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Enhancing Stability and Efficiency in Perovskite Solar Cells: Insights into Inorganic HTL Deposition and Interface Defect Passivation

Daniel I Amune^{1*}, W. Fidel², R. Koech³, Dahiru Sanni^{4*}, J. Botsoa², V. Anye^{1,5}, E. Nstoenzok²

1. Department of Materials Science and Engineering, African University of Science and Technology, Abuja, Nigeria
2. CEMHTI-CNRS, Site Cyclotron, Orléans, France
3. Department of Mathematics, Physics, and Computing, Moi University, Eldoret, Kenya
4. Department of Theoretical and Applied Physics, African University of Science and Technology, Abuja, Nigeria
5. Faculty of Engineering, Nile University of Nigeria, Plot 681, Cadastral Zone C-OO, Research & Institution Area Nigeria, Airport Rd, Jabi, 900001, Abuja, Nigeria

*Corresponding authors: damune@aust.edu.ng, dsanni@aust.edu.ng

Abstract: The perovskite solar cell (PSC) as an emerging and promising type of solar cell have been extensively studied, but instability is still a major challenge. Replacing the hygroscopic organic hole transport layer (HTL) in PSC can result in the improvement of the device stability. However, it is still difficult to deposit inorganic HTL onto underlying perovskite layer without eroding or distorting it, in the regular n-i-p architecture thereby inducing defects at the interface and reducing the performance of the device. In this study, performance of PSCs with inorganic HTL is modelled using SCAPS-1D. The perovskite-HTL interface defect density was varied from 1.0×10^{12} to $1.0 \times 10^{20} \text{ cm}^{-3}$. We realized that, for PSCs based on some hole transport materials (HTMs), the effect of interface defect density was not significant. We observed that the HTL/perovskite valence band offset (VBO) play a significant role in the phenomenon observed. Particularly, zero or slightly positive VBO results in the increase in both defect tolerance and device efficiency. This information provides insights into the fabrication of PSCs with improved interface defect passivation and also enable the fabrication of perovskite solar cell based on physically deposited inorganic charge transport materials.

Keywords: Perovskite, Stability, SCAPS 1D, Charge Transport Materials, Interface Defect

1. Introduction

The rapid rise in evolution of perovskite solar cell efficiency and the remarkable properties of perovskite solar cells such as high absorption coefficient, long carrier diffusion length, tunable bandgap and simple production methods have massively attracted researchers into this field, thereby leading to an achievement in high efficiency closer to the Shockley–Queisser limit.(1,2) However, the stability of perovskite solar cell devices is still a major issue. The stability issues arise from internal and external factors including moisture ingress due to the hygroscopic nature of the organic charge transport layers and the perovskite absorber, as well as thermal and photodegradation.(3–5)

These challenges have necessitated the search for alternative materials that will ensure the stability of solar cells, as stability is crucial to commercialization. To date, organic HTMs utilized in PSCs have demonstrated the highest efficiency.(6) Nevertheless, the intrinsic low charge mobility and poor conductivity associated with pristine organic HTMs adversely impact the photovoltaic performance of PSCs. This is primarily due to increased series resistance. and reduced recombination resistance. Dopants such as lithium bis(trifluoromethylsulphonyl)imide



(Li-TFSI) and 4-tert-butylpyridine (TBP) are added to suppress charge recombination and improve the contact between absorber and HTM. (7)

Besides the hygroscopic nature of these organic dopants, the cost of doped hole transport materials (HTMs) is relatively high compared to their inorganic counterparts.(8) Many strategies have been adopted to improve perovskite solar cell stability, including external encapsulation by lamination and internal passivation using organic or conductive polymers.(9) Early perovskite devices without encapsulation have shown stable operation up to hundreds of hours when stored in the dark and measured infrequently. However, these devices rapidly degraded after sustained exposure to sunlight.(10) In addition to light exposure, elevated temperature and humidity may accelerate degradation due to moisture-induced decomposition of perovskite crystals.(11) These stability issues are being addressed through, for example, the use of protective coatings.(12)

The stability of perovskite PV devices under high humidity and temperature conditions has been improved by employing moisture-resistant layers (e.g., carbon nanotubes or graphite) to prevent water ingress.(13–16) Improvements in device stability have also been achieved through encapsulation using glass or plastic lamination, with stability of over 3000 hours at 60 °C being recorded.(17) Furthermore, researchers have incorporated multiple cations, anions, and halides.(18,19) This has led PSCs towards higher stability and efficiency compared to pure archetypical perovskites, which are typically comprised of MAPbX₃, FAPbX₃, and CsPbX₃ (X = Br or I). Rather than pure MAPbI₃, mixed cation/halide perovskites, specifically those based on FAPbI₃, contribute to a slight reduction in the optical bandgap. For example, the bandgap (*E_g*) of FAPbI₃ is approximately 1.502 eV, whereas that of MAPbI₃ is around 1.563 eV. This tailored bandgap, combined with the enhanced environmental stability of FAPbI₃, results in improved performance of perovskite solar cells.(17)

While the addition of buffer layers or encapsulants increases device stability, it also adds to the cost and complexity. Another approach to improve PSC stability is replacing the hygroscopic organic hole transport layer with an inorganic HTL, which has been reported to be more stable than organic counterparts.(20) A major challenge in depositing inorganic HTMs via solution methods onto the active layer in a regular PSC structure is the erosion of the underlying active layer by the inorganic HTM precursor solvent. Researchers have tried to deposit inorganic HTMs onto the perovskite layer and have achieved improved efficiency and stability, but the moisture from the deposition can create degradation pathways that deteriorate device performance over time.(21) Depositing inorganic HTLs via solution method remains challenging because the solvents that ensure good solubility for the inorganic HTL materials typically dissolve the underlying perovskite layer.(22)

Physical deposition methods could provide a better improvement in the device performance and better control of the process parameter to ensure repeatability. Researchers have tried physical deposition methods, but interface defects between the HTL and perovskite layers, resulting from these deposition processes normally causes poor efficiency of the device.(23) High-energy ions from this process normally distort the interface between the HTL and the perovskite, causing an increase in interface defects, thereby increasing the barrier to transport of holes from the absorber to the anode electrode.(24)

To improve the solar cell performance by engineering the interface, emphasis have been placed on optimizing the HTM deposition process in order to minimize the interface defect for improved efficiency, however detailed understanding of the relationship between interface defect and other material properties and how these relationships consequently affect the performance of the solar cell is not well established. Thus, in this work, we focus on the material properties of different inorganic HTMs in order to understand the interplay of these properties and interfacial defects in relation to device performance. We investigated the effect of interface defect (N_t^i) on the efficiency of perovskite solar cell device based on some inorganic HTL by simulation using SCAPS 1D software.

The performance of perovskite solar cells is critically dependent on minimizing interface defects, which act as recombination centers. Emphasis is placed on the intrinsic band alignment between the perovskite absorber and the HTL which plays a central role in modulating charge carrier recombination without relying on external



chemical passivation strategies. In particular, a moderate positive VBO has been theoretically and experimentally linked to enhanced defect tolerance.(25,26) N_i^t

2. Methodology

2.1. Solar Cell Structure for the Study

The solar cell configuration for this study is FTO glass/compact TiO₂/mesoporous TiO₂/Tripple-cation (TC) perovskite absorber layer/Inorganic HTMs/Au. In this n-i-p device structure, the perovskite layer light harvesting material is sandwiched between n-type ETL (TiO₂) and p-type HTL (Inorganic HTL) as shown in *Figure 1a*, where the light passes through the ETL to reach the perovskite layer. The existence of the (ETL) and (HTL) in the PSC structure is essential because these layers play an important role in device performance by modifying the interface, adjusting the interfacial energy level, and enhances the transport of photo-generated charged carriers to the electrodes which in turn reduces the charge recombination.(27,28) The energy band diagram consisting of the various materials considered for this study is shown in *Figure 1b*.

2.2. Numerical Simulation

Numerical analysis describes the basic phenomena present in photovoltaic devices, allowing intuitive examination of each parameter in solar cells and thus identifying the optimal conditions for operating.(29) SCAPS-1D a simulation software developed by the University of Ghent, Belgium was used for this study. The simulation calculations conducted within SCAPS-1D are primarily based on two fundamental equations: the Poisson equation and the continuity equations for both electrons and holes.(30)

Figure 1. (a) Perovskite solar cell architecture (b) Energy bad diagram

$$\frac{dn}{dt} = G - R \quad (2.2)$$

$$\frac{dp}{dt} = G - R \quad (2.3)$$

$$J = J_n + J_p \quad (2.4)$$

$$J_n = qD_n \frac{dn}{dx} + q\mu_n n \frac{d\phi}{dx} \quad (2.5)$$

$$J_p = qD_p \frac{dp}{dx} + q\mu_p p \frac{d\phi}{dx} \quad (2.6)$$

$$\alpha(\lambda) = \left(A + \frac{B}{h\nu}\right) \sqrt{(h\nu - E_g)} \quad (2.7)$$

Where the symbols ϵ_0 and ϵ_r represent the vacuum dielectric constant and the relative dielectric constant, respectively. The variables n and p denote the concentrations of free carriers, specifically electrons and holes. N_D and N_A signify the concentrations of donor and acceptor impurities, respectively. Additionally, ρ_n and ρ_p refer to the charge densities of electrons and holes, while J_n and J_p correspond to the current densities associated with electrons and holes. The parameters R represent the recombination rates for electrons and holes, respectively, whereas G denote the generation rates of electrons or holes. ϕ symbolizes the electric field, q refers to electrical charge with a standard value of approximately 1.602×10^{-19} C. The mobilities of electrons and holes are represented by μ_n and μ_p , respectively. D_p and D_n stand for the diffusion coefficients for free holes and electrons, respectively. The variable ν represents the optical frequency, whereas A and B are arbitrary constants. E_g signifies the bandgap, h represents Planck's constant, and $\alpha(\lambda)$ is the absorption coefficient.

These simulations allow systematic tuning of one variable at a time or multiple variables simultaneously over a wide range that would otherwise be inaccessible experimentally. The simulation parameters in *Tables 1, 2, 3, and 4* below were obtained from previous reports. To accurately capture non-radiative losses originating at the



heterojunctions, defect states were incorporated at both the ETL/perovskite and perovskite/HTL interfaces within the SCAPS-1D structure. In practical devices, interface defects can originate from lattice mismatch, surface roughness, incomplete surface coverage, and structural discontinuities formed during film crystallization. Termination of the perovskite lattice at heterojunctions often results in under-coordinated ions and dangling bonds, which introduce electronic trap states.(31) Mechanical strain and mismatch between adjacent layers may further induce interfacial disorder and defect formation. Additionally, physical deposition processes, particularly sputtering of inorganic transport layers, can cause energetic particle bombardment damage, generating structural defects and increasing non-radiative recombination at the interface.(32–34) These mechanisms predominantly introduce recombination-active trap centers that degrade carrier lifetime and open-circuit voltage.

Although charged defect states may arise from ionized vacancies, antisite defects, or interfacial chemical reactions and can introduce fixed interfacial charge and band bending,(31) the present study intentionally assumes neutral defect states. This modelling choice isolates recombination-dominated losses from electrostatic effects such as band alignment shifts or ion migration phenomena.(35,36) By focusing on neutral recombination-active traps associated primarily with structural disorder rather than chemical degradation, the simulation framework enables direct evaluation of interface recombination sensitivity. This approach complements experimental passivation strategies aimed at chemically neutralizing charged defects and reducing trap densities. Consequently, the combined strategy of optimized band alignment and interface passivation provides a coherent pathway toward further enhancement of device performance.

We represented interface defects by a discrete deep trap level positioned 0.6 eV above the valence band maximum of the perovskite absorber. Deep-level traps located near mid-gap are known to efficiently capture both electrons and holes, thereby accelerating SRH recombination. (37) The electron and hole capture cross-sections were assigned values of $1 \times 10^{-19} \text{ cm}^2$, consistent with commonly reported simulation parameters for perovskite interfaces.(37) These values represent moderate recombination velocity and allow systematic evaluation of recombination kinetics as a function of defect density. The interface defect density (N_i^t) was varied over a wide range 10^{12} – 10^{20} cm^{-3} to represent different interface qualities, spanning well-passivated to severely damaged interface.

We varied N_i^t for different HTMs and the devices I-V characteristics were recorded. The effect of other factors including band offset, operating temperature, and acceptor concentration were also investigated. This approach allows us to assess the importance of each material parameter to the final solar cell performance and to elucidate the limits beyond which power conversion efficiency decreases. The resulting insights are then used to guide new materials selection criteria and design processes.(38)

3. Results and Discussion

3.1. Effect of Interface Defect Density on the Performance of the PSC for different inorganic HTMs

As seen in *Figure 2*, variation in the interface defect density significantly affect the I-V characteristics of the devices based on Cu_2O , CuSCN , or CuI . However, for devices based on CuO or MoS_2 , CuAlO_2 , and CuSbS_2 , it was observed that variation in the interface defect does not have significant effect on the device characteristics, which indicates that the N_i^t is not the main influencing factor for the solar cells' performances. As shown in *Figure 3* and *table 5*, the device based on CuSCN the efficiency, V_{oc} , J_{sc} and FF were reduced from 19.82%, 1.3 V, 23.3 mAcm^{-2} , and 65.3%, respectively, to 14.93 %, 1.14V, 23.05 mAcm^{-2} , and 56.96 %, as the interface defect density was increased from $1 \times 10^{12} \text{ cm}^{-3}$ to $1 \times 10^{20} \text{ cm}^{-3}$. Similarly, trend is observed in device based on CuI where the efficiency, V_{oc} , J_{sc} and FF decreased from 19.53%, 1.3V, 23.3 mAcm^{-2} , and 64.67 %, respectively, to 12.62 %, 1.09 V, 22.96 mAcm^{-2} , and 50.5 %. Similar trend is also observed for Cu_2O . However, devices based on CuO , MoS_2 , CuSbS_2 , and CuAlO_2 exhibits insignificant change in the I-V characteristics as the interface defect density increases from $1 \times 10^{12} \text{ cm}^{-3}$ to $1 \times 10^{20} \text{ cm}^{-3}$. In order to gain thorough insights into this tolerance behavior, we simulated the perovskite solar cell device based on these HTMs by varying the material properties of the HTMs as shown in the following sections.



3.2 Effect of Valence band offset

As the movement of photo-generated charge carriers is typically affected by energy band alignment, the valence band offset can be calculated using the Equation.(39):

$$VBO = (\chi_{HTL} + E_{g(HTL)}) - (\chi_{abs} + E_{g(Abs.)}) \quad (3.1)$$

where, χ_{Abs} and χ_{HTL} , are electron affinities of the absorber and HTL layers, respectively. $E_{g(Abs)}$ and $E_{g(HTL)}$ are the energy band gap values of the absorber and HTL layers, respectively.

The careful selection of the Conduction Band Offset (CBO) and Valence Band Offset (VBO) is essential for the effective transportation of photo-excited electrons and holes to the front electrode and back metal contact, respectively. This selection is critical for minimizing charge recombination within the absorber layer. The band offset can be modified by varying the electron affinity of the Hole Transport Material (HTM) and Electron Transport Material (ETM). In this study, our primary focus is on the effects of the VBO, as we are evaluating different HTMs. The band offsets are very important to define the barrier for transport of photo generated carriers from the absorber to the charge transport layers. Discontinuities at the interface generally play a crucial role in the charge transport phenomenon. Minority carriers are confronted with this barrier at the interface, favoring their accumulation and possibly causing recombination.(40) The electron affinities of the electron transport layer (ETL) and the hole transport layer (HTL) are essential in establishing the energy level alignment between the absorber layer and the charge transport layers (CTLs). This alignment significantly influences the efficiency of perovskite solar cells by enabling the formation of interfaces that are conducive to effective charge transport. Nevertheless, the fundamental physical mechanisms and loss characteristics related to this energy-level alignment remain unclear. Specifically, the dynamic influence of the energy-level alignment with varying interface defect density has not been thoroughly investigated. In this context, we examined the influence of VBO on the interface defect tolerance of solar cells utilizing each of the HTMs chosen for this analysis. The values of J_{sc} , V_{oc} , FF and the device efficiencies were recorded and plotted in *Figure 4*. The presence of cliffs and spikes at the edge of the absorber/HTL influences the performance of the specified solar cell. As illustrated in *Figure 5*, A positive VBO indicates the formation of a spike near the interface between the absorber and the hole transport layer (HTL), which creates obstacles for the movement of hole charge carriers. In contrast, a negative VBO creates an energy cliff at the absorber/HTL interface, facilitating the movement of hole charge carriers.(41) *Table 6* presents the computed valence band offsets (VBO) for each hole transport layer in relation to the perovskite absorber layer, utilizing Equation 3.1. From *Table 6*, it is evident that Cu₂O, CuI, and CuSCN exhibit a negative valence band offset (VBO), while CuO, MoS₂, CuSbS₂, and CuAlO₂ shows a positive VBO. The effect of the VBO is reflected in *Figure 6*, devices with negative VBO shows a decrease in efficiency while those having positive VBO exhibits a constant efficiency as N_i^t is increased from 10^{12} cm^{-3} to 10^{20} cm^{-3} .

From *Figure 4*, It was observed that the formation of a spike with a band offset of 0.04 eV provided beneficial effects, as reflected in the enhanced J-V characteristics and improved defect tolerance. We also noted that beyond certain thresholds above zero VBO, the efficiency of the devices begins to decrease.

Devices with moderate positive VBOs demonstrated superior open-circuit voltage (V_{oc}), short-circuit current density (J_{sc}), and fill factor (FF) retention with increasing N_i^t . CuSbS₂ and CuAlO₂ retained their efficiency even at high N_i^t ($\sim 1 \times 10^{19} \text{ cm}^{-3}$), validating the robustness of their band alignments. CuI and CuSCN showed steep efficiency drop-offs beyond $N_i^t \sim 1 \times 10^{15} \text{ cm}^{-3}$ due to higher recombination facilitated by less favorable Fermi level positioning. The small positive valence band offsets observed in CuO, MoS₂, CuSbS₂, and CuAlO₂ devices create a favorable energy barrier. This barrier slows hole transfer just enough to spatially separate charge carriers;



it prevents backflow of holes into the perovskite, stabilizing charge separation, thereby reducing recombination at the perovskite/HTM interface.(23,42)

The energy level diagram, which encompasses the electron transport layer (ETL), the perovskite absorber layer, and the HTL is presented in Figure 6. The quasi-Fermi levels for electrons ($E_{F,n}$) and holes ($E_{F,p}$) indicate the population distribution of photo-generated carriers under illumination. For HTLs with a positive VBO, the electron quasi-Fermi level $E_{F,n}$ stayed significantly separated from the valence band edge at the perovskite/HTL interfaces, as shown in Figure 6. The significant separation of the $E_{F,n}$ from the conduction band edge of the perovskite, HTL and their interfaces is distinct from the negative VBO based devices. An increase in the electron quasi-fermi level typically indicates an increase in the concentration of photogenerated electrons. Therefore, the separation may imply a decrease in electron concentration at the perovskite/HTL interface region available for non-radiative recombination.(30)

Also observed in Figure 6, the band bending induced by the positive VBO contributes to an electric field that repels electrons from the interface and attracts holes. This aids in spatial separation of carriers, reducing recombination and enabling defect states to remain electrically inactive.(43,44)

While negative VBOs (e.g., CuI, CuSCN, Cu₂O) promote seamless hole extraction, they may allow deeper penetration of holes into recombination-active regions. In contrast, HTMs with positive valence band offset impede such penetration, thus reducing defect-assisted recombination. This illustrates the fundamental trade-off between transport facilitation and defect suppression that must be balanced in HTM selection.(10,45) From the results obtained, it can be observed that MoS₂ offers an optimal balance between effective hole extraction and recombination suppression.

3.3 The effect of acceptor concentration on I-V characteristics:

As demonstrated, the materials with positive VBO exhibits high interface defect tolerance. However, due to their efficiencies are low compared to the devices based on negative valence band offset due to the interface barrier introduced by the spike formation at the interface, which obstruct carrier transport and increases recombination.

In order to optimize the efficiency of these devices, the effect of shallow acceptor density was analyzed by varying the acceptor densities of the HTMs in relation to the I-V characteristics of the devices. *Figure 7(a)* and *7(b)* shows the results of the optimization where the efficiency of the device based on CuO is increased from 12.18 % to 18.68 %. We selected CuO for the optimization because it has the lowest acceptor concentration used in the simulation. However, from the analysis, it can be seen that MoS₂ exhibit the best performance having shown high interface defect tolerance and higher efficiency when compared with other HTMs and at the same acceptor concentration. This is due to the fact that MoS₂ possess the least positive VBO which enhances its charge transport.

To investigate the cause of increased efficiency with respect to shallow acceptor concentration, we studied the effect of electric field on the device characteristics. From *Figure 7c* it can be observed that as the shallow acceptor concentration (N_A) increases, the electric field at the CuO/perovskite interface increases. This increase in electric field facilitates the transport of charge carriers across the interface hence the corresponding increase in efficiency at higher N_A .(46) This also corresponds with the decrease in recombination rate at increasing N_A as shown in *Figure 7d*. Furthermore, it can be seen that the increase in N_A has insignificant effect on the defect tolerance behavior of the device, as shown in *Figure 7e*.

3.4 Effect of change in temperature

Bandgap changes at higher temperature, thus the combined effect of temperature and defect density on the devices characteristics is investigated. The results in *Figure 8*, explains the interplay between bandgap changes at higher temperature during operation and the interface defect. It shows that irrespective of the temperature, the efficiency of the device follows similar trend in relation to the interface defect density. For the Cu₂O (*Figure 8(a)*), the



efficiency plateaus up to N_i^t of $1 \times 10^{15} \text{ cm}^{-3}$ for all temperatures. This is followed by a sharp decrease in efficiency at further increase in interface defect density. At higher temperature, the efficiency of the devices increases but their responses to change in interface defect density is similar. For the MoS_2 (Figure 8(a)), the positive valence band offset induces interface defect tolerance that is unaffected by temperature change.

3.5. Effect Band gap and electron Affinity on the device Performance

As explained previously, although the energy level alignment plays a crucial role in the stealth-like behavior of the interface defect. As the bandgap or electron affinity changes, one would expect the defect tolerance behavior to change since at change. Therefore, we investigated these for the CuI and MoS_2 based devices as shown in Figure 9. It was observed that as the bandgap increases towards the formation of a positive VBO, there is the emergence and extension of a plateau region from $1 \times 10^{12} \text{ cm}^{-3}$ to $1 \times 10^{16} \text{ cm}^{-3}$. This plateau region is caused by the band alignment at higher bandgap which is related to the defect tolerance effect. To further confirm these, we varied the electron affinity for the two selected HTMs, MoS and CuI while keeping the bandgap constant, and then study their effects on defect passivation. This results thus confirms that the primary factor responsible for the defect tolerance in the devices based on these different HTMs is the valence band offset.

3.6 Benchmarking against reported lead based Triple cation perovskite

To evaluate the realism of the simulated device performance, the optimized efficiency was compared with reported triple-cation (FA/Cs/MA) Pb-based mixed-halide perovskite solar cells as shown in Table 7. Saliba *et al.* first demonstrated 21.1% efficiency using a Cs-containing triple-cation perovskite in a mesoporous architecture.(47) Subsequent developments in compositional engineering and halide management have enabled efficiencies exceeding 22% in triple-cation Pb systems.(48) In this context, the present simulated maximum efficiency of 19.85% for a triple-cation Pb perovskite employing a mesoporous TiO_2 scaffold lies within the experimentally established performance envelope. The slightly lower efficiency compared to state-of-the-art certified devices may be attributed to conservative defect density assumptions and the absence of advanced interface passivation layers in the present model. This agreement supports the physical plausibility of the adopted recombination and material parameters.

4. Conclusion

In this work, a comprehensive SCAPS-1D numerical investigation of a lead based mixed-cation perovskite solar cell was performed. Systematic optimization of absorber thickness, defect density, band alignment, transport layer selection, temperature, and parasitic resistances was conducted to evaluate performance sensitivity and defect tolerance. At moderate positive valence band offset, a stealth-like behavior of the interface defects is observed evidenced by the near constant efficiency with respect to increase in interface defect density. The results indicated that the band alignment play a crucial role in the device interface defect tolerance. This exposes the critical role of materials selection in the reduction of the effect of interface defect on device performance. HTL/perovskite interface. We also noted that the efficiency of the device based on the well aligned HTMs can be further optimized by tuning the material parameters of the HTMs.

The interface modelling framework adopted in this work isolates recombination-dominated losses arising from structurally induced interfacial defects (e.g., surface roughness or deposition damage), independent of electrostatic effects associated with charged defects. This approach enables quantitative assessment of recombination sensitivity and complements experimental passivation strategies. The results highlight that appropriate band alignment can intrinsically improve tolerance to interfacial defects, suggesting that materials selection and interface engineering are critical design parameters for high-efficiency lead-free perovskite solar cells. While record efficiencies for triple-cation Pb perovskites exceed 22% experimentally, the present simulation



focuses on defect-limited behavior, providing insight into recombination-driven performance constraints rather than record-efficiency optimization.

The results provide insights that guides materials selection for the physical deposition of hole transport materials directly on perovskite absorber. Therefore, we conclude from our numerical results that appropriate band alignment in inorganic HTL based n-i-p perovskite solar cell structure provides a certain level of tolerance to interface defects.

Author contributions

Conceptualization: Daniel I Amune, E. Nstoenzok, V. Anye

Methodology: Daniel I Amune, E. Nstoenzok, V. Anye, R. Koech, Dahiru Sanni, W. Fidel, J. Botsoa,

Software: Daniel I Amune

Validation: Daniel I Amune, W. Fidel, R. Koech, Dahiru Sanni

Funding Acquisition: Daniel I Amune, E. Nstoenzok

Data curation: Daniel I Amune, E. Nstoenzok, V. Anye

Formal analysis: Daniel I Amune, E. Nstoenzok, W. Fidel, V. Anye, J. Botsoa, R. Koech, Dahiru Sanni

Writing – Original draft: Daniel I Amune

Investigation: Daniel I Amune

Supervision: E. Nstoenzok, V. Anye

Writing – Reviewing and Editing: Daniel I Amune, E. Nstoenzok, V. Anye, J. Botsoa, R. Koech, Dahiru Sanni

Conflicts of interest

There are no conflicts to declare

Data availability

Data for this article, including effect of interface defect density on IV characteristics for different HTM; Energy band for the various inorganic HTM based devices; Effect of N_A of the HTL on the I-V characteristics of the solar cell at (b) Effect of interface defect at higher acceptor concentration (c) Electric field across the device layers (d) Recombination rate across the device layers (e) Effect of defect density on the device characteristics at $1 \times 10^{19} \text{ cm}^{-3}$ shallow acceptor concentration; Effect of change in temperature on the I-V characteristics of the solar cell based on (a) Cu_2O (b) MoS_2 ; J-V characteristics of the CuI based device with respect to interface defect density for (a) Change in electron affinity (b) changes in bandgap, are available at Harvard Dataverse Repository at <https://doi.org/10.7910/DVN/VETUOV>

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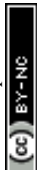
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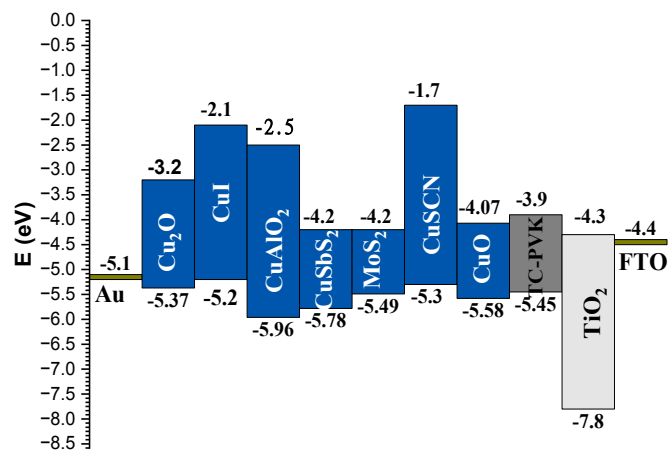
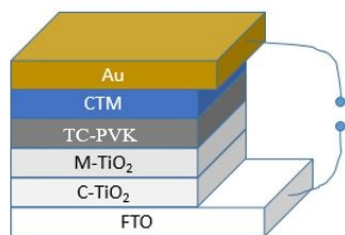


Figure 1. (a) Perovskite solar cell architecture (b) Energy band diagram



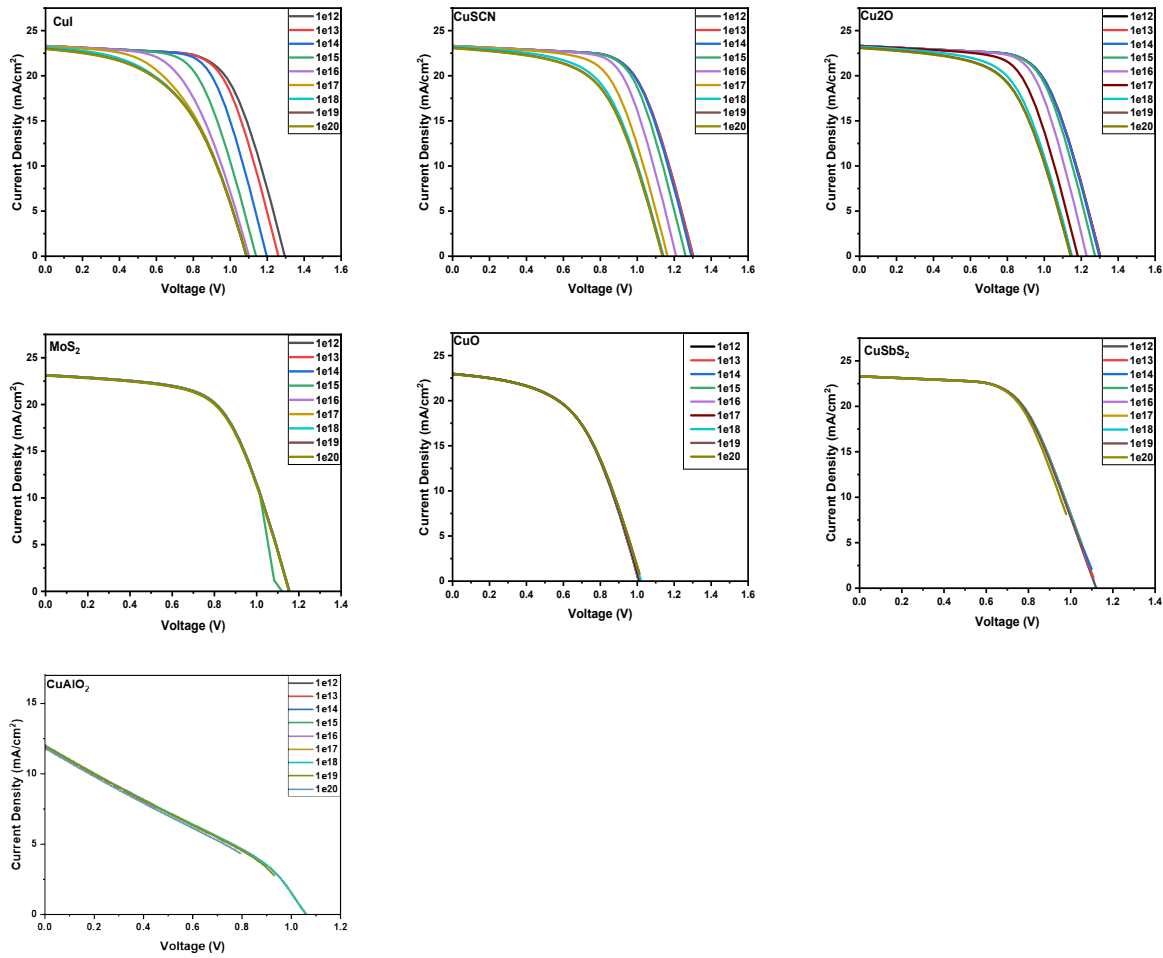


Figure 2. effect of interface defect density on IV characteristics for different HTM.



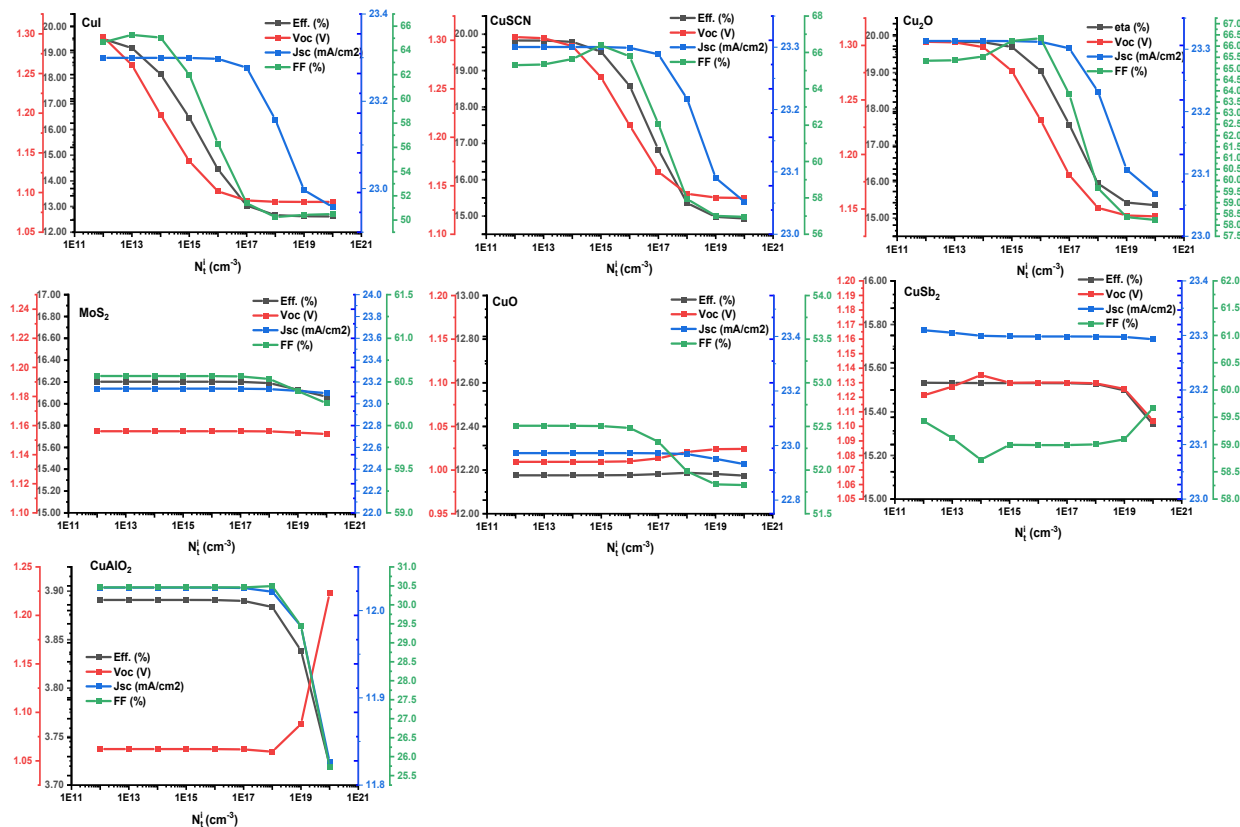


Figure 3. Effect of interface defect density on IV characteristics for different HTM.

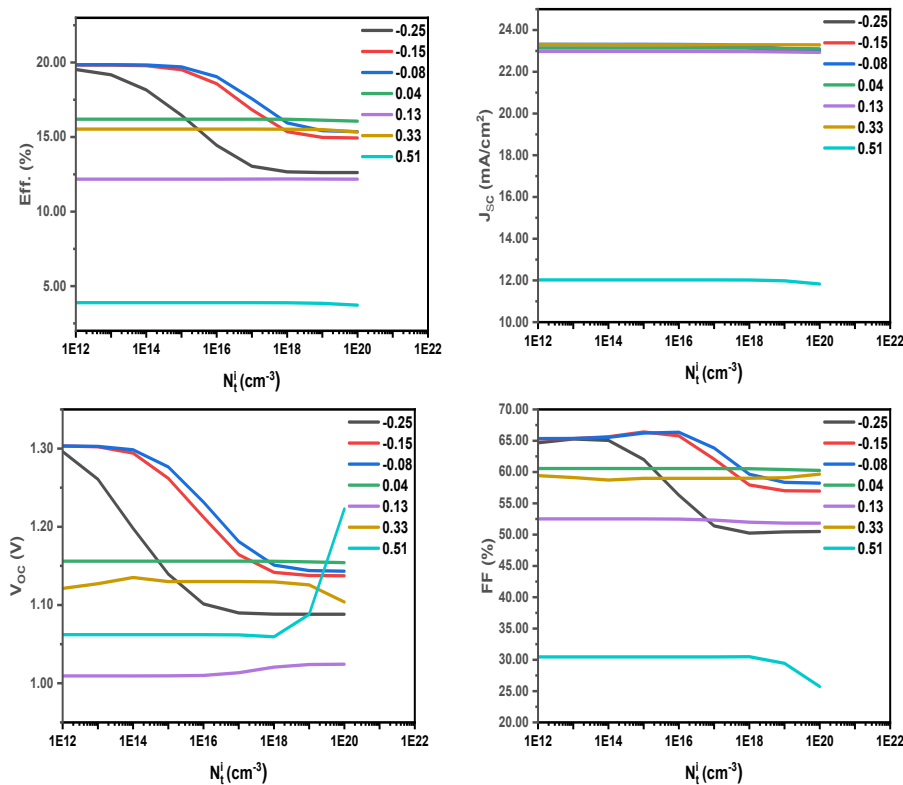


Figure 4. Effect of interface defect density on IV characteristics at different VBOs.





Figure 5. Schematics of cliff and spike at valence band

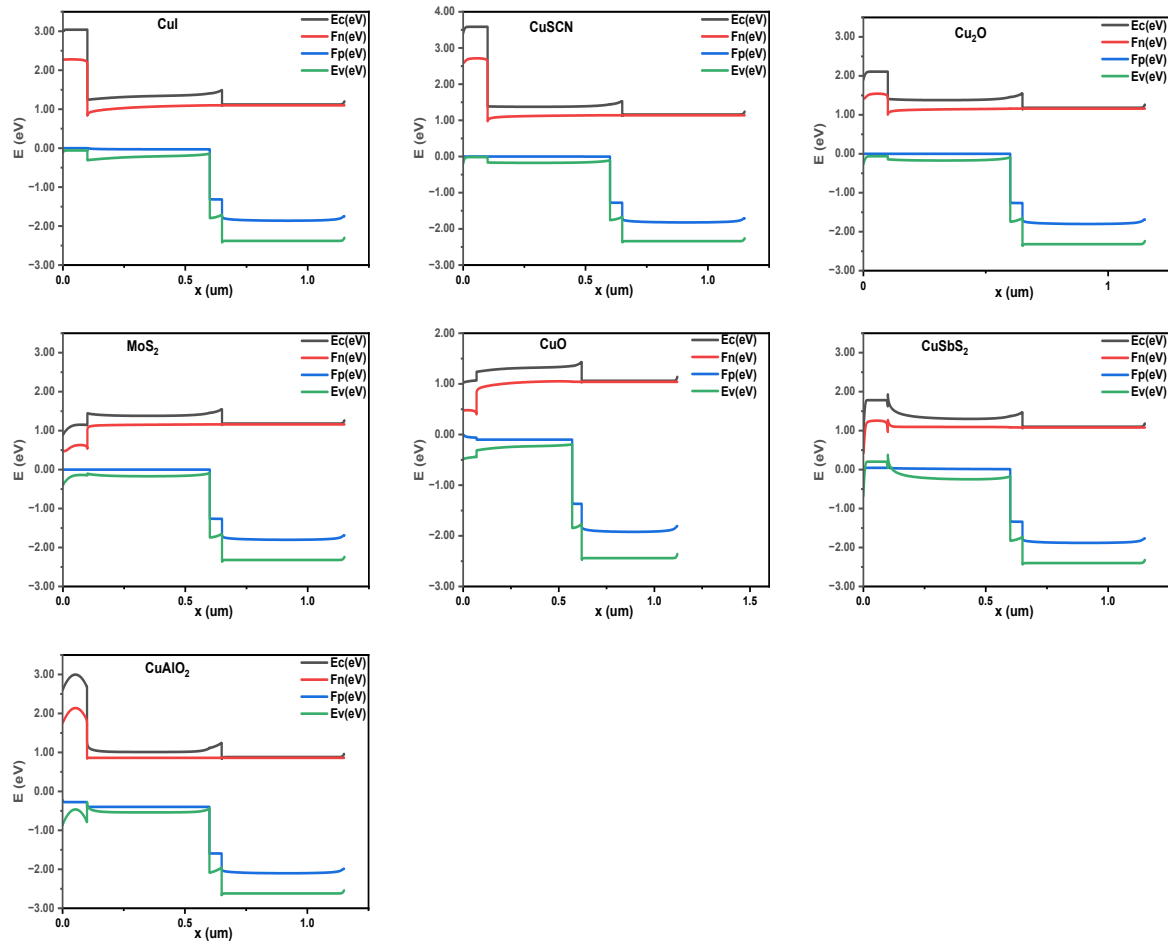


Figure 6. Energy band for the various inorganic HTM based devices



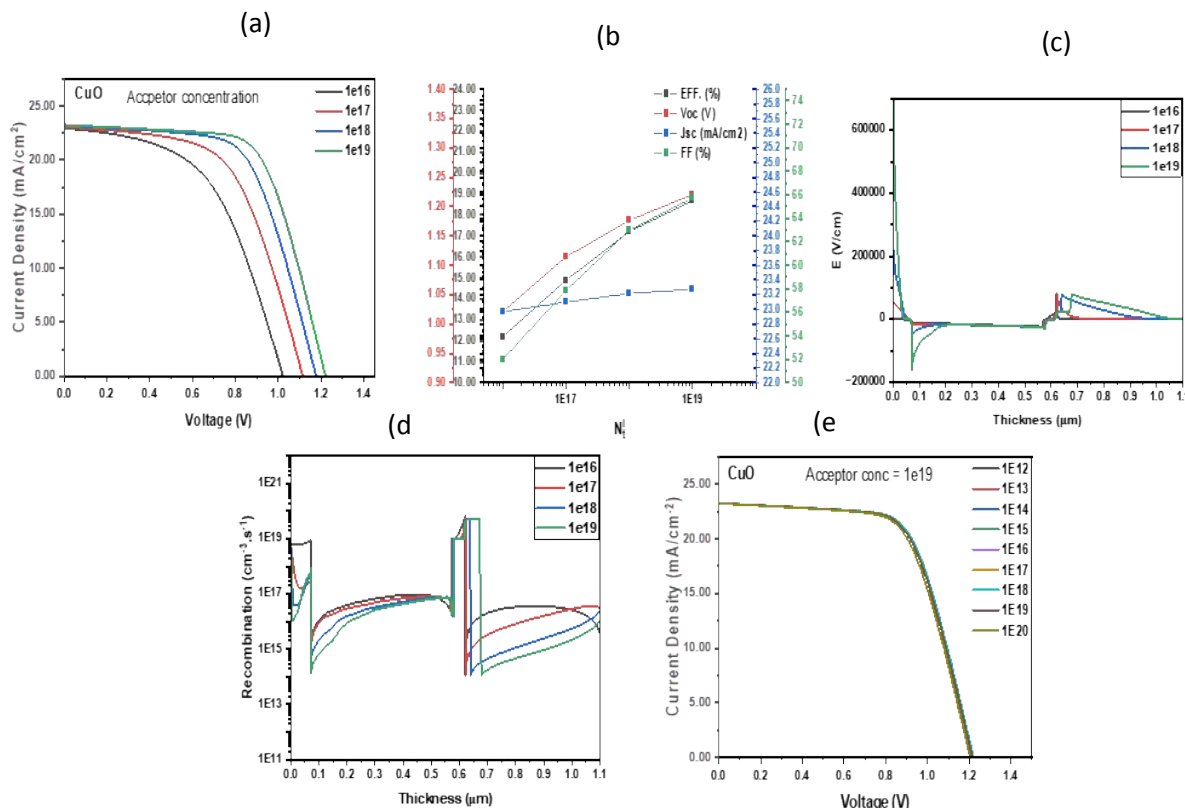


Figure 7. (a) Effect of N_A of the HTL on the I-V characteristics of the solar cell (b) Effect of interface defect at higher acceptor concentration (c) Electric field across the device layers (d) Recombination rate across the device layers (e) Effect of defect density on the device characteristics at $1 \times 10^{19} \text{ cm}^{-3}$ shallow acceptor concentration.

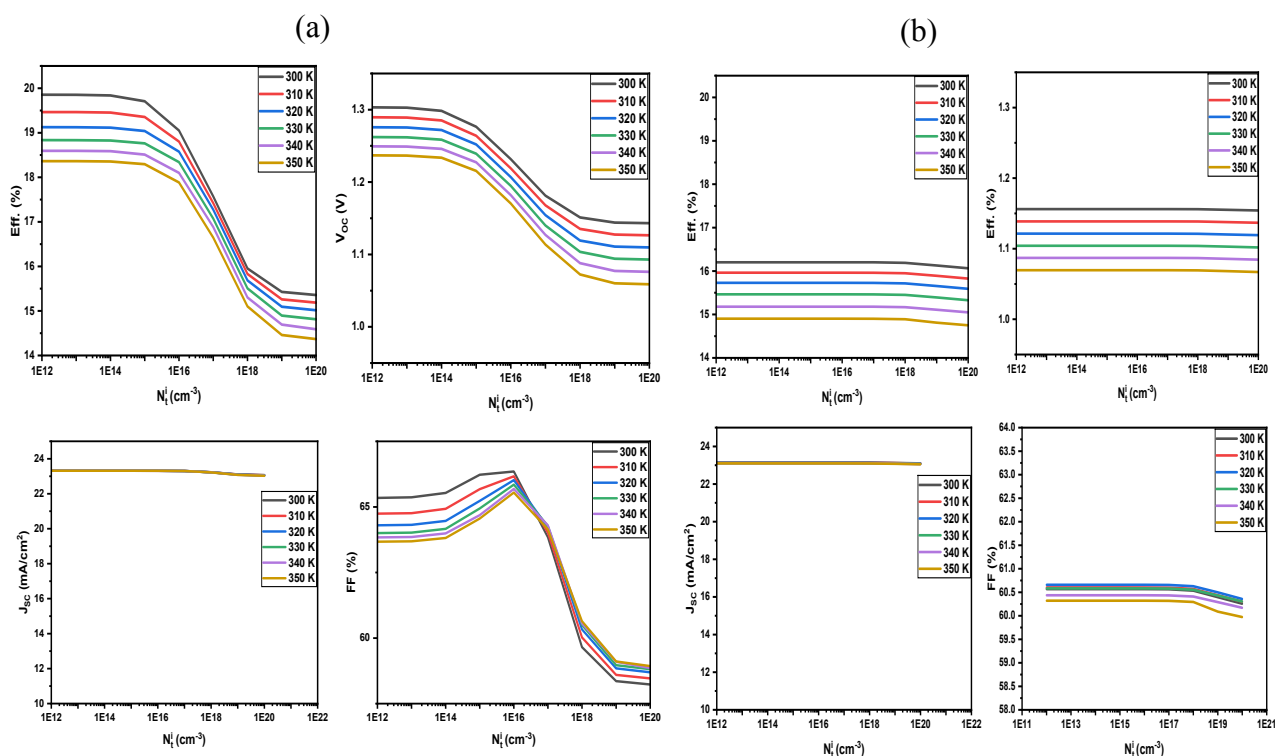


Figure 8. Effect of change in temperature on the I-V characteristics of the solar cell based on (a) Cu₂O (b) MoS₂

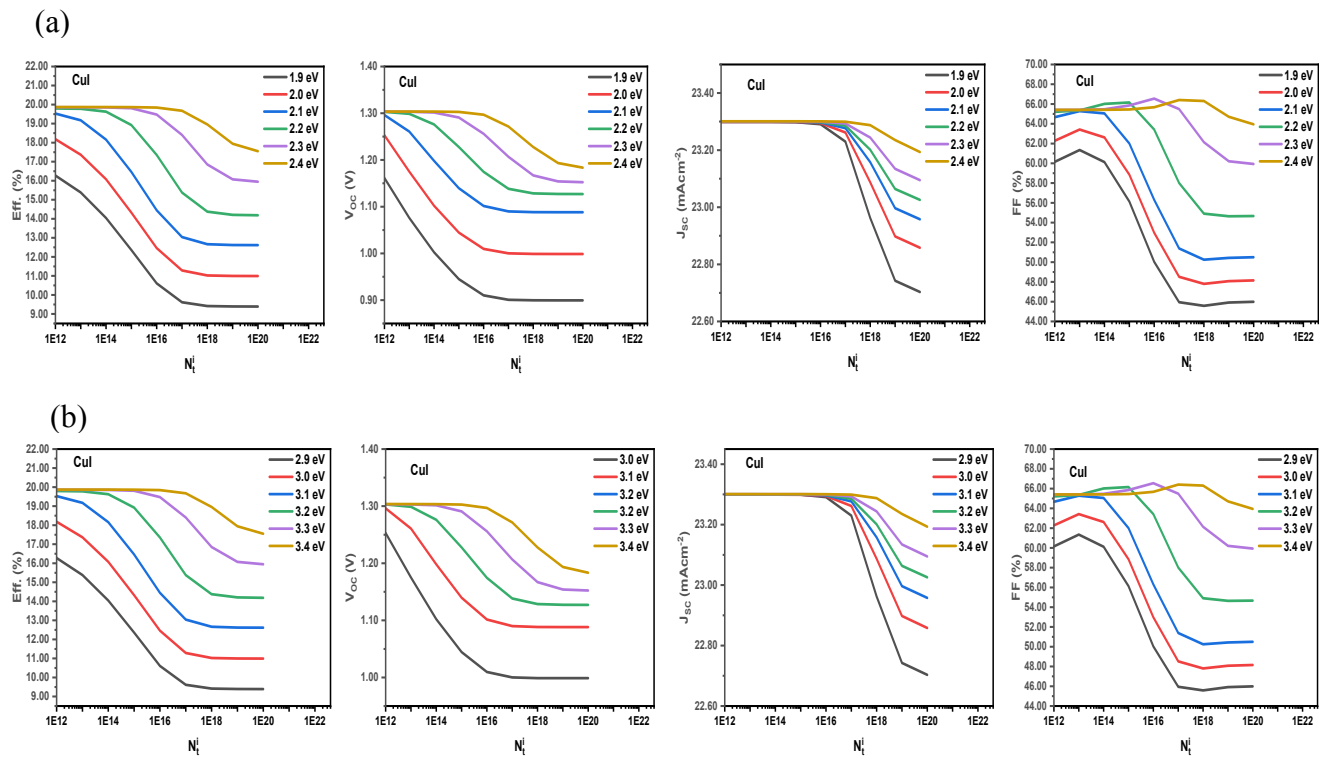


Figure 9. J-V characteristics of the CuI based device with respect to interface defect density for (a) Change in electron affinity (b) changes in bandgap.

Table 1. Physical parameter of different layers used for the simulation .(30,46,49–51)

Parameters	FTO	TC Perovskite	TiO ₂	CuO	CuSCN	CuI	CuAlO ₂	MoS ₂	CuSbS ₂	Cu ₂ O
Band gap/eV	3.5	1.55	3.2	1.51	3.60	3.10	3.46	1.29	1.58	2.17
Electron affinity/eV	3.3	3.9	4	4.07	1.7	2.1	2.50	4.2	4.20	3.2
Dielectric permittivity	9.0	6.6	9	18.1	6.5	7.11	60.00	8.9	14.6	10.7
Conduction band effective density of states/cm ⁻³	2.2X10 ¹⁸	1.0x10 ¹⁸	2.0x10 ¹⁸	2.2 x 10 ¹⁹	2.2 x 10 ¹⁹	2.8 x 10 ¹⁹	2.0x10 ²⁰	2.2 x 10 ¹⁸	2.0x10 ¹⁸	2.0 x 10 ¹⁷
Valence band effective density of states/cm ⁻³	1.8x10 ¹⁹	1.0x10 ¹⁸	1.8x10 ¹⁹	5.5 x 10 ²⁰	1.8 x 10 ¹⁸	1.0 x 10 ¹⁹	1.8x10 ²⁰	1.8 x 10 ¹⁹	1.0x10 ¹⁸	1.1 x 10 ¹⁹
Electron mobility/(cm ² .V ⁻¹ .s ⁻¹)	2.0x10 ³	2	20	100	100	100	2	100	49	200
Hole mobility/(cm ² _V ⁻¹ _s ⁻¹)	1.0x10 ²	2	10	0.1	25.0	43.9	8.6	150.0	49	80
Donor concentration/cm ⁻³	2.0x10 ¹⁹	0	1.0x10 ¹⁸							
Acceptor concentration/cm ⁻³	0	1.0x10 ¹⁵	0	1.0 x 10 ¹⁶	1.0 x 10 ¹⁸	1.0 x 10 ¹⁸	1.0x10 ¹⁹	1.0 x 10 ¹⁷	1.0x10 ¹⁹	1.0x10 ¹⁸



Table 2. Parameter for the contacts

Contacts	Anode properties	Cathode properties
Metal work function(eV)	5.1	4.4
Surface recombination velocity of electrons (cm/s)	1.0×10^7	1.0×10^7
Surface recombination velocity of holes (cm/s)	1.0×10^7	1.0×10^7

Table 3. Electric properties of defects in the materials.(52,53)

Parameter	Absorber	ETL	HTL
Defect type	Neutral	Single donor	Single donor
Capture cross section for electrons (cm^{-2})	1.0×10^{-15}	1.0×10^{-13}	1.0×10^{-14}
Capture cross section for holes (cm^{-2})	1.0×10^{-15}	1.0×10^{-14}	1.0×10^{-13}
Energetic distribution	Gauss	Guass	Gauss
Energy level with respect to E_v	0.6 (above E_v)	1.2 (above E_v)	0.65 (above E_v)
Defect density ($1/\text{cm}^3$)	1×10^{15}	1×10^{15}	1×10^{15}

Table 4. Electric properties of defects at the interfaces.(52,53)

Parameter	ETL/Absorber	Absorber/HTL
Defect type	Neutral	Neutral
Capture cross section for electrons (cm^{-2})	1.0×10^{-19}	1.0×10^{-19}
Capture cross section for holes (cm/s)	1.0×10^{-19}	1.0×10^{-19}
Energetic distribution	Single	Single
Energy level with respect to E_v	0.6	0.6
Characteristic energy (eV)	0.1	0.1
Interface defect density (cm^{-2})	1.0×10^{14}	$1.0 \times 10^{12} - 1.0 \times 10^{20}$

Table 5. I-V characteristics of the device based on various HTMs at 10^{12} cm^{-3} and 10^{20} cm^{-3}

HTMs	$N_t^i (\text{cm}^{-3})$	Eff. (%)	V_{oc} (V)	J_{sc} (mAcm^{-2})	FF (%)
CuI	10^{12}	19.52986	1.29615	23.2996	64.66876
	10^{20}	12.61576	1.0882	22.95758	50.49864
CuSCN	10^{12}	19.82882	1.30329	23.30042	65.29659
	10^{20}	14.93335	1.13738	23.05214	56.95612
Cu ₂ O	10^{12}	19.85439	1.3033	23.31324	65.34441
	10^{20}	15.3568	1.14322	23.06835	58.23095
CuO	10^{12} cm^{-3}	12.17611	1.00944	22.97219	52.50784
	10^{20} cm^{-3}	12.17463	1.02433	22.93198	51.8292
MoS ₂	10^{12} cm^{-3}	16.20221	1.15612	23.1383	60.56765
	10^{20} cm^{-3}	16.06413	1.15421	23.09789	60.25573
CuSbS ₂	10^{12} cm^{-3}	15.53302	1.12128	23.30958	59.43031
	10^{20} cm^{-3}	15.34221	1.10391	23.29305	59.66596
CuAlO ₂	10^{12} cm^{-3}	3.89095	1.06214	12.02633	30.46078
	10^{20} cm^{-3}	3.72049	1.22288	11.8264	25.72541

Table 6. Valence band offset at the Perovskite/HTL interface for devices based on various HTMs

Material	CuI	CuSCN	Cu ₂ O	MoS ₂	CuO	CuSbS ₂	CuAlO ₂
VBO	-0.25	-0.15	-0.08	0.04	0.13	0.33	0.51

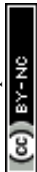


Table 7. Benchmarking for lead-based triple cation perovskite solar cell

Composition	Architecture	Halide System	PCE (%)	Study Type	Reference
FA/Cs/MA Pb(I/Br) ₃	n-i-p Mesoporous	I/Br	21.1%	Experimental	Saliba <i>et al</i> (47)
FA/Cs/MA Pb(I/Br) ₃	Mesoporous	I/Br	19.5% 20.1 %	Experimental	Tan <i>et al</i> (54)
FA/Cs/MA Pb(I/Br) ₃	Optimized triple cation	I/Br	18% 18.9%	Experimental	Conings <i>et al</i> (55)
FA/Cs/MA Pb(I/Br) ₃	Passivated triple cation	I/Br	19.21 21.71	Experimental	Yang <i>et al</i> (56)
This work (FA/Cs/MA Pb mixed halide)	Mesoporous TiO ₂	I/Br	19.85% (Max)	Simulation	This work



Data availability

Data for this article, including effect of interface defect density on IV characteristics for different HTM; Energy band for the various inorganic HTM based devices; Effect of N_A of the HTL on the I-V characteristics of the solar cell at (b) Effect of interface defect at higher acceptor concentration (c) Electric field across the device layers (d) Recombination rate across the device layers (e) Effect of defect density on the device characteristics at $1 \times 10^{19} \text{ cm}^{-3}$ shallow acceptor concentration; Effect of change in temperature on the I-V characteristics of the solar cell based on (a) Cu_2O (b) MoS_2 ; J-V characteristics of the CuI based device with respect to interface defect density for (a) Change in electron affinity (b) changes in bandgap, are available at Harvard Dataverse Repository at <https://doi.org/10.7910/DVN/VETUOV>

