RSC Advances



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. This Accepted Manuscript will be replaced by the edited, formatted and paginated article as soon as this is available.

You can find more information about *Accepted Manuscripts* in the **Information for Authors**.

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard <u>Terms & Conditions</u> and the <u>Ethical guidelines</u> still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.



RSC Advances

RSC Publishing

ARTICLE

Cite this: DOI: 10.1039/x0xx00000x

Received 9th November 2013, Accepted 00th March 2014

DOI: 10.1039/x0xx00000x

www.rsc.org/

Role of membrane disturbance and oxidative stress in the mode of action underlying the toxicity of differently charged polystyrene nanoparticles

Sourav Bhattacharjee, a,b,* Dmitry Ershov, Mohammed A. Islam, Angela M. Kämpfer, Katarzyna A. Maslowska, Jasper van der Gucht, Gerrit M. Alink, Antonius T. M. Marcelis, Han Zuilhof, Ivonne M. C. M. Rietjens

Abstract: Surface charge is often hypothesized to influence toxicity of nanoparticles (NPs) including polymeric nanoparticles (PNPs) while oxidative stress is considered to be an important mode of action (MOA) for such toxicity. In order to investigate the role of membrane disturbance and oxidative stress in the MOA of PNPs, the cytotoxicity and a range of related cellular endpoints induced by monodisperse, fluorescent, cationic and anionic polystyrene nanoparticles (PSNPs) of 50 and 100 nm sizes were investigated in vitro in macrophage NR8383 cells. Only amine-terminated cationic PSNPs exhibited cytotoxicity which was accompanied by induction of intracellular reactive oxygen species (ROS), increased levels of cytoplasmic free calcium, a reduced phagocytic index, a reduced mitochondrial membrane potential $(\Delta \Psi_m)$ and a decreased intracellular ATP content with the effects being more pronounced for 50 nm than 100 nm PSNPs. Both cationic and anionic PSNPs were found to increase the roughness of the cell membrane with the effect being more profound for cationic PSNPs. The pattern of protection by cellular antioxidants against the effects induced by positive PSNPs was similar to the pattern of protection against effects induced by the mitochondrial electron transport disrupting agent 2.4dinitrophenol (DNP) and dissimilar to that for protection against the model compound for oxidative stress, i.e. hydrogen peroxide (H₂O₂). Surface charge influences the cellular interaction for NPs. The results collectively indicated that membrane interaction, and disturbance of the mitochondrial electronic transport chain (ETC) may represent a principal mechanism of toxicity for cationic PSNPs resulting in ROS production and oxidative stress as secondary effects.

Keywords: Surface charge, polystyrene nanoparticles, reactive oxygen species, mitochondria, oxidative stress

Introduction

Nanotechnology in recent years has experienced unprecedented growth with its applications ranging from energy production to food technology and medicine. The increase in nanotechnology-based products also raises the concern for health related aspects [1, 2] with regards to the inevitable human exposure. One of the less understood topics related to nanotoxicological research is the factor(s) that influence the interaction between nanoparticles (NPs) and cells and the mode of action (MOA) underlying such cellular toxicity. Surface charge and particle size are often hypothesized to be the most important factors that influence cellular toxicity of NPs [3-5]. With successful application of polymer engineering, different polymeric nanoparticles (PNPs) with diverse physico-chemical properties (like biodegradability, fluorescence etc.) can be synthesized. This provides a unique tool for the toxicologists to investigate how surface charge and particle size might influence cellular interactions and toxicity of PNPs including their MOA. Hence, an investigation into the role of surface charge on the cellular interaction and MOA for cytotoxicity of PNPs is timely and justified.

Oxidative stress caused by the induction of intracellular reactive oxygen species (ROS) production by different NPs is a popular model to explain the cytotoxicity [6-8]. However, the source of cellular ROS still remains unclear. It is possible that NPs can react with a wide variety of biomolecules producing radicals, like ROS. Simultaneously, some recent reports identified intracellular mitochondria as target organs for different NPs. Charged NPs can interact with intracellular mitochondria and dissipate the mitochondrial membrane potential $(\Delta \Psi_m)$. In continuation to a possible deterioration of mitochondrial physiology after exposure to charged NPs, an increase in the intracellular calcium can be apprehended. Previously, some groups reported an increase in intracellular calcium after exposure to cationic NPs [9, 10]. This is an interesting finding as an increase in intracellular calcium can trigger apoptotic pathways. A disturbance in mitochondrial membrane physiology can also disrupt the electron transport chain (ETC) and induce the production of ROS in addition to causing cellular ATP depletion. Then the question may be raised whether the ROS production is the cause of the cytotoxicity or rather a secondary effect occurring after ATP depletion or other effects following disruption of the mitochondrial function? To investigate the

importance of ROS production and resulting oxidative stress as a mechanism of cytotoxicity of PSNPs, cytotoxic amine-terminated cationic PSNPs (PSNP-NH₂), and anionic acid-terminated PSNPs (PSNP-COOH) which did not show any cytotoxicity, were tested for ROS production, cytotoxicity and a range of related cellular endpoints in macrophage NR8383 cells in which antioxidant levels were artificially modified. For comparison, similar experiments as done with the PSNPs, were performed using the mitochondrial ETC disrupting agent, 2,4-dinitrophenol (DNP) and with the known inducer of oxidative stress, hydrogen peroxide (H₂O₂).

The main objectives of this article are therefore to systematically investigate the effect of surface charge in the cytotoxicity of well characterized PSNPs as well as to understand the MOA of such toxicity into detail. The importance of oxidative stress as MOA in such toxicities of charged NPs was also probed.

Results

Characterization of PSNPs

The characterization of the PSNPs was done by measuring the size and polydispersity index (PDI) by dynamic light scattering (DLS), surface potential (as an indication of surface charge) by ζ -potential measurements and imaging by atomic force microscopy (AFM). The results are shown in Figure 1 and Table 1, respectively.

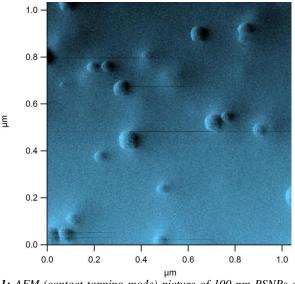


Fig 1: AFM (contact tapping mode) picture of 100 nm PSNPs on a mica surface. The different PSNPs are at different height levels on the mica surface.

Table 1: Characterization data of different PSNPs

		Surface charge (mV)		Particle size (nm)			
Polymer	PSNPs			water	DMEM	F12-K	PDI
	PSNP- NH ₃ ⁺	Positive	24 ± 5	50 ± 5	71 ± 5	78 ± 5	0.1
			23 ± 5	100 ± 5	129 ± 5	121 ± 5	0.1
Polystyrene	PSNP- COO		-21 ± 5	50 ± 5	64 ± 5	69 ± 5	0.1
		Negative	-24 ± 5	100 ± 5	116 ± 5	121 ± 5	0.1

From Figure 1 (AFM image), the PSNPs could clearly be visualized with mild variation of sizes arising due to embedding of the PSNPs in the soft layer. The dispersion of the PSNPs both in water and cell culture medium (DMEM/F12-K) containing 10% fetal calf serum (FCS) were confirmed by DLS and the PDI of the PSNPs in aqueous dispersions were found to be very low (Table 1). While in cell culture medium containing FCS, an increase in size for the PSNPs could be observed (after 4 h) which resulted from the surface adsorption of proteins added to the cell culture medium via FCS. However, despite the increase in size in the cell culture medium, the PSNPs continued to be quite monodisperse as indicated by the PDI and shown in Table 1. The PSNPs were quite stable both in aqueous and cell culture medium over at least a month (as determined by DLS) and did not show any sedimentation or aggregation.

Cytotoxicity measurement by the MTT and phagocytic index (PI) assay

The cytotoxicity of the PSNPs was measured by the MTT (Figure 2) and the PI (Figure 3) assays. At the concentrations tested (0-100 μ g/ml), only the cationic PSNPs showed signs of cytotoxicity in both these assays. For the anionic PSNPs, no such cytotoxicity could be observed. The EC50 values obtained from the MTT and PI assays are presented in Table 2. Although not always statistically significant (p < 0.05), a size-dependent effect on the cytotoxicity of PSNPs could be observed. In this case, the smaller cationic PSNPs (50 nm) were relatively more cytotoxic than the larger ones (100 nm).

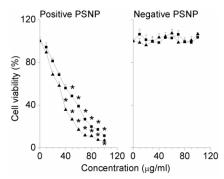


Fig 2: Cell viability measured by the MTT assay and expressed as % of negative control (0 μ g/ml) after 4 h exposure of NR8383 cells to positive and negative PSNPs of 50 (\blacktriangle) and 100 (\blacksquare) nm sizes (n=3). The asterisk (*) symbol signifies statistical difference from the negative control at p < 0.05.

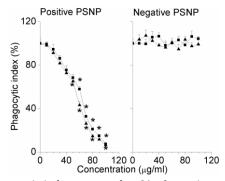


Fig 3: Phagocytic index expressed as % of negative control (0 μ g/ml) after 4 h exposure of NR8383 cells to positive and negative PSNPs of 50 (\blacktriangle) and 100 (\blacksquare) nm sizes (n=3). The asterisk (*) symbol signifies statistical difference from the negative control at p < 0.05.

Table 2: The EC50 values obtained from different experiments reported in this paper. No EC50 values were available for anionic PSNPs due to their non-toxicity in the tested concentration range.

Aggari	Figure	EC50 (µg/ml)			
Assay		50 nm PSNP	100 nm PSNP		
MTT	2	33	35		
PI	3	56	62		
$\Delta \Psi_m$	6	38	33		
DCFH-DA	7	54	68		
ATP	9	63	66		
Calcium	10	84	88		

Protective effects of vitamins E and C against the cytotoxicity of DNP, PSNP-NH₂ and H₂O₂.

To investigate the role of oxidative stress in the MOA of the PSNP-NH₂ the protective effects of vitamins E and C were investigated and compared to the protection by these antioxidants against the mitochondrial ETC disrupting agent 2,4-dinitrophenol (DNP) and the known inducer of oxidative stress, hydrogen peroxide (H_2O_2). The results obtained are shown in Figure 4. From the results on the protective effects of anti-oxidants compared to the toxicity of PSNP-NH₂, it could be seen that, although vitamins E and C were capable of providing some protection against cytotoxicity of PSNP-NH₂, the protection failed to be significant (p < 0.05). The pattern of protection by vitamins E and C against the cytotoxicity induced by positive PSNPs was similar to the pattern of protection against the cytotoxicity of the mitochondrial electron transport disrupting agent DNP but dissimilar to that for protection against cytotoxicity of the model compound for oxidative stress H_2O_2 .

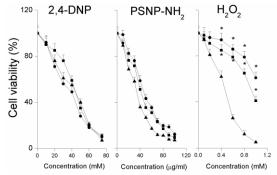


Fig 4: Effects of vitamins E (\blacksquare) and C (\bullet) on cell viability in NR8383 cells after 4 h exposure to serial dilutions of DNP (0-75 mM), PSNP-NH₂ (0-100 µg/ml) or H₂O₂ (0-1 µM) (\blacktriangle) as measured by the MTT assay (n=3). The asterisk (*) sign marks significant differences (p<0.05) compared to the corresponding values without vitamin E or C.

Effect of modulating intracellular levels of glutathione (GSH) on the cytotoxicity caused by DNP, PSNP-NH $_2$ and H $_2$ O $_2$.

The effect of modulating the intracellular levels of GSH by preexposure to buthionine sulphoximine (BSO) or N-acetyl cysteine (NAC) on the cytotoxicity caused by DNP, PSNP-NH₂ and H₂O₂ are shown in Figure 5. Both decreasing the intracellular GSH levels by incubation with BSO and increasing these levels of cellular antioxidants by incubation with NAC, did not significantly (p < 0.05) increase or decrease in cytotoxicity of DNP and PSNP-NH₂. However, for H_2O_2 the exposure to BSO and NAC significantly increased and decreased the cytotoxicity, respectively. Thus, also in these experiments the pattern of cytotoxicity and cellular protection observed for PSNP-NH₂ resembled that of the ETC uncoupler DNP more than that of the oxidative stress model compound H_2O_2 .

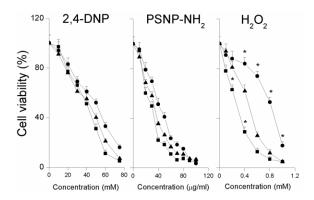


Fig 5: Effects of BSO (\blacksquare) and NAC (\bullet) on cell viability in NR8383 cells after 4 h exposure to serial dilutions of DNP (0-75 mM), PSNP-NH₂ (0-100 µg/ml) or H₂O₂ (0-1 µM) (\blacktriangle) as measured by the MTT assay (n = 3). The asterisk (*) sign marks significant differences (p < 0.05) compared to the corresponding values without BSO or NAC.

Mitochondrial membrane potential $(\Delta \Psi_m)$ assessment.

In order to investigate the probable effect of different PSNPs on intracellular mitochondria, the mitochondrial membrane potential $(\Delta \Psi_m)$ was measured. Normally, an electrochemical gradient of 130-140 mV potential difference exists across the mitochondrial membrane [11] with the outer side (towards the cytoplasm) being relatively positive compared to the inner side (towards the mitochondrial matrix). The results obtained are shown in Figure 6. The cationic PSNPs of both sizes (50 nm > 100 nm) caused a dose-dependent reduction in the $\Delta \Psi_m$ within the tested concentration range (0-100 µg/ml) in contrary to the anionic ones, which did not show any effect on the value of $\Delta \Psi_m$. The corresponding EC50 values are given in Table 2.

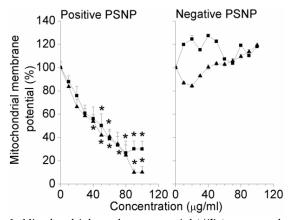


Fig 6: Mitochondrial membrane potential ($\Delta\Psi_m$) expressed as % of negative control (0 µg/ml) after 4 h exposure of NR8383 cells to positive and negative PSNPs of 50 (\blacktriangle) and 100 (\blacksquare) nm sizes (n=3). The asterisk (*) symbol signifies statistical difference from the negative control at p < 0.05.

Measurement of intracellular ROS production by the DCFH-DA assay.

The DCFH-DA (2',7'-dichlorofluorescein diacetate) assay was used to assess any induction in intracellular ROS production by different PSNPs. The results of the DCFH-DA assay are shown in Figure 7. In contrast to the anionic PSNPs, the cationic PSNPs of both sizes showed induction in ROS production, with the 50 nm PSNPs being more effective than the 100 nm. The corresponding EC50 values are given in Table 2.

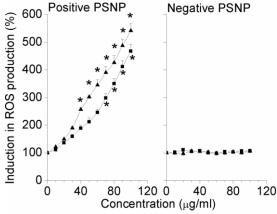


Fig 7: Induction in intracellular ROS production expressed as % of negative control (0 μ g/ml) after 4 h exposure of NR8383 cells to positive and negative PSNPs of 50 (\blacktriangle) and 100 (\blacksquare) nm sizes (n=3). The asterisk (*) symbol signifies statistical difference from the negative control at p < 0.05.

Induction of intracellular ROS measured by the DCFH-DA assay in presence of DNP, PSNP-NH₂ and H₂O₂.

Figure 8 shows the induction of ROS in NR8383 cells after 4 h exposure to serial dilutions of DNP, PSNP-NH $_2$ and H $_2$ O $_2$. In the absence of vitamins E or C, H $_2$ O $_2$ showed maximal induction of ROS production (>1100 %) followed by the PSNP-NH $_2$ (~563 %) and DNP (~500 %), all compared to the negative control. Vitamins E and C suppressed the inductions of ROS for all three exposure agents markedly, to a level hardly above control values. The pattern of ROS production induced by PSNP-NH $_2$ resembled that of the ETC uncoupler DNP more than that of the oxidative stress model compound H $_2$ O $_2$.

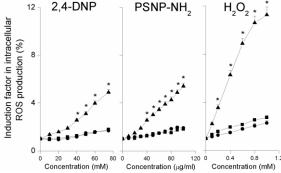


Fig 8: Effects of vitamins $E(\blacksquare)$ and $C(\bullet)$ on the intracellular ROS induction in NR8383 cells after 4 h exposure to serial dilutions of DNP (0-75 mM), PSNP-NH₂ (0-100 µg/ml) or H₂O₂ (0-1 µM) (\blacktriangle) as measured by the DCFH-DA assay (n=3). The asterisk (*) sign marks significant differences (p<0.05) compared to the negative control (0 µg/ml).

Measurement of intracellular ATP content.

The intracellular ATP was measured in NR8383 cells after 4 h exposure to different PSNPs in order to further characterize the possible interaction of different surface charge bearing PSNPs with intracellular mitochondria. The results are shown in Figure 9 with the corresponding EC50 values being shown in Table 2. For the cationic PSNPs, a dose dependent decrease in the cellular ATP content could be observed with the 50 nm PSNPs being more effective than the 100 nm, in contrast to the anionic PSNPs which failed to show any such decrease in cellular ATP content at the concentrations tested (0-100 $\mu g/ml$).

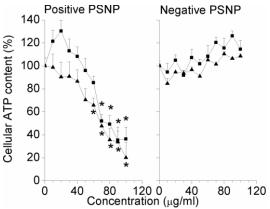


Fig 9: Cellular ATP content expressed as % of negative control (0 μg/ml) after 4 h exposure of NR8383 cells to positive and negative PSNPs of 50 (Δ) and 100 (■) nm sizes (n=3). The asterisk (*) symbol signifies statistical difference from the negative control at p < 0.05.

Measurement of intracellular calcium.

The cellular calcium content was measured in NR8383 cells after 4 h exposure to serial dilutions (0-100 $\mu g/ml)$ of different PSNPs. The results are shown in Figure 10 and the corresponding EC50 values are given in Table 2. As expected, only the cationic PSNPs (50 nm > 100 nm) caused a dose dependent increase in the cytoplasmic free calcium levels whereas no such effects were observed for anionic PSNPs.

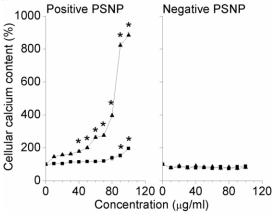


Fig 10: Cytoplasmic free calcium content expressed as % of negative control (0 μ g/ml) after 4 h exposure of NR8383 cells to positive and negative PSNPs of 50 (\blacktriangle) and 100 (\blacksquare) nm sizes (n=3). The asterisk (*) symbol signifies statistical difference from the negative control at p < 0.05.

Cellular uptake and cellular distribution of PSNPs as determined by CLSM.

The NR8383 cells were exposed to a non-toxic concentration (1 µg/ml) of both cationic and anionic 100 nm PSNPs for 4 h before checking the cellular uptake and cellular distribution of the PSNPs. The results are shown in Figure 11A. The CLSM technique was employed in order to characterize the uptake patterns as well as to perform quantitative assessment of the cellular uptake of PSNPs of different surface charges. The results from at least ten individual cells from three separate experiments were counted. The cationic PSNPs were taken up almost twice as much (Figure 11B) as the anionic PSNPs. With a detailed analysis of the data it was found that cationic PSNPs showed more interaction with the cellular periphery. For anionic PSNPs, no such enhanced interaction between the PSNPs and the cell membrane could be observed. Differences could also be seen in the intracellular distribution patterns of the two different surface charge bearing PSNPs. For anionic PSNPs, small vesicular structures could be seen inside the cells which were not seen for the cationic PSNPs. For the cationic PSNPs, the NPs were found to be diffusely distributed across the cytoplasm and also in the perinuclear regions.

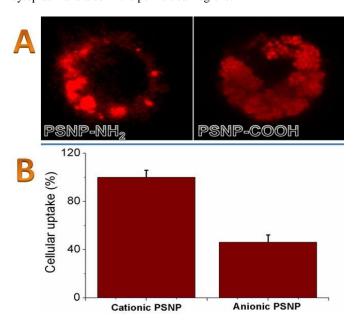


Fig 11: (A) CLSM images of NR8383 cells ($\lambda_{ex} = 543$ nm, $\lambda_{em} = 620$ nm) after 4 h exposure to positive (amine-terminated/PSNP-NH₂) and negative (acid-terminated/PSNP-COOH) 100 nm PSNP. (B) Relative cellular uptake of 100 nm cationic and anionic PSNP in NR8383 cells after 4 h exposure to non-toxic 1 µg/ml concentrations. Results are shown as mean \pm SEM. The uptake of the cationic PSNP was taken as 100% The results were expressed as an arithmetic mean of the CLSM data obtained from 20 individual cells from five separate experiments (n = 5). Scale bars show 10 µm.

Integrity of cell membrane after exposure to PSNPs.

Calcein-AM assay: To investigate the effect of charged PSNPs on cellular membrane integrity, the calcein assay was performed where non-fluorescent calcein-AM (acetomethoxy) salt was administered which after entering cells is hydrolyzed into calcein and becomes fluorescent. By CLSM, the calcein inside the cells could be visualized and this is a standard technique for testing the

integrity of the cell membrane [12]. From the results obtained (Figure 12), it can be seen that for the cationic PSNPs at a noncytotoxic concentration of 1 μ g/ml, pouches of calcein leaked out of the cell pointing at a compromised state of the cell membrane. In contrast, for the anionic PSNPs (also at 1 μ g/ml), the integrity of the cell membrane was comparable to that of a normal NR8383 cell (control). This indicates that cationic PSNPs could actually harm the structural integrity of the cell membrane.

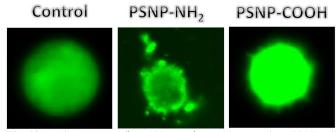


Fig 12: Calcein assay (λ_{ex} =488 nm, λ_{em} =538 nm) of NR8383 cells after 4 h exposure to non-toxic (1 µg/ml) concentration of 100 nm positive PSNP-NH₂ and negative PSNP-COOH. The positive PSNPs caused discontinuity on the cell surface with calcein coming out, which is absent for negative PSNPs. Scale bars show 10 µm.

AFM: The results of the AFM investigation are shown in Figure 13. The cationic PSNPs caused a higher disruption of cell surface and increased the overall roughness of the surface. The normal mean roughness of NR8383 cells was found to be ~5.3±1 nm whereas after exposure to the cationic PSNPs, the mean roughness showed an almost five-fold increase (~24.7±5 nm). Additionally, pores or holes of different sizes (at least ~150 nm in size) could be observed on the surface. In contrary, for anionic PSNPs, the roughness of the NR8383 cell surface did increase but only to a much lesser degree (~10.2±3 nm) and no such holes could be observed.

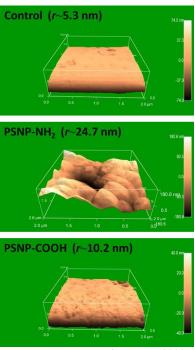


Fig 13: AFM (contact tapping mode) on NR8383 cells placed on mica surface after 4 h exposure to non-toxic (1 μ g/ml) concentrations of positive (amine terminated) and negative (acid terminated) 100 nm PSNPs. The positive PSNP created holes on the cell surface alike the negative PSNPs (n=5).

Discussion

The aim of this study was to investigate the role of membrane disturbance and oxidative stress in the MOA of the cytotoxicity of monodisperse, fluorescent anionic PSNPs of different charge and size. To this point, various cellular endpoints were identified in order to better understand the mechanism of such toxicity including the importance of oxidative stress as primary or secondary mechanism of cytotoxicity. The major observations were: 1) the cationic PSNPs showed higher cytotoxicity as well as cellular uptake compared to the anionic PSNP, indicating an effect of surface charge; 2) a sizeeffect was also found since smaller 50 nm cationic PSNP were found to be more toxic compared to the bigger 100 nm cationic PSNP; 3) positively charged PSNPs can cause disruption of the ETC which in turn can act as an inducer of intracellular ROS production, reduction of the mitochondrial membrane potential $(\Delta \Psi_m)$ and reduction of cellular ATP levels; 4) especially positive PSNP caused a disruption of the membrane integrity and pore formation could be visualized; 5) intracellular mitochondria were identified as a major target organelle in the cytotoxicity causing pathway; 6) the membrane disturbance as well as mitochondrial dysfunction were the principal mechanism of cytotoxicity (for cationic PSNPs) which further resulted into oxidative stress as secondary effect.

The PSNPs were either amine (PSNP-NH₂) or acid terminated (PSNP-COOH) rendering them to be positively (PSNP-NH₃⁺) or negatively (PSNP-COO⁻) charged in aqueous dispersions, respectively. Both the PSNP-NH₃⁺ and PSNP-COO gave stable dispersions in water as well as in cell culture medium. However, in medium, an increase in the sizes of both the PSNPs was found. This was due to surface adsorption of the protein present in the cell culture medium. The surface adsorption of protein on different engineered NPs has been investigated before and was found to influence the cellular interactions of PNPs [13, 14]. An interesting finding was that the cationic PSNPs overall showed a stronger increase in size as compared to the anionic ones. A possible explanation can be the electrostatic interactions between the surface charge of the PSNPs and the charge of these proteins [15]. The proteins, like albumin, which are present in FCS are often negatively charged at normal pH and thus experience electrostatic attraction towards cationic PSNPs. This can be an explanation for the dissimilar growth in the sizes of cationic and anionic PSNPs in cell culture medium. Our cytotoxicity data are in line with reports where cationic NPs showed greater cytotoxicity as compared to anionic ones, both for polymer [16] and inorganic NPs [17]. It is still not clear on mechanistic grounds why cationic NPs, in general, are more cytotoxic than the anionic ones. A hypothesis put forward by some groups is that the cationic NPs are attracted by electrostatic interaction towards the (overall) anionic cell membrane and even other lipid based biomembranes, which makes their interactions with biomembranes easier as compared to that of the anionic ones [18]. Similarly, size dependence in cytotoxicity of different NPs (with cytotoxicity inversely related to the sizes of NPs) has also been reported for other types of NPs. A clear understanding of the mode of action underlying this size-effect on the cytotoxicity of NPs is lacking although a hypothesis has been put forward that with decreasing size, NPs start becoming more reactive and hence, can react with a wider variety of biomolecules causing toxicity. It is also possible that with decreasing size, a higher % of the molecules in the NP get expressed on the surface, making smaller NPs relatively more reactive than the larger ones [19].

With reference to several recent reports, a surface charge-driven interaction between PSNP-NH₂ and the cellular membranes may be suggested as the major mode of action underlying the cytotoxicity of cationic PSNP-NH₂ [20-23]. Simultaneously, it was also mentioned

that oxidative stress failed to explain several findings related to the primary mechanism of cytotoxicity for NP. The experiments with DNP and H₂O₂ were therefore intentionally chosen in order to delineate the role of intracellular ROS and oxidative stress in the MOA of PSNP-NH₂. It was observed that the damage done by H₂O₂, which is a known inducer of oxidative stress, can be mitigated effectively by both the vitamins E and C. In contrast, these vitamins failed to significantly (p < 0.05) reduce cytotoxicity caused both by DNP or PSNP-NH₂. In fact, the protection patterns exhibited by the vitamins E and C against the cytotoxicity of DNP and PSNP-NH₂ showed an overall resemblance with each other and were clearly different from those obtained for H2O2. The data obtained by decreasing or augmenting the cellular GSH levels further supported this notion. It is reported that cationic PSNP-NH₂ caused a depletion of cellular GSH levels [24, 25]. Hence, external application of NAC should have largely curbed the cytotoxicity of PSNP-NH₂ if oxidative stress was the main driving event behind cytotoxicity Our results matched with the very small amount of investigations where cellular membranes and mitochondria have been identified as cellular targets for NPs with cationic NPs showing deleterious effects on the $\Delta \Psi_m$ [26]. Given that the mitochondrial membrane is the location of the ETC, any disturbance in mitochondrial membrane physiology can be expected to hamper the ETC, disturbing the creation of the proton motive force, and thus the mitochondrial membrane potential ($\Delta \Psi_m$), ATP generation and the flow of electrons, of which the latter may result in ROS production as a secondary effect. This induction of intracellular ROS can cause oxidative stress. Previously various groups had hypothesized that oxidative stress is the mechanism of cytotoxicity of different NPs including PNPs [27, 28]. However, there is still controversy about what may be the source of the production of intracellular ROS. Some groups have reported that intracellular mitochondria can be a target organ for NPs [29, 30].

Our data showed for the first time, that for cationic PSNP-NH₂, it is not oxidative stress but the membrane interaction with cationic PSNP-NH₂ (with possible ETC disruption) that propels the cytotoxicity pathway, including the induction of ROS as a secondary effect. With evidence pointing towards cationic PSNPs causing an effect on the mitochondrial membrane, it is possible that the mitochondrial calcium could leak out into the cytoplasm. This could result in an increase of cytoplasmic calcium levels and can trigger the apoptotic cascade via cell signaling mechanisms initiating apoptosis [31]. Previously, Xia et al. already reported such increase in cellular calcium levels in RAW 264.7 cells being exposed to cationic PSNPs of 60 nm sizes [27] and our data fit well with their results.

It is shown by different groups before that positive NPs, and not negative NPs, cause holes/pores in suspended lipid bilayers (as an in vitro mimic of lipid bilayer biomembranes like cell or mitochondrial membranes) [32, 33]. This is an important finding as creation of such pores in biomembranes could actually contribute to the cytotoxicity of NPs. However, so far creation of holes in the cell membrane of living cells upon exposure to positive NPs has not been observed. In this article, for the first time, with the help of atomic force microscopy (AFM), we were able to directly image the cell membrane topography in NR8383 cells after exposure to differently charged PSNPs. Why cationic PSNPs caused punctures in the cell membrane is not well understood. It has been argued by some groups, based on computational studies, that due to the electrostatic attraction between cationic PSNPs and the (overall) anionic cell membrane, part of the cell membrane protrudes to first cover the PSNPs and then gradually takes up the PSNPs into the lipid bilayers structure of the cell membrane. In contrast, anionic PSNPs get embedded within the membrane lipid bilayer but do not form holes

like the cationic ones [34, 35]. Although such explanations on the formation of the holes in the cell membrane by cationic PSNPs need further evidence, it can be said based on the current data that this pore formation phenomenon on cell membranes after exposure to cationic PSNPs may point at a possible mode of action for the adverse effects of cationic PSNPs on cellular integrity. The process of endocytosis might also contribute to the disturbance of membrane integrity. Although by calcein assay, an indirect assessment could be made on the compromised cell membrane integrity after exposure to especially cationic PSNPs, direct evidence for disturbance of membrane structure by PSNPs was obtained by AFM. Previously, some groups already advocated the use of AFM in imaging the surface topography of biological samples including cells [36, 37]. In addition, AFM can be performed in normal and water rich biological conditions making the technique compatible for imaging the surface characteristics of cells. The creation of hole-like structures on the cell membrane by cationic PSNPs can add to the mechanistic understanding of the cytotoxicity of PSNPs. The concentrations applied here for the investigations were non-toxic (1 µg/ml) and hence this discontinuity of the cell membrane may not reflect necrosis. By creating holes in the cell membrane, the cationic PSNPs can cause a disruption in the cell membrane permeability and thus disturb normal cell functioning. The cellular uptake of these fluorescent PNPs has been published before [38].

Taken together, the results of the present paper point towards membrane disruption as the primary cause for the cytotoxicity of PSNP-NH₂. This membrane disruption then leads to disruption of the ETC and ROS production, with oxidative stress as a secondary effect.

Experimental

PSNPs. The fluorescent PSNP-NH₂ and PSNP-COOH of 50 and 100 nm sizes were purchased from Sigma Aldrich Chemie BV. The sizes (while dispersed in water or DMEM/F12-K medium containing 10 % FCS after 4 h) and PDI of the PSNPs were checked by DLS and imaged by AFM through contact tapping mode on mica surface. The surface potentials of the PSNPs were measured by ζ-potential measurements (*Malvern Zetasizer*). The stabilities of these PSNPs suspensions both in water and DMEM/F12-K medium (with FCS) were checked by DLS.

NR8383 cells. Rat alveolar macrophage (NR8383) cells were obtained from ATCC (Manassas,VA). The NR8383 cells were cultured in 150 cm² cell culture flasks with 5 ml 1 -K culture medium (Gibco 21127) supplemented with 10 % (v/v) heat inactivated fetal calf serum (FCS) in a humidified atmosphere containing 5 % CO₂ at 37 °C.

MTT assav. An NR8383 cell suspension was centrifuged at 140 g for 5 min before re-suspending the cell pellet in F12-K medium followed by counting and adjusting the cellular concentration to $2 \times$ 10° cells/ml. The cells were then seeded in a 96-well plate (50 μl/well) and the plate was kept in a 5 % CO₂ incubator at 37 °C for 24 h. Subsequently, 50 µl of serial dilutions of different PSNPs in F12-K medium were added to the cells to obtain the required final concentrations (0-100 µg/ml). This was followed by incubation for another 4 h after which 5 µl of MTT solution in PBS (5 mg/ml) was added to each well and the plate was incubated for another 4 h. Then 100 µl of pure dimethysulfoxide (DMSO) was added to each well to dissolve the formazan crystals. The absorption reading of each well was measured at 562 nm in a 96-well plate reader and the background absorption reading at 612 nm was subtracted. Mitochondrial metabolic activity for each concentration of PSNPs was expressed as % of the negative control (0 μg/ml) reading.

Medium without PSNPs and medium with Triton-X (0.1%) were used as negative and positive controls, respectively. A 4 h exposure time was selected as the cationic PSNPs already showed significant cytotoxicity within the tested concentration range in 4 h.

MTT assay to investigate protective effects of vitamins E and C against the cytotoxicity of DNP, PSNP-NH2 and H2O2. The NR8383 cells were pre-incubated with 100 μM vitamin E or 1 mM vitamin C and then exposed to DNP (0-75 mM), PSNP-NH2 and H2O2 (0-1 mM). After 4 h, 5 μl of MTT solution in PBS (5 mg/ml) was added to each well and the plate was incubated for another 4 h. The purple formazan crystals were then dissolved in 100 μl /well DMSO and measured at 562 nm and 612 nm (as background). The reading of each well was expressed as % of the negative control (0 $\mu g/ml)$.

MTT assay with pre-exposure to buthionine sulphoximine (BSO) or N-acetyl cysteine (NAC) to investigate the protective role of cellular GSH. The NR8383 cells were plated with medium containing BSO (100 μ M/well) to reduce cellular GSH levels (Buchmiller-Rouiller et al., 1995) or NAC (10 mM/well) to increase cellular antioxidant levels for 24 h and 1 h, respectively, before being exposed to 2,4-DNP, PSNP-NH₂ and H₂O₂ and performing the MTT assay as described above.

Phagocytic index (PI) measurement. An NR8383 cell suspension (2 \times 10⁵ cells/ml) was seeded in a 96-well plate (50 µl/well) in F12-K medium, followed by addition of 50 μ l/well of serial dilutions of different PSNP in F12-K medium to obtain the required final test concentrations of PSNP (0-100 µg/ml) on the time of incubation. Plain F12-K medium without PSNPs and medium containing 100 μM CuSO₄ were used as negative and positive controls, respectively. After 4 h, the cells were exposed to yellow green fluorescent latex beads (1 µm size) at a ratio of beads to cells in each well of 50:1. After 4 h of incubation (total exposure time 8 h) counting samples were taken from the wells and viewed first under a fluorescent microscope to visualize the fluorescent beads, followed by bright field view to visualize the cells. Samples were also taken out of each well to assess the cell viability by trypan blue exclusion test. The trypan blue exclusion test was performed by adding trypan blue dye with cell suspension (1:1) before measuring the amount of viable cells under light microscope. The phagocytic index was determined by calculating the average number of fluorescent beads phagocytosed per viable cell and expressed as % of the negative control (0 µg/ml).

Measurement of intracellular ROS by the DCFH-DA assay. The NR8383 cell suspension was adjusted to 2×10^5 cells/ml and seeded in a 96-well plate (50 µl/well) in F12-K medium. 50 µl/well of serial dilutions of different PSNPs in F12-K medium were added to obtain the required final test concentrations of PSNPs (0-100 µg/ml). A concentration of 10 mM H₂O₂ was used as positive control and F12-K medium without PSNPs as negative control. Immediately after exposure to the PSNPs, 5 µl of a 20 mM solution of DCFH-DA (in DMSO) was added to each well and the plates were incubated for 4 h in a 5 % CO₂ atmosphere at 37 °C making the total exposure time of 4 h. The fluorescence was then measured on a spectrofluorometer at λ_{ex} =485 nm and λ_{em} =538 nm. The fluorescence induction factor for each concentration of PSNPs was calculated by dividing the reading of each well by the average reading of the negative control (0 μg/ml) and expressed as % of the control. Control experiments were performed by incubating the PSNPs at their test concentrations with DCFH-DA in the absence of cells to check the possibility of a positive fluorescence reading caused by reaction of DCFH-DA with PSNPs alone.

Measurement of intracellular ROS production by DCFH-DA assay in presence of DNP and H_2O_2 . 50 μl of an NR8383 cell suspension $(2 \times 10^5 \text{ cells/ml})$ in F12-K medium, containing 100 μ M vitamin E or 1 mM vitamin C, or no added antioxidants (negative control), was seeded in each well of a 96-well plate and incubated for 24 h. The vitamin C was added after 22 h of incubation, making the preincubation time for vitamin E and vitamin C, 24 h and 2 h, respectively. Different DNP, cationic PSNP-NH₂ and H₂O₂ dilutions were then added to obtain the test dilutions (0-75 mM for DNP, 0-100 μg/ml for PSNP-NH₂ and 0-1 mM for H₂O₂). 5 μl of a 20 mM DCFH-DA solution in DMSO was added to each well after 4 h and incubated for further 4 h (total exposure time of 8 h). The fluorescence was measured ($\lambda_{ex} = 485$ nm, $\lambda_{em} = 538$ nm) and the induction factors for each concentration of PSNP were obtained by expressing the readings of the wells as % of the negative control. Control experiments were performed by incubating the DNP, PSNP-NH₂ and H₂O₂ at their test concentrations with DCFH-DA to exclude any interference with the fluorescence.

Measurement of mitochondrial permeability transition pore opening. The NR8383 cell suspension was adjusted to 2×10^5 cells/ml and seeded in a 96-well plate (50 µl/well) in F12-K medium. A 50 µl/well of serial dilutions of different PSNPs in F12-K medium were added to obtain the required final test concentrations of PSNPs (0-100 µg/ml) on time of incubation. The mitochondrial membrane potential $(\Delta \Psi_m)$ was measured by a commercially available kit from Invitrogen (MitoProbeTM Transition Pore Assay Kit; catalogue no. M34153) and expressed as % of the negative control (0 µg/ml). The assay detects mitochondrial permeability transition pore opening as an indirect measurement for effects on mitochondrial membrane potential $(\Delta \Psi_m)$. In this kit, a nonfluorescent acetomethoxy derivative of calcein (calcein-AM) is administered which upon passively entering the cells and accumulating in the cytoplasm as well as in the mitochondria, gets cleaved by intracellular esterases to release ionic and fluorescent calcein. Due to strong ionic nature, this released calcein cannot cross the mitochondrial/plasma membrane and thus can be detected by fluorescence inside the cell. In this kit, the fluorescence coming from the calcein in cytoplasm is quenched with the addition of cobalt chloride (CoCl₂), so that only the fluorescence coming from the mitochondria can be detected. A 100 µM solution of ionomycin in DMSO (supplied with the kit) and F12-K medium without PSNPs were used as positive and negative controls, respectively.

Measurement of cellular ATP content. The NR8383 cell suspension was adjusted to 2×10^5 cells/ml and seeded in a 96-well plate (50 μl/well) in F12-K medium. A 50 μl/well of serial dilutions of different PSNPs in F12-K medium were added to obtain the required final test concentrations of PSNPs (0-100 μg/ml) on time of incubation. The cellular ATP was then measured by a commercially available kit from Invitrogen (catalogue no. FLASC) and expressed as % of negative control (0 μg/ml). In this kit, the ATP concentration is measured by the interaction between cellular ATP and luciferin in the presence of Mg^{2+} , producing adenyl luciferin which upon oxidation produces light. A 75 mM solution of DNP (in DMSO) and F12-K medium without PSNPs were used as positive (i.e. with known effects of ATP depletion) and negative controls, respectively.

Measurement of cytoplasmic calcium. The NR8383 cell suspension was adjusted to 2×10^5 cells/ml and seeded in a 96-well plate (50 µl/well) in F12-K medium. A 50 µl/well of serial dilutions of different PSNPs in F12-K medium were added to obtain the required final test concentrations of PSNPs (0-100 µg/ml) on time of incubation. The cytoplasmic free calcium was then measured by a

commercially available kit from Invitrogen (Fluo-4 Direct Calcium Assay Kit; catalogue no. F10472) and expressed as % of negative control (0 μ g/ml). F12-K medium without PSNPs was used as negative control.

CLSM. The NR8383 cell suspension was adjusted to 2×10^5 cells/ml and seeded in a 96-well plate (50 μl/well) in F12-K medium. A 50 μl/well of serial dilutions of different PSNPs in F12-K medium were added to obtain a required non-toxic concentration of PSNPs (1 μg/ml). After 4 h of exposure, samples were taken out and put on a glass slide before examining them under oil immersion microscope (100×) and then visualizing the cells under a Zeiss Axiovert 200M-Exciter confocal laser scanning microscope (λ_{ex} =543 nm; λ_{em} =620 nm). Different samples were measured with the same set up of similar resolution, aperture width and image gain. The relative cellular uptake of 100 nm cationic and anionic PSNPs were expressed as an arithmetic mean of data obtained from 20 individual cells from at least five different experiments (n=5). Cells exposed to only F12-K medium were used as control.

Calcein-AM assay. The NR8383 cell suspension $(2 \times 10^5 \text{ cells/ml})$ was exposed to 1 μ M calcein-AM for 1 h. The cells were then washed with PBS by repeated centrifugation and then exposed to non-toxic concentration of PSNPs (1 μ g/ml). After 4 h of exposure, samples were taken out and put on a glass slide before examining it under oil immersion microscope (100×) and then visualizing the cells under a Zeiss confocal laser scanning microscope (λ_{ex} =488 nm; λ_{em} =538 nm). Different samples were measured (n=5) with the same set up of similar resolution, aperture width and image gain. Cells exposed to only calcein were used as control.

AFM. After exposure to PSNPs (1 μg/ml) for 4 h, cellular samples (from different cultures) were taken from a NR8383 cell suspension (2×10^5 cells/ml) before placing it on mica slides. The samples were then checked by AFM in contact tapping mode with silicon nitride tip of an average diameter of 30 nm. Each time an area of 2×2 μm was scanned and then the height tracings as well as the three dimensional depictions of the surfaces were made. The roughness (r) was calculated for the area scanned by the computer software. In total, five cells (n=5) were measured for each type of PSNPs.

Statistical analysis. Each experiment was repeated three times (n=3) unless mentioned otherwise and the results are shown as mean \pm standard error of mean after analysis and plotting by Origin Pro (version 8.0) software. Data points were taken as statistically significantly different if p < 0.05 compared to the negative control (unless stated otherwise) and marked with an asterisk (*) symbol. The EC50 values were determined by the Origin Pro software after setting up a trendline to the curves.

Acknowledgements

The authors would like to thank graduate school VLAG, QNano and Wageningen UR strategic research programme Bionanotechnology (IPOP/BioNano) for funding. Dr. Marcel Giesbers is thanked for help with the AFM measurements.

Notes and references

^aLaboratory of Organic Chemistry, Dreijenplein 8, Wageningen University, 6703 HB Wageningen, The Netherlands

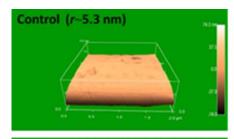
^bDivision of Toxicology, Tuinlaan 5, Wageningen University, 6703 HE Wageningen, The Netherlands

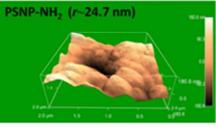
^cLaboratory of Physical Chemistry and Colloid Science, Dreijenplein 6, Wageningen University, 6703 HB Wageningen, The Netherlands

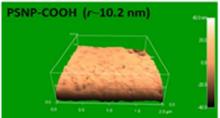
References

- M. J. D. Clift, S. Bhattacharjee, D. M. Brown and V. Stone, *Toxicol Lett*, 2010, **198**, 358-365.
- K. Donaldson and V. Stone, in *Nanotechnology:* Consequences for Human Healthand the Environment, The Royal Society of Chemistry, 2007, vol. 24.
- S. Bhattacharjee, L. H. J. de Haan, N. M. Evers, X. Jiang, A. T. M. Marcelis, H. Zuilhof, I. M. C. M. Rietjens and G. M. Alink, *Part Fibre Toxicol*, 2010, 7, 25.
- K. Greish, G. Thiagarajan, H. Herd, R. Price, H. Bauer, D. Hubbard, A. Burckle, S. Sadekar, T. Yu, A. Anwar, A. Ray and H. Ghandehari, *Nanotoxicology*, 2012, 6, 713-723.
- K. Kawata, M. Osawa and S. Okabe, Environ Sci Technol, 2009, 43, 6046-6051.
- S. M. Hussain, K. L. Hess, J. M. Gearhart, K. T. Geiss and J. J. Schlager, *Toxicol in Vitro*, 2005, 19, 975-983.
- C. Monteiller, L. Tran, W. MacNee, S. Faux, A. Jones, B. Miller and K. Donaldson, *Occup Environ Med*, 2007, 64, 609-615.
- K. Yu, C. Grabinski, A. Schrand, R. Murdock, W. Wang, B. Gu, J. Schlager and S. Hussain, *J Nanopart Res*, 2009, 11, 15-24
- M. Y. Wani, M. A. Hashim, F. Nabi and M. A. Malik, Adv Phys Chem, 2011, 450912-450927.
- H. Zhang, T. Xia, H. Meng, M. Xue, S. George, Z. Ji, X. Wang, R. Liu, M. Wang, B. France, R. Rallo, R. Damoiseaux, Y. Cohen, K. A. Bradley, J. I. Zink and A. E. Nel, ACS Nano, 2011, 5, 2756-2769.
- 11. A. E. Vercesi, R. Docampo and S. N. J. Moreno, *Mol Biochem Parasitol*, 1992, **56**, 251-257.
- A. Verma, O. Uzun, Y. Hu, Y. Hu, H.-S. Han, N. Watson, S. Chen, D. J. Irvine and F. Stellacci, *Nat Mater*, 2008, 7, 588-595.
- T. Cedervall, I. Lynch, S. Lindman, T. Berggård, E. Thulin, H. Nilsson, K. A. Dawson and S. Linse, *Proc Natl Acad Sci*, 2007, 104, 2050-2055.
- I. Lynch, T. Cedervall, M. Lundqvist, C. Cabaleiro-Lago, S. Linse and K. A. Dawson, Adv Colloid Interf Sci, 2007, 134– 135, 167-174.
- A. Alkilany and C. Murphy, J Nanopart Res, 2010, 12, 2313-2333.
- 16. E. Frohlich, Int J Nanomed, 2012, 7, 5577-5591.
- 17. C. Hoskins, A. Cuschieri and L. Wang, *J Nanobiotechnol*, 2012, **10**, 15.
- M. Schulz, A. Olubummo and W. H. Binder, *Soft Matter*, 2012, 8, 4849-4864.
- 19. M. Auffan, J. Rose, J. Y. Bottero, G. V. Lowry, J. P. Jolivet and M. R. Wiesner, *Nat Nanotechnol*, 2009, 4, 634-641.
- Y. Liu, W. Li, F. Lao, Y. Liu, L. Wang, R. Bai, Y. Zhao and C. Chen, *Biomaterials*, 2011, 32, 8291-8303.
- H. Fritz, M. Maier and E. Bayer, J Colloid Interf Sci, 1997, 195, 272-288.
- H. Chen, Q. C. Zou, H. Yu, M. Peng, G. W. Song, J. Z. Zhang, S. G. Chai, Y. Zhang and C. E. Yan, *Microchim Acta*, 2010, **168**, 331-340.
- S. Bhattacharjee, I. M. C. M. Rietjens, M. P Singh, T. M. Atkins, T. K. Purkait, Z. Xu, S. Regli, A. Shukaliak, R. J. Clark, B. S. Mitchell, G. M. Alink, A. T. M. Marcelis, M. J. Fink, J. G. C. Veinot, S. M. Kauzlarich and H. Zuilhof, *Nanoscale*, 2013, 5, 4870-4883.
- 24. W. H. de Jong and P. J. A. Borm, *Int J Nanomed*, 2008, 3, 133-149.
- 25. P. H. Hoet, I. Bruske-Hohlfeld and O. V. Salata, *J. Nanobiotechnol*, 2004, **2**, 12.

- T. Xia, M. Kovochich, M. Liong, J. I. Zink and A. E. Nel, ACS Nano, 2008, 2, 85-96.
- T. Xia, M. Kovochich, J. Brant, M. Hotze, J. Sempf, T. Oberley, C. Sioutas, J. I. Yeh, M. R. Wiesner and A. E. Nel, *Nano Lett*, 2006, 6, 1794-1807.
- T. Mocan, S. Clichici, L. Agoşton-Coldea, L. Mocan, Ş. Şimon, I. Ilie, A. Biriş and A. Mureşan, *Acta Physiologica Hungarica*, 2010, 97, 247-255.
- V. Salnikov, Y. O. Lukyánenko, C. A. Frederick, W. J. Lederer and V. Lukyánenko, *Biophys J*, 2007, 92, 1058-1071.
- 30. Ö. F. Karataş, E. Sezgin, Ö. Aydın and M. Çulha, *Colloid Surf B: Biointerfaces*, 2009, **71**, 315-318.
- 31. M. R. Duchen, J Physiol, 2000, **529**, 57-68.
- 32. J. Chen, J. A. Hessler, K. Putchakayala, B. K. Panama, D. P. Khan, S. Hong, D. G. Mullen, S. C. DiMaggio, A. Som, G. N. Tew, A. N. Lopatin, J. R. Baker, M. M. B. Holl and B. G. Orr, *J Phys Chem B*, 2009, 113, 11179-11185.
- P. R. Leroueil, S. A. Berry, K. Duthie, G. Han, V. M. Rotello, D. Q. McNerny, J. R. Baker, B. G. Orr and M. M. Banaszak Holl, *Nano Lett*, 2008, 8, 420-424.
- 34. Y. Li, X. Chen and N. Gu, *J Phys Chem B*, 2008, **112**, 16647-16653.
- 35. H. Lee and R. G. Larson, *J Phys Chem B*, 2008, **112**, 12279-12285.
- A. Raman, S. Trigueros, A. Cartagena, A. P. Z. Stevenson, M. Susilo, E. Nauman and S. A. Contera, *Nat Nanotechnol*, 2011, 6, 809-814.
- S. Nawaz, P. Sánchez, K. Bodensiek, S. Li, M. Simons and I. A. T. Schaap, *Plos One*, 2012, 7, e45297
- 38. S. Bhattacharjee, D. Ershov, J. van der Gucht, I. M. C. M. Rietjens, H. Zuilhof and A. T. M. Marcelis, *Nanotoxicology*, 2013, **7**, 71-84.







Positively charged polystyrene nanoparticles show higher cytotoxicity with oxidative stress and create holes/pores in the cell membranes.

Graphical Abstract 209x157mm (96 x 96 DPI)