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## [3+3] Annulation of Allylic Phosphoryl-Stabilized Carbanions / Phosphorus Ylides and Vinyl Azide: A Practice Strategy for Polyfunctionalized Anilines

Shen Liu,<sup>‡</sup> Wenteng Chen,<sup>‡</sup>Jin Luo and Yongping Yu<sup>\*</sup>

#### Tandem Michael addition and Witting or Horner-Wadsworth-Emmons olefination initiated [3+3] annulation between vinyl azides and allylic phosphorus ylide or allylic phosphoryl-stabilized carbanions has been developed. This one-pot protocol furnishes highly functionalized anilines in good to excellent yields under mild, room-temperature conditions. A rational mechanism is also proposed.

The aniline skeleton is an intriguing synthetic target because of its abundance in biologically active natural products, pharmaceuticals, pesticide, and dyes. <sup>[1-2]</sup> In industry, electrophilic nitration of arenes followed by a reduction is a key technology for aniline formation though it is limited by essential position directing and a relatively narrow substrates scope. <sup>[3]</sup> On the other hand, the Buchwald-Hartwig amination reaction has been extensively utilized for generating anilines. These coupling reactions are commonly catalysed by metal and phosphine ligands <sup>[4]</sup> either using an indirect nitrogen source <sup>[5]</sup> or by direct amination using ammonia. <sup>[6]</sup> In addition to these conventional modification on benzene rings, cross coupling of malononitrile and some activated moieties, such as amethylene ketones, ynones and nitrostyrenes are employed for multisubstituted aniline formation.<sup>[7]</sup> Despite these advances, the control of substituents on the aniline ring, the limited source of the starting materials and the development of milder reaction conditions still remains a challenge. Thus, the development of novel efficient strategies for the construction of polyfunctionalized anilines is highly desirable.

In the search of alternatives to the cross-coupling of saturated Nheterocycles, our group and others have reported the use of a vinyl azide as a synthon for the formation of nitrogen-containing heterocycles, such as pyrrolo  $[1, 2-\alpha]$  pyrazine, <sup>[8a]</sup> 4-aminopyridines, <sup>[8b]</sup> pyrazoles, <sup>[8c]</sup> or other heterocycles. <sup>[8d-8g]</sup> The continued interest in eletrophilicity reactivity of vinyl azides motivated us to investigate the reactions of vinyl azides with allylic phosphorus ylides or phosphoryl-stabilized carbanions, the latter of which is frequently investigated for its C-C bond formation. <sup>[9-10]</sup> Herein, a [3+3] cascade transformation is reported under mild reaction conditions, which provides a convenient method for the synthesis of polyfunctionalized anilines. Additionally the exceptional substrate scope of the reaction, which accepts heteroaryl, halogenated, nitro-containing substrates is reported.

Initially, the reaction of crotonate-derived phosphorus ylides (1a) and (Z)-2-azido-1-(4-nitrophenyl)-3-phenylprop-2-en-1-one (2a) was selected as a model to optimize the reaction conditions. Treatment of a mixture of vinyl azide 2a (1.0 equiv) and phosphorus ylides 1a (1.0 equiv) with EtONa (3.0 equiv) in *i*-PrOH at room temperature resulted in the desired aniline 3a with the conversion of 15% along with unidentified complex mixtures that included phosphazene via a Staudinger-Meyer reaction. <sup>[11]</sup> (Table 1, entry 1). When the reaction was carried out in other protic solvents, the conversion of 3a was slightly improved to 18% (n-butanol) and 52% (CH<sub>3</sub>OH), respectively (Table 1, entries 2-3). Assessment of the reaction with additional solvents revealed that THF and DCM worked more efficiently, giving 3a in 65% and 85% yield, respectively (Table 1, entries 4 and 9). The use of other bases, such as Et<sub>3</sub>N, DBU and NaOH were less effective for this transformation (Table 1, entries 10-11, 13). In fact when the reaction was carried out in the presence of Cs<sub>2</sub>CO<sub>3</sub>, only a trace amount of the desired product **3a** could be detected via LC-MS (Table 1, entry 12). As shown in Table 1, the highest yield (85%) was obtained when the reaction was performed in DCM at room temperature for 1.0 h using 3.0 equivalent of EtONa (Table 1, entry 9).

This facile annulation reaction encouraged us to explore the generality of the reaction for the synthesis of a variety of polyfunctionalized anilines. A series of substituted vinyl azides (2a-20) and phosphonium ylide (1a) were employed as cyclization

substrates and subjected to the previously optimized reaction conditions (Table 2).

#### Table 1. Optimization of the reaction condition <sup>[a]</sup>



Entry	Base	Solvent	T <sup>/o</sup> C	Conversion <sup>[b]</sup>
1	EtONa	i-propanol	rt	15
2	EtONa	n-butanol	rt	18
3	EtONa	CH <sub>3</sub> OH	rt	52
4	EtONa	THF	rt	$70(65)^{[c]}$
5	EtONa	DMF	rt	26
6	EtONa	CH <sub>3</sub> CN	rt	58
7	EtONa	1,4-dioxane	rt	45
8	EtONa	Toluene	rt	47
9	EtONa	DCM	rt	87 (85) <sup>[c]</sup>
10	Et <sub>3</sub> N	DCM	rt	19
11	DBU	DCM	rt	41
12	$Cs_2CO_3$	DCM	rt	trace
13	NaOH	DCM	rt	75(70) <sup>[c]</sup>

[a] Unless otherwise specified, reactions were performed using 1a (0.1 mmol) and 2a (0.1 mmol) in various solvent (1.0 mL) at rt in the presence of the base (0.3 mmol) for 1.0 h. [b] The conversion rate was determined by HPLC, based on the disappearance of the starting vinyl azide (2a). [c] Isolated yield

o

#### Table 2. Scope of vinyl azides 2

, or the second	⊖ PPh3 E	$B_{r}^{\ominus} + R_{1} + R_{2}$	EtONa(3.0 equiv)	
1a (1.0 equiv.)		2 (1.0 equiv.)	DCM/IT	NH2 3
Entry 2		R <sub>1</sub>	R <sub>2</sub>	Product / Isolated vield
1	2a	C <sub>6</sub> H <sub>5</sub>		3a (76%)
2	2b	4-MeSC <sub>6</sub> H <sub>5</sub>		3b (68%)
3	2c	4-MeOC <sub>6</sub> H <sub>5</sub>	4	3c (69%)
4	2d	3-MeC <sub>6</sub> H <sub>5</sub>		3d (73%)
5	2e	4- $MeC_6H_5$		3e (72%)
6	2f	$2-BrC_6H_5$		3f (83%)
7	2g	$3-BrC_6H_5$		3g (79%)
8	2h	$4-BrC_6H_5$	4- NO C H	3h (86%)
9	2i	$4-ClC_6H_5$	NO <sub>2</sub> C <sub>6</sub> II5	3i (82%)
10	2j	$2-FC_6H_5$		3j (78%)
11	2k	2-Cl-6-FC <sub>6</sub> H <sub>5</sub>		3k (81%)
12	21	3-(4-MeC <sub>6</sub> H <sub>5</sub> O)C <sub>6</sub> H <sub>5</sub>		31 (70%)
13	2m	$3-(C_6H_5O)C_6H_5$		3m (72%)
14	2n	$4-(C_6H_5O)C_6H_5$		3n (71%)
15	20	$4-(4-C_6H_5CH_2O)C_6H_5$		30 (67%)

Various vinyl azides 2a-2o, including those with nitro or halogen moieties, provided the corresponding multifunctionalized anilines 3a-3o in good to excellent yields. The reaction with vinyl azides, bearing electron-withdrawing group at R1 position, resulted in the corresponding [3+3] annulation products (Table 2, compounds 3f, 3g, **3h**, **3i**, **3j**, and **3k**) in good yields. Likewise, the  $\beta$ -electron-donating group substituted vinyl azides are capable of affording the desired

anilines, albeit in lower yields (Table 2, compounds 3b, 3c, 3d, 3e, **31**, **3m**, **3n** and **3o**). When varying the substitution position on the  $\beta$ aryl vinyl azides 2, the corresponding potential steric effects are tolerated by the present method and provided the desired products in good yields (Table 2, compounds 3f, 3j, 3k).

We then turned our attention towards the scope of viable  $R_2$  of vinvl azides 2 substitutions under the current conditions (Scheme 1). This turned out to be rather challenging (Scheme 1). When the  $R_2$ position of the vinyl azides 2 was substituted with a phenyl group, less than 10% of this transformation occurred under the reaction conditions (Scheme 1, eq 1). The poor result was probably due to the weak nucleophilicity of allylic phosphonium ylide.

To our knowledge, phosphoryl-stabilized carbanions are generally more reactive than phosphorus ylides. <sup>[12]</sup> We thus examined the use of phosphoryl-stabilized carbanions (1b) instead of phosphorus vlides (1a) (Scheme 1, eq 2) which increased the yield of the desired product to 50% (Table 3, entry 1). The desired product 3p was obtained in a higher isolated yield (Table 3, entry 2) when MeONa was utilized as the base. Further increasing the amount of base slightly improved the isolated yield (Table 3, entry 3). As shown in Table 3, MeONa and DCM seemed to be the best choice for the reaction of  $\alpha$ -phenyl substituted vinyl azide 2p with phosphorylsubstituted carbanions 1b providing an excellent yield of 78%.

#### Table 3. Optimization of the reaction condition <sup>[a]</sup>

0 0 0 0 0 P 0 1b (1.0 equiv.)	+ $N_3$ $DCM/rt$ 2p (1.0 equiv.)			
Entry	Base	Equivalent	Solvent	Conversion <sup>[b]</sup>
1	EtONa	3.0	DCM	56(50) <sup>[c]</sup>

2	MeONa	3.0	DCM	75(70) <sup>[c]</sup>
3	MeONa	5.0	DCM	83(78) <sup>[c]</sup>
[a] Reactions	were perform	ned using	<b>1b</b> (0.1 mmo	l) and <b>2a</b> (0.1
mmol) in vari	ous solvent (1	.0 mL) at r	t in the preser	nce of the base
for 1.0 h. [b]	The conversio	n rate was	determined by	HPLC, based
on the disapp	earance of the	e starting v	inyl azide (2a	a). [c] Isolated
vield				

#### Scheme 1. Generality of [3+3] Annulation



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 Table 4 Generality of [3+3] annulation of vinyl azides with allylic

 phosphoryl-stabilized carbanions

0 R <sub>3</sub> _0	0   	+ R <sub>1</sub> N <sub>3</sub> 2(1.0 e	O L R <sub>2</sub> <u>MeONa (5.0 equiv.)</u> DCM/rt	
Entry	2	<b>R</b> <sub>1</sub>	<b>R</b> <sub>2</sub>	Product / Isolated yield
1	2a	$C_6H_5$	$4-NO_2C_6H_5$	3a (85%)
1	2p	$C_6H_5$	$C_6H_5$	3p (78%)
2	2q	$2-BrC_6H_5$	$C_6H_5$	3q (79%)
3	2r	$3-BrC_6H_5$	$C_6H_5$	3r (80%)
4	2s	$4-BrC_6H_5$	$C_6H_5$	3s (83%)
5	2t	2-pyridinyl	$C_6H_5$	3t (65%)
6	2u	2-furyl	$C_6H_5$	3u (70%)
7	2v	$C_6H_5$	$4-MeC_6H_5$	3v (76%)
8	2w	$C_6H_5$	$4-BrC_6H_5$	3w (83%)
9	2x	$C_6H_5$	$C(CH_3)_3$	3x(trace)
10	2у	$C_6H_5$	OEt	3y (0%)
11	2z	$C_6H_5$	Н	3z (0%)

The results encouraged us to explore the expanded-scope of this annulation. The reaction of methyl (E)-4-(dimethoxyphosphoryl) but-2-enoate (1b) with various  $\alpha$ ,  $\beta$ - substituted vinyl azides (2) under the standard reaction conditions (given in Table 3) gave the multisubstituted anilines (3) in good to excellent yields (Table 4, isolated yields >70%). Both electron-rich and -deficient aryl groups at R<sub>2</sub> position were tolerated in this annulation. On the contrary, when  $\alpha$ -tributylcarbonyl-substituted vinyl azide 2x was reacted with 1b, only a trace amount of the corresponding aniline 3x was observed via LC-MS. Likewise, when the  $\alpha$ -ester substituted vinvl azide 2v was employed, no desired aniline 3v was observed. These results suggested that the poor eletrophilicity of these vinyl azides did not enable a HWE-olefination to form the desired anilines. While, the strong reactivity of aldehyde group at the  $\alpha$ -position of vinyl azide 2z favoured an intermolecular HWE olefination over a [3+3] annulation under the optimization reaction conditions. In general though the tandem reaction showed broad tolerance for various R<sub>1</sub> groups of vinyl azides, a selected substrates bearing phenyl (2p, 2v and 2w), electron-deficient aryl (2q, 2r and 2s), and heteroaryl (2t and 2u) at  $R_1$  groups, reacted efficiently with 1b to provide the corresponding anilines (3p-3u) in high yields at room temperature for 1.0 h.

The structures of the polyfunctionalized anilines were characterized by <sup>1</sup>H NMR, <sup>13</sup>C NMR and HRMS. The structures of one representative compound 3c were proved by X-ray crystal structural analysis as shown in Figure 1 (CCDC No.: 1005572).

The tandem reaction sequence outlined in Scheme 2 was proposed to explain this facile transformation. Nucleophilic attack of ylide **1a** or phosphoryl-stabilized carbanions **1b** with vinyl azides **2** initiates the formation of addcut **III** or **V** in the presence of base. Adduct **III** exclusively undergoes 1,6-proton shift to convert to intermediate **IV**, while intermediate **VI** is formed from adduct **V** via 1, 3-proton transfer. The annulation process is followed by an intramolecular Witting or HWE olefination and subsequent proton transfer to afford the corresponding aniline **3**.



Figure 1 X-Ray crystal structure of 3c

Scheme 2 Proposed mechanism for this annulation



#### Conclusions

In summary, a base-promoted [3+3] annulation between allylic phosphoryl-stabilized carbanions or phosphorus ylides with vinyl azides was developed under mild, room temperature conditions. This facile reaction allows construction of polyfunctionalized anilines in good yields. Both the substituents and positions on the anilines can be easily controlled by choosing appropriate and accessible starting materials. A rational mechanism was also proposed. It is anticipated that the present annulation is potential useful for synthesis of biologically and medicinally important anilines.

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#### Notes and references

Zhejiang Province Key Laboratory of Anti-Cancer Drug Research, College of Pharmaceutical Science, Zhejiang University, Hangzhou 310058, P. R. China. E-mail: yyu@zju.edu.en; Tel: +86-571-88208452.

Shen Liu and Wenteng Chen contributed equally to this work.
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