















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Photoluminescence of 2D and 3D quantum dots synthesized by laser-ultrasonic treatment on van der Waals materials

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A two-stage method for synthesis of van der Waals quantum dots in a liquid by laser ablation of the initial material followed by ultrasonic treatment of the samples is proposed and implemented. The use of various initial samples in the form of pressed powder targets or transition metal dichalcogenide crystals allows the fabrication of both three-dimensional and flat few-layer quantum dots that are capable of exhibiting a photoluminescent response in a wide spectral range. The possibility of straightforward control of photoluminescence spectra in ensembles of MoS₂, WS₂, MoSe₂, and WSe₂ quantum dots with wide-size dispersions by tuning the pump laser wavelength is shown.

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1 Introduction

At one time, the synthesis of quantum dots (QDs) in the form of bulk nanocrystals opened new horizons for experimental quantum physics and photonics.^{1,2} The introduction of new highly anisotropic materials into laboratory practice using the layered van der Waals (vdW) materials has raised non-trivial questions about reliable methods for obtaining quantum dots based on them and fundamental properties of their photoluminescence.³ Fundamentally, the nature of quantum confinement in vdW QDs is inherently anisotropic and can manifest in two distinct morphological archetypes.⁴ The first consists of planar, few-layer QDs (2D QDs), where confinement is strong in the out-of-plane direction (defined by the layer thickness) and tunable in the in-plane direction (defined by lateral dimensions). These structures are expected to exhibit PL across various sizes, with the emission energy governed by lateral confinement.^{5,6} The second archetype comprises quasi-spherical, polycrystalline nanocrystals (3D QDs). In these structures, the bulk vdW material's indirect bandgap persists unless the overall size is reduced to a few nanometers,

commensurate with only a few atomic layers, where the band structure begins to resemble that of a monolayer.^{7,8} The ability to selectively synthesize these distinct archetypes is critical, as the dimensionality of confinement dictates their fundamental optoelectronic behavior and suitability for specific applications, ranging from active flat optics (2D)^{9,10} to theranostics (3D).^{11–13}

Precise synthesis control over these morphologies is crucial for harnessing their distinct optical properties.⁴ Current synthesis methods, including modified liquid-phase exfoliation,¹⁴ and laser ablation of TMDC powders or crystals,^{15–18} often yield ensembles with ambiguous morphologies or require complex chemical procedures. Femtosecond laser ablation in liquids (LAL) is a versatile technique for nanoparticle synthesis; however, high-energy exposure can induce non-equilibrium processes, such as coulomb explosion, leading to the formation of QDs.^{8,11,19} Secondary treatments, such as ultrasonication (US), have been employed to exfoliate these nanoparticles or fragment bulk materials into planar QDs.^{20,21} Yet, a deterministic approach to selectively synthesize either 2D or 3D vdW QDs remains elusive.

In this work, we applied a two-stage method for the synthesis of vdW QDs with various shapes using femtosecond laser ablation followed by ultrasonic exposure on various samples of transition metal dichalcogenides, “fs+US” treatment, as shown in Fig. 1a and b. We put forward and validate the hypothesis that the physical form of the initial bulk TMDC target, namely, low-density pressed powder *versus* a high-density monolithic crystal, can serve as a macroscopic adjustor to determine the final, nanoscale morphology of the obtained QDs, yielding either planar 2D flakes or quasi-spherical 3D nanocrystals. To

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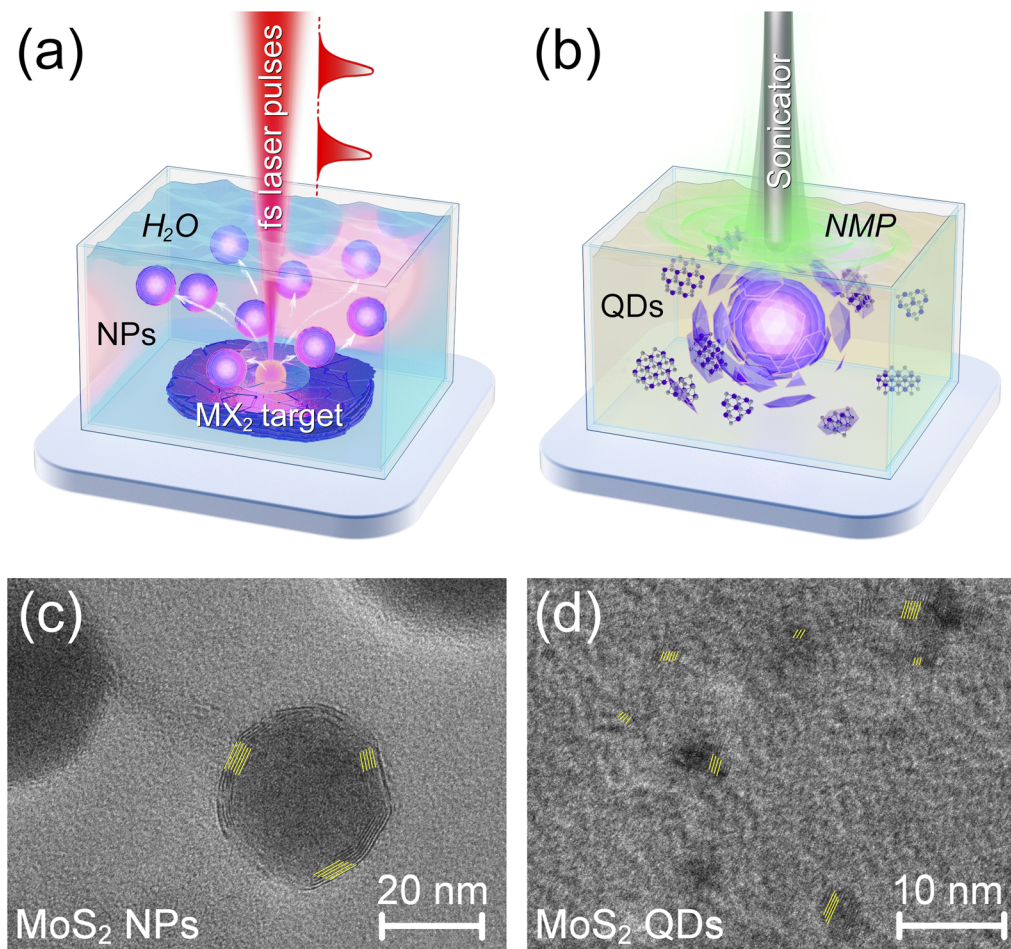


Fig. 1 Concept of two-stage synthesis of nanostructures using (a) femtosecond laser ablation of the target placed in water followed by transfer of NPs to NMP and centrifugation for subsequent (b) ultrasonication. TEM images of (c) core-shell NPs synthesized by laser ablation of the MoS₂ target in water and (d) MoS₂ QDs obtained after transfer of NPs to NMP and sonication. The average distances between lattice fringes in panels (c) and (d) are 0.61 nm and 0.28 nm, respectively.

characterize the obtained QDs and reveal their features, we employed the methods of Raman and infrared spectroscopy, transmission electron microscopy, and the numerical analysis of experimentally measured photoluminescence spectra. The developed approaches and methods can be used for the synthesis of photoluminescent vdW QDs with given spectral features.

2 Results and discussion

2.1 Synthesis and characterization of quantum dots

At the initial stage, we studied the influence of high-energy exposure on the solutions of vdW materials, as shown in Fig. 1a. A femtosecond Yb:KGW laser system TETA-10 (Avesta Project Ltd, Russia) was used as a radiation source, which provides the emission at a wavelength of 1030 nm with a pulse duration of 280 fs and pulse repetition rate of 10 kHz. Laser beam scanning was carried out using a galvanometer scan head RLA-1504 (RAYLASE GmbH, Germany); beam focusing was performed using an F-Theta lens with a working distance of 200 mm, and the beam diameter in the caustic region was about 70

μm. The samples of indirect band gap transition metal dichalcogenides were used as the starting material for treatment. For all materials, the laser ablation synthesis was carried out in deionized water; the samples were placed at the bottom of a cuvette filled with 10 ml of liquid, and the height of the liquid above the target surface was 4 mm. The pulse energy for ablation of MoS₂ and WS₂ materials was 50 μJ, and the fluence value F_{ab} was 2.6 J cm⁻². For ablation of MoSe₂ and WSe₂ materials, the pulse energy was 100 μJ with a fluence of 5.2 J cm⁻². For MoS₂ and WS₂ samples, the treatment area was a square with a side of 4 mm, the density of scanning tracks was 20 lines per mm, the scanning speed was 100 mm s⁻¹, and the process duration is 3 minutes. For MoSe₂ and WSe₂ samples, the treatment area was a square with a side of 2 mm, the density of scanning tracks was 20 lines per mm, the scanning speed was 20 mm s⁻¹, and the process duration is 10 minutes. Note that the ablation thresholds F_{th} for the used materials are 0.05, 0.3, 0.09, and 0.01 J cm⁻² corresponding to MoS₂, WS₂, MoSe₂, and WSe₂, respectively. Therefore, the coefficients $k = F_{ab}/F_{th}$, showing the relation between fluence and threshold values, are



52, 8.67, 57.78, and 520 for MoS₂, WS₂, MoSe₂, and WSe₂, respectively.

The use of various samples in the form of both pressed powder and crystals for laser ablation facilitated the synthesis of nanoparticles with different morphologies in the first step, as shown in Fig. S1.1. In particular, the use of pressed powder targets for ablation resulted in the formation of relatively friable MoS₂ and WS₂ nanoparticles with concentric arrangement of layers, as shown in Fig. 1c and S1.1b for such MoS₂ NPs.

These types of NPs can be obtained even by ablation from lower-density crystals of MoS₂ and WS₂ materials,²² and the use of pressed powder additionally reduces the density of the target material to form a thicker concentric-layered shell, as shown in Fig. 1c. In contrast, the higher-density crystals of MoS₂, MoSe₂ and WSe₂ materials provide the formation of only heavy polycrystalline NPs with various sizes without even a small shell. A comparison of NPs with similar sizes, fabricated from different materials and target types, is provided in Figs. S1.1a, b, and d. Note that in the case of laser exposure on the MoSe₂ crystal with the ratio *k* similar to the one of the MoS₂ material, the formation of a wide variety of NPs is observed, including crystallites shown in Fig. S1.1 and even spherulites, *i.e.*, particles consisting entirely of concentric-layered shells.

At the second stage, the ultrasonic exposure on the obtained colloidal solutions was performed, as shown in Fig. 1b. At this stage, *N*-methyl-2-pyrrolidone (NMP) was chosen as a high-fluidity compound with a small size of molecules. The intermediate stage was the centrifugation of colloidal solutions obtained by laser ablation synthesis in order to sediment NPs and their subsequent transfer to NMP. The stability of the nanoparticles during this solvent exchange relies on the inherent properties acquired during laser ablation; NPs synthesized *via* LAL typically possess a significant surface charge, providing electrostatic stabilization.²³ Centrifugation was carried out in centrifuge tubes (1 ml of solution in each tube) using a micro-centrifuge D3024 (DLAB Scientific Co., Ltd) at the relative centrifugal force (RCF) 10 000×*g*, and the process duration is 15 minutes. After the centrifugation, the supernatant was collected in such a way that the minimum possible amount of a liquid remains in the test tube. Next, 1 ml of NMP was immediately added to each test tube. To prevent agglomeration and ensure efficient redispersion of the “soft” pellet, the test tubes were promptly placed in a low-power ultrasonic bath for 2 minutes. Thus, the deionized water was replaced with NMP. The obtained solutions of NPs in NMP were transferred to 10 ml glass test tubes. Ultrasonic exposure on colloidal solutions was carried out by means of a probe sonicator USTA-0,1/28-O (U-Sonic, Biysk, Russia) with an intensity of US exposure of 240 W cm⁻². The probe was immersed in the test tube to a depth of 50 mm. To avoid overheating during US treatment, the glass test tube was placed in a bath filled with a cold water and ice. The duration of US treatment was 30 minutes.

The results of ultrasonic exposure on the solutions obtained after laser ablation significantly depend on the solutions of ablated NPs. Ultrasonic treatment of the solutions obtained from the crystals did not result in the modification of NPs, but contributed to their characteristic agglomeration, as shown in

Fig. S1.2a. At the same time, such agglomerates could contain both fairly large NPs and smaller 3D QDs. Moreover, the similar US treatment of solutions obtained from MoS₂ pressed powder targets led to the appearance of a large number of small particles, *i.e.*, quantum dots with a planar alignment of layers similar to that in Fig. 1d and S1.2b. We assume that the appearance of such flat 2D QDs can be due to the destructive effect of ultrasound on the shells of large concentric-layered NPs in NMP, as shown in Fig. 1b and d. This is also verified by the measured characteristic distances between lattice fringes of 0.61 nm in the shell of MoS₂ NPs immediately after femtosecond laser treatment as shown in Fig. 1c and 0.28 nm in QDs after US treatment as shown in Fig. 1d. In the former case, this distance corresponds to the interlayer distance in MoS₂, and in the latter case, it corresponds to the interplanar spacing for this material. In other words, when the concentric-layered shells break down, the small planar fragments, *i.e.*, quantum dots, are formed. The “fs+US” treatment of the MoS₂ target yields solutions containing both flat 2D QDs and residual large NPs, as evidenced in Fig. S1.3b. Thus, the final results of the two-stage treatment of various samples are the solutions containing both the large NPs in all fractions and smaller QDs of various types. We are talking about a small number of 3D (spherical) QDs in solutions obtained from the initial crystals, as well as the presence of flat 2D QDs in solutions prepared from pressed powder targets.

The detailed analysis of transmission electron microscopy (TEM) images of NPs in solutions obtained by two-stage treatment based on pressed powder MoS₂ and WS₂ targets, as well as based on MoSe₂ and WSe₂ crystals, shown in Fig. 2 and 3, revealed the presence of NP distributions with wide-size dispersion for all target materials. Based on the analysis of NP size distributions in the left insets of Fig. 2, the average sizes of MoS₂ and WS₂ NPs are 4.4 nm and 4.7 nm, respectively, whereas for MoSe₂ and WSe₂ NPs in Fig. 3 these parameters are 29.7 nm and 6.1 nm, respectively. Raman spectroscopy was used to analyze the chemical composition of the obtained NPs and QDs, as shown in Fig. 2c, d and 3c and d. Based on Raman analysis, it can be assumed that the MoS₂ and WS₂ particles in the solutions in Fig. 2 are in the crystalline phase. Besides, the characteristic distance between peaks indicates the predominant presence of a bulk material, which corresponds to the response of large bulk NPs dominating in the solution, as shown in histograms in the insets of Fig. 2a and b.

At the same time, TEM images of single QDs shown in Fig. 2a clearly indicate the presence of small MoS₂ quantum dots with planar nature. Moreover, the right insets of Fig. 2b show the large fragments of shells pulled from core-shell WS₂ nanoparticles during US exposure on them. Furthermore, it is possible to distinguish the pieces detached from the shells, which can become QDs.

In contrast, after fs+US treatment, nanoparticles made of selenide crystals remain spherical and either retain a concentric geometry, as for the MoSe₂ material in Fig. 3a, or become polycrystalline nanoparticles of various sizes, as for the WSe₂ material in Fig. 3b. Note that, according to the histograms in Fig. 3a and b, the average sizes of the obtained MoSe₂ NPs are significantly larger than those of WSe₂ NPs with almost the



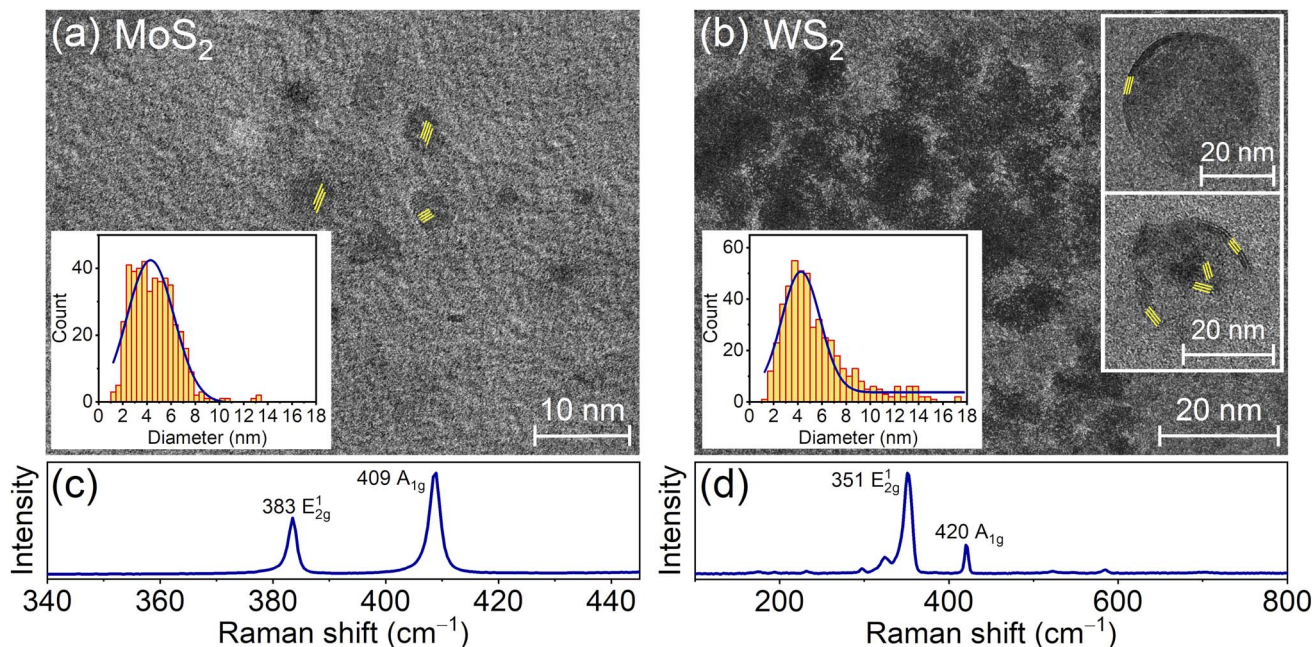


Fig. 2 (a and b) TEM images of NPs obtained by the complex method of laser ablation in water and subsequent ultrasonic (fs+US) treatment in NMP from pressed powder targets made of (a) MoS₂ and (b) WS₂ materials. (c and d) Raman spectra of drop-cast solutions presented in panels (a) and (b). The left insets of panels (a) and (b) show the histograms of size distributions for the obtained NPs and right insets of panel (b) show the images of shells pulled from nanoparticles (bottom inset), as well as the forming QD (top inset). The average distances between lattice fringes in panel (a) and in the insets of panel (b) are 0.28 nm and 0.62 nm, respectively.

same numbers of 3D QDs for these two materials. The point is that the two processes can occur during laser exposure on dense crystals. At low pump fluence, thermodynamic equilibrium melting with evaporation and formation of large NPs is realized.

At high pump fluence, a nonequilibrium phase (Coulomb) explosion can occur, which leads to the formation of small QDs.^{19,24} Apparently, under exposure on a moving crystal (the use of a magnetic stirrer), its illuminance changes over time

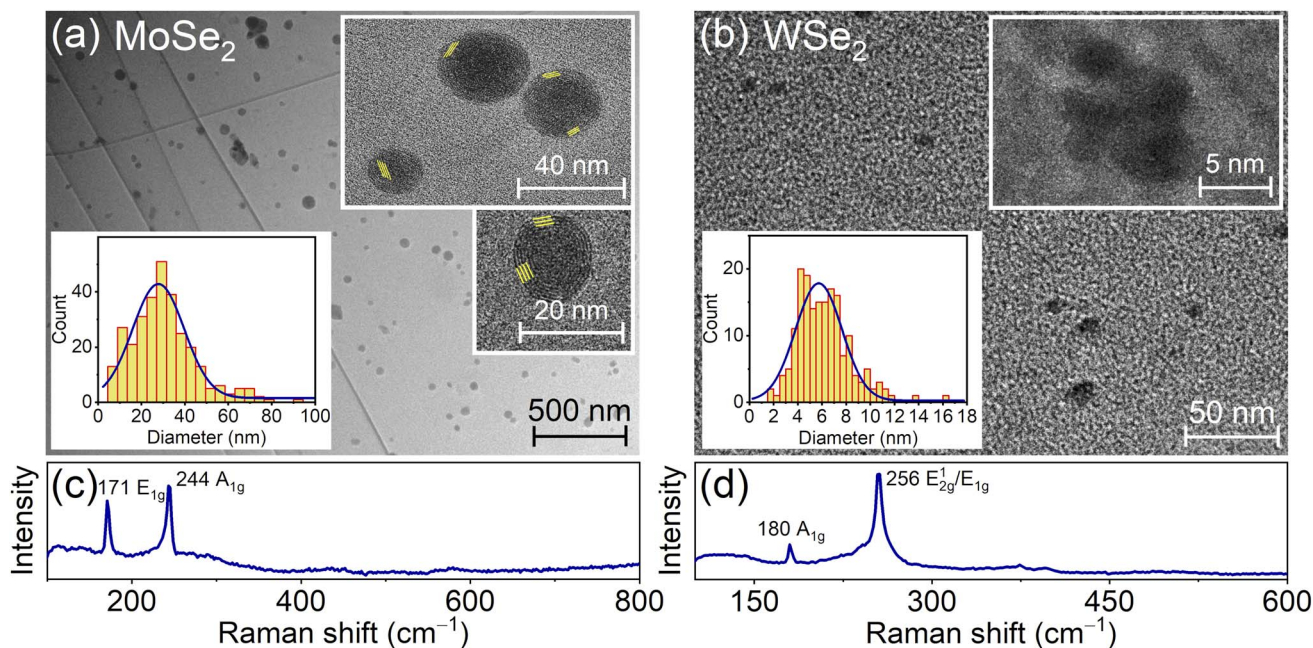


Fig. 3 (a and b) TEM images of NPs obtained by the complex method of laser ablation in water and subsequent ultrasonic (fs+US) treatment in NMP from (a) MoSe₂ and (b) WSe₂ crystals. (c and d) Raman spectra of solutions presented in panels (a) and (b). In panels (a) and (b), the left insets show the histograms of size distributions for the obtained NPs and right insets show the images of nanoparticles and 3D QDs. The average distance between lattice fringes in the insets in panel (a) is 0.68 nm.



even at a fixed pump fluence. Therefore, the formation of NPs and QDs occurs simultaneously. At the same time, the ratios between numbers and sizes of NPs and QDs can be determined by the exceptional properties of the used crystals. In particular, for the same sizes and numbers of QDs, the yields and sizes of NPs can be different, which can be clearly seen for the comparison of MoSe₂ and WSe₂ materials in Fig. 3. Besides, in the case of the MoSe₂ material, the large thickness of the shell in combination with the used power of US exposure does not allow the breakdown of NPs up to the QD state in contrast to the case shown in Fig. 2. In the case of the WSe₂ material, the NPs possess a polycrystalline structure immediately after femto-second laser treatment and retain it after US exposure. To verify the chemical composition of the obtained NPs and QDs, we performed the energy dispersive X-ray analysis of the solutions, which confirmed the agreement between chemical compositions of particles and initial samples for ablation, as shown in Fig. S1.3.

In general, fs+US treatment of vdW materials in a liquid leads to the formation of both vdW NPs and 2D and 3D vdW QDs in the solution. However, even without any separation, the photoluminescence properties of such solutions are determined solely by the concentration and morphological features of quantum-sized particles, including multilayer flakes of considerable area.

2.2 Photoluminescence control in QD ensembles with a wide-size dispersion

The study of the photoluminescence properties of the prepared solutions was carried out using a Varioskan LUX Multimode Microplate Reader (Thermo Fischer Scientific, USA). A double monochromator scheme was used for excitation control and emission analysis. The excitation wavelength was varied in the range from 300 nm to 400 nm while the detection range of the PL signal was from 325 nm to 600 nm. The QD solutions were

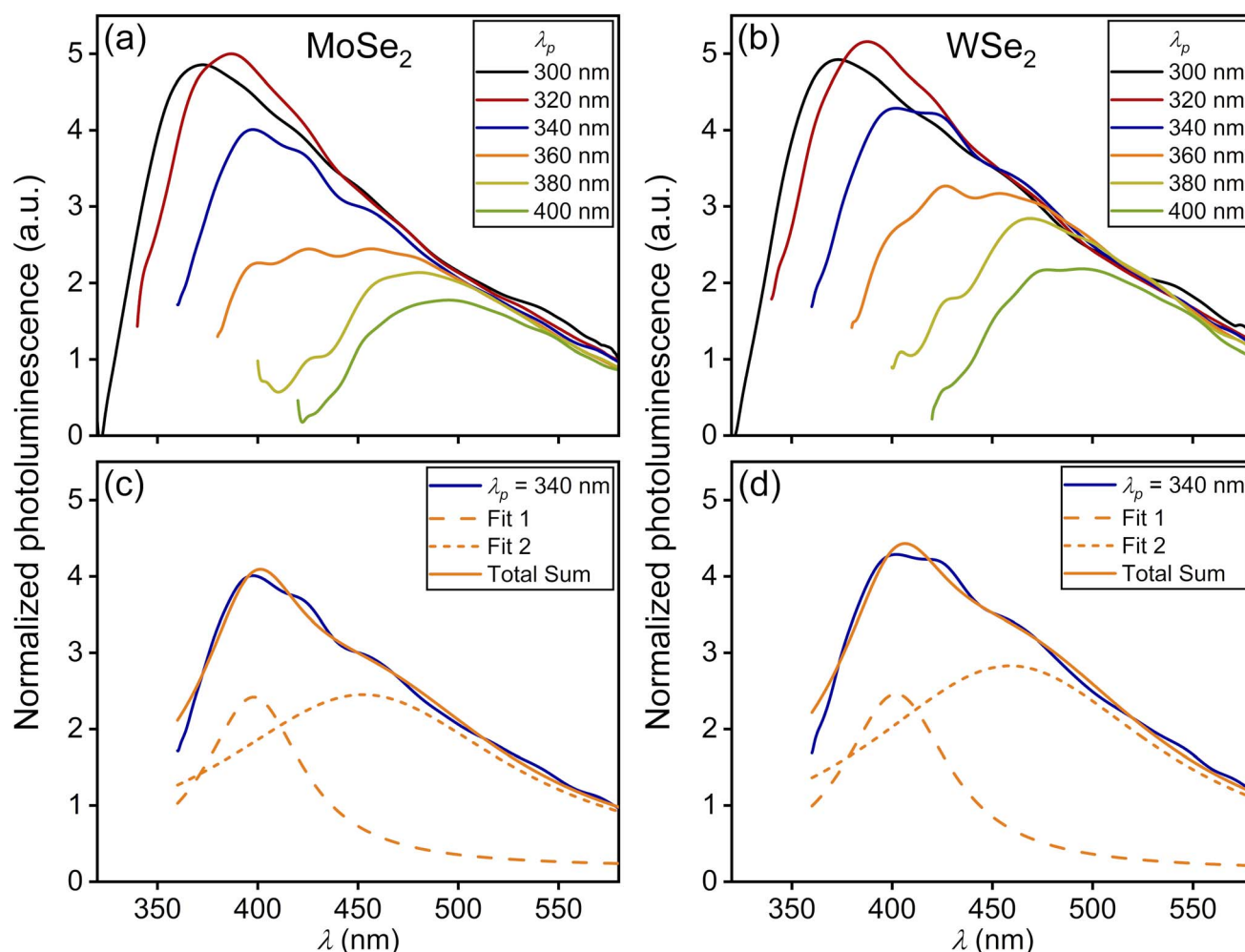


Fig. 4 Photoluminescence spectra for an ensemble of 3D QDs with a wide-size dispersion excited by various pump wavelengths $\lambda_p = 300, 320, 340, 360, 380, 400$ nm and exhibiting the corresponding photoluminescence maxima λ_{PL} (with QD sizes $D_{QD} = 2R$ calculated according to the formula (1)) for solutions obtained using the two-stage method fs + US on the basis of the (a) MoSe₂ material with $\lambda_{PL} = 373, 387, 398, 457, 481, 495$ nm ($D_{QD} = 1.73, 1.78, 1.82, 2.03, 2.12, 2.18$ nm) and (b) WSe₂ material with $\lambda_{PL} = 373, 388, 402, 427, 469, 495$ nm ($D_{QD} = 1.84, 1.90, 1.95, 2.05, 2.23, 2.35$ nm). (c) and (d) Deconvolution of photoluminescence spectra for MoSe₂ and WSe₂ QDs, respectively, under excitation by pump wavelength $\lambda_p = 340$ nm.



transferred to 96-well plastic plates with a UV-transparent well bottom prior to the PL spectral measurements. First, we consider the general properties of QD photoluminescence, which are independent of their morphology and the type of vdW material used for their fabrication. It is known that the spectral position of the PL peak of the QD ensemble with a narrow-size dispersion is almost independent of the pump wavelength λ_p , while its intensity first increases and then decreases with decreasing λ_p .^{14,17} However, for QD ensembles with a wide-size dispersion, as shown in Fig. 2 and 3, the PL peak position shows significant variation with pump wavelength, as shown in Fig. 4 and 5. The pump-dependent PL behavior in QDs originates from size-tunable interband transitions governed by quantum confinement effects.²⁵ Regardless of the QD type, the energy of the interband transition increases upon decreasing the QD size.^{26,27} Therefore, if the pump wavelength λ_p is greater than the wavelength λ_{IB} of the main $1S(e) \rightarrow 1S(h)$ transition for all QD subsets in the ensemble, then such an ensemble is not

excited and, consequently, does not emit the light, as shown in Section S2.5. As soon as the pump wavelength becomes less than λ_{IB} of the largest emitters in the ensemble, it begins to emit photoluminescence. A further decrease in λ_p results in engaging of more and more new subsets of QDs with smaller sizes. Thus, the shorter PL wavelengths begin to contribute to the PL process, which leads to a blue shift of the maximum of the PL spectrum, as shown in Fig. 4a, b and 5a and b.^{5,14,20} In this case, each pump wavelength can be associated with the most probable QD size, which mainly contributes to the photoluminescence, as shown in Table S3.1. However, the resulting PL spectrum is formed by not only such QDs with the maximum contribution to PL, but also all other QD fractions with larger sizes that determine the shape of the long-wavelength spectral range. As a result, all PL spectra for QD ensembles with a wide-size dispersion in Fig. 4 and 5 have very asymmetric shapes with steep short-wavelength and gently sloping long-wavelength sides. At the same time, there is an actual limitation on the

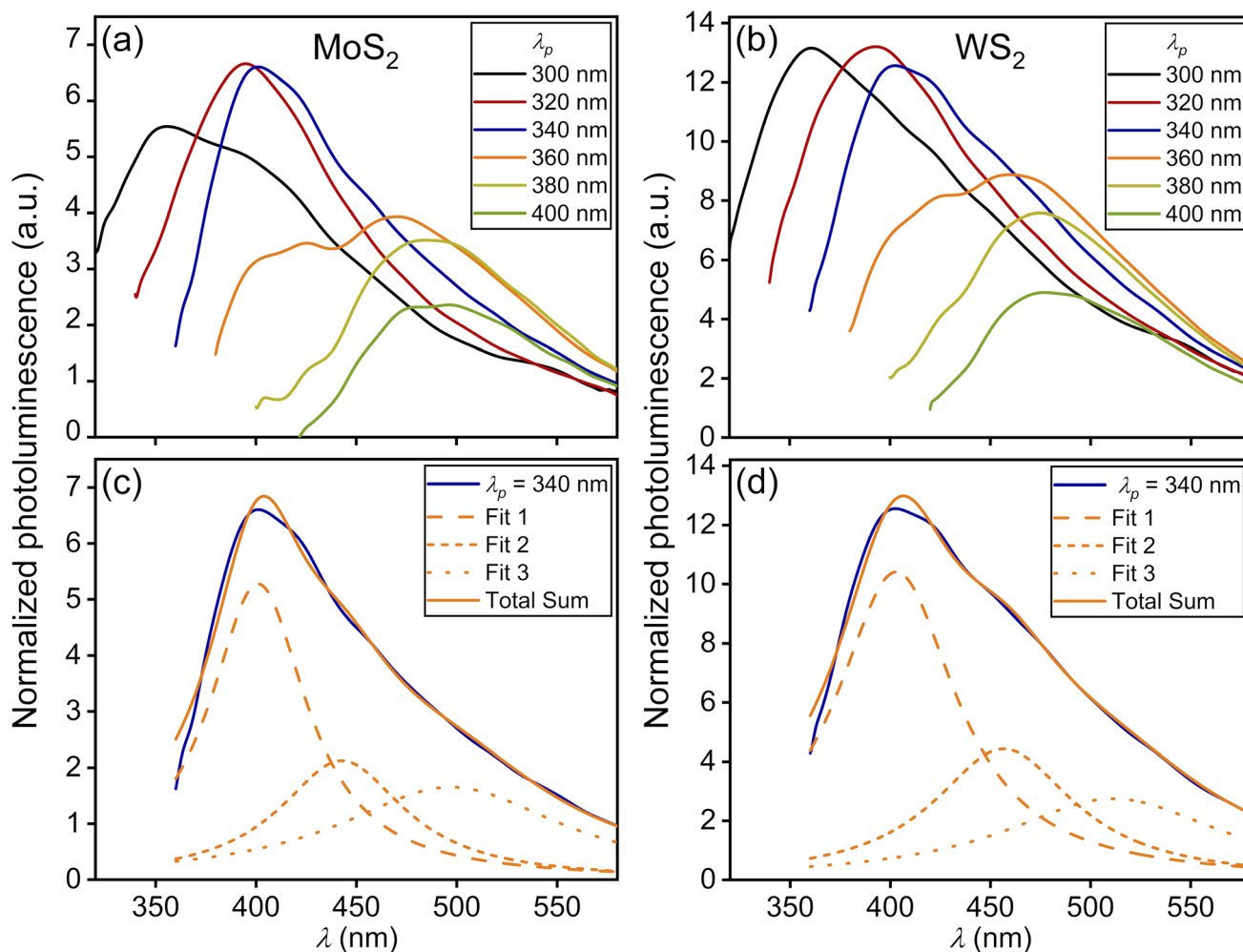


Fig. 5 Photoluminescence spectra for an ensemble of 2D QDs with a wide-size dispersion excited by various pump wavelengths $\lambda_p = 300, 320, 340, 360, 380, 400$ nm and exhibiting the corresponding photoluminescence maxima λ_{PL} (with QD sizes $D_{QD} = 2\rho_0$ calculated according to the formula (3)) for solutions obtained using the two-stage method fs + US on the basis of the (a) MoS₂ material with $\lambda_{PL} = 356, 395, 401, 471, 488, 498$ nm ($D_{QD} = 1.54, 1.72, 1.75, 2.17, 2.30, 2.38$ nm) and (b) WS₂ material with $\lambda_{PL} = 361, 393, 403, 459, 475, 477$ nm ($D_{QD} = 1.84, 2.06, 2.14, 2.69, 2.90, 2.93$ nm). (c) and (d) Deconvolution of photoluminescence spectra for MoS₂ and WS₂ QDs, respectively, under excitation by pump wavelength $\lambda_p = 340$ nm.



minimally available QD size in each ensemble, which together with a decrease in the quantum yield for the short-wavelength pump¹⁴ leads to a decrease in PL intensity under the $\lambda_p < 360$ nm condition for all spectra presented in Fig. 4 and 5.

Now we focus on the characteristic features that are inherent in QDs made of indirect band gap transition metal dichalcogenides being considered in this work. As can be seen in Fig. 3, the MoSe₂ and WSe₂ particles are 3D particles possessing a spherical shape.

The energy of the main interband transition in 3D QDs with a conventional spherical shape can be estimated using the formula (see Section S2)

$$E_g^{3D} = eE_g^{\text{bulk}} + \frac{\hbar^2 \pi^2}{2R^2 \mu^{\text{bl}}}, \quad (1)$$

where $\mu^{\text{bl}} = \frac{m_e^{\text{bl}} m_h^{\text{bl}}}{m_e^{\text{bl}} + m_h^{\text{bl}}}$ is the reduced mass of an electron and hole, $m_{e(h)}^{\text{bl}}$ is the effective mass of an electron (hole) for an isotropic (bulk) material, E_g^{bulk} is the bulk band gap for a QD material, R is the sphere radius, and e is the absolute value of electron charge, as shown in Section S2.3. However, the expression (1) is formally insensitive to the features of the interband transition in the wave-vector space. Both concentric MoSe₂ and polycrystalline WSe₂ particles exhibit light emission only below a critical diameter corresponding to 4–5 atomic layers, *i.e.*, when the transition actually becomes direct band gap in its nature.⁶ Thus, only the smallest particles of the entire ensembles of MoSe₂ and WSe₂ NPs with wide-size dispersions, which correspond to the left side of histograms for the NP size distributions in the insets of Fig. 3, are able to emit photoluminescence. In this regard, the total PL intensity of the ensemble of MoSe₂ and WSe₂ QDs is not so high, and the position of the long-wavelength side of the spectra in Fig. 4 is insensitive to the changes in pump wavelength.

Since the evaluated average size of 3D QDs emitting photoluminescence under excitation at pump wavelength $\lambda_p = 400$ nm takes the value 2.18 nm for MoSe₂ and 2.35 nm for WSe₂, the larger QDs, being indirect-band-gap QDs, are almost not engaged in the photoluminescence. For the mathematical description of QD PL, we use the deconvolution method^{28,29} with a set of Lorentz distribution functions for the intensity in the form:

$$I_{\text{PL}}(\lambda) = A \frac{\gamma_{\text{LD}}^{(\lambda)} / 2}{(\lambda - \lambda_0)^2 + \left(\gamma_{\text{LD}}^{(\lambda)} / 2\right)^2}, \quad (2)$$

where λ_0 is the central wavelength and $\gamma_{\text{LD}}^{(\lambda)}$ is the full width at half maximum (FWHM) in the wavelength scale, and A is the fitting parameter. Note that the calculation of the spontaneous relaxation rate for the considered QDs is presented in Section S2.8. However, we do not use these findings in our simulations because they do not take into account ensemble effects. At the same time, a clear tendency to increase the relaxation rate for single 2D QDs can be exploited for their coupling to planar photonic structures. For the case of PL from the QD ensembles with a wide-size dispersion pumped at a wavelength of 340 nm, the PL spectra for MoSe₂ and WSe₂ QDs are well approximated

by using two functions (2) with the parameters presented in Table S3.2. Moreover, the Lorentz distribution function at a wavelength of 398 nm (402 nm) for MoSe₂ (WSe₂) QDs with an average size of 1.82 nm (1.95 nm) well approximates the short-wavelength side of the PL spectrum, as shown in Fig. 4c and d. At the same time, the second function peaking at a wavelength of 452.6 nm (458.8 nm) for MoSe₂ (WSe₂) QDs corresponds to the PL from larger QDs with an average size of 2.02 nm (2.19 nm), which confirms the PL behavior for the ensemble with a wide-size dispersion.

Finally, the obtained sulfide-based 2D MoS₂ and WS₂ QDs exhibit a slightly different size dependence of the 1S(e) → 1S(h) transition compared to the 3D case (see Section S2):

$$E_g^{2D} = eE_g^{\text{ML}} + \frac{\hbar^2 2.4^2}{2\rho_0^2 \mu^{\parallel}}, \quad (3)$$

where $E_g^{\text{ML}} = eE_g^{\text{bulk}} + \frac{\hbar^2 \pi^2}{2h^2 \mu^{\perp}}$ is the band gap of the infinite monolayer, $\frac{1}{\mu^{\perp(\parallel)}} = \frac{1}{\mu_e^{\perp(\parallel)}} + \frac{1}{\mu_h^{\perp(\parallel)}}$ is the reduced effective mass, $m_e^{\perp(\parallel)}$ ($m_h^{\perp(\parallel)}$) is the effective mass of an electron (hole) across (\perp) and along (\parallel) layers of the material, and ρ_0 and h are the cylinder radius and height, respectively, if we consider a QD with cylindrical shape, as shown in Section S2.4. These anisotropic quantum emitters exhibit dual quantum confinement: one arising from their monolayer/few-layer thickness, and another from their in-plane QD dimensions. These few-layer QDs exhibit photoluminescence across all size scales, from nanoscale regions to macroscopic flakes, with emission corresponding to their band gap transition.⁶ Note that in further calculations we consider E_g^{ML} as the energy of the interband transition of the monolayer, as shown in Table S2.4.

Fig. 5 shows that the PL curves of MoS₂ and WS₂ QDs for various pump wavelengths do not form bundles of closely spaced curves at the long-wavelength side in addition to the increase in maximum PL intensity, which is especially noticeable for 2D WS₂ QDs. Actually, this is due to the efficient pump-dependent photoluminescence of large-area 2D NPs that mainly contribute to this part of spectra. Remarkably, the deconvolution method for 2D QDs yields results in good agreement between theory and experiment when fitting by using an already larger number of functions of type (2) with increasing central wavelengths, as shown in Fig. 5c and d, and Table S3.3. This verifies the fact of photoluminescence of large 2D QDs in the long-wavelength region.

It should be noted that US exposure in NMP can lead to both the formation of surface and edge defects in QDs,³⁰ as well as QD functionalization with carbon-based groups induced by laser-assisted solvent-QD interaction.^{16,20} These effects can contribute to pump-dependent behavior of the PL spectra of QDs. However, the complementary infrared spectroscopy studies of the solutions obtained after US treatment indicate the minimal or negligible functionalization of luminescent QDs, as shown in Section S5.

On a final note, the synthesis of vdW QDs with tunable photoluminescence requires further study using the entire variety of vdW materials. Further research will focus on



identifying clear criteria for sample preparation and energy impacts on the samples in order to obtain chromophores with a given morphology and size. In general, the differences between the PL spectra of 2D and 3D QD ensembles in the far-field region are not so significant, especially for the ensembles with a wide-size dispersion as in this work. However, the fundamental distinction lies not only in their spectral behavior (e.g., the indirect gap threshold in 3D QDs) but also in the anisotropy of their emission at the single-emitter level. The planar morphology of 2D QDs results in highly anisotropic transition dipole moments, predominantly oriented in-plane. This leads to distinct, directional radiation patterns compared to the relatively isotropic emission of 3D QDs.^{31–34} At the same time, the photoluminescent QDs possessing various morphologies may have considerably different applications. For example, the ability of 3D QDs to easily penetrate the bloodstream could be useful for theranostic applications.^{11,12} However, the ability of 2D QDs to be simply integrated with planar interfaces may find application in active flat optics, including active imaging lenses,⁹ microlasers and luminescent metamaterials,^{35–40} as well as integrated plasmonic devices^{41–43} and single photon sources.^{34,44} In the latter case, the use of single 2D QDs requires a detailed study of the near-field response and radiation patterns of such strongly anisotropic quantum emitters, including adaptation of methods of near-field diagnostics.⁴⁵

3 Conclusion

In this work, we developed and implemented the two-stage method of fs+US treatment for the fabrication of QD ensembles with a wide-size dispersion from indirect band gap TMDCs. Using transmission electron microscopy and Raman spectroscopy methods for characterization of materials, the chemical composition and topology of nanoparticles and quantum dots obtained from pressed MoS₂ and WS₂ powder targets, as well as MoS₂, MoSe₂ and WSe₂ crystals were investigated. It was found that quantum dots obtained from pressed powder targets have a crystalline structure and planar morphology (2D QDs). At the same time, QDs obtained from crystals can have both a polycrystalline structure and non-planar topology (3D QDs).

As a result, all the obtained quantum emitters demonstrate prominent photoluminescence, and control of the pump wavelength in an ensemble of chromophores with a wide-size dispersion allows us to tune the maximum photoluminescence to the desired spectral range. At the same time, 2D QDs obtained from pressed powder targets exhibit sensitivity to the pump wavelength over a wide spectral range, whereas long-wave photoluminescence of 3D QDs is insensitive to the pump. The observed behavior arises from a critical size threshold for 3D TMDC emitters, beyond which the transition to an indirect band gap quenches photoluminescence. A promising evolution of this methodology is the integration of ultrasonic and laser exposures in a single reactor, enabling simultaneous treatment. This “*in situ*” sono-laser approach could potentially enhance synthesis efficiency and yield through the synergistic interaction of laser ablation and

acoustic cavitation. We believe that the further combination of ultrasonic and laser exposures in one test tube, with the possibility of their parallel use, as well as the use of the entire class of van der Waals materials, can become the cornerstone of the technique for obtaining high-purity two-dimensional emitters with tunable photoluminescence spectra. In general, such bright anisotropic quantum emitters have broad prospects for practical use in lasing, imaging, sensing, and laser theranostics.

Author contributions

Alexei V. Prokhorov: conceptualization, project administration, formal analysis, writing-original draft, writing-review & editing, methodology. Anton S. Chernikov: investigation, formal analysis, writing-original draft. Gleb I. Tselikov: investigation, project administration, writing-original draft, writing-review & editing. Alexander V. Shesterikov: data curation, software. Mikhail Yu. Gubin: conceptualization, formal analysis, visualization, writing-original draft, writing-review & editing. Ivan S. Kazantsev: investigation. Alexander V. Syuy: investigation. Ilya A. Zavidovskiy: investigation. Elena S. Zhukova: investigation. Anton A. Popov: investigation. Kirill S. Khorkov: investigation, funding acquisition, validation. Dmitry A. Kochuev: investigation. Aleksey V. Arsenin: conceptualization, project administration, funding acquisition, validation, resources. Valentyn S. Volkov: conceptualization, project administration, funding acquisition, validation, writing-original draft, writing-review & editing, supervision.

Conflicts of interest

There are no conflicts of interest to declare.

Data availability

The data supporting this article have been included as part of the supplementary information (SI). Supplementary information: Additional experimental data and description of the used methods and equipment. See DOI: <https://doi.org/10.1039/d5na00885a>.

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