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Salicylic Acids as Readily Available Starting Materials for the Synthesis of meta-Substituted Biaryls

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Received 00th January 2012, Accepted 00th January 2012

Cite this: DOI: 10.1039/x0xx00000x

DOI: 10.1039/x0xx00000x

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Salicylic acids are shown as readily available and versatile starting materials that easily undergo a tandem arylation / protodecarboxylation process under Pd-catalysis. The corresponding meta-arylphenols can subsequently be easily transformed into a variety of meta-functionalized biaryls, highlighting the versatility of this approach to access this structural motif.

Over the last decade, transition metal-catalysed C-H arylation of aromatic rings has received great attention due to its power and efficiency for accessing diversely functionalized aromatic motives from simple starting materials.¹ Thus, controlling the regioselectivity of functionalization has become one of the most important challenges in the field.² Whereas great many methods are now available for the synthesis of ortho-arylated arenes,³ meta- and paraarylation methods are still significantly underdeveloped.^{4,5} However, meta-substituted biaryl motifs are widely found in drug candidates and other bioactive molecules (Scheme 1a).⁶ We recently reported a novel strategy for the meta-arylation of phenols in a one-pot operation involving a Kolbe-Schmitt carboxylation followed by a tandem arylation / protodecarboxylation process. 7,8 This methodology allowed the synthesis of meta-arylphenols from phenols containing moderately electron-rich or electron-poor substitution at C2 and C3. However, due to the intrinsic harsh requirements for the Kolbe-Schmitt carboxylation,⁹ these processes required the use of high pressures of CO2 (25 atm), high temperatures (190 °C) and, consequently, of specialized autoclave equipment. Furthermore, significantly electron-deficient phenols (such as 3-nitrophenol, or 3-trifluoromethylphenol) were not suitable substrates due to lack of reactivity towards carboxylation.

It is noteworthy that salicylic acids themselves are readily available starting materials, ¹⁰ and also easily synthesised from phenols via a variety of routes, including carboxylation,

carbonylation/oxidation, ¹¹ and ortho-lithiation of suitable Osubstituted phenols, followed by reaction with CO₂.¹² Furthermore, salicylic acids are also available in one step through Pd-catalyzed hydroxylation of benzoic acids.¹³ Therefore, we envisaged that an exploration of the suitability of salicylic acids for the general synthesis of meta-substituted biaryls would be of significant utility (Scheme 1b). In this report, we show that both electron-rich and electron-poor salicylic acids react smoothly under our tandem arylation / protodecarboxylation leading to the corresponding metaarylphenols. Furthermore, these substrates can then be easily functionalized at the C-O bond, resulting in a highly versatile and straightforward approach towards meta-biaryls.



Scheme 1. a) Representative examples of biologically active meta-substituted biaryls. b) This report: salicylic acids can be used as starting materials for the synthesis of a wide variety of meta-substituted biaryl motifs.

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Electronic Supplementary Information (ESI) available: Full experimental details are available. See DOI: 10.1039/c000000x/

Our optimized reaction conditions for the reaction of salicylic acid 1 (1.0 equiv) with aryl iodides 2 (3 equiv) involved the use of 2 mol % PEPPSI-IPr as a catalyst, 0.5 equiv each of K2CO3 and Ag₂CO₃, in AcOH at 150 °C (Scheme 1).¹⁴ Under these conditions a number of substituted salicylic acids were tested and both electronwithdrawing and electron-donating groups at C3 and C4 showed excellent compatibility with the reaction. It is important to note that in all cases the obtained yields were higher than those achieved in the one-pot methodology starting from phenols.⁷ Gratifyingly, the highly electron-withdrawing NO₂ (3ad) and CF₃ (3ae) groups, which had been shown to be unreactive in the one-pot methodology, now furnished 52% and 67% isolated yields, respectively. In all cases, the expected meta-arylphenol product was obtained with complete regioselectivity. This was the case even in the presence of an acetamido group (3aj), which has been shown to be a good orthodirecting group, affording **3aj** in 41% yield. The highly electron-rich 4-MeO-salicylic acid afforded only 22% yield of the desired product **3ak**, due to competitive protodecarboxylation of the starting material, highlighting one of the limitations of the methodology. Furthermore, substitution at C5 of the salicylic acid is poorly tolerated, with Me and Cl not reacting at all due to steric hindrance. On the other hand, the smaller F substituent allowed the arylation to proceed, affording 3an in 53% of yield.



Scheme 2. Scope of the tandem arylation / protodecarboxylation process on substituted salicylic acids (**1a-n**). Unless otherwise stated, all the reactions were carried out using 0.5 mmol of **1**, 1.5 mmol of **2a** and 0.5 mL of AcOH at 150 °C for 16 h. Yields are of isolated products. ^{*a*} The reaction was carried out at 160 °C. ^{*b*} 1.0 mL of AcOH were used. ^{*c*} K₂CO₃ was not used.

Substitution in the iodoarene coupling partner 2 was then examined (Scheme 3). The process is compatible with electron-withdrawing (**3ba-3ga**, **3ka-3na**) and electron-donating groups (**3ha**, **3oa**) in meta and para positions, leading to the corresponding meta-arylphenols in good yields. Remarkably, the monoarylated **3ga** was obtained selectively from *p*-diiodobenzene as coupling partner without any bisarylation product being observed.¹⁵ An aldehyde substituent (**3ia**) led to no reaction under the standard conditions. This may result

from the consumption of Ag_2CO_3 in an undesired aldehyde oxidation process. Surprisingly, when the salicylic acid was used in excess, **3ia** could be obtained in an excellent (85%) yield. Similarly, iodoarenes containing *p*-OMe (**3ha**), *p*-CH₂OH (**3ja**) and *m*-COMe (**3na**) proceeded in better yields when the iodoarenes were used as limiting reagents. Ortho-substitution at the iodoarene is not well tolerated, with only the smaller F substituent leading to appreciable reactivity (24% of **3qa**). Some heteroarenes such as iodoindole and iodopyridine were found to be compatible with the reaction, leading to the corresponding meta-heteroarylphenols **3ra** and **3sa** in 30% and 70% yields, respectively.



Scheme 3. Scope of the tandem arylation / protodecarboxylation process on substituted iodoarenes (**2b-s**). Unless otherwise stated, all the reactions were carried out using 0.5 mmol of **1a**, 1.5 mmol of **2** and 0.5 mL of AcOH at 150 °C for 16 h. Yields are of isolated products. ^{*a*} The reaction was carried out with 0.5 mmol of **1a** and 0.167 mmol of **2** at 130 °C. ^{*b*} Yield determined by ¹H NMR analysis using an internal standard. ^{*c*} *p*-iodobenzyl alcohol was used as starting material.

The great majority of reported methods for C-H arylation that use iodoarenes as coupling partners require the stoichiometric use of Ag-salts. ¹⁶ We have recently shown that for a range of such methodologies, the Ag-salt can be conveniently replaced by a cheap and readily available organic salt, NMe₄Cl, which makes these reactions increasingly attractive for large scale synthesis and industrial processes.¹⁷ We therefore tested if NMe₄Cl could also be a suitable replacement for Ag₂CO₃ in the present methodology (Scheme 4). Gratifyingly, good yields of the desired products could be obtained without any optimization in a silver-free process. Journal Name



Scheme 4. Silver-free method for the tandem arylation / protodecarboxylation of salicylic acids.

Taking advantage of the myriad of methodologies available for the functionalization of (and at) aromatic C-O bonds, the metaarylphenols here described are highly attractive intermediates towards the synthesis of meta-functionalized biaryls providing an efficient alternative to the most widely used Suzuki coupling. Thus, the meta-arylphenol **3ai** can be reacted with alkyl and aryl electrophiles in the presence of bases to form, in good yields, Oarylated **4a** and O-alkylated **4b**. Both of these motifs are common in natural products and pharmaceuticals. ¹⁸ On the other hand, transforming the OH into a triflate group, allowed a subsequent Buchwald-Hartwig amination¹⁹ to **4c** and Suzuki²⁰ coupling to the unsymmetrical meta-triaryl **4d** to occur in 92% and 93% yields, respectively. Finally, a cyano group could be easily installed in 90% yield by preforming the tosylate of **3ai**, followed by Pd-catalyzed coupling.



Scheme 5. Transformations of 3ai into a variety of meta-functionalized biaryls. Reagents and conditions: a) 2-chloro-5-nitropyridine, NaH, DMF, rt, 16 h; b) Mel, K₂CO₃, acetone, rt, 16 h; c) Tf₂O, Pyridine, DCM, rt, 1 h; Then, aniline, Pd(OAC)₂, BINAP, Cs₂CO₃, PhMe, 120 °C, 16 h; d) Tf₂O, Pyridine, DCM, rt, 1 h; Then, PhB(OH)₂, Pd(PPh₃)₄, Na₂CO₃, monoglyme/H₂O, 95 °C, 2.5 h; e) TsCl, Et₃N, MeCN, rt, 1 h; Then, Pd(OAc)₂, cm-Phos, K₄[Fe(CN)₆]•3H₂O, K₂CO₃, ^tBuOH/H₂O, 80 °C, 18 h.

In conclusion, we have demonstrated that salicylic acids can undergo facile Pd-catalyzed tandem arylation/decarboxylation leading to meta-arylated phenols with complete regioselectivity. These products can be further transformed into a variety of metafunctionalized biaryls highlighting salicylic acids as attractive starting materials for the synthesis of these structural motifs.

Financial support from the European Research Council for a Starting Grant (to I.L.), the Engineering and Physical Sciences Research Council (EPSRC) for a research grant, the China

Scholarship Council and Queen Mary University of London for a studentship (to J.L), and the Marie Curie Foundation for an Intra-European Fellowship (to S.P) are gratefully acknowledged. EPSRC National Mass Spectrometry Service (Swansea) is also acknowledged.

Notes and references

(a) J. Yamaguchi, A. D. Yamaguchi and K. Itami, Angew. Chem., Int. Ed., 2012, 51, 8960; (b) L. McMurray, F. O'Hara and M. J. Gaunt, Chem. Soc. Rev., 2011, 40, 1885; (c) C.-L. Sun, B.-J. Li and Z.-J. Shi, Chem. Rev., 2011, 111, 1293; (d) L. Ackermann, Chem. Rev., 2011, 111, 1315; (e) J. Wencel-Delord, T. Droge, F. Liu and F. Glorius, Chem. Soc. Rev., 2011, 40, 4740; (f) C. S. Yeung and V. M. Dong, Chem. Rev., 2011, 111, 1215; (g) Topics in Current Chemistry: C-H Activation, ed. J.-Q. Yu and Z. Shi, 1st edn, Springer, Berlin Heidelberg, 2010.

2 F. Juliá-Hernández, M. Simonetti and I. Larrosa, Angew. Chem. Int. Ed., 2013, 52, 11458.

3 For reviews on ortho-arylation see: (a) P. B. Arockiam, C. Bruneau, P. H. Dixneuf, Chem. Rev., 2012, 112, 5879; (b) D. A. Colby, R. G. Bergman, J. A. Ellman, Chem. Rev., 2010, 110, 624; (c) T. W. Lyons, M. S. Sanford, Chem. Rev., 2010, 110, 1147. For selected examples (since 2013), see: (d) B. Punji, W. Song, G. A. Shevchenko and L. Ackermann, Chem. Eur. J., 2013, 19, 10605. (e) B. Li, C. Darcel and P. H. Dixneuf, ChemCatChem, 2014, 6, 127. (f) R. K. Chinnagolla and M. Jeganmohan, Chem. Commun., 2014, 50, 2442. (g) Z. Fan, K. Wu, L. Xing, Q. Yao and A. Zhang, Chem. Commun., 2014, 50, 1682; (h) R. Feng, J. Yao, Z. Liang, Z. Liu and Y. Zhang, J. Org. Chem., 2013, 78, 3688; (i) D. Li, N. Xu, Y. Zhang and L. Wang, Chem. Commun., 2014, 50, 14862; (j) J.-H. Chu, H.-P. Huang, W.-T. Hsu, S.-T. Chen and M.-J. Wu, Organometallics, 2014, 33, 1190; (k) Z. Liang, J. Yao, K. Wang, H. Li and Y. Zhang, Chem. Eur. J., 2013, 19, 16825; (1) Z. Liang, R. Feng, H. Yin and Y. Zhang, Org. Lett., 2013, 15, 4544; (m) L. C. M. Castro and N. Chatani, Chem. Eur. J., 2014, 20, 4548; (n) C. Wan, J. Zhao, M. Xu and J. Huang, J. Org. Chem., 2014, 79, 4751; (o) W. H. Jeon, T. S. Lee, E. J. Kim, B. Moon and J. Kang, Tetrahedron, 2013, 69, 5152; (p) D. Li, N. Xu, Y. Zhang and L. Wang, Chem. Commun., 2014, 50, 14862; (q) L. Y. Chan, L. Cheong and S. Kim, Org. Lett., 2013, 15, 2186; (r) J. Han, P. Liu, C. Wang, Q. Wang, J. Zhang, Y. Zhao, D. Shi, Z. Huang and Y. Zhao, Org. Lett., 2014, 16, 5682; (s) J.-H. Chu, C.-C. Wu, D.-H. Chang, Y.-M. Lee and M.-J. Wu, Organometallics, 2013, 32, 272; (t) F. Yang, F. Song, W. Li, J. Lan and J. You, RSC Adv., 2013, 3, 9649; (u) Y. Aihara and N. Chatani, Chem. Sci., 2013, 4, 664; (v) P. B. Arockiam, C. Fischmeister, C. Bruneau and P. H. Dixneuf, Green Chem., 2013, 15, 67; (w) Z. Jiang, L. Zhang, C. Dong, X. Su,

H. Li, W. Tang, L. Xu and Q. Fan, *RSC Adv.*, 2013, **3**, 1025; (x) C. Arroniz, A. Ironmonger, G. Rassias and I. Larrosa, *Org. Lett.*, 2013, **15**, 910.

4 For examples of meta arylation see: (a) H.-X. Dai, G. Li, X.-G. Zhang, A. F. Stepan and J.-Q Yu, *J. Am. Chem. Soc.*, 2013, **135**, 7567; (b) L. Wan, N. Dastbaravardeh, G. Li and J.-Q Yu, *J. Am. Chem. Soc.*, 2013, **135**, 18056; (c) R. J. Phipps, M. J. Gaunt, *Science* 2009, **323**, 1593; (d) H. A. Duong, R. E. Gilligan, M. L. Cooke, R. J. Phipps, M. J. Gaunt, *Angew. Chem. Int. Ed.*, 2011, **50**, 463; (e) B. Chen, X.-L. Hou, Y.-X. Li, Y.-D. Wu, *J. Am. Chem. Soc.*, 2011, **133**, 7668.

5 For examples of para arylation see: (a) X. Wang, D. Leow and J.-Q. Yu, J. Am. Chem. Soc., 2011, 133, 13864; (b) C.-L. Ciana, R. J. Phipps, J. R. Brandt, F.-M. Meyer and M. J. Gaunt, Angew. Chem. Int. Ed., 2011, 50, 458; (c) Z. Wu, F. Luo, S. Chen, Z. Li, H. Xiang and X. Zhou, Chem. Commun., 2013, 49, 7653.

6 (a) M. Imanishi, S. Itou, K. Washizuka, H. Hamashima, Y. Nakajima, T. Araki, Y. Tomishima, M. Sakurai, S. Matsui, E. Imamura, K. Ueshima, T. Yamamoto, N. Yamamoto, H. Ishikawa, K. Nakano, N. Unami, K. Hamada, Y. Matsumura, F. Takamura and K. Hattori, *J. Med. Chem.*, 2008, **51**, 4002;
(b) L. Trinh, M. McCutchen, M. Bonner-Fraser, S. Fraser, L. Bumm and D. McCauley, *Biotechniques*, 2007, **42**, 756;
(c) M. Mor, S. Rivara, A. Lodola, P. V. Plazzi, G. Tarzia, A. Duranti, A. Tontini, G. Piersanti, S. Kathuria and D. Piomelli, *J. Med. Chem.*, 2004, **47**, 4998;
(d) P. Zhou, Y. Li, Y. Fan, Z. Wang, R. Chopra, A. Olland, Y. Hu, R. L. Magolda, M. Pangalos, P. H. Reinhart, M. J. Turner, J. Bard, M. S. Malamas and A. J. Robichaud, *Bioorg. Med. Chem. Lett.*, 2010, **20**, 2326.

7 J. Luo, S. Preciado and I. Larrosa, J. Am. Chem. Soc., 2014, 136, 4109.

8 For an alternative strategy for the meta-arylation of phenols, see: Ref. [4a] and [4b].

9 (a) H. Kolbe, *Justus Liebigs Ann. Chem.* 1860, **113**, 125. (b) A. Lindsey, H. Jeskey, *Chem. Rev.* 1957, **57**, 583.

10 Reaxys indicates over 1,700 salicylic acids are commercially available.

11 (a) M. Komiyama and H. Hirai, J. Am. Chem. Soc., 1983, 105, 2018; (b)
H. Wynberg, Chem. Rev., 1960, 60, 169; (c) U. N. Hofslokken, L. Skattebol, Acta Chemica Scandinavica, 1999, 53, 258; (d) D. Chakraborty, R. R. Gowda and P. Malik, Tetrahedron Lett., 2009, 50, 6553.

12 G. H. Posner and K. A. Canella, J. Am. Chem. Soc., 1985, 107, 2571.

13 Y.-H. Zhang and J.-Q. Yu, J. Am. Chem. Soc., 2009, 131, 14654.

14 See SI for extended optimization tables.

15 In contrast, PEPPSI-IPr mediated Negishi, Suzuki and Kumada crosscouplings on diiodo- and dibromo-benzenes have been shown to selectively lead to di- over mono-functionalization: I. Larrosa, C. Somoza, A. Banquy and S. Goldup, *Org. Lett.* 2011, **13**, 146. 16 (a) O. Daugulis, H.-Q. Do and D. Shabashov, Acc. Chem. Res., 2009, 42, 1074; (b) O. Daugulis, Top. Curr. Chem., 2010, 292, 57; (c) K. M. Engle, T.-

S. Mei, M. Wasa and J.-Q. Yu, Acc. Chem. Res., 2012, 45, 788.

17 C. Arroniz, J. G. Denis, A. Ironmonger, G. Rassias and I. Larrosa, *Chem. Sci.*, 2014, **5**, 3509.

18 (a) C. K.-F. Chiu, M. A. Berliner and Z. B. Li, In *Comprehensive Organic Functional Group Transformations*; A. R. Katritzky, R. J. K. Taylor, Eds.; Pergamon Press: New York, 1995; Vol. **2**, Chapter 2.13.; (b) C. L. E. Broekkamp, D. Leysen, B. W. M. M. Peeters and R. M. Pinder, *J. Med. Chem.*, 1995, **38**, 4615. (c) M. Palucki, J. P. Wolfe and S. L. Buchwald, *J. Am. Chem. Soc.*, 1997, **119**, 3395.

19 (a) A. O. Adeniji, B. M. Twenter, M. C. Byrns, Y. Jin, M. Chen, J. D.
 Winkler and T. M. Penning, *J. Med. Chem.*, 2012, **55**, 2311; (b) A. S. Guram,
 R. A. Rennels and S. L. Buchwald, *Angew. Chem Int. Ed.*, 1995, **34**, 1348. (c)

M. S. Driver and J. F. Hartwig, J. Am. Chem. Soc., 1997, 119, 8232.

20 (a) C. Y. HO, United States Patent Application Publication. US 2009/0105288 A1; (b) N. Miyaura and A. Suzuki, Chem. Rev., 1995, 95, 2457.