

Synthesis of cholic acid-core (ε-caprolactone-ran-lactide)-bpoly(ethylene glycol) 1000 random copolymer as chemotherapeutic nanocarriers for liver cancer treatment

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4	
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1 ABSTRACT

A star-shaped random copolymer, cholic acid functionalized (*ɛ*-caprolactone 2 *-ran*-lactide)-*b*-poly(ethylene glycol) 1000 (CA-(PCL-*ran*-PLA)-*b*-PEG_{1k}), was 3 4 synthesized by a core-first approach through three parts of chemical reaction, and characterized by hydrogen-1 nuclear magnetic resonance (¹H NMR), gel permeation 5 6 chromatography (GPC) and thermogravimetric analysis (TGA). The docetaxel-loaded 7 nanoparticles (NPs) were prepared by a modified nano-precipitation method. The 8 formation and characterization of these NPs were confirmed through dynamic light scattering (DLS), zeta potential measurements, field emission scanning electron 9 10 microscopy (FESEM), and transmission electron microscopy (TEM). The in vitro 11 release profiles indicated that CA-(PCL-ran-PLA)-b-PEG_{1k} NPs had an excellent sustained and controlled drug release property. Both confocal laser scanning 12 13 microscope (CLSM) and flow cytometric (FCM) results showed the coumarin-6 loaded CA-(PCL-ran-PLA)-b-PEG_{1k} NPs had the highest cellular uptake efficiency 14 15 compared with PEG_{1k}-b-(PCL-ran-PLA) NPs and CA-(PCL-ran-PLA) NPs in human hepatic carcinoma cells. The docetaxel-loaded CA-(PCL-ran-PLA)-b-PEG_{1k} NPs 16 17 were also proved to have the highest drug loading content, encapsulation efficiency, and the best anti-tumor efficacy both in vitro and in vivo. In conclusion, the 18 star-shaped CA-(PCL-ran-PLA)-b-PEG_{1k} copolymer was successfully synthesized 19 20 and could be used as a promising drug-loaded biomaterial for liver cancer chemotherapy. 21

22

1 1. Introduction

It is reported that live cancer (hepatocellular carcinoma) has resulted in 754,000 2 deaths as of 2010 worldwide, making it the 3rd leading cause of cancer death after 3 lung cancer and stomach cancer [1]. Current clinical treatments for liver cancer are 4 mainly including surgery followed by radiotherapy, chemotherapy, and some 5 6 emerging modalities (e.g. gene therapy, immunotherapy, phototherapy, and thermal 7 therapy). However, each of these single modalities could hardly provide complete 8 treatment because of their dose limitation and the resistance of cancer cells to the modality [2]. For example, chemotherapy is nonspecific and accompanied by side 9 effects, due to lack of drug targeting, resistance of cancer cells and problems in 10 permeability, solubility and stability of anticancer drugs [3]. 11

Nanomedicine, the application of nanotechnology to medicine, has shown 12 13 significant prospects solving these problems in cancer chemotherapy. Especially biodegradable polymeric NPs formulation, which could provide controlled and 14 15 sustained delivery of anticancer drugs, is expected to fundamentally to change the landscape of pharmaceutical and biotechnology industries[4-6]. Due to the 16 pathophysiological condition and anatomical changes caused by cancer, polymeric 17 NPs designed for drug delivery system can be exploited for passively targeting of 18 drugs [7]. Thus, an enhanced permeability and retention (EPR) effect of the polymeric 19 20 NPs can be found in tumors as a result of the increased vascular permeability coupled 21 with an impaired lymphatic drainage. Moreover, high drug loading capacity, high stability, excellent tolerability, protection of incorporated labile drugs from 22

1	degradation, controlled release and feasibility of variable routes of administration (e.g.
2	parenteral, oral, dermal, ocular, pulmonary, and rectal) are important technological
3	advantages brought by drug-loaded NPs [8]. Polymeric NPs could also reduce the
4	multidrug resistance characterized by many anti-cancer drugs (e.g. docetaxel) through
5	endocytosis of the drugs [9], and prevent drug efflux from cells mediated by the
6	P-glycoprotein [10]. Compared with conventional chemotherapy, polymeric NPs in
7	drug delivery system have advantages on overcoming annoying problems such as the
8	toxicity of drugs to normal tissues, limited aqueous solubility, short circulation
9	half-life in plasma and non-selectivity [11].
10	Biodegradable polymers approved by FDA, such as poly (ɛ-caprolactone) (PCL)
11	and poly(lactide) (PLA), are widely used in research of polymeric nanoformulations
12	nowadays [12-15]. PCL has attracted considerable interest owing to its permeability
13	to drugs, non-cytotoxicity, miscible properties with other polymers and
14	biocompatibility [16, 17]. However, PCL is easy to crystallize, which attenuates its
15	compatibility with soft tissues and makes it degrade much slower than PLA, thus
16	limiting the application of PCL in nanomedicine [18]. Nevertheless, this problem
17	could be settled through the copolymerization of ϵ -caprolactone with other monomers
18	[19, 20]. PLA is a thermoplastic aliphatic polyester and can be used as medical
19	implants for its biodegradable property [21, 22]. Copolymerization between lactide
20	enantiomers leading to PLA with variable stereoregularity, is a useful approach to
21	adjust degradability as well as mechanical and physical properties. It was reported

22 that PEGylation, chemical modification with poly(ethylene glycol) (PEG), is an

1	effective way to improve biological potencies of various pharmaceutical formulations
2	in vivo [23], and drugs conjugation to PEG or formulated in PEG copolymer
3	nanocarriers can enhance its solubility, permeability, stability and thus oral
4	bioavailability [24, 25]. Novel polymers with desired parameters between properties
5	of the parent polymers could be obtained by random copolymerization, which is
6	valuable for nanoparticle-based drug delivery system [26-28]. Cholic acid (CA),
7	composed of a steroid unit with three hydroxyl groups and one carboxyl group, is the
8	main bile acid in body. Due to its biological origin, CA was selected as the
9	polyhydroxy initiator, generating better biocompatibility for polymers incorporated
10	with CA moiety [2, 29]. Moreover, it is reported that CA functionalized copolymers
11	exhibited faster hydrolytic degradation rate compared with PCL, and the existence of
12	CA moiety in drug delivery system could also significantly enhance both cell
13	proliferation and adherence [30]. Branched polymers with useful rheological and
14	mechanical properties, such as star-shaped polymers, hyperbranched polymers and
15	dendrimers, have attracted a great deal of attention by research groups all over the
16	world [31-33]. With several branches extending from a single point [34], Star-shaped
17	polymers applied in drug carriers have advantages such as lower solution viscosity,
18	smaller hydrodynamic radius, encapsulation efficiency and higher loading content
19	compared with linear polymers in drug delivery system [32, 35-37]. Therefore,
20	star-shaped block copolymers based on PCL, PLA, PEG_{1k} and CA can provide
21	prospects for preparing excellent biodegradable drug-carriers [38, 39]. In this study,
22	we successfully synthesized a star-shaped random copolymer

CA-(PCL-*ran*-PLA)-*b*-PEG_{1k} with three branch arms for docetaxel (DTX) delivery.
 The DTX-loaded NPs exhibited satisfactory drug loading content, encapsulation
 efficiency, and achieved significant *in vivo* therapeutic effects for liver cancer
 treatment.

5

6 **2. Materials and methods**

7 2.1 Materials

8 Agents: D,L-lactide (3,6-dimethyl-1,4-dioxane-2,5-dione, $C_6H_8O_4$), cholic acid (CA), 1,3-diisopropylcarbodiimide (DCC), and 4-(dimethylamino)pyridine (DMAP) 9 were purchased from Sigma (St. Louis, MO, USA). E-caprolactone (CL) was obtained 10 from Acros Organics (Geel, Belgium). Poly(ethylene glycol) (PEG M_n 1000) was 11 12 purchased from Shanghai Yare Biotech, Inc. (Shanghai, China). DTX and commercial Taxotere® were provided by Shanghai Jinhe Bio-tech Co., Ltd. (Shanghai, 13 Stannous 14 China). octoate $(Sn(Oct)_2)$ and 15 3-(4,5-Dimethylthiazol-2-yl)-2,5-Diphenyltetrazolium Bromide (MTT) were supplied by Sigma (St. Louis, MO, USA). Acetonitrile and methanol were obtained from EM 16 17 Science (ChromAR, HPLC grade, Mallinckrodt Baker, USA). All other agents used were of analytical reagent grade. Ultrahigh pure water utilised throughout all 18 experiments was produced by Boon Environmental Tech. Industry Co., Ltd. (Tianjin, 19 20 China).

Cells and Animals: HepG2 cells (Human liver carcinoma cell line) were from
 American Type Culture Collection (ATCC, Rockville, MD, USA). The female severe

combined immunodeficient (SCID) mice were purchased from the Institute of
 Laboratory Animal Sciences, Chinese Academy of Medical Science. All the protocols
 for the proposed human cancer cell lines and animal experiments were approved by
 the Administrative Committee on Animal Research in the Tsinghua University.

5

6 2.2. Synthesis of star-shaped copolymer CA-(PCL-ran-PLA)-b-PEG_{1k}

The star-shaped copolymer CA-(PCL-*ran*-PLA)-*b*-PEG_{1k} was synthesized as
described in the literature [2,5], including three stages of chemical reaction: Ring
opening polymerization of ε-caprolactone and lactide monomer initiated by cholic
acid, Carboxylated reaction of PEG, and Esterification reaction of CA-(PCL-*ran*-PLA)
and CPEG.

12 2.2.1. Ring opening polymerization reaction

13 D,L-lactide (2.88g, 20 mmol), \varepsilon-caprolactone (9.12 g, 80 mmol), initiator cholic 14 acid (0.41 g, 1 mmol) were weighted in a vacuum sealed tube with four drops of 15 stannous octoate as catalyst. The mixture in the sealed tube was placed in the constant 16 temperature drier at 160 $^{\circ}$ C and allowed to react for about 12 h. After cooling to 17 room temperature, the sealed tube was opened, and the resulting copolymers were dissolved in DCM and then precipitated in excess cold ether, using methanol to 18 remove unreacted monomers. The final product, pure CA-(PCL-ran-PLA) was 19 collected by filtration and vacuum dried at 40 °C for 24 h. In addition, the linear 20 21 copolymer PEG_{1k}-b-(PCL-ran-PLA) was synthesized in the same way except the 22 initiator cholic acid was replaced by PEG.

2.2.2. Carboxylated reaction of PEG

2	PEG (M_n =1000, 10 g, 10.0 mmol), succinic anhydride (1.2 g, 12.0 mmol),
3	DMAP 1.22 g, 10.0 mmol), and TEA (1.01 g, 10.0 mmol) were dissolved in 40 mL of
4	anhydrous dioxane and stirred for 24 h at room temperature. The solvent was
5	completely evaporated with a rotary evaporator. The residue was dissolved in DCM,
6	washed with 10% hydrochloric acid and saturated brine respectively, dried by
7	anhydrous magnesium sulfate and filtered to remove unreacted succinic anhydride
8	and other substance. After filtration, the solution then precipitated in excess cold ether.

9 The final product (pure CPEG) was collected and vacuum dried at 35 $^{\circ}$ C for 24 h.

10 2.2.3. Esterification reaction between CA-(PCL-ran-PLA) and CPEG

Certain amounts of CA-(PCL-ran-PLA), CPEG (obtained through previous 11 12 reactions), dicyclohexyl carbodiimide (DCC), and DMAP were weighted in a dried 13 Schlenk tube connected to an argon-filled balloon, which can create an argon atmosphere. The mixture was allowed to react by magnetic stirring under room 14 temperature for 24 h. The reaction byproduct dicyclohexylcarbodiurea (DCU) was 15 removed by filtration and then precipitated in anhydrous ether. The obtained 16 star-shaped block polymer CA-(PCL-ran-PLA)-b-PEG_{1k} was purified by solvent 17 18 extraction using ether and benzene as a co-solvent. The final product (white powder) was collected and dried in vacuo at 40 $\,^{\circ}C$. 19

20

21 **2.3.** Characterization of star-shaped CA-(PCL-ran-PLA)-b-PEG_{1k} copolymer

22	The	structure	of	synthesized	CA-(PCL-ran-PLA)	and
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1	CA-(PCL-ran-PLA)-b-PEG _{1k} was confirmed by NMR (Bruker AMX 500) with
2	CDCl ₃ used as a solvent. Gel permeation chromatography (Waters GPC analysis
3	system with RI-G1362A refractive index detector, Milford, USA) was used to
4	measure the molecular weight and molecular weight distribution. THF was used as
5	the eluent at a flow rate of 1 ml/min to calibrate the data against polystyrene standards.
6	The thermal properties of star-shaped CA-(PCL-ran-PLA)-b-PEG _{1k} were studied by
7	thermogravimetric analysis (TGA, TGA Q500 thermogravimetric analyzer, USA).
8	During TGA measurement, approximately 10 mg of copolymer sample was heated
9	from 40 $^\circ\!\mathrm{C}$ to 600 $^\circ\!\mathrm{C}$ at a rate of 20 $^\circ\!\mathrm{C}/\!\mathrm{min}.$ The weight loss pattern in the
10	copolymer thermogram could be related to the composition of the polymer.

11

12 2.4. Formulation of DTX-loaded CA-(PCL-ran-PLA)-b-PEG_{1k} NPs

13 A modified nano-precipitation method with an acetone-water system was used to prepare DTX-loaded CA-(PCL-ran-PLA)-b-PEG_{1k} NPs [40]. In a word, 20 mg DTX 14 powder and 200 mg CA-(PCL-ran-PLA)-b-PEG_{1k} copolymer were dissolved in 16 ml 15 16 acetone, and then the organic solution was injected into 200 ml TPGS aqueous solution (0.03%, w/v) under gentle stirring. Afterwards, the mixture was stirred at the 17 18 speed of 800 rpm overnight to remove acetone completely (at room temperature). The 19 resulting particles suspension was centrifuged at 15,000 rpm for 20 min (4 °C), and the precipitates were washed three times to get rid of the emulsifier and 20 21 unencapsulated DTX. The resulting particles were resuspended in 10 ml deionized water, and placed in -80 °C refrigerator overnight. At last, the solid was freeze-dried 22

for two days. Drug-loaded CA-(PCL-*ran*-PLA) and PEG_{1k} -*b*-(PCL-*ran*-PLA) NPs
 were fabricated with the same method. The fluorescent coumarin-6 loaded
 CA-(PCL-*ran*-PLA)-*b*-PEG_{1k} NPs were prepared in the same way except coumarin-6
 was encapsulated instead of DTX.

5

6 **2.5. Nanoparticles Characterization**

7 2.5.1. Particle size, zeta potential and surface morphology

8 For the measurement of particle size and zeta potential, the NPs were 9 resuspended in deionized water before experiments, and Malvern Mastersizer 2000 10 (Zetasizer Nano ZS90, Malvern Instruments Ltd., UK) was used in this study. The 11 data were gained with the average of three times.

For observation of the surface morphology of NPs, a field emission scanning 12 13 electron microscopy (FESEM, S-4800, Hitachi, Tokyo, Japan), operated at a 15.0 kV accelerating voltage, was used in this experiment. The particles were fixed on a 14 15 copperplate by double-sided sticky conductive adhesives and then coated with platinum layer by JCF-1300 automatic fine platinum coater to prepare samples for 16 17 FESEM. Transmission electron microscopy (TEM, Tecnai G2 20, FEI Company, Hillsboro, Oregon, USA) was used to further observe the resultant NPs. Sample was 18 dropped onto a carbon-coated-on lacey support film and the lacey support film was 19 20 allowed to dry before characterization.

21 2.5.2. Drug loading and encapsulation efficiency

22

The drug loading content (LC) and encapsulation efficiency (EE) of DTX-loaded

1	NPs was determined by HPLC (LC 1200, Agilent Technologies, Santa Clara, CA)
2	with the same method previously published in our study [41]. 5 mg DTX-loaded NPs
3	were dissolved in 1 ml DCM under vigorous vortexing. Then the solution was
4	transferred to 5 ml of mobile phase consisting of acetonitrile and deionized water
5	(50:50, v/v). A nitrogen stream was introduced to evaporate the DCM for about 20
6	min, and then a clear solution was obtained for HPLC analysis. The flow rate of
7	mobile phase was set at 1.0 ml/min. A reverse-phase C_{18} column (150×4.6 mm, 5 μ m,
8	C ₁₈ , Agilent Technologies, CA, USA) was used in this study, and the column effluent
9	was detected at 227 nm with a UV/VIS detector. The drug loading content and
10	encapsulation efficiency of the DTX-loaded NPs were calculated by the following
11	equations respectively [20]. Each batch was conducted in three times.

12
$$LC (\%) = \frac{Weight of DTX in the nanoparticles}{Weight of the nanoparticles} \times 100\%$$

13
$$EE (\%) = \frac{Weight of DTX in the nanoparticles}{Weight of the feeding DTX} \times 100\%$$

14 *2.5.3. In vitro drug release*

In vitro DTX release from drug-loaded nanoparticles was performed as the followings. 5 mg lyophilized DTX-loaded NPs were dispersed in 5 mL Phosphate-Buffered Saline (PBS, pH = 7.4, containing 0.1%w/v Tween 80) to form a suspension. Tween 80 was used to raise the solubility of DTX in PBS and prevent it from adhering to the tube wall. Then the suspension was transferred to a dialysis bag (MWCO: 3500 Da, Spectra/Por[®] 6, Spectrum Laboratories, CA, USA), which was immerged in 15 mL of PBS release medium in a centrifuge tube. The tube was transferred into an orbital water bath and shaken at 120 rpm at 37°C. At designated time intervals, 10 mL of release medium was taken out for HPLC analysis and replaced with fresh PBS solution, and the tube was put back into the shaker for continuous analysis. The accumulative release of drug from DTX-loaded NPs was plotted against time.

6

7 2.6. Cellular uptake of fluorescent NPs

8 The HepG2 cells were cultivated in the chambered cover glass system in Dulbecco's Modified Eagle's Medium (DMEM) supplemented with 10% 9 heat-inactivated fetal bovine serum, 100 U/ml penicillin, and 100 mg/ml streptomycin 10 in 5% CO₂ at 37 $^{\circ}$ C. A fluorescent probe (coumarin-6) was used to replace the DTX 11 in the nanoformulation for the observation and analysis of cellular uptake of NPs [42, 12 43]. After the cells were incubated with 250 µg/ml coumarin-6 loaded 13 CA-(PCL-ran-PLA)-b-PEG_{1k} NPs at 37 °C for 4 h, the cells were rinsed with cold 14 15 PBS for three times and then fixed by cold methanol for 20 min. The nuclei were counterstained with 4',6-diamidino-2-phenylindole dihydrochloride (DAPI, Fluka, 16 17 Buche, Switzerland) for 10 min, then the stained cells were washed three times again with PBS to get rid of free DAPI. In order to visualize HepG2 cells, the chambers 18 were mounted onto the confocal laser scanning microscope (CLSM, Olympus 19 20 Fluoview FV-1000, Tokyo, Japan) with imaging software. Images of the cells were determined with a differential interference contrast channel. Nuclei of cells stained by 21 DAPI and coumarin-6 loaded NPs were recorded by the following channels: blue 22

1	channel with excitation at 340 nm and green channel with excitation at 485 nm.
2	For flow cytometric (FCM) assay, the HepG2 cells were incubated with 250
3	µg/ml coumarin-6 loaded CA-(PCL-ran-PLA), PEG _{1k} -b-(PCL-ran-PLA) and
4	CA-(PCL-ran-PLA)-b-PEG _{1k} NPs at 37 °C for 1 h, respectively. The cells were
5	collected and washed with PBS, and the intracellular fluorescence of coumarin-6 was
6	detected by FCM after excitation at 488 nm. Fluorescence emission at 530 nm from
7	10,000 cells were collected, amplified and scaled to generate single parameter
8	histogram.
9	For the quantitative analysis, HepG2 cells (initial density of 1×10^4 cells/well)
10	were plated in 96-well plates and cultivated overnight [44, 45]. Hank's buffered salt
11	solution (HBSS) was used to equilibrate the cells at 37 $^\circ C$ for 1 h, then coumarin-6
12	loaded NPs were added at concentrations of 100, 250 and 500 μ g/ml, respectively.
13	After 2 hour-incubation, the medium was removed and the wells were washed three
14	times with 50 μl cold PBS solution. At last, 50 μl 0.5% Triton X-100 in 0.2 N sodium
15	hydroxide was put into each sample well to lyse the cells.
16	

17 **2.7.** In vitro cell viability assay

HepG2 cells were seeded in 96-well plates at the density of 5000 viable cells per well, and incubated 24 h to allow cell attachment. After that, the cells were incubated with DTX-loaded CA-(PCL-*ran*-PLA)-*b*-PEG_{1k} NPs, PEG_{1k}-*b*-(PCL-*ran*-PLA) NPs suspension, commercial Taxotere[®] at 0.25, 2.5, 12.5 and 25 μ g/ml equivalent DTX concentrations and drug-free CA-(PCL-*ran*-PLA)-*b*-PEG_{1k} NPs suspension with the 1 same amount of NPs for 24, 48 and 72 h, respectively. The formulations were 2 replaced with DMEM containing MTT (5 mg/ml) at a determined time, and incubated cells for an additional 4 h. Then MTT was aspirated off and DMSO was added to 3 dissolve the formazan crystals (incubated for 2 h at 37 °C in dark). A microplate 4 5 reader (Bio-Rad Model 680, UK) was used to measure absorbance at 570 nm, and 6 untreated cells were taken as control with 100% viability, while cells without addition 7 of MTT were used as blank to calibrate the spectrophotometer to zero absorbance. IC_{50} , drug concentration at which inhibited of 50% cell growth, was calculated by 8 curve fitting of the cell viability data in comparison with that of the control samples. 9

10

11 **2.8.** Anti-tumor efficacy study with xenograft tumor model

HepG2 tumor cells $(2 \times 10^6 \text{ cells/mouse})$ in the culture medium were implanted 12 13 into the subcutaneous space of BALB/c nude mice (15-20 g) at right axilla and the tumor growth in each mouse was observed frequently. Tumor size was measured by a 14 vernier caliper and its volume (V) was calculated as $V=d^2 \times D/2$, where d and D 15 represent for the shortest and the longest diameter of the tumor in mm, respectively. 16 Animals were closely observed for clinical signs and behavior. At the moment the 17 tumor volume reached around 50 mm³ (designated as the 0 day), treatments were 18 performed. The mice were randomly divided into three groups (each group has 5 19 animals, n=5). With Saline as control, the DTX-loaded CA-(PCL-ran-PLA)-b-PEG_{1k} 20 NPs and Taxotere[®] were intraperitoneal injected at a single dose of 10 mg DTX/kg in 21 saline on days 0, 4, 8 and 12. Mice were sacrificed by cervical decapitation after 14 22

1	days' treatment. The terminal tumor weight was measured and utilized to evaluate the
2	antitumor activity.
3	
4	2.9. Statistical methodology
5	All experiments were repeated at least three times unless otherwise stated. The
6	results are expressed as mean \pm SD, and the statistical significance of all results was
7	determined by Student's t-test. $P < 0.05$ was considered significant.
8	
9	3. Results and discussion
10	3.1. Synthesis and characterization of CA-(PCL-ran-PLA)-b-PEG _{1k} copolymer
11	With $Sn(Oct)_2$ as a catalyst in bulk, the star-shaped random copolymer
12	CA-(PCL-ran-PLA) was synthesized via a ring-opening copolymerization of
13	ϵ -caprolactone (CL) and D,L-lactide (LA) initiated by multifunctional cholic acid (CA)
14	at 160 $^{\circ}$ C. Studies have demonstrated that ring-opening polymerization is an efficient
15	method for preparing aliphatic polyesters with high molecular weights [46]. With the
16	help of initiators possessing hydroxyl groups, we can achieve the controlled
17	architectures of copolymers, such as linear, comb-like and star-shaped [47]. As shown
18	in Fig.1, synthesized CA-(PCL-ran-PLA) was connected to the carboxylated PEG
19	through an esterification reaction with DCC and DMAP as catalysts, ultimately
20	generating the three-arm CA-(PCL- <i>ran</i> -PLA)- <i>b</i> -PEG _{1k} copolymer.
21	In order to confirm the formation of the star-shaped random copolymer, ${}^{1}H$ NMR
22	spectrum is recorded and the detailed assignment of the signals is shown in Fig. 2.

1	The typical signals from monomers CL and LA repeating units and CA moiety could
2	be observed in CA functionalized star-shaped polymer CA-(PCL-ran-PLA). ¹ H NMR
3	(CDCl ₃): a (δ = 5.21 ppm, LA repeating unit: -C <u>H</u> CH ₃), b (δ = 4.05 ppm, CL
4	repeating unit: -C <u>H</u> ₂ OCO-), c (δ = 2.31 ppm, CL repeating unit: -COC <u>H</u> ₂ -), d+g (δ =
5	1.55-1.70 ppm, CL repeating unit: $-(C\underline{\mathbf{H}}_2)_5$ - and LA repeating unit: $-C\underline{\mathbf{H}}_3$), e ($\delta = 1.38$
6	ppm, CL repeating unit: $-(C\underline{\mathbf{H}}_2)_5$ -). The successful coupling of star-shaped
7	CA-(PCL- <i>ran</i> -PLA) and CPEG was confirmed by the appearance of peak f (δ = 3.65
8	ppm, methoxyl protons of PEG: -C <u>H</u> ₂ C <u>H</u> ₂ O-) and peak h (δ = 3.38 ppm, methoxyl
9	protons of PEG: -OC $\underline{\mathbf{H}}_3$) in the spectra. The ¹ H NMR spectrum provides evidences
10	that star-shaped copolymers CA-(PCL-ran-PLA) and CA-(PCL-ran-PLA)-b-PEG _{1k}
11	with a well-defined three-branched structure were synthesized successfully. The
12	actual molecular weights (data shown in Table 1) of star-shaped copolymers
13	CA-(PCL-ran-PLA) and CA-(PCL-ran-PLA)-b-PEG _{1k} were calculated from the peak
14	areas integral ratio of a (δ = 5.21 ppm), b (δ = 4.05 ppm) and f (δ = 3.65 ppm).

15 For the purpose of investigating the thermal properties, TGA was conducted on the synthesized random copolymer CA-(PCL-ran-PLA)-b-PEG_{1k}. The typical thermal 16 decomposition profiles for carboxyl-terminated PEG, CA-(PCL-ran-PLA) and 17 18 CA-(PCL-ran-PLA)-b-PEG_{1k} were displayed in Fig.3. The thermal decomposition 19 process of star-shaped copolymer CA-(PCL-ran-PLA)-b-PEG_{1k} had two stages of 20 weight loss, whereas CA-(PCL-ran-PLA) and CPEG only had a single step of mass 21 loss. The combustion of a new component in the copolymer was marked by each turning point: the first stage (approximately 250-380°C) should be attributed to the 22

decomposition of CA-(PCL-*ran*-PLA) segments, while the second stage
(approximately 380-450°C) be assigned to the decomposition of PEG segments,
further proving the presence of two blocks in the star-shaped copolymer
CA-(PCL-*ran*-PLA)-*b*-PEG_{1k}.

The molecular weights of the obtained copolymers were measured by ¹H NMR 5 6 spectra and GPC, respectively. The detailed results are presented in Table 1. For 7 star-shaped copolymers CA-(PCL-ran-PLA) and CA-(PCL-ran-PLA)-b-PEG_{1k}, the $M_{\rm p}$ evaluating from ¹H NMR spectrum is larger than that of GPC (14,361.86 vs 8 12,045.32, 17,535.43 vs 14,972.69, respectively). This could be attributed to the 9 molecular weight estimated by GPC analysis using linear polymer as calibration. 10 11 Star-shaped copolymer has a smaller hydrodynamic volume than linear polymer with similar molecular weight, and can hardly be expanded in solution. Furthermore, the 12 13 molecular weight polydispersity of the copolymer is rather narrow.

14

15 **3.2. Formulation and characterization of NPs**

As displayed in Fig. 4 (A), the DTX-loaded CA-(PCL-*ran*-PLA)-*b*-PEG_{1k} NPs was formulated by a modified nano-precipitation method with acetone as acceptable solvent, which provided a facile, mild, and energy-saving pathway for drug encapsulation of polymeric NPs. Random copolymers and DTX without any chemical modification could be completely dissolved in acetone to generate a homogenous and clear solution. The formulation process could be described as followings: When the acetone solution was injected into the continuous stirred aqueous solution, the 1 water-insoluble DTX was immediately precipitated. In the meantime the hydrophobic 2 -(PCL-ran-PLA)- segment of the star-shaped CA-(PCL-ran-PLA)-b-PEG_{1k} was also precipitated rapidly, resulting in spontaneous formation of DTX into NPs [48]. In 3 order to obtain stable DTX-loaded CA-(PCL-ran-PLA)-b-PEG_{1k} NPs aqueous 4 5 dispersion, the mixed solution was stirred overnight to volatilize organic solvent 6 acetone. Eventually, the NPs possess a hydrophilic stabilization shell (PEG segment), 7 a core-shell structure (hydrophobic PCL-ran-PLA segment) and a hydrophobic core 8 (entrapped DTX).

9 3.2.1. Size, zeta potential and surface morphology of NPs

Cellular uptake, drug release, in vivo pharmacokinetics and biodistribution are 10 closely related to particle size and surface properties [49], a dynamic light scattering 11 (DLS) was used in order to measure the size and size distribution of the DTX-loaded 12 13 NPs, and the data is shown in Table 2. The mean hydrodynamic size of DTX-loaded NPs is about 90~140 nm in diameter. Within this size range, NPs can easily 14 15 accumulate in tumor vasculature under the influence of the enhanced permeability and retention effect [50]. The particle size 16 average of DTX-loaded CA-(PCL-ran-PLA)-b-PEG_{1k} NPs is about 95.1 nm and the polydispersity index (PDI) 17 value is 0.167. Compared with those of CA-(PCL-ran-PLA) NPs (~127.2 nm, PDI 18 19 0.203) and DTX-loaded PEG_{1k}-b-(PCL-ran-PLA) NPs (~140.6 nm, PDI 0.237), the 20 average particle size were much smaller and the size distribution were much narrower. 21 The data indicated that star-shaped and constrained geometry architecture of 22 copolymer may be attributed to small size and narrow size distribution of DTX-loaded

1	NPs, which is useful for nanomedical application. The size distribution of
2	DTX-loaded CA-(PCL-ran-PLA)-b-PEG _{1k} NPs is shown in Fig.4 (B).
3	The zeta potential of DTX-loaded CA-(PCL-ran-PLA)-b-PEG _{1k} NPs was
4	observed to be -8.3 mV, a slightly increase compared with that of DTX-loaded
5	PEG _{1k} -b-(PCL-ran-PLA) NPs (-12.5 mV) and CA-(PCL-ran-PLA) NPs (-20.3 mV),
6	as shown in Fig.2. The slightly negative charge of DTX-loaded
7	CA-(PCL-ran-PLA)-b-PEG _{1k} NPs can reduce the undesirable clearance by the
8	reticuloendothelial system (e.g. liver) and improve the blood compatibility, thus
9	facilitating NPs accumulation at the tumor sites [51].
10	The surface morphology of DTX-loaded CA-(PCL-ran-PLA)-b-PEG _{1k} NPs was
11	investigated by field emission scanning electron microscopy and transmission
12	electron microscopy. Fig. 4 illustrates the FESEM (C) and TEM (D) image of
13	DTX-loaded CA-(PCL-ran-PLA)-b-PEG _{1k} NPs. The NPs seemed to have a
14	near-spherical shape with rough surface and were around 60 nm in diameter, reaching
15	a significantly smaller size than that obtained from DLS experiments. This difference
16	may be caused by a tendency of shrink and collapse while the NPs are in dry state
17	[51].
18	3.2.2. Drug loading content and encapsulation efficiency

As displayed in Table 2, both the drug loading content (LC) and encapsulation
efficiency (EE) of star-shaped polymer CA-(PCL-*ran*-PLA)-*b*-PEG_{1k} NPs
(10.13±0.4%, 97.98±1.9%) and CA-(PCL-*ran*-PLA) (9.27±0.2%, 89.36±2.9%) were
higher than linear polymer PEG_{1k}-*b*-(PCL-*ran*-PLA) NPs (8.67±0.3%, 81.57±4.5%),

1	indicating the existence of a stronger binding affinity between hydrophobic DTX and
2	the star-shaped core region PCL-ran-PLA. In the meantime, compared with the linear
3	polymer NPs and CA-(PCL-ran-PLA) NPs, CA-(PCL-ran-PLA)-b-PEG _{1k} NPs can
4	wrap more drugs and be more efficient in the same amount, suggesting
5	CA-(PCL-ran-PLA)-b-PEG _{1k} can be a better biomaterial applied in nanotechnology
6	and drug delivery system. As shown in Fig.5, the average particle size and zeta
7	potential of the DTX-loaded NPs (redispersion in PBS) hardly changed during the
8	investigation period, i.e. the DTX-loaded NPs exhibited a good redispersion stability.

9 *3.2.3. In vitro DTX release from NPs*

Fig. 6 shows the cumulative in vitro release profiles of DTX-loaded 10 PEG_{1k}-b-(PCL-ran-PLA), CA-(PCL-ran-PLA) and CA-(PCL-ran-PLA)-b-PEG_{1k} NPs 11 12 in release medium (suspended in PBS containing 0.1% w/v Tween 80, pH 7.4) at 13 37 °C. As can be observed in the figure, the drug release from 14 CA-(PCL-ran-PLA)-b-PEG_{1k} NPs was found to be 69.46% and 79.26% of the 15 encapsulated DTX in the first 7 days and after 15 days, which was much faster than that of PEG_{1k}-b-(PCL-ran-PLA) NPs (39.82% and 45.13% in the same periods), and 16 that of CA-(PCL-ran-PLA) NPs (52.87% and 63.24% in the same periods), 17 18 demonstrating the star-shaped random polymer for nanoformulation may have faster 19 drug release rate than the linear polymer does at a similar molecular weight. The 20 fastest drug release speed of CA-(PCL-ran-PLA)-b-PEG_{1k} NPs could be attributed to 21 the shorter arms and smaller size. The release profiles of DTX-loaded NPs exhibited a typical biphasic release pattern: the DTX release from the PEG_{1k}-b-(PCL-ran-PLA) 22

NPs, CA-(PCL-*ran*-PLA) NPs and CA-(PCL-*ran*-PLA)-*b*-PEG_{1k} NPs showed an initial burst of 21.34%, 28.93% and 36.42% in the first day, respectively. At last the cumulative drug release of NPs approached 45~80% after 15 days. This pattern could be caused by drug poorly entrapped or just beneath the periphery of the NPs, while the subsequent sustained release was mainly caused by the diffusion of drug which was well encapsulated in the rigid core of NPs.

7

8 **3.3.** Cellular uptake of fluorescent NPs by HepG2 cells

9 In order to visualize and analyze cellular uptake of the NPs, a fluorescent probe called coumarin-6 was used to represent the DTX in the NPs and confocal laser 10 scanning microscopy (CLSM) was used to observe this process and distribution of 11 NPs in cells. Fig. 7 displays CLSM images of HepG2 cells after 4 h incubation with 12 13 the coumarin-6 loaded CA-(PCL-ran-PLA)-b-PEG_{1k} NPs suspension in the Dulbecco's modified Eagle medium (DMEM) at 250 µg/ml nanoparticle 14 15 concentration. The images were obtained from (A) the DAPI channel (blue); (B) the EGFP channel (green); (C) the overlay of the two channels. It can be seen from the 16 17 figure that the coumarin-6 loaded CA-(PCL-ran-PLA)-b-PEG_{1k} NPs (green) were closely located around the nuclei (blue, stained by DAPI), demonstrating the 18 19 fluorescent NPs had been internalized into the HepG2 cells.

The cellular internalization and sustained retention properties of the NPs were demonstrated to play an important role on the therapeutic effects of the drug-loaded nanoformulation [52]. For valuing the internalization and sustained retention

1	properties of PEG_{1k} -b-(PCL-ran-PLA), CA-(PCL-ran-PLA) and				
2	CA-(PCL-ran-PLA)-b-PEG _{1k} NPs, the cellular uptake efficiency of the coumarin-6				
3	loaded NPs was tested through 2 h incubation. As presented in Fig. 7 (D), the cellular				
4	uptake efficiency of all coumarin6-loaded NPs decreased with the increase of the				
5	incubated NPs concentration from 100 to 500 μ g/ml. It has been reported that NPs				
6	could be internalized into the cells, but the endocytic process of NPs could be				
7	influenced by particle size [53, 54]. In consistent with these reports, our data showed				
8	that the cellular uptake efficiency of CA-(PCL-ran-PLA)-b-PEG _{1k} NPs was 1.83-,				
9	2.17-, and 2.12-fold higher than that of PEG _{1k} -b-(PCL-ran-PLA) NPs at the incubated				
10	NPs concentration of 100, 250, and 500 μ g/ml, respectively. Furthermore, the cellular				
11	uptake efficiency of both CA-(PCL-ran-PLA)-b-PEG _{1k} and CA-(PCL-ran-PLA) NPs				
12	was higher than that of PEG_{1k} - <i>b</i> -(PCL- <i>ran</i> -PLA) NPs at any concentration.				
13	Cellular uptake of coumarin-6 loaded NPs was also confirmed by flow				
14	cytometric (FCM) assay. As shown in Fig. 7 (E), the fluorescence intensity of cells				
15	treated with CA-(PCL-ran-PLA)- b -PEG _{1k} NPs was the highest, followed by that of				
16	CA-(PCL-ran-PLA) NPs, and that of PEG _{1k} -b-(PCL-ran-PLA) NPs was the lowest				
17	after 1 h incubation, further indicating that CA-(PCL-ran-PLA)-b-PEG _{1k} NPs was in				
18	an appropriate size distribution easy for cellular internalization and sustained retention,				
19	which could also lead to excellent therapeutic effects.				
20					

21 **3.4. In vitro cell viability assay**

22 For the purpose of investigating the *in vitro* cytotoxicity of DTX-loaded

PEG_{1k}-*b*-(PCL-*ran*-PLA) NPs, DTX-loaded CA-(PCL-*ran*-PLA)-*b*-PEG_{1k} NPs and
drug-free CA-(PCL-*ran*-PLA)-*b*-PEG_{1k} NPs, MTT assay was conducted in the HepG2
cell line compared with clinically available DTX formulation, i.e. Taxotere[®]. The
DTX-loaded NPs were sterilized by gamma ray to exclude any contamination before
suspended and cultured with HepG2 cells.

Fig. 8 shows the in vitro cell viability of HepG2 cells after 24, 48, and 72 6 7 hour-incubation with DTX-loaded CA-(PCL-ran-PLA)-b-PEG_{1k} NPs, DTX-loaded PEG_{1k}-b-(PCL-ran-PLA) NPs and Taxotere[®] at equivalent DTX concentrations of 8 9 0.25, 2.5, 12.5 and 25 µg/ml, respectively. Drug-free CA-(PCL-ran-PLA)-b-PEG_{1k} NPs were used in control groups. As can be concluded from the figure, no obvious 10 cytotoxic activity was observed for drug-free CA-(PCL-ran-PLA)-b-PEG_{1k} NPs at 11 various concentrations from 0.25 μ g/ml to 25 μ g/ml, indicating the synthesized 12 13 star-shaped CA-(PCL-ran-PLA)-b-PEG_{1k} copolymer could be biocompatible and nontoxic to tissues and cells. The cellular viability decreased with the prolonged 14 incubation time for both Taxotere[®] and DTX-loaded NPs, exhibiting a 15 dose-dependent and time-dependent effect, especially for 16 drug-loaded CA-(PCL-ran-PLA)-b-PEG_{1k} NPs. DTX-loaded CA-(PCL-ran-PLA)-b-PEG_{1k} NPs 17 exhibited the best cytotoxicity efficacy against HepG2 cells, compared with 18 Taxotere[®] and PEG_{1k}-b-(PCL-ran-PLA) NPs. For example, the HepG2 cellular 19 viability after 72 hours incubation at the 12.5 µg/ml drug concentration was 43.96% 20 for Taxotere[®], 38.27% for PEG_{1k}-b-(PCL-ran-PLA) NPs, and 18.36% for 21 CA-(PCL-ran-PLA)-b-PEG_{1k} NPs. Furthermore, in contrast with Taxotere[®], the 22

1	cytotoxicity of HepG2 cell was larger than 30.36% (** $p < 0.01$, $n=5$) and 25.60%
2	(** $p < 0.01$, $n=5$) for DTX-loaded CA-(PCL-ran-PLA)-b-PEG _{1k} NPs after 72 hours
3	of incubation at the drug concentration of 2.5 μ g/ml and 12.5 μ g/ml, respectively. The
4	cytotoxicity of drug-loaded star-shaped copolymer CA-(PCL-ran-PLA)-b-PEG _{1k} NPs
5	also exhibited better cytotoxicity efficacy than HepG2 cells in comparison with
6	DTX-loaded linear copolymer PEG _{1k} -b-(PCL-ran-PLA) NPs, demonstrating
7	star-shaped copolymer may have more superior performance than linear copolymer in
8	drug delivery system.

9 Drug carriers based on star-shaped polymers have a lot of advantages, such as lower solution viscosity, smaller hydrodynamic radius, higher drug loading content 10 (LC) and higher drug encapsulation efficiency (EE) than that of linear polymers [20, 11 12 46]. In the meantime, star-shaped DTX-loaded CA-(PCL-ran-PLA)-b-PEG_{1k} NPs have faster drug release rate and higher cell uptake efficiency than linear 13 PEG_{1k} -b-(PCL-ran-PLA) NPs. Therefore it is reasonable that the *in vitro* therapeutic 14 15 effect of star-shaped DTX-loaded CA-(PCL-ran-PLA)-b-PEG_{1k} NPs is better than that of linear PEG_{1k}-*b*-(PCL-*ran*-PLA) NPs. 16

Table 3 lists the IC_{50} values of HepG2 cells after 24, 48 and 72 hours incubation with DTX formulation Taxotere[®], drug-loaded PEG_{1k}-*b*-(PCL-*ran*-PLA) NPs and CA-(PCL-*ran*-PLA)-*b*-PEG_{1k} NPs. The IC_{50} value, which could quantitatively evaluate the *in vitro* therapeutic effects of a pharmaceutical formulation, is defined as the drug inhibitory concentration needed to cause 50% tumor cell mortality in a designated period. As shown in the table, the IC_{50} value of DTX-loaded

1	CA-(PCL-ran-PLA)-b-PEG _{1k} NPs for HepG2 cells was 10.31 µg/ml, a little higher
2	than Taxotere [®] after 24 hours incubation. However, the IC_{50} value for HepG2 cells
3	was decreased from 7.04 to 4.25 $\mu g/mL$ for Taxotere®, from 4.64 to 2.30 $\mu g/ml$ for
4	DTX-loaded PEG _{1k} -b-(PCL-ran-PLA) NPs, and from 0.456 to 0.069 µg/ml for
5	DTX-loaded CA-(PCL-ran-PLA)-b-PEG _{1k} NPs after 48 and 72 hours incubation,
6	respectively. The advantages in tumor cell inhibition followed the order:
7	$CA-(PCL-ran-PLA)-b-PEG_{1k}$ NPs formulation > $PEG_{1k}-b-(PCL-ran-PLA)$ NPs
8	formulation $>$ Taxotere [®] formulation, thus demonstrating that the DTX-loaded NPs
9	had better in vitro therapeutic effects for HepG2 cells, and the star-shaped
10	CA-(PCL-ran-PLA)-b-PEG _{1k} was superior to linear PEG _{1k} -b-(PCL-ran-PLA) as drug
11	carrier for cancer chemotherapy.

12

13 **3.5.** *In vivo* anti-tumor efficacy study

Considering the satisfactory in vitro cytotoxicity against HepG2 cells, it is 14 possible that CA-(PCL-ran-PLA)-b-PEG_{1k} NPs could be developed as a promising 15 16 deliverable vehicle for the liver cancer chemotherapy. To further investigate the in vivo anti-tumor efficacy of CA-(PCL-ran-PLA)-b-PEG_{1k}NPs formulation of DTX vs 17 Taxotere[®], BALB/c nude mice were subcutaneously (s.c.) inoculated into the right 18 flank with 2×10^6 HepG2 cells in 100 µl culture medium, and the results showed that 19 95% of the mice injected developed a tumor with an average volume of \sim 50 mm³ 20 after one week. Then the mice were randomized into 3 groups (n = 5 for each group) 21 and treated with DTX-loaded CA-(PCL-ran-PLA)-b-PEG_{1k} nanoformulation or 22

Taxotere[®], respectively at 10 mg/kg DTX dose at day 0, 4, 8, 12 through
intraperitoneal injection. Physiological saline was used as control, and the tumors size
of the mice was recorded every 2 days until the 14th day. After two-week's therapy,
all mice were sacrificed and tumors were separated from the bodies at the
experimental terminal.

The morphology of the tumors of each group is presented in Fig. 9 (A). The 6 7 images clearly show that the tumor size for Saline group was the largest, in-between for the Taxotere[®] group, and smallest for CA-(PCL-*ran*-PLA)-*b*-PEG_{1k} NPs 8 formulation group. Fig. 9 (B) shows the tumor growth surveyed in the mice after the 9 intraperitoneal injection with DTX-loaded CA-(PCL-ran-PLA)-b-PEG_{1k} NPs, 10 Taxotere[®] and physiological Saline. The tumor volume for saline group increased 11 obviously during the experimental period, while size for 12 tumor CA-(PCL-ran-PLA)-b-PEG_{1k} nanoformulation group was significantly inhibited. The 13 weights of the tumors in each group showed in Fig. 9 (C) also indicated the 14 advantages of the DTX-loaded CA-(PCL-ran-PLA)-b-PEG_{1k} NPs vs Taxotere[®] in 15 suppressing tumors were significant. In summary, it can be confirmed that DTX 16 nanoformulation in the CA-(PCL-ran-PLA)-b-PEG_{1k} NPs could maintain its 17 pharmacological activity and significantly inhibit the tumor growth than Taxotere[®] at 18 the same dose. 19

20

21 **4.** Conclusions

22

In this research, a novel system of cholic acid functionalized star-shaped

1	CA-(PCL-ran-PLA)-b-PEG _{1k} NPs was developed for sustained and controlled
2	delivery of DTX applied in liver cancer therapy. The DTX-loaded
3	CA-(PCL-ran-PLA)-b-PEG _{1k} NPs were prepared by a modified nano-precipitation
4	technique. The hydrodynamic size of DTX-loaded CA-(PCL-ran-PLA)-b-PEG _{1k} NPs
5	was about 90 nm, much smaller than that of CA-(PCL-ran-PLA) and linear
6	$PEG_{1k}-b-(PCL-ran-PLA)$ NPs. Compared with $PEG_{1k}-b-(PCL-ran-PLA)$ and
7	CA-(PCL-ran-PLA) NPs, the drug loading content and encapsulation efficiency of the
8	star-shaped CA-(PCL-ran-PLA)-b-PEG _{1k} NPs were improved significantly. Moreover,
9	the DTX-loaded CA-(PCL-ran-PLA)-b-PEG _{1k} NPs could achieve much faster drug
10	release speed, as well as higher cellular uptake efficiency and cytotoxicity by human
11	liver carcinoma cells (HepG2). Both in vitro cell viability experiment and in vivo
12	anti-tumor efficacy demonstrated that the DTX-loaded CA-(PCL-ran-PLA)-b-PEG _{1k}
13	NPs exhibited significantly superior anti-tumor efficiency than linear
14	PEG _{1k} -b-(PCL-ran-PLA) NPs and commercially available Taxotere [®] formulation. In
15	conclusion, the star-shaped CA-(PCL-ran-PLA)-b-PEG _{1k} is a novel and promising
16	biodegradable copolymer for nanoformulations applied in liver cancer treatment.

17

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- 3

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1 Appendix

2 Abbreviation

Abbreviation	Description		
¹ H NMR	Proton nuclear magnetic resonance		
GPC	Gel permeation chromatography		
TGA	Thermogravimetric analysis		
DLS	Dynamic light scattering		
FESEM	Field emission scanning electron microscopy		
TEM	Transmission electron microscopy		
CLSM	Confocal laser scanning microscope		
FCM	Flow cytometric		
HPLC	High performance liquid chromatography		
LC	Loading content		
EE	Encapsulation efficiency		
PDI	Polydispersity index		
CA	Cholic acid		
PCL	Poly (ε-caprolactone)		
PLA	Poly(lactide)		
PEG	Poly(ethylene glycol)		
DTX	Docetaxel		
EPR	Enhanced permeability and retention		
SCID	Severe combined immunodeficient		
DCC	1,3-diisopropylcarbodiimide		
DCU	Dicyclohexylcarbodiurea		
DMAP	4-(dimethylamino)pyridine		
MTT	3-(4,5-Dimethylthiazol-2-yl)-2,5-Diphenyltetrazolium		
	Bromide		
DAPI	4',6-diamidino-2-phenylindole dihydrochloride		
HBSS	Hank's buffered salt solution		
DMEM	Dulbecco's Modified Eagle's Medium		

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2 Fig.1. Schematic diagram of synthesis of star-shaped CA-(PCL-ran-PLA)-b-PEG_{1k}

3 random copolymer.





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Fig. 3. Thermogravimetric analysis of CPEG, CA-(PCL-*ran*-PLA) and
CA-(PCL-*ran*-PLA)-*b*-PEG_{1k} copolymers.



2 Fig. 4. (A) Schematic representation of technique for preparation of DTX-loaded CA-(PCL-ran-PLA)-b-PEG_{1k} NPs; (B) Dynamic light scattering size distribution of 3 DTX-loaded CA-(PCL-ran-PLA)-b-PEG1k NPs; (C) Field emission scanning electron 4 microscopy image of DTX-loaded CA-(PCL-ran-PLA)-b-PEG_{1k} NPs (scale bar is 100 5 (D) Transmission nm); electron microscopy image of DTX-loaded 6 CA-(PCL-ran-PLA)-b-PEG_{1k} NPs(scale bar is 100 nm). 7



Fig. 5. *In vitro* stability of DTX-loaded NPs: (A) The size distribution and (B) The
zeta potential of DTX-loaded PEG_{1k}-*b*-(PCL-*ran*-PLA), CA-(PCL-*ran*-PLA) and
CA-(PCL-*ran*-PLA)-*b*-PEG_{1k} NPs for two-week storage at 4 °C (*p < 0.05, **p <
0.01, ***p < 0.001, n = 3).



9 Fig. 6. The *in vitro* release profiles of DTX-loaded PEG_{1k}-b-(PCL-ran-PLA) NPs,

1 CA-(PCL-ran-PLA) NPs and CA-(PCL-ran-PLA)-b-PEG_{1k} NPs at 37°C (*p < 0.05,

2 **
$$p < 0.01$$
, $n = 5$).





Fig. 7. Confocal laser scanning microscope images of HepG2 cells after 4 h 5 incubation with coumarin-6 loaded CA-(PCL-ran-PLA)-b-PEG_{1k} NPs at 37 °C. The 6 7 cells were stained by DAPI (blue) and the coumarin-6 loaded CA-(PCL-ran-PLA)-b-PEG_{1k} NPs are green. The cellular uptake was visualized by 8 overlaying images obtained by DAPI filter and EGFP filter: (A) image from DAPI 9 channel; (B) image from EGFP channel; (C) image from combined DAPI channel and 10 EGFP channel. (D) Cellular uptake efficiency of coumarin-6 loaded NPs (*p < 0.05, 11 **p < 0.01, ***p < 0.001, n = 5). (E) FCM histograms for coumarin-6 loaded NPs on 12



1 HepG2 cells after 1 h incubation.



2 Fig. 9. Anti-tumor efficacy of the DTX-loaded CA-(PCL-ran-PLA)-b-PEG_{1k} NPs, Taxotere[®] and saline on the BALB/c nude mice bearing HepG2 cells: (A) Images of 3 the tumors in each group taken out from the sacrificed mice at the end point of 4 research; (B) Tumor-growth curve of SCID mice after intraperitoneal injection with 5 DTX-loaded CA-(PCL-ran-PLA)-b-PEG_{1k} NPs, Taxotere[®] and saline (*p < 0.05, **p 6 < 0.01, n=5); (C) Tumor weight of each group taken out from the sacrificed mice at 7 the end of the study. Compared with saline and Taxotere® groups, significant 8 reduction in tumor weight for CA-(PCL-ran-PLA)-b-PEG_{1k} NPs could be observed 9 (*p < 0.05, ***p < 0.001, n = 5).10

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1 Table 1

- 2 Molecular weights of star-shaped copolymers CA-(PCL-ran-PLA) and
- 3 CA-(PCL-*ran*-PLA)-b-PEG_{1k}, linear copolymer PEG_{1k}-b-(PCL-*ran*-PLA).

Sample	$M_{\rm n}({\rm NMR}^{\rm a})$	$M_{\rm n}({\rm GPC}^{\rm b})$	$M_{\rm w}({ m GPC}^{ m b})$ PDI	с
CA-(PCL-ran-PLA)	14,361.86	12,045.32	15,177.47 1.26	5
CA-(PCL-ran-PLA)-b-PEG _{1k}	17,535.43	14,972.69	17,816.68 1.19)
PEG _{1k} -b-(PCL-ran-PLA)	18,142.09	18,376.71	22,602.07 1.23	;

- 4 a Determined by 1 H NMR.
- 5 b Determined by gel permeation chromatography.
- 6 c Polydispersity index (M_w/M_n) determined by gel permeation chromatography.
- 7 $M_{\rm n}$: Number-average molecular weight.
- 8 $M_{\rm w}$: Weight-average molecular weight.
- 9

10 **Table 2**

11 Characterization of DTX-loaded NPs.

Polymer	Size (nm)	PDI	ZP (mV)	LC (%)	EE (%)
PEG _{1k} - <i>b</i> -(PCL- <i>ran</i> -PLA)	140.6 ± 7.2	0.237	-12.5 ± 2.1	8.67 ± 0.3	81.57 ± 4.5
CA-(PCL-ran-PLA)	127.2 ± 6.1	0.203	-20.3 ± 2.5	9.27 ± 0.2	89.36 ± 2.9
CA-(PCL-ran-PLA)-b-PEG _{1k}	95.1 ± 4.3	0.167	-8.3 ± 1.6	10.13 ± 0.4	97.98 ± 1.9

- 12 PDI = Polydispersity index, ZP = Zeta potential, LC = Loading content, EE =
- 13 Encapsulation efficiency, n = 3.

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15 **Table 3**

- 16 IC₅₀ values of DTX formulation in the Taxotere[®], PEG_{1k}-*b*-(PCL-*ran*-PLA) NPs and
- 17 CA-(PCL-ran-PLA)-b-PEG_{1k} NPs on HepG2 cells following 24, 48 and 72 hours
- 18 incubation (*n*=5).

Insubstian	IC ₅₀ (µg/mL)			
time (h)	Taxotere®	PEG _{1k} - <i>b</i> -(PCL- <i>ran</i> -PLA)	CA-(PCL-ran-PLA)-b-	
time (ii)		NPs	PEG _{1k} NPs	
24	19.76 ± 1.83	18.63 ± 1.61	10.31 ± 1.54	
48	7.04 ± 0.48	4.64 ± 0.52	0.456 ± 0.36	
72	4.25 ± 0.43	2.30 ± 0.17	0.069 ± 0.05	

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The DTX-loaded CA-(PCL-*ran*-PLA)-*b*-PEG_{1k} NPs were prepared by a modified nano-precipitation method and had a near-spherical shape. These DTX-loaded NPs have great potential as drug delivery nanocarriers for cancer therapy.