ChemComm

Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this Accepted Manuscript with the edited and formatted Advance Article as soon as it is available.

You can find more information about *Accepted Manuscripts* in the **Information for Authors**.

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard <u>Terms & Conditions</u> and the <u>Ethical guidelines</u> still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.



www.rsc.org/chemcomm

Cite this: DOI: 10.1039/c0xx00000x

ARTICLE TYPE

Enantioselective cascade reaction between α , β -unsaturated aldehydes and malonic half-thioesters: rapid access to chiral δ -lactones

Qiao Ren,^{a,b} Shaofa Sun,^a Jiayao Huang,^b Wenjun Li,^b Minghu Wu,^a Haibing Guo^a and Jian Wang^{a,b}

Received (in XXX, XXX) Xth XXXXXXXX 20XX, Accepted Xth XXXXXXXX 20XX 5 DOI: 10.1039/b000000x

We disclose a novel efficient enantioselective organocatalytic cascade reaction for the preparation of δ -lactones in good to excellent yields (69-93%) and with high to excellent enantioselectivities (88-96% *ee*).

¹⁰ The assembly of *O*-heterocycles is an important field of research due to their prevalence in natural products and drugs.¹ Of the *O*-heterocycles, δ-lactones are well-known six-membered oxygen-



Figure 1: Examples of natural products or drugs contacting $\delta\mbox{-lactone}$ moiety.

containing heterocycles that form the structural scaffold of many biologically active molecules,² hence, they are broadly used in ¹⁵ medicinal chemistry as important structural elements. Notably, a large number of δ -lactones involved biologically active important molecules exist in a single enantiomer.^{2,3} As exemplified in Figure 1, Lovastatin⁴ **a** marketed by Merck under the trade name Mevacor, is a member of the drug class of statins, used in the ²⁰ treatment of dyslipidemia and the prevention of cardiovascular

disease. Artemisitene⁵ **b**, isolated from Artemisia annua L., has



Scheme 1: Efficient synthesis of chiral δ -lactones and related transformations.

been shown to possess antimalarial activity. Crassin acetate⁶ \mathbf{c} , a lactonic cembrane diterpene, has been shown to be the principal antineoplastic agent present in the marine invertebrates. ²⁵ Leiodermatolide⁷ \mathbf{d} is a structurally unique macrolide isolated

- ²⁵ Leiodermatolide **d** is a structurally unique macrolide isolated from the deep-water marine sponge *Leiodermatium* sp. which exhibits potent antiproliferative activity against a range of human cancer cell lines and drastic effects on spindle formation in mitotic cells.
- ³⁰ Notably, chiral pharmaceuticals are figuring largely in drug market. Most of beneficial drugs are sold as a single enantiomeric form because of the higher efficiency than its mirror image enantiomer and displaying better fit to its receptor. In addition, it may even be encountered with harmful results with the other
- $_{35}$ enantiomer.⁸ Given the fact that within a chiral surrounding two enantiomers often show distinct biological activity, the development of effective protocols to access optical pure δ lactones would be extremely desirable to further study the correlation between the chirality of these compounds and their

⁴⁰ propensities for biological activities to seek more potent and/or appropriate pharmaceutical candidates. In our continuing effort towards the development of new approaches for the stereoselective construction of enantiopure synthetically useful building blocks,⁹ we thought about expanding the scope of ⁴⁵ organocatalytic decarboxylative reactions^{10,11} to α,β -unsaturated aldehydes¹². Although many methods have been reported on chiral δ -lactone synthesis, most of them were prepared for the assembly of α,β - or γ,δ -unsaturated δ -lactones.^{13,14} Herein, we disclose a direct and efficient enantioselective cascade strategy

 $_{50}$ for the saturated functional δ -lactone synthesis (Scheme 1).

Scheme 2: Initial Experiments.



Reaction Conditions: **1a** (0.20 mmol), **2** (0.24 mmol), TEA (0.20 mmol), Cat. **II** (0.20 mmol), DCM (1.0 mL), 72 h, room temperature.

During our initial studies on the feasibility of amine-catalyzed intermolecular reactions of malonic acid mono-protected esters (2a-b) with cinnamaldehyde (1a), no decarboxylative adducts

Table 1: Screened catalysts.





- were observed (Scheme 2). In the context of exploring a plausible ⁵ catalytic process, we turned our attention to investigate malonic acid analogous. As shown in Scheme 2, the malonic half-thioester (**2c**) was found to be a suitable partner to allow the reaction to promote smoothly in the presence of prolinol catalyst **II** (Table 1). Interesting, an unexpected δ -lactone (**3ac**) was identified as a
- ¹⁰ major product (59% yield, 72% *ee*). Screening of catalysts revealed that the size of silyl ether moiety in catalysts (Cat. I-VI) is critical for the stereocontrol. As outlined in Table 1, the most bulky catalyst *tert*-butyltrimethylsilane ether VI (TBDMS) gave the best result (Table 1, 89% *ee*, 51% yield). Subsequent solvent ¹⁵ survey (Table 2, entries 1 and 4–6) indicated that DCM is an
- *ideal medium for this asymmetric transformation (entry 7, 89% ee). Other parameters (e.g. bases) gave no improvement on ee*

Table 2: Optimization of other reaction parameters.^a



^{*a*} Reactions were performed with **1a** (0.20 mmol), **2c** (0.24 mmol), additive (0.20 mmol) in the solvent (1.0 mL). ^{*b*} Isolated yield. ^{*c*} Determined by chiral HPLC analysis. ^{*d*} **2c** (0.50 mmol), TEA (0.50 mmol). ^{*e*} TEA dissolved in 0.50 mL of DCM was added dropwise in 10 min, 0 °C. ^{*f*} Isopropanol. ^{*g*} 1,5,7-Triazabicyclo[4.4.0]dec-5-ene.

value (Table 2, entries 2–3). It is noteworthy that acyclic 3phenyl 1-aldehyde 5-thioester **4** was formed as a major product ²⁰ when Na₂CO₃ instead TEA. Gratify, the dropwise addition pattern of TEA gave a better result (92% *ee*, 82% yield). Controlled experiment indicated that one-pot addition of TEA could cause a rapid decomposition of malonic half-thioester **2c**. Table 3: Substrate scope of α , β -unsaturated aldehydes.^{a-d}



^{*a*} Reaction conditions: **1a-k** (0.20 mmol), **2c-h** (0.50 mmol), TEA (0.50 mmol) and catalyst **VI** (0.04 mmol), DCM (1.0 mL), 0 °C for 24 h. ^{*b*} Isolated yield. ^{*c*} Determined by chiral HPLC analysis. ^{*d*} all dr > 20:1.

After having optimized the conditions for the cascade reaction, 25 the scope of the methodology was evaluated. Initially, various α,β -unsaturated aldehydes 1 with different substitution patterns and electronic properties were examined (Table 3). The substitution patterns of malonic half-thioesters had limited impact on the yields or enantioselectivities, thereby providing the only 30 regioisomer with ee values ranging from 91 to 94% (Table 3, **3ac–ah**, all dr > 20:1). Pleasingly, α,β -unsaturated aldehydes **1a–** h bearing different electron-withdrawing substituents on the aromic ring were easily reacted under the established reaction conditions (3ac-hc). In most of the cases, good yields and high ³⁵ steroselectivities were obtained.¹⁷ Additionally, α , β -unsaturated aldehydes 1i-k bearing heterocyclic rings and alkyl groups could be utilized as well in good yields with rigorous steroselectivities (3ic-kc)¹⁵ Additionaly, $\alpha, \beta, \gamma, \delta$ -unsaturated aldehyde 11 could react with 2c to afford 3-alkenyl \delta-lactone 3lc in 74% yield and 40 with 80% ee (eqn (3)).

$$0^{\text{Omega}} Ph + HO \frac{2c}{2c} SPh \frac{\text{Standard}}{74\%} \frac{Ph}{3lc} O (3)$$

Further application study was embarked on examining the applicability of our method to the concise synthesis of (-)-Paroxetine, marketed as Paxil/Seroxat (Scheme 3). The reaction of **1m** with **2c** catalyzed by **VI** and TEA gave the corresponding **3mc** in 72% yield and 91% *ee* (Scheme 3). Next, Ni-catalyzed ring opening reaction gave the corresponding 3-substituted 1,5-dicarbonyl **17**, and a subsequent reduction of **17** afforded the key intermediate **18**. According to reported procedure,¹⁶ the intermediate **18** can be converted to target (-)-Paroxetine, an ⁵⁰ antidepressant drug.

In summary, we have established a new method for the enantioselective cascade synthesis of δ -lactone in the presence of chiral prolinol or cinchona alkaloid amine organocatalysts. Our methodology provided a straightforward way to generate valuable ⁵⁵ chiral δ -lactone scaffold, which has potentially biological

significance in the field of medicinal chemistry. The ready accessibility of the starting materials makes the current methodology particularly attractive in organic synthesis. Studies towards expanding the synthetic utility of this strategy are ⁵ currently underway.

Scheme 3: Formal total synthesis of (-)-Paroxetine (Paxil/Seroxat).



Financial support from Singapore Ministry of Education (Academic Research Grant: R143000443112, R143000532112) and China Hubei Collaborative Innovation 2011 Project and National Nature Science Foundation of China (21202136) are 10 greatly appreciated.

Notes and references

^aHubei Collaborative Innovation Centre for Non-power Nuclear Technology, College of Chemistry and Biological Sciences, Hubei University of Science and Technology, Hubei Province 437100, People's 15 Republic China

^bDepartment of Chemistry,National University of Singapore, Block S15, Level 5, 3 Science Drive 3, Singapore 117543. Fax: 65-6779-1691; E-mail: <u>chmwangj@nus.edu.sg</u>

† Electronic Supplementary Information (ESI) available: [details of any
 ²⁰ supplementary information available should be included here]. See DOI: 10.1039/b000000x/

- For selected books, see: (a) K. C. Nicolaou, T. Montagnon, Molecules that changed the World, Wiley-VCH, Weinheim, 2008; (b)
 P. M. Dewick, Medicinal Natural Products, Wiley-VCH, Weinheim,
- 25 2009; (c) W. Francke, S. Schultz, *Comprehensive Natural Products Chemistry*, Elsevier, Amsterdam, 1999.
- For reviews, see: (a) P. Chiu, L. T. Leung, C. B. B. Ko, Nat. Prod. Rep. 2010, 27, 1066; (b) G. J. Florence, N. M. Gardner, I. Paterson, Nat. Prod. Rep. 2008, 25, 342; (c) V. Boucard, G. Broustal, J. M.
- 30 Campagne, Eur. J. Org. Chem. 2007, 225; (d) M. Mondon, J. P. Gesson, Curr. Org. Synth. 2006, 3, 41; (e) I. Collins, J. Chem. Soc. Perkin Trans. 1 1999, 1377.
- For selected recent reports, see: (a) S. Bonazzi, S. Güttinger, I. Zemp, U. Kutay, K. Gardemann, Angew. Chem. Int. Ed. 2007, 46, 8707; (b)
- T. Maauer, H. J. Martin, J. Mulzer, Angew. Chem. Int. Ed. 2009, 48, 6032; (c) J. Esteban, A. M. Costa, A. G ómez, J. Vilarrasa, Org. Lett. 2008, 10, 65; (d) H. Fukui, I. Shiina, Org. Lett. 2008, 10, 3153.
- 4. N. Gunde-Cimerman, A. Cimerman, *Exp Mycol.* 1995, **19**, 1. 5 (a) S. Ekthawathchai, S. Kamchonwongpaisan, P. Kongsaera
- (a) S. Ekthawathchai, S. Kamchonwongpaisan, P. Kongsaeree, B. Tarnchompoo, Y. Thebtaranonth, Y. Yuthavong, *J. Med. Chem.* 2001, 44, 4688; (b) M. A. Avery, M. Alvim-Gaston, J. A. Vroman, B. Wu, A. Ager, W. Peters, B. L. Robinson, W. Charman, *J. Med. Chem.* 2002, 45, 4321.
- (a) A. J. Weinheimer, C. W. J. Chang, J. A. Matson, *Fortschr. Chem.* Org. Naturst. 1979, 36, 285; (b) J. E. McMurry, R. G. Dushin, J. Am. Chem. Soc. 1990, 112, 6942.
- I. Paterson, S. M. Dalby, J. C. Roberts, G. J. Naylor, E. A. Guzmán, R. Isbrucker, T. P. Pitts, P. Linley, D. Divlianska, J. K. Reed, A. E. Wright, *Angew. Chem. Int. Ed.* 2011, **50**, 3219.

- 8. (a) R. A. O'Reilly, N. Engl. J. Med. 1976, 295, 354; (b) L. B. Wingard, R. A. O'Reilly, G. Levy, Clic. Pharmacol. Ther. 1978, 23, 212; (c) H. P. Rang, M. M. Dale, J. M. Ritter, P. Gardner, Pharmacology, 4th ed. Churchill Livingston. Philadephia. 2001.
- (a) Y. J. Gao, Q. Ren, H. Wu, M. G. Li, J. Wang, Chem. Commun. 2010, 46, 9232; (b) Q. Ren, Y. J. Gao, J. Wang, Chem. –Eur. J. 2010, 16, 13594; (c) Y. J. Gao, Q. Ren, J. Wang, Chem. –Eur. J. 2010, 16, 13068; (d) Q. Ren, W.-Y. Siau, Z. Du, K. Zhang, J. Wang, Chem. – Eur. J. 2011, 17, 7781; (e) Y. J. Gao, Q. Ren, W.-Y. Siau, J. Wang, Chem. Commun. 2011, 47, 5819; (f) H. R. Tan, H. F. Ng, J. Chang, J.
- Wang, Chem. –Eur. J. 2012, 18, 3865; (g) W.-Y. Siau, W.-J. Li, F. Xue, Q. Ren, M.-H. Wu, S.-F. Sun, H.-B. Guo, X.-F. Jiang, J. Wang, Chem. –Eur. J. 2012, 18, 9491; (h) W.-Y. Siau, J. Wang, Catal. Sci. Technol. 2011, 1, 1298; (i) W.-J. Li, H. Liu, X. F. Jiang, J. Wang, ACS Catal. 2012, 2, 1535.
- ⁶⁵ 10. For examples of decarboxylative reactions, see: (a) Y. Kobuke and J.-I. Yoshida, *Tetrahedron Lett.* 1978, **19**, 367; (b) G. Lalic, A. D. Aloise, M. D. Shair, *J. Am. Chem. Soc.* 2003, **125**, 2852; (c) D. Magdziak, G. Lalic, H. M. Lee, K. C. Fortner, A. D. Aloise, M. D. Shair, *J. Am. Chem. Soc.* 2005, **127**, 7284; (d) K. C. Fortner, M. D.
- 70 Shair, J. Am. Chem. Soc. 2007, **129**, 1032; (e) S. Orlandi, M. Benaglia, F. Cozzi, *Tetrahedron Lett.* 2004, **45**, 1747.
- For selected examples of enantioselective decarboxylative reactions, see: (a) J. Lubkoll, H. Wennemers, *Angew. Chem. Int. Ed.* 2007, 46, 6841; (b) A. Ricci, D. Pettersen, L. Bernardi, F. Fini, M. Fochi, R. P.
- Herrera, V. Sgarzani, Adv. Synth. Catal. 2007, 349, 1037; (c) F.
 Zhong, W. Yao, X. Dou, Y. Lu, Org. Lett. 2012, 14, 4018; (d) Y.
 Zheng, H. Xiong, J. Nie, M. Hua, J. Ma, Chem. Commun. 2012, 48, 4308; (e) Y. Pan, C. W. Kee, Z. Jiang, T. Ma, Y. Zhao, Y. Yang, H.
 Xue, C.-H. Tan, Chem. –Eur. J. 2011, 17, 8363; (f) N. Hara, S.
- Nakamura, M. Sano, R. Tamura, Y. Funahashi, N. Shibata, *Chem. Eur. J.* 2012, **18**, 9276; (g) M. Amere, M.-C. Lasne and J. Rouden, *Org. Lett.* 2007, **9**, 2621; (*h*) N. Hara, S. Nakamura, Y. Funahashi, N. Shibata, *Adv. Synth. Catal.* 2011, **353**, 2976; (*i*) H. –A. Yuan, S. Wang, J. Nie, W. Meng, Q.-W. Yao, J.-A. Ma, *Angew. Chem. Int. Ed.* 2013, **52**, 3869; (*j*) H. Y. Bae, J. H. Sim, J.-W. Lee, B. List, C. E.
- 2013, 52, 3809; (j) H. Y. Bae, J. H. Sim, J.-W. Lee, B. List, C. E. Song, Angew. Chem. Int. Ed. 2013, 52, 12143.
 2013, 51, 12143.
- 12 There is no reported enantioselective decarboxylative reaction of α , β unsaturated carbonyls and half-thioesters catalyzed by secondary amine.
- 90 13. For selected examples of α,β- or γ,δ-unsaturated δ-lactone synthesis, see: (a) H. Du, D. Zhao, K. Ding, Chem. –Eur. J. 2004, 10, 5964; (b) L. Lin, Z. Chen, X. Yang, X. Liu, X. Feng, X. Org. Lett. 2008, 10, 1311; (c) Oisaki, K.; Zhao, D.; Kanai, M.; Shibasaki, M. J. Am. Chem. Soc. 2007, 129, 7439; (d) J. M. Mo, X. K. Chen, Y.-G. Chi, J. Am. Chem. Soc. 2012, 134, 8810; (e) P. S. Tiseni, R. Peters, Angew. Chem. Int. Ed. 2007, 46, 5325; (f) P. S. Tiseni, R. Peters, Org. Lett. 2008, 10, 2019; (g) D. Belmessieri, L. C. Morrill, C. Simal, A. M. Z. Slawin, A. D. Smith, J. Am. Chem. Soc. 2011, 133, 2714; (h) L. Lin, Y. –L. Kuang, X.-H. Liu, X.-M. Feng, Org. Lett. 2011, 15, 3868; (i)
 100 S. J. Ryan, L. Candish, D. W. Lupton, J. Am. Chem. Soc. 2009, 131,
 - 14176; (*j*) G.-Q, You, L.-X. Dai, S.-L. You, *Org. Lett.* 2009, **11**, 1623;
 14. For examples of enantioselective synthesis of saturated δ-lactones,
 - For examples of enantioselective synthesis of saturated o-factories, see: (a) S. Brandau, A. Landa, J. Franz én, M. Marigo, K. A. Jøgensen, Angew. Chem., Int. Ed. 2006, 45, 4305; (b) A. Albrecht, F. Morana,
- A. Fraile, K. A. Jøgensen, *Chem. -Eur. J.* 2012, **18**, 10348; (c) K. Murai, A. Nakamura, T. Matsushita, M. Shimura, H. Fujioka, *Chem. -Eur. J.* 2012, **18**, 8448; (d) M. C. Dobish, J. N. Johnston, *J. Am. Chem. Soc.* 2012, **134**, 6068; (e) J. Moran, A. G. Smith, R. M. Carris, J. S. Johnson, M. J. Krische, *J. Am. Chem. Soc.* 2011, **131**, 18618.
- 110 15. CCDC 856259 (3ac) contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/ data_request/cif.
- 16. M. S. Yu, I. Lantos, Z.-Q. Peng, J. Yu, T. Cacchio, *Tetrahedron Lett.* 2000, **41**, 5647.
 - 17. For proposed reaction mechanism, see Supporting Information.