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Observation of nondispersing classical-like molecular rotation

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Using the technique of an optical centrifuge, we produce rotational wave packets which evolve in time along either classicallike or non-classical trajectories. After releasing O_2 and D_2 molecules from the centrifuge, we track their field-free rotation by monitoring the molecular angular distribution with velocity map imaging. Due to the dispersion of the created rotational wave packets in oxygen, we observe a gradual transition between "dumbbell"-shaped and "cross"-shaped distributions, both rotating with a classical rotation frequency. In deuterium, a much narrower rotational wave packet is produced and shown to evolve in a truly classical non-dispersing fashion.

1 Introduction

Microscopic quantum objects behaving in a classical manner are of great interest due to the fundamental aspects of quantum-classical correspondence. The well known examples include coherent states of the quantum harmonic oscillator, slowly spreading Rydberg wave packets¹ and nonspreading Trojan wave packets² in highly excited atoms. Classically behaving quantum wave packets preserve their minimum uncertainty shape as they move along the corresponding classical trajectories^{3–5}. In the case of rotating molecules, classicallike behavior implies nondispersing rotation of the molecular wave function with narrow distribution of angular momenta and strong localization around the internuclear axis.

Angular localization of molecular axes, well known as molecular alignment⁶, has been long recognized as an important tool for controlling chemical reactions in gaseous media⁷ and at gas-surface interfaces^{8,9}, for imaging molecular orbitals¹⁰ and generating extreme UV radiation^{11,12}, for deflecting molecular beams¹³ and separating molecular isotopes¹⁴. Although the popular non-adiabatic approach to molecular alignment with intense non-resonant laser pulses can produce very narrow angular distributions^{15–22}, the angular momentum of the aligned molecules is usually poorly defined, both in magnitude and direction. Hence, the created rotational wave packet quickly disperses in angle, undergoing oscillations between aligned and non-aligned distributions. Controlling molecular rotation with a sequence of two^{24,25} or more^{26,27} laser pulses offers better defined directionality, but suffers from the lack of selectivity and efficiency.

Quantum analogs of classical rotational dynamics, such as the gyroscopic destabilization in the rotation of an asymmetric top molecule, has been recently found in the absorption spectra of thermal molecular ensembles²³. In this work, we re-



Fig. 1 Time evolution of a wide (top) and a narrow (bottom) rotational wave packet, consisting of many or only two rotational states, respectively. The latter is an example of a quantum "cogwheel state" described in text. T_{rev} is the rotational revival time.

port on the direct imaging of the classically behaving molecular wave function. We use an optical centrifuge as a selective and efficient method of creating localized rotational wave packets^{28–31}. The centrifuge technique enables us to execute a complete transfer of molecules from the initial ground rotational level to a coherent superposition of rotational states $|N, M_N = N\rangle$, where N (either only even or only odd) and M_N are the rotational and magnetic quantum numbers, respectively. If this wave packet is initially localized along a certain spatial axis (upper left picture in Fig.1), it will rotate with an angular frequency of $\Omega_N = (E_{N+2} - E_N)/(2\hbar)$ (where E_N is the rotational energy) while simultaneously undergoing angular dispersion. For a homonuclear molecule with the rotational constant B, it will then revive to its initial shape, up to a rotation by a constant angle, at one quarter of the revival time $T_{rev} = h/(2B)$ (upper right picture in Fig.1). The so-called fractional revivals occur at intermediate times $t = \frac{p}{a}T_{rev}$, with $\frac{p}{a}$ being an irreducible fraction ^{32,33}. Provided the number of populated rotational states is large, at the time of a fractional revival the initial wave packet develops $\frac{q}{4}(3-(-1)^q)$ uniformly distributed lobes, which keep rotating with the same frequency Ω_N . For example, around $t = T_{rev}/8$, the angu-

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lar distribution exhibits four peaks in the plane of rotation, as shown in the top row of Fig. 1. As a *transient* effect, such multi-axial alignment has been observed experimentally ¹⁷ and used as a main ingredient in the new scheme for quantum logic gates based on the control of rotational wave packets ^{34,35}.

If the wave packet consists of only two states $|N, M_N = N\rangle$ and $|N+2n, M_N = N + 2n\rangle$, where n is an integer, the resulting "cogwheel state" rotates without spreading, preserving its shape at all times³⁶. The wave function of a quantum cogwheel is symmetric with respect to rotation by π/n and follows the classical-like nondispersing motion indefinitely³⁷. The simplest case of a cogwheel state with n = 1is shown in the lower row of Fig. 1. Similarly to the importance of transient cogwheel distributions in quantum information processing³⁵, it has been speculated that nondispersing cogwheel states could open new perspectives in the development of molecular machines, whereas in metrology, they could serve as "molecular stopwatches" on a femtosecond time scale with immediate applications in synchronizing ultrafast laser sources³⁸. Here, we report on the first direct observation of the nondispersing n = 1 cogwheel state in centrifuged D₂ molecules. In oxygen, we create a coherent superposition of more than two rotational states and demonstrate the transition between the classical-like rotation of a molecular stopwatch and that of a quantum cogwheel.

2 Experimental setup

We produce an optical centrifuge - a laser pulse, whose linear polarization undergoes an accelerated rotation along the propagation direction (see Appendix for technical details) following the original recipe of Karczmarek *et al.*²⁸. Through the interaction with an induced dipole moment of a molecule, the centrifuge field aligns molecular axes along the direction of its polarization and forces the molecules to spin along with it. The terminal frequency of this rotation can be accurately controlled and can reach as high as 10 THz for our laser system³¹. Key to this work is the ability to determine the initial polar angle ϑ_0 , at which the molecules are released from the centrifuge and start their field-free evolution. We achieve this by means of the time-resolved polarization gating of the centrifuge field, described in detail in Appendix.

We track the rotation of centrifuged molecules by means of the velocity map imaging (VMI) technique³⁹. In VMI, an intense femtosecond probe pulse dissociates the molecule via a multi-photon ionization process followed by a Coulomb explosion, as illustrated in Fig. 2 **a**. The method enables one to map out the direction of the ion recoil and, therefore, the angular distribution of molecular axes at the moment of explosion¹⁷. The recoiling ions are extracted with an electric field and projected onto a micro-channel plate detector equipped with a phosphorus screen, where they leave a spatially re-



Fig. 2 (a) Illustration of the operation principle of Velocity Map Imaging (VMI) employed for determining the molecular angle. E_{DC} indicates the direction of the permanent electric field used for accelerating ions toward the multi-channel plate (MCP). Only the signal from the ions recoiling in a particular direction ϑ_m is detected with a photomultiplier tube (PD) positioned behind a phosphorous screen and a small pinhole in an opaque mask. Brightness of the screen represents the strength of an ion signal. (b) Experimental setup. Centrifuge and probe pulses are combined on a beam splitter (BS) and sent into a vacuum chamber, where both beams are focused on a molecular jet with a concave mirror. Relative timing of the two pulses is controlled with a precision delay line. Dark green circle shows an example of a VMI image. 'PP' indicates the plane of molecular rotation, i.e. the plane of panel (a). solved fluorescence trace. Important to this work is the mapping of the molecular polar angle ϑ to the radial coordinate $r(\vartheta)$ of the ion image on the phosphorus screen.

The energy release in a Coulomb explosion is fully determined by the dissociation channel, i.e. by the final states of the recoiling ions. For any particular channel, all possible initial momenta reside on a sphere, with each point corresponding to a distinct angular orientation of the exploding molecule (Fig. 2 **a**, green dashed circle). Note, that fast molecular rotation may add a non-negligible tangential component to the momentum of recoiling ions. However, for a synchronously rotating molecules with a well-defined angular momentum, as produced by an optical centrifuge, this will simply rotate the whole orientation sphere, preserving the one-to-one correspondence between the location on the sphere and the ion image on the phosphorus screen.

In our experiment, we combine a beam of circularly polarized 35 fs ionizing probe pulses with a beam of centrifuge pulses. Circular probe polarization insures isotropic ionization in the plane of rotation. We focus both laser beams on a supersonically expanded gas of either O2 or D2 molecules between the charged plates of a standard VMI setup, as shown in Fig. 2 b together with a typical VMI image of Coulombexploded O₂. Multiple dissociation channels are represented by a number of concentric rings, whose center is shifted to the left from the vertical axis due to the initial molecular velocity in the direction of the supersonic jet. The rings are not uniform in the plane of the image because the ionization cross-section depends on the angle between the molecular axis and the polarization plane of a circularly polarized probe pulse (plane 'PP' in the figure). The rotational temperature in the supersonic jet is below 10 K, which in the case of oxygen means that the majority of molecules are in the ground rotational state, N = 1.

In general, a single point on the VMI image may originate from two different molecular polar angles ϑ , corresponding to the positive and negative vertical components of the recoil velocity, as illustrated by the blue and red lines in Fig. 2 a. To resolve the ambiguity, we introduce an opaque mask with a pinhole, centered on the circumference of the largest observable dissociation ring, which results from a single molecular orientation angle, labeled as ϑ_m . The phosphorescence signal, recorded behind this pinhole with a photomultiplier tube (PMT), is proportional to the molecular angular distribution along ϑ_m , $I_{\text{VMI}}(t) = |\Psi(\vartheta_m, t)|^2$. Since $\vartheta_m = \vartheta_0 + \Delta \vartheta$, where ϑ_0 is the initial angle at the time of the release from the centrifuge and $\Delta \vartheta$ is the relative angle with respect to ϑ_0 , one can extract the molecular angular distribution by varying ϑ_0 while recording $I_{\text{VMI}}(t)$ at a fixed ϑ_m . In contrast to the conventional VMI procedure, in which the molecular orientation is typically determined from the whole two-dimensional image, our method relies on a small region of the screen, which



Fig. 3 Normalized population of the rotational states with N = 37 (blue) and N = 39 (red) as a function of the centrifuge final angular velocity. The black arrow shows the terminal angular frequency of the centrifuge used in this work for creating rotational wave packets in O₂.

is essential for this work.

3 Results

To monitor the free evolution of the rotational wave packets in centrifuged oxygen, we truncate the spectral bandwidth of the centrifuge so as to match its terminal rotational frequency with that of an oxygen molecule occupying rotational states with N = 37 and N = 39. This is achieved by using the technique of "centrifuge spectroscopy" recently developed in our group⁴⁰. The method is based on resonance-enhanced multi-photon ionization (REMPI), which employs a tunable nanosecond dye laser for ionizing the centrifuged molecules via a multi-photon (2+1) resonant process.

The measured population of the two states is shown by the blue and red curves in Fig.3 as a function of the terminal angular frequency of the centrifuge. To make them equally populated, we fix the latter at the value indicated by the black arrow. Note, however, that since the distance between the two peaks and their half-widths are comparable, the two neighboring states with N = 35 and N = 41 are also populated by the centrifuge. The population spread among multiple *N*-states stems from an imperfect adiabaticity of the centrifugal excitation³¹.

Two VMI images of oxygen, one with and one without the centrifuge, are shown in Fig. 4 **a**. Pre-excitation with an optical centrifuge results in the visible narrowing of the ion distribution in space, owing to the localization of the rotational $|N, M_N = N\rangle$ wave functions in the plane perpendicular to the propagation direction of the centrifuge (plane 'PP' in Fig. 2 **b**). After binning the PMT signal into 11 intervals, uniformly distributed from $\vartheta_0 = 0$ to $\vartheta_0 = \pi$, we plot it as a function of the time delay between the centrifuge and the Coulomb-explosion



Fig. 4 (a) Ion images averaged over 10,000 shots for non-centrifuged (left) and centrifuged (right) oxygen molecules. **(b)** Probability density as a function of the molecular angle and the free propagation time. Time zero corresponds to approximately 100 ps since the release from the centrifuge. The blue dashed line marks the calculated trajectory of a "dumbbell" distribution rotating with the terminal angular frequency of the centrifuge, whose classical period is indicated with the white horizontal bar at the lower right corner. **(c)** Zoom-in to the region near $\frac{3}{8}T_{rev}$, taken with better averaging and angular resolution. Twice higher density of the tilted lines (4 per classical period indicated by the tilted arrows) stems from the emergence of a "cross"-shaped distribution with four peaks along two perpendicular directions.

pulses.

The result is shown in Fig. 4 **b**. Being aligned at the moment of the release from the centrifuge, the molecules first undergo free rotation with the expected classical frequency of 3.3 full rotations per picosecond, indicated by the blue dashed line at the beginning of the scan. The wave packet, however, gradually disperses and, although the rotational dynamics could still be identified by the overall linear tilt of the signal, its contrast decreases dramatically. This happens because of the admixture of the neighboring N = 35,41 states to the rotational wave packet created around N = 37 and N = 39.

As expected, the alignment reappears every quarter-revival time, $\frac{1}{4}T_{rev}$ (middle section in Fig. 4 **b**). In the vicinities of $\frac{1}{8}T_{rev}$ and $\frac{3}{8}T_{rev}$, another fractional revival is observed. After repeating the measurement around $\frac{3}{8}T_{rev}$ with better averaging and higher angular resolution, we plot the result in Fig. 4 **c**. Here, instead of two tilted traces per one classical period (marked by a horizontal white bar), reflecting a double-peaked "dumbbell"-shaped alignment geometry, four parallel traces per period can be seen, corresponding to the "cross"-shaped angular distribution peaked in four spatial directions.

This observation is well anticipated from the analysis of the rotational dynamics outlined in the introduction and illustrated in the top row of Fig. 1.

To create a truly nondispersing quantum cogwheel state, one needs to excite a narrower wave packet consisting of only two N-states. To achieve this, one can use a gas of lighter molecules with a larger energy spacing between the rotational levels. We have repeated the experiment with molecular deuterium, whose moment of inertia is ~ 20 times lower than that of oxygen. D₂ has two spin isomers, ortho and paradeuterium, with only even or only odd rotational quantum numbers in their rotation spectrum, respectively. The terminal frequency of the centrifuge was set at half the frequency of the $N = 2 \rightarrow N = 4$ transition. Under these conditions, orthodeuterium is excited to the coherent superposition of equally populated N = 2 and N = 4 states, while paradeuterium is transferred into a single state with N = 3. As the probability density of the latter does not dependent on the polar angle ϑ , it introduces a homogeneous background which does not affect the observed dynamics of orthodeuterium, shown in Fig. 5. One can see that during one revival time, the molecule com-



Fig. 5 Probability density as a function of the molecular angle and the the free propagation time of D_2 prepared in the equal-weight superposition of N = 2 and N = 4 states. The observed nondispersing behavior illustrates the main property of a quantum cogwheel state. The classical period is indicated with the white horizontal bar at the lower right corner.

pletes exactly $3\frac{1}{2}$ full rotations in agreement with the expected frequency of classical rotation $\Omega_2 = \frac{1}{2\pi} \frac{\omega_4 - \omega_2}{2} = 7\frac{B}{h} = \frac{7}{2}\frac{1}{T_{rev}}$. We observe no dispersion in the angular shape of the created wave packet, as anticipated for the time evolution of a cogwheel state.

4 Conclusion

In summary, we have experimentally created and observed rotational wave packets which mimic the classical motion of a rotating object, such as a dumbbell or a cross, which preserves its shape over an extended period of time. We have demonstrated that the classicality of the rotational dynamics depends on the distribution width of the molecular angular momentum, with narrower distributions showing closer resemblance to classical behavior. The technique of an optical centrifuge has enabled us to produce very narrow rotational wave packets in different molecular species. In case of deuterium, where the created rotational wave packet consisted of only two rotational states, a truly nondispersing classical-like dynamics of a cogwheel state has been observed. A slightly broader distribution of angular momentum in centrifuged oxygen resulted in richer dynamics, exhibiting gradual transitions between the molecular wave functions of different spatial symmetries.

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5 Appendix

To produce the electromagnetic field of an optical centrifuge, we use a pulse shaper schematically shown in Fig. 6 **a**. Mirror M, positioned in the Fourier plane of the input lens L, splits in half the spectrum of 800 nm, 35 fs (full width at half-maximum, FWHM) pulses from a Ti:Sapphire regenerative amplifier. Frequency chirps of equal magnitude ($\beta = 0.26$ ps⁻²) and opposite signs are then applied to the two spectral halves by means of the two "chirp boxes" (CB1 and CB2), described in Fig. 6 **b**. Circularly polarized with an opposite handedness, they are finally re-combined into a single beam to interfere and produce a linearly polarized pulse, whose plane of polarization undergoes an accelerated rotation along a "corkscrew"-shaped surface illustrated in Fig. 6 **c**.

Due to the lack of interferometric stability between the two long arms of the centrifuge shaper, the initial angle of the field polarization ϑ_s (and with it, angle ϑ_0 at which the molecules are released from the centrifuge and start their field-free evolution) changes randomly from pulse to pulse. We measure ϑ_s by recording the two projections ($\langle E_{0^\circ}^2 \rangle$ and $\langle E_{45^\circ}^2 \rangle$) of the centrifuge field vector for every centrifuge pulse, using an optical setup shown in Fig. 6 **d**.

First, the rotating polarization vector of the centrifuge is projected onto the vertical axis by means of a linear polarizer (P). We then measure the intensity of the transmitted vertical component at a fixed delay time τ with respect to the rising edge of the centrifuge, using the cross-correlation optical gating method. The latter is executed by frequency mixing the (polarization-filtered) centrifuge pulse with a transform limited ultrashort reference pulse (Ref) on a nonlinear BaB₂O₄ (BBO) crystal. The cross-correlation signal, recorded with a photo-diode (PD1) is proportional to $\cos^2 \vartheta(\tau) \propto 1 + \cos 2\vartheta(\tau) = I_c(\tau)$, where $\vartheta(\tau)$ denotes the angle of the centrifuge field polarization at time τ . Note, that since $\vartheta(\tau) = \vartheta_s + \beta \tau^2$, where ϑ_s is the initial field orientation and the second term reflects its accelerated rotation, the detected signal is

$$I_c(\tau) = 1 + \cos(2\vartheta_s + 2\beta\tau^2). \tag{1}$$

To avoid the ambiguity in extracting the value of ϑ_s , another cross-correlation is performed simultaneously at a slightly different delay time $\tau + \Delta \tau$. The extra time difference, $\Delta \tau$, is introduced with a thin glass plate retarder (RTR). The intensity recorded with a second photo-diode (PD2) is proportional to $I_c(\tau + \Delta \tau) \approx 1 + \cos(2\vartheta_s + 2\beta\tau^2 + 4\beta\tau\Delta\tau)$. We adjust τ with an optical delay line so as to satisfy $4\beta\tau\Delta\tau = \frac{\pi}{2}$, i.e. so as to allow the polarization vector to rotate by 45° . Then,

$$I_c(\tau + \Delta \tau) = 1 - \sin(2\vartheta_s + 2\beta\tau^2).$$
⁽²⁾

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Fig. 6 (a) Optical centrifuge shaper, consisting of the following elements: diffraction gratings (GR), lenses (L), mirrors (M), retro-reflectors (RR) and two "chirp boxes" (CB1 and CB2), shown in panel (b). (c)Illustration of the electric field of an optical centrifuge. Linear polarization of the field \vec{E} follows a "corkscrew" surface, adiabatically exciting molecules to high rotational states. The shape of the centrifuge is constant, while its orientation in the laboratory frame, and, therefore, the angle at which the molecules are released, ϑ_0 , is changing randomly from pulse to pulse. (d) Optical setup for measuring the orientation angle ϑ_0 . Labels indicate the following components: ultrashort reference pulse (Ref), beam splitter (BS), polarizer (P), lenses (L), photodiodes (PD1 and PD2), BaB₂O₄ crystals (BBO) and glass plate retarder (RTR). (e) Experimentally detected correlation between the two photodiode signals. Each point corresponds to a different ϑ_0 , as indicated.

From the correlation graph of the two photodiode signals, which at this value of τ becomes a circle, the orientation angle ϑ_s (and with it, ϑ_0) can be easily extracted up to a constant angle $\beta \tau^2$, as shown in Fig. 6 **e**.

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