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Probing the substrate specificity of Trypanosoma brucei GIcNAc-PI de- $N$-acetylase with synthetic substrate analogues.

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A series of synthetic analogues of 1-D-(2-amino-2-deoxy- $\alpha$-D-glucopyranosyl)-myo-inositol 1-(1,2-di-O-hexadecanoyl-sn-glycerol 3-phosphate), consisting of 7 variants of either the D-myo-inositol, D-Glc $p \mathrm{~N}$ or the phospholipid components, were prepared and following $N$-acylation were tested as substrates and inhibitors of GlcNAc-PI de- $N$-acetylase, a genetically validated drug target in Trypanosoma brucei, the causative agent of human African trypanosomiasis.











#### Abstract

A series of synthetic analogues of 1-D-(2-amino-2-deoxy- $\alpha$-D-glucopyranosyl)-myo-inositol 1-(1,2-di-O-hexadecanoyl-sn-glycerol 3-phosphate), consisting of 7 variants of either the D-myoinositol, D-GlcpN or the phospholipid components, were prepared and tested as substrates and inhibitors of GlcNAc-PI de- $N$-acetylase, a genetically validated drug target enzyme responsible for the second step in the glycosylphosphatidylinositol (GPI) biosynthetic pathway of Trypanosoma brucei. The D-myo-inositol in the physiological substrate was successfully replaced by cyclohexanediol and is still a substrate for $T$. brucei GlcNAc-PI de- $N$-acetylase. However, this compound became sensitive to the stereochemistry of the glycoside linkage (the $\beta$-anomer was neither substrate or inhibitor) and the structure of the lipid moiety (the hexadecyl derivatives were inhibitors). Chemistry was successfully developed to replace the phosphate with a sulphonamide, but the compound was neither a substrate or an inhibitor, confirming the importance of the phosphate for molecular recognition. We also replaced the glucosamine by an acyclic analogue, but this also was inactive, both as a substrate and inhibitor. These findings add significantly to our understanding of substrate and inhibitor binding to the GlcNAc-PI de- $N$-acetylase enzyme and will have a bearing on the design of future inhibitors.


## Introduction

The enzymes of the glycosylphosphatidylinositol (GPI) biosynthetic pathway are located in the endoplasmic reticulum, contain between one and thirteen predicted trans-membrane domains and are mostly present as components of multi-subunit complexes. ${ }^{1}$ No high-resolution structural data are available on any of these enzymes and our research group has been probing the specificities of several of the enzymes in the GPI pathway of the protozoan parasite Trypanosoma brucei, the causative agent of African sleeping sickness in humans and the related disease Nagana in cattle, using synthetic substrate analogues in vitro. ${ }^{2-8}$ One of the key enzymes of interest is an amidase, the GlcNAc-PI de- $N$-acetylase (EC $\underline{\text { 3.5.1.89 }}$ ) that de- $N$-acetylates 1-D-(2-acetamido-2-deoxy- $\alpha$-D-glucopyranosyl)-myo-inositol 1-(1,2-di-O-hexadecanoyl-sn-glycerol 3-phosphate) (1, $\alpha$-D-GlcpNAc-PI) to 1-D-(2-amino-2-deoxy- $\alpha$-D-glucopyranosyl)-myo-inositol 1-(1,2-di-O-hexadecanoyl-sn-glycerol 3-phosphate) (2, $\alpha$-D-Glc $p \mathrm{~N}-\mathrm{PI}$ ), Figure 1.





Figure 1 Some previously prepared GPI analogues

This enzyme catalyses the second step in the T. brucei GPI biosynthetic pathway, which is a prerequisite for all subsequent steps in the pathway. ${ }^{9}$ In earlier studies, we showed that $T$. brucei GlcNAc-PI de- $N$-acetylase is a zinc-dependent metalloenzyme ${ }^{10}$ and demonstrated, by construction of a condition-null mutant cell line, that it is essential for the bloodstream form of the parasite and, therefore, a genetically validated drug target. ${ }^{11}$ Previous studies with other substrate analogues showed that the phosphate, $2^{\prime}-\mathrm{NHAc}$ and $3^{\prime}-\mathrm{OH}$ groups of the natural substrate $\alpha$-D-Glcp $\mathrm{NAc}-\mathrm{PI}$ (1) are critical for recognition by the $T$. brucei GlcNAc-PI de- $N$-acetylase. ${ }^{2-4}$ In contrast, the
diacylglycerol moiety is not strictly required and may be efficiently replaced with an octadecyl chain, ${ }^{4}$ as shown in analogues $\mathbf{3}$ and 4. In the case of the T. brucei enzyme, we had hypothesised that one or more of the inositol 2, 3, 4 and $5-\mathrm{OH}$ groups is/are not required. ${ }^{2-4}$ We came to this hypothesis from the ability of the enzyme to recognise and process both $\alpha$-D-GlcpNAc-[L]-PI (5) and $\beta$-D-GlcpNAc-PI (6). Molecular dynamics simulations, showed that $\alpha$ and $\beta$ anomers can adopt conformations in which the phosphate, the $2^{\prime}$-amide and the $3^{\prime}$ '-OH overlay. Given the $2^{\prime}$ 'amide is where the reaction occurs, and evidence suggests that the $3^{\prime}-\mathrm{OH}$ and phosphate are important for recognition of the substrate with the enzyme, these conformations are likely to be the active enzyme-bound conformations. In these conformations the inositol 2, 3, 4 and 5-hydroxyls are in different orientations for the two $\alpha$ and $\beta$ anomers, implying the hydroxyls are not critical for interaction with the enzyme. Further evidence was obtained from a compound in which the inositol 2-hydroxyl was alkylated. This was also a substrate for the T. brucei enzyme. One of the goals of the study described in this paper was to investigate the hypothesis that the inositol 2, 3, 4 and 5-OH groups are not required.

Bearing in mind these key structural features, we have synthesised a variety of analogues to further probe the requirements for substrate recognition by the $T$. brucei GlcNAc-PI de- $N$-acetylase and, specifically, to test:

1. The hypothesis that the inositol $2,3,4$ and $5-\mathrm{OH}$ groups are not required for enzyme recognition, a series of pseudodisaccharides ( $\mathbf{7} \mathbf{- 1 0}$, Figure 2) containing a cyclohexanediol moiety in place of the inositol aglycone were prepared.
2. Whether the phosphate group can be replaced by more cell-permeable sulphonamide isosteres, ${ }^{12}$ compounds $\mathbf{1 1}$ and $\mathbf{1 2}$ (Figure 2) were prepared.
3. Whether, given the essentiality of the $2^{\prime}-\mathrm{NHR}$ and $3^{\prime}-\mathrm{OH}$ groups but non-essentiality of the $4^{\prime}-$ and $6^{\prime}-\mathrm{OH}$ groups for substrate recognition, the glucosamine residue might be simplified to a simple acyclic structure, as in compound 13.






11


12


13

Figure 2 Target molecules

The $N$-acetylated derivatives of the above analogues required for biological studies with the de- N -acetylase were prepared from the corresponding amines by standard procedures. ${ }^{13}$ All the analogues were examined for their recognition and processing by the T. brucei GlcNAc-PI de- N acetylase.

## Results and discussion

## Synthesis of analogues 7 - 13

The synthesis of the required $\alpha$ and $\beta$-glucosaminyl $\left(1^{\prime} \rightarrow 1\right)$ cyclohexandiol building blocks 16 and 17, respectively, began by reacting the known ${ }^{14}$ trichloroacetimidate 14 and the commercially available $1 R, 2 R$-trans-cyclohexandiol 15 , Scheme 1 , with a catalytic amount of trimethylsilyl trifluoromethanesulfonate (TMSOTf). The separation of anomers was achievable at this stage and these anomers were vital in providing the glucosamine-phosphodiester target analogues discussed herein. For the sake of brevity, we have chosen to describe in the main text the formation of the $\alpha$-anomers, while details for the corresponding $\beta$-anomers appears in the Supplementary Data Section. Therefore, the pseudodisaccharide 16 was coupled to the hydrogen
phosphonate $\mathbf{1 8}^{15}$, and the ensuing mixture of diastereoisomeric phosphonic diesters, was oxidised with iodine in pyridine-water ${ }^{16}$ to give the corresponding phosphodiester 19. The diester was transformed into the triol $\mathbf{2 0}$ after conventional $O$-deacylation of the latter compound with 0.05 M methanolic NaOMe in $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}$.


Scheme 1 Synthesis of 7

Our first attempt at hydrogenolysis of the triethylammonium (TEA) salt $\mathbf{2 0}$ over $\mathrm{Pd}(\mathrm{OH})_{2} / \mathrm{C}$ gave, surprisingly, the alkyl alcohol secondary amine 21. Initially, the azide of $\mathbf{2 0}$ was reduced to the primary amine, however, if there are peroxides present in the THF then the formation of THF hydroperoxide is, apparently, possible which could then lead to an amine-THF coupling via a free-radical-based mechanism. ${ }^{17}$ This intermediate is susceptible to a Pd-mediated THF ring opening reaction that gives an imine which is then further hydrogenated to an aminobutanol. ${ }^{17}$ Consequently, after purchasing a fresh bottle of anhydrous stabilised THF, the second hydrogenolysis attempt at $\mathbf{2 0} \rightarrow \mathbf{7}$ proceeded without incident.

The preparation of the dipalmitoyl glycerol pseudodisaccharide $\mathbf{9}$ was accomplished from the triacetate 16, Scheme 2. However, the acetate protecting groups in 16 are unsuitable because if they were left in place and removed by base at the final step of the synthesis, then those requisite
esters of the lipid fragment would likewise be saponified. Therefore, the acetates of $\mathbf{1 6}$ needed to be swapped to a more appropriate protecting group but first, the temporary tert-butyldimethylsilyl (TBDMS) protection of the $2-\mathrm{OH}, \mathbf{1 6} \boldsymbol{\rightarrow 2 2}$, was performed and then followed by conventional $O$ deacylation, as previously described for $\mathbf{1 9} \rightarrow \mathbf{2 0}$, furnished the triol 23.


Scheme 2 Synthesis of 9

The benzyl group was chosen as the $3^{\prime}, 4^{\prime}$ and $6^{\prime}-\mathrm{OH}$ protecting group because, from our past experiences synthesising GPI analogues, the benzyl group has been a very reliable protecting group via ease of installation and removal. Thus, the triol $\mathbf{2 3}$ was benzylated with benzyl bromide in the presence of NaH , as the base, to afford compound $\mathbf{2 4}$. We next turned our attention towards the reduction of the azide in 24 and because of the issues with $\mathrm{Pd}(\mathrm{OH})_{2}$ catalysed reduction in the presence of peroxidic THF discussed earlier, we chose to reduce the azide via the Staudinger reaction ${ }^{18}$ to give the amine $\mathbf{2 5}$ which was subsequently tert-butyl carbamate (Boc) protected to furnish 26. Desilylation of $\mathbf{2 6}$ using HF-pyridine conditions afforded the alcohol 27. The known hydrogen phosphonate $28^{19}$ was coupled, as already described, to the $2-\mathrm{OH}$ of $\mathbf{2 7}$ which, after oxidation, was isolated and characterised as the TEA salt 29. Hydrogenolysis of 29 over $\mathrm{Pd}(\mathrm{OH})_{2} / \mathrm{C}$ gave the Boc protected derivative 30 and subsequent cleavage of the Boc group produced the deprotected target analogue 9 .

Sulphonamides are potential isosteres for the phosphate group; they have the same tetrahedral shape and polar oxygen atoms. The synthesis of the sulphonamides $\mathbf{1 1}$ and $\mathbf{1 2}$, Scheme 3, was accomplished by reacting commercially available $1 R, 2 R-1$-amino-2-benzyloxycyclohexane 31 with 1-octadecanesulfonyl chloride in the presence of triethylamine and $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ to give the benzyl derivative 32. The benzyl group was removed by hydrogenolysis over $\mathrm{Pd}(\mathrm{OH})_{2} / \mathrm{C}$ to furnish the alcohol 33. Coupling of $\mathbf{3 3}$ with the known trichloroacetimidate $\mathbf{3 4}{ }^{20}$ resulted in an inseparable mixture of the $\alpha, \beta$-anomers 35 in the ratio of $\sim 1: 1$, as determined by ${ }^{1} \mathrm{H}$ NMR spectroscopy. Finally, hydrogenolysis of the aforementioned anomers $\mathbf{3 5}$ over $\mathrm{Pd}(\mathrm{OH})_{2} / \mathrm{C}$ and subsequent silica gel column chromatography $\left(9: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}\right)$ gave first the $\beta$-anomer 11 and then the $\alpha$ anomer 12.


Scheme 3 Synthesis of sulphonamides 11 and 12

We were also interested in seeing if we could replace the glucose ring with an acyclic moiety. From our knowledge of the SAR, retaining the 2-amino and 3-hydroxy groups are important for activity. The synthesis of the amino-phosphate 13, Scheme 4 , began by epoxide ringopening of cyclohexene oxide 36 with $p$-methoxybenzyl alcohol ( PMBOH ), using $\mathrm{Cu}\left(\mathrm{BF}_{4}\right)_{2} \cdot n \mathrm{H}_{2} \mathrm{O}$ as a catalyst ${ }^{21}$ to give the known racemic PMB monoprotected cyclohexanediol ${ }^{22}$ contaminated with unreacted PMBOH. Chromatographic separation of this PMB cyclohexyl derivative from the excess of PMBOH was not achievable, in our hands, and so the entire PMB reaction residue was
acetylated with acetic anhydride in the presence of pyridine and a catalytic amount of 4(dimethylamino)pyridine DMAP to furnish the acetate 37, which was easily separated from the acetate of $p$-methoxybenzyl alcohol by silica gel column chromatography. ${ }^{23}$ After deacetylation, the resulting alcohol ${ }^{22}$ residue was alkylated using sodium hydride and allyl bromide to give the allyl derivative 38. The epoxide 39 was prepared upon reacting 38 with 3 -chloroperbenzoic acid ( $m \mathrm{CPBA}$ ), and the subsequent hydrolysis of epoxide 39 with $\mathrm{DMSO}, \mathrm{H}_{2} \mathrm{O}$ and a catalytic amount of $\mathrm{KOH},{ }^{24}$ worked smoothly to furnish diol $\mathbf{4 0}$. The primary alcohol of $\mathbf{4 0}$ was protected using tertbutyl(chloro)diphenylsilane (TBDPSCl) and DMAP to give compound 41. The azido group of $\mathbf{4 3}$ was satisfactorily installed ( $51 \%$ yield over two steps) via the mesylate $\mathbf{4 2}$ obtained by reacting the secondary alcohol 41 with methanesulfonyl chloride in the presence of pyridine, followed by treatment of $\mathbf{4 2}$ with sodium azide under forcing conditions. The crude mesylate $\mathbf{4 2}$ was used directly in the displacement reaction but a small portion of 42 was purified for a full characterisation of this intermediate. The p-methoxybenzyl protecting group of $\mathbf{4 3}$ was removed with mild acid to give alcohol 44 and then it was phosphorylated, as previously described, to give the phosphoric diester 45 . Thereafter, the removal of the silyl protecting group of $\mathbf{4 5}$ with 1.0 M tetrabutylammonium fluoride (TBAF) in THF proceeded smoothly to give 46 which, after hydrogenolysis over $\mathrm{Pd}(\mathrm{OH})_{2} / \mathrm{C}$, provided the TEA salt 13 .


Scheme 4 Synthesis of racemic 13

## Biological results

## Substrate analogues:

The ability of the T. brucei GlcNAc-PI de- $N$-acetylase to recognise and process the synthetic pseudodisaccharides $\mathbf{7 - 1 3}$ was tested in $T$. brucei cell-free system using an LC-MS/MS assay. ${ }^{8,10}$ The $N$-acetylated analogues of these compounds were prepared as previously described to give compounds $\mathbf{4 7}$ - 53 (Figure 3). ${ }^{13}$






51


52


53

Figure 3 N -Acetylated analogues
Using LC-MS/MS, multiple reaction monitoring of characteristic transitions for the N acetylated and corresponding amine form of each compound was used to directly measure the rate of conversion of the $N$-acetylated compound to the free amine (Table 1). As no suitable transition was identified for the amine form of 13, the enzymatic turnover was accessed by reacting any free amine formed with $d_{6}-\mathrm{Ac}_{2} \mathrm{O}$, and measuring the formation of the $d_{3}-N$-acetylated form by LCMS/MS.

Over the range of enzyme concentration that gave a linear turnover for $\alpha$-D-GlcpNAc-PI (1) the substrate analogue $\alpha$-D-GlcpNAc-IPC ${ }_{18}$ (3) was de- $N$-acetylated at $450 \%$ the rate of $\alpha$-D-

GlcpNAc-PI (1). The increased turnover is most likely due to improved accessibility of the compound, conferred by the single alkyl chain, to the membranes that contain the de- N -acetylase enzyme in the cell-free system. Of the synthetic pseudodisaccharides ( $\mathbf{4 7} \mathbf{- 5 0}$ ) tested, only the $\alpha$ anomer of the dipalmitoylated compound 49 showed any appreciable turnover at $22 \%$ the rate of $\alpha$ -D-GlcpNAc-PI (1).

Table 1 Recognition of synthetic analogues by $T$. brucei GlcNAc-PI de-N-acetylase.

| Compound | $\mathrm{m} / \mathrm{z}$ | $\mathrm{m} / \mathrm{z}$ | Fragment | Turnover/ | Relative |
| :--- | :--- | :--- | :--- | :--- | :--- |
|  | transition for | transition | assignment | pMol/ $10^{6}$ | turnover $^{\mathrm{a}}$ |
|  | $\mathrm{NH}_{2}$ | for NHAc |  | cells equiv. |  |
| $\mathbf{1}$ | $1012>241$ | $972>241$ | $\mathrm{C}_{6} \mathrm{H}_{10} \mathrm{O}_{8} \mathrm{P}$ | $6.1 \pm 0.9$ | $100 \%$ |
| $\mathbf{3}$ | $673>223$ | $715>223$ | $\mathrm{C}_{6} \mathrm{H}_{8} \mathrm{O}_{7} \mathrm{P}$ | $27.0 \pm 6.0$ | $450 \%$ |
| $\mathbf{4 7}$ | $608>100$ | $650>100$ | $\mathrm{C}_{6} \mathrm{H}_{12} \mathrm{O}$ | ND | - |
| $\mathbf{4 8}$ | $608>100$ | $650>100$ | $\mathrm{C}_{6} \mathrm{H}_{12} \mathrm{O}$ | ND | - |
| $\mathbf{4 9}$ | $906>255$ | $948>255$ | $\mathrm{O}_{2} \mathrm{CC}_{15} \mathrm{H}_{31}$ | $1.3 \pm 0.3$ | $22 \%$ |
| $\mathbf{5 0}$ | $906>255$ | $948>255$ | $\mathrm{O}_{2} \mathrm{CC}_{15} \mathrm{H}_{31}$ | ND | - |
| $\mathbf{5 1}$ | $633>332$ | $592>332$ | $\mathrm{NHSO}_{2} \mathrm{C}_{18} \mathrm{H}_{37}$ | ND | - |
| $\mathbf{5 2}$ | $633>332$ | $592>332$ | $\mathrm{NHSO}_{2} \mathrm{C}_{18} \mathrm{H}_{37}$ | ND | - |
| $\mathbf{5 3}{ }^{\mathbf{b}}$ |  | $563>447$ | $\mathrm{C}_{24} \mathrm{H}_{47} \mathrm{O}_{5} \mathrm{P}$ | ND | - |

${ }^{\text {a }}$ turnover relative to $\alpha$-D-Glc $p$ NAc-PI (1), ${ }^{\text {b }}$ No suitable MRM, ND - turnover not detected. The
multiple reaction monitoring (MRM) transition is shown as [Parent ion $m / z$ ] > [Daughter ion $m / z$ ].

## Inhibitors:

Since the majority of the compounds were not processed by the $T$. brucei GlcNAc-PI de-Nacetylase, we tested their ability to inhibit the turnover of the $\alpha$-D-GlcpNAc-IPC ${ }_{18}$ (3) substrate by the T. brucei GlcNAc-PI de- $N$-acetylase in the LC-MS/MS assay. Most compounds showed no
inhibitory activity at $100 \mu \mathrm{M}$. However, compounds 47 and $\mathbf{4 8}$ showed significant inhibition, with $\mathrm{IC}_{50}$ values of $11 \pm 4 \mu \mathrm{M}$ and $37 \pm 20 \mu \mathrm{M}$, respectively, indicating that they can be recognised but not processed by the T. brucei GlcNAc-PI de- $N$-acetylase. Interestingly, the potency of 47 and 48 is comparable to that observed with the, hydroxamic acid pseudodisaccharide analogue 54 (Figure 4), where the $N$-acetyl group is replaced with the zinc-chelating hydroxamic acid $\left(\mathrm{IC}_{50}=19 \pm 0.5\right.$ $\mu \mathrm{M})^{8}$ and suggests that the zinc-chelating group may not be driving the potency of the latter compound.


Figure 4 Hydroxamic acid
The ability of $\mathbf{4 9}$ and $\mathbf{4 8}$ to act as a substrate and inhibitor, respectively, of the T. brucei GPI pathway was confirmed using the trypanosome cell-free system with $\left[{ }^{3} \mathrm{H}\right]$-mannose labelling (Figure 5). Priming the cell-free system with 49 produced three bands corresponding to the addition of 1-3 mannose residues (Figure 5A), and, consistent with this assignment, the bands were sensitive to jackbean $\alpha$-mannosidase. As these mannosylated compounds lack the inositol 2-OH group they cannot undergo inositol acylation, a prerequisite for the transfer of ethanolamine, and thus are not processed past the $\mathrm{Man}_{3}$-species. ${ }^{25}$ Priming the cell-free system with $\mathbf{3}$ ( $\alpha$-D-GlcpNAc-IPC ${ }_{18}$ at 10 $\mu \mathrm{M}$ ) was efficiently prevented by incubation with 48 at $100 \mu \mathrm{M}$ (Figure 5B), confirming that inhibition of the GlcNAc-PI de- $N$-acetylase is sufficient to prevent the formation of downstream GPI precursors.


Figure 5 The trypanosome GPI biosynthesis in the cell-free system. A. The T. brucei cell-free system was incubated without exogenous substrate, with 1 , $\alpha$-D-GlcpNAc-PI $(10 \mu \mathrm{M})$, or with 49 $(100 \mu \mathrm{M})$ in the presence of GDP- $\left[{ }^{3} \mathrm{H}\right]$ Mannose to stimulate the production of radiolabelled mannosylated GPI intermediates. B. Inhibition of the turnover of $\mathbf{3}, \alpha$-D-GlcpNAc-IPC $18(10 \mu \mathrm{M})$, in the presence of $48(100 \mu \mathrm{M})$. Glycolipid products were extracted, separated by high-performance thin-layer chromatography, and visualised by fluorgraphy. DPM - dolichol-phosphate-mannose, M1 - Man 1 species, M2 - Man 2 species, M3 - Man ${ }_{3}$ species, $A^{\prime}-E t N P M a n_{3}$ species, where the identity and migration of the species depends on the glycolipid substrate employed.

Previous studies (see the introduction for more detail) have shown that both $\alpha-\mathrm{D}-\mathrm{Glc} p \mathrm{NAc}-$ PI (1) and $\beta$-D-GlcpNAc-PI (6) are recognised and processed by the $T$. brucei GlcNAc-PI de- $N$ acetylase, leading to the hypothesis that one or more of the inositol $2,3,4$, and $5-\mathrm{OH}$ groups is/are not required. Our data supports this hypothesis with an important caveat; when these hydroxyl groups are removed, substrate recognition and turnover is dependent on both the stereochemistry of the glycosidic linkage and the lipid composition. With the diacylglycerol lipid containing compounds 49 and 50 , only the natural $\alpha$-anomer 49 is both recognised and processed, whereas the $\beta$-anomer 50 is neither a substrate nor an inhibitor. However, neither the $\alpha$ nor $\beta$-anomer of the octadecyl lipid containing compounds 47 and 48 is processed, although both appear to be recognised and act as inhibitors. These sets of diastereoisomers differ only in the identity of their lipid component, with the more flexible diacylglycerol moiety allowing the glycan to be recognised and processed. Thus, it appears that the requirement for the presence of inositol 2, 3, 4, and 5-OH groups for recognition by the $T$. brucei GlcNAc-PI de- $N$-acetylase is nuanced and may depend on the conformational flexibility of the substrate analogue.

The inability of the sulphonamide-containing compounds, $\mathbf{5 1}$ and $\mathbf{5 2}$, to act as substrates or inhibitors confirms the importance of the phosphate group in substrate recognition. It may be that the presence of the negatively charged phosphate is essential for binding at the enzyme active site.

The inactivity of compound $\mathbf{5 3}$ is difficult to interpret. It may be that removal of the inositol $2,3,4$, and $5-\mathrm{OH}$ groups is not compatible with having a modified glucosamine moiety, or that the entire gluosamine ring is required. Having said this, the glucosamine "replacement" in compound 53 is likely to have a considerable degree of conformational flexibility, which could allow it to take up multiple orientations within the active site.

## Conclusions

In summary, we have prepared a series of compounds to probe the substrate specificity and inhibition of enzymes involved at an early stage of GPI biosynthesis. The enzyme of interest to us, GlcNAc-PI de- $N$-acetylase, proved to be fastidious in its processing of variants of $\alpha$-D-GlcpN-PI. We conclude that the glucosamine and the phospholipid moieties are essential for binding and that, while the D-myo-inositol residue is the preferred aglycone for recognition by the enzyme, the dispensing of it entirely with a cyclohexanediol group is tolerated but should be used with caution. Further work on this enzyme should focus on using the emerging structure activity relationship data to develop less synthetically complex, cell permeable analogues, which will be valuable chemical tools and may serve as leads for a drug discovery programme.

## Experimental

## Synthesis general methods

${ }^{1} \mathrm{H},{ }^{13} \mathrm{C},{ }^{31} \mathrm{P}$ NMR spectra were recorded on a Bruker AVANCE 500 MHz spectrometer using tetramethylsilane or the residual solvent as the internal standard. High resolution electrospray ionisation mass spectra [HRMS (ESI)] were recorded with a Bruker microTof spectrometer.

Melting points were determined on a Reichert hot-plate apparatus and are uncorrected. Optical rotations were measured with a Perkin-Elmer 343 polarimeter. Thin layer chromatography (TLC) was performed on Kieselgel $60 \mathrm{~F}_{254}$ (Merck) plates with various solvent systems as developers, followed by detection under UV light or by charring with sulfuric acid-water-ethanol (15:85:5). Column chromatography was performed on Kieselgel $60(0.040-0.063 \mathrm{~mm})$ (Merck). Radialband chromatography (RBC) was performed using a Chromatotron (model 7924T, TC Research UK) with silica gel $\mathrm{F}_{254}$ TLC standard grade as the adsorbent. Iatrobeads (6RS-8060) were purchased from SES Analysesysteme. All reactions were carried out under argon in commercially available dry solvents, unless otherwise stated.

## 1R,2R-1-O-(2-Azido-3,4,6-tri- $O$-acetyl-2-deoxy- $\alpha$-D-glucopyranosyl)-cyclohexanediol 16 and the $\boldsymbol{\beta}$-anomer 17

After drying overnight over $\mathrm{P}_{2} \mathrm{O}_{5}$ in a vacuum desiccator, the glycosyl donor ${ }^{14} \mathbf{1 4}(731 \mathrm{mg}$, 1.54 mmol ) and the acceptor 15 (Sigma-Aldrich) were dissolved in 1:1 $\mathrm{Et}_{2} \mathrm{O}-\mathrm{CH}_{2} \mathrm{Cl}_{2}(10 \mathrm{~mL})$. To this solution was added activated $4 \AA$ molecular sieves ( 1 g ) and TMSOTf ( $5.4 \mathrm{uL}, 0.03 \mathrm{mmol}$ ) at rt under argon. The reaction mixture was stirred at rt overnight, whereafter it was neutralised with TEA, percholated through a short column of silica gel (further elution with EtOAc) and the subsequent eluent was concentrated under reduced pressure. RBC [elution first with PE (40-60 $)$ and then with 3:2 PE $\left(40-60^{\circ}\right)$ - EtOAc] of the residue gave first the $\alpha$-linked pseudodisaccharide $16(255 \mathrm{mg}, 39 \%)$ as a waxy solid; $[\alpha]_{\mathbf{D}}^{25}+83.7^{\circ}\left(c 2.37, \mathrm{CHCl}_{3}\right) ; \delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 5.42(\mathrm{dd}$, $\left.1 \mathrm{H}, J_{2^{\prime}, 3^{\prime}}=J_{3^{\prime}, 4^{\prime}}=10.3 \mathrm{~Hz}, \mathrm{H}-3^{\prime}\right), 5.09\left(\mathrm{~d}, 1 \mathrm{H}, J_{1^{\prime}, 2^{\prime}}=3.7 \mathrm{~Hz}, \mathrm{H}-1^{\prime}\right), 4.95\left(\mathrm{dd}, 1 \mathrm{H}, J_{3^{\prime}, 4^{\prime}}=J_{4^{\prime}, 5^{\prime}}=10.3\right.$ $\left.\mathrm{Hz}, \mathrm{H}-4^{\prime}\right), 4.18\left(\mathrm{dd}, 1 \mathrm{H}, J_{5^{\prime}, 6^{\prime} \mathrm{a}}=4.8, J_{6^{\prime}, 6^{\prime} \mathrm{b}}=12.2 \mathrm{~Hz}, \mathrm{H}-6^{\prime} \mathrm{a}\right), 4.08\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-5^{\prime}\right), 4.01\left(\mathrm{dd}, 1 \mathrm{H}, J_{5^{\prime}, 6^{\prime} \mathrm{b}}\right.$ $\left.=2.2, J_{6^{\prime}, 6^{\prime} \mathrm{b}}=12.2 \mathrm{~Hz}, \mathrm{H}-6^{\prime} \mathrm{b}\right), 3.58\left(\mathrm{dd}, 1 \mathrm{H}, \mathrm{H}-2^{\prime}\right), 3.44(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-2), 3.23(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-1), 3.05(\mathrm{~s}$, $1 \mathrm{H}, 2-\mathrm{OH}), 2.50-1.90\left(\mathrm{~m}, 11 \mathrm{H}, 3 \mathrm{x} \mathrm{CH}_{3}, \mathrm{H}-3 \mathrm{a}\right.$ and 6 a$), 1.64(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-4 \mathrm{a}$ and 5 a$), 1.35-1.15(\mathrm{~m}$, $4 \mathrm{H}, \mathrm{H}-3 \mathrm{~b}, 4 \mathrm{~b}, 5 \mathrm{~b}$ and 6 b$) ; \delta_{\mathrm{C}}\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 170.6-169.8\left(3 \times \mathrm{COCH}_{3}\right), 99.3(\mathrm{C}-1$ '), $87.5(\mathrm{C}-$
1), 74.1 (C-2), 71.7 (C-3'), 68.6 (C-4'), 67.8 (C-5'), 62.1 (C-2'), 61.9 (C-6'), 32.0, 31.7, 24.4, 23.8, $20.7\left(\mathrm{COCH}_{3}\right) 20.6\left(\mathrm{COCH}_{3}\right)$; HRMS (ESI) calcd. for $\mathrm{C}_{18} \mathrm{H}_{27} \mathrm{~N}_{3} \mathrm{NaO}_{9}[\mathrm{M}+\mathrm{Na}]^{+} 452.1640$, found 452.1622 and then the $\beta$-anomer $17(202 \mathrm{mg}, 31 \%)$ as a white solid; $\mathrm{mp} 120-122^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}^{25}+21.5^{\circ}$ (c 1.15, $\mathrm{CHCl}_{3}$ ); $\delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 4.91\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-3^{\prime}\right.$ and $\left.4^{\prime}\right), 4.43\left(\mathrm{~d}, 1 \mathrm{H}, J_{1^{\prime}, 2^{\prime}}=8.1 \mathrm{~Hz}, \mathrm{H}-1^{\prime}\right)$, $4.12\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H} 6^{\prime} \mathrm{a}\right.$ and $\left.6^{\prime} \mathrm{b}\right), 3.67\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-5^{\prime}\right), 3.45\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-2^{\prime}\right), 3.40-3.30(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-1$ and 2$)$, $2.50-1.93\left(\mathrm{~m}, 11 \mathrm{H}, 3 \times \mathrm{CH}_{3}\right.$ and 2 H -cyclitol), $1.70-1.60(\mathrm{~m}, 2 \mathrm{H}$, cyclitol) $1.35(\mathrm{~m}, 1 \mathrm{H}$, cyclitol), 1.17 (m, 3H, cyclitol); $\delta_{\mathrm{C}}\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 170.6$ - $169.6\left(3 \times \mathrm{COCH}_{3}\right), 102.1\left(\mathrm{C}-1^{\prime}\right), 88.2,73.2$, 72.1, 71.8 (C-5'), 68.3, 63.7 (C-2'), 61.8 (C-6'), 32.2, 30.9, 24.2, 23.6, $20.7\left(\mathrm{COCH}_{3}\right), 20.6$
$\left(\mathrm{COCH}_{3}\right) ;$ HRMS (ESI) calcd. for $\mathrm{C}_{18} \mathrm{H}_{28} \mathrm{~N}_{3} \mathrm{O}_{9}[\mathrm{M}+\mathrm{H}]^{+} 430.1820$, found 430.1831.
Triethylammonium 1R,2R-1-O-(2-azido-3,4,6-tri- $O$-acetyl-2-deoxy- $\alpha$-D-glucopyranosyl)cyclohexanediol 2-(n-octadecylphosphate) 19

Each of the compounds $\mathbf{1 6}(117 \mathrm{mg}, 0.27 \mathrm{mmol})$ and $\mathbf{1 8}^{15}(237 \mathrm{mg}, 0.54 \mathrm{mmol})$ were dried overnight over $\mathrm{P}_{2} \mathrm{O}_{5}$ in a vacuum desiccator, whereafter anhyd pyridine was evaporated therefrom. They were then dissolved in dry pyridine ( 10 mL ), pivaloyl chloride ( $216 \mu \mathrm{~L}, 1.76 \mathrm{mmol}$ ) was added and the resulting solution was stirred under argon at rt for 1 h . A freshly prepared solution of iodine ( $274 \mathrm{mg}, 1.08 \mathrm{mmol}$ ) in 9:1 pyridine - water was then added and stirring of the reaction mixture was continued for 45 min . After the addition of $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{~mL})$, the organic solution was washed successively with $5 \%$ aq $\mathrm{NaHSO}_{3}(25 \mathrm{~mL}$ ), water ( 25 mL ), 1M TEAB buffer solution (3 x 15 mL ), dried $\left(\mathrm{MgSO}_{4}\right)$ and concentrated under reduced pressure. RBC of the residue (elution first with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and then with 9:1 $\left.\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}\right)$ afforded the TEA phosphate derivative $\mathbf{1 9}$ (140 $\mathrm{mg}, 60 \%) ;[\alpha]_{\mathrm{D}}^{25}+68.6^{\circ}\left(c 1.07, \mathrm{CHCl}_{3}\right) ; \delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 5.40\left(\mathrm{dd}, 1 \mathrm{H}, J_{3^{\prime}, 4^{\prime}}=9.2 \mathrm{~Hz}, \mathrm{H}-3^{\prime}\right)$, $5.27\left(\mathrm{~d}, 1 \mathrm{H}, J_{1^{\prime}, 2^{\prime}}=3.7 \mathrm{~Hz}, \mathrm{H}-1^{\prime}\right), 4.94\left(\mathrm{t}, 1 \mathrm{H}, J_{4^{\prime}, 5^{\prime}}=9.6 \mathrm{~Hz}, \mathrm{H}-4^{\prime}\right), 4.24-4.13\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H} 6^{\prime} \mathrm{a}\right.$ and $\mathrm{H}-$ 1 or 2$), 4.07-3.98\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-5^{\prime}\right.$ and $\left.6^{\prime} \mathrm{b}\right), 3.84-3.76\left(\mathrm{~m}, 3 \mathrm{H}, \mathrm{OCH}_{2}\right.$ and $\mathrm{H}-1$ or 2$), 3.17(\mathrm{dd}, 1 \mathrm{H}$, , $\left.J_{2^{\prime}, 3^{\prime}}=10.6 \mathrm{~Hz}, \mathrm{H}-2^{\prime}\right), 2.83\left(\mathrm{q}, 6 \mathrm{H}, J=6.8 \mathrm{~Hz}, 3 \times \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 2.05-1.95\left(\mathrm{~m}, 10 \mathrm{H}, 3 \times \mathrm{COCH}_{3}\right.$ and 1 H -cyclitol), 1.87 (m, 1H, cyclitol), $1.62-1.45\left(\mathrm{~m}, 6 \mathrm{H}, \mathrm{OCH}_{2} \mathrm{CH}_{2}\right.$ and 4 H -cyclitol), $1.33-1.14$
( $41 \mathrm{H},\left[\mathrm{CH}_{2}\right]_{15}, 3 \mathrm{x} \mathrm{CH}_{2} \mathrm{CH}_{3}$ and 2 H -cyclitol), $0.81\left(\mathrm{t}, 3 \mathrm{H}, J=6.8 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}(125 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 169.5-168.7\left(3 \times \mathrm{COCH}_{3}\right), 96.9\left(\mathrm{C}-1\right.$ '), $76.4(\mathrm{C}-1$ or $\mathrm{C}-2), 73.9(\mathrm{C}-1$ or $\mathrm{C}-2), 69.4\left(\mathrm{C}-3^{\prime}\right)$, $67.9\left(\mathrm{C}-4^{\prime}\right), 66.7\left(\mathrm{C}-5^{\prime}\right), 65.2\left(\mathrm{OCH}_{2}\right), 61.1\left(\mathrm{C}-6^{\prime}\right), 60.0\left(\mathrm{C}-2^{\prime}\right), 44.4\left[\mathrm{~N}\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right)_{3}\right], 30.9,29.7,28.7$ - 28.1, 24.8, 21.7, 21.0, 20.3, 19.6, $13.1\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 7.5\left[\mathrm{~N}\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right)_{3}\right] ; \delta_{\mathrm{P}}\left(202 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)-0.05$ (with heteronuclear decoupling); HRMS (ESI) calcd. for $\mathrm{C}_{36} \mathrm{H}_{63} \mathrm{~N}_{3} \mathrm{O}_{12} \mathrm{P}\left[\mathrm{M}-\mathrm{NEt}_{3}-\mathrm{H}\right]^{-} 760.4155$, found 760.4154.

Triethylammonium 1R,2R-1-O-(2-azido-2-deoxy- $\alpha$-D-glucopyranosyl)-cyclohexanediol 2-(noctadecylphosphate) 20

To a solution of compound $\mathbf{1 9}(78 \mathrm{mg}, 0.09 \mathrm{mmol})$ in $1: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}(10 \mathrm{~mL})$ was added 5.4 M NaOMe in $\mathrm{MeOH}(0.10 \mathrm{~mL})$. The mixture was kept for 3 h at rt and was then neutralised with Amberlite IR-120 $\left(\mathrm{H}^{+}\right)$ion-exchange resin, filtered and the filtrate concentrated under reduced pressure. Column chromatography (elution first with 3:1 $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}$ and then with $2: 1 \rightarrow 1: 1$ ) of the residue furnished the TEA salt $\mathbf{2 0}(45 \mathrm{mg}, 67 \%)$ as a waxy solid; $[\alpha]_{\mathbf{D}}^{25}$ $+51.6^{\circ}(c 4.5,1: 1 \mathrm{THF}-\mathrm{MeOH}) ; \delta_{\mathrm{H}}\left(500 \mathrm{MHz}, 1: 1 \mathrm{CDCl}_{3}-\mathrm{MeOH}-\mathrm{d}_{4}\right) 5.18\left(\mathrm{~d}, 1 \mathrm{H}, J_{1^{\prime}, 2^{\prime}}=3.5 \mathrm{~Hz}\right.$, $\left.\mathrm{H}-1^{\prime}\right), 4.25(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-1$ or 2$), 3.96-3.65\left(7 \mathrm{H}, \mathrm{OCH}_{2}, \mathrm{H}-3^{\prime}, 5^{\prime}, 6^{\prime} \mathrm{a}, \mathrm{b}\right.$ and $\mathrm{H}-1$ or 2$) 3.40\left(\mathrm{t}, 1 \mathrm{H}, J_{3^{\prime}, 4^{\prime}}\right.$ $\left.=J_{4^{\prime}, 5^{\prime}}=9.6 \mathrm{~Hz}, \mathrm{H}-4^{\prime}\right), 3.13\left(\mathrm{q}, 6 \mathrm{H}, J=7.3 \mathrm{~Hz}, 3 \times \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 3.06\left(\mathrm{dd}, 1 \mathrm{H}, J_{2^{\prime}, 3^{\prime}}=10.5 \mathrm{~Hz}, \mathrm{H}-2^{\prime}\right)$, $2.03\left(\mathrm{~m}, 1 \mathrm{H}\right.$, cyclitol), $1.90\left(\mathrm{~m}, 1 \mathrm{H}\right.$, cyclitol), $1.64\left(\mathrm{~m}, 6 \mathrm{H}, \mathrm{OCH}_{2} \mathrm{CH}_{2}\right.$ and 4 H -cyclitol), $1.43-1.22$ ( $41 \mathrm{H},\left[\mathrm{CH}_{2}\right]_{15}, 3 \mathrm{x} \mathrm{CH} 2 \mathrm{CH}_{3}$ and 2 H -cyclitol), $0.88\left(\mathrm{t}, 3 \mathrm{H}, J=6.8 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{CH}_{3}\right.$ ); $\delta_{\mathrm{C}}(125 \mathrm{MHz}, 1: 1$ $\left.\mathrm{CDCl}_{3}-\mathrm{MeOH}-\mathrm{d}_{4}\right) 98.7\left(\mathrm{C}-1{ }^{\prime}\right), 76.3(\mathrm{C}-1$ or $\mathrm{C}-2), 74.2(\mathrm{C}-1$ or $\mathrm{C}-2), 73.872 .4,72.3,66.9\left(\mathrm{OCH}_{2}\right)$, 64.5 (C-2'), $62.7,47.4\left[\mathrm{~N}\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right)_{3}\right], 32.0,31.0-30.6,29.8,29.5,27.1,23.9,22.5,22.1,15.0$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 9.6\left[\mathrm{~N}\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right)_{3}\right] ; \delta_{\mathrm{P}}\left(202 \mathrm{MHz}, 1: 1 \mathrm{CDCl}_{3}-\mathrm{MeOH}-\mathrm{d}_{4}\right)-0.47$ (with heteronuclear decoupling); HRMS (ESI) calcd. for $\mathrm{C}_{30} \mathrm{H}_{57} \mathrm{~N}_{3} \mathrm{O}_{9} \mathrm{P}\left[\mathrm{M}-\mathrm{NEt}_{3}-\mathrm{H}\right]^{-}$634.3838, found 634.3850.

1R,2R-1-O-[2-(4-Hydroxybutyl)amino-2-deoxy- $\alpha$-D-glucopyranosyl]-cyclohexanediol 2-(noctadecylphosphate) 21

A solution of the azido compound $\mathbf{2 0}(45 \mathrm{mg}, 0.06 \mathrm{mmol})$ in 1:1 THF $-\mathrm{MeOH}(5 \mathrm{~mL})$ containing $10-20 \% \mathrm{Pd}(\mathrm{OH})_{2}$ on carbon ( 15 mg ) was stirred under a hydrogen atmosphere at rt for 30 min before it was percolated through a short column of Chelex 100 on a bed of Celite (further elution with 1:1 $\mathrm{THF}-\mathrm{MeOH}$ ). The eluent was concentrated under reduced pressure and the ensuing residue was purified by column chromatography (elution gradient 6:1 $\rightarrow 4: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2}-$ $\mathrm{MeOH})$ to give the hydroxybutylamino compound $21(15 \mathrm{mg}, 38 \%)$; $[\alpha]_{\mathrm{D}}^{25}+34.0^{\circ}\left(c 1.5,1: 1 \mathrm{CHCl}_{3}\right.$ $-\mathrm{MeOH}) ; \delta_{\mathrm{H}}\left(500 \mathrm{MHz}, 1: 1 \mathrm{CDCl}_{3}-\mathrm{MeOH}-\mathrm{d}_{4}\right) 5.43\left(\mathrm{~d}, 1 \mathrm{H}, J_{1^{\prime}, 2^{\prime}}=3.6 \mathrm{~Hz}, \mathrm{H}-1^{\prime}\right), 4.05(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-$ 1 or 2), $3.93\left(\mathrm{dd}, 1 \mathrm{H}, J_{3^{\prime}, 4^{\prime}}=10.4 \mathrm{~Hz}, \mathrm{H}-3^{\prime}\right), 3.90-3.50\left(8 \mathrm{H}, \mathrm{OCH}_{2}\right.$ butyl, $\mathrm{POCH}_{2}, \mathrm{H}-5^{\prime}, 6^{\prime} \mathrm{a}, \mathrm{b}$ and $\mathrm{H}-1$ or 2$), 3.38\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-4^{\prime}\right), 3.17\left(\mathrm{t}, 2 \mathrm{H}, J=7.6 \mathrm{~Hz}, \mathrm{NCH}_{2}\right), 3.00\left(\mathrm{dd}, 1 \mathrm{H}, J_{2^{\prime}, 3^{\prime}}=10.4 \mathrm{~Hz}, \mathrm{H}-2^{\prime}\right)$, $2.18(\mathrm{~m}, 1 \mathrm{H}$, cyclitol $), 2.08(\mathrm{~m}, 1 \mathrm{H}$, cyclitol $), 1.86\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}\right.$ butyl), $1.76-1.57\left(6 \mathrm{H}, \mathrm{POCH}_{2} \mathrm{CH}_{2}\right.$, $\mathrm{CH}_{2}$ butyl and 2H-cyclitol), $1.45-1.20\left(34 \mathrm{H},\left[\mathrm{CH}_{2}\right]_{15}\right.$ and 4 H -cyclitol), $0.89(\mathrm{t}, 3 \mathrm{H}, J=6.8 \mathrm{~Hz}$, $\left.\mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(125 \mathrm{MHz}, 1: 1 \mathrm{CDCl}_{3}-\mathrm{MeOH}-\mathrm{d}_{4}\right) 97.3\left(\mathrm{C}-1^{\prime}\right), 83.2(\mathrm{C}-1$ or $\mathrm{C}-2), 79.3$ (C-1 or $\left.\mathrm{C}-2\right)$, $74.0,71.8,71.7,67.1,62.3,61.6\left(\mathrm{C}-2^{\prime}\right), 54.8,50.6,48.7\left(\mathrm{NCH}_{2}\right), 34.2,33.6,31.9,30.9-30.6,27.0$, 25.3, 25.2, $15.0\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; \delta_{\mathrm{P}}\left(202 \mathrm{MHz}, 1: 1 \mathrm{CDCl}_{3}-\mathrm{MeOH}-\mathrm{d}_{4}\right) 0.66$ (with heteronuclear decoupling); HRMS (ESI) calcd. for $\mathrm{C}_{34} \mathrm{H}_{67} \mathrm{NO}_{10} \mathrm{P}[\mathrm{M}-\mathrm{H}]^{-} 680.4508$, found 680.4554 .

## 1R,2R-1-O-(2-Amino-2-deoxy-a-D-glucopyranosyl)-cyclohexanediol 2-(n-octadecylphosphate)

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A solution of the TEA salt $\mathbf{2 0}(52 \mathrm{mg}, 0.07 \mathrm{mmol})$ in $1: 1$ stabilised THF - MeOH ( 5 mL ) containing $10-20 \% \mathrm{Pd}(\mathrm{OH})_{2}$ on carbon $(5 \mathrm{mg})$ was stirred under a hydrogen atmosphere at rt for 1 h . Work-up as described for the derivative 21 gave, after column chromatography (5:1 $\mathrm{CH}_{2} \mathrm{Cl}_{2}-$ $\mathrm{MeOH})$, the amino compound $7(33 \mathrm{mg}, 77 \%) ;[\alpha]_{\mathrm{D}}^{25}+57.5^{\circ}\left(c 3.3,1: 1 \mathrm{CHCl}_{3}-\mathrm{MeOH}\right) ; \delta_{\mathrm{H}}(500$ $\left.\mathrm{MHz}, 1: 1 \mathrm{CDCl}_{3}-\mathrm{MeOH}-\mathrm{d}_{4}\right), 5.42\left(\mathrm{~d}, 1 \mathrm{H}, J_{1^{\prime}, 2^{\prime}}=3.9 \mathrm{~Hz}, \mathrm{H}-1^{\prime}\right), 4.05(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-1$ or 2$), 3.90-3.65$ ( $6 \mathrm{H}, \mathrm{OCH}_{2}, \mathrm{H}-3^{\prime}, 5^{\prime}, 6^{\prime} \mathrm{a}, \mathrm{b}$ ), $3.58(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-1$ or 2$) 3.32\left(\mathrm{~m}, 3 \mathrm{H}, \mathrm{H}-4^{\prime}\right.$ and $\left.\mathrm{MeOH}-\mathrm{d}_{4}\right), 3.05(\mathrm{dd}, 1 \mathrm{H}$, $\left.J_{2^{\prime}, 3^{\prime}}=10.5 \mathrm{~Hz}, \mathrm{H}-2^{\prime}\right), 2.10(\mathrm{~m}, 2 \mathrm{H}, 2 \mathrm{H}-\mathrm{cyclitol}), 1.74-1.57\left(4 \mathrm{H}, \mathrm{OCH}_{2} \mathrm{CH}_{2}\right.$ and 2 H -cyclitol), 1.44 $-1.21\left(34 \mathrm{H},\left[\mathrm{CH}_{2}\right]_{15}\right.$ and 4 H -cyclitol $), 0.90\left(\mathrm{t}, 3 \mathrm{H}, J=6.8 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(125 \mathrm{MHz}, 1: 1 \mathrm{CDCl}_{3}\right.$
$\left.-\mathrm{MeOH}-\mathrm{d}_{4}\right) 97.9\left(\mathrm{C}-1{ }^{\prime}\right), 81.8(\mathrm{C}-1$ or $\mathrm{C}-2), 79.9(\mathrm{C}-1$ or $\mathrm{C}-2), 74.5,71.8,71.7,66.7\left(\mathrm{OCH}_{2}\right), 62.4$ (C-6'), 55.8 (C-2'), 34.0, 33.4, 33.1, 31.9, 31.8, 30.8, 30.5, 26.9, 25.2, 25.1, 23.8, $14.5\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; \delta_{\mathrm{P}}$ (202 MHz, 1:1 $\mathrm{CDCl}_{3}-\mathrm{MeOH}-\mathrm{d}_{4}$ ) 0.54 (with heteronuclear decoupling); HRMS (ESI) calcd. for $\mathrm{C}_{30} \mathrm{H}_{61} \mathrm{NO}_{9} \mathrm{P}[\mathrm{M}+\mathrm{H}]^{+}$610.4078, found 610.4050.

## 1R,2R-1-O-(2-Azido-3,4,6-tri- $O$-acetyl-2-deoxy- $\alpha$-D-glucopyranosyl)-2-O-(tert-

## butyldimethylsilyl)-cyclohexanediol 22

The alcohol 16 ( $284 \mathrm{mg}, 0.66 \mathrm{mmol}$ ) was dried overnight in a desiccator over $\mathrm{P}_{2} \mathrm{O}_{5}$ under high vacuum and then dissolved in anhyd. $\mathrm{CH}_{2} \mathrm{Cl}_{2}(10 \mathrm{~mL})$. To this solution, at room temperature, was added 2,6 -lutidine $(154 \mu \mathrm{~L}, 1.32 \mathrm{mmol})$ and tert-butyldimethylsilyl trifluoromethane sulfonate ( $228 \mu \mathrm{~L}, 0.99 \mathrm{mmol}$ ). After $30 \mathrm{~min}, \mathrm{CH}_{2} \mathrm{Cl}_{2}(25 \mathrm{~mL})$ and brine $(25 \mathrm{~mL})$ were added and the organic layer separated. The aqueous layer was re-extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(25 \mathrm{~mL})$ and the combined organic layers were washed with brine $(2 \times 25 \mathrm{~mL})$, dried $\left(\mathrm{MgSO}_{4}\right)$ and concentrated under reduced pressure. RBC [elution first with $\mathrm{PE}\left(40-60^{\circ}\right)$ and then with $1: 1 \mathrm{PE}\left(40-60^{\circ}\right)-\mathrm{Et}_{2} \mathrm{O}$ ] of the residue gave the azide $22(284 \mathrm{mg}, 79 \%)$ as an oil; $[\alpha]_{\mathrm{D}}^{25}+90.2^{\circ}\left(c 1.52, \mathrm{CHCl}_{3}\right) ; \delta_{\mathrm{H}}(500$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) 5.40\left(\mathrm{t}, 1 \mathrm{H}, J_{2^{\prime}, 3^{\prime}}=J_{3^{\prime}, 4^{\prime}}=9.7 \mathrm{~Hz}, \mathrm{H}-3^{\prime}\right), 5.35\left(\mathrm{~d}, 1 \mathrm{H}, J_{1^{\prime}, 2^{\prime}}=3.3 \mathrm{~Hz}, \mathrm{H}-1^{\prime}\right), 5.20(\mathrm{t}, 1 \mathrm{H}$, $\left.J_{3^{\prime}, 4^{\prime}}=J_{4^{\prime}, 5^{\prime}}=9.7 \mathrm{~Hz}, \mathrm{H}-4^{\prime}\right), 4.26\left(\mathrm{dd}, 1 \mathrm{H}, J_{5^{\prime}, 6^{\prime} \mathrm{a}}=4.7, J_{6^{\prime}, 6^{\prime} \mathrm{b}}=12.1 \mathrm{~Hz}, \mathrm{H}-6^{\prime} \mathrm{a}\right), 4.10\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-5^{\prime}\right.$ and $6^{\prime} \mathrm{b}$ ), 3.75 (m, 1H, H-1 or 2), 3.57 (m, 1H, H-1 or 2), 3.17 (dd, 1H, H-2'), $2.10-2.03$ ( $3 \mathrm{x} \mathrm{s}, 9 \mathrm{H}, 3 \mathrm{x}$ $\left.\mathrm{COCH}_{3}\right), 1.89\left(\mathrm{~m}, 2 \mathrm{H}\right.$-cyclitol), $1.70-1.25(6 \mathrm{H}-\mathrm{cyclitol}), 0.89\left(\mathrm{~s}, 9 \mathrm{H}, 3 \times \mathrm{CH}_{3}\right), 0.12-0.08(2 \mathrm{x} \mathrm{s}$, $\left.6 \mathrm{H}, 2 \times \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 170.6-169.7\left(3 \times \mathrm{COCH}_{3}\right), 97.9\left(\mathrm{C}-1^{\prime}\right), 79.7(\mathrm{C}-1$ or 2$), 72.6$ (C-1 or 2), 70.4 (C-3'), 68.7 (C-4'), 67.6 (C-5'), 62.1 (C-6'), $61.0\left(\mathrm{C}-2^{\prime}\right), 32.5,29.8,25.9,22.1,20.7$ $\left(\mathrm{COCH}_{3}\right), 20.6\left(\mathrm{COCH}_{3}\right), 18.0,-4.1,-4.9$; HRMS (ESI) calcd. for $\mathrm{C}_{24} \mathrm{H}_{42} \mathrm{~N}_{3} \mathrm{O}_{9} \mathrm{Si}[\mathrm{M}+\mathrm{H}]^{+}$ 544.2685, found 544.2698.

1R,2R-1-O-(2-Azido-2-deoxy- $\alpha$-D-glucopyranosyl)-2-O-(tert-butyldimethylsilyl)cyclohexanediol 23

To a solution of the triacetate $22(185 \mathrm{mg}, 0.34 \mathrm{mmol})$ in $1: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}(92 \mathrm{~mL})$ was added 5.4 M NaOMe in $\mathrm{MeOH}(230 \mu \mathrm{~L})$. The mixture was kept for 30 min at rt and was then neutralised with Amberlite IR-120 $\left(\mathrm{H}^{+}\right)$ion-exchange resin, filtered and the filtrate concentrated under reduced pressure. The residue, so obtained, was percolated through a short silica-gel column (further elution with EtOAc ) and the eluent was concentrated under reduced pressure to afford the triol 23 ( $136 \mathrm{mg}, 96 \%$ ) as a white solid, $\mathrm{mp} 122-123{ }^{\circ} \mathrm{C}$ (from 10:1 hexane $-\mathrm{Et}_{2} \mathrm{O}$ ); $[\alpha]_{\mathrm{D}}^{25}+84.1^{\circ}$ (c $\left.1.63, \mathrm{CHCl}_{3}\right) ; \delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 5.22\left(\mathrm{~d}, 1 \mathrm{H}, J_{1^{\prime}, 2^{\prime}}=3.4 \mathrm{~Hz}, \mathrm{H}-1^{\prime}\right), 4.02\left(\mathrm{t}, 1 \mathrm{H}, J_{3^{\prime}, 4^{\prime}}=9.4 \mathrm{~Hz}\right.$, $\left.\mathrm{H}-3^{\prime}\right), 3.90\left(\mathrm{dd}, 1 \mathrm{H}, J_{5^{\prime}, 6^{\prime} \mathrm{a}}=2.3, J_{6^{\prime}, 6^{\prime} \mathrm{b}}=11.6 \mathrm{~Hz}, \mathrm{H}-6^{\prime} \mathrm{a}\right), 3.81\left(\mathrm{dd}, 1 \mathrm{H}, J_{5^{\prime}, 6^{\prime} \mathrm{b}}=2.2, J_{6^{\prime}, 6^{\prime} \mathrm{b}}=11.6 \mathrm{~Hz}\right.$, $\left.\mathrm{H}-6^{\prime} \mathrm{b}\right), 3.73\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-5^{\prime}\right.$ and $\mathrm{H}-1$ or 2$), 3.65\left(\mathrm{t}, 1 \mathrm{H}, J_{4^{\prime}, 5^{\prime}}=9.4 \mathrm{~Hz}, \mathrm{H}-4^{\prime}\right), 3.59(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-1$ or 2$)$, $3.15\left(\mathrm{dd}, 1 \mathrm{H}, J_{2^{\prime}, 3^{\prime}}=10.4 \mathrm{~Hz} \mathrm{H}-2^{\prime}\right), 1.85(\mathrm{~m}, 2 \mathrm{H}$, cyclitol $), 1.60(\mathrm{~m}, 2 \mathrm{H}$, cyclitol) $1.50-1.20(4 \mathrm{H}$, cyclitol), $0.89\left(\mathrm{~s}, 9 \mathrm{H}, 3 \times \mathrm{CH}_{3}\right), 0.12-0.08\left(2 \mathrm{x} \mathrm{s}, 6 \mathrm{H}, 2 \times \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 98.2\left(\mathrm{C}-1^{\prime}\right)$, 78.9 (C-1 or 2), 71.5, 71.4, 70.5 (C-4'), 62.9 (C-2'), 61.6 (C-6'), 31.9, 29.4, 25.9, 22.5, 21.7, 18.0, 4.3, -4.9; HRMS (ESI) calcd. for $\mathrm{C}_{18} \mathrm{H}_{36} \mathrm{~N}_{3} \mathrm{O}_{6} \mathrm{Si}[\mathrm{M}+\mathrm{H}]^{+} 418.2368$, found 418.2365.

## 1R,2R-1-O-(2-Azido-3,4,6-tri-O-benzyl-2-deoxy- $\alpha$-D-glucopyranosyl)-2-O-(tert-

## butyldimethylsilyl)-cyclohexanediol 24

To a stirred and cooled $\left(0^{\circ} \mathrm{C}\right)$ solution of the triol $23(70 \mathrm{mg}, 0.17 \mathrm{mmol})$ in DMF $(10 \mathrm{~mL})$ under argon was added $\mathrm{NaH}(19 \mathrm{mg}, 0.78 \mathrm{mmol})$ and the solution was stirred for 15 min before benzyl bromide ( $93 \mu \mathrm{~L}, 0.78 \mathrm{mmol}$ ) was added dropwise. The reaction mixture was stirred at rt overnight and then poured slowly and carefully into ice-cold water ( 50 mL ). After dilution with EtOAc ( 50 mL ), the EtOAc solution was washed with brine $(25 \mathrm{~mL})$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated under reduced pressure. RBC [elution gradient $\mathrm{PE}\left(40-60^{\circ}\right) \rightarrow 10: 1 \rightarrow 7: 1 \rightarrow 4: 1 \mathrm{PE}$ $\left.\left(40-60^{\circ}\right)-\mathrm{Et}_{2} \mathrm{O}\right]$ of the residue yielded the fully protected compound $24(86 \mathrm{mg}, 74 \%) ;[\alpha]_{\mathrm{D}}^{25}+57.9^{\circ}$ (c 1.78, $\left.\mathrm{CHCl}_{3}\right) ; \delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.33-7.03(15 \mathrm{H}, 3 \times \mathrm{Ph}), 5.17\left(\mathrm{~d}, 1 \mathrm{H}, J_{1^{\prime}, 2^{\prime}}=3.6 \mathrm{~Hz}, \mathrm{H}-1^{\prime}\right)$, $4.84-4.37\left(6 \mathrm{H}, 3 \times \mathrm{CH}_{2} \mathrm{Ar}\right), 3.93\left(\mathrm{t}, 1 \mathrm{H}, J_{3^{\prime}, 4^{\prime}}=9.0 \mathrm{~Hz}, \mathrm{H}-3^{\prime}\right), 3.83\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-5^{\prime}\right), 3.72-3.60(\mathrm{~m}$, $3 \mathrm{H}, \mathrm{H}-4^{\prime}, 6^{\prime} \mathrm{a}$ and 1 or 2$), 3.57\left(\mathrm{dd}, 1 \mathrm{H}, J_{5^{\prime}, 6^{\prime} \mathrm{b}}=2.0, J_{6^{\prime} \mathrm{a}, 6^{\mathrm{b}}}=10.7 \mathrm{~Hz}, \mathrm{H}-6^{\prime} \mathrm{b}\right), 3.48(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-1$ or 2$)$,
$3.26\left(\mathrm{dd}, 1 \mathrm{H}, J_{2^{\prime}, 3^{\prime}}=10.3 \mathrm{~Hz}, \mathrm{H}-2^{\prime}\right), 1.80(\mathrm{~m}, 2 \mathrm{H}$, cyclitol $), 1.32(\mathrm{~m}, 2 \mathrm{H}$, cyclitol $), 1.40-1.10(4 \mathrm{H}$, cyclitol), $0.82\left(\mathrm{~s}, 9 \mathrm{H}, 3 \times \mathrm{CH}_{3}\right), 0.02-0.00\left(2 \mathrm{x} \mathrm{s}, 6 \mathrm{H}, 2 \times \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 137.0-$ $136.8(\mathrm{Ph}), 127.4-126.7(\mathrm{Ph}), 97.0\left(\mathrm{C}-1{ }^{\prime}\right), 79.2\left(\mathrm{C}-3^{\prime}\right), 77.9(\mathrm{C}-1$ or 2$), 77.4,74.3,74.1,72.5,71.3$, 69.8 (C-5'), 67.3 (C-6'), 62.6 (C-2'), 31.3, 28.7, 24.5, 21.7, 21.0, 18.4, -5.2, -5.8; HRMS (ESI) calcd. for $\mathrm{C}_{39} \mathrm{H}_{54} \mathrm{~N}_{3} \mathrm{O}_{6} \mathrm{Si}[\mathrm{M}+\mathrm{H}]^{+}$688.3776, found 688.3780.

1R,2R-1-O-(2-Amino-3,4,6-tri- $O$-benzyl-2-deoxy- $\alpha$-D-glucopyranosyl)-2-O-(tert-

## butyldimethylsilyl)-cyclohexanediol 25

To a stirred solution of $\mathbf{2 4}(86 \mathrm{mg}, 0.12 \mathrm{mmol})$ in $10: 1 \mathrm{THF}$ - water $(5 \mathrm{~mL})$ at $60^{\circ} \mathrm{C}$ was added $\mathrm{Ph}_{3} \mathrm{P}(98 \mathrm{mg}, 0.38 \mathrm{mmol})$. After $3 \mathrm{~h}, \mathrm{TLC}$ showed the complete disappearance of the starting material. The reaction was then cooled to rt , poured into water ( 25 mL ) and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( $3 \times 25 \mathrm{~mL}$ ). The combined organics were washed successively with water ( 25 mL ), brine ( 25 mL ), dried $\left(\mathrm{MgSO}_{4}\right)$ and concentrated under reduced pressure. RBC [elution gradient hexane $\rightarrow 7: 1 \rightarrow$ $3: 1 \rightarrow 1: 1 \rightarrow 1: 3$ hexane -EtOAc$]$ of the residue afforded the amino derivative $\mathbf{2 5}(53 \mathrm{mg}, 62 \%) ;$ $[\alpha]_{\mathrm{D}}^{25}+65.6^{\circ}\left(c 1.09, \mathrm{CHCl}_{3}\right) ; \delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.35-7.09(15 \mathrm{H}, 3 \times \mathrm{Ph}), 4.98-4.44(7 \mathrm{H}, \mathrm{H}-$ $1^{\prime}$ and $\left.3 \times \mathrm{CH}_{2} \mathrm{Ar}\right), 3.87\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-5^{\prime}\right), 3.75\left(\mathrm{dd}, 1 \mathrm{H}, J_{5^{\prime}, 6^{\prime} \mathrm{a}}=3.6, J_{6^{\prime}, 6^{\prime} \mathrm{b}}=10.6 \mathrm{~Hz}, \mathrm{H}-6^{\prime} \mathrm{a}\right), 3.61(\mathrm{~m}$, $3 \mathrm{H}, \mathrm{H}-4^{\prime}, 6^{\prime} \mathrm{b}$ and 1 or 2 ), $3.55\left(\mathrm{t}, 1 \mathrm{H}, J_{3^{\prime}, 4^{\prime}}=9.2 \mathrm{~Hz}, \mathrm{H}-3^{\prime}\right), 3.45(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-1$ or 2$), 2.77(\mathrm{dd}, 1 \mathrm{H}$, $\left.J_{1^{\prime}, 2^{\prime}}=3.7, J_{2^{\prime}, 3^{\prime}}=9.7 \mathrm{~Hz}, \mathrm{H}-2^{\prime}\right), 1.98(\mathrm{~m}, 1 \mathrm{H}$, cyclitol $), 1.70(\mathrm{~m}, 1 \mathrm{H}$, cyclitol $), 1.63-1.16(6 \mathrm{H}$, cyclitol), $0.84\left(\mathrm{~s}, 9 \mathrm{H}, 3 \times \mathrm{CH}_{3}\right), 0.00\left(2 \mathrm{x} \mathrm{s}, 6 \mathrm{H}, 2 \times \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 138.7-138.0(\mathrm{Ph})$, 128.5 - 127.7 (Ph), 100.2 (C-1'), 84.1 (C-3'), 79.9 (C-1 or 2), 78.9, 75.6, 74.9, 73.5, 71.4 (C-5'), 71.0, 68.7 (C-6'), 56.4 (C-2'), 31.9, 29.4, 25.9, 22.4, 21.8, 18.0, -4.3, -4.4; HRMS (ESI) calcd. for $\mathrm{C}_{39} \mathrm{H}_{56} \mathrm{NO}_{6} \mathrm{Si}[\mathrm{M}+\mathrm{H}]^{+}$662.3871, found 662.3874 .

1R,2R-1-O-[2-N-(tert-Butoxycarbonyl)amino-3,4,6-tri-O-benzyl-2-deoxy- $\alpha-\mathrm{D}$-glucopyranosyl]-2-O-(tert-butyldimethylsilyl)-cyclohexanediol 26

The amine $\mathbf{2 5}$ ( $147 \mathrm{mg}, 0.22 \mathrm{mmol}$ ) was dissolved in EtOAc ( 10 mL ) at rt. Di-tertbutyldicarbonate ( $58 \mathrm{mg}, 0.26 \mathrm{mmol}$ ) was then added and the mixture was stirred overnight at rt .

Afterwards, the reaction mixture was diluted with EtOAc ( 25 mL ) and then washed successively with water $(25 \mathrm{~mL})$, brine $(25 \mathrm{~mL})$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated under reduced pressure. RBC [elution gradient hexane $\rightarrow 7: 1 \rightarrow 5: 1$ hexane -EtOAc ] of the residue afforded the Boc protected derivative $26(132 \mathrm{mg}, 79 \%) ;[\alpha]_{\mathrm{D}}^{25}+39.0^{\circ}\left(c 1.06, \mathrm{CHCl}_{3}\right) ; \delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.30-7.05(15 \mathrm{H}$, $3 \times \mathrm{Ph}), 4.95\left(\mathrm{~d}, 1 \mathrm{H}, J_{1^{\prime}, 2^{\prime}}=3.3 \mathrm{~Hz}, \mathrm{H}-1^{\prime}\right), 4.76-4.38\left(7 \mathrm{H}, \mathrm{NH}\right.$ and $\left.3 \times \mathrm{CH} \mathrm{H}_{2} \mathrm{Ar}\right), 3.90\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-2^{\prime}\right)$ $3.83\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-5^{\prime}\right), 3.69\left(\mathrm{dd}, 1 \mathrm{H}, J_{5^{\prime}, 6^{\prime} \mathrm{a}}=4.1, J_{6^{\prime} \mathrm{a}^{\prime} 6^{\prime} \mathrm{b}}=10.7 \mathrm{~Hz}, \mathrm{H}-6^{\prime} \mathrm{a}\right), 3.61\left(\mathrm{~m}, 3 \mathrm{H}, \mathrm{H}-3^{\prime}, 4^{\prime}\right.$, and $\left.6^{\prime} \mathrm{b}\right), 3.48$ (m, 1H, H-1 or 2), 3.37 (m, 1H, H-1 or 2 ), 2.01 (m, 1H, cyclitol), 1.74 (m, 1H, cyclitol), $1.52\left(\mathrm{~m}, 2 \mathrm{H}\right.$, cyclitol), $1.35\left(\mathrm{~s}, 9 \mathrm{H}, 3 \times \mathrm{BocCH}_{3}\right), 1.30-1.10\left(4 \mathrm{H}\right.$, cyclitol), $0.83\left(\mathrm{~s}, 9 \mathrm{H}, 3 \times \mathrm{XH}_{3}\right)$, $0.00\left(2 \mathrm{x} \mathrm{s}, 6 \mathrm{H}, 2 \times \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 155.3(\mathrm{C}=\mathrm{O}), 138.6-138.2(\mathrm{Ph}), 128.4-127.5$ (Ph), 99.1 (C-1'), 81.5, 81.3 (C-1 or 2), 79.5, 78.5, 75.3, 75.1, 73.4, 73.2 (C-1 or 2), $71.4\left(\mathrm{C}-5^{\prime}\right)$, 68.8 (C-6'), 54.6 (C-2'), 33.4, 30.8, 28.5, 26.1, 23.3, 22.9, 18.1, -3.9, -4.3; HRMS (ESI) calcd. for $\mathrm{C}_{44} \mathrm{H}_{64} \mathrm{NO}_{8} \mathrm{Si}[\mathrm{M}+\mathrm{H}]^{+} 762.4396$, found 762.4393.

## 1R,2R-1-O-[2-N-(tert-Butoxycarbonyl)amino-3,4,6-tri- $O$-benzyl-2-deoxy- $\alpha$-D-glucopyranosyl]cyclohexanediol 27

To a stirred solution of the silyl derivative $\mathbf{2 6}(114 \mathrm{mg}, 0.15 \mathrm{mmol})$ in THF $(10 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$ was added $\sim 70 \%$ HF-pyridine $(90 \mu \mathrm{~L})$. The solution was stirred overnight at rt whereafter a further aliquot of $\sim 70 \%$ HF-pyridine ( $90 \mu \mathrm{~L}$ ) was added and the solution was left to stir overnight; this process was continued on day 3 . On day 4 , TLC revealed the complete disappearance of the starting material, whereafter satd $\mathrm{NaHCO}_{3}(1 \mathrm{~mL})$ was added dropwise to quench the reaction and the resulting solution was poured into brine ( 25 mL ) and extracted with EtOAc ( $3 \times 25 \mathrm{~mL}$ ). The EtOAc extracts were combined and washed with brine ( 25 mL ), dried $\left(\mathrm{MgSO}_{4}\right)$ and concentrated under reduced pressure. RBC [elution gradient hexane $\rightarrow 2: 1$ hexane - EtOAc] of the residue furnished the alcohol $27(97 \mathrm{mg}, 49 \%) ;[\alpha]_{\mathrm{D}}^{25}+27.2^{\circ}\left(c 1.08, \mathrm{CHCl}_{3}\right) ; \delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.36-$ $7.10(15 \mathrm{H}, 3 \times \mathrm{Ph}), 5.12\left(\mathrm{~d}, 1 \mathrm{H}, J_{1^{\prime}, 2^{\prime}}=3.6 \mathrm{~Hz}, \mathrm{H}-1^{\prime}\right), 4.86-4.44\left(7 \mathrm{H}, \mathrm{NH}\right.$ and $\left.3 \times \mathrm{CH}_{2} \mathrm{Ar}\right), 3.95(\mathrm{dd}$, $\left.1 \mathrm{H}, J_{3^{\prime}, 4^{\prime}}=J_{4^{\prime}, 5^{\prime}}=9.7 \mathrm{~Hz}, \mathrm{H}-4^{\prime}\right), 3.86\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-2^{\prime}\right), 3.75-3.61\left(4 \mathrm{H}, \mathrm{H}-3^{\prime}, 5^{\prime}\right.$ and $\left.6^{\prime} \mathrm{a}, \mathrm{b}\right), 3.47(\mathrm{~m}, 1 \mathrm{H}$,
$\mathrm{H}-1$ or 2 ), 3.32 (m, 1H, H-1 or 2), $2.10-1.91$ (2H, cyclitol), 1.65 (m, 2H, cyclitol), 1.43 (s, 9H, $3 \times$ $\mathrm{CH}_{3}$ ), $1.33-1.16\left(4 \mathrm{H}\right.$, cyclitol); $\delta_{\mathrm{C}}\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 155.6(\mathrm{C}=\mathrm{O}), 138.4-138.0(\mathrm{Ph}), 128.4-$ 127.7 (Ph), 99.5 (C-1'), 85.0 (C-1 or 2), 80.5, 79.7, 78.5, 75.0, 74.0 (C-1 or 2), 73.4, 71.2 (C-4'), 68.7 (C-6'), 54.9 (C-2'), 32.9, 31.7, 28.4, 24.3, 23.9; HRMS (ESI) calcd. for $\mathrm{C}_{38} \mathrm{H}_{50} \mathrm{NO}_{8}[\mathrm{M}+\mathrm{H}]^{+}$ 648.3531, found 648.3527 .

Triethylammonium 1R,2R-1-O-[2-N-(tert-butoxycarbonyl)amino-3,4,6-tri- $O$-benzyl-2-deoxy-$\alpha$-D-glucopyranosyl]-cyclohexanediol 2-(1,2-di-O-hexadecanoyl-sn-glycerol 3-phosphate) 29 This compound was obtained from the alcohol $27(55.7 \mathrm{mg}, 0.086 \mathrm{mmol})$ and 1,2 -di- $O$ -hexadecanoyl-sn-glycerol 3-hydrogenphosphonate TEA salt $28^{19}(126 \mathrm{mg}, 0.17 \mathrm{mmol})$ in the presence of pivaloyl chloride ( $69 \mu \mathrm{~L}, 0.56 \mathrm{mmol}$ ) essentially as described for the 2 -( $n$-octadecyl phosphate) 19. After the oxidation with iodine ( $87 \mathrm{mg}, 0.34 \mathrm{mmol}$ ) in 9:1 pyridine - water followed by the same aqueous workup as described for $19, \mathrm{RBC}$ (elution first with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and then with 20:1 $\rightarrow$ 15:1 $\left.\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}\right)$ afforded the TEA phosphate derivative $29(57 \mathrm{mg}, 52 \%)$ as an opaque oil; $[\alpha]_{\mathbf{D}}^{25}+27.7^{\circ}\left(c 1.08, \mathrm{CHCl}_{3}\right) ; \delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 12.5$ (brs, $1 \mathrm{H}, \mathrm{NH}$ TEA salt), 7.36 - $7.10(15 \mathrm{H}, 3 \times \mathrm{Ph}), 5.24\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-2\right.$ glycerol), 4.96 ( $\left.\mathrm{s}, 1 \mathrm{H}, \mathrm{H}-1^{\prime}\right), 4.84-4.44\left(6 \mathrm{H}, 3 \times \mathrm{CH}_{2} \mathrm{Ar}\right)$, $4.39(\mathrm{~m}, 1 \mathrm{H}, 1-$ or 3-CHa glycerol), $4.17(\mathrm{dd}, 1 \mathrm{H}, J=6.6, J=12.0 \mathrm{~Hz}, 1-$ or $3-\mathrm{CHb}$ glycerol), 4.13 $-3.97\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{H}-2^{\prime}, 1\right.$ or 2 cyclitol and 1- or $3-\mathrm{CH}_{2}$ glycerol $), 3.95\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-4^{\prime}\right), 3.85\left(\mathrm{t}, \mathrm{J}_{2^{\prime}, 3^{\prime}}=\right.$ $\left.J_{3^{\prime}, 4^{\prime}}=9.9 \mathrm{~Hz}, \mathrm{H}-3^{\prime}\right), 3.74\left(\mathrm{dd}, 1 \mathrm{H}, J_{5^{\prime}, 6^{\prime} \mathrm{a}}=4.1, J_{6^{\prime} \mathrm{a}, 6^{\prime} \mathrm{b}}=10.7 \mathrm{~Hz}, \mathrm{H}-6^{\prime} \mathrm{a}\right), 3.66\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-5^{\prime}\right.$ and $\left.6^{\prime} \mathrm{b}\right)$, $3.50(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-1$ or 2$), 2.97\left(\mathrm{q}, 6 \mathrm{H}, J=7.1 \mathrm{~Hz}, 3 \times \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 2.27\left(\mathrm{~m}, 4 \mathrm{H}, 2 \times \mathrm{COCH}_{2}\right), 2.21-$ $1.95(2 \mathrm{H}$, cyclitol $), 1.72-1.50\left(\mathrm{~m}, 6 \mathrm{H}, 2 \times \mathrm{COCH}_{2} \mathrm{CH}_{2}\right.$ and 2 H cyclitol), $1.44\left(\mathrm{~s}, 9 \mathrm{H}, 3 \times \mathrm{CH}_{3}\right)$, $1.33-1.18\left(61 \mathrm{H}, 3 \times \mathrm{CH}_{2} \mathrm{CH}_{3}, 2 \times\left[\mathrm{CH}_{2}\right]_{12}\right.$ and 4 H cyclitol $), 0.88\left(\mathrm{t}, 6 \mathrm{H}, J=7.3 \mathrm{~Hz}, 2 \times \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$; $\delta_{\mathrm{C}}\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 172.4(\mathrm{C}=\mathrm{O}), 171.9(\mathrm{C}=\mathrm{O}), 155.4(\mathrm{C}=\mathrm{O}), 138.9-137.4(\mathrm{Ph}), 127.3-126.3$ (Ph), 98.9 (C-1'), 80.0, 76.8, 73.7, 72.7, 72.3, 70.3, 69.8, 69.5, 68.1, 67.6, 62.4, 61.8, 61.4, 53.5, $44.2\left[\mathrm{~N}\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right)_{3}\right], 33.3,33.1,30.9,30.2,28.7$ - 28.1, 27.5, 23.4, 21.7, 13.1, $7.40\left[\mathrm{~N}\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right)_{3}\right]$;
$\delta_{\mathrm{P}}\left(202 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 0.04$ (with heteronuclear decoupling); HRMS (ESI) calcd. for $\mathrm{C}_{73} \mathrm{H}_{115} \mathrm{NO}_{15} \mathrm{P}$ $\left[\mathrm{M}-\mathrm{NEt}_{3}-\mathrm{H}\right]^{-}$1276.8010, found 1276.8015.

## 1R,2R-1-O-[2-N-(tert-Butoxycarbonyl)amino-2-deoxy- $\alpha$-D-glucopyranosyl]-cyclohexanediol 2-

 (1,2-di-O-hexadecanoyl-sn-glycerol 3-phosphate) 30A solution of the benzylated compound $29(57 \mathrm{mg}, 0.041 \mathrm{mmol})$ in 1:1 THF - $n$-propanol $(10 \mathrm{~mL})$ containing $10-20 \% \mathrm{Pd}(\mathrm{OH})_{2}$ on carbon $(15 \mathrm{mg})$ was stirred under 3 atm of hydrogen for 3 $h$ before it was percolated through a short column of Chelex 100 on a bed of Celite (further elution with 1:1 THF - $n$-propanol). The eluent was concentrated under reduced pressure and the ensuing residue was purified by column chromatography (elution gradient 7:1 $\rightarrow 4: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}$ ) to give the Boc protected derivative $30(30 \mathrm{mg}, 73 \%)$; $[\alpha]_{\mathrm{D}}^{25}+39.1^{\circ}\left(c 3.00,1: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}\right) ; \delta_{\mathrm{H}}$ ( $\left.500 \mathrm{MHz}, 1: 1 \mathrm{CDCl}_{3}-\mathrm{MeOH}-\mathrm{d}_{4}\right) 5.25(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-2$ glycerol), $4.90(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}-1$ '), 4.45 (m, 1H, 1or 3-CHa glycerol), $4.20(\mathrm{dd}, 1 \mathrm{H}, J=6.7, J=11.7 \mathrm{~Hz}, 1-$ or $3-\mathrm{CHb}$ glycerol $), 4.00(\mathrm{~m}, 2 \mathrm{H}, 1-$ or 3$\mathrm{CH}_{2}$ glycerol $), 3.75\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-3^{\prime}\right.$ and $\left.4^{\prime}\right), 3.65(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-1$ or 2$), 3.55\left(\mathrm{dd}, 1 \mathrm{H}, J_{1^{\prime}, 2^{\prime}}=3.0, J_{2^{\prime}, 3^{\prime}}=\right.$ $\left.10.5 \mathrm{~Hz}, \mathrm{H}-2^{\prime}\right), 3.47$ (m, 1H, H-1 or 2), $3.40\left(\mathrm{~m}, 3 \mathrm{H}, \mathrm{H}-5^{\prime}\right.$ and $\left.6^{\prime} \mathrm{a}, \mathrm{b}\right), 2.40-2.00(\mathrm{~m}, 6 \mathrm{H}, 2 \mathrm{x}$ $\mathrm{COCH}_{2}$ and 2 H cyclitol), $1.74-1.57\left(\mathrm{~m}, 6 \mathrm{H}, 2 \times \mathrm{COCH}_{2} \mathrm{CH}_{2}\right.$ and 2 H cyclitol), $1.47(\mathrm{~s}, 9 \mathrm{H}, 3 \times$ $\left.\mathrm{CH}_{3}\right), 1.40-1.20\left(52 \mathrm{H}, 2 \times\left[\mathrm{CH}_{2}\right]_{12}\right.$ and 4 H cyclitol $), 0.89\left(\mathrm{t}, 6 \mathrm{H}, \mathrm{J}=7.1 \mathrm{~Hz}, 2 \times \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}(125$ $\left.\mathrm{MHz}, 1: 1 \mathrm{CDCl}_{3}-\mathrm{MeOH}-\mathrm{d}_{4}\right) 174.6(\mathrm{C}=\mathrm{O}), 174.1(\mathrm{C}=\mathrm{O}), 158.2(\mathrm{C}=\mathrm{O}), 98.9\left(\mathrm{C}-1^{\prime}\right), 82.0,78.0$, $73.3,73.0,72.0,71.1,64.0,63.1,62.0,57.1,56.4,34.7,34.5,33.0-29.5,28.5,25.4,24.3,23.5$, 23.1, 14.3; $\delta_{\mathrm{P}}\left(202 \mathrm{MHz}, 1: 1 \mathrm{CDCl}_{3}-\mathrm{MeOH}-\mathrm{d}_{4}\right)-0.28$ (with heteronuclear decoupling); HRMS (ESI) calcd. for $\mathrm{C}_{52} \mathrm{H}_{97} \mathrm{NO}_{15} \mathrm{P}[\mathrm{M}-\mathrm{H}]^{-} 1006.6601$, found 1006.6635 .

1R,2R-1-O-(2-Amino-2-deoxy- $\alpha$-D-glucopyranosyl)-cyclohexanediol 2-(1,2-di- $O$-hexadecanoyl-sn-glycerol 3-phosphate) 9

To a solution of the tert-butoxycarbonyl protected compound $\mathbf{3 0}(30 \mathrm{mg}, 0.030 \mathrm{mmol})$ in $1: 1$ $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}(1 \mathrm{~mL})$ was added 9:1 trifluoroacetic acid (TFA) - water ( 5 mL ). After stirring 3 h at rt , toluene ( 5 mL ) was added and the solvents were removed under reduced pressure. Toluene
( $2 \times 5 \mathrm{~mL}$ ) was evaporated off twice from the residue (to remove traces of TFA and water) to give the pseudodisaccharide phosphate derivative $\mathbf{9}(25 \mathrm{mg}, 93 \%)$ which did not require any further purification; $[\alpha]_{\mathbf{D}}^{25}+28.0^{\circ}\left(c 2.50,1: 1 \mathrm{CHCl}_{3}-\mathrm{MeOH}\right) ; \delta_{\mathrm{H}}\left(500 \mathrm{MHz}, 1: 1 \mathrm{CDCl}_{3}-\mathrm{MeOH}-\mathrm{d}_{4}\right) 5.28$ (d, 1H, $\left.J_{1^{\prime}, 2^{\prime}}=3.7 \mathrm{~Hz}, \mathrm{H}-1^{\prime}\right), 5.14(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-2$ glycerol $), 4.34(\mathrm{dd}, 1 \mathrm{H}, J=3.2, J=12.0 \mathrm{~Hz}, 1-$ or 3CHa glycerol), 4.10 (dd, $1 \mathrm{H}, J=6.6, J=12.0 \mathrm{~Hz}, 1$ - or $3-\mathrm{CHb}$ glycerol), $3.96(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-1$ or 2 ), $3.88\left(\mathrm{~m}, 2 \mathrm{H}, 1\right.$ - or $3-\mathrm{CH}_{2}$ glycerol), $3.71\left(\mathrm{~m}, 3 \mathrm{H}, \mathrm{H}-3^{\prime}\right.$ and $\left.6^{\prime} \mathrm{a}, \mathrm{b}\right), 3.62\left(\mathrm{dd}, 1 \mathrm{H}, \mathrm{J}_{4^{\prime}, 5^{\prime}}=9.4, J_{5^{\prime}, 6^{\prime}}=3.9\right.$ $\left.\mathrm{Hz}, \mathrm{H}-5^{\prime}\right), 3.45(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-1$ or 2$), 3.33\left(\mathrm{t}, 1 \mathrm{H}, \mathrm{J}_{3^{\prime}, 4^{\prime}}=9.4 \mathrm{~Hz}, \mathrm{H}-4^{\prime}\right), 2.96\left(\mathrm{dd}, 1 \mathrm{H}, J_{2^{\prime}, 3^{\prime}}=10.5 \mathrm{~Hz}, \mathrm{H}-\right.$ $\left.2^{\prime}\right), 2.24\left(\mathrm{~m}, 4 \mathrm{H}, 2 \times \mathrm{COCH}_{2}\right), 2.0\left(\mathrm{~m}, 2 \mathrm{H}\right.$, cyclitol), $1.68-1.47\left(\mathrm{~m}, 6 \mathrm{H}, 2 \times \mathrm{COCH}_{2} \mathrm{CH}_{2}\right.$ and 2 H cyclitol), $1.35-1.14\left(52 \mathrm{H}, 2 \times\left[\mathrm{CH}_{2}\right]_{12}\right.$ and 4 H cyclitol $), 0.79\left(\mathrm{t}, 6 \mathrm{H}, J=7.1 \mathrm{~Hz}, 2 \times \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}$ (125 MHz, 1:1 $\left.\mathrm{CDCl}_{3}-\mathrm{MeOH}-\mathrm{d}_{4}\right) 175.3(\mathrm{C}=\mathrm{O}), 174.8(\mathrm{C}=\mathrm{O}), 98.0(\mathrm{C}-1$ '), $82.5(\mathrm{C}-1$ or 2$), 80.2$ (C-1 or 2$), 74.2\left(\mathrm{C}-5^{\prime}\right), 71.6,71.5,71.4,64.8\left(\mathrm{CH}_{2}\right.$ glycerol), $63.8\left(\mathrm{CH}_{2}\right.$ glycerol), $62.3\left(\mathrm{C}-6^{\prime}\right), 55.7$ (C-2'), $35.5,35.4,34.0,33.5,33.2,31.0-30.4,26.2,25.3,23.9,15.1 ; \delta_{\mathrm{P}}\left(202 \mathrm{MHz}, 1: 1 \mathrm{CDCl}_{3}-\right.$ $\left.\mathrm{MeOH}-\mathrm{d}_{4}\right) 0.10$ (with heteronuclear decoupling); HRMS (ESI) calcd. for $\mathrm{C}_{52} \mathrm{H}_{97} \mathrm{NO}_{15} \mathrm{P}[\mathrm{M}-\mathrm{H}]^{-}$ 906.6077, found 906.6094.

## $N$-[(1R,2R)-2-(Benzyloxy)cyclohexyl]octadecane-1-sulphonamide 32

To a solution of $\mathrm{CH}_{2} \mathrm{Cl}_{2}(10 \mathrm{~mL})$ and triethylamine $(2.1 \mathrm{~mL})$ under argon was added ( $1 R$, 2R)-1-amino-2-benzyloxycyclohexane 31 ( $1.0 \mathrm{~g}, 4.87 \mathrm{mmol}$ ) and 1-octadecanesulfonyl chloride $(2.1 \mathrm{~g}, 5.95 \mathrm{mmol})$, purchased from Sigma-Aldrich and Alfa Aesar, respectively. The reaction mixture was stirred at rt overnight, whereafter it was diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(40 \mathrm{~mL})$, washed successively with water $(25 \mathrm{~mL})$, brine $(25 \mathrm{~mL})$, dried $\left(\mathrm{MgSO}_{4}\right)$ and concentrated under reduced pressure. RBC (elution first with hexane and then with a gradient of 5:1 $\rightarrow 3: 1$ hexane -EtOAc ) gave the sulphonamide $\mathbf{3 2}(1.9 \mathrm{~g}, 76 \%)$ as white needles; $\mathrm{mp} 63-64{ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}^{25}-35.6^{\circ}(c 1.00$, $\left.\mathrm{CHCl}_{3}\right) ; \delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.40-7.20(\mathrm{~m}, 5 \mathrm{H}, \mathrm{Ph}), 4.68\left(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=1.4 \mathrm{~Hz}, \mathrm{OCH}_{2}\right), 4.46(\mathrm{~d}, 1 \mathrm{H}$, $J=5.0 \mathrm{~Hz}, \mathrm{NH}), 3.23-3.11(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H} 1$ and 2$), 3.03-2.90\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{SO}_{2} \mathrm{CH}_{2}\right), 2.23(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H} 3 \mathrm{a}$ and 6a), $1.80-1.60\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{SO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}, \mathrm{H} 4 \mathrm{a}\right.$ and 5 a$), 1.35-1.11\left(\mathrm{~m}, 34 \mathrm{H}, 15 \times\left[\mathrm{CH}_{2}\right]_{15}, \mathrm{H} 3 \mathrm{~b}, 4 \mathrm{~b}\right.$,

5 b and 6 b$), 0.88\left(\mathrm{t}, 3 \mathrm{H}, J=6.8 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 138.1,128.5,127.8,127.7,80.1$
$(\mathrm{C} 1$ or 2$), 70.6\left(\mathrm{OCH}_{2}\right), 57.6(\mathrm{C} 1$ or 2$), 53.2\left(\mathrm{SO}_{2} \mathrm{CH}_{2}\right), 33.2(\mathrm{C} 3$ or 6$), 31.9,30.1(\mathrm{C} 3$ or 6$), 29.7$, 29.5, 29.4, 29.1, 28.3, 24.2, 23.8, 23.5, 22.7, $14.1\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right)$; HRMS (ESI) calcd. for $\mathrm{C}_{31} \mathrm{H}_{56} \mathrm{NO}_{3} \mathrm{~S}$ $[\mathrm{M}+\mathrm{H}]^{+} 522.3975$, found 522.3981.
$N$-[(1R,2R)-2-Hydroxycyclohexyl]octadecane-1-sulphonamide 33
A solution of the benzyloxysulphonamide $32(200 \mathrm{mg}, 0.38 \mathrm{mmol})$ in $5: 1 \mathrm{THF}$ - AcOH ( 6 mL ) containing $10-20 \% \mathrm{Pd}(\mathrm{OH})_{2}$ on carbon ( 50 mg ) was stirred under a slight over pressure of hydrogen at room temperature for 2 h before it was percholated through a short column of Celite on a bed of silica gel (further elution with EtOAc). The eluent was concentrated under reduced pressure to give the deprotected alcohol $\mathbf{3 3}(140 \mathrm{mg}, 85 \%)$ as a white solid which was used without any further purification; mp $101-102{ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}^{25}-7.3\left(c 3.40, \mathrm{CHCl}_{3}\right) ; \delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 4.54$ $(\mathrm{d}, 1 \mathrm{H}, J=7.4 \mathrm{~Hz}, \mathrm{NH}), 3.32(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H} 2), 3.15-3.02\left(\mathrm{~m}, 3 \mathrm{H}, \mathrm{SO}_{2} \mathrm{CH}_{2}\right.$ and H 1$), 2.54(\mathrm{brs}, 1 \mathrm{H}$, $\mathrm{OH}), 2.12-2.03(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H} 3 \mathrm{a}$ and 6 a$), 1.90-1.63\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{SO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}, \mathrm{H} 4 \mathrm{a}\right.$ and 5a), $1.50-1.10$ (m, $34 \mathrm{H},\left[\mathrm{CH}_{2}\right]_{15}, \mathrm{H} 3 \mathrm{~b}, 4 \mathrm{~b}, 5 \mathrm{~b}$ and 6 b$), 0.88\left(\mathrm{t}, 3 \mathrm{H}, J=6.8 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $73.8(\mathrm{C} 2), 59.8(\mathrm{C} 1), 53.5\left(\mathrm{SO}_{2} \mathrm{CH}_{2}\right), 34.1(\mathrm{C} 3$ or 6), 31.9, 29.7, 29.6, 29.5, 29.3, 29.1, 28.3, 24.8 (C4 or 5), $24.0\left(\mathrm{C} 4\right.$ or 5), 23.7, $14.1\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right) ;$ HRMS (ESI) calcd. for $\mathrm{C}_{24} \mathrm{H}_{50} \mathrm{NO}_{3} \mathrm{~S}[\mathrm{M}+\mathrm{H}]^{+}$ 432.3506, found 432.3487.
$N$-(1R,2R)-2-O-(2-Azido-3,4,6-tri-O-benzyl-2-deoxy-D-glucopyranosyl)-cyclohexyloctadecane-

## 1-sulphonamide 35

A mixture of trichloroacetimidate $\mathbf{3 4}^{20}(130 \mathrm{mg}, 0.21 \mathrm{mmol})$, acceptor $\mathbf{3 3}$ ( $109 \mathrm{mg}, 0.25$ mmol ) and activated $4 \AA$ molecular sieves ( 200 mg ) in dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}(15 \mathrm{~mL})$ was stirred under argon at room temperature for 15 min . Then TMSOTf ( $5.3 \mu \mathrm{~L}, 0.029 \mathrm{mmol}$ ) was added and the solution was stirred at room temperature for an additional 2 h . It was then percolated through a short column of silica gel (elution with EtOAc ) and the eluent was concentrated under reduced pressure. RBC (elution gradient 1:5 $\rightarrow$ 1:3 EtOAc - hexane) of the residue gave the pseudodisaccharide 35
(140 $\mathrm{mg}, 75 \%$ ) as an oily mixture of $\alpha, \beta$ anomers in the ratio of $\sim 1: 1$, as determined by ${ }^{1} \mathrm{H}$ NMR spectroscopy; $\delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.40-7.10(30 \mathrm{H}, 6 \times \mathrm{Ph} \alpha$ and $\beta), 5.62(\mathrm{~d}, 1 \mathrm{H}, J=2.5 \mathrm{~Hz}, \mathrm{NH}$ $\alpha$ or $\beta$ ), $5.46(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH} \alpha$ or $\beta), 4.98\left(\mathrm{~d}, 1 \mathrm{H}, J_{1^{\prime}, 2^{\prime}}=3.6 \mathrm{~Hz}, \mathrm{H} 1^{\prime} \alpha\right), 4.90-4.44\left(\mathrm{~m}, 12 \mathrm{H}, 6 \times \mathrm{CH}_{2} \mathrm{Ph}\right.$ $\alpha$ and $\beta), 4.30\left(\mathrm{~d}, 1 \mathrm{H}, J_{1^{\prime}, 2^{\prime}}=8.0 \mathrm{~Hz}, \mathrm{H}^{\prime} \beta\right.$ ), $3.91(\mathrm{~m}, 2 \mathrm{H}), 3.77-3.57(\mathrm{~m}, 7 \mathrm{H}), 3.47(\mathrm{~m}, 1 \mathrm{H}), 3.42-$ $3.27(\mathrm{~m}, 4 \mathrm{H}), 3.17(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H} 1 \alpha / \beta$ or $2 \alpha / \beta), 3.10-2.96\left(\mathrm{~m}, 4 \mathrm{H}, 2 \times \mathrm{XO}_{2} \mathrm{CH}_{2}\right), 2.37(\mathrm{~m}, 2 \mathrm{H}$, cyclitol), 2.13 ( $\mathrm{m}, 2 \mathrm{H}$, cyclitol), $1.90-1.18$ ( $\alpha$ and $\beta \mathrm{SO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2},\left[\mathrm{CH}_{2}\right]_{15}$ and H's cyclitol), $0.88(\mathrm{t}$, $6 \mathrm{H}, J=7.1 \mathrm{~Hz}, 2 \times \mathrm{CH}_{2} \mathrm{CH}_{3} \alpha$ and $\beta$ ); $\delta_{\mathrm{C}}\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 137.9,137.8,137.6,128.5-127.7$ (Ph), $101.2\left(\mathrm{C}^{\prime} \beta\right), 99.6\left(\mathrm{Cl}^{\prime} \alpha\right), 83.7,82.7,82.3,81.4,78.1,77.5,75.8,75.6,75.3,75.1,74.9,73.6$, $73.5,71.4,68.6,68.2,66.2,64.7,57.9(\mathrm{C} 1 \alpha / \beta$ or $\mathrm{C} 2 \alpha / \beta), 57.5,53.6,51.9,34.0,32.3,32.0,31.9$, $31.4,29.7,29.5,29.4,29.3,28.6,28.5,24.2,23.8,23.5,23.4,22.7,14.2\left(\mathrm{CH}_{2} \mathrm{CH}_{3} \alpha\right.$ and $\left.\beta\right)$; HRMS (ESI) calcd. for $\mathrm{C}_{51} \mathrm{H}_{76} \mathrm{~N}_{4} \mathrm{O}_{7} \mathrm{SNa}[\mathrm{M}+\mathrm{Na}]^{+} 911.5372$, found 911.5282.
$N$-(1R,2R)-2-O-(2-Amino-2-deoxy- $\beta$-D-glucopyranosyl)-cyclohexyloctadecane-1-sulphonamide 11 and the $\alpha$-anomer 12

A solution of the anomeric mixture $35(108 \mathrm{mg}, 0.12 \mathrm{mmol})$ in 5:1 THF - $\mathrm{AcOH}(6 \mathrm{~mL})$ containing $10-20 \% \mathrm{Pd}(\mathrm{OH})_{2}$ on carbon ( 25 mg ) was stirred under a slight over pressure of hydrogen at room temperature for 24 h before it was filtered through a bed of Celite. The catalyst was further washed with $1: 1 \mathrm{THF}-\mathrm{MeOH}(2 \times 10 \mathrm{~mL})$ and the washings were combined and concentrated under reduced pressure. Column chromatography $\left(9: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}\right)$ gave first the $\beta$ anomer $11(9.8 \mathrm{mg}, 14 \%)$ as a waxy solid; $[\alpha]_{\mathrm{D}}^{25}+2.5(c 0.98, \mathrm{MeOH}) ; \delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{MeOH}-\mathrm{d}_{4}\right)$ $4.44\left(\mathrm{~d}, 1 \mathrm{H}, J_{1^{\prime}, 2^{\prime}}=8.1 \mathrm{~Hz}, \mathrm{H}^{\prime}\right), 3.94\left(\mathrm{dd}, 1 \mathrm{H}, J_{5^{\prime}, 6^{\prime} \mathrm{a}}=2.1, J_{6 \mathrm{a}^{\prime}, 6^{\prime} \mathrm{b}}=11.7 \mathrm{~Hz}, \mathrm{H} 6^{\prime} \mathrm{a}\right), 3.65-3.50(\mathrm{~m}$, $2 \mathrm{H}, \mathrm{H} 1$ or 2 and $\left.6^{\prime} \mathrm{b}\right), 3.34\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H} 3^{\prime}\right.$ and $\left.5^{\prime}\right), 3.22\left(\mathrm{q}, 1 \mathrm{H}, J_{3^{\prime}, 4^{\prime}}=J_{4^{\prime}, 5^{\prime}}=9.2 \mathrm{~Hz}, \mathrm{H} 4^{\prime}\right), 3.09(\mathrm{~m}, 3 \mathrm{H}$, H 1 or 2 and $\left.\mathrm{SO}_{2} \mathrm{CH}_{2}\right), 2.63\left(\mathrm{t}, 1 \mathrm{H}, J_{2^{\prime}, 3^{\prime}}=9.0 \mathrm{~Hz}, \mathrm{H} 2^{\prime}\right), 2.13(\mathrm{~m}, 2 \mathrm{H}$, cyclitol $), 1.90-1.20(38 \mathrm{H}$, $\mathrm{SO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2},\left[\mathrm{CH}_{2}\right]_{15}$ and 6 H cyclitol), $0.89\left(\mathrm{t}, 3 \mathrm{H}, J=7.0 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(125 \mathrm{MHz}, \mathrm{MeOH}-\mathrm{d}_{4}\right)$ $100.7\left(\mathrm{C}^{\prime}\right), 80.5(\mathrm{C} 1$ or 2$), 78.5\left(\mathrm{C}^{\prime}\right.$ or $\left.5^{\prime}\right), 72.2\left(\mathrm{C}^{\prime}\right), 62.9\left(\mathrm{C}^{\prime}\right), 58.1\left(\mathrm{C} 1\right.$ or 2 or $\left.2^{\prime}\right), 58.0(\mathrm{C} 1$ or 2 or $2^{\prime}$ ), $54.1\left(\mathrm{SO}_{2} \mathrm{CH}_{2}\right), 35.4,32.2,30.8,30.6,30.5,25.4,25.0,24.7,14.5\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right) ;$ HRMS (ESI)
calcd. for $\mathrm{C}_{30} \mathrm{H}_{61} \mathrm{~N}_{2} \mathrm{O}_{7} \mathrm{~S}[\mathrm{M}+\mathrm{H}]^{+} 593.4194$, found 593.4178. Continued elution gave the $\alpha$ anomer $12(14.5 \mathrm{mg}, 20 \%)$ as an oil; $[\alpha]_{\mathrm{D}}^{25}+66.5(c 1.45, \mathrm{MeOH}) ; \delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{MeOH}-\mathrm{d}_{4}\right) 5.28(\mathrm{~d}, 1 \mathrm{H}$, $\left.J_{1^{\prime}, 2^{\prime}}=3.7 \mathrm{~Hz}, \mathrm{H} 1^{\prime}\right), 3.81\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H} 6^{\prime} \mathrm{a}\right), 3.71\left(\mathrm{~m}, 3 \mathrm{H}, \mathrm{H}^{\prime}, 5^{\prime}\right.$ and $\left.6^{\prime} \mathrm{b}\right), 3.42(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H} 1$ or 2$), 3.34$ $\left(\mathrm{m}, 1 \mathrm{H}, \mathrm{H}^{\prime}\right), 3.23(\mathrm{~m}, \mathrm{H} 1$ or 2$), 3.08\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}^{\prime}\right.$ and $\left.\mathrm{SO}_{2} \mathrm{CH}_{2}\right), 2.24(\mathrm{~m}, 1 \mathrm{H}$, cyclitol), $1.96(\mathrm{~m}, 1 \mathrm{H}$, cyclitol), $1.84-1.67\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{SO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2}\right.$ and 2 H cyclitol), $1.52-1.21\left(34 \mathrm{H},\left[\mathrm{CH}_{2}\right]_{15}\right.$ and 4 H cyclitol), $0.90\left(\mathrm{t}, 3 \mathrm{H}, J=7.0 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(125 \mathrm{MHz}, \mathrm{MeOH}-\mathrm{d}_{4}\right) 97.3\left(\mathrm{Cl}^{\prime}\right), 81.4(\mathrm{C} 1$ or 2$)$, $73.2\left(\mathrm{C} 3^{\prime}\right.$ or $\left.5^{\prime}\right), 70.4\left(\mathrm{C} 3^{\prime}\right.$ or $4^{\prime}$ or $\left.5^{\prime}\right), 70.3\left(\mathrm{C} 3^{\prime}\right.$ or $4^{\prime}$ or $\left.5^{\prime}\right), 60.8\left(\mathrm{C} 6^{\prime}\right), 56.6(\mathrm{C} 1$ or 2$), 54.9\left(\mathrm{C} 2^{\prime}\right)$, $52.6\left(\mathrm{SO}_{2} \mathrm{CH}_{2}\right), 32.7,32.3,31.6,29.4,29.3,29.2,29.0,28.8,27.8,24.2,23.5,23.3,22.3,13.0$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right)$; HRMS (ESI) calcd. for $\mathrm{C}_{30} \mathrm{H}_{61} \mathrm{~N}_{2} \mathrm{O}_{7} \mathrm{~S}[\mathrm{M}+\mathrm{H}]^{+} 593.4194$, found 593.4181.

## trans-2-(4-Methoxybenzyloxy)cyclohexyl acetate 37

$\mathrm{Cu}\left(\mathrm{BF}_{4}\right)_{2} \cdot n \mathrm{H}_{2} \mathrm{O}(42 \mathrm{mg}, 0.18 \mathrm{mmol})$ was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{~mL})$ and cyclohexene oxide $36(1.8 \mathrm{~mL}, 17.8 \mathrm{mmol})$ and 4-methoxybenzyl alcohol ( $10 \mathrm{~mL}, 80.2 \mathrm{mmol}$ ) were added. The reaction mixture was stirred for 24 h , diluted with water $(20 \mathrm{~mL})$, and the aqueous layer extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 30 \mathrm{~mL})$. The combined organic extracts were washed with brine, dried over $\mathrm{MgSO}_{4}$, filtered and the solvent was removed in vacuo. The resulting crude monoprotected PMB cycolhexanediol ${ }^{22}$ was used with no further purification in the next step, whereby it was dissolved in pyridine ( 7.5 mL ), cooled to $0^{\circ} \mathrm{C}$, before DMAP ( $3 \mathrm{mg}, 0.26 \mathrm{mmol}$ ) and acetic anhydride ( 4.5 $\mathrm{mL}, 48.0 \mathrm{mmol}$ ) were added. The reaction mixture was stirred overnight at rt , diluted with water $(100 \mathrm{~mL})$ and extracted with EtOAc (3 x 100 mL$)$. The combined organic extracts were washed with water ( 100 mL ), brine ( 100 mL ), dried over $\mathrm{MgSO}_{4}$, filtered and the solvent was removed under reduced pressure. Column chromatography ( $5: 1$ hexane $-\mathrm{Et}_{2} \mathrm{O}$ ) of the ensuing residue afforded the oily product $37(1.37 \mathrm{~g}, 55 \%) ; \delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.24(\mathrm{~d}, 2 \mathrm{H}, J=8.7 \mathrm{~Hz}, \mathrm{Ph}), 6.86$ (d, 2H, $J=8.6 \mathrm{~Hz}, \mathrm{Ph}), 4.83-4.79(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-1), 4.56-4.49\left(2 \mathrm{x} \mathrm{d}, 2 \mathrm{H}, J=11.7 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{Ar}\right), 3.80$ (s, $3 \mathrm{H}, \mathrm{OCH}_{3}$ ), $3.38-3.33(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-2), 2.04\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$ ), $2.02-1.98(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-3 \mathrm{a}$ and 6 a$), 1.70$ $-1.62(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-4 \mathrm{a}$ and 5 a$), 1.43-1.18(\mathrm{~m}, 4 \mathrm{H}, \mathrm{H}-3 \mathrm{~b}, 4 \mathrm{~b}, 5 \mathrm{~b}$ and 6 b$) ; \delta_{\mathrm{C}}\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 170.5$
$(\mathrm{C}=\mathrm{O}), 159.1,131.0,129.0,114.0,113.7,78.4(\mathrm{C}-2), 75.3(\mathrm{C}-1), 71.0\left(\mathrm{CH}_{2} \mathrm{Ar},\right), 55.3\left(\mathrm{OCH}_{3}\right), 30.0$, 29.9, 23.6, 23.3, $21.4\left(\mathrm{CH}_{3}\right)$; HRMS (ESI) calcd. for $\mathrm{C}_{16} \mathrm{H}_{22} \mathrm{NaO}_{4}[\mathrm{M}+\mathrm{Na}]^{+} 301.1410$, found 301.1397.

## 1-\{[trans-2-(Allyloxy)cyclohexyloxy]methyl\}-4-methoxybenzene 38

The acetate 37 ( $938 \mathrm{mg}, 3.37 \mathrm{mmol}$ ) was dissolved in $\mathrm{MeOH}(10 \mathrm{~mL})$ and $\mathrm{NaOMe}(5.4 \mathrm{M}$ in $\mathrm{MeOH}, 150 \mu \mathrm{~L}$ ) was added and the solution stirred for 1 h at rt . Afterwards, TLC revealed that there was still the presence of $\mathbf{3 7}$ and, thus, a further aliquot of $\mathrm{NaOMe}(5.4 \mathrm{M}$ in $\mathrm{MeOH}, 100 \mu \mathrm{~L}$ ) was added and the reaction mixture was stirred for an additional 24 h . After which, the reaction was neutralised with Amberlite IR-120 $\left(\mathrm{H}^{+}\right)$ion-exchange resin, filtered and the crude solution was passed down a short plug of silica gel (elution with EtOAc) to afford the known 2-PMB protected alcohol $^{22}$ as a pale yellow oil which was used without further purification in the next step. To a stirred and cooled $\left(0^{\circ} \mathrm{C}\right)$ solution of the alcohol ${ }^{22}(886 \mathrm{mg}, 3.75 \mathrm{mmol}$,$) in DMF (40 \mathrm{~mL})$ was added $\mathrm{NaH}(60 \%$ dispersion in mineral oil, $750 \mathrm{mg}, 18.7 \mathrm{mmol}$ ) and the solution was stirred for 30 min before allyl bromide ( $2.92 \mathrm{~mL}, 22.8 \mathrm{mmol}$ ) was added dropwise. The reaction mixture was stirred under argon for a further 18 h at rt , quenched with $\mathrm{MeOH}(50 \mathrm{~mL}), \mathrm{H}_{2} \mathrm{O}(250 \mathrm{~mL})$ was added and then the resulting solution was extracted with EtOAc (3 x 250 mL ). The combined organic extracts were washed with $\mathrm{H}_{2} \mathrm{O}(3 \times 250 \mathrm{~mL})$, brine ( 250 mL ), dried over $\mathrm{MgSO}_{4}$, filtered and the solvent was removed under reduced pressure. The resulting residue was passed down a short plug of silica gel (elution with EtOAc) and, after evaporation to dryness, purified by RBC (elution gradient hexane $\rightarrow 5: 1 \mathrm{Et}_{2} \mathrm{O}$ - hexane) to afford the allyl product $38(712 \mathrm{mg}, 69 \%)$ as a clear oil; $\delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.32(\mathrm{~d}, 2 \mathrm{H}, J=8.6 \mathrm{~Hz}, \mathrm{Ph}), 6.90(\mathrm{~d}, 2 \mathrm{H}, J=8.7 \mathrm{~Hz}, \mathrm{Ph}), 6.02-$ $5.94\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}=\mathrm{CH}_{2}\right), 5.32-5.17\left(4 \mathrm{x} \mathrm{m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}=\mathrm{CH}_{2}\right), 4.63(\mathrm{dd}, 2 \mathrm{H}, J=11.4 \mathrm{~Hz}$, $\left.\mathrm{CH}_{2} \mathrm{Ar}\right), 4.17\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}=\mathrm{CH}_{2}\right), 3.83\left(\mathrm{~s}, \mathrm{OCH}_{3}\right), 3.37-3.29(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-1$ and 2), 2.03-2.00 (m, 2 H , cyclitol), 1.69-1.67 ( $\mathrm{m}, 2 \mathrm{H}$, cyclitol), 1.39-1.20 ( $\mathrm{m}, 4 \mathrm{H}$, cyclitol); $\delta_{\mathrm{C}}\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right.$ ) 159.0, $135.8\left(\mathrm{CH}_{2} \mathrm{CH}=\mathrm{CH}_{2}\right), 131.5,129.1,116.1\left(\mathrm{CH}_{2} \mathrm{CH}=\mathrm{CH}_{2}\right), 113.7,81.0(\mathrm{C}-1$ or 2$), 80.8(\mathrm{C}-1$ or 2$)$,
$71.6\left(\mathrm{CH}_{2} \mathrm{CH}=\mathrm{CH}_{2}\right), 71.0\left(\mathrm{CH}_{2} \mathrm{Ar}\right), 55.3\left(\mathrm{OCH}_{3}\right), 30.33,30.31,23.59,23.58$; HRMS (ESI) calcd. for $\mathrm{C}_{17} \mathrm{H}_{24} \mathrm{NaO}_{3}[\mathrm{M}+\mathrm{Na}]^{+}$299.1618, found 299.1607.

## 2-\{[trans-2-((4-Methoxybenzyl)oxy)cyclohexyloxy]methyl\} oxirane 39

Compound $38(2.22 \mathrm{~g}, 8.02 \mathrm{mmol})$ was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(30 \mathrm{~mL})$ and $m \mathrm{CPBA}(4.15 \mathrm{~g}$, 24.1 mmol ) was added and the reaction mixture stirred for 18 h at rt . Afterwards, the reaction mixture was washed successively with $10 \%$ aq sodium sulfite ( 80 mL ), water ( 100 mL ), $10 \% \mathrm{aq}$ $\mathrm{NaOH}(80 \mathrm{~mL})$ and brine $(80 \mathrm{~mL})$. The organic phase was then filtered through cotton wool and the solvent was removed in vacuo. The crude material was passed down a short plug of silica gel (further elution with EtOAc ) and evaporated to dryness under reduced pressure. RBC (elution gradient 1:1 $\rightarrow 2: 1 \mathrm{Et}_{2} \mathrm{O}$ - hexane) furnished the epoxide $39(1.50 \mathrm{~g}, 64 \%)$ as a clear oil; $\delta_{\mathrm{H}}(500$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.20(\mathrm{dd}, 2 \mathrm{H}, J=8.7 \mathrm{~Hz}, \mathrm{Ph}), 6.77(\mathrm{~d}, 2 \mathrm{H}, J=8.7 \mathrm{~Hz}, \mathrm{Ph}), 4.50\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Ar}\right) 3.76$ - $3.40\left(5 \mathrm{H}, \mathrm{OCH}_{3}\right.$ and 1- or $3-\mathrm{CH}_{2}$ propyl), $3.25(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-1$ and 2$), 3.05(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-2$ propyl), 2.67 $-2.52\left(2 \mathrm{H}, 1-\right.$ or $3-\mathrm{CH}_{2}$ propyl), $1.90(\mathrm{~m}, 2 \mathrm{H}$, cyclitol), $1.57(\mathrm{~m}, 2 \mathrm{H}$, cyclitol), 1.22-1.06 (m, 4H, cyclitol); $\delta_{\mathrm{C}}\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right.$ ) 159.0, 131.4, 131.3, 129.14, 129.08, 113.7, 82.3 (C-1 or 2), 82.1 (C-1 or 2 ), $80.8(\mathrm{C}-1$ or 2$), 71.5\left(\mathrm{CH}_{2} \mathrm{Ar}\right), 71.2$ (1- or $3-\mathrm{CH}_{2}$ propyl), 70.4 (1- or 3- $\mathrm{CH}_{2}$ propyl), $55.2\left(\mathrm{OCH}_{3}\right), 51.3$ (C-2 propyl), 51.1 (C-2 propyl), 44.44 (1- or 3-CH2 propyl), 44.38 (1- or 3- $\mathrm{CH}_{2}$ propyl), $30.3,30.20,30.16,23.6,23.5$; HRMS (ESI) calcd. for $\mathrm{C}_{17} \mathrm{H}_{24} \mathrm{NaO}_{4}[\mathrm{M}+\mathrm{Na}]^{+}$315.1567, found 315.1553 .

## 3-\{[trans-2-((4-Methoxybenzyl)oxy)cyclohexyl]oxy\}propane-1,2-diol 40

To a solution of $\mathbf{3 9}(1.50 \mathrm{~g}, 5.13 \mathrm{mmol})$ in DMSO $(56.4 \mathrm{~mL})$ was added water $(10.8 \mathrm{~mL})$ and aq $0.3 \mathrm{M} \mathrm{KOH}(2.4 \mathrm{~mL})$. The reaction mixture was heated to $100^{\circ} \mathrm{C}$ for 18 h , and then diluted with water ( 200 mL ) followed by extraction with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 200 \mathrm{~mL})$. The combined organic extracts were washed with water $(100 \mathrm{~mL})$, brine $(100 \mathrm{~mL})$, dried over $\mathrm{MgSO}_{4}$, filtered and the solvent was removed under reduced pressure. The residue, so obtained, was percholated through a short column of silica gel (further elution with EtOAc ) and the subsequent eluent was concentrated under reduced pressure. RBC (elution first with hexane $\rightarrow 6: 1 \mathrm{EtOAc}$ - hexane) of the residue
afforded the diol $40(942 \mathrm{mg}, 65)$ as a clear oil; $\delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.28(\mathrm{~d}, 2 \mathrm{H}, J=8.6 \mathrm{~Hz}, \mathrm{Ph})$, $6.88(\mathrm{dd}, 2 \mathrm{H}, \mathrm{CH}, \mathrm{J}=8.6, \mathrm{Ph}), 4.59-4.50\left(2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Ar}\right), 3.82-3.47\left(8 \mathrm{H}, \mathrm{OCH}_{3}, \mathrm{H}-2\right.$ propyl, 1and $3-\mathrm{CH}_{2}$ propyl), $3.25(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-1$ and 2$), 2.50(\mathrm{bs}, 1 \mathrm{H}, \mathrm{OH}), 2.40(\mathrm{bs}, 1 \mathrm{H}, \mathrm{OH}), 2.11-2.01(\mathrm{~m}$, 2 H , cyclitol), 1.68 ( $\mathrm{m}, 2 \mathrm{H}$, cyclitol), 1.23 ( $\mathrm{m}, 4 \mathrm{H}$, cyclitol); $\delta_{\mathrm{C}}\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right.$ ) 159.3, 159.2, $130.5,130.4,129.5,129.4,113.84,113.83,83.40(\mathrm{C}-1$ or 2$), 82.39(\mathrm{CH}, \mathrm{C}-1$ or 2$), 80.86$ (C-1 or 2 ), 80.78 (C-1 or 2), 72.31 (1- or 3- $\mathrm{CH}_{2}$ propyl), 71.37 (C-2 propyl), 71.03 (1- or 3- $\mathrm{CH}_{2}$ propyl), 70.81 (1- or 3-CH2 propyl), 70.51 (C-2 propyl), 64.20 (1- or $3-\mathrm{CH}_{2}$ propyl), 63.83 (1- or $3-\mathrm{CH}_{2}$ propyl), $55.24\left(\mathrm{OCH}_{3}\right), 30.75,30.54,29.99,29.93,23.78,23.71,23.69$; HRMS (ESI) calcd. for $\mathrm{C}_{17} \mathrm{H}_{26} \mathrm{NaO}_{5}$ $[\mathrm{M}+\mathrm{Na}]^{+}$333.1672, found 333.1678.

1-[(tert-Butyldiphenylsilyl)oxy]-3-\{[trans-2-((4-methoxybenzyl)oxy)cyclohexyl]oxy\}propan-2ol 41

To a solution of the primary alcohol $40(942 \mathrm{mg}, 3.34 \mathrm{mmol})$ and DIPA ( $5.8 \mathrm{~mL}, 3.75$ mmol) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ was added TBDPSCl $(1.04 \mathrm{~mL}, 4.00 \mathrm{mmol})$ dropwise followed by DMAP ( $5 \mathrm{mg}, 0.038 \mathrm{mmol}$ ) and the reaction stirred for 24 h at rt . Afterwards, TLC revealed the presence of the starting material $\mathbf{4 0}$; thus an additional aliquot of $\operatorname{TBDPSCl}(0.521 \mathrm{~mL}, 2.00 \mathrm{mmol})$ was added and the reaction mixture was stirred for a further 3 h , whereafter it was quenched with water ( 60 mL ) and then extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 60 \mathrm{~mL})$. The combined organic extracts were washed successively with water ( 100 mL ), brine ( 50 mL ), filtered through cotton wool and the solvent was removed under reduced pressure. The residue so obtained was percolated through a short column of silica gel (further elution with EtOAc) and the subsequent eluent was concentrated under reduced pressure. RBC (elution first with hexane $\rightarrow 1: 2 \mathrm{EtOAc}$ - hexane) of the residue afforded the silyl protected product $41(1.33 \mathrm{~g}, 76 \%)$ as a pale yellow oil; $\delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.68-6.79(14 \mathrm{H}$, $\mathrm{Ph})$, 4.56-4.44 (m, $2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Ar}$ ), $3.90-3.51\left(8 \mathrm{H}, \mathrm{OCH}_{3}, \mathrm{H}-2\right.$ propyl, 1- and $3-\mathrm{CH}_{2}$ propyl), $3.25(\mathrm{~m}$, $2 \mathrm{H}, \mathrm{H}-1$ and 2 ), $3.12(\mathrm{bs}, 1 \mathrm{H}, \mathrm{OH}), 1.97(\mathrm{~m}, 2 \mathrm{H}$, cyclitol), $1.66(\mathrm{~m}, 2 \mathrm{H}$, cyclitol), 1.30-1.15 (m, 4H, cyclitol), 1.05 (s, $9 \mathrm{H}, 3 \times \mathrm{CH}_{3}$ ); $\delta_{\mathrm{C}}\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right.$ ) 159.1, 135.6, 135.6, 134.7, 133.5, 133.49, $133.42,130.9,130.8,129.73,129.72,129.3,129.2,127.74,127.71,113.79,113.77,82.9$ (C-1 or 2 ),
82.3 (C-1 or 2), 80.7 (C-1 or 2), 80.6 (C-1 or 2), 71.7, 71.6, 71.2, 71.1, 71.0, 70.4, 65.0, 64.8, 55.3, 55.2, 30.6, 30.4, 30.11, 30.06, 29.7, 26.9, 23.69, 23.66, 19.29, 19.28; HRMS (ESI) calcd. for $\mathrm{C}_{33} \mathrm{H}_{44} \mathrm{NaO}_{5} \mathrm{Si}[\mathrm{M}+\mathrm{Na}]^{+} 571.2856$, found 571.2860 .

1-[(tert-Butyldiphenylsilyl)oxy]-3-\{(trans-2-[(4-methoxybenzyl) oxy)cyclohexyl]oxy\}propan-2yl methanesulfonate 42

To the secondary alcohol $41(1.33 \mathrm{~g}, 2.54 \mathrm{mmol})$ in pyridine ( 5 mL ) was added mesyl chloride ( $0.63 \mathrm{~mL}, 8.14 \mathrm{mmol}$ ) dropwise. The reaction mixture was stirred at room temperature for 24 h and then quenched with saturated $\mathrm{NaHCO}_{3}(10 \mathrm{~mL})$ followed by extraction with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \mathrm{x}$ $20 \mathrm{~mL})$. The combined organic extracts were washed with water ( 60 mL ), brine ( 60 mL ), filtered through cotton wool and then evaporated to dryness; whereafter toluene ( 5 mL ) was added and evaporated therefrom. The resulting oil was passed through a short column of silica gel (elution with EtOAc) to give a clear yellow oil after evaporation to dryness under reduced pressure. This oil was used in the following step without further purification. However, a small sample of the product $(100 \mathrm{mg})$ was purified by $\operatorname{RBC}\left(1: 1 \mathrm{Et}_{2} \mathrm{O}\right.$ - hexane, $)$ to afford a clear, colourless oil of $\mathbf{4 2}$ for analytical analyses; $\delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.60-6.73(14 \mathrm{H}, \mathrm{Ph}), 4.70(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-2$ propyl), 4.61$4.35\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Ar}\right), 3.83-3.69\left(\mathrm{~m}, 7 \mathrm{H}, \mathrm{OCH}_{3}, 1-\right.$ and $3-\mathrm{CH}_{2}$ propyl), 3.18 ( $\mathrm{m}, 2 \mathrm{H}, \mathrm{H}-1$ and 2 ), $2.90\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{SO}_{2} \mathrm{CH}_{3}\right), 1.90(\mathrm{~m}, 2 \mathrm{H}$, cyclitol), $1.57(\mathrm{~m}, 2 \mathrm{H}$, cyclitol), $1.24-1.08(\mathrm{~m}, 4 \mathrm{H}$, cyclitol), $0.98\left(\mathrm{~s}, 9 \mathrm{H}, 3 \times \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 159.1,135.6,135.54,135.52,132.94,132.89,132.8$, 131.1, 131.0, 129.92, 129.90, 129.2, 129.1, 127.9, 113.78, 113.77, 82.5 (CH, CH propyl), 82.4 (CH, C-1 or 2), 82.1 (CH, C-1 or 2), 81.9 (CH, C-1 or 2), 80.5 (CH, C9), 71.2, 71.0, 68.9, 68.8, 63.5, $55.3\left(\mathrm{OCH}_{3}\right), 38.5\left(\mathrm{SO}_{2} \mathrm{CH}_{3}\right), 38.4,30.0,30.0,29.8,26.8,23.5,23.4,19.2$; HRMS (ESI) calcd. for $\mathrm{C}_{34} \mathrm{H}_{46} \mathrm{NaO}_{7} \mathrm{SSi}[\mathrm{M}+\mathrm{Na}]^{+}$649.2626, found 649.2628 .
\{2-Azido-3-[(trans-2-((4-methoxybenzyl)oxy)cyclohexyl)oxy] propoxy\} (tertbutyl)diphenylsilane 43

A solution of the mesylate $\mathbf{4 2}$ in DMF ( 10 mL ) containing sodium azide ( $496 \mathrm{mg}, 7.64$ $\mathrm{mmol})$ was heated and stirred at $125^{\circ} \mathrm{C}$ for 24 h , cooled and then poured into water ( 40 mL ). The
resulting aqueous solution was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 40 \mathrm{~mL})$ and the combined organic extracts were washed successively with water $(100 \mathrm{~mL})$, brine $(100 \mathrm{~mL})$, filtered through cotton wool and concentrated under reduced pressure. A solution of the residue in EtOAc was percolated through a short column of silica gel (elution with EtOAc) and the eluent concentrated under reduced pressure. RBC of the residue (elution first with hexane $\rightarrow 1: 1 \mathrm{Et}_{2} \mathrm{O}$ - hexane) gave the azide 43 $(1.30 \mathrm{~g}, 51 \%)$ as a clear oil; $\delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.62-6.73(14 \mathrm{H}, \mathrm{Ph}), 4.44\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Ar}\right), 3.71$ $-3.32\left(8 \mathrm{H}, \mathrm{OCH}_{3}, \mathrm{H}-2\right.$ propyl, 1- and $3-\mathrm{CH}_{2}$ propyl), $3.22(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-1$ and 2$), 1.86(\mathrm{~m}, 2 \mathrm{H}$, cyclitol), 1.55 ( $\mathrm{m}, 2 \mathrm{H}$, cyclitol), $1.25-1.04$ ( $\mathrm{m}, 4 \mathrm{H}$, cyclitol), $0.99\left(\mathrm{~s}, 9 \mathrm{H}, 3 \times \mathrm{CH}_{3}\right.$ ); $\delta_{\mathrm{C}}(125 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 159.0,135.9,135.8,135.6,135.3,134.8,133.10,133.06,131.31,131.29,129.8,129.7$, 129.2, 129.12, 129.08, 129.0, 127.8, 127.73, 127.67, 113.74, 113.70, 82.2 (C-1 or 2), 82.0 (C-1 or 2), 80.5 (C-1 or 2 ), 80.4 (C-1 or 2 ), 71.5, 71.4, 71.19, 69.5, 69.0, 64.08, 64.05, 63.2, 63.0, 55.3 $\left(\mathrm{OCH}_{3}\right), 30.0,29.9,26.9,26.8,26.6,23.43,23.40,23.35,19.2 ;$ HRMS (ESI) calcd. for $\mathrm{C}_{33} \mathrm{H}_{43} \mathrm{~N}_{3} \mathrm{NaO}_{4} \mathrm{Si}[\mathrm{M}+\mathrm{Na}]^{+} 596.2915$, found 596.2926 .

## trans-2-\{2-Azido-3-[(tert-butyldiphenylsilyl)oxy]propoxy\}cyclohexanol 44

The PMB derivative 43 ( $131 \mathrm{mg}, 0.240 \mathrm{mmol}$ ) was dissolved in a solution of $1 \% \mathrm{TFA}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(8.62 \mathrm{~mL})$. The reaction mixture was stirred at room temperature for 18 h ; whereafter an additional aliquot of TFA $(43 \mu \mathrm{~L})$ was added because TLC indicated the presence of the PMB protected starting material (43). After a further 4 h , TLC revealed the absence of any starting material (43) and the reaction mixture was diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(40 \mathrm{~mL})$ and washed with saturated $\mathrm{NaHCO}_{3}(40 \mathrm{~mL})$, water ( 40 mL ), brine ( 40 mL ) and then filtered through cotton wool. The $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ extract was concentrated under reduced pressure and the residue was percolated through a short column of silica gel (elution with EtOAc) and concentrated to dryness under reduced pressure. RBC purification of the residue (elution first with hexane $\rightarrow 1: 1 \mathrm{Et}_{2} \mathrm{O}$ - hexane) afforded the alcohol $44(67 \mathrm{mg}, 64 \%)$ as a clear oil; $\delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.60-7.31(10 \mathrm{H}, \mathrm{Ph}), 3.74-3.11$ ( $6 \mathrm{H}, \mathrm{H}-1$ or 2, H-2 propyl, 1- and 3-CH2 propyl), $2.99(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-1$ or 2$), 2.60(\mathrm{bd}, 1 \mathrm{H}, \mathrm{OH}), 1.96$ ( $\mathrm{m}, 2 \mathrm{H}$, cyclitol), 1.65 ( $\mathrm{m}, 2 \mathrm{H}$, cyclitol), 1.23-1.03 (m, 4H, cyclitol), $1.00\left(\mathrm{~s}, 9 \mathrm{H}, 3 \times \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}(125$
$\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) 134.75,134.58,134.56,132.2,131.91,131.87,128.9,126.81,126.79,83.6$ (C-1 or 2), 83.5 (C-1 or 2 ), $72.8,72.7,67.3,67.0,62.8,62.7,62.0,61.7,31.02,30.98,28.7,28.1,25.9,25.7$, 23.1, 22.9, 18.2; HRMS (ESI) calcd. for $\mathrm{C}_{25} \mathrm{H}_{35} \mathrm{~N}_{3} \mathrm{NaO}_{3} \mathrm{Si}[\mathrm{M}+\mathrm{Na}]^{+} 476.2340$, found 476.2351. Triethylammonium trans-2-\{2-azido-3-[(tert-butyldiphenylsilyl)oxy]propoxy\}cyclohexyl $\boldsymbol{n}$ octadecyl phosphate 45

This compound was obtained from the alcohol $44(280 \mathrm{mg}, 0.62 \mathrm{mmol})$ and the hydrogenphosphonate TEA salt $\mathbf{1 8}^{15}(537 \mathrm{mg}, 1.23 \mathrm{mmol})$ in the presence of pivaloyl chloride $(0.48 \mathrm{~mL}, 3.86 \mathrm{mmol})$ essentially as described for the TEA salt 19. After oxidation with iodine ( $623 \mathrm{mg}, 2.47 \mathrm{mmol}$ ) in 9:1 pyridine - water followed by the same aqueous workup as described for 19, column chromatography $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} \rightarrow 8: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}\right)$ of the residue afforded the octadecyl phosphate TEA salt $45(276 \mathrm{mg}, 50 \%)$ as a yellow paste; $\delta_{\mathrm{H}}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.63-7.30(10 \mathrm{H}$, $\mathrm{Ph}), 4.10(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-1$ or 2$), 3.92-3.33\left(7 \mathrm{H}, \mathrm{OCH}_{2}, \mathrm{H}-2\right.$ propyl, 1- and $3-\mathrm{CH}_{2}$ propyl), $3.28(\mathrm{~m}, 1 \mathrm{H}$, $\mathrm{H}-1$ or 2$), 3.00\left(\mathrm{~m}, 6 \mathrm{H}, 3 \times \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 1.95-1.73(\mathrm{~m}, 2 \mathrm{H}$, cyclitol $), 1.58-1.46\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{OCH}_{2} \mathrm{CH}_{2}\right.$ and 2 H cyclitol), $1.35-1.14\left(43 \mathrm{H},\left[\mathrm{CH}_{2}\right]_{15}, 3 \times \mathrm{CH}_{2} \mathrm{CH}_{3}\right.$ and 4 H cyclitol), $1.73\left(\mathrm{~s}, 9 \mathrm{H}, 3 \times \mathrm{CH}_{3}\right)$, $0.80\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{CH}_{3}, J=6.8 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 134.9,134.8,134.6,132.6,132.4$, 132.1, 132.0, 128.9, 128.8, 126.8, 126.7, 78.8 (C-1 or 2), 78.6 (C-1 or 2), 76.6 (C-1 or 2), 70.8, 70.1, 68.2, 67.7, 65.7, 65.6, 65.5, 63.0, 62.9, $44.5\left[\mathrm{~N}\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right)_{3}\right], 30.9,29.7,29.6,29.5,29.0,28.70$, 28.65, 28.4, 27.2, 26.0, 25.9, 25.7, 24.8, 24.7, 21.7, 21.1, 21.0, 18.2, $14.3\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 8.5$ $\left[\mathrm{N}\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right)_{3}\right] ; \delta_{\mathrm{P}}\left(202 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)-1.2$ (with heteronuclear decoupling); HRMS (ESI) calcd. for $\mathrm{C}_{43} \mathrm{H}_{71} \mathrm{~N}_{3} \mathrm{O}_{6} \mathrm{PSi}\left[\mathrm{M}-\mathrm{NEt}_{3}-\mathrm{H}\right]^{-} 784.4885$, found 784.4759.

Triethylammonium trans-2-(2-azido-3-hydroxypropoxy)cyclohexyl $\boldsymbol{n}$-octadecyl phosphate 46 Compound 45 ( $68 \mathrm{mg}, 0.077 \mathrm{mmol}$ ) was dissolved in THF ( 1 mL ) and 1.0 M TBAF in THF $(153 \mu \mathrm{~L}, 0.15 \mathrm{mmol})$ was added. The reaction mixture was stirred at rt for 16 h and then diluted with water ( 25 mL ), extracted $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 25 \mathrm{~mL})$ and the combined organic extracts were washed with aq 1.0 M TEAB ( $2 \times 10 \mathrm{~mL}$ ). The organic phase was filtered through cotton wool, the solvent was removed under reduced pressure and the resulting residue was purified by column
chromatography ( $8: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}$ ) to afford the alcohol $46(50 \mathrm{mg}, 100 \%)$ as a white paste; $\delta_{\mathrm{H}}$ ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $4.08-3.33\left(8 \mathrm{H}, \mathrm{OCH}_{2}, \mathrm{H}-1\right.$ or $2, \mathrm{H}-2$ propyl, 1- and $3-\mathrm{CH}_{2}$ propyl), $3.25(\mathrm{~m}$, $1 \mathrm{H}, \mathrm{H}-1$ or 2 ), $3.09\left(\mathrm{~m}, 6 \mathrm{H}, 3 \times \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 2.14-1.99(\mathrm{~m}, 2 \mathrm{H}$, cyclitol), $1.66-1.59(\mathrm{~m}, 4 \mathrm{H}$, $\mathrm{OCH}_{2} \mathrm{CH}_{2}$ and 2 H cyclitol), 1.30-1.19 (43H, $\left[\mathrm{CH}_{2}\right]_{15}, 3 \times \mathrm{CH}_{2} \mathrm{CH}_{3}$ and 4 H cyclitol), $0.88(\mathrm{t}, 3 \mathrm{H}$, $\left.\mathrm{CH}_{3}, J=6.5 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{CH}_{3}\right) ; \delta_{\mathrm{C}}\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 81.6(\mathrm{C}-1$ or 2$), 78.6(\mathrm{C}-1$ or 2$), 68.6,68.5,66.03$, $65.99,62.4,61.7,60.9,60.6,45.4\left[\mathrm{~N}\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right)_{3}\right], 32.2,32.0,31.9,30.8,30.74,30.68,29.73,29.67$, 29.42, 29.38, 25.8, 23.9, 23.8, 23.8, 23.7, 22.7, $14.1\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 8.5\left[\mathrm{~N}\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right)_{3}\right] ; \delta_{\mathrm{P}}(202 \mathrm{MHz}$, $\mathrm{CDCl}_{3}$ ) - 1.02 (with heteronuclear decoupling); HRMS (ESI) calcd. for $\mathrm{C}_{27} \mathrm{H}_{53} \mathrm{~N}_{3} \mathrm{O}_{6} \mathrm{P}\left[\mathrm{M}-\mathrm{NEt}_{3}-\right.$ H] 546.3677, found 546.3673.

Triethylammonium trans-2-(2-amino-3-hydroxypropoxy)cyclohexyl n-octadecyl phosphate 13
Pearlman's catalyst [10-20\% $\mathrm{Pd}(\mathrm{OH})_{2}$ on carbon, 15 mg ] was added to a solution of the azide $46(45 \mathrm{mg}, 0.069 \mathrm{mmol})$ in $1: 1 \mathrm{THF}-\mathrm{MeOH}(10 \mathrm{~mL})$ and the mixture was stirred under a hydrogen atmosphere at rt for 2 h . Processing as described for $\mathbf{2 1}$ gave the amino TEA salt $\mathbf{1 3}$ (29 $\mathrm{mg}, 44 \%)$ as a white paste, which did not require any chromatographic purification; $\delta_{\mathrm{H}}(500 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 4.00-3.31\left(8 \mathrm{H}, \mathrm{OCH}_{2}, \mathrm{H}-1\right.$ or 2, H-2 propyl, 1- and $3-\mathrm{CH}_{2}$ propyl), $3.10(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-1$ or 2$)$, $3.02\left(\mathrm{q}, 6 \mathrm{H}, J=7.4 \mathrm{~Hz}, 3 \times \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 2.04-1.93\left(\mathrm{~m}, 2 \mathrm{H}\right.$, cyclitol), $1.63-1.50\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{OCH}_{2} \mathrm{CH}_{2}\right.$ and 2 H cyclitol), 1.33-1.05 $\left(43 \mathrm{H},\left[\mathrm{CH}_{2}\right]_{15}, 3 \times \mathrm{CH}_{2} \mathrm{CH}_{3}\right.$ and 4 H cyclitol), $0.81\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{CH}_{3}, J=6.7\right.$ $\mathrm{Hz}, \mathrm{CH}_{2} \mathrm{CH}_{3}$ ); $\delta_{\mathrm{C}}\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 81.6(\mathrm{C}-1$ or 2$), 77.9(\mathrm{C}-1$ or 2$), 67.1,65.9,64.8,53.2,44.5$ $\left[\mathrm{N}\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right)_{3}\right], 33.0,31.8,30.9,29.7,29.6,29.33,29.29,29.0,28.70,28.65,28.43,28.35,27.9$, 24.9, 24.8, 24.8, 24.7, 23.2, 23.1, 22.8, 22.7, $13.1\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 7.5\left[\mathrm{~N}\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right)_{3}\right] ; \delta_{\mathrm{P}}(202 \mathrm{MHz}$, $\mathrm{CDCl}_{3}$ ) -0.23 (with heteronuclear decoupling); HRMS (ESI) calcd. for $\mathrm{C}_{27} \mathrm{H}_{55} \mathrm{NO}_{6} \mathrm{P}\left[\mathrm{M}-\mathrm{NEt}_{3}-\right.$ $\mathrm{H}]^{-} 520.3772$, found 520.3747 .

## Biological assays

## Materials

The synthesis of 1-D-6-O-(2-amino-2-deoxy- $\alpha$-D-glucopyranosyl)-myo-inositol 1(octadecyl phosphate), (4, $\alpha$-D-Glc $p \mathrm{NH}_{2}-\mathrm{IPC}_{18}$ ), ${ }^{15}$ has been described previously. The corresponding $N$-acetyl derivate $\alpha$-D-Glc $p N A c-I P C_{18}$ (3) was prepared by treatment with acetic anhydride ${ }^{13}$, and the concentration of stock solutions determined by measurement of the inositol content by selected ion-monitoring GC-MS. ${ }^{8}$ Bloodstream form Trypanosome brucei (variant MITat1.4) were isolated and membranes (cell-free system) prepared as described previously and stored at $-80{ }^{\circ} \mathrm{C} .{ }^{26}$

## Activity assays

Substrate recognition assays were performed using 500 pmol of $\alpha$-D-GlcpNAc-PI (1) in incorporation buffer ( 25 mM Tris $\mathrm{pH} 8.0,50 \mathrm{mM} \mathrm{KCl}, 50 \mathrm{mM} \mathrm{MnCl}_{2}$ ) and varying amounts of trypanosome cell-free system ( $0-15 \times 10^{6}$ cell equivalents per assay) in 96 -well plates containing $100 \mu \mathrm{~L}$ final volume, and incubated at $37{ }^{\circ} \mathrm{C}$ for 1 h . The reaction was quenched and the glycolipids enriched and analyzed by LC-MS/MS as described below.

Inhibition assays were performed in 96-well plates in $100 \mu \mathrm{~L}$ final volume, with $1 \% \mathrm{v} / \mathrm{v}$ DMSO with or without inhibitor. Trypanosome cell-free system $\left(2.5 \times 10^{6}\right.$ cell equivalents per assay) in incorporation buffer were added to wells containing $500 \mathrm{pmol} \alpha$-D-GlcpNAc-IPC ${ }_{18}$ (3) with or without inhibitor and incubated at $37{ }^{\circ} \mathrm{C}$ for 1 h . The reaction was quenched and the glycolipids enriched and analyzed by LC-MS/MS as described below.

## Glycolipid enrichment

Enrichment of glycolipids was performed in a 96-well plate format. Reactions were quenched by addition of $200 \mu \mathrm{~L}$ of $5 \%$ propan-1-ol, $5 \mathrm{mM} \mathrm{NH}_{4} \mathrm{OAc}$, and the glycolipids were bound to $\mathrm{C}_{18}$ resin ( 50 mg Isolute Array cartridge), washed three times with $200 \mu \mathrm{~L} 5 \%$ propan-1-ol, $5 \mathrm{mM} \mathrm{NH}_{4} \mathrm{OAc}$ and eluted with $100 \mu \mathrm{~L} 40 \%$ propan-1-ol, $5 \mathrm{mM} \mathrm{NH}_{4} \mathrm{OAc}$ into a 96 -well collection plate.

Prior to subsequent analysis, compound $\mathbf{1 3}$ was dried under nitrogen, resuspended in $\mathrm{MeOH}(100 \mu \mathrm{~L})$ and any free amine reacted with excess $d_{6}-\mathrm{Ac}_{2} \mathrm{O}(1.5 \mu \mathrm{~L})$ in the presence of pyridine $(10 \mu \mathrm{~L})$ for 15 min . The reaction was quenched with water ( $50 \mu \mathrm{~L}$ ), dried under nitrogen and resuspended in $100 \mu \mathrm{~L} 40 \%$ propan-1-ol, $5 \mathrm{mM} \mathrm{NH}_{4} \mathrm{OAc}$.

## Liquid chromatography - tandem mass spectrometery of glycolipids

Glycolipids were analyzed by liquid chromatography coupled to an electrospray tandem mass spectrometer (LC-MS/MS). Samples ( $40 \mu \mathrm{~L}$ ) were injected directly from a 96 -well plate onto a $10 \times 1 \mathrm{~mm} \mathrm{C}_{18}$ column (ACE, $5 \mu \mathrm{M}$ ) and then eluted using a binary gradient of $5-80 \%$ propan-1ol in $5 \mathrm{mM} \mathrm{NH}_{4} \mathrm{OAc}$ (Dionex Ultimate 3000). The gradient consisted of $2 \mathrm{~min} 0 \% \mathrm{~B}, 2-4 \mathrm{~min} 0-$ $100 \%$ B, 4-8 $\mathrm{min} 100 \%$, $8-9 \mathrm{~min} 100-0 \% \mathrm{~B}, 9-10 \mathrm{~min} 0 \%$ B where buffer A consisted of $5 \%$ propan-1-ol, $5 \mathrm{mM} \mathrm{NH}_{4} \mathrm{OAc}$ and buffer B $80 \%$ propan-1-ol, 5 mM NH 44 OAc . The glycolipids were analysed on an electrospray triple quadrapole mass spectrometer (Micromass Quattro Ultima) in multiple reaction monitoring mode.

For each pseudodisaccharide analogue (7-12), standards of the $N$-acetylated compound and corresponding free amine were analyzed separately in order to identify unique transitions for use in subsequent multiple reaction monitoring experiments (Table 1). The ratio of the integrals for these transitions was used to calculate the percentage of substrate conversion to product in a given sample. For compound 13, standards of the $N$-acetylated compound and the $d_{3}-N$-acetylated form were analyzed separately and found to produce a common fragment for use in subsequent multiple reaction monitoring experiments.

For inhibition assays, the turnover of the substrate $\alpha$-D-Glc $p \mathrm{NAc}-\mathrm{IPC} 18$ (3) was used to calculate the percentage of substrate conversion to product in a given sample. ${ }^{10}$ Inhibitor $\mathrm{IC}_{50}$ values were calculated using a four-parameter fit of eight-point potency curves derived from three independent experiments, and are quoted with a standard deviation.

## Trypanosome cell-free system assays

The formation of GPI precursors is monitored by following the incorporation of $\left[{ }^{3} \mathrm{H}\right]-$ mannose and then they were analysed using high-performance liquid chromatography and fluorography as described previously. ${ }^{8}$

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## Supplementary material

Supplementary data (additional experimental procedures and characterisation data for the $\beta$ anomers $\mathbf{8}$ and $\mathbf{1 0}$ plus ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra of all the compounds) associated with this article can be found, in the online version, at XXX.

## References

1. M. A. J. Ferguson, J. Cell Sci., 1999, 112, 2799-2809.
2. D. K. Sharma, T. K. Smith, C. T. Weller, A. Crossman, J. S. Brimacombe and M. A. J.Ferguson, Glycobiology, 1999, 9, 415-422.
3. T. K. Smith, M. J. Patterson, A. Crossman, J. S. Brimacombe and M. A. J. Ferguson, Biochemistry, 2000, 39, 11801-11807.
4. T. K. Smith, A. Crossman, C. N. Borrissow, M. J. Paterson, A. Dix, J. S. Brimacombe and M. A. J. Ferguson, EMBO J., 2001, 20, 3322-3332.
5. M. D. Urbaniak, A. Crossman and M. A. J. Ferguson, Chem. Biol. Drug Des., 2008, 72, 127-132.
6. M. D. Urbaniak, D. V. Yashunsky, A. Crossman, A. V. Nikolaev and M. A. J. Ferguson, ACS Chem. Biol., 2008, 3, 625-634.
7. N. Z. Abdelwahab, A. T. Crossman, M. D. Urbaniak and M. A. J. Ferguson, Carbohydr. Res., 2011, 346, 708-714.
8. N. Z. Abdelwahab, A. T. Crossman, L. Sullivan, M. A. J. Ferguson and M. D. Urbaniak, Chem. Biol. Drug Des., 2012, 79, 270-278.
9. D. K. Sharma, T. K. Smith, A. Crossman, J. S. Brimacombe and M. A. J. Ferguson, Biochem. J., 1997, 328, 171-177.
10. M. D. Urbaniak, A. Crossman, T. Chang, T. K. Smith, D. M. F. v. Aalten and M. A. J. Ferguson, J. Biol. Chem., 2005, 280, 22831-22838.
11. T. Chang, K. G. Milne, M. L. S. Guther, T. K. Smith and M. A. J. Ferguson, J. Biol. Chem., 2002, 277, 50176-50182.
12. N. A. Meanwell, J. Med. Chem., 2011, 54, 2529-2591.
13. T. K. Smith, S. Cottaz, J. S. Brimacombe and M. A. J. Ferguson, J. Biol. Chem., 1996, 271, 6476-6482.
14. G. Grundler and R. R. Schmidt, Liebigs Ann. Chem., 1984, 1826-1847.
15. A. Crossman, M. J. Patterson, M. A. J. Ferguson, T. K. Smith and J. S. Brimacombe, Carbohydr. Res., 2002, 337, 2049-2059.
16. A. V. Nikolaev, I. A. Ivanova, V. N. Shibaev and N. K. Kochetkov, Carbohydr. Res., 1990, 204, 65-78.
17. H. F. Russell, J. B. Bremner, J. Bushelle-Edghill, M. R. Lewis, S. R. Thomas and F. Bates, II, Tetrahedron Lett., 2007, 48, 1637-1639.
18. Y. G. Gololobov, I. N. Zhmurova, L. F. Kasukhin, Tetrahedron, 1981, 37, 437-472.
19. I. Lindh and J. Stawinski, J. Org. Chem., 1989, 54, 1338-1342.
20. S. Cottaz, J. S. Brimacombe and M. A. J. Ferguson, Carbohydr. Res., 1995, 270, 85-91.
21. J. Barluenga, H. Vázquez-Villa, A. Ballesteros, J. M. Gonzalez, Org. Lett. 2002, 4, 28172819.
22. A. Tschöp, A. Marx, A. R. Sreekanth, C. Schneider, Eur. J. Org. Chem. 2007, 14, 23182327.
23. A. S. Capes, A. T. Crossman, L. A. Webster, M. A. J. Ferguson and I. H. Gilbert, Tetrahedron Lett., 2011, 52, 7091-7094.
24. G. Berti, B. Macchia, F. Macchia, Tet. Lett. 1965, 38, 3421-3427.
M. L. S. Guther and M. A. J. Ferguson, EMBO J., 1995, 14, 3080-3093.
25. W. J. Masterson, T. L. Doering, G. W. Hart and P.W.Englund, Cell, 1989, 62, 73-80.
