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ARTICLE TYPE

Two act as one: unexpected dimers of catechol direct a solid-state [2+2] photodimerization in a six-component hydrogen-bonded assembly

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We describe hydrogen-bonded dimers of catechol that act collectively as a single template to direct an intermolecular [2+2] photocycloaddition in the solid state. The directed reactivity involves discrete, six-component hydrogen-bonded 10 assemblies and a photoreaction that occurs stereoselectively and in quantitative yield.

There is considerable interest in using small molecules to direct reactivity in the organic solid state.¹ The approach employs a ditopic molecule as a template - akin to an enzyme in Nature - to ¹⁵ assemble olefins into the strict geometry of Schmidt² for an intermolecular photocycloaddition reaction. Principles of supramolecular chemistry are used to stack olefins into zero-dimensional (0D), or discrete, three- and four-component

- hydrogen-bonded assemblies that effectively decouple solid-state ²⁰ reactivity from unwanted effects of long-range packing.³ In addition to synthesizing molecules difficult to achieve in solution, a major benefit of using the solid state as a medium to control reactivity – particularly, from a supramolecular perspective – is that one can effectively bypass structure effects of entropy and
- $_{25}$ dissolution of the liquid phase⁴ to investigate extents to which principles of supramolecular chemistry can direct reactivity in the solid state and beyond (*e.g.* solution).

In this context, a principle employed by Nature to control reactivity is the operation of multiple copies of macromolecules ³⁰ that collectively act as a single template for a chemical reaction.⁵

- In such a setting, two or more associated molecules utilize functional groups to both interact with each other and aid a substrate to assemble in an active site for reaction. With this in mind, we wish to report here an unexpected ability of multiple
- ³⁵ copies of the ditopic molecule catechol (cat) to act together as a template in the solid state (Scheme 1). The reactivity is achieved in a co-crystal of 4(cat)·2(2,2'-bpe) (where: 2,2-bpe = trans-1,2-bis(2-pyridyl)ethylene) wherein hydrogen-bonded dimers of cat pre-organize 2,2'-bpe into a novel, discrete, six-component
- ⁴⁰ assembly for an intermolecular [2+2] photodimerization (Scheme 1a). The olefin reacts to afford *rctt*-tetrakis(2pyridyl)cyclobutane (2,2'-tpcb) stereoselectively and in quantitative yield. Whereas a single molecule is typically employed to stack olefins into the geometry for a [2+2]
- ⁴⁵ photodimerization in the solid state (Scheme 1b), we are unaware of a case wherein two or more copies of a single molecule unite to act as one to direct a cycloaddition reaction in a discrete assembly of molecules in a solid.



50 Scheme 1 Comparison: a) supramolecular (six components) and b) unimolecular (four components) templates for a [2+2] photocycloaddition.

geometry requirements of Schmidt The for а 55 photodimerization provide foci for efforts to engineer covalent bonds in solids. Small ditopic hydrogen-bond-donor⁶ and acceptor⁷ molecules with spatially-oriented functional groups have been useful in this regard to reliably stack olefins in solids for the photoreaction. Earlier work involving ditopic cat, 60 however, suggested that the two hydroxyl groups were not suitably positioned to assemble and stack two olefins for the photoreaction.^{8,9} Specifically, in (cat)·(4,4'-bpe), the diol adopts an anti-anti conformation in an unreactive 1D polymer. The idea that two copies of cat could stack two olefins to react also 65 seemed unlikely given the disparity between the close proximity of the two hydroxyl groups of cat (~ 2.70 Å) and the larger



separation distance that two carbon-carbon double (C=C) bonds generally must assume to react in a solid (i.e. 4.20 Å). Scheme 2

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Addition of **cat** (0.11 g) to MeCN (3 mL) in the presence of **2,2'-bpe** (0.10 g) yielded, upon standing for a period of *ca* 2 d, colorless crystals of $4(cat) \cdot 2(2,2'-bpe)$ **1** suitable for X-ray analysis. The formulation of **1** was confirmed by ¹H NMR ⁵ spectroscopy and single-crystal X-ray diffraction.[†]

A perspective of **1** is shown in Figure 1. The asymmetric unit consists of one **2,2'-bpe** and two **cat** molecules. The components assemble to form a discrete, six-component molecular assembly sustained by eight hydrogen bonds based on four O-H…N [O…N 10 separations (Å): O(1)…N(1) 2.658(2), O(3)…N(2) 2.624(2)] and

four O-H···O [O···O (Å): O(4)···O(1) 2.006(2), O(3)···I(2) 2.024(2)] and four O-H···O [O···O (Å): O(4)···O(1) 2.701(2), O(2)···O(3) 2.732(2)] forces (Fig. 1a). In the arrangement, the **cat** molecules self-assemble to effectively form hydrogen-bonded dimers that



Fig. 1 Perspectives of 1: a) six-component assembly, b) π -stacked columns, and c) nearest-neighbor assemblies.

- ²⁰ interact with face-to-face π -stacked bipyridines. Each **cat** molecule adopts a different conformation within a dimer. Specifically, one diol adopts a *syn-anti* conformation (Scheme 2)¹⁰ wherein the *syn* and *anti* hydroxyl groups participate in intermolecular O-H···O and O-H···N hydrogen bonds, ²⁵ respectively. The second **cat** adopts a *syn-gauche* conformation¹¹
- wherein the former and latter participate in intermolecular O-H…O and O-H…N hydrogen bonds, respectively, with the latter being twisted from the aromatic plane (105.5°). As a result of the assembly process, the C=C bonds of the two stacked olefins lie 30 parallel and separated by 3.71 Å. The stacked geometry satisfies
- the requirements of Schmidt for a [2+2] photodimerisation in a solid.² The structure and composition of the six-component assembly contrasts those discrete four-component assemblies sustained by ditopic **res**.^{1a} The assemblies pack as face-to-face π -³⁵ stacked columns, with nearest-neighbor olefins being separated
- by 3.76 Å, which also satisfies the criteria of Schmidt for photoreaction.

The olefins of solid 1 are photoreactive. When a powdered crystalline sample of 1 was exposed to UV irradiation (broadband

- ⁴⁰ Hg lamp) for 50 h, **2,2'-bpe** reacted to form **2,2'-tpcb** quantitatively. The generation of the photoproduct was evidenced by the emergence of a cyclobutane peak at 4.93 ppm and complete disappearance of the olefin resonances.
- The solid **1** also undergoes a single-crystal-to-single-crystal ⁴⁵ (SCSC) reaction, ^{1b,c} with the cyclobutane ring forming within the discrete hydrogen-bonded structures. In particular, X-ray

diffraction data collected on a single crystal after being exposed to UV radiation using a curing lamp¹² after 35 h revealed electron density consistent with **2,2'-tpcb** having formed within the ⁵⁰ discrete hydrogen-bonded structures (cyclobutane occupancy: 10%).^{8,13} The components of the reacted solid maintain the sixcomponent assemblies compared to unreacted **1** [O···N (Å): O(1)···N(2B) 2.446(19), O(3)···N(1B) 2.437(18); O···O (Å): O(4)···O(1) 2.733(3), O(2)···O(3) 2.745(3)].



65 Fig. 2 SCSC transformation of 1 (unreacted olefins omitted for clarity).

Whereas **cat** interacts with **2,2'-bpe** and **2,2'-tpcb** in a sixcomponent assembly, a structure determination of a recrystallized sample of reacted **1** (100% yield of **2,2'-tpcb**) reveals hydrogen-70 bonding different than the discrete structure. Specifically, recrystallisation of reacted **1** from MeNO₂ (4:1) afforded single crystals of 3(**cat**)'(**2,2'-tpcb**) **2** suitable for X-ray analysis.

The asymmetric unit of **2** contains two one-half **2,2'-tpcb** and three **cat** molecules. Similar to **1**, the components assemble *via* 75 O-H…N and O-H…O hydrogen bonds. In contrast to **1**, the components form 1D chains [O…N (Å): O(5)…N(1) 2.682(2), O(1)…N(2) 2.753(2), O(4)…N(5) 2.761(3)] with **cat** molecules that self-assemble *via* O-H…O forces [O…O (Å): O(3)…O(1) 2.833(2), O(6)…O(3) 2.843(2), O(2)…O(5) 2.765(2)] to form trimers. Each trimer consists of one and two **cat** molecules in *syn-anti* and *syn-gauche* (-32.16°, 43.23°) conformations, respectively. Each trimer effectively donates two O-H…N hydrogen bonds to two 2-pyridyl groups in a *cisoid* conformation. A final hydroxyl group participates in an O-H…N force with a se neighboring and crystallographically different **2,2'-tpcb** molecule, with the pyridines being in a *transoid* conformation. The cyclobutane rotamers alternate along the backbone of the 1D



polymer.

⁹⁰ Fig. 3 Perspectives of 2: a) cat trimers with conformation as *syn-anti* (green) and *syn-gauche* (purple and orange) and b) extended packing with alternating cyclobutane rotamers.

From the assembly **1**, as well as **2**, it is evident that the close ⁹⁵ proximity of the hydroxyl groups of a single **cat** molecules are not in a geometry favorable to preorganize two olefins for a [2+2]

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photodimerization as compared to res. The capacity of cat to self-assemble as a dimer and/or trimer, however, provides a structural flexibility that enables the geometry criteria of Schmidt to be achieved in a solid. Moreover, these observations are

5 important since the assembly properties provide a further degree of structural freedom to utilize small-molecule templates to direct reactivity in the restricted environment of a crystalline solid.

In summary, we have demonstrated a [2+2] photodimerization in the solid state directed by assemblies of cat molecules that

10 function as templates. The multi-component templates comprise cat dimers, with the reactivities occurring within discrete, sixcomponent, hydrogen-bonded structures. We are now exploring the robustness of the self-assembly process involving dimers of cat, and related trimers, to afford reactive hydrogen-bonded

15 structures. The ability of cat to afford such assemblies can provide an added and reliable means to achieve chemical reactivity in organic solid-state materials.

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20 Notes and references

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† Electronic Supplementary Information (ESI) available: X-ray structure ¹H NMR spectra, and X-ray powder patterns. See 25 data, DOI: 10.1039/b00000x/

Crystal data for 1: monoclinic, space group $P2_1/c$, a = 7.3687(7), b =19.169(2), c = 14.7014(15) Å, $\beta = 102.277(5)$ °, V = 2029.1(4) Å³, $D_c =$ 1.317 g/cm³, Mo-Ka radiation ($\lambda = 0.71070$ Å) for Z = 4. Least-squares

30 refinement based on 3329 reflections with $I_{\text{net}} > 2.0\sigma(I_{\text{net}})$ (out of 4648 unique reflections) led to a final value of R = 0.043. CCDC reference number 1025307.

Crystal data for 1 SCSC (10% conversion): monoclinic, space group $P2_1/c$, a = 7.4665(8), b = 19.1328(19), c = 14.8336(15) Å, $\beta =$

35 102.249(5)°, V = 2070.81 Å³, $D_c = 1.291$ g/cm³, Mo-Ka radiation ($\lambda =$ 0.71073 Å) for Z = 4. Least-squares refinement based on 2685 reflections with $I_{\text{net}} > 2.0\sigma(I_{\text{net}})$ (out of 3631 unique reflections) led to a final value of R = 0.059 CCDC reference number 1025308.

Crystal data for 2: triclinic, space group $P\bar{1}$, a = 11.9841(12) Å, b =40 12.6106(13) Å, c = 13.7902(14) Å, $\alpha = 109.918(5)^{\circ}$, $\beta = 103.137(5)^{\circ}$, γ = 102.958(5) °, V = 1801.6(3) Å³, D_c = 1.281 g/cm³, Mo-Ka radiation (λ = 0.71070 Å) for Z = 2. Least-squares refinement based on 4235 reflections with $I_{\text{net}} > 2.0\sigma(I_{\text{net}})$ (out of 5711 unique reflections) led to a final value of R = 0.051. CCDC reference number 1025309.

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Graphical Abstract

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Catechol self-assemblies as a dimer to function as template that directs a [2+2] photodimerization in the solid state.

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