# ChemComm

# Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this Accepted Manuscript with the edited and formatted Advance Article as soon as it is available.

You can find more information about *Accepted Manuscripts* in the **Information for Authors**.

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard <u>Terms & Conditions</u> and the <u>Ethical guidelines</u> still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.



www.rsc.org/chemcomm

55

Cite this: DOI: 10.1039/c0xx00000x

www.rsc.org/xxxxx

# **ARTICLE TYPE**

## One pot and selective intermolecular aryl- and heteroaryltrifluoromethylation of alkenes by photoredox catalysis

Aude Carboni,<sup>a</sup> Guillaume Dagousset,<sup>a</sup> Emmanuel Magnier,<sup>b</sup> and Géraldine Masson\*<sup>a</sup>

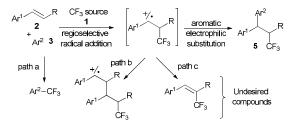
Received (in XXX, XXX) Xth XXXXXXXX 20XX, Accepted Xth XXXXXXXX 20XX 5 DOI: 10.1039/b000000x

We report herein the first photoredox-catalyzed intermolecular aryl- and heteroaryltrifluoromethylation of alkenes. Under the optimized conditions using Umemoto's reagent as CF<sub>3</sub> source, a wide range of styrenes can be readily <sup>10</sup> difunctionalized, affording the corresponding trifluoromethylated adducts in up to 99% yield.

Tandem vicinal difunctionalization of carbon-carbon double bonds has recently become an attractive and efficient strategy for the construction of two new  $\sigma$  bonds in one step.<sup>1</sup> Among these <sup>15</sup> methods, vicinal trifluoromethylfunctionalization of alkenes<sup>2</sup> has attracted much attention due to the CF<sub>3</sub> group which imparts profound changes in physical, chemical, and biological properties of organic molecules.<sup>3</sup> Thus, many radical trifluoromethylation reactions have been developed in the presence of various <sup>20</sup> nucleophiles, allowing the concomitant formation of C-O,<sup>4</sup> C-X (X = halogen),<sup>5</sup> C–N,<sup>6</sup> or C–H bonds<sup>7</sup> via the trapping of the radical intermediate (or its corresponding oxidized electrophilic cation). The carbotrifluoromethylation, which involves a carbon nucleophile in such radical-cationic processes, has also been 25 studied, especially with arenes as nucleophiles.<sup>8</sup> Although intramolecular aryltrifluoromethylation using photo-8a,b,d-f or metal-8h-j catalysis have been reported, the development of an intermolecular version of these reactions remains a significant challenge due to: (a) a potential competition between both 30 reactive species (arenes and alkenes) during the trifluoromethylation process (Scheme 1, path a);<sup>9</sup> (b) a problem of polymerization or elimination reaction of the radical or carbocation intermediate (Scheme 1, path b and c). Only one example of intermolecular aryltrifluomethylation of alkenes with 35 aryl boronic acid catalysed by copper with broad scope has been very recently described by Liu et al.<sup>10</sup> To our knowledge, the use nucleophiles in the intermolecular of arenes as trifluoromethylation has not been reported. Based on our experience in intermolecular cyanotrifluoromethylation of 40 alkenes using photoredox catalysis, 4k, 11, 12, 13 we envisioned a scenario in which the CF<sub>3</sub> radical generated by visible light

- photoreduction could selectively react with alkenes and the resulting radical or carbocation intermediate could be trapped by a suitable aryl nucleophile. To this end, we selected styrenes as 45 alkenes. Indeed, the more stable radical or carbocation in benzylic
- position would be favourably generated, thus allowing a regioand chemoselective tandem difunctionalization (Scheme 1). In addition, this approach would result in the efficient synthesis of

1,1-diarylalkane motif which is often present in various <sup>50</sup> biologically active compounds.<sup>14</sup> We report herein the first photomediated intermolecular aryltrifluoromethylation of a wide range of styrene derivatives. Importantly, extension of this efficient three-component process to the more challenging heteroaryltrifluoromethylation is furthermore presented.



Scheme 1 Synthetic approach and potential side-reactions

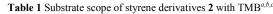
Firstly, we examined the reaction of 2-vinylnaphthalene (2a), 1,3,5-trimethoxybenzene TMB (3a) and Umemoto's reagent  $1^{15}$  as CF<sub>3</sub> source, in presence of Ru(bpy)<sub>3</sub>(PF<sub>6</sub>)<sub>2</sub> photocatalyst 4 <sup>60</sup> under visible light irradiation (5W blue LEDs). After surveying the reaction conditions by varying the catalyst, solvent, temperature, and stoichiometry, we were pleased to find that the reaction performed in CH<sub>2</sub>Cl<sub>2</sub> in the presence of 5 mol % of catalyst 4 afforded compound 5a in 69% yield (see the <sup>65</sup> Supporting Information for details).

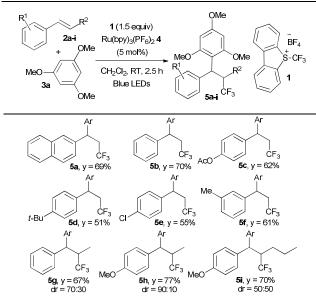
With these optimized reaction conditions, the scope of this photoredox-catalyzed intermolecular aryltrifluoromethylation was investigated and the results are summarized in Table 1. The reaction was first performed with styrene derivatives 2a-f bearing 70 various substituents on the aromatic ring, including electronwithdrawing and electron-donationg groups. Pleasingly, they were all suitable partners for this reaction, leading to the expected  $\alpha$ -aryl  $\beta$ -trifluoromethyl products **5a-f** in up to 70% yield.<sup>16</sup> Noteworthy is the perfect regioselectivity observed in the reaction 75 of internal alkenes, such as 2g-i. Moreover, these styrenes bearing a methyl or propyl group in  $\beta$  position exhibited a slightly higher reactivity compared to terminal alkenes, and reacted smoothly to afford the corresponding trifluoromethylarylated adducts 5g-i in good yields (67-77%) with moderate diastereoselectivity (from 80 50:50 to 90:10 dr). It is also worth noting that a variety of functional groups, such as halogen and ester, are compatible with the reaction conditions.

Pleasingly, this novel multi-component protocol was found to be successful with other various nucleophilic aromatic partners

MeC

MeC

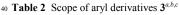




<sup>*a*</sup> Reaction conditions: styrene **2** (0.10 mmol), **1** (0.15 mmol), TMB **3a** (3 equiv), **4** (0.05 equiv), in CH<sub>2</sub>Cl<sub>2</sub> (2.0 mL) irradiated at RT for 2.5 h. <sup>*b*</sup> Yields referred to chromatographically pure product. <sup>*c*</sup> dr determined by 5 <sup>19</sup>F-NMR analysis of crude mixtures.

- (Table 2). Notably, 1,2-dimethoxybenzene (**3b**) readily participated in the reaction with terminal or internal alkenes and afforded the expected  $\alpha$ -aryl  $\beta$ -trifluoromethyl compounds **5j-m** in up to 65% yield as single *meta-para* regioisomers. Other <sup>10</sup> nucleophiles such as 1,3-dimethoxybenzene (**3c**) or *ortho*-
- methylanisole (**3d**) were also efficiently used in this photocatalyzed process, leading to the corresponding aryltrifluoromethylated adducts **5n-o** in 56-58% yield, once again with complete regioselectivity. Remarkably, this reaction could
- <sup>15</sup> be performed with the less nucleophilic anisole 3e, albeit with a lower yield (44%). Furthermore, even though an excess of arenes was used, only traces of trifluoromethylated arene were obtained (<15%), which demonstrates the high chemoselectivity and efficiency of our sequence.
- Encouraged by these results, we next sought to extend the methodology to "privileged" heteroaryl nucleophiles found in medicinal drugs and natural products such as indoles.<sup>17</sup> To the best of our knowledge, the use of heteroarene partners has never been reported in the case of aryltrifluoromethylation reactions.
- <sup>25</sup> The results are summarized in Table 3. To our delight, unprotected indole (**3f**) smoothly reacted with *trans*-anethole (**2i**) and Umemoto's reagent 1, and gave the 3-alkylated indole adduct **5q** in 62% yield. Various substituents in position 5 of the indole scaffold, such as bromine, methoxy or nitro groups, were
- <sup>30</sup> compatible with the reaction conditions. We observed that the more nucleophilic electron-rich indoles exhibited a better reactivity than electron-poor ones (52 to 66% yield). More interestingly, this photo-catalyzed process was even more efficient with indoles **3j** and **3k** bearing a phenyl or a methyl <sup>35</sup> group in position 2, respectively. We were pleased to see that the
- expected products **5u-v** were obtained in up to 99% yield.

Additionally, the reaction scope was broadened to other heteroarenes such as benzofuran (31) and pyrrole (3m), furnishing the desired trifluoromethylated adducts 5w-x in up to 50% yield.



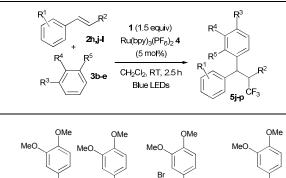
ĊF₃

ĊF₃

5n, y = 56% dr = 55:45 MeC

**5j**, y = 65%

= 65:35



ĊF3

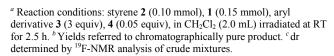
**5I**, y = 40%

ĊF₃

MeO

OMe

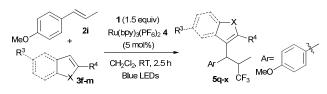
**5k**, y = 48% dr = 60:40

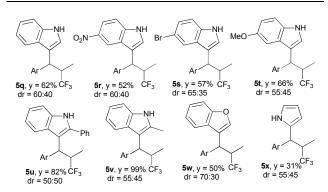


**50**, y = 58%

dr = 60:40

45 **Table 3** Scope of heteroaryl derivatives  $\mathbf{3}^{a,b,c}$ 





<sup>*a*</sup> Reaction conditions: styrene **2i** (0.10 mmol), **1** (0.15 mmol), aryl derivative **3** (3 equiv), **4** (0.05 equiv), in CH<sub>2</sub>Cl<sub>2</sub> (2.0 mL) irradiated at RT for 2.5 h. <sup>*b*</sup> Yields referred to chromatographically pure product. <sup>*c*</sup> dr determined by <sup>19</sup>F-NMR analysis of crude mixtures.

<sup>50</sup> Remarkably, even though an excess of heteroaryl nucleophiles was used, only traces of trifluoromethylated benzofuran, indole or pyrrole derivatives were observed (<10%), once again pointing out the high chemoselectivity of this multi-component process.

The following control experiments were carried out to gain ss some mechanistic insight. No reaction took place in the absence of irradiation and/or  $[Ru(bpy)_3(PF_6)_2]$  **4**. Moreover, the formation

ĊF<sub>3</sub>

5m, y = 61%

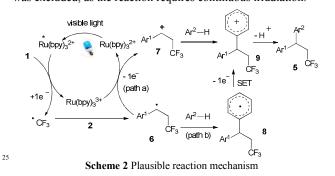
dr = 60:40 OMe

ĊF₂

**5p**, y = 44%

of **5** was inhibited in the presence of radical scavengers such as TEMPO, suggesting that a radical/cationic process is involved in this reaction. On the basis of the above results as well as other reports, a plausible reaction mechanism is shown in Scheme 2. <sup>5</sup> First, irradiation with visible light excites Ru(bpy)<sub>3</sub><sup>2+</sup> into a strong

- reductant species  $*Ru(bpy)_3^{2+}$ , which performs a single electron transfer (SET) to generate CF<sub>3</sub> from Umemoto's reagent 1.<sup>4a,6a,7a,18</sup> Subsequent regioselective addition of electrophilic CF<sub>3</sub> radical to styrene 2 furnishes the benzylic radical 6. This latter
- <sup>10</sup> species can either be oxidized into cation **7** by SET from  $\operatorname{Ru}(\operatorname{bpy})_3^{3+}$  (path a), or directly trapped by (hetero)aryl nucleophile **3**, giving radical species **8** (path b). Then, Friedel-Crafts-type alkylation of **3** by carbocation **7**, or oxidation of radical **8** by  $\operatorname{Ru}(\operatorname{bpy})_3^{3+}$ , leads to cation **9** which, after <sup>15</sup> rearomatization, affords the desired  $\alpha$ -(hetero)aryl  $\beta$ -trifluoromethyl adduct **5**. While the exact mechanism has not been fully elucidated, the failure with electron deficient arenes is in favour of the path a mechanism. Furthermore, the observed
- regioselective alkylation in C3-position of indoles **3f-i** also <sup>20</sup> supports this mechanism, as radical alkylation of indoles (path b) would generally lead to the C2-alkylated regioisomer.<sup>9a,19,20</sup> In addition, oxidation of radical species **6** or **8** by Umemoto's reagent **1** to regenerate CF<sub>3</sub> radical (radical chain propagation) was excluded, as the reaction requires continuous irradiation.



### Conclusions

In conclusion, we have successfully developed the first intermolecular aryltrifluoromethylation of styrenes using visible-<sup>30</sup> light-driven photoredox catalyst [Ru(bpy)<sub>3</sub>(PF<sub>6</sub>)<sub>2</sub>] under mild conditions. This difunctionalization protocol is suitable for the completely regioselective synthesis of a wide range of α-aryl βtrifluoromethyl products with good functional group compatibility. Terminal alkenes as well as internal alkenes are <sup>35</sup> compatible with this radical/ionic process. This method provides

a simple and efficient route to various carbotrifluoromethylated compounds which have many potential synthetic applications. Remarkably, this method was, for the first time, successfully extended to the preparation of valuable functionalized <sup>40</sup> heteroaromatic compounds in good to excellent yields.

### Aknowledgments

We thank ICSN and CNRS for financial support and doctoral fellowships to A.C.; G.D. thanks Labex Charmmat for a postdoctoral fellowship.

#### **45 Notes and references**

<sup>a</sup> Centre de Recherche de Gif, Institut de Chimie des Substances Naturelles, CNRS, 91198 Gif-sur-Yvette Cedex, France. Fax: (+) 33 1 69077247; E-mail: geraldine.masson@cnrs.fr <sup>b</sup> Latitut Laurining de Venzelles, UMD 8180, Linearcité de Venzelles

<sup>b</sup> Institut Lavoisier de Versailles, UMR 8180, Université de Versailles-50 Saint-Quentin, 78035 Versailles Cedex, France.

† Electronic Supplementary Information (ESI) available: Experimental procedures and spectroscopic data for all new compounds. See DOI: 10.1039/b000000x/

- (a) K. Muñiz, Chem. Soc. Rev, 2004, 33, 166; (b) J. P. Wolfe, Synlett,
  2008, 2913; (c) K. H. Jensen and M. S. Sigman, Org. Biomol. Chem,
  2008, 6, 4083; (d) F. Cardona and A. Goti, Nat. Chem, 2009, 1, 269;
  (e) R. I. McDonald, G. Liu and S. S. Stahl, Chem. Rev, 2011, 111,
  2981; (f) J. P. Wolfe, Angew. Chem., Int. Ed, 2012, 51, 10224.
- 2 (a) H. Egami and M. Sodeoka, Angew. Chem. Int. Ed, 2014, 53, 8294. (b) J. Xu, X. Liu and Y. Fu, Tetrahedron Lett, 2014, 55, 585; (b) E. Merino and C. Nevado, Chem. Soc. Rev, 2014, 43, 6598; (c) T. Koike and M. Akita, J. Fluorine. Chem, 2014, DOI: 10.1016/j.jfluchem.2014.06.025; (d) O. A. Tomashenko and V. V. Grushin, Chem. Rev, 2011, 111, 4475; (g) P. Chen and G. Liu, Synthesis, 2013, 45, 2919; (e) H. Liu, Z. Gu and X. Jiang, Adv. Synth. Catal, 2013,
- 355, 617; (f) T. Koike and M. Akita, *Top. Catal*, 2014, 57, 967.
  (a) D. Kirnel, M. Jung, *Flux constraints: Characteristics Constraints*, 2014, 57, 967.
- 3 (a) P. Kirsch, Modern Fluoroorganic Chemistry: Synthesis, Reactivity, Applications; Wiley-VCH: Weinheim, 2004; (b) J.-P. Bégué and D. Bonnet-Delpon, Bioorganic and Medicinal Chemistry of Fluorine; Wiley-VCH: Weinheim, 2008; (c) K. Müller, F. Fach and D. Diederich, Science, 2007, 317, 1881; (d) S. Purser, P. R. Moore, S. Swallow and V. Gouverneur, Chem. Soc. Rev, 2008, 37, 320; (e) J. Wang, M. Sanchez-Rosello, J. Luis Acena, C. del Pozo, A. E. Sorochinsky, S. Fustero, V. A. Soloshonok and H. Liu . Chem. rev, 2014, 114, 2432.
- 4 For photo-catalyzed examples, see: (a) Y. Yasu, T. Koike and M. Akita, *Angew. Chem. Int. Ed*, 2012, *51*, 9567; (b) E. Kim, S. Choi, H. Kim and E. J. Cho, *Chem. Eur. J*, 2013, *19*, 6209; (c) R. Tomita, Y. Yasu, T. Koike and M. Akita, *Angew. Chem. Int. Ed*, 2014, *53*, 7144;
- For other methods, see: (d) C. P. Zhang, Z. L. Wang, Q. Y. Chen, C. T. Zhang, Y. C. Gu and J. C. Xiao, *Chem. Commun*, 2011, 47, 6632;
  (e) R. Zhu and S. L. Buchwald, *J. Am. Chem. Soc*, 2012, 134, 12462;
  (f) C. Feng and T.-P. Loh, *Chem. Sci*, 2012, 3, 3458; (g) P. G. Janson, I. Ghoneim, N. O. Ilchenko and K. J. Szabó, *Org. Lett*, 2012, 14, 2882; (h) H. Egami, R. Shimizu and M. Sodeoka, *Tetrahedron Lett*,
- 2882; (h) H. Egami, R. Shimizu and M. Sodeoka, *Tetrahedron Lett*, 2012, *53*, 5503; (i) Y. Li and A. Studer, *Angew. Chem. Int. Ed*, 2012, *51*, 8221; (j) A. Deb, S. Manna, A. Modak, T. Patra, S. Maity and D. Maiti, *Angew. Chem. Int. Ed*, 2013, *52*, 9747; (k) A. Carboni, G. Dagousset, E. Magnier and G. Masson, *Org. Lett*, 2014, *16*, 1240.

<sup>90</sup> 5 For photo-catalyzed examples, see: (a) J. D. Nguyen, J. W. Tucker, M. D. Konieczynska and C. R. J. Stephenson, J. Am. Chem. Soc, 2011, 133, 4160; (b) C.-J. Wallentin, J. D. Nguyen, P. Finkbeiner and C. R. J. Stephenson, J. Am. Chem. Soc, 2012, 134, 8875; (c) S. H. Oh, Y. R. Malpani, N. Ha, Y.-S. Jung and S. B. Han, Org. Lett,

- 2014, 16, 1310; For other methods, see: (d) T. Xu, C. W. Cheung and X. Hu, Angew. Chem. Int. Ed, 2014, 53, 4910; (e) Z. Hang, Z. Li and Z.-Q. Liu, Org. Lett, 2014, 16, 3648.
- For photo-catalyzed examples, see: (a) Y. Yasu, T. Koike and M. Akita, Org. Lett, 2013, 15, 2136; For other methods, see: (b) H. Egami, S. Kawamura, A. Miyazaki and M. Sodeoka, Angew. Chem. Int. Ed, 2013, 52, 7841; (c) J.-S. Lin, Y.-P. Xiong, C.-L. Ma, L.-J. Zhao, B. Tan and X.-Y. Liu, Chem. Eur. J, 2014, 20, 1332; (d) F. Wang, X. Qi, Z. Liang, P. Chena and G. Liu, Angew. Chem. Int. Ed, 2014, 53, 1881.
- <sup>105</sup> 7 For photo-catalyzed examples, see: (a) S. Mizuta, S. Verhoog, K. M. Engle, T. Khotavivattana, M. O'Duill, K. Wheelhouse, G. Rassias, M. Medebielle and V. Gouverneur, *J. Am. Chem. Soc*, 2013, *135*, 2505; (b) D. J. Wilger, N. J. Gesmundo and D. A. Nicewicz, *Chem. Sci*, 2013, *4*, 3160; For other methods, see: (c) X. Wu, L. Chu and F. Qing, *Angew. Chem. Int. Ed*, 2013, *52*, 2198.
  - 8 For photo-catalyzed examples, see: (a) P. Xu, J. Xie, Q. Xue, C. Pan, Y. Cheng and C. Zhu, *Chem. Eur. J*, 2013, **19**, 14039; For other methods, see: (b) X. Mu, H.-Y. Wang, Y.-I. Guo and G. Liu, *J. Am. Chem. Soc*, 2012, **134**, 878; (c) H. Egami, R. Shimizu, S. Kawamura

75

and M. Sodeoka, *Angew. Chem. Int. Ed*, 2013, *52*, 4000; (d) H. Egami, R. Shimizu and M. Sodeoka, *J. Fluorine Chem*, 2013, *152*, 51; (e) W. Kong, M. Casimiro, N. Fuentes, E. Merino and C. Nevado, *Angew. Chem. Int. Ed*, 2013, *52*, 13086; (f) W. Kong, M. Casimiro,

- E. Merino and C. Nevado, J. Am. Chem. Soc, 2013, 135, 14480; (g)
  N. O. Ilchenko, P. G. Janson, K and J. Szabó, J. Org. Chem, 2013, 78, 11087; (h) H. Egami, R. Shimizu, Y. Usui and M. Sodeoka, Chem. Commun, 2013, 49, 7346; (i) X. Liu, F. Xiong, X. Huang, L. Xu, P. Li and X. Wu, Angew. Chem. Int. Ed, 2013, 52, 6962; (j) Z.-
- M. Chen, W. Bai, S.-H. Wang, B.-M. Yang, Y.-Q. Tu and F.-M. Zhang, *Angew. Chem. Int. Ed*, 2013, *52*, 9781; (k) Y. T. He, L.-H. Li, Y.-F. Yang, Z.-Z. Zhou, H.-L. Hua, X.-Y. Liu and Y.-M. Liang, *Org. Lett*, 2014, *16*, 270; (l) P. Gao, X.-B. Yan, F. Yang, T. He, X.-R. Song, X.-Y. Liu and Y.-M. Liang, *Chem. Eur. J*, 2013, *19*, 14420;
  (m) L. Zhang, Z. Li and Z.-Q. Liu, *Org. Lett*, 2014, *16*, 3688.
- Selected examples of trifluoromethylation of arenes: (a) D. A. Nagib and D. W. C. MacMillan, *Nature*, 2011, 480, 224; (b) R. Shimizu, H. Egami, T. Nagi, J. Chae, Y. Hamashima and M. Sodeoka, *Tetrahedron Lett*, 2010, 51, 5947; (c) T. Kino, Y. Nagase, Y.
- 20 Ohtsuka, K. Yamamoto, D. Uraguchi, K. Tokuhisa and T. Yamakawa, J. Fluorine Chem, 2010, 131, 98; (d) Y. Ye, S. H. Lee and M. S. Sanford, Org. Lett, 2011, 13, 5464; (e) R. N. Loy and M. S. Sanford, Org. Lett, 2011, 13, 2548; (f) A. Hafner and S. Bräse, Angew. Chem. Int. Ed, 2012, 51, 3713; (g) Y. Ji, T. Brueckl, R. D.
- Baxter, Y. Fujiwara, I. B. Seiple, S. Su, D. G. Blackmond and P. S. Baran, *Proc. Natl. Acad. Sci. USA*, 2011, *108*, 14411; (g) N. Iqbal, S. Choi, E. Ko and E. J. Cho, *Tetrahedron Lett*, 2012, *53*, 2005; For a review, see: (h) H. Liu, Z. Gu and X. Jiang, *Adv. Synth. Catal*, 2013, *355*, 617.
- 30 10 F. Wang, D. Wang, X. Mu, P. Chen and G. Liu, J. Am. Chem. Soc, 2014, 136, 10205.
- (a) K. Zeitler, Angew. Chem. Int. Ed, 2009, 48, 9785; (b) T. P. Yoon,
  M. A. Ischay and J. Du, Nat. Chem, 2010, 2, 527; (c) J. M. R.
  Narayanam, C.R. J. Stephenson, Chem. Soc. Rev, 2011, 40, 102; (d)
- F. Teplý, Collect. Czech. Chem. Commun, 2011, 76, 859; (e) S. Maity and N. Zheng, Synlett, 2012, 1851; (f) J. Xuan and W.-J. Xiao, Angew. Chem. Int. Ed, 2012, 51, 6828; (g) J. W. Tucker and C. R. J. Stephenson, J. Org. Chem, 2012, 77, 1617; (h) L. Shi and W. Xia, Chem. Soc. Rev, 2012, 41, 7687; (i) H. Y. Y. Xi and A. Lei, Org.
- Biomol. Chem, 2013, 11, 2387; (j) T. P. Yoon, ACS Catal, 2013, 3, 895; (k) C. K. Prier, D. A. Rankic and D. W. C. MacMillan, Chem. Rev, 2013, 113, 5322.
- 12 (a) T. Courant and G. Masson, *Chem. Eur. J*, 2012, *18*, 423; (b) G. Dagousset, A. Carboni, E. Magnier and G. Masson *Org. Lett*, 2014, *16*, 4340.
- 13 (a) G. Dagousset, J. Zhu and G. Masson, J. Am. Chem. Soc, 2011, 133, 14804; (b) A. Alix, C. Lalli, P. Retailleau and G. Masson, J. Am. Chem. Soc, 2012, 134, 10389; (c) L. He, G. Laurent, P. Retailleau, B. Folléas, J.-L. Brayer and G. Masson, Angew. Chem. Int. Ed, 2013,
- 52, 11088; (d) M. Bekkaye, Y. Su and G. Masson, *Eur. J. Org. Chem*, 2013, 3978.
- 14 For recent selected examples, see: (a) K. Gligorich, R. Vaden, D. Shelton, G. Wang, C. Matsen, R. Looper, M. Sigman and B. Welm, *Breast Cancer Res*, 2013, 15, R58; (b) S. Messaoudi, A. Hamze, O.
- Provot, B. Treguier, D. L. J. Rodrigo, J. Bignon, J.-M. Liu, J.; Wdzieczak-Bakala, S. Thoret, J. Dubois, J.-D. Brion and M. Alami, *ChemMedChem*, 2011, 6, 488; (c) A. V. Cheltsov, M. Aoyagi, A. Aleshin, E. C.-W. Yu, T. Gilliland, D.; Zhai, A. A. Bobkov, J. C. Reed, R. C. Liddington and R. Abagyan, *J. Med. Chem*, 2010, 53,
- 60 3899; (d) Q. Hu, L. Yin, C. Jagusch, U. E. Hille and R. W. Hartmann, J. Med. Chem, 2010, 53, 5049.
  - 15 Y. Macé and E. Magnier, E. Eur. J. Org. Chem, 2012, 2479.
- 16 Reaction of mono-substituted unactivated alkenes such as 1-octene led to the desired product only in poor yield (< 10%).
- 65 17 (a) E. C. Taylor, (senior Ed.); Saxton, J. E. (Ed.); *The Chemistry of Heterocyclic Compounds*, Vol. 25; Wiley-Interscience: New York, 1983, 1994; (b) R. J. Sundberg, *Indoles*; Academic Press: New York, 1996.
- (a) Y. Macé, C. Pradet, M. Popkin, J.-C. Blazejewski and E. Magnier, *Tetrahedron Lett*, 2010, *51*, 5388; (b) R. Tomita, Y. Yasu, T. Koike and M. Akita, *Beilstein J. Org. Chem*, 2014, *10*, 1099.

- 19 For examples of radical alkylation of indoles, see: (a) Y. M. Osornio, R. Cruz-Almanza, V. Jiménez-Montaño and L. D. Miranda, *Chem. Commun*, 2003, 2316; (b) L. Furst, B. S. Matsuura, J. M. R.
- Narayanam, J. W. Tucker and C. R. J. Stephenson, *Org. Lett*, 2010, *12*, 3104.
- 20 The C3 regioselectivity was unambiguously determined by COSY and NOESY experiments (see Supporting Information).