

Cite this: *Chem. Sci.*, 2026, 17, 7678

All publication charges for this article have been paid for by the Royal Society of Chemistry

Subunit fusion unlocks rapid *in vitro* maturation for slowly activating heterodimeric [FeFe]-hydrogenases

Jan Jaenecke, ^a Konstantin Bikbaev, ^b Miriam Malagnini, ^c Julia Bronold,^a Shanika Yadav, ^d Ulf-Peter Apfel, ^{de} Christophe Léger, ^c James A. Birrell, ^f Ingrid Span, ^b Nicolas Plumeré ^{*a} and Martin Winkler ^{*ag}

Hydrogenases offer a sustainable alternative to noble metals for catalyzing H₂-oxidation and H₂-production. The heterodimeric [FeFe]-hydrogenase of *Desulfovibrio desulfuricans* ATCC 7757 (*DdHydAB*) is most promising due to its exceptional catalytic activity and high-yield heterologous expression of its apo-form. Scalable production of the holo-form relies on *in vitro* maturation of the apo-enzyme using a chemically synthesized 2Fe_H cofactor mimic. However, the unusually slow *in vitro* maturation of *DdHydAB* raises mechanistic questions and limits its scalability. Through structural and sequence analysis, we identified the cause of this slow maturation and redesigned the enzyme *via* subunit fusion, inserting short peptide linkers near the active site. This modification facilitates the rearrangement of a critical locking element after cofactor uptake, increasing the maturation rate by up to 41-fold without compromising catalytic performance. Our findings elucidate a key step in the plug-lock-lid mechanism underlying maturation and promote the industrial applicability of *DdHydAB*.

Received 20th September 2025

Accepted 28th January 2026

DOI: 10.1039/d5sc07299a

rsc.li/chemical-science

Introduction

Hydrogenases are increasingly recognized as sustainable catalysts for H₂-production and utilization in green energy systems and the (bio)chemical industry.^{1–7} Among the three existing main classes, called [Fe]-, [NiFe]-, and [FeFe]-hydrogenases, the latter stand out for their bidirectional H⁺/H₂ interconversion at turnover frequencies that rival precious-metal catalysts.^{8–10} Previous studies highlight the superior features of individual [FeFe]-hydrogenases rendering them as promising candidates for future applications; this includes *DdHydAB* and *Cba5H*, both of which can reversibly adopt an inactive, O₂-resistant state (H_{inact}), permitting aerobic purification and facile sample

handling.^{11–15} Recent publications have demonstrated scalable heterologous expression and provide the basis for their implementation in electrochemical systems.^{6,7,16–18}

[FeFe]-hydrogenases offer a particularly convenient route to production at scales relevant for industry: their catalytic cofactor (H-cluster) comprises a [4Fe4S]-subcluster (4Fe_H) covalently linked *via* one of its Cys ligands to an unusual [2Fe]-subcluster (2Fe_H). In their apo-state, which lacks the 2Fe_H-subcluster, [FeFe]-hydrogenases can be effectively overexpressed in *Escherichia coli*. The purified apo-form can be matured *in vitro* to the active holo-enzyme by simply mixing the apo-hydrogenase with an excess of a synthetic 2Fe_H mimic complex (2Fe_H^{MIM}), which spontaneously enters the open cofactor-binding site (Fig. 1).^{19–21}

Upon *in vitro* maturation, the heterodimeric M2-type [FeFe]-hydrogenase *DdHydAB* of *Desulfovibrio desulfuricans* (identical to *Nitratidesulfovibrio vulgaris* and formerly labeled as *D. vulgaris*; see SI Discussion) reaches exceptionally high activities of 7400 s⁻¹ (H₂-evolution) and 56 000 s⁻¹ (H₂-oxidation),^{21–23} combined with the possibility of reversible sulfide-induced O₂-protection.^{14,15} However, the production of the active *DdHydAB* holo-enzyme still faces a major challenge.²¹ In contrast to the apo-states of monomeric [FeFe]-hydrogenases that fully activate within 15–30 min upon adding the 2Fe_H^{MIM} complex,²⁰ *in vitro* maturation of *DdHydAB* requires more than two days under similar conditions,²¹ lowering its application potential and motivating a mechanistic re-examination of the maturation process. *In vivo*, H-cluster maturation is assumed to be conserved among [FeFe]-hydrogenases of

^aProfessorship for Electrobiotechnology, Technical University of Munich, Campus Straubing for Biotechnology and Sustainability, Uferstrasse 53, 94315 Straubing, Germany. E-mail: martin-h.winkler@tum.de; nicolas.plumere@tum.de

^bBioinorganic Chemistry, Friedrich-Alexander-Universität Erlangen-Nürnberg, Egerlandstrasse 1, 91058 Erlangen, Germany

^cLaboratoire de Bioénergétique et Ingénierie des Protéines (BIP), CNRS, Aix-Marseille Université, 31 chemin Joseph Aiguier, 13009 Marseille, France

^dActivation of Small Molecules/Technical Electrochemistry, Faculty of Chemistry and Biochemistry, Ruhr-University Bochum, Universitätsstrasse 150, 44801 Bochum, Germany

^{de}Department of Electrosynthesis, Fraunhofer UMSICHT, 46047 Oberhausen, Germany

^fSchool of Life Sciences, University of Essex, Wivenhoe Park, Colchester, CO4 3SQ, UK

^gInorganic Spectroscopy, Energy Converting Enzymes, Max-Planck-Institute for Chemical Energy Conversion, Stiftstrasse 34-36, 45470 Mülheim an der Ruhr, Germany



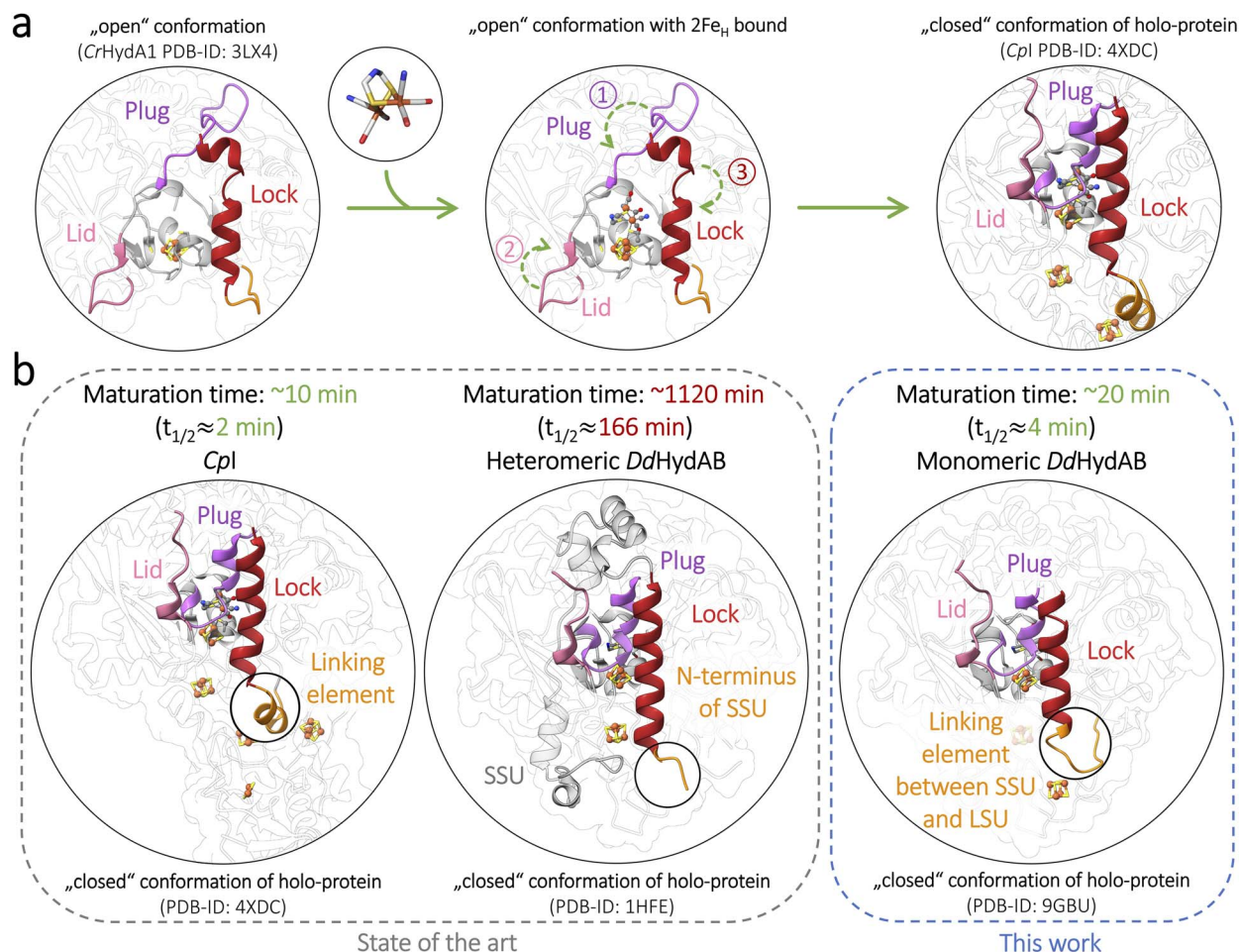


Fig. 1 Final steps of H-cluster assembly in [FeFe]-hydrogenases Cpl and DdHydAB after uptake of the 2Fe_H precursor. *In vivo*, the 2Fe_H precursor is biosynthesized by the three maturases HydE, HydG, and HydF (see Fig. S1). Alternatively, a mimic of the 2Fe_H precursor can be chemically generated and mixed with apo-hydrogenase to enable maturation of the H-cluster *in vitro*. (a) Upon insertion of the 2Fe_H precursor complex, the open H-cluster binding pocket is closed and locked by the “plug-lock-lid” system. (b) Closed-state structures of Cpl from *Clostridium pasteurianum* (PDB-ID: 4XDC) and DdHydAB (PDB-ID: 1HFE), illustrating that the gap between the small (SSU) and large (LSU) subunit is located right at the N-terminus of the lock-element. This potentially impacts the rate of binding site closure after cofactor insertion. In the monomeric Cpl, the lock-element is stabilized by a native linker-element. Fusion of the small subunit (SSU, grey) and the large subunit (LSU, transparent) in the heterodimeric DdHydAB via a linker peptide (orange) accelerates the maturation process by up to 41-fold (see $t_{1/2}$ reduction in Table S2).

eubacteria and eukaryotes.^{24–27} When heterologously expressed in its apo-form, DdHydAB is already equipped with the 4Fe_H -subcluster along with its two accessory [FeS]-clusters characteristic of the M2-type architecture, while exhibiting a vacant but accessible 2Fe_H binding site. The native 2Fe_H precursor arises via the contribution of three maturases (HydG, HydE, and HydF). HydG, one of the two radical-SAM enzymes involved, generates the mononuclear $[\text{Fe}^\text{II}(\text{CO})_2(\text{CN})(\text{Cys})]^-$ synthons. The other (HydE) combines two synthons to the 2Fe_H -pre-complex $[\text{Fe}_2(\mu\text{-SH})_2(\text{-CO})_4(\text{CN})_2]^{2-}$ before both of its Fe-sites are cross-bridged by the dithiomethylamine ligand, which is essential for H^+ -exchange between the proton transfer pathway and the active site (Fig. S1a).^{27–32} HydF serves as a scaffold protein for the biosynthesis of the 2Fe_H precursor before it is finally transferred from the mature HydF protein (holo-HydF) to the vacant cofactor-binding site of the apo-hydrogenase during a transient protein–protein interaction (Fig. 1 and S1b).^{28,33–36} Functional 2Fe_H insertion entails a rotation of the primary ligand sphere at the distal iron (Fe_d) of

the 2Fe_H precursor complex, favored by H-bond contacts between the two CN^- ligands and individual residues in the secondary ligand sphere,^{30,37} followed by a local reconfiguration of three structural elements first indicated by Mulder and coauthors in 2010.²⁶ One element turns into a plug-like structure that loosely closes the binding site over the inserted 2Fe_H precursor, thereby supporting the adoption of its rotated conformation. Next, this plug placement is barred in place from two sides by a restructuring of both remaining loop elements, termed lid and lock (Fig. 1a and S1b).^{24–26,30,31} The coupling of the $2\text{Fe}_\text{H}^{\text{MIM}}$ complex with the 4Fe_H -subcluster is enabled by one of its Cys ligands, which adopts a bridging configuration ($\mu\text{-SCys}$) between both subclusters (Fig. 1 and S1), yielding the transient $\text{H}_{\text{red}}\text{CO}$ state.^{25,27,38} A single oxidation step leads to the likewise inhibited but rather stable $\text{H}_{\text{ox}}\text{CO}$ state (Fig. S1c). Upon release of the apical CO ligand from Fe_d , the oxidized active-ready state H_{ox} is formed.^{25,27,38}

We previously hypothesized that DdHydAB's unusually slow maturation results from a higher propensity of the apo-form to



adopt the closed conformation.²¹ However, no attempts have been made to verify this idea or examine other potential reasons for the extremely low maturation rate of *DdHydAB*, let alone to speed up the maturation process. We propose an alternative, structure-based explanation centered on its heterodimeric state: a superposition of the X-ray structure of *DdHydAB* (1HFE)³⁹ with those of monomeric enzymes (*CrHydA1*, PDB ID: 3LX4; *CpI*, PDB ID: 4XDC)^{26,30} illustrates that the C-terminus of the small-subunit (SSU) coincides with the end of the lock-element (Fig. S2). In monomeric enzymes, the lock-helix straightens upon 2Fe_H uptake, bolting the plug in its place over the occupied binding site; in the case of *DdHydAB*, we hypothesize that the gap between SSU and LSU likely increases the flexibility of the lock-element at the free SSU N-terminus, hindering an efficient lock rearrangement and delaying the final closure of the pocket. Subunit fusion can be a powerful strategy to enhance the features of heterodimeric proteins, such as stability, yield, and catalytic efficiency.^{40,41} Linker-mediated coupling was also used to create chimeric catalysts with new functions.^{40–42}

Here, we investigated whether a linker-based monomerization approach positively impacts the unusually low *in vitro* maturation rate of *DdHydAB*. Employing FTIR spectroscopy and various kinetic measurements, we show that the rates of H-cluster formation and activity development of the monomeric variants increase by nearly two orders of magnitude relative to the heterodimeric wild-type.^{21,24} Protein-film electrochemistry and X-ray crystallography enable us to ensure the conservation of the advantageous traits of *DdHydAB* and to rationalize the differences in the variant-specific maturation rates.

Results and discussion

To investigate apo-*DdHydAB* for possible structural differences from monomeric [FeFe]-hydrogenases, particularly in the terminal regions of the subunits and the structural elements responsible for the closure and locking of the holo-protein, we crystallized *DdHydAB* in its unmaturation apo-form. The resulting crystal structure exhibited the closed state of the H-cluster binding site but did not show significant differences from the apo-state structure of the fast-maturing *CpI* enzyme, which was likewise crystallized in its closed state (PDB ID: 4XDD)³⁰ (Fig. S3a–e). As we were unable to support the hypothesis of a significantly higher tendency for premature binding-site closure in the case of *DdHydAB*, we focused instead on the fact that its heterodimeric state is the most prominent structural difference compared to monomeric enzymes such as *CpI* and *CrHydA1*. The gap between the two *DdHydAB* subunits coincides with the N-terminus of the lock-element, potentially affecting the 2Fe_H binding-site closure in the H-domain, and thus the rate of *in vitro* activation. To examine this hypothesis experimentally and with the aim of improving the maturation rate of *DdHydAB*, we engineered four SSU–LSU fusion constructs (variants L1–L4) by inserting linkers of defined length and sequence (Fig. 2).

For the linker design, we followed two strategies: (1) structural mimicry with *CpI*: based on a structural alignment between *DdHydAB* (1HFE)³⁹ and the monomeric *CpI* (PDB ID: 4XDC)³⁰ two linkers were designed (L1–2) with sequence homology and structural features corresponding to the native linker element in *CpI* (Fig. 2, L1–L2, and S4a, b). (2) Native linker sequences of monomeric homologs: we optimized the structural integration of the linker sequence into the protein environment to avoid destabilization and steric clashes. From a multiple sequence alignment, we selected two monomeric homologs of *DdHydAB* (M2-type [FeFe]-hydrogenases of *Solobacterium moorei* and *Veillonella atypica*; see corresponding sequences in Fig. S5) with a linker sequence coupling the parts that correspond to the SSU and LSU of *DdHydAB* (Fig. 2, L3–L4, and S4a, b). The resulting fusion proteins were heterologously expressed, purified (Fig. S6), and compared to the dimeric wild-type enzyme for their impact on the rate of *in vitro* maturation, comparing two independent approaches: (i) we used IR-spectroscopy to monitor the rate of H_{ox}CO formation as a product of 4Fe_H/2Fe_H-cluster coupling during maturation in a sealed sample of apo-enzyme upon 2Fe_H^{MIM} addition. (ii) We followed the activity development of the apo-hydrogenase variants upon addition of a 3-fold excess of 2Fe_H^{MIM} by periodically testing aliquots for the rate of H₂-evolution, which requires CO release from H_{ox}CO.

Electrochemical investigation of CO-binding kinetics and spectroscopic tracking of H-cluster formation rates

The production of the active enzyme from the H_{ox}CO state requires the final release of the apical CO ligand that blocks the substrate binding site (Fig. S1c). Among the best-characterized [FeFe]-hydrogenases, *DdHydAB* shows the lowest inhibition constant (K_i) for external CO.³⁹ We exploited this to track the amplitude increase of a characteristic peak at 2016 cm⁻¹ in the IR-spectrum of H_{ox}CO between 0 and 68 h (Fig. 3a; see Fig. S9 for full spectra) as a reporter for plotting the rate of H-cluster formation during the maturation process.

To ensure that all monomeric variants retain the high CO affinity of the wild-type enzyme, we employed protein film electrochemistry to monitor the kinetics of CO binding and release. In this classical chronoamperometry experiment, a film of holo-enzyme is prepared on the pyrolytic graphite surface of a rotating disk electrode.⁴³ The H₂-oxidation current is measured as a function of time before and after the enzyme is exposed to CO (Fig. S8). For all proteins, we determined and compared the Michaelis–Menten constants for H₂ (Fig. S7) and the inhibition constants for CO (K_i^{CO}). Both the binding affinity for the substrate (H₂) and the affinity for CO ($K_i = k_{\text{out}}/k_{\text{in}}$) are similar among all proteins (Table S1) with maximum differences among all proteins remaining well below a factor of two. The final CO-release step is associated with a K_i in the nM range, which confirms that the equilibrium of the reaction H_{ox}CO \rightleftharpoons H_{ox} + CO lies on the left side (H_{ox}CO). This especially applies to the conditions of the FTIR-spectroscopy experiment, as throughout the maturation process, the samples are kept in a tightly sealed cell, which maintains a high CO concentration



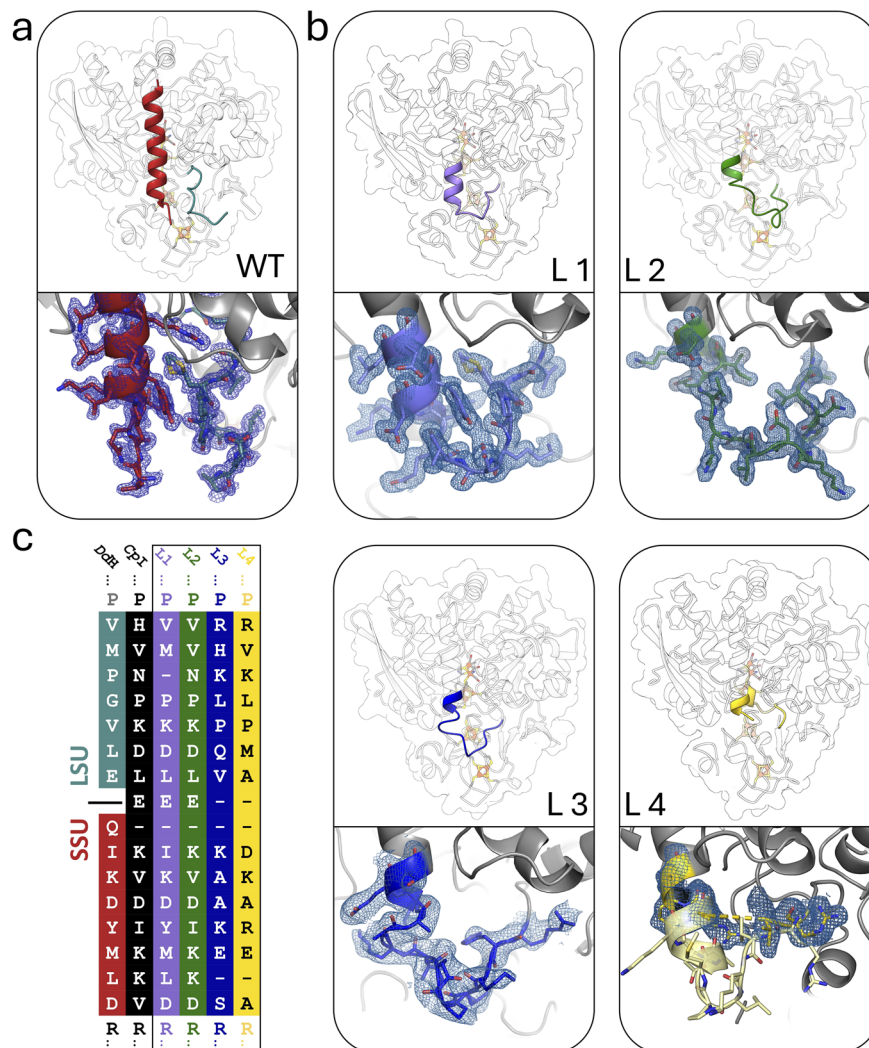


Fig. 2 Design of linker elements for subunit fusion in *DdHydAB*. (a) In the heterodimeric *DdHydAB* (PDB ID: 9QD6), the N-terminus of the small subunit is marked in red and the C-terminus of the large subunit in cyan. For holo-*DdHydAB*^{WT} and the monomeric fusion proteins L1–L3 (b) X-ray structures were solved including the linker region (PDB IDs: L1: 8RU6; L2: 9GNK; L3: GBU; L4: 8RYH). Linkers are colored according to the corresponding color code defined in (c) and shown as close-ups with their individual electron density as mesh contoured at 0.5 σ (L1–L3). For L4, the low local resolution in the linker section does not allow to derive the corresponding electron density mesh. Here, the linker section of a model structure of L4 generated *via* Boltz-2 was superposed with the X-ray structure obtained for L4. (c) L1 and L2 were derived from the sequence and structural alignment between *DdHydAB* and *CpI*. L3 and L4 were adopted from sequences of monomeric homologs of *DdHydAB* (see Fig. S4).

and excludes further competition with the substrate (H_2), thus ensuring a homogeneous accumulation of the $H_{ox}CO$ -state (Fig. S1c).²¹

Although *CpI* would be an ideal positive control for observing the H-cluster formation rate, its rapid activation and low enzyme-specific CO-ligand binding affinity prevent linear monitoring of the H-cluster formation rate using the state specific IR spectrum of $H_{ox}CO$.

For each protein, we determined the times required to reach 50% cofactor saturation (half-times; $t_{1/2}$) (Fig. 3c). All monomeric variants exhibited a significantly higher H-cluster formation rate compared to the heterodimeric wild-type, albeit to varying degrees. Based on the individual $t_{1/2}$ -values for the progression to full H-cluster occupancy, the rates of H-cluster formation can be ranked as follows: L3 > L2 \approx L4 >

L1 >> *DdHydAB*. With a 3.8-fold enhancement compared to *DdHydAB*^{WT}, the linker of variant L1 showed the lowest impact on the enzyme's maturation rate. Fusion proteins L2 and L4 displayed an 18- and 19-fold increase in the H-cluster formation rate, while variant L3 exhibited the highest impact with a 37-fold increase in the H-cluster formation rate relative to *DdHydAB*^{WT}.

Kinetics of enzyme activation monitored by solution assays

To assess *in vitro* maturation beyond H-cluster formation, we further monitored the activation to the catalytically competent holo-enzyme by periodically testing aliquots from an *in vitro* maturation mix for the enzyme's H_2 -evolution rate. This was achieved through a well-established sodium-dithionite-driven and methyl viologen (MV)-mediated activity assay measured



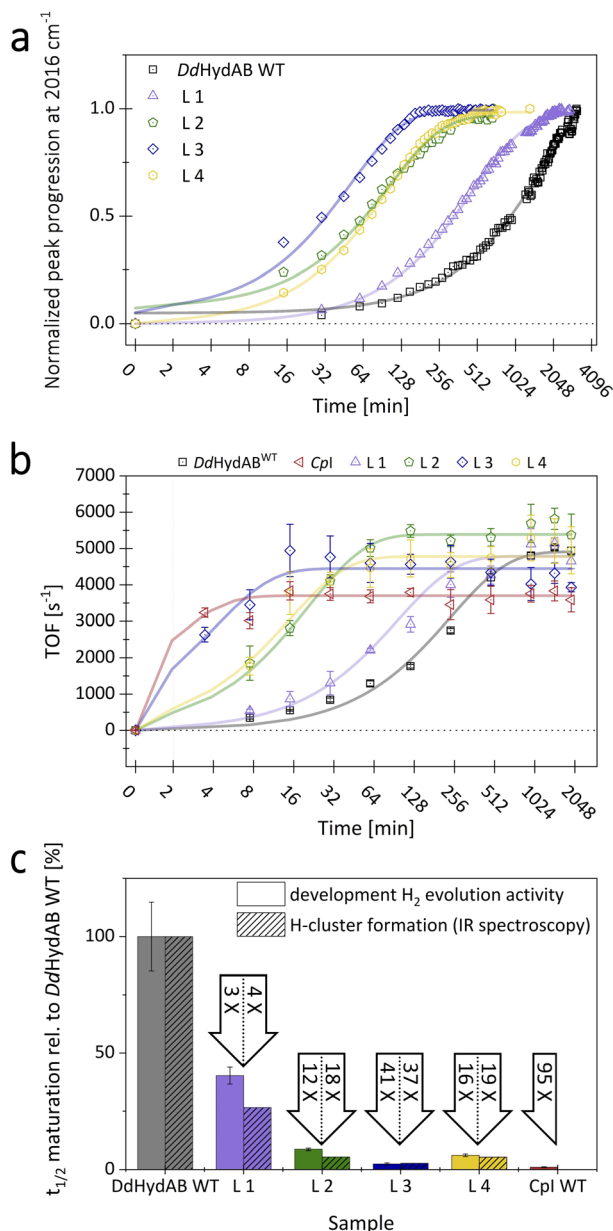


Fig. 3 Comparison of *in vitro* maturation rates for the apo-states of *DdHydAB*^{WT} and monomeric fusion variants L1–L4. (a) H-cluster formation was followed *via* FTIR spectroscopy by the progression of the amplitude of the indicator peak for H-cluster state $H_{ox}CO$ at 2016 cm^{-1} . (b) *In vitro* maturation was tracked by monitoring the development of H_2 -evolution activity over time in aliquots of *in vitro* maturation stock samples (error bars: standard deviation of biological duplicates ($n = 2$) and technical triplicates ($n = 3$)). (c) Comparison of the half-times for H-cluster formation and activity development during *in vitro* maturation for variants L1–L4 relative to the heterodimeric WT. The factor by which the half-time of the *in vitro* maturation process is decreased is indicated by the corresponding arrow above each bar (error bars: standard error of the fit parameters based on a covariance matrix of the parameters).

via gas chromatography (Fig. 3b).^{20,24} This approach enabled the inclusion of *Cpl* as a positive control. The hydrogen evolution assay mirrored the FTIR-based maturation trends, confirming its overall ranking among the compared enzymes. All proteins

reached their maximum activity within 34 h, consistent with previous data on the *in vitro* maturation of *DdHydAB*.²¹ While variant L1 exceeded the maturation rate of WT by reaching its full activity after approximately 7.4 h (see Table S2), it was nevertheless the slowest among the four monomeric *DdHydAB*-variants. In contrast, variant L3 showed an extraordinarily rapid activation with a half-time of merely 4 min, which nearly reached the activation rate of *Cpl* ($t_{1/2}$: 1.7 min), being one of the fastest-maturing hydrogenases reported to date (Fig. 3c).²⁰ Fusion variants L2 and L4 behaved similarly, peaking at a turnover frequency of around 5000 s^{-1} after 95 and 68 min, respectively, with variant L2 achieving a higher turnover rate than L4 and the heterodimeric wild-type. To verify that *in vitro* maturation is discontinued once the aliquot is added to the *in vitro* assay reaction mix, we prepared a control sample using the fastest-maturing *DdHydAB*-variant L3. $2Fe_H^{MIM}$ and the apo-form of variant L3 were separately diluted and combined in the assay mix right before sample incubation. Under these conditions even L3 showed negligible H_2 -production ($37 s^{-1}$), confirming that the *in vitro* maturation process is effectively stopped when added to the reaction mix of the assay.

Although the enzyme to $2Fe_H^{MIM}$ ratio (1 : 3) was the same for both approaches compared here, other experimental conditions differed due to the constraints of the respective methods (*e.g.*, enzyme concentration and incubation in open *vs.* closed vessels). These differences account for the 5–9-fold deviations in absolute $t_{1/2}$ -values for the same enzymes (compare absolute $t_{1/2}$ values for FTIR and the HER of the same protein in Table S2). We normalized the half-times of *in vitro* maturation for the different fusion proteins relative to the wild-type (set to 100%). For the same enzyme variants, the relative $t_{1/2}$ values (rate increase compared to wild-type) obtained on the one hand by monitoring H-cluster formation *via* FTIR and on the other hand by measuring H_2 -evolution activity development (HER) aligned closely for all variants (L1: 4-fold/3-fold; L2: 18-fold/12-fold; L3: 37-fold/41-fold; L4: 19-fold/16-fold). Remarkably, L3 shows, in both approaches, by far the highest relative increases compared to the other variants. Its relative HER rate increase is only 2.3-fold lower than the relative rate observed for the monomeric reference enzyme *Cpl* (95-fold; see Table S2). Overall, all linkers enabled a significant increase in the *in vitro* maturation rate, which was likewise reflected in the rates of H-cluster formation and activity increase. This suggests that the heterodimeric state either slows down subcluster coupling or affects an earlier step in the maturation process. This bottleneck could involve the uptake of the $2Fe_H$ precursor, which may be blocked by an unfavorable configuration of the free N-terminus at the lock-element or arise from steps further downstream. It may concern an early state, with the precursor complex $2Fe_H^{MIM}$ being embedded in the binding pocket but not yet covalently coupled to the $4Fe_H$ -subcluster. Such a state was postulated and experimentally supported to precede subcluster coupling as an inactive intermediate species by Megarity and coauthors.²⁵ Cluster coupling only happens if the applied potential is high enough to oxidize the $4Fe_H$ -subcluster, while the uncoupled intermediate state is conserved at potentials below -0.54 V. Megarity *et al.* described the $2Fe_H^{MIM}$ complex to be tightly



bound, withstanding several buffer exchanges even at potentials below -0.54 V, where no subcluster coupling occurred. This suggests that after 2Fe_H uptake, stable binding site closure occurs and is a prerequisite of subcluster coupling. Thus, the unbound $2\text{Fe}_\text{H}^{\text{MIM}}$ complex could be easily lost if binding site lockage *via* the plug, lock, and lid reconfiguration is incomplete.^{24,30} It can be assumed that subcluster-coupling *via* the formation of a bridging Cys ligand not only requires a suitable potential range but also a complete and stable H-cluster environment, granted by a sturdy closure of the open 2Fe_H binding site. The connection between the heterodimeric state and slow maturation rate supports this hypothesis. In $Dd\text{HydAB}^{\text{WT}}$, the flexible N-terminus of the small subunit slows down the locking of the plug-element over the 2Fe_H binding site after $2\text{Fe}_\text{H}^{\text{MIM}}$ uptake, as it lacks the stable resistance and anchoring of the terminal helix segment that is needed for a rapid structural rearrangement at the other end of the helix (Fig. 1a, b, and S1b).

Enzyme yields and O_2 protection

Considering that $Dd\text{HydAB}$ is a prime candidate for future biotechnological applications that include H_2 -driven or H_2 -producing processes, it is crucial to preserve the beneficial characteristics of the original enzyme in its monomeric variants. Therefore, we investigated the monomeric fusion proteins for their heterologous expression yields, [FeS]-cluster occupancy (Fig. 4), and the possibility to provide effective protection against O_2 by transiently turning the sample into the sulfide-bound H_{inact} state (Fig. 5).¹⁴

We determined differences in the yields from heterologous expression and protein purification between the monomeric variants and the heterodimeric wild-type (Fig. 4). Based on an average of three independent protein purifications, we calculated for L3 the lowest relative yield of 46% compared to $Dd\text{HydAB}^{\text{WT}}$. For fusion variants L4 and L1, the relative yields were slightly higher (50% and 58%). Variant L2, which already enabled an approximately 15-fold maturation rate increase, not

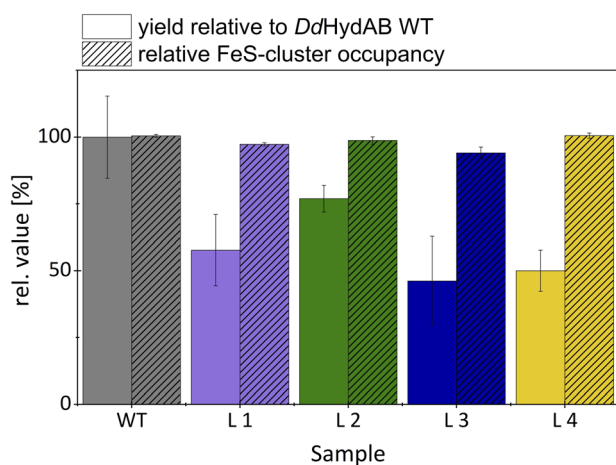


Fig. 4 Protein yields from heterologous expression and [FeS]-cluster occupancies of the monomeric $Dd\text{HydAB}$ -variants relative to wild-type. Error bars represent standard deviations from three independent purifications and technical replicates ($n = 3$).

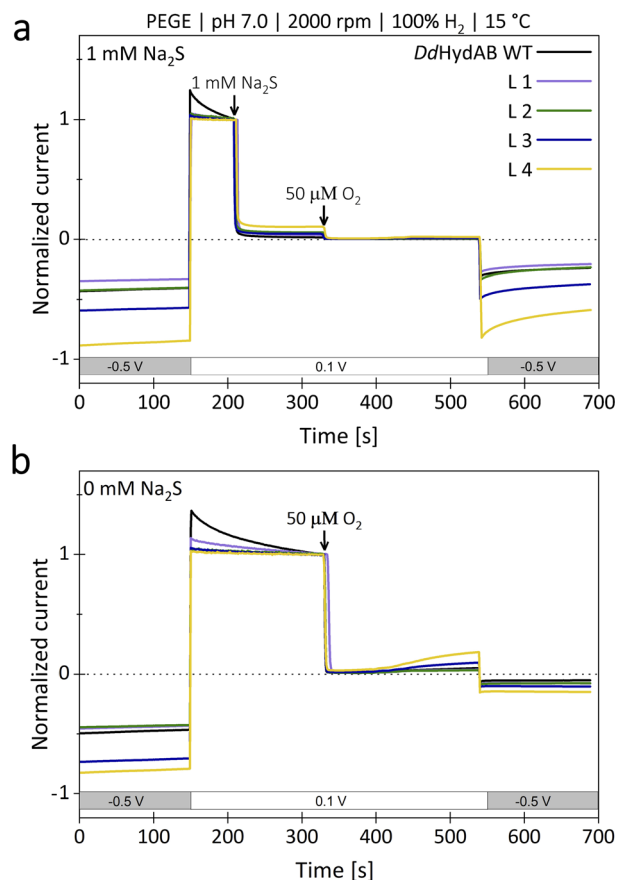


Fig. 5 Potential-step voltammetry of $Dd\text{HydAB}^{\text{WT}}$ compared to the monomeric variants L1–L4 (a). When adding Na_2S under oxidative conditions (0.1 V vs. SHE), all monomeric variants show the same, well-characterized inactivation as the heterodimeric $Dd\text{HydAB}^{\text{WT}}$. When adding O_2 -saturated buffer to a final O_2 concentration of 50 μM under these conditions, only a slight decrease in current can be observed for all proteins. By shifting the potential of the working electrode back to reducing conditions (0.5 V vs. SHE), all monomeric proteins show a residual activity comparable to that of $Dd\text{HydAB}^{\text{WT}}$. (b) If O_2 -saturated buffer is added without prior addition of Na_2S , almost no activity is recovered under reducing conditions, which likewise applies to all tested proteins.

only reached and maintained the highest H_2 -evolution rate (118% compared to wild-type) but also exhibited the highest relative expression yield of 77%. The retention of the protective effect of sulfide treatment against molecular O_2 was examined *via* potential-step voltammetry with (Fig. 5a) and without the addition of 1 mM Na_2S prior to an injection of 50 μM O_2 (Fig. 5b), as previously described by Rodríguez-Maciá and coworkers.^{14,15} After determining the original level of catalytic current resulting from H_2 -evolution at -0.5 V vs. SHE, the potential was stepped up to 0.1 V vs. SHE supporting H_2 -oxidation. Under such oxidative conditions, as soon as 1 mM Na_2S is added to the measuring solution, all four monomeric variants show a rapid and complete loss of catalytic current just like $Dd\text{HydAB}^{\text{WT}}$ (Fig. 5a), indicating for all proteins a likewise effective and potential-dependent inactivation *via* sulfide-binding to the active site. To further verify that the attached sulfide ligand protects the active site against O_2 -induced



damage, O₂-saturated buffer was injected to reach a final O₂ concentration of 50 μM in the measuring buffer. For the sulfide-treated enzyme sample, the addition of O₂-saturated buffer only caused a minimal loss of residual activity. When triggering the release of sulfide from the active site by switching back to −0.5 V vs. SHE, the activities of all investigated proteins recovered accordingly. A comparison between the initial H₂-evolution current and the reductive current after O₂-exposure suggests for WT and all monomeric variants a comparable decrease in activity, mostly due to film loss from the electrode surface, while in all cases only a very small fraction of the enzyme is irreversibly inactivated by O₂. To compare the impact of O₂-exposure on enzyme activity between sulfide protected and unprotected samples, a second series of experiments was performed in which O₂ exposure was carried out without prior sulfide addition (Fig. 5b). As expected, all enzyme variants show an instantaneous and complete loss of catalytic activity upon O₂-addition, which is not recovered upon stepping the potential back to −0.5 V vs. SHE. Thus, all monomeric proteins can be effectively protected against irreversible O₂-induced inactivation *via* pre-exposure to sulfide (Fig. 5a). When kept anaerobically, each variant retains a high catalytic activity comparable to DdHydAB^{WT} (Fig. 3b).

Structural analysis

When designing the four linker elements, we aimed at covering a range of lengths and amino acid compositions, considering

features such as bulkier side chains that influence the local flexibility based on preceding AlphaFold modelling.⁴⁴ Accordingly, different characteristics emerged in maturation efficiency, H₂-evolution activity, and heterologous expression. To elucidate the structural basis for the strong impact of subunit fusion on enzyme maturation of DdHydAB and to further identify the steric individualities responsible for differences in maturation rate increases between L1 and the other variants, we solved the crystal structures of variants L1–L4. Crystals of variants L1 and L4 were obtained under similar conditions to those previously established for the wild-type,⁴⁵ whereas variants L2 and L3 required additional optimizations of the crystallization conditions. Diffraction data were collected at beamline P11 at PETRA III (DESY, Hamburg, Germany)^{46,47} and structures of all monomeric variants were determined at resolutions of 1.05–1.78 Å (Table S3).

The linker regions in variants L1, L2, and L3 showed clear electron densities, enabling the modelling of all amino acid residues as well as the tracking of interactions with or within the corresponding loop region (Fig. 2b, S5f, k and p). However, the structures of variant L4 lacked sufficient electron density in the linker segment (Fig. S10), which indicates high mobility of the linker residues. Overall, the protein backbones of the variants are very similar to the wild-type protein, suggesting that the linkers do not affect the overall fold of the protein. We employed ChimeraX⁴⁸ to superpose a Boltz-2 model⁴⁹ of L4 with the well-resolved remaining part of the corresponding crystal

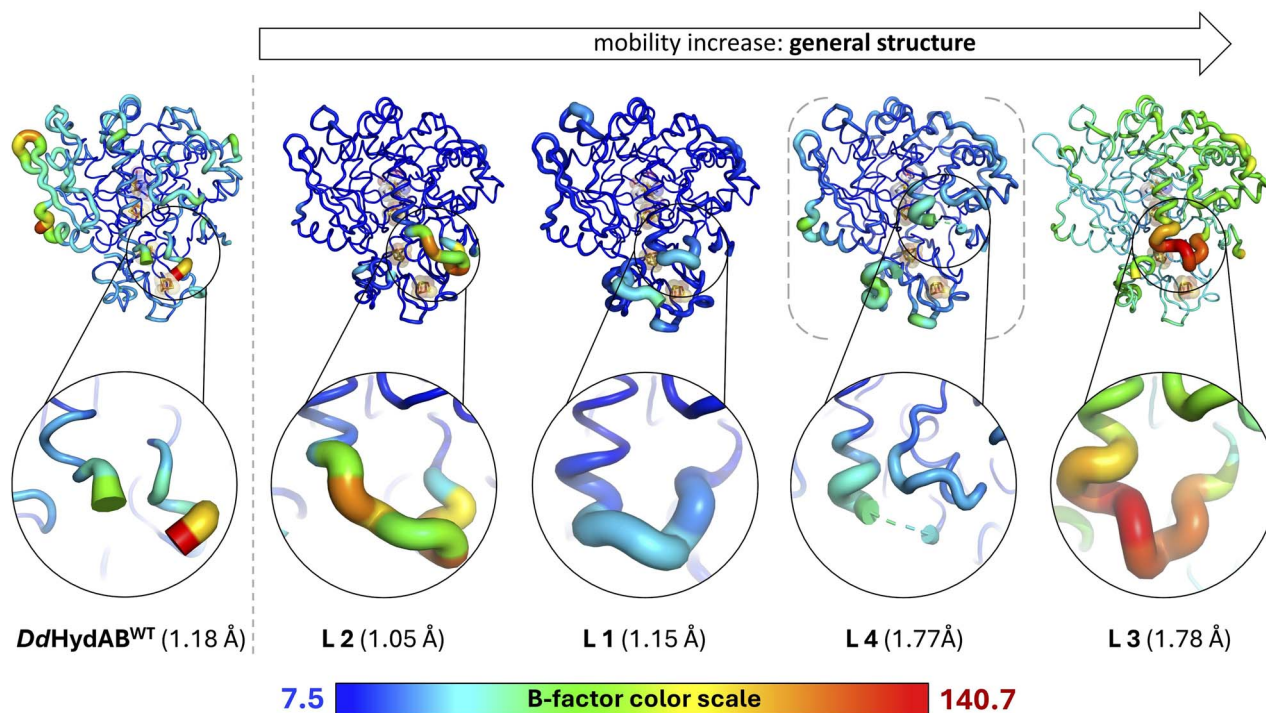


Fig. 6 Illustration of the impact of local *B*-factor differences on protein stability (general structure) and maturation-rate increase (linker). Structures of DdHydAB^{WT} (PDB ID: 9QD6) and SU-fusion variants L1–L4 (PDB IDs: 8RU6; 9GNK; 9GBU; 8RYH) are shown in the *B*-factor putty mode of PyMOL with close-ups of the linker elements. The highest *B*-factor among all structures is colored red and the lowest in dark blue, as indicated by the *B*-factor scale bar. The maximum and minimum values are presented at the corresponding ends of the color scale. The thickness of the protein backbone is also proportional to the *B*-factors. The H-cluster and accessory [FeS]-clusters are displayed as sticks and semi-transparent spheres. This image was generated with PyMOL 2.1.0.



structure, offering a plausible suggestion for the unresolved linker region in the L4 structure (see Fig. S10).

The 2Fe_H -subcluster is present in all structures, but with different occupancy levels. In variants L1 and L2, the occupancy of 2Fe_H has been refined to 0.9 to better match the experimental electron density, while in variants L3 and L4 the 2Fe_H -subcluster is fully occupied.

L1 and L2 show a well-defined diatomic ligand bound to the open coordination site of Fe_d with occupancies of 0.5 and 0.9, respectively. We propose that this diatomic ligand corresponds to a CO molecule derived from the synthetic complex used for the maturation of the enzymes. Electron density maps of L3 and L4 show very little positive density at the open coordination site in the $F_\text{o}-F_\text{c}$ maps, suggesting that most of the molecules in these crystals possess an open coordination site.

While the structures of all the variants are overall very similar, we observed significant differences in the local mobility between the variants, as highlighted by the visualization of the crystallographic displacement factors (B -factors) in Fig. 6. The displacement of atoms from their mean position in a crystal structure may be the result of temperature-dependent atomic vibrations or static disorder in a crystal lattice. Since all structures were determined from crystals with the same lattice, the different B -factors observed for wild-type and the monomeric variants are more likely to be the result of local mobility. As expected, the heterodimeric $Dd\text{HydAB}^{\text{WT}}$ structure shows increased mobility values at the termini of the two subunits, especially at the carboxy terminus of the LSU. The overall mobility of the structures for L1 and L2 is very low; however, the linkers of all variants show individual local flexibilities. The linker of the rather slowly maturing variant L1 is the least mobile one, while the longer linker of L2 is significantly more flexible. Interestingly, the linker of the fastest-maturing variant L3 shows the highest mobility. The L3 structure also shows the highest overall mobility range. However, as the structure of variant L3 was solved at a comparatively lower resolution, the increased B -factors may not be directly comparable with the B -factors determined for the high-resolution structures of L1 and L2. Crystal structures of L3, unlike those of L1 and L2, showed very low occupancies for the surplus CO ligand at the substrate binding site. This may suggest a lower CO-binding affinity or a faster release of the extra ligand, which could result from increased structural mobility compared to the wild-type and the other variants. Even though L3 indeed exhibits the highest k_{react} value among the five proteins, the difference to the other proteins is not very pronounced (see Table S1). Linker 1 showed by far the least impact on the enzyme maturation rate, being 4.5- to 10-fold lower compared to the other fusion proteins (Fig. 3c). According to a comparison of the side chain packing densities in the linker regions of X-ray structures obtained for L1, L2, and L3, the linker of L1 seems to be more tightly packed, while the side chains of linkers in L2 and L3 are comparatively more loosely embedded (Fig. S11). Thus, they may provide sufficient room for flexible movements (Fig. 2b, S5a, f, k and p). This is likewise reflected in the rather low B -factors assigned to linker 1 (Fig. 6), resulting in high local structural quality, while the higher B -factors for the linkers in

structures of L2 and L3 rendered the local quality in the linker regions rather low compared to the remaining structural parts (Fig. 6). These results suggest that stabilization of the N-terminal part of the lock-helix by a linker supports the transition to the stably locked configuration after 2Fe_H precursor uptake. However, a certain degree of residual flexibility may still be necessary to permit and buffer movements during the structural reconfiguration. In the final section, we examine whether the unusually slow maturation rate of the heterodimeric wild-type enzyme is an intrinsic property of $Dd\text{HydAB}$ or merely an artifact of the non-native conditions used for *in vitro* maturation.

Impact of the maturase HydF on maturation

So far, we have only investigated the change in maturation rate upon monomerization with different linkers of $Dd\text{HydAB}$ by activating the apo-enzyme *in vitro* with free $2\text{Fe}_\text{H}^{\text{MIM}}$. However, *in vivo*, the 2Fe_H precursor is assembled by the maturases HydE and G on the scaffold maturase HydF. Once loaded, HydF is

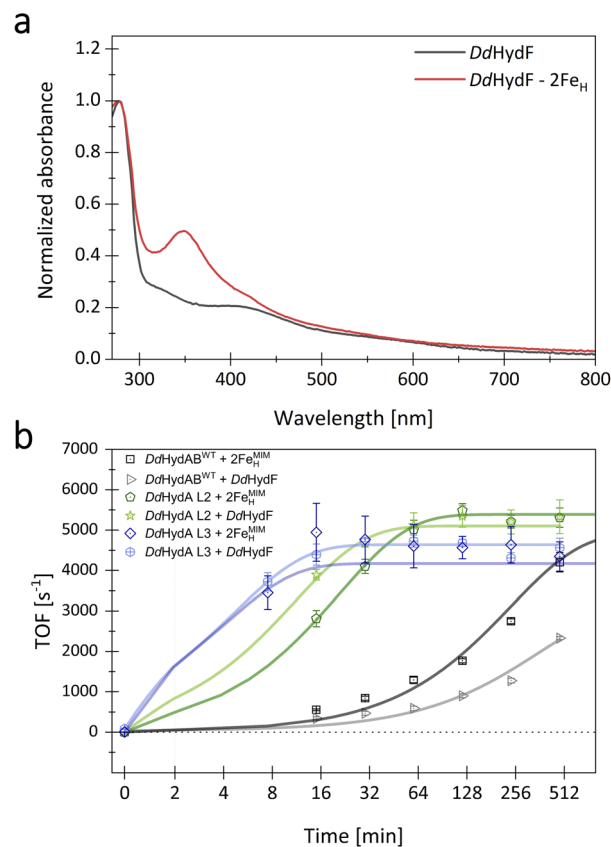


Fig. 7 Loading and use of $Dd\text{HydF}$ as a mediator for *in vitro* maturation of $Dd\text{HydAB}^{\text{WT}}$ and subunit-fusion variant L2. (a) UV-Vis-spectra of as-purified $Dd\text{HydF}$ and $Dd\text{HydF}$ loaded with $2\text{Fe}_\text{H}^{\text{MIM}}$. The spectra were normalized by setting the peak at 280 nm to 1.0. Protein samples were diluted with 0.1 M Tris-HCl buffer, pH 8. Both spectra are in line with those earlier reported for the apo- and cofactor-loaded holo-state of HydF from *Thermotoga maritima*.²⁸ (b) Monitoring the time-dependent increase of H_2 -evolution activity for $Dd\text{HydAB}^{\text{WT}}$ and subunit-fusion variants L2 and L3 during *in vitro* maturation of 0.16 mM hydrogenase, using either a 3-fold concentration of maturase-free $2\text{Fe}_\text{H}^{\text{MIM}}$ or holo-HydF loaded with $2\text{Fe}_\text{H}^{\text{MIM}}$.



proposed to transfer the 2Fe_H precursor to the apo-hydrogenase by forming a transient $\text{HydF}/2\text{Fe}_\text{H}^{\text{MIM}}/\text{apo-}Dd\text{HydAB}$ complex. Intermolecular interactions in this ternary complex may stabilize the N-terminus of the small subunit of $Dd\text{HydAB}$, thereby enabling fast 2Fe_H insertion, comparable to the situation in the monomeric $Dd\text{HydAB}$ versions. To investigate the influence of HydF-supported maturation, we expressed, purified, and loaded $Dd\text{HydF}$ of *D. desulfuricans* ATCC 7757 (for sequence of $Dd\text{HydF}$, see Fig. S5), as previously done to study the interactions of HydF with different [FeFe]-hydrogenases (Fig. S1).^{33–35} Loading of $Dd\text{HydF}$ was verified *via* electronic absorption spectroscopy (UV-Vis range) (Fig. 7a). HydF-supported activation was tested for the heterodimeric $Dd\text{HydAB}^{\text{WT}}$, the monomeric variant L3, showing the highest activation rate with free $2\text{Fe}_\text{H}^{\text{MIM}}$, and L2, as this variant showed an improved H_2 -evolution activity even in comparison to $Dd\text{HydAB}^{\text{WT}}$. The activation was tracked using the same assay as before for *in vitro* maturation, with free $2\text{Fe}_\text{H}^{\text{MIM}}$ replaced by equimolar amounts of $2\text{Fe}_\text{H}^{\text{MIM}}$ -loaded $Dd\text{HydF}$ (Fig. 7b). The maturation rates achieved by HydF-mediated activation of the apo-enzyme largely reflect the trends observed when activating the apo-enzyme with free $2\text{Fe}_\text{H}^{\text{MIM}}$. $Dd\text{HydAB}$ fusion variant L2 shows an even lower $t_{1/2}$ -value of 8.7 ± 0.5 min in the HydF-mediated approach, compared to 14.3 ± 0.9 min when being activated by free $2\text{Fe}_\text{H}^{\text{MIM}}$. L3 reaches a $t_{1/2}$ -value of approximately 4.7 ± 0.9 min in the HydF-supported assay, being only slightly higher than measured for the unmediated assay (2.3 ± 1.1 min). While the monomeric fusion variants were found to show largely comparable half-times for activation in the unmediated and HydF-supported assay, $Dd\text{HydAB}^{\text{WT}}$ even showed a two-fold increase in $t_{1/2}$ (341 ± 154 min compared to 166 ± 25 min). These results suggest that the slow *in vitro* maturation of $Dd\text{HydAB}^{\text{WT}}$ cannot be overcome by the formation of a transient protein–protein complex with $Dd\text{HydF}$ during the maturation process. Nevertheless, we cannot exclude alternative explanations based on the specific *in vivo* maturation conditions of periplasmic [FeFe]-hydrogenases such as $Dd\text{HydAB}^{\text{WT}}$ (see SI file S2). Our results, however, underscore the importance of subunit fusion, as exemplified by L2 and L3, for achieving high maturation rates, irrespective of the mediator function of HydF.

Conclusion

We demonstrated that using a linker with specific properties to connect the two subunits of the heterodimeric [FeFe]-hydrogenase $Dd\text{HydAB}$ significantly enhances the rate of *in vitro* maturation, while retaining the important catalytic features of this enzyme, such as high activity and affinity for the protecting inhibitor sulfide. Our data provide important insights into the mechanism and rate-limiting factors of cofactor maturation. Our results support the hypothesis of a maturation mechanism where $2\text{Fe}_\text{H}^{\text{MIM}}$ must first be tightly bound and stabilized in the binding pocket by reconfigurations in the plug-lock-lid system before subcluster coupling can occur. Consequently, efficient *in vitro* maturation depends on the right degree of flexibility of the structural elements involved. This flexibility must allow for a fully open conformation to

enable efficient and fast 2Fe_H uptake *via* the maturation channel to the 4Fe_H -moiety of the H-cluster. At the same time, the structural elements still need to be rigid enough to sufficiently support the conformational changes required to close and lock up the maturation channel before forcing the 2Fe_H -subcluster into its rotated, catalytically preferred conformation and enabling sub-cluster coupling. On the path toward establishing $Dd\text{HydAB}$ as a competitive and fully regenerative biocatalyst in future applications, subunit-fusion allowed us to overcome the hurdle of the extremely slow catalytic activation of the heterodimeric wild-type enzyme.

Author contributions

M. W., J. J., N. P., I. S. and C. L. conceived and planned the experiments. J. J., M. W., J. B., M. M. and K. B. carried out the experiments. U. P. A. and S. Y. provided the cofactor for the *in vitro* maturation experiments. M. W., J. J., N. P., I. S., K. B. and J. A. B. contributed to the interpretation of the results. M. W., J. J. and N. P. took the lead in writing the manuscript. All authors provided critical feedback and helped shape the research, analysis and manuscript.

Conflicts of interest

There are no conflicts to declare.

Data availability

The data supporting this article have been included as part of the supplementary information (SI). Supplementary information: File S1: materials and methods, text, Fig. S1–S12, and Tables S1–S4. A separate file (deposition reports) comprises all deposition reports associated with the newly published X-ray structure data summarized in Table S3. File S2: genomic data, polypeptide sequences, and further information on the origin and phylogenetic distribution of heterodimeric M2-type [FeFe]-hydrogenases. It also provides a hypothesis for the evolutionary pressures that may have led to the unusual heterodimeric configuration of periplasmic M2-type [FeFe]-hydrogenases and may help explain the low *in vitro* maturation rate of this M2 subtype. See DOI: <https://doi.org/10.1039/d5sc07299a>.

Acknowledgements

J. J. gratefully acknowledges support by the German Academic Scholarship Foundation. N. P. and M. W. acknowledge financial support by the BMBF funding program “Future Technologies for the Industrial Bioeconomy” (SynHydro3, Project No. 031B1123A and 031B1123C). M. W. was further funded by the Deutsche Forschungsgemeinschaft (DFG) – Project No. 561053011. N. P. was further funded by the FNR project SynergyFuels (16RK34003K) and by the Deutsche Forschungsgemeinschaft (DFG) – Project No. 490864939 (Prosecco). This work was financially supported by the Deutsche Forschungsgemeinschaft (DFG, German Research Foundation) Priority Program “Iron–Sulfur for Life: Cooperative Function of Iron–Sulfur Centers in Assembly,



Biosynthesis, Catalysis and Disease" (SPP 1927) Projects BI 2198/1-1 (J. A. B.) and IS 1476/4-1 (I. S.). I. S. further acknowledges support from COST Action FeSImmChemNet (CA21115) supported by COST (European Cooperation in Science and Technology). U.-P. A. was funded by the DFG under Germany's Excellence Strategy - EXC 2033 - 390677874 - RESOLV. CL acknowledges funding by Agence nationale de la recherche (ANR) (HYDRES, ANR-23-CE50-0016) and by the French government under the France 2030 investment plan, as part of the Initiative d'Excellence d'Aix-Marseille Université - A*MIDEX, AMX-22-RE-AB-097. A*Midex is a foundation of Aix Marseille University. We acknowledge DESY (Hamburg, Germany), a member of the Helmholtz Association HGF, for the provision of experimental facilities. Parts of this research were carried out at PETRA III. Data were collected using beamline P11 operated by DESY Photon Science. We would like to thank Johanna Hakanpää and Guillaume Pompidor for assistance during the experiments. Beamtime was allocated for proposal I-20230763.

Notes and references

- 1 M. Qiao, Catalyzing the chemical industry, *Nat. Chem. Eng.*, 2024, **1**, 270–272.
- 2 J. M. Le and K. L. Bren, Engineered Enzymes and Bioinspired Catalysts for Energy Conversion, *ACS Energy Lett.*, 2019, **4**, 2168–2180.
- 3 A. A. Karyakin, S. V. Morozov, E. E. Karyakina, N. A. Zorin, V. V. Perelygin and S. Cosnier, Hydrogenase electrodes for fuel cells, *Biochem. Soc. Trans.*, 2005, **33**, 73–75.
- 4 K. A. Vincent, J. A. Cracknell, A. Parkin and F. A. Armstrong, Hydrogen cycling by enzymes: electrocatalysis and implications for future energy technology, *Dalton Trans.*, 2005, 3397–3403, DOI: [10.1039/B508520A](https://doi.org/10.1039/B508520A).
- 5 K. A. Vincent, J. A. Cracknell, O. Lenz, I. Zebger, B. Friedrich and F. A. Armstrong, Electrocatalytic hydrogen oxidation by an enzyme at high carbon monoxide or oxygen levels, *Proc. Natl. Acad. Sci. U. S. A.*, 2005, **102**, 16951–16954.
- 6 A. A. Oughli, F. Conzuelo, M. Winkler, T. Happe, W. Lubitz, W. Schuhmann, O. Rudiger and N. Plumere, A redox hydrogel protects the O₂-sensitive [FeFe]-hydrogenase from *Chlamydomonas reinhardtii* from oxidative damage, *Angew Chem. Int. Ed. Engl.*, 2015, **54**, 12329–12333.
- 7 S. Hardt, S. Stapf, D. T. Filmon, J. A. Birrell, O. Rüdiger, V. Fourmond, C. Léger and N. Plumeré, Reversible H₂ oxidation and evolution by hydrogenase embedded in a redox polymer film, *Nat. Catal.*, 2021, **4**, 251–258.
- 8 S. V. Hexter, F. Grey, T. Happe, V. Climent and F. A. Armstrong, Electrocatalytic mechanism of reversible hydrogen cycling by enzymes and distinctions between the major classes of hydrogenases, *Proc. Natl. Acad. Sci. U. S. A.*, 2012, **109**, 11516–11521.
- 9 J. T. Kleinhaus, F. Wittkamp, S. Yadav, D. Siegmund and U.-P. Apfel, [FeFe]-Hydrogenases: maturation and reactivity of enzymatic systems and overview of biomimetic models, *Chem. Soc. Rev.*, 2021, **50**, 1668–1784.
- 10 K. Pandey, S. T. Islam, T. Happe and F. A. Armstrong, Frequency and potential dependence of reversible electrocatalytic hydrogen interconversion by [FeFe]-hydrogenases, *Proc. Natl. Acad. Sci. U. S. A.*, 2017, **114**, 3843–3848.
- 11 M. Heghmanns, A. Rutz, Y. Kutin, V. Engelbrecht, M. Winkler, T. Happe and M. Kasanmascheff, The oxygen-resistant [FeFe]-hydrogenase Cba5H harbors an unknown radical signal, *Chem. Sci.*, 2022, **13**, 7289–7294.
- 12 M. Winkler, J. Duan, A. Rutz, C. Felbek, L. Scholtyssek, O. Lampret, J. Jaenecke, U. P. Apfel, G. Gilardi, F. Valetti, V. Fourmond, E. Hofmann, C. Leger and T. Happe, A safety cap protects hydrogenase from oxygen attack, *Nat. Commun.*, 2021, **12**, 756.
- 13 S. Morra, M. Arizzi, F. Valetti and G. Gilardi, Oxygen Stability in the New [FeFe]-Hydrogenase from *Clostridium beijerinckii* SM10 (CbA5H), *Biochemistry*, 2016, **55**, 5897–5900.
- 14 P. Rodríguez-Maciá, L. M. Galle, R. Björnsson, C. Lorent, I. Zebger, Y. Yoda, S. P. Cramer, S. DeBeer, I. Span and J. A. Birrell, Caught in the Hinact: Crystal Structure and Spectroscopy Reveal a Sulfur Bound to the Active Site of an O₂-stable State of [FeFe] Hydrogenase, *Angew Chem. Int. Ed. Engl.*, 2020, **59**, 16786–16794.
- 15 P. Rodríguez-Maciá, E. J. Reijerse, M. van Gastel, S. DeBeer, W. Lubitz, O. Rüdiger and J. A. Birrell, Sulfide Protects [FeFe] Hydrogenases From O₂, *J. Am. Chem. Soc.*, 2018, **140**, 9346–9350.
- 16 A. A. Oughli, S. Hardt, O. Rudiger, J. A. Birrell and N. Plumere, Reactivation of sulfide-protected [FeFe] hydrogenase in a redox-active hydrogel, *Chem. Commun.*, 2020, **56**, 9958–9961.
- 17 D. T. Filmon, J. Jaenecke, M. Winkler, V. Fourmond, C. Leger and N. Plumere, Turning the FeFe hydrogenase from *Clostridium beijerinckii* into an efficient H₂ oxidation catalyst using a redox-active matrix, *Proc. Natl. Acad. Sci. U. S. A.*, 2025, **122**, e2514698122.
- 18 S. E. Cleary, S. J. Hall, R. Galan-Bataller, T. C. Lurshay, C. Hancox, J. J. Williamson, J. T. Heap, H. A. Reeve and S. Morra, Scalable Bioreactor Production of an O₂-Protected [FeFe]-Hydrogenase Enables Simple Aerobic Handling for Clean Chemical Synthesis, *ChemCatChem*, 2024, **16**, e202400193.
- 19 S. Yadav, R. Haas, E. B. Boydas, M. Roemelt, T. Happe, U.-P. Apfel and S. T. Stripp, Oxygen sensitivity of [FeFe]-hydrogenase: a comparative study of active site mimics inside vs. outside the enzyme, *Phys. Chem. Chem. Phys.*, 2024, **26**, 19105–19116.
- 20 J. Esselborn, C. Lambertz, A. Adamska-Venkatesh, T. Simmons, G. Berggren, J. Noth, J. Siebel, A. Hemschemeier, V. Artero, E. Reijerse, M. Fontecave, W. Lubitz and T. Happe, Spontaneous activation of [FeFe]-hydrogenases by an inorganic [2Fe] active site mimic, *Nat. Chem. Biol.*, 2013, **9**, 607–609.
- 21 J. A. Birrell, K. Wrede, K. Pawlak, P. Rodríguez-Maciá, O. Rüdiger, E. J. Reijerse and W. Lubitz, Artificial Maturation of the Highly Active Heterodimeric [FeFe] Hydrogenase from *Desulfovibrio desulfuricans* ATCC 7757, *Isr. J. Chem.*, 2016, **56**, 852–863.



- 22 E. C. Hatchikian, N. Forget, V. M. Fernandez, R. Williams and R. Cammack, Further characterization of the [Fe]-hydrogenase from *Desulfovibrio desulfuricans* ATCC 7757, *Eur. J. Biochem.*, 1992, **209**, 357–365.
- 23 B. R. Glick, W. G. Martin and S. M. Martin, Purification and properties of the periplasmic hydrogenase from *Desulfovibrio desulfuricans*, *Can. J. Microbiol.*, 1980, **26**, 1214–1223.
- 24 O. Lampret, J. Esselborn, R. Haas, A. Rutz, R. L. Booth, L. Kertess, F. Wittkamp, C. F. Megarity, F. A. Armstrong, M. Winkler and T. Happe, The final steps of [FeFe]-hydrogenase maturation, *Proc. Natl. Acad. Sci. U. S. A.*, 2019, **116**, 15802–15810.
- 25 C. F. Megarity, J. Esselborn, S. V. Hexter, F. Wittkamp, U. P. Apfel, T. Happe and F. A. Armstrong, Electrochemical Investigations of the Mechanism of Assembly of the Active-Site H-Cluster of [FeFe]-Hydrogenases, *J. Am. Chem. Soc.*, 2016, **138**, 15227–15233.
- 26 D. W. Mulder, E. S. Boyd, R. Sarma, R. K. Lange, J. A. Endrizzi, J. B. Broderick and J. W. Peters, Stepwise [FeFe]-hydrogenase H-cluster assembly revealed in the structure of HydA(DeltaEFG), *Nature*, 2010, **465**, 248–251.
- 27 B. Németh, M. Senger, H. J. Redman, P. Ceccaldi, J. Broderick, A. Magnuson, S. T. Stripp, M. Haumann and G. Berggren, [FeFe]-hydrogenase maturation: H-cluster assembly intermediates tracked by electron paramagnetic resonance, infrared, and X-ray absorption spectroscopy, *JBC, J. Biol. Inorg. Chem.*, 2020, **25**, 777–788.
- 28 R. Haas, O. Lampret, S. Yadav, U.-P. Apfel and T. Happe, A Conserved Binding Pocket in HydF is Essential for Biological Assembly and Coordination of the Diiron Site of [FeFe]-Hydrogenases, *J. Am. Chem. Soc.*, 2024, **146**, 15771–15778.
- 29 A. Pagnier, B. Balci, E. M. Shepard, W. E. Broderick and J. B. Broderick, [FeFe]-Hydrogenase In Vitro Maturation, *Angew. Chem., Int. Ed.*, 2022, **61**, e202212074.
- 30 J. Esselborn, N. Muraki, K. Klein, V. Engelbrecht, N. Metzler-Nolte, U. P. Apfel, E. Hofmann, G. Kurisu and T. Happe, A structural view of synthetic cofactor integration into [FeFe]-hydrogenases, *Chem. Sci.*, 2016, **7**, 959–968.
- 31 Y. Zhang, L. Tao, T. J. Woods, R. D. Britt and T. B. Rauchfuss, Organometallic Fe₂(μ-SH)₂(CO)₄(CN)₂ Cluster Allows the Biosynthesis of the [FeFe]-Hydrogenase with Only the HydF Maturase, *J. Am. Chem. Soc.*, 2022, **144**, 1534–1538.
- 32 R. D. Britt, T. B. Rauchfuss and G. Rao, The H-cluster of [FeFe] Hydrogenases: Its Enzymatic Synthesis and Parallel Inorganic Semisynthesis, *Acc. Chem. Res.*, 2024, **57**, 1941–1950.
- 33 B. Németh, H. Land, A. Magnuson, A. Hofer and G. Berggren, The maturase HydF enables [FeFe] hydrogenase assembly via transient, cofactor-dependent interactions, *J. Biol. Chem.*, 2020, **295**, 11891–11901.
- 34 S. E. McGlynn, E. M. Shepard, M. A. Winslow, A. V. Naumov, K. S. Duschene, M. C. Posewitz, W. E. Broderick, J. B. Broderick and J. W. Peters, HydF as a scaffold protein in [FeFe] hydrogenase H-cluster biosynthesis, *FEBS Lett.*, 2008, **582**, 2183–2187.
- 35 G. Berggren, A. Adamska, C. Lambertz, T. R. Simmons, J. Esselborn, M. Atta, S. Gambarelli, J. M. Mouesca, E. Reijerse, W. Lubitz, T. Happe, V. Artero and M. Fontecave, Biomimetic assembly and activation of [FeFe]-hydrogenases, *Nature*, 2013, **499**, 66–69.
- 36 G. Caserta, L. Pecqueur, A. Adamska-Venkatesh, C. Papini, S. Roy, V. Artero, M. Atta, E. Reijerse, W. Lubitz and M. Fontecave, Structural and functional characterization of the hydrogenase-maturation HydF protein, *Nat. Chem. Biol.*, 2017, **13**, 779–784.
- 37 O. Lampret, A. Adamska-Venkatesh, H. Konegger, F. Wittkamp, U. P. Apfel, E. J. Reijerse, W. Lubitz, O. Rudiger, T. Happe and M. Winkler, Interplay between CN(-) Ligands and the Secondary Coordination Sphere of the H-Cluster in [FeFe]-Hydrogenases, *J. Am. Chem. Soc.*, 2017, **139**, 18222–18230.
- 38 A. Adamska-Venkatesh, D. Krawietz, J. Siebel, K. Weber, T. Happe, E. Reijerse and W. Lubitz, New redox states observed in [FeFe] hydrogenases reveal redox coupling within the H-cluster, *J. Am. Chem. Soc.*, 2014, **136**, 11339–11346.
- 39 Y. Nicolet, C. Piras, P. Legrand, C. E. Hatchikian and J. C. Fontecilla-Camps, *Desulfovibrio desulfuricans* iron hydrogenase: the structure shows unusual coordination to an active site Fe binuclear center, *Structure*, 1999, **7**, 13–23.
- 40 N. Amet, W.-C. Lee Hf Fau-Shen and W. C. Shen, Insertion of the designed helical linker led to increased expression of tf-based fusion proteins, *Pharm. Res.*, 2009, **26**, 523–528.
- 41 J. Guo, Z. Cheng, J. Berdychowska, X. Zhu, L. Wang, L. Peplowski and Z. Zhou, Effect and mechanism analysis of different linkers on efficient catalysis of subunit-fused nitrile hydratase, *Int. J. Biol. Macromol.*, 2021, **181**, 444–451.
- 42 D. K. Patel, D. Vettiati, D. H. Patel and G. Dave, Linkers: A synergistic way for the synthesis of chimeric proteins, *Protein Expression Purif.*, 2022, **191**, 106012.
- 43 F. Leroux, S. Dementin, B. Burlat, L. Cournac, A. Volbeda, S. Champ, L. Martin, B. Guigliarelli, P. Bertrand, J. Fontecilla-Camps, M. Rousset and C. Léger, Experimental approaches to kinetics of gas diffusion in hydrogenase, *Proc. Natl. Acad. Sci. U. S. A.*, 2008, **105**, 11188–11193.
- 44 J. Jumper, R. Evans, A. Pritzel, T. Green, M. Figurnov, O. Ronneberger, K. Tunyasuvunakool, R. Bates, A. Židek, A. Potapenko, A. Bridgland, C. Meyer, S. A. A. Kohl, A. J. Ballard, A. Cowie, B. Romera-Paredes, S. Nikolov, R. Jain, J. Adler, T. Back, S. Petersen, D. Reiman, E. Clancy, M. Zielinski, M. Steinegger, M. Pacholska, T. Berghammer, S. Bodenstein, D. Silver, O. Vinyals, A. W. Senior, K. Kavukcuoglu, P. Kohli and D. Hassabis, Highly accurate protein structure prediction with AlphaFold, *Nature*, 2021, **596**, 583–589.



- 45 M. A. Martini, K. Bikbaev, Y. Pang, C. Lorent, C. Wiemann, N. Breuer, I. Zebger, S. DeBeer, I. Span, R. Bjornsson, J. A. Birrell and P. Rodríguez-Maciá, Binding of exogenous cyanide reveals new active-site states in [FeFe] hydrogenases, *Chem. Sci.*, 2023, **14**, 2826–2838.
- 46 A. Meents, B. Reime, N. Stuebe, P. Fischer, M. Warmer, D. Göries, J. Roeber, J. Meyer, J. Fischer, A. Burkhardt, I. Vartiainen, P. Karvinen and C. David, Development of an in-vacuum X-ray microscope with cryogenic sample cooling for beamline P11 at PETRA III, *Proc. SPIE*, 2013, **8851**, 88510K.
- 47 A. Burkhardt, T. Pakendorf, B. Reime, J. Meyer, P. Fischer, N. Stübe, S. Panneerselvam, O. Lorbeer, K. Stachnik, M. Warmer, P. Rödiger, D. Göries and A. Meents, Status of the crystallography beamlines at PETRA III, *Eur. Phys. J. Plus*, 2016, **131**, 56.
- 48 E. C. Meng, T. D. Goddard, E. F. Pettersen, G. S. Couch, Z. J. Pearson, J. H. Morris and T. E. Ferrin, UCSF ChimeraX: Tools for structure building and analysis, *Protein Sci.*, 2023, **32**, e4792.
- 49 S. Passaro, G. Corso, J. Wohlwend, M. Reveiz, S. Thaler, V. R. Somnath, N. Getz, T. Portnoi, J. Roy, H. Stark, D. Kwabi-Addo, D. Beaini, T. Jaakkola and R. Barzilay, Boltz-2: Towards Accurate and Efficient Binding Affinity Prediction, *bioRxiv*, 2025, preprint, DOI: [10.1101/2025.06.14.659707](https://doi.org/10.1101/2025.06.14.659707).

