

## CORRECTION

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Cite this: *Org. Chem. Front.*, 2026, **13**, 2657**Correction: Cu(I)-catalyzed 1,2-carbofluorination of unactivated alkenes enabled by *N*-fluorobenzamides via free radical relay**Hong Ji,<sup>a</sup> Haifeng Qiao,<sup>a</sup> Boyi Wang,<sup>a</sup> Tianyu Lu,<sup>a</sup> Weixing Chang,<sup>a</sup> Lingyan Liu\*<sup>a</sup> and Jing Li\*<sup>a,b</sup>

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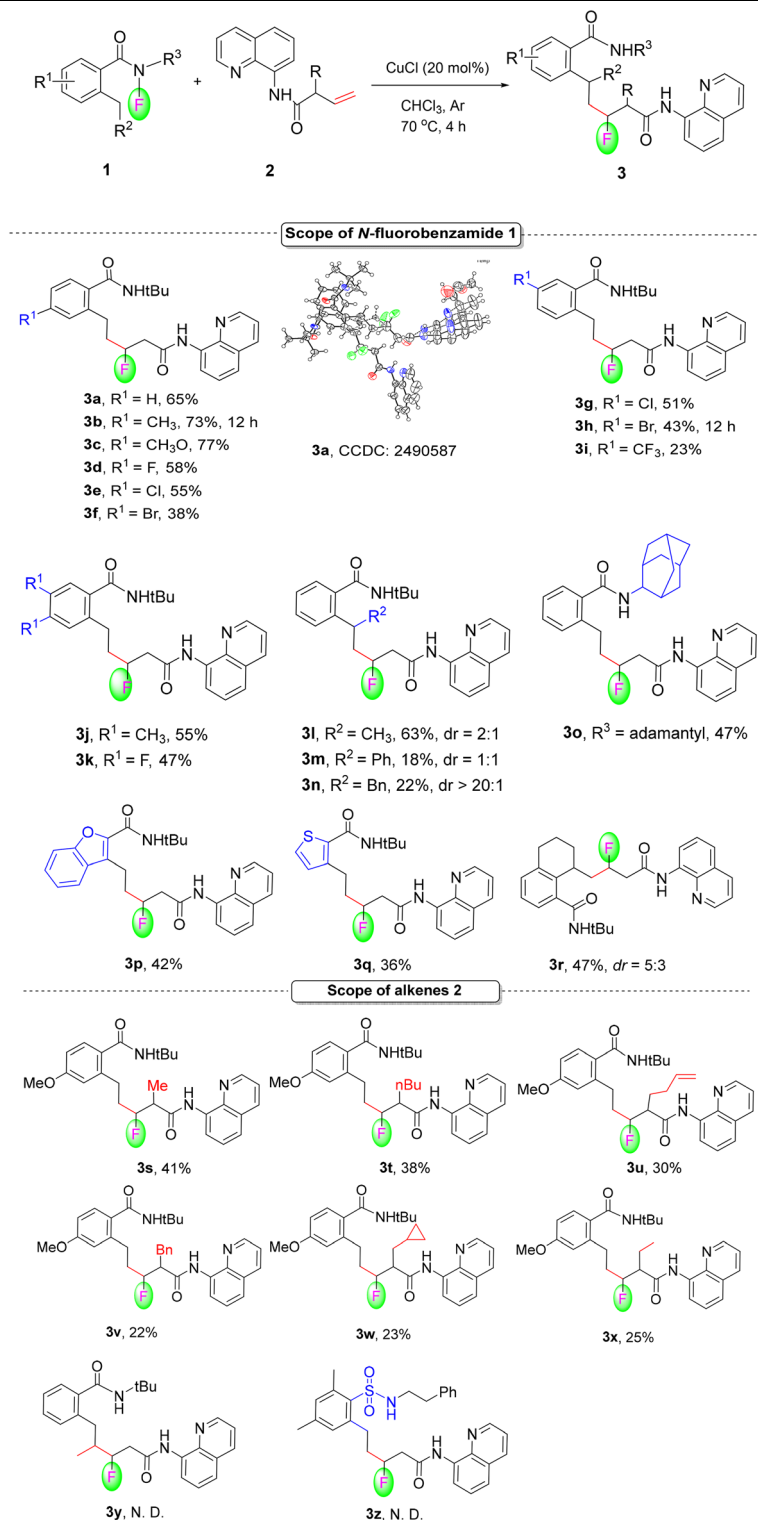
Correction for 'Cu(I)-catalyzed 1,2-carbofluorination of unactivated alkenes enabled by *N*-fluorobenzamides via free radical relay' by Hong Ji *et al.*, *Org. Chem. Front.*, 2026, <https://doi.org/10.1039/d5qo01763g>.

The authors regret that Table 2, Scheme 2d and Scheme 4 were incorrect in the original article. The correct table and schemes are shown here.

<sup>a</sup>State Key Laboratory and Institute of Elemento-Organic Chemistry, National Engineering Research Center of Pesticide, Frontiers Science Center for New Organic Matter, College of Chemistry, Nankai University, Weijin Road 94#, Nankai District, Tianjin, 300071, P. R. China. E-mail: liulingyan@nankai.edu.cn

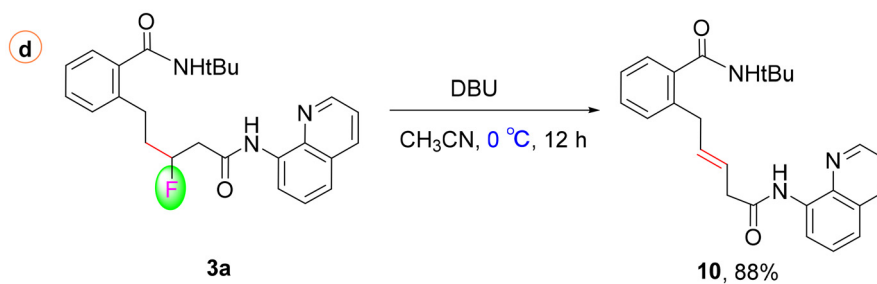
<sup>b</sup>Collaborative Innovation Center of Chemical Science and, Engineering (Tianjin), 300071, China. E-mail: lijing@nankai.edu.cn



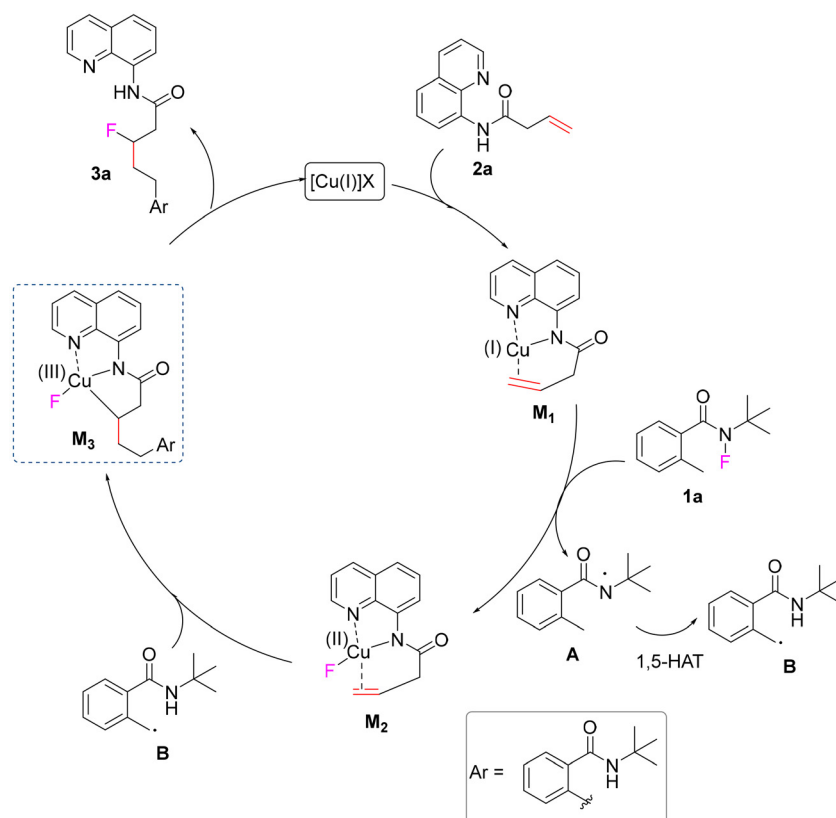
Table 2 The scope for the 1,2-carbofluorination reaction of unactivated alkenes<sup>a</sup>

<sup>a</sup> Under standard conditions, *N*-fluorobenzamide **1** (0.2 mmol, 2.0 equiv.), 8-aminoquinoline-protected vinyl acetamide **2** (0.1 mmol), and cuprous chloride (2.0 mg, 0.02 mmol, 20 mol%) were added to a 10 mL Schlenk tube equipped with a stirring bar. The mixture was protected under argon gas, and finally, 2.0 mL of CHCl<sub>3</sub> was added. Under the condition of 70 °C (oil bath), the mixture was stirred for 4 hours. Isolated yield.





Scheme 2 (d) Late transformation of 3a.

Scheme 4 The plausible mechanism of *N*-fluorobenzamide enabling 1,2-carbofluorination bifunctionalization of unactivated alkenes.

The Royal Society of Chemistry apologises for these errors and any consequent inconvenience to authors and readers.

