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Absorption of Caffeine in Fermented Pu-er tea is Inhibited in Mice

Ye-wei Huang^{1,2,3}, Huan-huan Xu^{2,3}, Su-min Wang^{2,3}, Yi Zhao^{1,2,3}, Yu-min Huang^{2,3}, Run-bo Li^{2,3}, Xuan-jun Wang^{2,3*}, Shu-mei Hao^{4*} and Jun Sheng^{2,3*}

¹College of Life Science, Jilin University, Changchun, 130012, People's Republic of China.

²Key Laboratory of Pu-erh Tea Science, Ministry of Education, Yunnan Agricultural University, Kunming, 650201, People's Republic of China.

³Tea research Center of Yunnan, Kunming, 650201, People's Republic of China.

⁴Yunnan University, Kunming, 650091, People's Republic of China.

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1 Abstract

Caffeine present in number of dietary sources consumed worldwide. 2 Although its pharmacokinetics has been intensively explored, little is known 3 about complexed caffeine (C-CAF) in aqueous extraction of fermented Pu-er 4 tea. The major components of C-CAF are oxidative tea polyphenols (OTP) and 5 caffeine. Furthermore, the C-CAF can be precipitated in low pH solution. After 6 7 administrating the same amount of total caffeine and comparing the peak level of plasma caffeine with the coffee (contains 0.11 ± 0.01 % C-CAF) group, the 8 9 results showed that the caffeine/OTP (contains 66.67 ± 0.02 % C-CAF) group and the instant Pu-er tea (contains 23.18 ± 0.02 % C-CAF) group were 33.39 10 % and 25.86 % lower, respectively. The concentration of the metabolites of 11 12 caffeine supports the idea that the absorption of the C-CAF was inhibited in mice. Congruent with this result, the amount of caffeine detected in mice 13 excrement showed that more caffeine was eliminated in the caffeine/OTP 14 group and the Pu-er tea group. The locomotor activity tests of mice 15 demonstrated that the caffeine/OTP group and Pu-er tea group were more 16 peaceful than coffee group. Our findings demonstrated that caffeine can be 17 combined with OTP and the absorption of C-CAF is inhibited in mice, thus 18 decreasing the irritation effect of caffeine. This may also be developing as a 19 slow release formulation of caffeine. 20

Key words: absorption, caffeine, complexed caffeine, fermented Pu-er
 tea, oxidative tea polyphenols

23 Introduction

Caffeine (1, 3, 7-trimethylxanthine), a naturally occurring plant xanthine 24 alkaloid¹, was first isolated in the eighteenth century². Caffeine is present in a 25 number of dietary sources consumed worldwide (i.e., coffee, tea, soft drinks -26 cola, cocoa beverages, and chocolate bars)³⁻⁵. About 87 % of the world's 27 population consumes caffeine with an average of 193 mg per day. Among 28 adults 18 years and older, the daily intake ranges between 166 and 336 mg per 29 day^{1,6,7}. Caffeine is considered as a stimulant of the central nervous system for 30 its ability to enhance alertness and can inhibit lipid absorption in 31 ovariectomized rats^{2,4,8}. The efficacy and side effects of caffeine have been 32 extensively studied^{3,6,9-12}, caffeine is generally recognized as safe by the Food 33 and Drug Administration. However, excessive caffeine use can result in 34 serious health problems and, in rare cases, death^{13,14}. 35

One of the most common side effects of caffeine consumption is sleep 36 loss. Some reported experiments about the negative effects of caffeine on 37 sleep have consistently found that caffeine shortens total sleep time, prolongs 38 sleep latency, and changes the duration of light and deep sleep¹⁵. Sleep loss is 39 a major reason that people are reluctant to consume caffeinated beverages, 40 namely tea and coffee. Fermented Pu-er tea is a unique fermented tea 41 produced in Yunnan province of China¹⁶, it has been consumed by Chinese 42 people for centuries and has recently become more popular functional 43 beverage in Asian, Europe and America^{17,18}. A mass of oxidative tea 44

45 polyphenols (OTP) in fermented Pu-er tea are produced during the 46 fermentation process^{19,20}. During the stage of fermentation, caffeine content 47 has been found to increase ²¹. Contrary to assumption, it is believed that the 48 influnce of fermented Pu-er tea on sleep is significantly less in comparison with 49 other kinds of tea or coffee. Unlike coffee or other kinds of tea, many people 50 can enjoy Pu-er tea before sleep.

The above discrepancy laid the groundwork for us to investigate the 51 effects of caffeine in fermented Pu-er tea. We previously found that caffeine in 52 fermented Pu-er tea can be divided into two kinds as complexed caffeine 53 (C-CAF) and free caffeine (F-CAF), and that the C-CAF content can be 54 evaluated by Low-pH Precipitate Method²². In this study, our primary objective 55 was to compare the content of C-CAF in different kinds of tea and coffee 56 samples, and then probe into the material basis of C-CAF formation. On 57 account that mouse stomach acidity has a pH of about 2.0, we postulated that 58 caffeine from Pu-er tea can be precipitated with other components in stomach 59 when administered orally. Resultantly, the absorption of caffeine in Pu-er tea 60 may be unavoidably influenced. To test this hypothesis, our second objective 61 was to investigate the difference between the absorption of caffeine in coffee, 62 fermented Pu-er tea and OTP. Our experiment results demonstrated that: one, 63 fermented Pu-er tea has a larger amount of C-CAF than non-fermented Pu-er 64 tea, green tea, black tea, oolong tea and coffee; two, C-CAF was principally 65 combined with OTP; and three, the absorption of C-CAF in fermented Pu-er 66

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tea was inhibited in mice and the effect of caffeine was influenced significantly.
Our findings provide a theoretical basis for the phenomenon of the mitigation
insomnia effects of fermented Pu-er tea in comparison with other kinds of tea
or coffee.

71 Materials and methods

Chemicals and reagents. Caffeine was purchased from Aladdin. 72 Theobromine was purchased from Shanghai Yuanye biological technology co., 73 LTD. Paraxanthine, theophylline and sodium dihydrogen phosphate were 74 75 purchased from Sigma-Aldrich. 8-chlorotheophylline was purchased from Tokyo chemical industry co., LTD. Methanol used in the mobile phases were of 76 HPLC-grade reagent purchased from Tedia Co. Inc.. Deionized water was 77 prepared using the Purifier (FST-UV-20, Shanghai Fushite instrument 78 equipment Co., Ltd. Shanghai, China) and was degassed by ultrasonic 79 cleaning machine (SK-18TC, Shanghai Kudos ultrasonic instrument Co., Ltd. 80 Shanghai, China). 81

Samples and preparation of HPLC infusion. Fourteen Chinese tea samples of various types, including three fermented Pu-er tea, two Non-fermented Pu-er tea, three black tea, two oolong tea, and four green tea, which were made into aqueous extract powder. Moreover, three coffee samples, one instant essence of Pu-er tea (Deepure, Yunnan tasly deepure biological tea group co., LTD.), one OTP, one TPS, one Pu-er tea theabrownin (PTTB), and one Pu-er tea polysaccharide (PTPS) sample were also made into aqueous

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extract power. Supporting information of all samples is listed in **Table 1**. The preparation process of OTP, TPS, PTPS and PTTB was showed in the "Supplementary Experimental Procedures" of Electronic Supplementary Information.

Preparation of HPLC infusion: 800mg of the sample was dissolved in 40 mL ultrapure water and was shocked by Vortex Mixer for 2 min and then was centrifuged at 800 ×g for 10 min. The resulting supernatant was diluted 10-folds and filtered through a 0.45µm membrane and subjected to HPLC to determine the content of caffeine and its metabolites of the sample. In addition, the 20 g/L sample solution would be used for Low-pH Precipitate Method and Chloroform Extraction as described below.

100 HPLC-VWD analysis. A 50 µL sample was analyzed using an autosampler (G1329B, 1260ALS, Agilent, USA), a ultraviolet detector (G1314F, 1260VWD, 101 Agilent, USA) at wavelength 274 nm, and HPLC pump (G1311B, 1260Quat 102 Pump, Agilent, USA) at 0.6 mL/min flow at 30°C (G1316A, 1260TCC, Agilent, 103 USA) through a C18 ODS column (ZORBAX SB-C18 4.6mm × 250mm, 5 104 Micron, Agilent, USA) with a guard column (ZORBAX Eclipse Plus-C18, 105 4-Pack, Analytical Guard Column, 4.6mm×12.5mm, 5-Micron, Agilent, USA). 106 The mobile phase was 73 % 0.1 mol/L sodium dihydrogen phosphate and 27 107 % methanol, filtered through a 0.45 micron filter. Concentrations of caffeine 108 and its metabolites were determined in mg/L for all of the samples³. 109

110 Determination of C-CAF content of the samples by Low-pH Precipitate

111 *Method.* The C-CAF content of the samples was evaluated by Low-pH 112 Precipitate Method as previously described²³. In brief, the poly-phenolic 113 hydroxyl contained components were precipitated under acidic conditions (pH 114 \leq 2) and then dissolved with alkaline solution. Conveniently, if caffeine was 115 combined with the Poly-phenolic hydroxyl contained components, the C-CAF 116 content of the samples can be determined in the precipitate dissolved solution 117 by HPLC.

Sample preparation: Hydrochloric acid was added to 12mL of 20 g/L 118 119 sample solution for adjusting acidity to pH 1.5 and was shocked by Vortex Mixer for 2min and centrifuged at 1600 \times g at 25 $^{\circ}$ C for 30 min. The precipitate 120 and supernatant were separated and the precipitate was weighed by sensitive 121 balance. The precipitate was dissolved NaOH solution (pH = 10). Then, the 122 supernatant and precipitate solution were diluted with deionized water to 13mL. 123 The caffeine content of the supernatant and precipitate was determined by 124 HPLC. 125

126 Checkout of C-CAF content of the samples through chloroform 127 extraction. The linear regression equation of the balance line of caffeine 128 distribution between water and chloroform was y = 15.02x at 25 °C. X and y 129 mean that caffeine concentration (g/L) of aqueous phase and chloroform 130 phase respectively when the distribution achieves equilibrium. The correlation 131 coefficient R = 0.9942. It has no significant effect on the caffeine extraction 132 between pH = 2.28 ~ 11.37 when x < 0.6 g/L²⁴. To checkout content of C-CAF

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of the samples, we determined caffeine distribution ratio of the samples 133 between chloroform phase and aqueous phase. 134 Sample preparation: 5mL chloroform was added to 5 mL of 20 g/L sample 135 solution and was shocked by Vortex Mixer for 2 min and centrifuged at 800 ×g 136 at 25 °C for 15 min. Then, the aqueous phase solution was diluted 10 folds with 137 deionized water. The caffeine content was determined by HPLC. 138 *Mice.* Healthy virgin inbred BALB/c mice (7–8 weeks old) composed of male 139 and female in equal numbers from Nanjing Peng-sheng biotechnology Co., Ltd, 140 141 China, were used in the animal experiments. Mice were maintained in a controlled environment (12 h light/12 h dark cycle; humidity 50-60 %; ambient 142 temperature 24°C ± 1°C) and were administered standard laboratory food and 143 144 water ad libitum. All mice experiments were performed in the animal facility

according to institutional guidelines and were approved by the
Institutional Animal Care and Use Committee of Yunnan Agricultural University.
Adverse events were not observed.

Group designations and caffeine administration. The BALB/c mice composed of male and female in equal numbers were divided at random into coffee group, fermented Pu-er tea group, and caffeine added OTP group. Coffee (85.91 g/L), fermented Pu-er tea (31.77 g/L) and OTP (31.77 g/L) with additional caffeine were prepared in deionized water (the caffeine content of the three samples are equivalent to 2.5 g/L) and administered via intragastric administration (0.2 mL/10g). The caffeine dose (50 mg/kg) was selected based

on its pharmacokinetic profile in mice compared with humans, which is well below the lethal dose of approximately 10 g in humans ¹¹. Caffeine doses were also based on the finding that 50 mg/kg of caffeine in rodents corresponds to approximately 10 cups of coffee for humans, but caffeine is metabolized up to 6 times faster in mice than in humans^{5,25}. The mice were feed-deprived for 12 hours before the intragastric administration.

Levels of caffeine and its metabolites in plasma. Levels of caffeine and its 161 metabolites in plasma were determined by HPLC as described by Youngberg³ 162 et al. with modifications. In brief, the blood samples were taken at various 163 intervals (5, 10, 20, 40, 60, 120, 240, 360, 480 min) after the caffeine 164 administration. As the internal standard, 3.33 mg/L of 8-chlorotheophylline was 165 added to each sample. 150 µl samples of subject plasma were extracted in 1 166 mol/L HCl and 3 mL methylene chloride. The organic layer was evaporated to 167 dryness and reconstituted in 0.1 mol/L HCl solution. Plasma levels of caffeine 168 and its metabolites were determined by HPLC with the same chromatographic 169 condition as "HPLC-VWD analysis" part. For caffeine, paraxanthine, 170 theophylline and theobromine in plasma: the limit of quantitation (mg/L) was 171 0.05, 0.10, 0.105 and 0.035; the limit of detection (mg/L) was 0.025, 0.045, 172 0.060, 0.015; the response factor was 1.005, 0.456, 0.514 and 0.906; and the 173 overall extraction recovery from human plasma was 85.28 ± 2.31 %, 57.13 ± 174 0.87 %, $60.10 \pm 0.39 \%$ and $72.84 \pm 0.87 \%$, respectively. The recovery for the 175 internal standard (8-chlorotheophylline) was 90.89 ± 1.07 %. The precision and 176

accuracy for the analytes were within the acceptable range (< 7 %). The assay
validation was showed in the "supplemental data" of Electronic Supplementary
Information.

Excrement sample extraction. The excrement samples were taken 8 hours after the caffeine administration, and then dried and ground to powder. 5 mL methanol (70 %) was added to a 200 mg powder sample, shocked by Vortex Mixer for 2min, and then extracted by ultrasound-assisted for 30 min at room temperature. Finally, the caffeine concentration was determined by HPLC.

Evaluation of locomotor activity. Locomotor activity was monitored with a 185 mouse activity monitor system (ZZ-6, Taimeng, Chengdu Technology & Market 186 Co. Ltd., Chengdu, China), which monitored the horizontal (locomotion) and 187 188 vertical (rearing) movements of the mice. The individual compartments (L=15; W=12; H=10 cm) were put in a dimly lit and quiet room. The mice were 189 feed-deprived for 12 hours before the experiment. Groups of female BALB/c 190 mice were first thoroughly habituated to the test environment over a 30 min 191 period. Then, they were removed from the open field, administrated with coffee, 192 fermented Pu-er tea, and caffeine added OTP aqueous solution (which were 193 equivalent to caffeine at 5 mg/kg) or water respectively via intragastric 194 administration (0.2 mL/10g), and replaced in the compartments for an 195 additional 90 min. The 5 mg/kg test dose of caffeine was chosen on the basis 196 of previous dose-effect studies performed in our laboratory. This dose is a little 197 higher than the threshold of stimulant effect of caffeine for the female mice. 198

Locomotor activity was recorded during the 90 min following the caffeine administration. The floor and walls of the chamber were cleaned with ethanol (70%) and dried with paper towels between each mouse exposure²⁶⁻²⁸.

Statistical analyses. All values were presented as mean ± the standard error
of the mean (SEM). Differences within groups were analyzed with repeated
measures one-way ANOVA, Two-tailed p < 0.05 was considered to be
statistically significant. All analyses were performed using SPSS 17.0 (Chicago,
IL, USA) and GraphPad Prism 5 (GraphPad Software, Inc., La Jolla, CA, USA).

207 **Results**

C-CAF content of the samples. Previous research has shown that tea 208 contains theobromine and theophylline in addition to caffeine²⁹. Therefore, in 209 the current study, we set out to observe plasma levels of caffeine and its 210 metabolites, by initially evaluating the content of caffeine, theophylline, 211 theobromine and paraxanthine of the samples (Supplemental Table 1). There 212 was a large amount of caffeine in the tea samples, whereas, there was slight 213 amount of caffeine in the OTP. In addition, the tea and coffee samples contain 214 small amounts of theophylline and theobromine. There was a relatively high 215 level of theobromine in the fermented Pu-er tea samples, but only trace 216 amounts of paraxanthine in the tea and coffee samples. 217

To evaluate and compare the C-CAF content of all samples, we next determined the C-CAF content by Low-pH Precipitate Method (**Table. 1**). Nearly a quarter of the caffeine in fermented Pu-er tea was C-CAF. What's

more, about three-quarters of the caffeine in the caffeine added OTP sample 221 was C-CAF. Only about one-tenth of caffeine in the black tea samples, 222 however, was C-CAF. In the non-fermented Pu-er tea, oolong tea, green tea 223 and caffeine added TPS there was just a moderate level of C-CAF. Curiously, 224 the C-CAF levels of PTPS and caffeine added PTTB were also high. In 225 addition to fermented Pu-er tea, the C-CAF percent of the caffeine added OTP, 226 PTPS and caffeine added PTTB samples were also obviously greater than the 227 percent of precipitate volume (Table. 1 and Supplemental Table 2). These 228 results indicate that, compared to the other samples, fermented Pu-er tea, 229 caffeine added OTP, PTPS, and caffeine added PTTB samples contained 230 more C-CAF, and caffeine in these samples abounded in the precipitate 231 232 components; while there was hardly any C-CAF in the coffee samples.

To further characterize the combination of caffeine with other components 233 of the samples, the fermented Pu-er tea, OTP and coffee samples were 234 examined through Chloroform Extraction (Fig. 1). The OTP were mainly 235 distributed in the aqueous phase. The caffeine concentration ratio of the 236 Chloroform phase to aqueous phase was determined with significant 237 differences (P < 0.001) among the coffee, fermented Pu-er tea and OTP 238 samples. For the fermented Pu-er tea and OTP samples, the balance of 239 caffeine distribution between water and chloroform was shifted severely. The 240 caffeine concentration ratio of the Chloroform phase to aqueous phase in 241 descending order was as follows: coffee samples, fermented Pu-er tea 242

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samples and OTP samples. These results indicate that caffeine in the OTP and fermented Pu-er tea samples was combined with OTP which belonged to the aqueous phase. *Effect of pH on the formation of caffeine contained precipitate.* To further probe into whether or not fermented Pu-er tea aqueous extract or OTP can be

probe into whether or not fermented Pu-er tea aqueous extract or OTP can be 247 precipitated in the condition of mouse stomach, we examined the association 248 between the formation of precipitate and solution acidity (Fig. 2 A, B). Plenty of 249 precipitate was formed at pH = 1.5 in the OTP and fermented Pu-er tea 250 251 solution (Fig. 2 A) and the C-CAF percent of the OTP and fermented Pu-er tea samples was high (Fig. 2 B). Compared to the pH of 1.5, in spite of the 252 precipitate quantity, the percentage of C-CAF was less at pH 3.5, but obvious 253 254 precipitate was still formed. Consequently, the fermented Pu-er tea and OTP might be precipitated by internal conditions of the mouse stomach. 255

Plasma levels of caffeine and its metabolites. The results above suggest 256 that the caffeine in fermented Pu-er tea was combined with OTP and the 257 C-CAF may be precipitated in mouse gastric juice. Nearly the same as pure 258 caffeine, there is hardly any C-CAF in coffee. Therefore, coffee was selected 259 as a control group to investigate whether the absorption of caffeine was 260 influenced by combining with OTP. Following the coffee, fermented Pu-er tea 261 and caffeine added OTP administration, plasma levels of caffeine, and its 262 metabolites were determined at different time points (Fig. 3 A, B, C, D). 263

Following the sample solutions administration, plasma levels were not

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significantly different 5 or 10 min later. However, compared to the coffee group, 265 plasma caffeine levels of the fermented Pu-er tea group and the caffeine 266 added OTP group were significantly reduced at 20 min and 40 min after 267 ingestion (P<0.05). Moreover, there were significant differences of plasma 268 caffeine levels between the coffee group and the caffeine added OTP group at 269 60 min and 120 min after ingestion (P<0.01). There were not significant 270 differences between the coffee group and the fermented Pu-er tea group 40 271 min after ingestion (Fig. 3 A). 272

273 For paraxanthine and theophylline, the plasma levels of the fermented Pu-er tea group and the caffeine added OTP group were higher than the coffee 274 group in the early stage (5 min to 60 min). In contrast, the plasma levels of the 275 276 two groups above were lower than the coffee group after 120 min (Fig. 3 B, C). Noteworthy, the plasma theobromine levels of the fermented Pu-er tea group 277 were far higher than the other two groups before 360 min (P < 0.05) (Fig. 3 D). 278 Compared to the coffee group, the plasma theobromine levels of the caffeine 279 added OTP group were higher in the early stage (5 min to 60 min), with the 280 most significant difference occurring 10 min after ingestion (P<0.01). 281 Interestingly, the plasma levels of theobromine of the caffeine added OTP 282 group were also lower than the coffee group after 120 min. 283

Excrement levels of caffeine. Given that the absorption of C-CAF in fermented Pu-er tea and OTP was decreased within a prescribed time limit, we speculated that the absorption of C-CAF must be either delayed or inhibited in

²⁸⁷ mice. To ascertain the cause of this, the excrement levels of caffeine were ²⁸⁸ examined (**Fig. 4**).

Excrement levels of caffeine from the fermented Pu-er tea group and caffeine added OTP group were significantly higher than that of the coffee group (P < 0.0001). The result indicates that more caffeine was excreted in the fermented Pu-er tea group and the caffeine added OTP group compared to the coffee group. In all probability, the absorption of C-CAF was inhibited.

Locomotor activity levels. The result above suggested that caffeine can be combined with OTP, and the absorption of C-CAF can be inhibited in mice. To better understand the biological significance of that, we then performed another experiment for monitoring the locomotor activity of mice to elucidate the effects of C-CAF (**Fig. 5**).

The locomotor activity levels of the water and caffeine added OTP groups were lower than the coffee group, significantly (P < 0.05). The fermented Pu-er tea group was also lower than the coffee group (P = 0.14). These results indicate that the effect of caffeine in vivo was inhibited in the fermented Pu-er tea group and the caffeine added OTP group.

304 **Discussion**

To clarify the absorption of C-CAF in mice, it is necessary to discuss the total caffeine, C-CAF and caffeine metabolites content of the samples, and probe into which components can combine with caffeine. Previous studies³⁴⁻³⁶ have reported that total sugar (14.47-27.62 %), caffeine (2.35-10.43 %) and

catechins (29.86-78.66 %) were found to be the main chemical components of 309 tea cream in the green tea concentrate. Furthermore, the effect of caffeine on 310 tea cream and haze of tea infusions is dependent on the presence of sufficient 311 substances within pyrogallol groups³⁷. Our previous study²² also showed that 312 the caffeine in fermented Pu-er tea can be categorized into F-CAF and C-CAF. 313 However, fermented Pu-er tea contained trace free tea catechins. The major 314 catechins were transformed to OTP due to the specific feature of the catechin 315 - a strong anti-oxidant and easily oxidated in the fermentation process. 316 Theaflavins, thearubigins and theabrownins are the main complex phenolic 317 compounds deriving from the oxidation of catechins and their gallates during 318 the fermentation process. Theaflavins undergo further oxidation during 319 fermentation to form more polymerized thearubigins, and then condensed 320 theabrownins^{18,38}. The OTP still contained a large amount of benzene groups 321 even though their chemical structure is not cleared, because the original 322 catechins retain at least one benzene ring and the absorption curve of the OTP 323 has characteristic peaks of benzene. The OTP was charactered as polymeric 324 organic acid, which can be precipitated by adjusting the pH value to 2. 325 Coincidentally, an increased level of C-CAF in tea is associated with the 326 increased degree of oxidization or fermentation during the processing of tea. 327 There is abundant OTP in fermented Pu-er tea, which is closely associated 328 with a high level of C-CAF³⁹. Accordant with previous studies²², the C-CAF 329 content of fermented Pu-er tea is far higher than black tea, green tea, and 330

instant coffee.

In order to investigate the material basis of the C-CAF we attempted to 332 separate PTPS and PTTB – two main components in Pu-er tea³³. In doing so, 333 we found that PTTB can be precipitated with PTPS during the alcohol 334 sedimentation process, and PTPS and PTTB are difficult to separate. In 335 essence, OTP is the principal component of the PTPS and PTTB samples. The 336 significantly lower C-CAF content of PTTB to that of OTP may result from the 337 fact that PTTB was mixed with other components, TPS for example. As 338 another major component in the extraction of tea, TPS is a kind of 339 poly-sacchride that may be, during the fermentation process, hydrolyzed to 340 oligosaccharides and monosaccharides. In keeping with this interpretation, 341 results of Chloroform Extraction for caffeine are consistent with the results of 342 the Low-pH Precipitate Method. These results highlight the importance of OTP 343 in the formation of C-CAF and demonstrated that OTP is the main material 344 basis for the formation of C-CAF in fermented Pu-er tea. 345

As the major stimulant compound to human central nervous system³⁰, caffeine is known to form complexes by self-association and with tea catechins. The Crystalline structure of the complexes of (-)-catechin (CA) and (-)-catechin-3-O-gallate (Cg) with caffeine have shown that CA formed a 1 : 1 complex with caffeine by intermolecular hydrogen bonds, whereas Cg formed a 2 : 4 complex with caffeine by face-to-face and offset π - π interactions and intermolecular hydrogen bonds^{31,32}. The ¹H-NMR spectra data indicated that

the chemical shift for caffeine between reference sample (free caffeine) and the complex of OTP & caffeine sample changed little (Supplemental Figure 2, Supplemental Table 3 and Supplemental Figure 3). These results suggest that OTP and caffeine formed a complex by intermolecular forces but without covalent bonds combination.

Moreover, we should note that the effect of pH on tea solids extraction 358 yield was significant. Total solids extracted between pH levels 2 to 11 varied 359 from 270 to 290 g/kg tea, but figures were doubled at pH 1.1⁴⁰. One can 360 speculate that pH is also a key factor to the precipitate quantity of the 361 fermented Pu-er tea and OTP aqueous solution. Considering that caffeine can 362 be combined with OTP and can form a precipitate with OTP in a Low-pH 363 solution, we sought to determine if the C-CAF can be precipitated in the 364 condition of mouse stomach. For this purpose, we examined the precipitate 365 quantity and percent of C-CAF of the samples at the pH similar to the stomach. 366 The results showed that the caffeine in fermented Pu-er tea and OTP samples 367 might be precipitated in the condition of mouse stomach. In addition, it's worth 368 noting that the content of C-CAF in the coffee samples was very low. 369

After ascertaining the above results, we examined whether the absorption of the C-CAF was influenced in mice. To address this issue, we investigated the plasma levels of caffeine and its metabolites after the mice were administrated caffeine contained samples. In agreement with previous reports⁴, our study showed that the mice plasma caffeine concentration of the coffee Food & Function Accepted Manuscript

375 group peaked 20min after the ingestion of caffeine (50 mg/kg). In marked 376 contrast, the plasma levels of caffeine of the fermented Pu-er tea and caffeine 377 added OTP group varied marginally between 10 and 40 min, and were 378 significantly lower than the coffee group. The simplest explanation for this 379 observation is that the absorption of caffeine in fermented Pu-er tea and OTP 380 was reduced, significantly.

Caffeine from coffee or other beverages is absorbed by the small intestine 381 within 45 min of ingestion and moves through cellular membranes with the 382 same efficiency that it is absorbed and circulated to tissues of the body⁴¹. It is 383 eliminated by first-order kinetics⁴². Caffeine is metabolised by CYP enzymes 384 (de-methylation), xanthine oxidase (formation of uric acid metabolites) and/or 385 N-acetyltransferase (acetylation) in the liver, and through enzymatic action 386 results in three metabolites: paraxanthine, theophylline, and theobromine⁴³. In 387 contrast to caffeine, the plasma levels of paraxanthine and theophylline in the 388 fermented Pu-er tea group and caffeine added OTP group were higher than 389 the coffee group in the early stage (5min to 60min), but were lower than the 390 coffee group after 120 min. Nevertheless, the paraxanthine and theophylline 391 intake from the fermented Pu-er tea group and caffeine added OTP group is 392 not more than the coffee group (Table 1). One postulate for the above results 393 is that some components (probably OTP) influenced the distribution or 394 metabolism of caffeine in the mice. However, owing to more caffeine was 395 absorbed and metabolized by the mice in the coffee group, the plasma levels 396

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of paraxanthine and theophylline in the fermented Pu-er tea group and caffeine added OTP group were lower than the coffee group after 120 min. As a result of the dramatically higher theobromine content of fermented Pu-er tea in comparison to the coffee and OTP groups (**Table 1**), the fermented Pu-er tea group plasma levels of theobromine were far higher than the other two groups before 360min.

To assist in determining if the significantly lower plasma levels of caffeine 403 found in the fermented Pu-er tea and caffeine added OTP group were the 404 result of caffeine metabolism, we tested the mice's excrement levels of 405 caffeine. Our findings confirmed that the absorption of C-CAF was inhibited in 406 mice and was excreted, in part. The popularity of investigating caffeine has 407 generated more interest from scientists of late due to the potentially harmful 408 effects it has on our health². Caffeine is a non-selective antagonist for 409 adenosine receptors²⁷. Several studies confirm that the activation of adenosine 410 receptors A₁ and A_{2A}, as well as the regulation of adenosine production and 411 degradation, is essential for sleep induction and proper control of the 412 sleep-wakefulness cycle⁴⁴. The results of locomotor activity of mice indicate 413 414 that the irritation effect of caffeine was inhibited when complexed with OTP.

To our knowledge, this is the first report that investigated the absorption of C-CAF in mice. Several main findings were obtained that might advance the theoretical framework by potentially illustrating why different sources of caffeine have different effects on humans. If the total caffeine intake is

419 consistent, the high level of C-CAF may be beneficial for people with sensitivity
420 to caffeine to reduce the side effects of caffeine, such as insomnia. Moreover,

421 this may be meaningful for developing some low F-CAF level products.

422 **Conclusions**

In conclusion, our results showed that OTP formed with caffeine to become the C-CAF contained in fermented Pu-er tea. The absorption of this C-CAF is inhibited in mice and may be the same in humans. What's more, the effect of caffeine was influenced significantly when it was combined with OTP. Taken together, this offers insight into the influences of fermented Pu-er tea on sleep loss are diminished in comparison with other forms of caffeine and/or caffeine containing products.

430 **Associated content**

431 **Abbreviations Used**

C-CAF, complexed caffeine; F-CAF, free caffeine; OTP, oxidative tea
polyphenols; PTPS, Pu-er tea polysaccharide; PTTB, Pu-er tea theabrownin;
TPS, tea polysaccharides.

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performed statistical analysis and wrote the manuscript; and J.S., S.H. and
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444	Supplementary Material
445	Refer to Web version on PubMed Central for the preparation process of OTP,
446	TPS, PTPS and PTTB; the content of caffeine, theophylline, theobromine and
447	paraxanthine of the samples (Supplemental Table 1); the percent of precipitate
448	volume of the samples (Supplemental Table 2); "Assay Validation" for Levels of
449	caffeine and its metabolites in plasma (Supplemental Figure 1); and $^{1}\mathrm{H}$ NMR
450	spectra data for caffeine (Supplemental Figure 2, Supplemental Table 3 and
451	Supplemental Figure 3).

Author information 452

Corresponding Author 453

^{*}(X.W.) +8615912579655. +8687165226711. Phone: Fax: E-mail: 454 wangxuanjun@gmail.com; haosm@sina.com (S. H.); shengj@ynau.edu.cn (J. 455 S.). 456

Notes 457

The authors declare no competing financial interest. 458

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sample type	sample name	F-CAF, <i>mg/L</i>	C-CAF, <i>mg/L</i>	C-CAF, %
	Gold bud tribute pu-erh tea	1716.13±0.93	486.54±3.81	22.09±0.14
fermented pu-erh tea	TAETEA pu-erh tea (7592)	1318.77±0.48	378.47±2.68	22.30±0.13
	Brick pu-erh tea	1370.85±0.22	267.63±1.13	16.33±0.06
non-fermented	Moonlight White pu-erh tea	2530.20±0.50	60.23±0.47	2.32±0.02
pu-erh tea	Bangwei pu-erh tea	1179.72±1.25	11.27±0.20	0.94±0.02
	Changning black tea(first grade)	1100.19±1.41	115.78±0.64	9.52±0.04
black tea	MAKEMY black tea	1830.36±1.22	261.60±0.52	12.50±0.02
	Dianhong black tea	1443.38±0.11	118.37±0.49	7.58±0.03
oolong too	Tieguanyin oolong tea	1033.37±7.51	6.29±0.47	0.60±0.04
	Da Hung Pao oolong tea	1460.97±4.83	69.08±0.51	4.52±0.05
	Longjing green tea	1474.15±0.46	13.63±0.24	0.92±0.02
groop too	Biluochun green tea	1399.68±2.39	20.78±0.84	1.46±0.06
greentea	Huilong green tea	1387.57±4.29	8.80±0.06	0.63±0.00
	Yunnan green tea	1411.13±1.57	10.91±0.27	0.77±0.02
	Nescafe coffee	550.58±0.40	0.63±0.06	0.11±0.01
coffee	Maxwell House coffee	664.36±0.19	0.98±0.02	0.15±0.00
	Yunnan Pasteral coffee	702.01±0.45	0.76±0.03	0.11±0.00
instant fermented	Deepure instant essence of	1162 04+0 65	250 62+0 62	22 10+0 02
pu-erh tea	pu-erh tea	1102.04±0.05	550.05±0.02	23.1010.02
OTP	OTP	45.92±0.44	146.95±0.61	76.19±0.16
PTPS	PTPS	262.13±0.73	165.80±2.06	38.74±0.34
mixture of TPS and	caffeine added TPS	1085.51±0.99	34.92±0.07	3.12±0.01
caffeine				
mixture of OTP and	caffeine added OTP	415.83+0.58	831 85+0 42	66.67+0.02
caffeine				
mixture of PTTB and	caffeine added PTTB	825.93+0.23	317.33+0.24	27.76+0.02
caffeine	· · · · · · · · · · · · · · · · · · ·			
caffeine	caffeine	1234.68±0.63	0.70±0.05	0.06±0.00

TABLE 1. The content a	and percent of C-CAF	of the samples ¹
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¹Data were presented as mean ± SEM of three independent experiments. C-CAF, complexed caffeine; F-CAF, free caffeine; OTP, oxidative tea polyphenols; PTPS, Pu-er tea polysaccharide; PTTB, Pu-er tea theabrownin; TPS, tea polysaccharides.

Figures

FIGURE 1



FIGURE 2



FIGURE 3



FIGURE 4



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FIGURE 5



Figure captions

FIGURE 1. Caffeine concentration ratio of Chloroform phase to aqueous phase. Data were presented as mean ± SEM and were analyzed with repeated measures (n=4) one-way ANOVA, Two-tailed p<0.05 was considered to be statistically significant (***P<0.001 vs. coffee samples). OTP, oxidative tea polyphenols.

FIGURE 2. Effect of pH on the formation of precipitate. Data were presented as mean ± SEM and were analyzed with repeated measures (n=3) one-way ANOVA. OTP, oxidative tea polyphenols.

FIGURE 3. Plasma levels of caffeine (A), paraxanthine (B), theophylline (C) and theobromine (D). Data were presented as mean ± SEM. Differences within groups at different time points were analyzed with repeated measures (n=6) one-way ANOVA, Two-tailed p<0.05 was considered to be statistically significant. (*P<0.05, **P<0.01 and ***P<0.001 vs. coffee group). OTP, oxidative tea polyphenols.

FIGURE 4. Excrement levels of caffeine. Data were presented as mean ± SEM. Differences within groups were analyzed with repeated measures (n=9) one-way ANOVA, Two-tailed p<0.05 was considered to be statistically significant. (***P<0.001 vs. coffee samples). OTP, oxidative tea polyphenols.

FIGURE 5. Locomotor activity levels after caffeine administration. Data were presented as mean ± SEM. Differences within groups were analyzed with repeated measures (n=6) one-way ANOVA, Two-tailed p<0.05 was considered to be statistically significant. (*P<0.05, vs. coffee group). OTP, oxidative tea polyphenols.