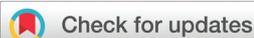


CORRECTION

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Correction: Interactions beyond H-bonding: unveiling the role of unorthodox noncovalent interactions in charged thiourea and its catalytic efficiencyPrabhakar Murugan,^{a,b} Parul Rathour,^{a,b} Dipankar Das,^{b,c} Brijesh Patel,^{a,b} Srinu Tothadi,^{b,d} Bishwajit Ganguly^{b,c} and Saravanan Subramanian^{*a,b}DOI: 10.1039/d6qo90023b
rsc.li/frontiers-organicCorrection for 'Interactions beyond H-bonding: unveiling the role of unorthodox noncovalent interactions in charged thiourea and its catalytic efficiency' by Prabhakar Murugan et al., *Org. Chem. Front.*, 2026, <https://doi.org/10.1039/d5qo01735a>.

The authors regret that Fig. 3 was not clearly readable in the original article. The correct version is shown here.

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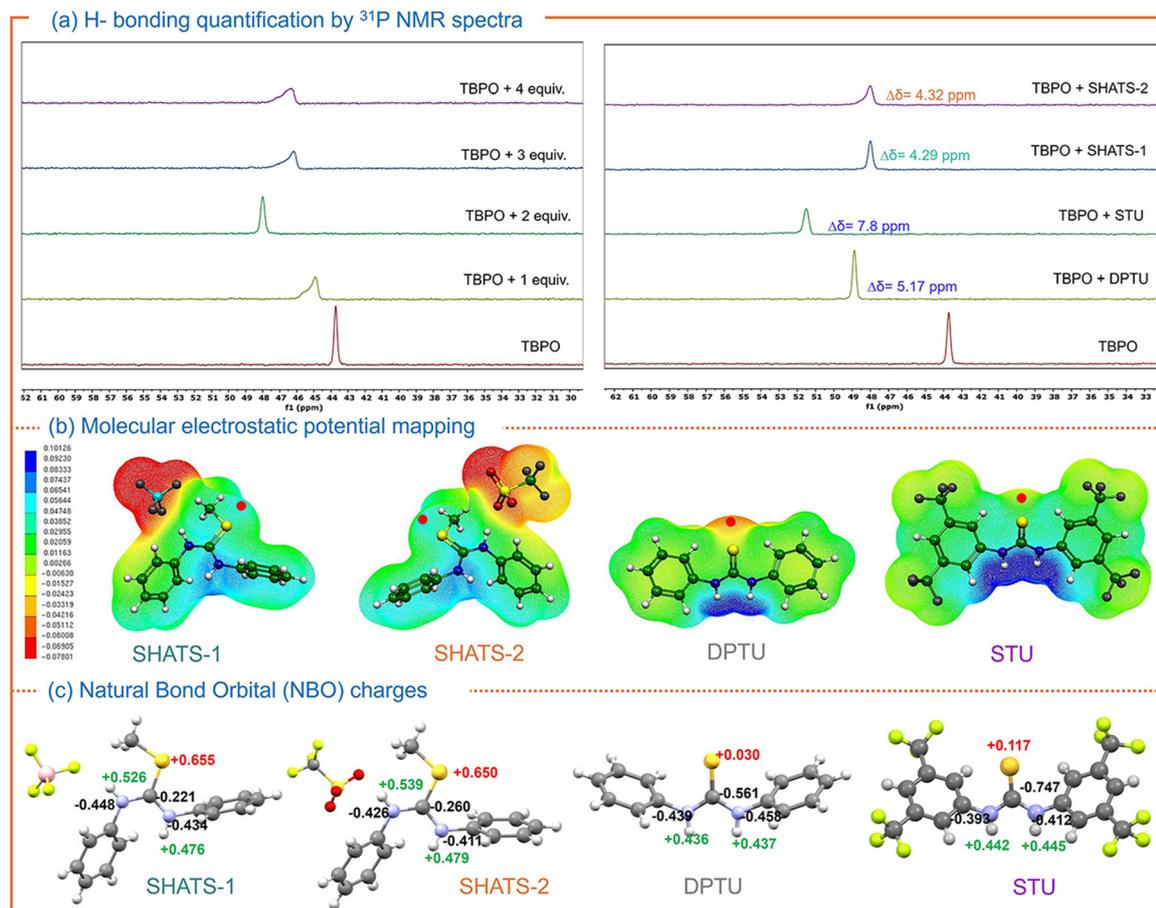


Fig. 3 Structural features of SHATS, diphenyl thiourea (DPTU) and Schriener's thiourea (STU). (a) ^{31}P NMR spectra representing titration experiments of TBPO and SHATS-1 (with varying concentrations, mol. eq.) in CCl_4 (left) and H-bonding interactions of different catalysts (right). (b) The molecular electrostatic potential surface of catalysts, wherein blue and red colors signify the positive and negative potential regions, respectively (isosurface value of 0.001 au). The σ -holes of the catalysts are shown as red-coloured dots, and the corresponding MESP values for the sulfur atom of SHATS-1, SHATS-2, DPTU and STU are 35.1, 34.3, -29.7 , and -12.6 kcal mol $^{-1}$, respectively. (c) Computed NBO charges of select atoms at the B3LYP/6-31G+(d,p) level of theory in DCM.

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

