

Cite this: *RSC Med. Chem.*, 2026, 17, 1811

Adamantane-based inhibitors of the influenza A M2 proton channel: structure-based design, biological evaluation, and synthetic approaches

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The influenza A matrix 2 (AM2) protein, a prototype of viroporins, conducts protons along a chain of water molecules and ionizable side chains, including histidine-37. Solid-state NMR (ssNMR) and high-resolution X-ray crystal structures have been obtained for AM2 wild-type (WT) constructs in complex with adamantanamines, as well as for mutant AM2 channels that confer resistance to amantadine and adamantanamines across different pH levels. For the structure of AM2 S31N channels in complex with second-generation adamantane derivatives, consisting of an amantadine analog linked to an aryl group via a methylene bridge, X-ray crystal structures are still unavailable. These complexes have been studied in some detail to date, primarily using solution NMR spectroscopy in micelles or ssNMR in lipid bilayers, providing insights into the inhibition mechanisms of these drugs. These findings, when combined with advances in computational methods, can inform the design and synthesis of adamantane-based blockers targeting WT and mutant AM2 channels. The most popular testing assays were presented. Selected synthetic chemistry routes leading to complex adamantanamines, other saturated polycyclic amines, and second-generation adamantane-based inhibitors were provided. Extensive and long-term research on the druggable M2 channel has provided the scientific society with fundamental tools of structure-based drug design, a synthetic chemistry toolbox, and a library of adamantane-based compounds that can be useful antivirals due to the frequent viral AM2 mutations.

Received 17th November 2025,
Accepted 16th February 2026

DOI: 10.1039/d5md01027f

rsc.li/medchem

Introduction

Purpose of the review article

Previous review articles have been published, including: (a) The structure and function of the influenza A M2 (AM2) proton channel^{1–7} and its role in virus assembly and budding,⁸ (b) The molecular basis of inhibition of AM2 viroporin function by adamantanamines,^{4,6,7,9,10} (c) The structure of adamantanamines and other saturated polycyclic amines that block AM2 wild-type (WT) (with serine at position 31), as well as adamantane-based drugs, that block mutant AM2-mediated proton currents studied through electrophysiology (EP),^{11–17} (d) The synthetic routes for selected adamantanamines,^{18,19} (e) The structures of other compounds that inhibit influenza A virus replication.^{20–22} We reviewed the development of adamantane- and other polycyclic amine-based blockers targeting either AM2 WT or mutant AM2 proteins and their associated viruses based on

key experimental structures of AM2 constructs. Our discussion included the assays used to measure potency and the structure-based drug design methods. Finally, we reported selected synthetic routes for adamantanamines, saturated polycyclic amines, and second-generation adamantane-based derivatives, that consist of an amantadine analog linked to an aryl group via a methylene bridge.

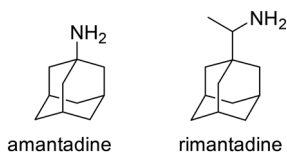
Functional role of influenza AM2 protein

The AM2 proton channel plays a crucial role in the life cycle of the influenza A virus. During the early phase of infection, the virus enters the host cell through endocytosis. The AM2 channel is activated in response to the lowered pH within the endosome, which leads to the transport of protons into the viral interior. This acidification of the interior of the virus triggers the dissociation of the viral RNA from the M1 protein,^{23,24} resulting in virus uncoating. This process, facilitated by a conformational change in hemagglutinin, leads to the fusion of the membranes of the virus and the host cell endosome.^{25,26} During the later stages of virus replication, during virus cellular exit, the AM2 protein maintains the integrity of the hemagglutinin during virus assembly in the *trans*-Golgi network. This function is inhibited by amantadine

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Scheme 1 Chemical structures of amantadine and racemic rimantadine.

or rimantadine (see Scheme 1).²⁵ The AM2 protein helps to maintain the high pH of the *trans*-Golgi network, thereby preventing the premature conformational change of hemagglutinin that occurs at low pH levels and triggers hemagglutinin-induced fusion.²⁷ The AM2 facilitates viral exit by curving the host cell membrane, enabling membrane scission and virus budding.^{28,29}

The AM2 is a 97-residue membrane protein that assembles and functions as a homo-tetramer, allowing it to conduct protons when activated by low pH, a process essential for the viral life cycle.^{30–35} It features a short unstructured N-terminal extracellular ectodomain, made up of residues 1–21,^{36,37} followed by a single transmembrane (TM) domain, made up of residues residues 22–46, known as AM2TM. Finally, it contains a long C-terminal intracellular tail, made up of residues 47–97, which is in contact with the M1 protein.³⁸

The proton-selective ion channel of AM2 is located within the AM2TM domain.³⁹ This domain is the minimal structure required for tetramerization, selective proton transport,^{30,32,33} and amantadine binding.^{40–44} The conductance rate of the AM2TM domain and the blocking of the proton current by amantadine, as measured with EP,^{45,46} correspond to those of the full-length AM2 protein (AM2FL).^{39,47} Near the C-terminal of the TM bundle, there is a region that features an interruption by the H37xxxW41 sequence in the C-terminal half of the TM helix. In this region, polar and single charged residues face the channel, while and hydrophobic residues are oriented towards the lipid bilayer.⁴⁸ In the AM2TM domain, residue H37 plays a key role in activation as the proton-conducting residue of the channel.^{49,50} At low pH in the viral endosome, the imidazole rings in the four H37 residues causes the channel to open due to electrostatic repulsion and destabilization of helix–helix packing. W41 is the residues that controls, the gate blocking, and proton flow when the channel is in the alkaline pH state.^{51,52} Besides the well known gating role of W41, V27 has been suggested to create a secondary gate.^{51,52} The cavity between V27 and H37 contains a binding site for amantadine.⁵³ Water molecules within the channel pore create a hydrogen-bonded water network along the 17 Å stretch between the V27 valve and H37 tetrad.^{54,55} Functional⁵⁶ and structural studies^{54,55} have demonstrated that the pore is formed by the TM-helix bundle and is lined by V27, S31, G34, H37, W41, D44, and R45, *i.e.*, includes all the polar residues of the TM sequence.

The amphipathic helix (AH), consisting of residues 47–62, is positioned between the TM helix and the M1-binding

segment of AM2. The AHs are located at the inner membrane interface and are essential for viral budding.⁸ It is believed that AHs create the membrane curvature necessary for viral budding.^{57,58} The TM and AH bundle comprises the AM2TM + AH, collectively known as the conductance domain (CD) AM2CD. In synthetic lipid bilayers, AM2CD appears to contain all the necessary residues and environmental interactions to achieve the detailed conductance properties of the native protein found in oocytes.³⁹

Experimental structures of AM2 bound with adamantane-based derivatives

Wild-type AM2 protein

The structure of the apo-AM2TM⁴⁸ or apo-AM2CD⁵⁹ was studied using oriented sample (OS) ssNMR in phospholipid bilayers at pH 7, showing a fourfold symmetric homotetramer; the structure with PDB ID 2L0J⁵⁹ was deposited for AM2CD. The charge of H37 tetrad was +2 charged between pH 6–7.5, and the charges may be extensively delocalized over a dimer-of-dimers structure. The kink in the TM helix occurs around G34, like the kinked TM-domain structure in the AM2TM protein, with the presence of amantadine.⁶⁰ At pH 8, the imidazole rings of the four H37 residues are unprotonated. The first protonation of H37 forms an imidazole–imidazolium dimer with a strong hydrogen bond between bridged imidazole rings. The second protonation forms two imidazole–imidazolium dimers.^{48,59} At pH 7.5, most of the H37 side chains existed as imidazole–imidazolium dimer. At pH 5.0, all the H37 are protonated.

The first X-ray crystal structures of samples of apo-AM2TM WT or AM2TM WT with amantadine,⁶¹ showed a tetrameric left-handed bundle, which was consistent with a previous solid-state NMR (ssNMR) structure of the apo-AM2TM (PDB ID 1NYJ⁶²) in phospholipid bilayers. While the apo-AM2TM WT crystal structure had a 2.05 Å resolution and an I33M mutation (PDB ID 3BKD⁶¹), the latter crystal structure (PDB ID 3C9J⁶¹), which was suggested to correspond to a complex with amantadine inside the TM pore, had a 3.5 Å resolution and the stabilizing mutation G34A.^{61,63} However, the 3.5 Å resolution of this structure did not ensure the presence of amantadine (having a diameter of ~3.4 Å). Additionally, the 3.5 Å resolution of PDB ID 3C9J⁶¹ did not allow for examining the critical role of water in drug binding. One of the proposed proton conduction mechanisms is mediated by ordered water molecules arranged in wires inside the channel, which play a role in the conduction of protons through the AM2 pore.^{64,65} In another work, a potential allosteric binding outside the pore was suggested in the complex of rimantadine with AM2CD in micelles (PDB ID 2RLF⁶⁶) using solution NMR in micelles. In this site, the ammonium group of rimantadine in the secondary binding site is in contact with the polar sidechain of D44, whereas the adamantyl group of the drug forms hydrophobic



interactions with I42 from one of the TM helices and with L40 and L43 from the adjacent helix. However, this structure was solved with a 200-fold excess of rimantadine, while the micelles did not allow proper folding, narrowing the tetrameric bundle in the N-end and disturbing the rimantadine entrance to the channel. Nevertheless, in the later ssNMR structure of a sample of AM2TM with a perdeuterated amantadine in phospholipid bilayers (PDB ID 2KQT⁶⁷) obtained using experimental restraints from a previous structure (PDB ID 2H95⁶⁰), amantadine was found inside the pore between V27 and H37, as previously suggested.^{42,52,60} Perdeuterated amantadine⁶⁸ was used in complex with AM2AH in dimyristoyl-*sn*-glycero-3-phosphocholine (DMPC) bilayers, and it was shown with OS ssNMR that amantadine binds close to S31 inside the pore, as was shown in the PDB ID 2KQT structure.⁶⁷ Additionally, it was also shown using ssNMR and CD₃-labeled rimantadine that rimantadine has its amino group towards the C-end.⁶⁹ Thus, the pharmacologically relevant drug-binding site is in the TM pore of the fully functional AM2CD. Other studies with AM2CD and rimantadine using ssNMR in phospholipid bilayers⁷⁰ or solution NMR in micelles (PDB ID 2LJC⁷¹) also showed the presence of rimantadine inside the AM2 pore.

Eventually, the most direct evidence of the pore binding site and water wire role for the stabilization of channel blockers by EP inside the pore came from higher resolution crystallographic studies of the AM2TM in the presence of adamantanamine. Indeed, water-mediated interactions in receptor binding sites can play key roles in drug binding and inhibitor design. Experimental structures with high resolution made the detection of crystallographically important waters possible. The detection of crystallographically important waters demonstrated the channel inhibition by inhibitors targeting these key waters. This was shown in structures of AM2TM WT bound to amantadine (PDB ID 6BKK;⁷² 2.00 Å), rimantadine (PDB ID

6BKL;⁷² 2.00 Å), or a spiro-adamantanamine inhibitor (PDB ID 6BMZ;⁷² 2.63 Å), see Fig. 1. These X-ray structures of AM2TM ion channel with bound inhibitors reveal that the ammonium group of the drugs is directed toward H37 and binds to water-lined sites that are hypothesized to stabilize transient hydronium ions formed in the proton-conduction mechanism. As supported by molecular dynamics (MD) simulations, the ammonium group is stabilized through three hydrogen bonds with proximal waters, creating a layer at the level of the A30 layer, followed by the G34-water layer and the H37 tetrad. The placement of the adamantanamine molecules within the pore requires a displacement of waters near V27 by the hydrophobic adamantane ring of the above adamantanamine molecules, thus disturbing the AM2 water network.⁷² The MD simulations⁷² of the AM2TM WT, with the three adamantanamine drugs predicted with accuracy the position of the ligands and waters inside the pore in the X-ray crystal structure of the AM2TM WT complexes. In another work, the X-ray structures of AM2TM WT pore bound to (*R*)-rimantadine (PDB ID 6US9⁷³) and (*S*)-rimantadine (PDB ID 6US8⁷³) were solved with a 2.00 Å resolution, showing only slight differences in the hydration of each enantiomer.

Mutant AM2 proteins

Amantadine resistance mutations are found close to the drug-binding site located at the pore-facing positions (V27A, A30T, S31N, and G34E), at the interhelical interfaces at the N-terminal half of the channel (L26F), and mutations outside the drug-binding site lying at the interhelical interfaces (L38F, D44). It was suggested⁷⁴ that a mutated, drug-resistant protein has a larger binding pocket for the drug, as shown with MD simulations.⁷⁵ Hence, despite binding to the channel, the drug remains sufficiently mobile so as not to exert a proton-blocking positive electrostatic hindrance as a simple mechanism for proton flow blocking in AM2 WT by amantadine.⁷⁵

A solution NMR structure of AM2 (19–49) S31N in complex with M2J332⁷⁶ in micelles (PDB ID 2LY0⁷⁶) was reported. M2J332⁷⁶ is a representative second-generation adamantane-based inhibitor that blocks M2 S31N-mediated current and replication of M2 S31N virus (Fig. 2); it can be considered a second-generation adamantane-based compound. Such compounds have been developed to inhibit influenza A M2 S31N which became the major epidemic virus after 2008. Indeed, in 2009, the emergence of the novel pandemic H1N1 influenza virus (A/H1N1/pdm09) led to a dramatic spike in hospitalizations and deaths in the United States, particularly among young adults.⁷⁷ These drugs consist of amantadine connected through a methylene linker with aryl groups, e.g., the amantadine-5-(thiophenyl)oxazolyl (M2J332⁷⁶), amantadine-5-phenyloxazolyl (M2J352⁷⁶), or amantadine-5-cyclopropyloxazolyl⁷⁸ conjugates. With the structure PDB ID 2LY0,⁷⁶ it was shown that these amantadine-aryl conjugates blocked AM2AH S31N with an orientation of the amantadine amino and aryl groups towards the N-end, as also

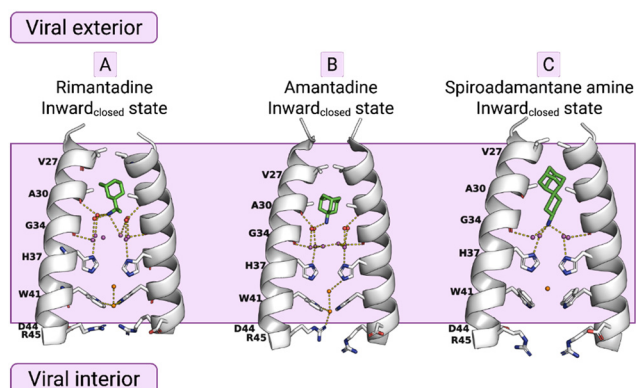


Fig. 1 X-ray crystal structures of the AM2TM proton channel bound to drugs and inhibitors.⁷² Top, left to right: (A) AM2TM bound to rimantadine (PDB ID 6BKL⁷²). (B) AM2TM bound to amantadine (PDB ID 6BKK⁷²). (C) AM2TM bound to a spiramic adamantanamine (PDB ID 6BMZ⁷²). This figure has been adapted from ref. 6 with permission from ELSEVIER, copyright 2025.



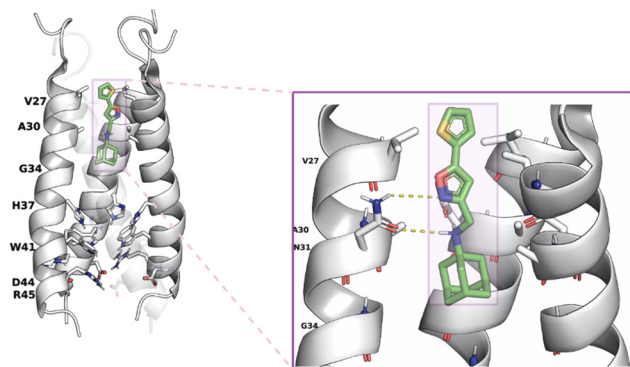


Fig. 2 AM2 (19–49) S31N drug inhibition mechanism.⁷⁶ Left part: binding site of M2WJ332 from the solution NMR structure (PDB ID 2LY0⁷⁶): the side chains V27, A30, H37, W41, and the molecule are shown as colored sticks; other side chains are shown as transparent space filling; the backbone of the protein (one monomer was omitted for clarity) is shown as opaque ribbons. Right part: zoom in showing the drug–protein interactions: one of the N31 side chains forms bidentate interactions with the drug (the carbonyls from another two N31 residues can form a water-mediated hydrogen bond with the ammonium from the drug, which is not shown). This figure has been adapted from ref. 6 with permission from ELSEVIER, copyright 2025.

demonstrated in the ssNMR study of M2J352,⁷⁹ compared to adamantanamines, which point their amino groups towards the C-end of the AM2 WT channel according to the observations from ssNMR⁸⁰ and X-ray structures.^{72,73} M2J332,⁷⁶ and M2J352⁷⁶ were able to block AM2 S31N by EP, and virus replication but did not block AM2 WT by EP and virus replication. Interestingly, the amantadine–3-bromo-thiophenyl conjugate (cmp11 in ref. 81) was a double AM2 S31N and WT blocker by EP. The solution NMR structure of the amantadine–3-bromo-thiophenyl conjugate with AM2 (19–49) WT (PDB ID MUW⁸¹) or AM2 (19–49) S31N (PDB ID 2MUV⁸¹) in micelles showed an orientation of the amantadine amino and aryl groups towards the N-end in AM2 (19–49) S31N and towards the C-end in AM2 WT.⁸¹

X-ray crystallography was used to solve the structures of a spiranic adamantanamine inhibitor bound to AM2TM V27A (PDB ID 6NV1⁸²), shown in Fig. 3. This spiranic adamantanamine inhibitor was a triple blocker by EP of AM2 WT, V27A, and L26F proteins.⁸³ The X-ray structure of the spiranic adamantanamine with AM2TM WT (PDB ID 6BMZ⁷²) is shown in Fig. 1. The X-ray structures with PDB IDs 6BMZ⁷² and 6NV1⁸² revealed the mechanism of amantadine resistance. As shown in the crystal structure, compared to the AM2TM WT (Fig. 3B), the AM2TM V27A channel pore was wider at its N-terminus because of the V27A mutation, which removed V27 side chain hydrophobic interactions that are important for binding of amantadine and rimantadine (Fig. 3A). The spiranic adamantanamine inhibitor, which is longer than amantadine, shifted its binding site in the pore depending on which residue is present at position 27, thus blocking effective proton conductance in the AM2 WT and V27A mutant channels. The MD simulations of the AM2TM V27A–spiranic adamantanamine complex agreed with the

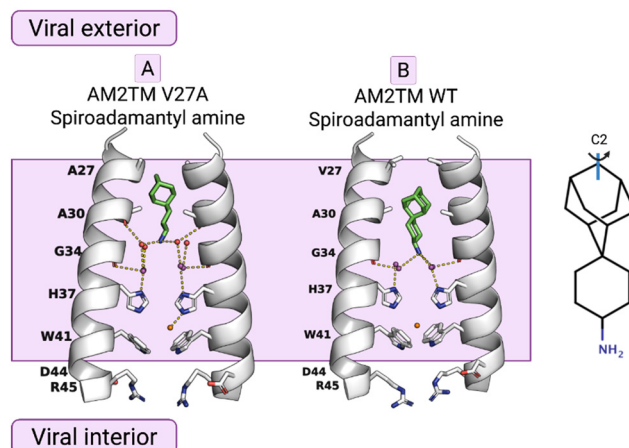


Fig. 3 The mechanism of amantadine drug resistance in the AM2 V27A mutant channel and the mechanism of dual inhibition by the shown spiranic adamantanamine can be rationalized through the crystal structures of this adamantanamine with (A) AM2TM WT (PDB ID 6BMZ⁷²) and (B) AM2TM V27A (PDB ID 6NV1⁸²). This figure has been adapted from ref. 6 with permission from ELSEVIER, copyright 2025.

experimental structures.⁸² The same argument can be applied to explain the EP-based blocking of the wide-diameter AM2TM L26F mutant by this spiranic adamantanamine.

Resistance of influenza A virus to adamantane-based drugs

The most common AM2 inhibitors are amantadine derivatives. In 1966, amantadine was first approved as a prophylactic drug against Asian influenza.^{84,85} After a few years, it was announced⁸⁶ that dyskinesia symptoms of a Parkinson's disease patient were remarkably improved while amantadine was administered for the prevention of flu. The mechanism of action of amantadine against these targets results from the blockage of ion channels either by direct interaction with the channel's pore (influenza AM2) or by stabilization of closed states of the channel (NMDA receptor).⁸⁷ Since 1985, AM2 has been identified as the primary target of amantadine.⁴⁰ Resistance to AM2 WT proton channel drugs, amantadine and rimantadine, is associated with mutations in the TM domain of the AM2 WT protein.¹⁰³ The homo-tetrameric structure of the AM2 WT channel places constraints on the types of drug-resistant mutations that can be accommodated.^{74,88} A few amantadine- or rimantadine-resistant mutations have been observed in transmissible viruses,^{89–91} which are L26F, V27A, A30T, G34E, and S31N, located in the TM pore of AM2 WT,^{89–94} although other mutations can easily be observed *in vitro*. The mutations that cause the greatest decrease in inhibition are S31N and V27A, which increase the polarity of the pore. The vast majority, 95% of resistant viruses bear the S31N substitution in AM2, 1% have V27A, and L26F, A30T, and G34E are rare.^{95,96} The AM2 S31N mutant is one of the most conserved in viral proteins among currently circulating influenza A viruses, which happens to maintain nearly



identical channel function as the AM2 WT but is resistant to amantadine, rimantadine, and adamantanamines. The presence of L26F, V27A, and particularly S31N in influenza A viruses circulating worldwide pushes the search for novel ion channel blockers with stronger, preferably resistance-overcoming activity.

Recently, the World Health Organization reported that influenza activity has globally increased since October 2025, with influenza A virus being predominant among the viruses (WHO, 10 December 2025, Disease Outbreak News; Seasonal influenza-Global situation). Currently circulating in humans are subtype A(H1N1) and A(H3N2) influenza viruses. Both subtypes include the S31N mutation in the M2 sequence. A recent work,⁹⁷ showed that prolonged treatment with the neuraminidase (NA) inhibitors like the commonly used for influenza treatment, oseltamivir, can cause resistant mutations at hemagglutinin (HA) alone or in combination with resistant mutants at NA as a compensatory mechanism. In that way, the virus reduces receptor binding and restores functional balance between HA and NA. It was shown⁹⁷ that these mutations can also confer cross-resistance to other NA inhibitors, including zanamivir and peramivir, in the A(H1N1)pdm09 virus. Considering (a) the prevalence of M2 S31N strains and (b) the ability of the virus to develop mutations in both HA and NA and become resistant to the currently used NA inhibitors, the design of new second-generation M2 inhibitors is becoming increasingly relevant. In another work,⁹⁸ the second-generation drug resistance mechanism of the virus towards second-generation M2 channel blockers was profiled *in vitro*. Serial viral passage experiments were performed⁹⁸ in the A/California/07/2009 (H1N1) viral strain under persistent drug selection pressure to identify resistant mutants against such blockers. The results suggested⁹⁸ that these S31N inhibitors have a higher genetic barrier to drug resistance than amantadine in cell culture. It was shown that the evolved mutations are all located at the N-terminal drug binding site of AM2 S31N, and they have either a direct or indirect effect on the channel

pore, which leads to either reduced or complete loss of drug sensitivity. More specifically, those mutations introduced either a structural hindrance for the new agents as their aryl groups lie in this region or change the polarity of that site of the tetramer, affecting their entrance to the pore.^{99,100} This information is quite important for optimizing the structure of new selective S31N inhibitors.

Methods to detect inhibition of wild-type and mutant AM2 viruses by adamantane-based drugs

Cell antiviral assays

Compounds have been tested as inhibitors of viral replication using cell assays. These are most often the miniplaque assay,^{101–104} and the cytopathic effect (CPE) inhibition assay,¹⁰⁵ performed in Madin-Darby canine kidney (MDCK) cells against influenza virus strains. Mutant AM2 viruses can be produced using reverse genetics.¹⁰⁶ For example, using a miniplaque assay, the (*R*)- and (*S*)-enantiomers of rimantadine showed indistinguishable potency against the amantadine-sensitive A/Solomon Island/3/2006 (H1N1) strain. Both (*R*)- and (*S*)-rimantadine inhibit viral replication with similar EC₅₀ values of 19.62 and 24.44 nM, respectively (Fig. 4).⁷³

An *Escherichia coli*-based assay was used to measure the H⁺ conductivity of AM2 WT¹⁰⁷ or mutants M2^{108,109} in the presence of a potential inhibitor, since bacteria that constitutively express a pH-sensitive green fluorescent protein—pHluorin, can be used to analyze the membrane permeation to H⁺. Specifically, the emission at 520 nm of pHluorin has two excitation maxima: 390 nm and 466 nm, whose ratio changes as a function of pH. The fluorescence obtained when exciting at 390 nm and 466 nm upon injection of a concentrated acid solution into the buffer was monitored. Any bacteria that express an H⁺-conducting channel will exhibit a dramatic change in the fluorescence

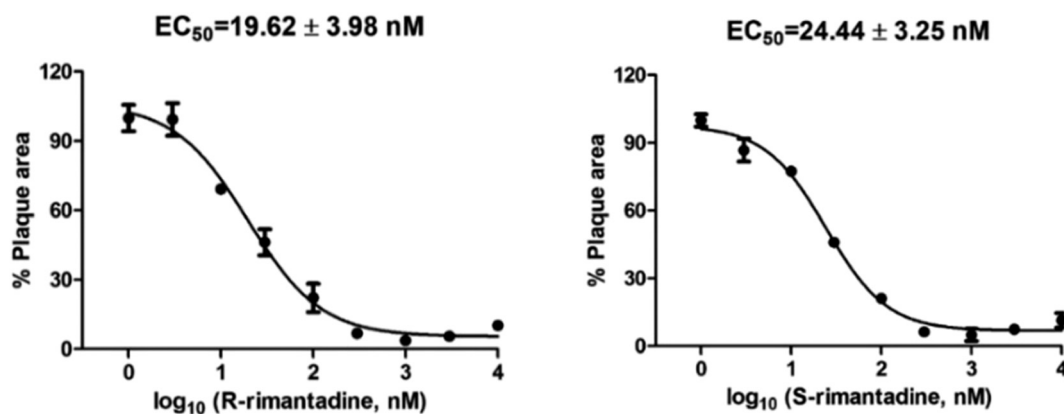


Fig. 4 Cellular antiviral assay results of (*R*)- and (*S*)-rimantadine against the amantadine-sensitive A/Solomon Island/3/2006 (H1N1) strain. The antiviral potency was determined in a plaque assay. The EC₅₀ values are the mean ± standard deviation of two independent repeats. This figure has been reproduced from ref. 73 with permission from the American Chemical Society, copyright 2021.



ratio, in contrast to control bacteria that contain an empty vector or when an inhibitor is present.

Assays that measure direct binding

Experiments that measure the thermodynamics of binding and proton flux blocking experiments. Assays using various biophysical methods have been applied to test the binding of drug molecules to the AM2 channel at pH 7.5–8 in dodecyl phosphocholine (DPC) micelles or phospholipid bilayers, where tetramers mainly exist in an AM2 monomer/lipid ratio of $\leq 1:25$. The AM2 construct system is titrated using a drug dissolved in the same lipid environment. The developed methods mostly explore the thermodynamics of direct ligand binding to AM2 binding. For example, they have applied circular dichroism (CD) spectroscopy, which monitors the ellipticity ratio at 223 vs. 209 nm ($\theta_{223}/\theta_{209}$),¹¹⁰ or analytic ultracentrifugation (AUC),¹¹⁰ or isothermal titration calorimetry (ITC) to AM2TM peptides in DPC micelles (Fig. 5).^{39,111}

An assay¹¹² was developed that uses the quenching of W41 fluorescence by H37 protonation below pH 6 in the AM2FL channel reconstituted in LDAO (*N,N*-dimethyldodecylamine-*N*-oxide) detergent. The quenching of W41 fluorescence was reversed by amantadine and analogs, and was used in testing adamantanamines against AM2 WT.¹¹³ Another assay measured the thiol-disulfide equilibria of AM2TM or AM2FL in phospholipid bilayers by CD¹¹⁴ or surface plasmon resonance (SPR).¹¹⁵ A series of SPR experiments was used to accurately measure the affinity of amantadine and rimantadine to AM2TM WT embedded in DMPC liposomes. It was found that this class of drugs can bind to AM2 with two different binding dissociation constants (K_d) in the order of 10^{-4} ($K_d = 0.91 \mu\text{M}$) and 10^{-7} M ($K_d = 0.40 \text{mM}$).¹¹⁵ It was shown that a high-affinity binding site corresponds to the AM2 ion channel pore-binding site, which is responsible for the pharmacological activity elicited by amantadine drug and its analogs, in agreement with

the ssNMR study of amantadine bound to AM2TM WT (PDB ID 2KQT⁶⁷). It was also found that the low-affinity site corresponds to the secondary binding site, which can be attributed to the lipid face of the pore, in agreement with the solution NMR structures of AM2CD WT in micelles (PDB IDs 2RLF⁶⁶ and 2LJC⁷¹).

In another assay, the proton flux mediated by AM2TM or AM2CD was measured in small liposomes composed of unilamellar 2:1 1-palmitoyl-2-oleoyl-*sn*-glycero-3-phosphocholine (POPC)/1-palmitoyl-2-oleoyl-*sn*-glycero-3-phosphoglycerate (POPG) vesicles.³⁹ An *in vitro* functional assay, which is an indirect liposome dye release assay for viroporin activity, was also reported.¹⁰³ However, the most straightforward way to determine the blocking effect of M2-mediated proton current is electrophysiology with AM2FL expressed in oocytes, published early for amantadine¹¹⁶ and further performed for amantadine analogs.¹¹⁷ The compounds are tested with a two-electrode voltage-clamp (TEVC) assay using *X. laevis* frog oocytes microinjected with RNA expressing the AM2 protein.^{118–120} Numerous studies have been performed.^{62,80,100,101,121–138} It was shown that drugs can block AM2-mediated proton current when the $k_{\text{off}}/k_{\text{on}}$ are, correspondingly, slow off/fast on or slow off/slow on, thus leading to favorable $K_d = k_{\text{off}}/k_{\text{on}}$ values.^{101,105,106,139} Only when K_d (TEVC) = $k_{\text{off}}/k_{\text{on}}$ is small enough, an *in vitro* antiviral activity is observed.^{101,106} If the k_{off} is too high, then while the compound can block the M2 channel, it does not inhibit the virus *in vitro*, and this was the case for compound 8 (see Table 2) against AM2 V27A.¹⁰¹ The potency of the inhibitors was usually expressed as the inhibition percentage of the AM2 current observed after 2 min. However, studies performed independently with second-generation amantadine-based drugs against AM2 S31N,^{106,139} rimantadine analogs against AM2 WT,¹⁰⁵ or spiranic adamantanamines against V27A¹⁰¹ showed that the 2-minute measurements of the %AM2 blocking effect (usually applied for testing a large number of compounds^{76,81,139–141}) and filtering the most important for

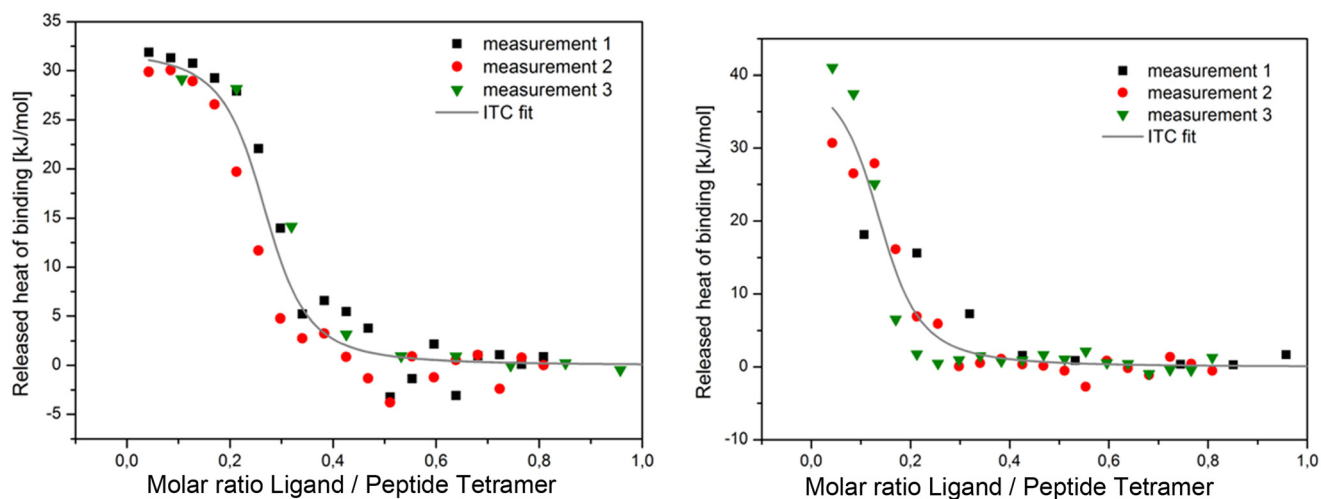


Fig. 5 Representative ITC plots for enantiomers 2-R, 2-S.



Table 1 Summary of the binding affinity of amantadine and rimantadine enantiomers against Udm M2 WT⁷³

	<i>rac</i> -Rimantadine	(<i>R</i>)-Rimantadine	(<i>S</i>)-Rimantadine	Amantadine
Concentration tested	50 μM	50 μM	50 μM	100 μM
k_{on} ($\text{min}^{-1} \text{M}^{-1}$)	$19\,600 \pm 300$	$20\,800 \pm 700$	$22\,500 \pm 300$	$20\,500 \pm 300$
k_{off} (min^{-1})	$(9.1 \pm 0.8) \times 10^{-4}$	$(9 \pm 2) \times 10^{-4}$	$(8.8 \pm 0.8) \times 10^{-4}$	$(119 \pm 2) \times 10^{-4}$
$K_{\text{d}} = k_{\text{off}}/k_{\text{on}}$ (nM)	46 ± 4	41 ± 9	39 ± 4	580 ± 20

subsequent antiviral assays were not efficient. This is due to the presence of compounds with, *e.g.*, very slow on- and slow off-rates for entry (measured as association and dissociation

rate constants, k_{on} and k_{off} , respectively), for which characterization of the blocking effect needs more than 2 min, *e.g.*, 5 min. For example, in ref. 105 for a rimantadine analog,

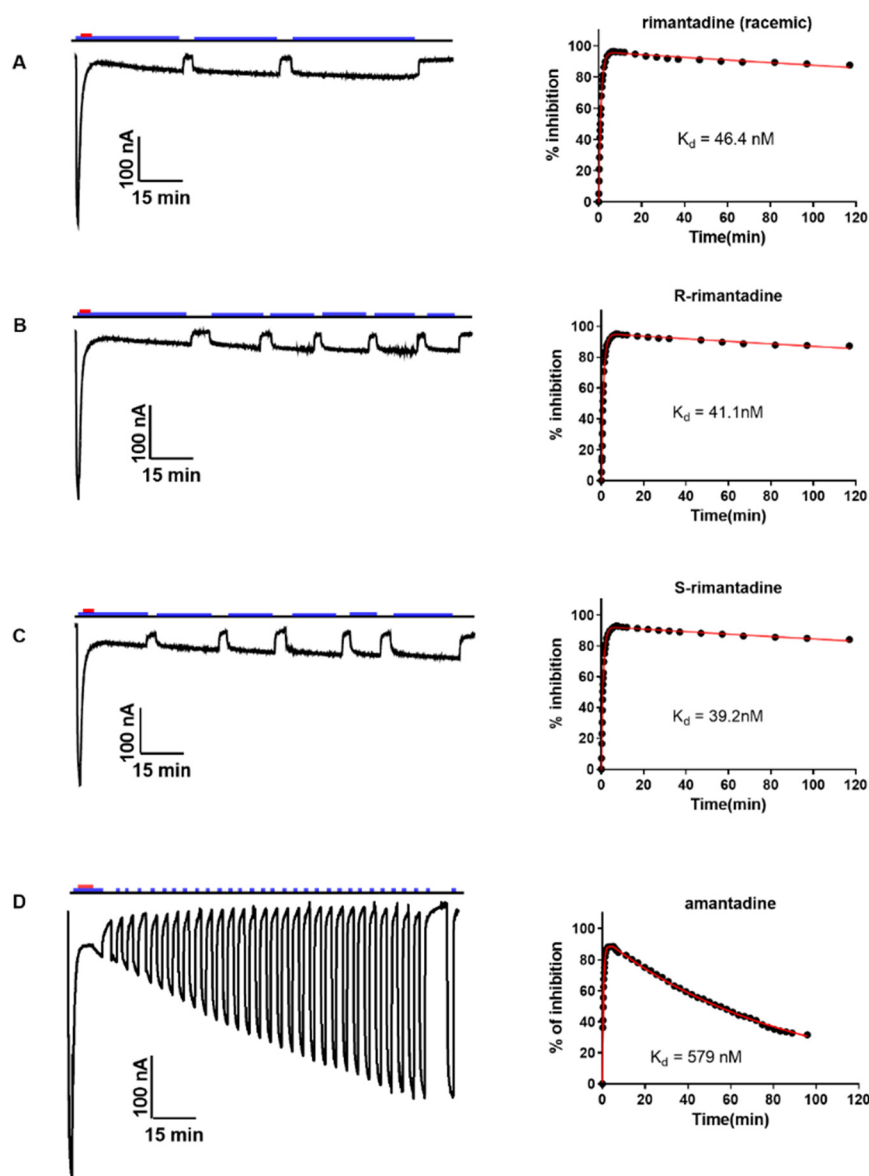


Fig. 6 Rimantadine enantiomers and amantadine binding kinetics against Udm M2 WT were determined using a combined application of inward current and washout procedure TEVC assay.⁷³ (A) Racemic rimantadine, (B) (*R*)-rimantadine, (C) (*S*)-rimantadine, or (D) amantadine was applied to oocytes for 5 to 7 min, and after the inward current reached its maximum, a washout protocol was applied to the oocytes. During the washout, pH 8.5, pulses were applied to make sure the current went to baseline to ensure the oocyte quality. The blue bar above the recording trace indicates the period in which the pH 5.5 Barth solution was applied; the red bar indicates the period in which compounds in the pH 5.5 Barth solution were applied. Representative recording traces are shown on the left side of each panel. The best-fit values are shown in Table 1. This figure has been reproduced from ref. 73 with permission from the American Chemical Society, copyright 2021.



only 27% blocking was observed at 2 min against Udorn AM2 WT, but low micromolar EC_{50} against M2 WT viruses. Despite the very slow binding k_{on} rate, the ligand exhibited high antiviral potency, which requires, as a biological effect, much longer exposure times than EP experiments. For such compounds, after 2 min or 5 min, the percentage of proton current inhibition was progressively increased. Thus, in ref. 105 it was striking to observe that the ligand against Udorn AM2 WT exhibited 27% blocking at 2 min, 38% at 5 min, and 61% at 10 min. It must be clarified that the measurements at 2, 5, or 10 min correspond to measurements before the establishment of equilibrium due to very slow on- and off-rates for entry, which characterizes bulky ligands, and the period can't be extended considerably due to the difficulty of maintaining oocytes at low pH for extended periods. Hence, the %-blocking/ IC_{50} values determined by the TEVC procedure can be significantly lower than those measured with *in vitro* antiviral assays, where the system can reach the equilibrium block state, or compared with K_d values at equilibrium measured with ITC¹⁰⁵ or SPR.¹¹⁵ ssNMR studies have also been performed to investigate the binding interactions of rimantadine and of amantadine and other adamantanamines with AM2 (22–62) WT.^{83,142–146}

An interesting work refers to cell-based antiviral assays and EP, which were applied to evaluate the biological potency of rimantadine enantiomers.^{105,147} It was observed that both enantiomers have similar %-channel blockage, binding k_{on} and k_{off} , K_d , and antiviral potency (EC_{50}) against AM2 WT.¹⁰⁵ These results showed no significant difference between the

two rimantadine enantiomers compared to what has been previously suggested with ssNMR at -10 °C, *i.e.*, that the two enantiomers have different affinity against AM2CD WT.¹⁴⁸ In a subsequent work,⁷³ much more accurate binding kinetics were performed compared to previous ones¹⁰⁵ measurements (Table 1, Fig. 6).

It was concluded that the slight differences in hydration for the (*R*)- and (*S*)-rimantadine enantiomers observed in structures of M2TM WT with PDB ID 6US9⁷³ and PDB ID 6US8⁷³ are not relevant to drug binding or channel inhibition. Additionally, it did not result in a difference in potency or binding kinetics as was shown by similar values of k_{on}/k_{off} , and K_d for (*R*)- and (*S*)-rimantadine in the TEVC assay (Table 1, Fig. 6).^{139,149}

Solid-state NMR experiments. Solid-state NMR was applied^{144,150} in AM2CD WT (residues 18–60), in the apo-form or in complex with rimantadine (drug:tetramer ratio ~ 3 : 1 mol), in DPhPC lipid bilayers at pH 7.8. In this alkaline pH, the His37 tetrad has a neutral charge. Proton resonances at 14.3, 11.7 ppm for the His37 N δ 1, N ϵ 2 of apo-AM2CD WT were recorded in the (H)NH spectra due to His-His⁺ dimer-of-dimers formation (N δ 1–H \cdots N ϵ 2 and N δ 1 \cdots H–N ϵ 2 forms) and showed an intermolecular coupling ($J_{NN} = 8.9 \pm 0.3$ Hz) due to hydrogen bonding between N ϵ and N δ of adjacent His37.^{144,150} When rimantadine was added to AM2CD WT in DPhPC bilayers or in VM bilayers or in DPhPC-30% cholesterol bilayers at 4 °C, the spectra did not show pore binding at all, as observed previously with AM2CD WT (residues 21–61) reconstituted in VM at 30 °C using a drug:

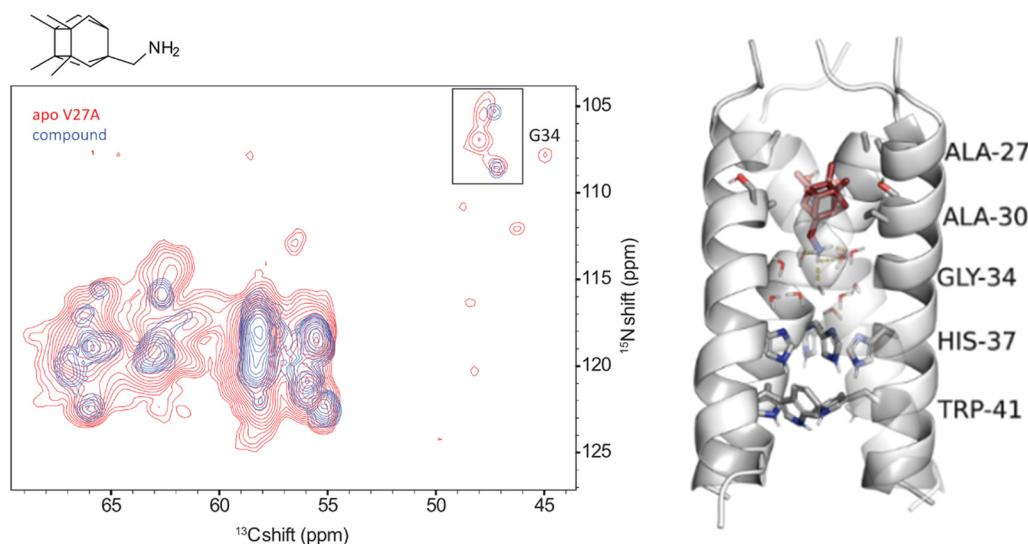


Fig. 7 Slices from 3D (H)C α NH spectra (^{15}N - $^{13}\text{C}\alpha$ projection) of AM2CD V27A and mutant proteins in complex with the cage amine triple blocker (protein was embedded in DPhPC bilayers with a lipid to protein tetramer molar ratio of 24 : 1 at pH 7.8) and results from 500 ns MD simulations for AM2TM-cage amine complexes embedded in POPC bilayers utilizing the CHARMM36m force field.¹⁵⁴ Panel on the left side: the NMR signals of the apo-protein are shown in red, and the signals of the complexes of AM2CD with the respective compound are shown in blue (at the top left side, in the inset, is shown clearly the pair of signals of G34, which suggests the presence of two conformational states: either due to a slow equilibrium between a kink and non-kink state at G34 or a dimer-of-dimers structure of M2CD). Panel on the right side: average structure of AM2TM V27A from MD simulations; grey cartoon is used to display the AM2TM tetramer, stick representations were used for ligands, waters, and amino acid side chains indicated and cartoon representations were used for the protein backbone (carbons are depicted with grey color for amino acids and ruby color for the ligand, red for oxygen, blue for nitrogen) while the yellow dotted line represents hydrogen bonding between the drug and water molecules. This figure has been adapted from ref. 6 with permission from ELSEVIER, copyright 2025.



tetramer ratio 5:1.⁸³ In the presence of rimantadine with AM2CD WT pore at 40 °C, compared to the apo-AM2CD WT signals, was observed intensity reduction of the side chain His37 proton peaks at 14.3 and 11.7 ppm in the (H)NH spectra and the His37-Ne2 shifted upfield, while disruption of the J_{NN} hydrogen bond between His37 Ne and Nδ was observed, in accordance with its pore-blocking mode of AM2CD WT inhibition. While binding took place over three days at 40 °C, complete binding was seen in just a few hours at 55 °C, offering a novel NMR assay for the detection of inhibition by adamantanamines. The slow kinetics reported for rimantadine by NMR are opposed to the fast blocking of AM2CD WT (residues 18–60)-induced proton currents by either rimantadine or amantadine observed in liposomal proton flux assays,^{116,151} respectively. At pH 6, however, *i.e.*, around the pH found in the Golgi lumen (pH 6.0–6.7),¹⁵² where the AM2 channel adopts its activated +3 charge state, the rimantadine pore-binding kinetics were faster.

3D (H)CαNH NMR was applied to study the potencies against influenza A with the M2 WT channel of several adamantanamines¹⁴² and some cage amine analogs, and the AM2 L26F, V27A, A30T, G34E mutants. For example, the compound shown in Fig. 7 exhibited a blockage of AM2 WT, L26F, and V27A channels by EP assays, while some adamantanamine analogs blocked AM2 WT and AM2 L26F channels by EP (see structures in Table 2).¹⁵³

The cage amine triple blocker in Fig. 7,¹⁵³ while having length equal to rimantadine, binds and blocks the AM2 V27A channel due to its larger volume, as well as AM2 L26F and AM2 WT, as was shown by 500 ns-MD simulations with the CHARMM36m force field¹⁵⁴ of AM2CD V27A, AM2CD L26F, and AM2CD WT embedded in a 150 POPC lipid bilayer. This compound's ammonium group was oriented steadily towards the C-terminus in AM2CD WT, V27A, and L26F channels during the MD simulations, and the ligand was stabilized within the AM2 pore by forming an average of three hydrogen bonds with the water that was present between the ligand's ammonium group and His37. Furthermore, by displacing loosely bound waters close to the top of the pore (N-terminal side), the large cage alkyl group stabilized the complex through interactions between the tetramethylcyclobutane ring and residue 27 side chains,^{44,155} and successfully blocked the passage of water molecules, preventing protons from entering the pore and generating conduction. The MAS ssNMR provided information on the contacts of the triple blocker in Fig. 7 with AM2CD V27A (residues 18–60), AM2CD L26F, and AM2CD WT. When the drug was added to the mutant channels, there was a noticeable increase in peak resolution as opposed to the relatively wide peaks in the apo-AM2 V27A and apo-AM2 L26F. This suggests that the protein's conformational flexibility was decreased during the formation of the drug-protein complex. Fig. 7 shows some representative results for the AM2CD V27A complex.

Selected reports of adamantanamines and assays used to measure AM2 inhibition

In Table 2, selected compounds and activities against AM2 WT and its AM2 mutant channels were measured with EP, and the *in vitro* cell-based activities against the corresponding viruses; the activities of amantadine (1) and rimantadine (2), BL-1743 (3),^{111,117,156} pinanamine (4),¹⁵⁷ and noramantadine 5, which all block the AM2 WT channel by EP and inhibit *in vitro* the replication of the influenza AM2 WT virus.

They also showed the activities of the tetramethyl-3-azapentacyclo[7.2.1.1^{5,8}.0^{1,5}.0^{7,10}] tridecane (6),¹⁵⁸ which blocks both the AM2 WT- and AM2 V27A-proton mediated current and inhibits *in vitro* AM2 WT, V27A, and L26F viruses. The piperidine derivative 8 blocks the AM2 WT and V27A by EP and inhibits *in vitro* the corresponding viruses.¹⁰¹ The azatetracyclo[5.2.1.1^{5,8}.0^{1,5}]undecane (7)¹²⁵ blocks AM2 V27A by EP, while the spiranic amine 9¹⁵⁹ and tetramethyl-3-azapentacyclo[7.2.1.1^{5,8}.0^{1,5}.0^{7,10}]tridecane (6)¹⁵⁸ are blockers of the AM2 WT, V27A, L26F channels and inhibited *in vitro* the corresponding viruses.

The rimantadine analog 10, the spiro pyrrolidine[2,2']adamantane (11), and the diamantanamine 12 block the AM2 WT and AM2 L26F channels by EP.¹⁴² The adamantanamines and saturated polycyclic amines 1–12 did not block the M2 S31N channel.

These adamantanamines are important agents since they can block AM2 WT, V27A, and L26F viruses. Other lipophilic and saturated polycyclic amines that block these AM2 channels have been reported.^{138,160–164} These compounds can be repurposed if the predominant influenza A M2 S31N, causing the current epidemics, can be changed to one of these channels that are sensitive to adamantanamines.

Structure-based drug design methods targeting wild-type and mutant AM2 channels

Protein channels and water-filled pores present particularly challenging targets for drug design.¹⁶⁸ MD simulations for adamantanamine binding to the AM2 protein have been performed, and some of them were applied for structure-based drug design purposes. For example, MD simulations of ligand-AM2TM complexes were used to design and synthesize spiro-piperidine inhibitors (*e.g.*, compound 9 in Table 2) that block the M2 WT, M2 V27A, and M2 L26F channels.¹⁵⁹ The design of AM2 blockers would benefit from the application of methods and the generation of models that can accurately calculate the relative binding energies of ligands targeting the AM2 channel.

Generally, the molecular mechanics-Poisson-Boltzmann surface area (MM-PBSA) method cannot calculate accurate binding free energies and can only provide a good correlation between calculated and experimental binding free energies for a set of ligands with a K_d value range of 1000.^{169,170} In contrast, binding free energy methods based on statistical



Table 2 Chemical structures of representative adamantanamines or saturated polycyclic amines 1–12 and *in vitro* potencies (from whole cell or TEVC assays) against viruses with AM2 WT or the amantadine-resistant AM2 V27A, AM2 L26F, and AM2 S31N mutant channels

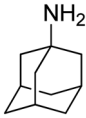
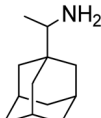
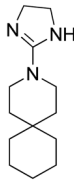
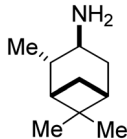
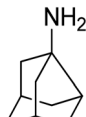
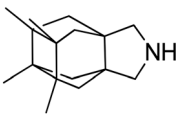
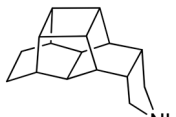
		Compound			
					
		1	2	3 BL-1743	4
Virus					
A/M2 WT (S31)	EC ₅₀ (or IC ₅₀ in TEVC)	0.78, ^{a,105} 0.48, μM ^{b,105} , 5.96, ^{e,165} 12.5 μM, ^{e,105} 8.8 μM ^{f,165}	0.48, ^{a,105} 0.04 μM ^{b,105} 10.8 μM, ^{e,105}	2 μM ^{a,111,117,156}	1.36 μM ^{e,165} , 6.8 μM ^{f,165}
A/M2 V27A		n.a.	n.a.	n.a.	n.a.
A/M2 S31N		n.a.	n.a.	n.a.	n.a.
A/M2 L26F		n.a.	n.a.	n.a.	n.a.
					

Table 3 Measurement of the potency of rimantadine enantiomers **2** and analogs **10**, **13**, **14**, which include progressively larger alkyl groups. ITC measured the binding affinity against AM2TM, the proton blockage against the AM2FL WT and S31N protein channels by EP, and cell-based antiviral potencies (CPE assay)¹³⁴

Complex	ITC	TEVC	CPE
	K_d (μM)	k_{on} , k_{off} ($\text{M}^{-1} \text{s}^{-1}$), K_d (μM)	EC_{50} (μM)
2-R + M2 WT	0.32	412, 0.0013, 3.2	0.05
2-S + M2 WT	0.34	407, 0.0016, 3.9	0.06
10 + M2 WT	0.13	230, 0.003, 13	0.01
13 + M2 WT	4.59	—	0.46
14 + M2 WT	3.43	34, 0.003, 13 μM	1.07

amantadine (**1**) with $K_d = 2.7 \mu\text{M}$ by ITC and rimantadine with $K_d \sim 0.33 \mu\text{M}$ by ITC, the dimethyl analog **10** had enhanced affinity against AM2TM WT (Udorn sequence) by ITC ($K_d = 0.13 \mu\text{M}$), and against AM2FL according to the kinetic binding measurements by EP. This was also confirmed by alchemical relative binding free energy calculations with the FEP/MD method, with OPLS2005 provided with the Desmond software. It was found that the diethyl and dipropyl rimantadine analogs showed reduced affinity and potency (Table 3).¹³⁴

Similarly, the antiviral potency (EC_{50}) evaluation showed the same ranking. In ref. 137 was also shown that the

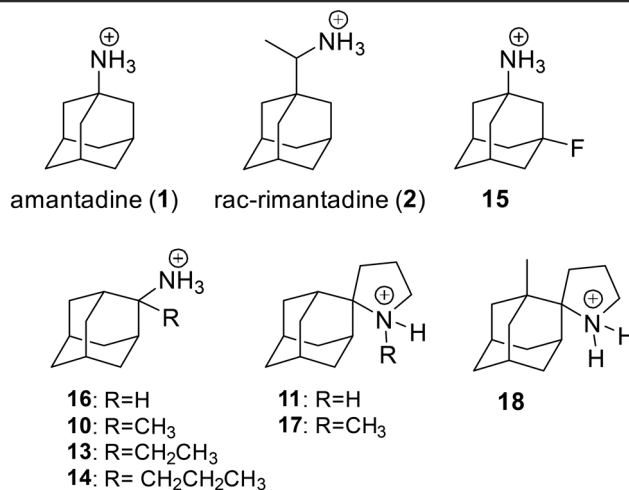
calculated relative binding free energies with FEP/MD calculations was $\Delta\Delta G_{\text{FEP/MD}} (3 \rightarrow 2\text{-R}) = 0.62 \pm 0.14 \text{ kcal mol}^{-1}$ with $\Delta\Delta G_{\text{ITC}} (3 \rightarrow 2\text{-R}) = 0.33 \pm 0.50 \text{ kcal mol}^{-1}$, while $\Delta\Delta G_{\text{FEP/MD}} (3 \rightarrow 2\text{-S}) = 0.68 \pm 0.15 \text{ kcal mol}^{-1}$ with $\Delta\Delta G_{\text{ITC}} (3 \rightarrow 2\text{-R}) = 0.42 \pm 0.48 \text{ kcal mol}^{-1}$. These FEP/MD calculation results supported that rimantadine enantiomers, although chiral molecules, have equal binding affinities against a chiral receptor, *i.e.*, the AM2TM WT protein. The MD simulations in ref. 134 showed that in diethyl and propyl analogs (**13** and **14**, respectively, see Table 3), the alkyl groups appear to better occupy the space between the ligand and the pore walls. For these compounds' binding, the ITC data indicated that the constrained motion and the resulting entropy cost of binding are important quantities and reduced the binding affinities of **13** and **14** compared to **10** (see Table 1 in ref. 134). The proton blocking activity of the adamantanes is consistent with the fact that no water molecules were detected in the area above the adamantane core (*i.e.*, toward the N-terminus) in the MD simulations.

A few adamantanes (**1**, 2-R, 2-S, 3-**10**; Table 4) in complex with AM2TM WT were studied at acidic pH¹⁷⁴ and alkaline pH¹⁷⁵ with alchemical calculations using the FEP/MD method with OPLS2005 force field provided with the Desmond software, and relative free energies were calculated using the Bennett acceptance ratio (BAR).^{171,172} The K_d values

Table 4 Binding constant, free energy, enthalpy, and entropy of binding derived from ITC measurements for AM2TM WT

Ligand	K_d	ΔG	ΔH	$T\Delta S$
1	2.17 ± 0.52	-7.77 ± 0.14	-6.66 ± 0.50	1.11 ± 0.52
2^a	0.51 ± 0.26	-8.64 ± 0.30	-7.60 ± 0.28	1.04 ± 0.41
2-R	0.34 ± 0.12	-8.88 ± 0.21	-7.73 ± 0.28	1.15 ± 0.35
2-S	0.32 ± 0.16	-8.97 ± 0.26	-7.54 ± 0.34	1.42 ± 0.43
15	6.33 ± 1.53	-7.14 ± 0.14	-3.60 ± 0.31	3.54 ± 0.34
16	1.60 ± 0.34	-7.96 ± 0.13	-7.03 ± 0.42	0.93 ± 0.44
10	0.89 ± 0.19	-8.31 ± 0.13	-6.79 ± 0.26	1.51 ± 0.29
13	0.62 ± 0.14	-8.52 ± 0.13	-7.14 ± 0.21	1.38 ± 0.25
14	0.63 ± 0.17	-8.53 ± 0.16	-7.62 ± 0.30	0.91 ± 0.34
11	0.36 ± 0.22	-8.90 ± 0.43	-5.02 ± 0.41	3.88 ± 0.60
17^b	0.93 ± 0.36	-8.28 ± 0.23	-3.82 ± 0.28	4.46 ± 0.36
18^a	1.30 ± 0.43	-8.08 ± 0.20	-5.08 ± 0.31	3.00 ± 0.37

^a Racemic mixture. ^b Racemic mixtures resulted from the protonation of *N*-methyl spiro[pyrrolidine-2,2'-adamantane].



were used as experimental probes for the FEP/MD calculations of alchemical relative binding free energies.

While the set of 12 compounds in Table 4 had a very narrow range of binding free energies of 8 kJ mol^{-1} ($1.9 \text{ kcal mol}^{-1}$), *i.e.*, K_d values for the close state of AM2TM WT (Udorn strain) at pH 8 that differed by a factor of 20 (Table 4), the FEP/MD method performed with high correlation found ($r = 0.94$, $p < 0.001$, $PI = 0.74$) between calculated and experimental binding free energies.¹⁷⁵ The correlation was also good in FEP/MD simulations of AM2TM at acidic pH using experimental K_d values measured at acidic pH against the open conformation of AM2FL.¹¹⁹ As mentioned previously, the experimental assay was based on inhibition by adamantanamines of the quenching of W41 fluorescence by H37 protonation below pH 6 in AM2FL in a detergent environment.¹¹⁸

Subsequently, the experimental binding free energies computed based on the K_d values measured with ITC were measured against the closed form of AM2TM tetramer in DPC micelles for a larger set of 27 adamantanamines, and the antiviral potencies (IC_{50}) with whole cell assays were also measured.¹⁶⁹ The range of the experimental K_d values was ~ 44 , and the range of the antiviral potencies (IC_{50}) values was ~ 750 . A good correlation ($r = 0.76$) was found between their corresponding binding free energy, computed using K_d or IC_{50} values. MD simulations with Amber19sb force field (ff19sb)¹⁷⁶ or CHARMM36m force field¹⁵⁴ and different experimental starting structures of AM2TM were used to investigate the binding mode of adamantanamines that bind strongly, moderately, or tightly to AM2TM embedded in DMPC, DPPC (dipalmitoyl-*sn*-glycero-3-phosphocholine), POPC, or a virus (VM) membrane (Fig. 8).

FEP/MD NPT simulations with the OPLS2005 force field provided with the Desmond software and the BAR

estimator,^{171,172} TI/MD NVT simulations with ff19sb¹⁷⁶ and the multistate BAR (MBAR) estimator¹⁷⁷ were applied for the AM2TM WT–adamantanamine complexes embedded in DMPC, DPPC, and POPC, to precisely predict experimental relative binding free energies measured with ITC. Dual topology and single topology alchemical transformations were used with TI/MD and FEP/MD, respectively. The pair of ligands considered bears subtle changes in the ligands' structures. With all lipids, it was found that both approaches exhibited a very good correlation between the calculated and experimental relative binding free energies ($r = 0.77$ – 0.87 , $MUE = 0.36$ – $0.92 \text{ kcal mol}^{-1}$), with FEP/BAR performing the worst in DMPC bilayers and TI/MBAR performing the best. The experimental binding free energies were also calculated using antiviral potencies, compared to K_d values, and both FEP/BAR ($r = 0.83$, $MUE = 0.75 \text{ kcal mol}^{-1}$) and TI/MBAR ($r = 0.69$, $MUE = 0.77 \text{ kcal mol}^{-1}$) performed well also in this case.¹⁶⁹

It should be noted that FEP/MD accuracy depends heavily on force fields and sampling, and that these methods are not a substitute for experimental validation, especially given the complexities of protonation states and water networks in AM2.

Synthetic routes to selected adamantane-based and saturated polycyclic amines targeting AM2 channels

General

Adamantane derivatives have been extensively used in medicinal chemistry and can be found in several drugs.^{19,178–181} After the original publication of amantadine's antiviral activity by du Pont de Nemours in 1964,¹⁸² early work

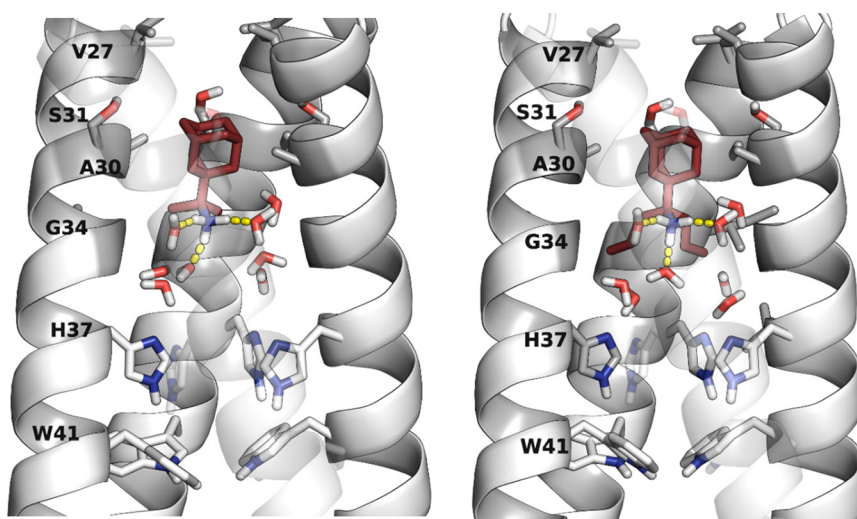


Fig. 8 Representative structures from MD simulations with ff19sb¹⁷⁶ of compounds (*R*)-rimantadine, its dimethyl and dipropyl adduct (see compounds 10, 14 in Table 4) in complex with AM2TM WT in DMPC bilayers. In the figure, the protein is shown with a white cartoon and the ligands with sticks and red color. The water molecules are shown with sticks. White is used for hydrogen, red for oxygen, and blue for nitrogen; yellow dotted lines show the hydrogen bonds.



in the synthesis of amantadine analogs was reported by the same company in 1970 and 1971^{183,184} and Philips-Duphar Research Laboratories in 1971.^{185,186} One of the most significant contributions was the synthesis of over 300 compounds from a lab in Greece, synthesized between 1994 and 2003, see, for example, ref. 113, 137, 142, 145 and 187–199. Most were tested initially for their *in vitro* potency against influenza A. We mentioned here a few works, including selected synthesis of adamantanamines, saturated polycyclic amines, including pinanamine or camphor analogs.^{111,134,146,153,159,160,162,163,200–208} We started with the presentation of the simpler adamantanamines synthesized with a more scalable route, followed by bulkier cage alkyl amines and saturated polycyclic amines to explore modification of the common adamantane ring. We presented in a distinct section the second-generation AM2 S31N inhibitors that can become available through simple synthetic procedures. Elegant syntheses, including radical functionalization methods, have been developed to access various substituted adamantanes, 1,2-didubstituted adamantanes, and diamondoids.^{209,210} Selected syntheses of various adamantanamines and other adamantane derivatives have been reviewed.^{18,19}

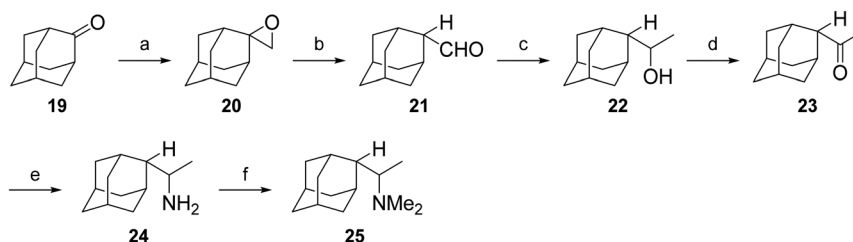
Rimantadine analogs

Extensive structure–activity relationships (SARs) about the *in vitro* potency have been performed for over 40 years, as

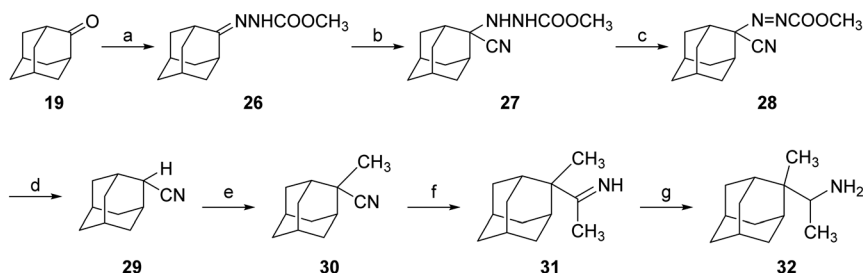
mentioned previously in numerous publications, see for example, ref. 208 and 211. These references include selected examples from the literature regarding the synthesis of amantadine and rimantadine analogs, which have been tested and found potent against influenza A. Adamantane derivatives have been extensively used in medicinal chemistry and can be found in several drugs.^{19,178–181} Elegant syntheses, including radical functionalization methods, have been developed to access various substituted adamantanes, 1,2-didubstituted adamantanes, and diamondoids.^{209,210} Nevertheless, selected syntheses of various adamantane derivatives have been recently reviewed in ref. 19.

For the preparation of 2-rimantadine analogs **24**, **25**, **32**, key intermediates were the 2-adamantyl methyl ketone **19** or the 2-cyanoadamantane **29**, as is described in Scheme 2.²¹²

The cyclic rimantadine analogs **37**, **38**, and **43**, **44** (Scheme 3) were also synthesized with key intermediates the cyclic alcohols **34** and **40**, obtained from the reaction of the acid chloride **33** with the dimagnesium reagent $\text{BrMg}(\text{CH}_2)_4\text{MgBr}$ ^{167,213,214} and with 1-adamantyl-lithium (generated from 1-bromoadamantane **39** with lithium) with cyclohexanone, respectively.^{18,215} Primary *tert*-alkyl amines **38**, **44** were obtained from azides **36**, **41**, respectively, and a work was published that provided routes to produce primary *tert*-alkyl amines from corresponding alcohols, improving steps d for the preparation of **36** from **41**.²¹⁴



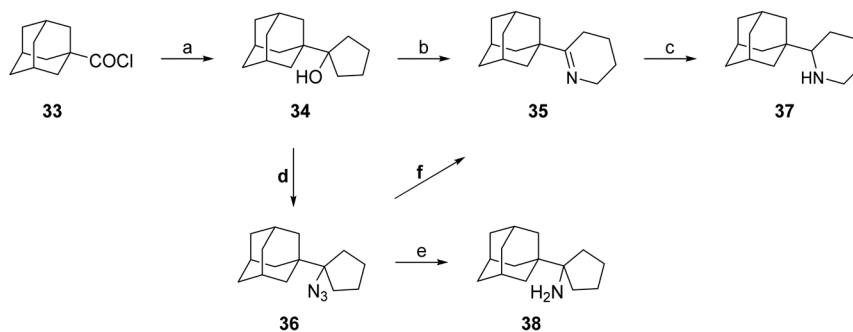
Reagents and conditions: (a) NaH, $(\text{CH}_3)_3\text{SO}^+ \text{I}^-$, DMSO, 25 °C, 1 h, then 60 °C, 2.5 h, (quant.); (b) i. $\text{BF}_3 \cdot \text{Et}_2\text{O}$, benzene, rt, 5 min, ii. H_2O ; (c) i. $\text{CH}_3\text{MgI}/\text{Et}_2\text{O}$, 6 h, rt, ii. aqueous NH_4Cl (93 % from adamantanone); (d) Jones reagent, rt, 24 h, then isopropyl alcohol, rt, 1 h (80 %); (e) NaCNBH_3 , AcONH_4 , MeOH, molecular sieves 3 Å, rt, 30 h (34 %); (f) NaCNBH_3 , CH_2O , acetonitrile, 1 h, then aqueous KOH (44 %).



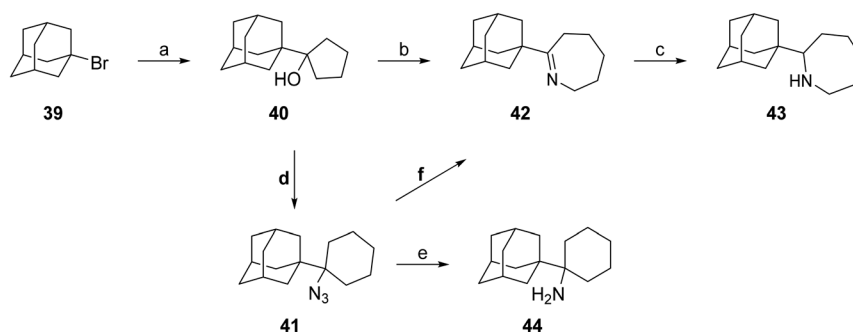
Reagents and conditions: (a) $\text{NH}_2\text{NHCOOCH}_3$, AcOH, MeOH, reflux, 1 h (quant.); (b) KCN, NH_4Cl , $\text{H}_2\text{O}/\text{MeOH}$, 30 °C, 24 h (80%); (c) Br_2 , NaHCO_3 , $\text{CH}_2\text{Cl}_2/\text{H}_2\text{O}$, rt (quant.); (d) MeONa/MeOH 1:1 (96 %); (e) i. LDA, ether, -80 °C, 1.5 h, ii. CH_3I , -65 °C and rt overnight, then water (97%); (f) i. MeLi, ether, -80 °C, then H_2O ii. 18% HCl, acetone, reflux (quant.); (g) LiAlH_4 , THF, reflux, 80 h (30%).

Scheme 2 Synthesis of acyclic 2-rimantadine analogs **24**, **25**, **32**.





Reagents and conditions: (a) $\text{BrMg}(\text{CH}_2)_4\text{MgBr}/\text{THF}$ and then NH_4Cl , H_2O (90 %); (b) $\text{NaN}_3/\text{conc. H}_2\text{SO}_4$, CHCl_3 , 0°C 1.5 h (73 %); (c) NaBH_4 , MeOH , 1h, 0°C , and then rt 20 h (92 %); (d) $^{214}\text{NaN}_3/\text{conc. H}_2\text{SO}_4$, CH_2Cl_2 , 0°C , 1.5 h (90 %); (e) $^{214}\text{LiAlH}_4$, THF , reflux (56 %).



Reagents and conditions: (a) i. Li/THF , sonication, ii. cyclohexanone, THF , 0°C , 5 h, and then $\text{MeOH}/\text{H}_2\text{O}$ 1:1 (72%); (b) $\text{NaN}_3/\text{conc. H}_2\text{SO}_4$, CHCl_3 , 0°C 1.5 h (30%) (c) NaBH_4 , MeOH , 0°C , 1h, and then rt, 2 h (73%); (d) $\text{NaN}_3/\text{conc. H}_2\text{SO}_4$, CH_2Cl_2 , 0°C , 1.5 h (65%); (e) LiAlH_4 , THF , reflux (48%).

Scheme 3 Synthesis of cyclic rimantadine analogs **37**, **38**, **43**, **44**.

Important intermediates for the preparation of **50** were the aminoalcohols **45** and **46** (Scheme 4). Compound **45** was obtained from the reaction of 2-pyridinyl lithium with 2-adamantanone **13**, and its protonated form was subjected to catalytic hydrogenation over PtO_2 catalyst to provide compound **46**.²¹⁶ Compounds **8** and **54** were prepared from adamantane carboxylic acid **51** by treatment with [bis(trifluoroacetoxy)iodo]benzene, followed by catalytic hydrogenation over H_2 , PtO_2 (Scheme 4).

Spiranic adamantanamines

Synthesis of BL-1743 (**3**) was accomplished with the reduction of commercially available 3,3-pentamethylene glutarimide (**55**) with LiAlH_4 in refluxing THF to give 3-azaspiro[5,5]undecane (**51**), which was subjected to a nucleophilic substitution with 2-methylthio-2-imidazoline to furnish BL-1743 (**3**).

The rigidity and crowding around the 2-position of the adamantane nucleus often make chemical transformations difficult and the synthesis of rigid adamantane derivatives challenging. A few synthetic pathways are described below.

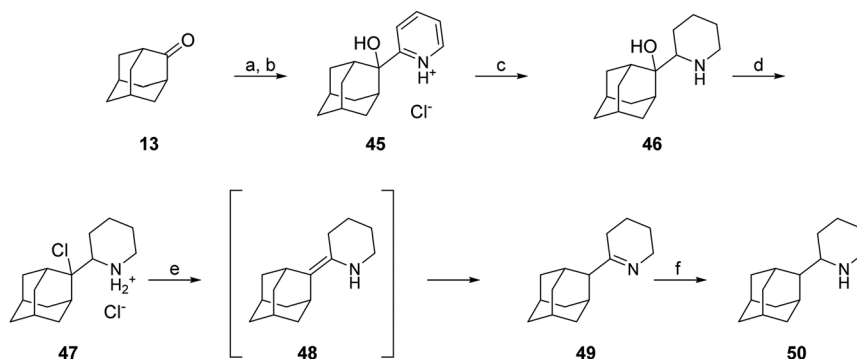
2-Nitroadamantane **57**, which was prepared originally as described in ref. 131, was used for the synthesis of

spiropyrrolidine **11**; the latter was prepared through a Michael condensation of **57** with ethyl acrylate to afford nitroester **58**, which was hydrogenated over Ni to afford spiro[pyrrolidine-2,2'-adamantan]-5-one (**59**) that was reduced with LiAlH_4 to compound **11**.^{217,218} (Scheme 5).

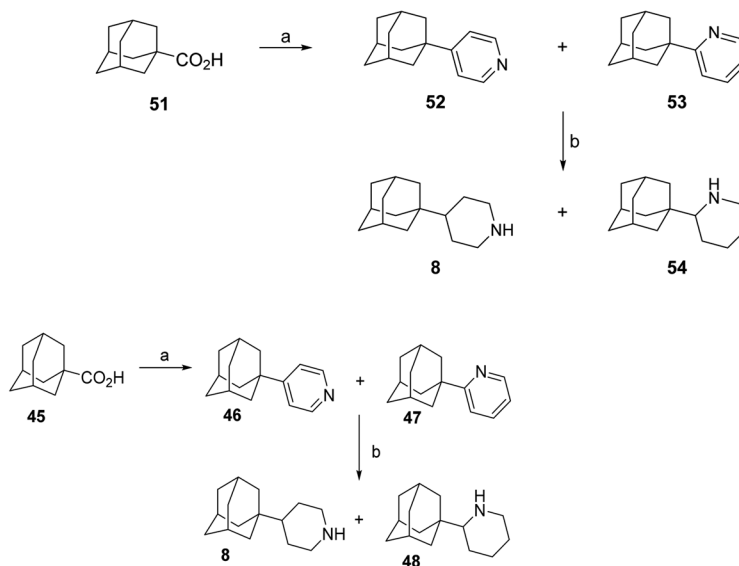
The key intermediate for the synthesis of the spiro[cyclopropane-1,2'-adamantan]-2-amines **63**, **64** and methanamines **66**, **67** (Scheme 5) was ethyl spiro[cyclopropane-1,2'-adamantan]-2-carboxylate **61**, obtained from the [2 + 1]cycloaddition reaction of ethyl diazoacetate with 2-methyleneadamantane **60** in the presence of copper-bronze and purified through saponification and conversion to acid **62**. Routine transformations were performed to prepare **63**, **64**, **66**, **67**.

The synthesis of spiro[cyclobutane-2,2'-adamantan]-2-amines **69** or **72** and spiro[cyclopentane-3,2'-adamantan]-2-amines **79** was also accomplished. For the synthesis of these amines, the spirocyclobutanone **68** or spirocyclopentanone **77** were used, respectively, the latter obtained from the dinitrile **75**, which was transformed to the cyanoenamine **76** through the Thorpe-Ziegler reaction (Scheme 5).²¹² Key reaction for the synthesis of cyclohexanamine **82** was a Diels-Alder reaction between 2-adamantanecarboxaldehyde and methyl vinyl ketone to afford cyclohexenone **80**.





Reagents and conditions. (a) 2-pyridinyl lithium, Et₂O/THF, -60 °C and then HCl 10% (82%); (b) gas HCl, EtOH; (c) H₂/PtO₂, EtOH and then Na₂CO₃ 10% (97%); (d) SOCl₂, CH₂Cl₂, 6h reflux; (e) KOH, EtOH, RT for 24 h under argon and then gentle reflux for 3 h (61% from **46**); (f) NaBH₄, MeOH, 0-5 °C and then RT for 5 h (94%).



Reagents and conditions: (a) pyridine, [bis(trifluoroacetoxy)iodo]benzene, benzene anh., reflux, overnight, 9% yield for pyridine (left side), 27% yield for pyridine (right side); (b) H₂, PtO₂, MeOH, 30 atm, 97% yield for **8**; 99% yield for **54**.

Scheme 4 Synthesis of cyclic rimantadine analogs **8** and analogs **48**, **50**, **54**.

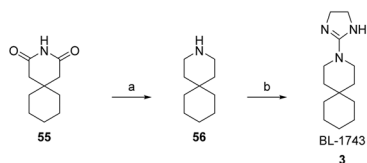
1,2-Annulated heterocycles

So far, we have presented adamantane derivatives with substitutions at the 1- or 2- position of the adamantane ring. Here are examples of 1,2-annulated adamantano-pyrrolidines and adamantano-piperidines that have been synthesized but not yet tested against influenza A strains. These synthetic ways show how to access disubstituted adamantane derivatives, which might be important for drug design, as different substitutions of the adamantane cage can affect the potency of M2 channel blockers. Synthetic approaches towards the 1,2-substitution pattern on the adamantane framework have been presented.^{210,219} Selected works containing a directed C-H functionalization step have been reviewed in ref. 210. Additionally, the synthesis of various

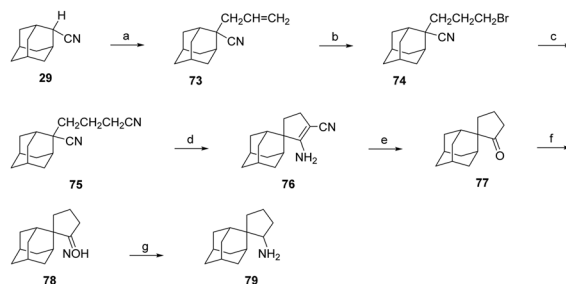
1,2-annulated derivatives with antiviral potency has been reviewed.²²⁰ As illustrated in Scheme 6, the 2-oxo-1-adamantane acetic acid **83** or 2-oxo-1-adamantane carboxylic acid **89** or 2-oxo-1-adamantanepropanoic acid **94** were the key starting materials to afford the 1,2-annulated adamantano-pyrrolidines **88** or **93** or the adamantano-piperidine **97** or **101** (Scheme 6). For the synthesis of **107**, the oxetane derivative **102** first reacted with the triphenyldibromophosphorane, which was prepared *in situ* by the addition of Br₂ to a solution of triphenylphosphine. The resulting dibromide derivative **103** was treated with sodium cyanide in DMSO to obtain the dinitrile **104**, which was then converted to lactam **106** and amine **107** (Scheme 6).

The synthesis of cyclopentanamine **113** in Scheme 6 starts from the tetrahydrofuran **108**, which was converted to

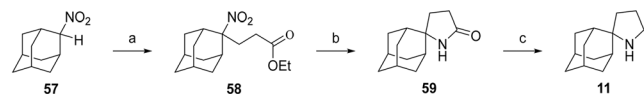




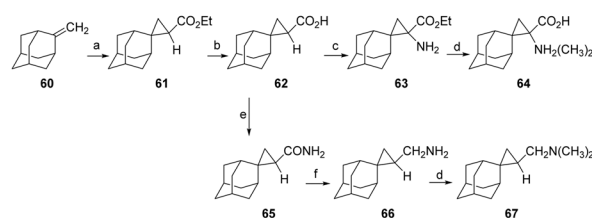
Reagents and conditions: (a) LiAlH₄, THF, reflux, overnight (75%); (b) 2-methylthio-2-imidazoline, MeOH, reflux, overnight (65%).



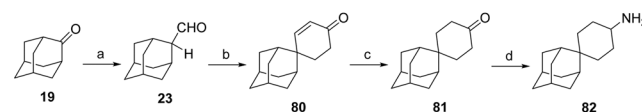
Reagents and Conditions: (a) LDA, -80 °C, THF, CH₂=CHCH₂Br, -65 °C, 1.5 h (88%); (b) (C₆H₅CO)₂O, petr. ether, HBr (g) (96%); (c) KCN, 18-crown-6, CH₃CN, reflux, 90 h (quant.); (d) LDA, -80 °C, THF, and rt overnight, then water (81%); (e) 33% H₂SO₄, AcOH, ethanol, reflux, 20h (quant.); (f) H₂NOH·HCl, AcONa·H₂O, ethanol, reflux, 30h (quant.); (g) Ni-Raney, ethanol, 50 lb/in², 70 °C, 3.5h (quant.); (h) ClCOEt, Et₃N, ether, rt, 24 h (quant.); (i) LiAlH₄, THF, rt, 24h (quant).



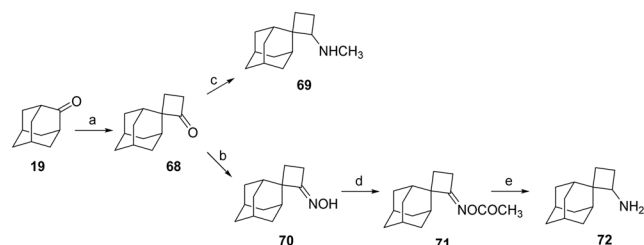
Reagents and conditions: (a) (i) ethyl acrylate, Triton-B, EtOH, 60 °C (70 %); (ii) NaOH, ethanol/water (94%); (iii) EtOH/HCl(g) (94%); (e) H₂, Ni/Raney, ethanol, 40lb/in², RT (quant.); (c) LiAlH₄, THF, 20-25 h reflux 95%.



Reagents and Conditions: (a) (i) N₂CH₂COOEt, Cu, heptane, heating (81%), (ii) NaOH, EtOH/H₂O (ii) H₂O* (81%) (c) (i) Et₃N, ClCO₂Et (ii) NaN₃ (iii) toluene, heat (iv) HCl, 18% (v) OH⁻ (92%); (d) NaBH₄, CH₂=O, MeOH (91%); (e) (i) SOCl₂, (ii) R₁R₂NH, THF (93%); (iii) LiAlH₄, THF (90%).



Reagents and conditions: (a) (i) nBuLi, MeOCH₂(PPh₃)⁺Cl⁻ (ii) HClO₄, H₂O, reflux (b) methyl vinyl ketone, H₂SO₄ (cat.), benzene, reflux; (c) Et₃SiH, PdCl₂, EtOH, reflux (d) (i) H₂NOH·HCl, AcONa (d) LiAlH₄, THF.



Reagents and Conditions: (a) (C₆H₅)₂SC₃H₅⁺BF₄⁻, KOH, DMSO, overnight, rt, then 1 M HBF₄ (quant.); (b) H₂NOH·HCl, AcONa·H₂O, ethanol, reflux, 10 h (quant.); (c) CH₃NH₂·HCl, KOH, NaCNBH₃, rt, 100 h (34%); (d) Ac₂O, py, 0 °C, 12 h, then 0 °C, overnight (92%); (e) NaBH₄/l₂, THF, reflux, 8 h (45%).

Scheme 5 Synthesis of 3-azaspiro[5.5]undecane (5) and its 4,5-dihydro-1H-imidazol-2-yl derivative (BL-1743; 3), spiropyrrolidine 11, spiro[cyclopropane-1,2'-adamantan]-2-amines 63, 64 and -methanamines 66, 67, spiro[cyclobutane-2,2'-adamantan]-2-amine 73, 69 and spiro[cyclopentane-3,2'-adamantan]-2-amine 79, and its and spiro[adamantane-2,1'-cyclohexan]-4'-amine (82).

dinitrile **110**. By employment of the Thorpe–Ziegler reaction, dinitrile **110** underwent an intramolecular condensation catalyzed by LDA to form the enamine **111**. Subsequent acid-promoted hydrolysis of the latter gave rise to the racemic cyclic ketone **112**, which was converted to the corresponding oxime, which was hydrogenated over RANEY® Ni to provide the cyclopentanamine **113**.

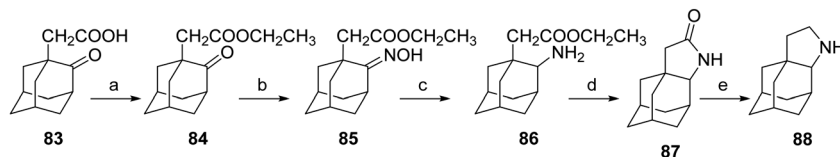
Saturated polycyclic amines

The saturated polycyclic amines, *e.g.*, 7,8,9,10-tetramethyl-3-azapentacyclo[7.2.1.1^{5,8}.0^{1,5}.0^{7,10}]tridecane (**6**)¹⁵⁸ and heptacyclo[8.6.1.0^{2,5}.0^{3,11}.0^{4,9}.0^{6,17}.0^{12,16}]heptadecane (**7**)¹⁶⁶ (Table 2) were potent triple AM2 WT, L26F, V27A blockers and inhibitors *in vitro* of the corresponding viruses. For their synthesis, fascinating reaction sequences were applied (Scheme 7).

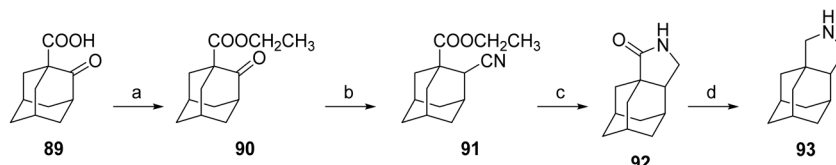
For the synthesis of the aesthetically appealing amines **6** that include one four-membered ring, three five-membered rings, and two six-membered rings, the known diacid **114** was used, easily available from the cycloaddition reaction between acetylene dicarboxylic acid and 2,3-dimethylbutadiene. The reaction of **114** with urea at 180 °C for 30 min yielded the corresponding imide, which was subjected to photolysis at rt in acetone with a 125 W Hg lamp for 2 days to furnish tetracyclic imide **116**, which was reduced with sodium bis(2-methoxyethoxy)aluminum hydride to amine **6**.¹⁵⁸

The starting material for the synthesis of **7** was the imide **118**, prepared from maleimide with cycloheptatriene. The Diels–Alder reaction between **118** and 5,5-dimethoxy-1,2,3,4-tetrachlorocyclopentadiene furnished **120**. Then, a one-pot dechlorination/acetal deprotection reaction with metallic sodium in liquid ammonia/acidic medium, and a one-pot decarbonylation produced **121** from which **7** was furnished.¹⁶⁶

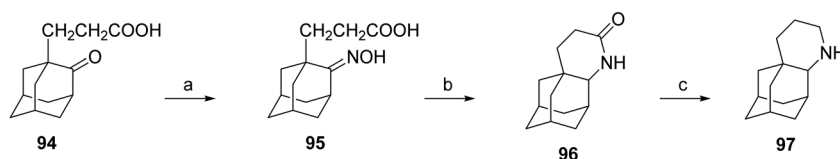




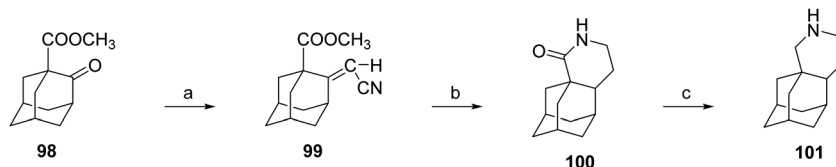
Reagents and conditions: (a) (i) SOCl_2 , 50 °C, 30 min, (ii) abs. EtOH (quant); (b) $\text{NH}_2\text{OH}\cdot\text{HCl}$, $\text{CH}_3\text{COONa}\cdot 3\text{H}_2\text{O}$, EtOH: H_2O (5:1), reflux, 1 h (97%); (c) $\text{H}_2/\text{Ni-Raney}$, EtOH, 55 psi, 100 °C, 3 h, (23%); (d) LiAlH_4 , THF, 5 h, reflux (94-96%); (e) xylene, reflux, 12 h (quant).



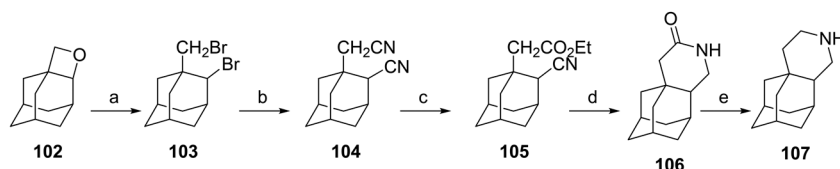
Reagents and conditions: (a) (I) SOCl_2 , 65 °C, 15 min, (II) abs. EtOH, 1 h, rt and 30 min, 70 °C (quant); (b) TosMIC, abs. EtOH, DME, $t\text{-BuOK}$, 0 °C, argon, 20 °C, 30 min, and 48 °C, 1 h (74%); (c) $\text{H}_2/\text{Ni-Raney}$, MeOH, 65 psi, 60 °C, 6 h; (d) xylene, reflux, 10 h (40%).



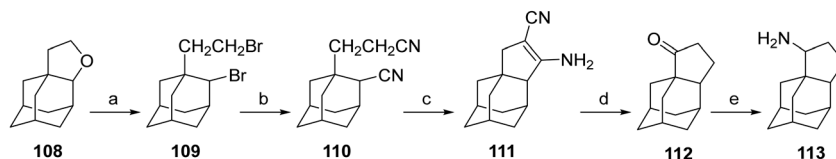
Reagents and conditions: (a) $(\text{EtO})_2\text{POCH}_2\text{CN}$, benzene, NaH , 1 h, 20 °C and then 15 min, 65 °C (95%); (b) $\text{H}_2/\text{Ni-Raney}$, EtOH, 50 psi, 150 °C, 10 h (40%); (c) LiAlH_4 , THF, 10 h, reflux (91%).



Reagents and conditions: (a) $\text{NH}_2\text{OH}\cdot\text{HCl}$, $\text{CH}_3\text{COONa}\cdot 3\text{H}_2\text{O}$, EtOH/ H_2O (9:1), reflux, 5 h (94%); (b) $\text{H}_2/\text{Ni-Raney}$, EtOH, 50 psi, 200 °C, 4 h (94%); (c) LiAlH_4 , THF, 20 h, reflux (98%).



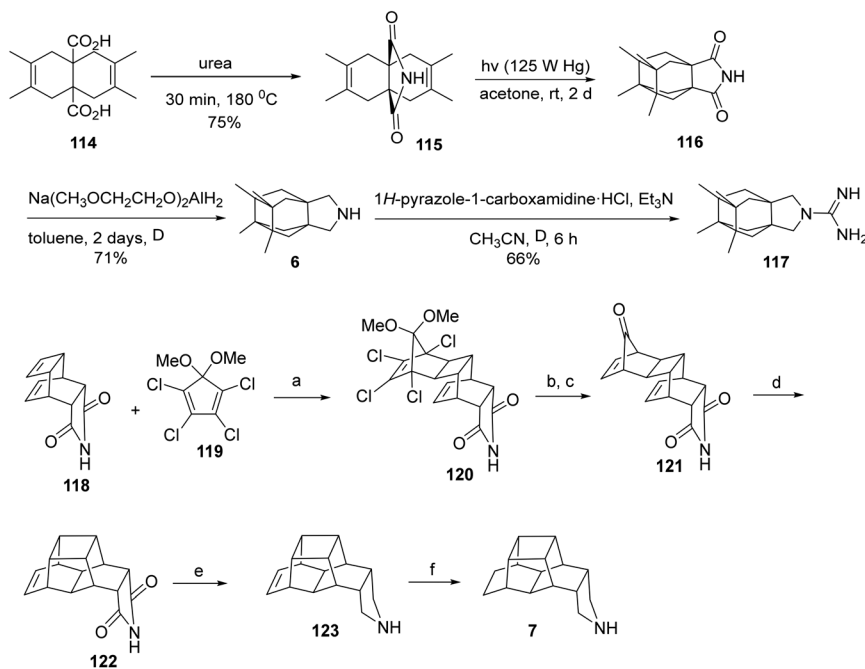
Reagents and conditions: (a) Br_2 , $\text{C}_6\text{H}_5\text{CN}$, Ph_3P , 122 °C, 4 h (38%); (b) NaCN , DMSO, 4 h, 170 °C (88%); (c) EtOH-HCl, 10 days, 30 °C (82%); (i) H_2O , HCl, ether, 24 h, 30 °C (86%); (d) $\text{H}_2/\text{Ni-Raney}$, EtOH, 55 psi, 140 °C, 7 h (21%); (e) LiAlH_4 , THF, 20h, reflux (98%).



Reagents and conditions: (a) Br_2 , $\text{C}_6\text{H}_5\text{CN}$, Ph_3P , 124 °C, 4 h (84%); (b) NaCN , DMSO, 115 °C, 1 h and 145 °C, 1 h (75%) and then NaCN , DMSO, 155 °C, 1 h (89%); (c) LDA, THF, -80 °C (quant); (d) H_2SO_4 (33%), glacial CH_3COOH , reflux, 20 h (quant); (e) (i) $\text{NH}_2\text{OH}\cdot\text{HCl}$, $\text{CH}_3\text{COONa}\cdot 3\text{H}_2\text{O}$, EtOH/ H_2O (14:1), 6 h, reflux (94%); (ii) EtOH, Ni-Raney, 50 psi, 70 °C, 4 h (86%).

Scheme 6 Synthetic procedures for the preparation of the 1,2-annulated adamantanepyrrolidines and piperidines **88** or **93** and **97**, **101**, **107**, and **113**.





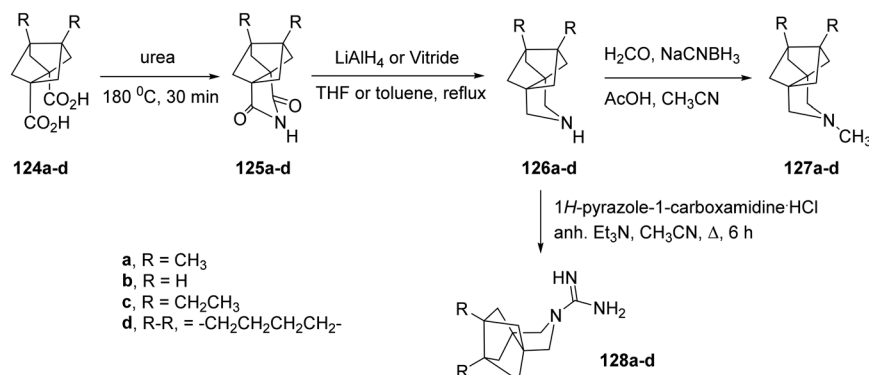
Reagents and conditions: (a) sealed tube, neat, 140 °C, overnight (89%); (b) sodium, liquid NH₃, dry THF, dry MeOH, 60 °C, 30min; (c) HCl 10%, water, rt, 1h (41% overall); (d) hydroquinone, sealed tube, xylene, 180 °C, 90h (44% overall); (e) Vitride, dry toluene, reflux, overnight (74%); (f) H₂, 1 atm, Pd/C, absolute EtOH, rt, overnight (79%).

Scheme 7 Synthetic procedure for the preparation of the polycyclic amines **6**, **7**.

Further synthetic studies aimed to develop polycyclic pyrrolidine scaffolds with potential inhibitory activity against the AM2 channel. The known diacids **124a,b** were heated with urea at 180 °C for 30 min to give the corresponding imides **125a,b**, which were subsequently reduced to the secondary amines **126a,b** in good overall yields.¹⁶⁶ Subsequent reductive alkylation of **126a,b** with formaldehyde and NaCNBH₃ produced the tertiary amines **127a,b**. Because guanidines derived from similar amine precursors had shown AM2 inhibitory properties, compounds **128a,b** were synthesized from **126a,b** using 1*H*-pyrazole-1-carboxamide hydrochloride (Scheme 8). Molecular and structural investigations have demonstrated that substituting V27 with

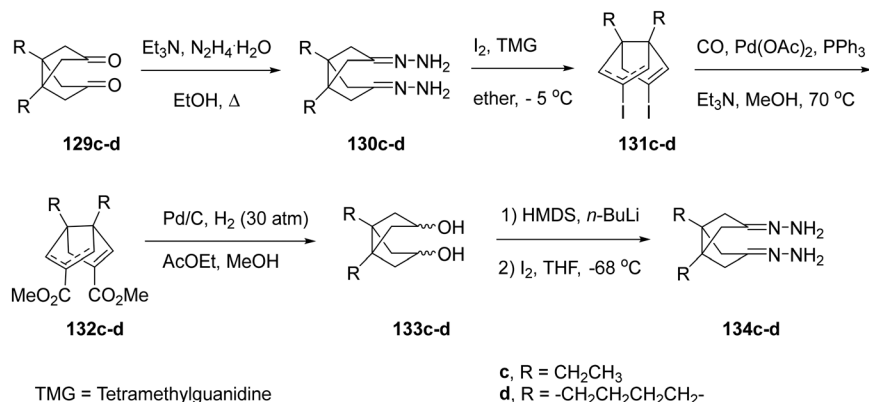
a smaller residue, such as alanine (AM2 V27A mutant), disrupts the hydrophobic constriction created by this position and expands the pore near the N-terminal end by roughly 2 Å.¹⁶⁶ Compounds **125a** and **127a** exhibited stronger inhibition of the M2-V27A mutant channel compared with their smaller analogues **125b** and **127b**. Molecular dynamics simulations predicted that bulkier, more hydrophobic ligands could occupy this enlarged pore more efficiently. Thus, two extended analogues of **127a**, namely **127c** and **127d**, were designed and synthesized following the same route.¹⁶⁶

To obtain the extended analogues **127c,d**, novel diacids **124c,d** were required. These were synthesized from readily available diketones **129c,d** through the multi-step sequence



Scheme 8 Synthetic procedure for the preparation of the polycyclic pyrrolidine and guanidine derivatives **124a-d-128a-d**.¹⁶⁶





Scheme 9 Synthetic procedure for the preparation of diacids **124c,d**.²⁰⁸

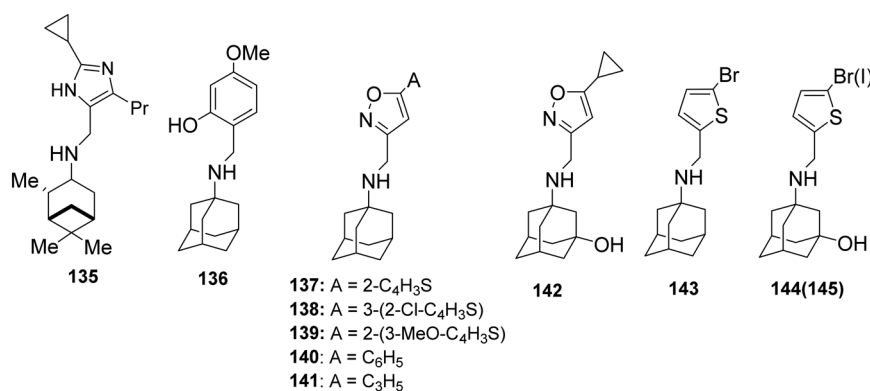
shown in Scheme 9.¹⁶⁶ The approach, previously optimized for bisnoradamantane scaffolds, involved conversion of **129c,d** to the corresponding bis-hydrazone **130c,d** under reflux in ethanol with triethylamine and hydrazine hydrate, followed by iodination with I₂/TMG (tetramethylguanidine) to yield the bis-vinyl iodides **131c,d**. Subsequent Pd(0)-catalyzed carbonylation (Pd(OAc)₂/PPh₃, CO, Et₃N, MeOH, 70 °C) furnished the diesters **132c,d** as mixtures of *syn/anti* isomers. Catalytic hydrogenation (Pd/C, H₂ 30 atm) produced diols **133c,d**, which, upon double deprotonation with HMDS/*n*-BuLi and treatment with I₂ at -68 °C in THF, afforded **134c,d**. Finally, hydrolysis of **134c,d** provided the target diacids **124c,d** in 63% and 80% yield, respectively.¹⁶⁶

Design and synthesis of adamantane-based drugs that lead to selective AM2 S31N inhibitors

After the observation that pinanamine (**4**)¹⁶⁵ blocks M2 WT-mediated proton current (Table 2), it was found^{128,130} that linking pinanamine with a heterocycle, *e.g.*, imidazole, through a CH= or CH₂ linker provided pinanamine-aryl conjugates, *e.g.*, compound **135**, which are very potent against influenza A M2 WT and M2 S31N viruses *in vitro* by

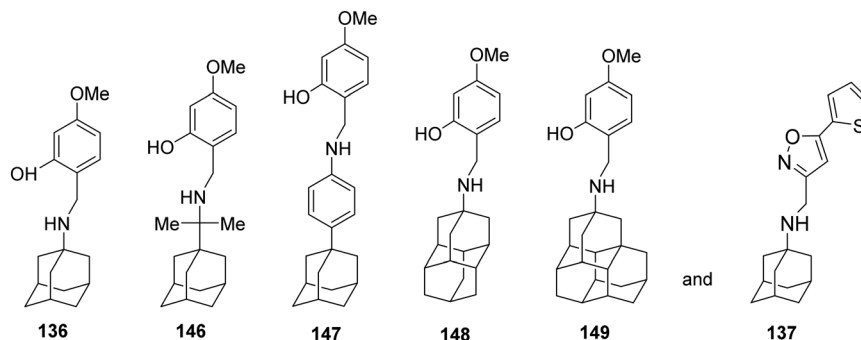
blocking the corresponding M2 channels (Scheme 10). The %-blocking efficiency at 2 min of 5-methyl-imidazole was less than 30%, suggesting a slow entry blocking compound.^{101,105,139}

It was then found that linking amantadine (**1**) with a heterocycle can produce potent compounds acting against AM2 S31N in both EP and antiviral assays. Using MD simulations and solution NMR in DPC micelles, it was shown that compound **137** (M2J332) in Scheme 10, with a 2-(2-thiophenyl)-isoxazol-3-yl group, blocks the AM2 S31N channel.⁷⁶ The adamantyl group of the drug is bound in the pore between AM2 N31 and G34. In contrast, the aryl group projects towards the N-terminus through the V27 side chains (Fig. 2). This orientation on the AM2 S31N channel is opposite to the orientation of amantadine on the M2 WT channel, *i.e.*, with the amino group facing H37 towards the C-end (Fig. 1).^{72,80,221} Using ssNMR in DMPC lipid bilayers, this position and orientation were also suggested in AM2TM S31N for compound **140** (M2J352).⁷⁹ It was also found that compounds **144**, **145**¹⁴⁰ bearing one of the simpler aryl groups used, *i.e.*, correspondingly the 3-bromothiophenyl, 3-iodothiophenyl, block both AM2 WT and AM2 S31N channels. Using NMR in micelles and MD simulations, it was found that the dual



Scheme 10 Chemical structure of pinanamine-aryl conjugate **135** (ref. 228) and amantadine-aryl conjugates **136–145**,^{76,81,140,229} which block both M2 WT and M2 S31N channels according to EP and inhibit *in vitro* the M2 WT and M2 S31N viruses (in **137** A is 2-thiophenyl, in **138** A is 2-chloro-3-thiophenyl, in **139** A is 3-methoxy-2-thiophenyl, in **140** A is phenyl and in **141** R is cyclopropyl).

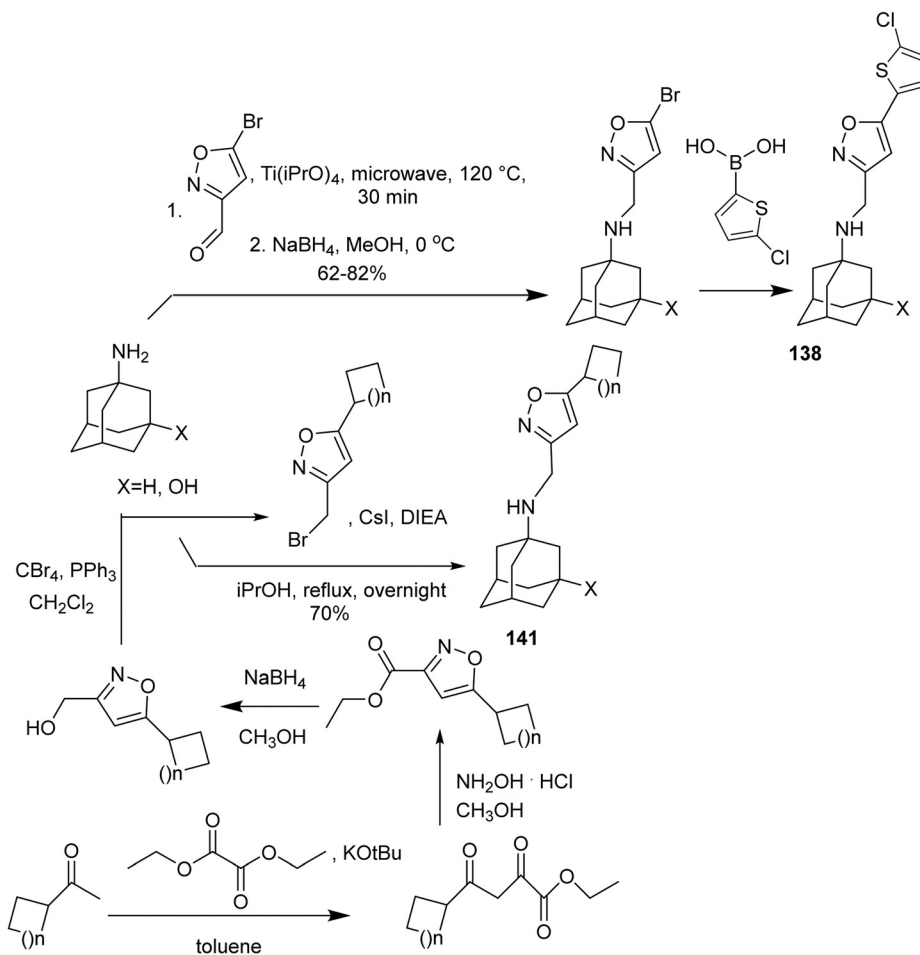




Scheme 11 Structures of compounds **136**, **146–149**, and **137** were used as chemical probes to investigate binding to M2 WT and M2 S31N channels.

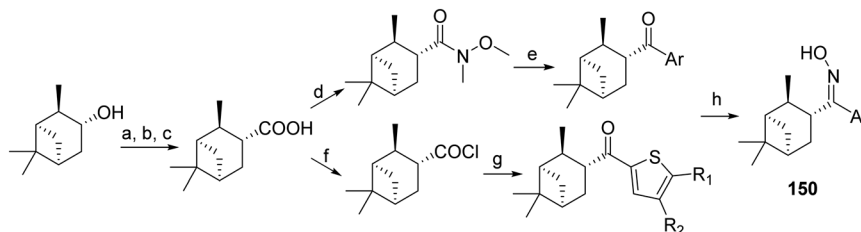
inhibitors **144** and **145**¹⁴⁰ are oriented with the aryl head group toward the N-end in the AM2AH S31N pore and toward C-end in the AM2AH WT pore.⁸¹ For these amantadine–aryl conjugates, between 2013 and 2018, extensive SAR investigations on *in vitro* activity and EP-based blocking potency were conducted,^{78,133,135,136,222–226} and/or binding kinetics^{78,136} were performed by modifying the adamantyl group and the aryl head group.^{76,81,139–141} The kinetics of a

ligand binding to its protein target are seen as increasingly important for *in vivo* efficacy in drug discovery²²⁷ and were critical for amantadine–aryl conjugates.^{139,141} Targeted optimization of binding kinetics was difficult to achieve and required systematic studies, as did those for AM2 channels, to increase understanding of the molecular interactions involved.¹³⁹ It was shown^{78,136} that, similarly to amantadine analogs against AM2 WT^{73,147} or against V27A,¹⁰¹ the

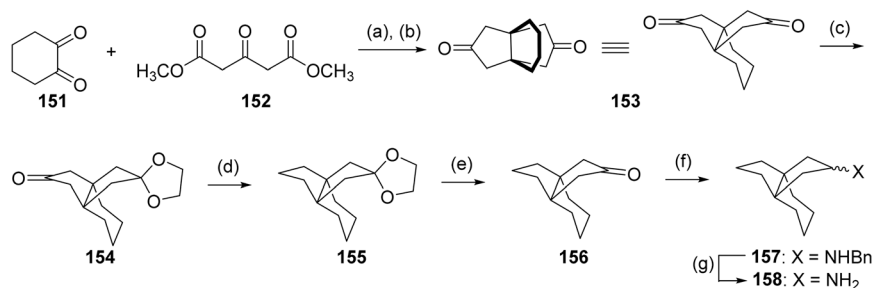


Scheme 12 Synthetic procedures leading to amantadine–aryl conjugates, e.g., **137** (M2J332),⁷⁶ **140** (M2J352),⁷⁶ **138**,⁸¹ **141**,²³⁵ as potent blockers of the M2 S31N channel and virus replication.





Scheme 13 Synthetic procedures lead to pinanyl-thiophenyl oximes, which are potent blockers of AM2 S31N channel and virus replication.¹³² Reagents and conditions: (a) PDC, CH₂Cl₂, rt, overnight; (b) TosMIC, *t*-BuOK, DMSO, 60 °C, 48 h; (c) H₂SO₄, CH₃COOH, reflux, overnight; (a) → (c) 36.4%; (d) *N,O*-dimethylhydroxylamine hydrochloride, TBTU, triethylamine, CH₃CN, rt, 1 h; (e) ArLi, ether, 0 °C, 30 min; -78 °C, 3 h; (f) SOCl₂, cat. DMF, reflux, 2 h; (g) substituted thiophene, SnCl₄, dry CH₂Cl₂, 0 °C, 30 min; (h) pyridine, NH₂OH·HCl, 80 °C, 24 h.



Reagents and conditions: (a) Citrate–phosphate buffer pH 5.6, MeOH (15% v/v), rt, 72 h; (b) HCl 6 M, 95 °C, 18 h (61% over two steps); (c) ethylene glycol (0.9 equiv), *p*-TsOH, toluene, Dean–Stark, reflux, 10 h (49 %); (d) N₂H₄·KOH, diethylene glycol, 136 °C 2 h then 200 °C 6 h (60%); (e) acetone, *p*-TsOH, 60 °C, 2 h (95%); (f) BnNH₂, NaBH(OAc)₃, HOAc, 1,2-dichloroethane, rt, 72 h (99%, *syn*-**157**: *anti*-**157** = 1:1); (g) HCO₂NH₄, Pd(OH)₂/C, MeOH/EtOAc, reflux, 3 h (75%, *syn*-**158**: *anti*-**158** = 1:1).

Scheme 14 Synthetic route toward diastereomeric [4.3.3]propellan-8-amines.¹⁶⁶

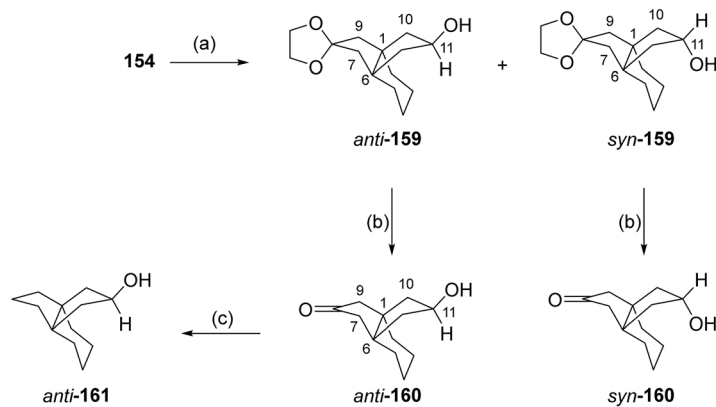
amantadine–aryl conjugates should have fast k_{on} , slow k_{off} (e.g., compound **141**) or slow k_{on} , slow k_{off} (e.g., compound **142**) that have low K_d ($= k_{off}/k_{on}$) values to achieve well *in vitro* antiviral potency.

Then, in 2020,¹⁰⁶ the effect of subtle modifications on the structure of an amantadine–aryl conjugate inhibitor on binding kinetics for AM2 WT and AM2 S31N channels was studied using synthetic amantadine variant-CH₂-aryl derivatives **136**, **143**, **146–149** (Scheme 11) as sensitive chemical probes for blocking AM2 S31N and AM2 WT channels as well as virus replication in cell culture. Notably, only the binding kinetics for the AM2 S31N channel were very dependent on the length between the adamantane moiety and the first ring of the aryl head group, as observed in **146** and **147**, and the girth and length of the adamantane adduct, as observed in **148** and **149**. The study of **136**, **137**, **146–149** with MD simulations in AM2TM S31N and binding free energy calculations (MM-PBSA) showed that all compounds bind in to the AM2 S31N channel with the adamantyl group positioned between V27 and G34 the aryl group projecting out of the channel (toward the N-end) with phenyl (or isoxazole in **137**) embedded in the V27 cluster. In this outward binding configuration, an elongation of the ligand by only one methylene in rimantadine **146** or using

diamantane as well as triamantane instead of adamantane in **148** and **149**, respectively, caused an incomplete entry and facilitated exit, abolishing effective block compared to the amantadine derivatives **136** and **137**. In the active M2 S31N blockers **136** and **137**, the phenyl and isoxazolyl head groups achieve a deeper binding position, corresponding to measured high k_{on} /low k_{off} and high k_{on} /high k_{off} rate constants, compared to inactive **146–149**, which have much lower k_{on} and higher k_{off} . The MD simulations also showed that compounds **136**, **146–149** each can block the M2TM WT channel by binding in the longer area from V27–H37, in the inward orientation, with high k_{on} and low k_{off} rate constants leading to insufficient block. Infection of cell cultures by influenza virus containing AM2 WT or AM2 S31N is inhibited by **136**, **146–149**, and **137**, respectively. While **136** and **137** block infection through the AM2 block mechanism in the AM2 S31N variant, **146–149** may block AM2 S31N virus replication in cell culture through the lysosomotropic effect.¹⁰⁶

For the most potent conjugates, it was shown that 3-hydroxylation of the 1-adamantyl group, e.g., in compound **142** compared to **141**, improved the drug metabolism and pharmacokinetics (DMPK) profile, e.g., the microsomal stability in rat and mouse microsomes against cytochromes,





Reagents and conditions: (a) L-Selectride, THF, -78 °C, 30 min, rt, 2 h (*anti*-159: *syn*-159 = 85:15; *anti*-159 69 %, *syn*-159 10 %); (b) acetone, *p*-TsOH, reflux, 2 h (*anti*-160 98 %, *syn*-160 97 %); (c) N₂H₄·KOH, diethylene glycol, 136 °C 2 h then 200 °C 6 h (60 %).

Scheme 15 Diastereoselective synthesis of alcohol *anti*-161 from monoketal 154.²⁰⁸

the membrane permeability in Caco-2 cells, and increased recovery. Two potent and promising lead candidates for further development as antiviral drugs are **144** and **145**¹⁴¹ which have high mouse and human liver microsomal stability ($T_{1/2} > 145$ min) and membrane permeability (>200 nm s⁻¹) also inhibit both currently circulating oseltamivir-sensitive and -resistant human influenza A viruses (H1N1 and H3N2) with EC₅₀ values ranging from 0.4 to 2.8 μM and a selectivity index of >100 .¹⁴¹

In a recent study,²³⁰ based on the structure of the amantadine-aryl conjugates **136**, **137** that inhibit AM2 S31N, amantadine was replaced by 16 other adamantanamines. Thus, 36 new compounds were synthesized and tested against AM2 WT and the five amantadine-resistant viruses bearing the AM2 with the mutation L26F, or V27A, or A30T, S31N, or G34E, aiming at identifying inhibitors against multiple M2 mutant-amantadine-resistant viruses. From this study, 16 compounds were identified that inhibited *in vitro* influenza A viruses with AM2 WT or L26F channels. However, compounds **146–149**, which are conjugates of a rimantadine analog or the diamantylamine or the 4-(1-adamantyl) benzenamine with the 2-hydroxy-4-methoxyphenyl group, were *in vitro* inhibitors against the three influenza A viruses M2 WT or L26F or S31N, while compound **147** also inhibited *in vitro* the M2 G34E virus, and compound **148** also inhibited *in vitro* the AM2 A30T virus. Using EP, it was shown that compound **147** was an efficient blocker of the AM2 WT and AM2 L26F channels, compound **148** blocked the AM2 WT channel, and compound **149** blocked the AM2 WT, L26F, and V27A channels. For these compounds, a preliminary drug metabolism and pharmacokinetics study was conducted.²³⁰

In another study²³¹ amantadine-aryl conjugates were tested against four amantadine resistant M2 mutants among avian and human influenza A H5N1 strains circulating between 2002 and 2019: the single AM2 S31N and V27A mutants, and the S31N/L26I and S31N/V27A double mutants S31N/L26I and S31N/V27A double mutants. Utilizing TEVC

assays, structurally diverse M2 inhibitors were screened against these single and double mutant channels. Compound **139**²³¹ (Scheme 10) was found to significantly block all three M2 mutant channels and *in vitro* replication: AM2 S31N, AM2 S31N/L26I, and AM2 S31N/V27A.

EP studies combined with MD simulations were applied to rationalize the resistance^{149,232} of AM2 S31N viruses to amantadine-aryl conjugates (*e.g.*, of compound **142**) and intriguingly^{149,232} to mutation L46P outside the M2 S31N channel. A few examples of procedures applied for the synthesis of amantadine-aryl conjugates^{76,81,139,141,233,234} are shown in Scheme 12.

Interestingly, AM2 S31N blockers, such as pinanyl-thiophenyl oximes **150**, have been synthesized as potent blockers of AM2 S31N, having a novel structure¹³² using (-)-isopinocampheol as raw material (Scheme 13).

Propellanamine analogs of adamantanamines

The tricyclic [4.3.3]propellan-8-amines were synthesized²⁰⁸ as conformationally constrained analogues of amantadine. These rigid polycyclic systems were conceived to explore how subtle variations in cage topology affect channel blocking and NMDA receptor binding. The synthesis began with a modified Weiss–Cook reaction between cyclohexane-1,2-dione (**151**) and dimethyl 3-oxoglutarate (**152**) to afford the diketone **153**, which was subsequently converted into the key propellanedione intermediate (Scheme 14).²⁰⁸ Diastereoselective reduction with L-selectride followed by Wolff–Kishner reduction and acetal manipulation furnished the diastereomerically pure alcohol *anti*-161 (Scheme 15).²⁰⁸ Further Mitsunobu inversion and S_N2 azidation yielded the diastereomeric primary amines *syn*-158 and *anti*-158, which were evaluated for both NMDA receptor affinity and anti-influenza A activity. Interestingly, despite structural resemblance to amantadine and comparable binding to the 1-(1-phenylcyclohexyl)piperidine (PCP) site ($K_i \approx 11$ μM),



Table 5 Summary table of representative adamantane-based AM2 inhibitors categorized by structural and mechanistic features (color in the 4th column describes a different class of an inhibitor)








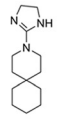




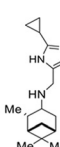
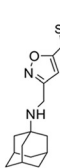
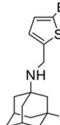

Scaffold	M2 target	Key assay	Potency range EC ₅₀ or IC ₅₀	Structural category	Structure-mechanism relationship confirmed by MD simulations or functional assays
 Amt 1	WT	Whole cell or TEVC assays	0.48–12.5 μM	Amt/Rmt analogs	The ammonium group interacts with pore waters close to H37. The adamantane ring displaces water near V27: proton-blocking mechanism-based inhibitors
 Rmt 2	WT	Whole cell or TEVC assays	0.02–10.8 μM	Amt/Rmt analogs	Mechanism of action similar to Amt: proton-blocking mechanism-based inhibitors
 5	WT	Whole cell or TEVC assays	13.5–38.2 μM	Amt/Rmt analogs	Mechanism of action similar to Amt
 10	WT/L26F	Whole cell or TEVC assays	0.01–9.3/0.55–0.79 μM	Amt/Rmt analogs	Mechanism of action similar to Rmt
 9	WT/V27A/L26F	Whole cell or TEVC assays	18.7/0.3/5.6 μM	Spiranic amines	Longer scaffold that fits the binding area in the V27A mutation: proton-blocking mechanism-based inhibitors
 8	WT/V27A	Whole cell or TEVC assays	0.14–4.1/3.6 μM	Amt/Rmt analogs	Longer scaffold that although having promising V27A channel blockade activity, failed to behave as a good antiviral in whole cell assays, likely due to a fast dissociation constant: mechanism of action similar to 9
 11	WT/L26F	Whole cell or TEVC assays	0.34–10.2/8.75 μM	Spiranic amines	A scaffold that fits the widening of the channel due to the L26F mutation: proton-blocking mechanism-based inhibitors; some polycyclic amines can inhibit S31N strains through an unknown mechanism



Table 5 (continued)

 BL-1743	WT	Whole cell or TEVC assays	2 μM	Adamantanamine analogs	Mechanism of action similar to 9
3 	WT	Whole cell or TEVC assays	1.36–6.8 μM	Adamantanamine analogs	Mechanism of action similar to Amt
4 	WT/V27A/L26F	Whole cell or TEVC assays	18/0.7/8.6 μM	Saturated polycyclic amines	Mechanism of action similar to 9
6 	WT/V27A	Whole cell or TEVC assays	6.7–0.54/17.2 μM	Saturated polycyclic amines	Mechanism of action similar to 9
7 	WT/L26F	Whole cell or TEVC assays	0.07/0.74 μM	Saturated polycyclic amines	Mechanism of action similar to 9
12 	WT/S31N	CPE assay	3.2–15.3/3.4–29.4 μM	Pinanamine derivatives	Precursor of AM2 S31N blockers
135 	S31N	TEVC assay	13–16 μM	Amantadine–aryl conjugates/selective S31N inhibitors	The adamantyl group of the drug is bound in the pore between N31 and G34, the aryl group projects towards the N-terminus embedded in the V27 cluster (the orientation is opposite to that of amantadine on the WT channel)
137 	WT, S31N	TEVC assay	0.4 to 2.8 μM	Amantadine–aryl conjugates/selective S31N inhibitors	3-Hydroxylation of the 1-adamantyl group, improved the drug metabolism and pharmacokinetics: blocking the WT or S31N channel through thiophene orientation towards C-terminus or N-terminus, respectively
144/145 					



neither diastereomer inhibited influenza A replication, demonstrating that modifying the cage geometry separates antiviral from NMDA antagonistic activity.

Conclusions

Combining structural studies with advancements in computational methods, can help design and synthesize adamantane-based blockers that target wild-type or/and mutant AM2 channels. The key structural design principles for saturated polycyclic amine-based blockers against AM2 WT, V27A, and L26F are grounded in advancements the fundamental mechanism of AM2 proton blocking. These principles involve incorporating features that enable a part of the molecule specifically, the carbon framework connected to an amine group-to function as a proton mimic. This allows these inhibitors to occupy the primary proton-binding site within the channel. Second-generation adamantanamines are known to occlude the AM2 S31 pore and block proton transport. Interestingly many saturated adamantanamines inhibit S31N strains, although the mechanism behind this inhibition has not yet been identified.¹⁹⁹ Examples of synthetic chemistry that have led to the development of complex adamantanamines, saturated polycyclic amines, and second-generation adamantane-based inhibitors are presented. An extensive and long-term study of the druggable M2 channel has provided the scientific community with essential tools for structure-based drug design, a synthetic chemistry toolkit, and a library of adamantane-based compounds that show promise as effective antivirals, especially given the frequent mutations of the AM2 protein in viruses (Table 5).

Conflicts of interest

The authors declare no conflict of interest.

Data availability

Data are available upon request.

Acknowledgements

AK thanks Chiesi Hellas for supporting this research (SERG No 10354).

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