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COMMUNICATION

The enantioselective Diels–Alder reaction of 1,2-dihydropyridines with α -acyloxyacroleins catalyzed by chiral primary ammonium salt†

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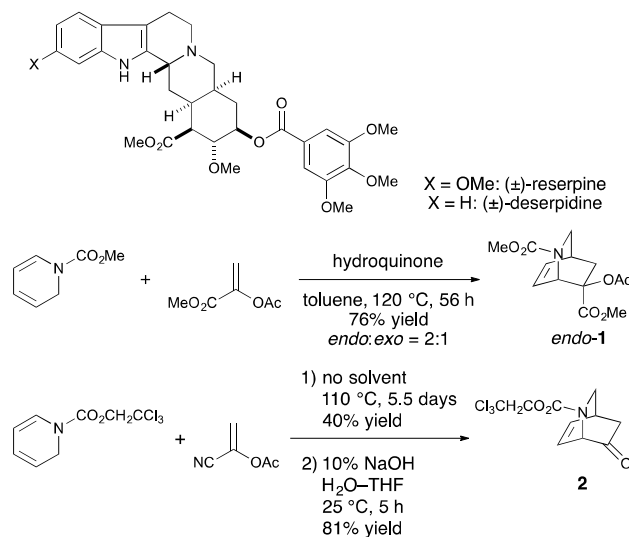
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The first enantioselective Diels–Alder reaction of 1,2-dihydropyridines with α -acyloxyacroleins catalyzed by chiral primary ammonium salt has been developed and offers more efficient routes to key synthetic intermediates of alkaloids, for which the direct preparations were unavailable before. The asymmetric induction can be understood through the optimized geometry of iminium salt aqua complex derived from the catalyst and the dienophile.

The isoquinuclidine skeleton is found in a wide range of alkaloids, and optically active functionalized isoquinuclidines, as chiral synthons, can be transformed to versatile alkaloids.¹ The enantioselective Diels–Alder reaction of 1,2-dihydropyridines with acryloyl derivatives is one of the most straightforward methods for the construction of optically active isoquinuclidines.

For example, in 1980, Wender *et al.* achieved the total synthesis of (\pm)-reserpine from (\pm)-endo-Diels–Alder adduct **1** of *N*-methoxycarbonyl-1,2-dihydropyridine with methyl α -acetoxyacrylate (Scheme 1).² In 1990, Mariano *et al.* achieved



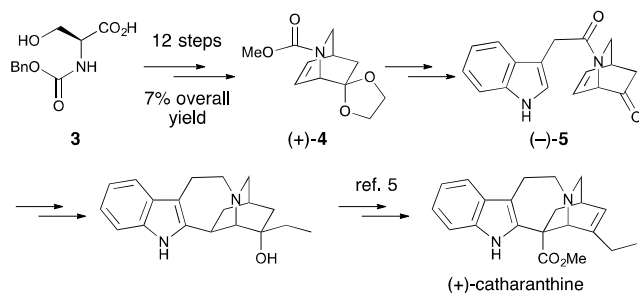
Scheme 1 Total synthesis of (\pm)-reserpine² and (\pm)-deserpine.³

the formal total synthesis of (\pm)-deserpine from (\pm)-Diels–Alder adduct **2** of *N*-(2,2,2-trichloroethoxycarbonyl)-1,2-dihydropyridine with α -acetoxyacrylonitrile. However, these Diels–Alder reactions were very slow and the *endo*-selectivity was low (Scheme 1).³

In 2006, Doris *et al.* reported the formal synthesis of (+)-catharanthine using optically active isoquinuclidine **4** as a key intermediate, which was derived from *N*-benzyloxycarbonyl-L-serine (**3**) in 7% overall yield over 12 steps (Scheme 2).^{4,5}

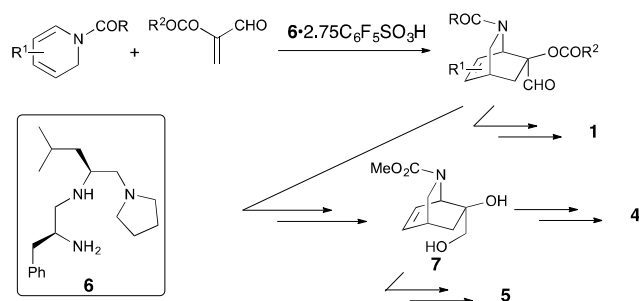
In 2002, Rawal *et al.* reported the first catalytic enantioselective Diels–Alder reaction of 1,2-dihydropyridine with *N*-acryloyloxazolidinone.^{6a} Since then, Nakano *et al.* have intensively studied on the development of an asymmetric catalysis.^{6b–k} In 2005, Fukuyama *et al.* reported that MacMillan's catalyst⁷ was also effective for the enantioselective Diels–Alder reaction of *N*-benzyloxycarbonyl-1,2-dihydropyridine with acrolein to give

optically active isoquinuclidine, and achieved the total synthesis of (–)-oseltamivir phosphate from the isoquinoline.⁸



Scheme 2 Doris's Formal Total Synthesis of (+)-Catharanthine.⁴

Against this background, we expected that we could prepare optically active synthons **1**, **2** and **4** through the enantioselective Diels–Alder reaction of 1,2-dihydropyridines with α -acyloxyacroleins as a key step. However, there have been no reports on the highly enantioselective Diels–Alder reaction of dihydropyridines with α -substituted acryloyl derivatives.⁹ Along these lines, we have developed the enantioselective Diels–Alder reaction of simple dienes with α -heterosubstituted acroleins catalyzed by chiral primary ammonium salts.¹⁰ In this report we describe the enantioselective Diels–Alder reaction of 1,2-dihydropyridines with α -acyloxyacroleins catalyzed by **6**·2.75C₆F₅SO₃H, and derivatization of Diels–Alder adducts to *endo*-**1**, **7**, **4**, and **5** (Scheme 3).



Scheme 3 This work.

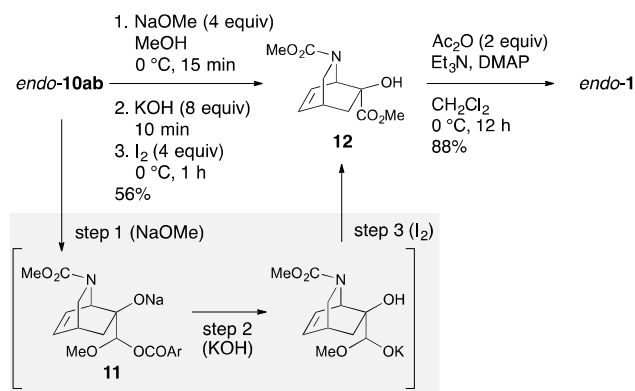
First, the Diels–Alder reaction of *N*-phenoxy carbonyl-1,2-dihydropyridine (**8a**, 0.11 mmol) with α -(benzyloxy)acrolein (**9a**, 0.1 mmol) was examined in the presence of 10 mol% of **6**·2.75C₆F₅SO₃H in nitroethane. The reaction temperature was gradually increased from -78 °C to 0 °C, and the reaction mixture was further stirred at 0 °C for 3 days. The desired isoquinuclidine **10aa** was obtained in 58% yield with >95% *endo*-selectivity. The enantioselectivity of *endo*-**10aa** was 88% ee (entry 1). Since a hydrogen-bonding interaction between an α -acyl moiety of **9** and an ammonium proton of the catalyst was expected, α -(*p*-methoxybenzyloxy)acrolein (**9b**) was examined as a dienophile under the same conditions. Interestingly, **9b** was much more reactive than **9a**. *Endo*-**10ab** was obtained in 88% yield with 91% ee (entry 2). When the reaction was started at 0 °C, the enantioselectivity was reduced to 69% ee (entry 3). Actually, the reaction proceeded slowly even at -20 °C, and the enantioselectivity was 95% ee (entry 4). The reaction of **8a** with **9b** was scalable, and could be increased up to 2.4 mmol scale without reducing chemical yield and enantioselectivity of **10ab** (entry 5).

Table 1 Enantioselective Diels–Alder Reaction of **8a** with **9**^a

entry	9 (Ar)	conditions	10	
			yield (%)	ee (%) ^b
1 ^c	9a	-78 °C, 5 min to	10aa , 58	88
	(Ph)	0 °C, 3 days		
2 ^c	9b	-78 °C, 5 min to	10ab , 88	91
	(C ₆ H ₄ (OMe)- <i>p</i>)	0 °C, 1.5 days		
3 ^c	9b	0 °C, 12 h	10ab , 88	69
4 ^d	9b	-78 °C, 5 min to	10ab , 61	95
		-20 °C, 4 days		
5 ^e	9b	-20 °C, 5 min to	10ab , 81	92
		0 °C, 2.5 days		

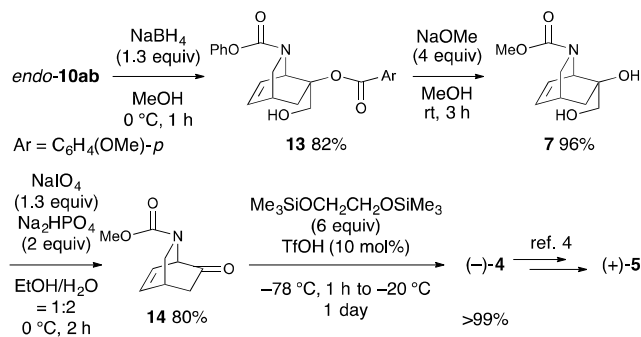
^a See the experimental section in detail. ^b Ee of *endo*-**10**. ^c 0.1 mmol scale of **9** (0.8 M) in EtNO₂. ^d 0.1 mmol scale of **9b** (0.4 M) in EtNO₂. 1.5 equiv of **8a** was used. ^e 2.4 mmol scale of **9b** (0.8 M) in EtNO₂.

Endo-**10ab** was transformed to **12** in 56% yield under basic oxidation conditions (Scheme 4). This is a one-pot 3-step sequence of the three different functional groups.¹¹ Deprotection of the *p*-methoxybenzoyl group of **10ab** was promoted by the adjacent oxyanion (Step 1).¹² The generation of **11** was detected by ¹H NMR analysis. The subsequent acetylation of **12** gave *endo*-**1** in 88% yield.



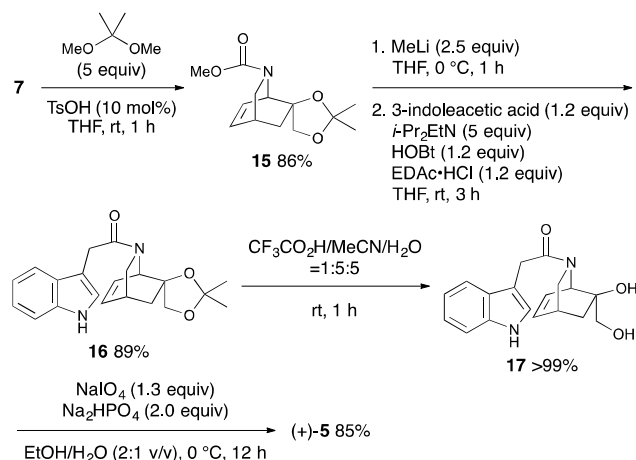
Scheme 4 Derivatization of *endo*-**10ab** to *endo*-**1**.

Endo-adduct **10ab** (91% ee) was converted to a key synthetic intermediate **4** of (+)-catharanthine in good yield by selective reduction of the formyl group of **10ab**, selective methanolysis of the ester moiety of **13**, oxidative cleavage of the vicinal diol moiety of **7**, and acetalization of **14** (Scheme 5).¹³ The absolute configuration of **4** was determined to be (–)-(1*R*,4*R*) by comparison of its optical rotation (91% ee, $[\alpha]_D^{26} = -65$ ($c = 0.4$, MeOH)).⁴ Spiroacetal **4** can be transformed to 2-(2-(1*H*-indol-3-yl)acetyl)-2-azabicyclo[2.2.2]oct-7-en-6-one (**5**) by Doris's synthetic method (Scheme 2).⁴ Thus, (+)-(1*R*,4*R*)-**5** can be synthesized from **7** in 49% yield over 5 steps.



Scheme 5 Derivatization of *Endo-10ab* to (-)-4.

As a new and more concise synthetic route to (+)-(1*R*,4*R*)-5, we developed a 4-step sequence from diol 7 which included the acetalization of 7, the *N*-deprotection/acylation of 15, the hydrolysis of 16, and the oxidative cleavage of 17 (Scheme 6). The overall yield of (+)-5 (91% ee, $[\alpha]_D^{24} = +88$ ($c = 1.2$, CHCl_3) from 7 was 65%.⁴



Scheme 6 Derivatization of 7 to (+)-5.

The substrate scope of the Diels–Alder reaction of 4-substituted *N*-phenoxyacetyl-1,2-dihydropyridines (**8b–e**) with **9b** was investigated under the optimized conditions (entry 2 in Table 1). The results are shown in Table 2. The reactions proceeded smoothly at $-20 \sim 0$ °C to give desired *endo*-adducts **10** in good to high yield with high enantioselectivity.

Table 2 Enantioselective Diels–Alder reaction of **8** with **9b**^a

entry	8 (R)	10	
		yield (%)	ee (%)
1 ^b	8b (Me)	10bb , 95	88
2	8c (<i>i</i> -Pr)	10cb , 86	80
3	8d (OCOC ₆ H ₄ (OMe)- <i>p</i>)	10db , 84	90
4 ^c	8d	10db , 81	95
5 ^c	8e (CH ₂ OSi- <i>t</i> -BuMe ₂)	10eb , 82	93

^a See the experimental section in detail. ^b 2.2 equiv of **8** was used. ^c The reaction was carried out at -20 °C for 3 days.

To understand the reaction mechanism, the optimized geometry of iminium salt **18** prepared from **6**·3C₆F₅SO₃H and **9a** was calculated using B3LYP/6-31G(d).^{14,15} However, no reasonable geometries that could explain the enantioface selectivity were obtained. When a key intermediate was postulated to be an aqua complex **18**·H₂O, the desired optimized geometry was obtained as shown in Fig. 1. One water molecule stabilizes the (*Z*)-iminium geometry through three hydrogen bonds. In this geometry, the *re*-face of the dienophile is effectively shielded by the benzyl moiety of **6**·3C₆F₅SO₃H. Thus, 1,2-dihydropyridine can approach the *si*-face of the dienophile to give (1*R*,4*R*)-adduct enantioselectively. While this mechanism is plausible, it is not supported by any experimental evidence. Nevertheless, some water molecules may contribute to the construction of key intermediates or transition-state assemblies. Further mechanistic studies are in progress and will be reported in the near future.

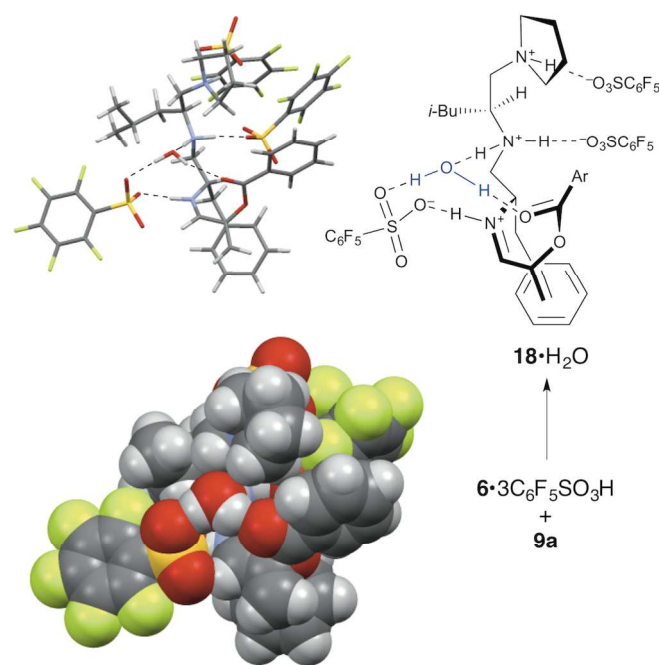


Fig. 1 Optimized geometry (B3LYP/6-31G(d)) of iminium salt **18**·H₂O prepared from **6**·3C₆F₅SO₃H and **9a** *in situ*.

In summary, we have developed the first example of a catalytic and highly enantioselective Diels–Alder reaction of 1,2-dihydropyridines with α -acyloxyacroleins catalyzed by **6**·2.75C₆F₅SO₃H, and their derivatization to some key synthetic intermediates of biologically active compounds. Furthermore, a plausible mechanism via an aqua complex such as **18**·H₂O can explain all previous results with the use of **6**·C₆F₅SO₃H.¹⁰

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Notes and references

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† Electronic Supplementary Information (ESI) available: Experimental procedures, characterization data, and copies of ¹H and ¹³C NMR spectra. See DOI: 10.1039/c000000x/

- (a) M. O. Farunk Khan, M. S. Levi, C. R. Clark, S. Y. Ablordeppey, S.-J. Law, N. H. Wilson, R. F. Borne, *Stud. Nat. Prod. Chem.*, 2008, **34**, 753; (b) E. M. P. Silva, P. A. M. M. Varandas, A. M. S. Silva, *Synlett*, 2013, 3053.
- (a) P. A. Wender, J. M. Schaus, A. W. White, *J. Am. Chem. Soc.*, 1980, **102**, 6157; (b) P. A. Wender, J. M. Schaus, A. W. White, *Heterocycles*, 1987, **25**, 263.
- E. W. Baxter, D. Labaree, H. L. Ammon, P. S. Mariano, *J. Am. Chem. Soc.*, 1990, **112**, 7682.
- L. Moisan, P. Thuéry, M. Nicolas, E. Doris, B. Rousseau, *Angew. Chem. Int. Ed.*, 2006, **45**, 5334. According to this report, the specific rotations ($[\alpha]_D^{20}$) of (+)-(1*S*,4*S*)-**4** (>99% ee) and (-)-(1*S*,4*S*)-**5** (>99% ee) are +70 (*c* = 0.4, MeOH) and -61 (*c* = 1.2, CHCl₃), respectively.
- G. Büchi, P. Kulsa, K. Ogasawara, R. L. Rosati, *J. Am. Chem. Soc.* 1970, **92**, 999.
- (a) N. Takenaka, Y. Huang, V. H. Rawal, *Tetrahedron*, 2002, **58**, 8299; (b) H. Nakano, N. Tsugawa, R. Fujita, *Tetrahedron Lett.*, 2005, **46**, 5677; (c) H. Nakano, N. Tsugawa, K. Takahashi, Y. Okuyama, R. Fujita, *Tetrahedron*, 2006, **62**, 10879; (d) Y. Nishiuchi, H. Nakano, Y. Araki, R. Sato, R. Fujita, K. Uwai, M. Takeshita, *Heterocycles*, 2009, **77**, 1323; (e) H. Nakano, K. Ozone, M. Takeshita, E. Kwon, C. Seki, H. Matsuyama, N. Takano, Y. Kohari, *Chem. Commun.*, 2010, **46**, 4827; (f) M. Hirma, Y. Kato, C. Seki, H. Nakano, M. Takeshita, N. Oshikiri, M. Iyoda, H. Matsuyama, *Tetrahedron*, 2010, **66**, 7618; (g) C. Stuttibut, Y. Kohari, K. Igarashi, H. Nakano, M. Hirma, C. Seki, H. Matsuyama, K. Uwai, N. Takano, Y. Okuyama, K. Ozone, M. Takeshita, E. Kwon, *Tetrahedron Lett.*, 2011, **52**, 4745; (h) C. Seki, M. Hirma, H. D. M. R. Hutabarat, J. Takada, C. Stuttibut, H. Takahashi, T. Takaguchi, Y. Kohari, H. Nakano, K. Uwai, N. Takano, M. Yasui, Y. Okuyama, M. Takeshita, H. Matsuyama, *Tetrahedron*, 2012, **68**, 1774; (i) Y. Okuyama, K. Ozone, H. Nakano, M. Takeshita, *Heterocycles*, 2012, **84**, 1209; (j) N. D. M. R. Hutabarat, C. Seki, T. Shimizu, M. Hirma, Y. Kohari, H. Nakano, K. Uwai, N. Takano, E. Kwon, H. Matsuyama, *Heterocycles*, 2012, **86**, 203; (k) Y. Sakuta, Y. Kohari, N. D. M. R. Hutabarat, K. Uwai, E. Kwon, Y. Okuyama, C. Seki, H. Matsuyama, N. Takano, M. Tokiwa, M. Takeshita, H. Nakano, *Heterocycles*, 2012, **86**, 1379.
- (a) K. A. Ahrendt, C. J. Borths, D. W. C. MacMillan, *J. Am. Chem. Soc.*, 2000, **122**, 4243; (b) G. Lelais, D. W. C. MacMillan, *Aldrichimica Acta*, 2006, **39**, 79.
- (a) N. Satoh, T. Akiba, S. Yokoshima, T. Fukuyama, *Angew. Chem. Int. Ed.*, 2007, **46**, 5734; (b) N. Satoh, T. Akiba, S. Yokoshima, T. Fukuyama, *Tetrahedron*, 2009, **65**, 3239.
- According to ref. 6a, Rawal *et al.* obtained *endo*-adduct in 89% yield with 67% ee through the enantioselective Diels–Alder reaction of *N*-phenyloxycarbonyldihydropyridine with methacrolein catalyzed by chiral Cr(III) Lewis acid.

- (a) K. Ishihara, K. Nakano, *J. Am. Chem. Soc.*, 2005, **127**, 10504 (13079 correction); (b) A. Sakakura, K. Suzuki, K. Nakano, K. Ishihara, *Org. Lett.* 2006, **8**, 2229; (c) A. Sakakura, K. Suzuki, K.; Ishihara, *Adv. Synth. Catal.*, 2006, **348**, 2457; (d) K. Ishihara, K. Nakano, *J. Am. Chem. Soc.* 2007, **129**, 8930; (e) K. Ishihara, K. Nakano, M. Akakura, *Org. Lett.*, 2008, **10**, 2893; (f) A. Sakakura, K. Ishihara, *Bull. Chem. Soc. Jpn.*, 2010, **83**, 313; (g) A. Sakakura, H. Yamada, K. Ishihara, *Org. Lett.*, 2012, **14**, 2972; (h) A. Sakakura, H. Yamada, K. Ishihara, *Asian J. Org. Chem.*, 2012, **1**, 133.
- For the oxidation of aldehydes with I₂/KOH, see: (a) S. Yamada, D. Morizono, K. Yamamoto, *Tetrahedron Lett.*, 1992, **33**, 4329; (b) B. D. Jones, J. L. Clair, C. E. Moore, A. L. Rheingold, M. D. Burkart, *Org. Lett.*, 2010, **12**, 4516.
- After the formyl group of **10ab** was oxidized to a methoxycarbonyl group, we failed to achieve selective deprotection of the *p*-methoxybenzoyl group to produce **12** under basic conditions.
- (a) T. Tsunoda, M. Suzuki, R. Noyori, *Tetrahedron Lett.*, 1980, **21**, 1357; (b) R. Noyori, S. Murata, M. Suzuki, *Tetrahedron*, 1981, **37**, 3899.
- M. J. Frisch *et al.*, Gaussian 09, Revision C01, Gaussian, Inc., Wallingford, CT, 2009.
- (a) A. D. Becke, *J. Chem. Phys.*, 1993, **98**, 5648; (b) Stevens, P. J.; Devlin, J. F.; Chablowski, C. F.; Frisch, M. J. *J. Phys. Chem.*, 1994, **98**, 11623; (c) J. Tirado-Rives, W. L. Jorgensen, *J. Chem. Theory Comput.*, 2008, **4**, 297.

Graphical abstract

The enantioselective Diels–Alder reaction of 1,2-dihydropyridines with α -acyloxyacroleins catalyzed by chiral primary ammonium salt has been developed.

