phenolic compounds, with contents ranging from 10.6 to 255.1 mg/100 g, 0.03 to 29 3.29 mg/100 g and 202.9 to 1010.7 mg/100 g, respectively. Boiling promoted losses in 30 different extensions considering both the different bioactive compounds and the distinct 31 vegetables analysed. Contrary to phenolics (more resistant), vitamin C was the most 32 affected compound. Boiling also originated significant losses in the antioxidant activity of the vegetables. Considering all the parameters analysed, the vegetables most affected 33 34 by boiling were broccoli and lettuce. The least affected ones were collard and tronchuda cabbage. 35

36

- 37 Keywords:
- 38 Green vegetables, boiling, carotenoids, phenolics, ascorbic acid, antioxidant activity.
- 39

40

41

42

43

44 Introduction

A characteristic of the so-called Mediterranean diet, recently recognized by UNESCO 45 as Intangible Cultural Heritage, is the inclusion of a large variety of vegetables in the 46 dishes. This behavior has, undoubtedly, a positive impact on the population's health. It 47 has already been scientifically established that the ingestion of large amounts of 48 vegetables reduces the risk of chronic injuries, particularly, cardiovascular diseases, 49 50 cancers, cataract and macular degeneration, obesity and type 2 diabetes, and several degenerative disorders.¹⁻⁵ Inevitably, all these effects are related to the chemical 51 composition of the consumed vegetables. Besides being rich in water and fiber and poor 52 in fat and carbohydrates, many raw vegetables are also naturally rich in minerals, 53 vitamins, and phytochemicals (carotenoids and phenolic compounds). Another feature 54 of the Mediterranean diet is the minimal processing of foods. Vegetables are mostly 55 ingested *in natura*, especially in salads or after being boiled in water. The main purpose 56 of cooking is to make vegetables more edible, palatable and digestible. Additionally, by 57 eliminating potential pathogens and reducing the intake of some anti-nutrients, boiling 58 contributes for the safety of these foods.⁶ The downside is that cooking may adversely 59 affect the levels of nutrients and bioactive compounds, especially the heat-sensitive and 60 water soluble ones. Although some studies reported the improvement of in vivo 61 bioavailability of some phytochemicals when vegetables are cooked^{7,8} many others 62 pointed out significant losses of minerals,⁹ vitamins,¹⁰ total soluble proteins and soluble 63 sugars,¹¹ carotenoids^{12,13} and phenolic compounds.^{14,15} 64

Nevertheless, the available data about boiling effects on the contents of vitamin C, carotenoids and phenolic compounds, as well as on the antioxidant activity, of green vegetables traditionally consumed in the Mediterranean diet are still scarce. The aim of this research was, therefore, to develop a comprehensive study about this topic, enabling a comparative analysis of 11 selected vegetables, encompassing 4 different genera.

Food & Function Accepted Manuscript

70

71 Results and Discussion

72 *Raw vegetables*

73 Chemical characterization of the eleven green vegetables analyzed in this study, is presented in **Table 1**. Our results reveal that all the vegetables are quite similar in terms 74 75 of moisture content (ranging from 96.2 to 98.0 %) and pH (values varied between 6.1 and 7.1). According to Morales et al.,¹⁶ higher moisture contents provide higher tender 76 and succulent properties in leafy vegetables. The presented results are in agreement with 77 those authors, since lettuce stands out with the highest moisture value (98.0 g/ 100 g) 78 79 and cabbage showed the lowest one (96.2 g/ 100 g). In what concerns to total soluble solids (TSS), the values are more dissimilar, ranging from 1.16 to 6.89 °Brix. It is 80 important to note that the highest and lowest values were determined in broccoli 81 82 samples (rabe buds and rabe leaves, respectively). A similar behavior can be described for ash content, except that lettuce presented the lowest ash value (0.66 %), reflecting a 83 comparatively lower mineral content. In turn, spinach presented the highest ash content 84 85 (1.31 %).

The similarity in terms of physicochemical parameters contrasts with the diversity observed relatively to the contents of the analyzed bioactive compounds. Vitamin C levels of the analysed vegetables were quite variable. Lettuce presented the lowest content (10.6 mg/ 100 g), while collard displayed the highest one (255 mg/ 100 g).

The variability in terms of carotenoids was even larger: cauliflower had only 0.03 mg/ 100 g while spinach presented a 100-fold higher value (**Table 1**). Phenolic compounds were the most representative phytochemicals (from 203 mg/ 100 g for cauliflower and 1011 mg/ 100 g for spinach). Notably, all vegetables presented significant differences in these bioactive compounds. **Figure 1** depicts the clusters classification of the selected

vegetables, considering the physicochemical parameters and contents of bioactive compounds, which reflects the described diversity.

97 The most similar samples, that constitute a well-defined cluster, are broccoli, broccoli 98 rabe leaves and broccoli rabe buds, all presenting analogous amounts of carotenoids and 99 phenolic compounds. The collard is closer to watercress than to the other Brassicaceae 100 because both have higher contents of vitamin C, carotenoids and phenolic compounds. 101 The cauliflower distinguishes itself from the other vegetables due to the lowest values 102 of natural pigments (carotenoids and phenolic compounds). Spinach is also a vegetable 103 apart because it has an exceptionally high carotenoid content.

104

95

96

105 Boiled vegetables

Consumers are aware to ingest a variety of vegetables and maximize the intake of 106 107 beneficial antioxidants. However, in Mediterranean gastronomy, vegetables are often boiled to become more edible. It is known that antioxidant composition and 108 bioavailability of vegetables are greatly affected by cooking methods.^{14,15} According to 109 Table 1, some of the raw vegetables are excellent sources of bioactive compounds. 110 111 However, cooking can drastically influence the content and bioavailability of phytochemical compounds. In this study, 10 minutes of boiling had a minor influence in 112 their pH and ash contents. The first parameter remained statistically unaltered (p > 0.05) 113 114 for all. Only the broccoli rabe leaves and watercress experienced a significant decrease (p < 0.05) in their ash contents (-10.6 % and -17.8 %, respectively). TSS values were 115 116 more affected by boiling, indicating the leaching of soluble sugars and organic acids 117 into the water. Just five of the vegetables experienced significant reductions of TSS (broccoli rabe buds -20.6 %; cabbage -22.1 %; cauliflower -30.5 %; spinach -23.3 %; 118 and watercress -16.9 %). 119

In the other hand, the boiling effect on the contents of bioactive compounds seemed to be somewhat detrimental. **Figure 2** shows significant losses of these compounds for all the studied vegetables after boiling. Another feature that becomes apparent is a different extension loss for the various bioactive compounds and, for a given compounds class, it depended upon the vegetable. Thermal degradation and leaching into the boiling water are expected to be the primarily responsible for the verified decreases, explaining the high losses observed in this and in similar studies for other foods.¹⁷

Overall the major loss occurs in the vitamin C content with 77.7 % decrease in spinach.
Somsub and colleagues¹⁸ reported losses in ascorbic acid content between 24 % and 95
% in 13 selected Thai vegetables subjected to boiling for just 4 minutes.

The carotenoid degradation may reach 40 %, in the case of lettuce, savoy cabbage, and 130 broccoli. Carotenoids are a class of lipophilic compounds, less susceptible to leaching 131 and also less heat sensitive than vitamin C. There are even studies referring the 132 bioavailability enhancement of some carotenoids with cooking.^{7,8} This can be attributed 133 to cell walls disruption with food processing procedures, facilitating their liberation 134 from proteins. But this behavior can vary with the food matrix, being reported increases 135 and decreases of these bioactive compounds.¹⁹ For instance, Chang, Prasad and Amin¹² 136 studied the effect of different domestic cooking methods on carotenoids retention in 7 137 138 commonly consumed leafy Malaysian vegetables. The authors concluded that 8 minutes 139 of boiling imply large variations of lutein retention (from 0 to 418 %) and β -carotene 140 (from 18 to 380 %). In a study with five tropical leafy vegetables from Africa, Djuikwo et al.²⁰ also recorded losses of total carotenoids from 5 to 20 % after 10 minutes boiling. 141 Kao and colleagues¹³ evaluated the effect of boiling in various carotenoid-rich green 142 leafy vegetables, including Thai basil leaves and cilantro, and noted that total 143 carotenoids content reached the maximum after boiling those vegetables for 5 minutes 144

Food & Function Accepted Manuscript

and 10 minutes, respectively. A negative effect on the total carotenoids contents of thevegetables was noticed with more boiling time.

147 Comparatively, total phenolic losses were smaller; nonetheless, they can account to 148 nearly 30% in the case of lettuce and 20 % in broccoli (**Figure 2**). Watercress, spinach, 149 savoy cabbage and the broccoli rabe leaves did not undergo substantial losses of 150 phenolic compounds (p > 0.05).

Turkmen et al.²¹ studied the effect of 5 minutes boiling on the contents of total phenolics of spinach and boccoli from Turkey and measured losses of about 6 % for the latter vegetable. Mazzeo and collaborators²² also detected losses of about 4 % when the spinach was boiled for 10 minutes. Although they measured a reduction of approximately 31 % in the phenolics content of cauliflower, a value quite superior to that recorded here (~ 14 %) it that may be correlated with differences on the chemical composition of the vegetable (edaphoclimatic, cultivation and post-harvest conditions).

As noted above, the effect of cooking depends on the type of vegetable.²³ This has been 158 mainly related with the fact that the morphology of the cells and organelles containing 159 the various phytochemicals differs among vegetables.²⁴ However, the remaining 160 chemical makeup and structure of the vegetables should also play a determinant role. 161 For example it is known that the leaves of collard and *tronchuda* cabbage are covered 162 163 by a relatively thick epicuticular waxy layer which may provide an additional barrier 164 reducing the wettability and the mass and heat transfer, thus hampering the leaching of 165 the compounds during boiling. Probably this is one of the reasons why such vegetables present lower losses than, for instance, lettuce. Anyway, a cluster analysis taking into 166 account the losses of bioactive compounds reported above, suggests that the studied 167 168 vegetables can be divided into two groups (Figure 3): a first one, comprising collard, tronchuda cabbage, cauliflower, savoy cabbage and watercress, i.e., vegetables that lost 169

Food & Function Accepted Manuscript

nearly half of the vitamin C and did not undergo significant losses in phenolic
compounds; and another group of vegetables that suffered substantial losses of all
bioactive compounds (cabbage, broccoli rabe leaves, broccoli rabe buds, broccoli and
lettuce). Spinach was not grouped, probably due to the higher decrease in vitamin C
contents.

175

176 *Effect of boiling on the antioxidant potential*

177 The antioxidant activity (AA) is a parameter that measures the combined effect of all antioxidants in preventing the harmful action of free radicals. Owing to the very 178 different chemical nature of the antioxidants, the activity should be assessed by 179 180 complementary methods. In this study, the DPPH[•] radical scavenging activity and the β -carotene linoleate model assays were chosen for that purpose. Among all the tested 181 vegetables, savoy cabbage, spinach and collard greens presented an exceptionally high 182 percentage of antioxidant potential (> 80 %). A moderate antioxidant potential (50-70 183 %) was shown by watercress and *tronchuda* cabbage, while cauliflower and lettuce 184 presented the lowest free radical scavenging activities (< 30 %) (Figure 4). 185

According to **Figure 4**, boiling can be associated with losses in the antioxidant activity 186 of the vegetables commonly used in the Mediterranean diet, reaching nearly 58 % in the 187 antioxidant capacity measured by the β -carotene bleaching (broccoli rabe buds) and 36 188 % against the DPPH[•] radical (cauliflower). These values are similar to those already 189 reported by other authors. For example, Faller and Fialho²⁵ in a study with vegetables 190 registered radical scavenging capacity variations between -93 % and +16 %. It should 191 be noted that the antioxidant activity may not be directly correlated with the total 192 concentration of antioxidant compounds, due to synergistic or antagonistic effects.²⁶ 193 The correlation results (Figure 5) have revealed that AA is directly related to the 194

phenolic compounds content (Pearson's $r_{\text{DPPH}} = 0.941$, Pearson's $r_{\text{BCL}} = 0.484$) and 195 moderately correlated with carotenoids content (Pearson's $r_{\text{DPPH}} = 0.399$, Pearson's r196 $_{BCL} = 0.580$) and vitamin C (Pearson's $r_{DPPH} = 0.394$; Pearson's $r_{BCL} = 0.224$). These 197 findings suggest that the reduction in antioxidant potential might be primarily related 198 with the depletion of phenolic compounds and secondly of carotenoids compounds. 199 200 Accordingly, vegetables that lost considerable amounts of phenolic and carotenoids 201 compounds (cauliflower, lettuce and broccoli rabe buds) also exhibit substantial 202 reduction in their antioxidant activity.

203

204 Experimental

205 Standards and reagents

206 2,6-Dichlorophenol-indophenol (Tillman's reagent), ascorbic acid, metaphosphoric acid, 207 acetone, petroleum ether, sodium carbonate, ethanol, 2,2-diphenyl-1-picrylhydrazyl 208 radical (DPPH[•]), β -carotene, chloroform, and Tween 40 emulsifier were all obtained 209 from Sigma-Aldrich (St. Louis, MO, USA). The Folin-Ciocalteu reagent, gallic acid, 210 and linoleic acid were purchased from Panreac Química S.L.U. (Barcelona, Spain). 211 Ultrapure water from a Simplicity 185 system (resistivity 18.2 M Ω .cm; Millipore, 212 Belford, USA) was used for all aqueous solutions preparation.

213

214 Samples and sample preparation

According to the high cultural relevance in the Mediterranean diet, eleven green vegetables (**Table 2**) where chosen to perform this study. They were all obtained from local markets, in the district of Porto, Portugal.

About 2 kg of each vegetable were sampled. In the same day, each vegetable wasprepared, cooked and submitted to the extraction process. All the material was washed

in running water and dried with absorbing paper before random sampling in batches of ~ 500 g. The edible parts of the vegetables (leaves and blooms) were cut into small pieces.

All batches were divided into two equal portions, one maintained raw and the other submitted to boiling. The boiling procedure was performed by adding each vegetable sample to boiling tap water (~ 100 °C) in a covered stainless-steel pot (1:5 vegetable/ water) and letting it there for exactly 10 minutes. This period was selected as the minimum cooking time needed for adequate sample palatability and taste. Afterwards the boiled samples were drained and dried carefully with absorbing paper prior to further analysis.

Before physicochemical and phytochemicals evaluation, samples (raw and cooked)
were homogenized in a grinder (Grindomix GM 200, Retsch, Haan, Germany).

Aqueous extracts were obtained and used to determine some parameters, like phenolics content and antioxidant capacity. In these cases, samples (~ 5 g) were stirred with 100 mL of water, at 25 °C, for 1 h, protected from light, and the solids separated from the extract by vacuum filtration. The extracts were stored at -20 °C prior analysis.

236

237 Physicochemical characterization

Moisture and ash contents, pH and total soluble solids (TSS) were determined in all samples. A gravimetric assay was performed to evaluate the moisture content. Samples (~5 g) were dried in a stove (WTC binder Klasse 2.0, Tuttlingen, Germany) at 105 °C \pm 1 °C, followed by regular weighing up to a constant weight. Results were expressed as water percentage (%). The mineral content was evaluated by incineration at 450 °C and results expressed in percentage (%). The pH value was measured in triplicate with a pHmeter (Microprocessor pH Bench-top HI 8417, Hanna Instruments). TSS were 10

245	quantified with an Atago, NAR-3T refractometer, properly adjusted and calibrated at 20
246	°C with distilled water and the results expressed as °Brix (an approximate value of total
247	sugar content).

248

249 Determination of antioxidant compounds

250 **Determination of vitamin C**

Ascorbic acid content was determined according to a previously described method.²⁷ Briefly, samples were mixed with metaphosphoric acid (0.1 g/ L) for 45 min at room temperature, and filtered through Whatman n° 4 filter paper. Then, 1 mL of filtrate was mixed with 2,6-dichlorophenol-indophenol and absorbance was measured at 515 nm. A calibration curve of ascorbic acid (linearity range: 0.5-100 μ g/ mL, r = 0.9994) was prepared and the results were expressed in mg of ascorbic acid per 100 g of fresh sample.

258

259 Total carotenoids content

Total carotenoids content was determined according to Wang and Liu²⁸ with some 260 modifications. Each vegetable (raw and boiled) was submitted to a previous extraction 261 before quantification. Briefly, 5 g of homogenized sample was added to 50 mL of 262 263 petroleum ether/acetone mixture (1:1, v/v), wrapped with aluminum foil and subjected to constant shaking during 30 min, at room temperature (25 °C). After filtration, 3 mL 264 of supernatant were collected and absorbance was measured at 445 nm. The total 265 carotenoids content (mg/ 100 g of fresh sample) was determined according to the 266 following equation: 267

Food & Function Accepted Manuscript

Total carotenoids (mg/ 100g) = $(A \ge y \ge 10^6)/(A_{1cm}^{\%} \ge 1000 \ge w)$, where *A* represents the absorbance of the extract at 445 nm; *y* is the volume of extract (mL); $A_{1cm}^{\%}$ represents the extinction coefficient of carotenoids ($A_{1cm}^{\%} = 2592$), and w is the sample weight (g).

271

272 Determination of total phenolic content

Total phenolic contents were determined according to Costa and colleagues.²⁹ Aliquots of 0.5 mL of extract were added to Folin-Ciocalteu reagent (2.5 mL, previously diluted with water 1:10 v/v) and 2 mL of a sodium carbonate solution (7.5 % m/v). After 30 min incubation at room temperature, absorbance readings were performed at 765 nm. Total phenolics were quantified by means of a calibration curve of gallic acid (linearity range = 2-200 μ g/ mL, r = 0.9989) and results expressed in mg per 100 g of fresh weight.

280

281 Antioxidant activity

282 **DPPH[•]** radical-scavenging activity

This assay was evaluated according to Vinha et al.³⁰ in a microplate reader (BioTek 283 Synergy HT, GENS5). The aqueous extracts (300 µL) were added to 2.7 mL of an 284 ethanolic DPPH[•] solution (6 x 10^{-5} mol/ L). The mixture was vigorously stirred and 285 286 absorbance determined at 515 nm, until a stable plateau was reached. DPPH[•] scavenging activity (RSA) was determined as the percentage of DPPH[•] discoloration 287 using the following equation: % RSA = $[(A_{DPPH} - A_S)/A_{DPPH}] \times 100$, where A_s 288 289 represents the absorbance of the sample with DPPH[•] and A_{DPPH} is the absorbance of the DPPH[•] solution. 290

292 Inhibition of β -carotene bleaching

 β -carotene bleaching inhibition by neutralization of linoleate free radicals was also 293 evaluated according to Vinha et al.²⁷ A solution of β-carotene (2 mg in 10 mL of 294 chloroform) was prepared and then, 2 mL pipetted into a 100 mL round-bottom flask. 295 296 After the chloroform removal under vacuum (40 °C), ~40 mg of linoleic acid, 400 mg of 297 Tween 40 emulsifier, and 100 mL of distilled water were added and vigorously shaken. 298 5 mL of this emulsion were transferred into different test tubes containing 1 mL of sample extracts. Immediately after adding the emulsion to the test tubes, the zero time 299 absorbance was read at 470 nm. The tubes were incubated in a water bath at 50 °C. 300 301 Measurement of absorbance continued until complete β -carotene bleaching and the antioxidant activity was obtained by the following equation: (β -carotene content after 2 302 h of assay/initial β -carotene content) x 100. 303

304

305 Statistical analysis

Statistical analysis was performed using SPSS v. 21 (IBM Corp., Armonk, NY, USA). 306 Data of all analysis, in triplicate, are expressed as mean \pm standard deviation. After 307 validating the assumptions of multivariate normality and homogeneity of variance-308 309 covariance, a MANOVA analysis was used to compare different vegetables. Whenever 310 the MANOVA analysis detected significant effects, an ANOVA, for each of the parameters experimentally determined, was performed followed by Tukey's HSD post-311 312 hoc test. The vegetables were grouped into homogeneous groups through a hierarchical 313 Cluster analysis by the method of least distance (Nearest Neighbor) using the squared 314 Euclidean distance as the measure of dissimilarity. A 0.05 significance level was 315 considered for all tests. *p*-values inferior to 0.05 were considered to be statistically316 significant.

Mean comparison between raw and boiled vegetables was made through independent samples t-test. Pearson correlation tests were used to ascertain the existence of linear relationships between the contents of bioactive compounds and antioxidant activity.

320

321 CONCLUSIONS

322 Upon boiling (10 min), the vegetables analysed in this study suffered considerable losses of vitamin C and carotenoids. In what concerns to phenolic compounds, the 323 324 decrease content was not so substantial and generalized. In addition to the intrinsic properties of the compounds, boiling effect seemed to be influenced by the nature of the 325 vegetable matrix, probably due to differences in chemical composition and 326 cellular/organelle structure. Globally, the vegetables most affected were broccoli rabe 327 leaves and buds, broccoli and lettuce. The least affected were collard and tronchuda 328 cabbage. A cluster analysis taking into account the losses of bioactive compounds 329 evidences two groups of vegetables. Collard, tronchuda cabbage, cauliflower, savoy 330 331 cabbage and watercress formed a group that lost nearly half of the vitamin C and did not undergo significant losses of phenolic compounds. The other group included vegetables 332 333 that lost substantial amounts of all bioactive compounds (cabbage, broccoli rabe leaves, 334 broccoli rabe buds, broccoli and lettuce). Since the antioxidant potential was directly 335 correlated with the contents of bioactive compounds, particularly with total phenolics, this parameter was also significantly reduced upon boiling, especially with cauliflower, 336 337 lettuce and broccoli rabe buds. Our results may be useful to consumers on the choice of 338 each vegetable, considering their losses after boiling. Moreover, this study improves the consumption of all vegetables, which may provide benefits to health. 339

340	Acknowledgements:
341	R. Alves is grateful to FCT for a post-doc grant (SFRH/BPD/68883/2010) financed by
342	POPH-QREN and subsidized by FSE and MCTES. This work received financial
343	support from the European Union (FEDER funds through COMPETE) and National
344	Funds (FCT) through project Pest-C/EQB/LA0006/2013, as well as from FEDER funds
345	under the framework of QREN through Project NORTE-07-0124-FEDER-000069.
346	
347	The authors state that there are no conflicts of interest.
348	
349	
350	
351	
352	
353	
354	
355	
356	
357	
358	
359	

360 REFERENCES

- 1. H. Boeing, A. Bechthold, A. Bub, S. Ellinger, D. Haller, A. Kroke, E. Leschik-
- Bonnet, M. J. Müller, H. Oberritter, M. Schulze, P. Stehle and B. Watzl, *Eur J Nutr*, 2012, **51**, 637-663.
- 364 2. T. Tanaka, M. Shnimizu and H. Moriwaki, *Molecules*, 2012, **17**, 3202-3242.
- 365 3. M. A. Asgar, *Int J Food Prop*, 2013, **16**, 91-103.
- 366 4. D. J. Williams, D. Edwards, I. Hamernig, L. Jian, A. P. James, S. K. Johnson
 367 and L. C. Tapsell, *Food Res Int*, 2013, **52**, 323-333.
- 368 5. A. P. Q. Larrosa, T. R. S. Cadaval Jr. and L. A. A. Pinto, LWT *Food Sci*369 *Technol*, 2015, **60**, 178-185.
- 370 6. N. O. A. Ilelaboye, I. A. Amoo and O. O. Pikuda, *Arch Appl Sci Res*, 2013, 5,
 371 254-260.
- 372 7. S. Aherne, T. Daly, M. Jiwan, L. O'Sullivan and N. O'Brien, *Food Res Int*,
 373 2010, 43, 1449–1454.
- M. Pasaporte, F. Rabaya, M. Toleco and D. Flores, *Food Chem*, 2014, **158**, 35 40.
- 376 9. Z. Lisiewska, P. Gębczyński, E. Bernaś and W. Kmiecik, *J Food Comp Anal*,
 377 2009, 22, 218-223.
- 378 10. E. Lešková, J. Kubíková, E. Kováčiková, M. Košická, J. Porubská and K.
 379 Holčíkova, *J Food Comp Anal*, 2006, 19, 252-276.
- 380 11. F. Xu, Y. Zheng, Z. Yang, S. Cao, X. Shao and H. Wang, *Food Chem*, 2014,
 381 161, 162-167.
- 382 12. S. K. Chang, N. K. Prasad and I. Amin, *Int Food Res J*, 2013, **20**, 457-465.

- 384 13. F. J. Kao, Y. S. Chiu, M. J. Tsau and W. D. Chiang, LWT *Food Sci Technol*,
 2012, 46, 485-492.
- M. Francisco, P. Velasco, D. A. Moreno, C. García-Viguera and M. E. Cartea, *Food Res Int*, 2010, 43, 1455-1463.
- 388 15. V. Perla, D. G. Holm and S. S. Jayanty, *LWT Food Sci Technol*, 2012, 45, 161389 171.
- P. Morales, I. C. F. R. Ferreira, A. M. Carvalho, M. C. Sánchez-Mata, M.
 Cámara, V. Fernández-Ruiz, M. Pardo-de-Santayana and J. Tradío, *LWT Food Sci Technol*, 2014, 55, 389-396.
- 393 17. A. Rawson, A. Patras, B. K. Tiwari, F. Noci, T. Koutchma and N. Brunton, *Food Res Int*, 2011, 44, 1875-1887.
- W. Somsub, R. Kongkachuichai, P. Sungpuag and R. Charoensiri, *J Food Comp Anal*, 2008, **21**, 187-197.
- 397 19. C. Arnold, U. Schwarzenbolz and V. Böhma, *LWT Food Sci Technol*, 2014, 57,
 398 442-445.
- 20. V. N. Djuikwo, R. A. Ejoh, I. Gouado, C. M. Mbofung and S. A Tanumihardjo,
 Food Nutr Sci, 2011, **2**, 793-802.
- 401 21. N. Turkmen, F. Sari and Y. S. Velioglu, *Food Chem*, 2005, **93**, 713-718.
- 402 22. T. Mazzeo, D. N'Dri, E. Chiavaro, A. Visconti, V. Fogliano and N. Pellegrini,
 403 *Food Chem*, 2011, **128**, 627-633.
- 404 23. D. C. Murador, D. T. Cunha and V. V. Rosso, *Food Res Int*, 2014, **65**, 177-183.
- 405 24. R. M. Schweiggert, D. Mezger, F. Schimpf, C. B. Steingass and R. Carle, *Food*406 *Chem*, 2012, **135**, 2736-2742.
- 407 25. A. L. K. Faller and E. Fialho, *Food Res Int*, 2009, **42**, 210-215.

- 408 26. E. Sanjust, G. Mocci, P. Zucca and A. Rescigno, *Nat Prod Res*, 2008, 22, 689409 708.
- 410 27. A. F. Vinha, R. C. Alves, S. V. P. Barreira, A. Castro, A. S. G. Costa and M. B.

411 P. P. Oliveira, LWT - *Food Sci Technol*, 2014, **55**, 197-202.

- 412 28. L. Wang and Y. Liu, *Nat Sci*, 2009, **1**, 23-29.
- 413 29. A. S. G. Costa, R. C. Alves, A. F. Vinha, S. V. P. Barreira, M. A. Nunes, L. M.
 414 Cunha and M. B. P. P. Oliveira, *LWT Food Sci Technol*, 2014, **53**, 350-357.
- 415 30. A. F. Vinha, S. V. P. Barreira, A. S. G. Costa, R. C. Alves and M. B. P. P.
- 415 30. A. F. Vinha, S. V. P. Barreira, A. S. G. Costa, R. C. Alves and M.
 416 Oliveira, *Food Chem Toxicol*, 2014, 67, 139-144.

433	Figure Captions
434	
435	Figure 1. Dendrogram resulting from a cluster analysis of the studied vegetables,
436	considering the physicochemical parameters and contents of bioactive compounds.
437	
438	Figure 2. Contents of bioactive compounds of the different vegetables before and after
439	boiling. [*] indicates significant differences ($p < 0.05$).
440	
441	Figure 3. Overall similarity of the vegetables in terms of bioactive compounds loss.
442 443	
444	Figure 4. Antioxidant activity (A.A.) of vegetables, before and after boiling, by the β -
445	carotene linoleate model system (β CL) and on DPPH [•] radical scavenging activity. *
446	indicates significant differences ($p < 0.05$) caused by boiling.
447	
448	Figure 5. Pearson correlation analysis between the antioxidant activity (β CL and
449	DPPH) and the bioactive contents (vitamin C, total carotenoids and phenolics). *
450	indicates that the correlation is significant at the 0.05 level.
451	
452	
453	
454	
455	

		Physicochemica	l Parameters		Bi	ioactive Compou	unds
Fresh Vegetable	Moisture (%)	рН	TSS (°Brix)	Ash (%)	Vitamin C (mg/100g)	Carotenoids (mg/100g)	Phenolics (mg/100g)
Broccoli	97.3±0.1 ^{a,b,c,d}	$6.37{\pm}0.06^{d,e,f}$	$3.09{\pm}0.06^{d}$	0.97 ± 0.02^{c}	49.6±1.0 ^h	$1.11{\pm}0.01^{f}$	455.9±1.6 ^h
Broccoli rabe buds	96.7±0.2 ^{c,d,e}	$6.36{\pm}0.10^{d,e,f}$	6.89±0.05 ^a	$1.04{\pm}0.03^{b,c}$	100.8±1.6 ^{d,e}	$1.09{\pm}0.01^{f}$	554.3 ± 0.8^{f}
Broccoli rabe leaves	97.7±0.4 ^{a,b,c}	6.52±0.11 ^{c,d,e,f}	1.16±0.07 ^g	$0.80{\pm}0.04^{d,e}$	115.0±7.9 ^{b,c}	1.21±0.01 ^e	403.6±1.6 ⁱ
Cabbage	96.2±0.3 ^e	$6.50{\pm}0.08^{c,d,e,f}$	$2.04{\pm}0.06^{f,e}$	$0.95{\pm}0.07^{c,d}$	90.8±3.4 ^{e,f}	$0.41{\pm}0.01^{i}$	561.8±0.9 ^e
Cauliflower	96.9±0.4 ^{b,c,d,e}	6.82±0.03 ^{a,b,c}	3.97±0.08 ^c	0.95±0.02 ^{c,d}	69.1±1.7 ^g	$0.03{\pm}0.01^{j}$	202.9 ± 0.6^{k}
Collard	96.5±0.3 ^{d,e}	6.13 ± 0.02^{f}	2.98±0.03 ^d	0.96±0.03 ^{c,d}	255.1±9.7 ^a	$2.50{\pm}0.07^{b}$	747.8±0.6 ^c
Tronchuda cabbage	96.5±0.6 ^{d,e}	6.93±0.10 ^{a,b}	2.20±0.13 ^e	$1.04{\pm}0.08^{b,c}$	105.8±8.0 ^{d,c}	1.39±0.01 ^d	608.9±0.9 ^d
Lettuce	98.0±0.4 ^a	$6.43 \pm 0.42^{d,e,f}$	$1.83{\pm}0.05^{f}$	0.66±0.03 ^e	10.6±0.9 ⁱ	$0.80{\pm}0.01^{h}$	393.0±1.5 ^j
Savoy cabbage	96.4±0.2 ^{d,e}	$6.37{\pm}0.03^{d,e,f}$	3.95±0.10 ^c	1.05±0.05 ^{b,c}	70.7±1.4 ^g	0.97±0.01 ^g	816.8±0.8 ^b
Spinach	97.8±0.3 ^{a,b}	7.10±0.04 ^a	$2.02{\pm}0.07^{f,e}$	1.31±0.04 ^a	40.8±1.6 ^h	3.29±0.05 ^a	1010.7±1.1ª
Watercress	96.4±0.6 ^{d,e}	$6.63 \pm 0.02^{b,c,d,e}$	$1.89{\pm}0.04^{\rm f}$	1.18±0.08 ^{a,b}	122.3±1.6 ^b	1.98±0.02 ^c	502.7±1.4 ^g

Table 1. Physicochemical parameters and bioactive compounds of 11 types of raw vegetables.

457 *Values expressed as mean \pm standard deviation obtained from 3 measurements per replicate. Within each column, different superscript letters represent significant differences 458 between samples (p < 0.05).

Table 2. List of the 11 samples selected for this study, identified by their common and scientific names, respectively.

Vegetables				
Common name Scientific name				
Broccoli	Brassica oleracea var. italica			
Broccoli rabe buds	Brassica rapa var. rapa			
Broccoli rabe leaves	Brassica rapa var. rapa			
Cabbage	Brassica oleracea var. capitata			
Cauliflower heads	Brassica oleracea var. botrytis			
Collard	Brassica oleracea var. acephala			
Tronchuda cabbage	Brassica oleracea L. var. costata DC			
Lettuce	Lactuca sativa var. latina			
Savoy cabbage	Brassica oleracea var. sabauda L.			
Spinach	Spinacia oleracea			
Watercress	Nasturtium officinale			

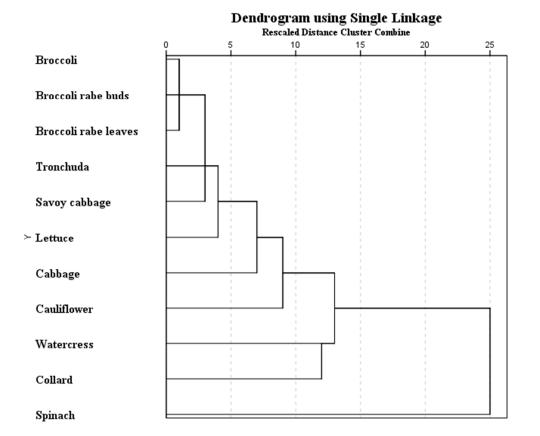


Figure 1.

Food & Function Accepted Manuscript

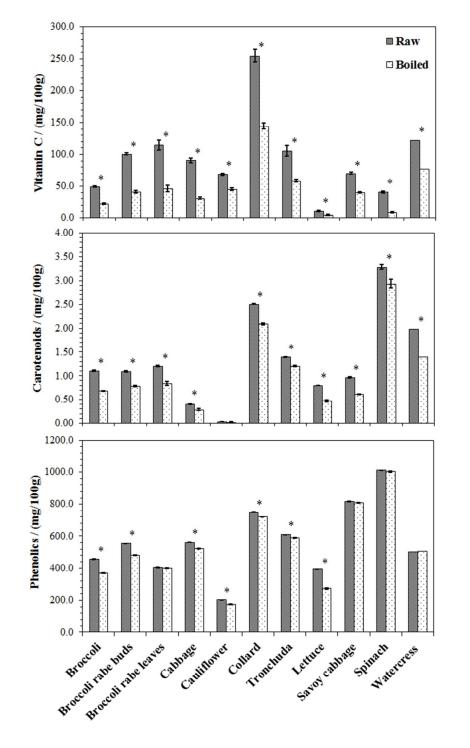


Figure 2.

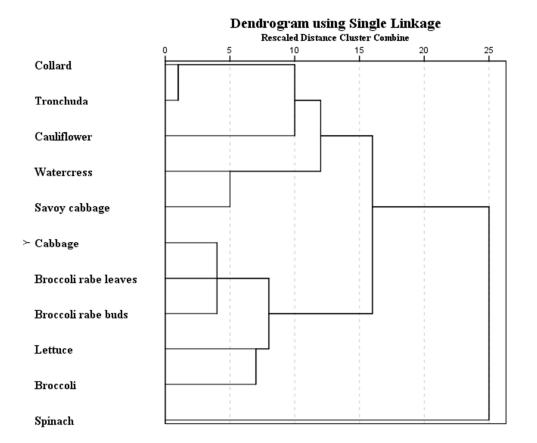


Figure 3.

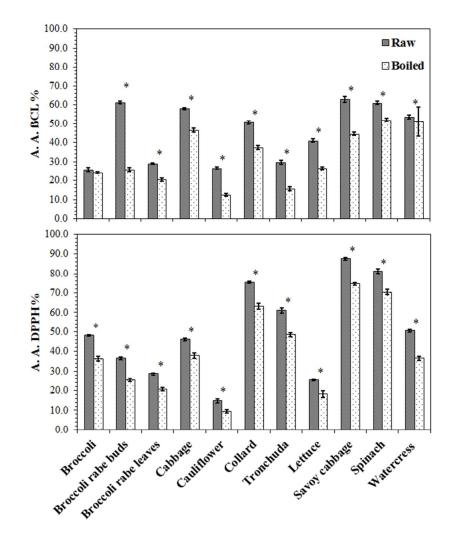


Figure 4.

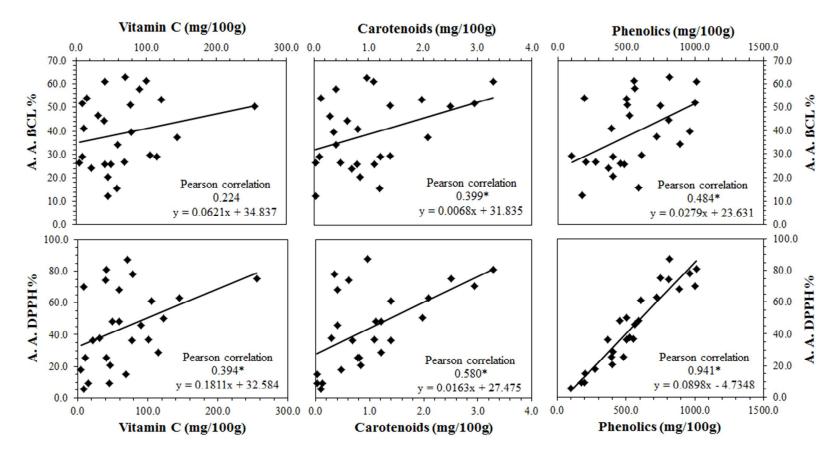


Figure 5.

TOC graphic/Graphical abstract

