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COMMUNICATION

Rhodium(III)-catalyzed intramolecular amidoarylation and hydroarylation of alkyne via C–H activation: Switchable synthesis of 3,4-fused tricyclic indoles and chromans ‡

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The controllable intramolecular amidoarylation and hydroarylation of alkynes has been achieved via rhodium(III)-catalyzed C–H activation. The merger of two distinct reaction pathways allows for the development of 10 atom- and step-economic protocols for the switchable synthesis of 3,4-fused indoles and chromans, respectively.

The 3,4-fused tricyclic indole structural motif forms the core of many natural products with pharmacological relevance, such asfargesine, ¹ dehydrobufotenine,² welwistatin,³ lysergic ¹⁵ acid,⁴dragmacidin E,⁵ decursivine,⁶ communesin F,⁷and indolactam V.⁸Theformation of the 3,4-fused indole framework generally involves building the third ring onto a preformed indole moiety.²⁻¹⁵ Very recently, Boger¹⁶ and Jia¹⁷ independently reported a palladium-catalyzed intramolecular Larock indole ²⁰ process for the preparation of suchpolycyclic indoles from 2-

bromo- or 2-iodoanilines with an alkyne tethered at 3-position (Eqn. 1).

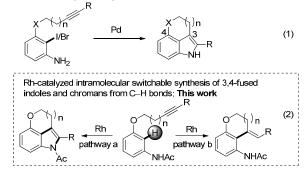
Due to its highatom- and step-economy, the rhodium(III)catalyzed C–H activation has received significant interest in ²⁵ recent years.¹⁸ In 2008, Fagnouand co-workers reported a novel strategy for indole synthesis via a Rh-catalyzed intermolecular amidoarylation of internal alkynes with acetanilides.¹⁹ Other directing strategies for the divergent synthesis of N-H and Nalkyl indoles were later developed.²⁰ On the other hand, ³⁰ (Cp*RhCl₂)₂ hasemerged as an efficient catalystfor alkyne hydroarylationwith aryl C–H bond via a novel concerted deprotonation–metalation pathway.²¹ Several examples of intermolecular hydroarylationof alkynes haverecently been

- reported.^{21,22} Although the Rh(III)-catalyzed intramolecular ³⁵ amidoarylation and hydroarylation of alkenes were documented last year,²³ to our knowledge, the controllable intramolecular version of above two pathways of alkyne have been remained unexplored. As part of our continuing interest in the development of Rh(III)-catalyzed C–H activation,²⁴ herein, we present the
- ⁴⁰ Rh(III)-catalyzed intramolecular amidoarylation and hydroarylation for the switchable synthesis of 3,4-fused tricyclic

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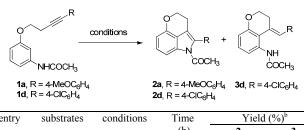
Pd-catalyzed intramolecular synthesis of 3,4-fused indoles from C–I/Br bonds (ref. 15,16): **Prior work**



indoles (pathway a) and chromans (pathway b) from alkyne tethered acetanilidesvia C–H activation (Eqn. 2). It is worth to ⁴⁵ mention that the construction of two distinct types of complex moleculesfrom the identical starting materials areachieved simply by a slight change of reaction conditions.

Initially, *N*-(3-((4-(4-methoxyphenyl)but-3-yn-1yl)oxy)phenyl)acetamide**1a** was subjected to the Fagnou's ⁵⁰ intermolecular reaction conditions^{19a} (Table 1, entry 1). We found that treatment of **1a** with (Cp*RhCl₂)₂ (1 mol%), AgSbF₆ (4

Table 1 Optimization of reaction conditions^a



entry	substrates	conditions	Time	Yield $(\%)^{\circ}$	
			(h)	2	3
1	1a	А	0.3	2a (87)	
2	1d	А	0.3	2d (26)	3d (51)
3	1d	В	3.0	2d (85)	
4	1d	С	0.3		3d(98)
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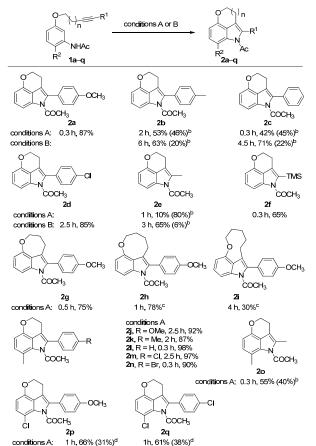
^aReactions conducted on 0.2 mmol scale. Conditions A: $(Cp*RhCl_2)_2$ (1 mol%), AgSbF₆ (4 mol%), Cu(OAc)₂•H₂O (2.1 equiv), *t*-AmOH (0.1 M), 120 °C; conditions B: $(Cp*RhCl_2)_2$ (5 mol%), AgSbF₆ (20 mol%), Cu(OAc)₂•H₂O (2.1 equiv), CH₃CN (0.1 M), 120 °C; conditions C: $(Cp*RhCl_2)_2$ (2.5 mol%), AgSbF₆ (10 mol%), PivOH (5.0 equiv), *t*-AmOH (0.1 M), 120 °C. ^bYields ofIsolated products.

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mol%) and Cu(OAc)₂•H₂O (210 mol%) in *t*-AmOH at 120°C gave the desired 3,4-fused tricyclic indole **2a** in 87%yield(Table 1, entry1, condition A). However, under the identical conditions the substrate N-(3-((4-(4-chlorophenyl)but-3-yn-1-

- ⁵ yl)oxy)phenyl)acetamide1d gave the amidoarylation product2d only in 26% yield along with the hydroarylation product 3d in 51% yield (Table 1, entry 2). These results indicated that the electronic effect of the aryl groups attached to the triple bond have a strong effect on the pathways of amidoarylation and hydroarylation.
- ¹⁰ Then, the reaction conditions of both the amidoarylation product **2d** and the hydroarylation product **3d** were further optimized. Aprotic solvent, acetonitrile increased the yield of tricyclic indole **2d** to 85% (Table 1, entry 3, conditions B), and replacement of $Cu(OAc)_2$ with pivalic acid (5.0 equiv)²¹ resulted in the ¹⁵ hydroarylation product **3d** exclusively (Table 1, entry 4,
- conditions C). With the optimal conditions in hand, we surveyed various substrates to determine the scope of the amidoarylation reaction. Under the reaction conditions A or B, the reactions proceeded
- ²⁰ smoothly to afford tricyclic indoles **2** in good to excellent yields (Table 2). The substituents on the alkyne (R¹ group) were well

Table 2Synthesis of tricyclic indoles 2a



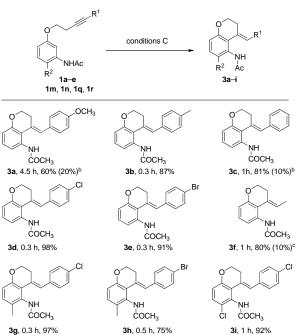
^aReactions conducted on 0.2 mmol scale. Conditions A: (Cp*RhCl₂)₂ (1 mol%), AgSbF₆ (4 mol%), Cu(OAc)₂•H₂O (2.1 equiv), *t*-AmOH (0.1 M), 120 °C; Conditions B: (Cp*RhCl₂)₂ (5 mol%), AgSbF₆ (20 mol%), Cu(OAc)₂•H₂O (2.1 equiv), CH₃CN (0.1 M), 120 °C. ^bYields of the corresponding hydroarylation products **3** are in parentheses. ^cConditions A, (Cp*RhCl₂)₂ (10 mol%), 0.01 M. ^dYields of products **2a** and **2d** are in parentheses (from the dechlorination of **2p** and **2q** respectively).

- ²⁵ tolerated with electron-rich and electron-deficient aryl, phenyl, alkyl and trimethylsilyl groups and gave the tricyclic indoles in good to high yield (**2a–f**)along with thehydroarylation by-products in the cases of **2b**, **2c** and **2e**. These results reveal that the amidoarylation pathwayis quite sensitive to the electronic ³⁰ effect of the substituents on the alkyne.In comparison, substrates with both electron-rich and electron-deficient R² groups resulted in the amidoarylation products exclusively in excellent yields under conditions A,no matter which substituents were attached to the alkyne group (**2j–q**). It is most likely that the steric effect of ³⁵ the R² groups facilitate this amidoarylation process. Compared
- with the palladium-catalyzed 3,4-fused tricyclic indole synthesis,¹⁷ the rhodium(III)-catalyzed protocol is more practical due to its low catalystloading (1 mol% in most cases vs 20 mol%) and without high dilution (0.1 M vs 0.01 M). In addition, The 40 intramolecular reaction could be extended togenerate 3,4-
- medium-ring fusedindoles (**2g–i**), which are especially difficult to prepare.²⁵

The intramolecular amidoarylation process mentioned above represents a very simple and efficient methodology for the ⁴⁵ construction of 3,4-fused tricyclic indoles which is highly atomand step-economic. Since the construction of distinct types of

- complex molecules from identical starting materials is an attractive and challenging task in organic synthesis,²⁶ the preparation of chromans **3** from the hydroarylation reactions of ⁵⁰ *N*-(3-((but-3-yn-1-yl)oxy)phenyl)acetamides**1**was also
- investigated. In the presence of $(Cp*RhCl_2)_2$ (2.5 mol%), AgSbF₆ (10 mol%) and pivalic acid (5.0 equiv), the reactions smoothly proceeded at 120 °C to afford chromans in good to excellent yields(Table 3). Substrates with different substituents on the ss alkyne (R¹), such as electron-rich and -deficient aryl, phenyl and

Table 3Synthesis of chromans 3



^aReactions conducted on 0.2 mmol scale. Conditions C: (Cp*RhCl₂)₂ (2.5 mol%), AgSbF₆ (10 mol%), PivOH (5.0 equiv), *t*-AmOH (0.1 M), 120°C. ^bPivOH (1.0 equiv), yields of the corresponding tricyclic indoles **2** are in parentheses. ^cConditions A, yields of tricyclic indoles **2e** is in parentheses.

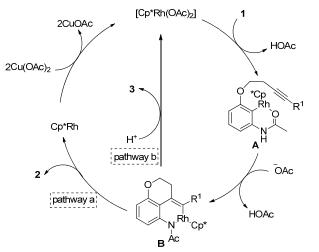
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alkyl groups were well tolerated (**3a–f**). Substrates with both electron-donating and electron-withdrawing R^2 groups participated in this reaction (**3g–i**).Unlike the amidoarylation, the hydroarylation pathway has slight effect on the electronic and

- s steric effectof the R¹ and R² substituents. Furthermore, these reactions gave the alkene products with *E* selectivity, and the results are consistent with those from the intermolecular version of the hydroarylation reaction.²¹ The configuration of **3f** was established by the X-ray single crystal analysis,²⁷ and others were
- ¹⁰ determined by analogy with their NMR spectra.Migratoryinsertion of alkyne to form an alkenylrhodium intermediate,

On the basis of the above results (Table 1–3) and the related work, 19,21,23 a mechanistic pathway is proposed (Scheme 1). First,

- ¹⁵ C–H bond cleavage of **1** occurs to produce a six-membered rhodacycle intermediate **A**. Next, Alkyne coordination to rhodium and migratory insertion of alkyne into the rhodium–carbon bond results in the formation of intermediate **B**. Then two pathways may exist, in pathway a, the carbon-nitrogen
- ²⁰ bond is formed to produce the tricyclic indoles **2** after reductive elimination, at which time the Rh(III) is reduced to Rh(I), reoxidation of the reduced catalyst with the copper(II) oxidant restores the catalytically active rhodium(III)-complex. In pathway b, intermediate **B** is protonated by the pivalic acid to give the
- ²⁵ corresponding alkene derivative **3**withregeneration of the catalyst.In addition, the hydroarylation product **3g** was treated with the amidoarylation conditions (Table 1 and 2, conditions A) for 10 h, the corresponding tricyclic indole product **2m** could not be detected. This result indicates that the hydroarylation product
- ³⁰ **3** is not the intermediate for the formation of tricyclic indole product **2** from the amidoarylation of **1** under these conditions.



Scheme 1 Proposed mechanism for the formation of 2 and 3.

- In summary, we have developed a Rh(III)-catalyzed ³⁵ intramolecular amidoarylation and hydroarylation for the switchable syntheses of 3,4-fused tricyclic indoles and chromans from alkyne tethered acetanilides via aryl C–H bonds activation. In this process, two distinct types of complex moleculesfrom the identical starting materials areachieved simply by a slight change
- ⁴⁰ of reaction conditions. The reaction features atom- and stepeconomy, high product yields and practical procedure. The synthetic applications of this protocol are being carried out in our

laboratory.

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