

# Nanoscale Advances

Accepted Manuscript

This article can be cited before page numbers have been issued, to do this please use: M. H. El-Dakdouki, C. Yang, A. K. M. A. Ullah, F. Fang, K. Liu, B. U. Hijazi and X. Huang, *Nanoscale Adv.*, 2026, DOI: 10.1039/D5NA00979K.



This is an Accepted Manuscript, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this Accepted Manuscript with the edited and formatted Advance Article as soon as it is available.

You can find more information about Accepted Manuscripts in the [Information for Authors](#).

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard [Terms & Conditions](#) and the [Ethical guidelines](#) still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this Accepted Manuscript or any consequences arising from the use of any information it contains.

# Selective Magnetic Particle Imaging of CD44 Expressing Cells Using Iron Oxide Nanoprobes Functionalized with Chemically Modified Hyaluronan

Mohammad H. El-Dakdouki<sup>a\*</sup>, Chia-wei Yang<sup>b</sup>, A. K. M. Atique Ullah<sup>b</sup>, Fei Fan<sup>b</sup>, Kunli Liu<sup>b</sup>, Baraah U. Hijazi<sup>b</sup>, and Xuefei Huang<sup>b,c,d\*</sup>

<sup>a</sup> Department of Chemistry, Faculty of Science, Beirut Arab University, P.O. Box 11-5020, Riad El Solh 11072809, Beirut, Lebanon

<sup>b</sup> Department of Chemistry, Michigan State University, 578 S. Shaw Lane, East Lansing, MI 48824, USA

<sup>c</sup> Institute for Quantitative Health Science and Engineering, Michigan State University, East Lansing, MI 48824, USA

<sup>d</sup> Department of Biomedical Engineering, Michigan State University, East Lansing, MI 48824, USA

**Corresponding authors:** [m.eldakdouki@bau.edu.lb](mailto:m.eldakdouki@bau.edu.lb); [huangxu2@msu.edu](mailto:huangxu2@msu.edu)

## Abstract

Hyaluronan (HA) is a naturally occurring biocompatible and non-immunogenic linear anionic glycosaminoglycan. In an attempt to enhance the selective binding of HA to the cluster of differentiation 44 (CD44), we recently identified a novel chemically modified HA derivative (referred to as G2HA) with improved affinity to CD44 compared to the native HA polymer. In this study, we aimed at conjugating G2HA to iron oxide nanoparticles (IONPs) and assessing the uptake of the G2HA-IONPs nanocomposite by CD44 expressing cells, namely 4T1 breast cancer cells. In addition, the utility of G2HA-IONPs as efficient contrast agents for Magnetic Particle Imaging (MPI) was studied. The IONPs were synthesized by the thermal decomposition method, and G2HA was conjugated onto the nanoparticles by the ligand exchange method. HA-IONPs were prepared similarly for comparison. The uptake of G2HA-IONPs by 4T1 breast cancer cells was HA-receptor dependent, and higher than HA-IONPs at the same iron concentration as validated in multiple assays. The cell viability assay verified that G2HA-IONPs were biocompatible at the highest tested dose. The suitability of G2HA-IONPs as MPI contrast agents was tested. MPI of cells that received the nanoparticles demonstrated that the improved uptake of G2HA-IONPs translated into more intense signals, with 2.4-fold MPI signal enhancement compared to those from cells that received HA-IONPs. Overall, the findings of this study validated the role of the modified HA derivative (G2HA) in enhancing the uptake of iron oxide



nanoparticles, and asserted the potential applicability of the developed G2HA-IONPs for selective imaging of CD44-expressing tissues and as a potential vehicle for drug delivery.

**Keywords:** Magnetic Particle Imaging; Contrast agents; Hyaluronan-Like Polysaccharides; Ligand Exchange.

## 1 Introduction

Magnetic particle Imaging (MPI) is an emerging non-invasive imaging modality that combines the high spatial and temporal resolutions of ultrasound, MRI, and CT, the high sensitivity of PET and SPECT, and the ability to track and quantify magnetic tracers, all while remaining more cost-effective.<sup>1-3</sup> MPI generates highly sensitive and quantitative three-dimensional (3D) images in real time by directly detecting a magnetic tracer (contrast agent) rather than the surrounding tissues, which appear transparent to MPI because of their diamagnetic properties.<sup>4</sup> Superparamagnetic iron oxide nanoparticles (SPION) are the preferred contrast agents for MPI because of their superparamagnetism, biocompatibility, ease of preparation and surface functionalization, and stability in different media.<sup>5-8</sup> Despite being an emerging technology, MPI holds huge promises in the medical diagnostics sector.<sup>9</sup> However, clinical maturation necessitates the development of improved imager design, enhanced tracer performance, and standardized imaging protocols. In fact, the practical implementation of MPI is strongly influenced by the intrinsic<sup>10</sup> and extrinsic<sup>11-14</sup> properties of the contrast agent. The generation of reproducible high-resolution images by MPI necessitates the fabrication of superparamagnetic magnetic nanoparticles with uniform size and size distribution. In this regard, our recent research has focused on the development of magnetic nanoprobe for enhanced MPI contrast and dual modality imaging in cancer models.<sup>15, 16</sup>

To facilitate enrichment of SPION at the desired sites, cell surface receptors can be selectively targeted by ligand modified-SPION. The cluster of differentiation 44 (CD44) is a receptor that has critical implications in the pathogenesis and progression of diseases such as cancer and atherosclerosis.<sup>17-20</sup> The natural polysaccharide, hyaluronan (HA), is the principal endogenous ligand for CD44.<sup>21</sup> HA is a vital constituent of the extracellular matrix and synovial fluid, and it mediates vital biological processes, including cell differentiation, wound healing, proliferation, and migration.<sup>22, 23</sup> While CD44 is expressed in low levels on some normal cells, it



exists in a quiescent state with low binding affinity to HA, and requires regulated activation before binding to HA. On the other hand, CD44 on cancer cells exists in an activated state with high binding affinity to HA due to alternative splicing and posttranslational modifications.<sup>24, 25</sup> This renders CD44 an interesting and viable target for selective targeted therapy and imaging of tumors while preserving normal tissue. To enhance HA's affinity and selectivity to CD44, we recently reported the preparation of a combinatorial library of HA derivatives using the Ugi reaction and identified a lead analogue (G2HA) with significant enhancement in binding to CD44 using surface plasmon resonance. The enhanced binding translated into improved uptake by cancer cells.<sup>26</sup>

Herein, we report the development of G2HA-coated iron oxide nanoparticles (G2HA-IONPs) as selective MPI contrast agents for targeted imaging and detection of CD44 expressing cancers cells. IONPs were synthesized via the thermal decomposition method, and HA or G2HA were anchored onto the nanoparticles' surface by the ligand exchange method. The targeted nanoformulations were thoroughly characterized by various analytical techniques to assess size, size distribution, surface charge, morphology, and magnetic properties. Various uptake studies demonstrated the role of the modified HA polymer in promoting the accumulation of nanoparticles in 4T1 breast cancer cells, yielding considerably higher contrast in MPI. The findings of this study will have tremendous impacts towards developing new MPI contrast agents for targeted imaging of CD44-mediated diseases such as cancer.

## 2 Materials and methods

### 2.1 Reagents and instrumentation

Reagent grade chemicals were used in this study as supplied by the manufacturers. Sodium hyaluronate (average molecular weight 31 kDa) was obtained from Lifecore Biomedicals. 1,2-hexadecanediol, nuclear fast red solution, benzyl ether, iron (III) acetylacetonate, fetal bovine serum (FBS), and ethyl-(3,3- dimethylaminopropyl) carbodiimide hydrochloride (EDCI) were obtained from Sigma-Aldrich. Mallinckrodt provided potassium ferrocyanide [K<sub>4</sub>Fe(CN)<sub>6</sub>] trihydrate, Acros Organics delivered n-hexanol and adipic dihydrazide, and Fluka supplied oleylamine. CellTiter 96®Aqueous One solution, containing 3-(4,5- dimethylthiazol-2-yl)-5-(3-carboxy-methoxyphenyl)-2-(4-sulfophenyl)-2*H*-tetrazolium, inner salt (MTS) and supplemented with phenazine ethosulfate (PES) as an electron coupling reagent, was acquired from Promega.



Dialysis tubing (MWCO 3500 Da) was procured from BioDesign Inc. Pen/Strep mixture, sodium pyruvate (100 mM), glutamine, phosphate buffered saline (PBS), and fluorescein isothiocyanate (FITC) were sourced from Invitrogen. 4T1-Luc2 mouse epithelial breast cancer cells (ATCC CRL-2539-LUC2) and MCF-7 epithelial human breast cancer cells (ATCC HTB-22) were cultured in RPMI 1640 medium (Gibco™) and DMEM (Gibco™), respectively. Both media were supplemented with 1mM sodium pyruvate, 1% Pen/Strep mixture, and 10% FBS. Cells were incubated under standard culture conditions of 5% CO<sub>2</sub> in a humidified atmosphere, and subcultured at 80% confluency using 0.25% trypsin. Transmission electron microscopy (TEM) micrographs were obtained on a JEOL 1400 Flash microscope equipped with a Lanthanum Hexaboride “Lab Six” emitter and operating at maximum accelerating voltage of 120 kV. A Malvern Zetasizer Nano zs instrument provided the hydrodynamic diameter and zeta potential measurements. Flow cytometry was performed on a Cytex Aurora spectral cytometer. The magnetic properties of the iron oxide nanoparticles were assessed on a superconducting quantum interference device (SQUID) from Quantum Design (Model MPMS3). The sample was placed inside plastic bag (10 mm x 15 mm), which was then inserted into a plastic straw (11 cm long), and measurements were recorded at 100 K. Metal ion concentrations were quantified on an Inductively Coupled Plasma-Optical Emission Spectroscopy (ICP-OES) apparatus from Varian (Model Varian 710). Optical density was measured on a Molecular Device plate reader (Model SpectraMax M3). A Nikon Eclipse TS2R inverted light microscope was used to collect images of Prussian blue stained cells.

## 2.2 Synthesis of the G2HA polymer

The G2HA polymer was synthesized following the Ugi reaction as described by Liu and colleagues.<sup>26</sup> In brief, 3-phenylpropyl amine (35 μL in 200 μL of ethanol) was mixed with an aqueous solution of HA (100 mg), and the pH was adjusted to 4 using hydrochloric acid (1 M). Formaldehyde (3.5% in ethanol, 100 μL) and cyclohexyl isocyanide (28 μL in 172 μL ethanol) were added to the mixture, which was then stirred for 1 hour at room temperature. The pH was subsequently increased to 12 by adding an aqueous solution of sodium carbonate (2 M) to hydrolyze the ester byproduct, and stirring was continued for 2 days. The mixture was dialyzed against water using dialysis tube with MWCO 3,500 Da, and freeze-dried to obtain G2HA as a white solid. The degree of functionalization was 28% per disaccharide as determined by <sup>1</sup>H-NMR.



### 2.3 Synthesis of iron oxide nanoparticles

The thermal decomposition method was employed to synthesize the oleic acid-coated iron oxide nanoparticles (OA-IONPs) as described by Sun and coworkers.<sup>27</sup> In brief, a mixture of Fe(acac)<sub>3</sub>, oleic acid, oleyl amine, 1,2-hexadecanediol (1:3:3:5 molar ratio, respectively) in benzyl ether was flushed with nitrogen gas at room temperature for 1 hour. Nucleation of the nanoparticles was achieved by heating the reaction mixture to 200 °C for 2 hours. Subsequent heating of the mixture to 320 °C for 1 hour ensured the growth of the iron oxide nanoparticles. The synthesized Fe<sub>3</sub>O<sub>4</sub> nanoparticles were first cooled to ambient temperature, followed by the addition of ethanol, and isolated via magnetic separation. The nanoparticles were washed with ethanol (3 times) and then dried in an oven set at 100 °C for 1 hour. The black powder (0.4 g) was collected and stored at room temperature for further analysis and use.

### 2.4 Synthesis of HA-IONPs and G2HA-IONPs

The biphasic ligand exchange method was utilized to coat the iron oxide nanoparticles with HA or G2HA.<sup>28</sup> The hydrophobic OA-IONPs (25 mg) were dissolved in 15 mL toluene, while HA or G2HA (50 mg) was dissolved in deionized water containing 0.5 mL of 0.5 N potassium hydroxide (KOH). The biphasic mixture was heated between 120-130 °C for 24 hours with vigorous stirring, after which it was cooled to room temperature and stirred for additional 6 hours. The aqueous layer containing HA-capped IONPs was collected from a separatory funnel. Excess HA and KOH were removed by Amicon® Ultra-15 Centrifugal Filter Units at a speed of 4,000 rpm. The concentrated nanoparticle solution was diluted to 20 mL and placed on an external magnet to remove large aggregates. The amounts of iron in HA-IONPs and G2HA-IONPs were then standardized using ICP-OES. The concentration of iron in either sample was extrapolated from a standard curve generated from five iron solutions prepared from standard iron solution (1,000 ppm).

### 2.5 Synthesis of FITC-G2HA-IONPs

The FITC-labelled G2HA-IONPs were prepared according to the procedure described in a previous report.<sup>28</sup> In brief, fluorescein isothiocyanate (FITC) (0.1 g, 0.26 mmol) was reacted with excess adipic dihydrazide (ADH) (0.25 g, 1.45 mmol) in 30 mL of bicarbonate/carbonate buffer (pH 9) for 1 hour at room temperature in the dark. Adjusting the pH of the reaction mixture to 3.5



through the dropwise addition of 0.1 N HCl solution caused the precipitation of the desired FITC-ADH conjugate, which was washed with water and dried by lyophilization. FITC-ADH was conjugated onto HA-IONPs or G2HA-IONPs using the EDCI coupling chemistry. In brief, 10 mg of the nanoparticles were dissolved in 20 mL of deionized water, and the pH was adjusted to 4.5 by 0.1 N HCl solution. EDCI (1 mg) dissolved in H<sub>2</sub>O was added, and the carboxyl groups were activated for 1 hour. FITC-ADH (5 mg) was added next, and the reaction mixture was stirred for 6 hours at room temperature, after which it was purified by dialysis (MWCO 3,500) and Amicon® Ultra-15 Centrifugal Filter Units (MWCO 100,000).

## 2.6 Assessing the suitability of G2HA-IONPs for MPI

The suitability of the synthesized nanoparticles as MPI tracer material was evaluated by imaging phantoms composed of serial dilutions of the hyaluronan-coated IONPs (50 µL in 1.5 mL Eppendorf tubes) on a MOMENTUM MPI scanner from Magnetic Insight Inc. Tubes were taped into a sample holder, and the optimal MPI operating parameters for obtaining images with high resolution and sensitivity were identified as follows: Scan type: 2D scan; Scan mode: Standard; Z FOV: 5.0 cm; 5.7 T/m gradient. MPI signals were subsequently extrapolated using VivoQuant's 3D ROI tool.

## 2.7 Cellular uptake of HA-IONPs and G2HA-IONPs by 4T1 breast cancer cells

### 2.7.1 Cellular localization of nanoparticles by TEM.

4T1 cancer cells were cultured in a 10 cm petri dish at  $5 \times 10^6$  cells/plate and incubated overnight at 37 °C and 5% CO<sub>2</sub>. Growth medium was removed, G2HA-IONPs were added at a working iron concentration of 10 µg/mL (in RPMI 1640), and the plate was incubated for 6 hours. The nanoparticle solution was then removed, cells were washed with PBS (three times), and trypsin (1 mL) was added to detach the cells for 5 min. FBS-containing RPMI 1640 was added (10 mL), and cells were transferred to 15 mL centrifuge tubes. Cells were pelleted by centrifugation, and washed successively with PBS. Cells were then transferred to 1.5 mL Eppendorf tube, and the fixing reagent (0.5 mL), comprised of paraformaldehyde/glutaraldehyde 2.5% each in 0.1 sodium cacodylate buffer pH 7.4, was added and incubated with the cells for 12 hours at 4 °C. This was followed by pelleting the cells by centrifugation (2500 rpm), embedding the pellet in 2% agarose, treatment with 1% OsO<sub>4</sub> in 0.1 M sodium cacodylate buffer, and dehydration by serial dilutions of acetone. Finally, Poly/Bed 812



resin embedded the dehydrated cells, and sections were placed on TEM copper grids. Cells were imaged by TEM.

**2.7.2 Quantitative uptake of nanoparticles by Prussian blue staining assay.** 4T1 breast cancer cells were cultured in a 24-well plate ( $3 \times 10^4$  cells/well) in FBS-containing RPMI 1640 for 24 hours at 37 °C and 5% CO<sub>2</sub>, after which the growth medium was removed and cells were washed with PBS three times. HA-IONPs or G2HA-IONPs were added to the cells at an iron concentration of 10 µg/mL (in RPMI 1640), and cells were incubated for 6 hours. The nanoparticles' solution was then removed, and 4T1 cells were washed with PBS, treated with trypsin, collected by centrifugation in 1.5 mL Eppendorf tubes. 6N HCl solution (750 µL) was added to the pelleted cells and the suspension was agitated for 2 hours to induce the lysing of the cells. The mixture was then centrifuged (10,000 rpm) to remove the debris, and the supernatant was transferred to 1.5 mL Eppendorf tubes. 4% K<sub>4</sub>FeCN<sub>6</sub> aqueous solution (750 µL) was added next, and the mixture was incubated in the dark at room temperature for 10 min to allow the development of blue color. Aliquots (100 µL) were transferred to a 96-well plate, and optical density (OD) was measured on an ELISA plate reader at 700 nm.

For competition experiments, 4T1 cells ( $5 \times 10^4$  cells/well) in a 6-well plate were first incubated with free HA (10 mg/mL, 1 mL) for 2 hours, followed by adding G2HA-IONPs in 1 mL of RPMI 1640 (Iron working concentration = 12.5 µg/ mL). Non-competed cells without pre-incubation with free HA received the G2HA-IONPs solution at the same iron working solution. Cells were incubated at 37 °C and 5% CO<sub>2</sub> for 5 hours, after which they were processed for analysis by the quantitative Prussian blue assay as described above.

For assessing the energy dependent uptake of G2HA-IONPs, 4T1 cells ( $5 \times 10^4$  cells/plate) were cultured in 35 mm plates at 37 °C and 5% CO<sub>2</sub>. After removing the culture medium and washing the cells with PBS, the plates received G2HA-IONPs in 1 mL of RPMI 1640 at a working iron concentration of 12.5 µg/ mL. Some plates were incubated at 37 °C while others at 4 °C in a refrigerator for 3 hours. Control plates received FBS-free RPMI 1640 medium and maintained at 37 °C. Following incubation, cells were collected and the amount of iron taken up by cells was determined by the quantitative Prussian blue assay as described above.



**2.7.3 Quantitative uptake of nanoparticles by ICP-OES.** 4T1 cancer cells ( $5 \times 10^4$  cells/well) were adhered to the bottom of a 6-well plate. Following the aspiration of the growth medium, each well received 2 mL of HA-IONPs or G2HA-IONPs at a working concentration of iron equivalent to  $10 \mu\text{g/mL}$ , and the plate was incubated at  $37 \text{ }^\circ\text{C}$  for 6 hours. The nanoparticle solutions were then removed, wells were washed with PBS, and cells were collected by trypsinization. Cells were transferred to 1.5 mL Eppendorf tubes by centrifugation at  $4 \text{ }^\circ\text{C}$  and washed with PBS. This process was repeated 3 times. The collected cells were digested with 70%  $\text{HNO}_3$  (0.5 mL) at  $70 \text{ }^\circ\text{C}$  for 3 hours, after which the cellular debris was removed by centrifugation. The collected supernatant was diluted by water to attain 2%  $\text{HNO}_3$  solution suitable for analysis on ICP-OES. The concentration of iron taken up by cells was extrapolated from a standard curve generated from five known concentrations of iron. A similar approach was used to evaluate the CD44-mediated cellular uptake of G2HA-coated nanoparticles. Briefly, 4T1 (CD44<sup>+</sup>) and MCF-7 cells (CD44<sup>-</sup>) were incubated with the nanoparticles at a working concentration of  $12.5 \mu\text{g-Fe/mL}$ , and the intracellular iron content was subsequently quantified by ICP-OES.

**2.7.4 Uptake of FITC-G2HA-IONPs by laser confocal microscopy.** 4T1 cancer cells ( $1 \times 10^4$  cells/well) were adhered to the bottom of a 4-well chambered plate suitable for confocal microscopy. The culture medium was removed, FITC-G2HA-IONPs in RPMI 1640 were added at an iron working concentration of  $10 \mu\text{g/mL}$  (0.5 mL), and cells were incubated for 5 hours. Negative control cells received RPMI 1640 only (0.5 mL). LysoTracker red DND-99 ( $1 \mu\text{M}$ ;  $50 \mu\text{L/well}$ ) was added one hour before the end of incubation. The supernatant was removed and cells were washed twice with PBS. Cells were fixed using 10% formalin ( $250 \mu\text{L/well}$ ) for 15 min, after which the fixing solution was removed, and cells were washed twice with PBS. DAPI ( $300 \text{ nM}$ ,  $200 \mu\text{L/well}$ ) was added, and cells were incubated for 5 minutes. DAPI was removed, and cells were washed with PBS, followed by water. The plate was stored in the dark at  $4 \text{ }^\circ\text{C}$  until analysis time on a Leica Stellaris 5 White Light Laser Confocal Microscope. A similar experimental protocol was followed to evaluate the uptake of the nanoparticles by MCF-7 cancer cells cultured in DMEM.

**2.7.5 Effect of  $\beta$ -cyclodextrin on the uptake of FITC-G2HA-IONPs by flow cytometry.** 4T1 cancer cells ( $5 \times 10^4$  cells/well) were cultured in a 24-well plate overnight at  $37 \text{ }^\circ\text{C}$  and 5%  $\text{CO}_2$ . The supernatant was aspirated, and cells were washed with PBS. Some of the wells received 0.5



mL of  $\beta$ -cyclodextrin in RPMI 1640 at a working concentration of 5 mM, while the other wells received RPMI 1640 only.<sup>29</sup> The plate was incubated at 37 °C for 30 minutes, after which FITC-G2HA-IONPs or G2HA-IONPs were added to the test wells at an iron working concentration of 5  $\mu$ g/mL, while negative control wells received RPMI 1640. The plate was incubated for additional 2 hours after which the supernatant was removed and cells were washed with PBS. Cells were collected by adding trypsin (0.2 mL), which was neutralized afterwards with FBS-containing RPMI 1640 (2 mL). Cells were pelleted by centrifugation and washed PBS three times, fixed with 10% formalin, and transferred in PBS to FACS tubes. FITC fluorescence was monitored on an Aurora flow cytometer.

## 2.8 MTS cell viability assay

4T1 cancer cells ( $5 \times 10^3$  cells/well) were cultured in a 96-well plate overnight at 37 °C and 5% CO<sub>2</sub>. The culture medium was removed, and G2HA-IONPs in RPMI 1640 were added to the test wells at different iron concentrations (0.00128 – 100  $\mu$ g/mL). Control wells received RPMI 1640 only. The plate was incubated at 37 °C for 6 hours, after which the nanoparticles were removed and cells were washed with PBS. FBS-containing RPMI 1640 was added to all wells (200  $\mu$ L/well) and the plate was incubated for 18 hours at 37 °C. MTS reagent was added (20  $\mu$ L/well), and the plate was incubated in the dark for 3 hours at 37 °C. Optical density at 490 nm was then measured on an ELISA plate reader. The percentage of viable cells was determined relative to control wells that did not receive nanoparticle treatment.

## 2.9 Imaging of 4T1 cells by MPI

4T1 cells ( $2 \times 10^5$  cells/well) immobilized in a 6 well-plate received HA-IONPs or G2HA-IONPs in RPMI 1640 at the same iron concentration (10  $\mu$ g/mL), and incubated at 37 °C and 5% CO<sub>2</sub> for 6 hours. Control cells received RPMI 1640 only. The treatment solution was then removed and cells were washed with PBS. Trypsin was added (300  $\mu$ L/well), and the detached cells were treated with FBS-containing RPMI 1640 (5 mL/well). Cells were collected by centrifugation (2,500 rpm, 4 °C), transferred to 1.5 mL Eppendorf tubes, and pelleted by centrifugation. The supernatant was removed, and cells were fixed in 10% formalin for 5 min, washed with PBS, and pelleted. 1% agarose solution, prepared by microwave heating of agarose in deionized water, was added to the tubes (0.3 mL/tube) and allowed to cool to room temperature. The prepared samples



were stored at 4 °C until imaging on the MPI scanner. MPI images were obtained on a MOMENTUM MPI scanner from Magnetic Insight Inc. Tubes were taped into a sample holder. The parameters applied for imaging were: Scan type: 2D scan; Scan mode: Standard; Z FOV: 5.0 cm; 5.7 T/m gradient. MPI signals were subsequently extrapolated using VivoQuant's 3D ROI tool.

## 2.10 Statistical analysis

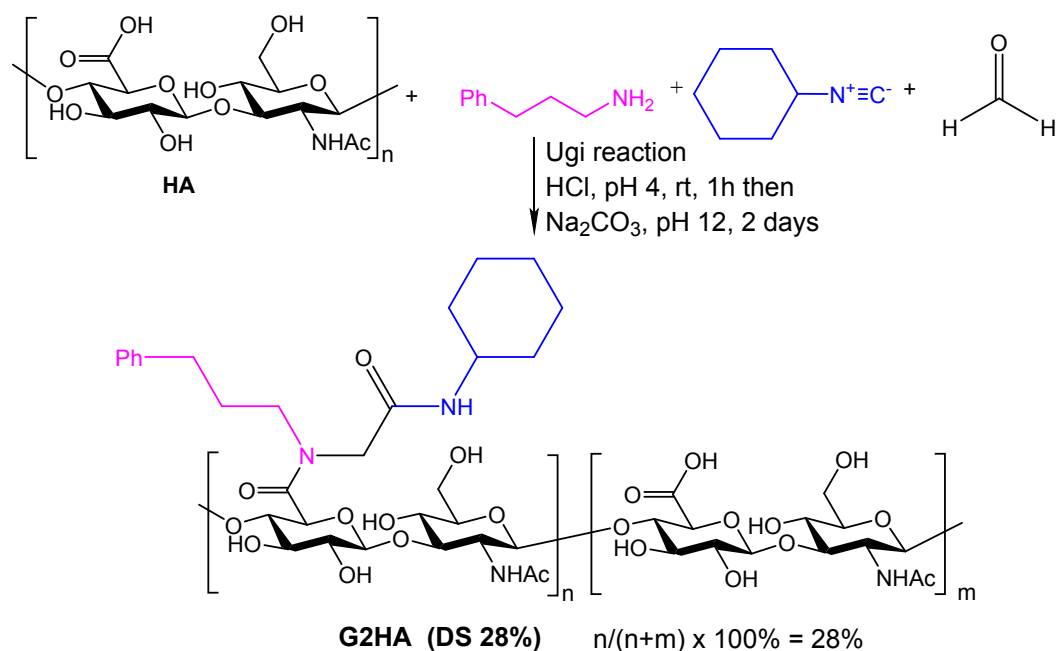
The assumption of normality was met and confirmed with Shapiro-Wilk test ( $p > 0.05$ ), validating the use of Tukey's HSD test for post-hoc analysis. Tukey's HSD was used to analyze pairwise differences between treatment groups following a significant ANOVA result. Most comparisons showed statistically significant differences, confirming that treatment effects differ across groups. Differences in the mean values among the treatment groups are considered statistically significant at  $p \leq 0.001$ .

## Results and Discussion

### 3.1 Synthesis of G2HA derivative

The G2HA analogue was synthesized via the Ugi reaction by reacting hyaluronan (31 kDa), 3-phenylpropyl amine, cyclohexyl isocyanate, and formaldehyde at pH 4.0 as represented in Scheme 1. The success of the reaction and the degree of the substitution was validated by NMR analysis (Figure S1). The appearance of aromatic protons in the  $^1\text{H}$ -NMR spectrum and three different carbonyl groups in the  $^{13}\text{C}$ -NMR spectrum of G2HA supported the successful functionalization. These peaks were absent in the spectra of unmodified hyaluronan. Integrating the aromatic protons in the  $^1\text{H}$ -NMR spectrum of G2HA relative to the anomeric protons of the disaccharide unit suggested that the degree of substitution (DS) is 28%.



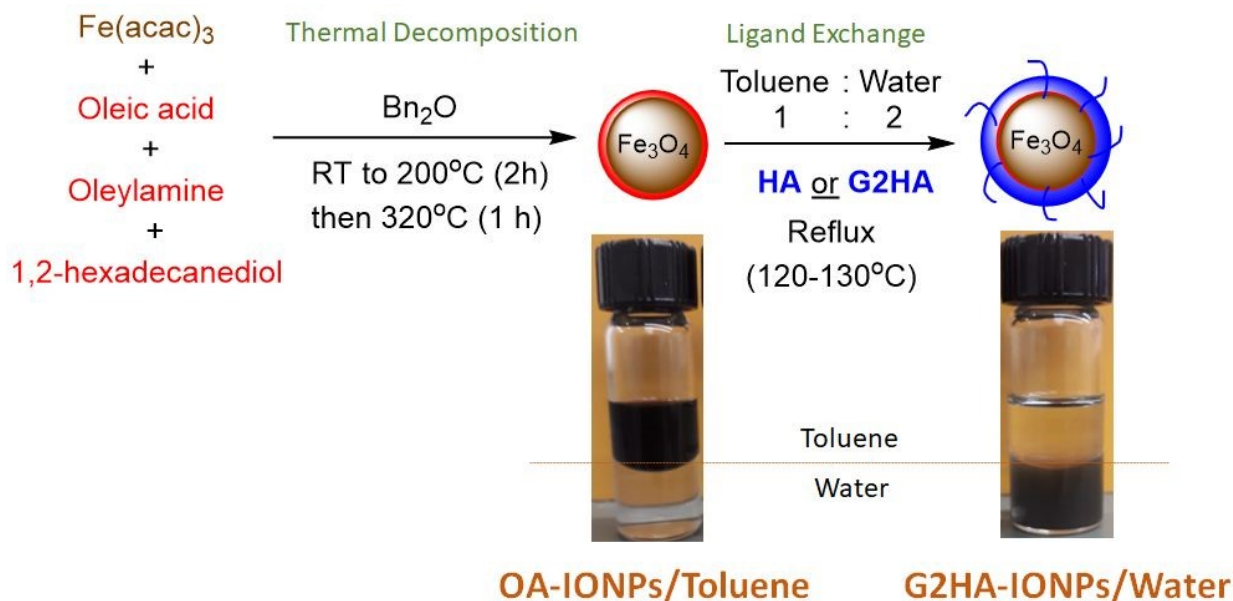


**Scheme 1.** Synthesis of the modified hyaluronan analogue G2HA under the Ugi reaction conditions. The reactants were reacted at pH 4 for 1 hour, after which the pH was increased to 12 to hydrolyze the ester byproduct resulting from the Passerini reaction.

### 3.2 Preparation of IONPs

The IONP core was prepared by the thermal decomposition method, which is well known for the assembly of monodisperse, uniformly sized (narrow size distribution), crystalline, spherical nanoparticles.<sup>27</sup> The experimental conditions summarizing the applied protocol are depicted in Figure 1. The resulting oleic acid (OA)-coated IONPs (OA-IONPs) are hydrophobic in nature and therefore are well dispersed in non-polar solvents such as toluene. The unmodified HA polymer or G2HA were anchored onto the iron oxide nanoparticles surface by the ligand exchange method where a biphasic system comprised of an aqueous solution of the hyaluronan adduct and a toluene solution containing the OA-IONPs was refluxed between 120-130 °C.<sup>28</sup> Functional groups like carboxyl and hydroxyl groups of HA can efficiently displace OA on IONPs, thus shifting their solubility from toluene to water (Figure 1). Unlike other methods, utilizing the ligand exchange method to attach HA analogues on the surface of the nanoparticles requires no prior functionalization of the biopolymer, thus helping maintaining optimal affinity of HA to its receptors such as CD44.





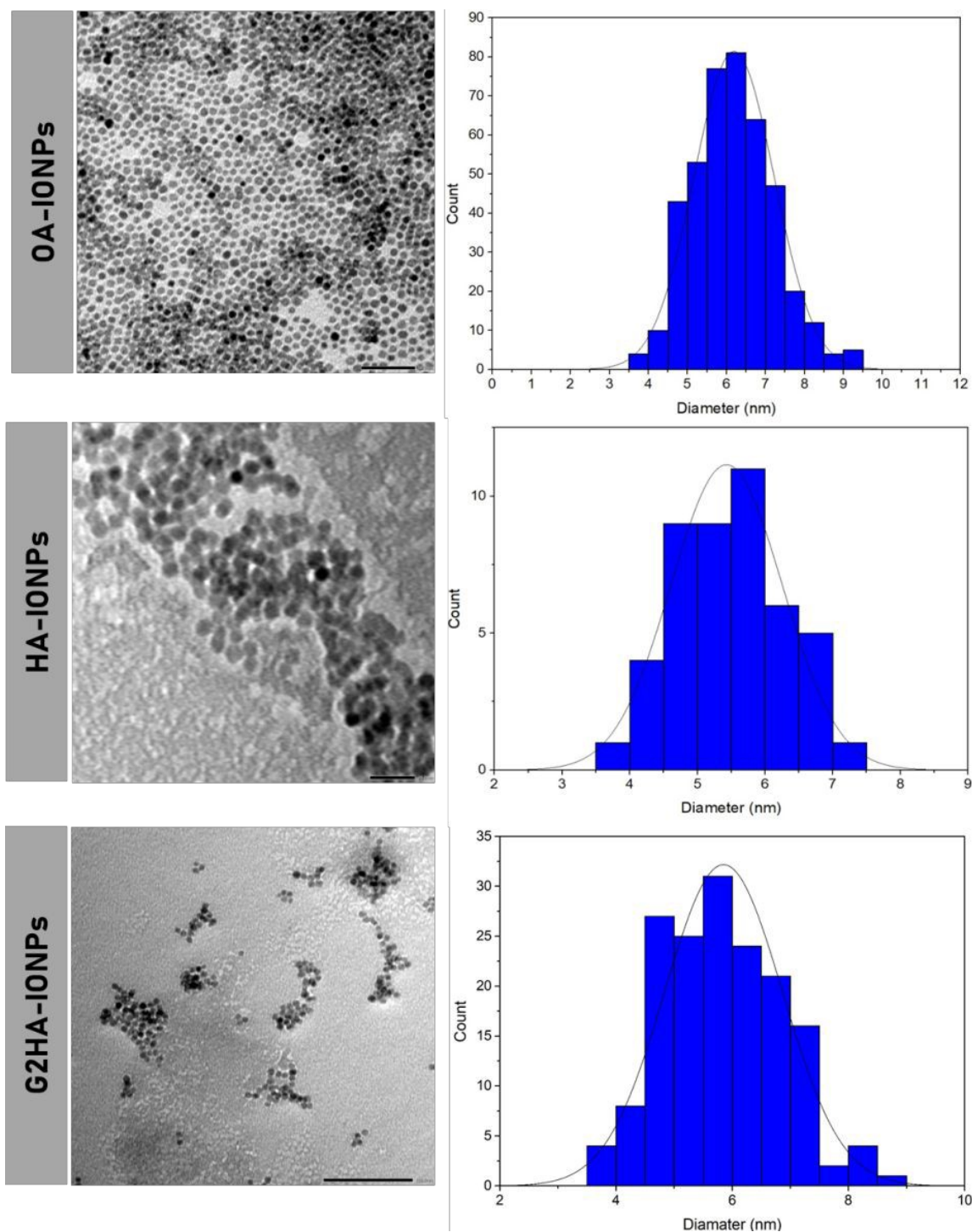
**Fig. 1** Preparation of the OA-IONPs by the thermal decomposition method and G2HA-IONPs by ligand exchange method.

TEM micrographs revealed the high monodispersity and size uniformity of the nanoparticles with an inorganic metal core  $5.7 \pm 2.0$  nm in size (Figure 2). The morphology and size of the inorganic core did not change following the incorporation of the HA analogues on IONPs, highlighting the suitability of the exchange method for preparing surface-modified nanocomposites.

The hydrodynamic diameter of the various nanoformulations was determined by dynamic light scattering (DLS), revealing sizes of  $113.53 \pm 0.49$  nm and  $135.03 \pm 1.53$  nm for HA-IONPs and G2HA-IONPs, respectively (Figure S2). Importantly, the polydispersity index of all nano-preparations ranged between 0.164 and 0.175 (below 0.4) highlighting the applicability of the utilized synthetic method for the preparation of close to monodispersed nanoparticles with minimal aggregation, thus corroborating the findings of the TEM images. The disparity in the nanoparticle size determined by TEM and DLS is due to the fact that TEM measures the size of the dried iron oxide nanoparticle core. On the other hand, DLS measures the larger effective size of the



nanoparticle in solution, which includes the particle core, its surface coating, and the surrounding solvent shell.



**Fig. 2** TEM images of OA-IONPs, HA-IONPs, and G2HA-IONPs and the corresponding size distribution histograms (Scales: 50 nm, 20 nm, and 100 nm, respectively).



All nanoformulations exhibited a highly negative zeta potential with values of  $-27.7 \pm 5.3$  mV and  $-44.2 \pm 7.5$  mV for HA-IONPs and G2HA-IONPs, respectively. Such negative values minimized aggregation of the nanoparticles and enhanced their dispersity in solution (Table 1; Figure S2). Monodispersity of nanoparticles is highly desired in biomedical applications such as MPI, ensuring reproducible and predictable biological profile, optimized cellular uptake, and consistent high-resolution imaging.

**Table 1** Summary of the iron core size, hydrodynamic diameter, size distribution (polydispersity index, PDI), and surface charge (zeta potential) of the various IONPs.

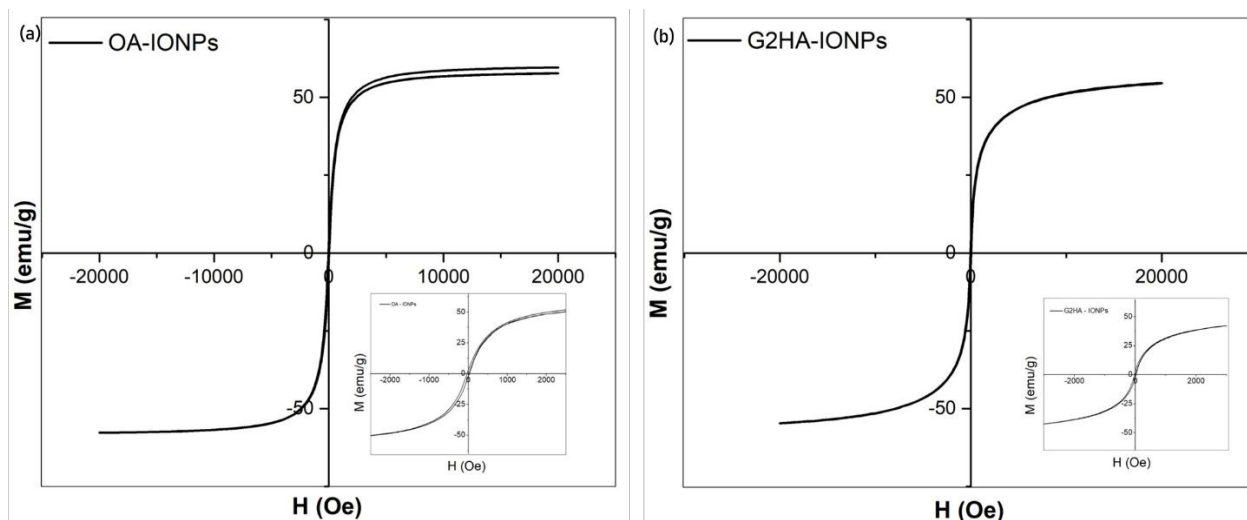
Nanoparticle	IONP core diameters (nm) (From TEM)	Hydrodynamic diameter (nm)	Polydispersity index	Zeta potential (mV)
OA-IONPs	$5.7 \pm 2.0$	-	-	-
HA-IONPs	$5.5 \pm 1.8$	$113.53 \pm 0.49$	$0.170 \pm 0.008$	$-27.7 \pm 5.3$
G2HA-IONPs	$5.5 \pm 2.3$	$135.03 \pm 1.53$	$0.185 \pm 0.012$	$-44.2 \pm 7.5$

### 3.3 Assessing the magnetic properties of the nanoparticles

A major goal of the current study is to prepare targeted IONPs that are efficient contrast agents for MPI. As such, it is essential to assess the magnetic properties of synthesized polysaccharide coated nanoparticles to justify its utility as tracers in MPI. A SQUID magnetometer was used to extrapolate the necessary magnetic parameters including magnetic saturation ( $M_s$ ), magnetic remanence ( $M_r$ ), and coercivity ( $H_c$ ), and their values are summarized in Table 2. OA-IONPs exhibited  $M_s$  value of 56.2 emu/g (Figure 3). The ligand exchange method utilized to anchor G2HA polymer onto the surface of IONPs marginally affected the magnetization saturation, and G2HA-IONPs recorded  $M_s$  value of 54 emu/g. Such high  $M_s$  values reflected the potential of the nanoparticles in various applications such as magnetic separation, magnetically targeted delivery, improved data storage density, and efficient hyperthermia therapy.<sup>30-32</sup> Specifically, high magnetization saturation  $M_s$  is an essential attribute of MPI contrast agents, providing optimal contrast, sharper spatial resolution, and higher sensitivity, and enabling linear quantitative imaging.<sup>33</sup> In addition, MPI contrast agents with high  $M_s$  values are characterized by minimal relaxation losses, thus maintaining the fidelity of the harmonic signal that is crucial for imaging by MPI.<sup>34</sup> Contrast agents with high  $M_s$  values generate considerable MPI signals, enabling the



usage of low doses for imaging, and reducing operational cost and the concerns of potential toxicities *in vivo*. Another important parameter for effective MPI contrast agents is magnetic remanence  $M_r$ , which reflects the ability of a material to retain magnetization after the removal of an applied external magnetic field. The  $M_r$  values of IONPs prepared in the current study are close to zero (2.65 emu/g for OA-IONPs and 2.92 emu/g for G2HA-IONPs), indicating that the magnetic characteristics of the nanoparticles are readily controllable (easily magnetized and demagnetized).<sup>35</sup> Magnetic nanoparticles with low  $M_r$  are highly desirable in biomedical applications. The fact that the prepared nanoparticles do not retain much residual magnetization (low  $M_r$ ) prevents aggregation of the nanoparticles, enhances the colloidal stability in solutions, and reduces potential toxicity.<sup>36</sup> Such attribute is essential for improved sensitivity and the generation of clean MPI signals when smaller amounts of the contrast agent are used. In addition to  $M_r$ , coercivity ( $H_c$ ) was also determined, which is defined as the external magnetic field required to bring  $M$  to zero following magnetization of the material to saturation.  $M_r$  and  $H_c$  values close to zero are characteristic of superparamagnetic materials that can act as ideal MPI tracers, which fits the magnetic profile of the IONPs prepared in this study (Figure 3). This is further supported by the low squareness ratio ( $M_r/M_s$ ) for OA-IONPs (0.044) and G2HA-IONPs (0.053) consistent with the narrow hysteresis loops and reflecting smaller hysteresis losses. These characteristics are typically important for the application of magnetic nanoparticles in magnetic imaging, which requires feasible and spontaneous relaxation of magnetization without the retention of magnetic memory.



**Fig. 3** Magnetic hysteresis curves of (a) OA-IONPs and (b) G2HA-IONPs showing the superparamagnetic properties of the synthesized magnetic nanoparticles. Magnetization measurements were recorded at 100 K and are expressed as electromagnetic unit per gram of IONPs (emu/g).

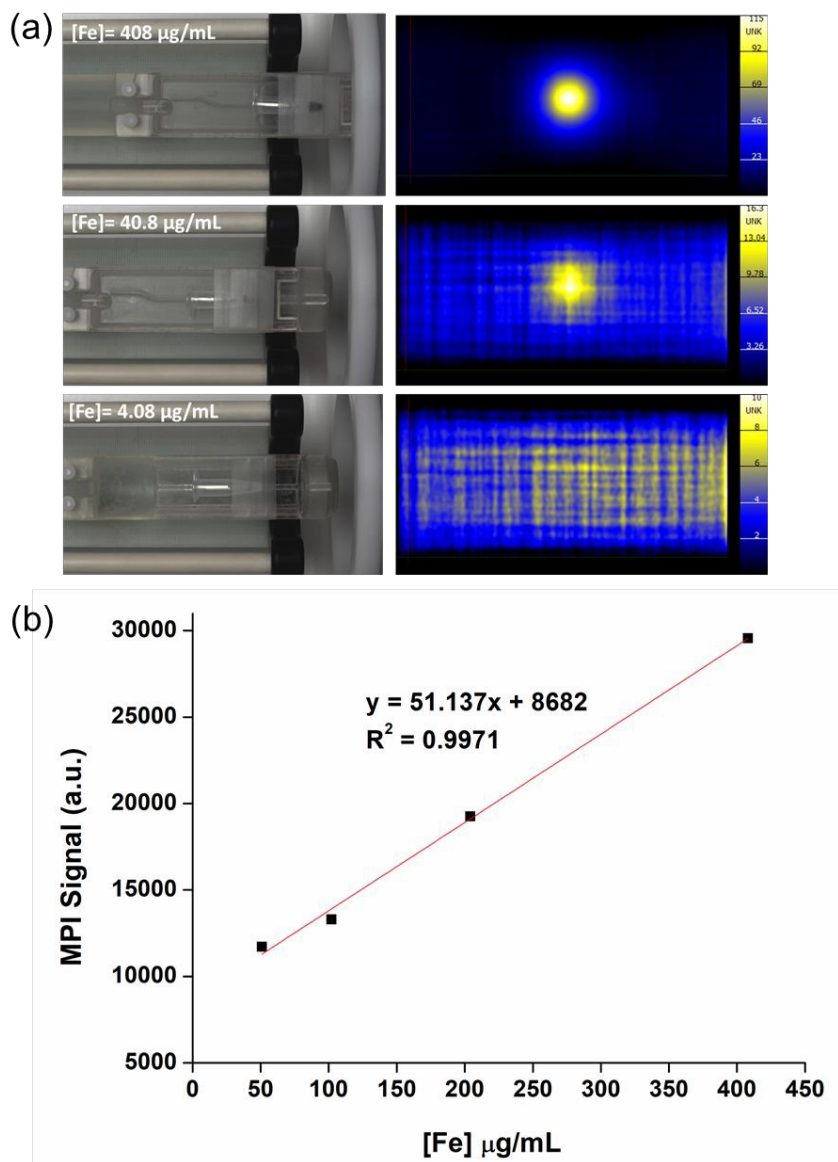
**Table 2** Magnetic parameters of the prepared nanoparticles. Magnetization measurements were recorded at 100 K.

Nanoparticle	Saturation Magnetization ( $M_s$ )	Magnetic remanence ( $M_r$ )	Coercivity ( $H_c$ )	Squareness ratio ( $M_r/M_s$ )
OA-IONPs	59.6 emu/g	2.65 emu/g	15.5 Oe	0.044
G2HA-IONPs	54.6 emu/g	2.92 emu/g	15.1 Oe	0.053

### 3.4 Hyaluronan-coated IONPs as MPI tracers

The desirable magnetic properties of the G2HA-coated IONPs enabled them to act as efficient MPI tracers. The suitability of the nanoparticles as MPI contrast agents and the linearity of the MPI signal at different concentrations of the nanoparticles solution were assessed, and results are depicted in Figure 4 for G2HA-IONPs and Figure S3 for HA-IONPs. Both nanoparticle formulations generated a linear dose-dependent signal on the MPI scanner, with coefficients of regression ( $R^2$ ) close to unity (0.9971 for G2HA-IONPs, and 0.9999 for HA-IONPs). Such observation highlighted the suitability of the method to determine the concentration of iron from the generated MPI calibration curve when desired. The observed signal linearity is consistent with the high magnetization saturation  $M_s$  of the nanoparticles enabling quantitative imaging.<sup>37</sup>





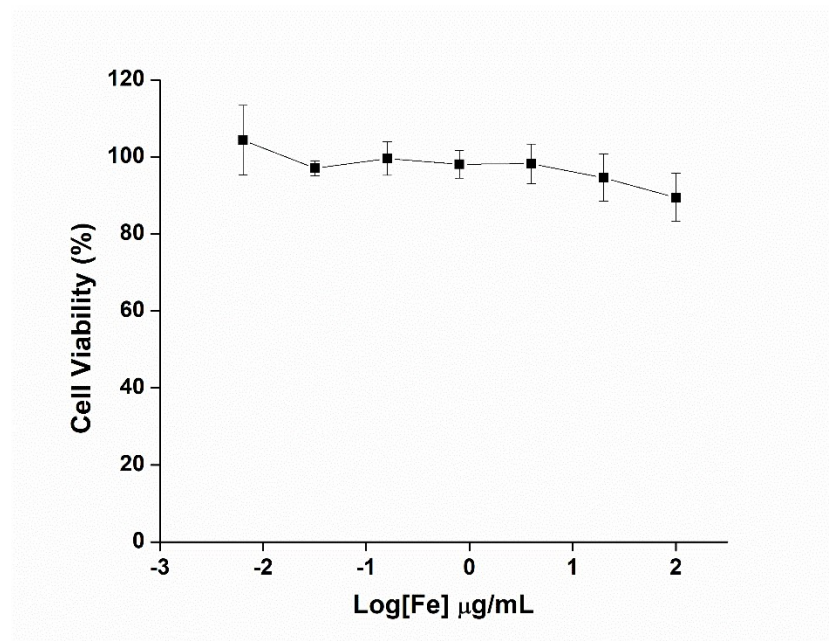
**Fig. 4** (a) Images showing sample holder of the MPI scanner with attached Eppendorf tubes containing G2HA-IONPs solutions at different concentrations, and the corresponding MPI signal; (b) Plot depicting the linear, dose-dependent MPI signal as a function of nanoparticle concentrations.

### 3.5 Biocompatibility of G2HA-IONPs

It is foundational to assess the biocompatibility of the iron oxide nanoparticles before using it in biomedical contexts to ensure safety of the nanoformulations, define the safe dose range, and avoid misinterpretation of experimental results.<sup>38</sup> In the current study, the biocompatibility of



G2HA-IONPs were evaluated using the MTS cell viability assay, and results are presented in Figure 5. The tested nanoparticles demonstrated high biocompatibility, with over 90% viability of treated cells even at the highest concentration tested (100  $\mu\text{g Fe/mL}$ ).



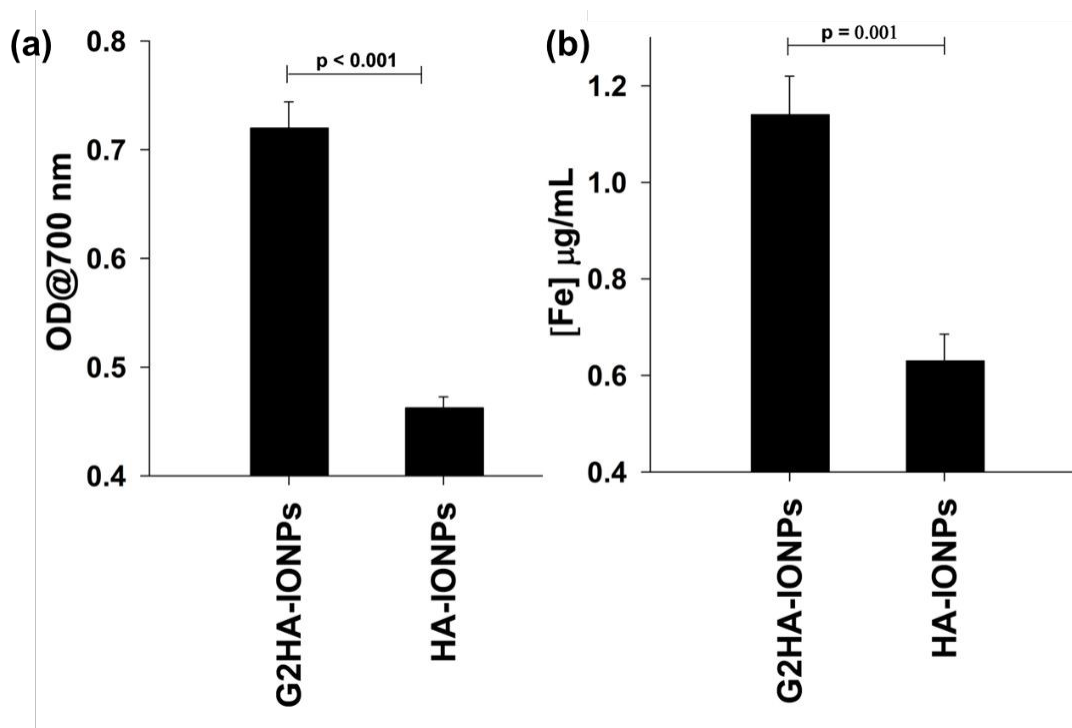
**Fig. 5** MTS cell viability assay of 4T1 breast cancer cells treated with different concentrations of G2HA-IONPs.

### 3.6 Uptake of HA-IONPs and G2HA-IONPs by 4T1 cancer cells

It has been shown that G2HA demonstrated significant enhancements in binding to CD44 compared to the native HA.<sup>26</sup> The expression of CD44 by the 4T1 cell line used in this study has been validated in our previous report.<sup>15</sup> The enhanced binding affinity translated into increased uptake of G2HA by CD44-overexpressing cancer cells. In this report, the G2HA was anchored onto IONPs, and its ability to increase the uptake of the nanoparticles compared to HA was assessed on 4T1 breast cancer cells, which are known to overexpress the CD44 receptor.<sup>15, 39</sup> The cells were incubated with G2HA-IONPs or HA-IONPs at 10  $\mu\text{g Fe/mL}$  for 6 hours. The internalization of the nanoparticles was validated visually by staining the iron oxide core with potassium ferrocyanide to produce the characteristic Prussian blue stain. Quantitation of the internalized nanoparticles was achieved by monitoring the absorbance of the Prussian blue stain at its characteristic wavelength (700 nm) on a UV-vis spectrophotometer, and the obtained results are presented in Figure 6a. Furthermore, the intracellular iron content was quantified by ICP-OES



as revealed in Figure 6b. The obtained results are presented in Figure 6a and Figure 6b, respectively. Interestingly, G2HA-IONPs showed remarkable uptake by the 4T1 cancer cells, demonstrating statistically significant enhancement over HA-IONPs as revealed by UV-vis spectrometry and ICP-OES, respectively. This implied that anchoring G2HA onto IONPs did not alter its superior binding profile to CD44 compared to HA. The enhancement in the relative uptake of nanoparticle formulations can further be intensified by avidity, defined as the simultaneous interaction between multiple ligands on the nanoparticle with cell surface receptors resulting in very strong overall binding even if affinity is moderate.<sup>40</sup> It has been validated that the stronger binding affinities can contribute to higher avidity.<sup>41</sup> Multivalent binding interactions trigger receptors clustering and effectively activate receptor-mediated endocytosis particularly in tissues that express high densities of the target receptor such as the cancerous tissues.<sup>42</sup> Therefore, the stronger binding of G2HA with CD44 compared to that by HA particularly in a nanoparticle formulation can induce higher avidity and improved uptake by CD44-overexpressing cells.

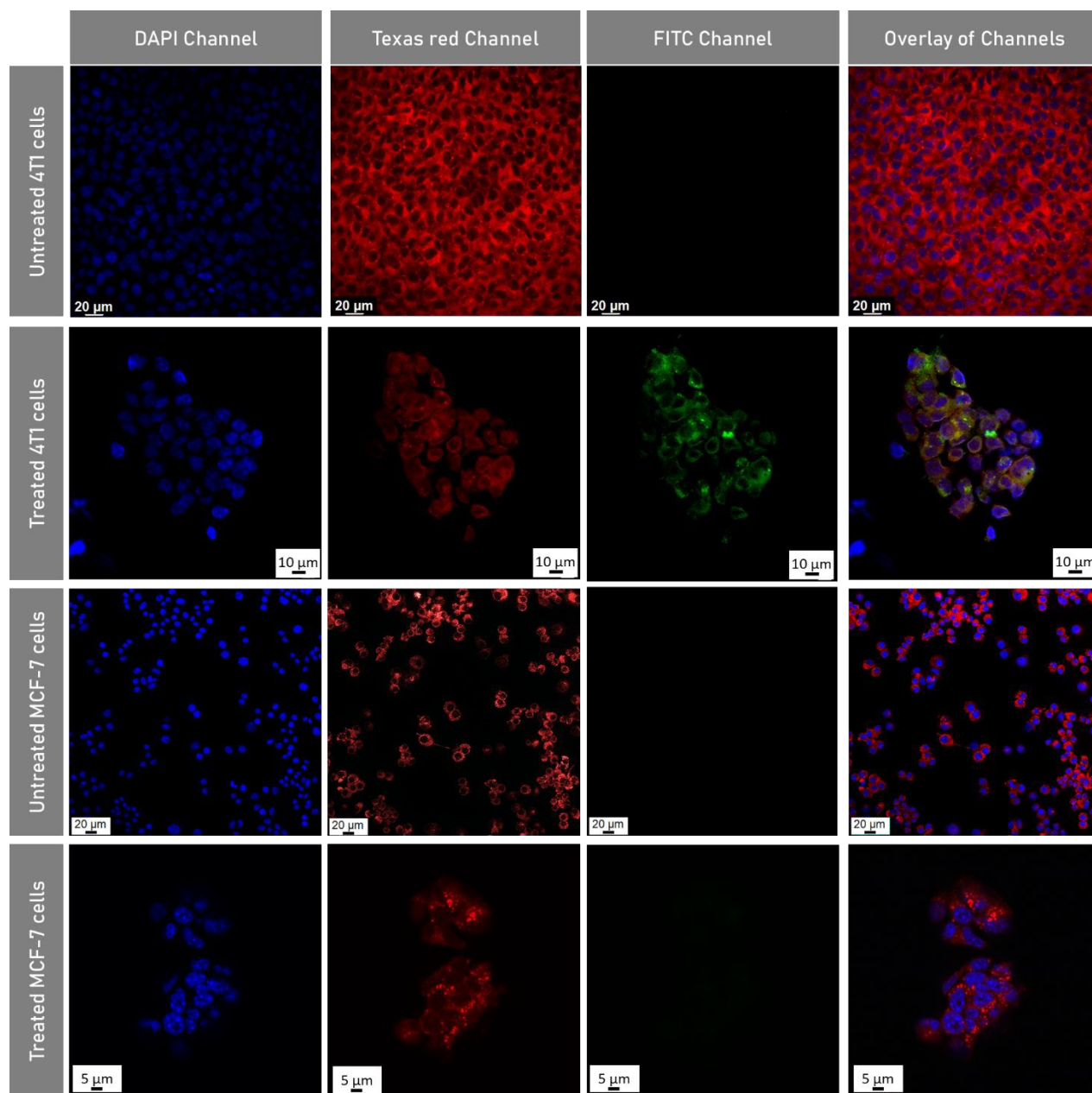


**Fig. 6** Quantification of the uptake of G2HA-IONPs and HA-IONPs by 4T1 cancer cells using (a) the Prussian blue staining assay and (b) ICP-OES.

### 3.7 Internalization of the G2HA-IONPs: Mechanistic insights into receptor-mediated endocytosis

The internalization of G2HA-IONPs by the 4T1 cancer cells and their intracellular localization were assessed by confocal microscopy, where laser confocal images were collected for 4T1 cancer cells treated with FITC-labelled G2HA-IONPs (Figure 7). The FITC dye was covalently attached to G2HA via EDCI chemistry as shown in Scheme S1. LysoTracker red dye was used to selectively stain lysosomes inside the cells and its fluorescence was registered on the Texas red channel. Simultaneously, the FITC dye on the nanoparticle was visualized on the FITC channel. No signal was collected on the FITC channel for the cells that received the growth media only without FITC-labeled nanoparticles. On the other hand, green fluorescence was registered in the FITC channel of the cells that received FITC-G2HA-IONPs, confirming that the nanoparticles were internalized by cells. Interestingly, remarkable co-localization of the LysoTracker red and FITC signals was recorded, producing purple color in the overlaid images. Such observation validated that G2HA-IONPs were internalized into lysosomes inside the cells, and supported a receptor mediated endocytosis mechanism for the intracellular trafficking the nanoprobe. Interestingly, the CD44-deficient MCF-7 breast cancer cells exhibited negligible fluorescence on the FITC channel under identical experimental conditions. The stark contrast in the uptake of the G2HA-IONPs by 4T1 and MCF-7 cells validated visually the targeting efficiency of the G2HA coating, and ascertained that the cellular internalization of the targeted nanoparticles is highly dependent on the presence of the CD44 receptors that mediated the selective active targeting of cancer cells through receptor mediated endocytosis.



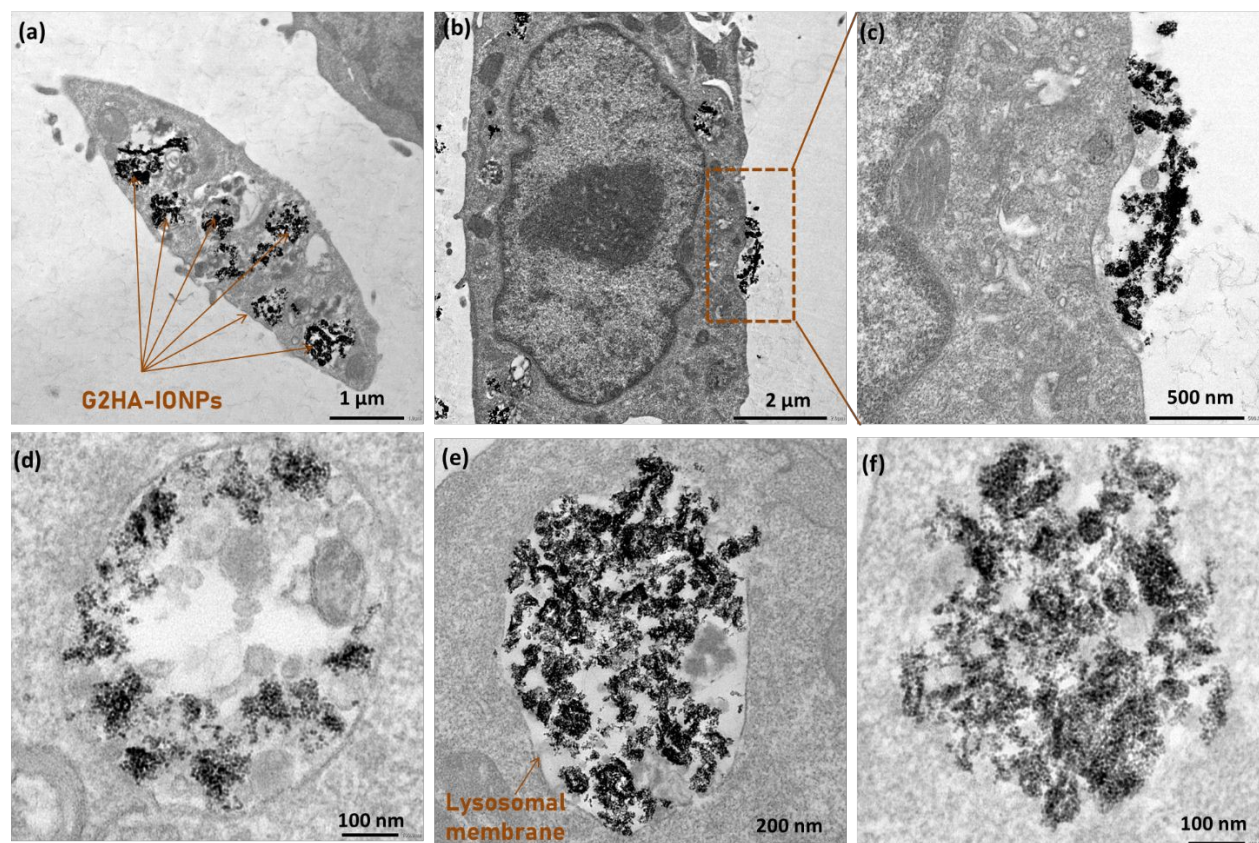


**Fig. 7** Laser confocal microscopy images for breast cancer cells that received FITC-labeled G2HA-IONPs (Treated 4T1 cells and treated MCF-7 cells) or culture medium (RPMI 1640 for untreated 4T1 cells or DMEM for untreated MCF-7 cells). DAPI, lysotracker red, and FITC dyes were used to stain nuclei, lysosomes, and the G2HA-IONPs, respectively. The overlay channels show the merged images captured from the DAPI, Texas red, and FITC channels.



TEM micrographs for cells treated with G2HA-IONPs (Figure 8) provided deeper mechanistic insights into the internalization of the nanoparticles by the cells. Figure 8a revealed the extensive uptake of the G2HA-coated nanoparticles by cells, highlighting the efficiency of the synthesized nanoformulation to act as an effective MPI tracer capable of efficiently labelling, tracking, and imaging cells. The hyaluronan-coated nanoparticles are presumably internalized through CD44-mediated endocytosis. Following the interaction of the G2HA coating on the nanoparticles with the CD44 receptors on cells, the endocytosis process is initiated by forming a groove on the cell membrane (Figures 8b and 8c). This groove could grow to encapsulate the HA-CD44 complex, and eventually pinch off into the cytoplasm forming an early endosome where nanoparticle clusters can be seen attached to the periphery of the formed vesicle (Figure 8d). As the pH decreases in the endosome to become a lysosome, the carboxyl groups of the hyaluronan derivative may become protonated which increases its hydrophobicity. As a result, the magnetic nanoparticles aggregate leading to the formation of clusters that fill the lysosome (Figure 8e). The nanoparticles appear to escape from the lysosome as evident from the disappearance of the lysosomal membrane that was encapsulating the nanoparticles cluster observed in some images (Figure 8f). It is possible that G2HA polymer acted as a proton sponge, soaking up the protons of the acidic lysosomal compartment through its carboxyl groups and disrupting its acidity balance, leading to its collapse and the release of IONPs into the cytoplasm.<sup>43, 44</sup>



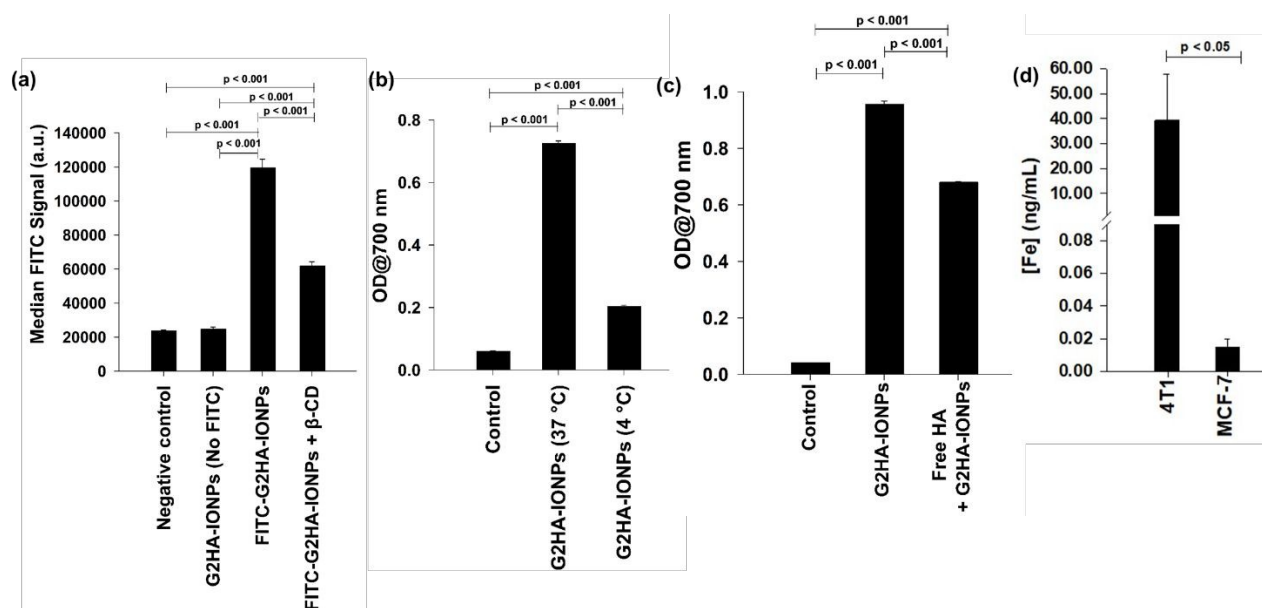


**Fig. 8** Visualization of the endocytosis of G2HA-IONPs. (a) Representative TEM image of a 4T1 cancer cell treated with G2HA-IONPs for 6 hours showing the extensive uptake of the nanoparticles by the cell; (b,c) Attachment of the nanoparticles to the surface of the cell by binding to the CD44 receptors, forming a groove that engulfs the nanoparticles, and pinches off to form an early endosome in the cytoplasm (d); (e) TEM micrograph showing the localization of the nanoparticles inside a lysosome; (f) Disappearance of the lysosomal membrane and release of the nanoparticles into the cytoplasm.

Several experiments were conducted to validate the proposed HA-CD44 mediated endocytosis mechanism. It has been reported that the endocytic CD44-mediated uptake of hyaluronan occurs through the lipid rich rafts, and not the clathrin-coated pits or caveolae.<sup>45, 46</sup> Therefore, adding a cholesterol solubilizer such as  $\beta$ -cyclodextrin would disrupt the cholesterol rich lipid rafts and adversely affect the CD44-mediated endocytosis of the nanoparticles. In fact, pre-incubation of the 4T1 cancer cells with  $\beta$ -cyclodextrin followed by the addition of the FITC-labeled G2HA-IONPs decreased the uptake of the nanoparticles by 48% as revealed by flow



cytometry (Figure 9a). Such observation supported the endocytosis of G2HA-IONPs and asserted that the lipid rich rafts played a pivotal role in the uptake mechanism.



**Fig. 9** (a) Flow cytometry analysis of the uptake of FITC-G2HA-IONPs in the absence and presence of  $\beta$ -cyclodextrin ( $\beta$ -CD). The uptake of the nanoparticles dropped by 48% when  $\beta$ -cyclodextrin was added; (b) Comparative study for the uptake of G2HA-IONPs by 4T1 breast cancer cells at 37 °C and 4 °C using the Prussian blue assay; (c) Uptake of the G2HA-IONPs in the absence and presence of free hyaluronan; (d) Cellular uptake of G2HA-coated nanoparticles by 4T1 (CD44-expressing) and MCF-7 (CD44-low expressing) breast cancer cell lines following incubation for 6 hours and quantification of iron content by ICP-OES. Significantly higher iron content was observed in 4T1 cells compared to MCF-7 cells.

A major characteristic of receptor-mediated endocytosis is that it is an energy dependent process.<sup>47</sup> In other words, endocytosis will dramatically slow down under low temperatures (e.g. 4 °C). To further validate that endocytosis is the mechanism of internalization of G2HA-IONPs, the uptake of the nanoparticles was studied at 37 °C and 4 °C, and cellular uptake was analyzed by the Prussian blue assay. Figure 9b summarizes the obtained results. Interestingly, 72% reduction in the uptake of G2HA-IONPs was observed when the experiments were performed at 4 °C relative to those conducted at 37 °C, suggesting that the IONPs uptake process was energy dependent and supporting the endocytosis mechanism.



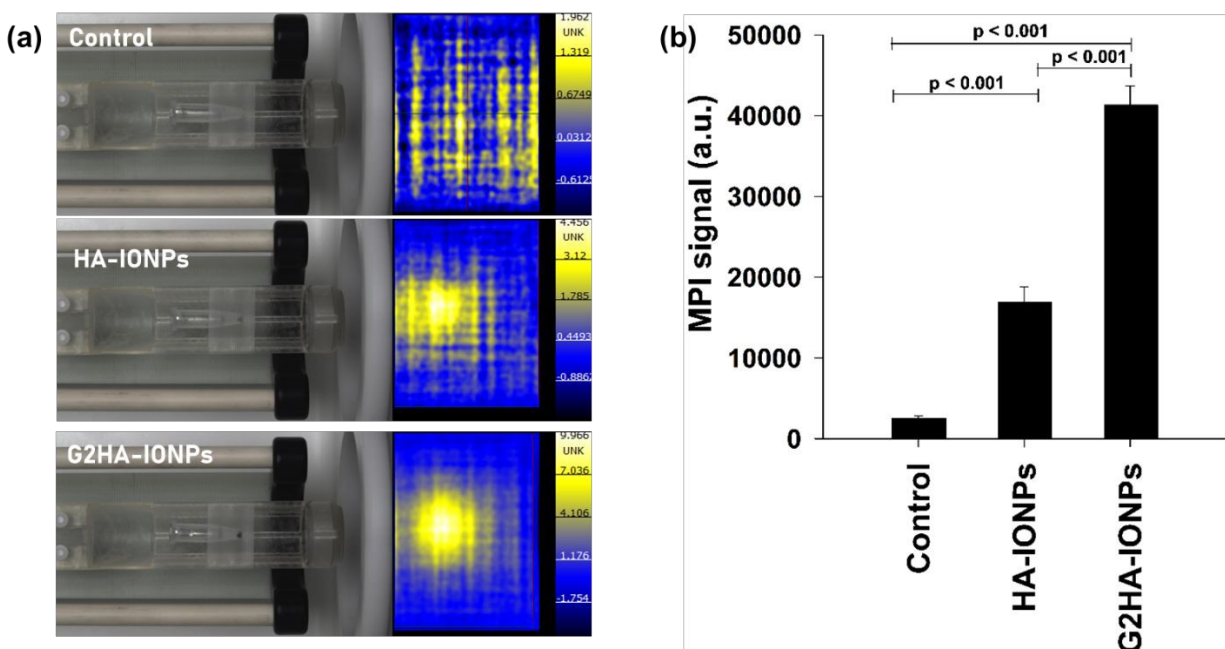
To confirm the role of HA receptor in mediating the uptake of the G2HA-IONPs, the uptake of the nanoparticles was conducted in the presence of excess free hyaluronan, and the amount of nanoparticle internalized by the 4T1 cancer cells was assessed by the Prussian blue assay. The uptake of the G2HA-IONPs was significantly reduced by 28% in the presence of free HA (Figure 9c), suggesting that HA and G2HA-IONPs competed for the same binding sites on the cells, and confirmed the vital role played by G2HA in mediating the endocytosis of the nanoparticles. The HA competition approach employed in the current study provided supportive evidence for the CD44-mediated endocytosis mechanism by blocking receptors on the same cell type, thus eliminating the influence of cell line-to-cell line variabilities in terms of metabolic and growth rates, membrane structures, and non-specific endocytosis.<sup>48-50</sup> The significant reduction obtained in the HA-competition experiments validated the selectivity of G2HA-IONPs for CD44.

To further validate the CD44-mediated cellular internalization of the G2HA-coated nanoparticles, the uptake was comparatively evaluated by 4T1 (CD44<sup>+</sup>) and MCF-7 (CD44<sup>-</sup>) breast cancer cell lines. The intracellular iron content in both cell lines is presented in Figure x. A striking difference in iron concentration was observed between the two cell lines, where the CD44-overexpressing 4T1 cells demonstrated substantial enhancement in intracellular iron (~39 ng/mL). On the other hand, the low CD44-expressing MCF-7 cells showed minimal uptake of the nanoparticles with iron content maintaining baseline values. The negligible uptake by the MCF-7 cells indicated that the non-specific internalization pathways contributed minimally to nanoparticles sequestration. This pronounced disparity in the uptake of the G2HA-coated nanoparticles supported the preferential accumulation in CD44-expressing cells, and reinforced the role of the CD44 receptor in mediating the internalization of the nanoparticles. The observed uptake profile therefore validated the selectivity of the CD44-expressing 4T1 cells for the G2HA-functionalized nanoparticles with marginal off-target accumulation. The enhanced uptake in 4T1 cells is consistent with active targeting which was likely prompted by the high affinity of the G2HA coating to the CD44 receptors triggering a receptor-mediated endocytic internalization of the nanoparticles. Overall, these findings underscored the role of the CD44 receptor as the primary gateway for the internalization of the G2HA-coated nanoparticles.



### 3.9 MPI imaging of 4T1 cancer cells

Having confirmed that G2HA remarkably enhanced the uptake of IONPs compared to unmodified HA, it was crucial to validate that the enhanced uptake translated into sharper and more intense signals in MPI. Therefore, 4T1 breast cancer cells were incubated with HA-IONPs or G2HA-IONPs at identical iron concentration. The cells were washed to remove unbound particles and imaged on an MPI scanner as shown in Figure 10. The control cells that did not receive any nanoparticle formulation produced faint background MPI signal (Figure 10a). On the other hand, cells that received G2HA-IONPs produced MPI signals 2.4 folds higher than those of the cells treated with HA-IONPs, highlighting that the enhanced uptake of G2HA-IONPs was concurrent with improved MPI signal (Figure 10b). Such attribute is a testament of the suitability of the designed nanoparticles as contrast agents for MPI. The superior magnetic properties of the nanoparticles generated intense MPI signals with higher contrast and improved spatial resolution, enabling the accurate localization and quantification of the magnetic tracer. The magnetic profile of the prepared IONPs renders it reliable for various biomedical applications such as cell tracking, early detection, and monitoring the progression of diseases.



**Fig. 10** (a) Images showing the loading of the samples on the MPI sample holder and the generated MPI signals; (b) Bar graph showing the quantification of the generated MPI signals.



## 4 Conclusion

The current report described the anchoring of a novel hyaluronan derivative, termed G2HA and characterized by superior binding affinity to CD44 compared to the native HA, onto superparamagnetic iron oxide nanoparticles. The thermal decomposition method was used to prepare the IONP core, and the ligand exchange method was suitable for coating the nanoparticle with the polysaccharides. The desirable magnetic properties of the prepared nanoformulation rendered them suitable targeted contrast agents for MPI. The enhanced binding affinity of G2HA to CD44 translated into prominent uptake by CD44-expressing cancer cells through receptor-mediated endocytosis, and generated stronger dose-dependent signals on the MPI scanner. The unique characteristics of the prepared G2HA-IONPs support their use in efficient cell tracking, early disease detection, and longitudinal monitoring of diseases. While the results of the current study supported efficient and selective uptake of the targeted nanoparticles by cancer cells *in vitro*, achieving comparable selectivity *in vivo* is more complex, and further investigation is needed to validate its clinical applicability.

### Author Contributions

Mohammad H. El-Dakdouki performed all experiments, conducted formal analysis, and wrote the original draft of the manuscript. Chia-wei Yang participated in MPI measurements, while A K M Atique Ullah was involved in the preparation of the nanoparticles. Baraah U. Hijazi contributed to the formal analysis of the results. Fei Fan and Kunli Liu developed and assessed the binding affinity of the modified hyaluronan. Xuefei Huang led conceptualization and supervision of the project, reviewed and edited the manuscript. All authors approve this version of the manuscript.

### Conflict of Interest

There are no conflicts to declare.

### Data Availability

The data supporting the findings of this study are included in the article and its ESI.

### Acknowledgments



The authors are grateful for the generous support of the Fulbright Visiting Scholar Program of the USA Department of State for funding and facilitating Prof. Mohammad H. El-Dakdouki research visit to Michigan State University, USA, as well as financial support by the National Institute of General Medical Sciences, NIH (R35GM158263, XH).

## References

1. L. C. Wu, Y. Zhang, G. Steinberg, H. Qu, S. Huang, M. Cheng, T. Bliss, F. Du, J. Rao, G. Song, L. Pisani, T. Doyle, S. Conolly, K. Krishnan, G. Grant and M. Wintermark, *Am. J. Neuroradiol.*, 2019, **40**, 206.
2. H. Arami, E. Teeman, A. Troksa, H. Bradshaw, K. Saatchi, A. Tomitaka, S. S. Gambhir, U. O. Häfeli, D. Liggitt and K. M. Krishnan, *Nanoscale*, 2017, **9**, 18723-18730.
3. P. W. Goodwill, E. U. Saritas, L. R. Croft, T. N. Kim, K. M. Krishnan, D. V. Schaffer and S. M. Conolly, *Adv. Mater.*, 2012, **24**, 3870-3877.
4. R. M. Ferguson, A. P. Khandhar, H. Arami, L. Hua, O. Hovorka and K. M. Krishnan, *Biomed. Tech. (Berl)*, 2013, **58**, 493-507.
5. Y. Du, X. Liu, Q. Liang, X.-J. Liang and J. Tian, *Nano Lett.*, 2019, **19**, 3618-3626.
6. A. Lindemann, K. Lütke-Buzug, B. M. Fräderich, K. Gräfe, R. Pries and B. Wollenberg, *Int. J. Nanomed.*, 2014, **9**, 5025-5040.
7. H. Paysen, N. Loewa, A. Stach, J. Wells, O. Kosch, S. Twamley, M. R. Makowski, T. Schaeffter, A. Ludwig and F. Wiekhorst, *Sci. Rep.*, 2020, **10**, 1922.
8. Z. W. Tay, S. Savliwala, D. W. Hensley, K. L. B. Fung, C. Colson, B. D. Fellows, X. Zhou, Q. Huynh, Y. Lu, B. Zheng, P. Chandrasekharan, S. M. Rivera-Jimenez, C. M. Rinaldi-Ramos and S. M. Conolly, *Small Methods*, 2021, **5**, 2100796.
9. A. Ali, T. Shah, R. Ullah, P. Zhou, M. Guo, M. Ovais, Z. Tan and Y. Rui, *Front. Chem.*, 2021, **9**, 629054-629054.
10. J. Guzy, S. Chakravarty, F. J. Buchanan, H. Chen, J. M. Gaudet, J. M. L. Hix, C. L. Mallett and E. M. Shapiro, *ACS Appl. Nano Mater.*, 2020, **3**, 3991-3999.
11. H. Arami, A. Khandhar, D. Liggitt and K. M. Krishnan, *Chem. Soc. Rev.*, 2015, **44**, 8576-8607.



12. H. Arami, R. M. Ferguson, A. P. Khandhar and K. M. Krishnan, *Med. Phys.*, 2013, **40**, 071904.
13. R. M. Ferguson, K. R. Minard, A. P. Khandhar and K. M. Krishnan, *Med. Phys.*, 2011, **38**, 1619-1626.
14. A. P. Khandhar, R. M. Ferguson, H. Arami and K. M. Krishnan, *Biomaterials*, 2013, **34**, 3837-3845.
15. C.-W. Yang, K. Liu, C.-Y. Yao, B. Li, A. Juhong, A. K. M. A. Ullah, H. Bumpers, Z. Qiu and X. Huang, *ACS Appl. Mater. Interfaces*, 2024, **16**, 27055-27064.
16. C.-W. Yang, K. Liu, C.-Y. Yao, B. Li, A. Juhong, Z. Qiu and X. Huang, *ACS Appl. Nano Mater.*, 2022, **5**, 18912-18920.
17. R. Marhaba and M. Zöller, *J. Mol. Histol.*, 2004, **35**, 211-231.
18. C. A. Cuff, D. Kothapalli, I. Azonobi, S. Chun, Y. Zhang, R. Belkin, C. Yeh, A. Secreto, R. K. Assoian, D. J. Rader and E. Puré, *J. Clin. Invest.*, 2001, **108**, 1031-1040.
19. M. Krolkoski, J. Monslow and E. Puré, *Matrix Biol.*, 2019, **78-79**, 201-218.
20. C. Chen, S. Zhao, A. Karnad and J. W. Freeman, *J. Hematol. Oncol.*, 2018, **11**, 64-64.
21. D. S. Bhattacharya, D. Svechkarev, J. J. Soucek, T. K. Hill, M. A. Taylor, A. Natarajan and A. M. Mohs, *J. Mater. Chem. B*, 2017, **5**, 8183-8192.
22. A. R. Jordan, R. R. Racine, M. J. P. Hennig and V. B. Lokeshwar, *Front. Immunol.*, 2015, **6**.
23. P. Heldin, E. Karousou, B. Bernert, H. Porsch, K. Nishitsuka and S. S. Skandalis, *Connect. Tissue Res.*, 2008, **49**, 215-218.
24. B. P. Toole, *Clin. Cancer Res.*, 2009, **15**, 7462-7468.
25. S. Misra, P. Heldin, V. C. Hascall, N. K. Karamanos, S. S. Skandalis, R. R. Markwald and S. Ghatak, *FEBS J.*, 2011, **278**, 1429-1443.
26. K. Liu, W. M. McCue, C.-w. Yang, B. C. Finzel and X. Huang, *Carbohydr. Polym.*, 2023, **300**, 120255.
27. S. Sun, H. Zeng, D. B. Robinson, S. Raoux, P. M. Rice, S. X. Wang and G. Li, *J. Amer. Chem. Soc.*, 2004, **126**, 273-279.
28. M. H. El-Dakdouki, K. El-Boubbou, D. C. Zhu and X. Huang, *RSC Adv.*, 2011, **1**, 1449-1452.
29. M. H. El-Dakdouki, E. Puré and X. Huang, *Nanoscale*, 2013, **5**, 3895-3903.



30. B. Rezaei, Z. W. Tay, S. Mostufa, O. N. Manzari, E. Azizi, S. Ciannella, H.-E. J. Moni, C. Li, M. Zeng, J. Gómez-Pastora and K. Wu, *Nanoscale*, 2024, **16**, 11802-11824.
31. L. M. Bauer, S. F. Situ, M. A. Griswold and A. C. S. Samia, *Nanoscale*, 2016, **8**, 12162-12169.
32. X. Han, Y. Li, W. Liu, X. Chen, Z. Song, X. Wang, Y. Deng, X. Tang and Z. Jiang, *Journal*, 2020, **10**.
33. H. T. K. Duong, A. Abdibastami, L. Gloag, L. Barrera, J. J. Gooding and R. D. Tilley, *Nanoscale*, 2022, **14**, 13890-13914.
34. X. Xie, J. Zhai, X. Zhou, Z. Guo, P.-C. Lo, G. Zhu, K. W. Y. Chan and M. Yang, *Adv. Mater.*, 2024, **36**, 2306450.
35. Y. Hadadian, H. Masoomi, A. Dinari, C. Ryu, S. Hwang, S. Kim, B. k. Cho, J. Y. Lee and J. Yoon, *ACS Omega*, 2022, **7**, 15996-16012.
36. P. Gong, Q. Chen, K. Shih, C. Liao, L. Wang, H. Xie and S. Deng, *J. Mater. Sci. Mater. Electron.*, 2019, **30**, 20970-20978.
37. M. Salimi, W. Wang, S. Roux, G. Laurent, R. Bazzi, P. Goodwill, G. Liu and J. W. M. Bulte, *Nanoscale Adv.*, 2025, **7**, 1018-1029.
38. Q. Feng, Y. Liu, J. Huang, K. Chen, J. Huang and K. Xiao, *Scientific Reports*, 2018, **8**, 2082.
39. M. Rezaeeyazdi, T. Colombani, L. J. Eggermont and S. A. Bencherif, *Mater. Today Bio*, 2022, **13**, 100207.
40. O. Zimmer and A. Goepferich, *Eur. J. Pharm. Biopharm.*, 2024, **198**, 114240.
41. S. Merminod, J. R. Edison, H. Fang, M. F. Hagan and W. B. Rogers, *Nanoscale*, 2022, **13**, 12602-12612.
42. J. E. Silpe, M. Sumit, T. P. Thomas, B. Huang, A. Kotlyar, M. A. van Dongen, M. M. Banaszak Holl, B. G. Orr and S. K. Choi, *ACS Chem. Biol.*, 2013, **8**, 2063-2071.
43. Y. Wang, V. Ukwattage, Y. Xiong and G. K. Such, *Mater. Horiz.*, 2025, **12**, 3622-3632.
44. Z. P. G. Xu, *Pharm. Res.*, 2022, **39**, 1035-1045.
45. T. Murai, *Front. Immunol.*, 2015, **6**, 420-428.
46. H. S. S. Qhattal and X. Liu, *Mol. Pharm.*, 2011, **8**, 1233-1246.
47. T.-G. Iversen, T. Skotland and K. Sandvig, *Nano Today*, 2011, **6**, 176-185.



### Data Availability

The data supporting the findings of this study are included in the article and its ESI.

Open Access Article. Published on 16 June 2026. Downloaded on 6/17/2026 6:00:24 AM.  
This article is licensed under a Creative Commons Attribution 3.0 Unported Licence.

