

ChemComm

Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this *Accepted Manuscript* with the edited and formatted *Advance Article* as soon as it is available.

You can find more information about *Accepted Manuscripts* in the [Information for Authors](#).

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard [Terms & Conditions](#) and the [Ethical guidelines](#) still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.

COMMUNICATION

Molecular photo-oscillators based on highly accelerated heterocyclic azo dyes in nematic liquid crystals

Cite this: DOI: 10.1039/x0xx00000x

Jaume Garcia-Amorós,^a Marta Reig,^a M. Cidália R. Castro,^b Alba Cuadrado,^a M. Manuela M. Raposo^b and Dolores Velasco^{a,*}

Received 00th February 2014,
Accepted 00th February 2014

DOI: 10.1039/x0xx00000x

www.rsc.org/

Benzothiazole-pyrrole-based azo dyes enhance greatly their thermal isomerisation rate up to 160 times when they are under the influence of the nematic mean field yielding the LC-based photochromic oscillators with the highest oscillation frequencies reported so far (2.6 kHz at 298 K).

Liquid crystals (LCs) combine the molecular order of crystalline solids with the fluidity of conventional liquids. Particularly, nematic LCs are characterised by the tendency of their constituting molecules to point along a common direction.¹ Doping of nematic LCs with light-sensitive molecules, like azo compounds, allows tuning the properties of the material in a quick and remote fashion by optical control. It is well known that azo chromophores exhibit a reversible photoisomerisation process between their *trans* (rod-like) and *cis* (bent-like) isomers; in addition, the metastable *cis* isomer also reverts back thermally in the dark.² When azo-LC mixtures are illuminated, the bent *cis*-form of the azo chromophore is generated in the probe inducing changes in the organization at the molecular level. This effect is reflected in variations of the properties of the material at the macroscopic scale, for instance, its ability to absorb light.³⁻⁴ This feature has given rise to many kinds of LC-based molecular photoswitchable materials.⁵⁻⁸

Molecular photochromic oscillators modify their optical properties periodically and extremely fast upon exposure to light. Oscillating materials have attracted a great deal of attention over the last years due to their huge applicability; e.g. artificial muscles for robotics, real-time optical information processors, rotary motors, switchable optical reflectors, etc...⁹⁻¹⁶

For an efficient photo-oscillation to be met, the azo dye used should undergo its thermal back conversion in a very quick fashion, most desirable within the microsecond timescale. Indeed, the overall performance of these materials rely on the control of the speed of such process which, in turn, is mainly determined by the chemical architecture of the azo derivative. However, in LC-based materials, the azo dye must be introduced in an ordered host matrix to operate. In this medium, both chemical nature and dynamics of the host LC might influence the rate of the isomerisation process and, therefore, the oscillation frequency of the final material. Accordingly, achieving knowledge about how the host nematic matrix affects the rate of the thermal back conversion for very fast-isomerising azo dyes is of great importance for an optimal design of LC-based molecular photo-oscillators.

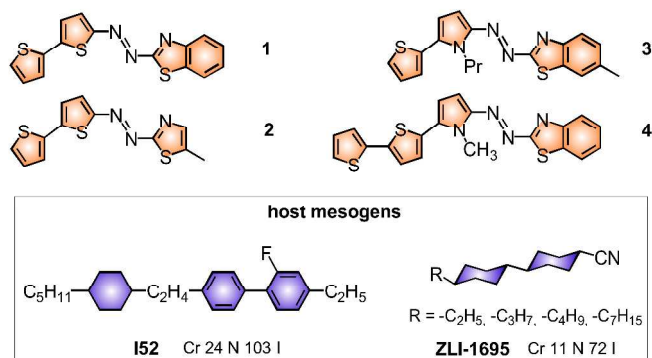
Positively-charged azo dyes have been recently described as the fastest molecular photochromic oscillators in neutral media since they exhibit thermal relaxation times as low as 33 μ s at 298 K yielding, therefore, oscillation frequencies up to 10 kHz.¹⁷ However, the low solubility of such azo chromophores in the most common host LC matrixes limits their use as light-sensitive guests for LC-based materials. Thus, the formulation of new azo-LC mixtures, where the azo moiety shows not only a high solubility but also a very fast thermal isomerisation, remains still as a challenge for achieving efficient LC-based photochromic oscillators.

Here we report on the role the nematic mean field plays on the thermal *cis*-to-*trans* isomerisation process for azo dyes that show relaxation times below the second timescale. We have found that benzothiazole-pyrrole-based azo compounds, which due to their neutral nature and rod-like geometry can be easily introduced as guests in a series of host LC matrixes, enhance drastically their thermal isomerisation rate up to 160 times in these media. In fact, such azo derivatives exhibit thermal relaxation times at room temperature as low as 130 μ s. This feature makes of them the fastest non-positively-charged azo dyes in nematic LCs reported so far. In turn, these novel azo-

^a Grup de Materials Orgànics, Institut de Nanociència i Nanotecnologia (IN²UB), Departament de Química Orgànica, Universitat de Barcelona, Martí i Franquès 1, E-08028, Barcelona, Spain. E-mail: dvelasco@ub.edu; Fax: + 34 93 339 78 78; Tel: + 34 93 403 92 60.

^b Centro de Química, Universidade do Minho, Campus de Gualtar, 4710-057, Braga, Portugal.

† Electronic Supplementary Information (ESI) available: experimental part for the kinetic analyses and physical data for the LCs used. See DOI: 10.1039/c000000x/



Scheme 1 Chemical structure of azo dyes **1-4** and nematic mesogens **I52** and **ZLI-1695**. The phase transformation temperatures for the different LCs are given in Celsius.

LC mixtures enable high oscillation frequencies in the optical density of the material up to 2.6 kHz under ambient conditions.

The kinetics of the thermal *cis*-to-*trans* isomerisation for azo dyes **1-4** (Scheme 1) in both isotropic and nematic solutions were analysed by means of the nanosecond laser flash-photolysis technique (ESI[†]). The thermal relaxation time of the *cis* isomers, τ ($\tau = 1/k$), was determined by fitting a monoexponential function to the data (Table 1). The maximum oscillation frequency of the corresponding photochromic oscillator was determined as $\nu_{\max} = 1/(3\tau)$.

Whereas the relaxation times for the thermal *cis*-to-*trans* isomerisation process are very similar for compounds **1** and **2** in all the isotropic and nematic media analysed (Table 1), for pyrrole-containing azo compounds (**3** and **4**), clearly faster thermal back reactions are registered in both ethanol and acetonitrile than in toluene. Indeed, and despite the high polarity of acetonitrile, which is given by its dielectric permittivity ($\epsilon = 37.5$ and 24.5 for acetonitrile and ethanol, respectively), the possibility of hydrogen bonding to be established in alcoholic solutions can induce an enhancement of the thermal isomerisation rate for such azo dyes. It is highly remarkable that the relaxation times found for compounds **3** and **4** in the nematic phase of mesogen **I52** (Scheme 1) were of 1.4 ms and 130 μ s, respectively (see Fig. 1, 2a and Table 1), which are very close to the ones registered in ethanol solution, even though the polarities of both media are far away from each other ($\epsilon = 2.91$ for **I52**).

A moderate increase of about 3–4 times in the thermal *cis*-to-*trans* isomerisation rate has been described previously for other azo compounds when dissolved in a LC matrix.¹⁸⁻²⁰ However, in this instance, the thermal isomerisation rate for azo compounds **3**

Table 1. Relaxation time for the *cis* isomer and maximum oscillation frequencies for azo dyes **1-4** in different media at 298 K. τ and ν_{\max} are given in milliseconds and Hertz, respectively. ^aNot determined.

	toluene		ethanol		acetonitrile		I52		ZLI-1695	
	τ	ν_{\max}	τ	ν_{\max}	τ	ν_{\max}	τ	ν_{\max}	τ	ν_{\max}
1	64	5.2	108	3.1	65	5.1	170	2.0	- ^a	- ^a
2	41	8.1	35	9.5	30	11	49	6.8	- ^a	- ^a
3	20	17	1.0	330	5.6	60	1.4	240	17	20
4	21	16	0.07	4830	0.44	760	0.13	2600	1.2	280

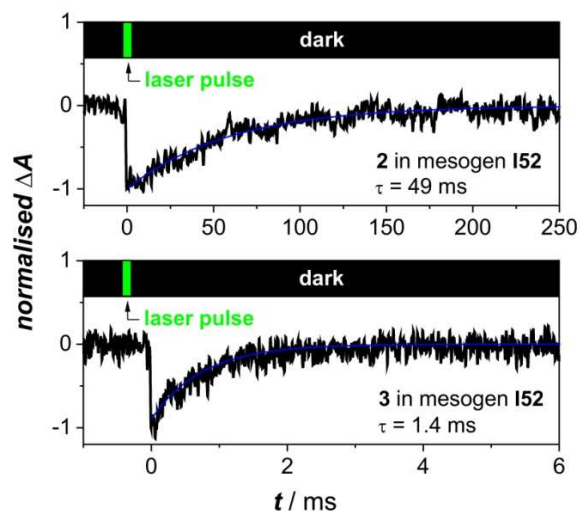


Fig. 1 Transient absorption change photoinduced by laser pulse irradiation ($\lambda_{\text{irrad}} = 532$ nm, Continuum Surelite I-10 Q-switched Nd-YAG laser with 5 ns pulse width and ca. 10 mJ per pulse) for azo dyes **2** and **3** in the mesogen **I52** at 298 K ($\lambda_{\text{obs}} = 500$ nm, [AZO] = 4 mM).

and **4** in the nematic mesogen **I52** was dramatically enhanced up to ca. 14 and 160 times with respect to the corresponding ones in toluene, respectively, although these two media exhibit identical polarities ($\epsilon = 2.38$ for toluene). Hence, the great enhancement of the thermal isomerisation rate observed for those azo compounds when dissolved in a nematic media should be clearly attributed mainly to the cooperative interaction between both azo dye and mesogen molecules that tend to recover their initial orientation along the nematic director once the illumination of the probe is stopped.

Previously reported systems thermally-isomerised *via* the inversion mechanism and, therefore, slow thermal back reactions were registered for them.¹⁸⁻²⁰ However, for the azo dyes described herein the rotational pathway operates instead and, thus, the relaxation time of the azo dye falls within the very same timescale than that for the reorganization of the host LC towards the director direction. Indeed, it can be nicely seen from data displayed in Table 1 that the enhancement induced in the thermal back reaction by the host LC matrix decreases as the relaxation time of the azo dye increases, most probably because the relaxation time of the azo moiety becomes more different from the one for the LC reorientation.

In order to test the role the dynamics of the host LC matrix plays in the relaxation kinetics of the azo dye, the thermal back conversion for azo derivatives **3** and **4** was studied in the nematic phase of mesogen **ZLI-1695** (Scheme 1). In this case, although the registered relaxation times fell within the same timescale than those found in mesogen **I52** (17 and 1.2 ms for **3** and **4** in **ZLI-1695**, respectively, see Table 1), the enhancement produced by the nematic mean field was ca. 10-fold lower than the one detected in **I52**, pointing out that the dynamics of the host nematic mesophase can influence the thermal relaxation kinetics of the azo dye and, therefore, on the oscillation frequency of the final material. The dynamics of the host LC matrix is

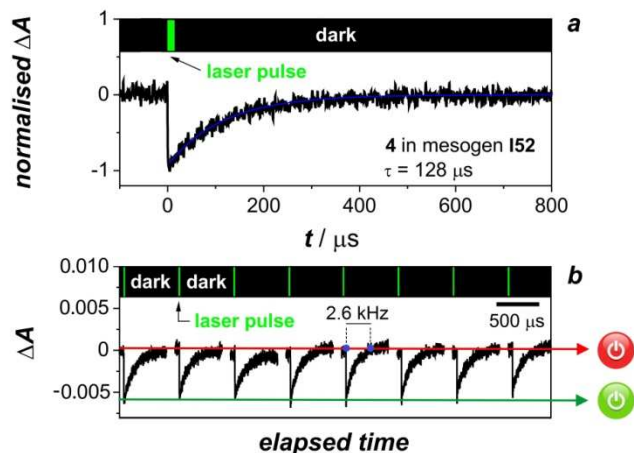


Fig. 2 (a) Transient absorption change photo-induced by laser pulse irradiation with green light ($\lambda_{\text{irrad}} = 532 \text{ nm}$) at 298 K ($\lambda_{\text{obs}} = 500 \text{ nm}$) and (b) oscillation in the optical density of azo dye **4** in mesogen **I52**.

characterized by the viscoelastic coefficient (K_{11}/γ_1),²¹⁻²² where K_{11} and γ_1 stand for the deformation elastic constant and the rotational viscosity, respectively. Mesogen **I52** shows a K_{11}/γ_1 ratio of $5.21 \times 10^{-10} \text{ m}^2 \cdot \text{s}^{-1}$ at 298 K, which is higher than the one exhibited by mesogen **ZLI-1695** ($1.22 \times 10^{-10} \text{ m}^2 \cdot \text{s}^{-1}$, see ESI[†] for full numerical data). The shorter K_{11}/γ_1 value for mesogen **ZLI-1695** is related with the longer time this mesogen needs to recover the initial orientation along the director after the illumination of the sample in comparison with mesogen **I52**. This feature stands for the higher enhancement that mesogen **I52** induces on the azo dye thermal back conversion (see Table 1).

The very fast thermal isomerisation rate registered for the benzothiazole-pyrrole based azo dye **4** in the nematic mesogen **I52** makes of this nematic mixture the best candidate to be applied as a light-controlled photochromic oscillator. Fig. 2b shows the oscillation of the optical density of the **4-I52** mixture with time upon continuous pulsed green light (532 nm)–dark cycles. The oscillation frequency of azo dye **4** in this nematogen is as high as 2.6 kHz at room temperature (see Table 1) and, thus, a full cycle can be completed for that system in less than the half of a millisecond. Moreover, the oscillating behaviour of such system remained unchanged after 12,000 continuous working cycles. It should be highlighted that the oscillation frequency achieved with this system falls in the very same range than that reached in alcoholic solutions (up to 4.8 kHz) as well as than those reported previously for positively-charged azobenzenes in isotropic solutions (10 kHz)¹⁷ evidencing thereby the great potential applicability of the neutral azo-LC mixtures reported herein in photonic devices.

In summary, benzothiazole-pyrrole-based azo dyes enhance their thermal isomerisation rate up to 160 times when they are doped in low molar mass liquid crystals due to the influence of the nematic mean field. This enhancement has enabled the development of stable LC-based photochromic molecular oscillators with oscillation frequencies as high as 2.6 kHz under ambient conditions for the first time.

Financial support for this research was obtained from the *Ministerio de Ciencia e Innovación* (Spain, CTQ2012-36074). J.G.-A. is grateful for a post-doctoral grant from the *Generalitat de Catalunya* (Spain, 2011 BP-A-00270). Thanks are also due to Fundação para a Ciência e Tecnologia (Portugal) for financial support to the portuguese NMR network (PTNMR, Bruker Avance III 400-Univ. Minho), FCT and FEDER (European Fund for Regional Development)-COMPETE-QREN-EU for financial support to the research centre CQ/UM PEst-C/QUI/UI0686/2013 (FCOMP-01-0124-FEDER-037302) and a PhD grant to M. C. R. C. (SFRH/BD/78037/2011).

Notes and references

- 1 G. W. Gray, *Handbook of Liquid Crystals*, Wiley-VCH, Weinheim, 1998.
- 2 H. Rau, *Photochemistry and Photophysics*; CRC Press: Boca Raton, FL, 1990.
- 3 T. Ikeda, *J. Mater. Chem.*, 2003, **13**, 2037.
- 4 Y. Yu and T. Ikeda, *J. Photochem. Photobiol. C*, 2004, **5**, 247.
- 5 X. Tong, M. Pelletier, A. Lasia and Y. Zhao, *Angew. Chem. Int. Ed.*, 2008, **47**, 3596.
- 6 Z. Alam, T. Yoshioka, T. Ogata, T. Nonaka and S. Kurihara, *Chem. Eur. J.*, 2007, **13**, 2641.
- 7 J. Ma, Y. Li, T. White, A. Urbas and Q. Li, *Chem. Commun.*, 2010, **46**, 3463.
- 8 U. A. Hrozhyk, S. V. Serak, N. V. Tabiryan, L. Hoke, D. M. Steeves and B. R. Kimball, *Optics Express*, 2010, **18**, 8697.
- 9 S. Kobatake, S. Takami, H. Muto, T. Ishikawa and M. Irie, *Nature*, 2007, **446**, 778.
- 10 T. J. White, N. V. Tabiryan, S. V. Serak, U. A. Hrozhyk, V. P. Tondiglia, H. Koerner, R. A. Vaia and T. J. Bunning, *Soft Matter*, 2008, **4**, 1796.
- 11 M. Klok, N. Boyle, M. T. Pryce, A. Meetsma, W. R. Browne and B. L. Feringa, *J. Am. Chem. Soc.*, 2008, **130**, 10484.
- 12 C. L. Van Oosten, C. W. M. Bastiaansen and D. J. Broer, *Nat. Mater.*, 2009, **8**, 677.
- 13 H. Koshima, N. Ojima and H. Uchimoto, *J. Am. Chem. Soc.*, 2009, **131**, 6890.
- 14 J. Horvath, I. Szalai, J. Boissonade and P. De Kepper, *Soft Matter*, 2011, **7**, 8462.
- 15 K. M. Lee and T. J. White, *Polymers*, 2011, **3**, 1447.
- 16 U. A. Hrozhyk, S. V. Serak, N. V. Tabiryan, T. J. White and T. J. Bunning, *Opt. Mater. Express*, 2011, **1**, 943.
- 17 J. Garcia-Amorós, S. Nonell and D. Velasco, *Chem. Commun.*, 2011, **47**, 4022.
- 18 J. Garcia-Amorós, M. Martínez, H. Finkelmann and D. Velasco, *J. Phys. Chem. B*, 2010, **114**, 1287.
- 19 J. Garcia-Amorós, H. Finkelmann and D. Velasco, *Phys. Chem. Chem. Phys.*, 2011, **13**, 11233.
- 20 J. Garcia-Amorós, H. Finkelmann and D. Velasco, *Chem. Eur. J.*, 2011, **17**, 6518.
- 21 S.-T. Wu, M. E. Neubert, S. S. Keast, D. G. Abdallah, S. N. Lee, M. E. Walsh and T. A. Dorschner, *Appl. Phys. Lett.*, 2000, **77**, 957.
- 22 S. Gauza, H. Wang, C.-H. Wen, S.-T. Wu, A. J. Seed and R. Dabrowski, *Jpn. J. Appl. Phys.*, 2003, **42**, 3463.