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Semi-crystalline and recyclable pressure sensitive adhesives from non-edible rapeseed oil-based hyperbranched polyester vitrimers

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Hyperbranched polymers exhibit distinctive properties attributed to their highly branched architecture and the abundant functional groups they carry. In this study, we synthesized innovative hyperbranched polyesters from vegetable oil derivatives without utilizing solvents. The reaction conditions were optimized and these polymers were comprehensively characterized based on size distribution, degree of branching and molecular structure. These hyperbranched polymers manifest as sticky viscous liquids with semi-crystalline behavior, featuring glass transition temperatures (T_g) and melting temperatures (T_m) as low as 18 °C. Subsequently, the latter were utilized as precursors for the design of unique thermosets with potential self-healing and recyclable pressure-sensitive adhesive properties.

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Sustainability spotlight

This work advances green chemistry by designing recyclable and self-healing pressure-sensitive adhesives (PSAs) from hyperbranched polyesters derived from erucic acid, a non-edible and renewable vegetable oil. The synthesis is performed solvent-free, valorizing a non-food biobased synthon and yielding vitrimers that are recyclable and reusable while maintaining adhesive performance.

Introduction

Manufacturing bio-based materials that offer both competitive properties and sustainability has emerged as a highly important industrial matter.¹ In the field of pressure sensitive adhesives (PSAs),² which are reversible adhesives *via* non-covalent bonding used in a large variety of applications from tapes to electronics, the market is dominated by oil-based and non-sustainable materials with extremely cheap costs. The most common systems are based on either polyacrylates, styrene-butadiene/isoprene block copolymers or polysiloxanes.² Consequently, recyclable materials³ and sustainable alternatives based on renewable resources have been developed.^{4–10}

Vegetable oils, which are triglycerides composed of various saturated and unsaturated fatty acid moieties, are considered potential resources for such applications due to their availability, low toxicity and low cost. Additionally, their fatty acid chains contribute to the flexibility, hydrophobicity and low glass-transition temperatures of PSAs, providing the tackiness such materials require.¹¹

Various methods are available for converting vegetable oils into pressure-sensitive adhesives (PSAs). One prevalent method involves the use of epoxidized vegetable oils, which are produced through the oxidation of double bonds with peroxide acids. Recently, Li *et al.* designed a resin based on epoxidized soybean oil (ESO) and a highly branched polyamine, which exhibits excellent adhesion and film-forming properties.¹² In a previous study, ESO was used by Wang *et al.* with rosin acid to obtain bio-based PSAs with high adhesivity.¹³ The authors also designed PSAs using carboxylic acid-terminated polyricinoleate and ESO as well as flame-retardant PSAs with ESO and phosphorus containing dicarboxylic acids.¹⁴ Epoxidized fatty acids have also been used by Li *et al.* to prepare bio-based PSAs by reaction with di-carboxylic acids or anhydrides.¹⁵ PSAs based on linseed oil were developed by Pan *et al.*¹⁶ Oleic methyl ester obtained by transesterification of the corresponding triglyceride was epoxidized with peracetic acid/H₂O₂ and acrylated with acrylic acid. The resulting acrylated monomer was then polymerized through a miniemulsion process to obtain PSAs with properties comparable to petroleum-based systems.¹⁷ Following a similar strategy, Badía *et al.* recently demonstrated that removable biobased waterborne PSAs could be obtained by forming bio-based 2-octyl acrylate-containing latexes by seeded semi-batch emulsion polymerization.¹⁸ This monomer contributed to the soft phase of the formulation. The properties of the resulting latexes were adjusted through the incorporation of

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piperylonyl methacrylate and isosorbide dimethacrylate, which constituted a harder phase.^{19,20} Using this strategy, high bio-content (71%) waterborne PSAs were thus obtained.

Nevertheless, due to their nature, thermoset-based adhesives are non-recyclable. In order to overcome this issue, bio-based vitrimers were developed by different teams for adhesive applications.^{21–31} Vitrimers are associative covalent adaptable networks that are built from reversible covalent bonds giving them a dynamic behavior and granting them both recyclability and thermoset properties such as mechanical strength and solvent resistance.²¹ The dynamic covalent network can be based on multiple types of bonds, among which boronic esters,²² which can present a high naturality index,^{23,24} are of particular interest in this study. Accessing recyclable thermosets is of utmost importance in today's context and many bio-based vitrimers were designed in anticipation of a post-petroleum era.²⁵ Some of these systems present adhesive properties such as the one synthesized from ESO and glycyrrhizic acid based on transesterification reactions²⁶ or an ESO based epoxy adhesive using dynamic borates designed by Qiu *et al.*²⁷ Similar to ESO, lignin is a highly available bio-resource and has been reported to manufacture vitrimer materials with adhesive properties.²⁸ Vitrimeric PSAs were developed as well^{29–31} but, to the best of our knowledge, no bio-based system has been reported.

The objective of this study was to enhance the adhesive properties of rapeseed oil-based hyperbranched polyesters (HBPEs) for the development of pressure-sensitive adhesives (PSAs) with customizable mechanical and adhesive characteristics. Hyperbranched polymers (HBPs) are synthesized in a single step and exhibit a highly branched, compact, and globular structure. This unique structure endows them with distinctive properties when compared to their linear counterparts, including lower viscosities in both molten and solution states, improved solubility, and high functionality.^{32,33} Their synthesis *via* self-condensation of AB_n type monomers was predicted by Flory in the 1950s³⁴ but the first intentional synthesis was reported in 1987 by Kim and Webster with polyphenlenes.³⁵ Our group explored the possibility to produce bio-based hyperbranched polyesters (HBPEs) with AB_n type monomers derived from vegetable oils.³⁶

There are examples of HBP based PSAs in the literature³⁷ but none of them are made from renewable resources. Some examples of bio-based HBPs with adhesive properties were, however, described.^{38,39}

In this study, a PSA was synthesized using a HBPE derived from erucic acid. The HBPE was prepared by transesterification polymerization of dihydroxylated methyl erucate at 160 °C under reduced pressure with a basic catalyst. The resulting HBPE exhibits high degrees of polymerization and branching, appearing as either a very viscous liquid or a visco-elastic solid at room temperature and turning into a flowing liquid at higher temperatures. Cold grinding of the HBPE with phenyl-diboronic acid and curing at 120 °C overnight under reduced pressure produced elastic materials with PSA properties. These materials were thermally recyclable at 120 °C through low pressure molding or injection molding.

Experimental section

Materials

TBD (triazabicyclodecene) and phenyl-1,4-diboronic acid were purchased from Sigma Aldrich and used as received. Dihydroxylated methyl erucate was kindly provided by ITERG (and was synthesized according to the procedure reported by Testud *et al.* with a GC purity of 85%).³⁶

Analyses

Nuclear magnetic resonance (NMR) spectroscopy and ¹H and ¹³C NMR experiments were conducted on a Bruker Avance 400 spectrometer (400.20 MHz or 400.33 MHz and 100.63 MHz for ¹H and ¹³C, respectively) at room temperature, in CDCl₃ as solvent unless otherwise mentioned. Chemical shifts (δ) are reported in parts per million relative to the known solvent residual peak (δ = 7.26 ppm). DEPT-135 (Distortion Enhanced Polarization Transfer) and two-dimensional analyses such as ¹H–¹H COSY (Homonuclear correlation spectroscopy), ¹H–¹³C-HSQC (Heteronuclear single quantum coherence) and ¹H–¹³C-HMBC (Heteronuclear multiple bond correlation) were also performed.

Size exclusion chromatography (SEC) analyses were performed in THF as the eluent (1 mL min^{−1}) at 40 °C, on a PL-GPC 50 plus Integrated GPC from Polymer Laboratories–Varian with a series of four columns from TOSOH [TSKgel TOSOH: HXL-L (guard column 6.0 mm ID × 4.0 cm L); G4000HXL (7.8 mm ID × 30.0 cm L); G3000HXL (7.8 mm ID × 30.0 cm L); and G2000HXL (7.8 mm ID × 30.0 cm L)]. The elution times of the filtered samples were monitored using RI detectors with a calibration curve based on low dispersity polystyrene standards (PSS). Trichlorobenzene was added as a flow marker.

Differential scanning calorimetry (DSC) measurements were carried out on a TA Instruments DSC Q200. For each sample, two cycles from −100 to 150 °C (unless otherwise mentioned) were performed at 10 °C min^{−1}. Glass transition and melting temperatures were calculated based on the second heating run. Thermogravimetric analyses (TGAs) were performed on a TA Q500 at a heating rate of 10 °C min^{−1} under a nitrogen atmosphere from room temperature to 700 °C.

Rheological analyses were performed using an Anton Paar MCR 302 rheometer equipped with parallel plates (Ø = 8 mm) and with 3% strain. The frequency dependence of the moduli was measured between 0.01 Hz and 100 Hz at 20 °C. The evolution of the moduli with temperature was measured at 10 rad s^{−1} and 2 °C min^{−1} between 20 and 120 °C.

The adhesion properties of the synthesized HBPE were assessed with a simple tack test. The upper plate of the rheometer was put in contact with the sample for 60 seconds in order for residual forces to disappear and then the plate was moved up at 1 mm s^{−1} and, the strength, imposed by the sample on the plate, was measured. This allowed us to obtain surface adhesion strengths in J m^{−2} and to compare the different materials. This experiment was performed multiple times in order to obtain reproducible results.

Stress relaxation tests were performed at 10% strain.



Synthesis of the HBPE

20 g of the monomer were heated and magnetically stirred under reduced pressure (*ca.* 35 mbar) at 90 °C for one hour in order to remove remaining traces of water. 300 mg of TBD were then added and the reaction mixture was heated at 120 °C under a nitrogen flux for one hour in order to eliminate the formed methanol. The reaction was then put under reduced pressure until no more bubbling was observed. The temperature was then increased to 160 °C for forty-nine hours. An increase of viscosity with time was observed but the product remained liquid at 160 °C. Upon cooling, a colorless sticky visco-elastic solid was obtained and used without any further purification.

The conversion of the reaction was followed by ^1H NMR (see SI Fig. S1–S6).

Synthesis of the HBPE – phenyl diboronic acid vitrimer

2 g of the HBPE and different amounts of phenyldiboronic acid (from 10.7 mg (0.06 eq. to diols) to 44.7 mg (0.25 eq. to diols)) were mixed and finely ground in liquid nitrogen. The resulting powder was cured under reduced pressure at 120 °C overnight and molded at 120 °C for 15 min. A sticky, elastic, non-colored and transparent material was obtained (Fig. 1 shows the image of the material produced by curing a mixture of HBPE with 8 equivalents of phenyldiboronic acid at 120 °C under vacuum overnight.)

Characterization

The degree of branching (DB) was expressed by Fréchet and coll. as a function of dendritic (*D*), terminal (*T*) and linear (*L*) unit proportions⁴⁰ but Frey and coll. lately simplified the equation for high molar mass HBPEs, which only takes into account the dendritic and linear units and is written as follows:⁴¹

$$\text{DB}_{\text{Frey}} = \frac{2D}{2D + L}$$

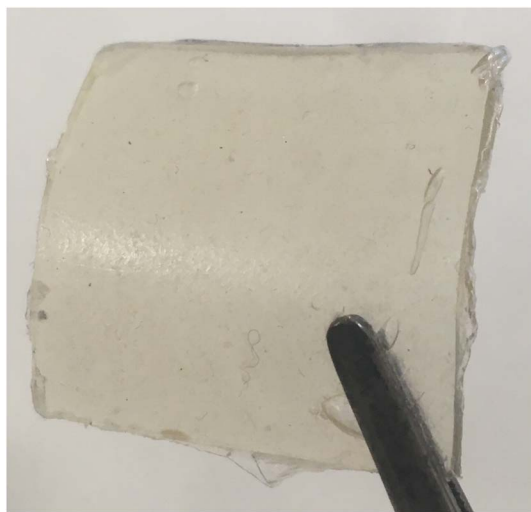


Fig. 1 Image of the material produced by curing a mixture of the HBPE with 8 equivalents of phenyldiboronic acid at 120 °C under vacuum overnight.

The Frey equation was used in this study. In order to calculate this DB, the proportions of dendritic and linear units can be obtained by ^1H NMR as the signals corresponding to dendritic (5.0 ppm) and linear (4.7 and 3.6 ppm) units are well defined. The degree of polymerization was assessed by NMR and the signal corresponding to the methyl ester is well separated (3.7 ppm). The dispersity of the polymer was assessed in SEC. It is important to note that the molar masses obtained by Size Exclusion Chromatography (SEC) using polystyrene standards were significantly underestimated. This discrepancy arises because these standards do not account for the dense structure of the hyperbranched polyester (HBPE), which consequently appears smaller than its real size in solution.

The thermal properties of both the HBPE and the vitrimers were analyzed. The melting temperature and the glass transition temperature of the different materials were measured by Differential Scanning Calorimetry (DSC) during the second heating and their thermal stability was assessed by Thermal Gravimetric Analysis (TGA). Moreover, the gel content of these materials was measured by immersing the sample in acetone for 48 h, drying overnight at 90 °C and measuring the resulting loss of weight, corresponding to the non-soluble fraction.

Results and discussion

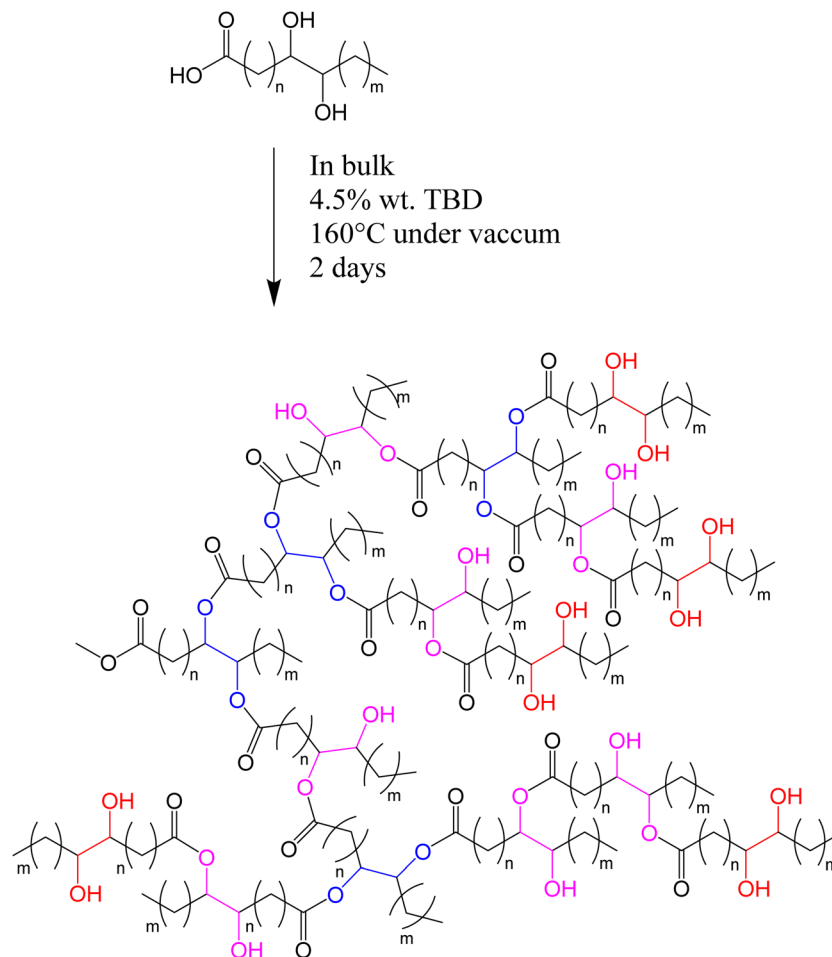
Due to their high availability and chemical versatility, vegetable oils have emerged as starting materials of high interest since they can serve as building blocks, the reactivity and properties of which can be adjusted through chemical modification. In addition, developing solutions that do not compete with the food industry is highly desirable. In this regard, erucic acid-enriched oils originating from rapeseed appear as relevant non-edible products, allowing further valorization by chemical transformation.

Successive transesterification, epoxidation, and hydrolysis of these erucic acid-enriched oils were performed, resulting in the production of dihydroxylated methyl erucate (Fig. S1). This compound, produced at the kg-scale at the technical institute, ITERG, was then polymerized in bulk by heating under vacuum with TBD as a basic catalyst to produce hyperbranched polyesters (HBPEs) (Scheme 1).

Characterization of the HBPE

The ^1H NMR analysis of the HBPE (Fig. 2) allowed the quantification of methyl ester moieties, dendritic units (*D*), linear units (*L*), and terminal units (*T*).³⁶ A DB_{Frey} of 0.46 was calculated from the integrals of the dendritic unit at 5 ppm and the linear unit at 4.6 ppm. The ratio between the terminal CH_3 of the fatty acid chain at 0.8 ppm and the focal methyl ester group at 3.7 ppm allowed calculation of the average molar mass of the resulting polymer, with a M_n of 40 600 g mol^{−1} (a degree of polymerization (DP_n) of 114.4). Finally, the ratio between the integrations of the focal methyl ester group at 3.7 ppm and the terminal units gave an average value of 30.6 (1,2-diol) units per HBPE molecule. Complementary NMR and FTIR characterization results of the HBPE are given in Fig. S2–S7.





Scheme 1 HBPE obtained by polymerization of dihydroxylated methyl erucate in bulk at 160 °C under vacuum for two days and catalysed by TBD.

The SEC analysis of this dihydroxylated methyl erucate-based HBPE (Fig. 3) in THF indicated an M_n value of 5200 g mol⁻¹ with a large dispersity value around 12, in agreement with the formation of a hyperbranched structure. While this M_n value could appear to be in disagreement with the NMR result, one should remember that the molar mass is given according to the linear PS standard and thus highly underestimated in this

specific case. This is mostly due to the limited swelling capacity of a densely constrained hyperbranched polymeric structure, making it appear smaller as one could expect.

The HBPE thus obtained was subsequently cross-linked with phenyldiboronic acid by means of cold grinding, followed by an overnight curing process at 120 °C under vacuum conditions (Scheme 2). FTIR analysis of the cross-linked materials reveals

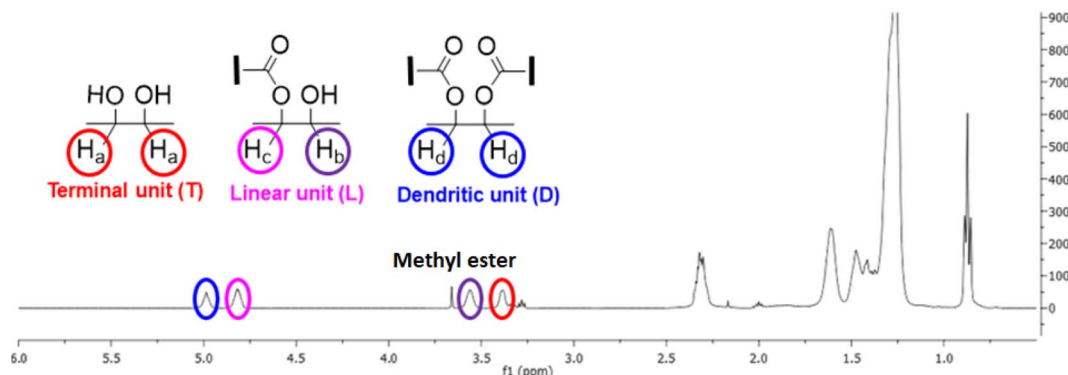


Fig. 2 ¹H NMR spectrum of the HBPE in CDCl₃, indicating methyl ester groups and terminal, linear, and dendritic units with their respective assignments.



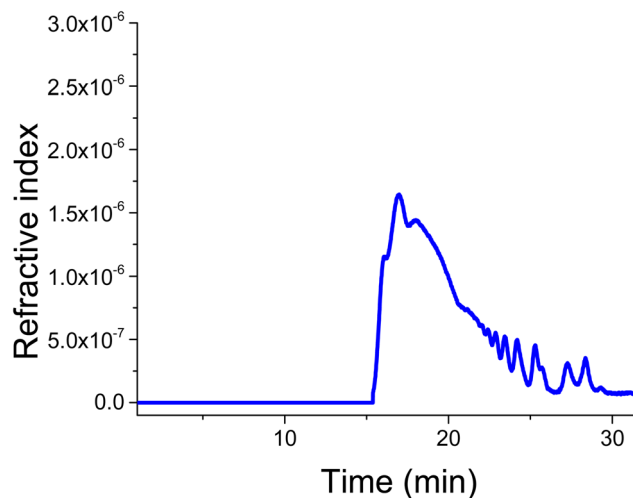


Fig. 3 Size exclusion chromatogram of the dihydroxylated methyl erucate-based HBPE in THF.

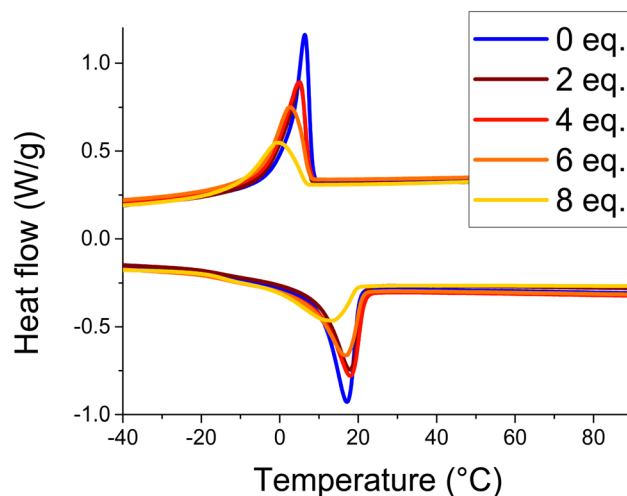


Fig. 4 DSC traces of the dihydroxylated methyl erucate-based HBPE (blue) and the cross-linked HBPE at different phenyldiboronic acid ratios.

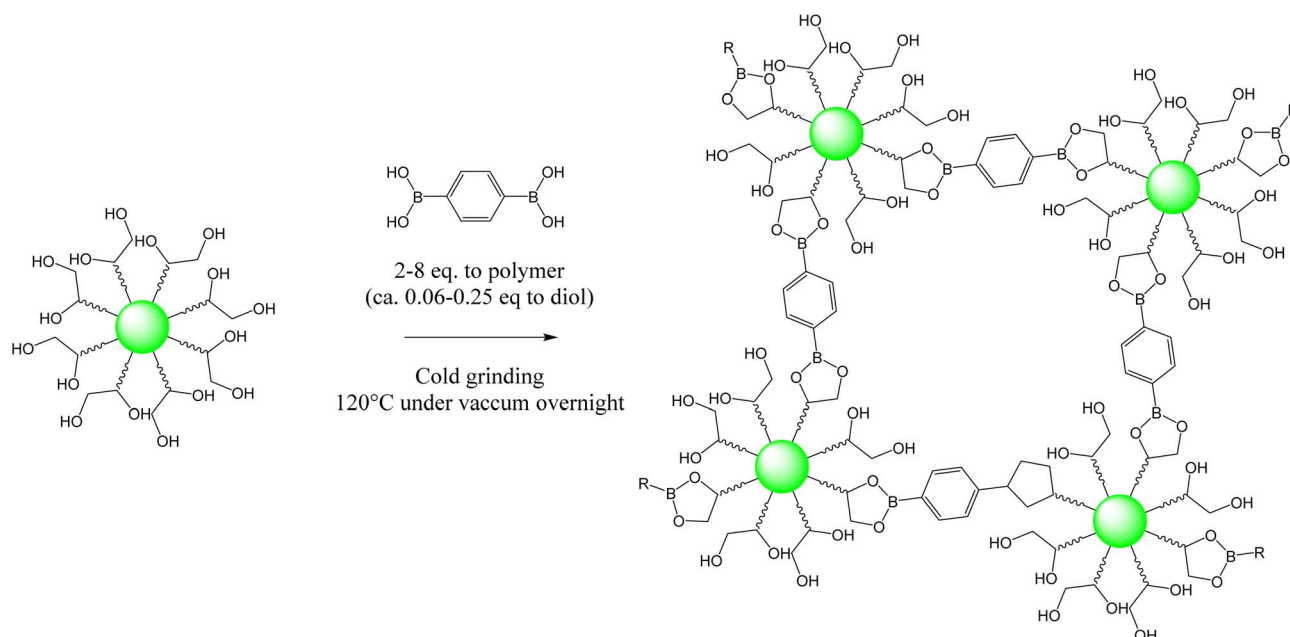
the appearance of bands at 650 cm^{-1} and 1350 cm^{-1} characteristic of the boronic ester function (see SI Fig. S7) without substantial variation according to the ratio of phenyldiboronic acid. In order to determine the influence of the cross-linking density on the properties of the material, different quantities of phenyldiboronic acid were used. The appearance of the resulting gel varies significantly according to differences in the phenyldiboronic acid ratio. Four materials ranging from 2 to 8 equivalents of phenyldiboronic acid to HBPE (*i.e.* 0.06 to 0.25 eq. to diol) were manufactured and characterized *via* thermal and rheological analyses.

To confirm the effective cross-linking, the gel content was measured by immersing the materials in acetone and recording

weight loss. The results indicated gel content values between 79 and 85%, irrespective of the quantity of phenyldiboronic acid used. The materials also demonstrated a swelling ratio in acetone of approximately 200% for 2 eq. of phenyldiboronic acid crosslinker against 100% for 8 eq.

Thermal properties

All materials showed similar thermal stability with $T_{d,5\%} = 326.1\text{--}331.7\text{ }^{\circ}\text{C}$ according to TGA under nitrogen (see SI Fig. S8). DSC analyses of the cross-linked materials (Fig. 4) reveal a decrease in the crystallinity ratio and the melting temperature



Scheme 2 Simplified view of dihydroxylated methyl erucate-based HBPE cross-linking with phenyldiboronic acid at $120\text{ }^{\circ}\text{C}$ under vacuum overnight.



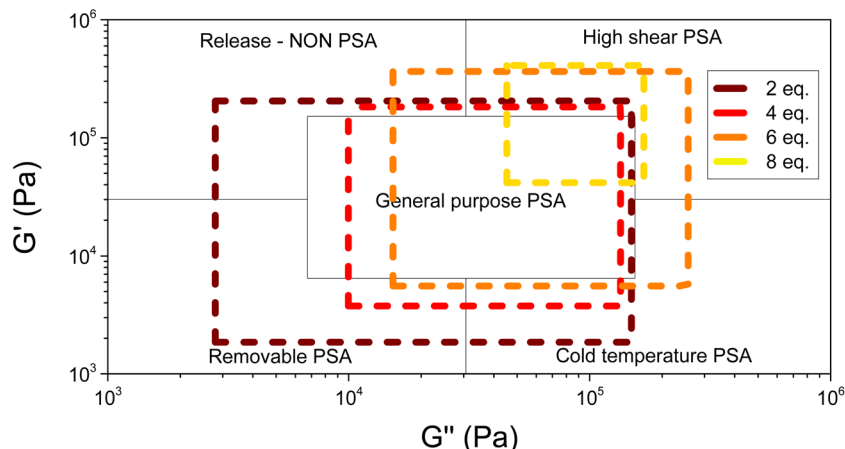


Fig. 5 Visco-elastic windows (visualization specific to PSA) of cross-linked dihydroxylated methyl erucate-based HBPE at different phenyl-diboronic acid ratios (from 2 eq. to 8 eq.).

with an increase in the phenylboronic acid ratio. The melting temperature ranges from 18 °C to 13.6 °C and the melting enthalpy ranges from 24 J g⁻¹ to 17 J g⁻¹. In addition, the glass

transition temperature decreases with the boronic acid content, from 5 °C for the HBPE precursor to -12 °C for the network with 8 eq. of boronic acid. The observation of uncommon remaining crystallinity in the networks is attributed to the alignment of pendant alkyl chains. This original feature observed with the erucic acid precursor may open up new functionalities such as thermomechanical actuation or controlled adhesion. Moreover, the increase in phenyldiboronic acid crosslinker content logically reduces the chain mobility, which decreases the ability of the alkyl chains to crystallize.

Mechanical properties

Rheological experiments were performed to measure the mechanical properties of the cross-linked HBPEs. The variations of materials' moduli with frequency were measured to obtain their visco-elastic window, which is a usual characteristic of pressure-sensitive adhesives, useful to predict the adhesive behavior of the materials.⁴² The visco-elastic window of the synthesized materials is represented in Fig. 5 (see SI Fig. S9). It represents the moduli window between a situation where the material is strongly solicited (10² Hz) and a situation where the material is weakly solicited (10⁻² Hz).

As the quantity of phenyldiboronic acid increases (the lighter the color of the visco-elastic window in Fig. 5), the window reduces and the moduli increase. This relationship indicates the potential to develop materials with specific properties ranging from general purpose PSAs (for low diboronic loading) to high shear PSAs (for high diboronic acid loading).

The variation of the moduli with temperature was measured to determine how phenyldiboronic acid affects the mechanical properties of the cross-linked HBPE upon heating (Fig. 6). The uncross-linked material (HBPE, black curve) shows a typical thermoplastic behavior, with a sharp decrease in G' over 30 °C, corresponding to the melting of its crystalline phase (see Fig. 4).

As expected, the loss of mechanical properties upon heating was directly related to the amount of cross-linker used. Highly cross-linked materials had higher and more stable moduli upon heating (Fig. 6). For cross-linked materials, all curves show

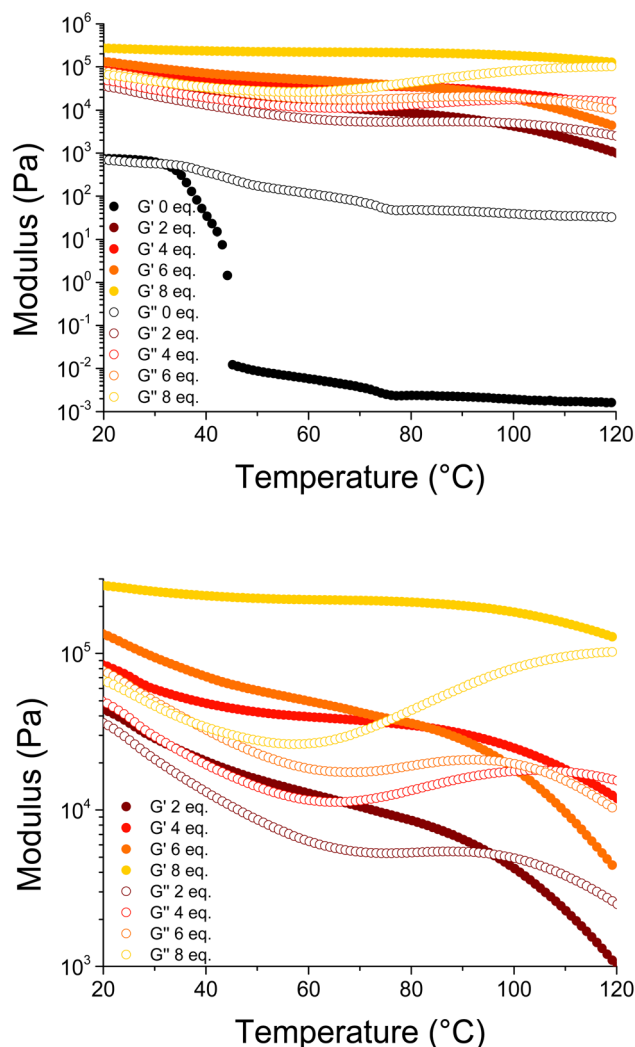


Fig. 6 Variation, with temperature, of the moduli of the cross-linked HBPE at different phenyldiboronic acid ratios (down: zoomed-in view).



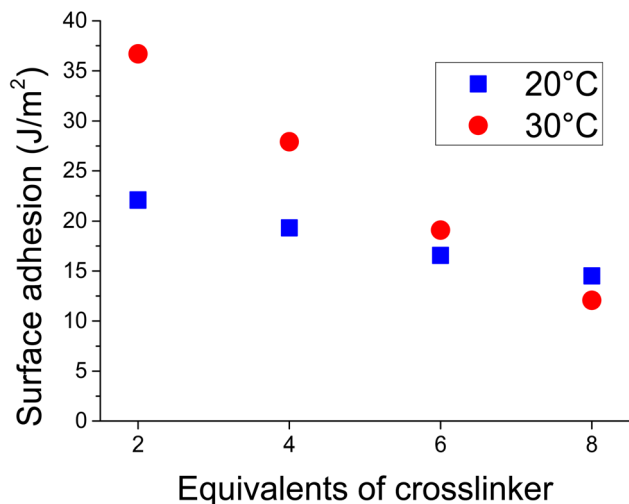


Fig. 7 Evolution of the surface adhesion of the cross-linked HBPE onto steel as a function of the quantity of phenyldiboronic acid cross-linker at 20 °C and 30 °C, respectively.

three temperature domains with different behaviors. The first domain shows a loss in both moduli (G'' faster than G'), with a steeper slope for the formulations including lower concentrations of cross-linker. The second domain exhibits a continued decrease in G' but an increase in G'' until it equals G' , indicating that the material starts flowing. The last domain shows a faster decrease in both G' and G'' , with a more liquid behavior (G'' is higher than G'). Finally, all materials exhibited vitrimeric behavior, with flowing temperatures increasing with higher cross-linker concentrations.

Adhesion properties

To evaluate the adhesion properties, the samples were placed in a rheometer between two steel plates under low stress at a given temperature and the upper plate was moved upward while measuring the adhesion force over time (see SI Fig. S10–S12).

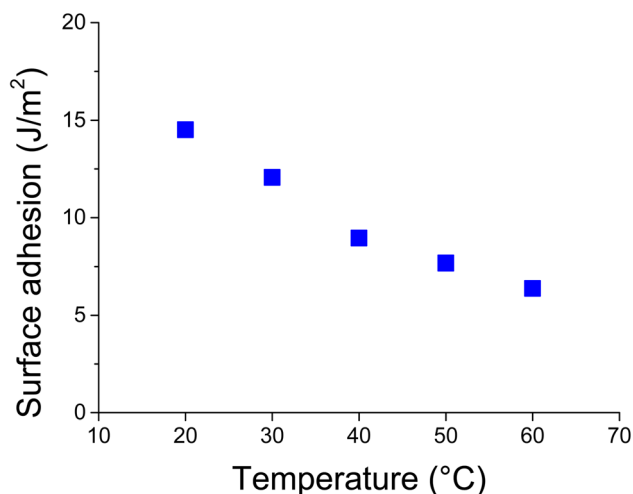


Fig. 8 Evolution of the surface adhesion with temperature of the HBPE cross-linked with 8 eq. of phenyl diboronic acid cross-linker.

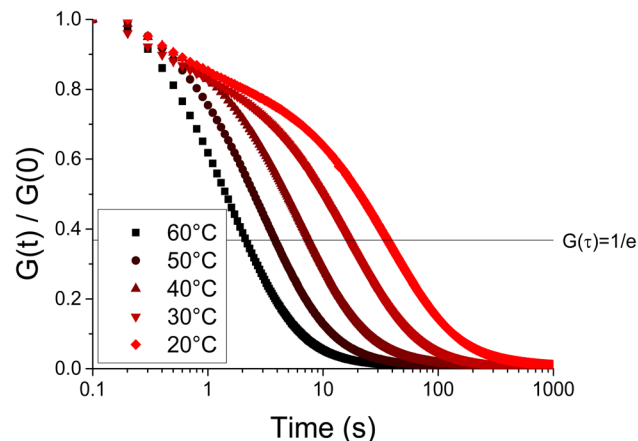


Fig. 9 Stress relaxation curves of the cross-linked HBPE with 8 eq. of phenyl diboronic acid cross-linker with respect to temperature.

The adhesion strength was determined by analyzing the variation area of the adhesion force as a function of time. The results are summarized in Fig. 7 and 8. An inversely proportional relationship between the amount of phenyldiboronic acid and the adhesion strength was observed. This could be attributed to the reduced number of hydroxyl groups that reacted with the crosslinker leading to fewer hydrogen bonding interactions.

Stress relaxation

Boronic ester, due to its constant chemical equilibrium, exhibits vitrimer properties and can relax stress within the network. This behavior is temperature-dependent, similar to the kinetics of the reactions occurring in the vitrimers. To assess the ability of these cross-linked materials, stress relaxation tests were performed on the HBPE crosslinked with 8 eq. of phenyl diboronic acid with a rheometer at different temperatures (Fig. 9) to measure the stress relaxation time τ ($G(\tau) = 1/e$). The principle involves applying stress (here 10% strain) to the

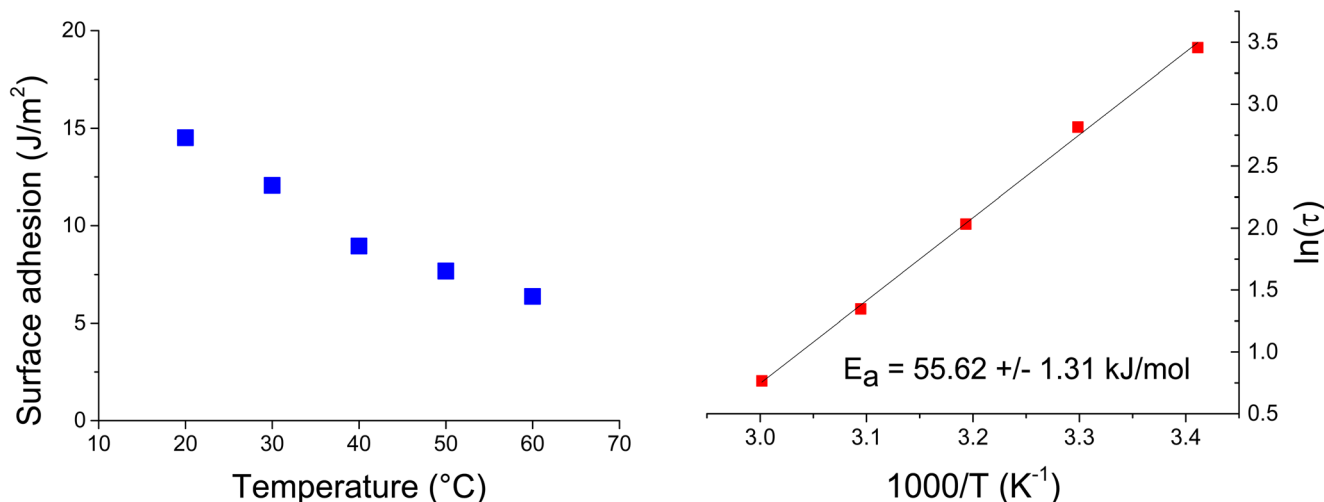


Fig. 10 Evolution of the relaxation time with the inverse of temperature.





Fig. 11 Pictures of the cut pristine (left) and recast (at 120 °C) (right) dihydroxylated methyl erucate-based cross-linked HBPEs with 8 eq. of phenyl diboronic acid used as the cross-linker.

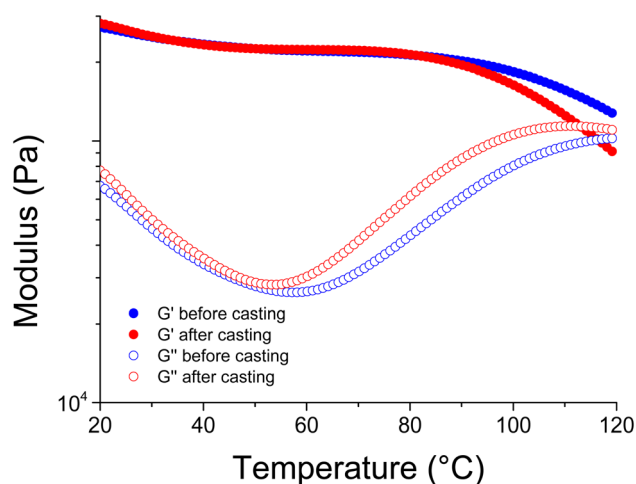


Fig. 12 Evolution of the moduli with temperature of the cross-linked HBPE with 8 eq. of phenyl diboronic acid used as the cross-linker, before and after mechanical recycling.

vitriemer and observing the evolution of this stress over time until complete relaxation. The evolution of the stress relaxation time with temperature followed an Arrhenius law (Fig. 10), thus demonstrating the vitrimeric feature of these materials. The latter exhibited an activation energy of $55.62 \pm 1.31 \text{ kJ mol}^{-1}$.

Mechanical recycling

Due to their vitriemer properties, these materials can be recycled. Mechanical recycling was performed on the highly cross-linked material (8 eq. of cross-linker). It can be reshaped at 120 °C using low pressure casting or injection molding (Fig. 11). The mechanical and adhesion properties of the recast material are comparable to those of the original. The moduli remain unchanged at low temperatures and slightly vary at higher temperatures (Fig. 12). Surface adhesion at 20 °C is consistent, maintaining 15 J m^{-2} (compared to 14.5 J m^{-2} for the original).

Conclusion

We have developed a bio-sourced, recyclable, and semi-crystalline pressure-sensitive adhesive (PSA) exhibiting vitrimeric behavior. This result leverages hyperbranched polyesters

(HBPEs), which offer multifunctionality with approximately 30 diol units per HBPE and flexibility due to their unique molecular structure. Notably, the HBPE utilized in this study is derived from erucic acid, a resource that is not in competition with the food industry, positioning it as a sustainable alternative.

Interestingly, the cross-linked networks display a degree of crystallinity that is unusual, a very specific feature that is attributed to the presence of long pendant alkyl chains derived from the fatty acid structure. This crystalline character may contribute to the materials' shape-memory behavior and may offer additional functionality worth exploring.

It has been established that the degree of crosslinking *via* boronic esterification can modulate both adhesive and mechanical properties, highlighting the interdependence of these characteristics. Owing to their vitrimeric nature, these materials can be reshaped and recycled mechanically, and their performance is maintained after reprocessing.

These PSA materials thus offer tunable properties and multiple recyclability pathways, supporting their potential in sustainable adhesive technologies.

Conflicts of interest

There are no conflicts to declare.

Data availability

The data supporting this article have been included as part of the supplementary information (SI). Supplementary information is available. See DOI: <https://doi.org/10.1039/d5su00868a>.

Acknowledgements

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